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## Characterization of Functional Heme Domains from Soluble Guanylate Cyclase<sup>Δ</sup>

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### Abstract

Soluble guanylate cyclase (sGC) is a heterodimeric, nitric oxide (NO)-sensing hemoprotein composed of two subunits,  $\alpha$ 1 and  $\beta$ 1. NO binds to the heme cofactor in the  $\beta$ 1 subunit, forming a 5-coordinate NO complex that activates the enzyme several hundred-fold. In this report the heme domain has been localized to the N-terminal 194 residues of the  $\beta$ 1 subunit. This fragment represents the smallest construct of the  $\beta$ 1 subunit that retains the ligand binding characteristics of the native enzyme, namely tight affinity for NO and no observable binding of O<sub>2</sub>. A functional heme domain from the rat  $\beta$ 2 subunit has been localized to the first 217 amino acids  $\beta$ 2(1-217). These proteins are ~40% identical to the rat  $\beta$ 1 heme domain, form 5-coordinate, low-spin NO complexes and 6-coordinate, low-spin CO complexes. Like sGC these constructs have a weak Fe-His stretch (208 and 207 cm<sup>-1</sup> for  $\beta$ 1(1-194) and  $\beta$ 2(1-217), respectively).  $\beta$ 2(1-217) forms a CO complex that is very similar to sGC and has a high  $\nu$ (CO) stretching frequency at 1994 cm<sup>-1</sup>. The autoxidation rate of  $\beta$ 1 (1-194) was 0.073 per minute, while the  $\beta$ 2(1-217) was substantially more stable in the ferrous form with an autoxidation rate of 0.003 per minute at 37 °C. This report has identified and characterized the minimum functional ligand binding heme domain derived from sGC providing key details towards a comprehensive characterization.

Soluble guanylate cyclase (sGC<sup>1</sup>) is a well-characterized NO-sensor, containing a heme-bound, N-terminal domain that binds NO, regulating a C-terminal guanylate cyclase. sGC catalyzes the conversion of GTP to cGMP, and it is a heterodimer, composed of  $\alpha$ 1 and  $\beta$ 1 subunits. The ferrous heme is ligated to the  $\beta$ 1 subunit via H105 (1), and when NO is bound, the activity of the enzyme is 300-fold elevated over basal activity (2). sGC has evolved to selectively bind NO and not O<sub>2</sub>. It does not form a stable, isolable complex with O<sub>2</sub>, permitting NO to bind selectively to the Fe<sup>+2</sup> heme in an aerobic environment. This selectivity is essential

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<sup>1</sup>Abbreviations: CO, carbon monoxide; sGC, soluble guanylate cyclase; NO, nitric oxide; *Tt*Tar4H, N-terminal 188 amino acids of *Thermoanaerobacter tengcongensis* Tar4 (heme domain); MCP, methyl-accepting chemotaxis protein; AxPDEA1H, *Acetobacter xylinum* phosphodiesterase A1 heme domain; Mb, myoglobin; H-NOX domain, Heme-Nitric oxide/Oxygen binding domain.

for sGC to trap nM concentrations of NO in the presence of a high competing concentration of O<sub>2</sub>.

sGC has been characterized in animals ranging from *Drosophila* to humans and, until recently, there had been no reports of any other NO sensors except for sGC. Notably, no heme-containing NO sensors had been found in prokaryotes, which could be exposed to both exogenous and endogenous NO (3). However, a PSI-BLAST search using  $\beta$ 1 (1-194) as the query sequence led to multiple hits in prokaryotic genomes (4,5). Two representative proteins were characterized—a 181 amino acid protein from *V. cholerae* (VCA0720) and the heme domain from the *T. tengcongensis* Tar4 protein (*TtTar4H*) (6). VCA0720 is encoded in a histidine kinase containing-operon, while *TtTar4H* is fused to a methyl-accepting chemotaxis domain. These proteins share spectroscopic properties, sequence homology and key conserved residues with the sGC heme domain and, therefore, were placed in the same family. Surprisingly, however, these proteins exhibited very different ligand-binding properties. For example, our spectroscopic data indicated that the predicted heme domain from *Vibrio cholerae* had ligand binding characteristics essentially identical to sGC, while *TtTar4H* was able to form a stable ferrous-oxy complex similar to the O<sub>2</sub>-sensor AxPDEA1 (6). Our results suggested that these sGC-like heme domains have evolved specific ligand binding properties that allow them to function as sensors for NO or O<sub>2</sub> and, therefore, we have named it the H-NOX (Heme-Nitric oxide/Oxygen binding) domain.

Previously, we had localized the heme-binding region of sGC to residues 1-385 of the  $\beta$ -subunit (1,7).  $\beta$ 1(1-385) includes a predicted coil-coil region and is isolated as a homodimer although the associated spectral properties are essentially the same as those in the native enzyme. In an effort to determine the minimum sequence needed to retain the unique ligand binding properties of the heme in sGC, we truncated and characterized several new constructs. In this report we show that the N-terminal 194 residues of the  $\beta$ 1 subunit retains the unique ligand binding properties of the full-length enzyme and appears to be the smallest functional  $\beta$ 1 fragment. To further probe the selectivity of ligand binding in the sGC H-NOX domain, we have characterized this 194 amino acid  $\beta$ 1 fragment representing the smallest sGC heme domain and the heme domain from the rat  $\beta$ 2 subunit.

The  $\beta$ 2 subunit is 43% identical to rat  $\beta$ 1(1-194) (Figure 1) and was cloned from rat and kidney cDNA libraries (8-10). Its primary localization appears to be the kidney and liver (8). Unlike the  $\beta$ 1 subunit, it contains an isoprenylation motif at its C-terminus which could lead to membrane localization (8). The functional relationship between the  $\beta$ 1 and  $\beta$ 2 subunits is not presently well characterized. Because of their sequence similarity, it has been hypothesized that both subunits function to sense NO, but this has not been definitively demonstrated. Two recent reports have described NO-stimulated  $\beta$ 2 cyclase activity in over-expressed cells, although the stimulation was much weaker compared to  $\alpha$ 1/ $\beta$ 1 heterodimeric sGC (10,11). Thus, it appears that this subunit is a catalytically active NO-sensing homodimer, although no studies have been performed on purified protein. In this report we show that  $\beta$ 2(1-217) is able to form a 5-coordinate NO complex and is unable to form a stable Fe<sup>+2</sup>-O<sub>2</sub> complex as has been observed for native sGC and the  $\beta$ 1 subunit.

## MATERIALS AND METHODS

### Construction of expression plasmids

PCR (PfuTurbo, Stratagene) was used to amplify  $\beta$ 1(1-194) and  $\beta$ 2(1-217) from  $\beta$ 1(1-385) and the full-length rat  $\beta$ 2 clone (containing the additional N-terminal 60 amino acids) (10), respectively. The upstream primers were 5'-ggaattccatgtacgggtttgtgaacctgccc-3' and 5'-ggaattccatgtatggattcatcaaacctgc-3' and the downstream primers were 5'-atagtttagcgccgctcaatctcataaaaaatcctcttctttg-3' and 5'-

atagttttagcggccgctcaaaggagtggtccctggagagcctc-3' for  $\beta$ 1(1-194) and  $\beta$ 2(1-217), respectively. Primers were obtained from Qiagen except for the upstream primer for  $\beta$ 2(1-217), which was obtained from GibcoBRL. Amplified PCR products were cloned into pET-20b (Novagen) and sequenced (UC Berkeley sequencing core and Elim Biopharmaceuticals, Inc).

### Protein expression

Expressions were performed as described previously (1) with the following modifications. Plasmids were transformed into Tuner DE3 plyS cells (Novagen). Cultures were grown to an OD<sub>600</sub> of 0.5-0.6 and cooled to 27 °C. IPTG (Promega) was added to 10  $\mu$ M and aminolevulinic acid to 1 mM. Cultures were grown overnight for 14-18 h and then harvested.

### Protein purification

$\beta$ 1(1-194) was purified in the following manner: Frozen cell pellets from 3 L of culture were thawed quickly at 37 °C and resuspended in 120 mL of Buffer A (50 mM DEA pH 8.5, 20 mM NaCl, 5 mM DTT, 1 mM Pefabloc (Pentapharm) and 5% glycerol). Resuspended cells were lysed with sonication and, subsequently, with an Emulsiflex-C5 high pressure homogenizer at 20,000 psi (Avestin, Inc). Lysed cells were centrifuged at 100,000  $\times$  g for 40 min. The supernatant was collected and reduced with  $\sim$ 500-fold excess dithionite, which was based on a final yield of 20 mg of pure protein (dithionite was used throughout the prep to keep the protein reduced because it oxidized over several hours and sometimes aggregated when oxidized. To minimize unwanted side effects of dithionite in aerobic conditions, we purged all of the chromatography buffers with nitrogen gas before use). The supernatant was applied to a 50 mL (2.5  $\times$  11 cm) Toyopearl SuperQ 650 M (Tosohaas) anion exchange column at 2.5 mL/min. The column was washed with 2 column volumes of Buffer A at 2.5 mL/min and eluted with a linear gradient of NaCl from 20 to 750 mM in a total volume of 340 mL at 3.5 mL/min. Fractions were selected on the basis of the intensity of the red/brown color. The eluate (50 mL) was concentrated to 4 mL using 15 mL 10K MWCO spin concentrators (Millipore) and reduced with  $\sim$ 100-fold excess dithionite. The concentrated proteins were applied to a pre-packed Superdex S75 HiLoad 26/60 gel filtration column (Pharmacia) that had been equilibrated with TEA pH 7.5, 50 mM NaCl, 5mM DTT and 5% glycerol. The flow rate was 1.5 mL/min. Fractions containing  $\beta$ 1(1-194) were pooled, reduced with  $\sim$ 100-fold excess dithionite and applied to a POROS HQ 7.9 mL (1  $\times$  10 cm, 10  $\mu$ m) anion exchange column (Applied Biosystems) that had been equilibrated with buffer B (50 mM TEA pH 7.5, 50 mM NaCl, 5 mM DTT and 5% glycerol). The column was washed with 3 column volumes of buffer B at 10 mL/min and developed with a linear gradient of NaCl from 50 to 500 mM in a total volume of 140 mL at the same flow rate.  $\beta$ 1(1-194) containing fractions were pooled and stored at -70 °C.  $\beta$ 2(1-217) was purified as described above except for the following modifications: Dithionite was not added before each chromatographic step as it did not readily oxidize and aggregate. Buffer A contained 50 mM DEA pH 8.5 and 25 mM NaCl, and the flow rate for loading in the first anion exchange step was 1.5 mL/min. The supernatant was applied to a 100 mL (2.5  $\times$  21 cm) Toyopearl DEAE 650M anion exchange column (Tosohaas) and washed with 3 column volumes of Buffer A. The column was developed with a linear NaCl gradient from 20 to 500 mM (800 mL) and the flow rate for loading, washing, and eluting was 1.2 mL/min. The high-resolution, anion exchange step included 5 mM DTT.

### UV/vis spectroscopy

All spectra were recorded in an anaerobic cuvette on a Cary 3E spectrophotometer equipped with a Neslab RTE-100 temperature controller set at 10 °C. Spectra were recorded from proteins in a solution of 50 mM TEA pH 7.5 and 50 mM NaCl (Buffer C). Heme concentration was  $\sim$ 7  $\mu$ M. Protein samples were placed into the anaerobic cuvette in an anaerobic glove bag (Coy). Fe<sup>+2</sup>-unligated protein samples were prepared in the following manner: Purified protein

was made anaerobic in an O<sub>2</sub>-scavenged gas train with 10 cycles of alternate evacuation and purging with purified argon and brought into an anaerobic glove bag. Because  $\beta$ 1(1-194) was susceptible to oxidation, it was first reduced anaerobically using dithionite (~100 equivalents). The dithionite was then removed using a PD10 desalting column that had been equilibrated with Buffer C. For consistency,  $\beta$ 2(1-217) was prepared in the same manner. The CO complexes were generated by adding Fe<sup>+2</sup>-unligated protein to a sealed Reacti-Vial (Pierce) that contained CO (Airgas). NO complexes were generated by preparing an anaerobic solution of Diethylamine NONOate (Caymen) in Buffer C (~10 mM) in a sealed Reacti-Vial. The head-space (1 mL) was then removed using a gas-tight syringe and delivered to a Fe<sup>+2</sup>-unligated protein sample contained in a sealed Reacti-Vial.

### Extinction coefficient determination

An anaerobic, Fe<sup>+2</sup>-unligated, UV/vis spectrum was recorded for each protein sample. These samples, along with the Mb standard, were prepared anaerobically as described above with the following modification: bound-O<sub>2</sub> was removed by ferricyanide oxidation (~100 equivalents). The ferricyanide was then removed using a PD10 desalting column that had been equilibrated with Buffer C and then the protein was reduced using dithionite (~100 equivalents). The dithionite was then removed in the same manner. For consistency,  $\beta$ 1(1-194) and  $\beta$ 2(1-217) were prepared in the same way. The heme concentration was then determined for each sample by HPLC (12) and used to determine the Fe<sup>+2</sup>-unligated extinction coefficient. This extinction coefficient was then used to determine the heme concentration and extinction coefficients for the CO and NO complexes from previously recorded UV/vis spectra. HPLC was used to determine the heme concentration using a protein C4 column (250 × 4.6 mm, 5  $\mu$ m, Vydac) and a Hewlett Packard Series II 1090 HPLC with a diode array detector. Each sample (25  $\mu$ L) was applied to the C4 column that had been equilibrated with 0.1% trifluoroacetic acid (TFA). The column was developed with a linear gradient of 0-75% acetonitrile over 15 min followed by a linear gradient of 75-95% acetonitrile over 5 min. The flow rate was 1 mL/min.

### Homology modeling

Homology models of  $\beta$ 1(1-194) and  $\beta$ 2(1-217) were generated using MODELLER following the methods described in (13).

### Resonance Raman spectroscopy

Spectra of all samples were collected using 413.1 nm laser excitation except for the NO complexes, which were excited at 406.7 nm. A microspinning sample cell was used to minimize photo-induced degradation. Raman scattering was detected with a cooled, back-illuminated CCD (LN/CCD-1100/PB, Roper Scientific) controlled by a ST-133 controller coupled to a subtractive dispersion, double spectrograph (14). Raman spectra were corrected for wavelength dependence of the spectrometer efficiency by using a white lamp and calibrated with cyclohexane. The reported frequencies are accurate to  $\pm 2$  cm<sup>-1</sup> and the resolution of the spectra is 8 cm<sup>-1</sup>. All Raman spectra were obtained by subtracting the buffer background and baseline correction. The laser power at the sample was 3 mW, except for the CO complexes where a power of 0.2 mW was used to avoid photolysis. Typical data acquisition times were 30 min.

All protein samples were placed into the Raman cell in an anaerobic glove bag. The heme concentration was 14 and 19  $\mu$ M for  $\beta$ 1(1-194) and  $\beta$ 2(1-217), respectively. Protein samples for Raman spectroscopy were prepared as described above for UV/vis spectroscopy with the following modifications: <sup>13</sup>CO (99% containing ~10% <sup>18</sup>O<sub>2</sub>, Cambridge Isotopes) was prepared in a similar manner to the <sup>12</sup>CO complexes. <sup>15</sup>NO complexes were made in the following manner: A 1 mL solution of 0.5 M K<sup>15</sup>NO<sub>2</sub> (Cambridge Isotopes) and a 1 mL solution of 1M KI/H<sub>2</sub>SO<sub>4</sub> were made anaerobic on an O<sub>2</sub>-scavenged gas train and delivered into the anaerobic glove bag. The potassium nitrite solution (100  $\mu$ L) was delivered to the KI solution

using a gas-tight syringe.  $^{15}\text{NO}$ -containing head-space ( $\sim 1$  mL) was then delivered to a protein sample contained in a sealed Reacti-Vial.

### Oxidation rates

Rat  $\beta 1(1-194)$  and rat  $\beta 2(1-217)$  ( $550$  and  $220 \mu\text{M}$ , respectively) were brought into an anaerobic chamber (Coy) and reduced at room temperature with sodium dithionite ( $20$  mM final). Both proteins were then desalted using a PD-10 column and eluted in  $50$  mM triethanolamine,  $50$  mM NaCl. Each protein was then added to an anaerobic cuvette and removed from the chamber. Prior to acquiring the first spectra, each sample was diluted 5-fold with  $50$  mM triethanolamine,  $50$  mM NaCl that had been fully equilibrated with room air and the cuvette was then left open to atmospheric oxygen. Immediately ( $<10$  second dead time) after dilution with the aerobic buffer, spectra were acquired ( $\beta 1$ : 1 scan/minute for 100 minutes,  $\beta 2$ : 1 scan/min for 60 minutes, 1 scan/10 minutes for 17 hours) at  $37^\circ\text{C}$ . Rate constants based on double difference plots ( $\beta 1$ :  $\Delta 404 - \Delta 434$ ;  $\beta 2$ :  $\Delta 390 - \Delta 432$ ) were then calculated and fit to single exponentials of the form  $f(x) = A*(1-e^{(-k*x)})$ .

## RESULTS

### Cloning, expression and purification

N-terminal sGC  $\beta 1$  heme-binding region fragments were generated by sub-cloning specific constructs into the *E. coli* expression vector pET-20b. The shortest region that was sub-cloned, had heme-bound and expressed well was  $\beta 1(1-194)$ . For example,  $\beta 1(1-189)$  was cloned, expressed and purified and relative to the longer constructs it was poorly expressed and unstable. Therefore  $\beta 1(1-194)$  was the specific construct that was chosen for further study.

For the  $\beta 2$  H-NOX domain, we initially attempted the expression of the full-length  $\beta 2$ -subunit (10) and it was cloned into the bacterial expression vector pET-20b. After expression and purification the results clearly showed that a soluble heme-bound protein was made but appeared to be truncated, as demonstrated by SDS-PAGE (data not shown). Mass spectral analysis yielded a heme-bound, N-terminal, 217 amino acid fragment confirming that the protein was indeed truncated (data not shown). It appears that an endogenous *E. coli* protease cleaved this fragment from the full-length protein. The 5651 bp of the rat  $\beta 2$ -subunit coding for residues 1-217 was then cloned into pET-20b and further characterized.

$\beta 1(1-194)$  and  $\beta 2(1-217)$  were expressed in *E. coli* and purified as described in materials and methods. These proteins were estimated to be  $>95\%$  pure on the basis of Coomassie staining of samples resolved by SDS-PAGE (data not shown). Yields were  $\sim 5$  and  $13$  mg/L of cell culture for  $\beta 1(1-194)$  and  $\beta 2(1-217)$ , respectively.  $\beta 2(1-217)$  was isolated in the  $\text{Fe}^{+2}$ -unligated state (Soret maximum  $\sim 431$  nm). Dithionite was needed to keep  $\beta 1(1-194)$  reduced throughout the purification, because it was found to oxidize readily (see Materials and Methods). UV/vis and resonance Raman spectroscopy were then used to characterize the heme environment and ligand binding characteristics of both proteins.

### UV/vis absorption spectroscopy

UV/vis spectra are shown in Figure 2 and the data with extinction coefficients summarized in Table 1 and compared to sGC (2) and hemoglobin (15). Details of the individual complexes are discussed where relevant below.

### $\text{Fe}^{+2}$ -unligated and CO complexes

These two  $\beta$ -subunit constructs have spectra that are similar to one another and to full-length sGC. The spectra of the  $\text{Fe}^{+2}$ -unligated hemes in both  $\beta$ -subunit constructs exhibit Soret bands at  $\sim 431$  nm and a single broad  $\alpha/\beta$  band at  $\sim 560$  nm. Addition of CO to the anaerobically-



reduced proteins resulted in CO complexes with sharp Soret bands at ~424 nm, increases in the extinction coefficient and observable  $\alpha/\beta$  bands. The shape and absorption maxima of the Soret and  $\alpha/\beta$  bands for both the  $\text{Fe}^{+2}$ -unligated and CO complexes are very similar to full-length sGC (2), suggesting that the reduced proteins are 5-coordinate, high-spin and the CO complexes are 6-coordinate, low-spin.

### NO complexes

The NO complexes of  $\beta 1(1-194)$  and  $\beta 2(1-217)$  are similar to those of full-length sGC, exhibiting broadened Soret bands at ~399 nm, decreases in the extinction coefficient and observable  $\alpha/\beta$  bands. These spectra are identical to previously characterized 5-coordinate, NO complexes (2).

### Resonance Raman spectroscopy

Raman spectra are shown in Figures 3 and 4 ( $\beta 1(1-194)$  and  $\beta 2(1-217)$ , respectively) and the data summarized in Tables 2, 3 and 4 ( $\text{Fe}^{+2}$ -unligated, CO adducts and NO adducts, respectively) along with comparisons to sGC (16,17), Mb (18-23), *Acetobacter xylinum* PDEA1H (the heme domain of AxPDEA1) (24), rat sGC  $\beta 1(1-385)$  (25), FixLN (the heme domain of FixL) (26), *Alcaligenes xylosoxidans* cyt c' (27,28), HemAT-Bs (29), and CoxA (30). Specific complexes are described below.

### $\text{Fe}^{+2}$ -unligated complexes

In the high frequency region the heme skeletal marker bands are observed. These include  $\nu_4$ , which is sensitive to the oxidation state of the heme, and  $\nu_3$  and  $\nu_2$ , which are sensitive to the spin and coordination state of the heme iron (31). The heme-skeletal markers are similar to the corresponding bands in histidine-ligated, 5-coordinate, high-spin,  $\text{Fe}^{+2}$  heme proteins like sGC. In the low frequency spectrum, bands at 208 and 207  $\text{cm}^{-1}$  are observed for  $\beta 1(1-194)$  and  $\beta 2(1-217)$ , respectively. These bands are assigned to the  $\nu(\text{Fe-His})$  stretching mode, on the basis of previous observations that in most histidine-ligated,  $\text{Fe}^{+2}$ -heme complexes, this stretching frequency is usually between 200 and 250  $\text{cm}^{-1}$  (32), and that in the NO and CO adducts this band was absent. We conclude that, when reduced, these proteins contain a 5-coordinate, high-spin heme.

### CO complexes

The heme skeletal markers in the high frequency region are similar to the corresponding bands from histidine-ligated, 6-coordinate, low-spin,  $\text{Fe}^{+2}$  heme proteins like sGC. In the low-frequency spectrum, isotope-sensitive bands are observed at 477 and 496  $\text{cm}^{-1}$  for  $\beta 1(1-194)$  and 476  $\text{cm}^{-1}$  for  $\beta 2(1-217)$ . These bands are assigned to the  $\nu(\text{Fe-CO})$  stretching mode on the basis of their isotopic shift and their similarity to the  $\nu(\text{Fe-CO})$  frequency for full-length sGC (472/487  $\text{cm}^{-1}$ ) and other 6-coordinate, low-spin heme proteins (Table 2). The two  $\nu(\text{Fe-CO})$  frequencies in full-length sGC are likely due to different Fe-CO conformations (25) and this may also account for the two  $\nu(\text{Fe-CO})$  frequencies in  $\beta 1(1-194)$ . In the high-frequency spectrum, isotope-sensitive bands are observed at 1968 and 1994  $\text{cm}^{-1}$  for  $\beta 1(1-194)$  and  $\beta 2(1-217)$ , respectively. These bands are assigned to the  $\nu(\text{CO})$  stretching mode on the basis of their isotopic shift and their similarity to the  $\nu(\text{CO})$  frequencies of sGC (1987  $\text{cm}^{-1}$ ) and of HemAT-Bs (1964  $\text{cm}^{-1}$ ) (Table 2). Like sGC,  $\beta 2(1-217)$  has a high  $\nu(\text{CO})$  stretching frequency. Therefore,  $\beta 2(1-217)$  forms a CO complex that is the most similar to sGC. These complexes are 6-coordinate, low-spin and are notable for their high  $\nu(\text{CO})$  stretching frequency. CO- $\beta 1(1-194)$  is also 6-coordinate, low-spin like sGC, but its  $\nu(\text{CO})$  frequency is closer to that of proteins with lower values, like HemAT-Bs (1964  $\text{cm}^{-1}$ ).

## NO complexes

For  $\beta 1(1-194)$  and  $\beta 2(1-217)$ , the high frequency spectra exhibit heme-skeletal marker bands that are similar to the corresponding bands in 5-coordinate NO-sGC. In the low-frequency spectra, isotope-sensitive bands are observed at 526 and 527  $\text{cm}^{-1}$  for  $\beta 1(1-194)$  and  $\beta 2(1-217)$ , respectively. These bands are assigned to the  $\nu(\text{Fe-NO})$  stretching mode based on their isotopic shift and similarity to the  $\nu(\text{Fe-NO})$  frequency in sGC (525  $\text{cm}^{-1}$ ). Isotope-sensitive bands are also observed in the high frequency spectra (1676 and 1677  $\text{cm}^{-1}$  for  $\beta 2(1-217)$  and  $\beta 1(1-194)$ , respectively). Based on their isotopic shift and similarity to the  $\nu(\text{NO})$  frequency in sGC (1677  $\text{cm}^{-1}$ ), these bands are assigned to the  $\nu(\text{NO})$  stretching mode. These data indicate that the NO complexes of  $\beta 1(1-194)$  and  $\beta 2(1-217)$  are 5-coordinate and similar to sGC.

## Oxidation rates

Rat  $\beta 1(1-194)$  was found to oxidize very quickly compared to full-length sGC which is very stable to oxidation (Figure 5). When averaged over three experiments the autoxidation rate of  $\beta 1(1-194)$  was 0.073 per minute. Rat  $\beta 2(1-217)$  was substantially more stable in the ferrous form with an autoxidation rate of 0.003 per minute at 37 °C. It should be noted that while  $\beta 1$  appears to cleanly convert from a 5-coordinate, high-spin Fe(II) complex (Soret  $\lambda_{\text{max}}$  at 431 nm) to a 5-coordinate, high-spin Fe(III) complex (Soret maximum at 390 nm), the  $\beta 2$  oxidation is more complex. The high-spin, ferrous unligated (Soret  $\lambda_{\text{max}}$  at 432 nm) protein appears to oxidize to a mixture of high and low spin ferric complexes (Soret  $\lambda_{\text{max}}$  at 393 and 415, respectively). To support this conclusion, after oxidation with potassium  $(\text{Fe}(\text{CN})_6)^{3-}$  the protein first forms a high spin complex (Soret  $\lambda_{\text{max}}$  at 393 nm) which then converts to a mixture of high and low spin complexes (broad Soret  $\lambda_{\text{max}}$  at 415 nm) (data not shown).

## DISCUSSION

The essential feature of sGC is the use of a ferrous heme with histidyl ligation to trap low concentrations of NO in the presence of a much higher  $\text{O}_2$  concentration. The heme in sGC shows no measurable affinity for  $\text{O}_2$ , hence there is no competition for NO against a severe concentration gradient. The molecular basis for this remarkable ligand discrimination against  $\text{O}_2$  has not been clear until recent findings that sGC-like heme domains are present in prokaryotes. Characterization of these proteins has provided a molecular basis for this ligand discrimination (5,6,33). Since some members of the family form a stable complex with  $\text{O}_2$ , we have named the family H-NOX (Heme-Nitric oxide/Oxygen) to more fully describe the family in terms of its ligand binding properties.

Prior to the discovery of the H-NOX family the strategy to characterize the sGC heme was focused on truncations of the  $\beta 1$ -subunit in an effort to identify a stable heme domain fragment. These studies had identified a 385 amino acid fragment of the  $\beta 1$  subunit that retained the unique ligand binding properties of native sGC (1,7). However, all attempts to crystallize this fragment failed and we surmised that this failure was due to the tendency of this protein to form an unnatural homodimer. Based on this, constructs were designed to determine the minimum size of the  $\beta 1$  fragment that retained the native ligand binding properties. Several constructs were generated and characterized; the smallest a 194 amino acid fragment. Any construct smaller than this (1-189 for example) could not be isolated with a full equivalent of heme and gave low yields upon purification, suggesting that ~190 amino acids is the minimum fragment for a stable heme domain.

$\beta 1(1-194)$  was examined further as reported here, but perhaps more importantly, this fragment served as a query sequence for genome searching that led us to the H-NOX family. In two recent papers we have described two of these H-NOXs. The first protein was a 181 amino acid protein from *V. cholerae* (VCA0720), and the other was a 188 amino acid heme domain

fragment from the *T. tengcongensis* Tar4 protein (*TtTar4H*) (6). The heme domain from *V. cholerae* has ligand binding properties like that of sGC, namely the ferrous protein does not form an O<sub>2</sub> complex but does form stable complexes with NO and CO. The *T. tengcongensis* heme domain on the other hand forms a very stable ferrous O<sub>2</sub> complex and like the globins will also form a CO and a NO complex. We have also determined the crystal structure of the heme domain from *T. tengcongensis* which, as mentioned above, provided a molecular explanation for the discrimination against O<sub>2</sub> in the sGC-like proteins (5,33). In fact, based on the structure of the *T. tengcongensis* H-NOX, we have recently converted  $\beta$ 1(1-385) into a protein that now forms a stable Fe(II)-O<sub>2</sub> complex (33).

Although the  $\beta$ 1(1-194) construct is the smallest fragment that was stable enough to characterize, it is clear that it is not as stable as sGC. The rate of oxidation to Fe(III) is very slow in sGC and as shown in Figure 5, occurs much more quickly in  $\beta$ 1(1-194). The Fe(II) unliganded, Fe(II)-NO, and Fe(II)-CO complexes are all identical to sGC, however, once the Fe(II) unliganded complex is exposed to air, oxidation occurs. A Fe(II)-O<sub>2</sub> complex is not observed during oxidation and the molecular steps involved in the oxidation are unknown.  $\beta$ 2(1-217) on the other is much more stable (Figure 5) and more comparable to sGC and the previously characterized  $\beta$ 1(1-385) in this respect.  $\beta$ 2(1-217) is 43% identical to the  $\beta$ 1 subunit and shares the key spectroscopic properties of sGC. Specifically, the Fe<sup>+2</sup>-unliganded forms of these proteins are 5-coordinate, high-spin and the CO complexes are 6-coordinate, low-spin.

Moreover, like sGC, both  $\beta$ 1(1-194) and  $\beta$ 2(1-217) form a CO complex with unusually high  $\nu(\text{CO})$  and low  $\nu(\text{Fe-CO})$  stretching frequencies. sGC has one of the highest reported  $\nu(\text{CO})$  stretching frequencies, and, based on this observation, it was previously suggested that the distal pocket of sGC has significant negative polarity (34). This is based on the observation that the  $\nu(\text{CO})$  and  $\nu(\text{Fe-CO})$  frequencies are inversely correlated and are sensitive to distal pocket polarity (35,36). Negative polarity in the distal pocket can inhibit Fe  $d_{\pi} \rightarrow \text{CO } \pi^*$  back-bonding, weakening the Fe-CO bond order and strengthening the C-O bond order. However, a homology model of  $\beta$ 1(1-194) and  $\beta$ 2(1-217) based on the structure of the *TtTar4H* H-NOX shows no significant region of negative polarity that is close to the bound ligand. Figure 6 shows this model when viewed from below the heme and looking up into the distal pocket. Tyr140 is clearly visible in the *T. tengcongensis* H-NOX distal site but there is no H-bonding donor or residue that would provide negative polarity in either model of  $\beta$ 1(1-194) or  $\beta$ 2(1-217) suggesting that the negative polarity hypothesis may not be correct. A  $\nu(\text{Fe-CO})/\nu(\text{CO})$  correlation, which was generated from model porphyrin and hemoprotein frequencies, is shown in Figure 7. The closest  $\nu(\text{CO})$  and  $\nu(\text{Fe-CO})$  frequencies to sGC are from a Mb mutant (H64V/V68T) which increases negative polarity in the distal pocket (37). The high  $\nu(\text{C-O})$  and low  $\nu(\text{Fe-CO})$  stretching frequencies observed in  $\beta$ 1(1-194) and  $\beta$ 2(1-217), while consistent with distal pockets containing negative polarity, would seem to be dependent on other factors.

$\beta$ 1(1-194) has higher  $\nu(\text{Fe-CO})$  and lower  $\nu(\text{CO})$  frequencies compared to sGC. For example, the  $\nu(\text{CO})$  frequency is observed at 1968 cm<sup>-1</sup> compared to 1987 cm<sup>-1</sup> for sGC. Moreover, a broad band with two peaks of roughly equal intensity at 477 and 496 cm<sup>-1</sup> were observed for the  $\nu(\text{Fe-CO})$  stretching frequencies. Two frequencies were also observed for sGC and for the previously characterized heme-binding region  $\beta$ 1(1-385) (25): An intense band at 472 cm<sup>-1</sup> and a very weak shoulder at 487 cm<sup>-1</sup> for sGC and an intense band at 478 cm<sup>-1</sup> and a weaker one at 494 cm<sup>-1</sup> for  $\beta$ 1(1-385). Although the two bands observed in sGC and  $\beta$ 1(1-385) have been attributed to different CO conformations, no definitive results have emerged to support this hypothesis.

$\beta$ 2(1-217) is very similar to the heme domain of sGC in that it is unable to form a stable Fe<sup>+2</sup>-O<sub>2</sub> complex but does form a 5-coordinate complex with NO. If we assume that the ligand binding characteristics of  $\beta$ 2(1-217) is representative of the full-length subunit, an assumption



that is consistent with reduced  $\beta 1(1-194)$  being spectrally similar to full-length sGC, then it is likely that the  $\beta 2$ -subunit also senses NO. These results are consistent with two reports that show NO-sensitive guanylate cyclase activity in  $\beta 2$ -expressing cells. The first showed NO-stimulated activity in rat  $\beta 2$ -expressing SF9 cell lysates (10) and the second in rat  $\beta 2$ -expressing COS 7 cell lysates (11). Although the enzyme was not purified, the conclusion was that homodimeric  $\beta 2$  was the active protein. This would be the first example of a sGC that contained two heme cofactors per dimer and raises interesting questions regarding heme occupancy and regulation.

In summary, this report has identified and characterized the minimum functional ligand binding heme domain derived from sGC. This 194 amino acid N-terminal fragment was then used to identify prokaryotic homologs of predicted hemoprotein sensor domains that probably function as NO sensors in some organisms and  $O_2$  sensors in others. In addition, the ligand binding properties of a heme domain derived from the  $\beta 2$ -subunit are reported. These results and others on this broadening family of proteins are providing key details towards a comprehensive characterization.

## ACKNOWLEDGEMENTS

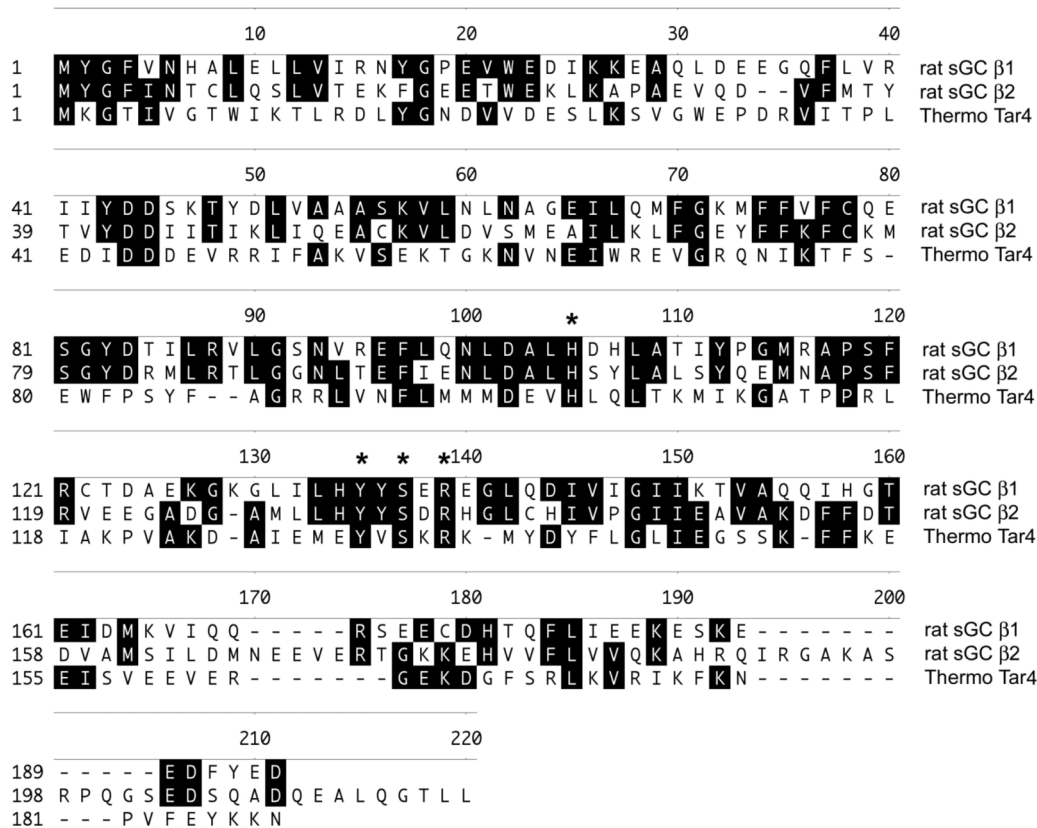
We thank Chinmay Majumdar for help with the initial characterization of  $\beta 1(1-194)$  and Dr. Elizabeth Boon for critical input.

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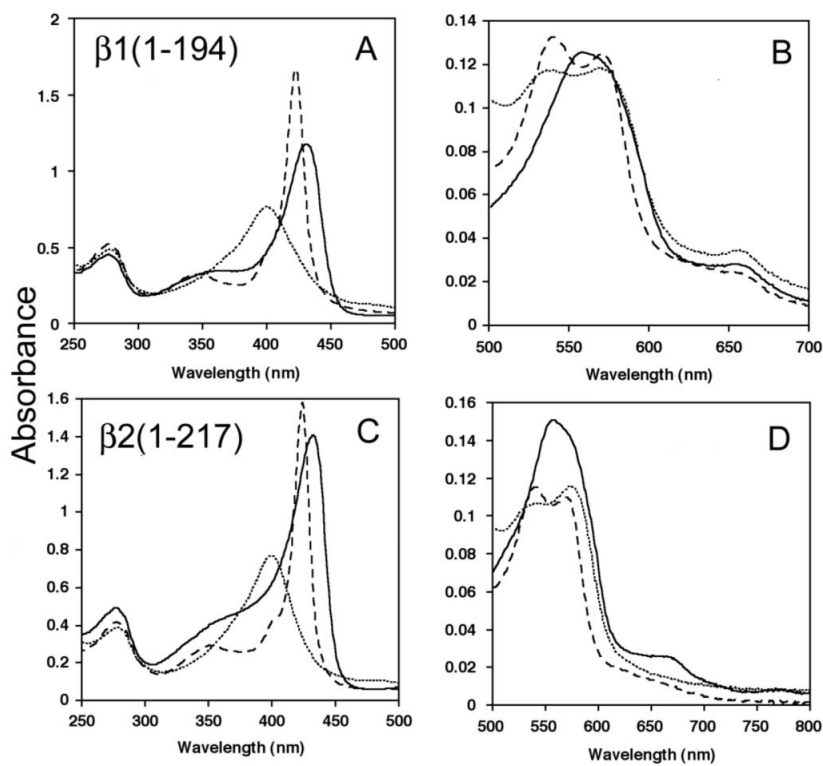
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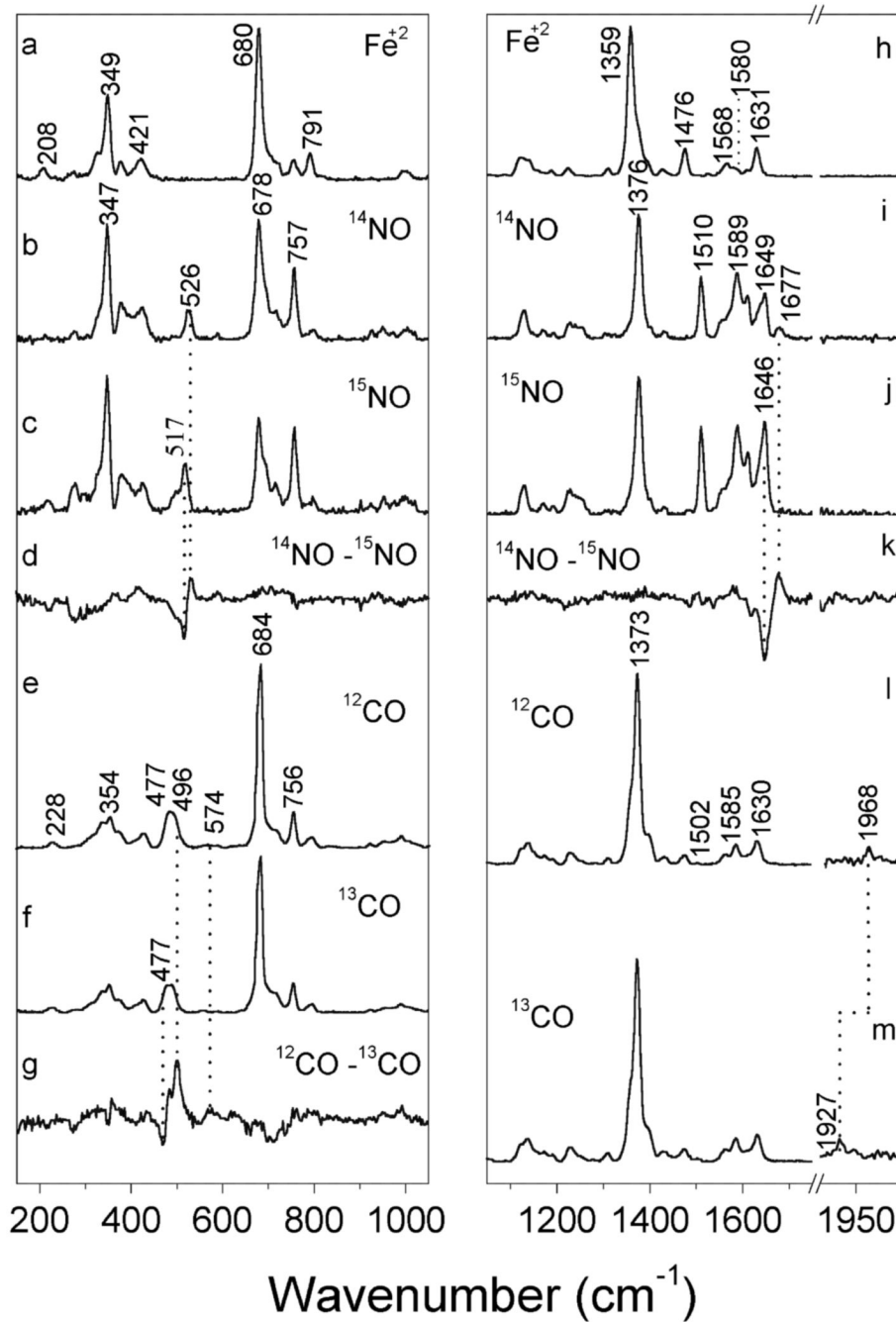


**Figure 1.** Alignment of rat β1(1-194) with rat β2(1-217) and the *Thermoanaerobacter tengcongensis* H-NOX domain. The alignment was generated using DNASTAR's Megalign program and the Clustal W algorithm. Conserved residues that are identical to the consensus sequence are shaded in black. Conserved residues discussed in the text are marked with an asterisk (His105, the heme ligand, and Tyr135, Ser137 and Arg139 using the rat β1 numbering).

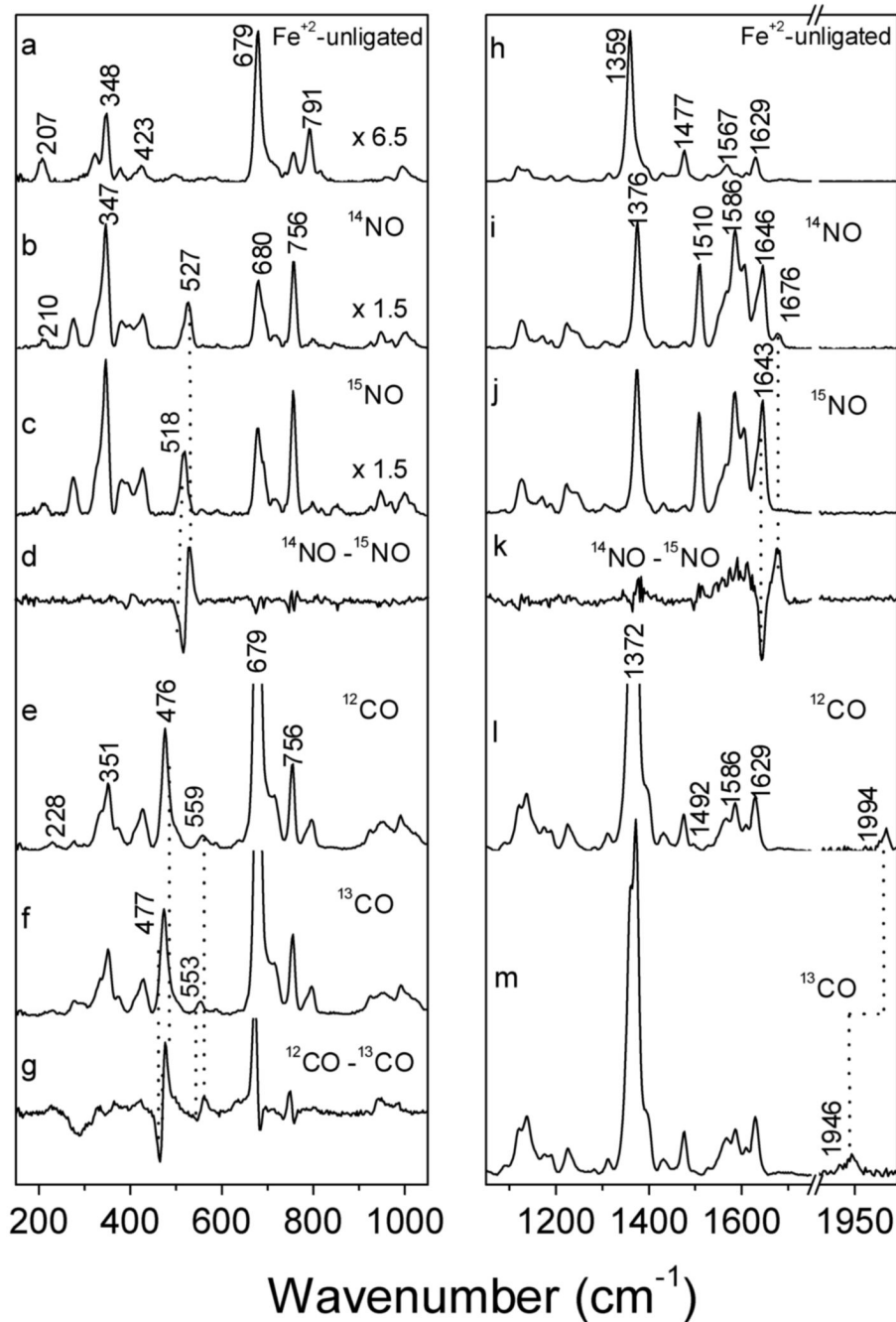


**Figure 2.** Electronic absorption spectra of  $\beta 1(1-194)$  and  $\beta 2(1-217)$  showing the Soret maximum and  $\alpha/\beta$  region. Reduced, unligated (—); CO complex (- - -); and NO complex (·····). Heme concentration was  $\sim 7 \mu\text{M}$ .

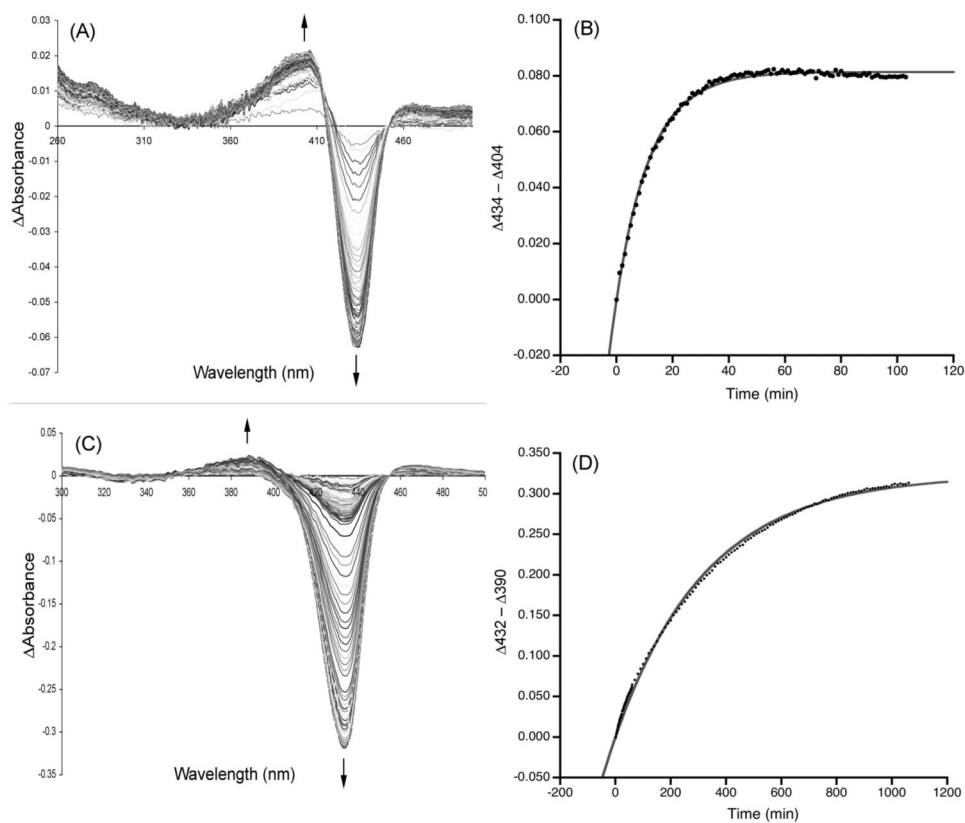




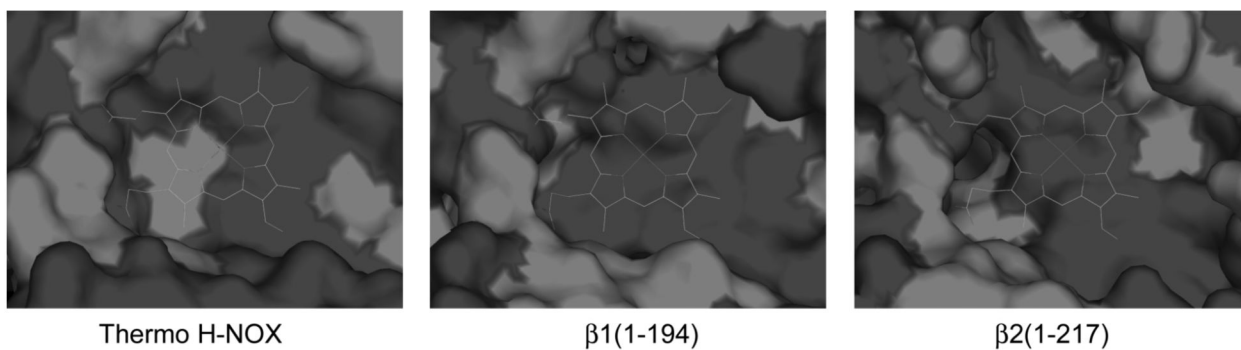
**Figure 3.**  $\beta 1(1-194)$  resonance Raman spectra.  $\text{Fe}^{2+}$ -unligated in the low (a) and high frequency region (h);  $^{14}\text{NO}$  and  $^{15}\text{NO}$  complexes and their difference spectra in the low (b, c and d) and high frequency regions (i, j and k), respectively;  $^{12}\text{CO}$  and  $^{13}\text{CO}$  complexes and their difference spectra in the low (e, f and g) and high frequency regions (l and m), respectively. Heme concentration was 14  $\mu\text{M}$ . The Raman intensity was normalized to the  $\nu_4$  mode. The asterisks indicate subtraction artifacts.



**Figure 4.**  $\beta 2(1-217)$  resonance Raman spectra.  $\text{Fe}^{+2}$ -unligated in the low (a) and high frequency region (h);  $^{14}\text{NO}$  and  $^{15}\text{NO}$  complexes and their difference spectra in the low (b, c and d) and high frequency regions (i, j and k), respectively;  $^{12}\text{CO}$  and  $^{13}\text{CO}$  complexes and their difference spectra in the low (e, f and g) and high frequency regions (l and m), respectively. Heme concentration was 19  $\mu\text{M}$ . The Raman intensity was normalized to the  $\nu_4$  mode. The asterisks indicate subtraction artifacts.

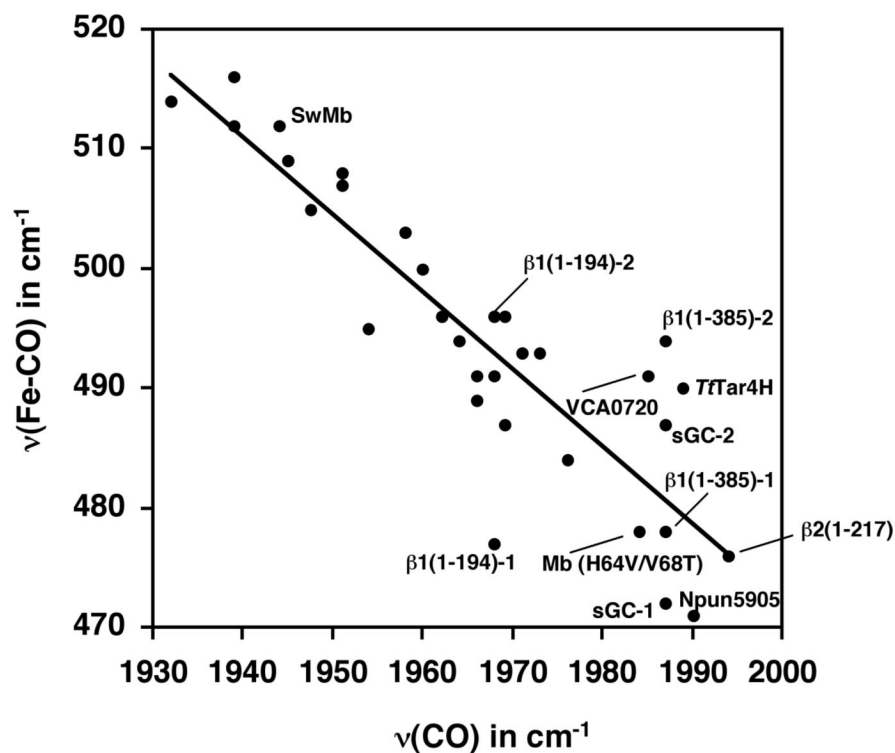


**Figure 5.** Autoxidation rates of  $\beta 1(1-194)$  and  $\beta 2(1-217)$ . Shown in panel A are difference spectra associated with the autoxidation of  $\beta 1(1-194)$  and panel B shows a plot of the  $\Delta 434 - \Delta 404$  versus time and a fit (grey line) to the data from which a rate constant was derived. Panel C shows the difference spectra for the autoxidation of  $\beta 2(1-217)$  and panel D shows a plot of  $\Delta 432 - \Delta 390$  versus time and a fit to those data (grey line).



**Figure 6.**

Homology models of the distal pockets. Homology models of the  $\beta 1$  and  $\beta 2$  were generated using MODELLER following the methods described in (13). The view in the model is from below the heme and looking up into the distal pocket. H-bonding donors or residues that would provide negative polarity are shown in light grey. Tyr140 in the *Tt*Tar4H distal site is clearly visible, however, there is none visible in either model of  $\beta 1(1-194)$  or  $\beta 2(1-217)$ .



**Figure 7.**  $\nu(\text{Fe-CO})/\nu(\text{CO})$  correlation for 6-coordinate CO adducts of model porphyrins and hemoproteins. Values for the metalloporphyrins and hemoproteins were obtained from: Kerr et al., Table 2, complexes 1-15 (38); Table 3 in this work; CooA (39,40); *EcDosH* (24); *BjFixL* (24); Heme oxygenase (41,42); Sperm whale (Sw) Mb (21); Mb (H64V/V68T) (37). For sGC,  $\beta 1(1-385)$  and  $\beta 1(1-194)$ , the numeric suffix refers to the CO conformation with the low and high  $\nu(\text{Fe-CO})$  frequency (1 and 2, respectively).



**Table 1**  
UV-vis Peak positions<sup>a</sup> and Extinction Coefficients<sup>b</sup>

protein	Soret	$\beta$	$\alpha$	ref
Fe <sup>2+</sup> -unligated				
sGC	431 (111)		555 (14)	(2)
$\beta$ 1(1-194)	431 (166)		558 (17.7)	*
$\beta$ 2(1-217)	433 (155)		557 (17.1)	*
Hb	430 (133)		555 (12.5)	(15)
CO-bound				
sGC	423 (145)	541 (14)	567 (14)	(2)
$\beta$ 1(1-194)	423 (235)	541 (18.7)	571 (17.6)	*
$\beta$ 2(1-217)	424 (248)	541 (18.1)	570 (17.2)	*
Hb	419 (191)	540 (13.4)	569 (13.4)	(15)
NO-bound				
sGC	398 (79)	537 (12)	572 (12)	(2)
$\beta$ 1(1-194)	400 (108)	537 (16.5)	570 (16.7)	*
$\beta$ 2(1-217)	399 (121)	543 (16.8)	574 (18.1)	*
Hb	418 (130)	545 (12.6)	575 (13.0)	(15)

<sup>a</sup>  
nm

<sup>b</sup>  
mM<sup>-1</sup> cm<sup>-1</sup>

\*  
this work

¶  
also a shoulder at 400 nm

**Table 2**  
Heme Skeletal Vibrations and Vibrational Modes<sup>a</sup>: Fe<sup>+2</sup>-unligated proteins

protein	cs <sup>b</sup>	$\nu_{10}$	$\nu_2$	$\nu_3$	$\nu_4$	Fe-His	ref.
sGC	5	1606	1562	1471	1358	204	(16,17)
$\beta$ 1(I-385)	5	1607	1563	1474	1358	206	(25)
$\beta$ 1(I-194)	5	1608	1568	1476	1359	208	*
$\beta$ 2(I-217)	5	1609	1567	1477	1359	207	*
T/Tar4H	5	1600	1575	1471	1354	218	(6)
Mb	5	1607	1563	1471	1357	220	(18)
HemA1-Bs	5	m <sup>c</sup>	1558	1469	1352	225	(29)
AAPDEA1H	5	1607	1557	1469	1354	212	(24)
Arcyt c'	5	1603	1577	1469	1351	231	(27)

<sup>a</sup>  $\text{cm}^{-1}$

<sup>b</sup> coordination state

\* this work

<sup>c</sup> not reported

**Table 3**  
Heme Skeletal Vibrations and Vibrational Modes<sup>a</sup>: CO complexes

protein	cs <sup>b</sup>	v <sub>10</sub>	v <sub>2</sub>	v <sub>3</sub>	v <sub>4</sub>	v(Fe-CO)	v(C-O)	δ <sup>c</sup>	ref.
sGC	6	1629	1583	1500	1371	472 <sup>d</sup> /487	1987	562	(16,17)
β1(1-385)	6	nd <sup>d</sup>	1582	1496	1373	478 <sup>d</sup> /494	1987	564	(25)
β1(1-194)	6	1631	1586	1502	1375	477/496	1968	574	*
β2(1-217)	6	1629	1586	1493	1372	476	1994	559	*
TTr4H	6	1620	1580	1496	1369	490	1989	567	(6)
Mb	6	1637	1587	1498	1372	512	1944	577	(21)
HemA1-Bs	6	nr <sup>e</sup>	1578	1495	1368	494	1964	nr	(29)
A <sub>2</sub> PDEA1H	6	nr	nr	nr	nr	493	1973	581	(24)
Axycyt c	6	no <sup>f</sup>	1596	nr	1368	491	1966	572	(27)

<sup>a</sup> cm<sup>-1</sup>

<sup>b</sup> coordination state

<sup>c</sup> Fe-C-O bending mode

<sup>d</sup> represents the dominant frequency

<sup>e</sup> not determined

\* this work

<sup>e</sup> not reported

<sup>f</sup> not observed

**Table 4**  
Heme Skeletal Vibrations and Vibrational Modes<sup>a</sup>: NO complexes

protein	cs <sup>b</sup>	$\nu_{10}$	$\nu_2$	$\nu_3$	$\nu_4$	$\nu(\text{Fe-NO})$	$\nu(\text{N-O})$	ref.
sGC	5	1646	1584	1509	1375	525	1677	(16,17)
$\beta 1(1-194)$	5	1647	1586	1510	1376	526	1677	
$\beta 1(1-385)$	5	1646	1585	1509	1376	526	1676	(25)
$\beta 2(1-217)$	5	1646	1586	1510	1376	527	1676	
T/Tar4H	6	1625	1580	1496	1370	553	1655	(6)
Mb	6	1638	1584	1501	1375	554	1624	(19,20)
Arcyt c'	5	1641	1592	1506	1373	526	1661	(27)
Arcyt c	6	1638	1596	1504	nr <sup>d</sup>	579	1624	(28)
FixLN (20 °C)	6	1632	nr	1498	nr	558	no <sup>e</sup>	(26)
FixLN (20 °C)	5	1646	nr	1509	nr	525	1676	(26)
A-rPDEA1H	6	nr	nr	nr	nr	560	1637	(24)
Mb(H64V)	6	nr	nr	nr	nr	557	1640	(22)
Mb(H93G)	5	nr	nr	nr	nr	535	1670	(22)
Mb(H64V/H93G)	5	nr	nr	nr	nr	534	1684	(22)
CooA	5	nr	nr	nr	nr	523	1672	(30)
Mb H93Y	5	nr	nr	nr	nr	524	1672	(23)

<sup>a</sup>  $\text{cm}^{-1}$

<sup>b</sup> coordination state

\* this work

<sup>d</sup> not reported

<sup>e</sup> not observed