

# NIH Public Access

Author Manuscript

J Org Chem. Author manuscript; available in PMC 2010 February 6

Published in final edited form as:

J Org Chem. 2009 February 6; 74(3): 1156–1162. doi:10.1021/jo8022582.

# Isolation, Structure Elucidation and Synthesis of Eudistomides A and B, Lipopeptides from a Fijian Ascidian Eudistoma sp.

Emily L. Whitson<sup>†</sup>, Anokha S. Ratnayake<sup>†,‡</sup>, Tim S. Bugni<sup>†</sup>, Mary Kay Harper<sup>†</sup>, and Chris M. Ireland<sup>†,\*</sup>

<sup>†</sup>Department of Medicinal Chemistry, University of Utah, Salt Lake City, Utah 84112

# Abstract

Eudistomides A (1) and B (2), two new cyclic peptides, were isolated from a Fijian ascidian *Eudistoma* sp. These five-residue cystine-linked cyclic peptides are flanked by a C-terminal methyl ester and a 12-oxo- or 12-hydroxy-tetradecanoyl moiety. The complete structures of the eudistomides were determined using a combination of spectroscopic and chemical methods. Chiral HPLC analysis revealed that all five amino acid residues in 1 and 2 had the L-configuration. Total synthesis of eudistomides A (1) and B (2) confirmed the proposed structures. Enantioselective lipase-catalyzed hydrolysis of a mixture of C-35 acetoxy epimers indicated a 35R absolute configuration for 2.

# Introduction

Ascidians are known to be a rich source of complex and structurally unique peptides—peptides such as ulithiacyclamides,  $^{1-3}$  patellamides,  $^{3-7}$  lissoclinamides,  $^{7-9}$  and didemnins.  $^{10-15}$  To date, however, peptides have not been reported from the genus *Eudistoma*. Previous investigations of *Eudistoma* have yielded eudistomins,  $^{16-20}$  eudistomidins,  $^{21-24}$  iejimalides,  $^{25,26}$  and many alkaloids,  $^{27-37}$  several of which are brominated.  $^{33-37}$  As part of the continuing search for structurally unique secondary metabolites from marine invertebrates, a detailed exploration of the morphologically distinct Fijian ascidian *Eudistoma* sp. was undertaken. The isolation, structure elucidation, and synthesis of two new *Eudistoma* derived lipopeptides, eudistomides A (1) and B (2), are described herein.

# **Results and Discussion**



The specimen (FJ04-12-071) of *Eudistoma* sp. was lyophilized, ground to a fine powder and exhaustively extracted with MeOH. The crude extract was subjected to an EtOAc/H<sub>2</sub>O partition, and the EtOAc soluble material was separated on HP20SS resin using a step gradient of H<sub>2</sub>O to acetone (10% steps, 11 fractions). The sixth (50/50 acetone/H<sub>2</sub>O) and seventh (60/40 acetone/H<sub>2</sub>O) fractions were combined and purified using several rounds of reversed-phase HPLC, resulting in the isolation of eudistomides A (1) and B (2).

<sup>\*</sup>To whom correspondence should be addressed. Tel: (801) 581-8305. Fax: (801) 585-6208. E-mail: E-mail: cireland@pharm.utah.edu. \*Present address: Wyeth Research, Pearl River, New York 10965

Eudistomide A (1) showed an  $[M+Na]^+$  ion at m/z 790.3897 in the HRESIMS, consistent with the molecular formula  $C_{37}H_{61}N_5O_8S_2$  (calcd for  $C_{37}H_{61}N_5O_8S_2Na$ , 790.3859;  $\Delta$  +4.8 ppm), and required ten degrees of unsaturation. Initial evaluation of the <sup>1</sup>H and <sup>13</sup>C spectra suggested that **1** was a peptide (Table 1). The peptide nature of the molecule was further supported by the presence of three exchangeable NH signals at  $\delta_{H}$  6.08 (d, 10.5), 7.20 (d, 9.3), and 9.42 (d, 8.1) in the <sup>1</sup>H NMR spectrum. The <sup>13</sup>C NMR spectrum showed six carbonyl signals (amides and/or esters) at  $\delta_C$  169.4, 170.1, 170.9, 171.5, 171.8, and 173.2. The presence of an ester group in 1 was confirmed by analysis of the IR spectrum which displayed a characteristic absorbance band at  $v_{max} = 1740 \text{ cm}^{-1}$ . The <sup>1</sup>H NMR and HMBC data corroborated the identity of the methyl ester; a methyl resonance at  $\delta_H$  3.73 (H-4) showed an HMBC correlation to a carbonyl at  $\delta_C$  170.9 (C-1). A carbon resonance at  $\delta$  212.1 (C-35) indicated the presence of a ketone in 1. The planar structure of eudistomide A (1) was assigned after extensive 1D and 2D NMR studies. Analysis of the 1D TOCSY, COSY, HSQC, and HMBC data established the presence of five amino acid residues; two Pro, two Cys, and one Leu (Table 1). The location of the C-35 ketone ( $\delta_{\rm C}$  212.1) was established based on the presence of a triplet methyl  $\delta_{\rm H}$  1.02 (H-37), a quartet methylene  $\delta_H$  2.38 (H-36), and a triplet methylene  $\delta_H$  2.36 (H-34) and their corresponding HMBC correlations to the ketone resonance. The combined NMR data was useful in assigning several methylene resonances to the aliphatic chain. However, the precise chain length could not be determined due to the extensive overlap of resonances within the methylene envelop. Fortunately, the molecular formula supported the proposed fourteen carbon lipid chain; the 12-oxo-tetradecanoyl portion of **1**. Nine of the ten degrees of unsaturation were accounted for with the seven carbonyls and the two Pro, which suggested that a cystine-linked ring was the remaining degree of unsaturation.

The amino acid sequence of eudistomide A(1) was determined using a combination of HMBC, ROESY, and MS/MS data. The location of the tetradecanoyl moiety and one of the Cys residues (Cys-2) in **1** was established based on HMBC correlations from the methylenes at  $\delta_{\rm H}$  2.10 and 1.55 (H-25 and H-26, respectively), the Cys-2  $\alpha$ -proton (H-22,  $\delta_H$  4.61), and the Cys-2 NH  $(\delta_{\rm H} 6.08)$  to a carbonyl at  $\delta_{\rm C}$  171.8 (C-24). The aforementioned methyl ester was assigned to the C-terminal Cys residue (Cys-1) based on a HMBC correlation from the Cys-1  $\alpha$ -proton (H-2,  $\delta_H$  4.86) to the methyl ester carbonyl (C-1,  $\delta_C$  170.9). The Leu residue was placed adjacent to Cys-1 (Cys-OMe) based on HMBC correlations from the Cys-1 NH ( $\delta_{\rm H}$  7.19), and the Leu  $\beta$ -protons (H-7a, H-7b,  $\delta_{\rm H}$  1.83, 1.48) to the carbonyl at  $\delta_{\rm C}$  171.5 (C-5). Pro-1 was positioned adjacent to the Leu based on HMBC correlations from the Leu NH ( $\delta_{\rm H}$  9.42), and the Pro-1  $\beta$ -protons (H-13a, H-13b,  $\delta_H$  2.35, 2.10) to the carbonyl at  $\delta_C$  173.2 (C-11). Due to their overlap, HMBC correlations from the Leu  $\alpha$ -proton (H-6,  $\delta_{H}$  4.49) and the Pro-1  $\alpha$ -proton (H-12,  $\delta_{\rm H}$  4.48) could not be utilized. HMBC correlations from the Pro-1  $\delta$ -proton (H-15,  $\delta_{\rm H}$ 3.56) and the Pro-2  $\beta$ -protons (H-18a, H-18b,  $\delta_{\rm H}$  2.28, 1.83) to the carbonyl at  $\delta_{\rm C}$  170.1 (C-16) placed Pro-2 adjacent to Pro-1. No HMBC correlations were observed between Pro-2 and Cys-2. ROESY data supported the proposed sequence of the peptide (Figure 1). Although the ROE correlations from  $\delta_H$  4.49 (H-6) and  $\delta_H$  4.48 (H-12) supported the peptide sequence, the overlap of the proton signals created ambiguity. Therefore, MS/MS studies were conducted to further support the proposed peptide sequence (Table 2). In order to simplify sequence analysis of 1, the linear peptide desthioeudistomide B (3), generated from a Raney Ni desulfurization of 1 (Cys  $\rightarrow$  Ala, ketone  $\rightarrow$  hydroxyl), was analyzed by MS/MS. Compound 3 showed fragment ions consistent with Leu-Ala-OMe, Pro-Leu-Ala-OMe, Pro-Pro-Leu-Ala-OMe, Pro-Leu, and Pro-Pro-Leu partial sequences. Eudistomide A (1) showed a similar fragmentation pattern to **3** (Pro-Leu and Pro-Pro-Leu were identical), with an addition of 64 Da for the Cys containing fragments, consistent with the addition of 2 sulfurs. Based on the HMBC, ROESY and MS/ MS data, the peptide sequence of eudistomide A (1) was assigned as cyclized Cys-Pro-Pro-Leu-Cys-OMe, N-acylated with a 12-oxo-tetradecanoyl fragment.



It is well documented that *cis-trans* conformational differences in proline amide bonds can be distinguished in solution by the chemical shift differential of the  $\beta$ - and  $\gamma$ -carbons ( $\Delta\delta_{\beta\gamma}$ ).<sup>38</sup> Typically, a *cis*-Pro has a chemical shift differential greater than 8 ppm and a *trans*-Pro has a differential less than 6 ppm. In eudistomide A (1), the shift differentials are 11.5 and 8.5 for Pro-1 and Pro-2, respectively, indicating that both Pro residues are in the *cis* conformation.

The absolute configuration of each amino acid in eudistomide A (1) was determined by chiral HPLC analysis of the acid hydrolyzate. Compound 1 was desulfurized using Raney Ni in refluxing MeOH<sup>39</sup> to generate desthioeudistomide B (3), which was then hydrolyzed with aqueous HCl. Chiral HPLC analysis of the hydrolyzate and authentic standards established the presence of two L-Pro, two L-Ala and one L-Leu in eudistomide A (1). Since the linear peptide 3 only contained L-Ala, the two cysteines in 1 were also assigned L configuration.

Eudistomide B (2) was assigned the molecular formula  $C_{37}H_{63}N_5O_8S_2$  based on HRESIMS analysis of the protonated molecular ion  $[M+H]^+$  at m/z 770.4160. When compared with the formula for 1, eudistomide B (2) showed an additional two H's, which suggested either the disulfide bond or the ketone was reduced in 2. The NMR spectra of eudistomides A (1) and B (2) are virtually identical, and indicated that both 1 and 2 contained the same five amino acids (Table 1). One obvious difference between the two compounds was the lack of a ketone in 2. Analysis of COSY and HMBC data showed that a secondary alcohol ( $\delta_C$  73.1) was present at C-35 in eudistomide B (2). MS/MS fragmentation of 2 confirmed that the disulfide bond was still present (Table 2). Since HMBC, ROESY and MS/MS data for 2 were comparable to 1, the peptide sequence of eudistomide A (2) was confirmed as cyclized Cys-Pro-Pro-Leu-Cys-OMe, *N*-acylated with a 12-hydroxy-tetradecanoyl fragment. The  $\Delta\delta_{\beta\gamma}$  for eudistomide B (2) was identical to 1, indicating that 2 also contained two *cis*-Pro residues. Since eudistomides A (1) and B (2) are biosynthetically related and all chemical shifts for the peptide portions are virtually indistinguishable, the amino acids in 2 were also assigned the L configuration.

A total synthesis of eudistomide A (1) was undertaken to confirm the proposed structure. To synthesize the 12-keto-tetradecanoic acid, the commercially available starting material dodecane-1,12-diol (4) was protected with TBSCl to give the monoprotected 12-(tertbutyldimethylsilyloxy)dodecan-1-ol (5) (Scheme 1). Catalytic TEMPO oxidation of compound 5, followed by Grignard addition of the ethyl group generated 14-(tertbutyldimethylsilyloxy)tetradecan-3-ol (6). Compound 6 was deprotected with TBAF to yield tetradecane-1,12-diol (7), and subsequently oxidized to form 12-oxo-tetradecanoic acid (8). The cyclized pentapeptide, Cys-Pro-Pro-Leu-Cys (9), was synthesized by the University of Utah Peptide Synthesis facility using standard solid phase peptide synthesis procedures. Coupling of the acid (8) to the cyclic pentapeptide (9), and simultaneous methylation was achieved using EDC, HOBt, DIPEA, and MeOH to yield eudistomide A (1) as the major product. Eudistomide acid (10) and the cyclic pentapeptide methyl ester (11) were identified as side products (Scheme 2). Interestingly, when attempting to form the methyl ester (11) of precursor peptide (9) with diazomethane prior to coupling, the primary amine of Cys-2 was concurrently methylated to form the quaternary amine. Thionyl chloride in MeOH proved to be the best means of introducing the methyl ester to this peptide, and can be used prior to coupling without undesired amine methylation. The side products of the coupling reaction (10 and 11) were converted to eudistomide A (1) in the following manner: compound 10 was

methylated using thionyl chloride in MeOH,  $^{40}$  and **11** was coupled to the acid (**8**) using EDC, HOBt, and DIPEA. The final synthetic product (**1**) was identical to the natural product obtained from *Eudistoma* sp. in HPLC retention time, MS/MS fragmentation, MS and NMR spectra, which confirmed the proposed structure for eudistomide A (**1**).

Determining the configuration of the C-35 alcohol in eudistomide B (2) was necessary to complete the structure. Several attempts were made to esterify the C-35 alcohol of 2 with MPA to determine the configuration by the modified Mosher method; however, none proved successful. Optimization of reaction conditions for MPA-derivitization was carried out using 6 from the synthesis, but the conditions were not favorable for 2, and no significant amounts of product were detected by LC-MS. To preserve the limited supply of eudistomide B (2), a scheme was proposed using synthetic eudistomide A (1) in an attempt to resolve the configuration of the C-35 alcohol in 2 (Scheme 3). Compound 1 was reduced using sodium borohydride in MeOH at -78 °C,<sup>41</sup> and the disulfide was re-oxidized using DMSO in acetic  $acid^{42}$  to give a mixture of C-35 alcohol epimers (2, 12). The HSQC data showed that the <sup>1</sup>H and <sup>13</sup>C chemical shifts of the C-35 hydroxy methines of the two epimers were very different; one showed a proton resonance at  $\delta_H$  3.52 attached to a carbon at  $\delta_C$  73.2, and the other showed a proton resonance at  $\delta_H$  4.84 attached to a carbon at  $\delta_C$  74.7. Efforts to separate the alcohol epimers (2, 12) were not successful. Literature precedents using similar substrates 43,44confirm that lipase B from Candida antarctica shows selectivity for R-secondary acetates as substrates, with % ee being  $\geq$  99% when conversion is less than 50%. Accordingly, lipasecatalyzed hydrolysis of the C-35 acetoxy epimers would primarily generate the *R*-alcohol. Therefore, the alcohol epimers were treated with acetic anhydride in pyridine to generate the C-35 acetoxy derivatives (13, 14). Lipase B from C. antarctica was added to the acetoxy epimers (13, 14) at 37 °C and the products were analyzed by HSQC. Only one alcohol signal appeared in the HSQC ( $\delta_H$  3.52,  $\delta_C$  73.2), and was assigned the *R*-configuration. The C-35 alcohol in eudistomide B (2) ( $\delta_{\rm H}$  3.49,  $\delta_{\rm C}$  73.1) was also assigned the *R*-configuration on the basis of identical chemical shifts to the lipase-derived *R*-isomer of 1.

Eudistomides A (1) and B (2) are interesting cyclic lipopeptides that contain several rare structural motifs. The acyl chain present in 1 and 2 has never been reported from a marine organism.<sup>45</sup> However, similar acyl groups such as unsubstituted tetradecanoyl,<sup>46–48</sup> 13-methyltetradecanoyl,<sup>49,50</sup> 2,3-hydroxy-tetradecanoyl,<sup>51</sup> 3-hydroxy-13-methyltetradecanoyl, <sup>52</sup> and 7-methoxytetradec-4-enoyl<sup>53–55</sup> chains have been reported. Interestingly, the microsporins<sup>56</sup> contain a 2-amino-8-oxodecanoic acid or a 2-amino-8-hydroxydecanoic acid that is similar to the ketone and hydroxyl derivatives seen in 1 and 2. And while marine organisms are the source of several peptides that contain cystine moieties such as the ulithiacyclamides, <sup>1–3</sup> thiocoralines, <sup>57</sup> microcionamides, <sup>39</sup> and neopetrosiamides, <sup>58</sup> eudistomides A (1) and B (2) are the first ascidian-derived peptides cyclized solely by a disulfide bridge.<sup>45</sup>

# **Experimental Section**

#### **Biological Material**

*Eudistoma* sp., sample FJ04-12-071, was collected by SCUBA near Namena Barrier Reef, Fiji Islands (17° 06.884' S, 179° 03.805' E); a voucher specimen is maintained at the University of Utah. This thick dark gray encrusting *Eudistoma* sp. was found in large patches (approximately 0.5 m) and expressed copious amounts of clear mucous.

#### **Extraction and Isolation**

The *Eudistoma* sp. specimen was lyophilized and ground to a fine powder. The powder was exhaustively extracted with MeOH to yield 15.4 g of crude extract. The crude extract was

subjected to an EtOAc/H<sub>2</sub>O partition to generate 1.7 g of the organic fraction. The EtOAc soluble material was separated on HP20SS resin using a gradient of H<sub>2</sub>O to acetone in 10% steps, and a final wash of 100% acetone, to yield 11 fractions. The sixth (50/50 acetone/H<sub>2</sub>O) and seventh fractions (60/40 acetone/H<sub>2</sub>O) were combined (34.5 mg) and chromatographed by HPLC using a Phenomenex Luna C<sub>18</sub> column (250 × 10 mm) employing a gradient of 50% CH<sub>3</sub>CN/H<sub>2</sub>O to 100% CH<sub>3</sub>CN at 2.5 mL/min over 30 min to yield 4 fractions, 29A–29D. Fraction 29C was further chromatographed using a Phenomenex Luna C<sub>18</sub> (150 × 4.6 mm) column using a gradient of 70% CH<sub>3</sub>CN/H<sub>2</sub>O to 85% CH<sub>3</sub>CN/H<sub>2</sub>O at 1.0 mL/min over 15 min to yield 2 fractions, 35A–35B. Fraction 35A was re-purified using a Phenomenex Luna C<sub>18</sub> (250 × 10 mm) column using a gradient of 50% CH<sub>3</sub>CN/50% H<sub>2</sub>O (+0.1% AcOH) to 98% CH<sub>3</sub>CN/2% H<sub>2</sub>O (+ 0.1% AcOH) at 4.5 mL/min over 20 min to yield eudistomide A (1, 0.2 mg) eluting at 12.9 min and eudistomide B (2, 0.35 mg) eluting at 11.2 min. Reisolation from associated fractions and extraction of the remaining crude material provided an additional 3.2 mg of 1.

**Eudistomide A (1)**—amorphous, white solid  $[\alpha]^{20}_D$ —<sub>1.3</sub> (*c* 0.2, MeOH); UV (MeOH)  $\lambda_{max}$  (log  $\varepsilon$ ) 206 (3.77) nm; IR (film, NaCl)  $\nu_{max}$  3284 (br), 2921, 2852, 1743, 1708, 1662, 1641, 1549, 1253, 702 cm<sup>-1</sup>; <sup>1</sup>H NMR and <sup>13</sup>C NMR data, see Table 1; HRESIMS *m*/*z* 790.3897 [M+Na]<sup>+</sup> (calcd for C<sub>37</sub>H<sub>61</sub>N<sub>5</sub>O<sub>8</sub>S<sub>2</sub>Na, 790.3859).

**Eudistomide B (2)**—amorphous, white solid  $[\alpha]^{20}_{D}$  –4.0 (*c* 0.03, MeOH); UV (MeOH)  $\lambda_{max}$  (log  $\epsilon$ ) 206 (3.61) nm; IR (film, NaCl)  $v_{max}$  3222 (br), 2921, 2852, 1745, 1662, 1641, 1549, 1259, 725 cm<sup>-1</sup>; <sup>1</sup>H NMR and <sup>13</sup>C NMR data, see Table 1; HRESIMS *m/z* 770.4160 [M +H]<sup>+</sup> (calcd for C<sub>37</sub>H<sub>64</sub>N<sub>5</sub>O<sub>8</sub>S<sub>2</sub>, 770.4191).

#### **Desulfurization of Eudistomide A (1)**

Approximately 100  $\mu$ L of Raney 2800 nickel (50% slurry in H<sub>2</sub>O; excess) was added to eudistomide A (1.0 mg, 1.3  $\mu$ mol) in MeOH (1.2 mL). Argon was bubbled through the solution to remove O<sub>2</sub>, the resulting black suspension was heated at 65 °C under argon for 4 h, and monitored by HPLC for the disappearance of starting material. Upon cooling, the solution was purified on a C<sub>18</sub> SPE cartridge using MeOH as eluant, yielding pure desthioeudistomide B (**3**, 0.5 mg, 54% yield).

**Desthioeudistomide B (3)**—amorphous white solid. HRESIMS m/z 708.49059 [M+H]<sup>+</sup> (calcd for C<sub>37</sub>H<sub>66</sub>N<sub>5</sub>O<sub>8</sub>, 708.49059); HRESI-MS/MS, see Table 2.

# Absolute Configuration of each Amino Acid in Desthioeudistomide B (3)

Compound **3** (0.5 mg) was dissolved in 6N HCl (500  $\mu$ L) and heated at 110 °C under argon for 21 h. The product mixture was lyophilized and analyzed by chiral HPLC [column, Phenomenex Chirex phase 3126 (D) (250 × 4.6 mm); solvent 1 mM CuSO<sub>4</sub>/CH<sub>3</sub>CN (95:5); flow rate, 1.0 mL/min, UV detection at 254 nm], comparing the retention times with those of authentic standards. The retention times (min) of the authentic amino acids were: L-Ala (6.9), p-Ala (8.4), L-Pro (9.5), p-Pro (20.0), L-Leu (51.3), and p-Leu (65.9). The retention times of the amino acid components in the acid hydrolyzate were 6.9, 9.4, and 51.3, indicating the presence of L-Ala, L-Pro, and L-Leu, respectively.

#### 12-(tert-butyldimethylsilyloxy)dodecan-1-ol (5)

0.4 g of NaH (60% dispersion in mineral oil, 10 mmol) was washed with anhydrous THF ( $2 \times 20 \text{ mL}$ ; to remove the mineral oil), and suspended in anhydrous THF (20 mL). 2.02 g (10 mmol) of dodecane-1,12-diol (4) was added to the solution and left to stir under argon at 55 °C. After 18 h, the mixture was cooled to room temperature, and 1.51 g of TBSCl (10 mmol) in anhydrous

THF (2.5 mL) was added, and allowed to stir for 2 h at rt. The mixture was diluted with Et<sub>2</sub>O, and washed successively with 10% K<sub>2</sub>CO<sub>3</sub> and brine. The combined organic extracts were dried (MgSO<sub>4</sub>) and concentrated *in vacuo*. Purification of the crude material by silica column chromatography: [solvent 1: 90% Hexanes/10% EtOAc (elutes the di-silyl ether product); solvent 2: 80% Hexanes/20% EtOAc] afforded the mono-silylated product **5** as a colorless solid (1.01 g, 57% yield). **5**: <sup>1</sup>H NMR (CD<sub>3</sub>OD, 400 MHz)  $\delta$  3.57 (2H, t, *J* = 6.4 Hz), 3.48 (2H, t, *J* = 6.7 Hz), 1.45 (4H, m), 1.28 (14H, br s), 0.84 (9H, s), 0.00 (6H, s); <sup>13</sup>C NMR (CD<sub>3</sub>OD, 400 MHz)  $\delta$  63.4, 63.1, 34.0, 33.8, 31.0–30.5, 27.1, 27.0, 26.6, 19.3, –5.0; HRESIMS *m/z* 299.2765 [M-H<sub>2</sub>O]<sup>+</sup> (calcd for C<sub>18</sub>H<sub>39</sub>OSi, 299.2770).

#### 14-(tert-butyldimethylsilyloxy)tetradecan-3-ol (6)

A 0.5 M aq solution of KBr (7.14 mg, 0.06 mmol) was added to a vigorously stirring solution of 5 (188.8 mg, 0.6 mmol) in CH<sub>2</sub>Cl<sub>2</sub> (0.7 mL) at 0 °C, followed by the addition of TEMPO (2,2,6,6-tetramethylpiperdine 1-oxyl) free radical (0.94 mg, 0.006 mmol). NaOCl (5% active chlorine) adjusted to pH = 8.6 using NaHCO<sub>3</sub> (1.5 mL) was added drop-wise to the solution and allowed to stir for 5 min at 0 °C until the hypochlorite was consumed and the solution went from a red-orange to a milky yellow color. The organic layer of the two phase mixture was separated, dried (MgSO<sub>4</sub>) and concentrated to dryness *in vacuo* at 0  $^{\circ}$ C. The crude aldehyde (160 mg, 0.51 mmol) was dissolved in anhydrous Et<sub>2</sub>O (3.2 mL) and cooled to -30 °C under argon. EtMgBr (202  $\mu$ L, 0.61 mmol, 3M in ether) was added drop-wise to the solution, and the mixture was left to stir at -30 °C. After 1.5 h, the reaction was quenched with saturated  $NH_4Cl$  (316 µL). The solution was warmed to room temperature, washed with brine, and extracted with Et<sub>2</sub>O. The ethereal extracts were combined, dried (MgSO<sub>4</sub>), filtered and concentrated in vacuo. The crude product was purified by silica column chromatography: [solvent: 83% Hexanes/17% MTBE] to afford 6 as a colorless solid (91.8 mg, 53% yield). **6**: <sup>1</sup>H NMR (CDCl<sub>3</sub>, 400 MHz)  $\delta$  3.55 (2H, t, J = 6.7 Hz), 3.47 (1H, m), 1.55–1.30 (6H, m), 1.24 (16H, br s), 0.89 (3H, t, J = 7.5 Hz), 0.84 (9H, s), 0.00 (6H, s); <sup>13</sup>C NMR (CDCl<sub>3</sub>, 400 MHz) & 73.5, 63.6, 37.2, 33.1, 30.4, 30.0-29.5, 26.2, 26.0, 25.9, 18.6, 10.1, -5.0; HRESIMS m/z 367.3009 (calcd for C<sub>20</sub>H<sub>44</sub>O<sub>2</sub>SiNa, 367.3008).

#### Tetradecane-1,12-diol (7)

TBAF (tetrabutylammonium fluoride, 124.5 mg, 0.48 mmol) in THF (0.5 mL) was added dropwise to a solution of **6** (81.8 mg, 0.24 mmol) in THF (3.1 mL) at 0 °C. The reaction mixture was warmed to room temperature and stirred for 4 h. Cold water was added to quench the reaction, and the resultant mixture was extracted with EtOAc. The combined organic extracts were washed with brine, dried (MgSO<sub>4</sub>), filtered and concentrated to dryness *in vacuo*. The crude product was purified by silica column chromatography: [solvent 1: 80% CH<sub>2</sub>Cl<sub>2</sub>/20% EtOAc (eluted remaining **6**); solvent 2: 50% CH<sub>2</sub>Cl<sub>2</sub>/50% EtOAc] to yield **7** as a colorless solid (52.0 mg, 95% yield). **7**: <sup>1</sup>H NMR (CDCl<sub>3</sub>, 400 MHz)  $\delta$  3.58 (2H, t, *J* = 6.6 Hz), 3.46 (1H, m), 1.60–1.30 (8H, m), 1.24 (14H, br s), 0.89 (3H, t, *J* = 7.4 Hz); <sup>13</sup>C NMR (CDCl<sub>3</sub>, 400 MHz)  $\delta$  73.5, 63.2, 37.1, 33.0, 30.3, 30.0–29.5, 25.9, 25.8, 10.1; HRESIMS *m/z* 213.2217 [M-H<sub>2</sub>O]<sup>+</sup> (calcd for C<sub>14</sub>H<sub>29</sub>O, 213.2218).

#### 12-oxo-tetradecanoic acid (8)

A solution of **7** (15 mg, 65.2  $\mu$ mol) in anhydrous DMF (700  $\mu$ L) was treated with PDC (197 mg, 0.52 mmol) under argon. The dark brown solution was allowed to stir at room temperature for 17 h. The reaction was quenched with water and extracted with EtOAc. The combined organic layers were washed successively with water and brine, dried (MgSO<sub>4</sub>), and concentrated to dryness *in vacuo* to yield **8** as a colorless solid (15.1 mg, 96% yield). **8**: <sup>1</sup>H NMR (CDCl<sub>3</sub>, 400 MHz) 2.50–2.20 (6H, m), 1.60–1.40 (4H, m), 1.23 (12H, br s) 1.01 (3H,

br t, J = 6.5 Hz); <sup>13</sup>C NMR (CDCl<sub>3</sub>, 400 MHz)  $\delta$  212.0, 179.7, 42.3, 35.7, 33.9, 29.5–28.5, 24.5, 23.8, 7.7; HRESIMS m/z 243.1946 [M+H]<sup>+</sup> (calcd for C<sub>14</sub>H<sub>27</sub>O<sub>3</sub>, 243.1955).

#### **Cyclic Pentapeptide (9)**

amorphous white solid. <sup>1</sup>H NMR and <sup>13</sup>C NMR data, see Supporting Info Table 1; HRESIMS m/z 552.1912 [M+Na]<sup>+</sup> (calcd for C<sub>22</sub>H<sub>35</sub>N<sub>5</sub>O<sub>6</sub>S<sub>2</sub>Na, 552.1926).

# Eudistomide A (1)

EDC (12.6 mg, 22.7 µmol) and HOBt (8.9 mg, 22.7 umol) were added to a stirred solution of 8 (5.1 mg, 22.7 µmol) in anhydrous 90% CH<sub>2</sub>Cl<sub>2</sub>/10% MeOH (1.8 mL) under argon for 3 h to activate the acid. DIPEA (3.3  $\mu$ L, 19.3  $\mu$ mol) was added to a stirred solution of **9** (10.2 mg, 19.3 µmol) in 85% CH<sub>2</sub>Cl<sub>2</sub>/15% MeOH (1.9 mL) and allowed to stir for 3 h to neutralize the peptide. After 3 h, the neutralized peptide (9) was added to the acid (8) and DIPEA was added to the solution  $(3.3 \,\mu\text{L}, 19.3 \,\mu\text{mol})$ . The reaction was monitored by HPLC for the disappearance of starting material. The reaction was quenched with H<sub>2</sub>O after 48h, extracted with CH<sub>2</sub>Cl<sub>2</sub>, and concentrated *in vacuo*. The reaction products were purified using a Phenomenex  $C_{18}$  (250  $\times$  10 mm) column using a gradient of 2% CH<sub>3</sub>CN/98% H<sub>2</sub>O (+0.1% AcOH) to 98% CH<sub>3</sub>CN/ 2% H<sub>2</sub>O (+ 0.1% AcOH) at 4.5 mL/min over 40 min to yield pure eudistomide A (1) (2.2 mg, 42%) eluting at 33.4 min, the peptide methyl ester (11) (1.0 mg, 27%) eluting at 11.2 min, and the eudistomide acid (10) (0.8 mg, 16%) eluting at 28.3 min. HPLC retention times of the natural eudistomide A (1) and the synthetic eudistomide A (1) were compared using a Phenomenex C<sub>18</sub> ( $250 \times 4.6$  mm) column employing a gradient of 50% CH<sub>3</sub>CN/50% H<sub>2</sub>O (+ 0.1% AcOH) to 98% CH<sub>3</sub>CN/2% H<sub>2</sub>O (+ 0.1% AcOH) at 1 mL/min over 20 min. Natural 1 had a retention time of 12.8 min, and synthetic 1 also had a retention time of 12.8 min. 1:  $[\alpha]^{20}$ <sub>D</sub> -3.5 (c 0.06, MeOH); <sup>1</sup>H NMR and <sup>13</sup>C NMR data identical to natural product, see Table 1; HRESIMS *m/z* 768.4034 [M+H]<sup>+</sup> (calcd for C<sub>37</sub>H<sub>62</sub>N<sub>5</sub>O<sub>8</sub>S<sub>2</sub>, 768.4040). ESI-MS/MS, see Table 2.

#### Eudistomide B Epimers (2, 12)

A solution of 1 (1.3 mg, 1.7 µmol) in anhydrous MeOH (250 µL) at -78 °C was treated with NaBH4 (1.0 mg, 26.4 µmol) under argon. The solution was allowed to stir for 7 h. The reaction was quenched with water, the MeOH was evaporated under argon, and the resulting aqueous layer was extracted with CH<sub>2</sub>Cl<sub>2</sub>. The combined organic extracts were dried under argon, and redissolved in 5% AcOH (400 µL, pH=6). DMSO (100 µL) was added to the solution that was left to stir overnight. The reaction products were concentrated *in vacuo* and purified using a Phenomenex C<sub>18</sub> (250 × 10 mm) column employing a gradient of 50% CH<sub>3</sub>CN/50% H<sub>2</sub>O (+ 0.1% AcOH) to 98% CH<sub>3</sub>CN/2% H<sub>2</sub>O (+ 0.1% AcOH) at 4.5 mL/min over 20 min to yield a mixture of C-35 alcohol epimers (**2**, **12**) co-eluting at 11.3 min (1.1 mg, 84%). HPLC retention times of the natural eudistomide B (**2**) and the synthetic C-35 alcohol epimers (**2**, **12**) were compared using a Phenomenex C<sub>18</sub> (250 × 4.6 mm) column employing a gradient of 50% CH<sub>3</sub>CN/2% H<sub>2</sub>O (+ 0.1% AcOH) to 98% CH<sub>3</sub>CN/2% H<sub>2</sub>O (+ 0.1% AcOH) at 1 mL/min over 20 min. Natural **2** had a retention time of 11.2 min, and the synthetic alcohol epimers (**2**, **12**) also had retention times of 11.2 min. LRESIMS *m*/*z* 770.4 [M+H]<sup>+</sup>; ESI-MS/MS, see Table 2

### Supplementary Material

Refer to Web version on PubMed Central for supplementary material.

# Acknowledgments

The authors wish to acknowledge Dr. Krishna Parsawar and Dr. Chad Nelson, University of Utah Mass Spectrometry and Proteomics Core Facility, for performing FTMS experiments. Ascidian taxonomy was graciously provided by

Gretchen Lambert, University of Washington, Friday Harbor Laboratories. The authors would also like to recognize Dr. Robert Schackmann and Scott Endicott of the University of Utah Peptide Synthesis Core Facility. We acknowledge the NIH, grants RR13030, RR06262 and RR14768, and NSF grant DBI-0002806 for funding NMR instrumentation in the University of Utah Health Sciences NMR Facility. This research was supported by NCI grant CA67786 and NIH grant CA36622.

#### **References and Notes**

- 1. Ireland CM, Scheuer PJ. J. Am. Chem. Soc 1980;102:5688-5691.
- 2. Williams DE, Moore RE, Paul VJ. J. Nat. Prod 1989;52:732-739. [PubMed: 2809606]
- 3. Fu X, Do T, Schmitz FJ, Andrusevich V, Engel MH. J. Nat. Prod 1998;61:1547–1551. [PubMed: 9868162]
- 4. Ireland CM, Durso AR, Newman RA, Hacker MP. J. Org. Chem 1982;47:1807-1811.
- 5. McDonald LA, Ireland CM. J. Nat. Prod 1992;55:376–379. [PubMed: 1593285]
- Rashid MA, Gustafson KR, Cardellina JH 2nd, Boyd MR. J. Nat. Prod 1995;58:594–597. [PubMed: 7623037]
- Degnan BM, Hawkins CJ, Lavin MF, McCaffrey EJ, Parry DL, van den Brenk AL, Watters DJ. J. Med. Chem 1989;32:1349–1354. [PubMed: 2724305]
- Morris LA, Kettenes van den Bosch JJ, Versluis K, Thompson GS, Jaspars M. Tetrahedron 2000;56:8345–8353.
- Hawkins CJ, Lavin MF, Marshall KA, van den Brenk AL, Watters DJ. J. Med. Chem 1990;33:1634– 1638. [PubMed: 2342056]
- 10. Rinehart KL, Gloer JB, Cook JC, Mizsak SA, Scahill TA. J. Am. Chem. Soc 1981;103:1857-1859.
- Rinehart KL, Kishore V, Bible KC, Sakai R, Sullins DW, Li KM. J. Nat. Prod 1988;51:1–21. [PubMed: 3373220]
- 12. Schmitz FJ, Yasumoto T. J. Nat. Prod 1991;54:1469-1490. [PubMed: 1812209]
- 13. Rinehart KL. Med. Res. Rev 2000;20:1-27. [PubMed: 10608919]
- 14. Boulanger A, Abou-Mansour E, Badre A, Banaigs B, Combaut G, Francisco C. Tetrahedron Lett 1994;35:4345–4348.
- 15. Sakai R, Stroh JG, Sullins DW, Rinehart KL. J. Am. Chem. Soc 1995;117:3734-3748.
- Rinehart KL, Kobayashi J, Harbour GC, Hughes RG, Mizsak SA, Scahill TA. J. Am. Chem. Soc 1984;106:1524–1526.
- 17. Kobayashi J, Harbour GC, Gilmore J, Rinehart KL. J. Am. Chem. Soc 1984;106:1526–1528.
- 18. Kinzer KF, Cardellina JH. Tetrahedron Lett 1987;28:925-926.
- 19. Kang H, Fenical W. Nat. Prod. Lett 1996;9:7–12.
- 20. Schupp P, Poehner T, Edrada R, Ebel R, Berg A, Wray V, Proksch P. J. Nat. Prod 2003;66:272–275. [PubMed: 12608864]
- 21. Kobayashi J, Nakamura H, Ohizumi Y, Hirata Y. Tetrahedron Lett 1986;27:1191-1194.
- 22. Kobayashi J, Cheng J, Ohta T, Nozoe S, Ohizumi Y, Sasaki T. J. Org. Chem 1990;55:3666–3670.
- 23. Murata O, Shigemori H, Ishibashi M, Sugama K, Hayashi K, Kobayashi J. Tetrahedron Lett 1991;32:3539–3542.
- 24. Rashid MA, Gustafson KR, Boyd MR. J. Nat. Prod 2001;64:1454–1456. [PubMed: 11720532]
- 25. Kobayashi J, Cheng J, Ohta T, Nakamura H, Nozoe SHY, Ohizumi Y, Sasaki T. J. Org. Chem 1988;53:6147–6150.
- 26. Kikuchi Y, Ishibashi M, Sasaki T, Kobayashi J. Tetrahedron Lett 1991;32:797-798.
- 27. Kobayashi J, Cheng J, Kikuchi Y, Ishibashi M, Yamamura S, Ohizumi Y, Ohta T, Nozoe S. Tetrahedron Lett 1990;31:4617–4620.
- 28. Rudi A, Kashman Y. J. Org. Chem 1989;54:5331–5337.
- 29. He HY, Faulkner DJ. J. Org. Chem 1991;56:5369-5371.
- 30. Viracaoundin I, Faure R, Gaydou EM, Aknin M. Tetrahedron Lett 2001;42:2669–2671.
- 31. Van Wagoner RM, Jompa J, Tahir A, Ireland CM. J. Nat. Prod 2001;64:1100–1101. [PubMed: 11520239]

- 33. Debitus C, Laurent D, Pais M. J. Nat. Prod 1988;51:799-801.
- 34. Adesanya SA, Chbani M, Pais M, Debitus C. J. Nat. Prod 1992;55:525–527. [PubMed: 1512602]
- Makarieva TN, Ilyin SG, Stonik VA, Lyssenko KA, Denisenko VA. Tetrahedron Lett 1999;40:1591– 1594.
- Van Wagoner RM, Jompa J, Tahir A, Ireland CM. J. Nat. Prod 1999;62:794–797. [PubMed: 10346975]
- Makarieva TN, Dmitrenok AS, Dmitrenok PS, Grebnev BB, Stonik VA. J. Nat. Prod 2001;64:1559– 1561. [PubMed: 11754612]
- 38. Siemion IZ, Wieland T, Pook KH. Angew. Chem. Int. Ed. Engl 1975;14:702-703. [PubMed: 812384]
- Davis RA, Mangalindan GC, Bojo ZP, Antemano RR, Rodriguez NO, Concepcion GP, Samson SC, de Guzman D, Cruz LJ, Tasdemir D, Harper MK, Feng X, Carter GT, Ireland CM. J. Org. Chem 2004;69:4170–4176. [PubMed: 15176844]
- 40. Culbertson SM, Porter NA. J. Am. Chem. Soc 2000;122:4032-4038.
- 41. Baag MM, Puranik VG, Argade NP. J. Org. Chem 2007;72:1009-1012. [PubMed: 17253824]
- 42. Tam JP, Wu CR, Liu W, Zhang JW. J. Am. Chem. Soc 1991;113:6657-6662.
- Ohtanib T, Nakatsukasaa H, Kamezawab M, Tachibanab H, Naoshimaa Y. J. Mol. Catal. B: Enzym 1998;4:53–60.
- 44. Baumann M, Hauer BH, Bornscheuer UT. Tetrahedron: Asymmetry 2000;11:4781-4790.
- 45. Blunt, JW.; Blunt, DA. MarinLit Database. Department of Chemistry, University of Canterbury; http://www.chem.canterbury.ac.nz/marinlit/marinlit.shtml
- 46. Rowley DC, Kelly S, Kauffman CA, Jensen PR, Fenical W. Bioorg. Med. Chem 2003;11:4263–4274. [PubMed: 12951157]
- 47. Smyrniotopoulos V, Abatis D, Tziveleka LA, Tsitsimpikou C, Roussis V, Loukis A, Vagias C. J. Nat. Prod 2003;66:21–24. [PubMed: 12542338]
- Zhou BN, Tang SB, Johnson RK, Mattern MP, Lazo JS, Sharlow ER, Harich K, Kingston DGI. Tetrahedron 2005;61:883–887.
- Bultel-Ponce VV, Debitus C, Berge JP, Cerceau C, Guyot M. J. Mar. Biotechnol 1998;6:233–236. [PubMed: 9852617]
- 50. Nagle DG, McClatchey WC, Gerwick WH. J. Nat. Prod 1992;55:1013-1017. [PubMed: 1402953]
- 51. Neuhof T, Schmieder P, Preussel K, Dieckmann R, Pham H, Bartl F, von Dohren H. J. Nat. Prod 2005;68:695–700. [PubMed: 15921412]
- 52. Kalinovskaya NI, Kuznetsova TA, Ivanova EP, Romanenko LA, Voinov VG, Huth F, Laatsch H. Mar. Biotechnol 2002;4:179–188. [PubMed: 14961278]
- 53. Orjala J, Nagle D, Gerwick WH. J. Nat. Prod 1995;58:764–768. [PubMed: 7623050]
- 54. Suntornchashwej S, Suwanborirux K, Koga K, Isobe M. Chem. Asian J 2007;2:114–122. [PubMed: 17441144]
- 55. Tan LT, Okino T, Gerwick WH. J. Nat. Prod 2000;63:952–955. [PubMed: 10924172]
- 56. Gu W, Cueto M, Jensen PR, Fenical W, Silverman RB. Tetrahedron 2007;63:6535-6541.
- Perez Baz J, Canedo LM, Fernandez Puentes JL, Silva Elipe MV. J. Antibiot 1997;50:738–741. [PubMed: 9360618]
- Williams DE, Austin P, Diaz-Marrero AR, Soest RV, Matainaho T, Roskelley CD, Roberge M, Andersen RJ. Org. Lett 2005;19:4173–4176. [PubMed: 16146380]







Scheme 1. Synthesis of 12-oxo-tetradecanoic acid (8)



Scheme 2. Synthesis of eudistomide A (1)





_	
_	
	Ľ
	P
	١.
_	
<u> </u>	
<b>_</b>	
0	
-	
•	
_	
<	
_	
0)	
-	
<u> </u>	
10	
0,	
0	
~	
1	
0	
t -	

Table 1	NMR data for eudistomides A (1) and B (2) (600 MHz, $CDCI_3$ )

			eudistomide A (1)	eudis	tomide B (2) <sup>c</sup>
	position	åc	$\delta_{ m H}$ mult (J, Hz)	δ <sub>C</sub>	$\delta_{ m H}$ mult (J, Hz)
Cys-1 (Cys-OMe)	_	170.9	1	170.9	1
	6	49.9	4.86, ddd (11.0, 9.3, 3.7)	49.9	4.87
	3a	41.5	3.32, dd (14.2, 3.7)	41.5	3.33
	3b		2.77, dd (14.2, 11.0)		2.77
	4	52.5	3.73, s	52.5	3.74
	NH	Ι	7.20, d (9.3)	Ι	7.18
Leu	S	171.5	1	171.5	Ι
	6	51.4	4.49, b ddd $(8.1, 5.3, 5.0)$	51.4	4.50
	7a	36.1	1.83, b ddd $(14.7, 10.8, 5.0)$	36.1	1.83
	7b		$1.48, b  { m ddd}  (14.7,  9.3,  5.3)$		1.48
	8	24.2	1.60, m (10.8, 9.3, 6.6, 6.4)	24.2	1.60
	6	21.0	0.84, d (6.4)	21.0	0.85
	10	23.0	0.94, d (6.6)	23.0	0.95
	NH	Ι	9.42, br d (8.1)	Ι	9.40
Pro-1	11	173.2	1	173.2	Ι
	12	60.9	$4.48,^{b}$ dd $(8.2, 2.5)$	6.09	4.49
	13a	32.5	2.35, b dddd (12.5, 6.5, 6.5, 2.5)	32.5	2.35
	13b		$2.10, ^b$ dddd (12.5, 10.5, 8.2, 1.9)		2.10
	14a	21.3	1.96, <sup>d</sup> m	21.3	1.96
	14b		1.80, <sup><i>a</i></sup> m		1.80
	15	46.5	3.56, b dd (7.2, 10.0)	46.5	3.56
Pro-2	16	170.1	1	170.1	Ι
	17	59.4	4.94, dd (8.7, 2.0)	59.4	4.94
	18a	30.7	$2.28,^b$ dddd $(8.1, 8.7, 12.3, 16.4)$	30.7	2.28
	18b		1.83, <sup><i>a</i></sup> m		1.83
	19a	22.4	1.94, <sup><i>a</i></sup> m	22.4	1.94

_
<b>_</b>
_
_
_
_
- U
~
-
-
_
_
-
<u> </u>
_
_
$\sim$
0
_
•
~
$\geq$
-
0
_
_
-
<u> </u>
CD
0
~
_
_
Ð
D

**NIH-PA Author Manuscript** 

			eudistomide A (1)	eudist	omide B (2) <sup>c</sup>
	position	åc	$\delta_{ m H}$ mult $(J,{ m Hz})$	δ <sub>c</sub>	$\delta_{\mathrm{H}}$ mult ( <i>J</i> , Hz)
	19b		1.87, <sup>a</sup> m		1.87
	20a 4	16.4	$3.73,^b$ ddd (11.9, 8.8, 4.6)	46.4	3.73
	20b		$3.53,^b$ ddd (11.9, 7.7, 7.7)		3.53
Cys-2	21 16	59.4	1	169.4	Ι
	22	8.8	4.61, ddd (10.5, 10.5, 5.1)	48.8	4.62
	23a	8.4	3.25, dd (12.9, 5.1)	38.4	3.25
	23b		2.56, dd (12.9, 10.5)		2.57
	HN		6.08, br d (10.5)	Ι	5.98
12-oxo-tetradecanoyl	24 17	71.8	I	171.8	I
	25	36.1	2.10, <sup>a</sup> m	36.1	2.10, <sup><i>a</i></sup> m
	26 2	24.9	1.55, <sup>a</sup> m	24.9	1.55, <sup>a</sup> m
	27–32	0.62	1.23, br s	29.0	1.23, br s
	33a 2	23.9	1.54, <sup>a</sup> m	25.8	1.38, <sup><i>a</i></sup> m
	33b	Ι	I		1.28, m
	34 4	12.4	2.36, t (7.5)	36.7	1.37, <sup><i>a</i></sup> m
	35 21	12.1	I	73.1	3.49, m
	36a 3	35.8	2.38, q (7.3)	29.7	1.47, m
	36b		I		1.40, m
	37	7.8	1.02, t (7.3)	10.1	0.92, t (7.5)
<sup>a</sup> Signals overlapped.					

J Org Chem. Author manuscript; available in PMC 2010 February 6.

 $^{\rm C}$  Coupling constants for peptide portion are similar.

b Obtained from 1D-TOCSY.

#### Table 2

Fragmentation ions observed in ESI-MS/MS data confirming the amino acid sequences of eudistomides A (1) and B (2), and desthioeudistomide B (3)

				_
Compound	Ions Observed	Identity	Molecular Formula	
1, 2	281 <sup><i>a</i>,<i>c</i></sup>	Leu-Cys-OMe + S	$C_{10}H_{21}N_2O_3S_2$	
1, 2	378 <sup><i>a</i>,<i>c</i></sup>	Pro-Leu-Cys-OMe + S	$C_{15}H_{28}N_3O_4S_2$	
1, 2	475 <sup><i>a</i>,<i>c</i></sup>	Pro-Pro-Leu-Cys-OMe + S	$C_{20}H_{35}N_4O_5S_2$	
1, 2	308 <sup><i>a</i>,<i>c</i></sup>	Pro-Pro-Leu	$C_{16}H_{26}N_3O_3$	
1, 2	211 <sup><i>a</i>,<i>c</i></sup>	Pro-Leu	$C_{11}H_{19}N_2O_2$	
3	217.15502 <sup>b</sup>	Leu-Ala-OMe	$C_{10}H_{21}N_2O_3$	
3	314.20796 <sup>b</sup>	Pro-Leu-Ala-OMe	$C_{15}H_{28}N_3O_4$	
3	411.26097 <sup>b</sup>	Pro-Pro-Leu-Ala-OMe	$C_{20}H_{35}N_4O_5$	
3	308.19740 <sup>b</sup>	Pro-Pro-Leu	$C_{16}H_{26}N_3O_3$	
3	211.14442 <sup>b</sup>	Pro-Leu	$C_{11}H_{19}N_2O_2$	

<sup>*a*</sup>Ions observed on a micro Q-tof using CID.

 $b_{\mbox{Ions}}$  observed on a LTQ-FT using IRMPD, all ions have <1.5 ppm accuracy.

 $^{\it C}$  Identical ions observed for synthetic eudistomides A (1) and B (2).