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A Methylated Phosphate Group and Four Amide-linked Acyl Chains in *Leptospira interrogans* Lipid A. The Membrane Anchor of an Unusual Lipopolysaccharide that Activates TLR2*

Nanette L. S. Que-Gewirth[‡], Anthony A. Ribeiro^{‡,§}, Suzanne R. Kalb[¶], Robert J. Cotter[¶], Dieter M. Bulach^{||}, Ben Adler^{||,‡‡}, Isabelle Saint Girons^{**}, Catherine Werts^{**}, and Christian R. H. Raetz[‡]

[‡] Department of Biochemistry, Durham, North Carolina 27710

[§] Duke NMR Spectroscopy Center and Department of Radiology, Duke University Medical Center, Durham, North Carolina 27710

[¶] Middle Atlantic Mass Spectrometry Laboratory, Department of Pharmacology and Molecular Sciences, The Johns Hopkins University School of Medicine, Baltimore, Maryland 21205

^{||} Australian Bacterial Pathogenesis Program, Department of Microbiology, Monash University, Clayton, Victoria 3800, Australia

^{**} Unité de Bactériologie Moléculaire et Médicale, Institut Pasteur, Paris, 75015, France

Abstract

Leptospira interrogans differs from other spirochetes in that it contains homologs of all the *Escherichia coli* *lpx* genes required for the biosynthesis of the lipid A anchor of lipopolysaccharide (LPS). LPS from *L. interrogans* cells is unusual in that it activates TLR2 rather than TLR4. The structure of *L. interrogans* lipid A has now been determined by a combination of matrix-assisted laser desorption ionization time-of-flight mass spectrometry, NMR spectroscopy, and biochemical studies. Lipid A was released from LPS of *L. interrogans* serovar Pomona by 100 °C hydrolysis at pH 4.5 in the presence of SDS. Following purification by anion exchange and thin layer chromatography, the major component was shown to have a molecular weight of 1727. Mild hydrolysis with dilute NaOH reduced this to 1338, consistent with the presence of four *N*-linked and two *O*-linked acyl chains. The lipid A molecules of both the virulent and nonvirulent forms of *L. interrogans* serovar Icterohaemorrhagiae (strain Verdun) were identical to those of *L. interrogans* Pomona by the above criteria. Given the selectivity of *L. interrogans* LpxA for 3-hydroxylaurate, we propose that *L. interrogans* lipid A is acylated with *R*-3-hydroxylaurate at positions 3 and 3' and with *R*-3-hydroxypalmitate at positions 2 and 2'. The hydroxyacyl chain composition was validated by gas chromatography and mass spectrometry of fatty acid methyl esters. Intact hexa-acylated lipid A of *L. interrogans* Pomona was also analyzed by NMR, confirming the presence a β -1',6-linked disaccharide of 2,3-diamino-2,3-dideoxy-D-glucopyranose units. Two secondary unsaturated acyl chains are attached to the distal residue. The 1-position of the disaccharide is derivatized with an axial phosphate moiety, but the 4'-OH is unsubstituted. ¹H and ³¹P NMR analyses revealed that the

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^{‡‡}To whom correspondence should be addressed: Dept. of Biochemistry, Duke University Medical Center, P.O. Box 3711, Durham, NC 27710. Tel.: 919-684-5326; Fax: 919-684-8885; E-mail: raetz@biochem.duke.edu.

1-phosphate group is methylated. Purified *L. interrogans* lipid A is inactive against human THP-1 cells but does stimulate tumor necrosis factor production by mouse RAW264.7 cells.

Nearly all of the diverse eubacteria that are enclosed by two membranes synthesize lipid A as the hydrophobic anchor of their outer membrane lipopolysaccharide (LPS)¹ (1,2). Several spirochetes of clinical importance, such as *Treponema pallidum*, *Treponema denticola*, and *Borrelia burgdorferi*, possess an outer membrane but do not make LPS (3–5). Accordingly, they lack the *lpx* genes (6,7), which are required for lipid A assembly in *Escherichia coli* and other Gram-negative organisms (2). The absence of lipid A in the outer membranes of *T. pallidum*, *T. denticola*, and *B. burgdorferi* may be compensated for by alternative lipids (8), lipoproteins (9), or other complex glycoconjugates (4). Whatever the explanation, spirochetes lacking LPS are not easily cultivated outside of their hosts (10,11).

Disease-causing serovars of *Leptospira interrogans* spp. are members of a distinct spirochete group (12,13) that can survive or proliferate either in a mammal or in the environment, typically in fresh water contaminated by the urine of infected animals (14). *L. interrogans* causes a hemorrhagic fever in humans known as Weil's disease, which may be fatal in untreated cases because of liver, kidney, or pulmonary damage (14,15). Early biochemical, serologic, and genetic studies showed that LPS is present in *Leptospira*, but its covalent structure has not been characterized (12,16–20). However, numerous studies have shown that leptospiral LPS possesses much lower endotoxic activity than typical Gram-negative LPS (21). Werts *et al.* (22) found that highly purified *L. interrogans* LPS is unusual because it activates TLR2 rather than TLR4. The latter is the classical signaling receptor of the innate immune system that detects the lipid A moiety of most other Gram-negative LPSs (23–26).

The recently completed sequencing of the *L. interrogans* serovar Lai genome strongly supports the idea that these spirochetes synthesize lipid A and LPS, because the genome encodes a complete set of *Lpx* orthologs and LPS-related glycosyl transferases (13). Likewise, the earlier studies of Adler and coworkers demonstrated the existence of typical O-antigen gene clusters in various strains of *L. interrogans*, indicating that some form of LPS must be present (12, 27).

Given the unusual bioactivity of *L. interrogans* LPS toward TLR2 (22) and the lack of structural studies, we now report methods for the purification and characterization of *L. interrogans* lipid A. A combination of mass spectrometry, NMR spectroscopy, bioinformatics, and enzymology was used to show that *L. interrogans* makes lipid A molecules in which the usual glucosamine units are replaced with the analog 2,3-diamino-2,3-dideoxy-D-glucopyranose (GlcN3N). As anticipated from studies with *Acidithiobacillus ferrooxidans*, described in the preceding manuscripts (28,29), the *L. interrogans* genome (13) contains significant full-length orthologs of the enzymes GnnA and GnnB. These proteins are needed to convert UDP-GlcNAc to UDP 2-acetamido-3-amino-2,3-dideoxy- α -D-glucopyranose (UDP-GlcNAc3N), a novel sugar nucleotide that is the key precursor of lipid A molecules containing GlcN3N units (28,29). The lipid A of *L. interrogans* contains a novel 1-phosphate residue that is capped with a methyl group. Methylated phosphate residues are uncommon in biology (30–32) and without precedent in lipid A biochemistry (2,33). *L. interrogans* lipid A lacks a 4'-phosphate moiety. Two unsaturated ester-linked secondary acyl chains are present on the distal unit. *L. interrogans* lipid A is inactive in the limulus lysate assay and against human THP-1 cells, but it does stimulate mouse macrophage tumor cells with about one-tenth the potency of *E. coli*

¹The abbreviations used are: LPS, lipopolysaccharide; Kdo, 2-keto-3-deoxy-D-manno-octulosonic acid; UDP-GlcNAc3N, UDP 2-acetamido-3-amino-2,3-dideoxy- α -D-glucopyranose; GlcN3N, 2,3-diamino-2,3-dideoxy-D-glucopyranose; COSY, correlation spectroscopy; HMQC, heteronuclear multiple-quantum coherence; NOE, nuclear Overhauser effect; MALDI-TOF, matrix-assisted laser desorption ionization time-of-flight.

lipid A. Our purification and proposed structure for *L. interrogans* lipid A should facilitate further pharmacological studies. A preliminary communication of our structure has appeared in abstract form (34).

EXPERIMENTAL PROCEDURES

Materials

Glass-backed 0.25-mm silica gel 60 TLC plates were obtained from Merck. Chloroform, ammonium acetate, and sodium acetate were from EM Sciences, whereas pyridine, methanol, and formic acid were from Mallinckrodt. Deuterated solvents (CD_3OD , CDCl_3 containing 0.1% tetramethylsilane, and D_2O) and 5-mm NMR tubes were from Sigma-Aldrich. Triton X-100 and the bicinchoninic assay kit were from Pierce.

Bacterial Strains

L. interrogans serovar Pomona (strain L170) (12,19) was grown in a special medium with added pyruvate at 30 °C (35). The cells from stationary phase cultures were harvested and lyophilized. *L. interrogans* serovar Icterohaemorrhagiae (strain Verdun) was from the Centre de Reference des Leptospires (Institut Pasteur, Paris, France). Both the virulent and the avirulent variants of the Verdun strain were cultured as described previously (22). The bacteria were grown in EMJH medium at 30 °C under aerobic conditions to a density of $\sim 5 \times 10^8$ cells/ml (36,37). The preliminary sequence data for *T. denticola* were obtained from the Baylor College of Medicine (www.hgsc.bcm.tmc.edu).

Purification of LPS

The method described by Westphal and Jann (38) was used to extract LPS from lyophilized cells of *L. interrogans* serovar Pomona (5 g dry weight) (19). LPS from *L. interrogans* serovar Icterohaemorrhagiae (strain Verdun) was extracted using a modification of the hot phenol-water method (22,38). Typically, 1 mg of LPS was recovered from 10^{12} bacteria. The LPS in the phenol phase was subjected to extensive dialysis against hot water (70 °C). The dialyzed material was clarified twice by low speed centrifugation ($3000 \times g$ for 15 min at 10 °C), and the LPS was collected by ultracentrifugation (3 h at $100,000 \times g$ at 10 °C). The pellet was resuspended in endotoxin-free water. The ultracentrifugation step was repeated two or three times until the resuspended LPS showed no absorbance at 260 and 280 nm, after which it was lyophilized and weighed. Because phenol-extracted LPS may retain some impurities that result in TLR2-dependent activation of cells, LPS preparations intended for biological studies are further purified using a procedure that removes LPS-associated proteins (39–41). However, these steps were omitted for the LPS used in the present work.

Purification of Lipid A Released from *L. interrogans* LPS

Lipid A was released from the *L. interrogans* LPS by 100 °C hydrolysis at pH 4.5 in the presence of SDS (42) as described previously, followed by Bligh-Dyer extraction (43). In a typical preparation, 150 mg of crude LPS was resuspended in 20 ml of 12.5 mM sodium acetate, pH 4.5, containing 1% SDS in a 150-ml Corex glass centrifuge bottle. The mixture was then placed in a boiling water bath for 30 min (42). The crude lipid A was extracted and then purified on a 2-ml DEAE-cellulose column (Whatman DE-52), equilibrated in the acetate form in $\text{CHCl}_3/\text{MeOH}/\text{H}_2\text{O}$ (2:3:1, v/v/v) (42,44–47). The entire lipid A sample was dissolved in 9 ml of $\text{CHCl}_3/\text{MeOH}/\text{H}_2\text{O}$ (2:3:1, v/v/v) and loaded onto the column. Most of the putative *L. interrogans* lipid A eluted with CHCl_3 , MeOH, 30 mM NH_4Ac (2:3:1, v/v/v), suggesting that it is not strongly anionic (42,45,48). Fractions from the 30 mM NH_4Ac wash were pooled and, following removal of the solvents (42), stored at -20 °C.

Further purification of the *L. interrogans* lipid A by preparative TLC was carried out as described by Que *et al.* (42,48) for *Rhizobium etli* lipid A. Briefly, the lipid A was redissolved in 2 ml of CHCl₃, MeOH (4:1, v/v), and a 0.2–0.5-mg sample was applied in 10- μ l spots along a line at the origin of four 20 \times 20-cm Silica Gel 60 analytical TLC plates (0.25-mm thickness), which were developed with the solvent CHCl₃, pyridine, 88% formic acid, MeOH, H₂O (60:35:10:5:2, v/v/v/v/v). The samples were eluted (42,48) and passed through another 0.5-ml DEAE column to remove residual silica chips. The purified preparations were stored dry at –20 °C.

O-Deacylation of *L. interrogans* Lipid A by Mild Alkaline Hydrolysis

Complete hydrolysis of all ester-linked fatty acids was achieved by resuspending ~0.1 mg of pure lipid A in a 1-ml glass vial equipped with a Teflon-lined cap in 120 μ l of CHCl₃, MeOH, 0.6 M NaOH (2:3:1, v/v/v) and allowing the hydrolysis to proceed at room temperature for 30–60 min. The final mild alkaline hydrolysis mixture was converted into an acidic two-phase Bligh-Dyer system by the addition of 220 μ l of CHCl₃, 210 μ l of MeOH, and 206 μ l of 0.1 M HCl. The lower phase was dried with a stream of N₂, and the sample was stored at –20°C. The deacylated lipid A was further purified using a 0.25-ml DEAE-cellulose column, equilibrated, and eluted as above. The *O*-deacylated lipid A was eluted with CHCl₃, MeOH, 30 mM NH₄Ac (2:3:1, v/v/v), was recovered by acidic two-phase Bligh-Dyer partitioning, and was stored dry at –20 °C prior.

To obtain lipid A that is partially *O*-deacylated, purified lipid A (~0.2 mg) was dissolved in 600 μ l of CHCl₃, MeOH, 0.6 M NaOH (2:3:1, v/v/v). After incubation for only 5 min at room temperature, the solution was converted into an acidic two-phase Bligh-Dyer mixture by the addition of 300 μ l of CHCl₃, 200 μ l of MeOH, and 350 μ l of 0.1 M HCl. After mixing, the lower phase was dried with a stream of N₂ and stored at –20 °C.

MALDI-TOF Mass Spectrometry of *L. interrogans* Lipid A

Spectra were acquired in the negative ion and the positive ion linear modes using a Kratos Analytical (Manchester, UK) MALDI-TOF mass spectrometer, equipped with a 337-nm nitrogen laser, a 20-kV extraction voltage, and time-delayed extraction (42). Each spectrum was the average of 50 shots. The lipid A samples were prepared for MALDI-TOF analysis by depositing 0.3 μ l of the sample dissolved in chloroform, methanol (4:1, v/v), followed by 0.3 μ l of the matrix, which was a mixture of saturated 6-aza-2-thiothymine in 50% acetonitrile and 10% tribasic ammonium citrate (9:1, v/v). The sample mixtures were allowed to dry at room temperature. Hexa-acylated lipid A 1,4'-bisphosphate from *E. coli* (Sigma) set at *m/z* 1797 was used as an external standard for calibration.

Fatty Acid Analysis by Gas Chromatography/Mass Spectrometry

Briefly, ~1 mg of purified *L. interrogans* lipid A was dissolved in 200 μ l of toluene and 400 μ l of fresh 1% sulfuric acid in methanol. The sample was then hydrolyzed for 2 h at 100 °C, after which 1 ml of hexane was added. The mixture was mixed for 30 s, followed by centrifugation for ~10 min at room temperature. The upper phase containing the methyl esters was removed and dried down under a stream of nitrogen. Fatty acid methyl esters were analyzed at the University of Minnesota Mass Spectrometry Facility Department of Chemistry using a Finnigan MAT 95 mass spectrometer coupled to a Hewlett-Packard Series II model 5890 gas chromatograph.

NMR Analysis

NMR spectroscopy was carried out at the Duke University NMR Spectroscopy Center (48, 49). The *L. interrogans* lipid A was dissolved in 0.6 ml of CDCl₃, CD₃OD, D₂O (2:3:1, v/v/v)

in a 5-mm NMR tube. Proton and carbon chemical shifts are reported relative to internal tetramethylsilane at 0.00 ppm.

NMR spectra were recorded on Varian Unity 500 or 600 NMR spectrometers, each equipped with a Sun Ultra 10 computer and a 5-mm Varian probe. Two-dimensional NMR experiments (COSY, NOE spectroscopy, total correlation spectroscopy, and HMQC) were performed at 600 MHz (48,50). Directly detected ^1H -decoupled ^{31}P NMR spectra were recorded at 202.37 MHz with a spectral window of 12143.3 Hz digitized into 25,280 data points (digital resolution of 1 Hz/point or ~ 0.005 ppm/point), a 60° pulse flip angle ($8 \mu\text{s}$), and a 1.6-s repeat time. ^{31}P chemical shifts were referenced to 85% H_3PO_4 at 0.000 ppm. Inverse decoupled difference spectra were recorded as ^1H -detected ^{31}P -decoupled heteronuclear NMR experiments (49, 50).

RESULTS

Weak Binding of *L. interrogans* Lipid A to DEAE-cellulose

The lipid A recovered from *L. interrogans* LPS after hydrolysis at 100°C in 12.5 mM sodium acetate, pH 4.5, migrated as a discrete band during TLC (Fig. 1). Given its elution from DEAE-cellulose with CHCl_3 , MeOH, 30 mM aqueous ammonium acetate (2:3:1, v/v/v), this substance is predicted to have a net negative charge of about -1 (42,45). Typical lipid A 1,4'-bisphosphate species from *E. coli* elute with CHCl_3 , MeOH, 240 mM ammonium acetate (2:3:1, v/v/v) (46,50), because they can carry up to four negative charges. The weak binding of the *L. interrogans* material to DEAE-cellulose resembles what is seen with *R. etli* lipid A in which both phosphate groups are missing and replaced with carboxylate-containing sugars (42).

Identical Lipid A Molecules in *L. interrogans* Serovars Pomona and Icterohaemorrhagiae

Negative ion mode MALDI-TOF mass spectrometry of the lipid A isolated from *L. interrogans* serovar Pomona (Fig. 2A) revealed a major peak at m/z 1726.4, likely representing $[\text{M} - \text{H}]^-$ of the major molecular species. Smaller peaks differing by 28 atomic mass units presumably reflect fatty acyl chain length heterogeneity. The lipid A of both the virulent and the avirulent forms *L. interrogans* strain Verdun (22) yielded virtually identical spectra (Fig. 2, B and C). Because much greater quantities of LPS were available from the Pomona serovar (12,19), all of the NMR studies were performed with Pomona lipid A.

One-dimensional NMR Spectroscopy *L. interrogans* Lipid A

The ^{31}P NMR spectrum of *L. interrogans* lipid A at 202 MHz revealed a single resonance at 0.676 ppm (Fig. 3), demonstrating that this material contains only one phosphate group. The chemical shift of the phosphorus signal is in the range of what is expected for a monophosphomonoester unit (-0.5 to $+2$ ppm) under these conditions (50). However, a monophosphodiester group cannot be excluded based solely upon the chemical shift position. The presence of only one phosphate group is consistent with the low affinity of *L. interrogans* lipid A for DEAE-cellulose.

The one-dimensional 600 MHz ^1H NMR spectrum of *L. interrogans* lipid A in CDCl_3 , CD_3OD , D_2O (2:3:1, v/v/v) revealed well resolved peaks in the sugar (3.5–5.5 ppm) and acyl chain (0.9–2.8 ppm) regions (Fig. 4, upper and left margins), similar to *E. coli* or *R. etli* lipid A (48–50). However, the *L. interrogans* spectrum contains an unusually prominent doublet at 3.61 ppm, which integrates to three protons and shows an apparent coupling constant of 10.9 Hz (Fig. 4, left arrow). This kind of signal had not been seen previously in lipid A from diverse sources (47–50).

NMR Evidence for a Methylated 1-Phosphate Moiety

The 3.61-ppm doublet in the ^1H NMR spectrum of *L. interrogans* lipid A does not show any cross-peaks in the ^1H - ^1H COSY analysis (Fig. 4). To determine whether or not this signal arises from a methylated phosphate group, selective inverse decoupling difference spectroscopy was used to detect possible heteronuclear coupling of the phosphorus atom to the 3.61-ppm doublet (49,50). The difference spectrum (Fig. 5C) of the off and on resonance ^{31}P -decoupled ^1H NMR spectra (Figs. 5, A and B) demonstrated the collapse of the 3.61 ppm doublet to a singlet during phosphorus irradiation. In addition, the difference spectrum showed simplification of a proton signal at 5.44 ppm to a doublet and of another proton at 4.05 ppm to a double-doublet (Fig. 5C, inset). The latter chemical shifts and peak shapes are consistent with phosphorus coupling to the H-1 and H-2 atoms, respectively, of a proximal α -linked glucopyranoside unit (49,50). The heteronuclear coupling data provide strong evidence that *L. interrogans* lipid A contains an unprecedented mono-phosphodiester unit bridging the C-1 atom of the proximal sugar and a methyl group.

Confirmation of a Methyl Group on the 1-Phosphate Moiety by MALDI-TOF Mass Spectrometry

The negative ion MALDI-TOF mass spectra of intact *L. interrogans* lipid A (Figs. 2 and 6A) show an intense peak with m/z near 1726 atomic mass units, interpreted as the molecular ion $[\text{M} - \text{H}]^-$. Additional structural information is contained in the positive ion mode spectrum (Fig. 6B). The B_1^+ oxonium ion (Fig. 6B) forms during fragmentation of the disaccharide glycosidic linkage (Fig. 7) (51) and reflects the mass of the distal sugar unit. The B_2^+ ion (Fig. 7) is generated by loss of the substituent attached to the 1-position in the proximal sugar (51). The mass of the group attached to the anomeric carbon is determined by comparing the m/z of the B_2^+ ion to the molecular weight derived from the $[\text{M} - \text{H}]^-$ ion (Fig. 6A). The mass of the proximal unit (without the substituent at the 1-position) is determined from the difference of B_1^+ and B_2^+ .

Fig. 6B shows prominent peaks at m/z 1003.0 and 1614.1, which are interpreted as the B_1^+ and B_2^+ ions, respectively, of *L. interrogans* lipid A (Fig. 7). Under the ionization conditions employed, the molecular ion $[\text{M} - \text{H}]^-$ ($m/z = 1727.4$) is not detected (Fig. 6B). The B_2^+ ion at m/z 1614.1 is 112.4 atomic mass units less than the molecular weight deduced from the negative mode spectrum in Fig. 6A (1726.5). If *L. interrogans* lipid A contained an unsubstituted phosphate group at the 1-position, as is present in *E. coli* lipid A, the B_2^+ ion should be 97.0 atomic mass units smaller than the molecular weight predicted from the negative mode spectrum (52). Given the experimental error of the MALDI-TOF measurements, the discrepancy of 15.4 atomic mass units (112.4 - 97.0) is consistent with the presence of a methyl substituent on the 1-phosphate group of *L. interrogans* lipid A, as deduced from the above NMR experiments (Fig. 5).

The difference in mass between the B_1^+ and B_2^+ ions is 611.1 atomic mass units (Fig. 6B), suggesting the presence of two acyl chains on the proximal unit of *L. interrogans* lipid A. Given the absolute selectivity of *L. interrogans* LpxA for 3-hydroxy lauroyl-ACP and UDP-GlcNAc3N (28), the GlcN3N 3-position must be acylated with 3-hydroxylaurate. To account for the 611.1 atomic mass units difference in the B_1^+ and B_2^+ ions (Fig. 6B), the acyl chain at the GlcN3N 2-position could be 3-hydroxypalmitate (Fig. 7). In fact, 3-hydroxylaurate and 3-hydroxypalmitate were predominant components in the fatty acid analysis (data not shown). The size of the B_1^+ ion at m/z 1003.0 atomic mass units suggests that four acyl chains are attached to the distal unit (Fig. 7).

Mild Alkaline Hydrolysis of *L. interrogans* Lipid A

Exposure of lipid A or lipid A precursors to aqueous triethylamine at 37 °C releases unsubstituted *O*-linked 3-hydroxyacyl chains (42,53). Mild triethylamine does not remove *O*-linked acyloxyacyl moieties or *O*-linked normal fatty acids under standard conditions. Exposure of *L. interrogans* lipid A to aqueous triethylamine at 37 °C did not alter its molecular weight, as judged by mass spectrometry (not shown), demonstrating the absence of unsubstituted, *O*-linked 3-hydroxyacyl chains.

Treatment of *L. interrogans* lipid A with 0.1 M NaOH for 30 min shifted $[M - H]^-$ from m/z 1725.5 to 1336.5 atomic mass units (Fig. 6, A and C), suggesting the release of two ester-linked acyl chains with the masses of C12:1 and C14:1 (Fig. 7A). A 5-min exposure to 0.1 M NaOH yielded partially *O*-deacylated intermediates, as judged by the appearance of peaks at m/z 1545.2 and 1517.4 atomic mass units (Fig. 6B), consistent with the presence of ester-linked C12:1 and C14:1 moieties.

The MALDI-TOF analyses of the partially and completely hydrolyzed *L. interrogans* lipid A samples in the positive mode (Fig. 6, D and F, respectively) are in accord with the negative mode data (Fig. 6, C and E). Importantly, the B_1^+ ion is shifted from m/z 1003.0 atomic mass units to m/z 614.1 atomic mass units after complete hydrolysis (Fig. 6, B versus F), demonstrating conclusively that both ester-linked acyl chains must be located on the distal unit of *L. interrogans* lipid A. The difference in mass between the B_1^+ and B_2^+ ions following complete hydrolysis is 613.2 atomic mass units (Fig. 6F), which is the essentially same as observed for the untreated material (Fig. 6B) and in good agreement with the expected value of 612.9 for the structure shown in Fig. 7.

The molecular weight of the dilute NaOH treated *L. interrogans* lipid A is 1337.5 (Fig. 6E). Given that the corresponding B_2^+ ion is observed at m/z 1227.3 atomic mass units (Fig. 6F), the molecular mass of the substituent present at the 1-position of this substance is 110.2 (*i.e.* 1337.5 - 1227.3). This result in good agreement with what is expected for the loss of a methylated phosphate residue (111.0 atomic mass units) (Fig. 7).

Two-dimensional ^1H NMR Analysis of *L. interrogans* Lipid A

All of the chemical shifts and coupling constants for *L. interrogans* lipid A are summarized in Table I, using the proposed structure and numbering scheme shown in Fig. 7A. Many of the protons assigned in the ^1H - ^1H COSY of *L. interrogans* lipid A (Fig. 4) appear at similar shifts as their *E. coli* counterparts (49,50). For instance, the H-1 anomeric proton of the proximal sugar at 5.44 ppm and the H-1' anomeric proton of the distal sugar at 4.47 ppm are easily recognized (Fig. 4) and serve as convenient entry points for the evaluation of the sugar region connectivity (3.5–4.5 ppm). The COSY (Fig. 4) and total correlation spectroscopy (not shown) data permit the sequential identification of H-2 through H-6a and H-6b for each hexose ring (Fig. 4 and Table I). The small $J_{1,2}$ coupling (3.2 Hz) and the large $J_{2,3}$, $J_{3,4}$ and $J_{4,5}$ couplings (9 to 11 Hz) suggest that the proximal pyranose ring is in the α -anomeric configuration with axially disposed H-2, H-3, H-4, and H-5 protons (Fig. 7). The large $J_{1',2'}$ coupling (7.1 Hz) shows that the distal sugar is in the β -configuration (Fig. 7). NOE spectroscopy analysis (not shown) demonstrates the following NOE dipolar interactions: 1) from the resolved H-1' to H-3' and to H-5' within the distal pyranose unit and to the H-6a and H-6b of the proximal sugar; 2) from H-1 to H-2 and from H-2 to H-4 in the proximal sugar; and 3) from H-5 to H-3 and to H-6a and H-6b in the proximal sugar. The NOE from H-1' to H-6a and H-6b is diagnostic for the β -1',6 linkage (49,50). The multiple 1,3 diaxial and single axial-equatorial (H-1 to H-2) intramolecular NOEs confirm that both sugar rings adopt the chair conformations with a β -linkage between the proximal α -glucopyranose and the distal β -glucopyranose rings.

The low field position of H-1 at 5.44 ppm (Fig. 4) is characteristic of an anomeric proton of a sugar 1-phosphate, similar to H-1 in *E. coli* lipid A, which resonates at 5.46 ppm (49,50). However, H-4' of *L. interrogans* lipid A (3.40 ppm) resonates at significantly higher field than the H-4' of *E. coli* lipid A (4.17 ppm), suggesting that the 4'-position of *L. interrogans* lipid A is not phosphorylated. The B₁⁺ ion detected by mass spectrometry in the positive ion mode (Fig. 6B) likewise suggests the absence of a phosphate group at position 4'. The ³¹P NMR data (Fig. 3) conclusively demonstrate only a single phosphate group in *L. interrogans* lipid A with heteronuclear coupling between the phosphorus atom and H-1 of the proximal sugar (Fig. 5).

H-3 and H-3' of *L. interrogans* lipid A resonate near 4.2 ppm and 3.9 ppm, respectively (Fig. 4 and Table I), significantly upfield of the H-3 and H-3' signals in *E. coli* lipid A at 5.25 and 5.18 ppm (49,50). Both the 3- and 3'-positions of *E. coli* lipid A are substituted with ester-linked acyl chains, and therefore H-3 and H-3' of *E. coli* lipid A are shifted considerably downfield relative to typical nonesterified sugar oxymethine groups (49,50). *L. interrogans* lipid A does not contain esterified sugar oxymethine groups at positions 3 and 3'.

¹³C NMR Evidence for a GlcN3N Disaccharide in *L. interrogans* Lipid A

In lipid A disaccharides consisting of two glucosamine units (49,50), two cross-peaks (originating from C-2 and C-2') are observed in the 52–57-ppm region of the HMQC spectrum. As shown in Fig. 8, the HMQC spectrum of *L. interrogans* lipid A reveals four sugar resonances between 52 and 58 ppm. Two of these cross-peaks are attributed to C-2 (53.5 ppm) and C-2' (55.1 ppm), because they correlate to H-2 at 4.05 ppm and H-2' at 3.75 ppm, respectively. The cross-peaks at 53.5 ppm and 56.7 ppm correlate with H-3 (4.21 ppm) and H-3' (3.90 ppm), demonstrating unequivocally that both C-3 and C-3' are substituted with nitrogen atoms in *L. interrogans* lipid A. The presence of aminomethine compared with oxymethine groups accounts for the large differences in the chemical shifts observed for H-3 and H-3' of *L. interrogans* lipid A versus *E. coli* lipid A. Moreover, the C-2 and C-3 shifts of *L. interrogans* lipid A agree with those reported for the α form of 2,3-diamino-2,3-dideoxyglucose in a 4:1 benzene-dimethyl sulfoxide mixture (54), whereas the C-2' and C-3' shifts are close to those of the β form of 2,3-diamino-2,3-dideoxyglucose (54).

The HMQC spectrum (Fig. 8) confirms two anomeric protons. H-1 of the proximal sugar at 5.44 ppm correlates with C-1 at 96 ppm, whereas H-1' of the distal sugar at 4.48 ppm connects to C-1' at 105 ppm. These C-1 and C-1' chemical shifts are characteristic of the α - and β -anomeric configurations, respectively (54), and are consistent with the ¹H NMR data (Fig. 4). The prominent three-proton doublet of *L. interrogans* lipid A at 3.61 ppm correlates to a carbon signal at 54.34 ppm, close to that of the CD₂HOD signal from the methanol solvent and in accord with the proposal that the methyl doublet arises from a methylated phosphate group (Fig. 7). The striking cross-peaks within the olefinic carbon region near 132 ppm (Fig. 8) correlate with proton signals at 5.35 and 5.40 ppm (Fig. 4), diagnostic for the presence of unsaturated acyl chains.

¹H NMR Analysis of the Acyloxyacyl Residues and Mono-unsaturated Acyl Chains in *L. interrogans* Lipid A

The *R*-3-hydroxyacyl chains that are the hallmark of all lipid A molecules are readily detected in *L. interrogans* lipid A by ¹H NMR (Fig. 4). The β -oxymethine protons of these acyl chains (Fig. 7A) resonate between 3.7 and 4.2 ppm when the β -OH group is not substituted, but they are shifted to about 5.2 ppm when a secondary acyl chain is present (48–50,55–57) (Fig. 4). The four α/β and four γ/β cross-peaks (Fig. 4) confirm that there are four β -hydroxyacyl chains in *L. interrogans* lipid A. Two of the four α/β cross-peaks overlap near 2.4 and 3.95 ppm ($\alpha 2, \beta 2$ and $\alpha 3, \beta 3$). Two of the four γ/β cross-peaks are detected near 1.5 and 3.95 ppm ($\gamma 2, \beta 2$ and $\gamma 3, \beta 3$). These signals are characteristic of α - and γ -methylene protons adjacent to β -

oxymethines of unsubstituted β -hydroxyacyl chains (48–50). The two remaining sets of α/β and γ/β cross-peaks (Fig. 4) are detected near 2.4–2.6 and 5.2 ppm and near 1.6 and 5.2 ppm, respectively. The downfield shift of the $\beta 2'$ and $\beta 3'$ protons versus the $\beta 2$ and $\beta 3$ protons (Fig. 4) confirms the presence of two acyloxyacyl moieties in *L. interrogans* lipid A (Fig. 7A).

Prominent cross-peaks are also observed near 5.38 and 2.1 ppm and near 5.42 and 2.05 ppm (Fig. 4). These signals arise from the spin coupling of olefinic protons to adjacent vinylic methylenes in the secondary acyl chains (Fig. 7A). The COSY analysis is therefore consistent with both the HMQC and the mass spectrometry in demonstrating the presence of unsaturated secondary acyl chains in *L. interrogans* lipid A. The exact location and stereochemistry of the double bonds remains to be determined. However, if *L. interrogans* generates fatty acid *cis*-double bonds by the same anaerobic pathway as *E. coli* (58–60), one would expect the double bonds of both the C12:1 and the C14:1 chains to be located at position ω -7 (Fig. 7A). The COSY analysis supports this idea, because it shows a cross-peak between at least one vinylic methylene and one β -methylene group (Fig. 4), as expected for the proposed structure of the C12:1 chain (Fig. 7A). The total correlation spectroscopy data (not shown) confirm the connectivity of olefinic protons to a subset of α - and β -methylenes within the secondary acyl chains, as well as showing the expected strong connectivity to vinylic and aliphatic methylenes.

DISCUSSION

The studies presented above document for the first time the existence of a lipid A molecule in *L. interrogans* with the proposed structure shown in Fig. 7A. The identification of this substance is consistent with the fact that the *L. interrogans* genome encodes a complete set of Lpx proteins (13), which catalyze the biosynthesis of the lipid A anchor of LPS in virtually all other Gram-negative bacteria (Fig. 9) (2). *L. interrogans* is the first spirochete shown to possess the *lpx* genes, in contrast to *T. pallidum*, *T. denticola*, and *B. burgdorferi*, which do not make lipid A (3,6,7). The absence of lipid A and LPS in the latter organisms (3–5) may explain their restricted ability to grow outside of their mammalian hosts. It would be of great interest to inactivate the *lpxA* gene (2,61–63) in *L. interrogans* to determine whether or not this organism is viable in the absence of its LPS. However, at present no mutagenesis system is available for pathogenic *Leptospira* spp. Lipid A is essential for growth in all Gram-negative bacteria examined to date (2,64) with the exception of *Neisseria meningitidis* strains containing a polysialic acid capsule (65,66). In the latter, the *lpxA* gene can be deleted with the consequence that the bacteria grow slowly and now require their polysialic acid capsule for viability (66).

The NMR studies shown in Figs. 4 and 8, in conjunction with the mass spectrometry of native and NaOH treated lipid A (Fig. 6), demonstrate unequivocally that *L. interrogans* lipid A consists of a hexa-acylated β -1',6-linked disaccharide in which the usual glucosamine residues are replaced with the more stable GlcN3N analog (28,29,67). The selectivity of *L. interrogans* LpxA for UDP-GlcNAc3N and 3-hydroxy-lauroylacyl carrier protein, documented in the preceding manuscript (28), independently supports this structural assignment. The two secondary acyl chains of *L. interrogans* lipid A appear to be unsaturated (Figs. 4,6,7A, and 8), which is unusual but not without precedent (68–70). Additional structural studies will be required to determine the exact location of the double bonds in the secondary acyl chains.

The 4'-position of *L. interrogans* lipid A is not phosphorylated. The latter finding indicates that a 4' phosphatase must be present in this organism, as in *R. etli* and *R. leguminosarum* in which the 4'-phosphate group is also missing (42,48,71,72). Removal of the 4'-phosphate group, when it occurs, appears to be a late step in lipid A biosynthesis, given that the 4'-phosphate residue is actually necessary for the attachment of the Kdo sugars (71,73,74). All bacteria with a 4'-phosphatase, including *L. interrogans*, retain the 4'-kinase encoded by *lpxK* (13,75) (Fig. 9). In preliminary studies, 4'-phosphatase activity was observed using

washed *L. interrogans* membranes (not shown) with the hexa-acylated substrate [4'-³²P] Kdo₂-lipid A from *E. coli* (76). No dephosphorylation of the tetra-acylated precursors [4'-³²P] lipid IV_A or [4'-³²P]Kdo₂-lipid IV_A (77) was detected.

The most unique aspect of *L. interrogans* lipid A is the finding that its 1-phosphate group is methylated (Figs. 4–6). This structural feature is without precedent in lipid A biochemistry (2,33). In fact, the enzymatic methylation of phosphate groups appears to be very rare in all of biology. To our knowledge, Kates *et al.* (31) have reported the only other example of a methylated lipid phosphate residue, found in the halophile *Halobacterium salinarium*, which synthesizes a methylated phosphatidylglycerophosphate analog. The enzymatic and genetic mechanisms for this type of lipid phosphate group methylation have not been explored. Similarly, the origin of the unusual methylated γ -phosphate cap found at the 5' end of the 7SK, B2, and U6 small RNAs in eucaryotic cells (32) has not been studied at the level of enzymology.

In contrast to the relatively uncommon biological methylation of phosphate residues, enzymes catalyzing the *N*-methylation of phosphatidylethanolamine (78,79) or the *O*-methylation of membrane proteins on selected carboxylate residues (80) are widely distributed. Reversible methylation of the methyl accepting chemotaxis proteins is essential for the proper response of *E. coli* to chemical signals (80,81). C-terminal methylation of Ras proteins in higher eucaryotic cells (82,83) or α -factor mating pheromone of yeast (84) is essential for membrane association and signaling. In these well documented examples of membrane lipid and membrane protein methylation, *S*-adenosyl-methionine serves as the methyl donor. We have recently found that membranes of *L. interrogans* catalyze the *S*-adenosyl-methionine-dependent methylation of Kdo₂-lipid A (85).

The characterization of the structure of *L. interrogans* lipid A sets the stage for the analysis of its biosynthesis and bioactivity. An initial survey of the lipid A described above demonstrates that it is inactive in the limulus lysate assay and against human THP-1 cells,² indicating that it is not contaminated with a classical endotoxin, such as *E. coli* lipid A. However, when assayed with mouse RAW 264.7 cells, *L. interrogans* lipid A induces tumor necrosis factor with about one-tenth the potency of *E. coli* lipid A.² We are currently evaluating *L. interrogans* lipid A with macrophages derived from various mouse TLR knockout strains. It may be that the robust TLR2 activating activity seen with intact *L. interrogans* LPS (22) requires more than just the lipid A moiety. Isolation of LPS from *L. interrogans* mutants blocked in defined steps of O-antigen and/or core biosynthesis (12) might address this question. In addition, chemically synthesized versions of *L. interrogans* lipid A need to be prepared to validate our proposed structure and to determine whether or not the activity seen with *L. interrogans* lipid A is real or is due to other biologically active impurities.

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Fig. 1. Thin layer chromatography of purified lipid A from *L. interrogans* serovar Pomona
The solvent was CHCl₃, pyridine, 88% formic acid/H₂O/MeOH (60:35:10:5:2, v/v/v/v/v). The lipid A band was visualized by charring on a hot plate (42).

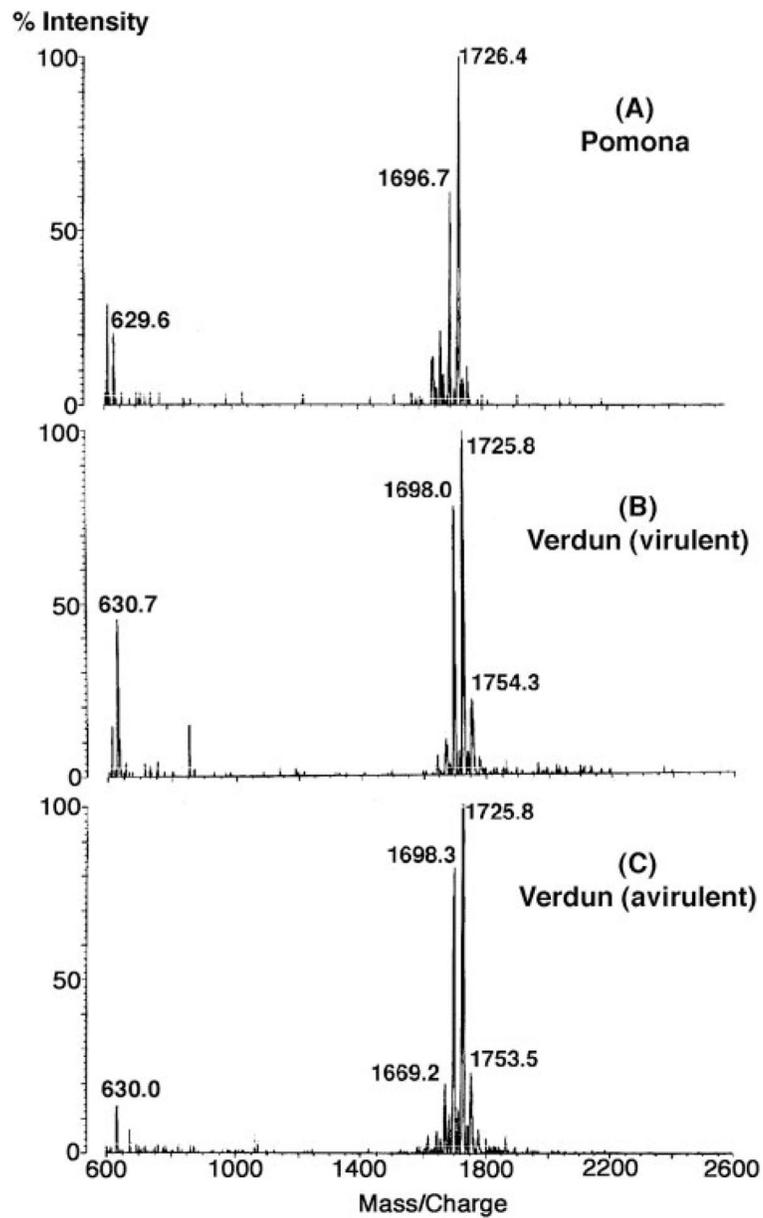


Fig. 2. Negative ion MALDI-TOF mass spectra of lipid A preparations from different *L. interrogans* strains
The major lipid A molecules of these strains appear to be identical.

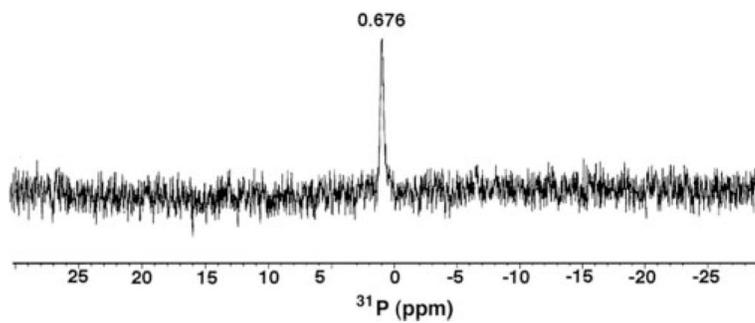


Fig. 3. ^{31}P NMR spectrum of *L. interrogans* serovar Pomona lipid A
The 2-mg lipid A sample was dissolved at 25 °C in 0.6 ml of CDCl_3 , CD_3OD , D_2O (2:3:1, v/v).

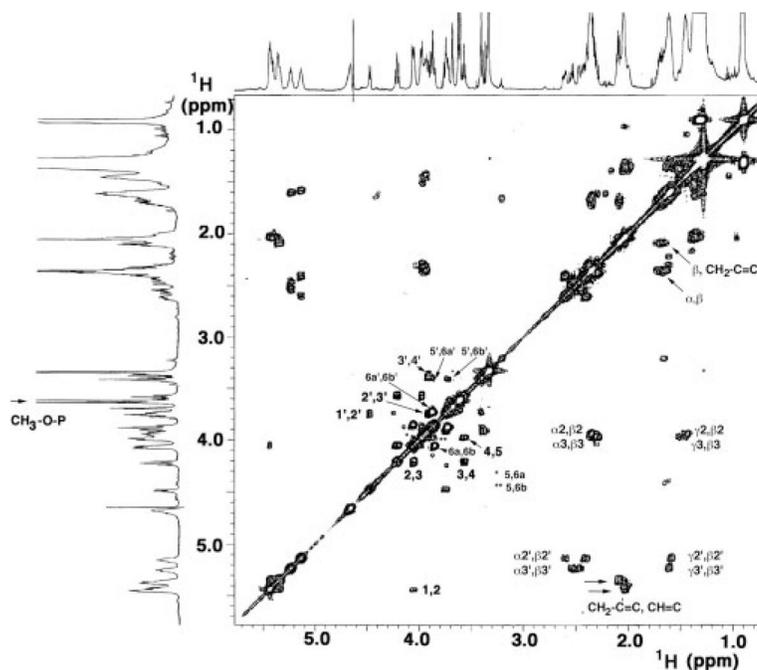


Fig. 4. ^1H - ^1H COSY of *L. interrogans* serovar Pomona lipid A at 600 MHz

The 2-mg lipid A sample was dissolved at 25 °C in 0.6 ml of CDCl_3 , CD_3OD , D_2O (2:3:1, v/v/v). The numbering scheme is shown in Fig. 7A. The arrow at the left highlights the doublet near 3.61 ppm that arises from the methylated 1-phosphate. The two arrows at the lower right highlight the strong cross-peaks between the olefinic and vinylic protons of the secondary acyl chains.

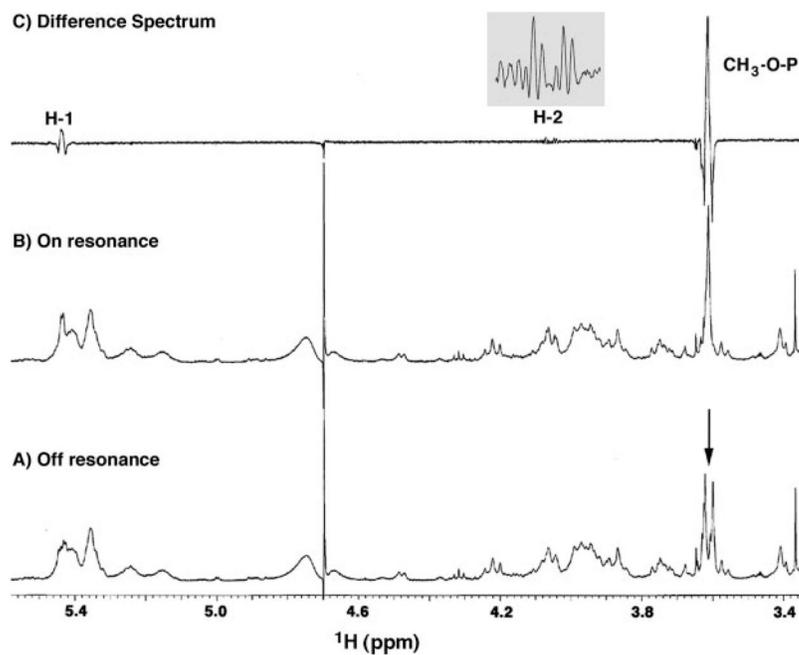


Fig. 5. Selectively ^{31}P -decoupled ^1H NMR spectra of *L. interrogans* serovar Pomona lipid A
 The sample used in Fig. 4 was subjected to ^1H NMR analysis with the decoupler either off resonance or on the ^{31}P atom resonance. The doublet at 3.61 ppm (*arrow* in *A*) collapses to a singlet (*B*) when the phosphorus atom is selectively irradiated. The difference spectrum (*C*) highlights additional coupling of the phosphorus atom to the H-1 and H-2 signals of the proximal sugar. The *inset* in *C* shows the H-2 difference spectrum in detail.

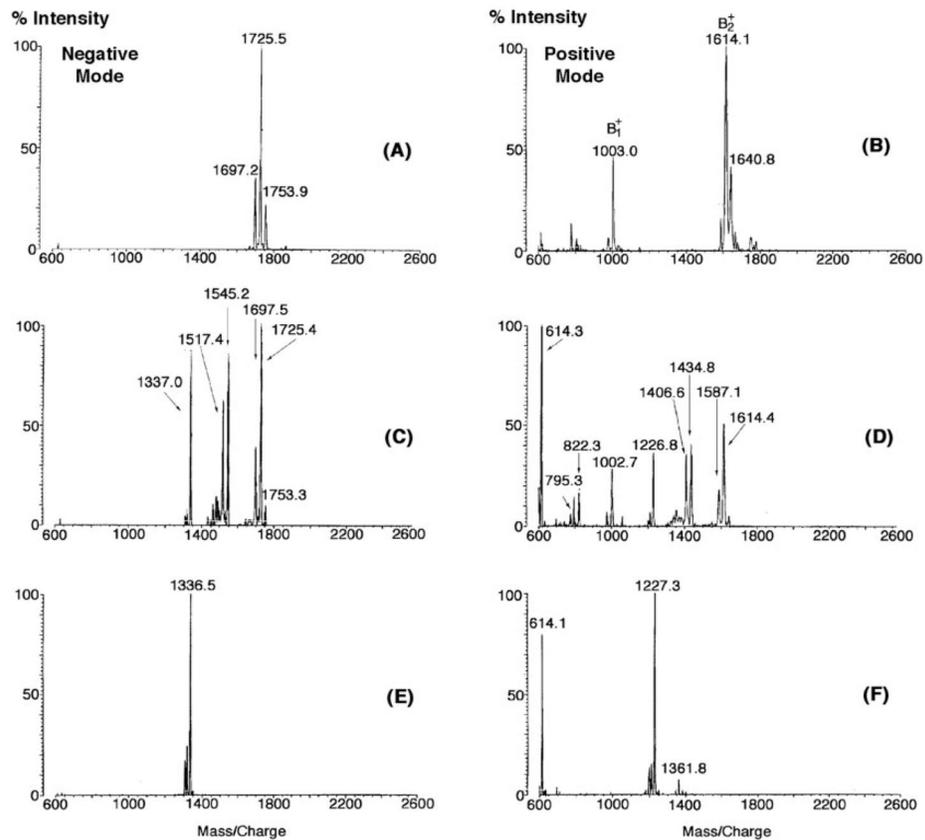


Fig. 6. MALDI-TOF mass spectrometry of *L. interrogans* serovar Pomona lipid A subjected to mild NaOH hydrolysis

The *left half* of the figure shows the negative ion mass analysis of *L. interrogans* lipid A before (A), after 5 min (C), or after 30 min of mild NaOH treatment (E). The *right half* of the figure shows the corresponding positive ion spectra (B, D, and F, respectively). The loss of two *O*-linked acyl chains from the distal unit with masses corresponding to C12:1 and C14:1 is complete after 30 min.

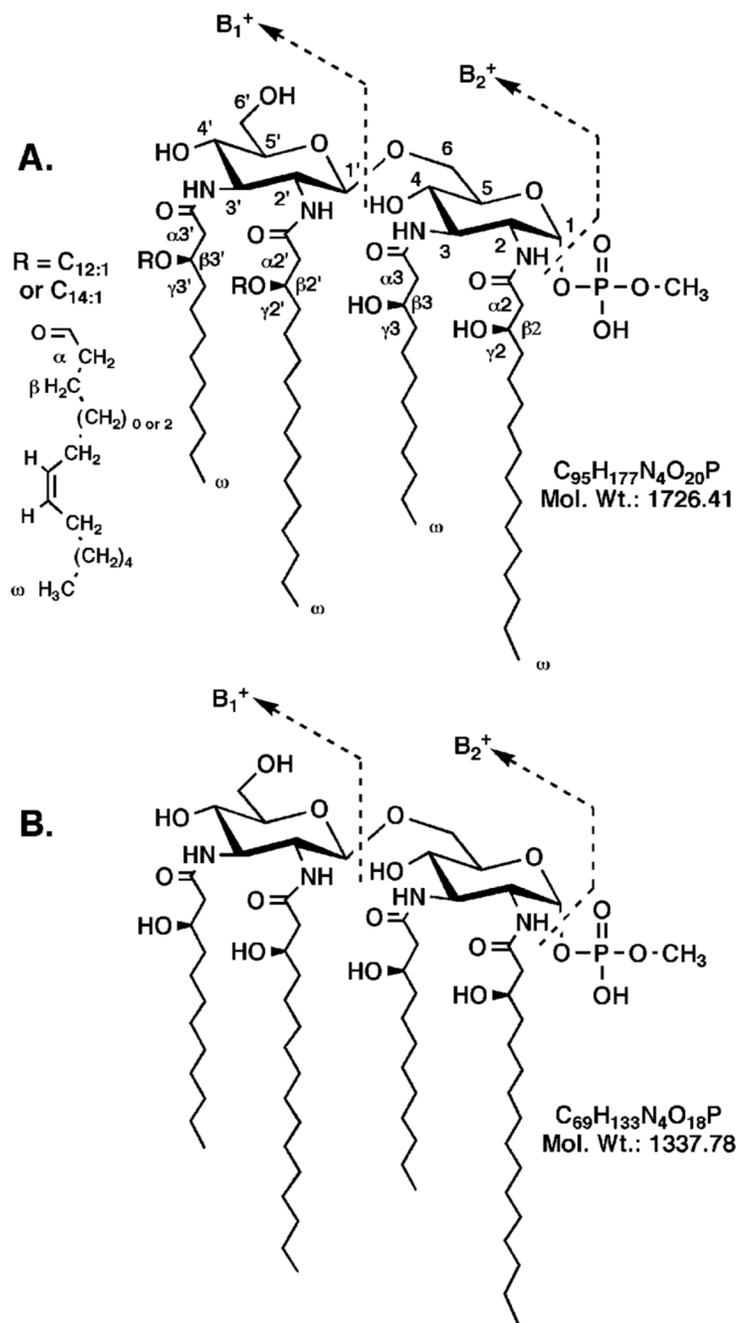


Fig. 7. Proposed structures of intact and *O*-deacylated *L. interrogans* serovar Pomona lipid A
The numbering scheme shown for the intact lipid A in A is used for the NMR assignments. The position of the *cis*-double bond is inferred based on the mechanism by which unsaturated fatty acids are synthesized in *E. coli* (58–60). The fully *O*-deacylated material is shown in B. Fatty acid analysis of *L. interrogans* lipid A shows equal amounts of 3-hydroxylaurate and 3-hydroxy-palmitate; 3-hydroxylaurate is attached at positions 3 and 3' (28).

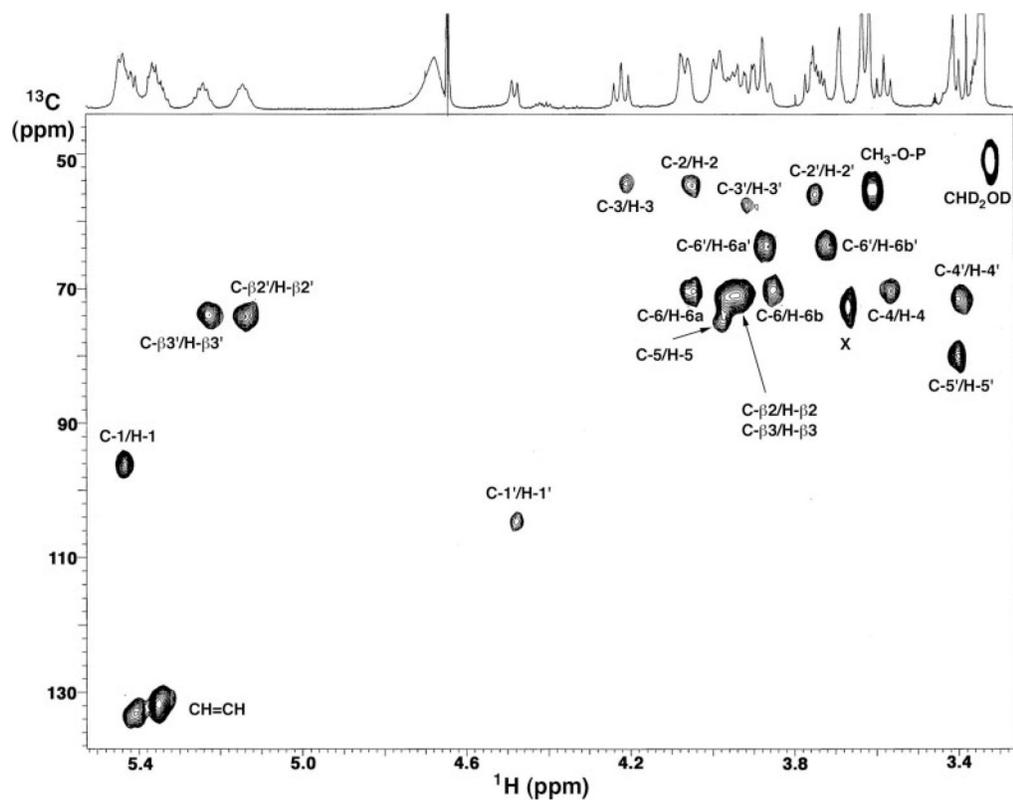


Fig. 8. Partial HMQC spectrum of *L. interrogans* serovar Pomona lipid A at 600 MHz
 The 2-mg lipid A sample was dissolved at 25 °C in 0.6 ml of CDCl₃, CD₃OD, D₂O (2:3:1, v/v/v). The cross-peaks are labeled according to the scheme shown in Fig. 7A.

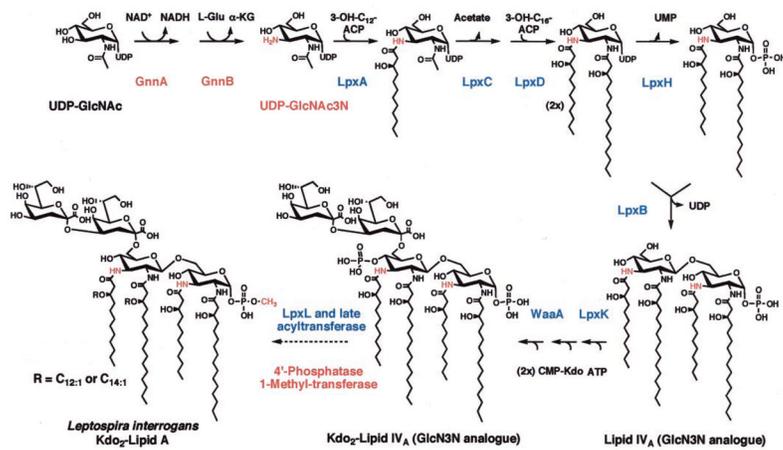


Fig. 9. Proposed biosynthetic pathway for *L. interrogans* lipid A

Highly significant orthologs of all of the indicated enzymes (2) are encoded in the *L. interrogans* serovar Lai genome (13). The methyl transferase responsible for the formation of the unusual 1-methylphosphate group of *L. interrogans* lipid A utilizes *S*-adenosyl-methionine as the donor (85).

Table I
 ^1H , ^{13}C , and ^{31}P NMR chemical shifts of *L. interrogans* lipid A
 ^1H and ^{13}C chemical shifts at 25 °C in $\text{CDCl}_3:\text{CD}_3\text{OD}:\text{D}_2\text{O}$ (2:3:1, v/v/v) relative to internal tetramethylsilane derived from the two-dimensional NMR experiments (Figs. 4 and 8). The ^{31}P chemical shift was relative to 85% H_3PO_4 . The numbering scheme is shown in Fig. 7A.

Position	δC	δH_i [(mult), J(Hz)]	δP
Proximal GlcN3N			
1	95.3	5.44 [dd, $J_{1,2} = 3.2$] ^a	+0.676
2	53.52	4.05 [ddd, $J_{2,3} = 11.23$] ^a	
3	53.46	4.21 [dd, $J_{3,4} = 9.8$]	
4	69.00	3.57 [dd, $J_{4,5} = 8.9$]	
5	73.63	3.98 [m]	
6a	69.29	4.04 [m]	
6b		3.85 [m]	
Distal GlcN3N			
1'	101.9	4.47 3dd, $J_{1,3,2,3} = 7.1$	
2'	55.08	3.75 [m]	
3'	56.71	3.90 [m]	
4'	70.31	3.39 [m]	
5'	79.05	3.40 [m]	
6'a	62.66	3.87 [m]	
6'b		3.73 [m]	
3-Hydroxyacyl			
$\alpha 2$	45.0	~2.35, 2.30	
$\beta 2$	69.8	~3.93 [m]	
$\gamma 2$	38.5	~1.44 [m]	
$\alpha 3$	42.3	~2.40	
$\beta 3$	70.0	~3.95 [m]	
$\gamma 3$	38.5	~1.44 [m]	
$\alpha 2'$	42.4	~2.58, 2.45	
$\beta 2'$	73.3	~5.13 [m]	
$\gamma 2'$		~1.63 [m]	
$\alpha 3'$	42.3	~2.52	
$\beta 3'$	73.5	~5.22 [m]	
$\gamma 3'$		~1.63 [m]	
Other			
α CH ₂	35.3	~2.34 [m]	
β CH ₂		~1.61 [m]	
(CH ₂)	30.4	~ 1.28 [m]	
ω CH ₃	15.2	0.891 [t]	
CH=CH	131.24	5.35	
CH=CH	132.5	5.40	
CH ₂ -C=C	28.06	2.08	
CH ₂ -C=C	28.57	2.03	
CH ₃ -OP	54.34	3.61 [d, $J_{\text{HP}} = 10.9$]	

^a Measured from the resolved H-2 double-doublet in the ^{31}P -decoupled ^1H NMR difference spectrum (Fig. 5C).