

## Cationic Composition of 22 Species of Bacteria Grown in Seawater Medium

GALEN E. JONES,<sup>†</sup>\* LESLIE G. ROYLE,<sup>‡</sup> AND LINDSAY MURRAY<sup>§</sup>

*Department of Oceanography, The University of Liverpool, Liverpool, England*

Received for publication 3 August 1979

Twenty-two species of bacteria of marine, estuarine, and terrestrial origin were analyzed for cationic content by atomic absorption spectrophotometry after growth in a basal seawater medium. *Alcaligenes marinus* was analyzed from eight separate but replicate determinations yielding the following cationic concentrations: Na,  $5,600 \pm 2,260$ ; Mg,  $1,580 \pm 740$ ; K,  $700 \pm 360$ ; Ca,  $790 \pm 390$ ; Mn,  $1.7 \pm 0.5$ ; Fe,  $256 \pm 57$ ; Ni,  $1.7 \pm 0.7$ ; Cu,  $14 \pm 4$ ; Zn,  $122 \pm 27$ ; Cd,  $2.8 \pm 0.7$ ; and Pb,  $10 \pm 3$  ppm/(dry weight). Washing *A. marinus* cells before analyses was necessary due to interstitial medium within the cell pellets after centrifugation and loose cationic retention by the cells. The principal source of error in the procedure was ascribed to variability due to washing cells with 0.5 M ammonium formate. The mean cationic concentrations for trace elements in the 22 bacterial cultures grown in the basal seawater medium to constant optical density and washed three times with 0.5 M ammonium formate were: Mn,  $2.4 \pm 3.8$ ; Fe,  $262 \pm 112$ ; Ni,  $2.3 \pm 1.8$ ; Cu,  $24 \pm 17$ ; Zn,  $146 \pm 72$ ; Cd,  $3.8 \pm 2.5$ ; and Pb,  $17 \pm 21$  ppm (dry weight). Major ions were concentrated only occasionally by the cells after washing, whereas Mn, Fe, Ni, Cu, Zn, Cd, and Pb were concentrated from the medium by the following factors on the average: 180, 1,600, 140, 1,200, 750, 1,900, and 900, respectively.

The elemental composition of bacteria has not been analyzed extensively. Although increasing interest has developed in the role of mineral elements in the physiology of microbial cells (7, 17, 37, 43) and their transformations in nature (12, 15, 33, 35), data on the bioaccumulation of mineral cations by bacteria have remained sparse (2, 18, 20, 27). Washing procedures have been variable, if practiced at all. The conditions of growth such as pH (24, 32), elemental composition of the medium (7, 37), growth rate (32, 37), and the source of the electron donor in photosynthetic organisms (40) have influenced the elemental composition of bacteria.

The bioaccumulation of cations by marine bacteria has not been studied (14). Marine plants and animals reportedly concentrate trace elements from seawater by several orders of magnitude (8, 18, 23, 25, 26, 29, 30, 34, 39, 41, 42). Undersaturation of many trace elements in seawater has been attributed partially to assimilation by organisms (9).

<sup>†</sup> Present address: Department of Microbiology and The Jackson Estuarine Laboratory, University of New Hampshire, Durham, NH 03824.

<sup>‡</sup> Present address: Severn-Trent Water Authority, Trent River Management Division, Gainsborough, Lincolnshire, England.

<sup>§</sup> Present address: Ministry of Agriculture and Fisheries, Burnham-on-Crouch, Essex, England.

Since a sufficient biomass of specific marine bacteria free of contaminating particles cannot be obtained directly from natural seawater, 22 pure cultures of bacteria were grown in a basal seawater medium for cationic analyses. The elemental composition of bacteria from a single medium should provide data for analytical and generic variability.

### MATERIALS AND METHODS

**Microorganisms.** Twenty-two cultures of bacteria were distributed into three groups: (i) five previously identified marine bacteria, namely, *Alcaligenes marinus* (formerly *Arthrobacter marinus* ATCC 25374) (J. B. Rake and G. E. Jones, Abstr. Annu. Meet. Am. Soc. Microbiol. 1977, 133, p. 160; 3), *Alteromonas haloplanktis* (B16 strain of R. A. MacLeod; ATCC 19855 and NCMB 19) (22, 28), *Pseudomonas putida* Roche (formerly *P. cuprodurans* ATCC 29735) (F. J. Passman, Ph.D. thesis, University of New Hampshire, Durham, 1977; 14), *Pseudomonas* sp. (NCMB 130), and *Vibrio fischeri* (ATCC 15382 and NCMB 1143), the facultative psychrophilic strain (4, 5, 10); (ii) 12 different, unidentified pure cultures of marine bacteria freshly isolated from the Great Bay estuarine complex, Durham, N.H.; and (iii) five identified pure cultures of nonmarine bacteria, namely, *Bacillus megaterium* (laboratory strain), *Escherichia coli* (NCLB 8114), *Pseudomonas fluorescens* (NCLB 8858), *Pseudomonas ovalis* (NCLB 9229), and *Staphylococcus aureus* (NCLB 6571).

**Culture conditions.** The basal seawater medium contained 1 g of yeast extract (Difco Laboratories, Detroit, Mich.); 1 g of tryptone (Oxoid Ltd., London); and 1,000 ml of 75% synthetic seawater (14). The final pH of the medium was 7.5 to 7.8. The calculated cationic composition of the basal medium in parts per million was: Na, 7,459.255; Mg, 995.426; K, 297.863; Ca, 300.778; Mn, 0.013; Fe, 0.166; Ni, 0.016; Cu, 0.020; Zn 0.196; Cd 0.002; and Pb 0.019 (14).

The inoculum, growth conditions, and harvesting and drying of cells have been described (14).

**Cell analyses.** Ashing and flame and atomic spectrophotometric techniques have been described (14).

## RESULTS

The reproducibility of the elemental analyses for *Alcaligenes marinus* grown under identical conditions on eight separate occasions was examined initially. Cells were grown in 500 ml of basal seawater medium contained in 1-liter Erlenmeyer flasks inoculated with 0.2 ml of inoculum (optical density at 420 nm [OD<sub>420</sub>] = 0.95 to 1.05) (14). The cells were harvested from three pooled flasks after 18 to 24 h (OD<sub>420</sub> = 0.96 to 1.03). Weights of washed cells were greater than 100 mg (dry weight) in all cases.

The coefficient of variation in single determinations of the major cation (Na, Mg, K, Ca) concentrations was high, 40 to 51% (Table 1). This was due to variability in washing techniques. The coefficient of variation in single determinations of the minor cations was lower, varying from 22% for Zn and Fe to 41% for Ni. The trace elements were retained more firmly than the major cations by the cells of *A. marinus*. Only those differences which lay outside the coefficient of variation of a single result as

determined by this experiment were considered significant in subsequent experiments.

To determine the effect of different washing solutions on elemental composition, *A. marinus* was grown in basal seawater medium at 21 ± 1°C for 18 h in 1,400-ml replicate amounts to the same cell density. The highest dry weight was obtained from unwashed cells, 780.7 mg (Table 2). Since some of this dry material was interstitial medium constituents and adsorbed substances, washing was essential. Three washes with distilled water reduced the dry weight to 332.6 mg, a reduction of 57%. One wash of 0.5 M ammonium formate reduced the dry weight by 26%, and three washes reduced it by 48%. Three washes of 0.01 M phosphate buffer reduced the dry weight by 39%, and three washes of phosphate buffer plus 10<sup>-4</sup> M ethylenediaminetetraacetic acid reduced it by 31%. Thus, washing was necessary to remove noncellular material. Some lysis may have occurred during the washing procedures.

The apparent removal of major ions by the washes was considerable when compared with the basal medium composition previously cited (Table 2). Thus, the unwashed cellular pellets retained appreciable amounts of these major ions by the negatively charged cells. Na was removed by all washing solutions to amounts less than in the medium. Mg was removed similarly by the washes except for distilled water and one wash of ammonium formate (Tables 1 and 2). K was only reduced drastically in the three washes of ammonium formate. Ca was retained by the cells or increased in most instances. Variability of results for the major ions was attributed to the type of washing solution,

TABLE 1. Cation composition of *A. marinus* grown in basal seawater medium at 20 ± 3°C to late logarithmic grown in eight separate experiments

Expt	Cellular cation concn (ppm, dry wt)										
	Na	Mg	K	Ca	Mn	Fe	Ni	Cu	Zn	Cd	Pb
1	4,600	1,850	600	950	1.5	328	2.7	16	84	3.9	13
2	2,000	1,390	300	530	2.2	227	1.7	22	119	2.6	12
3	8,000	1,890	1,100	820	2.2	292	1.5	14	131	3.4	14
4	4,700	1,090	200	230	1.1	176	0.8	14	94	1.9	7
5	3,700	700	300	370	1.3	257	1.2	15	153	1.9	11
6	7,600	2,800	900	1,350	2.1	272	2.0	9	146	3.3	5
7	8,300	2,180	900	1,170	2.0	313	2.8	11	150	2.2	10
8	6,200	700	1,000	900	1.4	184	1.0	14	97	3.0	7
Mean	5,600	1,580	700	790	1.7	256	1.7	14	122	2.8	10
Standard deviation <sup>a</sup>	2,260	740	360	390	0.5	57	0.7	4	27	0.7	3
Coefficient of variation <sup>b</sup>	40	47	51	49	29	22	41	29	22	25	30
Standard error <sup>c</sup>	800	260	130	140	0.2	20	0.3	1	10	0.3	1

$$^a \pm \sqrt{\frac{\sum y^2}{n-1}}$$

<sup>b</sup> (Standard deviation × 100)/mean.

<sup>c</sup> Standard deviation/√n.

TABLE 2. Cation composition of *A. marinus* grown in basal seawater medium at  $21 \pm 1^\circ\text{C}$  for 18 h, using different washing solutions

Washing treatment	Dry wt (mg)	% of wt re-tained during washing	Cellular cation concn (ppm, dry wt)										
			Na	Mg	K	Ca	Mn	Fe	Ni	Cu	Zn	Cd	Pb
None	780.7	100	30,330	3,200	13,800	3,880	2.1	189	6.1	11	63	2.3	13
3 × 200 ml of distilled water	332.6	43	3,100	3,550	3,000	3,090	1.2	280	3.9	24	89	2.4	17
1 × 200 ml of 0.5 M ammonium formate	580.4	74	4,700	1,260	570	910	1.4	238	3.8	9	86	0.9	10
3 × 200 ml of 0.5 M ammonium formate	410.3	52	2,000	390	80	530	2.2	296	5.3	22	119	2.6	27
3 × 200 ml of 0.01 M phosphate buffer	477.3	61	4,000	510	4,350	620	1.7	249	5.2	39	82	3.3	45
3 × 200 ml of 0.01 M phosphate buffer × $10^{-4}$ M EDTA <sup>a</sup>	534.9	69	4,100	500	6,210	260	1.5	217	4.7	18	60	3.2	18

<sup>a</sup> EDTA, Ethylenediaminetetraacetic acid.

inconsistencies during mechanical stirring of cells when washing, and maintenance of constant cold temperature.

Among the trace elements, firm retention by the cells during the various washing procedures was indicated. Mn, Fe, Ni, Cu, Zn, Cd, and Pb were retained by *A. marinus* cells within a half order of magnitude for each element regardless of washing solution. Where cells were only washed once with ammonium formate, the lower amount of the trace cations relative to three washings was ascribed to incomplete removal of other elements, resulting in a larger dry weight for the singly washed cells. The washing procedure was considered the greatest single source of variability in the elemental analyses of bacterial cells.

The cations retained by the identified marine and unidentified estuarine bacteria grown in basal seawater medium are presented in Table 3. Excluding widely anomalous amounts of Na, Mg, K, and Ca in a few samples, the concentration factors for these cations were 0.04, 0.6, 0.35, and 0.8, respectively, indicating less retention by the cells than the medium after three washes with chilled ammonium formate. The anomalous values were invariably higher than the mean; their inclusion in the calculations would tend to bring the mean closer to unity for major cations in cells and medium.

Other than the anomalous value of 16 ppm (dry weight) Mn in UNH 4, the mean concentration in marine bacteria was 0.9 ppm (dry weight). Mn concentration by the 15 other cultures indicated a concentration almost 70-fold higher than in the basal seawater medium.

Fe was concentrated on the average in the

marine cells to 288 ppm (dry weight), representing a concentration 1,700-fold higher than in basal seawater medium. Ni was concentrated on the average in the cells to 2.5 ppm (dry weight), a concentration of 160-fold higher than in the basal seawater medium. Cu, Zn, Cd, and Pb content in the cells averaged 28, 157, 4.3, and 16 ppm (dry weight), representing average concentrations 1,400-, 800-, 2,150-, and 840-fold higher than in the basal seawater medium. Thus, Mn and Ni were concentrated by about two orders of magnitude by the marine bacterial cells from the medium, and Fe, Cu, Zn, Cd, and Pb were concentrated by three orders of magnitude. The coefficient of variation ranged from 36% for Fe to 81% for Ca, excluding several indicated anomalous amounts (Table 3).

Cationic concentrations of five identified nonmarine bacterial cultures grown in basal seawater medium indicated wide coefficients of variation for the major cations (Table 4). Since only five cultures were analyzed, anomalous amounts were more difficult to detect. *E. coli* retained the highest concentrations of Na, Mg, and K and the lowest concentration of Ca. Major seawater cations remained in the nonmarine bacterial cells to a greater extent on average than in the marine bacteria, but with larger coefficients of variation.

Among the major cations in the dry weight of the nonmarine bacteria, Fe (178 ppm), Cu (18 ppm), Zn (113 ppm), and Cd (2.3 ppm) were reduced compared with marine bacteria. Mn (4.2 ppm) and Pb (22 ppm) indicated some tendency to be concentrated in the nonmarine bacteria. Within standard deviation, there was similarity of the cationic concentration in bacterial cells

TABLE 3. Cation composition of 16 marine bacteria grown in basal seawater medium at 22 ± 3°C to late logarithmic growth

Bacterium	Dry wt (mg)	Cellular cation concn (ppm, dry wt)										
		Na	Mg	K	Ca	Mn	Fe	Ni	Cu	Zn	Cd	Pb
Identified												
<i>Alteromonas haloplanktis</i> (NCMB 19)	281	299	420	36	267	0.4	262	1.5	22	159	3.9	3
<i>Pseudomonas putida</i> (ATCC 29735)	219	255	292	36	272	0.9	422	4.5	19	112	4.6	3
<i>Pseudomonas</i> sp. (NCMB 130)	147	327	156	55	286	0.5	365	1.5	49	180	8.8	2
<i>Vibrio fischeri</i> (NCMB 1143)	135	357	359	89	165	1.5	351	1.5	29	101	10.4	2
Unidentified												
UNH 1	641	11	440	47	80	2.0	343	3.4	24	168	2.0	15
UNH 3	470	192	536	85	216	0.4	290	4.7	33	206	4.3	15
UNH 4	287	(25,800) <sup>a</sup>	(8,939)	(4,184)	— <sup>b</sup>	(16)	137	1.5	19	44	4.9	3
UNH 5	172	(2,565)	705	(1,166)	186	0.3	314	3.5	50	227	6.4	45
UNH 7	105	531	698	123	266	1.0	462	8.5	84	308	6.6	96
UNH 11	739	433	391	202	70	0.3	132	0.7	10	57	1.4	1
UNH 14	70	315	(4,822)	143	830	1.4	436	2.9	24	302	1.7	11
UNH 15	228	281	737	114	150	2.2	205	0.9	21	153	1.8	4
UNH 16	86	232	672	46	104	1.2	243	1.7	16	118	3.5	27
UNH 23	207	290	663	87	136	0.4	289	1.0	17	161	1.7	11
UNH 25	316	392	1,362	57	596	0.5	147	1.5	8	53	2.6	4
UNH 26	613	227	463	326	160	0.7	222	0.8	22	169	3.4	20
Mean		296	564	103	252	0.9	288	2.5	28	157	4.3	16
Standard deviation		121	289	79	204	0.6	104	2.0	19	78	2.6	24
Coefficient of variation		40	51	76	81	66	36	80	67	50	60	66
Concn factor		0.04	0.6	0.35	0.8	69	1,700	160	1,400	800	2,150	840

<sup>a</sup> Numbers in parentheses are anomalous values not calculated.

<sup>b</sup> —, Not determined.

TABLE 4. Cation composition of five nonmarine bacteria grown in basal seawater medium at 20 ± 2°C to late logarithmic growth

Bacterium	Dry wt (mg)	Cellular cation concn (ppm, dry wt)										
		Na	Mg	K	Ca	Mn	Fe	Ni	Cu	Zn	Cd	Pb
<i>Bacillus megaterium</i> (laboratory strain)	143	298	1,178	149	1,655	10.5	339	2.4	36	186	1.0	13
<i>Escherichia coli</i> (NCB 8114)	335	3,341	1,351	1,789	42	2.7	111	0.6	11	76	3.1	11
<i>Pseudomonas fluorescens</i> (NCB 8858)	454	365	544	99	48	3.1	99	2.6	19	93	2.9	39
<i>P. ovalis</i> (NCB 9229)	379	221	660	26	834	0.5	256	2.8	12	108	0.4	18
<i>Staphylococcus aureus</i> (NCB 6571)	482	66	776	207	112	4.4	86	0.6	14	104	4.3	30
Mean		858	902	454	538	4.2	178	1.8	18	113	2.3	22
Standard deviation		1,392	347	749	708	3.8	113	1.1	10	42	1.6	12
Coefficient of variation		162	38	165	130	90	64	61	57	38	70	54
Concn factor		0.1	0.9	1.5	1.7	320	1,100	110	400	600	1,150	1,200

grown in the same medium under similar cultural conditions.

DISCUSSION

Chilled ammonium formate (0.5 M) was used in this study to wash the bacterial cell pellets since it was used by Riley and Roth (29) for washing phytoplankton, providing osmolarity and volatilizing during drying at 105 to 110°C.

Although the necessity for washing *A. marinus* cells was established to remove loosely bound cations (Table 2), the efficacy of ammonium formate treatment must be questioned. The major cations of seawater were reduced drastically from the cells during this procedure. Other investigators (1, 6, 7, 16, 17, 21, 31, 32) using a variety of procedures indicated higher retention of these major ions by bacterial cells. Kung et al. (17) reported that *E. coli* B/r grown expo-

nentially contained 2,800 ppm of Mg, 12,000 ppm of K, 2,300 ppm of Ca, and 150 ppm of Zn when washed with 10 ml of warm 0.3 M glucose solution. Jasper and Silver (11) reported that total cellular Mg was generally in the range of 360 to 840 ppm (wet cells), although the growth medium may have varied 100,000-fold. Thus, our values for major ions in the ammonium formate-washed cells on a dry weight basis are low, indicating removal during the washing procedure. Table 2, subsequent work in this laboratory (Passman, Ph.D. thesis, 1977), and studies by others (6, 31) have indicated distilled water as a more effective wash which still reduces the concentrations of major cations in the cells.

Comparing our data for major cations with those of Takacs et al. (36) for *Alteromonas haloplanktis* indicates that only bound Na and K remain after washing with ammonium formate. Distilled water-washed cells of *B. megaterium* grown in a chemostat also generally indicated greater retention of major cations than those in Table 4 (32). An increase in the rate of growth resulted in increased content of Ca, Mg, and Fe, with no change in K (32). Although the present investigation indicates little to no concentration of major ions by the cells from the basal seawater medium due to removal by the ammonium formate wash, concentrations of an order of magnitude are not uncommon (37).

Minor ions were retained quite comparably to other studies (6, 31, 32) due to their strong retention by cellular components during washing. For Mn, marine bacteria averaged about 1 ppm and nonmarine bacteria averaged 4 ppm (Tables 3 and 4). These Mn concentrations are lower than those of Rouf (31) (20 ppm for *E. coli*) and Curran et al. (6) (40 ppm for *B. megaterium* vegetative cells). Mn increased from 40 to 250 ppm in cells of *Chromatium vinosum* grown photoheterotrophically versus photoautotrophically, suggesting its relative importance in autotrophic evolution (40). Freshwater cyanobacteria averaged almost 40-fold the Mn of marine heterotrophic bacteria (13). Bacteria from 18 strains of microorganisms isolated from the Mn biogeochemical province of the Georgian U.S.S.R. were reported to contain from 40 to 100,000 ppm of Mn (19).

*C. vinosum* grown photoautotrophically contained 210 ppm of Fe, whereas photoheterotrophically grown cells contained 590 ppm of Fe (40). Five freshwater cyanobacteria grown with light energy averaged a greater than 10-fold increase in Fe compared with heterotrophically grown bacteria (13). The Fe concentration in vegetative bacteria was quite consistent: 30 to 560 ppm (6); 140 to 280 ppm, excluding *Sphaerotilus natans* (31); 20 to 3,000 ppm in *B. mega-*

*terium*, depending on pH of the medium and rate of growth (32); and 86 to 462 ppm (Tables 3 and 4).

Ni concentration by bacterial cells was about two orders of magnitude higher in bacterial cells than in the medium (Tables 3 and 4). Only slightly higher concentrations of Ni in bacterial cells were reported by Rouf (31).  $^{63}\text{Ni}$  appears to be transported by the system responsible for Mg accumulation by *E. coli* (11).

Cu uptake by *B. megaterium* was decreased from 22 to 3 ppm at pH 7 when the growth rate was increased in a chemostat from  $D = \mu = 0.2 \text{ h}^{-1}$  to  $D = \mu = 0.7 \text{ h}^{-1}$  (32). Cu concentration of the dry weight of vegetative bacterial cells by spectrographic analyses ranged only slightly higher than the concentrations in Tables 3 and 4 (6, 31).

The Zn content of bacterial cells ranged from 44 to 308 ppm (dry weight) (Tables 3 and 4). The mean concentrations in this study were slightly lower, about 100 ppm, than those of Rouf (31). Zn was shown to increase in the cells of *Alcaligenes marinus* as the concentration of Zn in the medium was elevated (14).

Although the Pb concentrations of bacterial cells reported here were only slightly higher than those of Rouf (31), *Micrococcus luteus* and *Azotobacter* sp. cells grown in a broth in contact with a dialysis membrane containing  $\text{PbBr}_2$  immobilized 490,000 and 310,000 ppm Pb on a dry weight basis, respectively (38).

The elemental composition of bacterial cells still requires considerable investigation. The data presented indicate remarkably similar cation concentrations from 22 different bacterial species grown under the same conditions, suggesting a reasonable unity for their elemental composition.

#### ACKNOWLEDGMENTS

We thank J. P. Riley, Department of Oceanography, University of Liverpool, Liverpool, England, for helpful advice and encouragement during this investigation. Appreciation is extended to J. M. Shewan, Torrey Research Station, Aberdeen, Scotland, for the previously identified marine bacterial cultures and to N. G. Carr, Department of Biochemistry, University of Liverpool, for providing the identified nonmarine bacterial cultures. The unidentified marine bacterial cultures were isolated by J. B. Rake and B. W. Rake, University of New Hampshire, Durham.

This work was supported by National Science Foundation grants GA-34101 and DES 75-04790 from the Oceanography Section.

#### LITERATURE CITED

1. Beveridge, T. J., and R. G. E. Murray. 1976. Uptake and retention of metals by cell walls of *Bacillus subtilis*. *J. Bacteriol.* **127**:1502-1518.
2. Bowen, H. J. M. 1966. Trace elements in biochemistry. Academic Press Inc., New York.
3. Cobet, A. B., C. Wirsén, and G. E. Jones. 1970. The effect of nickel on a marine bacterium, *Arthrobacter marinus* sp. nov. *J. Gen. Microbiol.* **62**:159-169.

4. Colwell, R. R., and M. Mandel. 1964. Base composition of deoxyribonucleic acid of marine and nonmarine vibrios deduced from buoyant-density measurements in cesium chloride. *J. Bacteriol.* **88**:1816-1817.
5. Colwell, R. R., and R. Y. Morita. 1964. Reisolation and emendation of description of *Vibrio marinus* (Russell) Ford. *J. Bacteriol.* **88**:831-837.
6. Curran, H. R., B. C. Brunstetter, and A. T. Meyers. 1943. Spectrochemical analysis of vegetative cells and spores of bacteria. *J. Bacteriol.* **45**:485-494.
7. Epstein, W., and S. G. Schultz. 1965. Cation transport in *Escherichia coli*. V. Regulation of cation content. *J. Gen. Physiol.* **49**:221-234.
8. Fukai, R., and W. W. Meinke. 1959. Trace analysis of marine organisms: a comparison of activation analysis and conventional methods. *Limnol. Oceanogr.* **4**:398-408.
9. Goldberg, E. D. 1965. Minor elements in sea water, p. 163-196. *In* J. P. Riley and G. Skirrow (ed.), *Chemical oceanography*, vol. 1. Academic Press Inc., New York.
10. Hendrie, M. S., W. Hodgkiss, and J. M. Shewan. 1971. A proposal that *Vibrio marinus* (Russell 1891) Ford 1927 be amalgamated with *Vibrio fischeri* (Beijerinck 1889) Lehmann and Neumann 1896. *Int. J. Syst. Bacteriol.* **21**:217-221.
11. Jasper, P., and S. Silver. 1977. Magnesium transport in microorganisms, p. 7-48. *In* E. D. Weinberg (ed.), *Microorganisms and minerals*. Marcel Dekker, Inc., New York.
12. Jernelöv, A., and A.-L. Martin. 1975. Ecological implications of metal metabolism by microorganisms. *Annu. Rev. Microbiol.* **29**:61-77.
13. Jones, G. E., L. Murray, and N. G. Carr. 1978. Trace element composition of five cyanobacteria, p. 967-973. *In* W. E. Krumbein (ed.), *Environmental biogeochemistry and geomicrobiology*, vol. 3. Ann Arbor Science Publishers, Ann Arbor, Mich.
14. Jones, G. E., L. G. Royle, and L. Murray. 1976. The assimilation of trace metal ions by the marine bacteria, *Arthrobacter marinus* and *Pseudomonas cuprodurans*, p. 889-898. *In* J. M. Sharpley and A. M. Kaplan (ed.), *Proceedings of the Third International Biodegradation Symposium*. Applied Science Publishers, London.
15. Konetzka, W. A. 1977. Microbiology of metal transformations, p. 317-342. *In* E. D. Weinberg (ed.), *Microorganisms and minerals*. Marcel Dekker, Inc., New York.
16. Koval'skii, V. V., T. F. Borovik-Romanova, S. V. Letunova, and E. O. Ginzburg. 1965. Data on the trace-element content of micro-organisms. *Microbiology* **34**:340-342.
17. Kung, F.-C., J. Raymond, and D. A. Glaser. 1976. Metal ion content of *Escherichia coli* versus cell age. *J. Bacteriol.* **126**:1089-1095.
18. Leland, H. V., S. N. Luoma, and J. F. Elder. 1978. Heavy metals and related trace elements. *J. Water Pollut. Control Fed.* **50**:1469-1514.
19. Letunova, S. V., M. V. Ulubekova, and V. I. Shcherbakov. 1978. Manganese concentration by microorganisms inhabiting soils of the manganese biogeochemical province of the Georgian SSR. *Microbiology* **47**:273-278.
20. Luria, S. E. 1960. The bacterial protoplasm: composition and organization, p. 1-34. *In* J. C. Gunsalus and R. Y. Stanier (ed.), *The bacteria*, vol. 1. Academic Press Inc., New York.
21. Lusk, J. E., R. J. P. Williams, and E. P. Kennedy. 1968. Magnesium and the growth of *Escherichia coli*. *J. Biol. Chem.* **243**:2618-2624.
22. MacLeod, R. A., E. Onofrey, and M. E. Norris. 1954. Nutrition and metabolism of marine bacteria. I. Survey of nutritional requirements. *J. Bacteriol.* **68**:680-686.
23. Martin, J. H., and G. A. Knauer. 1973. The elemental composition of plankton. *Geochim. Cosmochim. Acta* **37**:1639-1653.
24. Neyland, M., P. Dunkel, and A. L. Schade. 1952. The uptake of cobalt by *Proteus vulgaris*. *J. Gen. Microbiol.* **7**:409-416.
25. Nicholls, G. D., H. Curl, Jr., and V. T. Bowen. 1959. Spectrographic analyses of marine plankton. *Limnol. Oceanogr.* **4**:472-478.
26. Noddack, I., and W. Noddack. 1940. Die Häufigkeiten der Schwermetalle in Meerestieren. *Ark. Zool.* **32A**:1-52.
27. Perlman, D. 1965. Microbial production of metal-organic compounds and complexes, p. 103-138. *In* W. W. Umbreit (ed.), *Advances in applied microbiology*, vol. 7. Academic Press Inc., New York.
28. Reichelt, J. L., and P. Baumann. 1973. Change of name *Alteromonas marinopraesens* (ZoBell and Upham) Baumann et al. to *Alteromonas haloplanktis* (ZoBell and Upham) comb. nov. and assignment of strain ATCC 23821 (*Pseudomonas enalia*) and strain c-A1 of DeVoe and Oginsky to this species. *Int. J. Syst. Bacteriol.* **23**:438-441.
29. Riley, J. P., and I. Roth. 1971. The distribution of trace elements in some species of phytoplankton grown in culture. *J. Mar. Biol. Assoc. U.K.* **51**:63-72.
30. Riley, J. P., and D. A. Segar. 1970. The distribution of the major and some minor elements in marine animals. I. Echinoderms and Coelenterates. *J. Mar. Biol. Assoc. U.K.* **50**:721-730.
31. Rouf, M. A. 1964. Spectrochemical analysis of inorganic elements in bacteria. *J. Bacteriol.* **88**:1545-1549.
32. Sakharova, Z. V., L. F. Shaforostova, and G. A. Kochetkov. 1978. Content of metals in cells of *Bacillus megaterium* under various conditions of culturing. *Microbiology* **47**:46-50.
33. Saxena, J., and P. H. Howard. 1977. Environmental transformations of alkylated and inorganic forms of certain metals. *Adv. Appl. Microbiol.* **21**:185-226.
34. Segar, D. A., J. D. Collins, and J. P. Riley. 1971. The distribution of the major and some minor elements in marine animals. II. Molluscs. *J. Mar. Biol. Assoc. U.K.* **51**:131-136.
35. Summers, A. O., and S. Silver. 1978. Microbial transformations of metals. *Annu. Rev. Microbiol.* **32**:637-672.
36. Takacs, F. P., T. I. Matula, and R. A. MacLeod. 1964. Nutrition and metabolism of marine bacteria. XIII. Intracellular Na<sup>+</sup> and K<sup>+</sup> concentrations in a marine pseudomonad. *J. Bacteriol.* **87**:510-518.
37. Tempest, D. W. 1969. Quantitative relationships between inorganic cations and anionic polymers in growing bacteria, p. 87-111. *In* P. M. Meadow and S. J. Pirt (ed.), *Microbial growth*, 19th Symposium of the Society for General Microbiology. Cambridge University Press, Cambridge, England.
38. Tornabene, T. G., and H. W. Edwards. 1972. Microbial uptake of lead. *Science* **176**:1334-1335.
39. Turekian, K. K., A. Katz, and L. Chan. 1973. Trace element trapping in pteropod tests. *Limnol. Oceanogr.* **18**:240-249.
40. Udel'nova, T. M., V. I. Chudina, L. K. Osnitskaya, E. A. Boichenko, S. M. Chernogorova, and A. V. Karyakin. 1977. Content of polyvalent metals in the presence of a change in metabolism of *Chromatium vinosum*. *Microbiology* **46**:333-337.
41. Vinogradov, A. P. 1953. The elementary composition of marine organisms. *In* Sears Found. Mar. Res., Mem. II. Yale University, New Haven, Conn. (Translated from Russian.)
42. Webb, D. A. 1937. Studies on the ultimate composition of biological materials. Part II. Spectrographic analysis of marine invertebrates with special reference to the chemical composition of their environment. *Sci. Proc. R. Dubl. Soc.* **21**:505-539.
43. Weinberg, E. D. (ed.). 1977. *Microorganisms and minerals*. Marcel Dekker, Inc., New York.