

NIH Public Access

Author Manuscript

Bioorg Med Chem. Author manuscript; available in PMC 2009 May 1.

Published in final edited form as:

Bioorg Med Chem. 2008 May 1; 16(9): 5090–5102. doi:10.1016/j.bmc.2008.03.028.

*S***-Ribosylhomocysteine analogues with the carbon-5 and sulfur atoms replaced by a vinyl or (fluoro)vinyl unit**

Stanislaw F. Wnuka,* , **Jennifer Lalama**a,†, **Craig A. Garmendia**a, **Jenay Robert**a, **Jinge Zhu**b, and **Dehua Pei**b

a*Department of Chemistry & Biochemistry, Florida International University, Miami, FL 33199, USA*

b*Department of Chemistry and Ohio State Biochemistry program, The Ohio State University, Columbus, Ohio 43210, USA*

Abstract

Treatment of the protected ribose or xylose 5-aldehyde with sulfonyl-stabilized fluorophosphonate gave (fluoro)vinyl sulfones. Stannyldesulfonylation followed by iododestannylation afforded 5,6 dideoxy-6-fluoro-6-iodo-_{p-}ribo or *xylo*-hex-5-enofuranoses. Coupling of the hexenofuranoses with alkylzinc bromides gave ten-carbon ribosyl- and xylosylhomocysteine analogues incorporating a fluoroalkene. The fluoroalkenyl and alkenyl analogues were evaluated for inhibition of *Bacillus* subtilis S-ribosylhomocysteinase (LuxS). One of the compounds, 3,5,6-trideoxy-6-fluoro- p -erythrohex-5-enofuranose, acted as a competitive inhibitor of moderate potency $(K_I = 96 \mu M)$.

Keywords

LuxS enzyme; Negishi coupling; *S*-ribosylhomocysteine; vinyl fluorides; vinyl stannanes

1. Introduction

S-Adenosyl-L-homocysteine (SAH) is a byproduct of many methyltransferase reactions and a potent inhibitor of the methyltransferases. In eukaryotes and some bacteria, detoxification of SAH is mediated by SAH hydrolase (EC 3.3.1.1), which effects hydrolytic cleavage of SAH to L-homocysteine (Hcy) and adenosine (Figure 1).¹ Hcy appears to be a risk factor for coronary artery diseases.2 Alternatively, most bacteria utilize enzyme 5′-methylthioadenosine (MTA)/SAH nucleosidase (EC 3.2.2.9) to irreversibly cleave SAH yielding adenine and *S*ribosyl-L-homocysteine (SRH) .³ The SRH is then converted to Hcy and 2,4-dihydroxy-2,3pentadione (DPD) by a metalloenzyme *S*-ribosylhomocysteinase (LuxS).4 DPD5 spontaneously cyclizes and complexes with borate to form a furanosyl borate diester, which acts as a type 2 autoinducer for bacterial interspecies quorum sensing.⁶ Since quorum sensing regulates many bacterial behaviors such as virulence and biofilm formation, LuxS and other proteins involved in quorum sensing have been proposed as attractive targets for novel antibacterial drug design.7 Several substrate analogues of SRH (e.g., **1** and **2**) showed submicromolar inhibition of $LuxS$ ^{4e,h}

^{*}Corresponding author. Tel.: +1 305 348 6195; fax: +1 305 348 3772; e-mail: wnuk@fiu.edu. †Present address. Azopharma, Inc. Miramar, FL

Publisher's Disclaimer: This is a PDF file of an unedited manuscript that has been accepted for publication. As a service to our customers we are providing this early version of the manuscript. The manuscript will undergo copyediting, typesetting, and review of the resulting proof before it is published in its final citable form. Please note that during the production process errors may be discovered which could affect the content, and all legal disclaimers that apply to the journal pertain.

We have previously observed that SAH hydrolase is capable of the addition of water across 5',6' isolated double bond of adenosine analogues 3 and 4 (Figure 2).^{1c,8} The resulting adduct (or its derivative) caused covalent modification and inactivation $8b$ of the enzyme, a process which required the catalytic activity of the enzyme. Since LuxS catalyzes a similar reaction as SAH hydrolase (i.e., overall elimination of Hcy), we designed analogues of SRH with the vinyl or halovinyl moieties incorporated in place of the carbon-5 and sulfur atoms (e.g., **5**). We envisaged that these ribosyl analogues might serve as mechanistic probes to study the mechanism of action of LuxS and evaluate the similarities between SAH hydrolase and LuxS. As mentioned above, LuxS inhibitors may provide a novel class of antibacterial agents. We now describe the syntheses of SRH analogues with the carbon-5 and sulfur atoms replaced by vinyl or (6-fluoro)vinyl motifs and discuss their interactions with LuxS enzyme.

2. Chemistry

Our initial plan to prepare compound **5** and its congeners is illustrated in Scheme 1. Treatment of the diacetone 3-*O*-benzoylglucose **6** or allose **7** with periodic acid selectively removed the 5,6-*O*-isopropylidene group. Subsequent oxidative cleavage of the exposed vicinal diol^{9a} gave the corresponding 5-aldehydes **8** and **9**, respectively, in high yields (Scheme 1). Wittig olefination of aldehyde **8** with the ylide derived from commercially available [4-ethoxy-4 oxobutyl)triphenylphoshonium bromide provided a complex mixture of products. Column chromatography yielded protected 5,6,7,8-tetradeoxy-α-D-*xylo*-non-5(*Z*)-enofuranuronate **10** (18% yield). The stereochemistry was assigned as *Z*, based on the magnitude of the coupling constants for olefinic protons (${}^{3}J_{5-6} = 11.1$ Hz), and literature precedence for the Wittig condensations of aliphatic aldehydes with the nonstabilized ylides.^{9b} Similarly, Wittigtreatment of *ribo* 5-aldehyde **9** gave **11**; a nine-carbon analogue of SRH. Unfortunately, our attempts to add bromine (CH₂Cl₂/0 °C) across the double bond of **10** or **11** (as well as **16**) produced a complex mixture which did not give the desired SRH analogues of type **5** bearing a (6-bromo)vinyl unit when treated with DBU.¹⁰

In an alternative approach, we attempted a synthesis of 6-bromoalkenyl analogues $5 (X = Br)$ *via* Pd-catalyzed monoalkylation^{11–13} of the readily available (*gem*-dibromo)vinyl sugar precursors (e.g. **12**, **13**) with the corresponding alkylzinc reagents. Thus, dibromolefination of *5-aldehyde 8 by the Corey-Fuchs procedure¹⁴ gave 5-(dibromomethylene)-5*deoxyxylose **12** (81% from **6** ; Scheme 2). Analogous treatment of the *ribo* 5-aldehyde **9** afforded **13**. 15 Treatment of **12** with 3 equiv. of 4-ethoxy-4-oxobutylzinc bromide in the presence of Pd(PPh3)4 at 55 °C gave monoalkylated 5,6,7,8,9-pentadeoxy-α-D-*xylo*-dec-5 (E) -enofuranuronate 14 (18%, $\frac{3J_{5-6}}{2}$ = 15.4 Hz) and dialkylated 18 (48%) products, but did not yield the desired 6-bromoalkenyl product **15**. Analogous Negishi coupling of 5- (dibromomethylene)-5-deoxyribose **13** afforded only dialkylated product **19** (54%). Changing catalyst $[(Pd₂(dba)₃)]$, solvent (THF), reaction temperature (r.t. to 60 °C) as well as adding additives (CuI, tricyclohexylphosphine) did not lead to the formation of **15** or **17** but instead produced dialkylated byproducts **18** and **19** (3-49%) in agreement with a recent report.13b

We next explored stereoselective coupling of the *gem*-dihalovinyl sugars containing two different halogens. We chose 5-deoxy-5-(fluoroiodomethylene) hexenofuranoses **26** and **27** because the iodo and fluoro substituents are known to have quite different reactivity towards oxidative-addition in Pd-mediated couplings.11b,16,17 The precursors **26** and **27** were prepared employing McCarthy's stannyldesulfonylation methodology.18,19 Thus, treatment of the *xylo* aldehyde **8** with the enolate generated from the sulfonyl-stabilized fluorophosphonate20 gave (fluoro)vinyl sulphones **20** (*E/Z*, 7∶3; 76%; Scheme 3). The stereoselective radical-mediated stannyldesulfonylation of **20** with Bu3SnH produced (fluoro) vinyl stannanes **23** (*E/Z*, 7∶3; 95%). Iododestannylation of **23** with *N*-iodosuccinimide (NIS) quantitatively afforded 6-fluoro-6-iodo-*xylo*-hex-5-enofuranoses **26** with retention of the *E/Z*

configuration. The *ribo* analogue **27** (*E/Z*, 3∶2; 57% overall yield from **9**) was similarly prepared. The isomeric ratio for the fluorinated sugars could be distinguished by the magnitude of the $3J_{F-H5}$ in the NMR spectra.

Pd-mediated cross-coupling of the *xylo* analogue **26** (*E/Z*, 4∶1) with 2 equiv. of 4-ethoxy-4 oxobutylzinc bromide resulted in selective consumption of (*E*)-**26** to afford (*Z*)-**29** in 61% isolated yield or 76% based on consumption of (*E*)-**26** (Scheme 4). A small amount of (*E*)-**29** was also isolated, although monocoupling with *gem*-dihalovinyl substrates is considered to be *trans* selective.12,13b,16 Similar monoalkylation of the *ribo* analogue **27** (*E/Z*, 3∶2) with BrZn(CH₂)₃COOEt yielded (*Z*)-30 [54%, 90% based on the conversion of (E) -27]²¹ and (*E*)-**30** [12%, 30% from (*Z*)-**27**]. Coupling of the (iodo)vinyl (*E*)-**28**, prepared as depicted in Scheme 3 ($9 \rightarrow 22 \rightarrow 25 \rightarrow 28$), with BrZn(CH₂)₃COOEt gave the unfluorinated analogue (E) -16 (56%) with the retention of configuration. Treatment of (Z) -29 with NH₃/MeOH removed the benzoyl group and converted the ethyl ester into a methyl ester (*Z*)-**31** (74%). Subsequent removal of the isopropylidene group with aqueous trifluoroacetic acid (TFA) at 0 °C gave (Z) -33 (61%; α/β , 1:1). Successive treatment of (Z) -30 with NH₃/MeOH followed by TFA/H₂O gave (*Z*)-34 (52% overall yield; α/β , 3:7); a ten-carbon 6-fluoroalkenyl analogue of SRH.

The 5,6-dideoxy-6-fluorohex-5-enofuranoses 42 and 43 , depurinated analogues of $3 (X = F)$, were synthesized by protiodestannylation of the (fluoro)vinyl stannanes **23** and **24**. Thus, treatment of **23** (*E/Z*, 7∶3) with NH3/MeOH at 25 °C resulted in the removal of 3-*O*-benzoyl group to give **36** (Scheme 5). However, prolonged heating of **36** (or **23**) with NH3/MeOH at 65 °C for 48 h effected protiodestannylation to yield a separatable mixture of (*E*)-**39** (29%) and (*Z*)-39 (48%). Treatment of (*E*)-39 with TFA/H₂O at 0 °C gave (*E*)-42 (α/β , ~1:1). Analogous debenzoylation and protiodestannylation of **24** (*E/Z*, 1∶1) with NH3/MeOH yielded (*E*)-**40** (32%) and (*Z*)-**40** (26%). Acid-catalyzed removal of the isopropylidene group in (*E*)-**40** gave 5,6-dideoxy-6-fluoro-D-*ribo*-hex-5-enofuranose (*E*)-**43** (67%; α/β, ~1∶4). Alternatively, concomitant protiodestannylation and removal of acetone unit in **36** or **37** with TFA also afforded **42** and **43**.

The 3,5,6-trideoxy 6-fluorohex-5-enofuranose **44**, which lacks a hydroxyl group at C3 and therefore cannot participate in the second enolization step of the LuxS catalyzed reaction, ^{4b} was also prepared. Thus, oxidation of the diacetone 3-deoxyglucose²² with $H_5IO_6^{9a}$ and *in situ* treatment of the rather unstable 3-deoxyribose 5-aldehyde with the enolate generated from the sulfonyl-stabilized fluorophosphonate²⁰ gave the (fluoro)vinyl sulfones **35** (48%; *E/Z*, 2:1;). Subjection of **35** to the stannyldesulfonylation/protiodestannylation^{18b} sequence afforded 3-deoxy (6-fluoro)vinyl sugar **41**, which was deprotected to yield **44** (12% from **35**). Alternatively, treatment of vinyl stannanes **38** with TFA affected simultaneous protiodestannylation and removal of the acetone unit to give **44** (23% from **35**; *E/Z* ~1∶3, α/β $~1:4$).

3. Inhibition of LuxS

The (6-fluoro)vinyl *xylo*- (**42**) and *ribo*-hexofuranoses (**43**) and their 3-deoxy analogue **44** as well as (6-fluoro)vinyl *xylo*- and *ribo*-decofuranoses (33 and 34) were evaluated^{4h} as potential inhibitors of *Bacillus subtilis S*-ribosylhomocysteinase (LuxS). Compound **44** exhibited competitive inhibition of moderate potency, with a K_I value of $96 \pm 3 \mu$ M (Figure 3). None of the other compounds showed significant inhibition under the assay conditions.

4. Summary and Conclusions

We have developed synthesis of six-, nine- and ten-carbon analogues of ribosyl- and xylosylhomocysteines in which the carbon-5 and sulfur atoms are replaced by a vinyl or (fluoro)

vinyl unit. These fluoroalkenyl and alkenyl analogues of SRH were synthesized employing either the Wittig reaction or Pd-catalyzed coupling routes. They were evaluated against *Bacillus subtilis S*-ribosylhomocysteinase (LuxS). Only 3,5,6-trideoxy-6-fluoro-_{D-}*erythro*hex-5-enofuranose acted as competitive inhibitor of moderate potency with $K_I = 96 \mu M$.

5. Experimental Section

¹H (Me₄Si) NMR spectra were determined with solution in CDCl₃ at 400 or 600 MHz, ¹³C (Me₄Si) at 100.6 MHz and ¹⁹F (CFCl₃) at 376.5 MHz unless otherwise noted. Mass spectra (MS) were obtained by atmospheric pressure chemical ionization (APCI) and HRMS by electron impact techniques unless otherwise noted. Reagent grade chemicals were used as received. Solvents were dried by reflux over and distillation from $CaH₂$ under an argon atmosphere except THF (K/benzophenone). TLC was performed on Merck kieselgel 60 - F_{254} with MeOH/CHCl₃ (1:9) and EtOAc/MeOH (95:5) as developing systems, and products were detected with 254 nm light or by visualization with $Ce(SO₄)₂/(NH₄)₆Mo₇O₂₄•4H₂O/H₂SO₄/$ H2O reagent. Merck kieselgel 60 (230-400 mesh) was used for column chromatography. Elemental analyses were determined by Galbraith Laboratories, Knoxville, TN.

Ethyl 3-*O***-Benzoyl-5,6,7,8-tetradeoxy-1,2-***O***-isopropylidene-α-D-***xylo***-non-5(***Z***) enofuranuronate (10)**

Step (a). H_5IO_6 (150 mg, 0.66 mmol) was added to a stirred solution of 6 (200 mg, 0.55 mmol) in dried EtOAc at ambient temperature. A precipitate appeared within the first five minutes and the resulting solution was stirred for 90 min. The precipitate was filtered off and was washed with EtOAc (2×5 mL). The combined organic layer was washed with NaHCO₃/H₂O (10 mL), NaCl/H2O (10 mL), dried (Na2SO4) and evaporated to yield 3-*O*-benzoyl-1,2-*O*isopropylidene-α-D-*xylo*-pentodialdo-1,4-furanose (**8**; 160 mg, 95%; approx. 90% pure based on ¹H NMR): ¹H NMR δ 1.35 & 1.48 (2 × s, 2 × 3, 2 × CH₃), 4.76 (d, $J_{4-3} = 3.2$ Hz, 1, H4), 4.88 (d, *J*2-1 = 3.1 Hz, 1, H2), 5.77 (d, *J*1-2 = 3.1 Hz, 1, H1), 6.18 (d, *J*3-4 = 3.3 Hz, 1, H3), 7.42-8.01 (m, 5, Ar), 9.78 (s, 1, H5). Step (b). LHMDS (1M/THF; 0.69 mL, 0.69 mmol) was added dropwise to a stirred solution of $Ph_3PCH_2CH_2CH_2CO_2Et/Br$ (314 mg, 0.69 mmol) in anhydrous THF (4 mL) in a flame-dried flask under N_2 at ambient temperature. After 15 minutes, a solution of the crude, preferentially freshly prepared, aldehyde **8** (160 mg of the material from step a) in THF (2 mL) was added via syringe and stirring was continued overnight. EtOAc (30 mL) and NaHCO₃/H₂O (10 mL) was added and the separated organic was washed with NaCl/H₂O (10 mL), dried (Na₂SO₄), and evaporated. Column chromatography ($10 \rightarrow 30\%$ hexanes/EtOAc) gave 10 (39 mg, 18%) as an oil: ¹H NMR δ 1.24 $(t, J = 7.2 \text{ Hz}, 3, \text{ CH}_3), 1.37 \& 1.62 \ (2 \times s, 2 \times 3, 2 \times \text{CH}_3), 2.41 \ (t, J_{8-7/7'} = 6.9 \text{ Hz}, 2, \text{H8/8'}),$ 2.50 ("q", *J*7-6/8/8′ = 7.3 Hz, 2, H7/7′), 4.15 (q, *J* = 7.1 Hz, 2, CH2), 4.71 (d, *J*2-1 = 3.8 Hz, 1, H2), 5.22 (dd, *J*4-5 = 7.7 Hz, *J*4-3 = 2.8 Hz, 1, H4), 5.46 (d, *J*3-4 = 2.8 Hz, H3), 5.58 (dd, *J*5-6 = 11.1 Hz, *J*5-4 = 7.9 Hz, 1, H5), 5.68 (dt, *J*6-5 = 11.1 Hz, *J*6-7/7′ = 7.1 Hz, 1, H6), 6.05 (d, *J*1-2 = 3.7 Hz, 1, H1), 7.48-8.04, (m, 5, Ar); 13C NMR δ 14.62 (CH3), 24.08 (C7), 26.62 & 27.19 (CMe₂), 34.23 (C8), 60.90 (CH₂), 75.54 (C2), 78.59 (C3), 84.20 (C4), 105.02 (C1), 112.47 (*C*Me2), 123.90 (C6), 128.91 (Bz), 129.78 (Bz), 130.15 (Bz), 133.85 (Bz), 134.39 (C5), 165.64 (Bz), 173.09 (C9); MS *m/z* 391 (100, MH+). HRMS (AP-ESI) *m/z* calcd for $C_{21}H_{26}O_7Li$ (MLi⁺) 397.1839; found 397.1833.

Ethyl 3-*O***-Benzoyl-5,6,7,8-tetradeoxy-1,2-***O***-isopropylidene-α-D-***ribo***-non-5(***Z***) enofuranuronate (11)**

Step (a) Oxidation of **7** (200 mg, 0.55 mmol) with H_5IO_6 (150 mg, 0.66 mmol), as described for **10**, gave 3-*O*-benzoyl-1,2-*O*-isopropylidene-α-D-*ribo*-pentodialdo-1,4-furanose (**9**; 145 mg, 85%; approx. 90% pure, ¹H NMR): ¹H NMR δ 1.39 & 1.61 (2 × s, 2 × 3, 2 × CH₃), 4.64 (dd, *J*4-5 = 2.2 Hz, *J*4-3 = 9.2 Hz, 1, H4), 5.01 (t, *J*2-1/3 = 4.2 Hz, 1, H2), 5.13 (dd, *J*3-4 = 9.2

Hz, *J*3-2 = 4.6 Hz, 1, H3), 6.02 (d, *J*1-2 = 3.4 Hz, 1, H1), 7.48-8.03 (m, 5, Ar), 9.77 (d, *J*5-4 = 2.2 Hz, 1, H5). Step (b). Treatment of the crude $9(145 \text{ mg})$ with $Ph_3P(CH_2)_3CO_2Et/Br$ (275 mg, 0.60 mmol) and LHDMS (1M/THF; 0.60 mmol, 0.60 mL), as described for **10**, gave **11** (18 mg, 12%): ¹H NMR δ 1.26 (t, *J* = 7.1 Hz, 3, CH₃), 1.36 & 1.62 (2 × s, 2 × 3, 2 × CH₃), 2.38 (t, *J*8-7/7′ = 8.2 Hz, 2, H8/8′), 2.46-2.55 (m, 1, H7), 2.55-2.67 (m, 1, H7′), 4.15 (q, *J* = 7.1 Hz, 2, CH2), 4.74 (dd, *J*3-4 = 9.1 Hz, *J*3-2 = 4.8 Hz, 1, H3), 4.96 ("t", *J*2-1/3 = 4.3 Hz, 1, H2), 5.20 (t, *J*4-3/5 = 8.7 Hz, 1, H4), 5.50 (ddt, *J*5-6 = 10.9 Hz, *J*5-4 = 8.7 Hz, *J*5-7/7′ = 1.0 Hz, 1, H5), 5.72 (dt, *J*6-5 = 10.9 Hz, *J*6-7/7′ = 7.1 Hz, 1, H6), 5.93 (d, *J*1-2 = 3.8 Hz, 1, H1), 7.48-8.04, (m, 5, Ar); ¹³C NMR δ 14.30 (CH₃), 24.01 (C7), 26.99 & 27.04 (CMe₂), 34.53 (C8), 60.89 (CH₂), 73.39 (C4), 77.34 (C2), 77.56 (C3), 104.61 (C1), 113.40 (*CMe₂)*, 127.26 (C5), 128.90 (Bz), 129.80 (Bz), 130.28 (Bz), 133.80 (Bz), 135.14 (C6), 166.25 (Bz), 173.06 (C9). HRMS (AP-ESI) m/z calcd for $C_{21}H_{26}O_7Li$ (MLi⁺) 397.1839; found 397.1828.

3-*O***-Benzoyl-5,6-dideoxy-6,6-dibromo-1,2-***O***-isopropylidene-α-D-***xylo***-hex-5-enofuranose (12)**

(Dibromomethylene)triphenylphosphorane [generated *in situ* by stirring CBr4 (8.09 g, 24.5 mmol), Ph_3P (6.46 g, 24.5 mmol) and activated Zn (dust; 1.60 g, 24.5 mmol) in dried CH_2Cl_2 (100 mL) at 0 °C (ice-bath) for 30 min followed by stirring at ambient temperature under N2 for 3h] was added to the solution of freshly prepared aldehyde **8** [prepared as described for **10** (step a) from **6** (4.68 g, 12.9 mmol) and dried for 1 h under vacuum prior to use] in CH_2Cl_2 (75 mL). After stirring for 14 h at ambient temperature, the reaction mixture was partitioned (NaHCO₃/H₂O//CHCl₃), and the organic layer was washed (H₂O, brine), dried (MgSO₄), and the volatiles were evaporated. Column chromatography (15 \rightarrow 25% EtOAc/ hexane) gave **12** (4.68 g, 81% overall from **6**) as a solidifying viscous oil: 1H NMR δ 1.34 & 1.58 ($2 \times s$, 2×3 , $2 \times CH_3$), 4.68 (d, $J_{2-1} = 3.7$ Hz, 1, H2), 5.04 (dd, $J_{4-5} = 7.6$ Hz, $J_{4-3} = 3.0$ Hz, 1, H4), 5.54 (d, *J*3-4 = 3.0 Hz, 1, H3), 6.00 (d, *J*1-2 = 3.7 Hz, 1, H1), 6.60 (d, *J*5-4 = 7.6 Hz, 1, H5), 7.48 (t, *J* = 7.6 Hz, 2 Ar), 7.60 (tt, *J* = 1.3, 7.6 Hz, 1 Ar), 8.02 ("dd", *J* = 1.4, 7.7 Hz, 2 Ar); ¹³C NMR δ 24.96 & 25.51 (C*Me*₂), 75.71 (C3), 78.18 (C4), 82.04 (C2), 93.26 (C6), 103.30 (C1), 111.26 (*CMe₂)*, 127.31 (Bz), 127.73 (Bz), 128.44 (Bz), 130.49 (C5), 132.37 (Bz), 163.81 (Bz); MS m/z 451 (5, MH⁺ [⁸¹Br₂]), 449 (10, MH⁺ [^{81/79}Br₂]), 447 (5, MH⁺ [⁷⁹Br₂]).

Ethyl 3-*O***-Benzoyl-5,6,7,8,9-pentadeoxy-1,2-***O***-isopropylidene-α-D-***xylo***-dec-5(***E***) enofuranuronate (14) and Ethyl 3-***O***-Benzoyl-6-[3-(ethoxycarbonyl)propyl]- 5,6,7,8,9 pentadeoxy-1,2-***O***-isopropylidene-α-D-***xylo***-dec-5-enofuranuronate (18)**

Pd $[P(Ph)₃]$ ₄ (22 mg, 0.014 mmol) was added to a stirred solution of 12 (42 mg, 0.094 mmol) in anhydrous benzene (3 mL) in a flame dried flask under N_2 at ambient temperature. After 2 minutes, 4-ethoxy-4-oxobutylzinc bromide (0.5 M/THF; 0.56 mL, 129 mg, 0.28 mmol) was added and the resulting mixture was heated at 55 °C for 6 h. The reaction mixture was cooled down to ambient temperature and was partitioned between EtOAc (30 mL) and NaHCO $3/$ $H₂O$ (10 mL). The separated organic layer was washed with $H₂O$ (10 mL), NaCl/H₂O (10 mL), dried (Na₂SO₄), and evaporated. Column chromatography (10 \rightarrow 30% EtOAc/hexanes) gave recovered **12** (7 mg, 13%), **14** (7 mg, 18%) and **18** (19 mg, 48%). Compound **14** had: 1H NMR δ 1.23 (t, *J* = 7.1 Hz, 3, CH₃), 1.28 & 1.58 (2 × s, 2 × 3, 2 × CH₃), 1.68 (quint, *J*_{8-7/7'/9/9'} = 7.5 Hz, 2, H8/8′), 2.07 ("q", *J*7-6/8/8′ = 7.0 Hz, 2, H7/7′), 2.23 (t, *J*9-8/8′ = 7.4 Hz, 2, H9/9′), 4.15 (q, *J* = 7.1 Hz, 2, CH₂), 4.70 (d, *J*₂₋₁ = 3.7 Hz, 1, H₂), 4.86 (dd, *J*₄₋₅ = 7.1 Hz, *J*₄₋₃ = 2.8 Hz, 1, H4), 5.42 (d, *J*3-4 = 2.7 Hz, 1, H3), 5.56 (dd, *J*5-6 = 15.4 Hz, *J*5-4 = 7.3 Hz, 1, H5), 5.92 (dt, *J*6-5 = 15.4 Hz, *J*6-7/7′ = 6.9 Hz, 1, H6), 6.03 (d, *J*1-2 = 3.7 Hz, 1, H1), 7.46-8.05, (m, 5, Ar); ¹³C NMR δ 14.59 (CH₃), 23.34 (C8), 26.69 & 27.16 (CMe₂), 30.10 (C7), 34.05 (C9), 60.74 (CH2), 75.90 (C2), 81.58 (C3), 85.44 (C4), 105.00 (C1), 112.52 (*C*Me2), 128.35 (C5), 130.17 (Bz), 130.65 (Bz), 132.47 (Bz), 133.60 (Bz), 135.50 (C6), 165.89 (Bz), 173.74 (C10); MS *m/z* 405 (100, MH⁺). Compound **18** had: ¹H NMR δ 1.21 (t, *J* = 7.1 Hz, 3, CH₃), 1.29 (t, *J* = 7.1 Hz, 3, CH₃), 1.36 & 1.61 (2 × s, 2 × 3, 2 × CH₃), 1.69 (quint, *J* = 7.5 Hz, 2H), 1.71-1.80

(m, 2H), 2.05 (t, *J* = 7.5 Hz, 2H), 2.10-2.26 (m, 4H), 2.33 (t, *J* = 7.2 Hz, 2H), 4.09 (q, *J* = 7.1 Hz, 2, CH₂), 4.18 (q, *J* = 7.1 Hz, 2, CH₂), 4.70 (d, *J*₂₋₁ = 3.8 Hz, 1, H2), 5.14 (dd, *J*₄₋₅ = 8.7 Hz, *J*4-3 = 2.9 Hz, 1, H4), 5.35 (d, *J*5-4 = 8.4 Hz, 1, H5), 5.41 (d, *J*3-4 = 2.8 Hz, 1, H3), 6.02 (d, *J*₁₋₂ = 3.8 Hz, 1, H1), 7.47-8.06, (m, 5, Ar); ¹³C NMR δ 14.60 (CH₃), 14.65 (CH₃), 26.65 & 27.15 (C*Me*2), 23.32 & 23.84 (C8/8′), 30.53 & 36.22 (C7/7′), 34.04 & 34.08 (C9/9′), 60.59 (CH₂), 60.75 (CH₂), 75.90 (C4), 78.71 (C3), 84.24 (C2), 104.84 (C1), 112.36 (*CMe₂*), 119.09 (C5), 146.90 (C6), 128.92 (Bz), 128.92 (Bz), 129.81 (Bz), 130.17 (Bz), 133.83 (Bz), 165.70 (Bz), 173.62 & 173.75 (C10/10′); MS *m/z* 519 (100, MH+).

Ethyl 3-*O***-Benzoyl-5,6,7,8,9-pentadeoxy-1,2-***O***-isopropylidene-α-D-***ribo***-dec-5(***E***) enofuranuronate (16)**

Treatment (55 °C, 3 h) of **28** (*E*; 20 mg, 0.048 mmol) with 4-ethoxy-4-oxobutylzinc bromide (0.5 M/THF; 0.19 mL, 65 mg, 0.096 mmol) as described for **14**/**18** gave **16** (11 mg, 56%): 1H NMR δ 1.24 (t, *J* = 7.1 Hz, 3, CH₃), 1.35 & 1.59 (2 × s, 2 × 3, 2 × CH₃), 1.72 (quint, *J*8-7/7′/9/9′ = 7.4 Hz, 2, H8/8′), 2.13 ("q", *J*7-6/8/8′ = 7.2 Hz, 2, H7/7′), 2.28 (t, *J*9-8/8′ = 7.6 Hz, 2, H9/9'), 4.10 (q, *J* = 7.1 Hz, 2, CH₂), 4.65 (dd, *J*₄₋₅ = 7.6 Hz, *J*₄₋₃ = 8.9 Hz, 1, H4), 4.74 (dd, *J*3-4 = 9.2 Hz, *J*3-2 = 4.6 Hz, 1, H3), 4.96 (t, *J*2-1/3 = 4.3 Hz, 1, H2), 5.53 (dd, *J*5-6 = 15.4 Hz, *J*5-4 = 7.3 Hz, 1, H5), 5.89 (d, *J*1-2 = 4.0 Hz, 1, H1), 5.91 (dt, *J*6-5 = 15.8 Hz, *J*6-7/7′ = 6.8 Hz, 1, H6), 7.46-8.05, (m, 5, Ar); 13C NMR δ 14.62 (CH3), 24.40 (C8), 26.92 & 27.00 (C*Me*2), 31.99 (C7), 33.89 (C9), 60.65 (CH2), 76.92 (C2), 77.65 (C3), 78.60 (C4), 104.37 (C1), 113.35 (*C*Me2), 127.04 (C5), 128.85 (Bz), 129.83 (Bz), 130.28 (Bz), 133.76 (Bz), 136.28 (C6), 166.29 (Bz), 173.84 (C10); MS *m/z* 405 (100, MH+).

Ethyl 3-*O***-Benzoyl-5,6,7,8,9-pentadeoxy-6-[3-(ethoxycarbonyl)propyl]-1,2-***O***isopropylideneα-D-***ribo***-dec-5-enofuranuronate (19)**

Treatment of 13^{15} (42 mg, 0.094 mmol) with Pd[P(Ph)₃]₄ (22 mg, 0.014 mmol) and 4ethoxy-4-oxobutylzinc bromide (0.56 mL, 129 mg, 0.28 mmol) as described for **14**/**18** gave **19** (26 mg, 54%): ¹H NMR δ 1.22 (t, *J* = 7.1 Hz, 6, 2 × CH₃), 1.37 & 1.62 (2 × s, 2 × 3, 2 × CH3), 1.75 (quint, *J* = 7.5 Hz, 4H), 2.11 (t, *J* = 7.0 Hz, 2H), 2.24 (t, *J* = 9.0 Hz, 2H), 2.26 (t, *J* = 9.0 Hz, 2H), 2.32 (t, *J* = 7.2 Hz, 2H), 4.10 (q, *J* = 7.1 Hz, 2, CH₂), 4.16 (q, *J* = 7.1 Hz, 2, CH₂), 4.74 (dd, $J_{3-4} = 9.0$ Hz, $J_{3-2} = 4.8$ Hz, 1, H3), 4.95 (t, $J_{2-3} = 4.3$ Hz, 1, H2), 5.01 (t, *J*₄₋₅ = 9.0 Hz, 1, H4), 5.26 (d, *J*₅₋₄ = 8.9 Hz, 1, H5), 5.89 (d, *J*₁₋₂ = 3.9 Hz, 1, H1), 7.40-8.10 (m, 5, Ar); ¹³C NMR δ 14.55 (CH₃), 14.61 (CH₃), 21.48 (C8/8'), 23.11 & 27.03 (CMe₂), 30.08 & 30.12 (C7/7′), 32.34 & 34.04 (C9/9′), 60.73 (CH2), 60.83 (CH2), 73.86 (C4), 78.71 (C3), 84.24 (C2), 104.37 (C1), 113.28 (*C*Me2), 122.65 (C5), 128.92 (Bz), 128.92 (Bz), 129.81 (Bz), 130.17 (Bz), 133.83 (Bz), 143.90 (C6), 165.70 (Bz), 170.57 & 171.62 (C10/10′); MS *m/z* 519 $(100, \text{MH}^+).$

(*E/Z***)-3-***O***-Benzoyl-5,6-dideoxy-6-fluoro-1,2-***O***-isopropylidene-6-phenylsulfonyl-α-D-***xylo***hex-5-enofuranose (20)**

LHMDS (0.84 mL, 140 mg, 0.84 mmol) was added dropwise to a solution of diethyl fluoro (phenylsulfonyl)methylphosphonate²⁰ (260 mg, 0.84 mmol) in anhydrous THF (8 mL) in a flame dried flask under N₂ at −78 °C. After 30 minutes, a solution of **8** (265 mg, 0.82 mmol) in THF (4 mL) was added and stirring was continued for 1.5 h. EtOAc (30 mL) and $NH₄Cl$ / H2O (10 mL) were added and reaction mixture was allowed to warm to ambient temperature. The separated organic layer was washed with NaHCO₃/H₂O (10 mL), NaCl/H₂O (10 mL), dried (Na₂SO₄), and evaporated. Column chromatography ($10 \rightarrow 30\%$ EtOAc/hexanes) gave **20** (166 mg, 76%; *E/Z*, 7∶3) as inseparable mixture of isomers: HRMS (AP-ESI) *m/z*: calcd for C₂₂H₂₂FO₇S (MH⁺) 449.1065, found 449.1071; ¹⁹F NMR δ −110.25 (d, *J*_{F-H5} = 18.8 Hz, 0.30F, *Z*), −119.30 (d, *J*F-H5 = 32.1 Hz, 0.70F, *E*). Compound (*E*)-**20** had: 1H NMR δ 1.35 & 1.56 ($2 \times s$, 2×3 , $2 \times CH_3$), 4.73 (d, $J_{2-1} = 3.7$ Hz, 1, H2), 5.23 (dt, $J_{4-5} = 7.4$ Hz, $J_{4-3/F} = 2.3$

Hz, 1, H4), 5.49 (d, $J3-4 = 3.1$ Hz, 1, H3), 6.03-6.05 (m, 1, H1), 6.43 (dd, $J_{5-F} = 32.4$ Hz, *J*₅₋₄ = 7.2 Hz, 1, H5), 7.48-8.03 (m, 10, Ar); ¹³C NMR δ 26.55 & 27.10 (C*Me₂*), 73.78 (C4), 78.21 (C3), 83.77 (C2), 105.31 (C1), 113.13 (*CMe*₂), 112.10 (d, ²J_{5-F} = 3.3 Hz, C5), 128.99 (Ph), 129.08 (Bz), 129.13 (Bz), 129.79 (Ph), 129.94 (Ph), 130.12 (Ph), 130.17 (Bz), 134.25 (Bz), 135.02 (Ph), 156.00 (d, ¹J_{6-F} = 300.0 Hz, C6), 165.36 (Bz). Compound (*Z*)-20 had: ¹H NMR δ 1.38 & 1.69 (2 × s, 2 × 3, 2 × CH₃), 4.76 (d, *J*₂₋₁ = 3.7 Hz, 1, H₂), 5.68 (d, *J*₃₋₄ = 2.8 Hz, 1, H3), 5.99 (dd, J_{5-F} = 19.3 Hz, J_{5-4} = 8.6 Hz, 1, H5), 6.05-6.07 (m, 1, H1), 6.07-6.09 (m, 1, H4), 7.48-8.03 (m, 10, Ar); ¹³C NMR δ 26.98 & 27.32 (CMe₂), 73.15 (d, ³J_{4-F} = 10.14 Hz, C4), 79.06 (C3), 83.93 (C2), 105.37 (C1), 113.35 (CMe₂), 114.10 (d, ²J_{5-F} = 15.0 Hz, C5), 128.99 (Ph), 129.08 (Bz), 129.13 (Bz), 129.79 (Ph), 129.94 (Ph), 130.12 (Ph), 130.17 (Bz), 134.25 (Bz), 135.02 (Ph), 155.58 (d, ¹J_{6-F} = 292.3 Hz, C6), 165.36 (Bz). Note: Freshly prepared aldehyde **8**, dried under vacuum for 2 h at ambient temperature prior the use, gave the best results.

(*E/Z***)-3-***O***-Benzoyl-5,6-dideoxy-6-fluoro-1,2-***O***-isopropylidene-6-phenylsulfonyl-α-D-***ribo***hex-5-enofuranose (21)**

Treatment of **9** (200 mg, 0.68 mmol) with diethyl fluoro(phenylsulfonyl) methylphosphonate²⁰ (212 mg, 0.68 mmol) and LHMDS (0.68 mL, 114 mg, 0.68 mmol) as described for **20** gave **21** (216 mg, 71%; *E/Z*, 6∶4): HRMS (AP-ESI) *m/z*: calcd for C₂₂H₂₂FO₇S (MH⁺) 449.1065, found 449.1069; ¹⁹F NMR δ −108.98 (d, *J*_{F-H5} = 22.6 Hz, 0.40F, *Z*), −121.25 (d, *J*F-H5 = 30.1 Hz, 0.60F, *E*). Compound (*E*)-**21** had: 1H NMR δ 1.28 & 1.38 ($2 \times s$, 2×3 , $2 \times CH_3$), 4.85 (dd, $J_{3-4} = 9.0$ Hz, $J_{3-2} = 4.7$ Hz, 1, H3), 5.00 ("t", $J_{2-1/3} =$ 4.5 Hz, 1, H2), 5.10 (t, *J*4-3/5 = 8.2 Hz, 1, H4), 5.94 (d, *J*1-2 = 3.7 Hz, 1, H1), 6.37 (dd, *J*5-F = 31.3 Hz, *J*5-4 = 8.3 Hz, 1, H5), 7.44-8.20 (m, 10, Ar); 13C NMR δ 26.86 & 26.93 (C*Me*2), 71.36 $(d, {}^{3}J_{4-F} = 2.2 \text{ Hz}, \text{C4}), 76.64 (d, {}^{4}J_{3-F} = 1.8 \text{ Hz}, \text{C3}), 77.54 (\text{C2}), 104.96 (\text{C1}), 114.10$ $(CMe₂)$, 114.22 (d, $^{2}J_{5-F} = 3.1$ Hz, C5), 128.88 (Ph), 129.13 (Bz), 129.18 (Bz), 129.88 (Ph), 130.28 (Ph), 133.98 (Bz), 135.10 (Ph), 136.96 (Ph), 156.91 (d, ¹J_{6-F} = 306.0 Hz, C6), 165.96 (Bz). Compound (*Z*)-21 had: ¹H NMR δ 1.28 & 1.38 (2 × s, 2 × 3, 2 × CH₃), 4.84 (dd, J_{3-4} = 9.2 Hz, *J*3-2 = 4.6 Hz, 1, H3), 5.01 ("t", *J*2-1/3 = 4.6 Hz, 1, H2), 5.86 (dd, *J*5-F = 19.8 Hz, *J*5-4 = 9.9 Hz, H5), 5.96 (d, *J*1-2 = 3.7 Hz, 1, H1), 6.07 (t, *J*4-3/5 = 10.4 Hz, 1, H4), 7.44-8.20 (m, 10, Ar); ¹³C NMR δ 27.17 & 27.39 (CMe₂), 70.71 (d, ³J_{4-F} = 8.5 Hz, C4), 77.17 (C3), 77.86 $(C2)$, 104.82 $(C1)$, 114.37 $(CMe₂)$, 116.23 $(d, {}^{2}J_{5-F} = 16.2$ Hz, C5), 128.95 (Ph), 129.22 (Bz), 129.46 (Bz), 129.80 (Ph), 130.02 (Ph), 133.61 (Bz), 134.00 (Ph), 135.19 (Ph), 156.62 $(d, {}^{1}J_{6-F} = 296.3 \text{ Hz}, \text{C6}), 166.30 \text{ (Bz)}.$

(*E***)-3-***O***-Benzoyl-5,6-dideoxy-1,2-***O***-isopropylidene-6-phenylsulfonyl-α-D-***ribo***-hex-5 enofuranose (22)**

Treatment of **9** (150 mg, 0.50 mmol) with diethyl (phenylsulfonyl)methylphosphonate20 (146 mg, 0.50 mmol) and LHMDS (0.50 mL, 84 mg, 0.50 mmol) as described for **20** gave **22** (166 mg, 82%): ¹H NMR δ 1.33 & 1.55 (2 × s, 2 × 3, 2 × CH₃), 4.76 (dd, *J*₃₋₄ = 9.5 Hz, *J*₃₋₂ = 4.6 Hz, 1, H3), 4.92 (ddd, *J*4-5 = 3.7 Hz, *J*4-6 = 1.7 Hz, *J*4-3 = 9.5 Hz, 1, H4), 5.01 ("t", *J*2-3/1 = 4.2 Hz, 1, H2), 5.90 (d, *J*1-2 = 3.7 Hz, 1, H1), 6.79 (dd, *J*6-4 = 1.8 Hz, *J*6-5 = 15.0 Hz, 1, H6), 7.09 (dd, *J*5-6 = 15.0 Hz, *J*5-4 = 3.8 Hz, 1, H5), 7.50-8.05 (m, 10, Ar); MS *m/z* 431 (100, MH+).

(*E/Z***)-3-***O***-Benzoyl-5,6-dideoxy-6-fluoro-1,2-***O***-isopropylidene-6-tributylstannyl-α-D-***xylo***hex-5-enofuranose (23)**

Bu3SnH (407 mg, 0.38 mL, 1.4 mmol) was added dropwise to a degassed solution of **20** (300 mg, 0.70 mmol; *E/Z*, 7∶3) in anhydrous toluene (5 mL) in a flame-dried flask under N₂ at ambient temperature. After an additional 10 minutes of degassing with N_2 , AIBN (86 mg, 0.53) mmol) was added and the reaction mixture was refluxed at 110 °C with stirring for 5 h. The volatiles were evaporated and the residue was partitioned between EtOAc (50 mL) and

NaHCO₃/H₂O (30 mL). The organic layer was washed with NaCl/H₂O (30 mL), dried (Na_2SO_4) , and evaporated. Column chromatography (hexanes $\rightarrow 10\%$ EtOAc/hexanes) gave **23** (794 mg, 95%; *E/Z*, 7:3) as an inseparable mixture: MS *m/z* 599 (89, MH⁺, ¹²⁰Sn), 597 (63, MH⁺, ¹¹⁸Sn), 595 (33, MH⁺, ¹¹⁶Sn), 541 (100, M-57, ¹²⁰Sn), 539 (78, M-57, ¹¹⁸Sn), 537 (42, M-57, 116Sn); 19F NMR δ −87.67 (d, *J*F-H5 = 34.3 Hz, 84% of 0.30F, *Z*), −87.67 (dd, *J*F-Sn = 229.5 Hz, *J*F-H5 = 34.8 Hz, 16% of 0.30F, *Z*), −92.73 (d, *J*F-H5 = 52.7 Hz, 84% of 0.70F, *E*), −92.73 (ddd, *J*F-Sn = 213.1 Hz, *J*F-H5 = 52.7 Hz, *J*F-H4 = 4.9 Hz, 16% of 0.70F, *E*). Compound (*E*)-23 had: ¹H NMR δ 0.90-1.60 (m, 27, $3 \times$ Bu), 1.34 & 1.36 ($2 \times$ s, $2 \times$ 3, $2 \times$ CH₃), 4.71 (d, *J*₂₋₁ = 3.8 Hz, 1, H₂), 5.10 (dd, *J*_{5-F} = 52.6 Hz, *J*₅₋₄ = 7.4 Hz, 1, H₅), 5.32 (d, *J*₃₋₄ = 3.0 Hz, 1, H3), 5.47-5.49 (m, 1, H4), 6.02 (d, *J*1-2 = 3.8 Hz, 1, H1), 7.47-8.06 (m, 5, Ar); 13C NMR δ 10.33 (Bu), 11.15 (Bu), 17.90 (Bu), 27.42 & 27.54 (CMe₂), 28.24 (Bu), 70.56 (d, ³J_{4-F} = 17.6 Hz, C4), 77.21 (C3), 77.52 (C2), 104.31 (C1), 113.07 (*C*Me₂), 120.53 (d, ²J_{5-F} = 3.9 Hz, C5), 128.75 (Bz), 129.83 (Bz), 130.25 (Bz), 133.61 (Bz), 166.16 (Bz), 177.14 (d, ²J_{6-F} = 262.0 Hz, C6). Compound (*Z*)-23 had: ¹H NMR δ 0.90-1.60 (m, 27, 3 \times Bu), 1.38 & 1.69 (2 \times s, 2 \times 3, 2 × CH3), 4.69 (d, *J*1-2 = 3.9 Hz, 1, H2), 4.75 (d, *J*3-4 = 7.9 Hz, 1, H3), 5.47-5.49 (m, 1, H4), 5.98 (d, *J*1-2 = 3.8 Hz, 1, H1), 6.02 (dd, *J*5-F = 34.3 Hz, *J*5-4 = 9.2 Hz, 1, H5), 7.47-8.06 (m, 5, Ar); ¹³C NMR δ 10.33 (Bu), 11.15 (Bu), 17.90 (Bu), 27.42 & 27.54 (CMe₂), 28.24 (Bu), 74.73 $(d, {}^{3}J_{4-F} = 22.2 \text{ Hz}, \text{C}4), 77.38 (d, {}^{4}J_{3-F} = 1.4 \text{ Hz}, \text{C}3), 77.52 (\text{C}2), 104.53 (\text{C}1), 113.47$ (CMe_2) , 121.24 (d, $^2J_{5-F} = 9.5$ Hz, C5), 128.75 (Bz), 129.88 (Bz), 130.34 (Bz), 133.73 (Bz), 166.35 (Bz), 180.03 (d, $^{2}J_{6-F} = 254.3$ Hz, C6).

(*E/Z***)-3-***O***-Benzoyl-5,6-dideoxy-6-fluoro-1,2-***O***-isopropylidene-6-tributylstannyl-α-D-***ribo***hex-5-enofuranose (24)**

Treatment of **21** (300 mg, 0.70 mmol; *E*/Z, 3∶2) with Bu₃SnH (407 mg, 0.38 mL, 1.4 mmol) and AIBN (86 mg, 0.53 mmol) as described for **23** gave **24** (397 mg, 95%; *E/Z*, 3∶2): MS *m/* z 599 (89, MH⁺, ¹²⁰Sn), 597 (63, MH⁺, ¹¹⁸Sn), 595 (33, MH⁺, ¹¹⁶Sn), 541 (100, M-57, ¹²⁰Sn), 539 (78, M-57, 118Sn), 537 (42, M-57, 116Sn); 19F NMR δ −87.58 (d, *J*F-H5 = 33.1 Hz, 84% of 0.40F, *Z*), −87.58 (ddd, *J*F-Sn = 226.7 Hz, *J*F-H5 = 32.8 Hz, *J*F-H4 = 4.1 Hz, 16% of 0.40F), −94.80 (d, *J*F-H5 = 51.1 Hz, 84% of 0.60F, *E*), −94.80 (ddd, *J*F-Sn = 213.9 Hz, *J*F-H5 = 50.8 Hz, $J_{\text{F-H4}} = 4.5$ Hz, 16% of 0.60F, *E*). Compound (*E*)-24 had: ¹H NMR δ 0.70-1.70 (m, 27, 3 \times Bu), 1.24 & 1.26 (2 × s, 2 × 3, 2 × CH3), 4.62 (dd, *J*3-4 = 9.3 Hz, *J*3-2 = 4.7 Hz, 1, H3), 4.86-4.87 (m, 1, H2), 4.90 (dd, $J_{5-F} = 51.0$ Hz, $J_{5-4} = 8.4$ Hz, 1, H5), 5.28 ("t", $J_{4-3/5} = 8.9$ Hz, 1, H4), 5.79 (d, *J*1-2 = 3.9 Hz, 1, H1), 7.57-8.06 (m, 5, Ar); 13C NMR δ 11.15 (Bu), 14.03 (Bu), 29.96 (Bu) , 27.48 & 27.59 (CMe₂), 29.23 (Bu), 70.55 (d, ${}^{3}J_{4-F} = 18.1$ Hz, C4), 77.36 (C3), 77.48 $(C2)$, 104.50 $(C1)$, 113.54 $(CMe₂)$, 120.59 $(d, {}^{2}J_{5-F} = 3.8 \text{ Hz}, C5)$, 128.77 (Bz) , 129.81 (Bz) , 130.38 (Bz), 133.69 (Bz), 166.44 (Bz), 176.10 (d, ¹ *J*6-F = 260.0 Hz, C6). Compound (*Z*)-**24** had: ¹H NMR δ 0.70-1.70 (m, 27, $3 \times$ Bu), 1.28 & 1.30 ($2 \times$ s, $2 \times$ 3, $2 \times$ CH₃), 4.47 ("t", *J*₄₋₃/₅ = 9.3 Hz, 1, H4), 4.71 (dd, *J*₃₋₄ = 9.0 Hz, *J*₃₋₂ = 4.8 Hz, 1, H3), 4.85-4.86 (m, 1, H2), 5.81 (d, *J*₁₋₂ = 3.8 Hz, 1, H1), 5.83 (dd, *J*_{5-F} = 33.7 Hz, *J*₅₋₄ = 9.5 Hz, 1, H5), 7.57-8.06 (m, 5, Ar); ¹³C NMR δ 11.15 (Bu), 14.03 (Bu), 29.96 (Bu), 27.48 & 27.59 (CMe₂), 29.23 (Bu), 74.72 $(d, {}^{3}J_{4-F} = 22.0 \text{ Hz}, C4$, 77.16 (C3), 77.67 (C2), 104.28 (C1), 113.13 (*CMe₂)*, 121.18 $(d, {}^{2}J_{5-F} = 9.9 \text{ Hz}, \text{C5}), 128.77 \text{ (Bz)}, 129.76 \text{ (Bz)}, 130.28 \text{ (Bz)}, 130.80 \text{ (Bz)}, 166.24 \text{ (Bz)}, 177.00 \text{ (Bz)}$ $(d, {}^{1}J_{6-F} = 255.0 \text{ Hz}, \text{C6}).$

(*E***)-3-***O***-Benzoyl-5,6-dideoxy-1,2-***O***-isopropylidene-6-tributylstannyl-α-D-***ribo***-hex-5 enofuranose (25)**

Treatment of **22** (*E*; 300 mg, 0.70 mmol) with Bu3SnH (407 mg, 0.38 mL, 1.4 mmol) and AIBN (86 mg, 0.53 mmol) as described for **23** gave **25** (385 mg, 95%): 1H NMR δ 1.51-1.90 (m, 33, 3 × Bu & 2 × CH3), 4.66 (dd, *J*4-5 = 6.5 Hz, *J*4-3 = 8.4 Hz, 1, H4), 4.77 (dd, *J*3-4 = 9.2 Hz, *J*3-2 = 4.7 Hz, 1, H3), 4.97 (t, *J*2-1/3 = 3.6 Hz, 1, H2), 5.93 (d, *J*1-2 = 3.8 Hz, 1, H1), 6.10 (dd, *J*5-6 = 19.1 Hz, *J*5-4 = 6.5 Hz 1, H5), 6.50 (dd, *J*6-5 = 19.1 Hz, *J*6-4 = 0.9 Hz, 1, H6), 7.57-8.06

(m, 5, Ar); MS m/z 581 (89, MH⁺, ¹²⁰Sn), 579 (63, MH⁺, ¹¹⁸Sn), 577 (33, MH⁺, ¹¹⁶Sn), 523 $(100, M-57, \frac{120}{5}Sn), 521 (78, M-57, \frac{118}{5}Sn), 519 (42, M-57, \frac{116}{5}Sn).$

(*E/Z***)-3-***O***-Benzoyl-5,6-dideoxy-6-fluoro-6-iodo-1,2-***O***-isopropylidene-α-D-***xylo***-hex-5 enofuranose (26)**

A solution of NIS (50 mg, 0.23 mmol) in anhydrous CH_2Cl_2 (3 mL) was added dropwise to a stirred solution of **23** (90 mg, 0.15 mmol; E/Z , 7:3) in CH₂Cl₂ (5 mL) under N₂ at −20 °C. After 1h, CHCl₃ (30 mL) and diluted NaHSO₃/H₂O (10 mL) were added. The separated organic layer was washed with NaHCO₃/H₂O (10 mL), NaCl/H₂O (10 mL), dried (Na₂SO₄), and evaporated. Column chromatography (hexanes \rightarrow 30% EtOAc/hexanes) gave 26 (155 mg, 83%; *E/Z*, 8∶2) as an inseparable mixture: MS *m/z* 435 (100, MH+); 19F NMR δ −56.54 (d, *J*F-H5 = 15.8 Hz, 0.20F, *Z*), −60.98 (d, *J*F-H5 = 33.1 Hz, 0.80F, *E*). Compound (*E*)-**26** had: 1H NMR δ 1.35 & 1.38 (2 × s, 2 × 3, 2 × CH₃), 4.69 (d, *J*₂₋₁ = 3.8 Hz, 1, H2), 5.28 (ddd, *J*₄₋₅ = 9.8 Hz, *J*4-3 = 2.9 Hz, *J*4-6 = 1.6 Hz, 1, H4), 5.45 (d, *J*3-4 = 3.0 Hz, 1, H3), 5.58 (dd, *J*5-F = 33.0 Hz, *J*5-4 = 8.7 Hz, 1, H5), 5.99 (d, *J*1-2 = 3.7 Hz, 1, H1), 7.47-8.06 (m, 5, Ar); 13C NMR δ 26.60 & 27.12 (CMe₂), 74.04 (d, ³J_{4-F} = 4.4 Hz, C4), 77.88 (C3), 83.73 (C2), 104.74 (C1), 107.89 $(d, {}^{1}J_{6-F} = 338.7 \text{ Hz}, \text{C6}), 112.85 \text{ (CMe}_2), 116.80 \text{ (d, } {}^{2}J_{5-F} = 5.4 \text{ Hz}, \text{C5}), 129.04 \text{ (Bz)}, 129.50 \text{ }$ (Bz), 130.18 (Bz), 134.09 (Bz), 165.54 (Bz). Compound (*Z*)-**26** had: 1H NMR δ 1.35 & 1.68 $(2 \times s, 2 \times 3, 2 \times CH_3)$, 4.71 (d, *J*₁₋₂ = 3.7 Hz, 1, H2), 4.84 (dd, *J*₄₋₅ = 8.7 Hz, *J*₄₋₃ = 2.7 Hz, 1, H4), 5.48 (d, $J_{3-4} = 3.0$ Hz, 1, H3), 5.77 (dd, $J_{5-F} = 15.4$ Hz, $J_{5-4} = 8.8$ Hz, 1, H5), 6.03 (d, *J*₁₋₂ = 3.7 Hz, 1, H1), 7.47-8.06 (m, 5, Ar); ¹³C NMR δ 26.78 & 27.27 (C*Me₂*), 77.88 (C3), 80.06 (d, ³J_{4-F} = 8.2 Hz, C4), 83.84 (C2), 105.04 (C1), 112.94 (d, ²J_{5-F} = 16.9 Hz, C5), 112.95 (*C*Me₂), 114.78 (d, ¹J_{6-F} = 332.0 Hz, C6), 129.04 (Bz), 129.40 (Bz), 130.18 (Bz), 134.14 (Bz), 165.54 (Bz).

(*E/Z***)-3-***O***-Benzoyl-5,6-dideoxy-6-fluoro-6-iodo-1,2-***O***-isopropylidene-α-D-***ribo***-hex-5 enofuranose (27)**

Treatment of **24** (250 mg, 0.42 mmol; *E/Z*, 3∶2) with NIS (142 mg, 0.63 mmol) as described for **26** gave **27** (155 mg, 85%; *E/Z*, 3∶2) as an inseparable mixture: HRMS (AP-FAB) *m/z*: calcd for C16H16FIO5Li (MLi+) 441.0181; found 441.0192; 19F NMR δ −56.42 (d, *J*F-H5 = 15.1 Hz, 0.40F, *Z*), −63.30 (d, *J*F-H5 = 33.5 Hz, 0.60F, *E*). Compound (*E*)-**27** had: 1H NMR δ 1.36 & 1.60 (2 × s, 2 × 3, 2 × CH3), 4.75 (dd, *J*3-4 = 9.2 Hz, *J*3-2 = 4.7 Hz, 1, H3), 4.98 (t, *J*2-1/3 = 4.5 Hz, 1, H2), 5.16 (t, $J_{4-3/5} = 9.0$ Hz, 1, H4), 5.49 (dd, $J_{5-F} = 32.7$ Hz, $J_{5-4} = 8.7$ Hz, 1, H5), 5.89 (d, *J*1-2 = 3.8 Hz, 1, H1), 7.50-8.10 (m, 5, Ar); 13C NMR δ 26.96 & 27.11 (C*Me*2), 72.45 $(d, {}^{3}J_{4-F} = 4.3 \text{ Hz}, C4)$, 76.74 $(d, {}^{4}J_{3-F} = 2.1 \text{ Hz}, C3)$, 77.32 (C2), 104.39 (C1), 113.78 (CMe_2) , 114.96 (d, ¹J_{6-F} = 331.5 Hz, C6), 119.90 (d, ²J_{5-F} = 5.5 Hz, C5), 128.92 (Bz), 129.46 (Bz), 130.50 (Bz), 133.94 (Bz), 166.21 (Bz). Compound (*Z*)-**27** had: 1H NMR δ 1.38 & 1.64 $(2 \times s, 2 \times 3, 2 \times CH_3)$, 4.72-4.75 (m, 1, H4), 4.84 (dd, $J_{3-4} = 9.2$ Hz, $J_{3-2} = 4.6$ Hz, 1, H3), 4.98 (t, *J*2-3/1 = 4.5 Hz, 1, H2), 5.68 (dd, *J*5-F = 15.3 Hz, *J*5-4 = 8.9 Hz, 1, H5), 5.91 (d, *J*1-2 = 3.8 Hz, 1, H1), 7.50-8.14 (m, 5, Ar); ¹³C NMR δ 26.96 & 27.11 (CMe₂), 76.85 (d, ⁴J_{3-F} = 2.1 Hz, C3), 77.63 (C2), 78.13 (d, ³J_{4-F} = 8.3 Hz, C4), 104.54 (C1), 108.75 (d, ¹J_{6-F} = 339.4 Hz, C6), 113.90 (*C*Me₂), 115.77 (d, ²J_{5-F} = 16.2 Hz, C5), 128.91 (Bz), 129.49 (Bz), 130.37 (Bz), 133.95 (Bz), 166.21 (Bz).

(*E***)-3-***O***-Benzoyl-5,6-dideoxy-6-iodo-1,2-***O***-isopropylidene-α-D-***ribo***-hex-5-enofuranose (28)**

Treatment of **25** (150 mg, 0.25 mmol) with NIS (85 mg, 0.38 mmol) as described for **26** gave **28** (93 mg, 87%): ¹H NMR δ 1.35 & 1.62 (2 × s, 2 × 3, 2 × CH₃), 4.67 (dd, $J_{2-3} = 9.2$ Hz, *J*2-1 = 4.6 Hz, 1, H2), 4.77 (dd, *J*3-4 = 3.4 Hz, *J*3-2 = 9.2 Hz, 1, H3), 4.97 (t, *J*4-3/5 = 4.2 Hz, 1, H4), 5.91 (d, J_{1-2} = 3.8 Hz, 1, H1), 6.61-6.70 (m, 2, H5/6), 7.49-8.08 (m, 5, Ar); ¹³C NMR δ 26.95 & 26.96 (CMe₂), 76.37 (C4), 77.59 (C3), 79.61 (C2), 81.32 (C6), 104.43 (C1), 113.68

(*C*Me2), 128.89 (Bz), 129.60 (Bz), 130.94 (Bz), 133.89 (Bz), 141.58 (C5), 166.08 (Bz); MS m/z 417 (100, MH⁺).

Ethyl 3-*O***-Benzoyl-5,6,7,8,9-pentadeoxy-6-fluoro-1,2-***O***-isopropylidene-α-D-***xylo***-dec-5-(***E/Z***) enofuranuronate (29)**

Pd[P(Ph)₃]₄ (5 mg, 0.004 mmol) was added to a stirred solution of **26** (30 mg, 0.07 mmol; $E/$ *Z*, 4∶1) in anhydrous benzene (3 mL) under N₂ at ambient temperature. After 2 minutes, 4ethoxy-4-oxobutylzinc bromide (0.5M/THF; 0.28 mL, 65 mg, 0.14 mmol) was added and the resulting mixture was heated at 55 °C for 5 h. EtOAc (30 mL) and NaHCO₃/H₂O (10 mL) were added and the separated organic layer was washed with $H_2O(10 \text{ mL})$, NaCl/H₂O (10 mL), dried (Na₂SO₄), and then was evaporated. Column chromatography (10 \rightarrow 30% EtOAc/ hexanes) gave (*Z*)-**29** (18 mg, 61%, 76% based on the conversion of the *E*-isomer), (*E*)-**29** (2 mg, 7%, 35% based on the conversion of the *Z*-isomer) and more polar byproduct tentatively assigned as 3-*O*-debenzoylated-(*Z*)-26 [~5%, TLC; ¹⁹F NMR δ –57.21 (*J*_{F-H5} = 16.4 Hz)]. Compound (*Z*)-**29** had: ¹H NMR δ 1.22 (t, *J* = 7.1 Hz, 3, CH₃), 1.36 & 1.61 (2 × s, 2 × 3, 2 × CH₃), 1.81 ("quint", $J_{8-7/7/9/9}$ = 7.4 Hz, 2, H8/8'), 2.26 (dt, J_{7-F} = 18.1 Hz, $J_{7-8/8'}$ = 7.4 Hz, 2, H7/7′), 2.30 (t, *J*9-8/8′ = 7.4 Hz, 2, H9/9′), 4.09 (q, *J* = 7.1 Hz, 2, CH2), 4.69 (d, *J*2-1 = 3.7 Hz, 1, H2), 4.84 (dd, *J*5-F = 35.8 Hz, *J*5-4 = 8.4 Hz, H5), 5.33 (dd, *J*4-5 = 8.5 Hz, *J*4-3 = 2.8 Hz, 1, H4), 5.45 (d, $J_{3-4} = 2.8$ Hz, 1, H3), 6.00 (d, $J_{1-2} = 3.7$ Hz, 1, H1), 7.48-8.04, (m, 5, Ar); ¹³C NMR δ 14.59 (CH₃), 21.49 (C8), 26.64 & 27.11 (CMe₂), 31.61 (d, ²J_{7-F} = 25.4 Hz, C7), 33.34 $(C9)$, 60.75 (CH_2) , 73.37 (d, $3J_{4-F} = 6.6$ Hz, C4), 77.63 (C2), 78.33 (C3), 100.03 (d, $2J_{5-F} =$ 10.9 Hz, C5), 104.78 (C1), 112.60 (*C*Me2), 128.97 (Bz), 129.70 (Bz), 130.16 (Bz), 133.93 (Bz), 162.94 (d, ¹J_{6-F} = 260.7 Hz, C6), 165.61 (Bz), 173.32 (C10); ¹⁹F NMR δ −99.93 (dt, $J_{\text{F-H5}}$ = 35.8 Hz, J = 18.0 Hz); HRMS (AP-FAB⁺) m/z calcd for C₂₂H₂₇FO₇Li (MLi⁺) 429.1910; found 429.1900. Compound (*E*)-**29** had: 1H NMR δ 1.28 (t, *J* = 7.1 Hz, 3, CH3), 1.35 & 1.61 $(2 \times s, 2 \times 3, 2 \times CH_3)$, 1.85-1.95 (m, 2, H8/8'), 2.38 (t, $J_{9-8/8'} = 7.2$ Hz, 2, H9/9'), 2.39-2.50 (m, 2, H7/7'), 4.15 (q, *J* = 7.1 Hz, 2, CH₂), 4.69 (d, *J*₂₋₁ = 3.8 Hz, 1, H2), 4.93 ("dt", $J_{4-5} = 9.3$ Hz, $J_{4-F/3} = 2.5$ Hz, 1, H4), 5.30 (dd, $J_{5-F} = 18.6$ Hz, $J_{5-4} = 9.4$ Hz, H5), 5.38 (d, *J*3-4 = 2.9 Hz, 1, H3), 6.00 (d, *J*1-2 = 3.8 Hz, 1, H1), 7.48-8.04 (m, 5, Ar); 19F NMR δ −94.53 ("q", *J* = 22.1 Hz). HRMS (AP-FAB⁺) m/z calcd for C₂₂H₂₇FO₇Li (MLi⁺) 429.1910; found 429.1903.

Ethyl 3-*O***-Benzoyl-5,6,7,8,9-pentadeoxy-6-fluoro-1,2-***O***-isopropylidene-α-D-***ribo***-dec-5(***E/Z***) enofuranuronate (30)**

Treatment of **27** (42 mg, 0.097 mmol; *E/Z*, 3∶2) with Pd[P(Ph)3]4 (22 mg, 0.01 mmol) and 4 ethoxy-4-oxobutylzinc bromide (0.5M/THF; 0.30 mmol, 0.60 mL) as described for **29** followed by column chromatography ($10 \rightarrow 40\%$ EtOAc/hexanes) gave (Z)-30 (22 mg , 54% ; 90% based on the conversion of *E*-isomer), (*E*)-**30** (5 mg, 12%; 30% based on the conversion of the *Z*isomer), and more polar 3-*O*-debenzoylated-(*Z*)-**27** (3 mg, 10%). Compound (*Z*)-**30** had: 1H NMR δ 1.24 (t, *J* = 7.1 Hz, 3, CH₃), 1.37 & 1.60 (2 × s, 2 × 3, 2 × CH₃), 1.84 ("quint", *J*_{8-7/7'/9/9' = 7.4 Hz, 2, H8/8'), 2.25 (dt, *J*_{7-F} = 17.6 Hz, *J*_{7-8/8'} = 7.5 Hz, 2, H7/7'), 2.32 (t,} *J*9-8/8′ = 7.4 Hz, 2, H9/9′), 4.09 (q, *J* = 7.1 Hz, 2, CH2), 4.72 (dd, *J*3-4 = 9.2 Hz, *J*3-2 = 4.7 Hz, 1, H3), 4.75 (dd, *J*5-F = 35.0 Hz, *J*5-4 = 8.9 Hz, 1, H5), 4.95 (t, *J*2-1/3 = 4.3 Hz, 1, H2), 5.19 (t, *J*4-3/5 = 9.1 Hz, 1, H4), 5.89 (d, *J*1-2 = 3.8 Hz, 1, H1), 7.48-8.09 (m, 5, Ar); 13C NMR δ 14.60 $(CH₃), 21.45 (C8), 26.95 \& 27.00 (CMe₂), 31.63 (d, ²J_{7-F} = 26.5 Hz, C7), 33.32 (C9), 60.79$ $(CH₂)$, 71.39 (d, ³ $J_{4-F} = 6.3$ Hz, C4), 77.54 (C2), 73.63 (C3), 103.40 (d, ² $J_{5-F} = 11.8$ Hz, C5), 104.39 (C1), 113.52 (*C*Me2), 128.84 (Bz), 129.71 (Bz), 130.33 (Bz), 133.77 (Bz), 164.13 (d, ¹ *J*6-F = 272.7 Hz, C6), 165.33 (Bz), 173.30 (C10); 19F NMR δ −102.14 (dt, *J*F-H5 = 34.1 Hz, $J_{\text{F-H7/7'}} = 17.6 \text{ Hz}$). HRMS (AP-ESI) m/z calcd for $\text{C}_{22}\text{H}_{28}\text{FO}_7$ (MH⁺) 423.1814; found 423.1815. Compound (*E*)-**30** had: 1H NMR δ 1.26 (t, *J* = 7.1 Hz, 3, CH3), 1.36 & 1.61 (2 × s, 2×3 , $2 \times CH_3$), 1.90 ("quint", $J_{8-7/7/9/9}$ = 6.9 Hz, 2, H8/8'), 2.38 (t, $J_{9-8/8'}$ = 6.9 Hz, 2, H9/9'), 2.50 (dt, *J*7-F = 23.0 Hz, *J*7-8/8' = 7.3 Hz, 2, H7/7′), 4.14 (q, *J* = 7.1 Hz, 2, CH2), 4.75-4.80 (m,

2, H3/4), 4.95 (t, $J_{2-1/3}$ = 4.0 Hz, 1, H2), 5.17 (dd, J_{5-F} = 19.2 Hz, J_{5-4} = 8.7 Hz, 1, H5), 5.88 (d, J_{1-2} = 3.8 Hz, 1, H1), 7.48-8.09 (m, 5, Ar); ¹³C NMR δ 14.63 (CH₃), 21.99 (C8), 26.92 & 27.00 (CMe₂), 28.49 (d, ²J_{7-F} = 27.0 Hz, C7), 33.67 (C9), 60.78 (CH₂), 73.65 (d, ³J_{4-F} = 14.7 Hz, C4), 77.50 (C2), 77.61 (C3), 104.58 (d, ²J_{5-F} = 25.8 Hz, C5), 104.37 (C1), 113.39 (*C*Me₂), 128.88 (Bz), 129.64 (Bz), 130.25 (Bz), 133.83 (Bz), 165.15 (d, ¹J_{6-F} = 256.5 Hz, C6), 165.99 (Bz), 173.21 (C10); ¹⁹F NMR δ - 94.73 ("q", *J*_{F-H5/7/7'} = 22.8 Hz). HRMS (AP-ESI) *m/z* calcd for C₂₂H₂₈FO₇ (MH⁺) 423.1814; found 423.1819. The 3-O-debenzoylated-(*Z*)-27 had: ¹H NMR δ 1.27 & 1.58 (2 × s, 2 × 3, 2 × CH₃), 3.82-3.84 (m, 1, H3), 4.18 (t, *J*_{4-3/5} = 8.8 Hz, 1, H4), 4.60-4.62 (m, 1, H2), 5.61 (dd, $J_{5-F} = 15.1$ Hz, $J_{5-4} = 8.9$ Hz, 1, H5), 5.83 (d, *J*₁₋₂ = 3.7 Hz, 1, H1); ¹³C NMR δ 26.99 & 27.07 (C*Me*₂), 76.63 (C3), 78.56 (C2), 80.47 $(d, {}^{3}J_{4-F} = 8.0 \text{ Hz}, C4)$, 104.16 (C1), 115.85 $(d, {}^{1}J_{6-F} = 344.1 \text{ Hz}, C6)$, 113.70 (*CMe*₂), 115.85 $(d, {}^{2}J_{5-F} = 15.7 \text{ Hz}, \text{C5})$; ¹⁹F NMR δ –56.50 (d, $J_{\text{F-H5}} = 15.1 \text{ Hz}$); MS m/z 331 (100, MH⁺).

Methyl 5,6,7,8,9-Pentadeoxy-6-fluoro-1,2-*O***-isopropylidene-α-D-***xylo***-dec-5(***Z***) enofuranuronate (31)**

Compound (Z) -29 (26 mg, 0.062 mmol) was dissolved in MeOH (6 mL) and saturated NH₃/ MeOH (3 mL) was added at 0 °C (ice bath). The resulting mixture was stirred for 48 h (0 °C \rightarrow ambient temperature). The volatiles were evaporated and the residue was column chromatographed (15 \rightarrow 50% EtOAc/hexanes) to give 31 (14 mg, 74%): ¹H NMR δ 1.35 & 1.55 (2 × s, 2 × 3, 2 × CH3), 1.90 (quint, *J*8-7/7'/9/9' = 7.2 Hz, 2, H8/8′), 2.23-2.40 (m, 2, H7/7′), 2.41 (t, *J*9-8/8′ = 7.2 Hz, 2, H9/9′), 3.70 (s, 3, CH3), 4.17 (d, *J*3-4 = 2.6 Hz, 1, H3); 4.59 (d, *J*₂₋₁ = 3.7 Hz, 1, H₂), 4.84 (dd, *J*_{5-F} = 37.6 Hz, *J*₅₋₄ = 7.7 Hz, 1, H₅), 5.08 ("dm", *J*₄₋₅ = 7.6 Hz, 1, H4), 5.95 (d, *J*₁₋₂ = 3.7 Hz, 1, H1); ¹³C NMR δ 21.40 (C8), 26.59 & 27.12 (C*Me*₂), 31.65 (d, ²J_{7-F} = 26.6 Hz, C7), 33.29 (C9), 52.08 (CH₃), 75.29 (d, ³J_{4-F} = 4.9 Hz, C4), 76.74 $(d, {}^{4}J_{3-F} = 1.0 \text{ Hz}, \text{C3}), 85.51 \text{ (C2)}, 101.15 \text{ (d, } {}^{2}J_{5-F} = 11.0 \text{ Hz}, \text{C5}), 104.74 \text{ (C1)}, 112.08$ (*C*Me2), 161.86 (d, ¹ *J*6-F = 261.8 Hz, C6), 174.02 (C10); 19F NMR δ −100.23 (dt, *J*F-H5 = 38.0 Hz, $J_{\text{F-H7}}$ = 18.0 Hz); MS m/z 305 (100, MH⁺).

Methyl 5,6,7,8,9-Pentadeoxy-6-fluoro-1,2-*O***-isopropylidene-α-D-***ribo***-dec-5(***Z***) enofuranuronate (32)**

Saturated NH3/MeOH (3 mL) was added to a solution of (*Z*)-**30** (26 mg, 0.062 mmol) in MeOH (3 mL) and the mixture was stirred at 0 \degree C for 48 h to ambient temperature. The volatiles were evaporated and the residue was column chromatographed ($15 \rightarrow 60\%$ EtOAc/hexanes) to give **32** (13 mg, 69%): ¹H NMR δ 1.39 & 1.62 (2 × s, 2 × 3, 2 × CH₃), 1.90 (quint, $J_{8-7/7'}/9/9' = 7.3$ Hz, 2, H8/8'), 2.31 (dt, *J*_{7-F} = 17.6 Hz, *J*_{7-8/8}^{*c*} = 7.4 Hz, 2, H7/7'), 2.40 (t, *J*_{9-8/8}^{*c*} = 6.9 Hz, 2, H9/9′), 3.75 (s, 3, CH3), 4.56-4.72 (m, 4, H2/3/4/5), 5.82 (d, *J*1-2 = 3.9 Hz, 1, H1); 13C NMR δ 21.44 (C8), 26.81 & 26.94 (C*Me*2), 31.74 (d, ² *J*7-F = 26.1 Hz, C7), 33.20 (C9), 52.07 $(CH₃), 73.94$ (d, $³J_{4-F} = 5.1$ Hz, C4), 76.80 (C2), 78.62 (C3), 103.63 (d, $²J_{5-F} = 11.6$ Hz, C5),</sup></sup> 104.00 (C1), 113.07 (*C*Me₂), 163.91 (d, ¹J_{6-F} = 262.70 Hz, C6), 173.90 (C10); ¹⁹F NMR δ -100.23 (dt, *J*_{F-H5} = 37.1 Hz, *J*_{F-H7/7}′ = 18.1 Hz). HRMS (AP-ESI) m/z calcd for C₁₄H₂₂FO₆ (MH+) 305.1395; found 305.1396.

Methyl 5,6,7,8,9-Pentadeoxy-6-fluoro-α/β-D-*xylo***-dec-5(***Z***)-enofuranuronate (33)**

A solution of **31** (17 mg, 0.056 mmol) in TFA/H₂O (9:1, 3 mL) was stirred for 45 min at 0 $^{\circ}$ C and was evaporated and coevaporated [toluene $(3\times)$, CH₃CN $(2\times)$]. The residue was dissolved in H₂O and the aqueous layer was extracted with ether $(2\times)$. The water layer was evaporated to give **33** (9 mg, 61%; α/β, 1:1): ¹H NMR (MeOH- d_4) δ 1.81-1.94 (m, 2, H8/8'), 2.27-2.38 (m, 2, H7/7'), 2.38-2.45 (m, 2, H9/9'), 3.67 (m, 3, CH₃), 3.92 (dd, $J_{3-4} = 3.8$ Hz, $J_{3-2} = 1.8$ Hz, 0.5, H3), 3.97 (dd, $J_{3-4} = 3.9$ Hz, $J_{3-2} = 2.6$ Hz, 0.5, H3), 4.00-4.04 (m, 1, H2), 4.91 (d, $J_{4-5} =$ 8.9 Hz, 0.5, H4), 4.99 (d, *J*4-5 = 9.1 Hz, 0.5, H4), 5.04-5.12 (m, 1, H5), 5.08 (s, 0.5, H1β), 5.37 (d, $J_{1-2} = 4.0$ Hz, 0.5, H1α); ¹⁹F NMR δ –106.29 (dt, $J_{\rm F-H5} = 37.2$ Hz, $J_{\rm F-H7} = 17.6$ Hz, 0.5F),

−106.87 (dt, *J*F-H5 = 37.8 Hz, *J*F-H7 = 18.2 Hz, 0.5F). HRMS (AP-ESI) *m/z* calcd for $C_{11}H_{18}FO_6$ (MH⁺) 265.1082; found 265.1088.

Methyl 5,6,7,8,9-Pentadeoxy-6-fluoro-α/β-D-*ribo***-dec-5(***Z***)-enofuranuronate (34)**

A solution of **32** (12 mg, 0.04 mmol) in TFA/H₂O (9:1, 3 mL) was stirred for 30 min at 0 $^{\circ}$ C and was evaporated and coevaporated [toluene $(3\times)$]. The residue was dissolved in H₂O and the aqueous layer was extracted with ether $(2\times)$. The water layer was evaporated to give 34 (8) mg, 76%; α/β, 3[:]7): ¹H NMR (D₂O) δ 1.72-1.76 (m, 2, H8/8'), 2.20 (dt, *J*_{7-F} = 18.1 Hz, *J*7-8/8′ = 8.4 Hz, 2, H7/7′), 2.31-2.38 (m, 2, H9/9′), 3.59 (d, *J*2-3 = 2.4 Hz, 0.7, H2), 3.82-3.85 (m, 0.3, H2), 3.90-3.93 (s, 0.7, H3), 3.90-4.06 (m, 4.3, H3/H4/CH3), 4.64-4.77 (m, 1, H5), 5.12 (s, 0.7, H1), 5.28 (d, *J*1-2 = 3.7 Hz, 0.3, H1); 19F NMR δ −104.06 (dt, *J*F-H5 = 36.4 Hz, *J*F-H7/7′ = 17.8 Hz, 0.3, α); −105.04 (dt, *J*F-H5 = 35.8 Hz, *J*F-H7/7′ = 18.8 Hz, 0.7F, β). HRMS (AP-ESI) m/z calcd for $C_{11}H_{18}FO_6$ (MH⁺) 265.1082; found 265.1090.

(*E/Z***)-3,5,6-Trideoxy-6-fluoro-1,2-***O***-isopropylidene-6-phenylsulfonyl-α-D-***erythro***-hex-5 enofuranose (35)**

Step (a). Treatment of diacetone 3-deoxyglucose²² (204 mg, 0.83 mmol) with H_5IO_6 (228 mg, 1.00 mmol), as described for **10** (Step *a*, except no aqueous workup was performed) gave 3 deoxy-1,2-*O*-isopropylidene-α-D-*erythro*-pentdialdo-1,4-furanose [~85% pure; 1H NMR δ 9.75 (d, $J_{5-4} = 4.8$ Hz, H5)] which was directly used in the next step. Step (b) Treatment of the crude aldehyde with diethyl fluoro(phenylsulfonyl)methylphosphonate²⁰ (297 mg, 0.96 mmol) and LHMDS (0.96 mL, 0.96 mmol), as described for **20**, gave **35** (150 mg, 48%; *E/Z*, 2∶1) as an inseparable mixture of isomers. Compound (*E*)-**35** had: 1H NMR δ 1.30 & 1.47 (2 \times s, 2 \times 3, 2 \times CH₃), 1.71 (ddd, J_{3-4} = 10.9 Hz, $J_{3-3'}$ = 15.5 Hz, J_{3-2} = 4.6 Hz, 1, H3), 2.28 (dd, *J*3′-4 = 4.5 Hz, *J*3′-3 = 13.4 Hz, 1, H3′), 4.77 (t, *J*2-1/3 = 4.0 Hz, 1, H2), 4.93-4.97 (m, 1, H4), 5.85 (d, *J*₁₋₂ = 3.6 Hz, 1, H1), 6.31 (dd, *J*_{5-F} = 32.4 Hz, *J*₅₋₄ = 7.5 Hz, 1, H5), 7.56-7.97 (m, 5, Ar); ¹³C NMR δ 26.42 & 27.00 (CMe₂), 39.25 (d, ⁴J_{3-F} = 2.0 Hz, C3), 71.26 (d, ³J_{4-F} = 2.4 Hz, C4), 80.69 (C2), 105.90 (C1), 112.01 (*C*Me₂), 116.77 (d, ²J_{5-F} = 3.7 Hz, C5), 129.17 (Ph), 129.91 (ph), 135.08 (Ph), 137.22 (Ph), 155.54 (d, ¹ *J*6-F = 301.9 Hz, C6); 19F NMR δ −122.72 (d, $J_{\text{F-H5}}$ = 32.5 Hz, 0.66). Compound (*Z*)-35 had: ¹H NMR δ 1.34 & 1.60 (2 × s, 2 × 3, 2 × CH₃), 1.68 (ddd, *J*₃₋₄ = 10.5 Hz, *J*₃₋₃^{*,*} = 15.2 Hz, *J*₃₋₂ = 4.7 Hz, 1, H3), 2.46 (ddd, *J*_{3'-4} = 4.6 Hz, *J*3′-3 = 13.2 Hz, 1, H3′), 4.79 (t, *J*2-1/3 = 3.9 Hz, 1, H2), 5.71 (ddd, *J*4-5 = 8.7 Hz, *J*4-3 = 10.6 Hz, $J_{4-3'} = 4.5$ Hz, 1, H4), 5.84-5.85 (m, 1, H1), 5.86 (dd, $J_{5-F} = 20.1$ Hz, $J_{5-4} = 8.6$ Hz, H5), 7.56-7.97 (m, 5, Ar); 19F NMR δ −114.04 (d, *J*F-H5 = 20.0 Hz, 0.33); MS *m/z* 329 (100, MH^+).

(*E/Z***)-3,5,6-Trideoxy-6-fluoro-1,2-***O***-isopropylidene-6-tributylstannyl-α-D-***erythro***-hex-5 enofuranose (38)**

Treatment of **35** (128 mg, 0.39 mmol) with Bu3SnH (0.21 mL, 228 mg, 0.78 mmol,) and AIBN (481 mg, 0.29 mmol), as described for **23**, gave **38** (83 mg, 44%; *E/Z*, 1∶1): 19F NMR δ −92.83 (d, *J*F-H5 = 33.9 Hz, 84% of 0.50F, *Z*), −92.83 (ddd, *J*F-Sn = 230.4 Hz, *J*F-H5 = 34.7 Hz, *J*F-H4 = 5.2 Hz 16% of 0.50F, *Z*), −96.75 (d, *J*F-H5 = 52.7 Hz, 84% of 0.50F, *E*), −96.75 (ddd, *J*F-Sn $= 221.1$ Hz, $J_{\text{F-H5}} = 52.7$ Hz, $J_{\text{F-H4}} = 4.9$ Hz, 16% of 0.50F, *E*); MS m/z 479 (100, MH+, ¹²⁰Sn), 477 (73, MH+, ¹¹⁸Sn), 475 (48, MH+, ¹¹⁶Sn). Compound (*E*)-**38** had: 1H NMR δ 0.98-1.70 (m, 34, 3 × Bu/2 × CH3/H3), 2.26 (dd, *J*3′-4 = 4.3 Hz, *J*3′-3 = 13.4 Hz, 1, H3), 4.45-4.55 (m, 1, H4), 4.75 (t, $J_{2-1/3} = 4.2$ Hz, 1, H2), 4.96 (dd, $J_{5-F} = 52.9$ Hz, $J_{5-4} = 7.5$ Hz, 1, H5), 5.83 (d, *J*1-2 = 3.8 Hz, 1, H1); 13C NMR δ 10.31 (Bu), 11.00 (Bu), 13.97 (Bu), 27.04 & 27.08 (CMe₂), 27.50 (Bu), 40.19 (d, ²J_{3-F} = 1.6 Hz, C3), 71.25 (d, ³J_{4-F} = 17.3 Hz, C4), 81.06 (C2), 105.47 (C1), 111.41 (CMe₂), 123.42 (d, ²J_{5-F} = 3.7 Hz, C5), 174.29 (d, ¹J_{5-F} = 323.5 Hz, C6). Compound (*Z*)-38 had: ¹H NMR δ 0.98-1.70 (m, 34, 3 \times Bu/2 \times CH₃/H3), 2.11 (dd, *J*3′-4 = 4.3 Hz, *J*3′-3 = 13.4 Hz, 1, H3′), 4.73 (t, *J*2-1/3 = 4.2 Hz, 1, H2), 5.21 (ddd, *J*4-5 =

7.5 Hz, *J*4-3 = 4.4 Hz, *J*4-3′ = 15.2 Hz, 1, H4), 5.81 (d, *J*1-2 = 3.7 Hz, 1, H1), 5.84 (dd, *J*5-F = 34.2 Hz, *J*5-4 = 9.2 Hz, 1, H5); 13C NMR δ 10.31 (Bu), 11.00 (Bu), 13.97 (Bu), 27.04 & 27.08 (CMe_2) , 27.50 (Bu), 41.00 (d, ²J_{3-F} = 1.8 Hz, C3), 74.73 (d, ³J_{4-F} = 21.7 Hz, C4), 80.83 (C2), 105.69 (C1), 111.17 (*C*Me₂), 123.26 (d, ²J_{5-F} = 8.1 Hz, C5), 177.16 (d, ¹J_{6-F} = 316.5 Hz, C6).

(*E/Z***)-5,6-Dideoxy-6-fluoro-1,2-***O***-isopropylidene-α-D-***xylo***-hex-5-enofuranose (39)**

Step (a). Compound **23** (200 mg, 0.34 mmol; *E/Z*, 7∶3) was dissolved in saturated NH3/MeOH (20 mL) and the resulting solution was stirred overnight at ambient temperature. The volatiles were evaporated to give **36** in quantitative yield of sufficient purity to use in the subsequent reaction. Step (b). Compound **36** (crude from step *a*, 0.34 mmol) was dissolved in NH3/MeOH (20 mL) and the resulting mixture was heated in a pressure Ace tube at 65 \degree C for 18 h. The volatiles were evaporated and the residue was column chromatographed (hexanes/EtOAc, 8∶2 → 3∶7) to give (*E*)-**39** (20 mg, 29% from **23**) and (*Z*)-**39** (33 mg, 48% from **23**). Compound (*E*)-39 had: ¹H NMR δ 1.35 & 1.58 (2 × s, 2 × 3, 2 × CH₃), 1.78 (br s, 1, OH3), 4.32 (d, J_{3-4} $= 2.5$ Hz, 1, H3), 4.59 (d, $J_{2-1} = 3.7$ Hz, 1, H2), 4.69 ("dm", $J_{4-5} = 7.0$ Hz, 1, H4), 5.53 (ddd, *J*_{5-F} = 18.1 Hz, *J*₅₋₆ = 11.2 Hz, *J*₅₋₄ = 7.1 Hz, 1, H5), 5.94 (d, *J*₁₋₂ = 3.7 Hz, 1, H1), 6.86 (ddd, *J*6-F = 82.9 Hz, *J*6-5 = 11.2 Hz, *J*6-4 = 1.0 Hz, 1, H6); 13C NMR δ 26.47 & 27.06 (C*Me*2), 76.48 $(d, {}^4J_{3-F} = 2.0 \text{ Hz}, \text{C3}), 76.80 (d, {}^3J_{4-F} = 12.6 \text{ Hz}, \text{C4}), 85.31 (C2), 104.87 (C1), 106.20$ $(d, {}^{2}J_{5-F} = 13.7 \text{ Hz}, \text{C5}), 112.25 \text{ } (\text{CMe}_2), 153.79 \text{ } (d, {}^{1}J_{6-F} = 262.6 \text{ Hz}, \text{C6}); {}^{19}\text{F} \text{ NMR } \delta - 122.18$ (dd, $J_{F-H5} = 17.8$ Hz, $J_{F-H6} = 82.9$ Hz). Compound (*Z*)-39 had: 1.35 & 1.54 (2 × s, 2 × 3, 2 × CH₃), 1.81 (br s, 1, OH3), 4.22 (br s, 1, H3), 4.58 (d, $J_{2-1} = 3.7$ Hz, 1, H2), 5.07 (ddd, $J_{5-F} =$ 40.1 Hz, *J*5-6 = 4.9 Hz, *J*5-4 = 7.5 Hz, 1, H5), 5.12-5.15 (m, 1, H4), 5.96 (d, *J*1-2 = 3.7 Hz, 1, H1), 6.63 (dd, *J*_{6-F} = 82.7 Hz, *J*₆₋₅ = 4.8 Hz, *J*₆₋₄ = 1.2 Hz, 1, H6); ¹³C NMR δ 26.57 & 27.10 (CMe_2) , 74.49 (d, ³ J_{4-F} = 5.1 Hz, C4), 76.74 (d, ⁴ J_{3-F} = 1.9 Hz, C3), 85.47 (C2), 104.80 (C1), 106.24 (C5), 112.24 (*C*Me₂), 150.20 (d, ¹J_{6-F} = 265.2 Hz, C6); ¹⁹F NMR δ −121.02 (dd, $J_{\text{F-H5}} = 41.1 \text{ Hz}, J_{\text{F-H6}} = 83.3 \text{ Hz}$). MS (APCI⁺) m/z 205 (100, MH⁺). Anal. Calcd for $C_9H_{13}FO_4$ (204.19): C, 52.94; H, 6.42. Found: C, 53.19; H, 6.63.

(*E/Z***)-5,6-Dideoxy-6-fluoro-1,2-***O***-isopropylidene-α-D-***ribo***-hex-5-enofuranose (40)**

Step (a). Compound **24** (200 mg, 0.34 mmol; *E/Z*, 1∶1) was dissolved in NH3/MeOH (20 mL) and stirred overnight at ambient temperature. The volatiles were evaporated to give **37** in quanitative yield of sufficient purity to use in the subsequent step. Step (b). Treatment of **37** (crude, 0.34 mmol) with NH₃/MeOH (20 mL) at 65 °C, as described for **39**, gave unchanged **37** (17 mg, 10% from **24**; *E/Z*, 2∶3) and **40** as separable isomers (*E*; 22 mg, 32% from **24**) and (*Z*; 18 mg, 26% from 24). Compound (*E*)-40 had: ¹H NMR δ 1.40 & 1.60 (2 × s, 2 × 3, 2 × CH3), 2.37 (d, *J*OH3-3 = 10.0 Hz, 1, OH3), 3.70-3.73 (m, 1, H3), 4.12 (t, *J*4-5/3 = 8.3 Hz, 1, H4), 4.60 (t, *J*2-1/3 = 4.6 Hz, 1, H2), 5.53 (ddd, *J*5-F = 17.1 Hz, *J*5-6 = 11.2 Hz, *J*5-4 = 7.8 Hz, 1, H5), 5.84 (d, $J_{1-2} = 3.9$ Hz, 1, H1), 6.82 (ddd, $J_{6-F} = 82.4$ Hz, $J_{6-5} = 11.1$ Hz, $J_{6-4} = 0.7$ Hz, 1, H6); ¹³C NMR δ 26.75 & 26.84 (CMe₂), 76.55 (d, ³J_{4-F} = 17.0 Hz, C4), 76.49 (C3), 78.57 (C2), 104.08 (C1), 109.44 (d, $^2J_{5-F} = 12.9$ Hz, C5), 113.11 (*CMe₂)*, 152.81 (d, $^1J_{6-F} = 262.1$ Hz, C6); 19F NMR δ −123.67 (dd, *J*F-H5 = 17.1 Hz, *J*F-H6 = 82.5 Hz). Compound (*Z*)-**40** had: 1.40 & 1.60 ($2 \times s$, 2×3 , $2 \times CH_3$), 2.39 (d, $J_{OHA-3} = 11.0$ Hz, 1, OH3), 3.74 (ddd, $J_{3-OHA} =$ 10.9 Hz, *J*3-4 = 8.9 Hz, *J*3-2 = 5.1 Hz, 1, H3), 4.60 (t, *J*2-1/3 = 4.5 Hz, 1, H2), 4.70 (t, *J*4-3/5 = 8.8 Hz, 1, H4), 4.89 (ddd, *J*5-F = 40.0 Hz, *J*5-6 = 4.9 Hz, *J*5-4 = 8.9 Hz, 1, H5), 5.84 (d, *J*1-2 = 3.9 Hz, 1, H1), 6.69 (ddd, *J*6-F = 82.7 Hz, *J*6-5 = 4.9 Hz, *J*6-4 = 0.8 Hz, 1, H6); 13C NMR δ 26.79 & 26.92 (CMe₂), 73.18 (d, ³J_{4-F} = 5.1 Hz, C4), 76.74 (d, ⁴J_{3-F} = 2.0 Hz, C3), 78.58 (C2), 104.20 (C1), 108.69 (${}^{2}J_{5-F}$ = 1.9 Hz, C5), 113.17 (CMe₂), 153.71 (d, ${}^{1}J_{6-F}$ = 265.8Hz, C6); ¹⁹F NMR δ −123.90 (dd, *J*_{F-H5} = 40.1 Hz, *J*_{F-H6} = 82.6 Hz): MS (APCI⁺) *m/z* 205 (100, MH⁺). Anal. Calcd for C₉H₁₃FO₄ (204.19): C, 52.94; H, 6.42. Found: C, 53.07; H, 6.67.

(*E/Z***)-3,5,6-Trideoxy-6-fluoro-1,2-***O***-isopropylidene-α-D-***erythro***-hex-5-enofuranose (41)**

Treatment of **38** (100 mg, 0.21 mmol; E/Z , 1:1) with NH₃/MeOH (15 mL) and C_sF (51 mg, 0.33 mmol) at 65 °C for 4 h, as described for **39** (Step b), gave **41** (16 mg, 40%; *E/Z*, ~45∶55): 19F NMR δ −124.79 (dd, *J*F-H5 = 41.8 Hz, *J*F-H6 = 83.0 Hz, 0.55F), −125.95 (dd, *J*_{F-H5} = 16.7 Hz, *J*_{F-H6} = 82.9 Hz, 0.45F); MS (APCI⁺) m/z 189 (100, MH⁺). Compound (*E*)-41 had: ¹H NMR δ 1.35 & 1.55 ($2 \times s$, 2×3 , $2 \times CH_3$), 1.55-1.68 (m, 1, H3), 2.20 (dd, *J*3′-3 = 13.5 Hz, *J*3′-4 = 4.3 Hz, 1, H3′), 4.62 (ddd, *J*4-3 = 11.5 Hz, *J*4-5 = 8.2 Hz, *J*4-3′ = 4.3 Hz, 1, H4), 4.75-4.79 (m, 1, H2), 5.42 (ddd, $J_{5-F} = 16.8$ Hz, $J_{5-6} = 11.2$ Hz, $J_{5-4} = 8.3$ Hz, 1, H5), 5.82-5.84 (m, 1, H1), 6.80 (dd, *J*6-F = 82.7 Hz, *J*6-5 = 11.2 Hz, 1, H6). Compound (*Z*)-**41** had: 1.28 & 1.57 (2 × s, 2 × 3, 2 × CH3), 1.55-1.68 (m, 1, H3), 2.28 (dd, *J*3′-3 = 13.5 Hz, *J*3′-4 = 4.3 Hz, 1, H3'), 4.75-4.79 (m, 1, H2), 4.92 (ddd, $J_{5-F} = 41.3$ Hz, $J_{5-4} = 8.1$ Hz, $J_{5-6} = 4.8$ Hz, 1, H5), 5.13 (ddd, *J*4-3 = 11.5 Hz, *J*4-5 = 8.0 Hz, *J*4-3' = 4.1 Hz, 1, H4), 5.82-5.84 (m, 1, H1), 6.53 $(dd, J_{6-F} = 82.8 \text{ Hz}, J_{6-5} = 4.8 \text{ Hz}, 1, \text{ H6}.$

(*E***)-5,6-Dideoxy-6-fluoro-α/β-D-***xylo***-hex-5-enofuranose (42)**

A solution of (E) -39 (13 mg, 0.064 mmol) in TFA/H₂O (9:1; 3 mL) was stirred for 50 min at 0 °C (ice bath). The volatiles were evaporated, coevaporated [toluene (3 \times) and MeCN (2 \times)], and the residue was flash column chromatographed ($50 \rightarrow 95\%$ EtOAc/hexanes) to give **42** (4 mg, 38%; α/β, 1∶1): 1H NMR (MeOH-*d*4) δ 3.91-4.03 (m, 2 H2/3), 4.56-4.63 (m, 1, H4), 5.07 (s, 0.5, H1β), 5.37 (d, $J_{1-2} = 4.0$ Hz, 0.5, H1α), 5.51 (ddd, $J_{5-F} = 17.8$ Hz, $J_{5-6} = 11.2$ Hz, *J*₅₋₄ = 8.9 Hz, 0.5, H5), 5.65 (ddd, *J*_{5-F} = 17.9 Hz, *J*₅₋₆ = 11.1 Hz, *J*₅₋₄ = 8.9 Hz, 0.5, H5), 6.87 (dd, $J_{6-F} = 84.0$ Hz, $J_{6-5} = 11.0$ Hz, 0.5, H6), 6.90 (dd, $J_{6-F} = 83.7$ Hz, $J_{6-5} = 11.0$ Hz, 0.5, H6); ¹³C NMR (MeOH-*d*₄) δ 75.29 (d, ³J_{4-F} = 13.8 Hz, C4), 77.00 & 77.20 (C3), 77.73 (C2), 78.17 (d, ³ *J*4-F = 13.7 Hz, C4), 81.26 (C2), 96.72 (C1α), 103.17 (C1β), 108.71 (d, ² *J*5-F = 12.0 Hz, C5), 109.32 (d, ²J_{5-F} = 11.7 Hz, C5), 152.14 (d, ¹J_{6-F} = 258.7 Hz, C6), 152.24 (d, ¹J_{6-F} = 259.2 Hz, C6); 19F NMR (MeOH-*d*4) δ - 126.55 (dd, *J*F-H5 = 17.8 Hz, *J*F-6 = 83.9 Hz, 0.5F), −126.85 (dd, *J*F-H5 = 18.1 Hz, *J*F-H6 = 84.0 Hz, 0.5F); MS (APCI−) *m/z* 163 (100, MH−).

Analogous treatment of **39** (*E/Z*, 1∶1; 20 mg, 0.040 mmol) gave **42** (5 mg, 76%) as a mixture (*E/Z*, ~1∶1; α/β, ~1∶1). Compound (*E/Z*)-**37** had: 19F NMR (MeOH-*d*4) δ −126.55 (dd, *J*F-H5 = 17.3 Hz, *J*F-H6 = 84.0 Hz; *E*, 0.25, β), −126.85 (dd, ³ *J*F-H5 = 17.5 Hz, ² *J*F-H6 = 84.0 Hz; *E*, 0.25, β), −127.66 (dd, *J*F-H5 = 41.5 Hz, *J*F-H6 = 84.9 Hz; *Z*, 0.25, β), −128.34 (dd, *J*F-H5 = 42.2 Hz, *J*F-H6 = 84.8 Hz; *Z*, 0.25, α); MS(APCI−) *m/z* 163 (100, MH−).

Treatment of the crude 36 [from step (a) for the preparation of 39] with TFA/H₂O (1 h, 0 °C) followed by evaporation, coevaporation [toluene (2×) and MeCN (1×)], partition (H₂O/ethyl ether), and evaporation of the aqueous layer also gave **42** (55% from **23**, α/β, 1∶1).

(*E***)-5,6-Dideoxy-6-fluoro-α/β-D-***ribo***-hex-5-enofuranose (43)**

Treatment of (E) -40 (13 mg, 0.064 mmol) with TFA/H₂O (9:1, 3 mL), as described for 42, gave **43** (7 mg, 67%; α/β, 1∶4): 1H NMR (MeOH-*d*4) δ 3.77 (t, *J*3-2/4 = 6.1 Hz, 0.2, H3), 3.87 (d, *J*2-1 = 4.5 Hz, 0.8, H2), 4.01-4.05 (m, 1, H2α & H3β), 4.20 (t, *J*4-3/5 = 8.0 Hz, 0.8, H4), 4.30 (dd, *J*4-5 = 8.2 Hz, *J*4-3 = 6.4 Hz, 0.2, H4), 5.12 (br s, 0.8, H1), 5.28 (d, *J*1-2 = 4.1 Hz, 0.2, H1), 5.42 (ddd, *J*5-F = 17.5 Hz, *J*5-6 = 11.1 Hz, *J*5-4 = 8.3 Hz, 0.2, H5), 5.49 (ddd, *J*5-F = 17.6 Hz, *J*₅₋₆ = 11.1 Hz, *J*₅₋₄ = 8.5 Hz, 0.8, H5), 6.86 (dd, *J*_{6-F} = 83.9 Hz, *J*₆₋₅ = 11.1 Hz, 0.8, H6), 6.87 (dd, $J_{6-F} = 83.7$ Hz, $J_{6-5} = 11.0$ Hz, 0.2, H6); ¹³C NMR (MeOH- d_4) δ 70.90 (C2 α), 75.32 (d, ${}^4J_{3-F} = 2.6$ Hz, C3a), 75.5 (d, ${}^4J_{3-F} = 2.5$ Hz, C3 β), 76.13 (C2 β), 77.55 (d, ${}^3J_{4-F} = 13.6$ Hz, C4α), 77.88 (d, ³J_{4-F} = 13.7 Hz, C4β), 96.63 (C1α), 102.09 (C1β), 111.19 (d, ²J_{5-F} = 11.4 Hz, C5α), 112.65 (d, ²J_{5-F} = 10.6 Hz, C5β), 151.98 (d, ¹J_{6-F} = 258.6 Hz, C6β), 152.17 (d, ¹J_{6-F} = 258.7 Hz, C6α); 19F NMR (MeOH-*d*4) δ −128.55 (dd, *J*F-H5 = 17.4 Hz, *J*F-H6 = 83.5 Hz, 0.2F, α), −129.00 (dd, *J*F-H5 = 17.3 Hz, *J*F-H6 = 83.7 Hz, 0.8F, β); MS (APCI−) *m/z* 163 (100, $MH⁻$).

Analogous treatment of **40** (*E/Z*, 1∶1; 16 mg, 0.032 mmol) gave **43** (3 mg, 57%) as a mixture (*E/Z*, ~3∶1; α/β, ~1∶4 for *E* isomer and α/β, ~1∶15 for *Z* isomer): 19F NMR (MeOH-*d*4) δ −128.01 $(\text{dd}, J_{\text{F-H5}} = 41.3 \text{ Hz}, J_{\text{F-H6}} = 83.4 \text{ Hz}; Z, 0.02 \text{ F}, \alpha$) −128.55 (dd, $J_{\text{F-H5}} = 17.6 \text{ Hz}, J_{\text{F-H6}} = 84.2$ Hz; *E*, 0.14F, α), −129.00 (dd, *J*_{F-H5} = 17.5 Hz, *J*_{F-H6} = 83.8 Hz; *E*, 0.60F, β), −129.69 (dd, *J*F-H5 = 40.8 Hz, *J*F-H6 = 84.4 Hz; *Z*, 0.24F, β); MS(APCI−) *m/z* 163 (100, MH−).

Treatment of the crude **37** [from step (a) for the preparation of **40**] with TFA/H2O (1 h, 0 °C) followed by evaporation, coevaporation [toluene (2×) and MeCN (1×)], partition (H₂O/ethyl ether), and evaporation of the aqueous layer also gave **43** (45% from **24**, α/β, 1∶3).

(*E/Z***)-3,5,6-Trideoxy-6-fluoro-α-D-***erythro***-hex-5-enofuranose (44)**

Treatment of **38** (62 mg, 0.13 mmol; *E/Z*, 3∶2) with TFA/H2O (9∶1, 1mL; 1 h, 0 °C) followed by evaporation, coevaporation [toluene (2×) and MeCN (1×)], partition (H₂O/ethyl ether), and evaporation of the aqueous layer gave 44 (10 mg, 52%; $E/Z \sim 1:3$, α/β , ~1:4): ¹H NMR (D₂O) δ 1.85-2.10 (m, 2, H3,3'), 4.05-4.25 (m, 1, H2), 4.58-4.75 (m, 1, H4), 4.86 (ddd, *J*5-F = 41.7 Hz, *J*5-4 = 8.9 Hz, *J*5-6 = 4.7, Hz, 0.15, H5), 4.92 (ddd, *J*5-F = 41.6 Hz, *J*5-4 = 8.9 Hz, *J*5-6 = 4.7 Hz, 0.6, H5), 5.12 (s, 0.15, H1β), 5.13 (s, 0.6, H1β), 5.24 (d, *J*1-2 = 3.5 Hz, 0.05, H1α), 5.25 (d, *J*1-2 = 3.8 Hz, 0.2, H1α), 5.39 (ddd, *J*5-F = 17.4 Hz, *J*5-6 = 11.2 Hz, *J*5-4 = 9.3 Hz, 0.2, H5), 5.45 (ddd, *J*5-F = 17.6 Hz, *J*5-6 = 11.1 Hz, *J*5-4 = 9.1 Hz, 0.05, H5), 6.52 (ddd, *J*6-F = 83.7 Hz, *J*₆₋₅ = 4.7 Hz, *J*₆₋₄ = 1.0 Hz, 0.15, H₆), 6.55 (dd, *J*_{6-F} = 83.7 Hz, *J*₆₋₅ = 4.7 Hz, *J*₆₋₄ = 1.0, 0.60, H6), 6.77 (dd, *J*6-F = 83.9 Hz, *J*6-5 = 10.7 Hz, 0.05, H6), 6.78 (dd, *J*6-F = 83.9 Hz, *J*6-5 = 11.0 Hz, 0.20, H₆); ¹⁹F NMR (D₂O) δ −126.35 (dd, *J*_{F-H5} = 17.3 Hz, *J*_{F-H6} = 83.6 Hz; *E*, 0.20F, β), −126.45 (dd, *J*F-H5 = 17.2 Hz, *J*F-H6 = 82.8 Hz; *E*, 0.05F, α), −126.81 (dd, *J*F-H5 = 42.0 Hz, *J*F-H6 = 83.7 Hz; *Z*, 0.15F, α), −127.55 (dd, *J*F-H5 = 42.0 Hz, *J*F-H6 = 83.8 Hz; *Z*, 0.60F, β); HRMS (LCT-ESI) m/z : calcd for $C_6H_9FO_3$ [M + Na]⁺ 171.0433; found 171.0434.

Analogous treatment of **41** (10 mg, 0.053 mmol; *E/Z*, ~45∶55) with TFA/H2O gave **44** (5 mg, 64%; $E/Z \sim 1:2$, $\alpha/\beta \sim 1:3$).

Enzymatic assay

Inhibition assays were performed in a buffer containing 50 mM HEPES (pH 7.0), 150 mM NaCl, 150 μ M 5,5'-dithio-bis-(2-nitrobenzoic acid), 23 and various concentrations of SRH (0– 55 μ M) and inhibitors (0–1 mM). The reactions were initiated by the addition of Co²⁺substituted LuxS from *Bacillus subtilis* (final concentration 0.4–0.5 µM) and monitored continuously at 412 nm ($\varepsilon = 14150 \text{ M}^{-1} \text{ cm}^{-1}$) in a Perkin-Elmer λ 25 UV-VIS spectrophotometer at room temperature. The initial rates recorded from the early regions of the progress curves were fitted into the Lineweaver-Burk equation $1/V = K_M$ [']/(*k*cat [E]₀) × 1/ $[S] +1/(kcat [E]0)$ and the Michaelis-Menten equation $V = kcat [E]0 [S]/(K_M' + [S])$ using KaleidaGraph 3.5 to determine the inhibition pattern. K_I values were calculated from the equation $K_M' = K_M \times (1 + [I]/K_I)$, where $K_M = 2.2 \mu M$.

Acknowledgements

This work was partially supported by grants from the National Institutes of Health (S06GM08205 and R01AI062901). C.A.G and J.R were sponsored by MBRS RISE program (NIH/NIGMS; R25GM61347). C.A.G is also thankful to R. E. McNair Program for summer support. The support of US Army Research Office (W911NF-04-1-0022) in the purchase of 600 MHz NMR spectrometer is gratefully acknowledged.

References and Notes

1. Yuan, C-S.; Liu, S.; Wnuk, SF.; Robins, MJ.; Borchardt, RT. Advances in Antiviral Drug Design. De Clercq, E., editor. Vol. 2. Greenwich: JAI Press; 1996. p. 41-88. (b) Turner MA, Yang X, Yin D,

Kuczera K, Borchardt RT, Howell PL. Cell Biochem. Biophys 2000;33:101. [PubMed: 11325033] (c) Wnuk SF. Mini-Rev. Med. Chem 2001;1:307. [PubMed: 12369977]

- 2. (a) Schneyder G, Roffi M, Pin R, Flammer Y, Lange H, Eberli FR, Meier B, Turi ZG, Hess OM. N. Engl. J. Med 2001;345:1593. [PubMed: 11757505] (b) Langheinrich AC, Braun-Dullaeus RC, Walker G, Jeide I, Schiling R, Tammoscheit K, Dreyer T, Fink L, Bohle RM, Haberbosch W. Atherosclerosis 2003;171:181. [PubMed: 14644386]
- 3. (a) Lee JE, Cornel KA, Riscoe, Howell PL. Structure 2001;9:941. [PubMed: 11591349] (b) Lee JE, Cornel KA, Riscoe MK, Howell PL. J. Biol. Chem 2003;278:8761. [PubMed: 12496243]
- 4. (a) Chen X, Schauder S, Potier N, Van Dorsselaer A, Pelczer I, Bassler BL, Hughson FM. Nature 2002;415:545. [PubMed: 11823863] (b) Zhu J, Hu X, Dizin E, Pei D. J. Am. Chem. Soc 2003;125:13379. [PubMed: 14583032] (c) Zhu J, Dizin E, Hu X, Wavreille AS, Park J, Pei D. Biochemistry 2003;42:4717. [PubMed: 12705835] (d) Zhu J, Patel R, Pei D. Biochemistry 2004;43:10166. [PubMed: 15287744] (e) Alfaro JF, Zhang T, Wynn DP, Karschner EL, Zhou ZS. Org. Lett 2004;6:3043. [PubMed: 15330583] (f) Pei D, Zhu J. Curr. Opin. Chem. Biol 2004;8:492. [PubMed: 15450491] (g) Rajan R, Zhu J, Hu X, Pei D, Bell CE. Biochemistry 2005;44:3745. [PubMed: 15751951] (h) Shen G, Rajan R, Zhu J, Bell CE, Pei D. J. Med. Chem 2006;49:3003. [PubMed: 16686542]
- 5. (a) Meijler MM, Hom LG, Kaufmann GF, McKenzie KM, Sun C, Moss JA, Matsushita M, Janda KD. Angew. Chem. Int. Ed 2004;43:2106. (b) Semmelhack MF, Campagna SR, Federle MJ, Bassler BL. Org. Lett 2005;7:569. [PubMed: 15704896] (c) De Keersmaecker SCJ, Varszegi C, van Boxel N, Habel LW, Metzger K, Daniels R, Marchal K, De Vos D, Vanderleyden J. J. Biol. Chem 2005;280:19563. [PubMed: 15790567] (d) Frezza M, Soulere L, Balestrino D, Gohar M, Deshayes C, Queneau Y, Forestier C, Doutheau A. Bioorg. Med. Chem. Lett 2007;17:1428. [PubMed: 17169556]
- 6. (a) Waters CM, Bassler BL. Annu. Rev. Cell Dev. Biol 2005;21:319. [PubMed: 16212498] (b) Bassler BL, Losick R. Cell 2006;125:237. [PubMed: 16630813]
- 7. (a) Geske GD, Wezeman RJ, Siegel AP, Blackwell HE. J. Am. Chem. Soc 2005;127:12762. [PubMed: 16159245] (b) Pomianek ME, Semmelhack MF. ACS Chem. Biol 2007;2:293. [PubMed: 17518429] (c) Higgins DA, Pomianek ME, Kraml CM, Taylor RK, Semmelhack MF, Bassler BL. Nature 2007;450:883. [PubMed: 18004304]
- 8. (a) Wnuk SF, Yuan C-S, Borchardt RT, Balzarini J, De Clercq E, Robins MJ. J. Med. Chem 1994;37:3579. [PubMed: 7932585] (b) Yuan C-S, Wnuk SF, Robins MJ, Borchardt RT. J. Biol. Chem 1998;273:18191. [PubMed: 9660780] (c) Wnukx SF, Mao Y, Yuan CS, Borchardt RT, Andrei J, Balzarini J, DeClercq E, Robins MJ. J. Med. Chem 1998;41:3078. [PubMed: 9685247]
- 9. (a) Xie M, Berges DA, Robins MJJ. Org. Chem 1996;61:5178. (b) Maryanoff BE, Reitz AB. Chem. Rev 1989;89:863.
- 10. Analogous bromination-dehydrobromination strategy applied to the adenosine derivatives bearing a C5′–C6′ double bond produced a single (5′-bromo)vinyl SAH analogue: Andrei D, Wnuk SF. Org. Lett 2006;8:5093. [PubMed: 17048851]
- 11. (a) Although Pd-catalyzed cross-coupling reactions are powerful methods for the formation of new carbon-carbon bonds, 11b monocross-coupling reactions of 1,1-dihalovinyl electrophiles with C_{sp}2 and C_{sp} nucleophiles are less common¹² and monocouplings between 1,1-dihalovinyl electrophiles and $C_{\text{sp}}^{\text{sp}}$ 3 nucleophiles are scarce.¹³ de MeijereADiederichFMetal-Catalyzed Cross-Coupling Reactions2004WeinheimWiley-VCH
- 12. Negishi E-I, Hu Q, Huang Z, Qian M, Wang G. Aldrichim. Acta 2005;38:71.; references cited therein.
- 13. For successful Pd-catalyzed monoalkylation of vinyl dihalides via *trans*-selective monobutylation of 1,1-dichloro-2-phenylethene with *n*-C4H9ZnCl in 81% yield see: Minato A, Suzuki K, Tamao K. J. Am. Chem. Soc 1987;109:1257. For studies on differentiation of two chlorine atoms in stepwise double alkylation reactions of 1,1-dichloroalkenes see: (b) Tan Z, Negishi E-i. Angew. Chem. Int. Ed 2006;45:762.For Suzuki-Miyaura protocol to the selective monocoupling of 1,1-dichloroalkenes with 9-alkyl-9-BBN see: (c) Liron F, Fosse C, Pernolet A, Roulland E. J. Org. Chem 2007;72:2220. [PubMed: 17311458]
- 14. (a) Corey EJ, Fuchs PL. Tetrahedron Lett 1972;36:3769. (b) Tronchet JMJ, Bonenfant AP, Perret F, Gonzalez A, Zumwald JB, Martinez EM, Baehler B. Helv. Chim. Acta 1980;63:1181.
- 15. Robins MJ, Wnuk SF, Yang X, Yuan C-S, Borchardt RT, Balzarini J, De Clercq E. J. Med. Chem 1998;41:3857. [PubMed: 9748360]

- 16. Pd-catalyzed Negishi monoalkylation of 1-fluoro-1-(iodo, or bromo, or chloro)alkenes with alkylzincs give stereoselective access to the internal fluoroalkenes: Andrei D, Wnuk SF. J. Org. Chem 2006;71:405. [PubMed: 16388671]
- 17. For Pd-catalyzed couplings of 1-fluoro-1-haloalkenes with $C_{sp}2$ and C_{sp} nucleophiles see: (a) Chen C, Wilcoxen K, Huang CQ, Strack N, McCarthy JR. J. Fluorine Chem 2000;101:285. (b) Xu J, Burton DJ. Tetrahedron Lett 2002;43:2877–2879. (c) Xin Z, Burton DJ. J. Fluorine Chem 2001;112:317.
- 18. (a) McCarthy JR, Huber EW, Le T-B, Laskovics M, Matthews DP. Tetrahedron 1996;52:45. (b) McCarthy JR, Matthews DP, Stemerick DM, Bey P, Lippert BJ, Snyder RD, Sunkara PS. J. Am. Chem. Soc 1991;113:7439.
- 19. For other examples of synthetic sequences involving introduction of alkenes substituted with F/ SO2Ph, F/SnBu3, or F/H into amino acids or nucleoside frames see: (a) Berkowitz DB, De la Salud-Bea R, Jahng W-J. Org. Lett 2004;6:1821. [PubMed: 15151423] (b) Karukurichi KR, De la Salud-Bea R, Jahng W-J, Berkowitz DB. J. Am. Chem. Soc 2007;129:258. [PubMed: 17212389] (c) Pan Y, Calvert K, Silverman RB. Bioorg. Med. Chem 2004;12:5719. [PubMed: 15465348] (d) Shen Y. J. Organomet. Chem 2006;691:1452. (e) Rapp M, Haubrich TA, Perrault J, Mackey ZB, McKerrow JH, Chiang PK, Wnuk SF. J. Med. Chem 2006;49:2096. [PubMed: 16539398]
- 20. (a) Appel RB. Synth. Commun 1995;25:3593. (b) Wnuk SF, Bergolla LA, Garcia PI Jr. J. Org. Chem 2002;67:3065. [PubMed: 11975568] (c) Wnuk SF, Garcia PI Jr, Wang Z. Org. Lett 2004;6:2047. [PubMed: 15176815]
- 21. Wnuk SF, Lalama J, Robert J, Garmendia CA. Nucleosides Nucleotides Nucleic Acids 2007;26:1051. [PubMed: 18058535]
- 22. Robins MJ, Wilson JS, Hansske F. J. Am. Chem. Soc 1983;105:4059.
- 23. Ellman GL. Arch. Biochem. Biophys 1959;82:70. [PubMed: 13650640]

Figure 1.

Reaction pathways for SAH detoxification in eukaryotes (a) and the majority of bacteria (b). The latter is also utilized by the bacteria to produce the type 2 autoinducer.

Figure 2.

Inhibitors of LuxS enzyme (**1**, **2**).4e,h 5′-Deoxy-5′-(halomethylene)adenosine analogues (**3**, 4) which serve as suicide substrates for SAH hydrolase⁸ and targeted SRH analogues (5) in which the sulfur and C5 atoms are replaced by a vinyl unit.

Figure 3.

Inhibition of Co^{2+} -substituted *B. subtilis* LuxS by compound 44. (A) Plot of remaining LuxS activity (relative to that in the absence of inhibitor) as a function of [I]. (B) Lineweaver-Burke plot of data from part A to show the competitive inhibition mode.

Scheme 1.

(a) H_5IO_6 /EtOAc; (b) $Ph_3PCH_2CH_2CH_2CO_2Et/HMDS/THF$.

20 $X = OBz$, $Y = H$, $Z = F$
21 $X = H$, $Y = OBz$, $Z = F$
22 $X = H$, $Y = OBz$, $Z = H$

26 $X = OBz$, $Y = H$, $Z = F$
27 $X = H$, $Y = OBz$, $Z = F$
28 $X = H$, $Y = OBz$, $Z = H$

23 $X = OBz$, $Y = H$, $Z = F$
24 $X = H$, $Y = OBz$, $Z = F$
25 $X = H$, $Y = OBz$, $Z = H$

(*a*) PhSO₂CHFPO(OEt)₂ or PhSO₂CH₂PO(OEt)₂/LHMDS/THF/-78 °C; (*b*) Bu₃SnH/AIBN/ toluene/85 °C; (c) NIS/CH₂Cl₂

Scheme 5.

(*a*) NH3/MeOH/25 °C; (*b*) Bu3SnH/AIBN/toluene/85 °C;(*c*) NH3/MeOH/65 °C or NH3/ MeOH/CsF/65 °C; (d) TFA/H₂O