

# Formate as the Main Branch Point for Methylotrophic Metabolism in *Methylobacterium extorquens* AM1<sup>∇</sup>

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**In serine cycle methylotrophs, methylene tetrahydrofolate (H<sub>4</sub>F) is the entry point of reduced one-carbon compounds into the serine cycle for carbon assimilation during methylotrophic metabolism. In these bacteria, two routes are possible for generating methylene H<sub>4</sub>F from formaldehyde during methylotrophic growth: one involving the reaction of formaldehyde with H<sub>4</sub>F to generate methylene H<sub>4</sub>F and the other involving conversion of formaldehyde to formate via methylene tetrahydromethanopterin-dependent enzymes and conversion of formate to methylene H<sub>4</sub>F via H<sub>4</sub>F-dependent enzymes. Evidence has suggested that the direct condensation reaction is the main source of methylene H<sub>4</sub>F during methylotrophic metabolism. However, mutants lacking enzymes that interconvert methylene H<sub>4</sub>F and formate are unable to grow on methanol, suggesting that this route for methylene H<sub>4</sub>F synthesis should have a significant role in biomass production during methylotrophic metabolism. This problem was investigated in *Methylobacterium extorquens* AM1. Evidence was obtained suggesting that the existing deuterium assay might overestimate the flux through the direct condensation reaction. To test this possibility, it was shown that only minor assimilation into biomass occurred in mutants lacking the methylene H<sub>4</sub>F synthesis pathway through formate. These results suggested that the methylene H<sub>4</sub>F synthesis pathway through formate dominates assimilatory flux. A revised kinetic model was used to validate this possibility, showing that physiologically plausible parameters in this model can account for the metabolic fluxes observed in vivo. These results all support the suggestion that formate, not formaldehyde, is the main branch point for methylotrophic metabolism in *M. extorquens* AM1.**

*Methylobacterium extorquens* AM1 is a facultative methylotroph that has served as a model system for understanding methylotrophic metabolism for many decades (1, 3, 33). One outstanding question in methylotrophic metabolism involves the route by which C<sub>1</sub> compounds are incorporated into assimilatory metabolism. In this bacterium, methanol or methylamine is oxidized via formaldehyde and formate to CO<sub>2</sub> for energy metabolism and carbon is assimilated via the serine cycle (Fig. 1) (1, 3, 33). The entry point into the serine cycle is methylene tetrahydrofolate (H<sub>4</sub>F). In the 1960s and 1970s, it was proposed that the main route for generating methylene H<sub>4</sub>F is the spontaneous (nonenzymatic) condensation of formaldehyde with H<sub>4</sub>F (12, 19). However, a potential alternate route exists, involving the conversion of formaldehyde to formate and then to methylene H<sub>4</sub>F (Fig. 1). This set of interconversions was shown to function in the reductive direction (25, 26), and it was suggested that this pathway might contribute to production of biomass from formate (25, 30, 38). This alternate route uses an additional ATP compared to the direct condensation route (Fig. 1), suggesting that it is more expensive energetically. The relative fluxes of these two potential routes were assessed using an in vivo deuterium labeling technique (27). This study suggested that the direct condensation of

formaldehyde with H<sub>4</sub>F is the main source of methylene H<sub>4</sub>F in cells growing on methanol (27).

However, the meaning of these results remained uncertain because mutants defective in any of the three enzymes (formyl H<sub>4</sub>F ligase [FtfL], methenyl H<sub>4</sub>F cyclohydrolase [Fch] and methylene H<sub>4</sub>F dehydrogenase [MtdA]) involved in the pathway converting formate to methylene H<sub>4</sub>F cannot grow on methanol (7, 25, 26, 32). These mutant results suggested that this alternate route of methylene H<sub>4</sub>F synthesis should play a significant role in methylotrophic metabolism. The present study addressed the role of this pathway in methylotrophic growth using a combination of flux analysis and modeling approaches.

## MATERIALS AND METHODS

**Bacterial strains and growth conditions.** *M. extorquens* AM1 was routinely grown at 28°C in the minimal medium described previously (11), either in batch cultures, with 120 mM methanol or 15 mM succinate, or in chemostat cultures in a benchtop fermentor essentially as described previously (9, 37), with a dilution rate of 0.100 h<sup>-1</sup> and a substrate concentration in the feed of either 25 mM methanol or 3.7 mM succinate plus 12.5 mM methanol. Methanol was measured in the effluent as described previously (23). Wild-type chemostat cultures maintained a steady-state optical density at 600 nm (OD<sub>600</sub>) of approximately 1.0. Because of the unique growth requirements of the  $\Delta mtdA$  mutant (26), this strain was grown in chemostat cultures to a steady-state OD<sub>600</sub> of approximately 0.6 with 3.7 mM succinate plus 12.5 mM methanol. The  $\Delta ftfL$  strain (25) was grown in a batch culture on 15 mM succinate and then pelleted, washed, and exposed to 120 mM methanol for 3 to 4 h to promote induction of C<sub>1</sub> enzymes. The following antibiotic concentrations were used: kanamycin, 25 to 50  $\mu$ g/ml, and rifamycin, 25 to 50  $\mu$ g/ml.

**<sup>14</sup>C-labeling experiments.** The rate of assimilation of labeled carbon from [<sup>14</sup>C]methanol was determined by incubating cell samples (OD<sub>600</sub> between 0.2 and 1.0; high-flux samples diluted to an OD<sub>600</sub> of 0.2 to 0.3 to avoid oxygen limitation) with 1 mM labeled methanol (1.4  $\mu$ Ci/ $\mu$ mol final specific activity) in

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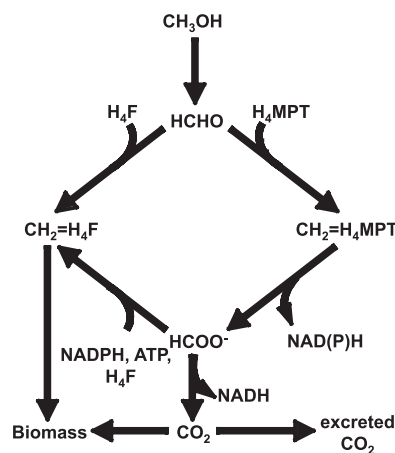


FIG. 1. Traditional scheme of methylotrophic metabolism in serine cycle methylotrophs (12, 19).  $\text{CH}_2=\text{H}_4\text{F}$ , methylene tetrahydrofolate;  $\text{CH}_2=\text{H}_4\text{MPT}$ , methylene  $\text{H}_4\text{MPT}$ .

2-ml autosampling vials (Kimble) for 12 min at 28°C and then pipetting samples onto 0.2- $\mu\text{m}$  polyvinylidene difluoride filters (Millipore), submerging the filters in scintillation fluid, and quantifying the radioactivity with one of two scintillation counters (Beckman LS3801 and Perkin-Elmer 2800TR). Experiments to determine the relative contributions of labeled methylene  $\text{H}_4\text{F}$  and labeled  $\text{CO}_2$  to labeled biomass were performed by modifying a previous method (18). Prior to incubation, cells were diluted with medium bubbled with an  $\text{N}_2\text{-O}_2$  gas mixture containing either 0%  $\text{CO}_2$  or 5%  $\text{CO}_2$  and then injected into vials containing the same gases.

**$^2\text{H}$ -labeling experiments.** Incorporation of deuterium ( $^2\text{H}$ ) into serine was quantified as previously described (27). Briefly, concentrated cell suspensions were incubated with 1 mM fully deuterated methanol ( $\text{CD}_3\text{OD}$ ) for 20 s (unless otherwise noted) and then lysed via the addition of 3 volumes of boiling ethanol and derivatized with ethyl chloroformate and trifluoroacetic anhydride (ECF/TFAA). Fragments of ECF/TFAA-derivatized serine containing either 0, 1, or 2 deuterium atoms were quantified according to their differences in mass by gas chromatography-mass spectrometry. Data were corrected for the natural abundance of heavy isotopes in the derivatized serine fragments.

**Flux calculations.** Methanol input rate for methanol-limited chemostat cultures is 41.0 nmol/(ml  $\cdot$  min). The methanol concentration in the effluent was below the detection limit ( $\sim 50 \mu\text{M}$ ) of our assay (23), indicating essentially complete consumption of the methanol. For a steady-state  $\text{OD}_{600}$  of 0.99 (the average of nine cultures), the consumption rate therefore is 41.5 nmol/(ml  $\cdot \text{OD}_{600} \cdot \text{min}$ ). This methanol consumption generates a biomass flux of 18.3 nmol/(ml  $\cdot \text{OD}_{600} \cdot \text{min}$ ), assuming 47.33% of biomass is carbon (8), using measured dry-weight values (0.278 mg biomass per ml of culture at an  $\text{OD}_{600}$  of 1.0 [X. Guo and M. Lidstrom, unpublished data]) and the chemostat growth rate of 0.1  $\text{h}^{-1}$ . Biomass fluxes were corrected for  $\text{CO}_2$  fixation using a value of 47% of biomass carbon from  $\text{CO}_2$  (see Results).

**Conversion of flux rates per culture volume to flux rates per intracellular volume.** Calculations were made assuming  $4 \times 10^8$  cells per ml of culture at an  $\text{OD}_{600}$  of 1.0 (9) and that an *M. extorquens* AM1 cell may be approximated as a cylinder with a radius of 0.4  $\mu\text{m}$  and length of 3  $\mu\text{m}$  such that the volume of a cell is approximately  $1.5 \times 10^{-15}$  liter (T. Strovas, personal communication).

**Measurement of the rate constant for the spontaneous reaction of formaldehyde with  $\text{H}_4\text{F}$ .** The reaction of formaldehyde with  $\text{H}_4\text{F}$  was monitored at pHs of 6.0, 6.5, 6.7, 7.0, and 8.0 at temperatures of 26 to 28°C.  $\text{H}_4\text{F}$  powder (Sigma) was dissolved in 100 mM potassium phosphate buffer (degassed with  $\text{N}_2$ ) and used immediately. The final concentration of  $\text{H}_4\text{F}$  was determined by measuring the  $A_{297}$ , applying an extinction coefficient of 27  $\text{mM}^{-1} \text{cm}^{-1}$  (41). Solutions of formaldehyde (5 to 20 mM) were also made fresh shortly before use by heating paraformaldehyde suspended in 100 mM potassium phosphate buffer (22). The assay was started by adding 100  $\mu\text{l}$  of 5 to 20 mM formaldehyde to 900  $\mu\text{l}$  of 10 to 20  $\mu\text{M}$   $\text{H}_4\text{F}$  in a 1-ml cuvette. An increase in  $A_{295}$  was taken to reflect formation of methylene  $\text{H}_4\text{F}$ , and the slope of this increase ( $\Delta A_{295}/\text{min}$ ) and the corresponding extinction coefficient of 3.0  $\text{mM}^{-1} \text{cm}^{-1}$  were used to calculate the rate of methylene  $\text{H}_4\text{F}$  production (40).

**Kinetic model of  $\text{C}_1$  metabolism.** The previous kinetic model for growth on methanol (27) was modified in several ways. Changes A through E described below represent the new “default” version of the model. In change A, concentrations of energy and redox cofactors (ADP, ATP,  $\text{NAD}^+$ ,  $\text{NADH}$ ,  $\text{NADP}^+$ , and  $\text{NADPH}$ ) as measured in *Escherichia coli* were replaced with newly published values (9) for chemostat cultures of *M. extorquens* AM1. In change B, kinetic rate constants ( $V_{\text{max}}$ s) were recalculated from enzyme activities reported in previous publications (5, 6, 10, 25, 31, 32, 39, 40) using a conversion factor of 1  $\mu\text{mol}/(\text{min} \cdot \text{mg protein}) = 3.86 \text{ mM/s}$ , with mM referring to intracellular volume in this case. This conversion factor is based on (i) the assumption that 50% of biomass is protein and (ii) the numbers listed above under “Conversion of flux rates per culture volume to flux rates per intracellular volume.” In change C, the spontaneous reaction rate constant for the condensation of formaldehyde with  $\text{H}_4\text{F}$  was changed to the maximum measured rate of 0.08  $\text{mM}^{-1} \text{ s}^{-1}$ . The spontaneous condensation of formaldehyde with tetrahydromethanopterin ( $\text{H}_4\text{MPT}$ ) was ignored because of its modest contribution of methylene  $\text{H}_4\text{MPT}$  relative to the enzyme-catalyzed reaction (40). In change D, formylhydrolase, part of the formyltransferase/hydrolase complex (Fhc), was removed from the model because its activity is not limiting to the flux through Fhc (30) and because no experimental data on  $V_{\text{max}}$  or  $K_m$  are available for this reaction. In change E, new modeling was used to represent formyltetrahydrofolate ligase (FtlL). This enzyme forms a bond between formate and  $\text{H}_4\text{F}$ , hydrolyzing ATP to make the reaction energetically favorable. In this respect, the enzyme is analogous to pyruvic carboxylase, which catalyzes the formation of oxaloacetate from  $\text{HCO}_3^-$  and pyruvate at the expense of ATP. Based on this analogy, we used equation IX-337 in Segel’s enzyme kinetics text (34) to represent the flux through FtlL. The  $K_{\text{eq}}$  value used in the modeling was taken from previous measurements (13).

One model feature retained from the previous version was the use of a formate dehydrogenase (FDH) activity that is double the published value. The published value is only for the NAD-linked FDH activity, but *M. extorquens* AM1 also has two other functional FDHs whose activities are not measurable, since their electron acceptors are unknown (4, 5).

To explore conditions that would predict fluxes similar to those measured, additional changes were then made in an iterative manner to generate a version that predicted measured fluxes. Changes were made solely to parameters for which some uncertainty existed for the published values. The changes made were as follows. The first change was made because total  $\text{H}_4\text{F}$  and  $\text{H}_4\text{MPT}$  published concentrations (39) are minimum values because (i) the method used did not capture formyl  $\text{H}_4\text{F}$  or formyl  $\text{H}_4\text{MPT}$  and (ii) capture of the other molecular species was likely incomplete due to the thermodynamics of the reactions involved (J. Vorholt, personal communication). Therefore, the values for total  $[\text{H}_4\text{F}]$  and  $[\text{H}_4\text{MPT}]$  were increased from 0.15 mM to 0.2 mM and from 0.4 mM to 0.5 mM, respectively. In the second change, involving formaldehyde-activating enzyme (Fae), the  $K_m$  for  $\text{H}_4\text{MPT}$  was reduced to 15% of the published value. The published value is dependent on  $\text{H}_4\text{MPT}$  from another species, which is chemically different (6). In the third change, involving Ftr hydrolase complex (Fhc), the activity was increased fivefold from the published value. The enzyme is unstable, and the cofactor used is from a different species. The cofactor in *M. extorquens* is expected to be different (6, 30). In the fourth change, involving methenyl- $\text{H}_4\text{F}$  cyclohydrolase (Fch), the value of the  $K_{\text{eq}}$  is pH dependent because  $\text{H}^+$  participates in the reaction (17). Since the intracellular pH of *M. extorquens* may be close to 6.5 (see Results), a value determined at this pH (21) was used in the model.

**Statistics.** Values reported are means  $\pm$  standard deviations unless otherwise indicated.

## RESULTS

**Reexamination of the deuterium assay for direct condensation.** As previously described (27), in *M. extorquens* AM1 the serine generated from fully deuterated methanol contains two deuteriums when synthesized via the direct condensation route but only one deuterium when synthesized via formate and formyl  $\text{H}_4\text{F}$  (Fig. 2A). Therefore, the +2/+1 ratio in serine provides an estimate of the ratio of flux through each route. However, a potential limitation of this estimate is the fact that if the NADPH pool becomes contaminated with deuterium as a result of NADPH production from formaldehyde oxidation via the  $\text{H}_4\text{MPT}$  pathway, some +2 serine could be produced via the route involving formate and formyl  $\text{H}_4\text{F}$  (27) (Fig. 2B).

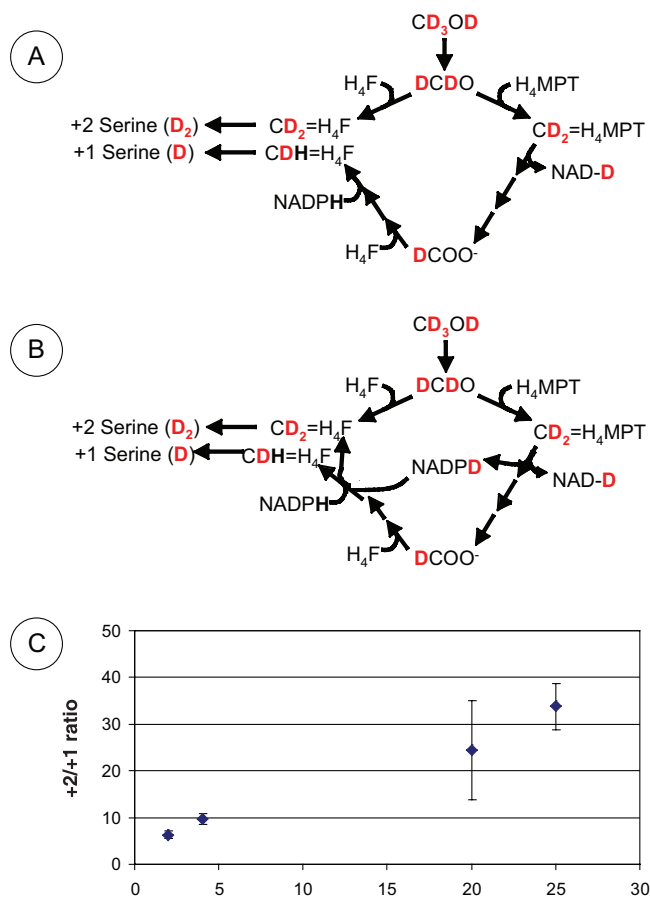


FIG. 2. The assay for methylene tetrahydrofolate synthesis routes in whole cells using deuterated methanol (CD<sub>3</sub>OD). (A) Spontaneous condensation consists of joining formaldehyde (in this case, DCDO) and H<sub>4</sub>F to form doubly deuterated methylene-tetrahydrofolate (CD<sub>2</sub>=H<sub>4</sub>F). In the alternate pathway, the one-carbon unit is joined to H<sub>4</sub>MPT, released as formate (DCOO<sup>-</sup>), and then joined to H<sub>4</sub>F, leading to the formation of CDH=H<sub>4</sub>F. In this scheme, no significant NADPH (NADPD in this experiment) is produced from the H<sub>4</sub>MPT pathway, and the ratio of doubly deuterated (D<sub>2</sub>) serine to singly deuterated (D) serine reflects relative fluxes through the spontaneous and alternate pathways. (B) In this scheme, it is assumed that significant NADPH (NADPD in this experiment) is generated by the H<sub>4</sub>MPT pathway. Contamination of the NADPH pool with deuterated NADPH produced after metabolism of deuterated methanol will increase the amount of +2 serine at the expense of +1 serine, leading to an overestimation of flux through spontaneous condensation. (C) Effect of assay incubation time on the ratio of +2 serine to +1 serine in methanol-grown cells.

In the original study, it was assumed that the pool turnover would not be sufficiently rapid to affect the results of the 20-s assay. Since NADH production by methylene H<sub>4</sub>MPT dehydrogenase (MtdB) cannot be replaced by the NADPH-specific enzyme (MtdA) when overexpressed (24), it appeared that the major pyridine nucleotide produced during methanol oxidation to CO<sub>2</sub> is NADH, not NADPH, thereby limiting the amount of deuterated NADPH that could contaminate the assay. However, since it is now known that the intracellular levels of NADPH (9) are small (<1 mM) relative to the levels of methanol consumed per second (>1 mM), rapid contami-

nation of the NADPH pool may occur if a significant amount of NADPH is generated by MtdA/MtdB.

We reasoned that if the NADPH pool became contaminated during the assay time, the +2/+1 ratio of serine should increase with incubation time, whereas a constant or declining +2/+1 ratio would argue against a contamination problem. Figure 2C shows that in methanol-grown cells, the ratio increases roughly linearly over the 25-s time period assessed, consistent with contamination of the NADPH pool with deuterium over time. These results suggested that flux through the direct condensation reaction might be lower than previously suggested.

**Biomass assimilation flux in mutants.** To directly test the capacity for flux through the direct condensation reaction, we examined serine production and biomass flux in the  $\Delta$ *fffL* mutant strain of *M. extorquens* AM1, which is defective in formyl H<sub>4</sub>F ligase. In this mutant, the alternate pathway for synthesis of methylene H<sub>4</sub>F is interrupted and the only route for methylene H<sub>4</sub>F synthesis is direct condensation. This strain is unable to grow on methanol, but it was grown on succinate and then exposed to methanol, conditions known to induce methylotrophic enzymes in this mutant (15).

If the direct condensation reaction is quantitatively important *in vivo*, production of +2 serine and positive biomass fluxes should be observed in the  $\Delta$ *fffL* mutant. However, biomass flux was not significantly different from zero (0.0005 ± 0.0032 mM/s; *n* = 6). In addition, +2 serine was very low when CD<sub>3</sub>OD was provided to these strains, constituting only 2.6% ± 1.6% (*n* = 4) of the total (+0, +1, and +2) serine detected as compared to percentages of ≥70% for wild-type cells grown on methanol. This is not inconsistent with previous results demonstrating a drop in +2 serine in this mutant compared to the wild type (27). That experiment involved cells grown on succinate, and the much lower fluxes make direct comparisons difficult. Similar results were obtained with another mutant defective in this pathway, the  $\Delta$ *mtdA* strain, for which biomass flux was 0.0013 ± 0.0037 mM/s (*n* = 8) and +2 serine was 1.1% ± 0.5% (*n* = 3). These data are consistent with the results presented above suggesting that little methylene H<sub>4</sub>F is produced via the direct condensation route under these conditions.

**Flux measurements in chemostat-grown cultures.** If the direct condensation reaction plays a minor role during methylotrophic growth, then the alternate route, involving the H<sub>4</sub>MPT and H<sub>4</sub>F pathways (Fig. 1), must be able to accommodate nearly all of the formaldehyde flux. In order to test this prediction, we carried out new flux measurements using methanol-limited chemostat-grown cells. The use of chemostat cultures allows a direct calculation of fluxes to biomass occurring in the growing cells using the substrate input rate, the dilution rate, and the amount of cells in the culture (see Materials and Methods). For methanol utilization, these calculations must be corrected for incorporation of CO<sub>2</sub> via the serine cycle and other carboxylation reactions in the cell. To determine the proportion of biomass carbon from CO<sub>2</sub>, radioactive labeling of biomass was measured in the presence of 5% unlabeled CO<sub>2</sub> as compared to a CO<sub>2</sub>-free environment. In the absence of external CO<sub>2</sub>, all or nearly all of the CO<sub>2</sub> incorporated into biomass should be derived from labeled methanol and therefore should be labeled, while in high unlabeled CO<sub>2</sub>, all or nearly all of the CO<sub>2</sub> incorporated should be unlabeled. There-

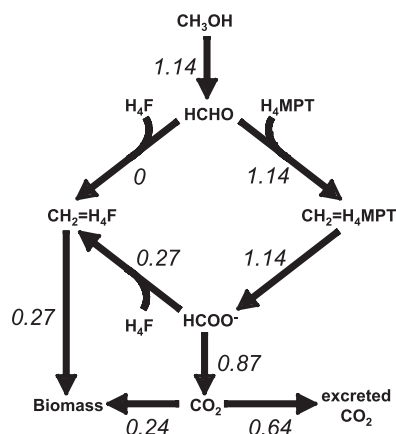


FIG. 3. Distribution of fluxes (in mM/s) in wild-type *M. extorquens* AM1 chemostat cells grown on methanol, assuming that no methylene  $H_4F$  is produced via spontaneous condensation.

fore, the difference between the two conditions should reflect the proportion of biomass derived from  $CO_2$ . The average values for five replicate  $^{14}C$ -labeling experiments indicated that at least 47% of the biomass carbon comes from  $CO_2$ , with the remainder coming from methylene  $H_4F$ . These results are in keeping with previous labeling studies of flask-grown cells, suggesting approximately half of the biomass carbon comes from  $CO_2$  (18).

It has previously been shown that methanol-limited cells grown at a dilution rate equivalent to 80% of the batch culture growth rate are generally similar to batch culture cells in terms of enzyme activities and nucleotide concentrations, suggesting that these conditions are roughly comparable (9).

By assuming that all methanol not converted to methylene  $H_4F$  is converted to  $CO_2$ , a complete flux distribution for the  $C_1$  network can be calculated for cells in a methanol-limited chemostat (Fig. 3). In order to evaluate whether the  $H_4MPT$  and  $H_4F$  pathways could accommodate such fluxes, a kinetic model was utilized.

**Predictions from kinetic modeling.** The new data reported in this paper along with other recently published data were used to update and enhance the previous kinetic model for formaldehyde and formate partitioning during methylotrophic growth (27). One issue with the previous model was that the fitted rate constant for the spontaneous condensation reaction was  $2.64 \text{ mM}^{-1} \text{ s}^{-1}$  (27), which is significantly higher than literature values. However, the literature values were determined at  $20^\circ\text{C}$  (2), while growth temperatures are 26 to  $28^\circ\text{C}$ . To explore this issue further, the nonenzymatic rate constant for the condensation of formaldehyde with  $H_4F$  was measured at normal growth temperatures for *M. extorquens*. The rate equation for flux through this reaction is  $\text{rate} = V_6 \cdot [H_4F] \cdot [\text{formaldehyde}]$ , referring to the rate constant as  $V_6$  for consistency with the earlier publication (27).  $V_6$  was measured at 26 to  $28^\circ\text{C}$  at pH values from 6 to 8. The range of values was 0.02 to  $0.08 \text{ mM}^{-1} \text{ s}^{-1}$ , with the maximum value at pH 6.7. These rate constants are in reasonable agreement with the data of Blakley (2), from which a rate constant of  $\sim 0.026 \text{ mM}^{-1} \text{ s}^{-1}$  can be estimated at  $20^\circ\text{C}$  and pH 7.2. Therefore, the

maximum rate constant measured ( $0.08 \text{ mM}^{-1} \text{ s}^{-1}$ ) was used in the model, to provide a high-flux scenario.

In keeping with the other results presented in this study, this model predicted that even with the use of a relatively high rate constant for spontaneous formaldehyde condensation with  $H_4F$ , this rate constant does not allow significant flux through this route unless formaldehyde rises to levels known to inhibit growth (concentrations greater than 1 mM). A number of simulations were run to identify conditions that would accommodate the measured fluxes to  $CO_2$  and biomass. Parameters for one such simulation are shown in Tables 1 and 2. In this case, published values for kinetic parameters and concentrations were used aside from the exceptions noted in Materials and Methods. Each of these changes results in values that are within expected tolerances, given the uncertainties of in vitro enzyme measurements with regard to in vivo conditions (see Materials and Methods). Using these parameters, the model predicted fluxes similar to the measured fluxes (Table 2).

## DISCUSSION

In this study, the significance in methylotrophic metabolism of the direct condensation route for synthesis of methylene  $H_4F$  was assessed in *M. extorquens* AM1. We first obtained evidence that the deuterated methanol assay might overestimate the extent of the direct condensation reaction. Exact determinations with deuterated substrates are difficult, due to the possibility of differential isotope effects, but the data are consistent with an overestimation. An alternative approach to measuring flux through this reaction is to measure it in the  $\Delta\text{ftfL}$  mutant defective in the alternate route involving the  $H_4F$  pathway. The results from these analyses demonstrate that under these conditions, the flux through the direct condensation route is minor. However, the  $\Delta\text{ftfL}$  mutant does not grow on methanol and must be tested in cells that have been grown on succinate, washed, and incubated with methanol. These conditions are known to induce methylotrophic enzymes in this mutant (16), but the flux capacity through the assimilatory pathways is unknown. Removal of methylene  $H_4F$  by the serine cycle is likely to be important to sustain flux through this route (27). In addition, if an enzyme exists that converts formaldehyde plus  $H_4F$  to methylene  $H_4F$ , it may not be fully induced under these conditions. Therefore, these results do not rule out a higher flux through this route in the wild type during growth on methanol. However, even if the contribution of this route to the synthesis of methylene  $H_4F$  increases an order of magnitude during growth on methanol, the alternate route, including the enzymes Fae, MtdA/MtdB, Mch, Fhc, FtfL, Fch, and MtdA (Fig. 4), should dominate

TABLE 1. Concentrations of the compounds used

Compound(s)	Concn in mM (reference)
ATP.....	2.88 (9)
ADP.....	0.60 (9)
NAD.....	1.69 (9)
NADH.....	0.53 (9)
NADP.....	0.62 (9)
NADPH.....	0.42 (9)
$H_4F$ + methylene $H_4F$ + methenyl $H_4F$ .....	0.2 (39)
$H_4MPT$ + methylene $H_4MPT$ + methenyl $H_4MPT$ .....	0.5 (39)
$P_i$ .....	3 (29)



TABLE 2. Parameters used in the final model and predicted fluxes

Enzyme(s) or reaction(s)	Parameters (reference[s]) <sup>a</sup>	Flux (mM/s):	
		Predicted by kinetic model	Calculated from exptl data <sup>b</sup>
Formaldehyde-activating enzyme (Fae)	$V_{\max}$ , 5.4 mM/s; $K_{\text{eq}}$ , irrev.; $K_m$ s, 0.2 mM for formaldehyde and 0.1 mM for H <sub>4</sub> MPT (40)	1.13	1.14
Methylene H <sub>4</sub> MPT dehydrogenases MtdB	$V_{\max}$ , 2.3 mM/s; $K_{\text{eq}}$ , 174.2; $K_m$ s, 0.05 mM for methylene H <sub>4</sub> MPT and 0.2 mM for NAD <sup>+</sup> (10)	1.13	1.14
MtdA	$V_{\max}$ , 10.0 mM/s; $K_{\text{eq}}$ , 174.2; $K_m$ s, 0.02 mM for methylene H <sub>4</sub> MPT and 0.03 mM for NADP <sup>+</sup> (39)		
Methenyl H <sub>4</sub> MPT cyclohydrolase (Mch)	$V_{\max}$ , 11.1 mM/s; $K_{\text{eq}}$ , 0.137; $K_m$ , 0.03 mM for methenyl H <sub>4</sub> MPT (6, 32)	1.13	1.14
Formyltransferase (Ftr)	$V_{\max}$ , 9.9 mM/s; $K_{\text{eq}}$ , 0.204; $K_m$ s, 0.05 mM for formyl MFR and 0.03 mM for H <sub>4</sub> MPT (31)	1.13	1.14
Formaldehyde + H <sub>4</sub> F	$K_{\text{eq}}$ , NA; $V_0$ , 0.08 mM <sup>-1</sup> s <sup>-1</sup> (this study)	0.01	0
Methylene H <sub>4</sub> F dehydrogenase (MtdA)	$V_{\max}$ , 6.8 mM/s; $K_{\text{eq}}$ , 4; $K_m$ s, 0.01 mM for NADP <sup>+</sup> and 0.03 mM for methylene H <sub>4</sub> F (6, 39)	0.24	0.27
Methenyl H <sub>4</sub> F cyclohydrolase (Fch)	$V_{\max}$ , 1.9 mM/s; $K_{\text{eq}}$ , 0.54; $K_m$ , 0.08 mM for methenyl H <sub>4</sub> F (21, 32)	0.24	0.27
Formyl H <sub>4</sub> F ligase (FtfL)	$V_{\max}$ , 1.6 mM/s; $K_{\text{eq}}$ , 41; $K_m$ s, 22 mM for formate, 0.8 mM for H <sub>4</sub> F, 0.021 mM for ATP, and 2 mM for P <sub>i</sub> (13, 25)	0.24	0.27
FDH	$V_{\max}$ , 1.2 mM/s; $K_{\text{eq}}$ , irrev.; $K_m$ , 1.6 mM for formate (5, 20)	0.88	0.87
Serine hydroxymethyltransferase (GlyA)	$V_{\max}$ , 20 mM/s; $K_{\text{eq}}$ , NA; $K_m$ , 0.16 mM for methylene H <sub>4</sub> F (fitted as in reference 27)	0.25	0.27

<sup>a</sup>  $K_{\text{eq}}$ s and  $V_{\max}$ s of reversible reactions are for the directions shown in Fig. 1, 2, 3, and 4. irrev., irreversible; NA, not available.

<sup>b</sup> Shown in Fig. 3.

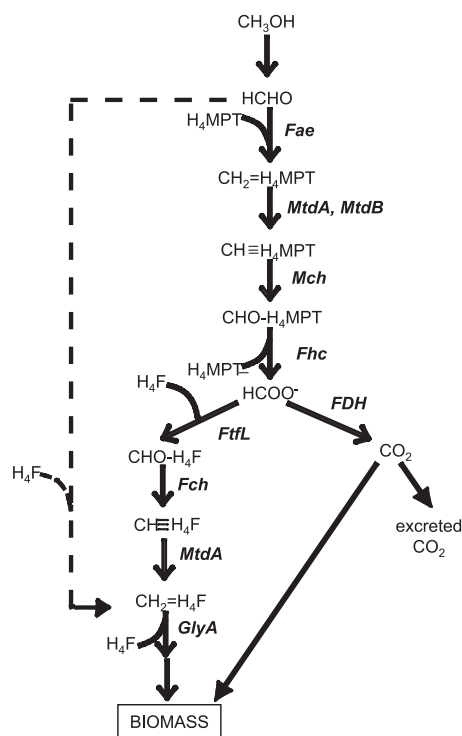


FIG. 4. Proposed scheme of methylotrophic metabolism in *M. extorquens* AM1 showing formate (HCOO<sup>-</sup>) as the branch point between assimilatory and dissimilatory metabolism.

the total biomass production. This suggestion is supported by the kinetic model generated in this study, which predicts that this alternate route can support the measured fluxes during methylotrophic growth using kinetic and cofactor concentration parameters that are within known or predicted ranges. In addition, these results explain the finding that FtfL, Fch, and MtdA mutants are all unable to grow on methanol.

Although our results suggest that the spontaneous condensation reaction is not an important source of methylene H<sub>4</sub>F under standard lab culture conditions, it is still possible that it serves as an overflow valve for transient episodes of formaldehyde overproduction. Formaldehyde overproduction might occur following transient exposure to methanol, due to the high methanol dehydrogenase activity in methanol-limited cells (9). Transient methanol exposure is known to occur under natural conditions for these bacteria. *Methylobacterium* strains are commonly found on leaf surfaces, apparently growing on methanol emitted from stomata (35), and methanol emission from leaves is episodic (28). It is possible that under such conditions the spontaneous condensation reaction would play a role in formaldehyde detoxification.

These results suggesting that, under laboratory growth conditions, methylene H<sub>4</sub>F is mainly formed by the route involving the H<sub>4</sub>MPT and H<sub>4</sub>F pathways lead to the suggestion that formate represents the primary metabolic branch point between assimilation of C<sub>1</sub> units into biomass (via methylene H<sub>4</sub>F

and the serine cycle) and dissimilation to CO<sub>2</sub> for energy generation in this bacterium (Fig. 4). The linear conversion of formaldehyde to formate by the high-capacity enzymes of the H<sub>4</sub>MPT pathway is a straightforward but effective method to maintain high fluxes of formaldehyde without allowing its accumulation. The disadvantage is that it requires an extra ATP for every formaldehyde assimilated, compared to a direct condensation route. Since growth on methanol is predicted to be limited by reducing power rather than ATP (36), it is possible that the tradeoffs are positive for metabolism as a whole.

Optimal partitioning of formate between these two branches may require extensive control of the relevant branch point enzymes, namely, FtfL, Fch, and MtdA for the H<sub>4</sub>F pathway, and FDH. The importance of FDH is suggested by the presence of four isoforms in *M. extorquens* AM1, which vary greatly in terms of their expression patterns, dependence on cofactors (NAD<sup>+</sup>) and metal ions (molybdenum and tungsten), and mutant phenotypes (4, 5). These results now direct attention to these enzymes as the likely main control points for balancing of metabolism during methylotrophic growth in *M. extorquens* AM1.

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