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Determination of torque generation from the power stroke of *Escherichia coli* F₁-ATPase

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Abstract

The torque generated by the power stroke of *Escherichia coli* F₁-ATPase was determined as a function of the load from measurements of the velocity of the γ -subunit obtained using a 0.25 µs time resolution and direct measurements of the drag from 45 to 91 nm gold nanorods. This result was compared to values of torque calculated using four different drag models. Although the γ -subunit was able to rotate with a 20× increase in viscosity, the transition time decreased from 0.4 ms to 5.26 ms. The torque was measured to be 63±8 pN nm, independent of the load on the enzyme.

Keywords

F1-ATPase; Gold nanorod; Molecular motor; High-speed data acquisition; Single molecule

1. Introduction

The F_1F_0 ATP synthase is required for energy conversion in almost all living organisms. It is composed of two complexes, the F_1 -ATPase (*Escherichia coli* subunits α , β , γ , δ and ϵ) that is peripheral to bio-energetic membranes, and F_0 (*E. coli* subunits a, b, and c) that is an integral membrane complex. In F_1 , a ring of three $\alpha\beta$ -subunit heterodimers surrounds the γ -subunit coiled-coil domain while the γ -subunit 'foot' domain anchors F_1 to the c_{10} -subunit ring of *E. coli* F_0 [1,2]. Functionally F_1F_0 is a rotary machine, where rotor subunits γ , ϵ and c rotate relative to the stator composed of subunits a, b, α , β and δ [3,4].

Each $\alpha\beta$ heterodimer contains a site that catalyzes ATP synthesis/hydrolysis. The synthesis of ATP is driven by $\Delta\mu_{H^+}$ driven rotation the c-ring of F_0F_0 that forces conformational changes in the $(\alpha\beta)_3$ ring [4]. Conversely, ATP hydrolysis can drive γ -subunit rotation in the opposite direction [3]. The three catalytic sites operate in an alternating site mechanism in which binding of ATP to the empty catalytic site induces release of ADP and phosphate from an adjacent site concurrent with rotation [5].

Direct observation of ATPase-driven rotation by microscopy is possible when F_1 is mounted on a slide and a visible probe is attached to the γ -subunit. Rotation of the γ -subunit occurs in discrete 120° steps, each of which is associated with hydrolysis of an ATP molecule [3,6-8]. The torque generated by the molecule is dependent upon the strength of the power stroke associated with each hydrolysis event. The power stroke is dependent upon the velocity of the γ -subunit when it rotates. At saturating ATP concentrations, the velocity is determined by the transition time; i.e. the time required for γ to move from one dwell location to the next. By

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Limitations of the temporal resolution of these methods resulted in using the number of revolutions per second to approximate rotational velocity [3,10]. This method failed to account for the fact that the motor spends most of the time in a state where γ is not rotating. The number of rotations per second gives an average velocity which is the integral of all the instantaneous velocities that occur during one 360° rotation. When the motor is in a dwell state it has a velocity of zero. Thus, the velocity of the power stroke is faster than the average value resulting in an underestimation of the torque. A related, but less significant, problem occurs with the actin filament curvature approach. Since the filament relaxes during each dwell, the net curvature is less than what would occur if the motor rotated continuously.

labeled with a gold nanosphere [10].

Recently, the rotational events of the γ -subunit were measured with a time resolution of 0.25 µs using gold nanorods as probes such that the speed of rotation could be distinguished from the dwell time [11]. Based on these measurements, values of torque as high as 61 pN nm were calculated by estimating the drag [11]. However, the accuracy of these torque calculations was limited by the ability to estimate the drag on the gold nanorod. Several different models have been proposed to estimate the drag on the probe based on the size, shape, viscosity, and binding orientation [3,11-13].

We now report the torque generated by the power stroke of F_1 -ATPase as a function of the load on the enzyme. Measurements of the velocity were obtained using a 0.25 µs time resolution and directly measuring the drag. This result was compared to values of torque calculated using four different drag models. Rotational assays were performed as described recently [11,14, 15] as a function of polyethylene glycol (PEG 400) which increases the viscosity of the reaction medium. Although the γ -subunit was able to rotate with a 20× increase in viscosity, the transition time decreased from 0.4 ms to 5.26 ms and the torque remained 63±8 pN nm, independent of the load on the enzyme.

2. Results

Fig. 1 shows the measured viscosity of the assay buffer as a function of the percentage (vol/ vol) of PEG 400. These data were used to calculate the shear stress as a function of the shear rate (Fig. 2). The linear dependence between these parameters indicates that the assay buffer containing PEG 400 behaves as a Newtonian fluid. Thus, PEG 400 molecules are too small to be pulled along by the rotating nanorod, and do not make secondary nonlinear contributions to the drag.

The drag coefficient was determined directly as described by Hunt et al. [16] by measuring the amount of angular change in individual gold nanorods near the surface in the absence of F_1 (Fig. 3). Nanorods were selected which displayed a color change indicative of rotational movement, and which did not move significantly in the *z* direction, i.e. remained in focus. The dynamic range of scattered light intensity from a nanorod was measured by rotating the polarizer 360° in 10° increments. The angular change is related to the intensity by: $I=A \quad \sin \quad (\phi) + B,$ (1)

$$D = \frac{\Delta \phi^2}{2\Delta t}.$$
 (2)

The change in the angle of gold nanorods near the surface was measured at data acquisition rates from 10 to 200 kHz in 10 kHz increments, which made the determination of the diffusion constant, D, independent of time artifacts. Using Einstein's relation (Eq. (3)) the drag coefficient, Γ , was determined from the diffusion coefficient,

$$\Gamma = \frac{k_{\rm B}T}{D}.$$
(3)

Fig. 3 also shows a comparison of the various drag values calculated from models that assume that the gold nanorod has the properties of a propeller [12], a nanosphere [13], a rod [11], and a long filament [3]. Based on the model for a propeller, the drag force is approximated by Eq. (4):

$$\Gamma = \frac{4\pi\mu \left(L_1^3 + L_2^3 \right)}{3 \cosh^{-1} \left(h/r \right)},\tag{4}$$

where L_1 and L_2 are the lengths of the propeller extending from the rotational axis, r is the radius of the rod, μ is the viscosity of the medium, and h is the height of the cylinder axis relative to the surface. The height of the propeller was estimated to be 17 nm based on the dimensions of F₁ and avidin from crystal structures [17,18]. When h is small, most energy dissipation is believed to occur in the gap between the propeller and the surface [19].

The models used to estimate the drag force generated by a nanoscale rod, a nanosphere, and a filament are described by Eqs. (5)-(7), respectively:

$$\Gamma = 8\pi\mu r^3 \tag{5}$$

$$\Gamma = 16\pi\mu r^3 + 12\pi\mu r L_1^2 + 12\pi\mu r L_2^2 \tag{6}$$

$$\Gamma = \frac{4\pi\mu \left(L_1^3 + L_2^3\right)}{3 \ln \left(\frac{L}{2r} - 0.447\right)}.$$
(7)

The results in Fig. 3 show that the propeller model provides the closest approximation to the experimental data. Other models estimated the drag to be as much as 4-fold higher than the measured value at 60% PEG.

To determine the velocity of the power stroke generated during F_1 -ATPase-driven rotation, transition times were measured by tracking a 91×45 nm gold nanorod, through 90° of a rotation as described in Spetzler et al. [11], and the velocity plotted as a function of the viscosity (Fig. 4). The velocity of the transition decreased with increasing viscosity.

Fig. 5 shows the torque values calculated from the drag and the velocity by: $T=\Gamma w$, (8)

where *w* is the angular velocity as a function of viscosity. The value calculated from the direct measurements of the drag was 63 pN nm \pm 8 nm, and was independent of the load applied to the motor. The estimated value for the torque based on the propeller model (~55 pN nm) was similar to the measured result. However, the estimated values based on the sphere, the rod, and

filament models were substantially different, indicating that the models provide poor estimates of the drag on the nanorod. For the models of the sphere and rod, the torque values were greater than the maximum amount of energy available to the enzyme in the absence of PEG 400 and increased linearly as a function of the viscosity violating the second law of thermodynamics.

3. Discussion

The torque value obtained from measured values for both the velocity and the drag reported here was 63 pN nm±8 nm. As the load increased, there was a corresponding decrease in the velocity indicating that the torque was constant over the range of viscous conditions examined. This value is consistent with the highest measured values for the torque generated by the F₁-ATPase [11], but is significantly higher than values obtained elsewhere [3,9,10]. Independent evidence that supports the torque reported here can be derived from measurements of the $H^+/$ ATP ratio of proton transport-coupled ATP synthesis and hydrolysis catalyzed by liposomes that contain chloroplast F_0F_1 [20]. In this study, the rates of ATP synthesis and hydrolysis were measured with the luciferin/luciferase assay after an acid-base transition at various [ATP]/ ([ADP][Pi]) ratios as a function of ΔpH . By determining the point at which the free energy from the [ATP]/([ADP] [Pi]) ratio was in equilibrium with the transmembrane electrochemical potential difference such that there was no net synthesis or hydrolysis of ATP, H⁺/ATP ratios of 3.9-4.0 and standard Gibbs free energies of ATP synthesis of 37±2 kJ/mol and 36±3 kJ/mol at pH 8.45 and 8.05, respectively, were obtained. If we assume that the equilibrium point (when there is no net ATP synthesis or hydrolysis) occurs when $\Delta \mu_{\rm H^+}$ is equivalent and opposite to the force generated by F_1 -ATPase hydrolysis, then the Gibbs free energy is equal to the energy used to generate torque. Based on this assumption the chloroplast F_1 -ATPase generates 61.4 ±3 pN nm and 59.8±5 pN nm of torque at pH 8.45 and 8.05, respectively. These values are comparable to those reported here for the torque generated by E. coli F_1 -ATPase.

Previous estimates of the torque F_1 produces against the friction of the viscous medium range from 40 to 50 pN nm [9,10]. From these estimates, it was calculated that the work F_1 can do in a 120° step is about 80-90 pN nm or ~20 kBT, calculated as 40-44 pN nm multiplied by (2/3) π radians. This is close to the physiological $|\Delta G|$ for ATP hydrolysis, and was interpreted to mean that the efficiency of the enzyme to convert chemical energy into mechanical energy is nearly 100% under physiological conditions [21]. Since the experiments were not performed under physiological conditions, not all sources of energy were taken into consideration, thus, the efficiency would be lower, consistent with predictions made by Karplus et al. [22].

The results presented here show that the enzyme is capable of doing more work than would be predicted by the amount of free energy under physiological conditions, suggesting that there are additional sources of energy. Gao et al. [23] and Yang et al. [24] calculated that the differential in binding constants could yield ~80 pN nm worth of work using cellular concentrations of 3 mM ATP, 0.4 Mm ADP, and 6 mM Pi. In the experiments reported here, there were only trace amounts, <1%, of ADP and Pi in the 1 mM ATP solution used to induce rotation allowing as much as 60 pN nm of free energy to be generated from the concentration gradient between [ATP]/([ADP][Pi]). Based upon the data presented here, the work done by the enzyme is ~130 pN nm. The energy available from substrate binding and sub-sequent product release, and the concentration gradient yield a net amount of 140 pN nm of free energy, suggesting that the enzyme is at least 90% efficient.

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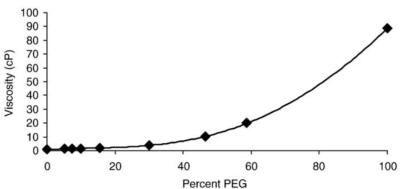


Fig. 1.

The viscosity of the buffer used to make single molecule rotation measurements (Rotation Buffer (50 mM Tris-Cl/pH 8.0, 10 mM KCl)) as a function of its percent composition of PEG 400 (vol/vol). The measurements were made using a Brookfield DV-E viscometer with a UL adapter at 28 °C.

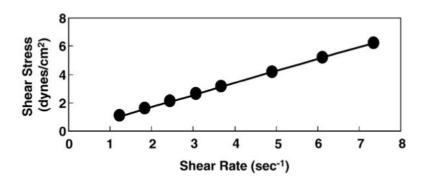


Fig. 2.

The shear stress as a function of the shear rate of the Rotation Buffer containing PEG 400. This was determined by measuring the effect of increasing the velocity on a rotating probe in the Brook field DV-E viscometer.

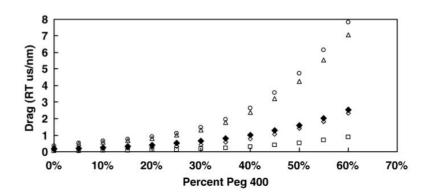


Fig. 3.

Drag as a function of viscosity. The solid symbol (\spadesuit) is the experimental value, while the open symbols are the theoretical values for a rod (Δ) , a sphere (\circ) , a filament (\Box) , and a propeller (\diamondsuit) calculated from Eqs. (4)-(7).

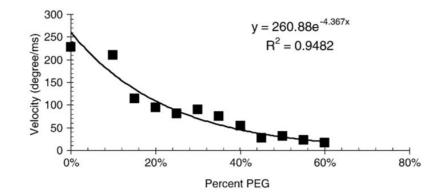


Fig. 4. The velocity of the γ -subunit power stroke as a function of viscosity.

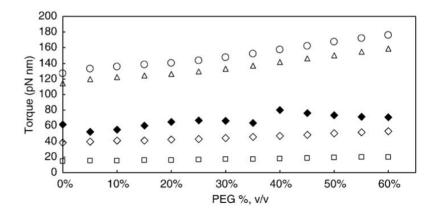


Fig. 5.

The torque generated by the F_1 as a function of the load on the motor. The solid symbol (\blacklozenge) is the experimental value, while the open symbols are the theoretical values for a rod (Δ), a sphere (\circ), a filament (\Box), and a propeller (\diamondsuit) calculated from Eqs. (4)-(7).