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Gold-Catalyzed 1,2-Migration of Silicon, Tin, and Germanium *en* route to C-2 Substituted Fused Pyrrole-Containing Heterocycles

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Abstract

An efficient method for the synthesis of fused pyrroloheterocycles from diverse propargyl-substituted heterocycles in the presence of Au-catalyst has been developed. The cascade transformation proceeds via alkyne-vinylidene isomerization with concomitant 1,2-shift of hydrogen, silyl and stannyl groups. Remarkably, it was also shown that previously unknown 1,2-migration of a germyl group upon alkyne-vinylidene rearrangement occurs under these reaction conditions. This method allows for mild and efficient synthesis of diverse C-2 substituted N-containing heterocycles.

Alkyne-vinylidene isomerization is a mechanistically interesting 1 and synthetically useful transformation. For example, McDonald used this transformation as the key step in efficient synthesis of heterocycles (eq 1). It has also been shown that various groups (G) can undergo 1,2-migration upon alkyne-vinylidene isomerization, as demonstrated by Iwasawa (G = Hal, M = W), Fürstner (G = Hal, M = Au), Katayama ($G = SiR_3$, M = Ru), and Kawakami ($G = SnR_3$, M = Ru), eq 2). However, to the best of our knowledge, no examples of synthesis of heterocycles with 1,2-migration of groups other than H have ever been reported. Thus, we reasoned that development of alternative routes toward heterocycles, which proceed with 1,2-group migration, would be desirable, as they would allow for the synthesis of densely substituted molecules. Herein, we wish to report a new Au-catalyzed cascade cycloisomerization of propargylic derivatives of N-containing heterocycles into fused pyrrolecontaining heterocycles. The cascade transformation involves 1,2-migration of silyl-, stannyl-, and even the *previously unknown migration of a germyl group*, and allows for efficient synthesis of various fused pyrroloheterocycles functionalized at position C-2 (eq 3).

We have recently reported the cycloisomerization of alkynylpyridines into indolizines (eq 4). ⁸ The reaction proceeds via a base-assisted propargyl-allenyl isomerization to intermediate i, followed by its cyclization into the indolizine core. This transformation presumes two formal hydrogen migrations, and thus is limited to the preparation of C-1, 2 unsubstituted indolizines.

$$\begin{array}{c|c}
H & H \\
R & base
\end{array}$$

$$\begin{array}{c|c}
H & H \\
\hline
N & R
\end{array}$$

$$\begin{array}{c|c}
H & H \\
\hline
N & R
\end{array}$$

$$\begin{array}{c|c}
H & H \\
\hline
N & R
\end{array}$$

$$\begin{array}{c|c}
H & (4)
\end{array}$$

Naturally, as we were interested in developing approaches toward C-1 substituted heterocycles, we turned our attention to the cycloisomerization of easily available non-conjugated propargylpyridine 1 (eq. 5). After catalyst optimization, 10 it was found that 1, in the presence of Au(I) or Au(III) salts, 11 undergoes smooth cycloisomerization into C-1 substituted indolizine 2. It is reasonable to propose that this transformation operates through allenyl intermediate i (e.g. via another mode of the propargyl-allenyl cycloisomerization depicted in eq. 4). Alternatively, this reaction may proceed via isomerization of terminal alkyne 1 into gold-vinylidene intermediate ν , which subsequently cycloisomerizes into the heteroaromatic structure 2 (eq. 5).

OTBS

formal
$$[1,3]H^{\sim}$$
 $[Au]$
 $[1,2]H^{\sim}$
 $[1,2]H^{\sim}$
 $[1,2]H^{\sim}$
 $[Au]$
 $[Au]$

In order to clarify whether this reaction proceeds via an allenyl (i) or vinylidene (v) intermediate, we examined cycloisomerization of TMS-substituted propargylpyridine 3a in the presence of Au-catalyst. It was hypothesized that a prototropic isomerization (via intermediate i)¹² would lead to indolizine with the silyl group attached to the C-3 position, whereas the silyl group would be at C-2 if alkyne-vinylidene isomerization operates (via intermediate v). To our great delight, it was found that 3a, in the presence of AuBr₃ (2.0 mol %) in toluene at 50°C, underwent smooth cycloisomerization to afford indolizine 4a with TMS group migration to the C-2 position, as the sole regioisomer in 63% yield (Table 1, entry 1). It deserves mentioning that in indolizines, the C-2 site is an unfunctionalizable position, and its substituent has to be introduced prior to cyclization. 13

Motivated by the importance of differently substituted fused pyrroloheterocycles, ¹⁴ and encouraged by the successful cycloisomerization of TMS-containing substrate **3 a**, we examined various propargyl heterocycles in this transformation (Table 1). Gratifyingly, the stannyl group, known to undergo migration upon alkyne-vinylidene isomerization ⁷ (entry 2), underwent smooth migration to give 2-stannyl indolizine **4b** in good yield. Remarkably, we also found that unprecedented 1,2-germyl migration can also occur to produce 2-germylindolizine **4c** in excellent yield (entry 3). Notably, this cycloisomerization appeared to

be general with regard to the heterocyclic core. Other heterocyclic systems, such as isoquinoline (entry 5), quinoxaline (entries 6 and 7), pyrazine (entries 8 and 9) and thiazole (entry 10), reacted smoothly, producing fused pyrroloheterocycles in good to excellent yields.

We propose the following mechanistic rationale for this novel transformation. First, isomerization of alkyne 3 results in the formation of vinylidene v^{15} (Scheme 1), followed by nucleophilic attack of the nitrogen lone pair at the vinylidene carbon, resulting in formation of zwitterion 5. The latter can either undergo a series of 1,2-hydride shifts (Path A), or a deprotonation-protonation sequence (Path B) to furnish $4.^{16}$ In order to verify which mechanism operates, we performed a deuterium-labeling experiment utilizing isotopically homogeneous propargyl pyridine 3k. It was found that under standard cycloisomerization conditions, the reaction produced indolizine 4k with equal distribution of deuterium between positions C-2 and C-3, 17 thus strongly supporting Path A. 18

In summary, we have developed a new mild cascade cycloisomerization of propargyl *N*-containing heterocycles into various types of *N*-fused pyrroloheterocycles in the presence of gold catalyst. The reaction proceeds via alkyne-vinylidene isomerization with concomitant 1,2-migration

of H, silyl-, and stannyl groups, as well as previously unknown 1,2-migration of a germyl group, giving easy access to a variety of C-2 functionalized heterocycles.

Supplementary Material

Refer to Web version on PubMed Central for supplementary material.

Acknowledgment

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- 15. Au(I) is most likely the active catalyst. In the case of employment of $AuBr_3$ as a precatalyst, the latter can be reduced to Au(I) species via various redox processes. See ref 11b for discussion.
- 16. A referee pointed out possible Au-catalyzed C-3→C-2 TMS group migration after cyclization, and brought to our attention a reference on Aucatalyzed migration of alkyl group in indole series: Alfonsi M, Arcadi A, Aschi M, Bianchi G, Marinelli F. J. Org. Chem 2005;70:2265. [PubMed: 15760214] However, a test experiment with pyrroloquinoxaline 8, C-3 TMS analog of 4g, ruled out this possibility. 9
- 17. Control experiment indicated no deuterium scrambling between 4h and D₂O occurred under the same reaction conditions. For deuterium scrambling in N-unsubstituted pyrrole ring in the presence of Au(I) complexes, see ref ^{11h}.
- 18. Equal distribution of deuterium between positions C-2 and C-3 is possible via Path A if there is no H/D kinetic isotope effect (transformation 6 to 4). Obviously, Path B cannot explain the observed scrambling of deuterium at C-2.

Scheme 1. Proposed Mechanism for Cascade Cycloisomerization

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Synthesis of Fused Pyrrole-Containing Heterocycles

	o d	_
OTBS)_Z D-Q N-C	4a-
TO (0)	AuBr ₃ 2 mol% C c c c c c c c c c c c c c c c c c c	4a-

Yield, %a	63 <i>b</i>	64^d	62^{b}	62	81	94	78	72	87^e	56
	4a	4b	94	4d	4 e	4f	48 8	4h	.4	į.
Product	OTBS SiMe ₃	OTBS Subus	N OTBS	OTBS H	Pa Substitution of the sub	N N N N N N N N N N N N N N N N N N N	T OTBS	SiMe ₃ OTBS	N OTBS	OTBS S N S N S
Time, h	1.5	0.5	0.5	1.5	0.5	2.0	4.0	4.5	0.5	3.5
T (°C)	50	50	25	50	25	50	50	09	25	50
9	SiMe ₃	SnBu_3	GeMe ₃	н	н	Н	$SiMe_3$	н	$SiMe_3$	SiMe ₃
#		71	8	4	١٨	9	7	∞	6	10

 $^{^{\}rm C}$ Reaction was performed in 5.0 mmol scale in the presence of AuCl catalyst (0.5 mol %).

 $^{^{\}it a}$ Isolated yield; reactions performed in 0.5 mmol scale.

 $^{^{\}it b}$ Yield over 2 steps.

 $[^]d$ NMR yield.

eAuCl was used as a catalyst.