

# NIH Public Access

Author Manuscript

J Phys Chem B. Author manuscript; available in PMC 2009 September 18.

Published in final edited form as:

J Phys Chem B. 2008 September 18; 112(37): 11827–11837. doi:10.1021/jp8054679.

## Kinetic-Dynamic Model for Conformational Control of an Electron Transfer Photocycle: Mixed-Metal Hemoglobin Hybrids

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## Abstract

It is becoming increasingly clear that the transfer of an electron across a protein-protein interface is coupled to the dynamics of conformational conversion between and within ensembles of interface conformations. ET reactions in conformationally mobile systems provide a 'clock' against which the rapidity of a dynamic process may be measured, and we here report a simple kinetic (master equation) model that self-consistently incorporates conformational dynamics into an electron transfer (ET) photocycle comprised of a photoinitiated 'forward' step and thermal return to ground. This kinetic/ dynamic (KD) model assumes an ET complex exists as multiple interconverting conformations which partition into an ET-optimized (reactive; R) population and a less-reactive population (S). We take the members of each population to be equivalent by constraining them to have the same conformational energy, the same average rate constant for conversion to members of the other population, and the same rate constants for forward and back ET. This model successfully describes the changes in the ET photocycle within the 'pre-docked' mixed-metal hemoglobin (Hb) hybrid,  $\left[\alpha\right]$ (Zn),  $\beta$ (Fe<sup>3+</sup>N<sub>3</sub><sup>-</sup>)], as conformational kinetics are modulated by variations in viscosity (n = 1-15 cP; 20 °C). The description reveals how the conformational 'routes' by which a hybrid progresses through a photocycle differ in different dynamic regimes. Even at  $\eta = 1$  cP the populations are not in fast exchange and ET involves a complex interplay between conformational and ET processes; at intermediate viscosities the hybrid exhibits 'differential dynamics' in which the forward and back ET processes involve different initial ensembles of configurational substates; by  $\eta = 15$  cP the slowexchange limit is approached. Even at low viscosity the ET-coupled motions are fairly slow, with rate constants of  $< 10^3 \text{s}^{-1}$ . Current ideas about Hb function lead to the testable hypothesis that ET in the hybrid may be coupled to allosteric fluctuations of the two  $[\alpha_1, \beta_2]$  dimers of Hb.

## INTRODUCTION

The study of long-range electron transfer (ET) occupies a central place in modern biology and chemistry. <sup>1,2</sup> However, it is becoming increasingly clear that the transfer of an electron across a protein-protein interface almost invariably is coupled to the dynamics of conformational conversion between and within ensembles of interface conformations.<sup>3-17</sup> This is true for large multi-component redox systems that involve tightly-bound and precisely arranged redox centers, such as the [reaction center, cytochrome  $c_2$ ] complex, <sup>12</sup> the [electron transferring flavoprotein, medium chain acyl-CoA dehydrogenase] complex, <sup>18</sup> and the [plastocyanin, photosystem I] complex. <sup>19</sup> Indeed, such systems often exhibit ET reactions that are gated<sup>3</sup>, <sup>6,8,9</sup> – controlled by the dynamics rather than the ET process. Such coupling also occurs in simpler complexes of small, weakly-bound redox protein partners. In these cases, ET-active configurations are only rarely sampled during many diffusional encounters, and the complexes can exhibit conformational 'phase space' control by a process denoted dynamic docking.<sup>4</sup>, <sup>20,21</sup> In response to the experimental investigations of these phenomena there have been

numerous theoretical studies, with diverse approaches that range from diffusion-dynamics<sup>22</sup>,  $^{23}$  to kinetic/master-equation treatments.<sup>24</sup>

In the case of photoinitiated ET, both experimental and theoretical studies generally have focused on the photostimulated 'forward' ET process (rate constant  $k_t$ ) and infrequently<sup>10</sup>, <sup>25,26</sup> considered an entire ET photocycle (Scheme 1) comprised of both forward and 'back' ET(rate constant k<sub>b</sub>). However, our recent work suggests that detailed studies of a full photocycle might give a more complete picture of the dynamics of a protein interface, in part at least because the forward and back ET processes should involve different initial ensembles of configurational substates (Fig 1).<sup>5,27</sup> The forward ET reaction is initiated with the system in its equilibrium ensemble of configurations. When most - or the most stable - ensemble members exhibit low ET rate constants, then forward ET is controlled by conformational conversion to a reactive state. In contrast, because the photoinitiated ET process necessarily generates an ET intermediate that is in a reactive configuration, the back reaction begins in a non-equilibrium conformation and is controlled by the competition between direct back ET and deactivation from this conformation, Fig 1. Evidence for such 'differential dynamics' effects within a photocycle was obtained from examination of the effects of viscogens and interface mutations on the forward and back ET reactions of four ET complexes encompassing a wide range of energy landscapes and stabilities: the weakly-bound and highly dynamic [hemoglobin (Hb), cytochrome  $b_5$ ] and [myoglobin, cytochrome  $b_5$ ] complexes; the moderately bound [cytochrome c peroxidase, cytochrome c] complex; and the structurally defined, mixed-metal Hb hybrids,  $\left[\alpha(\text{Zn}), \beta(\text{Fe}^{3+}L)\right]^{5}$ 

Mixed-metal Hb hybrids in particular provide an ideal system with which to explore dynamic control of an ET photocycle involving pre-docked protein redox partners.<sup>28-31</sup> The hybrid is an  $[\alpha(Zn)_2, \beta(Fe)_2]$  tetramer and ET occurs within two independent pairs of  $[\alpha_1(Zn), \beta_2(Fe)]$  ET partners, across distances of about 17 Å between heme edges and 25 Å between metal ions, Fig 2.<sup>32-35</sup> However even this 'pre-docked' tetramer can access multiple configurations, as shown by recent NMR<sup>36-38</sup> and x-ray<sup>39-42</sup> studies. Previous work revealed a coupling of ET to conformational conversion among these configurations through the failure of the simple kinetic equations for the photocycle of a rigid complex (Scheme 1) to describe the viscosity ( $\eta$ ) dependence of the hybrid photocycle.<sup>5,27</sup> Instead the forward and back ET steps exhibited 'differential dynamics'; as the interfacial dynamics were slowed with increasing  $\eta$  the apparent forward ET rate constant decreased, whereas the apparent back ET rate constant increased.<sup>5</sup>, 27

Although this description in terms of viscosity-dependent ET rate constants provided qualitative insights into the coupling of the ET photocycle to conformational dynamics, the kinetic equations used did not explicitly include dynamic processes, and indeed did not represent any well-defined kinetic mechanism. Therefore we have developed, and herein describe, a simple kinetic (master equation) model that explicitly incorporates such processes into a photocycle. This 'KD' model has been applied to experiments in which we monitor the photocycle of a mixed-metal Hb hybrid as conformational kinetics are modulated by variations in viscosity at fixed temperature, thereby avoiding changes in the microscopic rate-constants that would come with variations in temperature. We find that the experiments are welldescribed with the KD model, and that there are conditions under which the behavior of the Hb hybrids in fact corresponds to the postulated differential dynamics paradigm for dynamically controlled ET that is sketched in Fig 1. The analysis further introduces the idea of a conformational 'route' by which a hybrid progresses from a photo-activated triplet state, through an ET intermediate, then back to ground, and reveals how these routes evolve as the system traverses through distinct dynamic regimes. The possible nature of the ET-coupled conformational motions are considered in light of current ideas about Hb function.<sup>43</sup> Finally,

the advantages offered by studying an ET photocycle are discussed and limitations of the KD model are examined.

## EXPERIMENTAL PROCEDURES

#### Materials

Hemoglobin was acquired from outdated human erythrocytes obtained from Life Source (Skokie, IL). NaN<sub>3</sub>, K<sub>3</sub>Fe(CN)<sub>6</sub>, IHP (inositol hexaphosphate), and Sephadex G25 were obtained from Sigma-Aldrich. Glycerol, K<sub>2</sub>HPO<sub>4</sub> and KH<sub>2</sub>PO<sub>4</sub> were obtained from VWR and zinc protoporphyrin IX (ZnP) from Frontier Scientific. H<sub>2</sub>O was purified by a MilliQ water purification system (Millipore). Prepurified N<sub>2</sub>(g) from Air Gas was used for all deoxygenating operations.

## Preparation of $[\alpha(Zn), \beta(Fe^{3+}N_3)]$

Hemoglobin was purified by anion exchange chromatography, and separated into its component chains as described previously.<sup>44,45</sup> After removal of bound mercurial with mercaptoethanol, the heme was extracted from the  $\alpha$ -chains and the apo- $\alpha$ -protein was reconstituted with ZnP and recombined with the native  $\beta$ -chains to give the [ $\alpha$ (ZnP),  $\beta$ (FeCO)] hybrid as described earlier.<sup>44-46</sup> The [ $\alpha$ (ZnP),  $\beta$ (FeCO)] hybrid was oxidized at 4°C with a 5-fold excess of K<sub>3</sub>Fe(CN)<sub>6</sub> at pH 6 with a N<sub>2</sub> stream over the solution. Excess ferricyanide was removed and the hybrid hemoglobin converted to the azidomet form by passage over G-25 resin equilibrated with a phosphate-buffered solution containing 100 mM sodium azide at pH 7.

Cryosolutions of varying viscosity were prepared at selected compositions (w/w) of glycerol and deionized water. Stock solutions of 50 mM monobasic and dibasic phosphate were prepared by dissolving the potassium salts in the cryosolvent at the desired w/v composition. Cryobuffers were prepared at pH 7 by titrating the stock monobasic cryosolution with the stock dibasic cryosolution. IHP (50  $\mu$ M) was included in the cryobuffers to ensure that the hybrid is in the T-conformation<sup>28</sup> and 100 mM sodium azide was included to ensure that the hybrid remains in the azidomet form. The viscosities of the glycerol/water solutions were taken from tables.<sup>47</sup> Sucrose/water cryobuffers were prepared similarly, with the viscosities also taken from tables.<sup>48</sup>

Samples for ET measurements were prepared in the dark under nitrogen. Small volumes of an aqueous stock solution of the azidomet hybrid were added to the anaerobic cryobuffer with a gastight syringe. Protein concentrations were determined spectrophotometrically using a diode array spectrophotometer (Agilent):  $\in$  (424 nm) = 166 M<sup>-1</sup>cm<sup>-1</sup> for the [ $\alpha$ (Zn),  $\beta$ (FeCO)] derivative.<sup>49</sup>

## Laser Flash Photolysis

In a simple photocycle, laser-flash excitation of the Zn porphyrin (ZnP) to its lowest excited triplet state (**T**) initiates ET to the Fe<sup>3+</sup> with an ET rate constant, k<sub>t</sub>; this forms the charge-separated intermediate complex (**I**) which subsequently returns to the ground state with ET rate constant, k<sub>b</sub>, Scheme 1. In the absence of ET the <sup>3</sup>ZnP decays with the intrinsic decay rate constant, k<sub>D</sub>. **T** decays exponentially with decay constant, k<sub>p</sub> = k<sub>t</sub> + k<sub>D</sub>; **I** appears with rate constant k<sub>b</sub> and disappears with k<sub>p</sub>. Flash photolysis measurements were made with an LKS60 photolysis unit (Applied Photophysics) equipped with a pulsed Xe arc lamp (75 W). Samples were excited with pulses from the second harmonic (532 nm, > 60 mJ, 10 ns) of a Q-switched Nd:YAG laser (Continuum). Triplet decays were monitored as absorbance changes at 475nm; absorbance changes of the ET intermediate, **I**, were monitored at ~ 440nm, the triplet/ground isobestic. Typically 10-25 transients were averaged for triplet decays and 100 transients for

intermediate signals. For initial analyses triplet decays and the progress curves for I were fit to the equations for simple decay to of T to ground through the intermediate,  $T \Rightarrow I \Rightarrow G$ ,

$$\mathbf{T}(\mathbf{t}) = \mathbf{T}_0 \mathbf{e}^{\mathbf{k}_p \mathbf{t}} \tag{1}$$

$$\mathbf{I}(t) = \mathbf{T}_0 \frac{\mathbf{k}_t}{\mathbf{k}_b \mathbf{k}_p} (\mathbf{e}^{\mathbf{k}_b \mathbf{t}} \mathbf{e}^{\mathbf{k}_b \mathbf{t}})$$
(2)

 $(k_p = k_t + k_D)$  using either Sigma Plot or lab-written routines.

#### **Kinetic Modeling**

For comparison of experiment and simulations with the KD model, presented in Results and illustrated in Scheme 2, we either used the solution to the differential equations for Scheme 2 that were presented previously<sup>6</sup> and are repeated in Supporting Information., or generated progress curves for the electron-transfer intermediate (I) and its component species (I<sup>R</sup> and I<sup>S</sup>) by constructing the KD kinetic model of Scheme 2 with the Simulation Control Program (ScoP, version 3.51) available from Simulation Resources, Inc; the overall progress curves are identical.

## RESULTS

## I. Kinetic-Dynamic (KD) Model for Conformational Control of ET

**Conformational Kinetics**—Consider a protein-protein ET complex that functions on an idealized energy landscape that has many (N<sup>S</sup>) poorly/non-reactive conformations,  $S_i$ ; and a smaller number (N<sup>R</sup>) of highly reactive ones,  $R_j$ , (Fig 1). The conformations are presumed to partition into two populations because the minority of ET-optimized  $R_j$  conformations are distinguished by a shorter ET distance<sup>24,50-54</sup> and/or the presence of effective pathways created by specific H-bonds;<sup>55</sup> together, such effects would introduce a 'conformational reactivity filter'.

The kinetic equations for conformational interconversion between  $R_j$  and  $S_i$  can be written in terms of conformational conversion rate constants,  $k_{ii}^u$ ,  $k_{ij}^d$ ,

$$\frac{\mathrm{d}R_j}{\mathrm{dt}} = \sum_i k_{ji}^{\mathrm{u}} S_i \sum_i k_{ij}^{\mathrm{d}} R_j \quad \frac{\mathrm{d}S_i}{\mathrm{dt}} = \sum_j k_{ij}^{\mathrm{d}} R_j \sum_j k_{ji}^{\mathrm{u}} S_i \tag{3}$$

In applying these equations and linking them to an ET photocycle we shall take the members of each population to be equivalent by constraining them to have the same conformational energy, the same *average* rate constant for conversion to members of the other population, and the same rate constants for forward and back ET.<sup>56</sup> This leads to landscapes such as the simple one illustrated by Fig 3A for N<sup>R</sup> = 1, N<sup>S</sup> = 7; an ET complex would be characterized by three such sheets, one for each of the excitation/redox states:  $k = \mathbf{T}$  (photoexcited triplet); I (ET intermediate); G (ground).

By summing Eq 3 (*right*) over *j* and Eq 3 (*left*) over *i*, defining total R and S populations as  $S = \Sigma S_i$ ,  $R = \Sigma R_j$ , and inserting *R*, *S* into Eqs 3, these become differential equations for the *R* and *S* populations,

$$\frac{\mathrm{d}R}{\mathrm{d}t} = \frac{\mathrm{d}S}{\mathrm{d}t} = \mathbf{k}_{\mathrm{u}} S \, \mathbf{k}_{\mathrm{d}} R \tag{4}$$

upon use of the constraint that each conformation within a population has the same *average* rate constants,  $k_d$  and  $k_u$ , for conversion to the other,

$$\mathbf{k}_{\mathrm{d}} \equiv \mathbf{N}^{\mathrm{S}} \overline{\mathbf{k}^{\mathrm{d}}} \equiv \sum_{i} \mathbf{k}_{ij}^{\mathrm{d}} \quad \mathbf{k}_{\mathrm{u}} \equiv \mathbf{N}^{\mathrm{R}} \overline{\mathbf{k}^{\mathrm{u}}} \equiv \sum_{j} \mathbf{k}_{ji}^{\mathrm{u}}$$
(4a)

These equations map the ensemble of interconverting  $R_j$  and  $S_i$  configurations of state k onto the simple 2-D energy surface for two substate populations, R and S, Fig 3B. The equilibrium ratio of S configurations to R becomes:

$$K^{s} = \frac{k_{d}}{k_{u}} = \left[\frac{N^{S}}{N^{R}}\right] e^{-(E^{S} - E^{R})/RT} = e^{-\Delta A^{RS}/RT} \ge 1$$
(5)

where  $E^S$  and  $E^R$  are the energies of the *S* and *R* configurations, R the ideal gas constant, and T the temperature. Eq 5 further recognizes that the ratio,  $N^S/N^R$ , corresponds to a configurational entropy difference:  $\Delta S^{RS} = Rln(N^S/N^R)$ , and defines a Helmholtz free-energy difference between populations,  $\Delta A^{RS} = \Delta E^{RS} - T\Delta S^{RS}$ , which typically will differ little from the Gibbs free-energy difference:  $\Delta G^{RS} \approx \Delta A^{RS}$ . With the appropriate choice of conformational parameters this surface can describe the full range of energy landscapes and full spectrum of ET-coupled dynamics.

**Conformationally coupled ET**—As described in the introduction, to explicitly generate equations for conformationally coupled ET, we assume that ET rate constants are the same in each member of a population, but differ between populations because they exhibit different ET pathways.<sup>24,55</sup> The [T<sup>R</sup>, T<sup>S</sup>] populations are assigned ET rate constants [ $k_t^R$ ,  $k_t^S$ ]; the [I<sup>R</sup>, I<sup>S</sup>] population are assigned rate constants [ $k_b^R$ ,  $k_b^S$ ]. A possible coupling between excitation/redox state and conformational dynamics is allowed for by allowing the interconversion rate constants to depend on the state: [ $k_{dT}$ ,  $k_{uT}$ ] and [ $k_{dI}$ ,  $k_{uI}$ ]. The resulting kinetic-dynamic (KD) scheme for the photocycle (Scheme 2) incorporates interconversions between reactive (*R*) and slow (*S*) conformational populations in each of the three excitation/redox states: **G**, **T**, and **I**; the corresponding differential equations are:

$$\frac{\mathrm{d}\mathrm{T}^{\mathrm{s}}}{\mathrm{d}\mathrm{t}} = (\mathrm{k}_{\mathrm{t}}^{\mathrm{s}} + \mathrm{k}_{\mathrm{D}} + \mathrm{k}_{\mathrm{uT}})\mathrm{T}^{\mathrm{s}} + \mathrm{k}_{\mathrm{dT}}\mathrm{T}^{\mathrm{R}} \tag{6}$$

$$\frac{\mathrm{d}T^{\mathrm{R}}}{\mathrm{d}t} = (k_{\mathrm{t}}^{\mathrm{R}} + k_{\mathrm{D}} + k_{\mathrm{dT}})T^{\mathrm{R}} + k_{\mathrm{uT}}T^{\mathrm{S}}$$
<sup>(7)</sup>

$$\frac{\mathrm{d}I^{S}}{\mathrm{d}t} = (k_{\mathrm{uI}} + k_{\mathrm{b}}^{S})I^{S} + k_{\mathrm{dI}}I^{R} + k_{\mathrm{t}}^{S}T^{S}$$
(8)

$$\frac{dI^{R}}{dt} = k_{ul}I^{S}(k_{dl} + k_{b}^{R})I^{R} + k_{t}^{R}T^{R}$$
(9)

with  $k_D$  being the triplet decay constant in the absence of ET.

The solutions to Eqs 6-9 were given earlier<sup>6</sup> and for convenience are repeated with the current notation as Supporting Information. The decay curve of the photo-generated triplet ( $\mathbf{T} = T^{S} + T^{R}$ ) is a biexponential:

$$T(t) = a_1 e^{-ft} + a_2 e^{-gt}$$
(10a)

The associated progress curve of the ET intermediate ( $\mathbf{I} = \mathbf{I}^{S} + \mathbf{I}^{R}$ ) is a sum of four exponentials in which two of the rate constants correspond to the triplet-decay rate constants (*f* and *g*) and two correspond to back rate constants (*m* and *n*).

$$\mathbf{I}(t) = c_1 e^{-ft} + c_2 e^{-gt} + c_3 e^{-mt} + c_4 e^{-nt}$$
(10b)

The progress curve for **T** can be decomposed into the individual curves for the conformational species,  $T^R$  and  $T^S$ ; likewise, the progress curve for **I** can be decomposed into those for  $I^R$  and  $I^S$ . Alternatively, the function I(t) can be decomposed into at most three kinetic phases of the form of Eq 2, although these do not, in general, correspond to the progress curves of an individual species.

The four rate constants of Eqs 10 (*f*, *g*, *m*, and *n*) and the coefficients ( $a_x$  and  $c_y$ ) of their exponentials are functions of both the four microscopic ET rate constants ( $k_t^S$ ,  $k_t^R$ ,  $k_b^S$ , and  $k_b^R$ ) and the constants which describe the dynamics of the system.(Supporting Information) Each sheet of the energy surface is characterized by two independent dynamic parameters, which we choose to be  $k_d$  and  $K^S = k_d/k_u$ , defining  $k_u = k_d/K^S$ . For simplicity, for the remainder of this report we take the conformational rates and equilibrium constants to be independent of excitation/redox state ( $k_{dT} = k_{dI} = k_d$ ,  $k_{uT} = k_{uI} = k_u$ , and  $K^S_T = K^S_I = K^S$ ) in which case the model employs only two independent conformational parameters,  $k_d$  and  $K^S$ .

We finally note that while the KD model as embodied in Eqs 6-9 is defined in terms of the temporal evolution of electronic/conformational populations or 'species' – the *S* and *R* conformational populations of the **T** and **I** (and **G**) electronic states – Fig 1 was inspired by the complementary picture of a conformationally controlled competition between different ET 'routes', where a route comprises the sequence of excitation/redox/conformational states through which the system passes during an ET photocycle. As we discuss below, the limiting cases and important regimes of Eqs 6-9 display four ET routes of particular interest (Fig 4).

**Detectability**—Contributions of the two component phases to an experimental triplet decay curve are simply determined by their relative weights (which are not necessarily identical to their statistical populations). However, the contribution of a component kinetic phases to an

experimental progress curve for **I**, which is best thought of as its maximum amplitude, does not simply correspond to its weight, for it is strongly influenced by the *relative* values of the rate constants for the appearance and disappearance of the phase. This is discussed in detail in the Supporting Material, and is of great importance in determining the appearance of experimental curves, as noted below.

## **II. Modulating Interconversion Dynamics**

The final link between the KD model and a system in which ET kinetics are coupled to conformational interconversion is to fix the function that relates interconversion rates and solution properties. When viscosity is the important solution variable we may take changes in viscosity as affecting  $k_u$  and  $k_d$  equally, leaving  $K^S$  invariant, and write the  $k_u$  and  $k_d$  in the Kramers-like form,<sup>57-59</sup>

$$\mathbf{k}_{\mathrm{d},\mathrm{u}}(\eta) = \mathbf{k}_{\mathrm{d},\mathrm{u}}^0 / \eta^\alpha \quad \alpha \le 1 \tag{11}$$

although alternate forms are available.<sup>60,61</sup> If osmotic effects are important the  $k_{d,u}^0$  become functions of the molality of the solute. They also are functions of the number of waters released during the conformational changes, so osmotic effects can change the two rate constants differently, and hence change  $K^S$ . We show below that Eq 11 is applicable to data for the Hb hybrids.

## III. Limits and Regimes

If the properties of the KD model of Eqs 10, 11 are examined over ranges of the controlling parameters, one finds a quite complicated 'phase diagram' comprised of distinct regimes (parameter ranges) of dynamical behavior. Considering the forward and back ET processes individually, one recognizes known regimes such as the fast-exchange (*FE*) and slow-exchange (*SE*) limits, as well as the gating (*G*) regime for forward ET. When examining the complete photocycle, limiting behaviors also appear, but novel ones also arise. To facilitate analysis of the experimental results for the hybrids we briefly describe the two limiting photocycle regimes, as well as one in which back ET is initiated within a non-equilibrium ensemble of conformers as envisaged in Fig 1. It is important to emphasize that such limits/regimes are not simply characteristics of an ET system, but represent the behavior of a system whose coupling between ET and conformational dynamics has been 'tuned' by a particular set of solution conditions. In general, variations of temperature/viscosity/osmotic pressure/pH/ionic strength can shift the system smoothly from one regime to another along a continuum of behaviors, while mutations can shift it discontinuously.

(i) Fast Exchange Photocycle—In the *fast-exchange* (*FE*) limit for forward ET,  $[k_u, k_d] \gg [k_t^R, k_t^S]$ , the *S* and *R* populations interconvert so rapidly that equilibrium populations are maintained at all times. The decay of **T** is described by a single forward ET rate constant  $k_{teq}$ , with observed decay constant  $k_{obs}^{eq} = k_D + k_{teq}$ , where  $k_{teq}$  is a weighted average of  $k_t^S$  and  $k_t^R$ ,

$$k_{teq} = F^{S} k_{t}^{S} + F^{R} k_{t}^{R}; \quad F^{S} = K^{S} / (K^{S} + 1); \quad F^{R} = 1F^{S}$$
(12)

The weighting factors (F<sup>S</sup> and F<sup>R</sup>) correspond to the equilibrium fractional populations of the *S* and *R* substates. In the *FE* limit for back ET, defined by  $[k_u, k_d] \gg [k_b^R, k_b^S]$ , the I<sup>R</sup> and I<sup>S</sup> populations also interconvert so rapidly that an equilibrium ratio of these conformers is

maintained throughout the progress curve for **I** and back ET also is described by a weighted average of the back ET rates for the individual conformational ensembles,  $k_b = k_{bea}$ ,

$$k_{beq} = F^S k_b^S + F^R k_b^R \tag{13}$$

As a result, for the *FE* photocycle, in which both forward and back ET are in the *FE* limit, the progress curve for **I** is described by a single kinetic phase (Eq 2) with average ET rate constants,  $k_{teq}$  and  $k_{beq}$ .

The 'conformationally coupled'<sup>62</sup> photocycle has the same type of dependence on ET driving force as if ET took place on a simple landscape, as  $k_{teq}$  and  $k_{beq}$  are proportional to microscopic ET rate constants. However, the photocycle has observed ET rate constants that are independent of the individual conformational rate constants. Thus, within the limit, a change in viscosity would have no effect on the photocycle and would not reveal the presence of coupling between ET and conformational dynamics.

The dominant route in the *FE* photocycle depends on the explicit values of the rate constants. We show below that for the Hb hybrid at low viscosity the microscopic rate constants correspond to a well-defined majority '*FE* route' for ET (Fig 4):  $T^S \Rightarrow T^R \Rightarrow I^R \Rightarrow I^S \Rightarrow G$ .

(ii) Slow Exchange Photocycle—The opposite, *slow-exchange* (*SE*), limits for forward and back ET occur respectively if  $[k_u, k_d] \ll [k_t^R, k_t^S]$  and  $[k_u, k_d] \ll [k_b^R, k_b^S]$ . If both forward and back processes are in *SE* the *R* and *S* populations react independently, with their own forward and back ET rate constants, and with weights/'yields' of the two phases of T(t) and I (t) simply given by the equilibrium populations,  $[F^S, F^R]$ . The photo-excited *R* hybrids follow the '*R*-route' ( $T^R \Rightarrow I^R \Rightarrow G$ ); photo-excited *S* hybrids follow the '*S*-route' ( $T^S \Rightarrow I^S \Rightarrow G$ ) (Fig 4). As with the *FE* photocycle, the ET rate constants in the *SE* photocycle depend on ET driving force and, of course, are independent of viscosity.

(iii) Photocycle with a Non-Equilibrium Distribution of I—Situations of this type arise when forward ET is in the *gating* (*G*) regime, which occurs when  $[k_t^S \ll k_u < k_d \ll k_t^R]$ . The observed decay rate of **T** then is determined by the rate of conformational conversion,  $k_{obs} = k_u + k_D$ , not an ET rate. As a result, this rate constant varies with addition of viscogen, not with changes in ET driving force. In principle, the decay of **T** actually is biphasic, with the small fraction F<sup>R</sup> forming T<sup>R</sup> upon laser excitation and promptly decaying with the viscosity independent rate constant,  $k_{obs} = k_t^R + k_D$ . However, for K<sup>S</sup>  $\gg$  1 ( $k_u \ll k_d$ ), F<sup>R</sup>  $\ll$  1 and the contribution of this 'prompt' or 'burst' phase is small.

When forward ET is in the *G* regime the major route for forward ET produces the I<sup>R</sup> conformer, even though the *R* conformation is highly disfavored by a large value of K<sup>S</sup>. As a result, the back reaction begins in a non-equilibrium state, as envisaged in Fig 1. If the conformational deactivation of I<sup>R</sup> is slower than back ET from I<sup>R</sup> ( $k_b^R \gg k_d$ ), then I<sup>R</sup> undergoes I<sup>R</sup>  $\Rightarrow$  G back ET without conformational deactivation to I<sup>S</sup> and the observed back ET rate constant will be  $k_b^R$ . We call this back ET regime 'anti-gated (*A*)' because the relative importance of the I<sup>R</sup>  $\Rightarrow$ G process is governed by the value of  $k_d$ , a conformational rate constant, yet this back ET step occurs with an ET rate constant which varies with ET driving force, unlike gated (*G*) forward ET.

With the gating and anti-gating inequalities both satisfied, the majority of the complexes follows the route,  $T^S \Rightarrow T^R \Rightarrow I^R \Rightarrow G$  (*G/A* route in Fig 4), which precisely corresponds to the photocycle schematized in Fig 1: the forward ET process occurs from an initial equilibrium

ensemble, while the reverse ET process occurs from a non-equilibrium ensemble, with conformational dynamics controlling both ET processes. As noted above, if the *R* population is not vanishingly small there would be a second, minority route: the fraction  $F^R$  of complexes that are photoexcited to the  $T^R$  state would undergo 'prompt' ET along the *R*-route (Fig 4):  $T^R \Rightarrow I^R \Rightarrow G$ .

## IV. Modeling Experimental Data for the $[\alpha(Zn), \beta(FeN_3)]$ hybrids

Figure 5A shows representative progress curves for the appearance and disappearance of the ET intermediate, **I**, as the range of viscosities,  $\eta = 1 \rightarrow 15$  cP is traversed by addition of either glycerol or sucrose at 20° C. The most apparent change in the curves with increasing  $\eta$  is an ~ 60% decrease in the maximum absorbance of **I** ( $\Delta A_{max}$ ). In addition, a concomitant shift of  $\Delta A_{max}$  from 7.5 ms to 5.1 ms reflects an apparent increase in the rate of appearance of **I** and a decrease in the rate of its disappearance.

Attempts to fit the viscosity dependence of the progress curves for **T** and **I** to the equations for the simple photocycle of Scheme 1, Eqs 1,2 (Fig S1), with viscosity-independent value values for  $[k_d, k_b]$  fail (Fig S1). The calculated and experimental decays of **I** differ substantially even at  $\eta = 1$  cP, with the discrepancy being biggest in the appearance of **I**, and the discrepancy gets progressively worse as  $\eta$  increases. This behavior implies the type of coupling between conformational dynamics and ET embodies in the KD model. Application of this model to the hybrid dataset involves determination of the microscopic parameters in Scheme 2 optimize the description of the progress curves for both **T** and for **I**.

There are seven KD parameters in Scheme 2, but the intrinsic decay constant ( $k_D = 50\pm5 \text{ s}^{-1}$ ) is known from independent measurements of the triplet decay for the [ $\alpha$ (Zn),  $\beta$ (Fe<sup>2+</sup>)] hybrid]. <sup>63</sup> This leaves four microscopic ET rate constants ( $k_t^R, k_b^R, k_t^S, k_b^S$ ) and two conformational variables, K<sup>S</sup> and  $k_d$ , to be determined. For simplicity we take the conformational rates and equilibrium constants to be independent of excitation/redox state, which means that each trace is described by two independent conformational variables: the viscosity-invariant conformational equilibrium constant, K<sup>S</sup>;  $k_d$  for that trace/viscosity. Upon determination of a  $k_d$  for each viscosity/trace, a fit of this set of parameters to Eq 12 provides a global description of the viscosity dependence in terms of three conformational parameters: K<sup>S</sup>,  $k_d^0$  and  $\alpha$ .

Direct determination of these parameters by a direct, global least-squares fit to a set of progress curves for **T** and **I** generated over a range of viscosities would not be instructive and might not be wholly credible; fortunately, it is not necessary. Instead, we have devised a rapidly converging, iterative procedure that begins by approximating the microscopic ET constants and generating four constraints; this reduces the problem to a determination of K<sup>S</sup> for the system,  $k_d$  at each viscosity and their subsequent use to parameterize Eq 11.

We began the analysis by first fitting the progress curves to the kinetic equations for a "relaxed photocycle (RP)", defined through a rewriting of Eqs 1,2:

$$T(t) = T_0 e^{-k_{obs}t}$$
(14)

$$I(t) = T_0 \frac{k_t}{k_b - k_p} (e^{-k_p t} - e^{-k_b t})$$
(15)

with  $k_{obs} = k_t + k_D$ . These differ from Eqs 1, 2 in that the triplet decay constant,  $k_{obs}$ , is not required to equal a rate constant associated with **I**, and is replaced by an additional fitting parameter associated with an individual trace,  $k_p$ . This approach, which we have used before, 5,27,32,34 provides a useful, heuristic description of the progress curves, as shown by the moderately successful overlay of calculated and experimental traces in Fig 5A. There is, however, no kinetic scheme which gives rise to these heuristic equations.

Fig 5B plots the  $\eta$  dependencies of the parameters,  $[k_{obs}, k_b, k_p]$ , obtained in this fashion. It shows that  $[k_{obs}, k_p]$  *decrease* modestly, while  $k_b$  *increases* two-fold as  $\eta$  is increased with either viscogen. When the ET rate constants for the datasets obtained through the addition of glycerol and of sucrose are plotted versus viscosity and molality (Fig S2), only those plotted against viscosity show a correlation for the two solutes, consistent with our earlier suggestion<sup>5,27</sup> that the dynamics are responding to viscosity rather than osmotic effects. <sup>64-66</sup> Thus, the RP model of the photocycle at fixed temperature confirms that the forward and reverse ET processes respond differentially to changes in conformational dynamics caused by addition of viscogen, even though there is no kinetic mechanism which would produce the equations that define the RP model, Eqs 14,15.

Four initial constraints on the KD model were derived from the plots of  $k_t (k_t = k_{obs} - k_D)$  and  $k_b$  (Fig 5B), by provisionally assuming that the behaviors at low and high viscosity approach the *FE* and *SE* limits, respectively. The values of  $[k_t, k_b]$  at the highest viscosity employed ( $\eta = 15 \text{ cP}$ ) were extrapolated (by eye) to yield approximations for two of the four microscopic ET rate constants:  $k_t^S = 5s^{-1}$ ;  $k_b^R = 500s^{-1}$ . The values of  $[k_t, k_b]$  at the lowest viscosity ( $\eta = 1 \text{ cP}$ ) were taken as estimates of the *FE* rate constants:  $k_{teq} = 18s^{-1}$ ;  $k_{beq} = 230s^{-1}$ . Eqs 12,13 were then inverted to obtain expressions for the two remaining microscopic ET rate constants ( $k_t^R$  and  $k_b^S$ ) as functions which incorporate the four constraints as parameters ( $k_t^S, k_b^R$ ,  $k_{teq}, k_{beq}$ ) and have K<sup>S</sup> as the *one* independent variable.

If a value of  $K^S$  is specified for the dataset of traces collected over a range of viscosities, the optimized values of  $k_d$  for each progress curve of **I** then completes a set of microscopic constants that can be input into the KD model and used to simulate this curve either through direct use of Eq 10 or with the SCoP package. At this initial stage, we used as "goodness-of-fit" criteria, the time at which the concentration of **I** was maximal ( $\tau_{max}$ ) and the  $\Delta A_{max}$  at that time. These features were readily reproduced with a globally optimized K<sup>S</sup> for the set of traces and individual values of  $k_d$  for each trace; these latter then were used to parameterize Eq 11. The values for the constraints and the resulting initial set of KD parameter are compiled in Table 1.

Next, we refined these values by iterating the constraints. The hybrid cannot be poised completely into the *FE* limit at  $\eta = 1$  cP, because if it were then Eqs 1, 2 would be adequate to describe the at  $\eta = 1$  cP progress curves, but as stated above this is not so. Therefore, the rate constants obtained by fitting the progress curves of **T** and **I** at  $\eta = 1$  cP to Eqs 14, 15 provide only a lower bound to  $k_{teq}$  (18s<sup>-1</sup>) and an upper bound to  $k_{beq}$  (230s<sup>-1</sup>). We refined the constraints by systematically increasing  $k_{teq}$  and decreasing  $k_{beq}$  while further varying the extrapolated value for  $k_b^R$ . At this final stage, goodness-of-fit was determined by the ability of the simulations to accurately reproduce the complete timecourses of **I**.

This iterative exploration of parameter space quickly converged to a set of four values for the constraints, none of which differ from the original set of four by more than ~10% (Table 1). The progress curves calculated for **I** with the final set of KD parameters are in excellent agreement with the experimental curves, as shown in Fig 6A-D for four different viscosities. In particular, the early-time 'rise' of the intermediate signal is improved by the iteration (Fig S3). Fig 6E shows how the traces for **I** evolve as the dynamics progressively traverse the full

range from *FE* to *SE*, relating them to experiment by including the traces for the lowest and highest viscosities studied. This figure clearly shows that at  $\eta = 1$  cP the system does *not* approach *FE*, while at  $\eta = 15$  cP it is rather near to *SE*.

**Decomposition and Analysis of Simulated Hybrid Progress Curves**—In addition to showing the experimental and final calculated progress curves, Fig 6 gives decompositions of I(t) into the progress curves of the I<sup>R</sup> and I<sup>S</sup> components for  $\eta = 1, 1.7, 6.3$  and 12 cP; Fig 6E shows the analogous decomposition for the extrapolated *SE* limit. These decompositions show that the major consequence of increasing  $\eta$  is the suppression of the I<sup>S</sup> phase as a slowing of conformational conversions suppresses the  $S \rightarrow R$  transition. Other consequences include: an increasing importance and sharpening of the I<sup>R</sup> phase, with  $\Delta A_{max}$  increasing slightly and shifting to earlier time; the appearance of a 'tail' associated with I<sup>S</sup> becomes visible at longer times. The increased importance of I<sup>R</sup> relative to I<sup>S</sup> corroborates our intuitive expectation that I is largely formed as the reactive conformer (I<sup>R</sup>) and that increasing the viscosity slows the conversion to the less-reactive conformer (I<sup>S</sup>) thereby increasing the contribution of a fast return to ground within the I<sup>R</sup> conformation.

The fact that the fit to the experimental progress curves has significant contributions from two kinetic phases with quite different time courses explains why even the relaxed photocycle model does not adequately describe the results, most especially failing to match the curves for I at elevated viscosities (Fig 5A): in *any* 'simple' photocycle I(t) is necessarily a single kinetic phase. The similarity of the forward and back ET rate constants for the *R* population (Table 1) is expected, as the driving forces for the two reactions are similar.<sup>67</sup>

The values of the forward ET rate constants differ from each other by approximately two orders of magnitude (Table 1,  $k_t^R = 550 \text{ s}^{-1}$  and  $k_t^S = 5 \text{ s}^{-1}$ ), while the back ET rate constants differ only by a factor of ~2.5 ( $k_b^R = 517 \text{ s}^{-1}$  and  $k_b^S = 200 \text{ s}^{-1}$ ); further, the forward and back ET rate constants for the *S* population (Table 1) differ appreciably. We speculate that these difference reflects a limitation of the two-population approach. The large difference in the forward rates may occur because the system is distributed over all phase space prior to excitation, with the majority in a slow configuration, while the smaller difference in the back ET rate constants suggests that the system samples a much more limited volume of phase space before returning to the ground state by back ET. In addition, these effects may reflect conformational averaging on a more rapid timescale that differs between the populations.

## DISCUSSION

The Hb hybrids provide a convenient system with which to explore dynamic control of an ET photocycle within a pre-docked protein-protein complex, and we have examined how the hybrid photocycle responds to changes in conformational dynamics induced by changes in viscosity at fixed temperature. This photocycle exhibits 'differential dynamics' in the sense that a fit of the progress curves for **T** and **I** to the 'relaxed photocycle' of Eqs 14, 15 showed that as  $\eta$  increases and conformational dynamics slow, the apparent forward ET rate constant decreases while the apparent back ET rate constant increases, Fig 5B.<sup>5,27</sup> However, the progress curves for **I** in fact are not well fit by this model, Fig 5A, and in any case the model is inadequate in that it does not explicitly incorporate conformational processes and represents no underlying kinetic mechanism. As a result we have developed the simple 'KD' model for the ET photocycle of a system in which ET is coupled to conformational dynamics and applied it to the viscosity dependence of the [ $\alpha(Zn)$ ,  $\beta(Fe^{3+}N_3^-)$ ] hybrid photocycle.

The model assumes the complex exists as multiple interconverting conformations. These partition into two populations,  $N^R$  reactive conformations, *R* and  $N^S$  non(poorly)-reactive *S* (Fig 3B) because the *R* conformations are ET-optimized by a shorter ET distance and/or the

presence of effective pathways created by the presence of specific H-bonds. Taking the conformations within a population to be equivalent imposes the constraint that the members of each population have the same *average* rate constant for conversion to the other along the single (effective) conformational coordinate (Eq 4a). Fig 3A illustrates the underlying energy landscape for the simple case,  $N^R = 1$ ,  $N^S = 7$ . The solutions of the resulting differential equations for the KD model of a conformationally-coupled ET photocycle (Eq 10; Supporting Information) were presented previously and their application to forward ET discussed in detail. <sup>6</sup> However, their general applicability was not recognized, and their application to a full ET photocycle was at best rudimentary.

Examination of the solutions showed they can describe the full spectrum of ET-coupled dynamics through an appropriate choice of the difference in ensemble energies,  $\Delta E^{RS}$ , of the ratio of the number of members of *R* and *S* ensembles, N<sup>S</sup>/N<sup>R</sup>, and of relative values for ET and conformational kinetic rate constants. As conformational interconversion is 'turned on' the model leaves the limit (*SE*) in which dynamics play no role, either because only a single state or population is energetically accessible or because conformational changes are much slower than ET. It passes through a 'gating/anti-gating' (*G/A*) regime, in which the two conformational populations convert on a time-scale competitive with ET, arriving at a limit in which ET occurs within a conformationally averaged state (*FE*) in which the timecourses of both **T** and **I** are monophasic. The model further can describe the 'Dynamic Docking' regime, in which phase-space control of ET in a weakly-bound complex is engendered by a landscape with numerous conformations of similar energy, only a few of which are ET-active (N<sup>S</sup>  $\gg$  N<sup>R</sup>).

The analysis further led to the recognition that when treating intermediate dynamic regimes, it is instructive to consider not merely the timecourses of the states involved, **T** and **I**, but also the *routes* by which a complex evolves from its initial triplet conformation to the ET intermediate, and finally back to the ground conformation, a route being defined as the sequence of excitation/redox/conformational states traversed during this process. Fig 4 illustrates routes associated with the *FE* and *SE* limits, but it also illustrates a particularly interesting route traversed when the conformational and ET rate processes compete. Under conditions when the forward ET reaction is gated and the back ET process 'anti-gated', **I** can follow the *GA* route,  $T^S \Rightarrow T^R \Rightarrow I^R \Rightarrow G$ , which corresponds closely to the intuitive idea of a photocycle in which the forward reaction occurs from an equilibrium ensemble of conformations but the back reaction occurs from a non-equilibrium reactive ensemble, Fig 1.

Such consideration of ET routes highlight the fact that ET reactions in conformationally mobile systems provide a 'clock' against which the rapidity of a dynamic event is measured, and that each 'branch point' in a route represents a competition between an ET and conformational process. In addition to direct deactivation to ground, the triplet populations of Scheme 1 each have two channels for reaction, one ET and one conformational,  $[T^S \leftarrow T^R \Rightarrow I^R]$ ,  $[T^R \leftarrow$  $T^S \Rightarrow I^S$ ]; each of the populations of **I** has only two,  $[I^S \leftarrow I^R \Rightarrow G]$ ,  $[I^R \leftarrow I^S \Rightarrow G]$ . The partition of each state into the alternate reaction channel is determined by the relative values of the respective competing ET and conformational rate constants. As there are four ET rate constants in the two-population KD model, Scheme 1, all of which can in principle differ, there are four 'clocks' and the photocycle overall should be sensitive to motions on the four timescales set by them. As the rate constants for the effective conformational coordinate are progressively slowed with increasing viscosity the partitions at the four branch points should progressively respond, starting with the branch point that has the largest ET rate constant. Regarding the two T populations, one must further note that if an ET rate constant is not much larger than k<sub>D</sub>, then direct deactivation provides the clock against which conformational events must compete.

## The KD Model and the ET Photocycle of the Hb hybrids

As a first application of the KD model we analyze the viscosity dependence at 20°C of the ET photocycle of the pre-docked [ $\alpha$ (Zn),  $\beta$ (Fe<sup>3+</sup>N<sub>3</sub><sup>-</sup>)] hybrid, Fig 6. Measurements performed with glycerol and sucrose as solutes give equivalent results when correlated by solution viscosity, not solute molality, which supports the earlier suggestion that solutes indeed exert their influence on viscosity, rather than through osmotic effects.<sup>68-70</sup> A robust, iterative procedure for fitting the model to experiment has been devised and simulations of the hybrid ET photocycle based on the resulting ET and conformational parameters reproduce experiment quite well across the range of viscosities (Fig 6;Table 1).

The timecourses for the hybrid at 20°C are well described with the model's assumption of two conformational populations, the minority ensemble being more ET active. Comparison of the experimental and calculated progress curves for I shows that the trace for  $\eta = 1$  cP deviates appreciably from that for the FE limit, Fig 6E. Beginning with  $\eta = 1$ , and through most of the viscosity range, multiple routes contribute to the timecourse for I and this precludes simple analysis; we do note significant involvement of a route involving gated forward ET and conformational coupling in the back. By the highest viscosity employed,  $\eta = 15$  cP, the simulated trace (Fig 6E) suggests that the photocycle is close to the SE limit, but in fact there is still a significant contribution from the G/A route (Table S3). The fraction of T<sup>S</sup> that follows the S route produces  $I^S$ , while the remainder of the  $T^S$  population that follows the G/A route produces I<sup>R</sup>. The small amount of T<sup>R</sup> formed directly by flash excitation, F<sup>R</sup>, undergoes 'prompt' ET along the R route; this process, which is similar to that for  $T^{R}$  in the SE limit, does not contribute significantly to the triplet decay, but is important for the progress curve of I. The total fraction of back ET that proceeds along routes that pass through I<sup>R</sup> is given by the sum of the contributions from 'prompt' ET along the R route and from the G/A route ( $\phi^{R}$  +  $\phi^{G/A}$ ). This increases from 8% in the *FE* limit to 49% at  $\eta = 15$  cP, and then decreases to  $\phi^{R}$ ~ 24% as the contribution of the G/A-route falls to zero ( $k_u \Rightarrow 0$ ) and the SE limit is reached, Table S3.

## Nature of the ET-coupled Motions in the $[\alpha(Zn), \beta(Fe^{3+}N_3)]$ Hybrid

The KD analysis of the hybrid ET photocycle shows that the ET-coupled motions are fairly slow,  $k_d < 10^3 s^{-1}$ ; what are they? Hb is the archetypical allosteric protein, and Hb allosteric conformational rearrangements occur on the microsecond – second timescales.<sup>71</sup> Our current hypothesis is that ET is coupled to the large-scale allosteric motions which play an integral role in the O<sub>2</sub>-binding mechanism of the Hb tetramer.<sup>43</sup> In recent years it has been shown that the low-affinity 'T' and the high affinity 'R' structures of Hb are in fact ensembles of structures, and within each there is a near-continuous variation in structure.  $^{40,41,72}$  The [ $\alpha$ (Zn),  $\beta$ (Fe<sup>3+</sup>L)] hybrid behaves as a diliganded T tetramer, <sup>28,35,73,74</sup> One possibility is that the ETcoupled motions involve conversions between ensembles, and in fact T  $\rightarrow$ R conversion rates for diliganded hybrids fall into the range of interest here.<sup>75,76</sup> Alternatively the conversion could be within the T ensemble. In either case, an accompanying change in the contacts at the interface between the  $[\alpha_1(Zn), \beta_2(Fe^{+3}N_3)]$  ET partners upon conversion between S and R configurations would lead to the differing ET reactivities of S and R. Perhaps these changes might involve the formation and collapse of an inter-subunit network of H-bonded waters, as has been proposed for inter-protein ET in other systems.<sup>52</sup> Future experiments will test these ideas through the study of Hb hybrids in the presence of allosteric effectors, 77-79 and of hybrids prepared with mutations at the  $\alpha_1/\beta_2$  interface, 80,81 both of which systematically alter the allosteric conformational equilibria/dynamics

## **Benefits of a Photocycle**

Although numerous studies have described conformational control of a 'forward' ET reaction that is initiated either by rapid mixing or by photolysis, 60,82-90 the results for the hybrid and

their analysis with the KD model illustrate the benefits of studying a photocycle when possible. The simplest one is practical and diagnostic. In the absence of conformational influences the timecourse of I necessarily is a single 'rise-and-fall' kinetic phase that is self-consistently tied to the decay of T through its decay constant, Eqs 1<sup>,</sup> 2. The simple observation either of a lack of consistency (different  $k_{obs}$ ,  $k_p$ , Eqs 14, 15) as shown for the hybrids in Fig 5B, or of the need to fit I(t) with multiple kinetic phases<sup>91</sup> implies conformational influences even before performing measurements as a function of  $\eta$ . For the hybrids, not only are multiple phases needed to describe I(t) without addition of viscogen, but the  $\eta$  dependence of the timecourse of I gives additional indications of ET coupling to conformational dynamics, including: *i*) a shift in  $\Delta A_{max}$  to shorter time and decrease in maximum absorbance change as  $\eta$  is increased; *ii*) poor fits of the actual progress curves even to the 'relaxed photocycle' model, Eqs 14, 15.

Specific analysis highlights the limited ability to characterize both the ET and conformational processes when only the kinetics for the photoinitiated 'forward' reaction are studied. When the timecourse of **T** can be studied throughout the entire dynamic range from *FE* to *SE* by variation of viscosity, and both kinetic phases of Eq 10a, can be observed, it is possible to obtain all the ET and conformational constants for the forward reaction. If, however, the timecourse of **T** is essentially monophasic over the accessible range of  $\eta$ , as is true for the hybrid, then only some of the parameters are accessible. *i*) If **T** is independent of  $\eta$  throughout the accessible range of  $\eta$ , it could be that there is no conformational coupling, but also it could be that the system is in *FE* throughout the range. In this case the measured rate constant is a conformationally weighted average of microscopic ET constants and conformational coupling is masked. *ii*) If, on the other hand, k<sub>obs</sub> for the single phase simply decrease with increasing viscosity, reaching a limiting value at high  $\eta$ , then one can only assume that the decay of **T** corresponds to the decay of **T**<sup>S</sup>, which involves a competition between deactivation to ground, ET from this state, and gated ET governed by the conformational conversion to **T**<sup>R</sup>. In this case,

$$k_{obs}(\eta) = k_{p} + k_{t}^{S} + k_{u}(\eta)$$
(16)

with  $k_u(\eta) = k_d(\eta)/K^S$ . A fit of  $k_{obs}(\eta)$  with  $k_d(\eta)$  given by Eq 11 would then give values for  $[k_t^S, k_d^0, \alpha]$ , as  $k_D$  is known. In the case of the hybrids,  $k_{obs}$  does decrease with  $\eta$  (Fig 5B), and a fit to Eq 16 gives  $[k_t^S, k_u^0, \alpha]$ ;  $k_t^S$  and  $\alpha$  are within ten percent of those from the KD fit (Table 1), while  $k_u^0$  is within a factor of two. Thus treatment with Eq 16 indeed is useful in defining the role of conformational coupling on forward ET through determination of values for  $[k_t^S, k_u^0, \alpha]$ . However, measurements of forward ET never can verify that the gating inequalities given above are satisfied, and in fact the KD analysis of the hybrid photocycle shows that the key requirement,  $[k_d \ll k_t^R]$ , is violated at low viscosities,  $k_d(1 \text{ cP}) = k_d^0 = 1017 \text{ s}^{-1} > k_t^R = 517 \text{ s}^{-1}$ ; thus the assumption leaves a false impression about the values of  $k_d^0$  and  $k_t^R$ . This treatment naturally cannot give the remaining four parameters that come from treatment of the back reaction in the KD model,  $[k_t^R, k_b^S, k_h^R, k_d^0 \text{ or } K^S]$ .

#### Applicability of the KD model

We have shown that this model can describe the full 'phase diagram' of ET-coupled dynamics through an appropriate choice of parameters, but one must consider what is lost by the restriction to a single effective conformational coordinate. Within the model the decay of **T** is biexponential under all conditions except *FE*, while the timecourse for **I** can be decomposed into at most three of the 'rise-and-fall' kinetic phases that each correspond to those of the simple photocycle of Scheme 1. If members of the *R* and *S* populations were not constrained to be equivalent, then the decay of **T** would be distributed, a characteristic of more complex models of ET.<sup>27-29,73</sup> However, if the distribution of rate constants in a population is not too broad

this difference is more apparent than real, as it would not be possible to distinguish between bi-exponential and distributed kinetics in **T** unless the experimental traces have exceptionally (and usually unreachably) high signal/noise ratios.<sup>92</sup> Similarly it would be hard to convincingly establish that an experiment demands a more complicated equation for I(t), one decomposable into four or more kinetic phases or showing distributed behavior. We therefore suggest that the two-population KD model will usefully describe the basic features of ET coupled to dynamics in most photocycles, although in some experimental situations more complex models may well be required.

## Supplementary Material

Refer to Web version on PubMed Central for supplementary material.

#### Acknowledgements

This work was supported by the National Institutes of Health (HL63203).

## References

- 1. Marcus RA, Sutin N. Biochim Biophys Acta 1985;811:265-322.
- 2. Balzani, V. Electron transfer in chemistry. Wiley-VCH: Weinheim; New York: 2001.
- 3. Hoffman BM, Ratner MR. J Am Chem Soc 1987;109:6237-6243.
- Liang Z-X, Nocek J, Huang K, Hayes RT, Kurnikov IV, Beratan DN, Hoffman BM. J Am Chem Soc 2002;124:6849–6859. [PubMed: 12059205]
- Hoffman BM, Celis LM, Cull DA, Patel AD, Seifert JL, Wheeler KE, Wang J, Yao J, Kurnikov IV, Nocek J. PNAS 2005;102:3564–3569. [PubMed: 15738411]
- Hoffman, BM.; Ratner, MA.; Wallin, SA. Advances in Chemistry Series. Johnson, MK.; King, RB.; Kurtz, DM., Jr; Kutal, C.; Norton, ML.; Scott, RA., editors. Vol. 226. American Chemical Society; Washington, D.C.: 1990. p. 125-146.
- Johnson, MK.; King, RB.; Kurtz, DM.; Kutal, C.; Norton, ML.; Scott, RA. Electron Transfer in Biology and the Solid State: Inorganic Compounds with Unusual Properties. American Chemical Society; Washington, D.C.: 1990.
- Kosek-Dygas MM, Hoffman BM, Matkowsky BJ, Nitzan A, Ratner MA, Schuss Z. J Chem Phys 1989;90:1141–1148.
- 9. Davidson VL. Acc Chem Res 2000;33:87-93. [PubMed: 10673316]
- Ivkovic-Jensen MM, Ullmann GM, Crnogorac MM, Ejdebaeck M, Young S, Hansson O, Kostic NM. Biochemistry 1999;38:1589–1597. [PubMed: 9931026]
- 11. Balabin IA, Onuchic JN. Science 2000;290:114-117. [PubMed: 11021791]
- Miyashita O, Okamura MY, Onuchic JN. Proc Natl Acad Sci U S A 2005;102:3558–3563. [PubMed: 15738426]
- Volkov AN, Worrall Jonathan AR, Holtzmann E, Ubbink M. Proc Natl Acad Sci U S A 2006;103:18945–18950. [PubMed: 17146057]
- 14. Bendall, DS. Protein Electron Transfer. BIOS Scientific Publishers; Oxford: 1996.
- 15. Kang SA, Crane BR. Proc Natl Acad Sci U S A 2005;102:15465–15470. [PubMed: 16227441]
- 16. Jeuken LJC. Biochim Biophys Acta, Bioenerg 2003;1604:67-76.
- 17. Toogood HS, Leys D, Scrutton NS. FEBS Journal 2007;274:5481–5504. [PubMed: 17941859]
- Toogood HS, van Thiel A, Basran J, Sutcliffe MJ, Scrutton NS, Leys D. J Biol Chem 2004;279:32904– 32912. [PubMed: 15159392]
- 19. Crowley PB, Ubbink M. Acc Chem Res 2003;36:723-730. [PubMed: 14567705]
- Wheeler KE, Nocek J, Cull DA, Yatsunyk LA, Rosenzweig AC, Hoffman BM. J Am Chem Soc 2007;129:3906–3917. [PubMed: 17343378]
- Liang Z-X, Kurnikov IV, Nocek JM, Mauk AG, Beratan DN, Hoffman BM. J Am Chem Soc 2004;126:2785–2798. [PubMed: 14995196]

- 22. Sumi H, Marcus RA. J Chem Phys 1986;84:4894–4914.
- Medvedev ES, Kotelnikov AI, Goryachev NS, Psikha BL, Ortega JM, Stuchebrukhov AA. Molecular Simulation 2006;32:735–750.
- 24. Kawatsu T, Beratan DN. Chem Phys 2006;326:259-269.
- 25. Shihab MS, Kubota K, Takahashi T, Ohga Y, Asano T. Chem Lett 2004;33:1542-1543.
- 26. Barzykin AV, Frantsuzov PA. J Chem Phys 2001;114:345-354.
- 27. Patel AD, Nocek JM, Hoffman BM. J Am Chem Soc 2005;127:16766-16767. [PubMed: 16316205]
- 28. Blough NV, Zemel H, Hoffman BM, Lee TCK, Gibson QH. J Am Chem Soc 1980;102:5683–5685.
- 29. Huang Y, Yonetani T, Tsuneshige A, Hoffman BM, Ackers GK. Proc Natl Acad Sci U S A 1996;93:4425–4430. [PubMed: 8633083]
- Hui HL, Kwiatkowski LD, Karasik E, Colby JE, Noble RW. Biochemistry 2004;43:7843–7850. [PubMed: 15196027]
- Samuni U, Juszczak L, Dantsker D, Khan I, Friedman AJ, Perez-Gonzalez-De-Apodaca J, Bruno S, Hui HL, Colby JE, Karasik E, Kwiatkowski LD, Mozzarelli A, Noble R, Friedman JM. Biochemistry 2003;42:8272–8288. [PubMed: 12846576]
- 32. Natan MJ, Kuila D, Baxter WW, King BC, Hawkridge FM, Hoffman BM. J Am Chem Soc 1990;112:4081–4082.
- 33. Gingrich DJ, Nocek JM, Natan MJ, Hoffman BM. J Am Chem Soc 1987;109:7533-7534.
- Kuila D, Natan MJ, Rogers P, Gingrich DJ, Baxter WW, Arnone A, Hoffman BM. J Am Chem Soc 1991;113:6520–6526.
- Miyazaki G, Morimoto H, Yun K-M, Park S-Y, Nakagawa A, Minagawa H, Shibayama N. J Mol Biol 1999;292:1121–1136. [PubMed: 10512707]
- 36. Lukin JA, Ho C. Chem Rev (Washington, DC, United States) 2004;104:1219-1230.
- Wiltrout ME, Giovannelli JL, Simplaceanu V, Lukin JA, Ho NT, Ho C. Biochemistry 2005;44:7207– 7217. [PubMed: 15882059]
- 38. Gong Q, Simplaceanu V, Lukin JA, Giovannelli JL, Ho NT, Ho C. Biochemistry 2006;45:5140–5148. [PubMed: 16618103]
- 39. Park S-Y, Yokoyama T, Shibayama N, Shiro Y, Tame JRH. J Mol Biol 2006;360:690–701. [PubMed: 16765986]
- 40. Kavanaugh JS, Rogers PH, Arnone A. Biochemistry 2005;44:6101–6121. [PubMed: 15835899]
- 41. Mueser TC, Rogers PH, Arnone A. Biochemistry 2000;39:15353-15364. [PubMed: 11112521]
- 42. Silva MM, Rogers PH, Arnone A. J Biol Chem 1992;267:17248–17256. [PubMed: 1512262]
- Noble RW, Hui HL, Kwiatkowski LD, Paily P, DeYoung A, Wierzba A, Colby JE, Bruno S, Mozzarelli A. Biochemistry 2001;40:12357–12368. [PubMed: 11591155]
- 44. Naito A, Hui HL, Noble RW, Hoffman BM. Biochemistry 2001;40:2060–2065. [PubMed: 11329273]
- 45. Naito, NR. Chemistry. Northwestern University; Evanston, IL: 1999.
- 46. McGourty JL, Peterson-Kennedy SE, Ruo WY, Hoffman BM. Biochemistry 1987;26:8302–8312. [PubMed: 3442655]
- 47. Miner, CS.; Dalton, NN. Glycerol. Reinhold Pub Corp.; New York: 1953.
- 48. Mathlouthi, M.; Reiser, P., editors. Sucrose: Properties and Applications. 1995.
- 49. Naito N, Huang H, Sturgess W, Nocek JM, Hoffman BM. J Am Chem Soc 1998;120:11256-11262.
- 50. Page CC, Moser CC, Dutton PL. Curr Opin Chem Biol 2003;7:551–556. [PubMed: 14580557]
- 51. Moser CC, Keske JM, Warncke K, Farid RS, Dutton LP. Nature 1992;355:796–802. [PubMed: 1311417]
- 52. Lin J, Balabin IA, Beratan DN. Science (Washington, DC, United States) 2005;310:1311-1313.
- 53. Jones ML, Kurnikov IV, Beratan DN. J Phys Chem A 2002;106:2002–2006.
- Onuchic JN, Beratan DN, Winkler JR, Gray HB. Annu Rev Biophys Biomol Struct 1992;21:349– 377. [PubMed: 1326356]
- 55. Beratan DN, Onuchic JN, Winkler JR, Gray HB. Science 1992;258:1740–1741. [PubMed: 1334572]
- 56. Because of the assumption of equivalence, eqs 3 need not include terms for conversion within an ensemble, as they have no physical consequence.

- 57. Steinfeld, JI.; Francisco, JS.; Hase, WL. Chemical Kinetics and Dynamics. Vol. 2. Prentice-Hall, Inc.: Upper Saddle River; New Jersey: 1999.
- 58. Schlitter J. Chem Phys 1988;120:187-197.
- Beece D, Bowne SF, Czege J, Eisenstein L, Frauenfelder H, Good D, Marden MC, Marque J, Ormos P, Reinisch L, Yue KT. Photochem Photobiol 1981;33:517–522.
- 60. Ivkovic-Jensen MM, Kostic NM. Biochemistry 1997;36:8135–8144. [PubMed: 9201962]
- 61. Jas GS, Eaton WA, Hofrichter J. J Phys Chem B 2001;105:261-272.
- 62. Davidson VL. Biochemistry 2000;39:4924-4928. [PubMed: 10769151]
- Peterson-Kennedy SE, McGourty JL, Kalweit JA, Hoffman BM. J Am Chem Soc 1986;108:1739– 1746.
- 64. Colombo MF, Rau DC, Parsegian VA. Science (Washington, DC, United States) 1992;256:655-659.
- 65. Parsegian VA, Rand RP, Rau DC. Methods Enzymol 1995;259:43-94. [PubMed: 8538466]
- 66. Parsegian VA, Rand RP, Rau DC. Proc Natl Acad Sci U S A 2000;97:3987–3992. [PubMed: 10760270]
- Natan MJ, Baxter WW, Kuila D, Gingrich DJ, Martin GS, Hoffman BM. Adv Chem Ser 1991;228:201–213.Electron Transfer Inorg Org Biol Syst
- 68. Roche CJ, Guo F, Friedman JM. J Biol Chem 2006;281:38757-38768. [PubMed: 17057250]
- Courtenay ES, Capp MW, Anderson CF, Record MT Jr. Biochemistry 2000;39:4455–4471. [PubMed: 10757995]
- 70. Furukawa Y, Matsuda F, Ishimori K, Morishima I. J Am Chem Soc 2002;124:4008–4019. [PubMed: 11942839]
- 71. Boehr DD, Dyson HJ, Wright PE. Chem Rev (Washington, DC, United States) 2006;106:3055-3079.
- 72. Lukin JA, Kontaxis G, Simplaceanu V, Yuan Y, Bax A, Ho C. Proc Nat Acad Sci USA 2003;100:517– 520. [PubMed: 12525687]
- 73. Fiechtner MD, McLendon G, Bailey MW. Biochem Biophys Res Comm 1980;96:618–625. [PubMed: 7426003]
- 74. Daugherty MA, Shea MA, Ackers GK. Biochemistry 1994;33:10345-10357. [PubMed: 8068671]
- 75. Goldbeck RA, Esquerra RM, Kliger DS, Holt JM, Ackers GK. Biochemistry 2004;43:12065–12080. [PubMed: 15379546]
- Goldbeck RA, Esquerra RM, Holt JM, Ackers GK, Kliger DS. Biochemistry 2004;43:12048–12064. [PubMed: 15379545]
- 77. Lalezari I, Lalezari P, Poyart C, Marden M, Kister J, Bohn B, Fermi G, Perutz MF. Biochemistry 1990;29:1515–1523. [PubMed: 2334712]
- 78. Schay G, Smeller L, Tsuneshige A, Yonetani T, Fidy J. J Biol Chem 2006;281:25972–25983. [PubMed: 16822864]
- 79. Noble RW, DeYoung A, Vitale S, Cerdonio M, DiIorio EE. Biochemistry 1989;28:5288–5292. [PubMed: 2548609]
- Bavydov R, Kofman V, Nocek J, Noble RW, Hui H, Hoffman BM. Biochemistry 2004;43:6330– 6338. [PubMed: 15147217]
- Samuni U, Roche CJ, Dantsker D, Juszczak LJ, Friedman JM. Biochemistry 2006;45:2820–2835. [PubMed: 16503637]
- 82. Pletneva EV, Fulton DB, Kohzuma T, Kostic NM. J Am Chem Soc 2000;122:1034-1046.
- Nocek JM, Stemp EDA, Finnegan MG, Koshy TI, Johnson MK, Margoliash E, Mauk AG, Smith M, Hoffman BM. J Am Chem Soc 1991;113:6822–6831.
- Mei H, Wang K, Peffer N, Weatherly G, Cohen DS, Miller M, Pielak G, Durham B, Millett F. Biochemistry 1999;38:6846–6854. [PubMed: 10346906]
- Feng C, Kedia RV, Hazzard JT, Hurley JK, Tollin G, Enemark JH. Biochemistry 2002;41:5816– 5821. [PubMed: 11980485]
- 86. Lasey RC, Liu L, Zang L, Ogawa MY. Biochemistry 2003;42:3904–3910. [PubMed: 12667081]
- Schlarb-Ridley BG, Mi H, Teale WD, Meyer VS, Howe CJ, Bendall DS. Biochemistry 2005;44:6232–6238. [PubMed: 15835911]

- Wang H, Lin S, Allen JP, Williams JC, Blankert S, Laser C, Woodbury NW. Science (Washington, DC, United States) 2007;316:747–750.
- 89. Ren Y, Wang W-H, Wang Y-H, Case M, Qian W, McLendon G, Huang Z-X. Biochemistry 2004;43:3527–3536. [PubMed: 15035623]
- 90. Hazzard JT, Poulos TL, Tollin G. Biochemistry 1987;26:2836–2848. [PubMed: 3038167]
- 91. Everest AM, Wallin SA, Stemp EDA, Nocek JM, Mauk AG, Hoffman BM. J Am Chem Soc 1991;113:4337–4338.
- 92. Marshall DB. Anal Chem 1989;61:660-665.



#### Figure 1.

Cartoon of the 'differential dynamics' energy landscape for forward and back ET, with  $R_j$  reactive substates and  $S_i$  less reactive substates. Dot (•) represents the system point for the equilibrium ensemble of configurations that exist at the start of a photo-initiated ET process and diamond (•) represents the system point subsequent to forward ET and prior to return of ET intermediate to ground.

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**Figure 2.** One  $[\alpha_1, \beta_2]$  ET 'complex' of a T-structure hemoglobin tetramer; hemes shown in red.

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## Figure 3.

(A) Two views of a 3D energy landscape with  $R_j = 1$  and seven equivalent  $S_i$  conformations; rate constants for conversion from  $S_i$  to R and R to  $S_i$ ,  $k_u$  and  $k_d$  respectively. (B) Gating energy landscape;  $S = \Sigma S_i$  and  $R = \Sigma R_j$ .

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#### Figure 5.

(a) Experimental progress curves of I for  $[\alpha(Zn)\beta(Fe)(N_3^{-1})]$  at  $\eta = 1$  cP (red),  $\eta = 1.7$  cP (green),  $\eta = 6.3$  cP (blue), and  $\eta = 12$  cP (purple); overlaid lines represent fits to Eq 2 for a relaxed photocycle,  $k_{obs} \neq k_p$ . The agreement between the long-time tails of the experimental and calculated traces has been slightly improved by inclusion in the latter of a slight contribution from  $[\alpha_1(ZnP^+)\beta_2(Fe^{3+})/\alpha_2(ZnP)\beta_1(Fe^{2+})]$  dimer pairs that form by dissociation of tetrameric I; these do not recombine to tetramers and undergo back ET on the timescale of these experiments (Supporting Information). (b) ET rate constants,  $k_b$  (red),  $k_p$  (blue), and  $k_{obs}$ (green) as a function of  $\eta$  for a relaxed photocycle:  $k_{obs}$  from fits of T to Eq 15;  $k_b$  and  $k_p$  from fits of timecourse for I to Eq 15. Lines are to guide the eye.

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## Figure 6.

(*a-d*) Experimental I timecourses (black) at  $\eta = 1, 1.7, 6.3$ , and 12 cP overlaid with simulated KD traces (solid lines) calculated with final parameters in Table 1. Below each trace is its decomposition into conformers, I<sup>S</sup> (short dashed line) and I<sup>R</sup> (dotted line). As in Fig 5A, agreement between the long-time tails of experimental and calculated traces requires a slight contribution from  $[\alpha_1(ZnP^+)\beta_2(Fe^{3+})/\alpha_2(ZnP)\beta_1(Fe^{2+})]$  dimer pairs that form by dissociation of tetrameric I; these do not recombine to tetramers and undergo back ET on the timescale of these experiments (see Supporting Information). (*e*) Evolution of I progress curves from *FE* limit to *SE* limit; calculations use final parameters in Table 1.



Scheme 1.





Scheme 2.

## Table 1

KD Parameters that Describe the  $[\alpha(Zn), \beta(Fe^{3+}N_3^{-})]$  ET Photocycle.

Parameter	Initial	Final
$k_{teq} (s^{-1})^a$	18	21
$k_{beq} (s^{-1})^a$	230	210
$k_t^{S} (s^{-1})^a$	5	5
$k_{t}^{R}(s^{-1})$	409	517
$k_b^{S}(s^{-1})$	221	200
$\mathbf{k}_{\mathbf{b}}^{\mathbf{R}} (\mathbf{s}^{-1})^{a}$	500	550
K <sup>S</sup>	30	31
α	0.83	0.91
$k_{d}^{0}(s^{-1})$	3000	1017

<sup>a</sup>Constraints generated as described in text