Efficiency of Histidine-Associating Compounds for Blocking the Alzheimer's A β Channel Activity and Cytotoxicity

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ABSTRACT The opening of the Alzheimer's $A\beta$ channel permits the flux of calcium into the cell, thus critically disturbing intracellular ion homeostasis. Peptide segments that include the characteristic histidine (His) diad, His¹³ and His¹⁴, efficiently block the $A\beta$ channel activity, blocking $A\beta$ cytotoxicity. We hypothesize that the vicinal His-His peptides coordinate with the rings of His in the mouth of the pore, thus blocking the flow of calcium ions through the channel, with consequent blocking of $A\beta$ cytotoxicity. To test this hypothesis, we studied $A\beta$ ion channel activity and cytotoxicity after the addition of compounds that are known to have His association capacity, such as Ni²⁺, imidazole, His, and a series of His-related compounds. All compounds was correlated with the number of imidazole side chains in the blocker compounds. These data reinforce the premise that His residues within the $A\beta$ channel to the cytotoxicity of exogenous $A\beta$.

INTRODUCTION

A consensus on the primary mechanisms that cause neuronal damage in Alzheimer's disease (AD) remains elusive, but numerous reports have associated the cytotoxicity of amyloid beta peptides (A β) with the neurodegeneration observed in specific brain areas of AD patients (1–4). It has been shown that addition of fresh aggregates of A β to cell cultures generates a potentially toxic increase in the intracellular calcium concentration (5–9). Years of research supports the concept that disturbances of intracellular calcium homeostasis may play a pathological role in the neurodegeneration associated with AD (10–13).

A mechanism for the A β peptide-induced increase in intracellular calcium was originally proposed based on the formation of an independent tromethamine and aluminumsensitive A β channel (14). This A β channel, which permits the entrance of extracellular calcium ions into the cell (15,16), has been confirmed in a variety of membranes by many researchers over the past decade (17-22), has been observed with atomic force microscopy (AFM) (23,24), and has been subjected to theoretical modeling (25-27). The asymmetry in one of the models (25) explains our finding that zinc preferentially binds and blocks only one side of the A β channel (28). It has been frequently demonstrated in studies of different metalloproteases, as well as in the A β molecule, that sites rich in histidine (His) and anionic residues are associated with Zn^{2+} binding (29–33). Because of the unique chemical nature of the His residue, it has a strong metal affinity. His residues

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act as a ligand to a metal center (34) bridging imidazole groups from the side chains of His residues (33). In the theoretical models of Durell et al. (25), the least energy calculations for the full size $A\beta$ channels, imbedded in a lipid environment, position the rings of His¹³ and His¹⁴ of the A β molecule around the entrance of the putative pore. To test this prediction for the modeling algorithm, we recently showed that peptide segments that copy sectors of the sequence of the A β molecule containing the two neighboring His¹³ and His¹⁴ residues effectively block $A\beta$ channel activity in planar lipid bilayers (16,35,36). When the His-His diad is substituted with residues that lack a propensity to interact with His residues, the peptides lose their effectiveness to both block the A β channel and prevent A β peptide cytotoxicity (36). Furthermore, methylation of the imidazole side chains of His residues in the $A\beta$ peptide prevents the formation of His bridges, and also results in abolition of A β peptide neurotoxicity (37). We have interpreted these data to indicate that the blocker peptides containing the His-His diad form conventional coordination complexes with complementary His-His diads in the other A β subunits of the A β channel. Disruption of these complexes by the His-His diadic peptide segments prevents the formation of the potentially toxic $A\beta$ channels. Therefore, we now hypothesize that compounds that are known to have His coordinating capacity, such as Ni²⁺, imidazole, His, and Hisrelated compounds should interfere with the currents and cytotoxicity of A β channels. We report here that the hypothesis is supported by the experimental data. We found that His-coordinating and His-related compounds can efficiently block A β channels incorporated in artificial membranes, and can also entirely prevent A β cytotoxicity. Therefore, we believe that these results reinforce the concept that the toxicity of the A β peptide on cells is basically exerted through the

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capacity of A β peptide to form A β ion channels in the cell surface membrane.

MATERIALS AND METHODS

Planar lipid bilayer methodology

A suspension of palmitoyloleoyl phosphatidylserine and palmitoyloleoyl phosphatidylethanolamine (Avanti, Alabaster, AL), 1:1, in *n*-decane was prepared. This suspension was applied to an orifice of ~100–120 μ m diameter with a Teflon film separating two compartments, 1.2 mL volume each. The ionic solutions in the compartments contained asymmetrical concentrations of CsCl (200 cis/50 trans mM) and symmetrical 0.5 mM CaCl₂ and 5 mM K-HEPES, pH 7. The two ionic compartments were electrically connected to the input of a voltage clamp amplifier. Current was recorded using a patch clamp amplifier and data were stored on computer disk memory. Offline analysis of the channel activity was carried out using the software package pClamp (Axon Instruments, Foster City, CA). Incorporation of A β peptide into the bilayer was obtained by adding an aliquot of proteoliposome (A β -liposome) suspension to the solution in the *cis* side of the planar lipid bilayer chamber and stirring.

Preparation of proteoliposomes

Liposomes were prepared by hydration of air-dried palmitoyloleoyl phosphatidylserine (10 mg) with 1 M potassium aspartate, pH 7.0 (1 mL), followed by water sonication for 5 min. The liposome suspension (50 μ L) was mixed with a stock aqueous solution of A β peptide (1 mg/mL, obtained from Bachem, Torrance, CA, and AnaSpec, San Jose, CA), followed by sonication.

Materials and His-related compounds

NiCl₂, imidazole, and L-His were purchased from Sigma-Aldrich (St. Louis, MO). The His-related compounds NAHIS01 (Ac–His–CONH2), NAHIS02 (Ac–His–His–CONH2), and NAHIS04 (Ac–His–His–His–His–CONH2), which contains one, two, and four His residues, respectively, were synthesized as amide and capped in the amino terminal with acetic anhydride. NAHIS02-(π -Met) (Ac–His– π -Met–His– π -Met–CONH2) was synthesized combining Fmoc-His(3-Me)-OH (Fmoc-His(π -Me)-OH; Bachem), which possesses the imidazole group methylated at the π position.

Cell culture

PC12 cells, derived from a transplantable rat pheochromocytoma (ATCC No. CRL 1721), were cultured in the recommended ATCC medium. Primary cultures of hippocampal and cortical neurons from P18-P21 rat brains were grown in neurobasal medium/B27 (GIBCO, Carlsbad, CA). For neuron preparation, pregnant rats were anesthetized and killed to extract the fetuses. For pain alleviation, the animals were anesthetized following recommendations in the 2000 Report of the American Veterinary Medical Association Panel on Euthanasia. Brains from the fetuses were dissected out and neuronal cell cultures were prepared as in a previously described protocol (9).

Cell viability assays

The percentage of cells protected from $A\beta$ -induced cell death by various treatments was evaluated by means of a colorimetric XTT assay (Cell Proliferation Kit II, Roche, Mannheim, Germany). The cytotoxicity was also directly measured by the release of lactate dehydrogenase (LDH) from the cytosol into the media (Cytotoxicity Detection Kit (LDH), Roche).

RESULTS

Effect of Ni²⁺ and imidazole on the A β ion channel electrical activity

Based on the theoretical model proposed by Durell et al. (25), the predicted pore region of the A β channel is made of the hydrophilic structure composed of residues 1-16. The structure formed by a radial polymer of four to six $A\beta$ subunits predicts that rings of His residues surround and form the path for the ions passing through the pore. The results from our previous work (36,37) show that the common feature of the active A β channel blocker peptides is that they all possess in their sequence the two vicinal His residues that have been modeled as lining the entry to the pore. It is therefore possible that compounds of known His coordinating capacity will interact with His in the mouth of the pore. This interaction will block the entrance to the A β pore, and consequently affect the flow of current through the A β channel. We studied various compounds of known His coordinating capacity for their interaction with $A\beta$ channels incorporated into planar lipid bilayers. No membrane potential difference was applied to the lipid bilayers, to avoid any disturbing effect the membrane potential might have on the channel activity and the conformations of the channel. The experiments depicted in Fig. 1 A show that when incorporated into a lipid bilayer, the A β channel operates between multiple conductance levels. One channel is responsible for the multiple conductance levels and each channel incorporation may show a different pattern of conductance transitions, as has been described elsewhere (14,35). The insert in Fig. 1 A shows that discrete jumps of current to different levels (1.46, 2.92, 5.03, and 10.17 pA) can be observed throughout. The addition of Ni²⁺, which coordinates with imidazole with high affinity, and imidazole, which has a preference to have an interplanar interaction with other aromatic residues, blocks the current activity of $A\beta$ channels incorporated in lipid bilayers. After the ionic current through the incorporated A β channels appeared stable for several minutes, either Ni²⁺ or imidazole was added to the experimental chamber. The current records in the figure display 8 s of activity from the A β channels, maintained at zero membrane potential, before (control) and several seconds after the channels were exposed to either Ni^{2+} (Fig. 1 A) or imidazole (Fig. 1 C). The channel activity slowly reduced to an undetectable level, suggesting a full block of the channel. The blockage by both Ni²⁺ (not shown) and imidazole (bottom record of Fig. 1 C) was irreversible since the channels remained fully blocked after the chamber was washed. Amplitude histograms of the current events during same time intervals of channel activity before and at different times after the addition of the test compounds were elaborated and are displayed in the right panels of Fig. 1, B and D. The histograms show that both Ni^{2+} and imidazole are effective at quickly abolishing the number of observations of the most frequent peak current values (0.8, 3.2, 5.2, 7.5, and 10 pA).

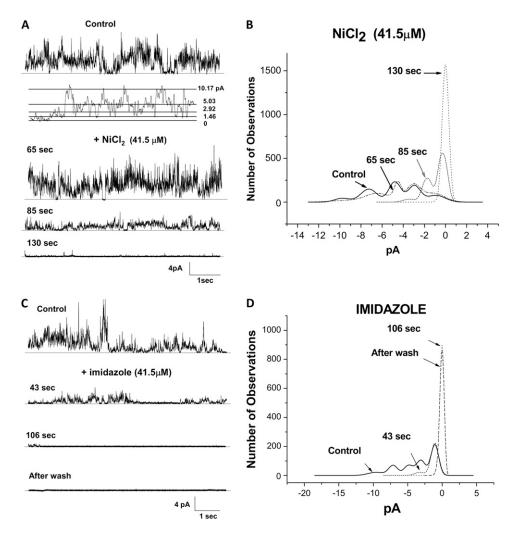


FIGURE 1 Nickel and imidazole irreversibly block the membrane-bound A β ion channel. (A and C) Electrical activity across a lipid bilayer with incorporated $A\beta$ channels, recorded before (control) and in the presence of either Ni^{2+} (A) or imidazole (C). The membrane electrical potential in the experiment was maintained at zero level. B and D display the current amplitude histograms of the channel activity from the current traces in A and C. Five main current peaks of 0.8, 3.2, 5.2, 7.5, and 10 pA characterize the A β channel activity. Both compounds quickly abolish the most frequent current peaks.

Effect of His on the $A\beta$ ion channel electrical activity

We have previously shown that His residues are essential components in the sequence of active $A\beta$ ion channel blocker peptides (36,37). His is an aromatic amino acid that contains a heteroaromatic imidazole ring available for interaction. To study the contribution of the imidazole ring in the His interaction with the $A\beta$ ion channel, we investigated the channel blocking efficiency of a modified, end-capped His (NAHIS01) and compared it with the blocking efficiency of the unmodified, amino and carboxyl ends-free His. The end-capped His, with the carboxyl and amine group amidated and acetylated, respectively, would leave the imidazole side chain as the sole group available for interaction with other reactive groups in the $A\beta$ channel.

The current activity from $A\beta$ channels illustrated in Fig. 2 indicates that both unmodified His and modified His NAHIS01 affect the peaks of ionic current from an $A\beta$ channel incorporated in a lipid bilayer. The current records in Fig. 2, *A* and *C*, and the current amplitude histograms in the Fig. 2, *B* and *D*, show that unmodified His mildly reduces the $A\beta$

channel activity and is relatively ineffective at producing an irreversible full block of the $A\beta$ channels. The peaks of ionic current corresponding to larger channel conductance (5.8 and 4.2 pA) are blocked in a concentration-dependent manner, but the smaller current peaks remain unaffected at the highest concentration of His. As shown in Fig. 2, *C* and *D*, although the modified His NAHIS01 failed as a full channel blocker, it more effectively reduced the $A\beta$ channel activity at lower concentrations.

The A β channel blocking efficiency of His-related compounds increases with the number of imidazole side chains

In the previous experiments we showed that NAHIS01, in which the imidazole is the sole group available for interaction, is very efficient at blocking the $A\beta$ ion channel. Here we studied the ability of two His-related compounds, NAHIS02 and NAHIS04, to block $A\beta$ channels. These compounds possess two and four imidazole side chains, respectively. The current records and current amplitude histograms from $A\beta$

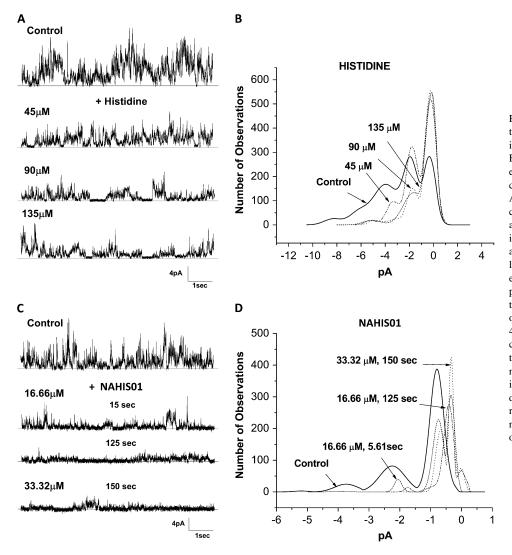


FIGURE 2 Efficacy of His to block the A β channel is improved by capping its amino and carboxyl ends. (A and C) Electrical activity from two lipid bilayers with incorporated A β channels, recorded before (control) and while the A β channel was exposed to various concentrations of unmodified, amino and carboxyl ends-free His, and a modified His, NAHIS01, with the carboxyl and amine groups amidated and acetylated. The A β channels were exposed to each concentration of the blocker compounds for 2-min periods. Increases in the concentration of the unmodified His only affect the current peaks of 5.8 and 4.2 pA, corresponding to larger channel conductance. NAHIS01 is more effective at reducing all current peaks at much lower concentrations than unmodified His. B and D display the amplitude distribution histograms of the ionic current peaks recorded while the $A\beta$ channel was exposed to each concentration of His and NAHIS01, respectively.

channels, before and after exposure to the two His-related compounds, are illustrated in Fig. 3. During the current recording the bilayer was maintained at zero membrane potential. The current records in Fig. 3, A and C, show that both compounds efficiently block the multiple conductances exhibited by the $A\beta$ channels. However, one of the distinguished observations is an apparent higher affinity by NAHIS04 for the channel that stayed fully blocked after the chamber was washed. The lower current record in Fig. 3 C shows that NAHIS04, which possesses four imidazole side chains, is the most effective at irreversibly blocking the A β channel activity. The lower current record in Fig. 3 A shows some recovery of channel activity after NAHIS02 was washed off. No effect of membrane potential in the blocking capacity of these compounds was observed (not shown). We previously reported that the level of the membrane potential has no effect on the ability of similar His-related compounds to block the A β channel (35). The current amplitude histograms in Fig. 3, B and D, show a gradual reduction in the number of current peaks after the A β channels are exposed to either one of these His-related compounds. The results reveal that the number of imidazole side chains in the blocker compounds has a substantial influence on their efficiency for blocking the A β channel conductance. The experiment illustrated in Fig. 3 *C* shows that NAHIS04 (16.66 μ M), which possesses four imidazole side chains, reduced the A β channel activity to occasional channel openings just 20 s after addition. By contrast, at the same concentration, NAHIS02, which possesses two imidazole groups, takes several minutes to produce similar levels of channel blockage.

The contribution of the imidazole side chains in the Hisrelated compounds to block the A β channel activity was verified with the experiments shown in Fig. 4. Here the effectiveness of the His-coordinating and His-related compounds was studied in terms of the time required to block the A β channel currents. The top current record in Fig. 4 A displays the time course of the A β channel activity before and after the addition of the channel blocker NAHIS02. The bottom current record displays the channel activity from a similar experiment in which increasing concentrations of

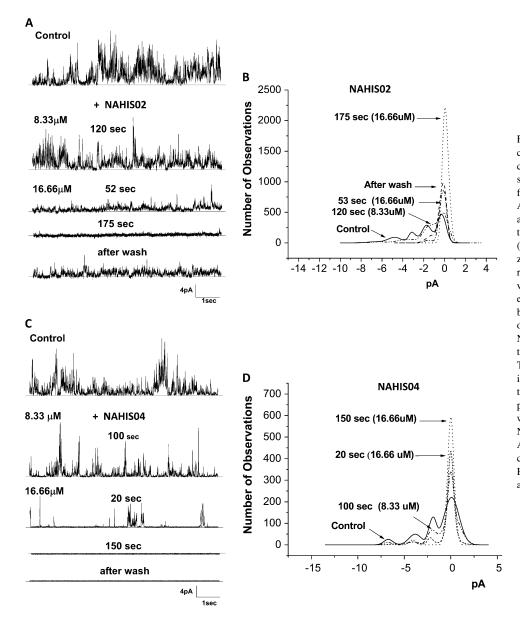


FIGURE 3 $A\beta$ channel blocking efficiency of His-related compounds increases with the number of imidazole side chains. (A and C) Electrical activity from two lipid bilayers with incorporated A β channels, recorded before (control) and while the A β channels were exposed to two compounds that possess two (NAHIS02) and four (NAHIS04) imidazole side chains, respectively. The current records show the channel activity at various times after the channels were exposed to two concentrations of the blocker compounds. After full blockage of the A β channel, the blocking action of NAHIS02 reversed when the experimental chamber was washed of the blocker. The blockage by NAHIS04 was irreversible. B and D display the amplitude distribution histograms of the ionic current peaks recorded while the $A\beta$ channel was exposed to various concentrations of NAHIS02 and NAHIS04, respectively, At a concentration of 16.66 μ M both compounds removed all current peaks. However, NAHIS04 produced full blockage more quickly.

a modified NAHIS02, NAHIS02-(π -Met), were added. NAHIS02-(π -Met) is a modified NAHIS02 in which the imidazole side chains are methylated. The unmodified NAHIS02, at 16.66 μ M, totally blocks the A β channel activity in <3 min. In contrast, NAHIS02-(π -Met) is unable to block the A β channel activity even at a fourfold higher concentration, indicating that the methylation of the imidazole groups reduces the affinity of NAHIS02 for the A β molecule and consequently the ability to block the A β channel. When A β channels are incorporated into artificial membranes, they form multiconductance systems. This is manifested by frequently fluctuating conductance between specific levels. To comparatively evaluate the blocking strength of the test compounds in terms of the time required to stop the A β channel activity, experiments similar to that displayed in Fig. 4 A were performed with each one of the test compounds. The

results were analyzed by an alternative procedure that quantifies the total ionic current flowing through the A β channel incorporated into the artificial lipid membrane at any given time. For this purpose, we integrated the total ionic current flowing through the membrane and averaged the amount of charge conducted in consecutive time intervals of 8 ms duration. The integration was initiated after the incorporated channel achieved stable activity, and also after the addition of the test compounds. The left panel plot in Fig. 4 B shows that imidazole is extremely efficient at promptly (30 s) blocking the flow of ionic charges, in contrast to His, which was very slow-acting or had almost no channel-blocking capacity. Ni²⁺ ions, as also shown in Fig. 2 A, fully blocked the A β channel within a few seconds (120 s). The plot in the right panel in Fig. 4 B shows that the efficiency for stopping the A β channel activity of the His-related compounds ap-

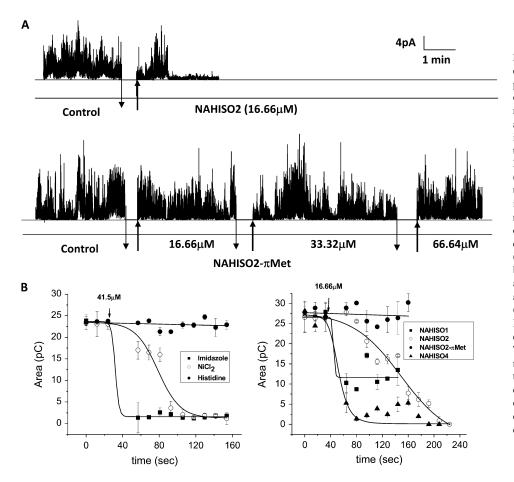


FIGURE 4 Time course of ionic charge conducted by membrane-incorporated A β channels, showing the effect of A β channel blockers. (A) The current records show the time course of the activity of $A\beta$ channels incorporated into planar lipid bilayers before and after the addition of the A β channel blocker NAHIS02 (top record) and NAHIS02- $(\pi$ -Met) (bottom record). The concentration of NAHIS02-(π -Met) was gradually increased as indicated, at the times signaled by the arrows. NAHIS02-(π -Met) did not affect the A β channel activity even at a fourfold higher concentration. (B) The amount of charge conducted by bilayer-incorporated A β channels before and after the addition of imidazole, Ni2and His at concentrations of 41.5 µM (left plot), and NAHIS01, NAHIS02, NAHIS02-(π -Met), and NAHIS04 at a concentration of 16.6 µM (right plot). The ionic current flowing through the membrane was integrated in consecutive time intervals of 8-ms duration. The efficiency to stop $A\beta$ channel activity increases as the number of imidazole side chains in the blocker compounds is increased. pC: picoCoulombs.

pears to increase as the number of imidazole side chains in the His-related compound is increased. Hence NAHIS04, which has four imidazole side chains, reduces the flow of ionic charges through the membrane by 50% in 15 s. This reducing effect is eightfold faster compared to the 125 s it takes for NAHIS02, which has two imidazole side chains, to achieve the same level of reduction. In contrast, NAHIS01, which only has one imidazole side chain, does not fully prevent the flow of ionic charges, and NAHIS02-(π -Met), from which imidazole reactivity is removed by methylation, showed no blocking capacity.

Protection of cells from the A β cytotoxicity increases in efficiency with the number of imidazole groups in the A β channel blocker

We have previously shown that $A\beta$ cytotoxicity can be prevented when cells are incubated in media containing $A\beta$ channel blockers (9,16,37). Since Ni²⁺, imidazole, His, and His-related compounds affect, to different degrees, $A\beta$ channels incorporated into artificial membranes, we examined the ability and relative strength of these compounds to protect cells from $A\beta$ peptide-induced cell death. Cell viability was examined using two different assays. These were the colorimetric XTT assay that quantifies metabolically active cells, and an assay that measures the release of LDH from the cytosol into the media, which evaluates the cell membrane integrity. The panels in Fig. 5 show the viability of PC12 cells (A and B), and cortical (C and D) and hippocampal neurons (E and F) observed after 3 days of incubation in the presence of A β peptide (5 μ M) and A β peptide (5 μ M) plus Ni²⁺, imidazole, or His. The left panels (A, C, and E) show the results from the XTT assay expressed as a percentage of cytotoxicity, and the right panels (B, D, and F)show the results from the measurements of LDH released from cells into the media expressed as a percentage of protection of cells. The performance of imidazole, Ni²⁺, and His to maintain cell viability and protect cells from $A\beta$ peptideinduced cell death strongly corresponds to what one would expect from their effects displayed on the ionic current flowing through the A β channels incorporated into artificial membranes. Imidazole and Ni²⁺ were found to fully protect the cells against A β cytotoxicity, but Ni²⁺ showed a celldependent toxicity at high concentrations. On the other hand, His, which mildly reduced the A β channel activity, performed poorly in protecting cells from A β toxicity. The panels in Fig. 6 show the viability of PC12 cells (A and B), and cortical (C and D) and hippocampal neurons (E and F) observed after 3 days of incubation in the presence of $A\beta$ peptide (5 μ M) and A β peptide (5 μ M) plus the His-related

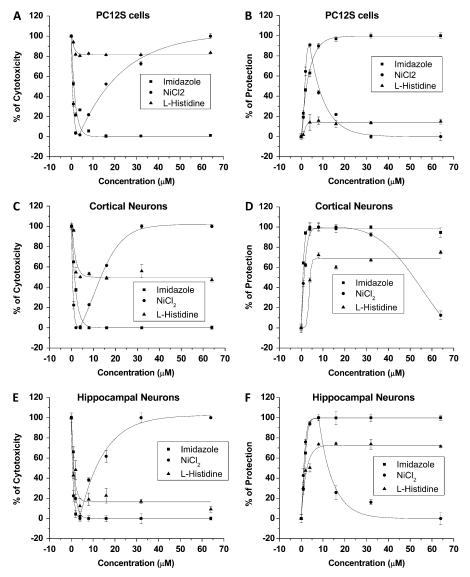


FIGURE 5 Protection of cells from $A\beta$ cytotoxicity by compounds that coordinate and associate with A β channels. Viability of PC12S cells (A and B), and cortical (C and D) and hippocampal neurons (E and F) observed after 3 days of incubation in the presence of $A\beta$ peptide (5 μ M) and A β peptide (5 μ M) plus Ni²⁺, imidazole, or His. The results from the XTT assay are expressed as a percentage of cytotoxicity (A, C, and E). The LDH released from cells into the media is expressed as a percentage of protection of cells (B, D, and F). Imidazole and Ni²⁺ fully protect cells against A β cytotoxicity, but Ni²⁺ showed a cell-dependent toxicity at high concentrations. His performed poorly in protecting cells from $A\beta$ toxicity.

compounds NAHIS01, NAHIS02, NAHIS02-(π -Met), or NAHIS04. The left panels (A, C, and E) show the results from the XTT assay as a percentage of cytotoxicity, and the right panels (B, D, and F) show the results from the measurements of LDH release into the media as a percentage of protection of cells. The performance of the His-related compounds to protect cells from the A β peptide-induced cell death also strongly corresponded to what one would expect from their effects displayed on the ionic current flowing through the A β channels incorporated into artificial membranes. NAHIS02 and NAHIS04 fully protected the three different types of cells against A β . NAHIS01, which did not fully block A β channels in artificial membranes (see Fig. 3), showed partial protection. The level of protection for all these compounds correlated with the number of imidazole side chains in these compounds. For instance, the concentration required for 50% protection (half maximal effective concentration, EC_{50}) for the three different types of cells for the compounds NAHIS02

and NAHIS04 was always below 1 μ M. The EC₅₀ for imidazole was between 2 and 3 μ M (see Fig. 6). The compound NAHIS02-(π -Met) was very inefficient at protecting against A β peptide cytotoxicity. The mild protection observed with this compound was cell-type dependent and always plateaued at much higher concentrations than that observed with the unmethylated form. Full methylation of NAHIS02-(π -Met) is never achieved during the synthesis process. Therefore, a mild protection is not totally unexpected to be observed in the results. However, the loss of protection strongly suggests that the methylation of NAHIS02 is affecting at the site of the molecule that interacts with the A β channel.

DISCUSSION

The results described in this investigation show that Hiscoordinating and His-related compounds can efficiently block $A\beta$ channels incorporated into artificial membranes,

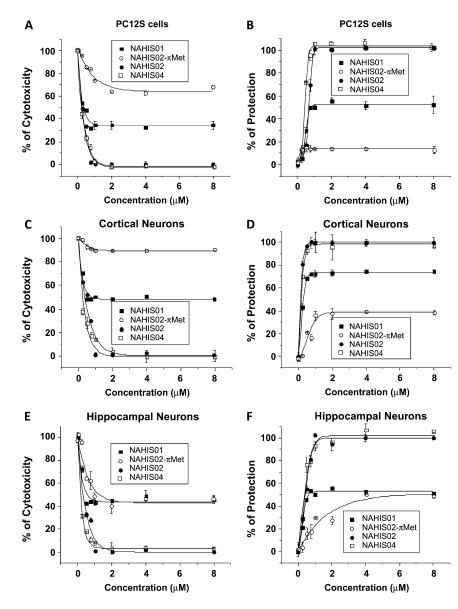


FIGURE 6 Protection of cells from $A\beta$ cytotoxicity by His-related compounds that establish aromatic interaction with $A\beta$ channels. Viability of PC12 cells (A and B), and cortical (C and D) and hippocampal neurons (E and F) observed after 3 days of incubation in the presence of A β peptide (5 μ M) and A β peptide (5 μ M) plus NAHIS01, NAHIS02, NAHIS02-(π -Met), and NAHIS04. The results from the XTT assay are expressed as a percentage of cytotoxicity (A, C, and E). The LDH released from cells into the media is expressed as a percentage of protection of cells (B, D, and F). The level of protection for all these compounds correlated to the number of reactive imidazole side chains. NAHIS02 and NAHIS04 fully protected the cells against A β . NAHIS02-(π -Met) was very inefficient at protecting against A β peptide cytotoxicity.

and can also entirely prevent $A\beta$ cytotoxicity produced by the incorporation of $A\beta$ channels in the cell surface membrane. Therefore, we interpret these data to support the hypothesis that His residues within the $A\beta$ channel sequence are in the pathway of ion flow and contribute to define the ion channel selectivity. Additionally, the data confirm the contribution of the $A\beta$ channel to the cytotoxicity of $A\beta$.

We previously reported that small peptides designed to closely match the sequence of the $A\beta$ peptide that includes His¹³ and His¹⁴ were the most effective at blocking ion currents through $A\beta$ channels (35,36). The results of the investigation presented here strongly suggest that $A\beta$ channel blockage by compounds of known His-coordinating capacity, such as Ni²⁺ and imidazole, occurs by the interaction with the His residues located in the pathway of the ions in the $A\beta$ channel. This conclusion is supported by the enhanced

blocking efficiency observed after increasing the number of imidazole reactive side chains in His-related compounds. Additionally, we observed that perturbing the bonding of the imidazole side chains by selective methylation, which prevents the interplanar interaction of imidazole with the aromatic His residues, results in abolition of A β channel current activity and neurotoxicity.

Binding sites for the A β channel blockers

The highly effective block of the $A\beta$ channels observed in our experiments after the application of either nickel (Ni²⁺) or imidazole is expected if the His in the $A\beta$ subunits of the $A\beta$ channels are the participating residues. Information collected from the Protein Data Bank reveals that among the aromatic residues His can be found in various chemical environments

in protein structures, sometimes behaving as an aromatic residue, or as a metal ligand, and at other times forming salt bridges with acidic groups (38). The interaction between His and Ni^{2+} is so profound that the His tag is globally the most used tag in the preparative purification of proteins. Immobilized metal affinity chromatography, which is used to purify His-tagged proteins, exploits the ability of His to bind chelated transition metal ions, and Ni²⁺ has generally been proven to be the most successful of the metal ions. Competitive interaction is the most common method used to recover the purified protein fractions. Imidazole competitively interacts with immobilized Ni²⁺ ions to reverse the binding of the protein. With the exception of nickel, the compounds that we found in this investigation to be efficient for blocking the A β channels were aromatic residues containing the heteroaromatic imidazole ring. As a heteroaromatic moiety, imidazole can interact with other aromatic and nonpolar groups, since it can exist in the neutral or positively charged form at the physiological pH. Additionally, imidazole can form the most conspicuous hydrogen bonds with polar and charged (both negative and positive) residues (39,40). Depending on the protonation state, imidazole can also be involved in salt bridges with acidic groups. For these reasons, it is expected that effective interactions can be established between the imidazole-containing compounds that block A β channels and the charged His⁶, His¹³, His¹⁴, Glu²², and Lys²⁸ residues in the A β subunits in the A β channel. However, among the charged residues in the A β subunit sequence, His is the only aromatic residue and is the most likely candidate to carry the preferred face-to-face geometric interactions with imidazole (38). Imidazole is a five-membered planar ring that consists of one π electron from the =N-atom, one from each carbon atom, and two from the NH nitrogen. This resonance structure makes imidazole an excellent nucleofile that would be attracted to a full or partial positively charged form of the His in the A β subunits of the A β channels. Therefore, the propensities of imidazole to interact with His residues vastly overcome the propensities to interact with the other charged residues, such as Lys or Glu, in the A β subunit (38,40,41). In the case of the His-related compounds tested for blocking the A β channel, we observed a remarkable increase in blocking efficiency because the blocker compounds possess more available reactive imidazole side chains. In contrast, NAHIS02, which behaved as a highly efficient blocker, not only lost its capacity to block the A β channels, but additionally became ineffective in protecting from $A\beta$ cytotoxicity when the imidazole side chains were perturbed by methylation. A large number of studies have dealt with the subject of aromatic-aromatic interaction in proteins. The vast majority of medicinal agents contain aromatic substituents and their differential recognition by proteins is likely to be facilitated by the noncovalent interactions involving aromatic residues (42). Among the aromatic residues. His has the highest propensity to interact with planar groups and forms a distinct class separate from all other residue types (40,41).

The function of the imidazole side chain in the His interaction with the $A\beta$ ion channel

We observed that the efficacy for blocking the A β channel by the amino acid His is improved when the free ends of His are modified by amidation and acetylation of the carboxyl and amine groups, respectively. This modification, in addition to making the peptide resistant to proteases degradation, reduces the peptide unspecific reactivity. Our interpretation of the improved blocking efficacy of NAHIS01 compared to the ends-free His is that in the unmodified His, in addition to the imidazole side chain, the free ends increase the possibility for His to unspecifically interact with any other reactive residues present in the A β subunits of the A β channel. Therefore, the unspecific interactions with other regions of A β reduce the probability of specific interactions with His that form the ionconducting path of the A β channel. When the His ends are capped and are unable to chemically react, as is the case with NAHIS01, the reactivity of His is restricted to the specific interaction that may be established by its imidazole side chain. Thus, the probability for aromatic interactions between the exceptionally nucleofilic imidazole in the ends-capped His and the imidazole side chain in the participating His in the A β channel will be increased. We believe that the improved efficacy of the ends-capped NAHIS01 compared to ends-free His, as an A β channel blocker, is further proof of the concept that the aromatic interaction between the imidazole side chains contributes to the blocking of A β channels.

Blocking of $A\beta$ channels preserves cells from $A\beta$ cytotoxicity

The efficacy of His coordinating compounds for preserving cells from A β cytotoxicity was tested in three different types of cells. Initially, the preservation of the viability of cells observed in cases where specific $A\beta$ channel blockers were used in combination with $A\beta$ confirmed the participation of the A β channels in A β cytotoxicity (9,36). Those compounds that were observed to fully block $A\beta$ channels incorporated into artificial membranes were also found to be efficient at preserving cell viability during exposure to the toxic $A\beta$ peptide. The results were obtained from three different cell types by the application of two different viability assays. It has been shown that the interaction between A β peptides and the cell surface membrane is followed by activation of an intracellular signaling cascade that leads to the death of cells by apoptosis (43). The finding that A β channel blockers fully prevent cell death corroborates the formation of $A\beta$ ion channels as the initial step in the changes associated with the A β -induced apoptosis (9), and suggests that channel formation is the main mechanism by which $A\beta$ exerts its toxicity. Additionally, the data shown here confirm that exposure of cells to A β results in the formation of oligometric A β aggregates, which assemble the A β channels as they interact with the cell surface membrane. This is the first step in the signaling cascade that leads to cell death. The results from the application of viability assays on the three different types of cells show that the efficacy of the His-related compounds to preserve cells from A β cytotoxicity is correlated to the intensity of the interactions with the His residues in the A β channel. By increasing the number of reactive imidazole side chains in the blocker compound, the capacity to block the active A β channels incorporated into the membranes of the cells is also increased. Our results in cultured cells showed that the blocking efficacy of His-related A β channel blockers was improved to EC₅₀ values below the micromolar levels, as was the case for the blockers NAHISO2 and NAHISO4. This represents a considerable improvement compared to the block-ing effect obtained with previously published A β channel blockers (16,35,36).

Our experimental results and the theoretical models of the $A\beta$ channel

The channel-like annular structure of A β oligomers, as has been observed by electron microscopy (44) and by AFM (23,24), suggests that to form an A β channel, A β subunits assemble in a polymeric transmembrane structure. Based on this structure, a series of theoretical models have been designed to shape the different forms that oligomers of A β assemble to form ion channels when incorporated into a lipid membrane (25-27). The most recently developed models, which illustrate the atomistic structure of the A β channel in annular topology, follow molecular-dynamics simulations based on nuclear magnetic resonance data of the oligomers and use the universal U-shaped (strand-turn-strand) motif for the truncated $A\beta^{17-42}$ (26,27). Although these models do not include the first 16 residues of the A β subunits, which contain the His residues, the negatively charged side chains of Glu²² are arranged circularly in the pore, inducing a cationic ring by the binding of cations such as Mg^{2+} , Ca^{2+} , K^+ , and Zn^{2+} (26,27). Former models were developed based on least-energy calculations, and assume that the channels are formed from an assembly of A β subunits arranged symmetrically around the axis of a pore lined by an amphipathic array of alternated charged residues. In this case the resulting models present rings of His¹³ and His¹⁴ residues of the A β molecule around the entrance of the putative pore (25). If not all the His are protonated, this annular arrangement will model pores with a net negative charge to explain the cation selectivity of the A β channels. The theoretical models for the truncated $A\beta^{17-42}$ include the charged residues Glu^{22} and Lys^{28} of the A β subunit. The old model on energetic grounds for the fulllength $A\beta^{1-40}$ includes in addition the charged residues His⁶, His¹³, and His¹⁴. Although the molecular-dynamics simulations successfully reproduce the channel dimensions, shapes, and subunit organization of the A β channels observed with AFM (23,24), our results from using His-containing A β channel blockers could be better explained by a model that will represent the full-length A β subunit. Such a model must include His residues in addition to the Glu²² and Lys²⁸ charged residues. There are examples of the amino end of the Lys side chain and Glu acid side chain interacting with His residues. However, His prefers to interact in a face-to-face stacked orientation with His rings (38,40). We speculate that this may provide stability to the assembly of A β subunits in a polymeric transmembrane channel structure. In this respect, disruption of this His-His interaction may prevent the functioning of A β channels. Recent experiments showed that perturbing the hydrogen bonding of the imidazole side chains of His residues of A β by selective methylation prevents the formation of His bridges and results in abolition of the A β neurotoxicity (37).

In conclusion, the addition of compounds of known Hiscoordinating capacity, such as Ni²⁺, imidazole, His, and a series of His-related compounds, to A β channel incorporated in membranes show that the His residues located in the N-terminal branch of the A β peptide sequence are essential to form the functional structure that constitutes the selectivity filter and the ion pathway of the A β channel. To achieve the functional characteristic of the A β channels, the theoretical models, constructed to shape the different forms that oligomers of A β assemble to form ion channels, have to include the segment of the A β peptide that contains the charged His residues.

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