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## Two Palladium-Catalyzed Domino Reactions from One Set of Substrates/Reagents: Efficient Synthesis of Substituted Indenes and *cis*-Stilbenoid Hydrocarbons from the Same Internal Alkynes and Hindered Grignard Reagents

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### Abstract

Two types of domino reactions from the same internal alkynes and hindered Grignard reagents based on carbopalladation, Pd-catalyzed cross-coupling reaction and C-H activation strategy are described. The realization of these domino reactions relied on the control of the use of the ligand and the reaction temperature. Our study provides an efficient access to useful polysubstituted indenenes and *cis*-substituted stilbenes, and may offer new means to the development of tandem/domino reactions in a more efficient way.

The development of transition metal-catalyzed tandem or “domino” reactions, which combine two or more bond-forming reactions into one synthetic operation, represents one of the most attractive subjects in synthetic organic chemistry.<sup>1,2</sup> Such tandem/domino reactions allow the concomitant formation of two or more bonds with rapid increase in molecular complexity with minimized separation/purification efforts. Since arranging two or more bond-forming reactions to occur in a tandem or domino fashion is always challenging, it is not surprising to observe that almost all tandem/domino reactions were developed on a basis of one type of tandem/domino reaction per one set of substrates/reagents.<sup>1–3</sup> Developing two or more types of tandem/domino reactions from the same substrates and reagents, which represents a strategy that could further heighten the efficiency of conducting reactions in a tandem/domino fashion, is apparently very attractive, but remains to be largely unexplored.<sup>4</sup>

We have recently documented the palladium-catalyzed tandem reaction of 1,2-dhalobenzenes and 2-haloaryl tosylates with hindered Grignard reagents to form substituted fluorenes,<sup>5</sup> in which palladium-associated arynes were believed to be the intermediates when the reaction was carried out in the absence of phosphines or *N*-heterocyclic carbenes ligands.<sup>5b</sup> The triple bond nature of arynes led us to consider that alkynes might also function similarly as *in situ* generated arynes. We thus envisioned that carbopalladation of alkynes could generate vinylpalladium(II)X complexes **I** (Scheme 1).<sup>6</sup> **I** could then (a) undergo cyclization via C-H activation<sup>7,8</sup> to afford substituted indenenes, which are structural constituents of metallocene-based catalysts for olefin polymerizations, of biologically active compounds and of functional materials;<sup>9,10</sup> and (b) undergo transmetalation followed by reductive elimination (cross-coupling process) to yield *cis*-stilbenoid hydrocarbons, which are potentially useful in the fields of molecular sensors and molecular electronics.<sup>11,12</sup> Therefore, two types of domino reactions, namely domino carbopalladation-cyclization to form polysubstituted indenenes and

domino carbopalladation–cross-coupling to form *cis*-stilbenoid hydrocarbons containing highly substituted phenyl groups, might be developed from the same alkynes and hindered Grignard reagents if the two competing pathways could be controlled (Scheme 1). Herein, we report our successful realization of these two types of reactions by controlling the use of ligand and the reaction temperature.

Based on the consideration that the activation of C–H bond would involve the interaction of C–H bond with Pd(II) center and such interaction would be disfavored at higher reaction temperature and/or in the presence of ligands, we surmised the cyclization via *sp*<sup>3</sup> C–H activation process would be favored in the absence of ligands and at lower reaction temperature. We thus began our study by examining the reaction of diphenylacetylene with 2-mesitylPd(II) (OAc), *in situ* generated from 2-mesitylmagnesium bromide with Pd(OAc)<sub>2</sub>. We found the domino carbopalladation–cyclization product 4,6-dimethyl-2,3-diphenylindene was the major product with only Pd(OAc)<sub>2</sub> as the promoter, either at room temperature, 60 °C or refluxing (Table 1, entries 1, 2, 5). The use of PPh<sub>3</sub> as a ligand decreases the formation of the cyclization product as well as slowed down the reaction (Table 1, entries 2–4). By using 4 equiv. of PPh<sub>3</sub> and in refluxing THF, the domino carbopalladation–cross-coupling product became the major product, along with the self-coupling of Grignard reagent as the main side reaction (Table 1, entry 7). Our results suggested that by controlling the use of ligand and reaction temperature, it is possible to control the domino reaction pathways.

As Pd(II)X<sub>2</sub> would be reduced to Pd(0) species after every reaction cycle, after establishing factors that influence the reaction competing pathways, we next turned our attention to develop the catalytic version of these two types of domino reactions by identifying oxidants that could oxidize Pd(0) species to Pd(II) species. We have tested several commonly available oxidants and found 1,2-dibromoethane can be served as an excellent oxidizer (Table 2). By using a stoichiometric amount of 1,2-dibromoethane and 3% Pd(OAc)<sub>2</sub>, the domino carbopalladation–cyclization process proceeded smoothly to give 4,6-dimethyl-2,3-diphenylindene in excellent yield (Table 2, entry 6).

With 1,2-dibromoethane as the oxidant, a number of alkynes were examined for the Pd(OAc)<sub>2</sub>-catalyzed domino carbopalladation–cyclization reaction and our results are listed in Table 3. We found that diaryl-, dialkyl- and alkylarylacetylenes were all suitable substrates. When unsymmetrical alkylarylacetylenes were employed as the substrates, we found the domino reaction occurred mainly from the alkyl sides of alkylarylacetylenes,<sup>13</sup> as evidenced by the ratio of two isomeric products (Table 3, entries 9–13). To determine whether other types of hydrogens (nonbenzylic 1° hydrogens, benzylic 2°, and 3° hydrogens) could also be activated under our condition, we have tested 2-ethyl-6-methylphenylmagnesium bromide and 2-isopropyl-6-methylphenylmagnesium bromide for the domino reaction. We found that the *sp*<sup>3</sup> C–H activation exclusively occurred at the benzylic methyl group, suggesting that nonbenzylic 1° hydrogens (nonbenzylic methyl group), 2° (ethyl group) and 3° (isopropyl group) benzylic hydrogens could not be activated (Table 3, entries 14, 15). This was further confirmed by the fact that no reaction was observed for 2,6-diethylphenylmagnesium bromide with diphenyl-acetylene (Table 3, entry 16).

By using 1,2-dibromoethane as the oxidant, 4 equivalent of PPh<sub>3</sub> relative to Pd(OAc)<sub>2</sub> and in refluxing THF, we were also able to realize the second type of domino reaction, the carbopalladation followed by cross-coupling, to form *cis*-stilbenes,<sup>11,12</sup> in a catalytic fashion. *cis*-Substituted stilbenes containing highly substituted phenyl groups were obtained in good yields from the same alkynes and hindered Grignard reagents that form polysubstituted indenenes (Table 4). Our results also suggested that Pd(PPh<sub>3</sub>)<sub>2</sub>Cl<sub>2</sub>-catalyzed reactions of *trans*-1,2-dibromoalkenes with Grignard reagents in refluxing THF to give *cis*-substituted stilbenes<sup>12a</sup> most likely also proceeded with alkynes and **I** as the reaction intermediates.<sup>14</sup>

In summary, we developed two types of Pd-catalyzed domino reactions from the same alkynes and hindered Grignard reagents by controlling the use of ligand and the reaction temperature. We also showed that only benzylic methyl hydrogens might be activated by Pd(II) species. Our study provided an efficient access to useful polysubstituted indenenes and *cis*-substituted stilbenes from simple, commercially available starting materials/reagents. The ligand and temperature factors for controlling the domino reaction pathways identified in this study may also be applicable for other cross-coupling and C-H activation-based tandem/domino reactions. Work toward this direction is underway.

## Supplementary Material

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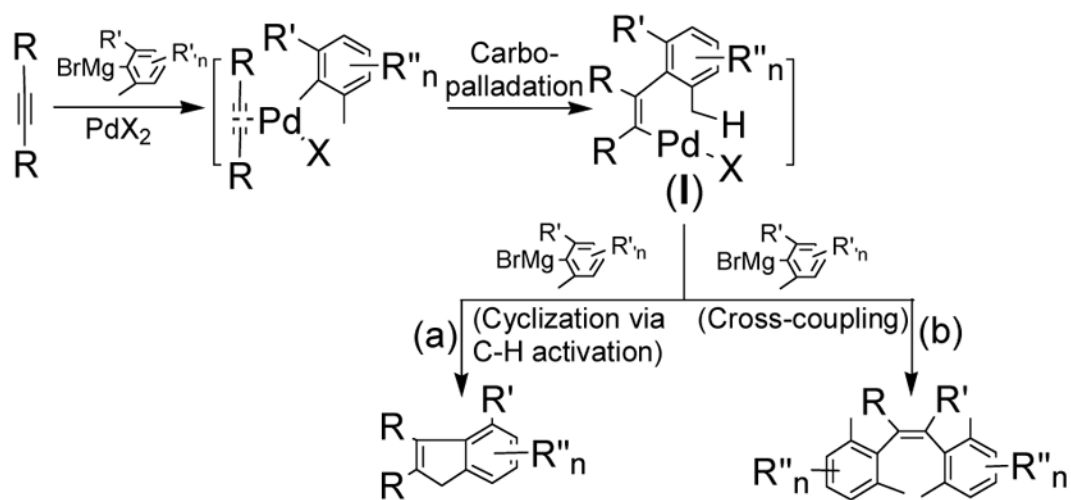
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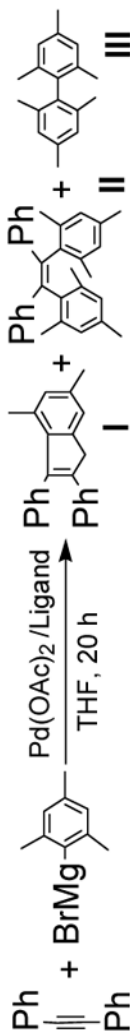
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**Scheme 1.**  
Domino Carbopalladation-Cyclization via  $sp^3$  C-H Activation vs. Domino Carbopalladation-Cross-Coupling Reaction

Pd(OAc)<sub>2</sub>-Promoted Domino Reaction of Diphenylacetylene with 2-Mesitylmagnesium Bromide<sup>a</sup>

Table 1



entry	ligand	temperature	conversion	ratio <sup>b</sup>		
				I	II	III
1	None	R. T.	85%	97:	2:	1
2	None	60	99%	90:	2:	8
3	2 equiv PPh <sub>3</sub>	60	75%	81:	9:	10
4	4 equiv PPh <sub>3</sub>	60	60%	20:	24:	56
5	None	Reflux	99%	93:	3.5:	3.5
6	2 equiv PPh <sub>3</sub>	Reflux	99%	69:	12:	19
7	4 equiv PPh <sub>3</sub>	Reflux	81%	2:	67:	31

<sup>a</sup> Reaction conditions: diphenylacetylene (1.0 equiv), Grignard reagent (2.5 equiv), Pd(OAc)<sub>2</sub> (1 equiv), THF (2 mL), 20 h.

<sup>b</sup> Based on <sup>1</sup>H NMR.

**Table 2**Pd(II)-Catalyzed Domino Reaction of Diphenylacetylene with Mesitylmagnesium Bromide <sup>a</sup>

entry	additives	yield (%)
1	None	< 2 <sup>b</sup>
2	CuCl <sub>2</sub>	< 2 <sup>b</sup>
3	CuSO <sub>4</sub>	< 2 <sup>b</sup>
4	FeCl <sub>3</sub>	15
5	Ag <sub>2</sub> CO <sub>3</sub>	5
6	Br-CH <sub>2</sub> -CH <sub>2</sub> -Br	87

<sup>a</sup> Reaction conditions: diphenylacetylene (1.0 equiv), Grignard reagent (2.5 equiv), THF (2 mL).<sup>b</sup> Conversion based on <sup>1</sup>H NMR

Table 3

Pd(OAc)<sub>2</sub>-Catalyzed Cyclizations of Internal Alkynes with Hindered Grignard Reagents<sup>a</sup>

entry	alkyne	ArMgBr	product	yield(%) <sup>b</sup>
1	Ph≡Ph	BrMg-2,4,6-trimethylphenyl	Ph-2,4,6-trimethylindene	87
2	Ph≡Ph	BrMg-2,4,6-trimethylphenyl	Ph-2,4,6-trimethylindene	85
3	Ph≡Ph	BrMg-2,4,6-trimethylphenyl	Ph-2,4,6-trimethylindene	81
4	Ph≡Ph	BrMg-2,4,6-trimethylphenyl-OMe	Ph-2,4,6-trimethyl-OMe-indene	87
5	MeO-C <sub>6</sub> H <sub>4</sub> -C≡C-C <sub>6</sub> H <sub>4</sub> -OMe	BrMg-2,4,6-trimethylphenyl	Ar-2,4,6-trimethylindene	97
6	MeO-C <sub>6</sub> H <sub>4</sub> -C≡C-C <sub>6</sub> H <sub>4</sub> -OMe	BrMg-2,4,6-trimethylphenyl	Ar-2,4,6-trimethylindene	94
7	C <sub>2</sub> H <sub>5</sub> -C≡C-C <sub>2</sub> H <sub>5</sub>	BrMg-2,4,6-trimethylphenyl	C <sub>2</sub> H <sub>5</sub> -2,4,6-trimethylindene	71
8	C <sub>2</sub> H <sub>5</sub> -C≡C-C <sub>2</sub> H <sub>5</sub>	BrMg-2,4,6-trimethylphenyl	C <sub>2</sub> H <sub>5</sub> -2,4,6-trimethylindene	77
9	Me-C <sub>6</sub> H <sub>4</sub> -C≡C-C <sub>4</sub> H <sub>9</sub>	BrMg-2,4,6-trimethylphenyl	C <sub>4</sub> H <sub>9</sub> -2,4,6-trimethylindene + <i>p</i> -C <sub>6</sub> H <sub>4</sub> -C <sub>4</sub> H <sub>9</sub> -2,4,6-trimethylindene (91:9) <sup>c</sup>	78
10	Ph-C≡C-CH <sub>3</sub>	BrMg-2,4,6-trimethylphenyl	Ph-2,4,6-trimethylindene + Ph-2,4,6-trimethylindene (92:8) <sup>c</sup>	72, <sup>de</sup>
11	Ph-C≡C-CH <sub>3</sub>	BrMg-2,4,6-trimethylphenyl	Ph-2,4,6-trimethylindene + Ph-2,4,6-trimethylindene (90:10) <sup>c</sup>	64, <sup>df</sup>
12	Ph-C≡C-C <sub>2</sub> H <sub>5</sub>	BrMg-2,4,6-trimethylphenyl	C <sub>2</sub> H <sub>5</sub> -2,4,6-trimethylindene + Ph-2,4,6-trimethylindene (89:11) <sup>c</sup>	85
13	Ph-C≡C-C <sub>2</sub> H <sub>5</sub>	BrMg-2,4,6-trimethylphenyl	C <sub>2</sub> H <sub>5</sub> -2,4,6-trimethylindene + Ph-2,4,6-trimethylindene (85:15) <sup>c</sup>	67
14	Ph≡Ph	BrMg-2,4,6-trimethylphenyl	Ph-2,4,6-trimethylindene	69
15	Ph≡Ph	BrMg-2,4,6-trimethylphenyl	Ph-2,4,6-trimethylindene	78
16	Ph≡Ph	BrMg-2,4,6-trimethylphenyl	Ph-2,4,6-trimethylindene	<2% <sup>g</sup>

<sup>a</sup>Reaction conditions: alkyne (1.0 equiv), Grignard reagent (2.5 equiv), Pd(OAc)<sub>2</sub> (3%), 1,2-dibromoethane (1.0 equiv.), THF (2 ml), 60 °C.<sup>b</sup>Isolated yields.<sup>c</sup>Ratio based on <sup>1</sup>H NMR.<sup>d</sup>Reaction condition: room temperature, 30 h.<sup>e</sup>15% Cross-coupling product was observed.<sup>f</sup>21% Cross-coupling product was observed.<sup>g</sup>Reaction time: 45 h.



**Table 4**Pd(OAc)<sub>2</sub>-Catalyzed Domino Carbopalladation-Cross-coupling of Internal Alkynes and Hindered Grignard Reagents<sup>a</sup>

$$R-C\equiv C-R' + BrMg-\text{Ar}-R'' \xrightarrow[\text{Br-CH}_2\text{-CH}_2\text{-Br, THF, reflux, 30-40 h}]{5\%Pd(OAc)_2/20\%PPh_3} R''-\text{Ar}-C(R)=C(R')-\text{Ar}-R''$$

entry	R≡R	BrMg-Ar-R'	R-Ar-C(R)=C(R')-Ar-R''	yield(%) <sup>b</sup>
1	Ph≡Ph	BrMg-3,5-Me <sub>2</sub> -C <sub>6</sub> H <sub>3</sub> -R'	Ph-3,5-Me <sub>2</sub> -C <sub>6</sub> H <sub>3</sub> -C(Ph)=C(Ph)-3,5-Me <sub>2</sub> -C <sub>6</sub> H <sub>3</sub> -R''	71
2	Ph≡Ph	BrMg-2,4,6-Me <sub>3</sub> -C <sub>6</sub> H <sub>2</sub> -R'	Ph-2,4,6-Me <sub>3</sub> -C <sub>6</sub> H <sub>2</sub> -C(Ph)=C(Ph)-2,4,6-Me <sub>3</sub> -C <sub>6</sub> H <sub>2</sub> -R''	65
3	H <sub>3</sub> CO-C <sub>6</sub> H <sub>4</sub> ≡C <sub>6</sub> H <sub>4</sub> -OCH <sub>3</sub>	BrMg-3,5-Me <sub>2</sub> -C <sub>6</sub> H <sub>3</sub> -R'	p-H <sub>3</sub> COC <sub>6</sub> H <sub>4</sub> -3,5-Me <sub>2</sub> -C <sub>6</sub> H <sub>3</sub> -C(p-H <sub>3</sub> COC <sub>6</sub> H <sub>4</sub> -C <sub>6</sub> H <sub>4</sub> -OCH <sub>3</sub> )=C(p-H <sub>3</sub> COC <sub>6</sub> H <sub>4</sub> -C <sub>6</sub> H <sub>4</sub> -OCH <sub>3</sub> )-3,5-Me <sub>2</sub> -C <sub>6</sub> H <sub>3</sub> -R''	72
4	H <sub>3</sub> CO-C <sub>6</sub> H <sub>4</sub> ≡C <sub>6</sub> H <sub>4</sub> -OCH <sub>3</sub>	BrMg-2,4,6-Me <sub>3</sub> -C <sub>6</sub> H <sub>2</sub> -R'	p-H <sub>3</sub> COC <sub>6</sub> H <sub>4</sub> -2,4,6-Me <sub>3</sub> -C <sub>6</sub> H <sub>2</sub> -C(p-H <sub>3</sub> COC <sub>6</sub> H <sub>4</sub> -C <sub>6</sub> H <sub>4</sub> -OCH <sub>3</sub> )=C(p-H <sub>3</sub> COC <sub>6</sub> H <sub>4</sub> -C <sub>6</sub> H <sub>4</sub> -OCH <sub>3</sub> )-2,4,6-Me <sub>3</sub> -C <sub>6</sub> H <sub>2</sub> -R''	69
5	C <sub>2</sub> H <sub>5</sub> ≡C <sub>2</sub> H <sub>5</sub>	BrMg-3,5-Me <sub>2</sub> -C <sub>6</sub> H <sub>3</sub> -R'	C <sub>2</sub> H <sub>5</sub> -3,5-Me <sub>2</sub> -C <sub>6</sub> H <sub>3</sub> -C(C <sub>2</sub> H <sub>5</sub> )=C(C <sub>2</sub> H <sub>5</sub> )-3,5-Me <sub>2</sub> -C <sub>6</sub> H <sub>3</sub> -R''	60
6	C <sub>2</sub> H <sub>5</sub> ≡C <sub>2</sub> H <sub>5</sub>	BrMg-2,4,6-Me <sub>3</sub> -C <sub>6</sub> H <sub>2</sub> -R'	C <sub>2</sub> H <sub>5</sub> -2,4,6-Me <sub>3</sub> -C <sub>6</sub> H <sub>2</sub> -C(C <sub>2</sub> H <sub>5</sub> )=C(C <sub>2</sub> H <sub>5</sub> )-2,4,6-Me <sub>3</sub> -C <sub>6</sub> H <sub>2</sub> -R''	81
7	Ph≡CH <sub>3</sub>	BrMg-3,5-Me <sub>2</sub> -C <sub>6</sub> H <sub>3</sub> -R'	Ph-3,5-Me <sub>2</sub> -C <sub>6</sub> H <sub>3</sub> -C(Ph)=C(CH <sub>3</sub> )-3,5-Me <sub>2</sub> -C <sub>6</sub> H <sub>3</sub> -R''	78
8	Ph≡CH <sub>3</sub>	BrMg-2,4,6-Me <sub>3</sub> -C <sub>6</sub> H <sub>2</sub> -R'	Ph-2,4,6-Me <sub>3</sub> -C <sub>6</sub> H <sub>2</sub> -C(Ph)=C(CH <sub>3</sub> )-2,4,6-Me <sub>3</sub> -C <sub>6</sub> H <sub>2</sub> -R''	74
9	Ph≡CH <sub>3</sub>	BrMg-2,4,6-Me <sub>3</sub> -C <sub>6</sub> H <sub>2</sub> -R'	Ph-2,4,6-Me <sub>3</sub> -C <sub>6</sub> H <sub>2</sub> -C(Ph)=C(CH <sub>3</sub> )-2,4,6-Me <sub>3</sub> -C <sub>6</sub> H <sub>2</sub> -R''	81

<sup>a</sup>Reaction conditions: alkyne (1.0 equiv), Grignard reagent (4.0 equiv), 1,2-dibromoethane (1.5 equiv), Pd(OAc)<sub>2</sub> (5%), PPh<sub>3</sub> (20%), THF (2 mL), refluxing, 30–40 h.

<sup>b</sup>Isolated yields.