



Published in final edited form as:

Free Radic Biol Med. 2008 May 1; 44(9): 1700–1711.

## Oxidative modification of cytochrome c by singlet oxygen

Junhwan Kim<sup>a</sup>, Myriam E. Rodriguez<sup>b</sup>, Ming Guo<sup>c</sup>, Malcolm E. Kenney<sup>c,d</sup>, Nancy L. Oleinick<sup>a,b,d</sup>, and Vernon E. Anderson<sup>a,c,d,\*</sup>

<sup>a</sup> Department of Biochemistry, School of Medicine, Case Western Reserve University, Cleveland, OH, USA

<sup>b</sup> Department of Radiation Oncology, School of Medicine, Case Western Reserve University/University Hospitals of Cleveland, Cleveland, OH 44106, USA

<sup>c</sup> Department of Chemistry, Case Western Reserve University, Cleveland, OH, USA

<sup>d</sup> Case Comprehensive Cancer Center, Case Western Reserve University, Cleveland, OH, USA

### Abstract

Singlet oxygen ( $^1\text{O}_2$ ) is a reactive oxygen species that may be generated in biological systems. Photodynamic therapy generates  $^1\text{O}_2$  by photoexcitation of sensitizers resulting in intracellular oxidative stress and induction of apoptosis.  $^1\text{O}_2$  oxidizes amino acid side chains of proteins and inactivates enzymes when generated *in vitro*. Among proteogenic amino acids, His, Tyr, Met, Cys, and Trp are known to be oxidized by  $^1\text{O}_2$  at physiological pH. However, there is a lack of direct evidence of oxidation of proteins by  $^1\text{O}_2$ . Because  $^1\text{O}_2$  is difficult to detect in cells, identifying oxidized cellular products uniquely derived from  $^1\text{O}_2$  could serve as a marker of its presence. In the present study,  $^1\text{O}_2$  reactions with model peptides analyzed by tandem mass spectrometry provide insight into the mass of prominent adducts formed with the reactive amino acids. Analysis by MALDI-TOF and tandem mass spectrometry of peptides of cytochrome *c* exposed to  $^1\text{O}_2$  generated by photoexcitation of the phthalocyanine Pc 4 showed unique oxidation products, which might be used as markers of the presence of  $^1\text{O}_2$  in the mitochondrial intermembrane space. Differences in the elemental composition of the oxidized amino acid residues observed with cytochrome *c* and the model peptides suggest the protein environment can affect the oxidation pathway.

### Keywords

singlet oxygen; amino acids; proteins; cytochrome-*c*; tandem mass spectrometry; protein modification; histidine; tryptophan; phenylalanine; methionine; oxidation; phthalocyanine; photodynamic therapy

## Introduction

### Intracellular generation of singlet oxygen

Singlet oxygen ( $^1\text{O}_2$ ) is a unique reactive oxygen species (ROS) in that its chemical reactivity derives from its characteristic electronically excited state. The lowest energy singlet state

\* Corresponding author: 10900 Euclid Avenue, Cleveland, OH 44106-4935, Ph: (216) 368-2599, fax (216) 368 3419, E-mail vea@case.edu.

<sup>†</sup>This work was supported by NIH/NIA grant (P01 AI55739) to V.E.A and by NIH/NCI grants (R01 CA083917, R01 CA106491 and P01 CA48735) to N.L.O.

**Publisher's Disclaimer:** This is a PDF file of an unedited manuscript that has been accepted for publication. As a service to our customers we are providing this early version of the manuscript. The manuscript will undergo copyediting, typesetting, and review of the resulting proof before it is published in its final citable form. Please note that during the production process errors may be discovered which could affect the content, and all legal disclaimers that apply to the journal pertain.

( $^1\Delta_g$ ) of  $O_2$  is  $94 \text{ kJ}\cdot\text{mol}^{-1}$  above the ground state triplet and is the common form of  $^1O_2$  [1]. The lifetime of  $^1O_2$  depends on its environment, but in most solvents it falls in the range of 1-100  $\mu\text{s}$ . In neutral-pH aqueous solution the lifetime is 2-4  $\mu\text{s}$  in  $H_2O$  and  $53 \pm 5 \mu\text{s}$  in  $D_2O$  [2]. In biological systems,  $^1O_2$  may be efficiently generated by either endogenous or exogenous sensitizers that transfer their excited state energy to ground state molecular oxygen [3] or as one of the products of peroxidase enzymes [4]. An alternative source of  $^1O_2$  of potential pathological significance is the decomposition of lipid peroxidation products which are formed during ischemia-reperfusion injury [5,6].

$^1O_2$  may be intentionally generated in biological tissues by photodynamic therapy (PDT) which employs a photosensitizing drug and visible light to produce an oxidative stress in cells and ablate cancerous tumors [7,8]. PDT is also used for treating certain non-cancerous conditions that are generally characterized by the overgrowth of unwanted or abnormal cells [8]. Photosensitizers employed for PDT are most commonly porphyrins or certain porphyrin-related macrocycles, such as phthalocyanines or pheophorbides. Most photosensitizers for PDT are efficient producers of  $^1O_2$  through Type II photochemistry, which is considered to be the dominant mechanism for PDT in cells and tissues [7,9].

Pc 4, a silicon phthalocyanine (Pc) photosensitizer [10], effectively generates  $^1O_2$  when exposed to red light in the presence of  $O_2$ . Of particular importance for this study, Pc 4 absorbs red light and produces  $^1O_2$  by electronic energy transfer without forming superoxide by electron transfer. These and other virtues have led to clinical trials at University Hospitals Case Medical Center with Pc 4 being used in PDT for the treatment of cancer.

### Reactions of singlet oxygen

The primary target of  $^1O_2$  generated in biological systems may be proteins. Proteins have a high rate constant for reaction, a high effective concentration in cells [11], and proximity to the lipid membranes proposed as the site of  $^1O_2$  generation. The reactions of  $^1O_2$  with amino acids and the reaction products have been reviewed [11,12]. Among the proteogenic amino acids, His, Tyr, Met, Cys, and Trp are known to undergo rapid chemical reaction with  $^1O_2$  at physiological pH [12,13]. Model reactions of singlet oxygen with His [14-16] and Trp [17] have resulted in the identification of numerous different products, but the major reaction pathways are summarized in Figure 1. While numerous studies have shown  $^1O_2$  can inactivate enzymes [13,18], classical studies identified the modified residue types by losses detected following acid hydrolysis and amino acid analysis [13] [19]. Typically, His and Trp decrease with occasional increases of Asp/Asn. The identification of the actual products of the reaction of  $^1O_2$  with proteins is less well studied, but it is anticipated that these five residues may be most frequently modified. These five residues may also react with  $\cdot\text{OH}$ ,  $\text{ONOO}^-$ , and  $\text{H}_2\text{O}_2$ . Consequently, determination of which residues are modified will be insufficient to implicate  $^1O_2$ ; this would require the characterization of unique oxidative modifications.

The primary chemical reactions of  $^1O_2$  with mitochondrial proteins are important since confocal microscopy indicates Pc 4 localizes in mitochondrial and ER/Golgi membranes [20-23]. PDT with Pc 4, as well as with other photosensitizers that localize to mitochondria, induces apoptosis in many types of cells and tumors [8,24,25]. The primary apoptotic mechanism triggered by Pc 4-PDT is the mitochondrial (intrinsic) pathway, wherein photooxidation damage leads to opening of the permeability transition pore, loss of the mitochondrial membrane potential, release of cytochrome-*c* (Cyt-*c*) from mitochondria into the cytosol [21,26,27], and the activation of a cascade of apoptosis-mediating caspases [8]. Upon photoirradiation of Pc 4-loaded cells, a sub-set of mitochondrial and ER proteins undergoes immediate photodamage, as revealed on Western blots as the loss of native protein and the appearance of high-molecular weight complexes [22,28,29].

## Role of Cytochrome-c in response to $^1\text{O}_2$

The precise mechanism for the PDT-induced release of Cyt-*c* from the mitochondrial intermembrane space remains unclear. Fluorescence resonance energy transfer studies indicate that some Pc 4 colocalizes with cardiolipin [23], a phospholipid of the inner mitochondrial membrane. Cardiolipin has an affinity for Cyt-*c*, placing Pc 4 in the vicinity of Cyt-*c*. Oxidation of cardiolipin could disrupt electrostatic and hydrophobic interactions to create a soluble pool of Cyt-*c* that can pass into the cytosol in a Bax-dependent process [30]. Oxidized Cyt-*c* may enhance its own release through its acquired peroxidase activity, which leads to increased cardiolipin oxidation [31,32].

As an abundant 12-kDa protein localized to the intermembrane space of mitochondria, Cyt-*c* is situated in the region where Pc 4-PDT would generate  $^1\text{O}_2$ . Characterization of the chemical modification will provide insight into the loss of function reported following exposure of Cyt-*c* to  $^1\text{O}_2$ . Further identifying modifications of amino acid residues in Cyt-*c* that would be uniquely attributable to reaction with  $^1\text{O}_2$  could provide a specific marker for this ROS and its mitochondrial production. Previous studies have shown that  $^1\text{O}_2$  modifies Cyt-*c* by oxidizing its ferro-form and inactivating its function as an electron carrier [33,34]. Amino acid analysis of Cyt-*c* following visible light irradiation of a hematoporphyrin solution resulted in destruction of His, Trp, Tyr, and Met residues [19]. Direct evidence for modification of amino acid side chains of Cyt-*c* or other mitochondrial proteins by  $^1\text{O}_2$  has not been reported to our knowledge [35]; however, the chemical modification of Cyt-*c* attendant to various forms of oxidative stress has been well studied [36-40].

Because  $^1\text{O}_2$  is difficult to detect in cells, there is an interest in identifying oxidized cellular products that are uniquely derived from that oxidant, thereby directly indicating its presence, and potentially serving as a surrogate marker of photodynamic dose. One such product is  $3\beta$ -hydroxy- $5\alpha$ -cholest-6-ene-5-hydroperoxide derived from  $^1\text{O}_2$  reaction with cholesterol [41]. Since there is little cholesterol in mitochondrial membranes, direct targets of  $^1\text{O}_2$  attack present in mitochondria would be more useful for studying PDT with Pc 4 or similar photosensitizers on this organelle. Either cardiolipin or Cyt-*c* might be directly oxidized by PDT-generated  $^1\text{O}_2$  to yield a signature product.

In this study, Cyt-*c* and two model peptides, P824 and tryptophan cage (Trp-cage) [42], were irradiated with 670–675 nm light in the presence of the phthalocyanine Pc 4-malate salt (Pc 4-m) and analyzed by MALDI-TOF-MS and LC-ESI-MS<sup>2</sup>. P824, an 8-residue peptide (ASHLGLAR) with a single His residue, and Trp-cage, a 20-residue peptide (NLYIQWLKDGGPSSGRPPPS) with single Tyr and Trp residues, were chosen to characterize the reaction of  $^1\text{O}_2$  with these residues in their context as residues in a peptide. Then, photooxidation of Cyt-*c* by Pc 4 and light was examined by mass spectrometry revealing evidence of oxidative modification of Trp and His residues, with the His modification potentially diagnostic for oxidation by  $^1\text{O}_2$ .

## Experimental

### Materials

Horse heart Cyt-*c* was purchased from Calbiochem (La Jolla, CA, USA),  $\alpha$ -cyano-4-hydroxycinnamic acid and P824 from Sigma (St. Louis, MO, USA), sequencing-grade modified trypsin from Promega (Madison, WI, USA) and ZipTips from Millipore (Bedford, MA, USA). Trp-cage was synthesized by Biomer Technology (Hayward, CA, USA). Millipore water (18 M $\Omega$ ) was used throughout.

## Synthesis of Pc 229

Pc 229, herein referred to as Pc 4-malate,  $[\text{HOSiPcOSi}(\text{CH}_3)_2(\text{CH}_2)_3\text{NH}(\text{CH}_3)_2]^+ [\text{HO}(\text{C}(\text{O})\text{CH}_2\text{CHOHC}(\text{O})\text{O})^-]$ . A mixture of Pc 4,  $\text{HOSiPcOSi}(\text{CH}_3)_2(\text{CH}_2)_3\text{N}(\text{CH}_3)_2$ , (10 mg, 14  $\mu\text{mol}$ ) synthesized as described [43] and L-(–)-malic acid (2.3 mg, 17  $\mu\text{mol}$ ) in ethanol (20 mL) was stirred for 30 min, and then dried by rotary evaporation (room temperature). The solid was dissolved in toluene and chromatographed (Bio-Beads S-X3; Bio-Rad Laboratories, Hercules, CA) with toluene elution, dried, washed ( $\text{CH}_3\text{CN}$ ), air dried, and weighed (5.1 mg, 43 %). UV-vis (toluene)  $\lambda_{\text{max}}$ , nm (log  $\epsilon$ ): 670 (5.0). NMR ( $\text{CDCl}_3$ ):  $\delta$  9.35 (m, 8H, 1, 4-Pc H), 8.30 (m, 8H, 2, 3-Pc H), 3.92 (d, H,  $\text{C}(\text{O})\text{OCHOH}$ ), 2.55 (m, 2H,  $\text{C}(\text{O})\text{OCHOHCH}_2$ ), 1.95 (s, 6H,  $\text{NCH}_3$ ), 1.21 (t, 2H,  $\text{SiCH}_2\text{CH}_2\text{CH}_2$ ), -1.20 (m, 2H,  $\text{SiCH}_2\text{CH}_2$ ), -2.29 (t, 2H,  $\text{SiCH}_2$ ), -2.94 (s, 6H,  $\text{SiCH}_3$ ). HR-MALDI ( $m/z$ ):  $[\text{M-OH-C}(\text{O})\text{OCHOHCH}_2\text{C}(\text{O})\text{OH}]^+$  calcd for M as  $\text{C}_{43}\text{H}_{41}\text{N}_9\text{O}_7\text{Si}_2$ , 700.2425; found, 700.2452, 700.2421.

Pc 4-malate is a blue solid which dissolves in  $\text{C}_2\text{H}_5\text{OH}$  (4.2 mg/mL) or in 2%  $\text{C}_2\text{H}_5\text{OH}:\text{H}_2\text{O}$  (v:v) solution (0.2 mg/mL). It is also soluble in  $\text{CH}_2\text{Cl}_2$ , dimethylformamide, moderately soluble in toluene, and insoluble in hexanes.

## $^1\text{O}_2$ generation

Pc 4-malate was dissolved in ethanol and diluted in  $\text{H}_2\text{O}$  or  $\text{D}_2\text{O}$  to give a final concentration of 6  $\mu\text{M}$  and a residual ethanol concentration 1-2%. The final concentration of Pc 4-malate was determined by measuring the absorbance at 674 nm ( $\epsilon = 230,000 \text{ M}^{-1}\text{cm}^{-1}$ ). To these solutions, Cyt-*c* or the peptides were added to final concentrations of 5  $\mu\text{M}$  and 10  $\mu\text{M}$ , respectively, at pH 6.5. Sample solutions (2 mL) were pipetted into 3.5-cm diameter tissue culture dishes, resulting in solution depths of about 2 mm, and exposed to 100  $\text{mJ}/\text{cm}^2$  red light produced by a light-emitting diode array (EFOS, Mississauga, Ontario, Canada,  $\lambda_{\text{max}}$  670–675 nm) at room temperature over varying times up to 50 min. To determine whether peroxides were present, thiourea was added to 1 mM following irradiation in one set of experiments and during the irradiation in a second set of experiments [44]

## Tryptic digestion

Irradiated Cyt-*c* solutions were dried in a Speed Vac concentrator, redissolved in 20 mM Tris-HCl, pH 7.4, and filtered through cotton to remove the precipitated Pc 4. The Cyt-*c* in the filtrate was reduced with DTT, alkylated with iodoacetamide and digested with trypsin at 37 °C for 16 h. The digest was desalted and eluted with 60% acetonitrile and 40% TFA (0.1%) onto a stainless steel MALDI sample block from ZipTips according to the manufacturer's instructions. Alternatively, the digest was concentrated to 80  $\mu\text{M}$  for LC-MS<sup>2</sup> analysis.

## MALDI-TOF mass spectrometry

MALDI-TOF mass spectrometric analyses of the oxidized peptides and Cyt-*c* were performed with a Bruker BiFlex III MALDI-TOF-MS equipped with a pulsed  $\text{N}_2$  laser (3 ns pulse at 337 nm) and XTOF acquisition software. The reflectron mode was used with an accelerating voltage of -20 kV and a laser power attenuation ranging from 70 to 85 with an average of 3000 shots for digested Cyt-*c* and 500 shots for the peptides. The data were processed using  $m/z$  software [45].

## Electrospray mass spectrometry

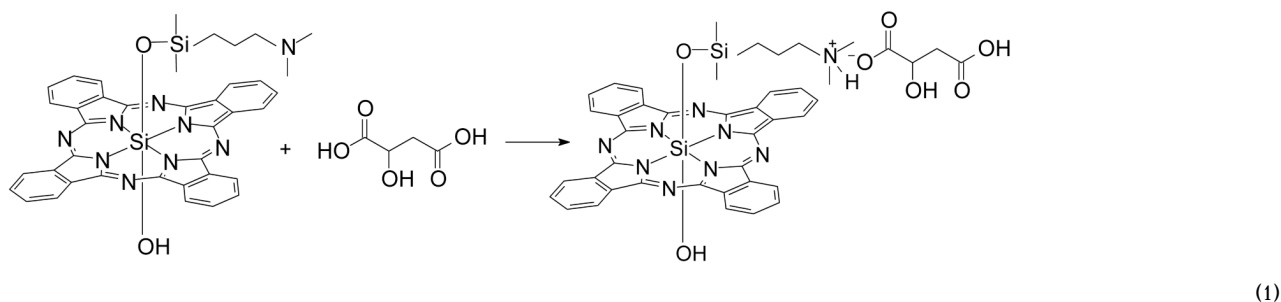
LC-MS<sup>2</sup> analyses of the Cyt-*c* tryptic digests were performed using a ThermoFinnigan LCQ Advantage spectrometer in the Case Center for Proteomics (Cleveland, OH, USA). The peptides were analyzed by data-dependent acquisition of full scan mass spectra and MS<sup>2</sup> scans for the most abundant ion, after separation on a reverse-phase Vydac C18 column (4.6 mm  $\times$  15 cm) with a flow rate of 200  $\mu\text{L}/\text{min}$  eluted with a linear gradient from 80% buffer A (5%

aqueous ethanol containing 0.05% TFA) to 100% buffer B (95% aqueous ethanol containing 0.05% TFA) over 63 min. The peptides modified by  $^1\text{O}_2$  were verified by direct comparison between the experimental spectra and theoretical peptide sequences using Xcalibur 2.0 software. The predicted fragment ions of peptides containing modified amino acid residues were calculated using an Excel spreadsheet program which permits the elemental composition of a modified residue to be specified. To determine the extent of Cyt-*c* modification, electrospray ionisation-TOF mass spectra were obtained with Applied Biosystems Q-STAR XL hybrid quadrupole-TOF mass spectrometer and analyzed with Bioworks software.

## Results and Discussion

### Pc 4 malate

The amino nitrogen of the siloxy ligand of Pc 4 is basic with an estimated  $\text{pK}_a$  of 9.6, thus the formation of Pc 4-malate from Pc 4 and malic acid,  $\text{pK}_a$  3.4, is a simple acid-base reaction (Equation 1):



In physiologic pH buffers, Pc 4 should exist as a minimally soluble dissociated salt, although there is a potential for formation of dimers and/or higher oligomers. It is possible to solubilize Pc 4-malate in 1-2 % ethanol in water, the water solubility facilitates the work in biological media. The dissolution of Pc 4-malate gave a clear blue solution, and in the conditions used, the Beer-Lambert law was obeyed, enabling the use of the Pc 4-malate salt in this and other biological studies of Pc 4 where, as in the present study, the solubility facilitates the work.

A particular virtue of silicon phthalocyanines is that their reactivity is largely restricted to type II photochemistry, i.e. energy transfer to  $\text{O}_2$  to generate  $^1\text{O}_2$ . Type I photochemistry, electron transfer to  $\text{O}_2$  generating superoxide directly and other reactive oxygen species indirectly, usually results in photobleaching of the photosensitizer. After irradiation of Pc4-malate under our experimental conditions, no photobleaching was observed [46]. The photochemistry of Pc 4 is characterized; it has a triplet lifetime expected for Pcs ( $> 100 \mu\text{s}$ ), a near diffusion limited second order rate constant for reaction with  $\text{O}_2$ , which results in a quantum yield of  $^1\text{O}_2$  of  $\sim 0.5$  in the absence of exogenous quenchers [46,47].

### Modification of peptides by $^1\text{O}_2$

His is known to be the most reactive of the proteogenic amino acids with  $^1\text{O}_2$ , with several observed products as shown in Figure 1A [11, 48]. The initial product appears to be the cyclic endoperoxide shown in Figure 1A which can be trapped at  $-100 \text{ }^\circ\text{C}$  but decomposes to the hydroperoxide at  $-78 \text{ }^\circ\text{C}$  and subsequently to the 2-imidazolone which can be hydrated at any of the carbon positions [16]. Similar modifications of the His residue in P824 were anticipated following red light irradiation of P824 in the presence of Pc 4-malate. Analysis by MALDI-MS identified only two new peaks corresponding to  $M+14$  and  $M+32$  (Figure 2A). HPLC- $\text{MS}^2$  analysis confirmed that these new peaks were appropriately assigned to modification of



the His residue. The structures of His+14 and His+32 have been previously characterized as the 2-imidazolone and as the hydrated imidazolone species, respectively [11, 48]. Since  $^1\text{O}_2$  reacts with the unprotonated imidazole, the rate constant of His increases at higher pH consistent with our observation that the relative intensity of the M+14 and M+32 peaks increased when subjected to irradiation at higher pH (data not shown).

The red-light irradiation of Trp-cage in the presence of Pc 4-malate gave four new peaks, M-59, M+16, M+32, and M+48, when analyzed by MALDI-MS (Figure 2B). Since the rate constant for reaction of Trp ( $3 \times 10^7 \text{ M}^{-1}\text{s}^{-1}$ ) with  $^1\text{O}_2$  is only 4-fold faster than that of Tyr ( $0.8 \times 10^7 \text{ M}^{-1}\text{s}^{-1}$ ), it was anticipated that the modified peaks were derived from oxidation of the both Tyr and Trp residues. However, LC-MS<sup>2</sup> analysis found that all of the mass altered peptides were derived solely from oxidative modification of Trp. When the reactions of Trp containing dipeptides with  $^1\text{O}_2$  were studied, structures corresponding to mass increments of +4, +16 and +32 were identified as suggested in Figure 1B [11] [49]. Consistent with our observation of +48 and -59 photochemical products, additional peaks attributed to addition of two  $\text{O}_2$  molecules and subsequent fragmentations were observed when Trp-Phe was reacted with  $^1\text{O}_2$  [49] however mass spectrometry does not provide sufficient information to unequivocally assign structures to these modifications. The LC-MS<sup>2</sup> spectrum of M+48 peak is shown in Figure 3. It is possible that +48 precursor ion could result from the incorporation of a total of 3 oxygen atoms in the Trp and Tyr residues; however, this possibility is discounted by the presence of  $b_3$ ,  $b_4$ ,  $y_{15}+48$ , and  $y_{16}+48$  ions which indicate that the extra mass is not present on the Tyr residue but is contained in product ions containing modified Trp residues (Figure 3). No alterations in the mass spectra were observed following treatment of the oxidized peptides with thiourea (data not shown). If hydroperoxides were present, a decrease in the M+32 peak would be anticipated with a concomitant increase of the M+16 peak. Thiourea reduction of endoperoxides would shift the M+32 peak to M+34. The absence of these shifts is consistent with *N*-formylkynurenine being the source of the M+32 peak.

### Reaction of $^1\text{O}_2$ with Cyt-c

Cyt-*c* is a heme protein with 104 amino acids containing one Trp, four Tyr, three His, and two Met residues. Following our Pc 4-malate irradiation procedure, electrospray-time of flight mass spectra revealed several modifications (Figure 4). The predominance of +16 and +32 modifications is consistent with most modified proteins having a single modification, which are predominantly the addition of one or two oxygen atoms. Samples of Cyt-*c* subjected to irradiation with Pc 4-malate in  $\text{D}_2\text{O}$  were digested with trypsin and analyzed by both MALDI TOF-MS and LC-MS<sup>2</sup>. Consistent with the electrospray mass spectrum of the intact Cyt-*c*, several new peaks corresponding to mass increments +14, +16, and +32 Da relative to unmodified peptides appeared in the MALDI-TOF spectrum of the tryptic digest. In addition to determining which residue is modified in the peptides identified by MALDI, the fragmentation data from LC-MS<sup>2</sup> made it possible to recognize additional peptides with oxidative modifications. These peptides, the mass increments, and the identity of the modified residues are presented in Table 1.

### Characterization of Met and Trp modification

Cyt-*c* has two Met residues, Met65 and Met80. LC-MS<sup>2</sup> spectra showed that both of these residues were modified as M+16, the common sulfoxide oxidation product of this amino acid. Generation of  $^1\text{O}_2$  resulted in both +32 and +16 oxidation products of Met80. The unusual +32 adduct may be attributed to either a peroxo species or a sulfone [50]. Met residues are also oxidized by peroxides,  $\cdot\text{OH}$ , and  $\text{ONOO}^-$  [51], so the presence of modified Met residues in general, and Met80 in particular, can not be used to identify the responsible ROS. However since oxidation of Met80 is subject to oxidation by all of these ROS, it could potentially serve as a generalized monitor of ROS *in vivo*. Furthermore modification of Met80 may be

pharmacologically relevant as it is associated with activation of Cyt-*c*'s peroxidase activity [52,53]

A unique Trp present in Cyt-*c* is also modified by exposure to  $^1\text{O}_2$ . Although four different types of modification were observed for Trp modification by  $^1\text{O}_2$  in Trp-cage, only the Trp +16 adduct was found in the reaction with Cyt-*c*. The MS<sup>2</sup> data were specifically interrogated for the +32, +48 and -59 modifications by examining the appropriate mass range ion chromatograms and by adding the appropriate masses to the data dependent include mass list. In spite of these specific efforts, no evidence for these modifications was found. The difference between Trp-cage where the +32 adduct predominates and Cyt-*c* where the +16 adduct predominates, suggests that the protein environment alters the oxidation pathway initiated by  $^1\text{O}_2$ , favoring the reduction of the initial peroxide or delaying the formation of the endoperoxide shown in Figure 1B. Exposure of rat heart Cyt-*c* to  $\gamma$ -irradiation also resulted in a mass increment of +16 in the Trp-containing peptide, so detection of Trp oxidation by mass spectrometry cannot serve as a specific marker for  $^1\text{O}_2$  [54].

### Characterization of His modifications

There are three His residues, His18, His26, and His33, present in Cyt-*c*. Both His26 and His33 were found to be modified. While an unmodified tryptic peptide containing His18 was identified, this peptide contains the Cys residues covalently attached to the heme, which can confound any interpretation of observed and unobserved modifications. The solvent accessibility of His26 and His33 is shown in Figure 5 adapted from the crystal structure.

Five different modifications of His33 with mass increments of -22, +14, +26, +32 and +34 Da were detected relative to the unmodified K.TGPNLHGLFGK<sup>38</sup> tryptic peptide. The relative intensities of the modified peaks were <2% of the unmodified peptide peak in the MALDI spectrum (Figure 6). The proposed structures of the His-22, His+14 and His+32, adducts are shown in Figure 1A. The structure and mechanism of the His+32 adduct following reaction with  $^1\text{O}_2$  has been established by MS<sup>2</sup> and MS<sup>3</sup>. A key observation was identifying that the hydroxyl incorporated  $^{17}\text{O}$  when  $^1\text{O}_2$  was generated in  $\text{H}_2^{17}\text{O}$  [15]. An imidazole endoperoxide has been isolated at -78 °C but decomposes on warming. Since thiourea will not react with the hydroxyimidazolone our results are consistent with this structure as well. The loss of water (His+14) may occur during the ionization and not be present in solution. The LC-MS<sup>2</sup> spectrum of the peptide identified as the -22 adduct is shown in Figure 7A. This spectrum confirms the sequence of the peptide and is consistent with a multiple oxidation of the His residue to asparagine [14]. Ozonolysis of His results initially in a +48 adduct identified by MS<sup>2</sup> as 2-amino-4-oxo-4-(3-formylureido)butanoic acid followed by decomposition to asparagine, the -22 adduct [55,56]. This conversion is also consistent with the earliest reports of His photooxidation [14], but does not represent the major product detected following minimal reaction with  $^1\text{O}_2$  with the products detected by MS. With its relative intensity and unique mass increment, the peptides derived from Cyt-*c* containing a His33 +14 modification have the potential to be used as a marker for the presence of mitochondrial  $^1\text{O}_2$ .

The peptide containing His18 is clearly modified as +14 and +28, but it is not clear whether the modification was derived from His18. Two Cys residues present in this peptide make the LC-MS<sup>2</sup> spectrum more complicated. Both Cys residues form thioether linkages to the side chains of the heme. It is not certain when these thioethers are cleaved but MALDI-TOF spectra routinely confirmed the existence of the free thiol groups in the peptide (mass of the peptide matches with theoretical value). However, they when analyzed by LC-MS<sup>2</sup> the reduction in mass by 2 Da suggested the formation of an intramolecular disulfide bond.

Modification of His26 can be detected only when the trypsin cleavage at Lys27 is missed, with the resulting peptide containing both His26 and His33 (K<sup>25</sup>.HKTGPNLHGLFGK<sup>38</sup>). In every

MALDI-TOF spectrum of peptides obtained following exposure of Cyt-*c* to  $^1\text{O}_2$ , the HKTGPNLHGLFGK +32 Da peptide was always the most intense modified peptide, and sometimes the only one observed. LC-MS<sup>2</sup> confirmed that from the precursor ion  $m/z$  734.27, the only detectable product ions were derived from oxidation of His26 alone (Figure 7B). The unmodified precursor peptide peak with one missed cleavage has not been observed. We attribute these observations to the potential for His26 oxidized to the 2-imidazolone to form a crosslink by addition of the  $\epsilon$  amine of Lys27 to the electrophilic imine-one. Previous studies of His or imidazole oxidation have provided significant precedents for this addition [14] [57] [15]. This intramolecular crosslink would prevent the tryptic digestion following Lys27, and eliminate the potential to observe MS<sup>2</sup> peaks corresponding to cleavage of the His26-Lys27 peptide bond. Because His26 was shown to be the most sensitive to  $^1\text{O}_2$ , there may be other modifications present, as observed for His33. However, only the M+14 and M+32 species prevent the tryptic cleavage. In the absence of crosslink formation, the resulting dipeptide generated by tryptic digestion and containing the modified His residue, H\*K, would be too small to detect.

### Potential use of His 26 modification as a marker of $^1\text{O}_2$

The +14 and +32 modifications appear to be specific to  $^1\text{O}_2$ . His modification by  $\bullet\text{OH}$  results in +16 adducts and does not induce the formation of the observed Lys-His crosslink. The specificity of this modification and the absence of any background peptide should make this peptide a marker for the presence of  $^1\text{O}_2$  *vis-a-vis* its reaction with Cyt-*c* in biological samples. This sequence within Cyt-*c* is 100% conserved in chicken, rat, mouse, bovine and human proteins, suggesting this modified peptide could be observed in most species. The intensity of this peak relative to that of an unmodified Cyt-*c* peptide at low-modest extents of modification is proportional to the irradiation time and hence  $^1\text{O}_2$  exposures (Figure 8).

### Modification of Phe and Tyr

An unanticipated observation is the modification of Phe residues and absence of Tyr modification. Cyt-*c* has four Phe residues, Phe10, Phe36, Phe46, and Phe82. Both Phe36 and Phe82 were found to be modified as Phe+16. Because the aromatic ring of Phe is not inherently reactive with  $^1\text{O}_2$ , benzene and toluene have been used as solvents for  $^1\text{O}_2$  reactions [58]. The apparent hydroxylation of these Phe residues reflects either an enhanced reactivity of the aromatic ring or the production of a secondary oxidant during the generation of and exposure to  $^1\text{O}_2$ . Despite the presence of Phe36 and Phe82 in peptides containing His33 and Met80, respectively, which have been modified, the MS<sup>2</sup> spectra strongly implicate Phe oxidation due to the presence of  $y_3+16$ ,  $y_4+16$ , and  $y_5+16$  as well as  $b_6$  and  $b_8$  product ions (Figure 9). The two Phe residues modified following generation of  $^1\text{O}_2$  were also modified by the same mass increment following exposure to  $\bullet\text{OH}$  generated by radiolysis. However modifications of Phe10 and Phe46 were not detected, although they are predominant modifications generated by  $\bullet\text{OH}$  [36,59].

Because Tyr residues are known to react with  $^1\text{O}_2$ , coupled with their solvent accessibility, the absence of any oxidized Tyr residues is notable. The unmodified peptides can be detected, but +14, +16, and +32 Da ion chromatograms relative to the unmodified peptide, in +1, +2, or +3 charge states, failed to reveal any candidate molecular ions for further analysis.

### Cyt-*c* oxidation by $^1\text{O}_2$ in $\text{H}_2\text{O}$

The lifetime of  $^1\text{O}_2$  in  $\text{H}_2\text{O}$  is  $\sim 10$  fold smaller than in  $\text{D}_2\text{O}$ , thus decreasing its effective reactivity. Oxidation of Cyt-*c* in  $\text{H}_2\text{O}$  gave only one distinguishable modified peak in the MALDI spectrum corresponding to peptide HKTGPNLHGLFG +32. LC-MS<sup>2</sup> analysis identified only six of the oxidatively modified peaks that were present following oxidation in  $\text{D}_2\text{O}$ . The list of modified peptides following generation of  $^1\text{O}_2$  in  $\text{H}_2\text{O}$  is shown in Table 1.



The greater oxidation in the presence of D<sub>2</sub>O provides support for these modifications to be derived from the action of <sup>1</sup>O<sub>2</sub>.

## Summary

<sup>1</sup>O<sub>2</sub> generated by photoexcitation of Pc4 modifies numerous amino acids of Cyt-*c*, predominantly His, Trp and Met. The reaction with these residues can be influenced by the protein environment, with different oxidatively generated modifications observed when the residues are present in different environments. The reaction with His generates both +32 and +14 adducts that are not observed with other ROS and may serve as markers for the in vivo production of <sup>1</sup>O<sub>2</sub>. This is particularly true for His26 in Cyt-*c* and for PDT studies where the photosensitizer is known to accumulate in mitochondria.

## References

1. Schmidt R. Photosensitized generation of singlet oxygen. *Photochem Photobiol* 2006;82:1161–1177. [PubMed: 16683906]
2. Lindig BA, Rodgers MAJ, Schaap AP. Determination of the lifetime of singlet oxygen in water-d<sub>2</sub> using 9,10-anthracenedipropionic acid, a water-soluble probe. *J Am Chem Soc* 1980;102:5590–5593.
3. Cadet J, Ravanat JL, Martinez GR, Medeiros MH, Di Mascio P. Singlet oxygen oxidation of isolated and cellular DNA: Product formation and mechanistic insights. *Photochem Photobiol* 2006;82:1219–1225. [PubMed: 16808595]
4. Davies MJ. Reactive species formed on proteins exposed to singlet oxygen. *Photochem Photobiol Sci* 2004;3:17–25. [PubMed: 14743273]
5. Girotti AW. Lipid hydroperoxide generation, turnover, and effector action in biological systems. *J Lipid Res* 1998;39:1529–1542. [PubMed: 9717713]
6. Miyamoto S, Martinez GR, Medeiros MH, Di Mascio P. Singlet molecular oxygen generated from lipid hydroperoxides by the Russell mechanism: Studies using <sup>18</sup>O-labeled linoleic acid hydroperoxide and monomol light emission measurements. *J Am Chem Soc* 2003;125:6172–6179. [PubMed: 12785849]
7. Dougherty TJ, Gomer CJ, Henderson BW, Jori G, Kessel D, Korbek M, Moan J, Peng Q. Photodynamic therapy. *J Natl Cancer Inst* 1998;90:889–905. [PubMed: 9637138]
8. Oleinick NL, Morris RL, Belichenko I. The role of apoptosis in response to photodynamic therapy: What, where, why, and how. *Photochem Photobiol Sci* 2002;1:1–21. [PubMed: 12659143]
9. Weishaupt KR, Gomer CJ, Dougherty TJ. Identification of singlet oxygen as the cytotoxic agent in photoinactivation of a murine tumor. *Cancer Res* 1976;36:2326–2329. [PubMed: 1277137]
10. Oleinick NL, Antunez AR, Clay ME, Rihter BD, Kenney ME. New phthalocyanine photosensitizers for photodynamic therapy. *Photochem Photobiol* 1993;57:242–247. [PubMed: 8451285]
11. Davies MJ. Singlet oxygen-mediated damage to proteins and its consequences. *Biochem Biophys Res Commun* 2003;305:761–770. [PubMed: 12763058]
12. Wright A, Bubb WA, Hawkins CL, Davies MJ. Singlet oxygen-mediated protein oxidation: Evidence for the formation of reactive side chain peroxides on tyrosine residues. *Photochem Photobiol* 2002;76:35–46. [PubMed: 12126305]
13. Miki T, Yu L, Yu CA. Hematoporphyrin-promoted photoinactivation of mitochondrial ubiquinol-cytochrome c reductase: Selective destruction of the histidine ligands of the iron-sulfur cluster and protective effect of ubiquinone. *Biochemistry* 1991;30:230–238. [PubMed: 1846289]
14. Tomita M, Irie M, Ukita T. Sensitized photooxidation of histidine and its derivatives. Products and mechanism of the reaction. *Biochemistry* 1969;8:5149–5160. [PubMed: 5365801]
15. Au V, Madison SA. Effects of singlet oxygen on the extracellular matrix protein collagen: Oxidation of the collagen crosslink histidinohydroxylysinonorleucine and histidine. *Arch Biochem Biophys* 2000;384:133–142. [PubMed: 11147824]
16. Kang P, Foote CS. Photosensitized oxidation of <sup>13</sup>C, <sup>15</sup>N-labeled imidazole derivatives. *J Am Chem Soc* 2002;124:9629–9638. [PubMed: 12167059]

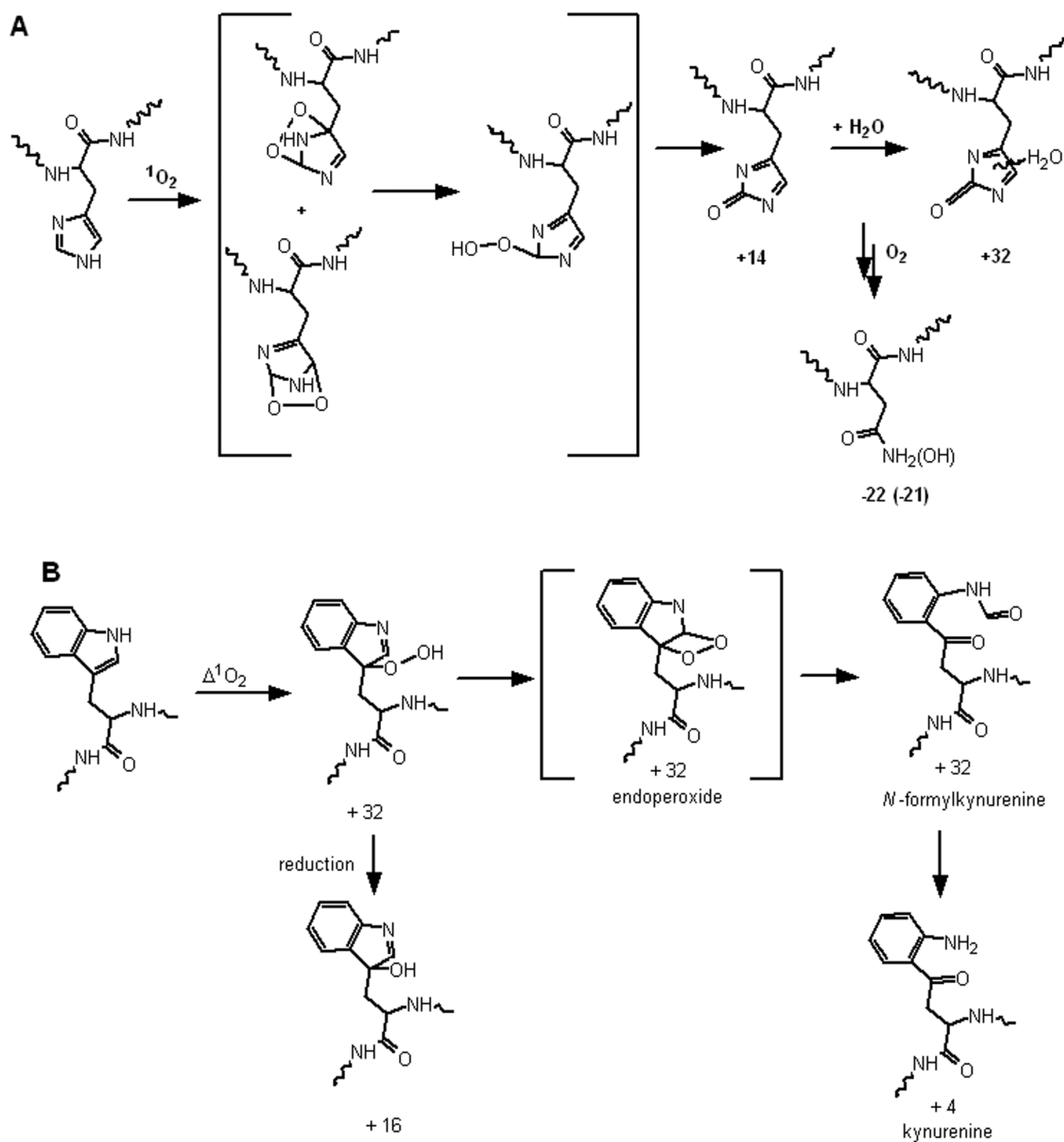
17. Nakagawa M, Watanabe H, Kodato S, Okajima H, Hino T, Flippen JL, Witkop B. A valid model for the mechanism of oxidation of tryptophan to formylkynurenine-25 years later. *Proc Natl Acad Sci U S A* 1977;74:4730–4733. [PubMed: 16592461]
18. Kukreja RC, Kearns AA, Zweier JL, Kuppusamy P, Hess ML. Singlet oxygen interaction with Ca(2+)-ATPase of cardiac sarcoplasmic reticulum. *Circ Res* 1991;69:1003–1014. [PubMed: 1657435]
19. Jori G, Gennari G, Galiazzo G, Scoffone E. Photo-oxidation of horse heart cytochrome c. Evidence for methionine-80 as a heme ligand. *FEBS Lett* 1970;6:267–269. [PubMed: 11947391]
20. Trivedi NS, Wang HW, Nieminen AL, Oleinick NL, Izatt JA. Quantitative analysis of Pc 4 localization in mouse lymphoma (ly-r) cells via double-label confocal fluorescence microscopy. *Photochem Photobiol* 2000;71:634–639. [PubMed: 10818795]
21. Lam M, Oleinick NL, Nieminen AL. Photodynamic therapy-induced apoptosis in epidermoid carcinoma cells. Reactive oxygen species and mitochondrial inner membrane permeabilization. *J Biol Chem* 2001;276:47379–47386. [PubMed: 11579101]
22. Usuda J, Chiu SM, Murphy ES, Lam M, Nieminen AL, Oleinick NL. Domain-dependent photodamage to Bcl-2. A membrane anchorage region is needed to form the target of phthalocyanine photosensitization. *J Biol Chem* 2003;278:2021–2029. [PubMed: 12379660]
23. Morris RL, Azizuddin K, Lam M, Berlin J, Nieminen AL, Kenney ME, Samia AC, Burda C, Oleinick NL. Fluorescence resonance energy transfer reveals a binding site of a photosensitizer for photodynamic therapy. *Cancer Res* 2003;63:5194–5197. [PubMed: 14500343]
24. Oleinick NL, Evans HH. The photobiology of photodynamic therapy: Cellular targets and mechanisms. *Radiat Res* 1998;150:S146–156. [PubMed: 9806617]
25. Kessel D, Luo Y. Photodynamic therapy: A mitochondrial inducer of apoptosis. *Cell Death Differ* 1999;6:28–35. [PubMed: 10200545]
26. Usuda J, Chiu SM, Azizuddin K, Xue LY, Lam M, Nieminen AL, Oleinick NL. Promotion of photodynamic therapy-induced apoptosis by the mitochondrial protein smac/diablo: Dependence on bax. *Photochem Photobiol* 2002;76:217–223. [PubMed: 12194220]
27. Chiu SM, Oleinick NL. Dissociation of mitochondrial depolarization from cytochrome c release during apoptosis induced by photodynamic therapy. *Br J Cancer* 2001;84:1099–1106. [PubMed: 11308261]
28. Xue LY, Chiu SM, Fiebig A, Andrews DW, Oleinick NL. Photodamage to multiple Bcl-xl isoforms by photodynamic therapy with the phthalocyanine photosensitizer Pc 4. *Oncogene* 2003;22:9197–9204. [PubMed: 14681679]
29. Xue LY, Chiu SM, Oleinick NL. Photochemical destruction of the Bcl-2 oncoprotein during photodynamic therapy with the phthalocyanine photosensitizer Pc 4. *Oncogene* 2001;20:3420–3427. [PubMed: 11423992]
30. Iverson SL, Enoksson M, Gogvadze V, Ott M, Orrenius S. Cardiolipin is not required for bax-mediated cytochrome c release from yeast mitochondria. *J Biol Chem* 2004;279:1100–1107. [PubMed: 14551208]
31. Vladimirov YA, Proskurnina EV, Izmailov DY, Novikov AA, Brusnichkin AV, Osipov AN, Kagan VE. Cardiolipin activates cytochrome c peroxidase activity since it facilitates H<sub>2</sub>O<sub>2</sub> access to heme. *Biochemistry (Mosc)* 2006;71:998–1005. [PubMed: 17009954]
32. Bayir H, Fadeel B, Palladino MJ, Witasp E, Kurnikov IV, Tyurina YY, Tyurin VA, Amoscato AA, Jiang J, Kochanek PM, DeKosky ST, Greenberger JS, Shvedova AA, Kagan VE. Apoptotic interactions of cytochrome c: Redox flirting with anionic phospholipids within and outside of mitochondria. *Biochim Biophys Acta* 2006;1757:648–659. [PubMed: 16740248]
33. Chernomorsky S, Wong C, Poretz RD. Pheophorbide a-induced photo-oxidation of cytochrome c: Implication for photodynamic therapy. *Photochem Photobiol* 1992;55:205–211. [PubMed: 1311860]
34. Giulivi C, Sarcansky M, Rosenfeld E, Boveris A. The photodynamic effect of rose bengal on proteins of the mitochondrial inner membrane. *Photochem Photobiol* 1990;52:745–751. [PubMed: 2089421]
35. Estevam ML, Nascimento OR, Baptista MS, Di Mascio P, Prado FM, Faljoni-Alario A, Zucchi Mdo R, Nantes IL. Changes in the spin state and reactivity of cytochrome c induced by photochemically generated singlet oxygen and free radicals. *J Biol Chem* 2004;279:39214–39222. [PubMed: 15247265]

36. Nukuna BN, Sun G, Anderson VE. Hydroxyl radical oxidation of cytochrome-c by aerobic radiolysis. *Free Radic Biol Med* 2004;38:1203–1213. [PubMed: 15451060]
37. Deterding LJ, Barr DP, Mason RP, Tomer KB. Characterization of cytochrome c free radical reactions with peptides by mass spectrometry. *J Biol Chem* 1998;273:12863–12869. [PubMed: 9582316]
38. Chen YR, Deterding LJ, Sturgeon BE, Tomer KB, Mason RP. Protein oxidation of cytochrome c by reactive halogen species enhances its peroxidase activity. *J Biol Chem* 2002;277:29781–29791. [PubMed: 12050149]
39. Isom AL, Barnes S, Wilson L, Kirk M, Coward L, Darley-Usmar V. Modification of cytochrome c by 4-hydroxy-2-nonenal: Evidence for histidine, lysine, and arginine-aldehyde adducts. *J Am Soc Mass Spectrom* 2004;15:1136–1147. [PubMed: 15276160]
40. Williams MV, Wishnok JS, Tannenbaum SR. Covalent adducts arising from the decomposition products of lipid hydroperoxides in the presence of cytochrome c. *Chem Res Toxicol* 2007;20:767–775. [PubMed: 17407328]
41. Girotti AW. Photosensitized oxidation of membrane lipids: Reaction pathways, cytotoxic effects, and cytoprotective mechanisms. *J Photochem Photobiol B* 2001;63:103–113. [PubMed: 11684457]
42. Verma A, Schug A, Lee KH, Wenzel W. Basin hopping simulations for all-atom protein folding. *J Chem Phys* 2006;124:044515. [PubMed: 16460193]
43. Li, YS.; Kenney, ME. Methods of syntheses of phthalocyanine compounds. U.S. Patent 5,763,602. 1998.
44. Adam W, Erden I. Synthesis and characterization of 7-spirocyclopropyl-2,3-dioxabicyclo[2.2.1]hept-5-ene. *J Org Chem* 1978;43:2737–2738.
45. Jensen PH, Weilguny D, Matthiesen F, McGuire KA, Shi L, Hojrup P. Characterization of the oligomer structure of recombinant human mannan-binding lectin. *J Biol Chem* 2005;280:11043–11051. [PubMed: 15653690]
46. He J, Larkin HE, Li YS, Rihter BD, Zaidi SIA, Rodgers MAJ, Mukhtar H, Kenney ME, Oleinick NL. The synthesis, photophysical and photobiological properties and in vitro structure-activity relationships of a set of silicon phthalocyanine PDT photosensitizers. *Photochem Photobiol* 1997;65:581–586. [PubMed: 9077144]
47. Anula HM, Berlin Jeffrey C, Wu H, Li YS, Peng X, Kenney Malcolm E, Rodgers Michael AJ. Synthesis and photophysical properties of silicon phthalocyanines with axial siloxy ligands bearing alkylamine termini. *J Phys Chem A* 2006;110:5215–5223. [PubMed: 16610845]
48. Chang SH, Teshima GM, Milby T, Gillece-Castro B, Canova-Davis E. Metal-catalyzed photooxidation of histidine in human growth hormone. *Anal Biochem* 1997;244:221–227. [PubMed: 9025937]
49. Posadz A, Biasutti A, Casale C, Sanz J, Amat-Guerri F, Garcia NA. Rose bengal-sensitized photooxidation of the dipeptides l-tryptophyl-l-phenylalanine, l-tryptophyl-l-tyrosine and l-tryptophyl-l-tryptophan: Kinetics, mechanism and photoproducts. *Photochem Photobiol* 2004;80:132–138. [PubMed: 15339205]
50. Ray WJ Jr. Photochemical oxidation. *Methods Enzymol* 1967;11:490–497.
51. Nukuna BN, Sun G, Anderson VE. Hydroxyl radical oxidation of cytochrome c by aerobic radiolysis. *Free Radic Biol Med* 2004;37:1203–1213. [PubMed: 15451060]
52. Kagan VE, Tyurina VA, Jiang J, Tyurina YY, Ritov VB, Amoscato AA, Osipov AN, Belikova NA, Kapralov AA, Kini V, Vlasova II, Zhao Q, Zou M, Di P, Svistunenka DA, Kurnikov IV, Borisenko GG. Cytochrome c acts as a cardiolipin oxygenase required for release of proapoptotic factors. *Nat Chem Biol* 2005;1:223–232. [PubMed: 16408039]
53. Kagan VE, Borisenko GG, Tyurina YY, Tyurina VA, Jiang J, Potapovich AI, Kini V, Amoscato AA, Fujii Y. Oxidative lipidomics of apoptosis: Redox catalytic interactions of cytochrome c with cardiolipin and phosphatidylserine. *Free Radic Biol Med* 2004;37:1963–1985. [PubMed: 15544916]
54. Shi WQ, Hu J, Zhao W, Su XY, Cai H, Zhao YF, Li YM. Identification of radiation-induced cross-linking between thymine and tryptophan by electrospray ionization-mass spectrometry. *J Mass Spectrom* 2006;41:1205–1211. [PubMed: 16924597]
55. Kotiaho T, Eberlin MN, Vainiotalo P, Kostianen R. Electrospray mass and tandem mass spectrometry identification of ozone oxidation products of amino acids and small peptides. *J Am Soc Mass Spectrom* 2000;11:526–535. [PubMed: 10833026]

56. Lloyd JA, Spraggins JM, Johnston MV, Laskin J. Peptide ozonolysis: Product structures and relative reactivities for oxidation of tyrosine and histidine residues. *J Am Soc Mass Spectrom* 2006;17:1289–1298. [PubMed: 16820303]
57. Shen HR, Spikes JD, Kopecekova P, Kopecek J. Photodynamic crosslinking of proteins. I. Model studies using histidine- and lysine-containing n-(2-hydroxypropyl)methacrylamide copolymers. *J Photochem Photobiol B* 1996;34:203–210. [PubMed: 8810538]
58. Arnbjerg J, Johnsen M, Frederiksen PK, Braslavsky SE, Ogilby PR. Two-photon photosensitized production of singlet oxygen: Optical and optoacoustic characterization of absolute two-photon absorption cross sections for standard sensitizers in different solvents. *J Phys Chem A Mol Spectrosc Kinet Environ Gen Theory* 2006;110:7375–7385. [PubMed: 16759125]
59. Nukuna, BN. PhD. Case Western Reserve University; Cleveland, OH: 2001. The hydroxyl radical: Its role in oxidative damage and as a biochemical tool to probe solvent accessible surfaces.
60. Qi PX, Beckman RA, Wand AJ. Solution structure of horse heart ferricytochrome c and detection of redox-related structural changes by high-resolution  $^1\text{H}$  NMR. *Biochemistry* 1996;35:12275–12286. [PubMed: 8823161]
61. Valadon, P.; Bernstein, HJ. La Jolla, CA 92039: 2001. RasTop, 1.3.1. <http://www.bernstein-plus-sons.com/software/rastop/RasTop.zip>

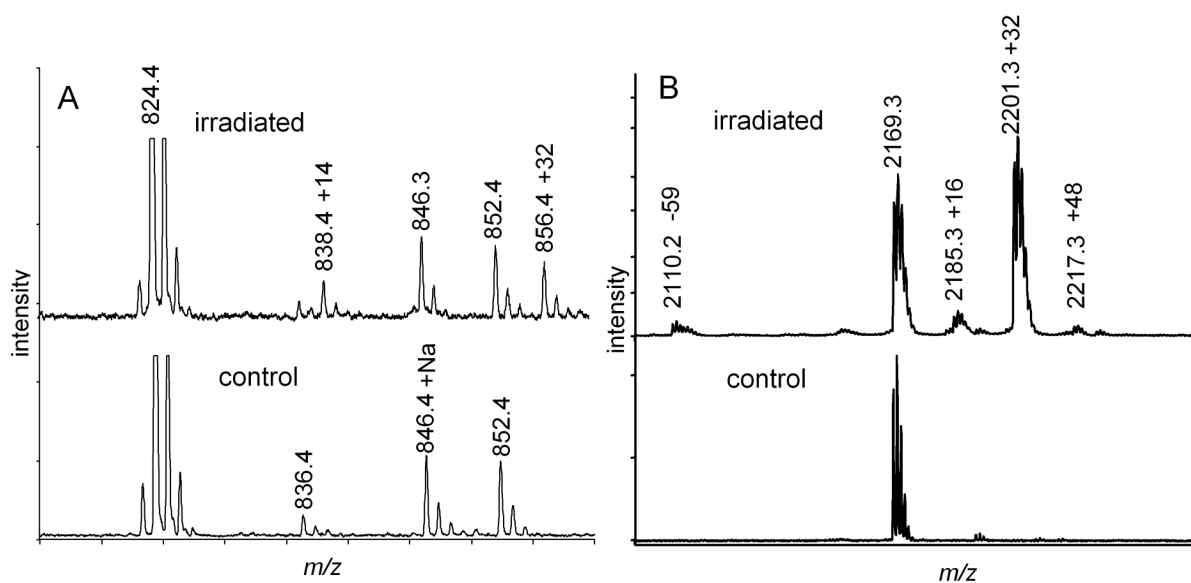
## List of abbreviations

<b>Cyt-c</b>	cytochrome <i>c</i>
<b>LC-MS<sup>2</sup></b>	liquid chromatography- tandem mass spectrometry
<b>P824</b>	peptide ASHLGLAR
<b>Pc</b>	phthalocyanine
<b>PDT</b>	photodynamic therapy
<b>ROS</b>	reactive oxygen species
<b>Trp-Cage</b>	tryptophan cage peptide

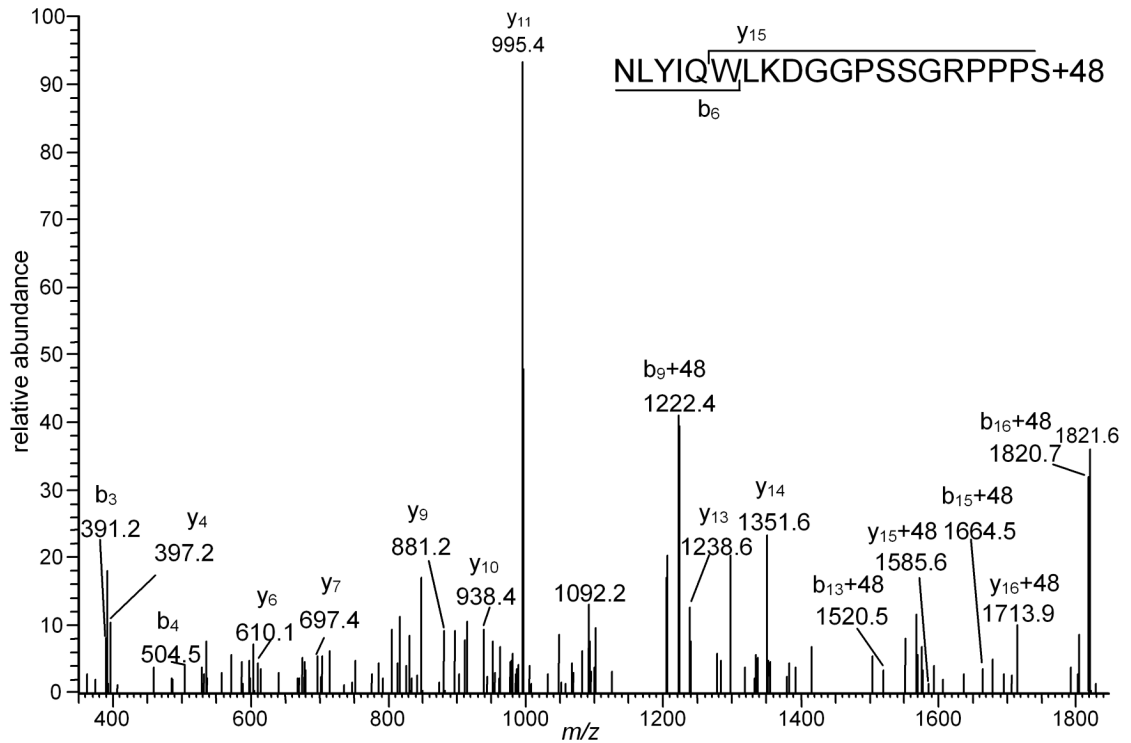


**Figure 1.** Potential oxidation products and their integral mass increments relative to the unmodified amino acid residue are given for (A) His and (B) Trp by  $^1\text{O}_2$ . The endoperoxide and hydroperoxide forms of the His modification are shown in brackets since they were shown to decompose above  $0^\circ\text{C}$  [16]. Water addition to the imidazolone may occur at any of the three imidazolone carbons. The scheme for Trp oxidation to *N*-formylkynurenine is based on [17]



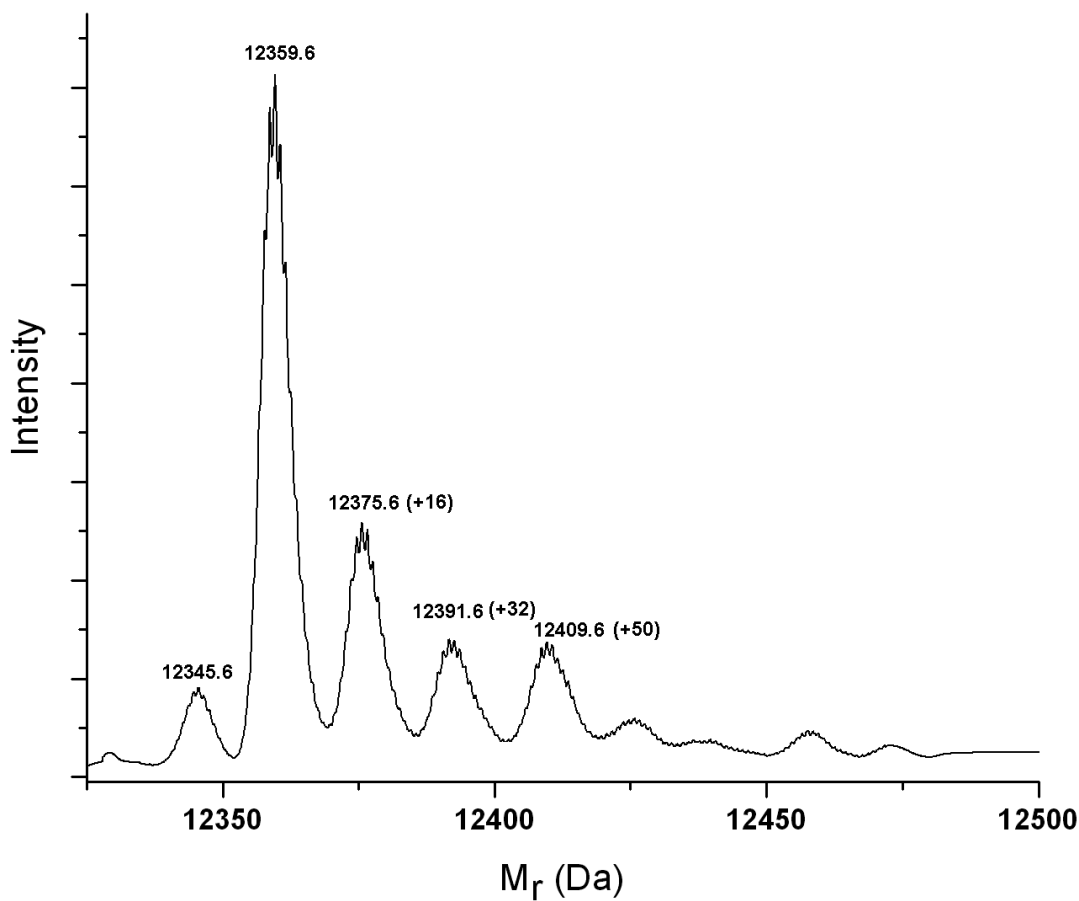


**Figure 2.** MALDI-TOF spectra of A) P824 and B) Trp-cage. Lower spectra are of the control peptides and upper spectra are derived from the same peptide exposed to  $^1\text{O}_2$  generated by photoexcitation of Pc 4. The observed oxidatively generated modifications are identified by their mass increments, A) +14 and +32 and B) -59, +16, +32 and +48, are indicated by the increment in mass of the monoisotopic peaks.

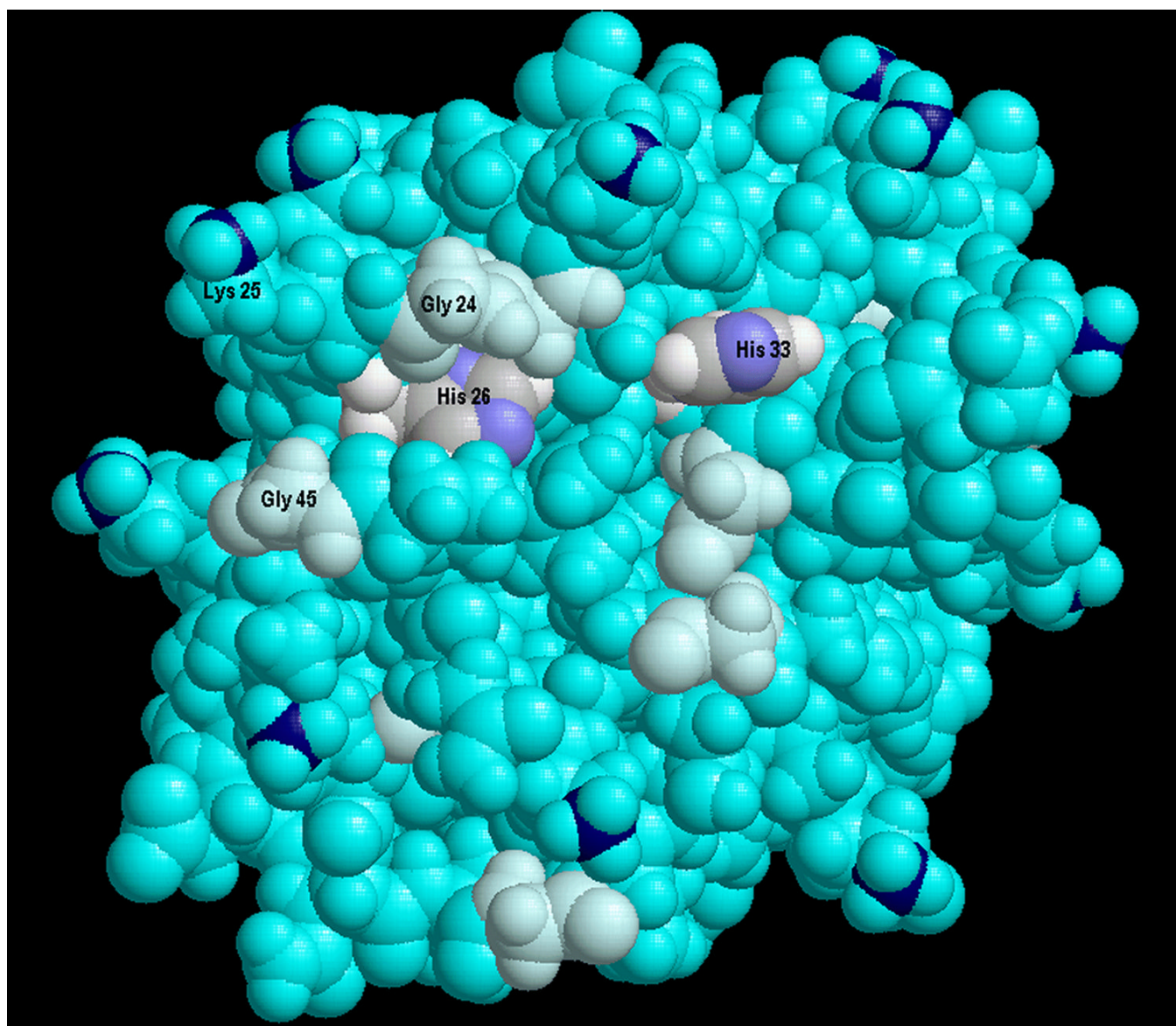


**Figure 3.**

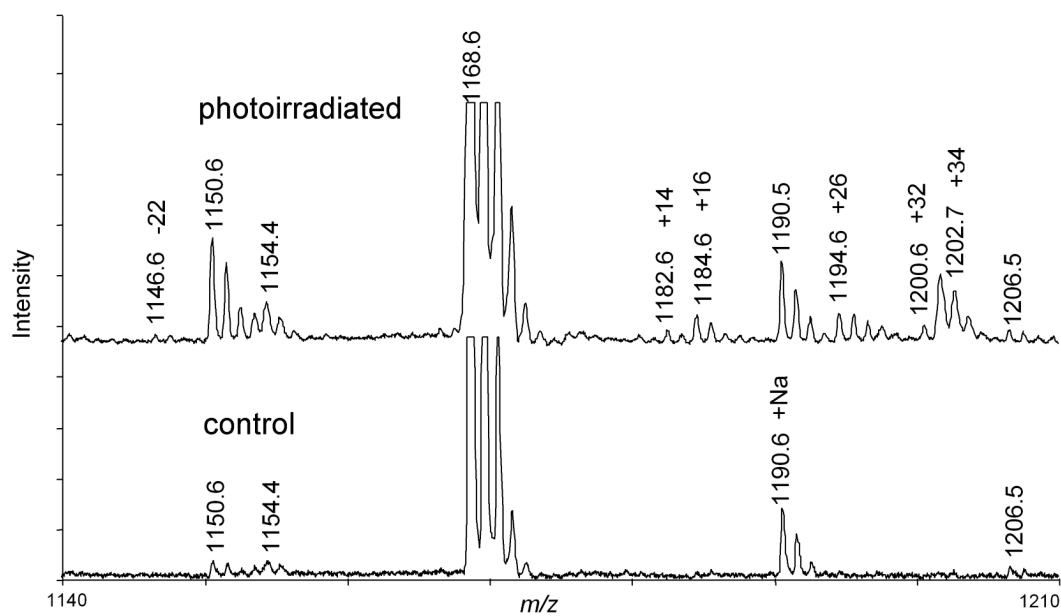
LC- MS<sup>2</sup> spectrum of a precursor ion whose  $m/z$  corresponds to the  $M_r$  of Trp-cage peptide +48. The inset shows the sequence of the peptide and indicates the sequence number of the first product ions in the b- and y-series that contain the modified tryptophan residue. The presence of unmodified b3-b4 and y4-y14 ions confirms the absence of modifications in residues other than the tryptophan.



**Figure 4.** Deconvoluted electrospray-time of flight mass spectrum of Cyt-*c* exposed to  $^1\text{O}_2$ . The prominent +16, +32 and +50 adducts of Cyt-*c* not present in control samples indicate that the reaction of  $^1\text{O}_2$  with Cyt-*c* results in oxidation of the protein. While evidence for multiple oxidations are present, the most intense +16 and +32 peaks correspond to single oxidative events and even the +50 adduct can correspond to the addition of a single  $\text{O}_2$  and  $\text{H}_2\text{O}$ .

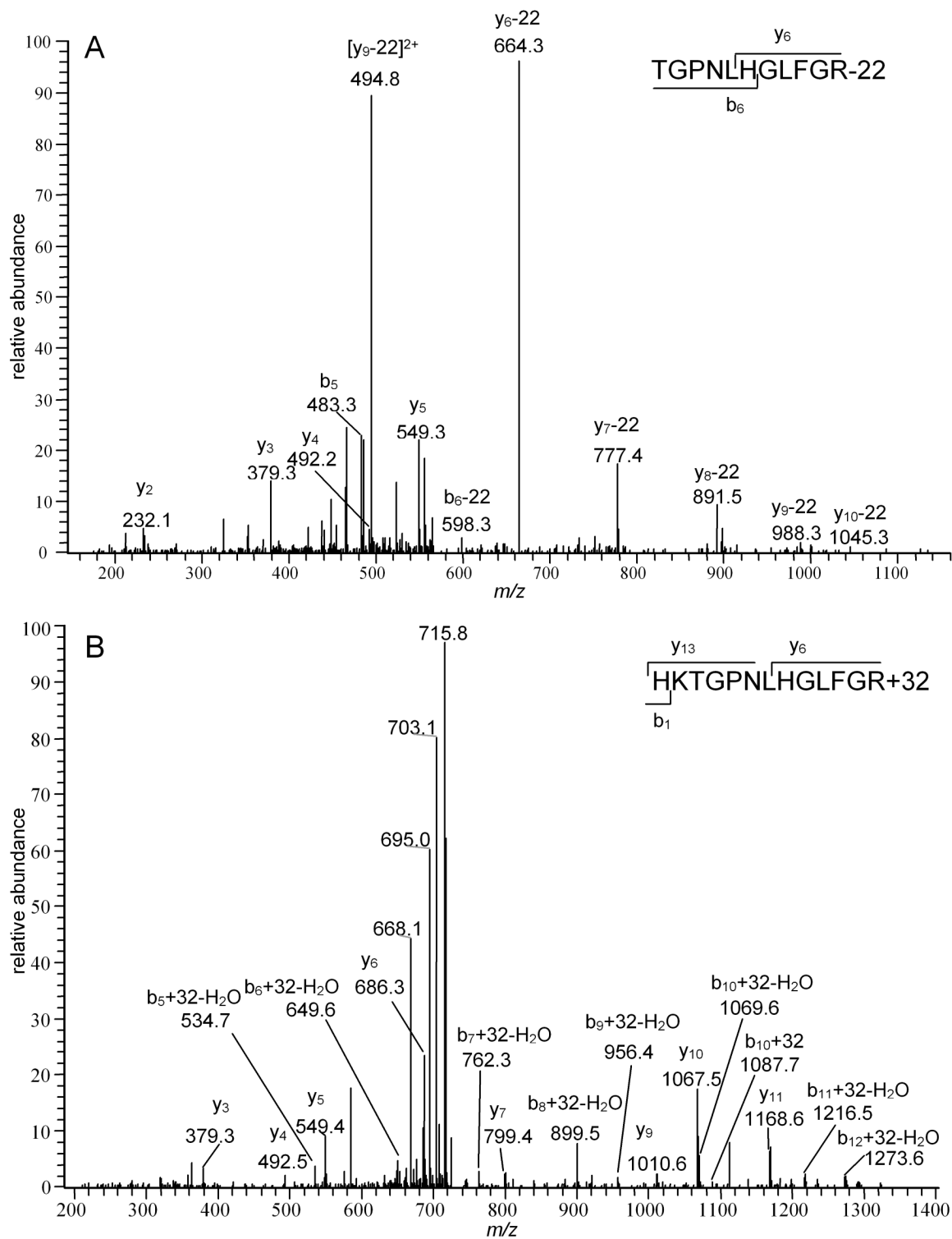


**Figure 5.** The crystal structure of horse heart Cyt-*c* (coordinates obtained from 2FRC in the Protein Data Bank [60]) represented as spheres drawn at 90% of the CPK radius with all atoms colored cyan except as noted. His28 and His33 which are modified by  $^1\text{O}_2$  are differentially colored with carbons gray, hydrogens white, and nitrogen blue. To emphasize the potential mobility around His26, Gly residues were colored lighter cyan. To permit identification of Lys residues, the  $\epsilon$ -amino nitrogens are dark blue. The view was created with RasTop [61].



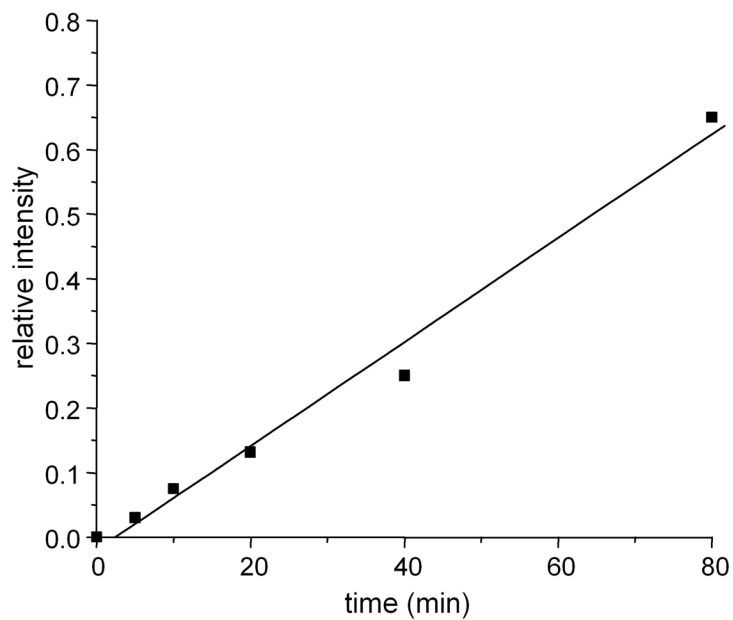
**Figure 6.** MALDI spectra of TGPNLHGLFGR peptide (lower) and the same peptide exposed to  $^1\text{O}_2$  generated by photoexcitation of Pc 4 (upper). The observed modifications, -22, +14, +16, +26, +32, +34 are indicated by the increment in mass of the monoisotopic peaks.





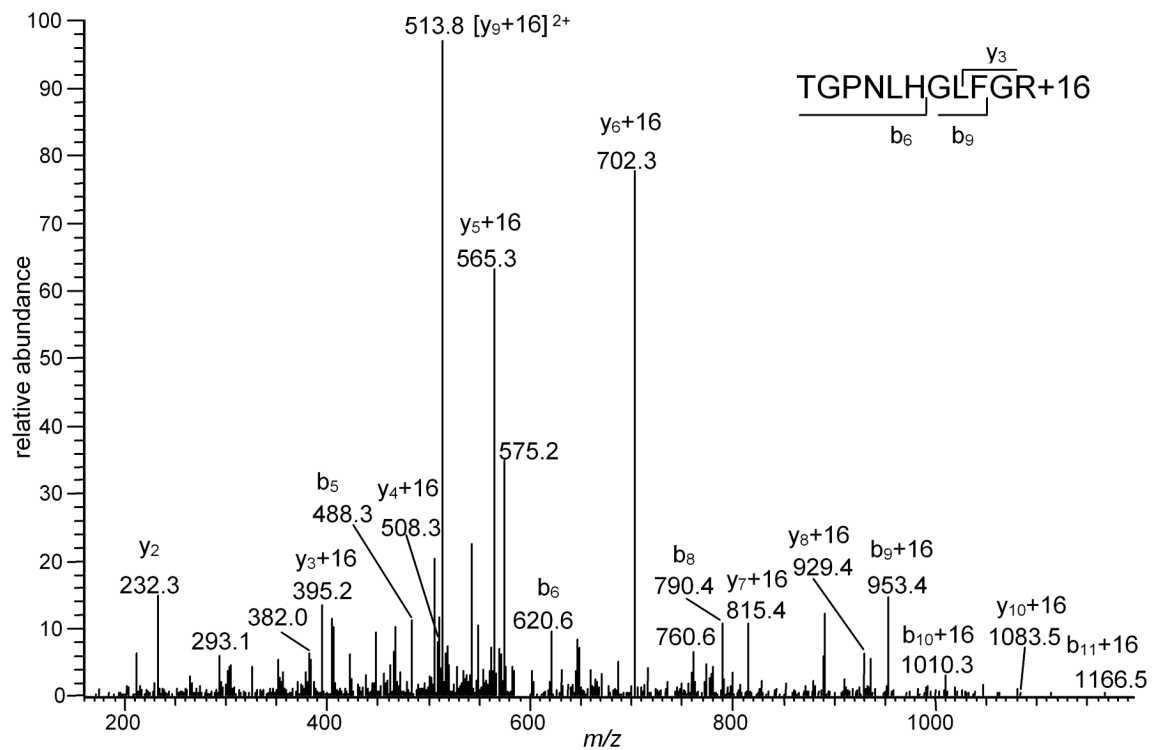
**Figure 7.** LC-MS<sup>2</sup> spectra of precursor ions A) whose  $m/z$  corresponds to the  $M_r$  of TGPNLH<sup>33</sup>GLFGR peptide-22 (574.1) and B) whose  $m/z$  corresponds to the  $M_r$  of H<sup>26</sup>KTGPNLHGLFGR peptide +32 (734.2). The insets show the sequences of the peptides and indicate the sequence number

of the first product ions in the b- and y-series that contain the modified A) His33 residue and B) His26 residue. The presence of B)  $b_5+32+H_2O$ ,  $b_6+32+H_2O$ ,  $b_7+32+H_2O$ ,  $y_6$ ,  $y_7$ ,  $y_9$ ,  $y_{10}$  and  $y_{11}$  ions confirms the absence of modifications of His33 in  $H^{26}KTGPNLHGLFGR$  peptide.



**Figure 8.**

Relative intensity of the  $H^{26}KTGPNLHGLFG+32$  peak to the unmodified  $TGPNHGLFG$  M3 peak as a function of irradiation time determined from MALDI-TOF spectra of digested Cyt-c irradiated for 0, 5, 10, 20, 40, and 80 min. The linear relationship indicates that the relative intensity of  $H^{26}KTGPNLHGLFG+32$  peak is proportional to the amount of  $^1O_2$  generated by photoexcitation.



**Figure 9.**

LC-MS<sup>2</sup> spectrum of a precursor ion whose  $m/z$  corresponds to the  $M_r$  of TGPNLHGLF<sup>36</sup>GR +16 (592.8). The inset shows the sequence of the peptide and indicates the sequence number of the first product ions in the b- and y-series that contain the modified Phe residue. The presence of  $b_6$ ,  $b_8$  and  $y_3$ - $y_5$  ions confirms the absence of modifications of His33.