

# Virtual Coupling of Pyran Protons in the $^1\text{H}$ NMR Spectra of C- and N-Glucuronides: Dependence on Substitution and Solvent

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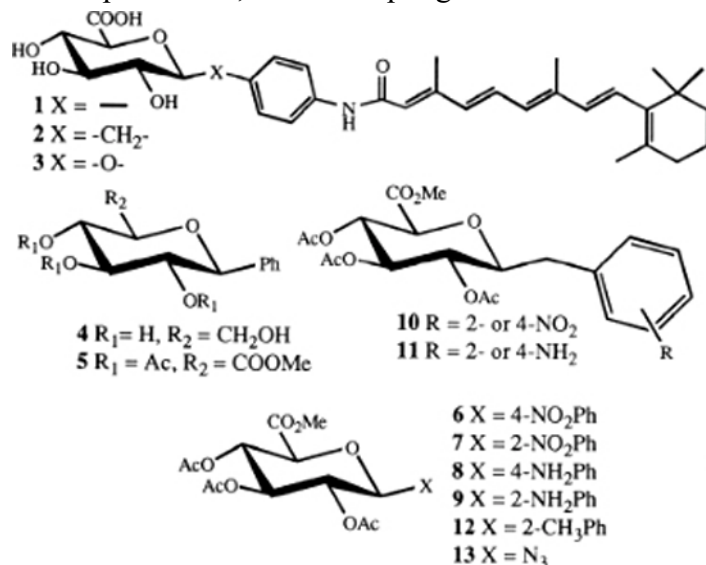
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**ABSTRACT** We have observed that certain C- and N-glucuronides prepared as intermediates for breast cancer preventives demonstrate non-first order  $^1\text{H}$  NMR spectra that are not the result of impurities or degradation but are instead due to virtual coupling in the pyran proton network. This virtual coupling shows the expected dependence on solvent and field strength and, more importantly, on the nature of the C-1 substitution. Although the hybridization of the atom bonded to C-1 may play a role, it appears that steric and/or electronic factors, which have the effect of increasing  $\Delta\nu/J$  for H-3 and H-4, are critical for eliminating the spectral complexity. These observations, which appear to be fairly general, suggest that this phenomenon should be considered when addressing the purity of pharmaceutical agents containing these types of structural units.

**Key Words:**  $^1\text{H}$  NMR, glucuronides, breast cancer, chemoprevention, virtual coupling



**Figure 1.** C- and N-linked glucuronides investigated

## INTRODUCTION

The O-glucuronide metabolites of retinoic acid and certain of its natural and synthetic analogues have been suggested to be biologically active forms of the parent molecule (1). As a class, these retinoids regulate epithelial tissue differentiation and show utility in treating dermatological diseases as well as promise for the treatment and prevention of cancer (2). Because of the relative chemical and metabolic instability of these glucuronides, we have been synthesizing C- and N-glucuronosyl analogues of some of these metabolites in an effort to improve the activity of these compounds and/or to determine whether these metabolites are active themselves or are hydrolyzed to the active parent retinoid (3). Thus, we have prepared C-glucuronosyl analogues 1 and 2 (Figure 1) of the O-glucuronide 3 of the semisynthetic retinoid N-(4-hydroxyphenyl) retinamide. Our results suggest these compounds show promise as mammary tumor chemopreventive agents (4,5).

In the course of synthesizing 1, selective PtO<sub>2</sub>-mediated oxidation (6, 7) of the 6-hydroxymethyl group of glucosylbenzene (4) followed by esterification and acetylation produced a product 5 that showed unusual complexity in the  $^1\text{H}$  NMR spectrum in the region of the pyran ring protons. This was true for all resonances except that assigned for the H-1 proton. Since the Adams' catalyst that promoted oxidation had not to our knowledge been previously employed for the oxidation of C-glycosyl compounds into their glucuronide analogues, and given that this aryl-C-glycoside contains a tertiary carbon and benzylic ether unit (carbohydrate position 1), both of which may be prone to oxidation, we were concerned that other products might have been produced during the reaction that would compromise the purity of the materials and hence the validity of bioactivity assays performed with them.

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After careful chromatographic purification and recrystallization of 5 to apparent homogeneity, while its  $^1\text{H}$  NMR spectrum remained unchanged, other available spectroscopic evidence ( $^{13}\text{C}$  NMR, IR, and MS) was consistent with a single compound assigned the structure 5. The possibility that the complexity of the  $^1\text{H}$  NMR spectrum resulted from long-range virtual  $^1\text{H}$ - $^1\text{H}$  coupling was thus considered (8). Spin simulation of the spectrum using PANIC (Parameter Adjustment in NMR by Iterative Calculation) appeared to confirm this explanation.

Prompted by the report of Saito et al (9) on their observation of virtual  $^1\text{H}$ - $^1\text{H}$  coupling in glucuronosyl moieties within *O*-disaccharides and their conjugates, we wish to report our interesting observations of similar phenomena in *C*- and *N*-glucuronosyl compounds, which appears to depend on the structure of the pyran C-1 substituent and the solvent employed in NMR measurements. This observation of deceptively complex spectra appears to be surprisingly general and should be considered when evaluating the purity, including the stereochemical purity, of potential pharmaceutical agents containing these structural units.

## MATERIALS AND METHODS

Fourier-transformed  $^1\text{H}$  NMR spectra were obtained on sample solutions in glass 175 x 5 mm sample tubes (Wilmad; Buena, NJ). Spectra were collected for 20 mg/mL solutions at 250, 400, 600, and 800 MHz on AC250 or DPX250, DRX400, DMX600, and DMX800 instruments, respectively (Bruker Instruments; Billerica, MA). Samples were dissolved in  $\text{CDCl}_3$ ,  $\text{CD}_2\text{Cl}_2$ , acetone- $d_6$ , benzene- $d_6$ ,  $\text{DMSO-}d_6$ ,  $\text{CD}_3\text{OD}$ , pyridine- $d_5$ , and tetrahydrofuran- $d_8$  as appropriate (Cambridge Isotope Laboratories; Andover, MA) and spectra referenced to the residual protio solvent (relative to TMS) in the deuterated solvents. Spectra were collected at ambient temperature using  $90^\circ$  pulse widths and transformed after exponential multiplication (LB = 0.2 Hz). Spectral simulation (see Table 1) was performed using PANIC version 840419 implemented on an ASPECT 3000 computer (Bruker Instruments).

The compounds studied were prepared as previously published (3, 10, 11). Entries 2 and 9 (Table 2) were prepared by methods identical to those used for entries 1 and 10 using the appropriate Grignard reagents, while entries 17 and 18 were prepared by

methods identical to those used in entry 19 using acetyl and benzoyl chloride respectively.

**Table 1. Chemical Shifts and Coupling Constants Simulated for H-1 to H-5 of 5**

Proton	Chemical Shift (ppm)	Coupling Constant (Hz)
H-1	4.40	$J_{1,2} = 9.895$
		$J_{1,3} = -0.172$
		$J_{1,4} = 0.0$
		$J_{1,5} = 0.0$
		$J_{2,3} = 9.294$
H-2	5.15	$J_{2,4} = 0.0$
		$J_{2,5} = 0.0$
		$J_{3,4} = 9.800$
		$J_{3,5} = 0.012$
H-3	5.34	H-4 5.37 $J_{4,5} = 9.485$
		H-5 4.16

**Table 2. Virtual Coupling Dependence on C-1 Substituent**

Entry No.	C-1 B Substituent	Virtual Coupling <sup>a</sup>
1	Ph-(5)	YES <sup>b</sup>
2	1-Naphthyl-	NO
3	4-NO <sub>2</sub> Ph-(6)	YES
4	2-NO <sub>2</sub> Ph-(7)	NO
5	4-NH <sub>2</sub> Ph-(8)	YES
6	2-NH <sub>2</sub> Ph-(9)	NO
7	4-CH <sub>3</sub> Ph-	YES
8	2-CH <sub>3</sub> Ph-(12)	YES <sup>c</sup>
9	CH <sub>3</sub> -	YES
10	PhCH <sub>2</sub> -	NO
11	4-NO <sub>2</sub> PhCH <sub>2</sub> -(10a)	NO
12	2-NO <sub>2</sub> PhCH <sub>2</sub> -(10b)	NO
13	4-NH <sub>2</sub> PhCH <sub>2</sub> -(11a)	NO
14	2-NH <sub>2</sub> PhCH <sub>2</sub> -(11b)	NO
15	N <sub>3</sub> -(13)	YES <sup>d</sup>
16	H <sub>2</sub> N-	NO
17	CH <sub>3</sub> CONH-	NO
18	PhCONH-	NO
19	Retinoyl NH-	NO
20	CH <sub>3</sub> COO-	NO

<sup>a</sup>In  $\text{CDCl}_3$  at 250 MHz; <sup>b</sup>Eliminated in acetone- $d_6$  and at 800 MHz (see Appendix); <sup>c</sup>Weakly present; <sup>d</sup>Eliminated in acetone- $d_6$ , benzene- $d_6$ , pyridine- $d_5$ , tetrahydrofuran- $d_8$ ,  $\text{CD}_2\text{Cl}_2$ ,  $\text{CD}_3\text{OD}$ , and  $\text{DMSO-}d_6$  and at 400 MHz (see Appendix).

## RESULTS AND DISCUSSION

The **5** used in this study was prepared as previously described (3). The 250 MHz  $^1\text{H}$  NMR spectrum of this compound in  $\text{CDCl}_3$ , in the region of the pyran protons, is shown in Figure 2. The surprising complexity of this spectrum, which is still present at 400 MHz (but is reduced at 600 MHz and eliminated at 800 MHz), and the possibility that it arose from virtual coupling between H-2 and H-5, led us to simulate the spectrum using PANIC, as is also shown in Figure 2. The chemical shifts and calculated coupling constants derived from simulating the spectrum of **5** are shown in Table 1. For this simulation, the apparent coupling constants  $J_{1,4}$ ,  $J_{1,5}$ ,  $J_{2,4}$ , and  $J_{2,5}$  are sufficiently small that they can be set to zero and a satisfactory simulation can be obtained. Nonetheless, the H-2 and H-5 nuclei appear to show the observed complexity by virtue of being coupled as X parts of ABX spectra to H-3 and H-4, which themselves form a strongly coupled AB system with  $\Delta\nu/J = 0.82$  at 250 MHz. As might be expected, this phenomenon can be eliminated by recording the  $^1\text{H}$  NMR spectrum of **5** in different solvents. As also shown in Figure 2, the spectrum of **5** in acetone- $d_6$  can be analyzed as first order, with  $\Delta\nu/J$  for H-3 and H-4 now being 2.96.

Interestingly, our chemistry to further elaborate **5** to **1** produced intermediates that show virtual coupling that depends on both the nature and site of aromatic ring substitution. Nitration of **5** produced a 3:2 mixture of isomers **6** and **7**, which were difficult to separate (3). In one instance, small quantities of pure **6** and **7** were obtained by preparative TLC. Their 250 MHz  $^1\text{H}$  NMR spectrum in  $\text{CDCl}_3$  showed virtual coupling comparable to that of **5** for **6** but not to that of **5** for **7** (Data not shown). Reduction of the nitroaromatic isomer mixture produced the readily separable *O*- and *p*-anilines **8** and **9** (3). In this instance, the *para* substituted aniline **8** also shows strong virtual coupling that was not simulated but appears likely to result from the even smaller  $\Delta\nu/J_{3,4}$  ratio (Figure 3). For the *ortho* regioisomer **9**, this virtual coupling observed for **5** and **8** is also absent. Homonuclear decoupling and NOE difference spectra established that H-2 in **9** has moved substantially downfield to 5.61 ppm. More importantly, the chemical shift of H-3 and H-4 has reversed relative to **5** (5.37 and 5.29 ppm respectively) and  $\Delta\nu/J_{3,4}$  has increased to 1.91, which

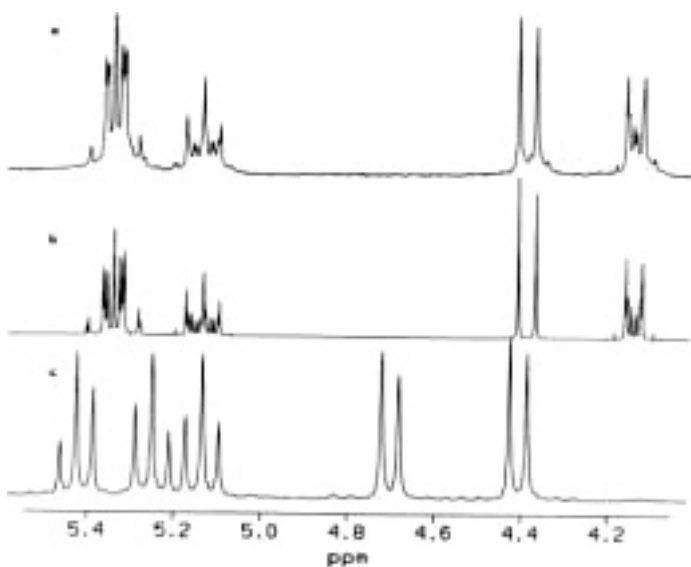


Figure 2. Partial 250 MHz  $^1\text{H}$  NMR of **5** in a)  $\text{CDCl}_3$ , b) simulated, and c)  $\text{CD}_3\text{CO}$

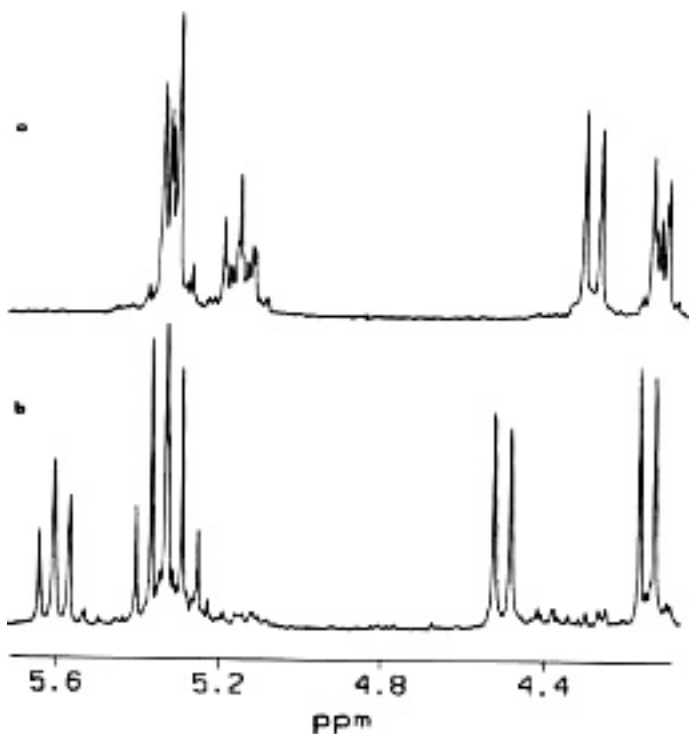


Figure 3. Partial 250 MHz  $^1\text{H}$  NMR  $\text{CDCl}_3$  spectrum of a) **8** and b) **9**

appears to be sufficient to eliminate this coupling phenomenon.

Because both the *O*-nitrophenyl and *O*-aminophenyl isomers 7 and 9 fail to show the virtual coupling present in 5, 6, and 8, which bear a C<sub>2</sub>-symmetric substituent at C-1, it seems plausible that this lack of virtual coupling results from steric interactions of the *O*-substituent with the axial H-1 or H-2 protons. This results in a different favored rotamer about the C-1-Ar bond and/or causes subtle changes in the conformation of the pyran ring, changes that have the effect of increasing  $\Delta\nu/J_{3,4}$ . In support of this concept, none of the *ortho* nitro or amino *C*-benzyl analogues 10 or 11 (3) (which we required for the preparation of 2) that have an interposed methylene unit show evidence of virtual coupling in the 250 MHz <sup>1</sup>H NMR spectra in CDCl<sub>3</sub> (see Table 2 for a summary of the compounds we investigated to determine whether the phenomenon is observed). That other more subtle influences such as electronics may also play a role is suggested by inspection of the spectrum of the *O*-tolyl analog 12, which we prepared serendipitously during efforts to synthesize 2 (10). In the CDCl<sub>3</sub> <sup>1</sup>H NMR spectrum of 12, the H-2, H-3, and H-4 resonances overlap extensively, unlike any of the other compounds reported here. However, the H-5 resonance at 4.16 ppm shows some evidence of much less extensive virtual coupling than for 5, implying that the impact of the *O*-methyl substituent is insufficient to change  $\Delta\nu/J_{3,4}$  enough to eliminate virtual coupling under these spectroscopic conditions. Furthermore, we observed that the 1- $\beta$ -azido glucuronide 13 we previously prepared (11) demonstrated virtual coupling in the <sup>1</sup>H NMR spectrum in CDCl<sub>3</sub>, which is nearly identical to that of 5. This coupling is absent at 400 MHz and in the 250 MHz acetone-d<sub>6</sub>, benzene-d<sub>6</sub>, CD<sub>2</sub>Cl<sub>2</sub>, CD<sub>3</sub>OD, pyridine-d<sub>5</sub>, and tetrahydrofuran-d<sub>8</sub> DMSO-d<sub>6</sub> spectra of 13 and also in the CDCl<sub>3</sub> spectrum of the amine prepared by reduction of 13 as well as its acylated derivatives (11). Once again, linear, symmetrical azide substitution results in virtual coupling while reduction products do not show this property, suggesting, perhaps, that the hybridization of the C-1 attached atom may play a role in causing this phenomenon. However, as shown in entry 9 of Table 2, the spherically symmetrical, sterically undemanding methyl substituted compound also demonstrates this virtual coupling.

Thus, with the limited set of examples explored here, while those with atoms with sp<sup>2</sup>-like character bonded to C-1 demonstrate this coupling, steric and electronic effects from the C-1 substituent are likely to be more important contributors to the complexity of the observed spectra than is hybridization.

It might be expected that homonuclear decoupling experiments would allow elimination of this observed virtual coupling in many instances. In the present case, this is only a partially successful strategy because the phenomenon is driven by the small value of  $\Delta\nu/J_{3,4}$  and thus selective irradiation of H-3 or H-4 is not possible. As shown for compound 13 in the Appendix, irradiation of H-5 and H-2 (4.1 and 4.95 ppm respectively) still leaves some significant evidence of a noN-first order spectrum. More successful in this case is the impact of raising the temperature on spectral appearance (also see Appendix). Interestingly, we have observed this virtual coupling for *C*- and *N*-glucuronides only when samples are dissolved in CDCl<sub>3</sub>. Thus, it appears that in this solvent a unique pyran ring conformation and fortuitous <sup>1</sup>H chemical shifts create the observed phenomenon. Given the high volatility of CDCl<sub>3</sub>, limits are placed on routine use of elevated temperature experiments. Nonetheless, raising the temperature for 13 in CDCl<sub>3</sub> by 20°C above ambient clearly alters spectral appearance in a manner consistent with movement toward a first order spectrum.

## CONCLUSIONS

Thus, as in some  $\beta$ -D-glucopyranosuronate systems (9), certain *C*- and *N*-glucuronides can show surprisingly complex <sup>1</sup>H NMR spectra. These appear to be the result of long-range virtual coupling and are not caused by the presence of isomer mixtures at C-1 or in substitution of the aromatic ring in *C*-aryl glucuronides. The phenomenon shows sensitivity to substituents at the *O*-position of *C*-aryl glucuronides, but this is observed strongly only when the *O*-positions are unsubstituted. Both solvent and field strength dependences are observed. Changing the solvent from CDCl<sub>3</sub> to other solvents causes a greater chemical shift dispersion, thereby removing virtual coupling effects in these <sup>1</sup>H NMR spectra. By increasing the spectrometer magnetic field, the value of  $\Delta\nu/J$  becomes sufficiently large to no longer exhibit virtual coupling effects. The relatively high

frequency with which this spectral phenomenon is observed in these types of structural units suggest it should be considered when the purity of potential pharmaceutical agents containing these structural units is in doubt based on  $^1\text{H}$  NMR analysis.

## ACKNOWLEDGEMENTS

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## Appendix

*Additional Spectra for Table 2, Entry 1 (5) and Entry 15 (13)*

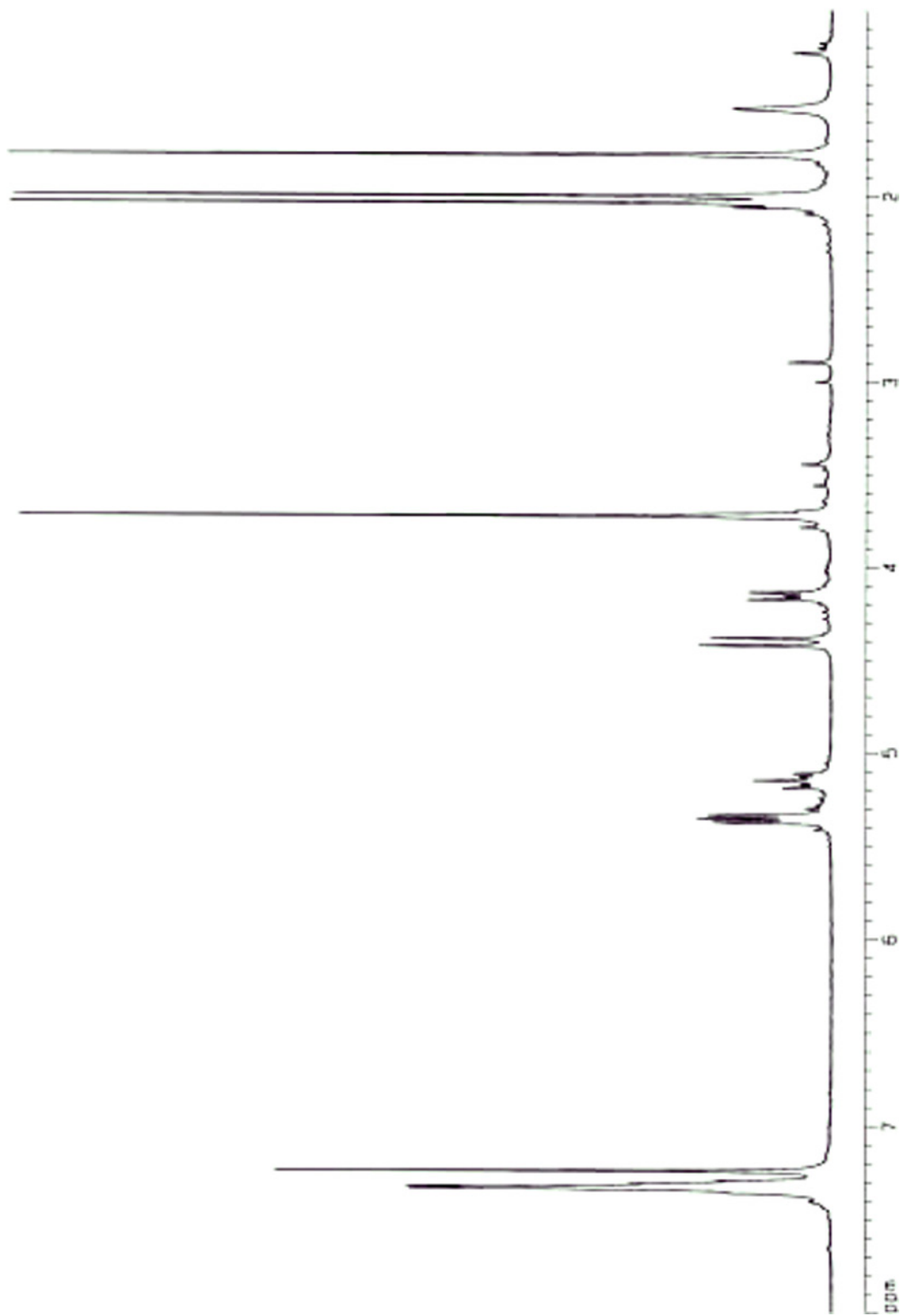
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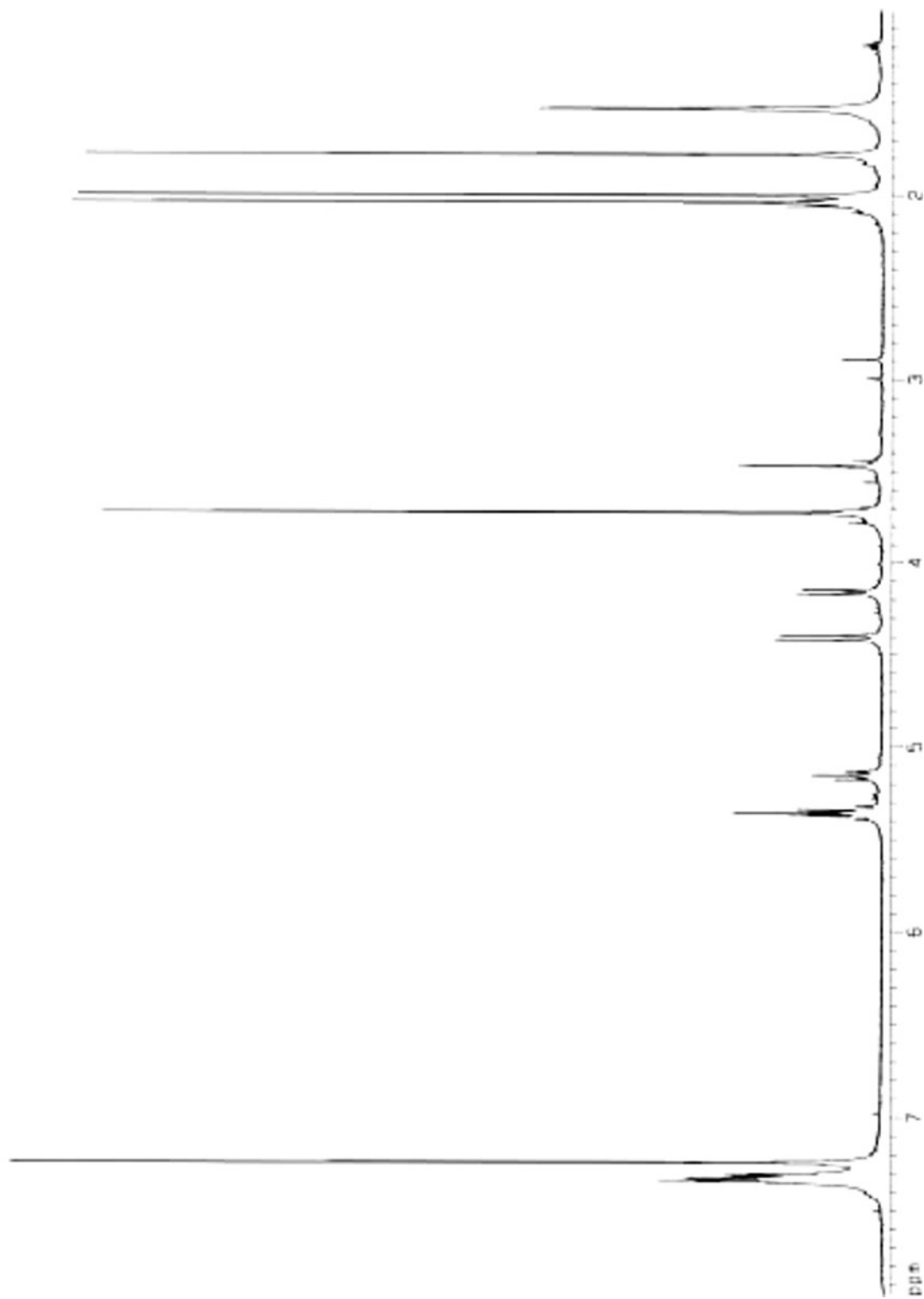
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1H phenylglucuronide in CDCl3



<sup>1</sup>H phenylglucuronide in CDCl<sub>3</sub> @ 250MHz

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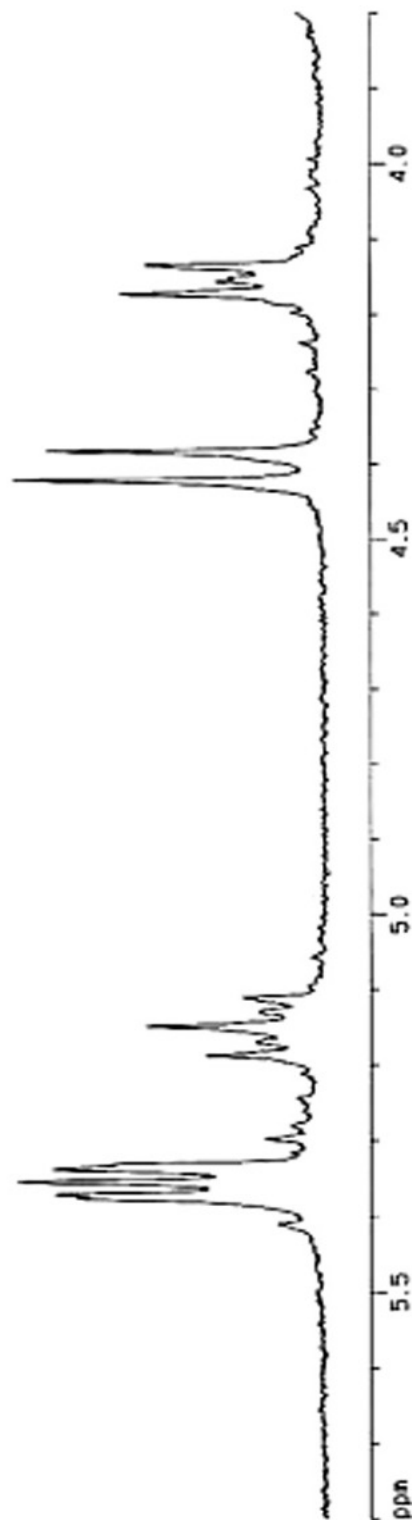
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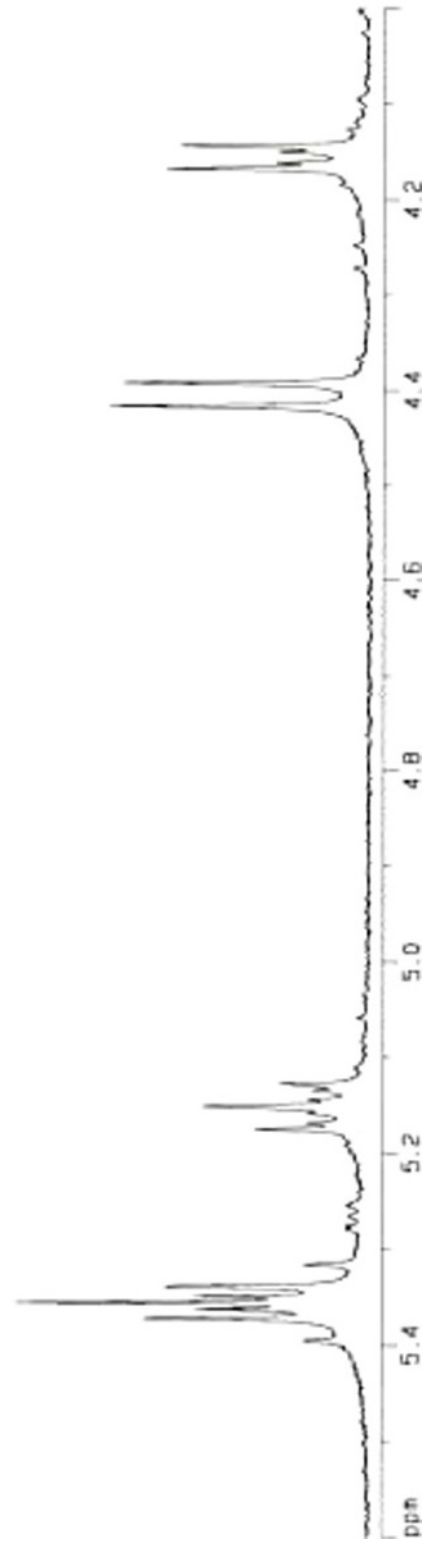
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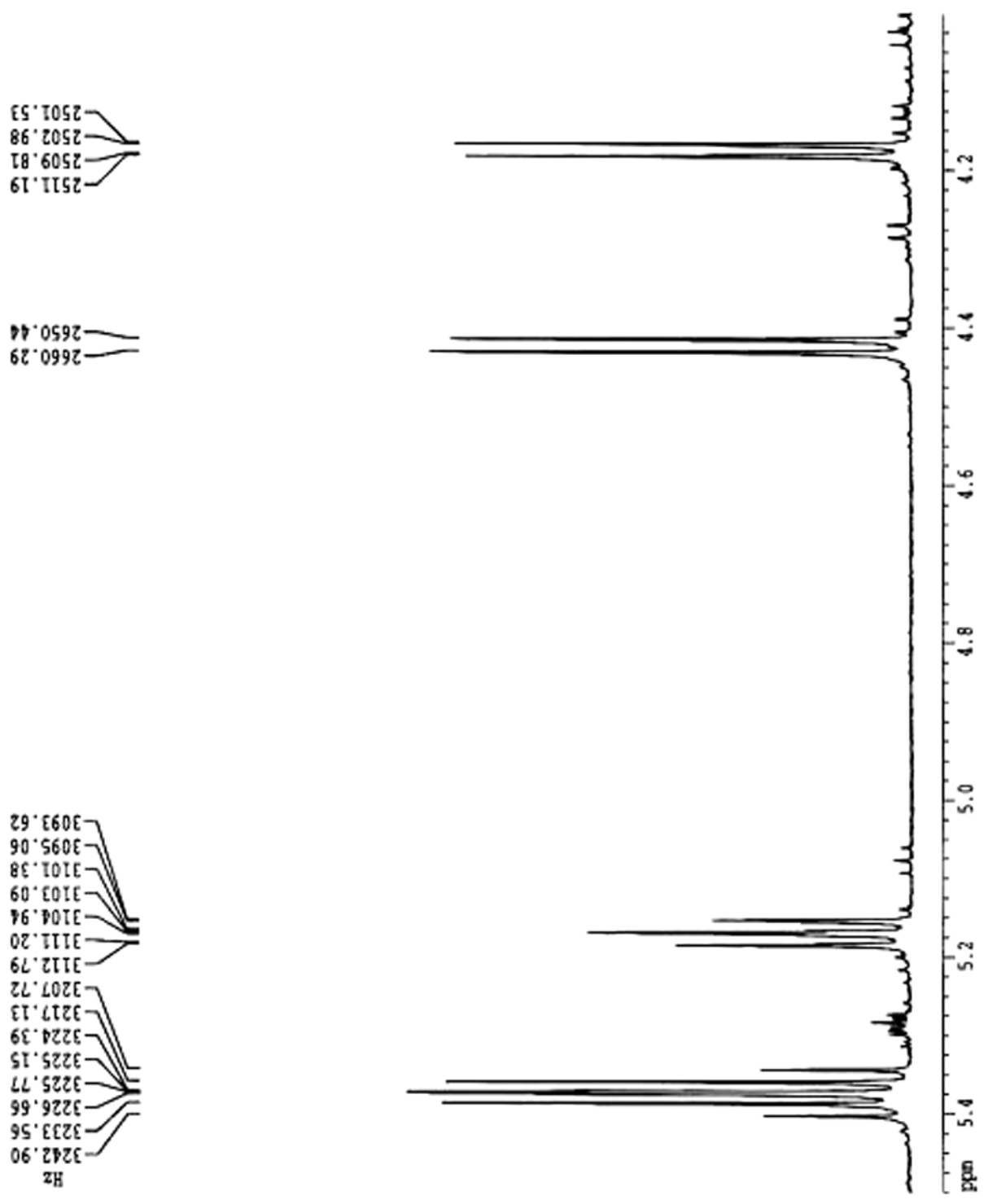
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PHGLUC at 600 MHz 4/27/99

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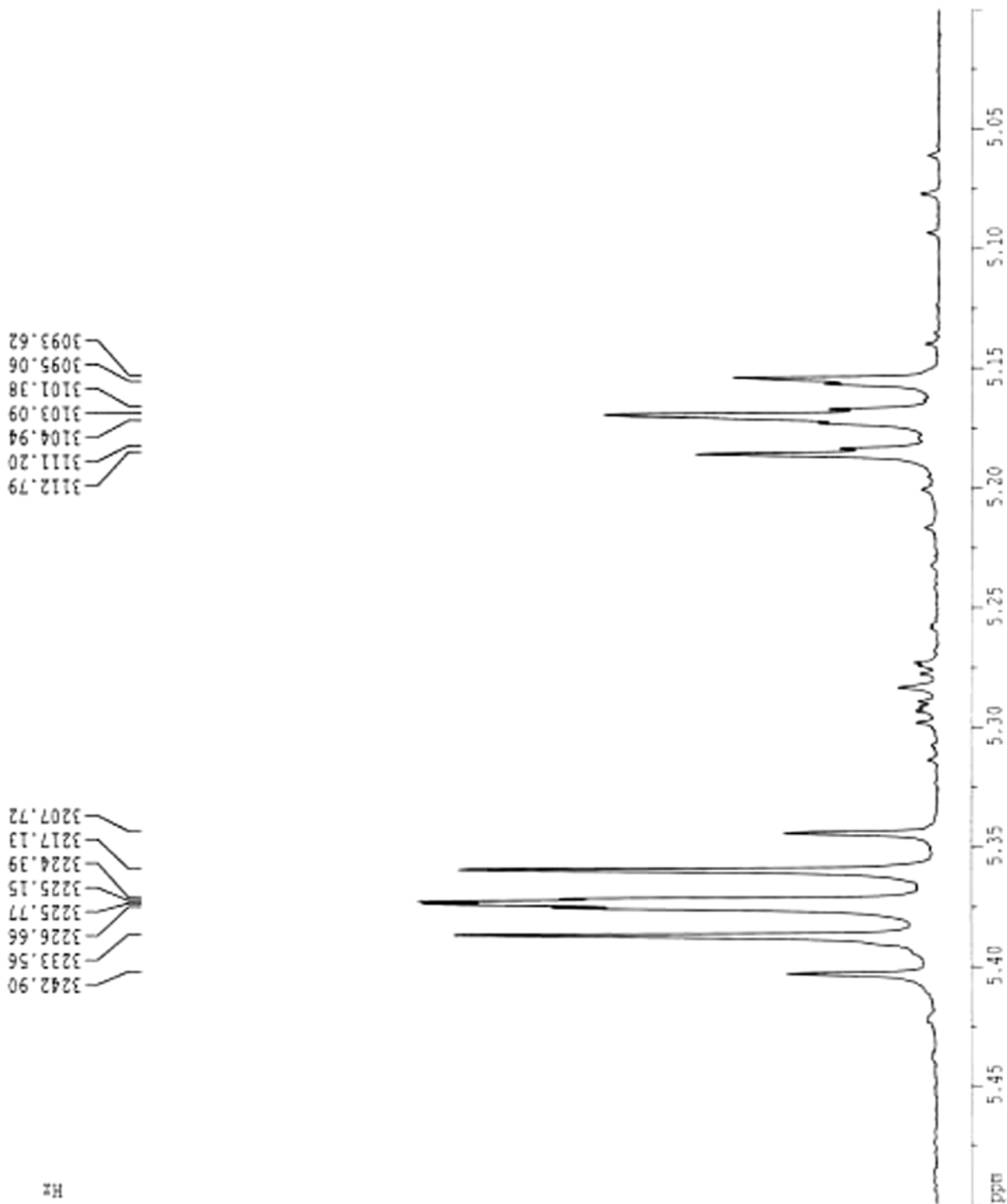
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F2 - Processing parameters
SI        32768
SF        600.1700270 MHz
WDW       EM
SSB       0
LB        0.10 Hz
GB        0
PC        1.00

1D NMR plot parameters
CX        20.00 cm
FIP       5.500 ppm
F1        3300.94 Hz
F2P       5.500 ppm
F2        3000.85 Hz
PFMCHX    0.02500 ppm/cm
HDCM      15.00425 Hz/cm
    
```



PHGLJC 800 MHz

Current Data Parameters  
 NAME phgluc800  
 EXPNO 4  
 PROCNO 1

F2 - Acquisition Parameters

Date\_ 390503  
 Time 15.05  
 INSTRUM spect  
 PROBO 5 mm QNP 1H1  
 PULPROG zg  
 TD 65536  
 SOLVENT CDCl3  
 NS 16  
 DS 4  
 SWH 10416.667 Hz  
 FIDRES 0.158846 Hz  
 AQ 3.1457779 sec  
 RG 256  
 CW 48.000 usec  
 CB 6.00 usec  
 TE 323.0 K  
 D1 1.2000005 sec

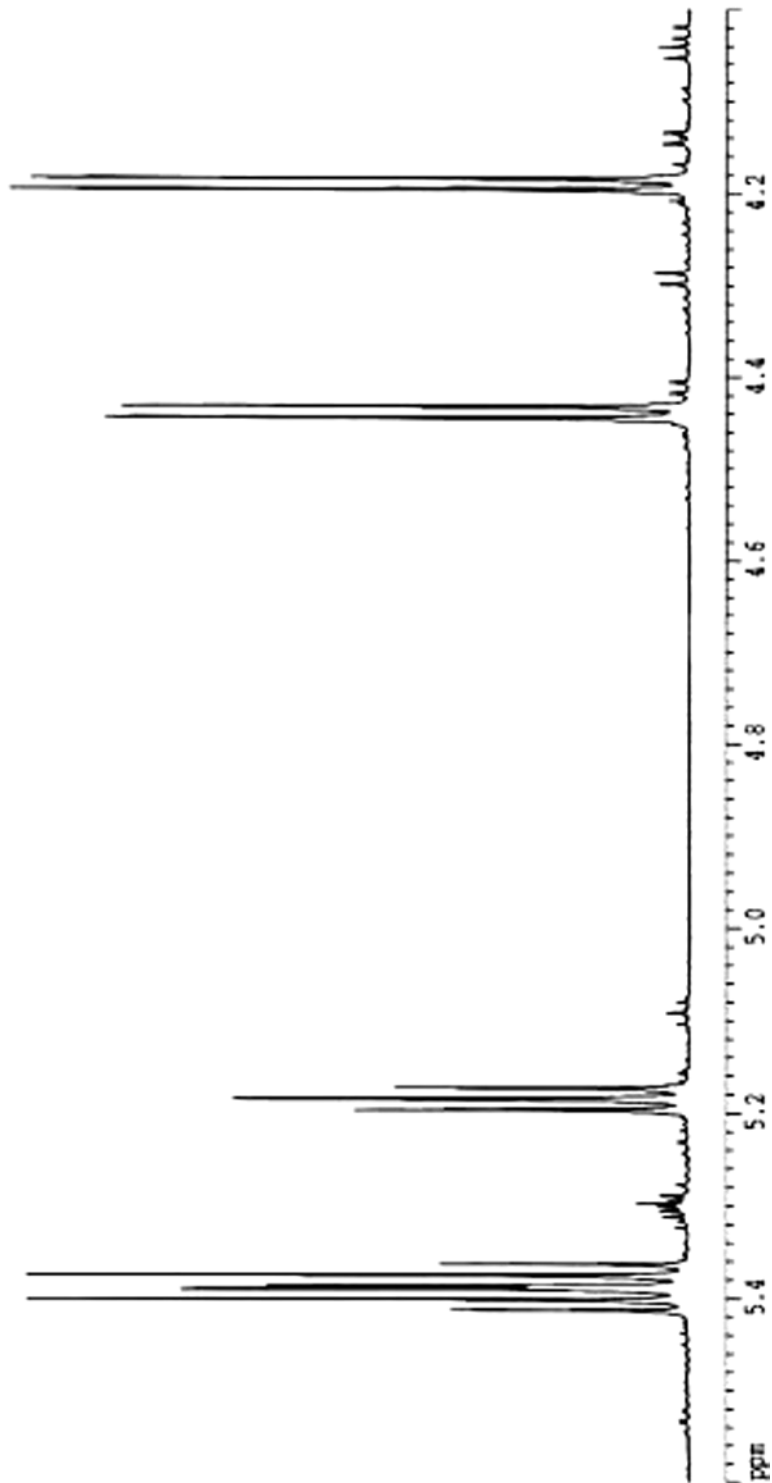
===== CHANNEL f1 =====  
 NUC1 1H  
 P1 4.00 usec  
 PL1 -3.00 dB  
 SFO1 800.1337464 MHz

F1 - Processing parameters

SI 65516  
 SF 800.1330000 MHz  
 ICW EM  
 SSB 0  
 LB 0.10 Hz  
 GB 0  
 PC 1.00

1D NMR plot parameters

CK 20.00 cm  
 F1P 5.600 ppm  
 F1 4480.73 Hz  
 F2P 4.000 ppm  
 F2 3200.52 Hz  
 FREQM 0.08000 ppm/cm  
 HSCM 64.01040 Hz/cm



PHGLDC 800 MHz

Current Data Parameters  
 NAME pbj:luc800  
 EXPNO 4  
 PROCNO 1

F1 - Acquisition Parameters

Date\_ 990503  
 Time 15.05  
 INSTRUM spect  
 PROBED 5 mm QXI no1  
 PULPROG zg  
 TD 65536  
 SOLVENT CDCl3  
 NS 16  
 DS 4  
 SMH 10416.667 Hz  
 FIDRES 0.158945 Hz  
 AQ 3.1457779 sec  
 RG 256  
 DW 48.000 usec  
 DE 6.00 usec  
 TE 323.0 K  
 D1 1.20000005 sec

===== CHANNEL f1 =====

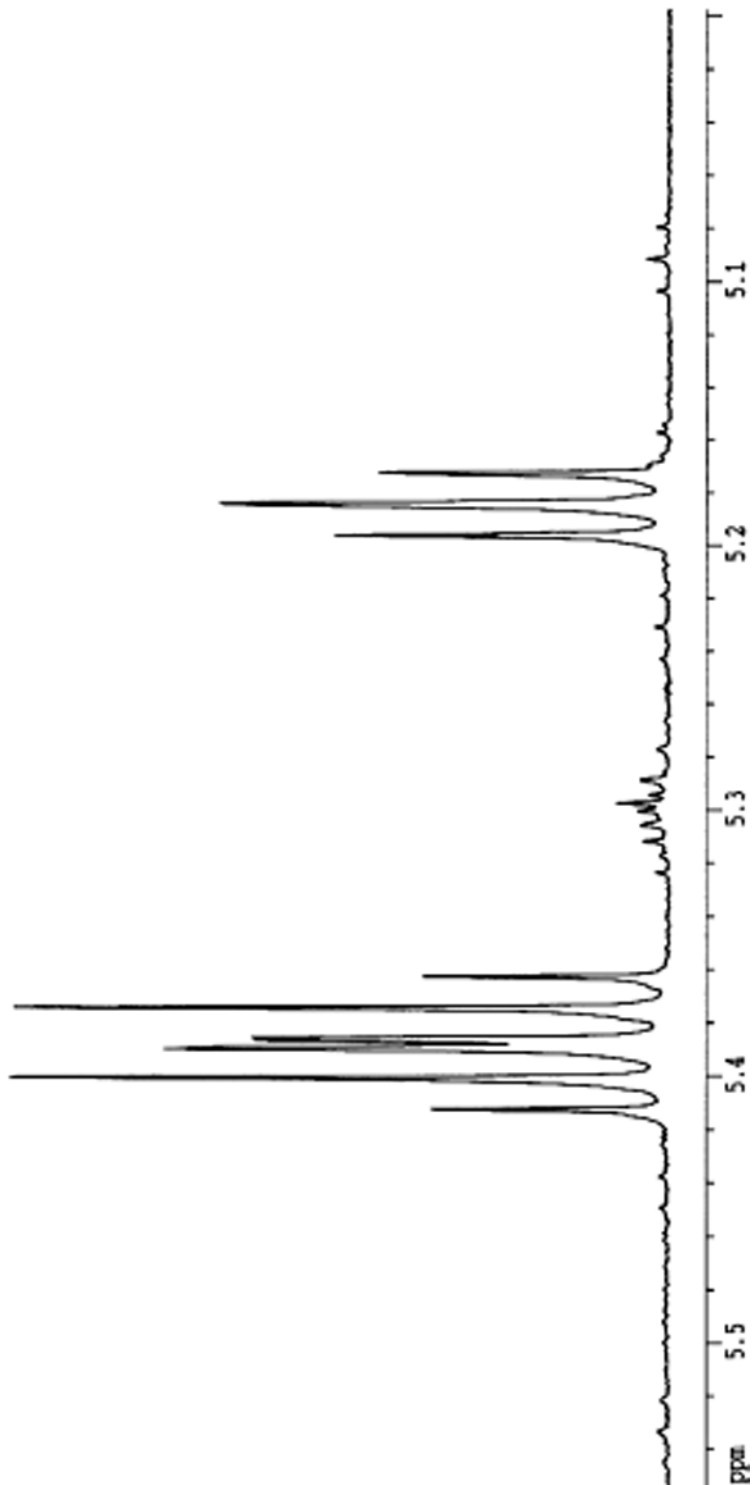
NUC1 1H  
 P1 4.00 usec  
 PL1 -3.00 dB  
 SFO1 800.137464 MHz

F2 - Processing parameters

SI 65536  
 SF 800.1300000 MHz  
 NDM EM  
 SSB 0  
 LB 0.10 Hz  
 GB 0  
 PC 1.00

ID NMR plot parameters

CK 20.00 cm  
 F1P 5.554 ppm  
 F1 4443.97 Hz  
 F2P 4.397 ppm  
 F2 3998.42 Hz  
 PPMCN 0.02784 ppm/cm  
 HzCN 22.27728 Hz/cm



<sup>1</sup>H azidoogluc in cdcl3 @ 250MHz

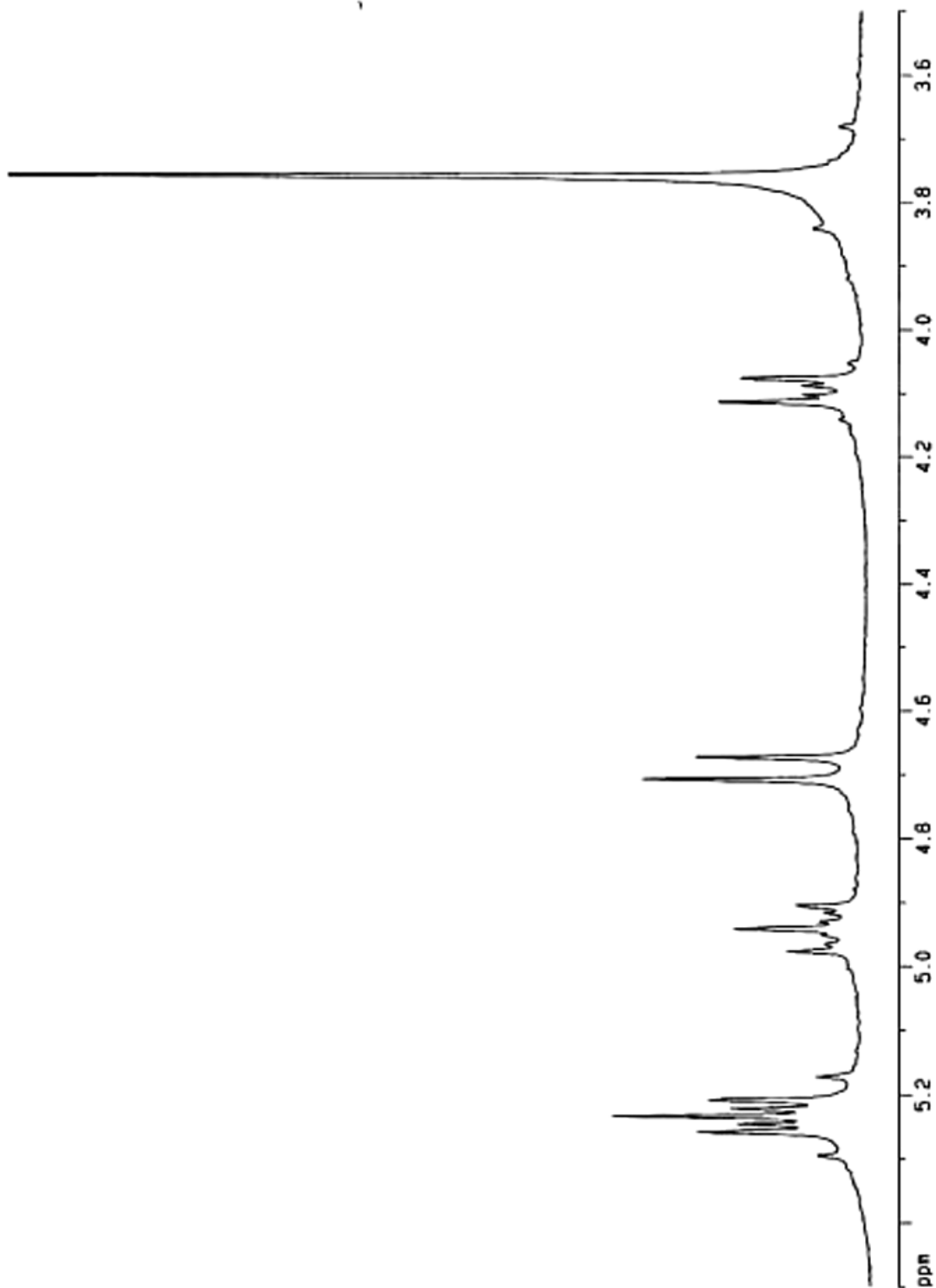
```

Current Data Parameters
NAME      azidoogluccl3
EXPNO     1
PROCNO    1

F2 - Acquisition Parameters
Date_     990203
Time      9.03
INSTRUM   spect
PROBHD    5 mm QNP 1H
PULPROG   zg
TD         32768
SOLVENT   CDCl3
NS         32
DS         2
SWH        2003.205 Hz
FIDRES     0.061133 Hz
AQ         8.1789427 sec
RG         912.3
DM         249.600 usec
DE         6.00 usec
TE         300.0 K
P1         2.0000000 sec
P2         8.00 usec
SFO1      250.1310005 MHz
NUC1       1H
PL1        -3.00 dB

F2 - Processing parameters
SI         16384
SF         250.1300130 MHz
WDW        EM
SSB        0
LB         0.20 Hz
GB         0
PC         1.00

1D NMR plot parameters
CX         20.00 cm
F1P        5.500 ppm
F1         1375.72 Hz
F2P        3.500 ppm
F2         875.46 Hz
PRNCH      0.10000 ppm/cm
HZ/CM      25.01300 Hz/cm
    
```



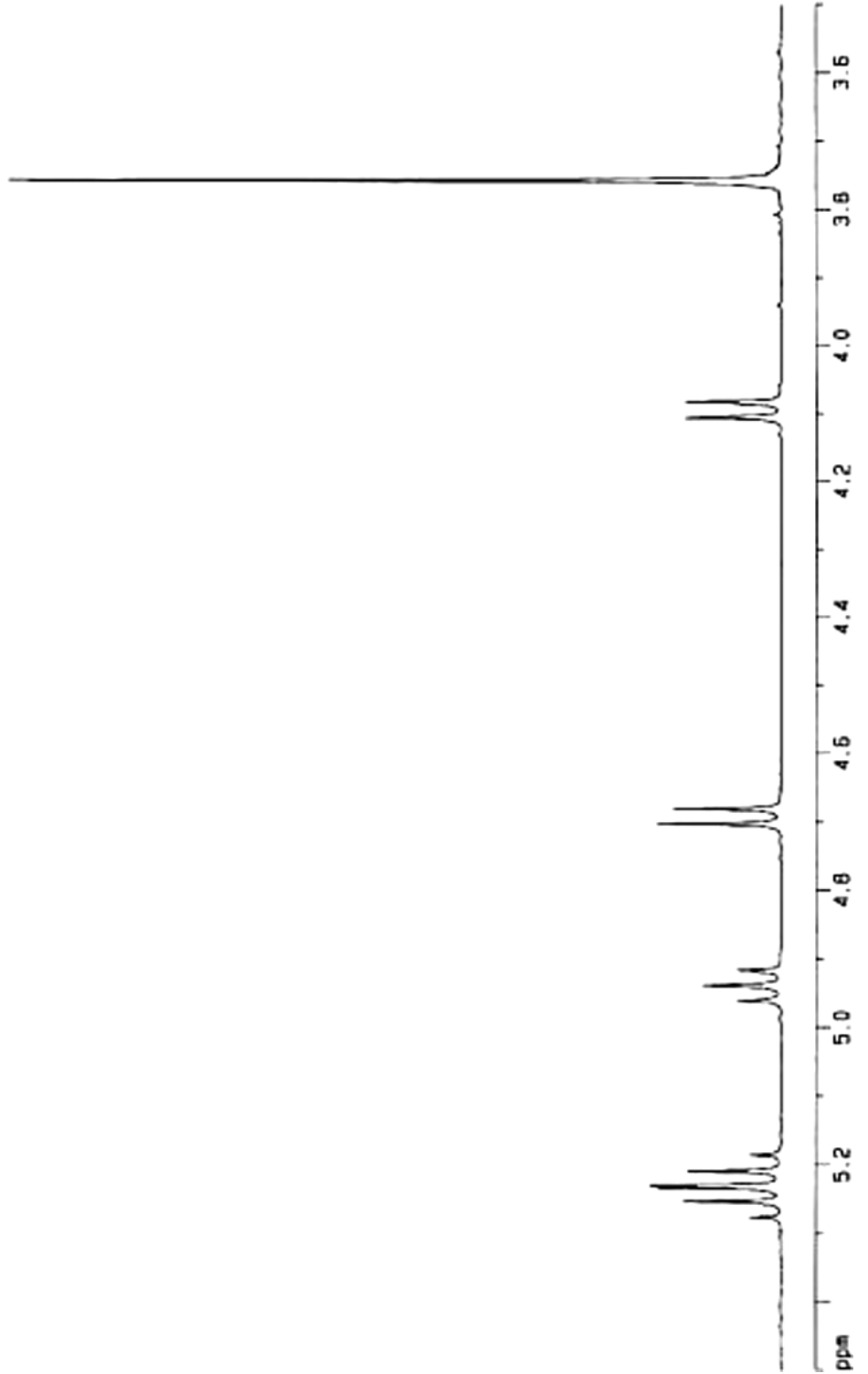
Current Data Parameters  
 NAME azgluc  
 EXPNO 1  
 PROCNO 1

F2 - Acquisition Parameters  
 Date\_ 980928  
 Time\_ 14.05  
 INSTRUM spect  
 PROBHD 5 mm TBO

PULPROG zg  
 TD 32768  
 SOLVENT CDCl3  
 NS 8  
 DS 2  
 SWH 3205.128 Hz  
 FIDRES 0.097813 Hz  
 AQ 5.118579 sec  
 RG 256  
 DM 156.000 usec  
 DE 6.00 usec  
 TE 300.0 K  
 O1 2.0000000 sec  
 P1 10.00 usec  
 SF01 400.1316005 MHz  
 NUC1 1H  
 PL1 -6.00 dB

F2 - Processing parameters  
 SI 16384  
 SF 400.1300170 MHz  
 MDW EM  
 SSB 0  
 LB 0.30 Hz  
 GB 0  
 PC 1.00

ID NMR plot parameters  
 CX 20.00 cm  
 FIP 5.500 ppm  
 F1 2200.72 Hz  
 F2P 3.500 ppm  
 F2 1400.46 Hz  
 SPMCM 0.50000 ppm/cm  
 HZCM 40.01300 Hz/cm





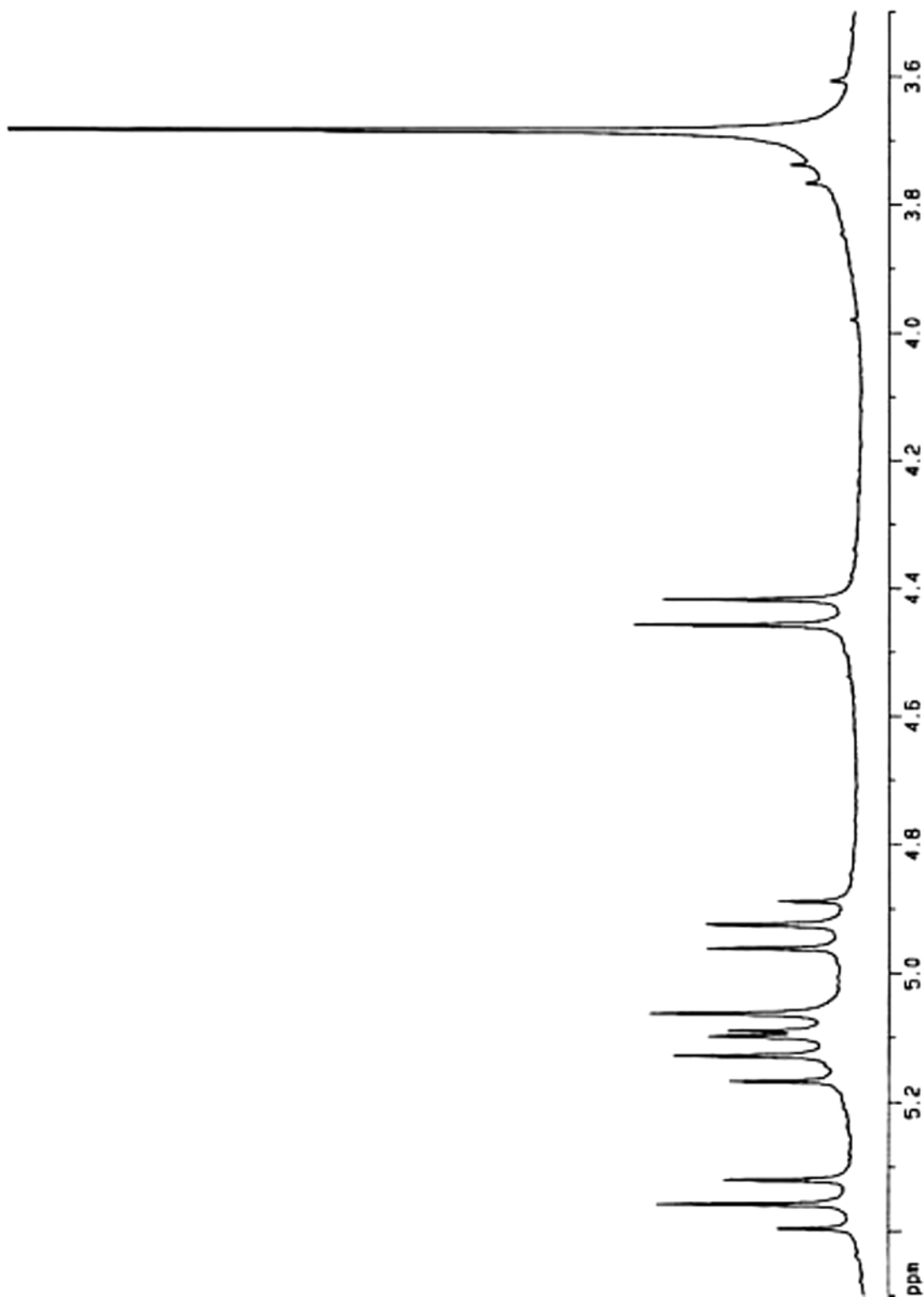
Current Data Parameters  
 NAME az1doglucdmk  
 EXPNO 1  
 PROCNO 1

F2 - Acquisition Parameters  
 Date\_ 990203  
 Time 11.41  
 INSTRUM spect  
 PROBHD 5 mm QNP 1H  
 PULPROG zg  
 TD 32768  
 SOLVENT Aceton  
 NS 32  
 DS 2  
 SWH 2003.205 Hz  
 FIDRES 0.061133 Hz  
 AQ 0.1789427 sec  
 RG 645.1  
 DM 249.600 usec  
 DE 5.00 usec  
 TE 300.0 K  
 O1 2.00000000 sec  
 P1 0.00 usec  
 SF01 250.1310005 MHz  
 NUC1 1H  
 PL1 -3.00 dB

F2 - Processing parameters  
 SI 16384  
 SF 250.1300130 MHz  
 WDW EM  
 SSB 0  
 LB 0.20 Hz  
 GB 0  
 PC 1.00

1D NMR plot parameters  
 CX 20.00 cm  
 F1P 5.500 ppm  
 F1 1375.72 Hz  
 F2P 3.500 ppm  
 F2 875.46 Hz  
 PPMCN 0.10000 ppm/cm  
 HZCN 25.01300 Hz/cm

<sup>1</sup>H az1dogluc @ 250MHz in DMK-d6



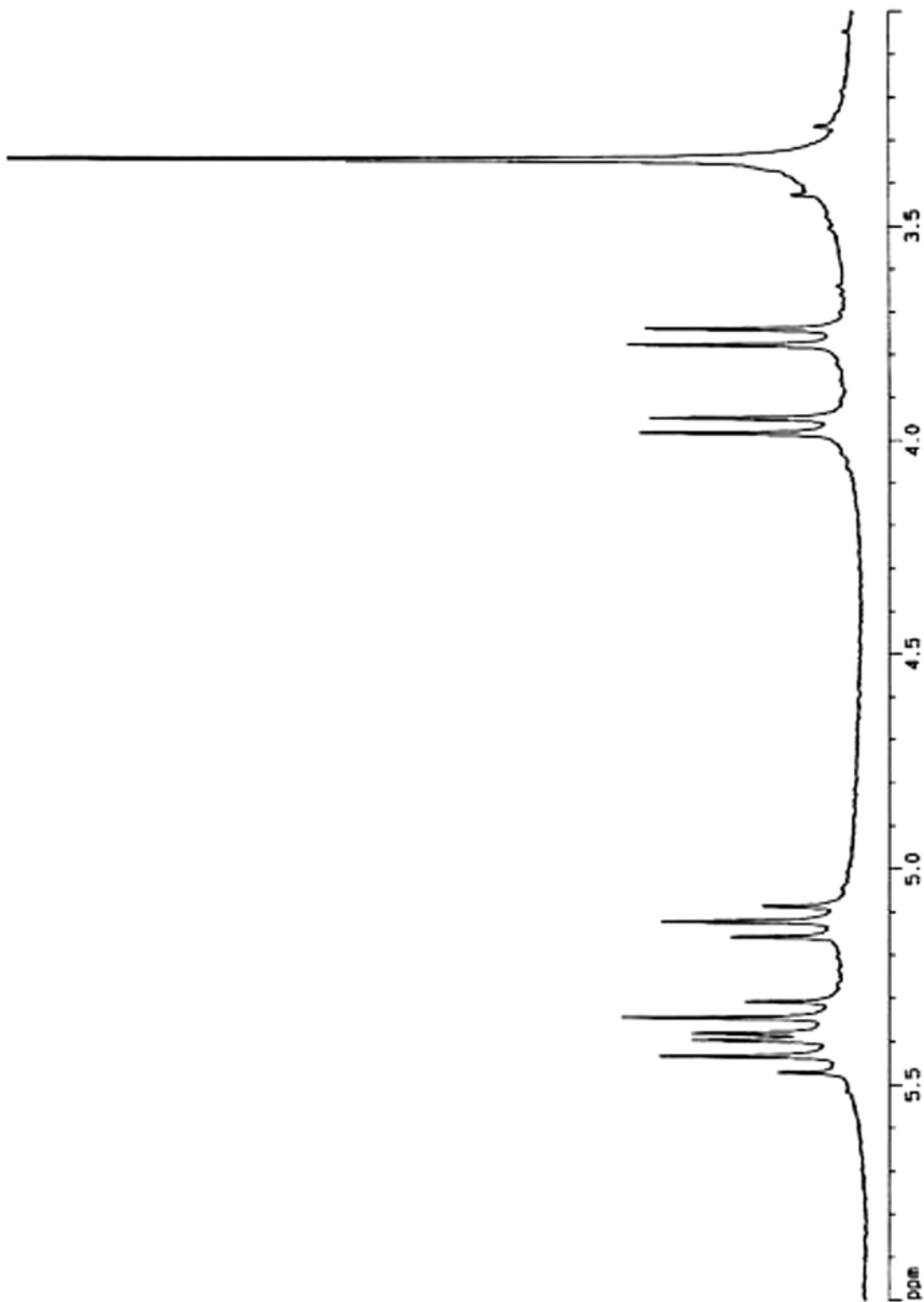
<sup>1</sup>H azidogluguc @ 250MHz in C606

Current Data Parameters  
 NAME azidoglugucbenz  
 EXPNO 1  
 PROCNO 1

F2 - Acquisition Parameters  
 Date\_ 990203  
 Time 17.23  
 INSTRUM spect  
 PROCNO 5  
 PULPROG 1H  
 TO 32768  
 SOLVENT C606  
 NS 32  
 DS 2  
 SWH 2003.205 Hz  
 FIDRES 0.061133 Hz  
 AQ 8.1789427 sec  
 RG 1624  
 DR 249.600 usec  
 DE 6.00 usec  
 TE 300.0 K  
 D1 2.00000000 sec  
 P1 8.00 usec  
 SF01 250.1310005 MHz  
 NUC1 1H  
 PL1 -3.00 dB

F2 - Processing parameters  
 SI 16384  
 SF 250.1300130 MHz  
 WDM EM  
 SSB 0  
 LB 0.20 Hz  
 GB 0  
 PC 1.00

10 NMR plot parameters  
 CK 20.00 cm  
 F1P 6.000 ppm  
 F1 1500.78 Hz  
 F2P 3.000 ppm  
 F2 750.39 Hz  
 PPMCM 0.15000 ppm/cm  
 HZCM 37.51950 Hz/cm



<sup>1</sup>H azidoaluc @ 250MHz in MeOH-d4

Current Data Parameters  
 NAME azidoalucmeth  
 EXPNO 1  
 PROCNO 1

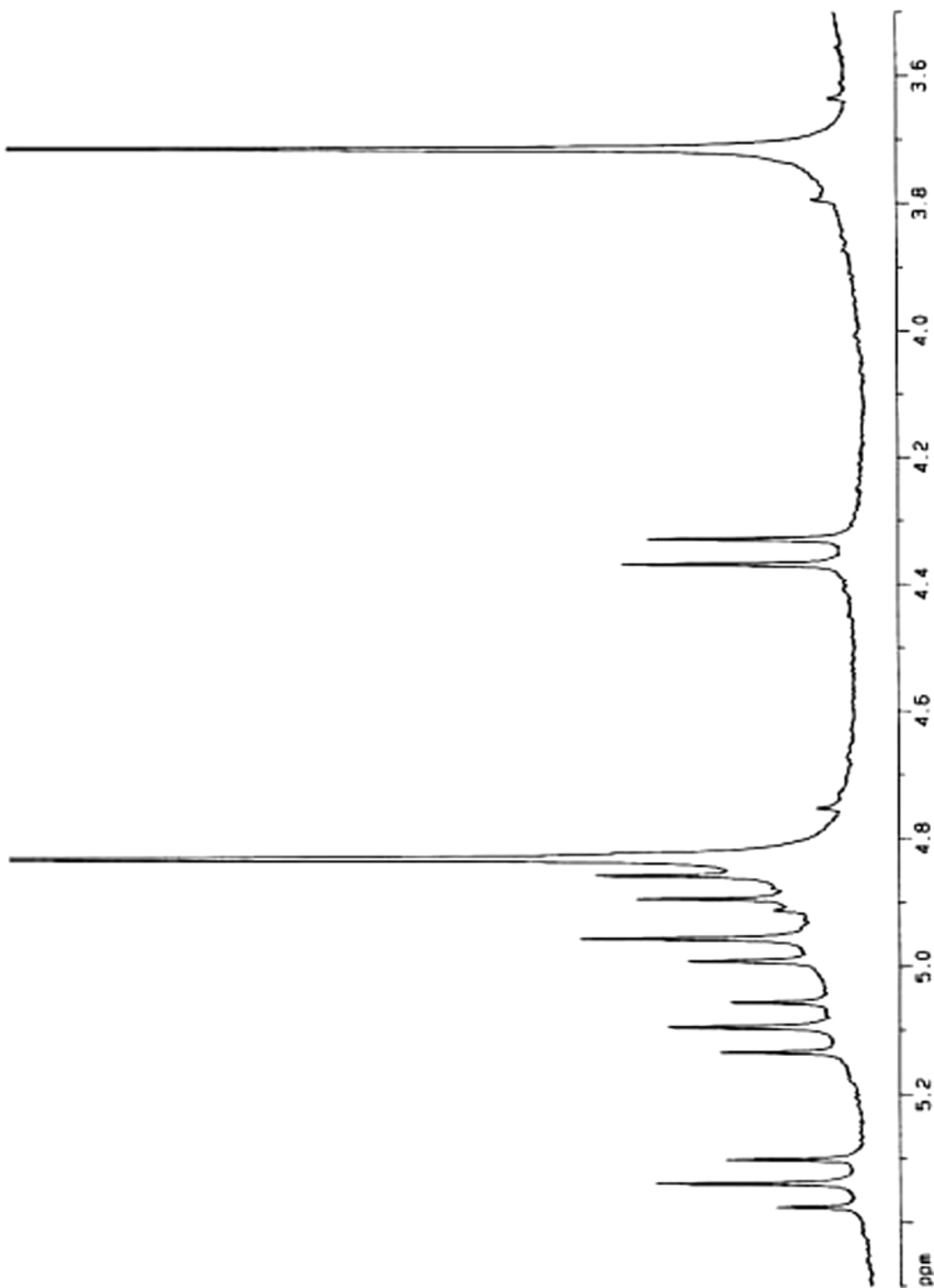
F2 - Acquisition Parameters

Date\_ 990203  
 Time 15.02  
 INSTRUM spect  
 PROBR0 5 mm QNP 1H  
 PULPROG zg  
 TD 32768  
 SOLVENT MeOH  
 NS 32  
 DS 2  
 SWH 2003.205 Hz  
 FIDRES 0.081133 Hz  
 AQ 0.1709427 sec  
 RG 1024  
 DM 249.600 usec  
 DE 6.00 usec  
 TE 300.0 K  
 D1 2.0000000 sec  
 P1 8.00 usec  
 SF01 250.1310005 MHz  
 NUC1 1H  
 PL1 -3.00 dB

F2 - Processing parameters

SI 16384  
 SF 250.1300130 MHz  
 MDW CM  
 SSB 0  
 LB 0.20 Hz  
 GB 0  
 PC 1.00

3D NMR plot parameters  
 CX 20.00 cm  
 F1P 5.500 ppm  
 F1 1375.72 Hz  
 F2P 3.500 ppm  
 F2 875.46 Hz  
 PPM0H 0.10000 ppm/cm  
 MZCN 25.01300 Hz/cm



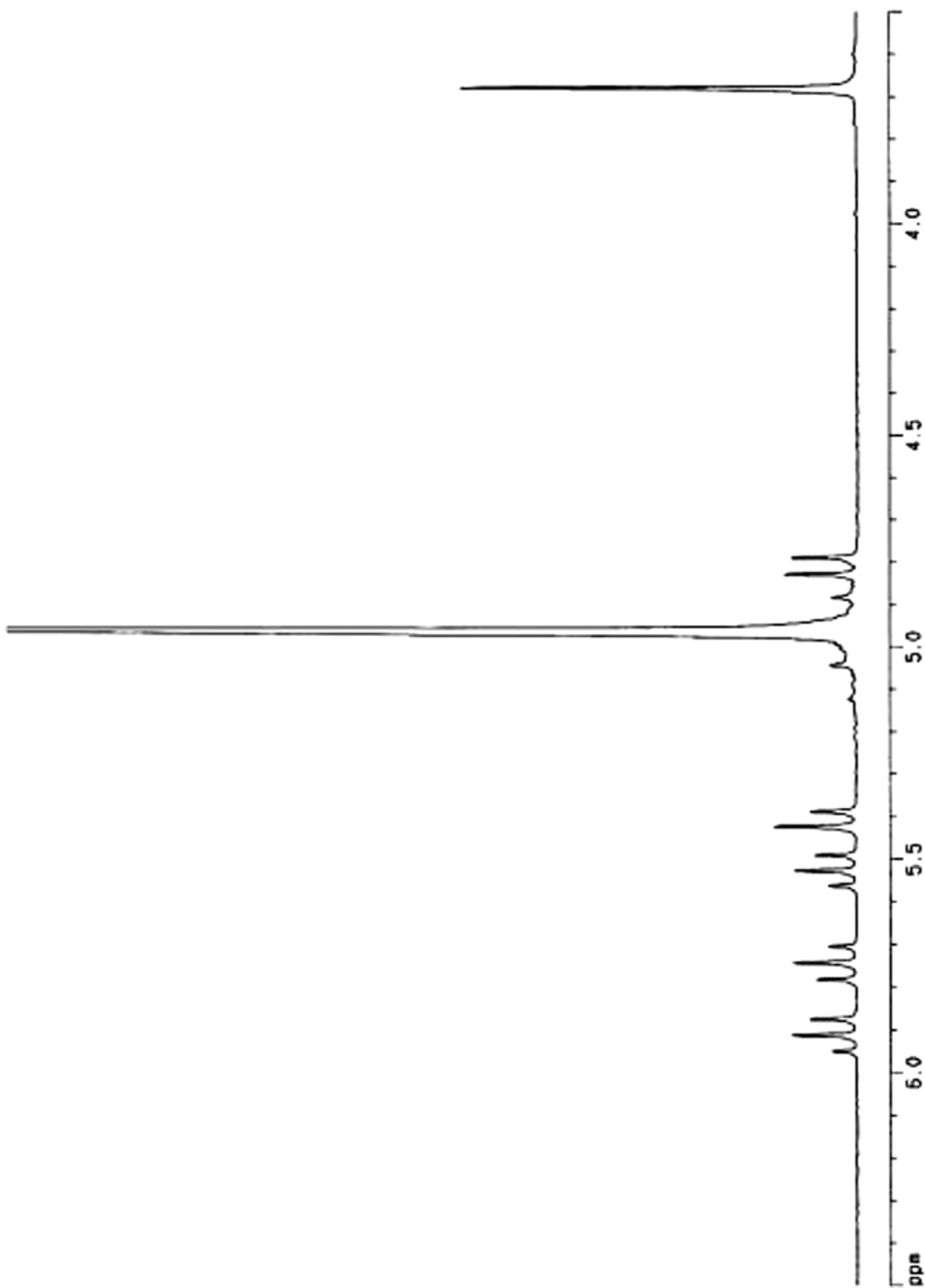
<sup>1</sup>H azidoogluc @ 250MHz in pyr-d5

Current Data Parameters  
 NAME azidooglucpyr  
 EXPNO 1  
 PROCNO 1

F2 - Acquisition Parameters  
 Date\_ 990204  
 Time 9.06  
 INSTRUM spect  
 PROBN0 5 mm GNP 1H  
 PULPROG zg  
 TD 32768  
 SOLVENT Aceton  
 NS 32  
 DS 2  
 SWH 2495.010 Hz  
 FIDRES 0.076142 Hz  
 AQ 6.5667572 sec  
 RG 400  
 ON 200.400 usec  
 DE 6.00 usec  
 TE 300.0 K  
 P1 2.0000000 sec  
 P1 8.00 usec  
 SFO1 250.1312505 MHz  
 NUC1 1H  
 PL1 -3.00 dB

F2 - Processing parameters  
 SI 16384  
 SF 250.1303781 MHz  
 WDW EM  
 SSB 0  
 LB 0.20 Hz  
 GB 0  
 PC 1.00

1D NMR plot parameters  
 CX 20.00 cm  
 F1P 6.500 ppm  
 F1 1625.85 Hz  
 F2P 3.500 ppm  
 F2 875.46 Hz  
 PPMON 0.15000 ppm/cm  
 HQDM 37.51925 Hz/cm



<sup>1</sup>H azidooglucuronide in CD2Cl2

```

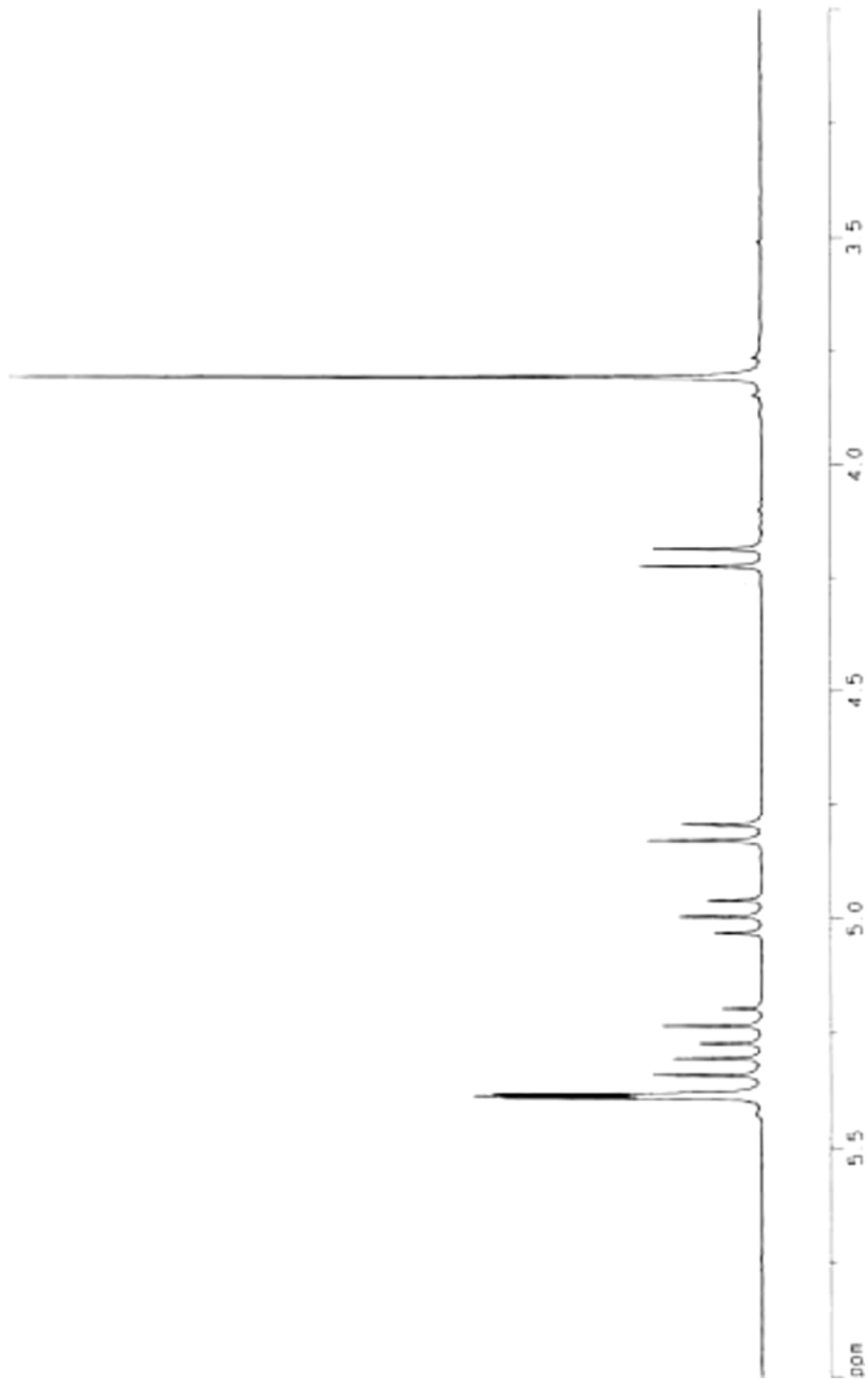
Current Data Parameters
NAME      azidoogluc13c
EXPNO    6
PROCNO    1

F2 - Acquisition Parameters
Date_     20001201
Time      11:07
INSTRUM   spect
PROBHD    5 mm QNP 1H
PULPROG   zg
TD         32768
SOLVENT   CD2Cl2
NS        8
DS        2
SWH        1755.618 Hz
FIDRES     0.053577 Hz
AQ         9.3323765 sec
RG         912.3
DM         284.800 usec
DE         6.00 usec
TE         300.0 K
D1         2.5000000 sec

***** CHANNEL f1 *****
NUC1       1H
P1         11.00 usec
PL1        -3.00 dB
SFO1       250.131256 MHz

F2 - Processing parameters
SI         32768
SF         250.129942 MHz
WDW        EM
SSB        0
LB         0.20 Hz
GB         0
PC         1.40

1D NMR plot parameters
CK         20.00 cm
F1P        6.000 ppm
F1         1500.78 Hz
F2P        3.000 ppm
F2         750.39 Hz
RGCM       0.15000 ppm/cm
HZCM       37.51950 Hz/cm
    
```



1H azidogluconide in THF-d8

Current Data Parameters  
 NAME azidoglucon13c  
 EXPNO 7  
 PROCNO 1

F2 - Acquisition Parameters

Date\_ 20001201  
 Time 11 18  
 INSTRUM spect  
 PULPROG 5 ms GMP 1H  
 TD 79  
 FID 32768  
 SOLVENT Aceton  
 NS 8  
 DS 2  
 SWH 1755.618 Hz  
 FIDRES 0.053577 Hz  
 AQ 9.3323785 sec  
 RG 812.7  
 DM 284.600 usec  
 DE 6.00 usec  
 TE 300.0 K  
 D1 2.5000000 sec

\*\*\*\*\* CHANNEL f1 \*\*\*\*\*

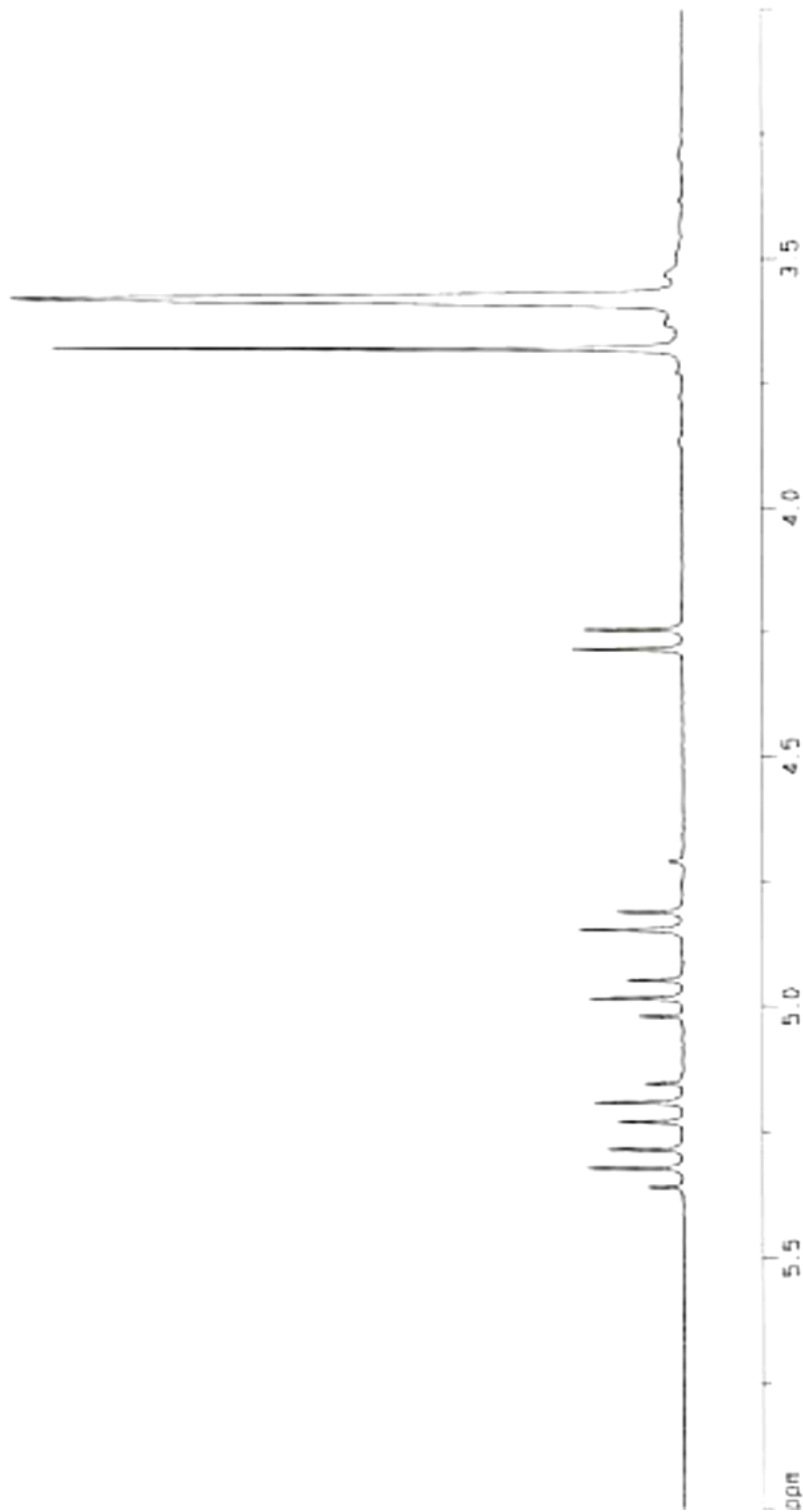
NUC1 1H  
 P1 11.00 usec  
 PL1 -3.00 dB  
 SFO1 250.131256 MHz

F2 - Processing parameters

SI 32768  
 SF 250.1300102 MHz  
 MDW EM  
 SSB 0  
 LB 0.20 Hz  
 GB 0  
 PC 1.40

3D NMR plot parameters

CX 20.00 cm  
 FIP 6.000 ppm  
 F1 1500.78 Hz  
 F2P 3.000 ppm  
 F2 750.39 Hz  
 PPMCM 0.15000 ppm/cm  
 HZCM 37.51950 Hz/cm



<sup>1</sup>H azidogluconide at 250MHz in CDC13 undecoupled

```

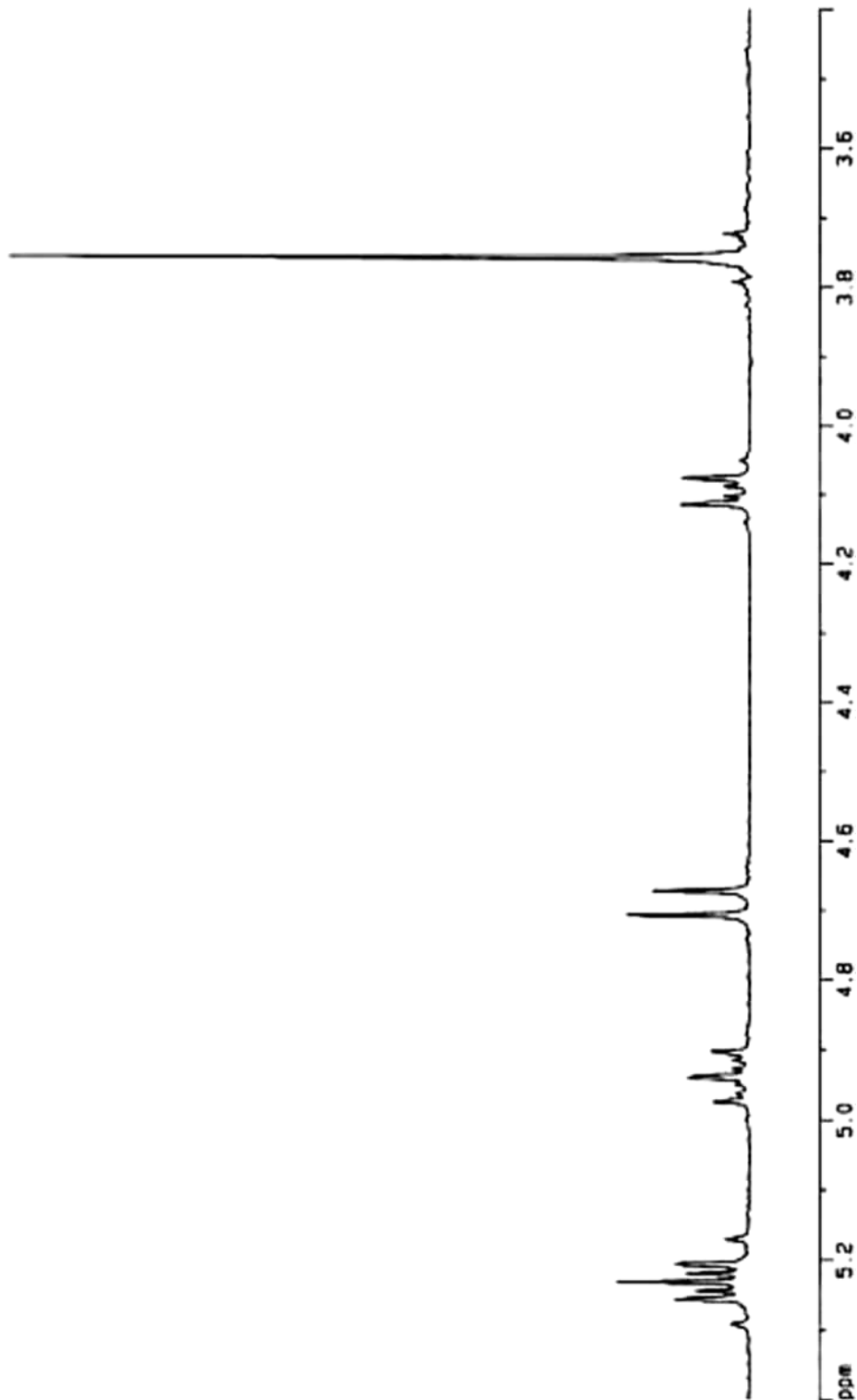
Current Data Parameters
NAME azidogluconide
EXPNO 9
PROCNO 1

F2 - Acquisition Parameters
Date_ 20001031
Time 9.14
INSTRUM spect
PROBHD 5 mm QNP 1H
PULPROG zg
TD 32768
SOLVENT CDC13
NS 1
DS 0
SWH 1745.810 Hz
FIDRES 0.053278 Hz
AQ 9.3848047 sec
RG 645.1
DM 286.400 usec
DE 6.00 usec
TE 300.0 K
D1 2.5000000 sec

***** CHANNEL f1 *****
NUC1 1H
P1 11.00 usec
PL1 -3.00 dB
SFO1 250.1311256 MHz

F2 - Processing parameters
SI 32768
SF 250.1300126 MHz
WDW EM
SSB 0
LB 0.20 Hz
GB 0
PC 1.40

1D NMR plot parameters
CX 20.00 cm
FIP 5.400 ppm
F1 1350.70 Hz
F2P 3.400 ppm
F2 850.44 Hz
PPMCM 0.10000 ppm/cm
HZCM 25.01300 Hz/cm
    
```



<sup>1</sup>H azidoglucuronide at 250MHz in CDCl<sub>3</sub> decoupled at 4.1ppm

Current Data Parameters  
 NAME azidogluc  
 EXPNO 10  
 PROCNO 1

F2 - Acquisition Parameters

Date\_ 20001031  
 Time 9.37  
 INSTRUM spect  
 PROBO 5 mm QNP 1H  
 PULPROG zgpg  
 TO 32768  
 SOLVENT CDCl<sub>3</sub>  
 NS 32  
 DS 2  
 SWH 1745.810 Hz  
 FIDRES 0.053278 Hz  
 AQ 9.3848047 sec  
 RG 645.1  
 DM 2865.400 usec  
 DE 6.00 usec  
 TE 300.0 K  
 D1 2.50000000 sec  
 d12 0.00002000 sec

\*\*\*\*\* CHANNEL f1 \*\*\*\*\*  
 NUC1 1H  
 P1 11.00 usec  
 PL1 -3.00 dB  
 SF01 250.131256 MHz

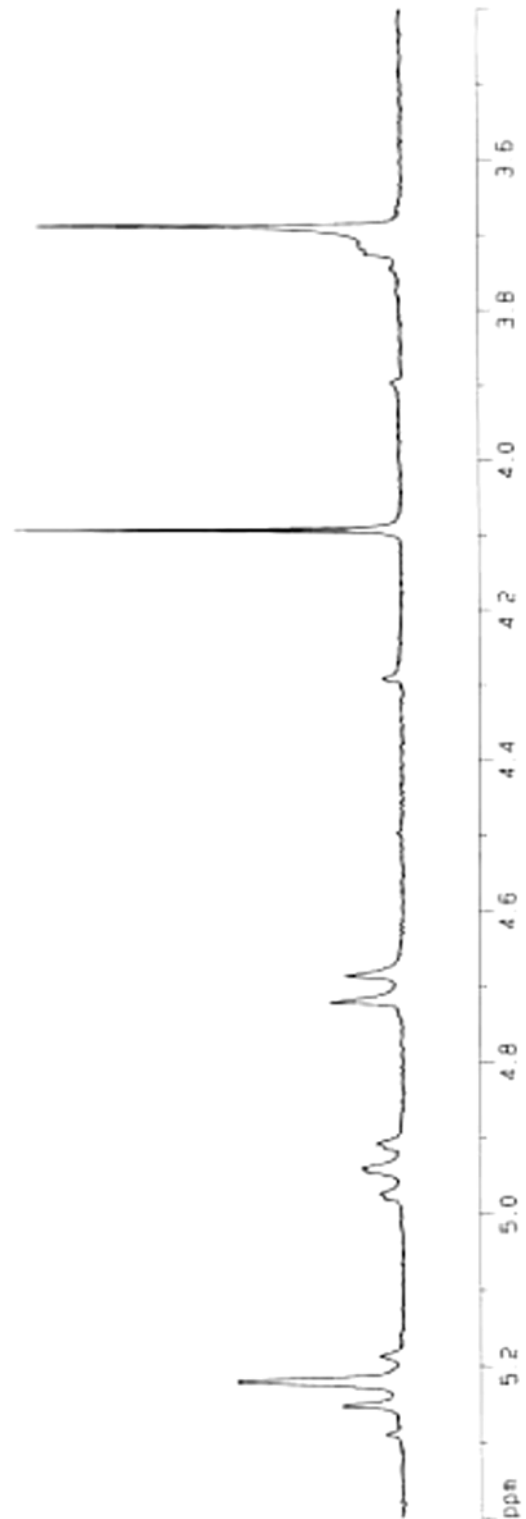
\*\*\*\*\* CHANNEL f2 \*\*\*\*\*  
 NUC2 1H  
 PL2 0.00 dB  
 PL24 40.00 dB  
 SF02 250.1310410 MHz

F2 - Processing parameters

SI 32768  
 SF 250.1300172 MHz  
 WDW EM  
 SSB 0  
 LB 0.20 Hz  
 GB 0  
 PC 1.40

10 NMR plot parameters

CK 20.00 cm  
 F1P 5.400 ppm  
 F1 1350.70 Hz  
 F2P 3.400 ppm  
 F2 850.44 Hz  
 PRGCM 0.10000 ppm/cm  
 HZCM 25.01300 Hz/cm





<sup>1</sup>H azidoglucuronide at 250MHz in CDCl<sub>3</sub> decoupled at 4.9ppm

Current Data Parameters  
 NAME azidogluc  
 EXPNO 11  
 PROCNO 1

F2 - Acquisition Parameters

Date\_ 20001031  
 Time 9.50  
 INSTRUM spect  
 PROBN0 5 mm QNP 1H  
 PULPROG zgpg30  
 TO 32768  
 SOLVENT Aceton  
 NS 32  
 DS 2  
 SWH 1745.810 Hz  
 FIDRES 0.053278 Hz  
 AQ 9.3048047 sec  
 RG 645.1  
 DM 286.400 usBC  
 DE 6.00 usBC  
 TE 300.0 K  
 D1 2.50000000 sec  
 d12 0.00062000 sec

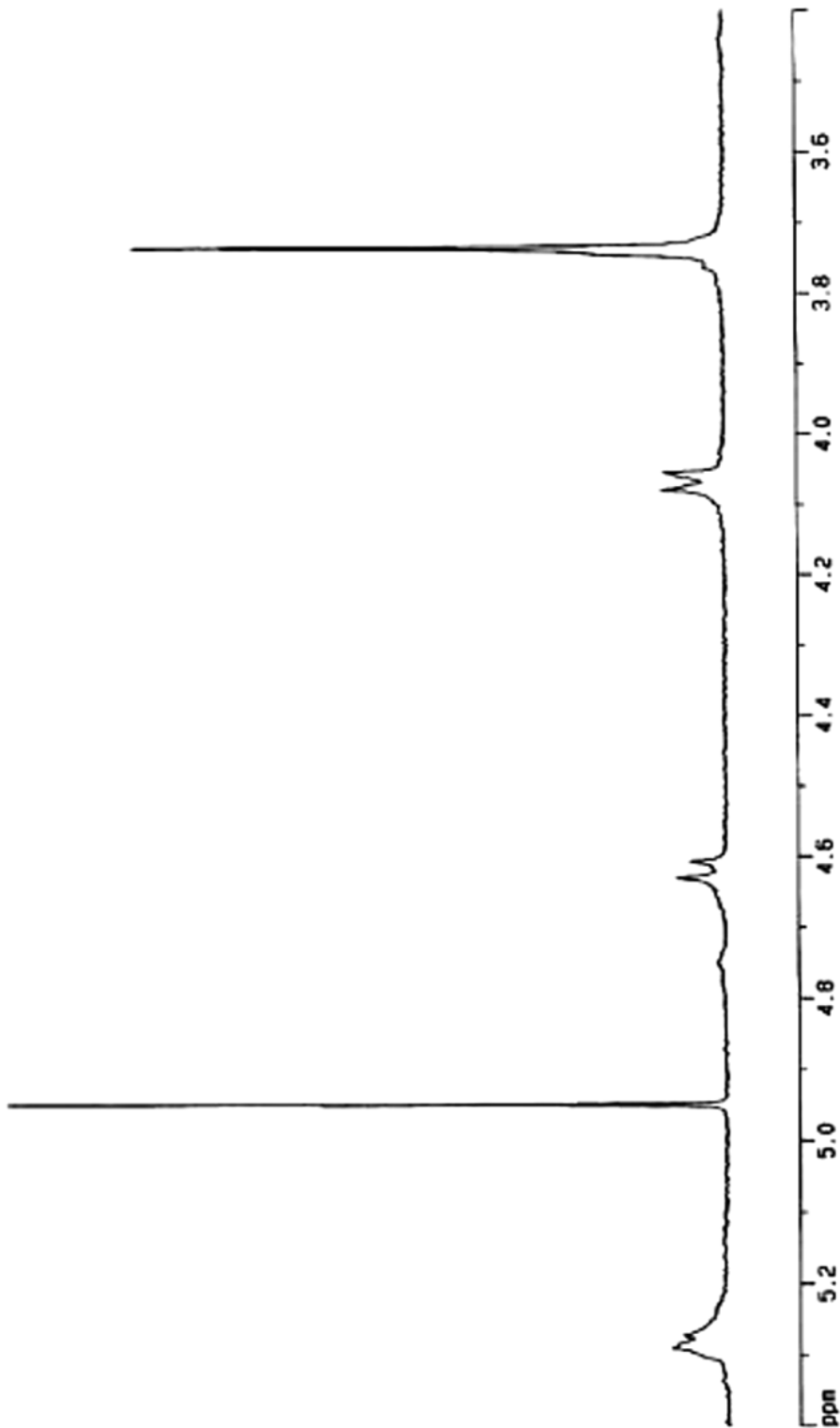
\*\*\*\*\* CHANNEL f1 \*\*\*\*\*  
 NUC1 1H  
 P1 11.00 usec  
 PL1 -3.00 dB  
 SF01 250.131255 MHz

\*\*\*\*\* CHANNEL f2 \*\*\*\*\*  
 NUC2 1H  
 PL2 0.00 dB  
 PL24 40.00 dB  
 SF02 250.131250 MHz

F2 - Processing parameters

SI 32768  
 SF 250.1300143 MHz  
 MDW EN  
 SSB 0  
 LB 0.20 Hz  
 GB 0  
 PC 1.40

1D NMR plot parameters  
 CX 20.00 cm  
 FJP 5.400 ppm  
 F1 1350.70 Hz  
 F2P 3.400 ppm  
 F2 850.44 Hz  
 PPH0H 0.10000 ppm/cm  
 HZCN 25.01300 Hz/cm



<sup>1</sup>H azidoogluc in CDCl<sub>3</sub> at 25 degrees

Current Data Parameters  
 NAME azidoogluc  
 EXPNO 5  
 PROCNO 1

F2 - Acquisition Parameters

Date\_ 20001012  
 Time 13 55  
 INSTRUM spect  
 PROBHD 5 mm DNP 1H  
 PULPROG zg  
 TD 32768  
 SOLVENT CDCl<sub>3</sub>  
 NS 8  
 DS 2  
 SWH 1755.618 Hz  
 FIDRES 0.052577 Hz  
 AQ 9.3323765 sec  
 RG 512  
 DM 294.800 usec  
 DE 6.00 usec  
 TE 300.0 K  
 D1 2.5000000 sec

\*\*\*\*\* CHANNEL f1 \*\*\*\*\*

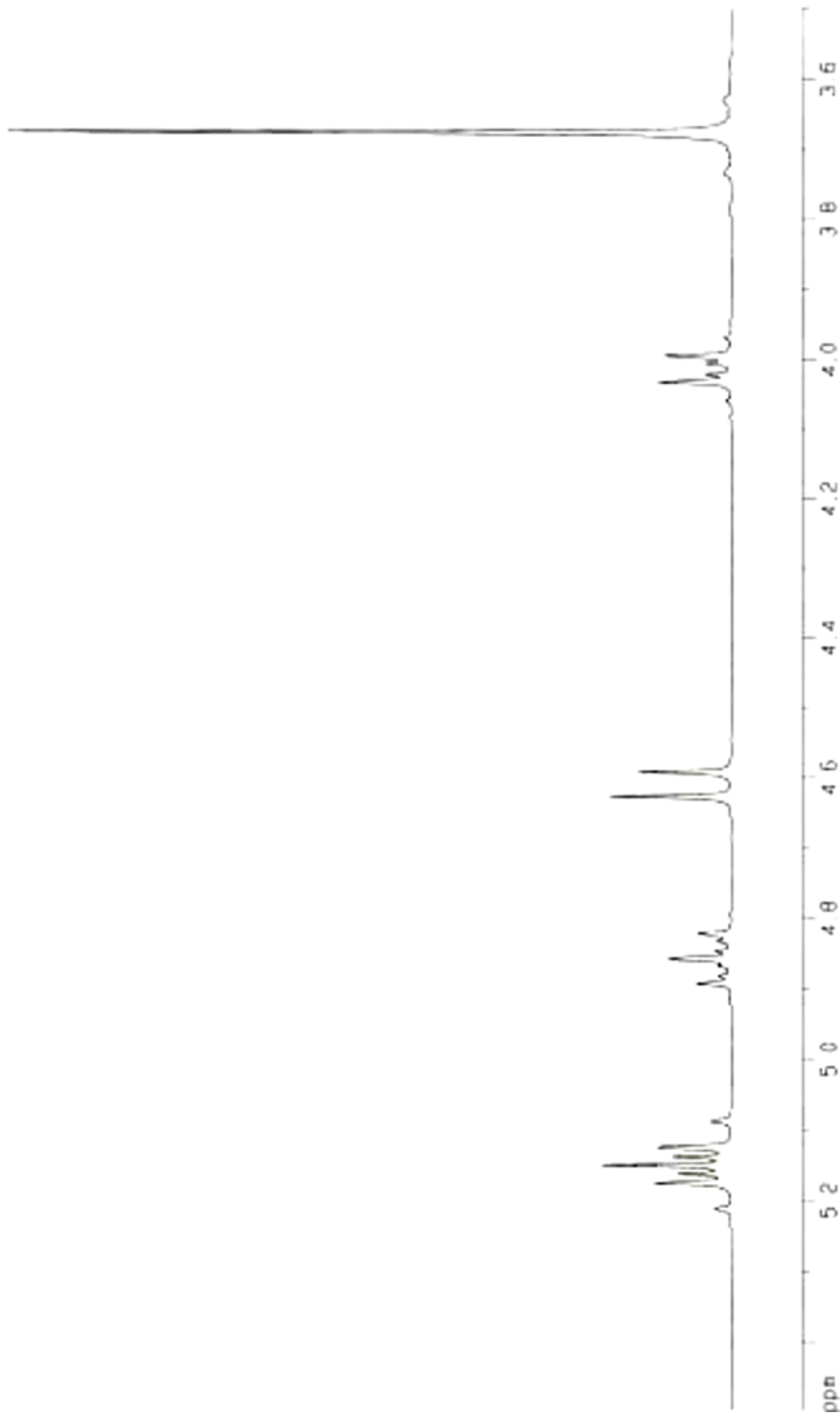
NUC1 1H  
 P1 11.00 usec  
 PL1 -3.00 dB  
 SFO1 250.131255 MHz

F2 - Processing parameters

SF 250.130028 MHz  
 WDW EM  
 SSB 0  
 LB 0.20 Hz  
 GB 0  
 PC 1.40

1D NMR plot parameters

CK 20.00 cm  
 F1P 5.500 ppm  
 F1 1375.71 Hz  
 F2P 3.500 ppm  
 F2 875.66 Hz  
 PRGCM 0 10000 dppm/cm  
 HZCM 25 01300 Hz/cm



<sup>1</sup>H azidogluguc in CDCl<sub>3</sub> at 35 degrees

Current Data Parameters  
 NAME azidogluguc  
 EXPRNO 6  
 PROCNO 1

F2 - Acquisition Parameters

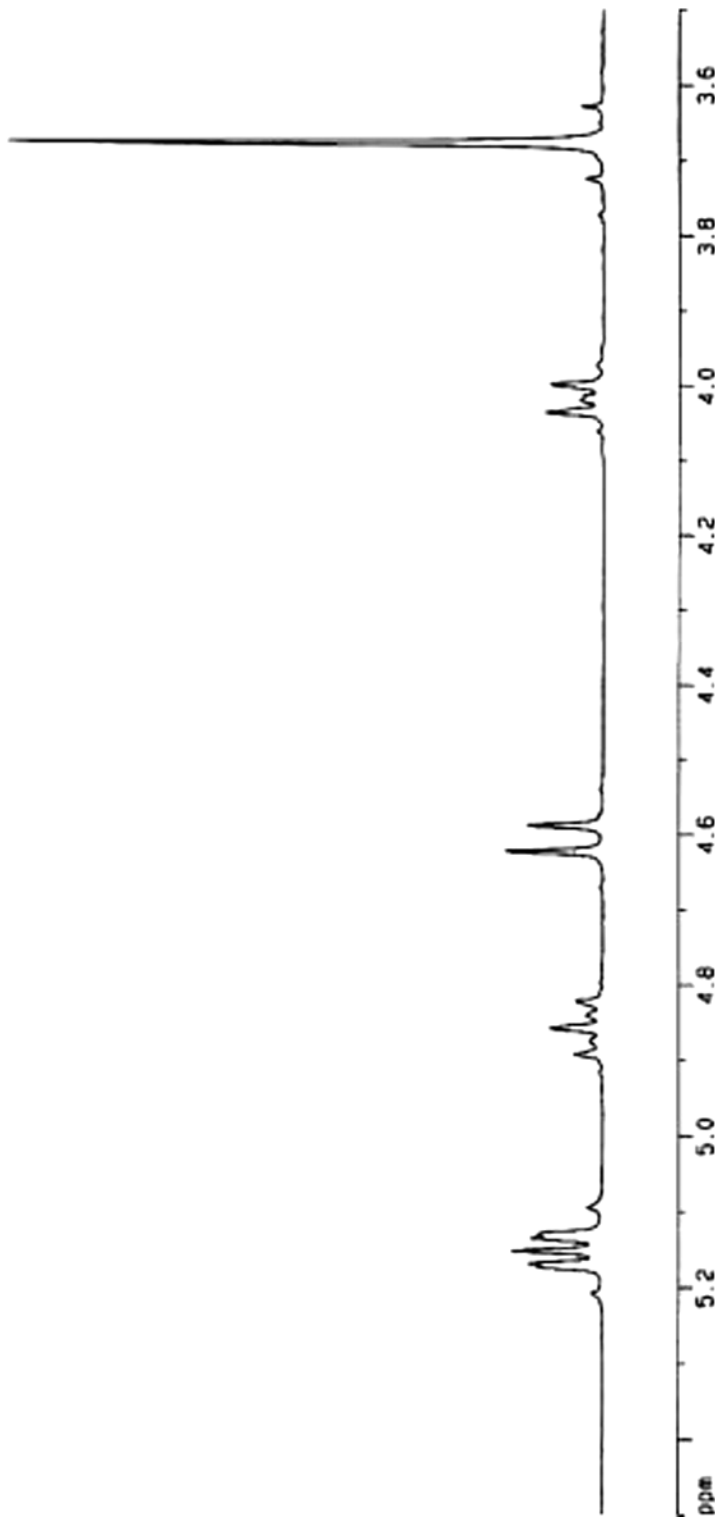
Date\_ 20001012  
 Time 13.59  
 INSTRUM spect  
 PROBHD 5 mm DNP 1H  
 PULPROG zg  
 TD 32768  
 SOLVENT CDCl<sub>3</sub>  
 NS 8  
 DS 2  
 SMH 1755.618 Hz  
 FIDRES 0.003577 Hz  
 AQ 9.3323765 sec  
 RG 512  
 DM 284.800 usec  
 DE 6.00 usec  
 TE 300.0 K  
 D1 2.30000000 sec

\*\*\*\*\* CHANNEL f1 \*\*\*\*\*

NUC1 1H  
 P1 11.00 usec  
 PL1 -3.00 dB  
 SF01 250.1311256 MHz

F2 - Processing parameters

SI 32768  
 SF 250.1300328 MHz  
 MCM EM  
 SSB 0  
 LB 0.20 Hz  
 GB 0  
 PC 1.40  
 ID MHz pilot parameters  
 CX 20.00 cm  
 FIP 5.500 ppm  
 F1 1375.72 Hz  
 F2P 3.500 ppm  
 F2 875.46 Hz  
 PPMCH 0.10000 ppm/cm  
 HZDM 25.01300 Hz/cm



<sup>1</sup>H NMR of azido-gluc in CDCl<sub>3</sub> at 45 degrees

Current Data Parameters  
 NAME azidogluc  
 EXPNO 7  
 PROCNO 1

F2 - Acquisition Parameters

Date\_ 20001012  
 Time 14.03  
 INSTRUM spect  
 PROCNO 5 nu DNP 1H  
 PULPROG zg  
 TD 32768  
 SOLVENT CDCl<sub>3</sub>  
 NS 8  
 DS 2  
 SHH 1755.618 Hz  
 FIDRES 0.053577 Hz  
 AQ 9.3323765 sec  
 RG 512  
 DM 284.800 usec  
 DE 6.00 usec  
 TE 300.0 K  
 D1 2.5000000 sec

\*\*\*\*\* CHANNEL f1 \*\*\*\*\*  
 NUC1 1H  
 P1 11.00 usec  
 PL1 -3.00 dB  
 SF0: 250.1311256 MHz

F2 - Processing parameters

SF 32768  
 SF 250.1300320 MHz  
 WDW EM  
 SSB 0  
 LB 0.70 Hz  
 GB 0  
 PC 1.40  
 ID MR plot parameters  
 CX 20.00 cm  
 FIP 5.500 ppm  
 F1 1375.72 Hz  
 F2P 3.500 ppm  
 F2 875.46 Hz  
 PPMCN 0.10000 ppm/cm  
 HZCN 25.01300 Hz/cm

