

Published in final edited form as:

Biochem Soc Trans. 2005 February ; 33(Pt 1): 80–82. doi:10.1042/BST0330080.

Molecular dynamics and experimental investigation of H₂ and O₂ diffusion in [Fe]-hydrogenase

Jordi Cohen¹, Kwiseon Kim², Matthew Posewitz³, Maria L. Ghirardi², Klaus Schulten¹, Michael Seibert², and Paul King^{2,*}

¹ Beckman Institute, University of Illinois at Urbana-Champaign, Urbana, Illinois 61801, U.S.A

² National Renewable Energy Laboratory, Golden, Colorado 80401, U.S.A

³ Department of Environmental Science and Engineering, Colorado School of Mines, Golden CO 80401, U.S.A

Abstract

The [Fe]-hydrogenase enzymes are highly efficient H₂ catalysts found in ecologically, and phylogenetically diverse microorganisms, including the photosynthetic green alga, *Chlamydomonas reinhardtii*. Although these enzymes can occur in several forms, H₂ catalysis takes place at a unique [FeS] prosthetic group, or H-cluster, located at the active site. Significant to the function of hydrogenases is how the surrounding protein structure facilitates substrate-product transfer, and protects the active site H-cluster from inactivation. To elucidate the role of protein structure in O₂ inactivation of [Fe]-hydrogenases, experimental and theoretical investigations have been performed. Molecular dynamics was used to comparatively investigate O₂ and H₂ diffusion in [Fe]-hydrogenase CpI. The results are compared to initial investigations of H₂ diffusion in [NiFe]-hydrogenase [1]. Our preliminary results suggest that H₂ diffuses more easily and freely than O₂, which is restricted to a small number of allowed pathways to and from the active site. These O₂ pathways are located in the conserved active site domain, shown experimentally to have an essential role in active site protection.

Keywords

hydrogenase; gas diffusion; oxygen sensitivity

Introduction

Hydrogen production in the green alga, *C. reinhardtii*, is a strictly anaerobic process. Coupling to photosynthesis can be achieved under conditions of sulfur deprivation that results in a series of physiological events leading to reduced PSII activity, the respiratory consumption of residual O₂, and establishment of an anaerobic state [2,3]. Induction of hydrogenase gene expression and enzyme synthesis ultimately results in H₂-production. Although metabolic sequestration of O₂ overcomes the severe sensitivity of algal [Fe]-hydrogenase to O₂, the H₂-production efficiency is lower than theoretical maximum [2,3]. In order to develop an efficient, large-scale, photobiological H₂-production system, the O₂ sensitivity of algal [Fe]-hydrogenases needs to be addressed [2].

To whom correspondence should be addressed (paul_king@nrel.gov).

Inactivation of [Fe]-hydrogenase by O₂ involves diffusion of the diatomic gas from solvent to the active site, and subsequent chemical oxidation of the catalytic H-cluster [4]. The initial stages of diffusion involve molecular interactions between O₂ and protein, suggesting that the protein structure influences how O₂ approaches the active site. Previous studies of H₂ diffusion in [NiFe]-hydrogenase suggest that H₂ diffusion in the enzyme is not a random process, but rather tends to occur through a specific pathway, the H₂-channel [1]. To further investigate gas diffusion in hydrogenase, and to elucidate the initial events in the O₂-inactivation process, molecular mechanics methods were used to study O₂ and H₂ diffusion in [Fe]-hydrogenase CpI.

Methods and Results

Expression and O₂-sensitivity determination of [Fe]-hydrogenase

The expression of active the [Fe]-hydrogenases HydA1, HydA and HydAΔN were carried out in *E. coli* as previously reported [5]. Whole cell samples were solubilized, exposed to air, and aliquots assayed for residual H₂ production activity using the reduced methyl viologen assay [5].

Early biochemical studies of O₂-inactivation of [Fe]-hydrogenase suggested that differences exist in the O₂ sensitivities of enzymes isolated from different sources. The partially purified [Fe]-hydrogenase from *C. reinhardtii* exhibited an I₅₀ value of ~1 second when exposed to atmospheric levels of O₂ [6], whereas under similar conditions purified CpI exhibited an I₅₀ of 120–300 seconds [7] (Table 1). Our recombinantly expressed *C. reinhardtii* HydA1 [Fe]-hydrogenase was ~415-fold more sensitive to atmospheric O₂ (21%) than the *C. acetobutylicum* HydA (Table 1). To date the algal enzymes represent the simplest [Fe]-hydrogenases known, consisting solely of an active site domain. In contrast, the homologous bacterial enzymes, CpI and HydA, are more complex, possessing an additional electron-transfer domain [8]. This domain may also contribute to protection of the active site, and may explain the higher observed O₂-tolerance levels of these enzymes. Removal of the electron-transfer domain from HydA (HydAΔN), however, resulted in only a 3-fold decrease in O₂ tolerance (Table 1), and remained ~140-fold more tolerant to O₂ than HydA1. The differences in enzyme sensitivities conferred by the conserved active site domain strongly suggests that the amino acid composition of this domain is critical to protection of the H-cluster from O₂.

Simulation setup and methods

Our CpI model is based on the X-ray crystal structure of [Fe]-hydrogenase CpI from *Clostridium pasteurianum* [8]. Missing H-cluster atoms from the CpI structure are modeled as a di(thiomethyl)amine as in [9]. The partial charges for the rest of H-cluster atoms were based on Ref. [10], with modifications of up to ±0.02e to preserve charge neutrality. The model was then embedded in a water box, resulting in a 57,000-atom system consisting of 9,000 hydrogenase atoms, 16,000 water molecules and 15 sodium ions. The full system was then equilibrated at constant temperature (310K) and pressure (1 atm) for 1ns.

Oxygen and hydrogen gas diffusion in CpI were investigated by all-atom molecular dynamics simulations of the outward diffusion of either O₂ or H₂ from the active site, in a similar manner to previous studies of gas diffusion across proteins [1,11]. A charge-free model for O₂ and H₂ is used. In addition, for reasons of numerical stability, and to facilitate comparison between the two gases, H₂ with equivalent mass to O₂ (hH₂) was used in place of H₂. This way, O₂ and H₂ differ solely by their van der Waals (Lennard-Jones) parameters, molecular bond lengths, and spring constants.

In order to increase the sampling, the method of locally enhanced sampling (LES), otherwise known as the time-dependent Hartree approximation [12], was used to simulate 1,000 mutually invisible, simultaneous copies of the gas molecules (hH_2 or O_2) within a single protein-water system. Gas temperatures were regulated independently at 310K. Each of the 1,000 copies of either O_2 or hH_2 was initially placed at the same location as the H-cluster bound CO molecule in the Cpl structure [4]. For hH_2 , our simulations correspond to the realistic transport of H_2 out of the protein, but for O_2 , we are observing a reverse of the normal O_2 diffusion from the bulk solvent to the active site.

LES simulations were performed with NAMD2 [13] using the CHARMM22 force-field [14]. A constant volume and temperature were used. The system was simulated with periodic boundary conditions, and Particle-Mesh Ewald method.

MD simulation of H_2 and O_2 diffusion in Cpl

During the 1 ns equilibration, we observed a permanent and almost continuous tunnel-shaped cavity (using a 1 Å-radius probe) connecting the active site binding location to the solvent outside the protein at the location of the “ H_2 -channel” detailed for the *Desulfovibrio desulfuricans* [Fe]-hydrogenase reported by Nicolet *et. al.* [9,15].

Our preliminary simulations (6 or 4 independent 2 ns simulations, for either O_2 or H_2 diffusion respectively) suggest that the diffusion pathways of O_2 and hH_2 between the active site of Cpl and the protein-solvent interface are different. More specifically, on one hand, O_2 is observed to transit across the protein barrier through a limited set of precisely defined channels, of which one is displayed in Fig 1a. We have observed O_2 motion across only two such channels, and in only one simulation did O_2 reach the solvent within the simulated time of 2 ns. On the other hand, in all four H_2 simulations, H_2 was observed exiting the protein. A sample of the allowed hH_2 pathways is shown in Fig 2a. In addition, hH_2 is occasionally observed (in 2–15% of all observed exits per simulation) to cut through the bulk protein by unique paths where no channel has been detected. A second important difference between H_2 and O_2 diffusion is that, during LES simulations with 1000 copies of the gas molecule sharing a unique protein trajectory, the different simultaneous hH_2 molecules are observed to spread out in the protein and take a multitude of exits (see Fig 2b–c). Oxygen molecules, in stark contrast, stay clumped together for most or all of the duration of the trajectories (see Fig 1b–c).

Conclusion

In summary we have shown that the diffusion of H_2 and O_2 gases inside of an [Fe]-hydrogenase is governed by the physical properties of both the gas itself and the protein structure. The dynamic process of gas migration dictates that static representations of structures can unsatisfactorily predict pathway selection. In this study, although a majority of H_2 was found to migrate through a series of conserved hydrophobic cavities, or a “ H_2 -channel” [15], H_2 is clearly able to diffuse during the same time period through a number of alternative routes. This is somewhat inconsistent with previous results shown for H_2 diffusion in [NiFe]-hydrogenase [1].

In contrast to H_2 diffusion which displays a degree of randomized behavior, O_2 diffusion is clearly limited to specific regions located within the conserved active site domain. The importance of this domain in O_2 diffusion agrees with the experimental results shown here. The range of sensitivity levels found for algal and bacterial enzymes are also characterized by diversity among the amino acids that comprise the diffusion pathways. From the standpoint of engineering O_2 tolerance in [Fe]-hydrogenases, the sequences and structures of naturally occurring enzymes offer important clues on how to overcome the O_2 -sensitivity of algal [Fe]-hydrogenases.

Acknowledgements

Research at NREL (MLG, MS and PK), and the Colorado School of Mines (MP) is supported by the Division of Energy Biosciences, Office of Science, U.S. Department of Energy, and by the Hydrogen, Fuel Cells and Infrastructure Technologies Program, U.S. Department of Energy.

Research at UIUC (JC and KS) is supported by National Institute of Health grant P41-RR05969, with computational time support by National Resource Allocations Committee grant MCA93S028.

Abbreviations used

CpI	[Fe]-hydrogenase I from <i>Clostridium pasteurianum</i>
HydA	<i>Clostridium acetobutylicum</i> [Fe]-hydrogenase I
hH₂	heavy di-hydrogen
LES	locally enhanced sampling

References

1. Montet Y, Amara P, Volbeda A, Vernede X, Hatchikian EC, Field MJ, Frey M, Fontecilla-Camps JC. *Nat Struc Biol* 1997;4:523–526.1997
2. Ghirardi ML, Zhang L, Lee JW, Flynn T, Seibert M, Greenbaum E, Melis A. *Trends Biotech* 2000;18:506–511.
3. Melis A, Zhang L, Forestier M, Ghirardi ML, Seibert M. *Plant Physiol* 2000;122:127–136. [PubMed: 10631256]
4. Lemon BJ, Peters JW. *Biochem* 1999;38:12969–12973. [PubMed: 10529166]
5. Posewitz MC, King PW, Smolinski SL, Zhang L, Seibert M, Ghirardi ML. *J Biol Chem* 2004;279:25711–25720. [PubMed: 15082711]
6. Erbes DL, King D, Gibbs M. *Plant Physiol* 1979;63:1138–1142. [PubMed: 16660871]
7. Adams MWW, Mortenson LE. *J Biol Chem* 1984;259:7045–7055. [PubMed: 6327705]
8. Peters JW, Lanzilotta WN, Lemon BJ, Seefeldt LC. *Science* 1998;282:1853–1858. [PubMed: 9836629]
9. Nicolet Y, Piras C, Legrand P, Hatchikian CE, Fontecilla-Camps JC. *Struc* 1999;7:12–23.
10. Torres RA, Lovell T, Noodleman L, Case DA. *J Am Chem Soc* 2003;125:1923–1936. [PubMed: 12580620]
11. Hofacker I, Schulten K. *Prot Struct Func Gen* 1997;30:100–107.
12. Elber R, Karplus M. *J Am Chem Soc* 1990;112:9161–9175.
13. Kalé L, Skeel R, Bhandarkar M, Brunner R, Gursoy A, Krawetz N, Phillips J, Shinozaki A, Varadarajan K, Schulten K. *J Comp Phys* 1999;151:283–312.
14. MacKerell AD Jr, et al. *FASEB J* 1992;6:A143–A143. MacKerell AD Jr, et al. *J Phys Chem B* 1998;102:3586–3616.
15. Nicolet Y, Cavazza C, Fontecilla-Camps JC. *J Inorg Biochem* 2002;91:1–8. [PubMed: 12121756]
16. Humphrey W, Dalke A, Schulten K. *J Mol Graphics* 1996;14:33–38.

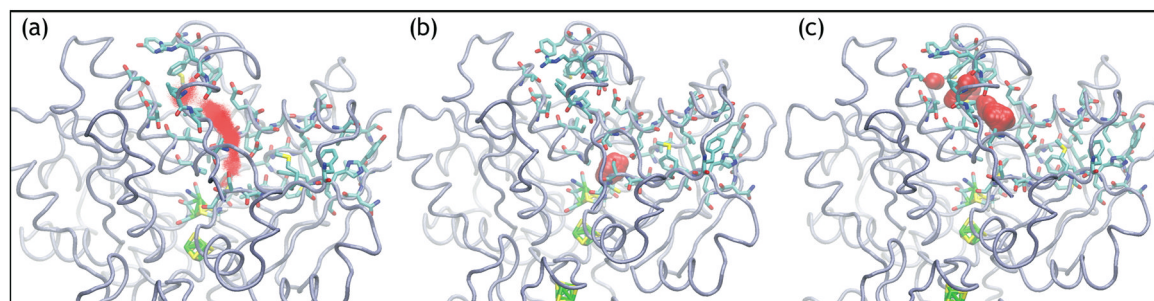


Figure 1.
Caption Sample trajectory of molecular diffusion of 1000 simultaneous and mutually invisible copies of di-oxygen (O_2) from the active site of CpI hydrogenase. Shown is (a) a superposition of all the O_2 positions over the 2.3 ns trajectory and snapshots of the O_2 trajectories after (b) 20 ps and (c) 2.3 ns. Figures were made using VMD [16].

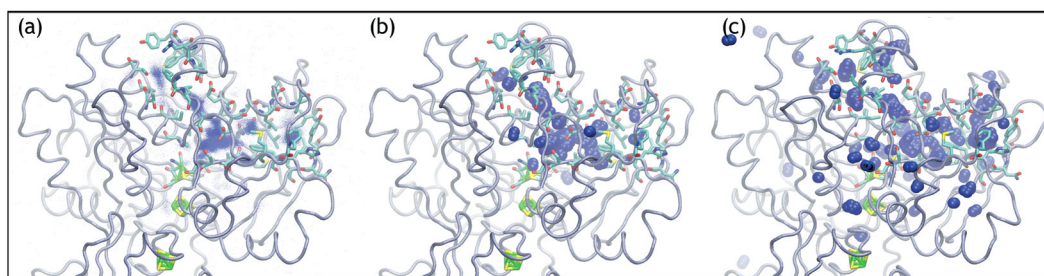


Figure 2.

Caption Same as in Figure 1 for hH₂. Shown is (a) a superposition of all the hH₂ positions of the 2.3 ns trajectory and snapshots of the hH₂ trajectories after (b) 300 ps and (c) 2.2 ns.

Table 1Comparison of algal and bacterial [Fe]-hydrogenase O₂ sensitivities.

[Fe]-hydrogenase	I50 value (seconds) in atmospheric O ₂
<i>C. reinhardtii</i> hydrogenase	<1 ¹
<i>C. reinhardtii</i> HydA1	<1
<i>C. pasteurianum</i> CpI	120 to 300 ²
<i>C. acetobutylicum</i> HydA	415 ± 115
<i>C. acetobutylicum</i> HydAΔN	145 ± 45

¹From [6].²From [7].