

Ultraslow oligomerization equilibria of p53 and its implications

Eviatar Natan^a, Daniel Hirschberg^b, Nina Morgner^b, Carol V. Robinson^b, and Alan R. Fersht^{a,b,1}

^aMedical Research Council Centre for Protein Engineering, Cambridge CB2 0QH, United Kingdom; and ^bDepartment of Chemistry, Cambridge University Chemical Laboratory, Lensfield Road, Cambridge CB2 1EW, United Kingdom

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The tumor suppressor p53 is in equilibrium at cellular concentrations between dimers and tetramers. Oncogenic mutant p53 (mut) exerts a dominant-negative effect on co-expression of p53 wild-type (wt) and mut alleles in cancer cells. It is believed that wt and mut form hetero-tetramers of attenuated activity, via their tetramerization domains. Using electrospray mass spectrometry on isotopically labeled samples, we measured directly the composition and rates of formation of p53 complexes in the presence and absence of response element DNA. The dissociation of tetramers was unexpectedly very slow ($t_{1/2} = 40$ min) at 37 °C, matched by slow association of dimers, which is approximately four times longer than the half-life of spontaneous denaturation of wt p53. On mixing wt tetramers with the oncogenic contact mutant R273H of low DNA affinity, we observed the same slow formation of only wt₄, wt₂mut₂, and mut₄, in the ratio 1:2:1, on a cellular time scale. On mixing wt and mut with response element DNAs P21 and BAX, we observed only the complexes wt₄.DNA, wt₂mut₂.DNA, and mut₄.DNA, with relative dissociation constants 1:4:71 and 1:13:85, respectively, accounting for the dominant-negative effect by weakened affinity. p53 dimers assemble rapidly to tetramers on binding to response element DNA, initiated by the p53 DNA binding domains. The slow oligomerization of free p53, competing with spontaneous denaturation, has implications for the possible regulation of p53 by binding proteins and DNA that affect tetramerization kinetics as well as equilibria.

mass spectrometry | protein-protein interactions | slow association | tetramerization domain

The tumor suppressor p53 is a transcription factor that is inactivated by mutation in some 50% of human cancers (1, 2). p53 consists of an unstructured N-terminal domain (residues 1–94) which points away from the rest of the protein (3), a folded core domain (residues 94–292), a linker to the tetramerization domain (residues 325–355) and an unstructured C-terminal domain (356–394). Nearly all of the mutations reside in its DNA-binding domain. Mutations can be either ‘contact,’ which remove residues that interact with DNA and lower affinity, or ‘structural,’ which destabilize the protein, sometimes with concomitant conformational changes (4). The transcriptionally active form of p53 is a homo-tetramer, linked via its tetramerization domains. Since most cancer mutations are located outside the tetramerization domain, in the core domain, the mutant (mut) proteins retain wild-type (wt) ability to form tetramers. Consequently the formation of hetero-tetramer complexes between wt and mut p53 is observed both in vivo and in vitro (5). It is thought that formation of hybrids between the wt and the mut protein is the cause of the *trans*-dominant effect of the mut over the wt in heterozygous cells (dominant-negative effect), either following somatic mutation or from birth as in the case of Li-Fraumeni syndrome (5, 6). In such events, the chances of initiating cancer are increased (7), and the p53 heterozygous cell may enhance invasion and migration of the tumor (8). The extent of the dominant-negative effect is mutant-specific (9). The exact natures of wt and mut homo-tetramers, as well as the dynamics of tetramer formation are only poorly understood. The equilib-

rium constant for the formation of the tetramers, approximately 20 nM, is such that there is an equilibrium between dimers and tetramers in the cell, and dimer-tetramerization phenomena may be important in the regulation of the activity of p53 (10–15).

It is difficult to analyze equilibria between complexes of p53 and DNA in mixtures of wt and mut proteins using standard bulk measurements. However, mass spectrometry (MS) gives another route for direct study of discrete populations of molecules. Intact protein complexes of masses up to megaDaltons can be studied using soft ionization electro-spray (ESI-MS) techniques (16–18). The kinetics of formation of homo-protein complexes and the exchange of subunits can be monitored using ESI-MS and isotopically labeled samples, and mass spectrometry can be used as a quantitative tool for characterizing protein complexes (19–21). Here, we have used ESI-MS to determine the dissociation and assembly kinetics of the subunits of the wt p53 tetramer in a framework, stabilized by four mutations in the core domain that do not perturb the structure (22, 23) that allows the protein to be studied over extended periods (22, 23). All experiments were performed on the core plus tetramerization domain construct (residues 94–356) which has the same quaternary structure as full-length protein (24). Parallel experiments were done on the R273H contact mutant (25), which has the same stability as the other construct (26).

We measured the composition and the kinetics of formation of complexes formed on mixing wt p53 with the most frequent contact mutant, R273H. We also measured directly the complexes formed between response element DNAs in the homo- and hetero-hybrids of wt and mut. We discovered that the formation and exchange of complexes are very slow in the absence of DNA, and that only a subset of the possible combinations of wt and mut hybrids were formed. We were also able to quantify the affinity for different compositions of wt and contact mut hybrids for two representative response element DNAs.

Results

Exchange Rate of p53 Subunits. The exchange of p53CT subunits was studied using an equimolar solution of an unlabeled protein (¹²C–¹⁴N = “wtL”) with expected molecular weight of 120.2 kDa and a uniformly labeled protein with expected molecular weight of 127.0 kDa (¹³C–¹⁵N = “wtH”). The measured values were identical, and the experimental errors were less than ± 0.020 kDa. The initial concentration of wtL₄ and wtH₄ were 30 μ M each, in monomers (= 7.5 μ M tetramers) as verified by several techniques. The equilibrium constant for the formation of

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¹To whom correspondence should be addressed. E-mail: arf25@cam.ac.uk.

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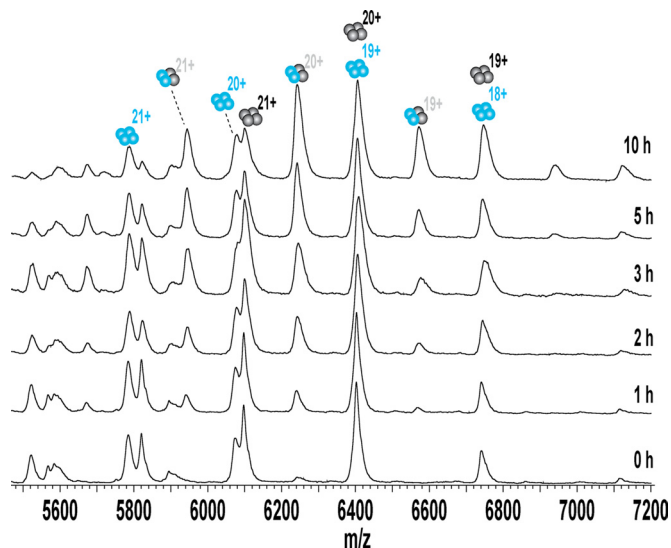


Fig. 1. Exchange of p53CT subunits during incubation at 20 °C as determined by ESI-MS. Selected MS spectra of ^{12}C - ^{14}N (blue) and ^{13}C - ^{15}N (black) p53CT at different time points. The numbers represent the charge states of the respective peaks. No species other than wtL₂wtH₂ was observed for up to 10 days.

tetramers from dimers is approximately 20 nM, so the starting species were nearly 100% tetrameric.

The spectra at different time points in Fig. 1, illustrate the subunit exchange of wt at room temperature. The natural and heavy isotopically labeled p53CT, wtL₄, and wtH₄ homotetramers, were essentially the only species present immediately after mixing and had their expected molecular weights. A new set of peaks corresponding to wtL₂wtH₂ emerged in the spectrum taken after 0.5 h, which increased in intensity until equilibrium was reached after approximately 10 h. No further hybrids were seen up to 10 days of incubation. To determine the relative abundance of the different species, we simulated the mass spectra with an in-house program based on LabVIEW, which deconvolutes each unresolved peak into the sum of its components (27). The relative abundances for each species were then plotted against time and fitted to a simple first-order exponential (Fig. 2A). The half time for exchange was 2.3 h. The equilibrium constants between wtL₄, wtL₂wtH₂ and wtH₄ were directly calculated from the ratios of intensities at equilibrium. The same exchange pattern and rates were observed using the R273H mutant (Fig. 2B). The same results were observed for incubation of p53 wt with p53 R273H. Again, only 2:2 hetero-tetramers were formed during co-incubation, with the same rate constants as found for the homo-tetramer. Repetition of the experiments between 4–65 μM (monomers) using different labeling regimes gave the same rate constant, indicating that data were independent of concentration in a range well above the dissociation constant between wt₄ and wt₂.

The same experiments were repeated at 37 °C. After ≈2 h, the system reached equilibrium for the primary dimer exchange (Fig. 3), 5-fold faster than incubation at 20 °C with $t_{1/2}$ of 40 min. But, sets of peaks corresponding to L₃H₁, and L₁H₃ were also observed. The protein was not stable for long enough to reach the theoretical equilibrium distribution of 1:4:6:4:1 for mutL₄, mutL₃mutH₁, mutL₂mutH₂, mutL₁mutH₃, mutH₄, respectively. It should be noted that wild-type protein that does not contain the four stabilizing mutations has a half life of only 9–16 min at 37 °C (28).

It can be shown for the reaction:

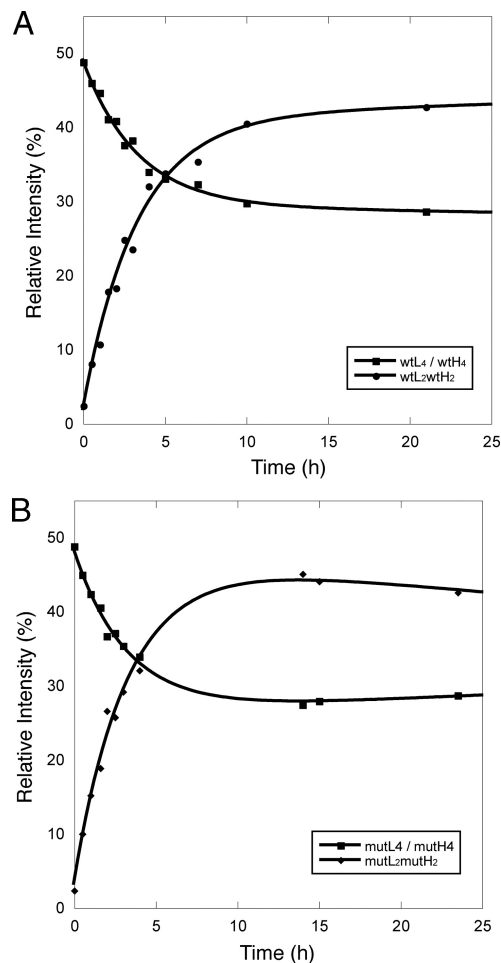
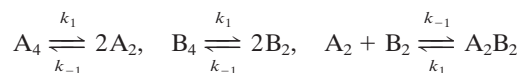


Fig. 2. Exchange of p53CT wt-wt and mut-mut 20 °C. Both have similar kinetics and patterns. (A) wtL (^{12}C - ^{14}N)/wtH (^{13}C - ^{15}N) exchange, as shown in Fig. 1. The relative intensities of the species were calculated from the spectra and plotted against time. (B) mutL/mutH exchange. The relative intensities of the species were calculated from the spectra and plotted against time. For both A and B, we plotted the average of L₄ and H₄ (L₄/H₄).

where, for example, A and B are different forms of p53 monomers, at concentrations well above the dissociation constant for formation of tetramers, the rate constant for exchange is k_1 .

Binding of DNA to p53 Homo- and Heterotetramers. The exchange of subunits is immediately quenched by the addition of response element DNA so binding of DNA may be measured without perturbing the equilibrium. We measured the binding of an equimolar mixture of the preequilibrated wt and R273H mut tetrameric p53 to a 29-residue, double-stranded oligomer containing the P21 DNA response element sequence, which is of high affinity (29), and to a 31 residue oligomer containing the low affinity BAX sequence (29). The mixture of wt and R273H might simulate the hybrids found in vivo in heterozygous cells. Incubation was at 4 °C for 24 h to form a 1:2:1 distribution of active protein. Then, various concentrations of response element DNA were added, the binding of which equilibrated within the 5 min of sample preparation time. The center of gravity of the spectra was progressively shifted with increasing DNA concentration. First, the peaks corresponding to wt₄.DNA appeared followed by wt₂mut₂.DNA and, finally, mut₄.DNA. Sequentially, the peaks of wtL₄ disappeared first, followed by wt₂mut₂. DNA-free mut₄, was still present at the highest concentration of DNA, even for the high affinity response element P21. The mut₄.DNA

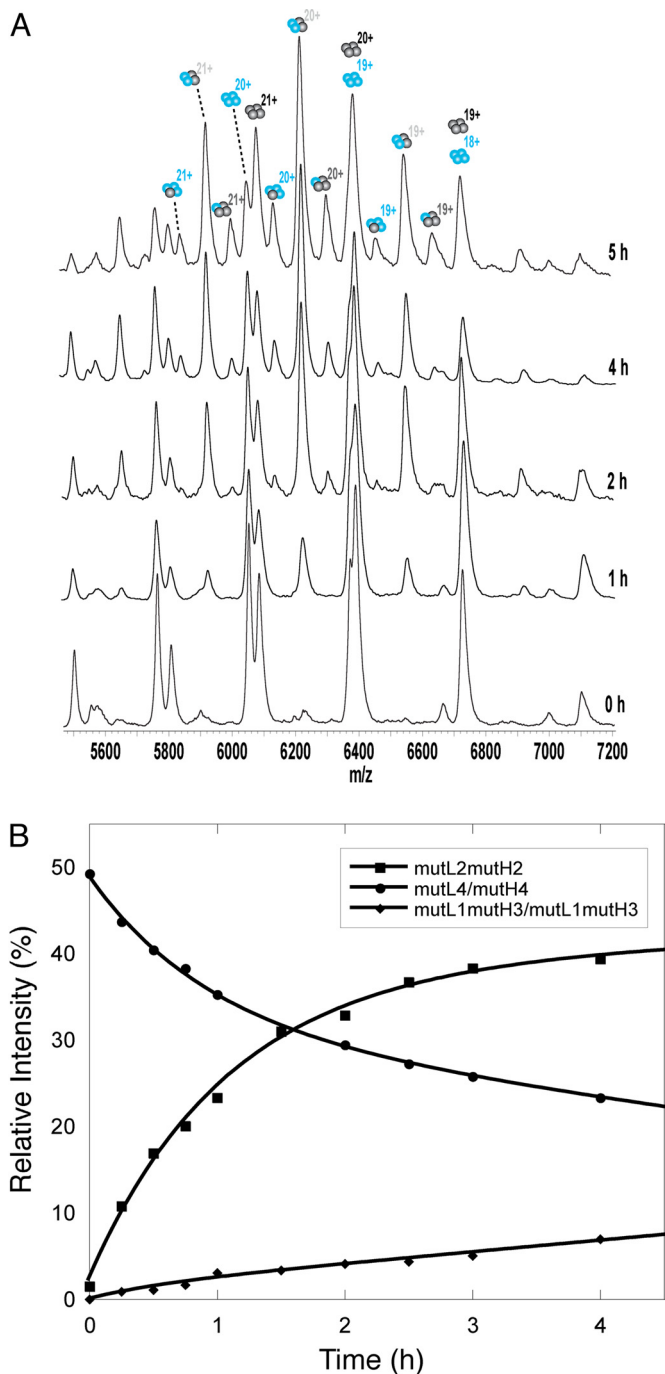


Fig. 3. Exchange of p53CT R273H subunits during incubation at 37 °C. (A) MS spectra over time of ^{12}C - ^{14}N (blue) with ^{13}C - ^{15}N (black). Additionally 1:3, 3:1 hetero-tetramer species appear that were not observed at 20 °C. (B) The relative intensities of each species, mutL₄, mutL₃mutH₁, mutL₂mutH₂, mutL₁mutH₃, and mutH₄, were extracted from the spectra using simulation and then plotted against time. Equilibrium was not achieved because of aggregation of the protein after 6 h. The same results were found for wt-mut.

complex did not form until sufficient DNA had been added that all of the wt₄ and wt₂mut₂ had been bound, leading to a lag in the binding curve.

Calculation of Absolute K_D s of wt₄, wt₂mut₂, and mut₄ Complexes with DNA. The concentrations of protein used in the MS-ESI experiments (~1 μM in tetramers) were too far above the K_D s of the

wt₄-P21DNA complexes for accurate determination of dissociation constants. The low concentrations of added DNA were nearly all bound to the wt₄ protein and the wt₂mut₂ so that the concentration of unbound DNA was hard to calculate accurately (Fig. 4A). The concentration of p53 in the sample was verified by a stoichiometric titration of P21 DNA into the preincubated mixture of p53, which was about 100 times above its dissociation constant of tetramers from DNA (Fig. 5). The concentration thus measured was ~80% of that calculated from the dilutions, which is very reasonable given that there could be handling losses and some denaturation. The fractions bound were plotted against the concentration of free protein, after subtraction of the bound protein from the total (Fig. 4B and C), and fitted to a simple Michaelis-Menten equation to calculate the K_D s.

The weak binding to mut₄ allowed a reasonable measurement of the binding isotherm since it was effectively titrated after all of the other two complexes were saturated with DNA and when there was excess unbound DNA over protein. The K_D for the R273H tetramer bound to P21 DNA was $1.47 \pm 0.17 \mu\text{M}$. Calculation of the K_D s for the other two complexes required accurate measurement of the amount of DNA bound to the protein and subtracting it from the total added DNA to give the free DNA. The values thus calculated for P21 bound to wt₄ and wt₂mut₂ were very approximately, because of the uncertainties, $8 \pm 5 \text{ nM}$ and $57 \pm 25 \text{ nM}$ (Fig. 4B). The K_D s for binding to BAX DNA were more accurately determined for the weaker binding of BAX DNA (Fig. 4C), and were $184 \pm 5 \text{ nM}$, $3.4 \pm 0.3 \mu\text{M}$, and $18 \pm 1.5 \mu\text{M}$ for wt₂, wt₂mut₂, and mut₄, respectively.

Calculation of Relative K_D s of wt₄, wt₂mut₂, and mut₄ Complexes with DNA.

We used a method for the direct calculation of the relative affinities of the different proteins for the response elements that avoided any estimation of the concentrations of unbound DNA under conditions when most was bound to protein. For a simple 1:1 binding of tetrameric p53 to a response element DNA containing four subsites: $K_w = [\text{wt}_4][\text{DNA}]/[\text{wt}_4.\text{DNA}]$; $K_{w/m} = [\text{wt}_2\text{mut}_2][\text{DNA}]/[\text{wt}_2\text{mut}_2.\text{DNA}]$; and $K_m = [\text{mut}_4][\text{DNA}]/[\text{mut}_4.\text{DNA}]$. In a mixture of all of the protein components and DNA, the concentration of free DNA is the same for all equilibria. Therefore, we can calculate the ratios of all of the equilibrium constants from taking the ratios of the free and bound proteins thus: $K_{w/m}/K_w = [\text{wt}_2\text{mut}_2][\text{wt}_4.\text{DNA}]/[\text{wt}_4][\text{wt}_2\text{mut}_2.\text{DNA}]$ and $K_m/K_{w/m} = [\text{mut}_4][\text{wt}_2\text{mut}_2.\text{DNA}]/[\text{wt}_2\text{mut}_2][\text{mut}_4.\text{DNA}]$. Further, we can calculate the ratios of $[\text{wt}_4]/[\text{wt}_4.\text{DNA}]$, $[\text{wt}_2\text{mut}_2]/[\text{wt}_2\text{mut}_2.\text{DNA}]$, and $[\text{mut}_4]/[\text{mut}_4.\text{DNA}]$ at each point quite simply. If the amplitude of, say, the signal for wt₄.DNA at any DNA concentration, [D], is x_D , and the amplitude at saturating DNA is x_∞ , then $[\text{wt}_4]/[\text{wt}_4.\text{DNA}] = (x_\infty - x_D)/x_D$. Accordingly, we do not need to know the absolute concentrations of DNA or protein to calculate the ratio of K_D s. The method is most accurate for the regions where x_D is significantly above zero and below x_∞ , that is, [DNA] is close to the K_D values. The range of [DNA] used, $0, 2.8 \times 10^{-8}$ to 10^{-5} M , was enough to span the K_D s and have binding to wt₄ and mut₄ close to saturation. But the weak binding of DNA to the mut₄ species precluded direct measurement of the saturating amplitude so we assume that it was the same as for wt₄. (The amplitude of the wt₂mut₂ was twice that of wt₄, as expected.)

For P21 DNA: $K_{w/m}/K_w = 3.8 \pm 0.3$; $K_m/K_{w/m} = 18.7 \pm 8$; and so $K_m/K_w = 71 \pm 31$.

The values calculated from the ratios of individual K_D s were in acceptable agreement to those measured directly, $K_{w/m}/K_w = \sim 7$; $K_m/K_{w/m} = \sim 26$. For BAX DNA: $K_{w/m}/K_w = 12.9 \pm 1.9$; $K_m/K_{w/m} = 6.6 \pm 0.8$; and so $K_m/K_w = 85 \pm 16$. The values calculated from the ratios K_D s are in reasonable agreement, $K_{w/m}/K_w = \sim 18$; $K_m/K_{w/m} = \sim 5$.

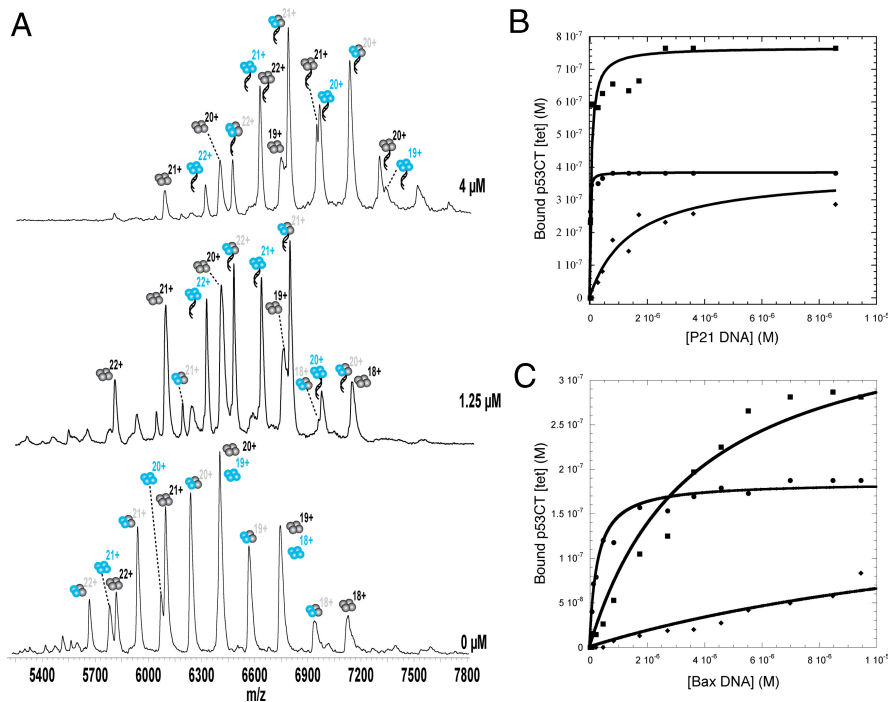


Fig. 4. Binding of P21 and BAX DNA response elements to preequilibrated wt p53CT (blue) and p53CT R273H (black). (A) MS spectra of wt and mut at different P21 DNA concentrations. (B and C) Fitting of intensities to simple binding isotherms: wt₄.DNA (filled circles); wt₂mut₂.DNA; and (filled squares) mut₄.DNA (filled diamonds).

Discussion

The use of ESI-MS has allowed us to measure the association-dissociation kinetics of p53 subunits and also the affinity of mutant tetrameric protein for DNA. The rate constant for the dissociation of tetrameric protein was surprisingly and unexpectedly low. The half-life for the dissociation into dimers at 20 °C was 2.3 h and the only species of mixed hybrids formed between wild-type and mutant p53 was wt₂mut₂, without any of the 3:1 and 1:3 species. The half-life was 40 min at 37 °C. The dissociation constant of tetrameric p53 into dimers is approximately 20 nM, which means that at a concentration of approximately 20 nM

p53 (20 nM of dimers) the half-life for association of dimers into tetramers is also 40 min. The second-order rate constant for association of dimers into tetramers is only approximately $1.4 \times 10^4 \text{ s}^{-1} \text{ M}^{-1}$, some one to two orders of magnitude lower than for most protein-protein interactions.

These experiments would not have been possible using wild-type p53 but required the superstable quadruple mutant that has been key to our structural studies (22, 24, 25, 30–32): a half-life of 40 min at 37 °C for formation of tetramers is longer than the half-life of 9–16 min for the spontaneous denaturation of wild-type p53 at 37 °C (28). The combination of the high instability of wild-type p53 and its slow oligomerization has profound implications: unless there are special factors within the cell, p53 will spontaneously denature faster than it will form active tetramers. However, there are routes for p53 to associate other than the initial dimerization of a pair of dimers via their tetramerization domains. For example, very dilute solutions of p53 that are mainly dimeric form DNA-bound tetramers within seconds when added to response element DNA (33). This reaction must occur by the rapid binding of 1 dimer to one pair of DNA binding sites, followed by the second dimer to the second pair of sites—DNA is thus a catalyst for association. The finding of slow association rate constants shows that the cooperative binding of p53 dimers to DNA is the result of a stepwise addition of dimers rather than resulting from a fast preequilibrium between dimeric and tetrameric p53.

We speculate that the slow association and dissociation rate constants for oligomerization might be part of a regulatory mechanism of p53 for processes where the presence of either dimers or tetramers is important for function or trafficking. It is possible that there may be proteins (or other DNAs) that can catalyze the association of p53. For example, we have recently found that 14–3-3γ enhances the association equilibrium of p53 by binding to its C-termini (34). So, there may be a further mode of regulation of p53 by proteins altering the rate of p53 association.

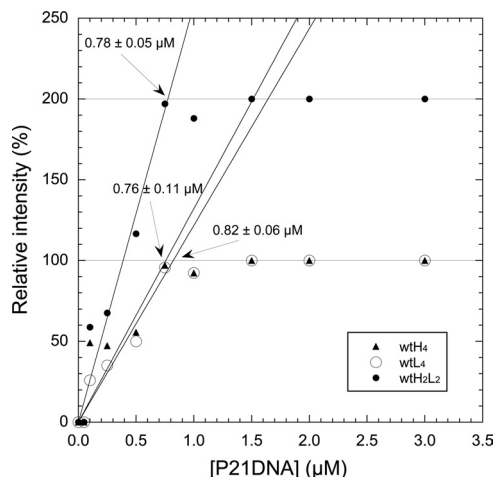


Fig. 5. Determination of active p53 from stoichiometric binding of P21 DNA. The DNA was added to a nominal concentration of 1 μM total of tetramers of wtH₄ and wtL₄ that had been preincubated to equilibrate. Each resultant complex could be followed independently. The concentration of total tetramers at the equivalence point was $0.785 \pm 0.019 \mu\text{M}$.

The slow rate of shuffling of subunits also has implications for the dominant negative effect of mutants where wild-type and mutant alleles are present. For contact mutants that are as stable as wild type, there will be formed just the species wt₄, wt₂mut₂, mut₄ in the ratio 1:2:1, which, given the affinities we have measured here, will lower the effective concentration of wt₄ tetramers competent for binding by about a factor of 4, rather than closer to 16 expected for complete shuffling of subunits in the ratio 1:4:6:4:1. Unstable structural mutants that denature far faster than they can oligomerize with wild type will effectively lower the concentration of wild-type p53 by a factor of 2.

Although we do not know the full reasons for the high instability and slow rates of association of p53, it would seem that they are part of a complex scheme for regulation of p53 that invokes a myriad of active networks with secondary features of inherent safeguards for the spontaneous denaturation of p53. The slow rates of equilibration of p53 has to be considered in interpreting biophysical data.

Materials and Methods

Materials. The double-stranded P21 and BAX oligonucleotides TGAGGAACAT-GTCCCAACATGTTGAGCTC and TGGGCTCAAGTTAGAGACAAGCCTGGCGC, respectively, were purified by high-pressure liquid chromatography.

Protein Expression and Purification. The p53 constructs were expressed in *Escherichia coli* BL21 and purified as described below. Human p53CT construct (94–356) and p53CT R273H were cloned into a pET24a-HLTV plasmid containing a 5' extension expressing a His₆-tag, a lipoamyl domain and a TEV protease cleavage site. p53CT is a superstable pseudowild-type mutant with the following mutations in the core domain: M133L/V203A/N239Y/N268D.

p53CT and p53CT R273H were grown at 37 °C in 2 × TY medium with 30 mg/mL kanamycin, and 0.1 mM Zn²⁺. Isopropyl 1-thio-β-D-galactopyranoside (IPTG) was added to a final concentration of 1 mM to induce expression. The cells were incubated for 15 h at 25 °C before they were harvested and the proteins were purified as described (24, 30) except expression of ¹³C- and ¹⁵N-labeled p53 was in M9 minimal medium supplemented with a vitamin mix together with 1 g/L ¹⁵NH₄Cl and ¹³C D-glucose (30). Protein concentration was determined spectrophotometrically using a molar extinction coefficient ε₂₈₀ = 20,400 M⁻¹cm⁻¹. The concentration was verified by amino acid analysis (Protein and Nucleic Acid Chemistry Facility, Department of Biochemistry, University of Cambridge).

Mass Spectrometry. The buffer was exchanged to 500 mM ammonium acetate, pH 6.9, before spectrometry, using Micro Bio-Spin 2–4 columns (Bio-Rad). Nanoflow electrospray mass spectra (nESI-MS) were recorded on a Synapt HDMS system and an LCT Premier instrument (Waters Corp.) optimized for the transmission of noncovalent complexes (35). Typically, 3 μL solution contain-

ing p53 were electrosprayed from gold-coated glass capillaries (36). To preserve the non-covalent interactions in the p53 tetramer, the MS parameters used for the Synapt were: capillary voltage, 1.3–2.0 kV; sample cone, 150 V; trap and transfer collision energy, 100 V; backing pressure 5 mbar, source pressure 6–7 × 10⁻² mbar; trap and IMS pressure 5 × 10⁻² and 5 × 10⁻¹ mbar, respectively; time-of-flight analyser pressure 2 × 10⁻⁶ mbar. The following parameters were used or the LCT Premier: capillary voltage, 1.1–2.0 kV; sample cone, 200 V; ion guide 1, 150 V; aperture 1, 150 V; ion energy, 55 V; source pressure 2.8 mbar; time-of-flight analyser pressure 4.5 × 10⁻⁷ mbar. All spectra were calibrated internally using a solution of cesium iodide (100 mg/mL). Data were processed with MassLynx 4.0 software (Waters/Micromass) with minimal smoothing and without background subtraction.

Spectra Simulation and Kinetic Modeling. The spectra were quantitatively analyzed with the LabVIEW program (27). We generated a simulated charge-state distribution for each of the component complexes, which were modeled using three parameter Gaussians. The observed *m/z* values of the complexes were used as midpoints and the peak width was chosen according to the observed broadness of the experimental peaks. The peak heights were scaled with a three-parameter Gaussian curve that was fitted for every complex across the tops of the corresponding peaks in the spectra. The derived simulations of the components were summed and the parameters optimized to achieve maximal agreement of the summed simulation with the experimental mass spectra (Fig. S1). The individual simulations give the peak areas for the whole charge distribution of each complex, even where peaks belonging to different species overlap. These values were used to calculate the relative abundance of the different species, which were expressed as a percentage of the total peak intensity and plotted against time.

DNA Binding Measurements by Mass Spectrometry. The wild-type and mutant homo-tetramer samples were mixed and incubated at 4 °C rather than room temperature to prevent any loss of DNA binding as a consequence of denaturation of the core. The incubation lasted until the binomial distribution of 1:2:1 for wt₄:wt₂mut₂:mut₄ was reached, as verified by MS. DNA binding assays were then performed at equilibrium. A concentration of 55–65 μM protein was diluted 5 times in 500 mM ammonium acetate with or without DNA oligomers. The final concentration of the DNA oligomers ranged from 25 nM to 10 μM.

Relative DNA-Binding Analysis. The relative abundance of the bound p53 forms, that is, wt₄.DNA, wt₂mut₂.DNA, mut₄.DNA, to the response elements was assessed from the sum of the associated peak intensities of all charge states for each mass spectrum. Then, this value was divided by the total peak intensity in the spectra and presented as a percentage. Each spectrum represents one concentration point, and the plot was the percentage bound species versus concentration of ligand.

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