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# **Control of Ion Distortion in Field Asymmetric Waveform Ion** Mobility Spectrometry via Variation of Dispersion Field and Gas **Temperature**

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#### Abstract

Field asymmetric waveform ion mobility spectrometry (FAIMS) has emerged as an analytical tool of broad utility, especially in conjunction with mass spectrometry. Of particular promise is the use of FAIMS and 2-D ion mobility methods that combine FAIMS with conventional IMS to resolve and characterize protein and other macromolecular conformers. However, FAIMS operation requires a strong electric field and ions are inevitably heated by energetic collisions with buffer gas molecules. This may induce ion isomerization or dissociation that distort the separation properties of FAIMS (and subsequent stages) and/or reduce instrumental sensitivity. As FAIMS employs a periodic waveform, whether those processes are controlled by ion temperature at maximum or average field intensity has been debated. Here we address this issue by measuring the unfolding of compact ubiquitin ion geometries as a function of waveform amplitude (dispersion field,  $E_{\rm D}$ ) and gas temperature, T. The field heating is quantified by matching the dependences of structural transitions on  $E_D$  and T: increasing  $E_D$  from 12 to 16 or from 16 to 20 kV/cm is equivalent to heating the  $(N_2)$ gas by  $\sim 15 - 25$  °C. The magnitude of field heating for any  $E_D$  can be estimated using the twotemperature theory, and raising  $E_D$  by 4 kV/cm augments heating by  $\sim$ 15 – 30 °C for maximum and ~4 – 8 °C for average field in the FAIMS cycle. Hence, isomerization of ions in FAIMS appears to be determined by the excitation at waveform peaks.

#### Introduction

Separation and structural elucidation of gas-phase ions by mass spectrometry (MS) or ion mobility spectrometry (IMS) is central to many analytical applications. <sup>1,2</sup> To accurately characterize the present species and particularly solution structures, one usually seeks to minimize their destruction or distortion during ionization and subsequent analysis. As in MS the determination of ion m/z is performed in vacuum and thus does not change ion structure (except for spontaneous decay), one generally needs to preserve species only during ionization and transfer through high pressure regions. This has motivated development of electrospray (ESI), <sup>3,4</sup> sonic spray, <sup>5</sup> and other sources that can ionize large biomolecules, including proteins, protein complexes, and DNA "softly" - without changes to the primary or quaternary structure (i.e., dissociation) and with limited distortion of secondary and tertiary structure (i.e., isomerization). The resulting ions often retain key aspects of their solution conformation. 6–11

In IMS, ions are separated while pulled through gas by electric field, which may influence their structure. Ion - molecule chemistry is normally avoided by use of He, N2, or other inert gases that do not react with ions under experimental conditions, but "field heating" of ions by above-thermal collisions with molecules is not preventable. The ion-molecule collision velocity  $(v_{I-M})$  is the vector sum of the random component due to Brownian motion of both particles  $(v_{Br})$  and ion drift motion (v) directed by electric field, hence in IMS exceeds that for freely diffusing ions defined by gas temperature, T. As v is constant for all collisions, the

distribution of  $v_{\text{I-M}}$  (unlike  $v_{\text{Br}}$ ) is not rigorously Maxwellian, but may be approximated as such when  $v \ll \overline{v}_{Br}$ . This underlies the two-temperature theory <sup>12,13</sup> where the "ion temperature",  $T_{\text{I}}$ , is

$$T_{\rm I} = T + T_{\rm H} = T + M(KE)^2 / (3k_{\rm B}) \tag{1}$$

Here M is the gas molecule mass, K is the ion mobility, E is the electric field intensity, and  $k_{\rm B}$  is the Boltzmann constant. While eq (1) has been derived for the translational temperature, on the timescale of IMS and particularly FAIMS separations all degrees of freedom are equilibrated and the ion vibrational temperature is also set by eq (1). For polyatomic ions, this heating by  $T_{\rm H}$  may cause dissociation or isomerization that would change the IMS separation properties.

In drift tube IMS (DTIMS) at conventional pressures (P) of >~1 Torr,<sup>2,12</sup> the field is usually so weak that  $T_{\rm H}$  is insignificant. For example, the reduced mobility  $K_0$  [equal to K at standard T and P] is <3 cm²/(V×s) even for smallest ions and ~1 – 2 cm²/(V×s) for most medium-size ones, and typical E in atmospheric-pressure DTIMS systems is ~200 – 700 V/cm. Then  $T_{\rm H}$  is a negligible ~0.1 °C even for highest  $K_0$  and E, and routinely is lower. Somewhat stronger but still generally immaterial heating may occur in reduced-pressure IMS used in some IMS/MS systems. For example, the IMS stage in tandem quadrupole - drift tube instruments <sup>14,15</sup> is commonly operated at P ~5 Torr and E ~15 V/cm, meaning  $T_{\rm H}$  = 0.15 – 1.3 °C for  $K_0$  = 1 – 3 cm²/(V×s).

The mobility of ions in gases is a function of E. This allows field asymmetric waveform IMS (FAIMS) $^{16}$  or differential IMS $^{17}$  to separate and identify species based on the difference between mobilities at two unequal E. In practice, FAIMS employs a periodic field E(t) that alternates segments of higher E and lower E of opposite polarity. This field and a superimposed compensation field  $E_C$  are established by applying a waveform U(t) and "compensation voltage"  $U_C$  to a gap between two electrodes through which ions are carried by gas flow. The amplitudes of E(t) and U(t) are termed the "dispersion field",  $E_D$ , and "dispersion voltage",  $U_D$ . At any given  $E_C$ , only species with certain K(E) profile remain balanced and pass to the detector or following analytical stage. The deviation of K from K(0) approximately scales as  $E^3$  and thus is negligible at moderate E typical of DTIMS, but rapidly rises at high E. The change of E at higher E is due to several factors. The major is that E equals: E

$$K = K_0 \frac{N_0}{N} = \frac{3}{16} \left( \frac{2\pi (m+M)}{mMk_B T_I} \right)^{1/2} \frac{q}{N\Omega(T_I)}$$
 (2)

where m and q are the ion mass and charge, N is the gas number density,  $N_0$  is N at STP, and  $\Omega$  is the ion-molecule collision cross section that depends on the  $v_{\text{I-M}}$  distribution and thus on  $T_{\text{I}}$  in a way governed by the interaction potential. Hence, by requiring K to materially depend on E, FAIMS operation necessitates  $T_{\text{I}}$  in one or both E(t) segments to substantially differ from T. Thus field heating is unavoidable during at least the high-field segment.

The heating may fragment ions entering the gap, with FAIMS characterizing the product ions  $^{18-21}$  Though this regime may improve the specificity similarly to in-source decay in MS, one usually desires the ability to study intact precursors. In FAIMS, the ion activation is caused by the separation mechanism itself and thus cannot be switched off without altering its outcome. The trivial solution of decreasing E is hardly acceptable  $^{22}$  because it reduces the resolving power (scaling as  $^{\sim}E^3$ ) more than heating that scales as  $E^2$  by eq (1). In FAIMS/MS systems, one may instantly detect the ion dissociation in FAIMS by switching E(t) off. More subtle is

the thermal isomerization, particularly for macroions such as most proteins that denature (i.e., significantly change conformation) upon modest collisional or thermal heating.  $^{22-28}$  This process is not easily assessed by MS but alters FAIMS separation parameters, which is of concern for the characterization of conformers including misfolded proteins relevant to biomedical problems.  $^{28-34}$  A dramatic expansion of FAIMS separation space and thus peak capacity, ascribed to the alignment of ion macrodipoles by electric field, was recently reported  $^{33}$  for proteins with  $m > \sim 30$  kDa, and might be extendable to lower m. This raises the utility of FAIMS for analysis of macromolecular conformations, motivating the effort to better understand their distortion in FAIMS and devise approaches to its mitigation.

Unlike in DTIMS, |E| in FAIMS is not fixed but oscillates from 0 to  $E_{\rm D}$  and the proper choice of E in eq (1) is not obvious. In the studies of ubiquitin and cytochrome c, where isomerization of ions induced by field heating in FAIMS was first discussed,  $^{30,31}$  the process was assumed to be controlled by mean |E| over the E(t) cycle. This and the adoption of too low K values  $^{22}$  produced  $T_{\rm H}$  <10 °C, and conformations were not expected to change significantly.  $^{30,31}$  This issue could not be probed using only FAIMS or FAIMS/MS operating at fixed T for the lack of a priori means to deduce ion structures from FAIMS data.

Positive-mode ESI produces protonated proteins with a distribution of z sensitive to the solution composition. For ubiquitin ( $m \sim 8.6$  kDa) in acidic media,  $^{22-24,28,35}$  charge states of  $z \sim 5-13$  are observed. Proteins in either solution or the gas phase may adopt many 3-D structures depending on the present and past conditions. Proteins with no disulfide links like ubiquitin are free to adopt any conformation. In the gas phase, they assume compact near-native geometries at low z and elongated geometries lacking tertiary structure at high z because the Coulomb repulsion between charged sites (scaling as  $\sim z^2$  for fixed geometries) overcomes the hydrogen bonds.  $^{22-26,28,35}$  At intermediate z, one usually encounters the broadest diversity of geometries, including compact, partly folded, and unfolded conformers.

Compact and partly folded proteins gradually unfold at higher  $T_{\rm I}$  as the thermal energy exceeds the dissociation barrier of specific bonds, while elongated conformers change little. The needed  $T_{\rm I}$  decreases for higher z as the folds are destabilized by Coulomb repulsion, so the isomer population over a range of z covering the transition from compact to elongated geometries provides a reliable ion thermometer with laddered overlapping scales. Prominent structural differences between intermediates on the protein unfolding pathway frequently make them distinguishable (and thus the thermometer readable) by either DTIMS or FAIMS. This allows extracting  $T_{\rm H}$  in high-field IMS by matching the resulting populations with those produced by thermal heating at low E where  $T_{\rm H}$  is negligible. In general, heat-denatured protein ions do not spontaneously refold upon cooling,  $^{22,23,25,26,28}$  and, as medical thermometers for measuring the body temperature, the "protein unfolding thermometer" registers the highest temperature ever experienced rather than the instantaneous temperature. This feature permits characterizing  $T_{\rm H}$  in high-field IMS by analyses of conformer populations in subsequent stages producing less or no ion heating.

In contrast to FAIMS data, the drift times  $(t_d)$  in DTIMS can be related to ion geometries by physical modeling: greater  $t_d$  signifies more open structures. At room T, ubiquitin ions generated by ESI under gentle conditions are  $^{22-24,35}$  compact for  $z \le 6$  and elongated for  $z \le 11$ . For z = 7 - 10, both families and/or partly folded species co-exist. Insertion of a FAIMS stage prior to DTIMS raises  $t_d$  (summed over the FAIMS spectra), indicating some unfolding by FAIMS. The resulting  $t_d$  distributions resemble those obtained without FAIMS for ions heated  $^{24}$  to  $^{20}$  to  $^{20}$  c, i.e., by  $^{20}$  c over room T. This value greatly exceeds  $T_H$   $^{20}$  c calculated using eq (1) for average |E| but is only slightly less than  $T_H$   $^{20}$  c obtained for  $E = E_D$ , suggesting that ion isomerization in FAIMS is governed by the maximum  $^{20}$  rather than the average E. It was proposed to avoid the isomerization or

dissociation caused by such field heating via cooling the buffer gas by relevant  $T_{\rm H}$  below  $T_{\rm I}$  of entering ions, so that  $T_{\rm I}$  during separation never exceeds the initial value. <sup>22</sup> However, the FAIMS (the "Selectra" device of dome geometry previously manufactured by Ionalytics) and DTIMS stages in those experiments had dissimilar ion inlets - a curtain plate and capillary, respectively. So the distortion caused by FAIMS might have reflected stronger ion excitation at the source inlet. <sup>22</sup> Also, ions spend a substantial time in FAIMS: ~200 ms in Selectra vs. ~60 – 90 ms in DTIMS and <100 ms in the ion funnel trap at DTIMS entrance. <sup>22</sup> Hence adding the FAIMS stage raises the "age" of ions characterized by DTIMS by several times, which might promote unfolding by delayed (spontaneous) rather than a heating-driven mechanism. <sup>22</sup>

The recent introduction of heated FAIMS devices  $^{36}$  enables analyses at T up to  $\sim 110$  °C. This allows comparing isomer populations as a function of E or T without involving other stages, which excludes the effects of inlet conditions and different reaction time scales. The field heating for any ion also depends on the gas identity through the value of K in eq (1): in general, K in gases of smaller molecules is higher,  $^{12}$  leading to greater  $T_H$ . Here we explore the ubiquitin ion conformers resolved by FAIMS in  $N_2$  as a function of both  $E_D$  and T, and confirm that the ion structure in FAIMS is largely determined by  $T_L$  given by eq (1) using the maximum E.

### **Experimental method**

Measurements were performed using the TSQ Quantum Ultra triple quadrupole MS system with FAIMS option<sup>36</sup> and Ion Max ESI source (Thermo Fisher, San Jose, CA). A 10 μM solution of bovine ubiquitin (Sigma Aldrich) in 50:49:1 water/methanol/acetic acid was infused at 2 μL/min to the ESI emitter biased at 4 kV. The FAIMS unit has transverse or "side-to-side" cylindrical configuration with the gap width of g = 2.5 mm (inner and outer radii of 6.5 and 9 mm) and allows  $|U| \le 5$  kV. With proper  $U_D$  sign, curved FAIMS gaps focus ions to the median, which provides high sensitivity but limits resolution.<sup>37</sup> All ubiquitin conformers with z = 5 - 14 generated by ESI belong<sup>28–30</sup> (with N<sub>2</sub> carrier gas) to type C,<sup>38</sup> meaning a negative derivative of K(E). Such species are focused in the P2 mode<sup>37</sup> ( $U_D < 0$ ;  $U_C < 0$ ). Though the  $E_D$  value in curved FAIMS varies across the gap,<sup>38</sup> here we define  $E_D = U_D/g$  (approximating  $E_D$  at the median)<sup>39</sup> to compare the data at different  $E_D/N$ . We used  $U_D$  of -3, -4, and -5 kV that correspond to  $E_D$  of -12, -16, and -20 kV/cm. The total (carrier and curtain plate) gas flow was 3 L/min. The FAIMS spectra for z = 6 - 13 were acquired by stepping  $U_C$  in 0.3 V increments with MS in the selected ion monitoring mode scanning 5 amu-wide windows around pertinent m/z. All data were collected over the T = 35 - 80 °C range with 5 °C step. The FAIMS electrodes were at equal T, creating no thermal gradient across the gap.

#### **Results and Discussion**

#### Apparent and intrinsic temperature dependences of FAIMS spectra

The FAIMS spectra for ubiquitin ions at  $E_{\rm D}=-20~{\rm kV/cm}$  and minimum  $T=35~{\rm ^{\circ}C}$  (Figure 1a) are close to those measured  $^{28,37}$  at same  $E_{\rm D}$  and  $^{\sim}20~{\rm ^{\circ}C}$  using Selectra with  $g=2~{\rm mm}$ . The spectra for z=6 and 7 are broad, with major peaks for compact geometries at  $E_{\rm C}\sim-60~{\rm V/cm}$  for 6+ and  $^{\sim}-75~{\rm V/cm}$  for 7+ and minor features for elongated conformers at  $^{\sim}-35-40~{\rm V/cm}$ . For 7+, there also are wide features for partly folded structures at intermediate  $E_{\rm C}$ . The results for 8+ resemble those for 6+, but the peak at  $^{\sim}-40~{\rm V/cm}$  becomes dominant as the protein unfolds. For z=9-13, a single peak at  $E_{\rm C}\sim-(40-45)~{\rm V/cm}$  indicates full unfolding.  $^{28,37}$  The absence of more compact geometries at higher  $|E_c|$  for 9+ and abundance of unfolded species for 7+ and 8+ point to  $T_{\rm I}$  exceeding that in earlier work,  $^{28}$  possibly reflecting higher T in present experiments.

With T increasing to 80 °C, the raw spectra move to higher  $|E_c|$  for z of 8 – 13 but not 6 or 7, and the spectrum for 7+ changes from the signature three-peak shape seen in FAIMS or DTIMS to a single feature (Figure 1b). This pattern remains when data are plotted vs.  $E_C/N$  (Figure 1 c, d). The non-uniformity of changes indicates a structural transition upon heating for some z, including 7. To isolate it, we must remove the thermal shift of  $E_C$  due to the dependence of N on T. Indeed,  $K_0$  for rigid ions generally depends not on E but on E/N:

$$K_0(E/N) = K_0(0)[1 + a_1(E/N)^2 + a_2(E/N)^4 + \dots + a_n(E/N)^{2n}]$$
 (3)

where  $a_{\rm n}$  are fixed coefficients characterizing the ion/gas pair. <sup>12</sup> Hence, graphing the spectra acquired at equal  $E_{\rm D}/N$  vs.  $E_{\rm C}/N$  eliminates <sup>36</sup> the trivial effect of T within the relative error of ideal gas law which, for N<sub>2</sub> over T=35-80 °C, is a negligible <0.1 % (according to the Van der Waals equation of state). Presently,  $E_{\rm D}$  was not scaled with inverse temperature. In this case, we can exploit the dependence <sup>40</sup> of  $E_{\rm C}/N$  on  $E_{\rm D}/N$ :

$$E_{C}/N = -\left(\sum_{n=1}^{\infty} a_{n} (E_{D}/N)^{2n+1} \langle F_{2n+1} \rangle\right) / \left(1 + \sum_{n=1}^{\infty} (2n+1) a_{n} (E_{D}/N)^{2n} \langle F_{2n} \rangle\right)$$
(4)

where  $\langle F_n \rangle$  are dimensionless properties of the E(t) profile. For bisinusoidal E(t), eq (4) for n=1 and 2 is equivalent to: 16

$$\frac{E_c}{N} = -\frac{a_1}{N^3} \left( \frac{E_D^3}{9} + \frac{5E_D^2 E_C}{6} + E_C^3 \right) - \frac{a_2}{N^5} \left( \frac{55E_D^5}{486} + \frac{55E_D^4 E_C}{72} + \frac{10E_D^3 E_C^2}{9} + \frac{25E_D^2 E_C^3}{9} + E_C^5 \right)$$
(5)

The highest  $|E_c|/E_D|$  value for present data is  $\sim 0.004$  (with  $E_C \sim -80$  V/cm at  $E_D = -20$  kV/cm for 7+), hence omitting the terms containing  $E_C$  on the rhs of eq (5) changes the quantity in either parentheses by <3%. Further, all  $E_C$  values are negative at any  $E_D$  with  $|E_c|$  increasing at higher  $E_D$  (Figure 1), which confirms that all ubiquitin ion species in  $N_2$  are  $^{38}$  type C where  $a_1 < 0$ . For such species, the terms with  $n \ge 2$  usually contribute little to the  $K_0(E/N)$  and thus  $E_C(E_D)$  functions (at moderate E/N of present experiments), and eq (5) may be approximated as

$$E_C/N = -a_1 \langle F_3 \rangle (E_D/N)^3 \tag{6}$$

Indeed, for two most abundant conformers for 11+ or 12+ that are "unfolded" even at  $E_{\rm D}=0$  and thus should largely retain the same structure at any  $E_{\rm D}$ , the  $E_{\rm C}(E_{\rm D})$  curves measured  $^{30}$  (in N<sub>2</sub>) up to  $|E_{\rm D}|=22$  kV/cm can be fit by  $a_1=-(3.2-3.7)\times 10^{-11}$  (cm/V)<sup>2</sup> and  $a_2=(0\pm3)\times 10^{-21}$  (cm/V)<sup>4</sup>: the n=2 term in eq (3) is <4% of the n=1 term even at highest  $|E_{\rm D}|$  used here.

Thus the quantity  $(E_{\rm C}/N)/(E_{\rm D}/N)^3$  (which we call "normalized  $E_{\rm C}$ " or  $E_{\rm CN}$ ) for various spectral features should be largely independent of T to the extent  $a_1$  is independent of T. That is never rigorous because  $K_0$  by eq (2) depends on T (i) directly as  $T_{\rm I}^{-1/2}$  and (ii) through  $\Omega(T_{\rm I})$  function. However, the effect of (i) decreases  $K_0$  between 35 and 80 °C by just 7%, which is partly offset by (ii) because  $\Omega$  for fixed geometries decreases at higher T. So  $K_0$  values for rigid mediumsize and large ions in  $N_2$  tend to decrease by just a few percent over the  $T_{\rm I} \sim 50 - 150$  °C range sampled here. Further, the  $K_0(T)$  dependence is approximately additive  $^{12}$  to  $K_0(E/N)$  and thus mostly cancels from the ratio of K at high and low E measured in FAIMS. In contrast, unfolding of proteins commonly raises  $\Omega$  and thus drops  $K_0$  by several times.  $^{22-26,28,35}$ 

Hence, most of the temperature-dependent variation of FAIMS spectra plotted vs.  $E_{CN}$  must reflect changes of ion structure.

The data for ubiquitin ions at T=35-80 °C support this hypothesis. The redrawn spectra at T=35 °C (Figure 2) appear the same as in Figure 1a, with  $E_{\rm CN}$  (in  $10^{-7}$  Td<sup>-2</sup>) of ~(5 – 5.5), ~(4 – 4.5), ~(3.5 – 4.5), and ~(2.5 – 3.5) for compact (7+), compact (6+), partly folded (7+), and elongated (z=8-13) conformers, respectively. However, the temperature dependence differs from that in Figure 1. For z=9-13 that are already unfolded at T=35 °C, the peaks are nearly fixed with a tiny shift to the right at higher T for z=10-13 (Figure 2a). For z=6-8 (Figure 2b), the peaks shift to lower  $E_{\rm CN}$  and greatest changes occur for 6+ and 7+: the compact and partly folded conformers disappear by T ~80 °C, leaving only the features at ~(2.5 – 3) ×  $10^{-7}$  Td<sup>-2</sup> that indicate elongated geometries. As the "isomer population thermometry" requires observable interconversions over the relevant temperature range, we shall focus on structures with z=6-8 that unfold in present experiments.

#### Quantification of field heating in FAIMS

As discussed above, no FAIMS separation is possible at E low enough for negligible  $T_{\rm H}$ . Thus, unlike with DTIMS,  $^{22}$  comparisons of the spectra in Figure 2 with those at other  $E_D$  and Tvalues cannot determine absolute  $T_H$ . However, they can reveal the difference ( $\Delta T$ ) between  $T_{\rm H}$  at unequal  $|E_{\rm D}|$  values, such as 12, 16, and 20 kV/cm used here. The temperature dependences of spectra at 12 and 16 kV/cm (Figure 3) resemble that at 20 kV/cm described above, but are dwarfed by systematic increase of  $|E_C|$  at greater  $|E_D|$ . That increase is standard for rigid geometries and must be offset to isolate the spectral changes reflecting isomerization due to stronger field heating at higher  $E_{\rm D}$ . Within the approximations made to isolate the structural part of  $E_{\mathbb{C}}(T)$  function, we can express the trivial  $E_{\mathbb{C}}(E_{\mathbb{D}})$  dependence using eq (6) and, again, remove it by plotting the spectra vs.  $E_{\text{CN}}$  (Figure 4). This largely aligns the peaks for same isomers at different  $E_D$ , as is most clearly seen for 8+. For 6+ and 7+ with multiple features, the same spectral shape changes and  $E_C$  shifts occur at higher T as  $|E_D|$  drops. For example, in the spectra for 7+ that comprise three peaks (Figure 5), that at the highest  $|E_C|$  for compact geometries vanishes, but that at the lowest  $|E_C|$  for elongated geometries grows at higher T (at least on the relative scale). While this happens at any  $|E_D|$ , the two peaks reach equal height at different T: ~40 – 45 °C in the data for 20 kV/cm (Figure 5a) but ~65 °C for 16 kV/cm (Figure 5b), suggesting  $\Delta T \sim 20 - 25$  °C.

When the shift of  $E_{\rm C}$  upon isomerization during FAIMS analysis compares to or exceeds the peak width set by instrumental resolving power for the remaining ion residence time in the gap, the products may be filtered out in part or in full.<sup>22</sup> Hence spectral changes upon heating may reflect either the reduction/elimination of reactants and growth/emergence of products or just the reactant loss, with other features appearing to increase in relative terms. As the total ion count in these experiments depends on the gas temperature (because of varying efficiency of ion desolvation in front of FAIMS and of focusing in the analytical gap), the two effects are hard to disentangle. However, this does not matter for our goal of measuring the extent of ion heating by electric field via comparison of its consequences by whatever mechanism with those due to known thermal heating.

In most cases, the diversity of protein conformers and their consequent poor resolution by FAIMS precludes robust peak fitting to extract isomer abundances. However, given a uniform decrease of  $E_{\rm C}$  with protein unfolding, its overall extent is quantifiable by mean  $E_{\rm CN}$  values,  $E_{\rm CN}$  (Figure 6a). As expected,  $E_{\rm CN}$  decreases with increasing T for all z=6-8 at any  $E_{\rm D}$ , but the  $T(E_{\rm CN})$  curves move to the right and up at lower  $E_{\rm D}$ : the geometries are more folded at equal T and higher T is needed for equal unfolding. The value of  $\Delta T$  for two  $E_{\rm D}$  values can be deduced by transposing one curve along the T axis to minimize its deviation from the other. All curves are near-linear for T 50 °C, but those for  $|E_{\rm D}|=12$  and 16 kV/cm have breaks of

unclear origin at  ${\sim}45$  °C. Regardless of whether the data for T 45 °C are included in the sets, the linear regressions at  $|E_{\rm D}|$  of 20 and 12 kV/cm match those at 16 kV/cm upon shifting up and down by  ${\sim}15-25$  °C, respectively (Figure 6b). This is consistent with  $\Delta T {\sim}20-25$  °C obtained by matching intensities of specific features, as illustrated above. We will now compare these estimates with the two-temperature theory.

#### Modeling of field heating and comparison with measurements

By eq (1), changing E from  $E_1$  to  $E_2$  produces:

$$\Delta T = M[(K_2 E_2)^2 - (K_1 E_1)^2]/(3k_{\rm B})$$
(7)

where  $K_1$  and  $K_2$  are mobilities at  $E_1$  and  $E_2$ . When  $E_1$  and  $E_2$  are  $E_D$  values for two FAIMS waveforms, eq (7) provides the maximum  $\Delta T$  at waveform peaks ( $\Delta T_{\rm max}$ ). The variation of K over the sampled E range for ubiquitin species studied is ~1%, which is negligible compared to the difference between  $E_D$  values used. Hence eq (7) can be reduced to:

$$\Delta T_{\text{max}} = MK[E_2^2 - E_1^2]/(3k_{\text{B}})$$
(8)

The average  $\Delta T$  in FAIMS depends on the integral of eq (8) over E(t) period and thus on the E(t) profile. The TSQ FAIMS system employs a bisinusoidal waveform comprising two harmonics with 2:1 frequency ratio: 39,42-44

$$E(t) = E_{\rm D} [f \sin wt + \sin(2wt - \pi/2)] / (f+1)$$
(9)

where f is the ratio of harmonic amplitudes. Substituting this into eq (8) and averaging, we obtain

$$\Delta \overline{T} = \frac{f^2 + 1}{2(f+1)^2} M K^2 (E_2^2 - E_1^2)$$
(10)

As most studies, present experiments use a near-optimum  $^{42-44} f = 2$  and eq (10) converts to

$$\Delta \overline{T} = 5MK^2(E_2^2 - E_1^2)/18 \tag{11}$$

Hence  $\Delta \overline{T}$  under any conditions equals  $\Delta T_{max}$  scaled by  $\approx$ 0.278. This factor determined by averaging  $\Delta T$  is more accurate than the published value<sup>22,30,31</sup> of  $\approx$ 0.218 derived<sup>30</sup> by substituting the mean |E| into eq (8), i.e., the mean of the square is not the square of the mean. To calculate  $\Delta T_{max}$  and  $\Delta \overline{T}$  for specific ions, we need to know the values of K.

The mobilities of ubiquitin ions with z=6-8 in  $N_2$  span a broad range,  $^{22,28}$  but only compact and partly folded geometries can unfold upon field heating and thus only their mobilities are relevant here. At  $T=20\,^{\circ}\text{C}$ , the measured  $^{22}$  values for those species are  $K_0\sim0.9-1.2\,\text{cm}^2/(\text{V}\times\text{s})$  or  $K\sim0.97-1.29\,\text{cm}^2/(\text{V}\times\text{s})$ . For macroions such as proteins, the cross section of a rigid geometry may be assumed independent of temperature. Then K is proportional to  $T^{1/2}$  by eq (2), and the values of K,  $T_{\rm H}$  by eq (1),  $\Delta T_{\rm max}$  by eq (8), and by eq (11) are listed in Table 1. Reflecting a narrow temperature range, the results at T=35 and 80 °C are close and we can use average values, especially as unfolding mostly occurs at intermediate  $T\sim50-60\,^{\circ}\text{C}$  (Figure 2). The computed  $\Delta T_{\rm max}$  or  $\Delta T$  for  $|E_{\rm D}|$  of 12 and 16 kV/cm are lower than the

corresponding values for 16 and 20 kV/cm by 22%. This difference is also not critical to the present discussion, and we can estimate  $\Delta T_{\rm max}$  as  $\sim 13-30$  °C while equals  $\sim 4-8$  °C only.

This  $\Delta T_{\rm max}$  range is in agreement with measured  $\Delta T \sim 15-25$  °C, particularly as the maximum  $T_{\rm H}$  and thus  $\Delta T_{\rm max}$  apply instantly at  $E=E_{\rm D}$ , while isomerization takes finite time and  $T_{\rm H}$  and  $\Delta T$  available over finite segments around E(t) peaks are lower. <sup>22</sup> An (admittedly arbitrary) discount of 10% in terms of E would reduce  $\Delta T_{\rm max}$  at any two  $E_{\rm D}$  by  $\sim 20\%$  - presently, to  $\sim 10-25$  °C. The calculated  $\Delta T$  values are much lower than measured  $\Delta T$  and such small shifts of  $T(E_{CN})$  curves would hardly even be noticed in present experiments with T incremented in 5 °C steps. Accordingly, isomerization of protein ions in FAIMS is controlled by maximum rather than average ion temperature.

The magnitude of ion heating depends on the value of K for particular species, but those for compact ions of proteins generated by ESI are rather insensitive to the protein size and stay close to  $1 \text{ cm}^2/(\text{V}\times\text{s})$  over a broad mass range. 22-26,28 This happens because the cross sections of macroions scale as or slightly slower than (volume) $^{2/3}$  and thus  $m^{2/3}$ , while the typical charging of near-spherical protein geometries by ESI is proportional  $^{45}$  to  $^{25}$  to  $^{25}$ . Then K by eq (2) would scale as or slower than  $m^{-1/6}$ , meaning a mobility change of  $^{25}$  times over the range of  $m^{25}$  to  $^{25}$  times over the range of  $m^{25}$  times, e.g., from the maximum of  $^{25}$  C (Table 1) to  $^{25}$  °C.

#### **Conclusions**

We have explored the dependence of FAIMS spectra for ubiquitin ions on the temperature (T) of the N<sub>2</sub> carrier gas and waveform amplitude. For charge states of 6+ to 8+, those ions adopt compact geometries at room temperature but unfold at elevated internal temperatures that may be raised by either heating of the gas or collisional (field) heating of ions by rapid drift in electric field of high intensity (E) that separates ions in FAIMS. This allows quantifying the changes  $(\Delta T)$  in field heating by comparing the increases of E at waveform peaks  $(E_D)$  and gas temperature needed for equal extent of unfolding determined from FAIMS spectra adjusted for trivial (non-structural) shifts as a function of both gas temperature and  $E_D$ . The field heating for a given E may be calculated using the two-temperature theory. However, as E in FAIMS cycle oscillates from zero to  $E_D$ , the field heating (that scales as  $E^2$ ) and thus  $\Delta T$  similarly oscillate and their maxima exceed the means by several (here 3.6) times. The measurements of  $\Delta T$  over a broad  $E_{\rm D}$  range for all three charge states match the computed maximum but not mean values. Hence, the structure of ions in FAIMS is dictated primarily by their highest rather than average internal temperatures. This conclusion is consistent with findings from FAIMS/ IMS experiments, <sup>22</sup> but is more reliable as it excludes possible artifacts due to unequal ion excitation in FAIMS and IMS inlets, different timescales of IMS and FAIMS/IMS analyses, as well as comparisons with thermal heating measured on other platforms.

It has been suggested that isomerization or dissociation of ions in FAIMS can be avoided if the gas is cooled by the magnitude of field heating of ions.  $^{22}$  Though here the gas was heated, we have directly demonstrated that the nature of ions is controlled by the sum of gas temperature and (maximum) field heating. This should equally apply with a cooled gas, opening the path to non-destructive FAIMS and FAIMS/IMS analyses of fragile macromolecules. At the highest dispersion field used here, cooling to  $\sim\!\!-60~^\circ\mathrm{C}$  should suffice for most protein ions.

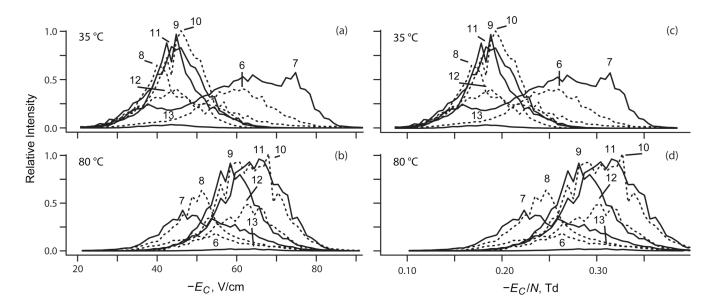
## **Acknowledgments**

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#### References

- 1. Gross, JH. Mass Spectrometry. Springer: a Textbook; 2006.
- 2. Eiceman, GA.; Karpas, Z. Ion Mobility Spectrometry. Boca Raton: CRC Press; 2004.
- 3. Cole, RB., editor. Electrospray Ionization Mass Spectrometry. New York: Wiley; 1997.
- 4. Pramanik, BN.; Ganguly, AK.; Gross, ML., editors. Applied Electrospray Mass Spectrometry. Boca Raton: CRC Press; 2002.
- 5. Hirabayashi A, Sakairi M, Koizumi H. Sonic Spray Mass Spectrometry. Anal. Chem 1995;67:2878.
- 6. van den Heuvel RH, Heck AJR. Curr. Opin. Chem. Bio 2004;8:519. [PubMed: 15450495]
- 7. Ruotolo BT, Robinson CV. Curr. Opin. Chem. Bio 2006;10:402. [PubMed: 16935553]
- 8. Krishnaswamy SR, Williams ER, Kirsch JF. Protein Sci 2006;15:1465. [PubMed: 16731980]
- 9. Patrikkson A, Marklund E, van der Spoel D. Biochemistry 2007;46:933. [PubMed: 17240977]
- 10. Baker ES, Bernstein SL, Bowers MT. J. Am. Soc. Mass Spectrom 2005;16:989. [PubMed: 15908229]
- 11. Yamaguchi K. J. Mass Spectrom 2003;38:473. [PubMed: 12794868]
- 12. McDaniel, EW.; Mason, GA. Transport Properties of Ions in Gases. New York: Wiley; 1988.
- 13. Viehland LA, Mason EA. At. Data Nucl. Data Tables 1995;60:37.
- 14. Bowers MT, Kemper PR, Von Helden G, Van Koppen PAM. Science 1993;260:1446. [PubMed: 17739800]
- 15. Shvartsburg AA, Schatz GC, Jarrold MF. J. Chem. Phys 1998;108:2416.
- 16. Guevremont R. J. Chromatogr. A 2004;1058:3. [PubMed: 15595648]
- 17. Kolakowski BM, Mester Z. Analyst 2007;132:842. [PubMed: 17710259]
- 18. Veasey CA, Thomas CLP. Analyst 2004;129:198. [PubMed: 14978520]
- Nazarov EG, Coy SL, Krylov EV, Miller RA, Eiceman GA. Anal. Chem 2006;78:7697. [PubMed: 17105161]
- Kendler S, Lambertus GR, Dunietz BD, Cos SL, Nazarov EG, Miller RA, Sacks RD. Int. J. Mass Spectrom 2007;263:137.
- 21. Borsdorf H, Nazarov EG, Miller RA. Talanta 2007;71:1804. [PubMed: 19071526]
- 22. Shvartsburg AA, Li F, Tang K, Smith RD. Anal. Chem 2007;79:1523. [PubMed: 17297950]
- 23. Valentine SJ, Counterman AE, Clemmer DE. J. Am. Soc. Mass Spectrom 1997;8:954.
- 24. Li J, Taraszka JA, Counterman AE, Clemmer DE. Int. J. Mass. Spectrom 1999;185/186/187:37.
- 25. Shelimov KB, Clemmer DE, Hudgins RR, Jarrold MF. J. Am. Chem. Soc 1997;119:2240.
- 26. Shelimov KB, Jarrold MF. J. Am. Chem. Soc 1997;119:2987.
- 27. Kohtani M, Jones TC, Sudha R, Jarrold MF. J. Am. Chem. Soc 2006;128:7193. [PubMed: 16734471]
- 28. Shvartsburg AA, Li FM, Tang K, Smith RD. Anal. Chem 2006;78:3304. [PubMed: 16689531]
- 29. Purves RW, Barnett DA, Guevremont R. Int. J. Mass Spectrom 2000;197:163.
- 30. Purves RW, Barnett DA, Ells B, Guevremont R. J. Am. Soc. Mass Spectrom 2001;12:894. [PubMed: 11506222]
- 31. Purves RW, Ells B, Barnett DA, Guevremont R. Can. J. Chem 2005;83:1961.
- 32. Borysik AJH, Read P, Little DR, Bateman RH, Radford SE, Ashcroft AE. Rapid Commun. Mass Spectrom 2004;18:2229. [PubMed: 15384141]
- 33. Shvartsburg AA, Bryskiewicz T, Purves RW, Tang K, Guevremont R, Smith RD. J. Phys. Chem. B 2006;110:21966. [PubMed: 17064166]
- 34. Guevremont, R.; Purves, RW. US Patent. 7,250,306. 2007.
- 35. Myung S, Badman ER, Lee YJ, Clemmer DE. J. Phys. Chem. A 2002;106:9976.
- 36. Barnett DA, Belford M, Dunyach JJ, Purves RW. J. Am. Soc. Mass Spectrom 2007;18:1653. [PubMed: 17662612]
- 37. Shvartsburg AA, Li FM, Tang K, Smith RD. Anal. Chem 2006;78:3706. [PubMed: 16737227]
- 38. Guevremont R, Purves RW. Rev. Sci. Instrum 1999;70:1370.
- 39. Shvartsburg AA, Tang K, Smith RD. J. Am. Soc. Mass Spectrom 2004;15:1487. [PubMed: 15465362]
- 40. Buryakov IA. Talanta 2003;61:369. [PubMed: 18969196]

- 41. Karpas Z. Anal. Chem 1989;61:684.
- 42. Krylov EV. Inst. Exp. Tech 1997;40:628.
- 43. Krylov EV, Nazarov EG, Miller RA. Int. J. Mass Spectrom 2007;266:76.
- 44. Shvartsburg AA, Tang K, Smith RD. J. Am. Soc. Mass Spectrom 2005;16:2. [PubMed: 15653358]
- 45. De la Mora JF. Anal. Chim. Acta 2000;406:93.



**Fig. 1.** Divergent temperature dependences of FAIMS spectra for protein ions of low and high charge states. Data for ubiquitin with z = 6 - 13 measured using  $E_D = -20$  kV/cm at T = 35 °C (top row) and 80 °C (bottom row).

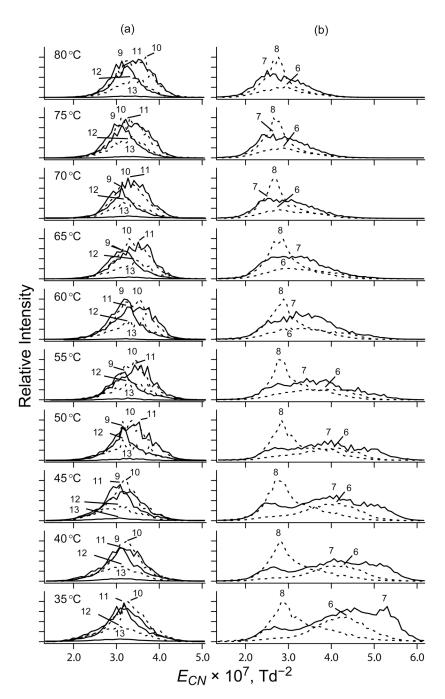


Fig. 2. Removing the dependence of spectra on the gas number density. Normalized spectra of ubiquitin ions with z=9-13 (a) and 6-8 (b) at  $E_D=-20$  kV/cm as a function of gas temperature (labeled).

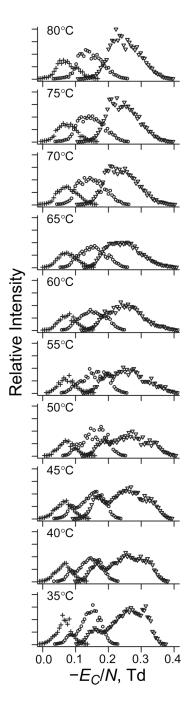


Fig. 3. Spectra of ubiquitin 6+ ions as a function of gas temperature (labeled) at  $E_D = -20 \text{ kV/cm}$  ( $\nabla$ ), -16 kV/cm ( $\circ$ ), and -12 kV/cm (+).

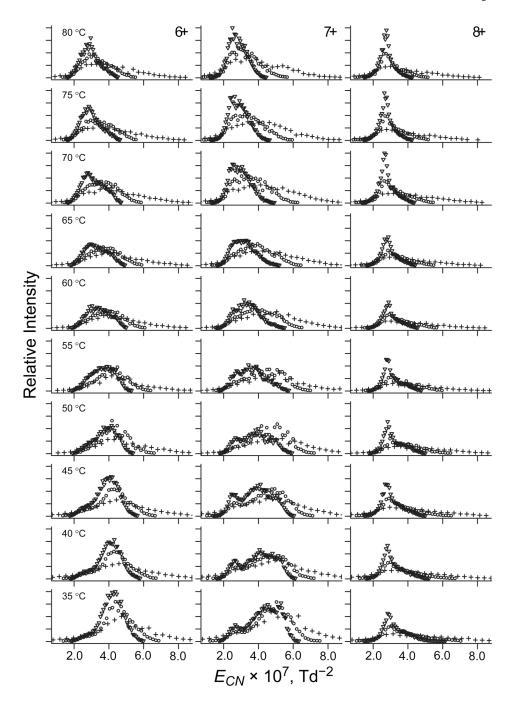


Fig. 4. Removing the dependence of spectra for same geometries on the dispersion field. Normalized spectra of ubiquitin ions with z = 6 - 8 as a function of gas temperature (labeled) at  $E_D = -20$  kV/cm ( $\nabla$ ), -16 kV/cm ( $\circ$ ), and -12 kV/cm (+).

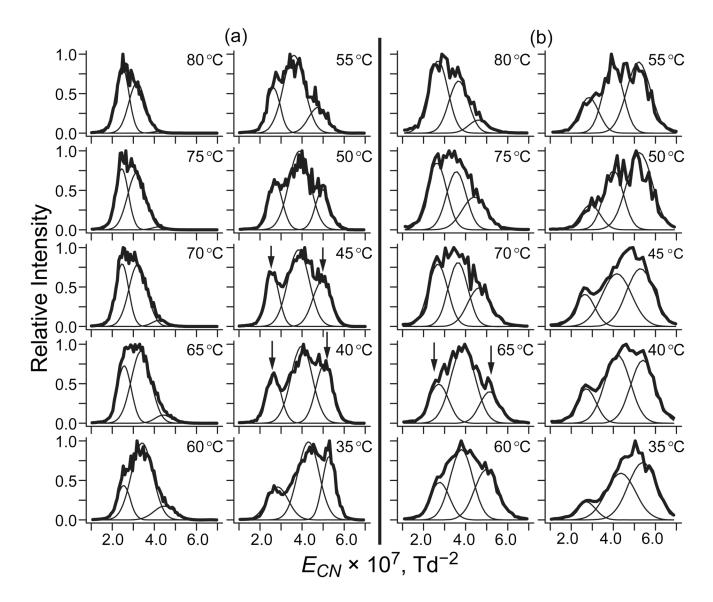


Fig. 5. Normalized spectra of ubiquitin 7+ at  $E_{\rm D} = -20~{\rm kV/cm}$  (a) and  $-16~{\rm kV/cm}$  (b) as a function of gas temperature, fitted assuming three conformers with Gaussian peak profiles. The compared features are marked by arrows.

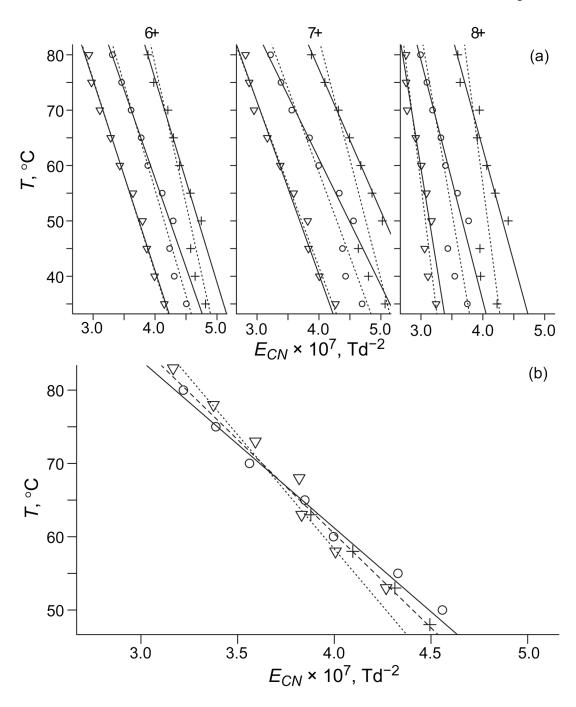


Fig. 6. Comparison of the effects of thermal and field heating on FAIMS spectra measured for ubiquitin ions. (a) Dependences of mean normalized  $E_{\rm C}$  on gas temperature for z=6-8 at  $E_{\rm D}=-20$  kV/cm ( $\nabla$ ), −16 kV/cm ( $\circ$ ), and −12 kV/cm (+). First-order regressions through each set including points for T=35-80 °C and 50-80 °C are marked by solid and dashed lines, respectively. (b) For z=7, the minimum scatter of data for three  $E_{\rm D}$  values is achieved via displacements by +18 °C for −20 kV/cm and −17 °C for −12 kV/cm. The needed shifts are 16-19 °C for 6+ and 24-26 °C for 8+. Lines are  $1^{\rm st}$ -order regressions through sets for T=50-80 °C, at  $E_{\rm D}$  of −20 kV/cm (dotted), −16 kV/cm (solid), and −12 kV/cm (dashed).

**Table 1** Mobilities (in  $N_2$  at 1 Atm) and calculated magnitudes of field heating in FAIMS for compact and partly folded conformers of ubiquitin ions (z = 6 - 8) at  $E_1 = -12$ ,  $E_2 = -16$ , and  $E_3 = -20$  kV/cm.

$T, {}^{\circ}\mathrm{C}$	$T, {}^{\circ}C \mid K, \operatorname{cm}^2/(V \times s)$		$T_{ m H^{*}}^{\circ}{ m C}$		$\Delta T_{ m max}~(E_2-E_1)$	$\Delta T_{\max} \left( E_2 - E_1 \right) \left  \Delta T_{\max} \left( E_3 - E_2 \right) \right  \Delta T \left( E_2 - E_1 \right) \left  \Delta T \left( E_3 - E_2 \right) \right $	$\Delta T(E_{2}-E_{1})$	$\Delta T(E_{3}-E_{2})$
		$E_1$	$E_1$ $E_2$ $E_3$	$E_3$				
35	0.99–1.32   16–28   28–50   44–78	16–28	28–50	44–78	12–22	16–28	3.4–6.1	4.4–7.8
08	1.03–1.38   17–31   30–55   48–85	17–31	30–55	48-85	13–24	17–31	3.7–6.6	4.8–8.6