

# Highly sensitive quantification of serum malonate, a possible marker for de novo lipogenesis, by LC-ESI-MS/MS

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**Abstract** We describe a new sensitive and specific method for the quantification of serum malonate (malonic acid, MA), which could be a new biomarker for de novo lipogenesis (fatty acid synthesis). This method is based upon a stable isotope-dilution technique using LC-MS/MS. MA from 50  $\mu$ l of serum was derivatized into di-(1-methyl-3-piperidinyl)malonate (DMP-MA) and quantified by LC-MS/MS using the positive electrospray ionization mode. The detection limit of the DMP-MA was approximately 4.8 fmol (500 fg) (signal-to-noise ratio = 10), which was more than 100 times more sensitive compared with that of MA by LC-MS/MS using the negative electrospray ionization mode. The relative standard deviations between sample preparations and measurements made using the present method were 4.4% and 3.2%, respectively, by one-way ANOVA. Recovery experiments were performed using 50  $\mu$ l aliquots of normal human serum spiked with 9.6 pmol (1 ng) to 28.8 pmol (3 ng) of MA and were validated by orthogonal regression analysis. The results showed that the estimated amount within a 95% confidence limit was  $14.1 \pm 1.1$  pmol, which was in complete agreement with the observed  $\bar{X}_0 = 15.0 \pm 0.6$  pmol, with a mean recovery of 96.0%. **■** This method provides reliable and reproducible results for the quantification of MA in human serum.—Honda, A., K. Yamashita, T. Ikegami, T. Hara, T. Miyazaki, T. Hirayama, M. Numazawa, and Y. Matsuzaki. **Highly sensitive quantification of serum malonate, a possible marker for de novo lipogenesis, by LC-ESI-MS/MS.** *J. Lipid Res.* 2009. 50: 2124–2130.

**Supplementary key words** acetyl-CoA carboxylase • carnitine palmitoyl transferase 1 • fatty acid synthase • liquid chromatography-electrospray ionization-tandem mass spectrometry • malonic acid • malonyl-CoA • malonyl-CoA decarboxylase

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Acetyl-CoA carboxylase (ACC) is the rate-controlling enzyme in the fatty acid biosynthetic pathway, and catalyzes the formation of malonyl-CoA from acetyl-CoA plus bicarbonate. Malonyl-CoA is not only substrate for fatty acid synthase (FAS) but is also a potent inhibitor of carnitine palmitoyl transferase 1 (1), the rate-limiting enzyme of fatty acid  $\beta$ -oxidation. Therefore, malonyl-CoA is a key molecule that controls fatty acid metabolism in the body. In addition, recent studies have shown that the level of hypothalamic malonyl-CoA is dynamically regulated by fasting and feeding and that it alters subsequent feeding behavior (2).

To determine ACC activity in tissues, an invasive tissue biopsy is necessary. However, whole body synthesis of fatty acid may be evaluated by the quantification of serum malonyl-CoA metabolites. This concept originates from our previous studies, which showed that serum concentrations of the immediate products of the rate-controlling enzymes in cholesterol and bile acid biosynthetic pathways reflected the activities of the rate-controlling enzymes and whole body cholesterol and bile acid biosynthesis (3). Furthermore, patients with malonyl-CoA decarboxylase (MCD) deficiency, who must have increased tissue malonyl-CoA concentrations, are characterized by markedly elevated urinary malonic acid (MA), called “malonic aciduria” (4). This phenomenon suggests that malonyl-CoA is easily hydrolyzed into MA by an unidentified tissue thioesterase(s). Therefore, we thought that serum MA concentrations might well reflect total body FAS.

Abbreviations: ACC, acetyl-CoA carboxylase; DMP-MA, Di-(1-methyl-3-piperidinyl)malonate; FAS, fatty acid synthase; MA, malonic acid (malonate); MCD, malonyl-CoA decarboxylase; MMA, methylmalonic acid (methylmalonate); N-ESI, ESI in negative mode; P-ESI, ESI in positive mode; SA, succinic acid (succinate); SRM, selected reaction monitoring.

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Although methodological reports for the quantification of serum MA are not available, there have been some reports that describe the methods for the determination of urinary MA levels in patients with MCD deficiency by gas chromatography (5, 6) or gas chromatography-mass spectrometry (7). In these methods, urinary organic acids were extracted with ethyl acetate and converted into trimethylsilyl derivatives before analysis. Alternatively, blood malonylcarnitine has been measured for the diagnosis of MCD deficiency using liquid chromatography-tandem mass spectrometry coupled with electrospray ionization mode (LC-ESI-MS/MS) (8). However, because all of these methods were developed to diagnose markedly elevated MA levels in patients with MCD deficiency, the authors did not pay significant attention to the sensitivities of the methods.

The aim of this study was to measure serum MA concentrations in normal human subjects with sufficient sensitivity and specificity. For this purpose, serum MA was derivatized into di-(1-methyl-3-piperidinyl)malonate (DMP-MA) and quantified using positive LC-ESI-MS/MS (LC-P-ESI-MS/MS).

## MATERIALS AND METHODS

### Chemicals

MA and [ $^{13}\text{C}_3$ ]MA were purchased from Sigma-Aldrich Chemical Co. (St. Louis, MO). 3-Hydroxy-1-methylpiperidine and 2-methyl-6-nitrobenzoic anhydride were purchased from Tokyo Kasei Kogyo (Tokyo, Japan) and 4-dimethylaminopyridine and formic acid were obtained from Wako Pure Chemical Industries (Osaka, Japan). Additional reagents and solvents were of analytical grade.

### Sample collection

Blood samples were collected from healthy human volunteers. After coagulation and centrifugation at 1,500 *g* for 10 min, serum samples were stored at  $-20^\circ\text{C}$  until analysis. Informed consent was obtained from all subjects, and the experimental procedures

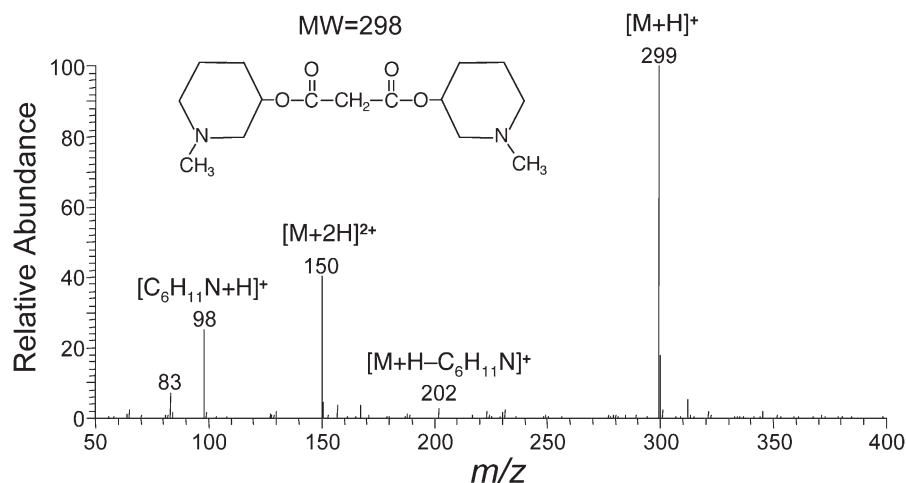
were conducted in accordance with the ethical standards of the Helsinki Declaration. Rat serum was prepared in our previous study (9) and had been stored at  $-20^\circ\text{C}$  until it was used in the present experiments.

### Sample preparation

Fifty  $\mu\text{l}$  of serum was placed in a microcentrifuge tube (1.5 ml, Eppendorf, Hamburg, Germany), and 19.2 pmol (2 ng) of [ $^{13}\text{C}_3$ ] MA in 100  $\mu\text{l}$  of acetonitrile as an internal standard. The sample tube was vortexed for 1 min and centrifuged at 2,000 *g* for 1 min. The solution of internal standard in acetonitrile led to deproteinization of the sample and the liquid phase was collected and evaporated to dryness at  $80^\circ\text{C}$  under a nitrogen stream. Derivatization of MA into DMP-MA was performed according to the Shiina method for the synthesis of carboxylic esters (10) with some modifications. The reagent mixture for derivatization consisted of 2-methyl-6-nitrobenzoic anhydride (67 mg), 4-dimethylaminopyridine (20 mg), pyridine (900  $\mu\text{l}$ ), and 3-hydroxy-1-methylpiperidine (100  $\mu\text{l}$ ). The freshly prepared reagent mixture (100  $\mu\text{l}$ ) was added to the serum extract and the reaction mixture was allowed to stand at room temperature for 30 min. After the addition of 2 ml of *n*-hexane, the mixture was vortexed for 30 s and centrifuged at 700 *g* for 2 min. The clear supernatant was collected and evaporated at  $80^\circ\text{C}$  under nitrogen. The residue was redissolved in 50  $\mu\text{l}$  of 1% formic acid in water and an aliquot (1  $\mu\text{l}$ ) was injected into the following LC-MS/MS system.

### Determination of DMP-MA by LC-P-ESI-MS/MS

The LC-MS/MS system consisted of a TSQ Quantum Ultra quadrupole mass spectrometer (Thermo Fisher Scientific, San Jose, CA) equipped with an H-ESI probe and a Nanospace SI-2 HPLC system (Shiseido, Tokyo, Japan). Chromatographic separation was performed using a Hypersil GOLD aQ column (150  $\times$  2.1 mm, 3  $\mu\text{m}$ , Thermo Fisher Scientific) at  $40^\circ\text{C}$ . Initially, the mobile phase was comprised of 0.2% formic acid in water and was used at a flow rate of 200  $\mu\text{l}/\text{min}$  for 5 min, and it was then switched to 0.2% formic acid in acetonitrile at a flow rate of 300  $\mu\text{l}/\text{min}$  for an additional 3.5 min. The general LC-MS/MS conditions were as follows: spray voltage, 1000 V; vaporizer temperature,  $350^\circ\text{C}$ ; sheath gas (nitrogen) pressure, 50 psi; auxiliary gas (nitrogen) flow, 40 arbitrary units; ion transfer capillary temperature,  $350^\circ\text{C}$ ; collision gas (argon) pressure, 1.5 mTorr; collision energy, 15 V; and ion polarity, positive.



**Fig. 1.** Typical P-ESI mass spectrum of the DMP-MA. The general LC-MS/MS conditions were as described in Materials and Methods.

### Determination of MA by LC-N-ESI-MS/MS

LC-negative (N)-ESI-MS/MS analysis of MA was carried out using the same LC-MS/MS instrument described above. Hypersil GOLD column (150 × 2.1 mm, 3 μm, Thermo Fisher Scientific) was used at 40°C. The mobile phase consisted of methanol-water (5:95, v/v) containing 0.2% formic acid and was used at a flow rate of 200 μl/min. The general LC-MS/MS conditions were as follows: spray voltage, 4000 V; vaporizer temperature, 350°C; sheath gas (nitrogen) pressure, 50 psi; auxiliary gas (nitrogen) flow, 30 arbitrary units; ion transfer capillary temperature, 300°C; collision gas (argon) pressure, 1.5 mTorr; collision energy, 15 V; and ion polarity, negative.

### Statistics

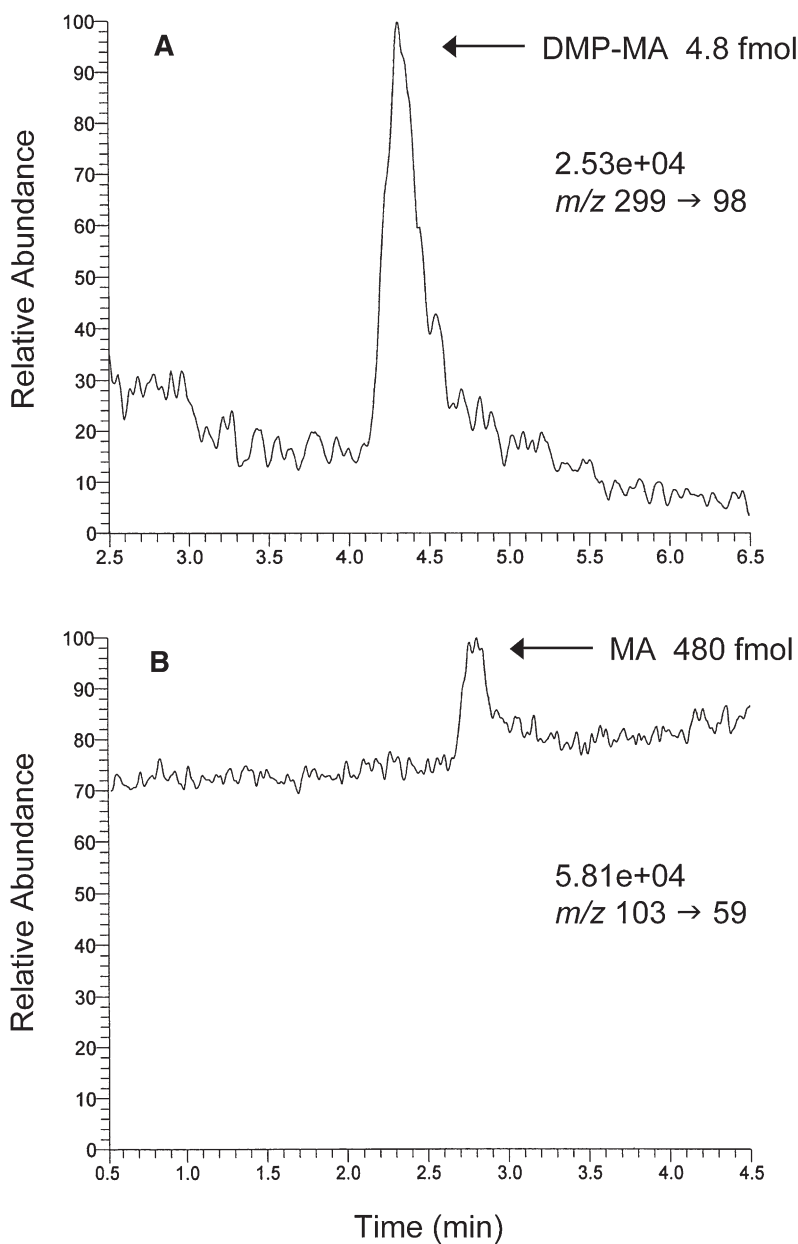
Data are reported as the mean ± SD. Linearity of the calibration curve was analyzed by simple linear regression. Reproducibility was analyzed by one-way ANOVA (JMP software, SAS Institute, Inc., Cary, NC). The estimated amount ± 95% confidence limit was obtained as an index of precision (11). To calcu-

late the values, orthogonal regression analysis was performed in the recovery study by using JMP software. For all analyses, significance was accepted at the level of  $P < 0.05$ .

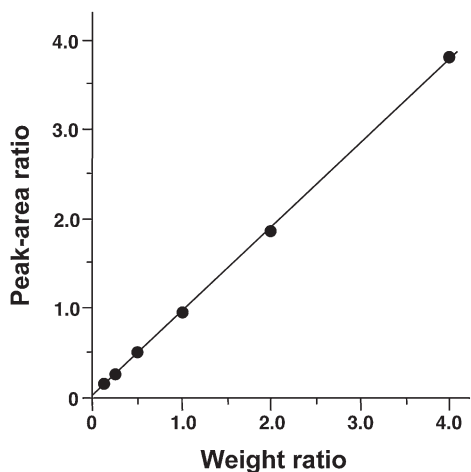
## RESULTS

### Selected reaction monitoring

A typical ESI positive mass spectrum of the DMP-MA is shown in **Fig. 1**. This DMP ester derivative exhibited  $[M+H]^+$  ion at  $m/z$  299 as the base peak. In the MS/MS spectrum using  $m/z$  299 as a precursor ion, the  $[C_6H_{11}N+H]^+$  ion was observed at  $m/z$  98 as the most prominent peak. The selected reaction monitoring (SRM) was conducted using  $m/z$  299 →  $m/z$  98 for the DMP-MA and  $m/z$  302 →  $m/z$  98 for the  $[^{13}C_3]$  variant. We also monitored  $m/z$  299 →  $m/z$  202, a product ion containing the MA molecule



**Fig. 2.** Comparison of the detection limit of DMP-MA by LC-P-ESI-MS/MS at  $m/z$  299 →  $m/z$  98 (A) with that of MA by LC-N-ESI-MS/MS at  $m/z$  103 →  $m/z$  59 (B). Authentic standard of DMP-MA (4.8 fmol) or MA (480 fmol) was injected into the HPLC. The numbers written above the SRM ion pair represent the full scale of the chromatogram.



**Fig. 3.** Calibration curve for the weight ratio of MA to the corresponding deuterated internal standard. Linearity was checked by simple linear regression and the equation for the line of best fit was  $y = 0.948x + 0.021$  ( $n = 6$ ;  $r = 1.000$ ;  $P < 0.0001$ ).

but the former showed much better signal-to-noise ratio than the latter.

By N-ESI mode, authentic MA exhibited  $[M-H]^-$  ion at  $m/z$  103 as the base peak. In the MS/MS spectrum, the  $CH_3COO^-$  ion was observed at  $m/z$  59 as the most prominent peak. The SRM was conducted using  $m/z$  103  $\rightarrow$   $m/z$  59 for the MA.

#### Comparison of the sensitivities between P-ESI and N-ESI methods

To compare the sensitivity of DMP-MA by LC-P-ESI-MS/MS with that of MA by LC-N-ESI-MS/MS, the standard DMP-MA or MA solution was diluted and injected into the LC-MS/MS system. As shown in **Fig. 2A**, the DMP-MA was easily detected to 4.8 fmol by LC-P-ESI-MS/MS, with a signal-

to-noise ratio of 10, whereas the conventional LC-N-ESI-MS/MS was barely able to detect 480 fmol of MA (**Fig. 2B**).

#### Calibration curve

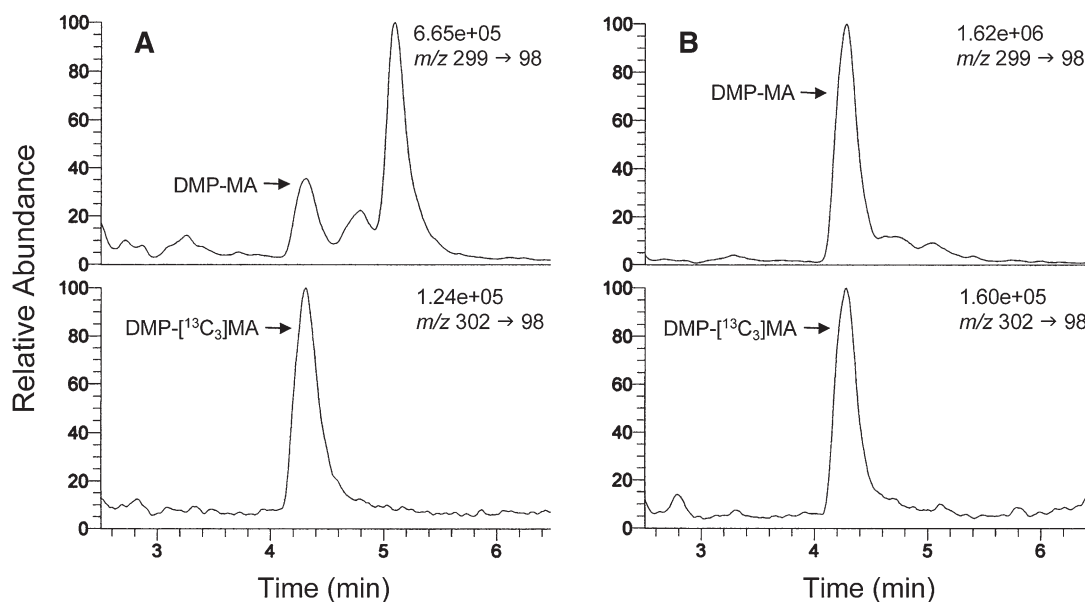
A calibration curve was established for MA (**Fig. 3**). Each of different amounts (2.4, 4.8, 9.6, 19.2, 38.5, and 76.9 pmol) of authentic MA was mixed with 19.2 pmol of  $[^{13}C_3]$ MA, derivatized to the DMP ester and quantified as described in the Materials and Methods. The weight ratio of MA, relative to the corresponding  $^{13}C$ -labeled internal standard, was plotted on the abscissa and the peak-area ratio of the DMP-MA to the  $[^{13}C_3]$  variant measured by LC-P-ESI-MS/MS was plotted on the ordinate. The linearity of the standard curve, as determined by simple linear regression, was excellent for weight ratios between 0.125 and 4.0 ( $n = 6$ ;  $r = 1.000$ ;  $P < 0.0001$ ).

#### Representative SRM

**Figure 4** shows typical SRM chromatograms for DMP-MA and the  $[^{13}C_3]$  variant obtained with 50  $\mu$ l sera from a normal human (A) and a control rat (B). The peak-area ratio of the DMP-MA to the  $[^{13}C_3]$  variant was calculated from the chromatograms, and MA amount was determined by applying the ratio to the calibration curve. The peaks of DMP-MA in chromatograms A and B correspond to  $\sim 0.66$  pmol (0.66  $\mu$ M) and  $\sim 4.43$  pmol (4.43  $\mu$ M), respectively.

#### Precision and accuracy of the LC-P-ESI-MS/MS method

The following studies were performed to determine the precision and accuracy of the present method using the same serum obtained from a normal human subject. Reproducibility was investigated by analyzing four samples in triplicate by LC-P-ESI-MS/MS (**Table 1**). The results were analyzed by a one-way ANOVA in which the analytical errors were divided into two sources, sample preparation



**Fig. 4.** Representative SRM chromatograms of DMP-MA and its  $^{13}C_3$  variant (internal standard) obtained from 50  $\mu$ l sera of a normal human (A) and a rat (B). The peaks of DMP-MA in chromatograms A and B correspond to  $\sim 0.66$  pmol (0.66  $\mu$ M) and  $\sim 4.43$  pmol (4.43  $\mu$ M), respectively. The numbers written above the SRM ion pair represent the full scale of the chromatogram.

TABLE 1. Reproducibility in the quantification of MA in human serum: analytical data

Sample	Individual Values			Mean $\pm$ SD
	<i>pmol</i>			
A	15.0	15.7	15.0	15.2 $\pm$ 0.38
B	14.4	15.6	15.5	15.2 $\pm$ 0.67
C	14.7	14.4	14.2	14.5 $\pm$ 0.29
D	15.6	15.6	14.7	15.3 $\pm$ 0.48
Mean $\pm$ SD				15.0 $\pm$ 0.58

MA was quantified in 50  $\mu$ l of normal human serum.

and SRM measurement. The variances were not considered to be attributable to the sample preparation because the errors during sample preparation were not significantly larger than those between the measurements (**Table 2**). The inter-assay coefficients of variation for the between- and within-sample variations were 4.4% and 3.2%, respectively.

For the recovery experiments, known amounts of MA (a, 2a, 3a; a = 9.6 pmol) were spiked into 50  $\mu$ l aliquots of the serum samples (n = 2). After the clean-up and derivatization procedures, SRM was carried out in triplicate for each sample. The recoveries of the known spiked amounts of MA ranged from 94.5% to 99.0%, with a mean of 96.0% (**Table 3**). In addition, the amount of endogenous MA found in unspiked 50  $\mu$ l serum aliquots was within the 95% confidence limit for the estimated amount of MA calculated by orthogonal regression analysis, which also constituted an index for the precision and accuracy of the present method.

### The circadian rhythm of MA levels in human sera

**Figure 5** depicts the circadian rhythm of the serum concentrations of MA in a healthy male. Postprandial increases of MA concentrations (maximum 235% after dinner) were observed and the levels peaked between 2.5 and 6.5 h post-meal. The increase of MA concentration disappeared after skipping breakfast on the second day, which supports the idea that the diurnal pattern of serum MA concentrations is controlled mainly by food intake.

## DISCUSSION

We describe a sensitive new LC-P-ESI-MS/MS method for the quantification of MA in serum. LC-N-ESI-MS/MS may be more suitable for the determination of negatively charged compounds, such as organic acids because the method does not require a derivatization step. However,

as shown in Fig. 2, the sensitivity of N-ESI was not sufficient to quantify MA concentrations in a small volume of normal human serum.

Recently, we derivatized another organic acid, mevalonate, into mevalonyl-(2-pyrrolidin-1-yl-ethyl)-amide and measured it using LC-P-ESI-MS/MS (12). In this method, mevalonate was lactonized into mevalonolactone and then a tertiary amine moiety was introduced by a characteristic amidation reaction with a primary alkylamine. As a result, the tertiary amine moiety markedly promoted protonation and attomole levels of mevalonate were detected. In the present study, tertiary amine moieties were successfully introduced to MA by esterification with 3-hydroxy-1-methylpiperidine. Thus, the reaction for the synthesis of carboxylic esters by Shiina et al. (10) appears to be useful not only for the derivatization of alcohols (13) but also for that of carboxylic acids. This derivative, DMP-MA, exhibited  $[M+H]^+$  as the base peak by P-ESI-MS and the detection limit by SRM was more than 100 times lower than that of underivatized MA by SRM with N-ESI mode.

The derivatization and purification steps in this method are very simple but it should be mentioned that there are two pitfalls to obtaining reliable and reproducible results. First, use of the anion exchange column cartridge gave unexpectedly high values of MA concentrations. Serum MA was extracted by this cartridge and interfering peaks on SRM chromatograms were markedly reduced by the addition of this purification step. However, the recoveries of known amounts of MA from this cartridge were always more than 100%, and additional experiments suggested that a significant amount of MA was produced from unknown substance(s) in organic solvents by this anion exchange column (data not shown). Plasma methylmalonic acid (MMA) and its isomer succinic acid (SA) are also known to be extracted by this column (14). We have derivatized MMA and SA into DMP-MMA and DMP-SA, respectively, and analyzed them by the same HPLC condition as that for DMP-MA. The SRM was conducted using  $m/z$  313  $\rightarrow$   $m/z$  98 for both DMP-MMA and DMP-SA. The results showed that DMP-MMA and DMP-SA were much more hydrophobic than DMP-MA and both compounds were eluted during washout phase with 0.2% formic acid in acetonitrile (after 6 min).

Second, pH of the final sample solution should not be more than 7 because an alkaline condition easily hydrolyzes DMP-MA. After the derivatization step, most of the excess reagents and hydrophilic impurities were

TABLE 2. Reproducibility in the quantification of MA in human serum: ANOVA

Source	S	f	V	$F_0$	Relative SD
					%
Sample preparation	1.293	3	0.431	1.89	4.4
Error (SRM)	1.820	8	0.228		3.2
Total	3.113	11			
$F(3,8,0.05)=4.07$					

S, residual sum of squares; f, number of degrees of freedom;  $f_1$ ,  $f_{\text{sample preparation}}$ ;  $f_2$ ,  $f_{\text{error}}$ ; V, unbiased variance;  $F_0$ , observed value following F distribution variance ratio ( $V_{\text{sample preparation}}/V_{\text{error}}$ );  $F(f_1, f_2, \alpha)$ , density function of F distribution with  $f_1$  and  $f_2$  degrees of freedom.

TABLE 3. Recovery of MA from human serum

Sample ( $X_0 + na$ ) ( $n = 0, 1, 2, 3$ )	Amount Added	Amount Found			Recovery <sup>b</sup>	Estimated Amount $\pm$ 95% Confidence Limit <sup>c</sup>
		<i>pmol</i>			%	<i>pmol</i>
$X_0$	0	$\bar{X}_0 \pm SD = 15.0 \pm 0.6^a$				$14.1 \pm 1.1$
$X_0 + a$	9.6	23.6	25.2	25.0		
$X_0 + a$	9.6	24.2	23.6	23.3	$94.5 \pm 8.2$	
$X_0 + 2a$	19.2	33.2	32.1	32.0		
$X_0 + 2a$	19.2	33.9	34.1	34.0	$94.6 \pm 5.1$	
$X_0 + 3a$	28.8	43.8	43.9	43.0		
$X_0 + 3a$	28.8	44.1	43.6	43.2	$99.0 \pm 1.5$	

Known amounts of MA were spiked into 50  $\mu$ l of normal human serum before sample preparation.

<sup>a</sup>The value was obtained from Table 1.

<sup>b</sup>Recovery (%) = (amount found -  $\bar{X}_0$ ) / amount added  $\times$  100.

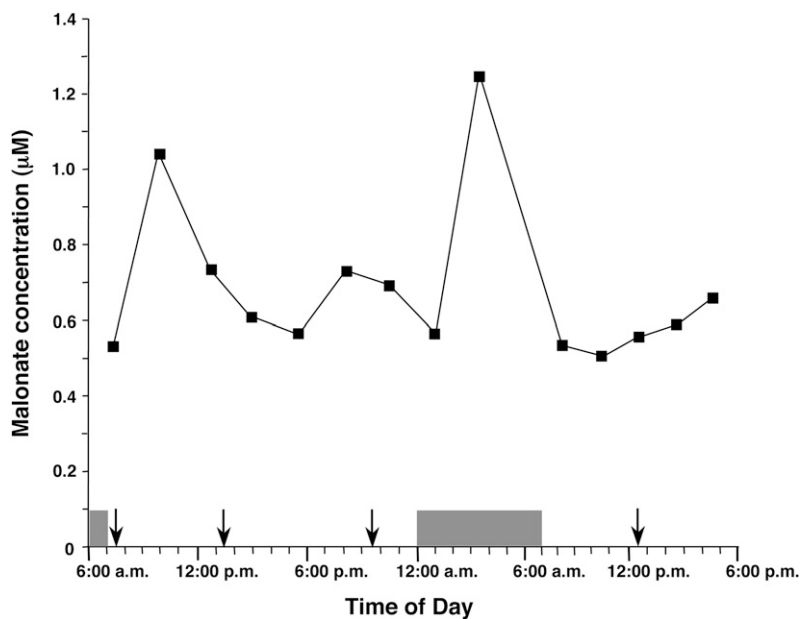
<sup>c</sup>The estimated amount was calculated by orthogonal regression.

precipitated by the addition of n-hexane but significant amounts of 3-hydroxy-1-methylpiperidine and 4-dimethylaminopyridine were recovered with DMP-MA in the final residue of the extract. Therefore, it was necessary to dissolve the final residue in 1% formic acid in water to keep the pH of the solution less than 7. The mobile phase of the HPLC (0.2% formic acid in water) was not sufficient to neutralize the final extract.

The highly sensitive quantification of serum MA can be useful for monitoring of de novo FAS, also called de novo lipogenesis, in normal humans. The diurnal variation of serum MA levels in a healthy human (Fig. 5) was similar to the variation of de novo FAS determined in humans by continuous intravenous infusion of sodium [ $1-^{13}$ C]acetate and mass isotopomer distribution analysis (15, 16). According to Timlin et al. (16), de novo FAS peaked 4.2 h after ingestion of a meal whereas lipoprotein-triacylglycerol concentrations peaked at 2.0 h postmeal. Another study, by Hudgins et al. (15), showed that the maximum values of de novo FAS occurred in the evening, 3.0–9.0 h after the last meal, although the peak after every meal was


not detected because a limited number of postprandial data points were obtained. In our data, postprandial increases of MA concentrations peaked between 2.5 h and 6.5 h after meals and the maximum value was observed in the night 6.0 h after dinner. In addition, the increase of MA concentration disappeared after skipping the meal. Thus, serum MA concentrations are regulated by food intake and appear to be a good marker that reflects de novo FAS in normal humans.

Because serum MA concentrations correlate well with de novo FAS, the most important enzyme that determines serum MA concentration is thought to be ACC, the rate-limiting enzyme in the fatty acid biosynthesis. In mammals, two ACC isoforms exist. Cytosolic ACC1 synthesizes malonyl-CoA, which participates in both de novo FAS and negative regulation of  $\beta$ -oxidation. In contrast, malonyl-CoA synthesized by mitochondrial ACC2 acts mainly as an inhibitor of  $\beta$ -oxidation (17). We cannot clarify at present which ACC contributes to serum MA concentration but both ACCs regulate de novo lipogenesis in a coordinated and complementary manner (18).



**Fig. 5.** The circadian rhythm of the serum levels of MA in a healthy volunteer. Blood samples were taken every 2–3 h. On the first day the volunteer consumed a normal hospital diet at 7:30 AM, 1:30 PM, and 9:30 PM (indicated by the arrows), and slept from 12:00 AM to 7:00 AM (indicated by the shaded box). On the second day the volunteer did not eat breakfast but consumed a normal hospital diet at 12:30 PM.

Under special conditions, however, other enzymes, MCD, and FAS can also be determinants of tissue malonyl-CoA levels and serum MA concentrations. For example, when MCD activity is reduced, such as with MCD deficiency, serum MA concentrations are elevated. Alternatively, when FAS is blocked by any drugs, such as C75 and cerulenin (2), MA concentrations increase in spite of reduced de novo FAS. Therefore, it is important to rule out the presence of such special conditions when we use serum MA as a biomarker for de novo FAS.

In summary, we developed a new method for the quantification of MA in human serum, which can be a good marker for de novo FAS. Derivatization of MA into DMP-MA allowed it to be quantified by LC-P-ESI-MS/MS with excellent sensitivity. Recovery and reproducibility experiments verified that this method provided highly reliable and reproducible analytical results. 

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