

NIH Public Access **Author Manuscript**

Org Lett. Author manuscript; available in PMC 2008 August 12.

Published in final edited form as: *Org Lett*. 2006 February 2; 8(3): 427–430.

Synthetic Studies on Amphidinolide B1

Amit K. Mandal†, **John S. Schneekloth Jr.**‡, **Kouji Kuramochi**†, and **Craig M. Crews***,†,‡,§

†*Department of Molecular, Cellular, and Developmental Biology, Yale University, New Haven, Connecticut 06520-8103*

‡*Department of Chemistry, Yale University, New Haven, Connecticut 06520-8103*

§*Department of Pharmacology, Yale University, New Haven, Connecticut 06520-8103*

Abstract

The syntheses of three fragments, 2, 3, and 4, of amphidinolide B1 have been accomplished. The 1,3-isomerization of allylic alcohol 10 was accomplished via rhenium oxo catalysis and has been applied successfully in the synthesis. (–)-MIB-catalyzed asymmetric vinylzinc addition to aldehyde 31 and the regio- and stereoselective epoxidation of unsymmetrical divinyl methanol 32 were key steps.

> Amphidinolide B1 (**1**), a polyketide-based 26-membered macrolide, was isolated from a culture of the symbiotic marine dinoflagellate *Amphidinium* sp. (strain Y-5) in 1987 by Kobayashi et al.1,2 There have been several reports of partial syntheses toward **1**; however, no completed total synthesis has been disclosed yet.³

> In our laboratory, we undertook a multipronged approach toward the synthesis of amphidinolide B1 (1) (Scheme 1). One strategy was communicated earlier, $\frac{4}{3}$ whereas this paper represents an alternative strategy. Our previous strategy involved three major C–C bondforming reactions and one C–O bond-forming reaction. This new retrosynthetic analysis involves two C–C and one C–O bond-forming coupling reactions: Suzuki–Miyaura coupling, 5 an aldol reaction,6 and macrolactonization using the three fragments **2**, **3**, and **4**. As C–C bond-forming reactions are traditionally more difficult than C–O bond-forming reactions, we hope that this new strategy will increase the efficiency of the synthesis.

> The synthesis of fragment **2** began with Brown's crotylation reaction (Scheme 2).7 The reaction of *trans*-crotonaldehyde and (*E*)-crotyldiisopinocamphenylborane (prepared from *trans*butene, *n*-BuLi, *t*-BuOK, and $(+)$ -Ipc₂BOMe) in the presence of BF₃·OEt₂ in THF-Et₂O at -78 °C produced the allyl alcohol **6** in 50% yield and 14:1 anti/syn selectivity. Silyl protection of the hydroxyl group and a regioselective hydroboration–oxidation sequence furnished alcohol **7**, which was oxidized to the corresponding aldehyde **8** using Dess–Martin periodinane. The asymmetric methylation of **8** with Me₂Zn and $Ti(OⁱPr)₄$ was examined with the chiral ligands $\rm BINOL$ and bissulfonamide. $\rm\frac{8}$ Unfortunately, the conversions were unacceptable in both cases (25% and 29%, respectively). However, diastereoselective methylation to obtain **9** was achieved with Seebach's method⁹ [Me₂Zn and Ti(O^{*i*}Pr)₄ in the presence of (−)-TADDOL] in 92% yield and a 5–7:1 diastereomeric ratio (based on 1 H NMR). Protection of the free alcohol as a TIPS ether (the minor isomer can be separated at this stage) and selective cleavage of the allylic TBS ether furnished **10** in 70% yield over two steps. At this point, the stage was set to

craig.crews@yale.edu.

Supporting Information Available: Experimental procedures and spectroscopic data. This material is available free of charge via the Internet at [http://pubs.acs.org.](http://www.pubs.acs.org)

manipulate allylic alcohol **10** to the requisite α,β-unsaturated ketone **12** and to install the remaining stereogenic centers. The 1,3-isomerization of **10** and subsequent oxidation promised to be an atom-efficient method to access ketone 12. $Ph_3SiOReO_3^{10}$ is currently the most effective catalyst for 1,3-isomerization.¹¹ When we treated **10** in the presence of 1 mol % catalyst in ether at −60 °C, complete isomerization to regioisomeric allylic alcohol **11** was observed in 5 min. Oxidation of the alcohol **11** under Dess–Martin periodinane conditions furnished the α,β-unsaturated ketone **12**. Swapping TIPS protection with TBS, Sharpless asymmetric dihydroxylation¹² followed by acetonide formation completed the synthesis of fragment **2**.

The synthesis of fragment **3** started with known vinyl iodide **13**13 (Scheme 3). Transmetalation with *t*-BuLi in Et₂O and capture of the resulting vinyllithium species with acetaldehyde yielded (\pm) -allylic alcohol 14. Sharpless kinetic resolution¹⁴ of (\pm) -14 gave the desired epoxide 15 in 85–90% enantiomeric excess (ee) and unreacted alcohol in 25–30% ee as a mixture that proved difficult to separate. The crude mixture was subjected to a Parikh–Doering oxidation to yield easily separable epoxy ketone **16** and the corresponding α,β-unsaturated ketone (not shown). The α ,β-unsaturated ketone was subjected to a Luche reduction to produce (\pm)-allylic alcohol **14**, which was recycled under Sharpless kinetic resolution conditions. The Horner– Wadsworth–Emmons condensation of epoxy ketone **16** with phosphonate in the presence of NaHMDS produced enyne **17** [82% yield, *E/Z* 6:1]. Simultaneous cleavage of TBS ether and the TMS group by TBAF followed by regioselective opening of the epoxide in the presence of LiAlH₄ in Et₂O furnished diol 18, which was separated from the minor (*Z*) isomer using column chromatography. Functionalization of the enyne **18** to the 1,3-diene iodide was quite problematic. After exploring several possibilities, we found that the triple bond could be silylstannylated regio- and stereoselectively employing *n*-Bu₃SnSiMe₂Ph/Pd(PPh₃)₄ to produce functionalized diene 19.¹⁵ TBAF-mediated removal of the PhMe₂Si group produced stannane 20. Reaction with I₂ and selective TBS protection completed the synthesis of fragment **3**.

The synthesis of fragment **4** began with aldehyde **21** (Scheme 4). A nonchelation-controlled S_E ^{2'} reaction between aldehyde 21 and methallylsilane in the presence of BF_3 ·OEt₂ furnished homoallylic alcohol 22 in 90% yield and \geq 95:5 diastereomeric ratio (dr).¹⁶ Hydroxyl group protection as a TBS ether **23**, followed by oxidative deprotection of the PMB ether, revealed the primary alcohol. The alcohol was then converted to the aldehyde, which was subjected to vinyl-magnesium bromide addition to provide allyl alcohol **24** as a 1.5:1 diastereomeric mixture (based on ${}^{1}H$ NMR). The allylic alcohol 24 underwent Johnson ortho ester Claisen rearrangement to provide the homologated ethyl ester exclusively as the (*E*) isomer. Reduction to the aldehyde using *i*-Bu2AlH in THF and the Wittig reaction furnished fragment **4** with exclusive (*E*) selectivity.

With all three fragments securely in hand, we sought to explore the possibility of setting up the C11 stereogenic centers by hydroboration and a subsequent (*B*)-alkyl Suzuki-Miyaura coupling reaction⁵ to assemble the 1,3-diene. We hoped to induce 1,3-stereocontrol on the basis of Evans' alkyl-directed hydroboration model.¹⁷ We have been able to form the C12– C13 bond successfully using the (*B*)-alkyl Suzuki–Miyaura coupling reaction between (*B*) alkylborane (derived from regioselective hydroboration of fragment **4**) and (±)-1,3-vinyl iodide fragment **3** (synthesis not shown) to put together the C1–C18 portion of the molecule. However, investigations revealed that 1,3-stereocontrol cannot be achieved with this substrate, as hydroboration of fragment **4** with 9-BBN provided a 1:1 inseparable mixture of diastereoisomers at C11 (amphidinolide B1 numbering).

Considering A values, an alkyl group (i.e., $Me = 1.74$ kcal/mol) imparts more nonbonding interaction than an $OSiR_3$ group (i.e., $OSiMe_3 = 0.74$ kcal/mol), presumably because the

 $\sin 2$ group can be turned away to avoid the steric interaction.¹⁸ This explains the loss of π facial selectivity in our substrate. As separation of the diastereoisomers was not easy, we revised our strategy and decided to use compound **25** as a new fragment for the C1–C12 segment in which the C11 methyl stereogenic center is preinstalled (Scheme 5). Instead of coupling the 1,3-diene iodide fragment **3** to (*B*)-alkylborane (derived by hydroboration from fragment **4**), we decided to couple it to the ate complex **26** derived from the corresponding iodide easily accessible from compound 25 in two steps.¹⁹ Preliminary studies involving \pm)alkenyliodide fragment **3** and a model alkyl boronate to assemble the 1,3-diene corroborate the feasibility of this strategy. 20

The synthesis of compound **25** commenced with isoprene monoxide (Scheme 6). TiCl4 mediated regioselective and stereoselective opening of the epoxide provided the alcohol, which was protected as a TIPS ether **28**. It was then coupled with propargylmagnesium bromide in the presence of Pd-(PPh3)4 to give acetylene **29** in 83% yield. In parallel, aldehyde **31** was synthesized in two steps via the Wittig reaction, followed by reduction with i -Bu₂AlH. The key strategy is to employ asymmetric addition of alkenylzinc to aldehyde **31** in the presence of Nugent's21 isoborneol-based (−)-MIB ligand.22 Hydroboration of the terminal acetylene **29** with freshly prepared dicyclohexylborane proceeds regio-selectively to afford the alkenylborane. Transmetalation of this alkenylborane with Me₂Zn generates the reactive alkenylzinc reagent in situ. Attempts to carry out the transmetalation at 0 °C, as described in the literature, 23 led to decomposition of the substrate. After extensive experimentation, we found that the reaction could be performed successfully by adding Me₂Zn followed by $(-)$ -MIB at −78 °C to alkenylborane. The mixture was then warmed to −20 °C over 10 min. The aldehyde **31** in hexane was added via syringe pump over 20 min while warming the mixture to 0 °C to furnish divinyl methanol **32** in 43% yield and ≥95% diastereomeric excess (de). This assembled the full carbon backbone of compound **25**. The next challenge was to functionalize stereoselectively the C7–C8 double bond of this unsymmetrical divinyl methanol **32**.

It is documented that the simple (\pm) -unsymmetrical divinyl methanols can be subjected to the Sharpless kinetic resolution conditions to synthesize the optically active monoepoxide.²⁴ However, to the best of our knowledge, Sharpless asymmetric epoxidation has not been applied to desymmetrize a complex unsymmetrical divinyl methanol such as ours in the context of natural product synthesis. We were pleased to find that Sharpless' asymmetric epoxidation provided the C7–C8 monoepoxide in 71% yield (6:1 dr). TIPS protection went uneventfully to furnish **25** in 90% yield.

In conclusion, we have accomplished the synthesis of three fragments, **2**, **3** and, **4**. Compound **25** will be used as the new fragment for the C1–C12 segment. 1,3-Isomerization of allylic alcohol via rhenium oxo catalysis has been applied successfully in our synthesis. (−)-MIBcatalyzed asymmetric vinylzinc addition to aldehyde **31**, highly selective late-stage epoxidation of divinyl methanol **32**, and regio- and stereo-selective silylstannylation to synthesize stannane **20** have been used as key steps. Continued advancement of these intermediates toward the eventual total synthesis of **1** are currently ongoing in our laboratory.

Acknowledgment

We gratefully acknowledge financial support from NIH (GM062120). J.S.S. acknowledges the American Chemical Society, Division of Medicinal Chemistry, and Aventis Pharmaceuticals for a pre-doctoral fellowship.

References

1. Ishibashi M, Ohizumi Y, Hamashima M, Nakamura H, Hirata Y, Sasaki T, Kobayashi J. J. Chem. Soc., Chem. Commun 1987:1127.. For reviews: (a) Kobayashi J, Ishibashi M. Chem. Rev 1993;93:1753. (b) Chakraborty T, Das S. Curr. Med. Chem.: Anti-Cancer Agents 2001;1:131. (c) Kobayashi J,

Org Lett. Author manuscript; available in PMC 2008 August 12.

Shimbo K, Kubota T, Tsuda M. Pure Appl. Chem 2003;75:337. (d) Kobayashi J, Tsuda M. Nat. Prod. Rep 2004;21:77. [PubMed: 15039836]

- 2. Kobayashi J, Ishibashi M, Nakamura H, Ohizumi Y, Yamasu T, Hirata Y, Sasaki T, Ohta T, Nozoe S. J. Nat. Prod 1989;52:1036. [PubMed: 2607346]
- 3. (a) Kobayashi J, Tsuda M. Nat. Prod. Rep 2004;21:77. [PubMed: 15039836] (b) Cid MB, Pattenden G. Synlett 1998:540. (c) Cid MB, Pattenden G. Tetrahedron Lett 2000;41:7373. (d) Ishiyama H, Takemura T, Tsuda M, Kobayashi J. Tetrahedron 1999;55:4583. (e) Ishiyama H, Takemura T, Tsuda M, Kobayashi J. J. Chem. Soc., Perkin Trans 1999;1:1163. (f) Ohi K, Shima K, Hamada K, Saito Y, Yamada N, Ohba S, Nishiyama S. Bull. Chem. Soc. Jpn 1998;71:2433. (g) Ohi K, Nishiyama S. Synlett 1999:573. (h) Ohi K, Nishiyama S. Synlett 1999:571. (i) Chakraborty TK, Thippeswamy D, Suresh VR, Jayaprakash S. Chem. Lett 1997:563. (j) Chakraborty TK, Thippeswamy D. Synlett 1999:150. (k) Chakraborty TK, Thippeswamy D, Jayaprakash S. J. Ind. Chem. Soc 1998;75:741. (l) Chakraborty TK, Suresh VR, Vayalakkada R. Chem. Lett 1997:565. (m) Eng HM, Myles DC. Tetrahedron Lett 1999;40:2275. (n) Eng HM, Myles DC. Tetrahedron Lett 1999;40:2279. (o) Lee D-H, Lee S-W. Tetrahedron Lett 1997;38:7909. (p) Lee D-H, Rho M-D. Bull. Korean Chem. Soc 1998;19:386. (q) Lee D-H, Rho M-D. Tetrahedron Lett 2000;41:2573. (r) Zhang W, Carter RG, Yokochi AFT. J. Org. Chem 2004;69:2569. [PubMed: 15049660]
- 4. Mandal AK, Schneekloth JS Jr, Crews CM. Org. Lett 2005;7:3645. [PubMed: 16092840]
- 5. Miyaura N, Ishiyama T, Sasaki H, Ishikawa M, Satoh M, Suzuki A. J. Am. Chem. Soc 1989;111:314.
- 6. Evans DA, Carter PH, Carreira EM, Charette AB, Prunet JA, Lautens M. J. Am. Chem. Soc 1999;121:7540.
- 7. Brown HC, Bhat KS. J. Am. Chem. Soc 1986;108:293.
- 8. (a) Takahashi H, Kawakita T, Ohno M, Yoshioka M, Kobayashi S. Tetrahedron 1992;48:5691. (b) Kitamoto D, Imma H, Nakai T. Tetrahedron Lett 1995;36:1861.
- 9. (a) Schmidt B, Seebach D. Angew. Chem., Int. Ed. Engl 1991;30:99. (b) von dem Bussche-Hunnefeld JL, Seebach D. Tetrahedron 1992;48:5719.
- 10. Schoop T, Roesky HW, Noltemeyer M, Schmidt H-G. Organometallics 1993;12:571.
- 11. Bellemin-Laponnaz S, Gisie H, Le Ny J-P, Osborn JA. Angew Chem., Int. Ed. Engl 1997;36:976.
- 12. Walsh PJ, Sharpless KB. Synlett 1993:605.
- 13. Ashimori A, Bachand B, Calter MA, Govek SP, Overman LE, Poon DJ. J. Am. Chem. Soc 1998;120:6488.
- 14. Gao Y, Hanson RM, Klunder JM, Ko SY, Masamune H, Sharpless KB. J. Am. Chem. Soc 1987;109:5765.
- 15. (a) Mitchell TN, Wickenkamp R, Amamria A, Dieke R, Scheider U. J. Org. Chem 1987;52:4868. (b) Chenard BL, Van Zyl CM. J. Org. Chem 1986;51:3561.
- 16. Reetz MT, Kesseler K. J. Org. Chem 1985;50:5434.
- 17. Evans DA, Bartroli J, Godel T. Tetrahedron Lett 1982;23:4577.
- 18. Eliel, EL.; Wilen, SH.; Mander, LN. Stereochemistry of Organic Compounds. New York: John Wiley & Sons; 1993. p. 696
- 19. Miyaura N, Suzuki A. Chem. Rev 1995;95:2457.
- 20. Unpublished results
- 21. Nugent WA. J. Chem. Commun 1999:1369.
- 22. (a) Oppolzer W, Radinov RN. Hev. Chim. Acta 1992;75:170. (b) Pu L, Yu H-B. Chem. Rev 2001;101:757. [PubMed: 11712502]Review:
- 23. Chen YK, Lurain AE, Walsh PJ. J. Am. Chem. Soc 2002;124:12225. [PubMed: 12371863]
- 24. Honda T, Mizutani H, Kanai K. J. Chem. Soc., Perkin Trans. 1 1996;14:1729.

Fragment 4

Scheme 1.

Scheme 2.

Scheme 3.

Org Lett. Author manuscript; available in PMC 2008 August 12.

Scheme 5.

Org Lett. Author manuscript; available in PMC 2008 August 12.

Scheme 6.