

Production of 1-Carbon Units from Glycine Is Extensive in Healthy Men and Women^{1,2}

Yvonne Lamers,³ Jerry Williamson,³ Douglas W. Theriaque,⁴ Jonathan J. Shuster,^{4,5} Lesa R. Gilbert,⁶ Christine Keeling,⁶ Peter W. Stacpoole,^{4,6,7} and Jesse F. Gregory III^{3*}

³Food Science and Human Nutrition Department, Institute of Food and Agricultural Sciences, ⁴General Clinical Research Center, ⁵Division of Biostatistics, Department of Epidemiology and Health Policy Research, ⁶Division of Endocrinology and Metabolism, Department of Medicine, and ⁷Department of Biochemistry and Molecular Biology, College of Medicine, University of Florida, Gainesville, FL 32611-0370

Abstract

Glycine undergoes decarboxylation in the glycine cleavage system (GCS) to yield CO₂, NH₃, and a 1-carbon unit. CO₂ also can be generated from the 2-carbon of glycine by 10-formyltetrahydrofolate-dehydrogenase and, after glycine-to-serine conversion by serine hydroxymethyltransferase, from the tricarboxylic acid cycle. To evaluate the relative fates of glycine carbons in CO₂ generation in healthy volunteers (3 male, 3 female, aged 21–26 y), primed, constant infusions were conducted using 9.26 μmol·h⁻¹·kg⁻¹ of [1,2-¹³C]glycine and 1.87 μmol·h⁻¹·kg⁻¹ of [5,5,5-²H₃]leucine, followed by an infusion protocol using [1-¹³C]glycine as the glycine tracer. The time period between the infusion protocols was >6 mo. In vivo rates of whole-body glycine and leucine flux were nearly identical in protocols with [1,2-¹³C]glycine and [5,5,5-²H₃]leucine and with [1-¹³C]glycine and [5,5,5-²H₃]leucine tracers, which showed high reproducibility between the tracer protocols. Using the [1-¹³C]glycine tracer, breath CO₂ data showed a total rate of glycine decarboxylation of 96 ± 8 μmol·h⁻¹·kg⁻¹, which was 22 ± 3% of whole-body glycine flux. In contrast, infusion of [1,2-¹³C]glycine yielded a glycine-to-CO₂ flux of 146 ± 37 μmol·h⁻¹·kg⁻¹ (P = 0.026). By difference, this implies a rate of CO₂ formation from the glycine 2-carbon of 51 ± 40 μmol·h⁻¹·kg⁻¹, which accounts for ~35% of the total CO₂ generated in glycine catabolism. These findings also indicate that ~65% of the CO₂ generation from glycine occurs by decarboxylation, primarily from the GCS. Further, these results suggest that the GCS is responsible for the entry of 5,10-methylenetetrahydrofolate into 1-carbon metabolism at a very high rate (~96 μmol·h⁻¹·kg⁻¹), which is ~20 times the demand for methyl groups for homocysteine remethylation. J. Nutr. 139: 666–671, 2009.

Introduction

Glycine is a precursor for the synthesis of purines, protein, glutathione, and 1-carbon units in folate-dependent 1-carbon metabolism. Through the mitochondrial glycine cleavage system (GCS),⁸ glycine undergoes decarboxylation to yield CO₂, ammonia, and a 1-carbon unit in the form of 5,10-methylenetetrahydrofolate (methyleneTHF) (1). This 1-carbon unit from glycine is used in reactions including nucleotide synthesis, regeneration of methionine from homocysteine, and methylation of many biological compounds. One-carbon metabolism is linked to the transsul-

furation pathway in which homocysteine undergoes vitamin B-6-dependent catabolism leading to cysteine, whose availability governs the formation of the antioxidant glutathione. Nutritional or genetic conditions that impair 1-carbon metabolism are associated with elevation in plasma homocysteine concentration and increased risk of vascular disease (2).

In addition to glycine cleavage, CO₂ can be formed from glycine carbons through mitochondrial and cytosolic pathways of folate or pyruvate metabolism in healthy individuals (Fig. 1). MethyleneTHF synthesized via GCS can be enzymatically oxidized to 10-formylTHF by methyleneTHF dehydrogenase. 10-FormylTHF dehydrogenase converts 10-formylTHF to tetrahydrofolate (THF) and CO₂, which appears to be a mechanism of achieving a regeneration of THF and regulating 1-carbon pools (3). Glycine and serine are interconvertible through mitochondrial and cytosolic serine hydroxymethyltransferase (SHMT). We previously showed that glycine-to-serine conversion via SHMT accounts for ~41% of whole-body glycine flux, a sum of mitochondrial and cytoplasmic processes (4). Serine synthesis with a GCS-derived 1-carbon unit (Fig. 1, left) accounted for 46% of total serine synthesis rate (4). Serine dehydratase transforms

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⁸ Abbreviations used: AUC, area under the curve; Ep, plateau enrichment; 10-formylTHF, 10-formyltetrahydrofolate; GCRC, General Clinical Research Center; GCS, glycine cleavage system; I, infusion rate; methyleneTHF, 5,10-methylenetetrahydrofolate; PLP, pyridoxal 5'-phosphate; Q, a measurement of flux; SHMT, serine hydroxymethyltransferase; TCA, tricarboxylic acid; THF, tetrahydrofolate; VCO₂, rate of CO₂ production.

* To whom correspondence should be addressed. E-mail: jfgy@ufl.edu.

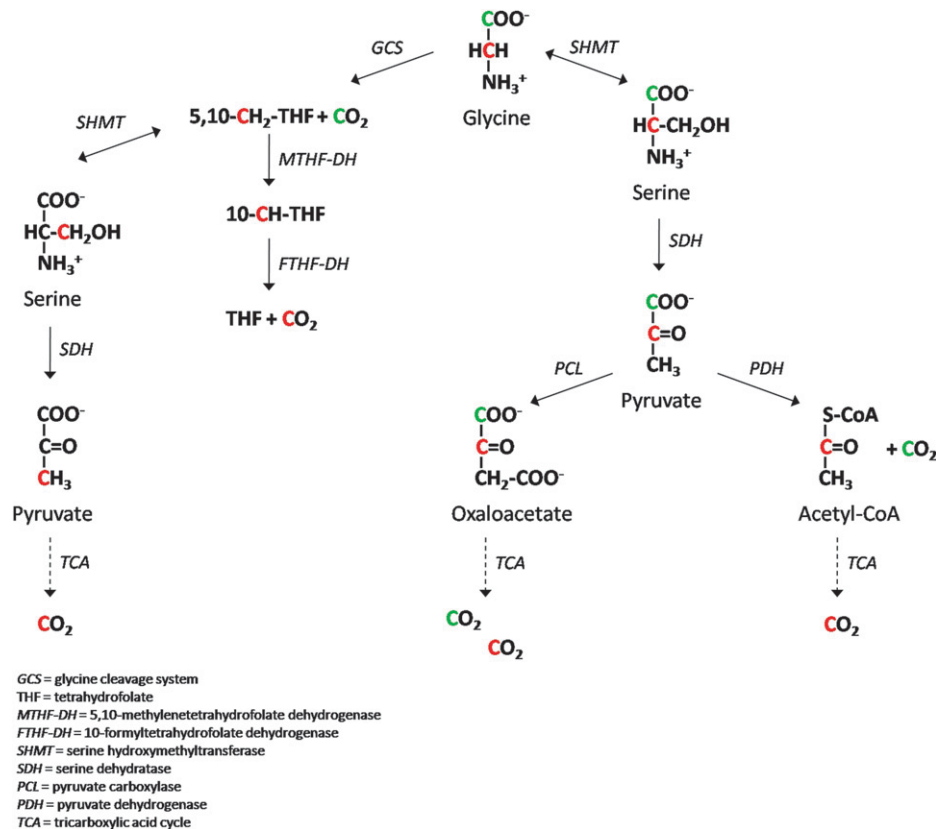


FIGURE 1 Schematic of possible fates of glycine carbons as CO₂.

serine to pyruvate that enters the tricarboxylic acid (TCA) cycle either as oxaloacetate or acetyl-CoA. The carbons of serine, following conversion to pyruvate, can be converted to CO₂ via the pyruvate dehydrogenase reaction and metabolism of the resulting acetyl-CoA and through the TCA cycle. Alternatively, serine carbons (via pyruvate) also can be converted to CO₂ after the formation of oxaloacetate by pyruvate carboxylase, followed by metabolism in the TCA cycle.

With the use of stable isotope tracer protocols, we aimed to quantify postprandial rates of in vivo glycine turnover and the relative fates of glycine carbons in CO₂ generation in healthy, adequately nourished men and women. The tracer protocols included primed, constant infusions of [1,2-¹³C]glycine or [1-¹³C]glycine and [²H₃]leucine with a time period between both protocols of >6 mo. [²H₃]Leucine was included to determine any changes in protein turnover (5,6). The results of whole-body glycine flux after infusion with [1,2-¹³C]glycine and [²H₃]leucine tracers were part of a recently published protocol (7). Measurement of breath ¹³CO₂ enrichment (8) was used for quantification of the fates of glycine carbons with respect to the generation of CO₂. Formation of CO₂ from the [1-¹³C]glycine tracer is hypothesized to occur primarily from glycine decarboxylation through the GCS. The difference between CO₂ formation after infusion with [1,2-¹³C]glycine and CO₂ production from [1-¹³C]glycine tracer infusion would account for the fate of the 2-carbon of glycine as CO₂.

Methods

Materials

[1,2-¹³C]Glycine, [1-¹³C]glycine, L-[5,5,5-²H₃]leucine, and sodium [¹³C]bicarbonate were purchased from Cambridge Isotopes Laborato-

ries. Parenteral solutions were prepared in isotonic saline, filter sterilized and analyzed to ensure lack of pyrogenicity and microbial contamination.

Human participants

Adult male and nonpregnant female participants underwent a physical examination and were screened by standard clinical measures of hematological, renal, hepatic, and thyroid function before each infusion protocol. Medical history, dietary habits, and demographic data were assessed by a questionnaire. All participants met the following inclusion criteria: age between 20 and 40 y, no history of gastrointestinal surgery, abnormal kidney, or thyroid function, or any other chronic disease; no smoking or chronic drug use or alcoholism; no vitamin, amino acid, or protein supplementation; no chronic consumption of a high-protein diet; and a BMI of <28 kg/m². Adequate nutritional status for folate and vitamins B-12 and B-6 was defined as serum folate >7 nmol/L, serum vitamin B-12 >200 pmol/L, and plasma pyridoxal 5'-phosphate (PLP) >30 nmol/L, respectively, and plasma total homocysteine concentration <12 μmol/L. All participants gave written informed consent. The University of Florida Institutional Review Board and the General Clinical Research Center (GCRC) Scientific Advisory Committee approved this protocol.

Dietary treatment

All meals were prepared by the Bionutrition Unit of the GCRC. Participants consumed nutritionally adequate meals with standardized composition for 2 d prior to the infusion day to minimize dietary variation immediately prior to the study.

Analytical methods

Screening measurements. The analytical procedures used for data collection are as described previously (4). Serum folate and vitamin B-12 were analyzed with the use of a commercial chemiluminescence-based assay (Elecsys, Roche Diagnostics). The plasma PLP concentration was measured as the semicarbazone-derivative by reverse-phase HPLC with

fluorescence detection (9). After incubation with the reducing agent tris(2-carboxyethyl)phosphine before protein precipitation, the plasma total homocysteine concentration was measured as the ammonium 7-fluorobenzo-2-oxa-1,3-diazole-4-sulfonate derivative by reverse-phase HPLC with fluorescence detection (10).

GC-MS analysis of amino acid isotopic enrichment. Plasma free amino acids were isolated, derivatized, and analyzed as previously described (6). The N-heptafluorobutyramide-n-propyl ester derivatives were dried, dissolved in ethyl acetate, and stored at -20°C until analysis. Isotopic enrichment was determined by negative chemical ionization GC-MS with the use of a Thermo-Finnigan DSQ GC-MS and a 30-m poly (5% diphenyl and 95% dimethylsiloxane) fused silica capillary column (Equity 5, Supelco). The relative abundance of specific ions was determined by selected-ion monitoring at the following mass:charge ratios: glycine (293–295) and leucine (349–352). Isotopic enrichments are expressed as molar ratios of labeled to nonlabeled isotopomers after correction for the natural abundance of stable isotopes essentially as performed by Storch et al. (11).

Breath CO_2 . For determination of the isotopic enrichment of breath $^{13}\text{CO}_2$, samples were collected in Exetainer tubes provided by Metabolic Solutions and shipped to Metabolic Solutions for isotope ratio-MS analysis. Total CO_2 production rate (VCO_2) was determined with the use of a metabolic cart (TrueMax 2400; ParvoMedics). Measurements were taken at 30-s intervals for ~ 5 min until 4 consecutive time points differed by no more than ± 0.01 L/min.

Infusion protocol

The infusion protocol has been previously reported (4) and was identical for both interventions except for the glycine tracer used. The time period between both protocols was >6 mo. Participants were admitted to the GCRC on the evening before the infusion protocol and consumed no food and drinks, except water, between 2100 and initiation of the infusion. On the morning of the infusion, an angiocatheter was inserted in the antecubital vein of each arm; 1 for the tracer infusion and 1 for blood collection. Fasting blood samples were taken 2 h before infusion (at ~ 0700) for measurement of folate and vitamin B-12 and B-6 status and plasma total homocysteine concentrations. Infusions were initiated at ~ 0900 with a 5 min, ~ 20 mL priming dose that delivered $9.26 \mu\text{mol/kg}$ $[1,2-^{13}\text{C}]$ glycine or $9.26 \mu\text{mol/kg}$ $[1-^{13}\text{C}]$ glycine and $1.87 \mu\text{mol/kg}$ $[5,5,5-^2\text{H}_3]$ leucine, respectively, along with a simultaneous priming dose of $2.15 \mu\text{mol/kg}$ of $\text{NaH}^{13}\text{CO}_3$. The 9-h constant infusion followed immediately after the priming dose and delivered ~ 20 mL/h infusion solution that contained $9.26 \mu\text{mol/kg}$ $[1,2-^{13}\text{C}]$ glycine or $9.26 \mu\text{mol/kg}$ $[1-^{13}\text{C}]$ glycine and $1.87 \mu\text{mol/kg}$ $[5,5,5-^2\text{H}_3]$ leucine.

Blood samples were taken at 0, 0.5, 1, 1.5, 2, 2.5, 3, 4, 5, 6, 7.5, and 9 h of the infusion. These samples were placed immediately on ice and were centrifuged within 15 min after the blood draw at $1500 \times g$ (10 min at 4°C). Plasma was stored in microcentrifuge tubes at -80°C .

To measure $^{13}\text{CO}_2$ production, breath samples were collected into Exetainer tubes at times 0, 1, 2, 3, 4, and 6 h of infusion in both protocols and additionally at 5 h of infusion in the protocol using $[1,2-^{13}\text{C}]$ glycine and at 7.5 and 9 h of infusion in the protocol using $[1-^{13}\text{C}]$ glycine. Measurements of the total VCO_2 were conducted at 0, 2, 4, 6, and 8 h of infusion. Starting 2 h before infusion after drawing of the fasting blood sample, the participants received hourly a nutritive formula to maintain a fed state (11). This formula provided an intake of energy and a balanced pattern of amino acids at a rate based on requirements of $0.8 \text{ g protein}\cdot\text{kg}^{-1}\cdot\text{d}^{-1}$ [30 and $31 \text{ kcal}\cdot\text{kg}^{-1}\cdot\text{d}^{-1}$ (126 and $130 \text{ kJ}\cdot\text{kg}^{-1}\cdot\text{d}^{-1}$) for women and men, respectively], which equals an hourly protein dose of $\sim 0.03 \text{ g/kg}$ with 1.25 and $1.30 \text{ kcal}\cdot\text{kg}^{-1}\cdot\text{d}^{-1}$ (5.23 and $5.44 \text{ kJ}\cdot\text{kg}^{-1}\cdot\text{d}^{-1}$) for women and men, respectively.

Kinetic principles and analysis

As previously described (4), the combined use of $[1,2-^{13}\text{C}]$ glycine or $[1-^{13}\text{C}]$ glycine and $[^2\text{H}_3]$ leucine permitted the determination of the kinetics of whole-body glycine turnover, the rate of glycine decarboxylation, and its role as a source of 1-carbon units in 1-carbon metabolism. $[^2\text{H}_3]$ Leucine is included to evaluate any effects on protein

turnover (5,6). During the decarboxylation and catabolism of a glycine tracer via the GCS, the original glycine ^{13}C -labeled carbon at the 2-carbon position is transferred to THF to yield methyleneTHF. The major focus of the study reported here was to determine the quantitative aspects of the different fates of glycine carbons as CO_2 through GCS and folate cycle (Fig. 1).

The plateau enrichment (E_p) of all infused amino acid tracers was calculated as the mean of the isotopic enrichment for the ~ 1.5 – 9 -h time points for the infused $[^{13}\text{C}_2]$ glycine, $[^{13}\text{C}_1]$ glycine, and $[^2\text{H}_3]$ leucine tracers. The E_p of breath $^{13}\text{CO}_2$ was approximated by fitting enrichment data to a single exponential curve defined by the equation

$$E_t = E_f(1 - e^{-kt}). \quad (1)$$

In this equation, E_t is the enrichment at time t (h), whereas E_f and k are the enrichment at infinity (i.e. E_p) and rate constant (h^{-1}) from the fitted curve, respectively (12). Data were fit to a single regression exponential equation with the use of SigmaPlot 2002 (for Windows version 8.02).

Steady-state kinetics of amino acid tracers were calculated using standard equations (13), including correction for overestimation of intracellular enrichment from plasma enrichment data (11–13), as discussed below. The flux of an amino acid is the rate of appearance of that amino acid from endogenous production (de novo synthesis and protein breakdown), absorption, and the tracer infusion, and is calculated from the E_p of the corresponding amino acid tracer. Specifically, the flux (Q) of leucine (Q_{Leu}) in the plasma pool is calculated as:

$$Q_{\text{Leu}} = I_{\text{Leu}} \cdot [(E_{\text{Leu}})/(E_{p,\text{Leu}}) - 1], \quad (2)$$

where I_{Leu} is the $[^2\text{H}_3]$ leucine infusion rate, E_{Leu} is the enrichment of the $[^2\text{H}_3]$ leucine tracer and $E_{p,\text{Leu}}$ is the E_p of $[^2\text{H}_3]$ leucine in plasma. The E_p of plasma leucine was not corrected for overestimation of intracellular enrichment, consistent with previous studies using leucine flux as a relative indicator of protein turnover (5,14,15).

Glycine flux (Q_{Gly}) was calculated from plasma $[1,2-^{13}\text{C}]$ glycine enrichment after correcting for the overestimation of the intracellular $[1,2-^{13}\text{C}]$ glycine enrichment that occurs when the plasma E_p of the glycine tracer is used. This prediction of intracellular $[1,2-^{13}\text{C}]$ glycine enrichment ($E_{p',\text{Gly}}$) was accomplished by multiplying the observed plasma $[1,2-^{13}\text{C}]$ glycine enrichment by a correction factor of 0.4 derived from previous glycine tracer infusion studies in humans (16,17). The same calculation was done for the $[1-^{13}\text{C}]$ glycine tracer.

$$Q_{\text{Gly}} = I_{\text{Gly}} \cdot [(E_{\text{Gly}})/(E_{p',\text{Gly}}) - 1] \quad (3)$$

The rate of production of $^{13}\text{CO}_2$ provided a direct measurement of the whole-body flux of the decarboxylation of the glycine tracer. This was measured in standard fashion as in amino acid oxidation studies (11,18). In this procedure, the rate of $^{13}\text{CO}_2$ release ($V^{13}\text{CO}_2$; in units of $\mu\text{mol}\cdot\text{h}^{-1}\cdot\text{kg}^{-1}$ body weight) and the rate of glycine catabolism via decarboxylation (C_{Gly} ; $\mu\text{mol}\cdot\text{h}^{-1}\cdot\text{kg}^{-1}$ body weight) were calculated as follows:

$$V^{13}\text{CO}_2 = E^{13}\text{CO}_2 \cdot \text{VCO}_2 / 0.81 \cdot 1/W, \quad (4)$$

where $E^{13}\text{CO}_2$ is breath CO_2 enrichment at plateau, VCO_2 is the rate of total CO_2 production, 0.81 is the assumed fraction of CO_2 release from the body pool of bicarbonate, and W is body weight (18).

$$C_{\text{Gly}} = V^{13}\text{CO}_2 \cdot [(1/E_{p',\text{Gly}}) - (1/E_{i,\text{Gly}})], \quad (5)$$

where $E_{p',\text{Gly}}$ is the E_p of plasma $[^{13}\text{C}_2]$ glycine corrected for intracellular overestimation and $E_{i,\text{Gly}}$ is the enrichment of the infused glycine tracer.

The fraction of Q_{Gly} occurring via glycine decarboxylation was calculated as

$$F_{\text{GCS}} = C_{\text{Gly}}/Q_{\text{Gly}}. \quad (6)$$

Statistical analysis

Descriptive statistics are presented as means and SD. The 2 primary outcome measurements were the yield of breath $^{13}\text{CO}_2$ enrichment after $[1,2-^{13}\text{C}]$ glycine and $[1-^{13}\text{C}]$ glycine infusion. The primary dependent variable for each of these outcomes was constructed as the area under the

curve (AUC) based on the trapezoidal rule for times 1, 2, 3, 4, and 6 h postinfusion, the times where both protocols collected the same information, dividing this result by 6 to obtain a mean value for the 6 h. Differences between infusion trials differing in the glycine tracer employed were tested via 2-sided paired *t* tests with differences considered significant at *P* < 0.05. All other statistical comparisons, including time-specific comparisons of the 2 primary outcome measures, were conducted in the same manner. The target sample size was 10 participants, the number of participants from the [1,2-¹³C₂]glycine protocol potentially available. The actual number of participants that participated in the [1-¹³C]glycine infusion protocol, approximately one-half a year after the first, was 6.

Results

All participants (3 male, 3 female, age 21–26 y) had a BMI < 25 kg/m². Their serum folate, vitamin B-12, and plasma PLP concentrations were in the normal range and their plasma total homocysteine (<12 μmol/L) indicated normal 1-carbon metabolism and adequate nutritional status at each infusion protocol (Table 1).

In vivo rates of whole-body glycine and leucine turnover were nearly identical in both tracer protocols (Table 2), which indicates high reproducibility and comparability between the protocols using differently labeled glycine tracers. The plasma enrichment of the infused amino acid tracers is illustrated (Fig. 2).

Infusion of [1,2-¹³C]glycine tracer yielded a significantly higher glycine-to-CO₂ flux than the [1-¹³C]glycine tracer (Table 2). Glycine-to-CO₂ flux after infusion with [1-¹³C]glycine indicated that GCS accounted for 22 ± 3% of whole-body glycine flux. The difference of glycine-to-CO₂ flux from [1,2-¹³C]glycine vs. [1-¹³C]glycine tracer implies a rate of CO₂ formation from the glycine 2-carbon of 51 ± 40 μmol·h⁻¹·kg⁻¹, which accounts for ~35% of the CO₂ generated in glycine catabolism. The AUC for breath ¹³CO₂ enrichment (Fig. 3) and the rate of ¹³CO₂ production (V¹³CO₂) was significantly higher from the [1,2-¹³C]glycine infusion compared with that of [1-¹³C]glycine tracer (Table 3).

Discussion

Glycine is catabolized through the mitochondrial GCS and its 2-carbon is accepted by THF entering the folate cycle as methyleneTHF. The metabolic importance of this glycine cleavage reaction is illustrated by severe clinical symptoms, e.g. lethargy, apnea, seizures, and cognitive impairment in patients with loss-of-function mutations for genes encoding the components of the GCS. In these inborn errors, glycine accumulates to pathologic concentrations in cerebrospinal fluid and plasma

TABLE 1 Variables related to 1-carbon metabolism and body weight of healthy men and women prior to each infusion protocol¹

	Protocol [1,2- ¹³ C]glycine	Protocol [1- ¹³ C]glycine
Plasma PLP, nmol/L	55 ± 9	63 ± 15
Serum folate, nmol/L	29 ± 8	28 ± 3
Serum vitamin B-12, pmol/L	395 ± 74	363 ± 80
Plasma tHcy, ² μmol/L	7.9 ± 1.5	7.5 ± 1.5
Body weight, kg	69 ± 11	70 ± 13

¹ Values are mean ± SD, *n* = 6.

² tHcy, total homocysteine.

TABLE 2 Plateau enrichment of stable isotope-labeled amino acids and their metabolic products and the corresponding flux values after infusion with [1,2-¹³C]glycine or [1-¹³C]glycine and [²H₃]leucine¹

	Protocol [1,2- ¹³ C]glycine		Protocol [1- ¹³ C]glycine	
	Ep, mol % excess	Whole-body Q, μmol·h ⁻¹ ·kg ⁻¹	Ep, mol % excess	Whole-body Q, μmol·h ⁻¹ ·kg ⁻¹
Infused AA		Q _{infused AA}		Q _{infused AA}
[² H ₃]leucine	1.69 ± 0.15	109 ± 11	1.86 ± 0.11	99 ± 6
[¹³ C ₂]glycine	5.54 ± 1.26	423 ± 79	5.36 ± 0.78	430 ± 62
CO ₂ generation		Q _{Gly→CO₂}		Q _{Gly→CO₂}
[¹³ CO ₂]	0.046 ± 0.006	146 ± 37	0.033 ± 0.003*	96 ± 8 [‡]

¹ Values are mean ± SD, *n* = 6. Symbols indicate different from [1,2-¹³C]glycine protocol: **P* = 0.0003; [‡]*P* = 0.026 (paired *t* test). AA, amino acids.

leading to the conditions of nonketotic hyperglycinemia or glycine encephalopathy (19,20).

In the present study, 2 stable isotope tracer protocols were designed to quantify the rate of CO₂ generation from infused glycine to evaluate more extensively the quantitative role of glycine cleavage in folate-dependent 1-carbon metabolism and whole-body glycine metabolism. The use of both single- and double-labeled glycine tracers allowed a selective assessment of the relative fates of glycine carbons in CO₂ generation. Glycine is decarboxylated in the mitochondrial GCS, which accounted for 22% of the whole-body glycine flux. In addition to glycine decarboxylation via GCS, the 1-carbon of glycine can be released as CO₂ by pyruvate dehydrogenase and in the TCA cycle (Fig. 1). These reactions might have contributed partially to ¹³CO₂ formation after labeled glycine infusion. However, the activity of serine dehydratase forming pyruvate is lower in humans than other mammalian species (21) and, thus, pyruvate oxidation would contribute minimally to CO₂ formation from glycine. Further, because glycine is the direct substrate for GCS, we propose that CO₂ formation through the GCS would be substantially faster and quantitatively greater than from pyruvate dehydrogenase or the TCA cycle.

The results of this study also indicate that the formation of CO₂ from the glycine 2-carbon accounts for ~35% of the total CO₂ generation in glycine metabolism. The 2-carbon of glycine can be released as CO₂ by 10-formylTHF dehydrogenase and in

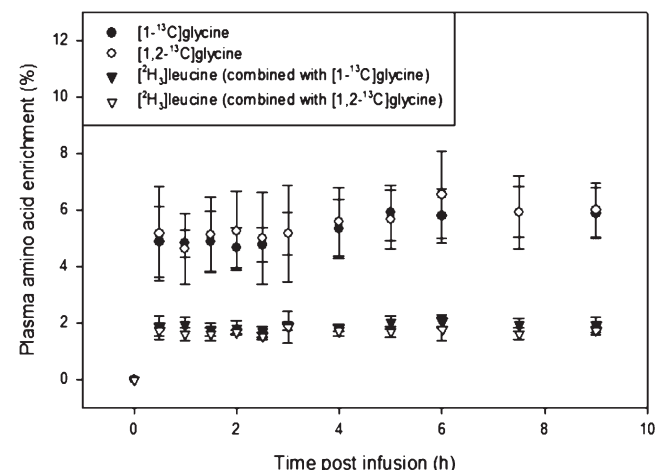


FIGURE 2 Plasma enrichment of infused amino acids in healthy men and women during primed, constant infusion with 9.26 μmol·kg⁻¹·h⁻¹ [1,2-¹³C]glycine or 9.26 μmol·kg⁻¹·h⁻¹ [1-¹³C]glycine and 1.87 μmol·kg⁻¹·h⁻¹ [²H₃]leucine. Values are means ± SEM, *n* = 6.

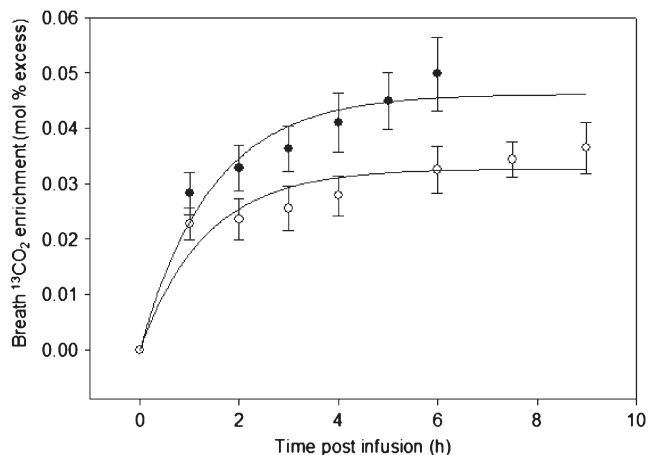


FIGURE 3 Breath $^{13}\text{CO}_2$ enrichment in healthy men and women during primed, constant infusion with $[1,2-^{13}\text{C}]$ glycine (closed circles) or $[1-^{13}\text{C}]$ glycine (open circles). Values are means \pm SEM, $n = 6$.

the TCA cycle. The TCA cycle is further downstream in the metabolism where the isotopic enrichment of oxaloacetate and other TCA cycle constituents become increasingly diluted. We therefore speculate that the $^{13}\text{CO}_2$ enrichment of TCA cycle components would be very low. If this assumption is true, it appears that $^{13}\text{CO}_2$ formation from the glycine 2-carbon occurs primarily from the conversion of 10-formylTHF to THF and CO_2 by 10-formylTHF dehydrogenase. Recent mathematical modeling predicted that 10-formylTHF dehydrogenase would contribute to CO_2 generation at a rate of $\sim 30\%$ of the rate of GCS in hepatic metabolism (22). This study appears to represent, to our knowledge, the first in vivo evaluation of 10-formylTHF dehydrogenase flux with a rate of $\sim 51 \mu\text{mol}\cdot\text{h}^{-1}\cdot\text{kg}^{-1}$. The rate of CO_2 formation from the glycine 2-carbon greatly varied among participants. 10-FormylTHF dehydrogenase was reported to be downregulated in tumor tissues (23), but the basis of its regulation in healthy adults and factors responsible for interindividual variability is unclear. The physiologic significance of variability in whole-body 10-formylTHF dehydrogenase flux needs further investigations.

The mean glycine cleavage rate of $96 \mu\text{mol}\cdot\text{h}^{-1}\cdot\text{kg}^{-1}$ implies that GCS produces methyleneTHF at a very high rate for use in folate-dependent 1-carbon metabolism. To put this into the context of overall demand for 1-carbon units, total homocysteine remethylation flux employs 5-methylTHF (derived from methyleneTHF) as the 1-carbon donor at a rate of $2\text{--}8 \mu\text{mol}\cdot\text{h}^{-1}\cdot\text{kg}^{-1}$ (14). These results provide clear evidence that the GCS produces 1-carbon units as methyleneTHF at a rate of

~ 20 times that needed for homocysteine remethylation and, thus, for methylation demand. In patients with nonketotic hyperglycinemia, the elevated homocysteine concentration in cerebrospinal fluid might indicate a lack of GCS-derived 1-carbon units for homocysteine remethylation (24). The high rate of methyleneTHF production from glycine cleavage suggests that the GCS might supply methyleneTHF in support of a very high flux of purine and thymidylate synthesis. However, this hypothesis appears to be incorrect. In the $[1,2-^{13}\text{C}]$ glycine infusion protocol, the incorporation of GCS-derived methyleneTHF in serine synthesis via SHMT occurred at a rate of $\sim 92 \mu\text{mol}\cdot\text{kg}^{-1}\cdot\text{h}^{-1}$ (7). In comparison, the $[1-^{13}\text{C}]$ glycine tracer indicated that the GCS reaction occurred at a whole-body rate of $96 \mu\text{mol}\cdot\text{kg}^{-1}\cdot\text{h}^{-1}$. We conclude that nearly all GCS-derived 1-carbon units are consumed in serine synthesis while a much smaller percentage enters all other aspects of 1-carbon metabolism, including nucleoside synthesis, homocysteine remethylation, and S-adenosylmethionine-dependent methylation reactions (7). We acknowledge that this conclusion is based on whole-body flux measurements. The contribution of glycine to nucleoside synthesis and methylation reactions is likely to be cell-type or tissue-specific and cannot be determined with the present stable isotope tracer protocol. The quantitative role of glycine in providing 1-carbon units for nucleotide synthesis and the quantitative role of serine synthesis in gluconeogenesis need further investigation.

The recently published protocol using $[1,2-^{13}\text{C}]$ glycine tracer infusion (7) was designed for quantification of whole-body glycine flux, rate of glycine cleavage, serine synthesis rate, and contribution of glycine in nucleoside synthesis (7). We recognized the potential for overestimation of GCS flux because of the smaller but not inconsequential generation of CO_2 from the glycine 2-carbon. The present study provides further insight regarding the interpretation of the quantitative aspects of glycine catabolism and the role of glycine in 1-carbon metabolism. We conclude from this kinetic analysis that the total rate of glycine decarboxylation is $\sim 30\text{--}35\%$ less than the estimate based on total $^{13}\text{CO}_2$ generation from $[1,2-^{13}\text{C}]$ glycine tracer infusion protocol reported here and in our previous article (7).

These tracer protocols were performed in participants in the fed state to accelerate pathways by substrate loads (11). Whole-body glycine turnover was $\sim 430 \mu\text{mol}\cdot\text{h}^{-1}\cdot\text{kg}^{-1}$ in these healthy men and women. This is consistent with a glycine flux of $458 \mu\text{mol}\cdot\text{h}^{-1}\cdot\text{kg}^{-1}$, as determined by Gersovitz (25) also in fed participants. Robert et al. (18) measured a glycine flux of $240 \mu\text{mol}\cdot\text{h}^{-1}\cdot\text{kg}^{-1}$ in young, healthy men who were fasting. Other influencing factors on amino acid flux and their metabolic pathways are hormonal variations, e.g. glucagon levels. In rat hepatocytes, glucagon activated GCS by a phosphorylation-mediated cell-signaling mechanism (26,27).

In conclusion, the results of this study provide evidence for and quantitative insight into the major role of glycine and the mitochondrial GCS as a source of methyleneTHF 1-carbon units for supplying serine synthesis and folate-dependent 1-carbon metabolism. We have recently reported that mild vitamin B-6 restriction has little effect on GCS flux in healthy adults (7). In view of its quantitative importance in human metabolism, other aspects of physiological, genetic, and nutritional regulation of the GCS in health and disease should be evaluated.

TABLE 3 AUC of breath $^{13}\text{CO}_2$ enrichment and $V^{13}\text{CO}_2$ after infusion with $[1,2-^{13}\text{C}]$ glycine or $[1-^{13}\text{C}]$ glycine and $[^2\text{H}_3]$ leucine¹

	Protocol $[1,2-^{13}\text{C}]$ glycine	Protocol $[1-^{13}\text{C}]$ glycine	Difference
AUC of $^{13}\text{CO}_2$ enrichment, mol % excess over 6 h	0.0347 ± 0.0041	0.0244 ± 0.0032	$0.0103 \pm 0.0034^*$
AUC of $V^{13}\text{CO}_2$, $\mu\text{mol}\cdot\text{h}^{-1}\cdot\text{kg}^{-1}$ over 6 h	2.46 ± 0.52	1.62 ± 0.23	$0.84 \pm 0.65^\ddagger$
$V\text{CO}_2$, $\mu\text{mol}\cdot\text{h}^{-1}\cdot\text{kg}^{-1}$	5383 ± 460	5772 ± 678	

¹ Values are mean \pm SD, $n = 6$. Symbols indicate difference: * $P < 0.001$; $^\ddagger P = 0.025$ (2-sided paired t test).

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