Formation of the Stable Structural Analog of ADP-sensitive Phosphoenzyme of Ca²⁺-ATPase with Occluded Ca²⁺ by Beryllium Fluoride

STRUCTURAL CHANGES DURING PHOSPHORYLATION AND ISOMERIZATION*

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As a stable analog for ADP-sensitive phosphorylated intermediate of sarcoplasmic reticulum Ca²⁺-ATPase E1PCa₂·Mg, a complex of $E1Ca_2$ ·BeF_x, was successfully developed by addition of beryllium fluoride and Mg^{2+} to the Ca^{2+} -bound state, $E1Ca_2$. In $E1Ca_2 \cdot BeF_x$, most probably $E1Ca_2 \cdot BeF_3^-$, two Ca^{2+} are occluded at high affinity transport sites, its formation required Mg²⁺ binding at the catalytic site, and ADP decomposed it to E1Ca₂, as in E1PCa₂·Mg. Organization of cytoplasmic domains in $E1Ca_2$ ·BeF_x was revealed to be intermediate between those in $E1Ca_2 \cdot AIF_4^-$ ADP (transition state of $E1PCa_2$ formation) and $E2 \cdot BeF_3 \cdot (ADP - insensitive phosphorylated intermediate$ E2P·Mg). Trinitrophenyl-AMP (TNP-AMP) formed a very fluorescent (superfluorescent) complex with $E1Ca_2 \cdot BeF_r$ in contrast to no superfluorescence of TNP-AMP bound to E1Ca₂·AIF_x. $E1Ca_2$ ·BeF_x with bound TNP-AMP slowly decayed to $E1Ca_2$, being distinct from the superfluorescent complex of TNP-AMP with E2·BeF₃, which was stable. Tryptophan fluorescence revealed that the transmembrane structure of *E*1Ca₂·BeF_x mimics $E1PCa_2$ ·Mg, and between those of $E1Ca_2$ ·AlF₄·ADP and E2·BeF₃. $E1Ca_2 \cdot BeF_x$ at low 50–100 μ M Ca²⁺ was converted slowly to $E2 \cdot BeF_3^-$ releasing Ca²⁺, mimicking $E1PCa_2 \cdot Mg \rightarrow E2P \cdot Mg +$ 2Ca²⁺. Ca²⁺ replacement of Mg²⁺ at the catalytic site at approximately millimolar high Ca^{2+} decomposed $E1Ca_2 \cdot BeF_x$ to E1Ca₂. Notably, E1Ca₂·BeF_x was perfectly stabilized for at least 12 days by 0.7 mm lumenal Ca²⁺ with 15 mm Mg²⁺. Also, stable $E1Ca_2 \cdot BeF_x$ was produced from $E2 \cdot BeF_3^-$ at 0.7 mM lumenal Ca²⁺ by binding two Ca²⁺ to lumenally oriented low affinity transport sites, as mimicking the reverse conversion E2P. $Mg + 2Ca^{2+} \rightarrow E1PCa_2 \cdot Mg.$

Sarcoplasmic reticulum Ca²⁺-ATPase (SERCA1a),² a representative member of the P-type ion transporting ATPases, cat-



Various intermediate structural states in the transport cycle were fixed as their structural analogs produced by appropriate ligands such as AMP-PCP (non-hydrolyzable ATP analog) or metal fluoride compounds (phosphate analogs), and their crystal structures were solved at the atomic level (15-22). The three cytoplasmic domains, N, P, and A, largely move and change their organization state during the transport cycle, and the changes are coupled with changes in the transport sites. Most remarkably, in the change from $E1Ca_2 \cdot AlF_4 \cdot ADP$ (the transition state for *E*1PCa₂ formation, *E*1PCa₂·ADP·Mg[‡]) to *E*2·BeF₃⁻ (the ground state E2P·Mg) (23-25), the A domain largely rotates by more than 90° approximately parallel to the membrane plane and associates with the P domain, thereby destroying the Ca²⁺ binding sites, and opening the lumenal gate, thus releasing Ca²⁺ into the lumen (see Fig. 2). E1PCa₂·Ca·AMP-PN formed by CaAMP-PNP without Mg²⁺ is nearly the same as

rylation; A, actuator; AMP-PNP, adenosine 5'- $(\beta, \gamma$ -imido)triphosphate; AMP-PCP, adenosine 5'- $(\beta, \gamma$ -methylene)triphosphate; AMP-PN, adenosine 5'-diphosphoramidate.



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² The abbreviations used are: SERCA1a, adult fast-twitch skeletal muscle sarcoplasmic reticulum Ca²⁺-ATPase; SR, sarcoplasmic reticulum; EP, phosphoenzyme; E1P, ADP-sensitive phosphoenzyme; E2P, ADP-insensitive phosphoenzyme; MOPS, 3-(*N*-morpholino)propanesulfonic acid; TG, thapsigargin; TNP-AMP, 2'(3')-O-(trinitrophenyl)-AMP; C₁₂E_a, octaethylene glycol monododecyl ether; N, nucleotide binding; P, phospho-



 $E1Ca_2 \cdot AIF_4^- \cdot ADP$ and $E1Ca_2 \cdot CaAMP-PCP$ in their crystal structures (17, 18, 22).

Despite these atomic structures, yet unsolved is the structure of $E1PCa_2$ ·Mg, the genuine physiological intermediate $E1PCa_2$ with bound Mg²⁺ at the catalytic site without the nucleotide. Its stable structural analog has yet to be developed. E1PCa₂·Mg is the major intermediate accumulating almost exclusively at steady state under physiological conditions. Its rate-limiting isomerization results in Ca²⁺ deocclusion/release producing E2P·Mg as a key event for Ca^{2+} transport. In E1Ca₂·CaAMP-PCP, $E1Ca_2 \cdot AlF_4 \cdot ADP$, and $E1PCa_2 \cdot Ca \cdot AMP - PN$, the N and P domains are cross-linked and strongly stabilized by the bound nucleotide and/or Ca²⁺ at the catalytic site, thus they are crystallized (17, 18, 22). Kinetically, E1PCa2 •Ca formed with CaATP is markedly stabilized due to Ca²⁺ binding at the catalytic Mg²⁺ site, and its isomerization to E2P is strongly retarded in contrast to $E1PCa_2$ ·Mg (26, 27). Thus, the bound Ca^{2+} at the catalytic Mg²⁺ site likely produces a significantly different structural state from that with bound Mg^{2+} .

Therefore, it is now essential to develop a genuine E1PCa₂·Mg analog without bound nucleotide and thereby gain further insight into the structural mechanism in the Ca²⁺ transport process. It is also crucial to further clarify the structural importance of Mg²⁺ as the physiological catalytic cation. In this study, we successfully developed the complex $E1Ca_2 \cdot BeF_x$, most probably $E1Ca_2 \cdot BeF_3^-$, as the $E1PCa_2 \cdot Mg$ analog by adding beryllium fluoride (BeF_x) to the $E1Ca_2$ state without any nucleotides. For its formation, Mg²⁺ binding at the catalytic site was required and Ca²⁺ substitution for Mg²⁺ was absolutely unfavorable, revealing a likely structural reason for its preference as the physiological cofactor. In $E1Ca_2 \cdot BeF_3^-$, two Ca²⁺ ions bound at the high affinity transport sites are occluded. It was also produced from $E2 \cdot BeF_3^-$ by lumenal Ca²⁺ binding at the lumenally oriented low affinity transport sites, mimicking $E2P \cdot Mg + 2Ca^{2+} \rightarrow E1PCa_2 \cdot Mg$. All properties of the newly developed $E1Ca_2 \cdot BeF_3^-$ fulfilled the requirements as the *E*1PCa₂·Mg analog, and hence we were able to uncover the hitherto unknown nature of E1PCa2·Mg as well as structural events occurring in the phosphorylation and isomerization processes. Also, we successfully found the conditions that perfectly stabilize the $E1Ca_2 \cdot BeF_3^-$ complex.

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EXPERIMENTAL PROCEDURES

Preparation of SR Vesicles and Treatment with BeF_{x} , AlF_{x} , and AlFx: ADP—SR vesicles were prepared from rabbit skeletal muscle as described (28). The content of the phosphorylation site in the vesicles determined according to Barrabin et al. (29) was 4.49 ± 0.22 nmol/mg of vesicle protein (n = 5). The Ca²⁺dependent ATPase activity determined at 25 °C in a mixture containing 5 μ g/ml microsomal protein, 1 mM ATP, 1.7 μ M A23187, 7 mм MgCl₂, 0.1 м KCl, 50 mм MOPS/Tris (pH 7.0), and 0.6 $\rm mm~CaCl_2$ with 0.5 $\rm mm~EGTA$ (or 2 $\rm mm~EGTA$ without added CaCl₂) was $1.87 \pm 0.14 \,\mu \text{mol/min/mg}$ of vesicle protein (n = 3). The Ca²⁺-ATPase was purified from the vesicles by deoxycholate as described (30, 31). The E1Ca₂ state ATPase was incubated with fluoride compounds, 2 mM potassium fluoride and 100 $\mu{\rm M}$ BeSO4 or AlCl3, at 25 °C for 30 min in the presence of 0.1 mм Ca²⁺, 15 mм MgCl₂, 0.1 м KCl, 30 mм MOPS/Tris buffer (pH 7.0), unless stated otherwise. E1Ca₂· $AlF_4^- ADP$ was formed by including 50 μ M ADP in the above AlF_x incubation mixture as described (32). $E2 \cdot BeF_3^-$, $E2 \cdot AlF_4^-$, and $E2 \cdot MgF_4^{2-}$ were produced as described (23–25).

Determination of *EP*—*E*P formation was performed with 3 μ_{M} [γ -³²P]ATP in 100 (or 50) μ_{M} Ca²⁺ at 0 °C for 3 s, and terminated by trichloroacetic acid containing carrier P_i. The amount of *E*P formed was determined as described previously (28). The background level determined with excess 5 mM EGTA was less than 0.5% of the phosphorylation sites.

 Ca^{2+} Binding and Occlusion—⁴⁵Ca²⁺ binding and occlusion at the transport sites was determined at 25 °C with 2 ml of the SR vesicle mixture (0.2 mg/ml protein) with a 0.45- μ m nitrocellulose membrane filter (Millipore) as described (31). In some cases, the vesicles on the filter were washed for 10 s by perfusion with 2 ml of a washing solution containing 5 mM EGTA. The amount of Ca²⁺ bound at the transport sites was obtained by subtracting the nonspecific Ca²⁺ binding level determined as described in the figure legends.

Proteolysis—SR vesicles (0.45 mg/ml protein) were treated at 25 °C with trypsin (0.3 mg/ml) or proteinase K (0.1 mg/ml) as described (23, 24) and as noted in the figure legends. The samples were subjected to Laemmli SDS-PAGE (33) and densitometric analyses of the Coomassie Brilliant Blue R-250-stained gels (23, 24).

Fluorescence Measurements—The TNP-AMP fluorescence and intrinsic tryptophan fluorescence of Ca^{2+} -ATPase (0.06 mg/ml protein) were measured on a RF-5300PC spectrofluorophotometer (Shimadzu, Kyoto, Japan) with excitation and emission wavelengths 408 and 540 nm for TNP-AMP (with band widths 5 and 10 nm), and 290 and 338.4 nm for tryptophan (with bandwidth 1.5 and 5 nm), unless otherwise described (28).

Miscellaneous—Trypsin (L-1-tosylamido-2-phenylethyl chloromethyl ketone-treated) and proteinase K were obtained from Sigma. TNP-AMP was synthesized according to Hiratsuka (34). Protein concentrations were determined by the method of Lowry *et al.* (35) with bovine serum albumin as a standard. Free Ca²⁺ concentrations were calculated by the Calcon program. Data were analyzed by nonlinear regression using the program Origin (Microcal Software, Inc., Northampton,



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MA). Three-dimensional models of the enzyme were reproduced by the program VMD (36).

RESULTS

Formation of E1PCa₂ Analogs by Fluoride Compounds-The Ca²⁺-ATPase of SR vesicles in 100 µM Ca^{2+} and 15 mm Mg^{2+} was incubated with beryllium fluoride (BeF_r) or aluminum fluoride (AIF_r) for 30 min at 25 °C. The ability of EP formation from ATP was almost completely lost (actually within 1 min) (Fig. 3), showing the stable complex formation with BeF_x and AlF_x . No inhibition occurred with Mg²⁺ and fluoride without beryllium and aluminum $(E1Ca_2 + (MgF_x))$. Thus, MgF_x (MgF_4^{2-}) was not able to produce a complex with *E*1Ca₂, in contrast to $E2 \cdot MgF_4^{2-}$ formation from *E*2, the $E2 \cdot P_i$ product analog in *E*2P hydrolysis (19, 25). This finding agrees with the in-line phosphorylation of E1Ca₂ to E1PCa₂ (37), in which there is no state with non-covalently bound P_i.

The binding of Mg²⁺ at the catalytic site as a physiological cation is nevertheless required for EP formation. Actually in Fig. 3, Mg²⁺ was required for complex formation with BeF_x in 100 μ M Ca²⁺. The apparent Mg^{2+} affinity ($K_{0.5}$ of 5 mm, supplemental Fig. S1) was consistent with that of the catalytic site in phosphorylation from ATP or P_i (e.g. Refs. 11, 12, and 38-41). The BeF_x-induced inhibition also occurred with Mn²⁺ with apparent affinity ($K_{0.5}$ of 0.6 mM) significantly higher than that of Mg^{2+} , as found in E1PCa₂ formation and ATP hydrolysis with Mn^{2+} (40, 42).

When Ca^{2+} over millimolar amounts was added in place of Mg^{2+} , the *EP* formation was not inhibited by BeF_x (*BeF_x*(5 *mM Ca*)). Therefore, Ca^{2+} substitution probably at the catalytic Mg^{2+} site abolished complex formation with BeF_x . Although CaATP as a substrate and Ca^{2+} bound at the catalytic Mg^{2+} site are able to function for *E*1PCa₂ formation (26, 27), Ca^{2+} bound at the catalytic site likely produces different structure from the Mg^{2+}







FIGURE 3. Inhibition of EP formation by binding of BeF_x and AIF_x to E1Ca₂. The E1Ca₂ state ATPase of SR vesicles in 0.1 or 5 mM free Ca²⁺ was treated for 30 min at 25 °C with fluoride compounds in the presence of the indicated concentrations of the ligands. Subsequently, the samples were cooled to 0 °C and phosphorylated for 3 s by 3 $\mu M (\gamma^{-32}P)$ ATP. The amount of EP formed was shown as the percentage of the control value obtained without F⁻, Be²⁺, and Al³⁺. The EP formation was not inhibited when the incubations were performed with Be²⁺ or Al³⁺ but without F⁻. The error bars show the S. D. of five independent experiments.

bound structure (as in fact found, see below). The complex formation of $E1Ca_2$ by AIF_x in 100 μ M Ca^{2+} also required Mg^{2+} or Mn^{2+} with somewhat higher apparent affinities than those for BeF_x-induced complex formation, and was abolished by Ca^{2+} binding at the catalytic site.

 Ca^{2+} Binding and Occlusion at Transport Sites—In Fig. 4, Ca²⁺ binding and occlusion in formation of the Ca²⁺-ATPase complexes with fluoride compounds were determined in 15 mM Mg²⁺ with and without a 10-s EGTA filter washing. In all the cases without washing, Ca²⁺ was bound to the Ca²⁺-ATPase with high affinities; $K_{0.5}$ at sub-micromolar to millimolar ranges, Hill coefficient ~2, and maximum levels of 9–10 nmol/mg of protein, *i.e.* the stoichiometry of two Ca²⁺ per phosphorylation site (*inset* at 50 μ M Ca²⁺). Therefore, $E1Ca_2 \cdot BeF_x$ and $E1Ca_2 \cdot AIF_4^- \cdot ADP/E1Ca_2 \cdot AIF_x$ were produced by cooperative binding of two Ca²⁺ ions at high affinity transport sites. This finding agrees with the property of the sites for Ca²⁺ binding and the resulting enzyme activation for phospho-

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rylation by ATP as nicely demonstrated for the first time by Inesi *et al.* (14). Upon the EGTA washing of $E1Ca_2$ that is complexed with BeF_x, the two bound Ca²⁺ were not removed, and therefore occluded in the complex as " $E1Ca_2$ ·BeF_x." The two Ca²⁺ are occluded also in $E1Ca_2$ ·AlF₄⁻·ADP and less strongly in $E1Ca_2$ ·AlF_x.

Note that the Ca²⁺ affinity became 2–3-fold lower for BeF_x and AlF_x. This may be because the Ca²⁺-free *E*2 state produces $E2 \cdot BeF_3^-$ and $E2 \cdot AlF_4^-$ (25), and therefore competes with Ca²⁺ binding for formation of $E1Ca_2 \cdot BeF_x$ and $E1Ca_2 \cdot AlF_x$. On the other hand, the observed ~3-fold Ca²⁺-affinity increase in formation of $E1Ca_2 \cdot AlF_4^- \cdot ADP$ is probably brought about by the fact that ADP together with AlF_x strongly stabilizes the cross-linked N-P domains (17, 18), which is unfavorable for formation of the Ca²⁺-free *E*2 and $E2 \cdot AlF_4^-$, because for these structures the A domain should rotate into the opened space between the N and P domains and associate with them (19, 23, 24).

Cytoplasmic Structure in $E1Ca_2 \cdot BeF_x$ Is Intermediate between Those in $E1Ca_2 \cdot AlF_4^- \cdot ADP$ and $E2 \cdot BeF_3^-$ Proteolytic analysis was made to reveal the organization state of the cytoplasmic domains in the newly developed E1PCa₂·Mg analog $E1Ca_2 \cdot BeF_x$, and to compare with $E1Ca_2 \cdot AlF_4 \cdot ADP/$ $E1Ca_2 \cdot AlF_x$ and $E2 \cdot BeF_3^-$ ($E2P \cdot Mg$) (see the typical cleavage in supplemental Fig. S2). The initial rate of the "A1" appearance upon cleavage at the T2 site (Arg¹⁹⁸ on the Val²⁰⁰ loop of the A domain) in $E1Ca_2 \cdot BeF_x$ was substantially slower than the rapid cleavage of $E1Ca_2 \cdot AIF_4^- \cdot ADP$ and $E1Ca_2 \cdot AIF_x$ as well as $E1Ca_2$ (Table 1). The slowed T2 cleavage was also observed when $E1Ca_2 \cdot BeF_x$ was formed with 3 mM Mn^{2+} in place of 15 mM Mg^{2+} (data not shown). Also important was the slow but definitely occurring T2 cleavage in $E1Ca_2$ ·BeF_x, in sharp contrast to its complete resistance in $E2 \cdot BeF_3^-$. Therefore A-P domain organization at the Val²⁰⁰ loop in $E1Ca_2$ ·BeF_x is intermediate between those in $E1Ca_2 \cdot AlF_4 \cdot ADP/E1Ca_2 \cdot AlF_x$ and $E2 \cdot BeF_3^-$.

All complexes were almost completely resistant to proteinase K at the major site of Thr²⁴² on the A/M3-linker that produces the "p83" fragment. Therefore, in $E1Ca_2$ ·BeF_x, the A domain is rotated perpendicular to the membrane plane from its position in $E1Ca_2$ thereby causing the A/M3-linker strain, as in $E1Ca_2$ ·CaAMP-PCP, $E1Ca_2$ ·AlF₄·ADP (18, 19, 24), and $E1Ca_2$ ·AlF_x.

These analyses revealed that in the change $E1Ca_2 \cdot AIF_4^- \cdot ADP \rightarrow E1Ca_2 \cdot BeF_x$ (*i.e.* upon the ADP release from the transition state), the A domain moves, *i.e.* probably rotates to some extent parallel to the membrane plane likely due to the A/M3-linker strain, and thereby Arg¹⁹⁸ on the Val²⁰⁰ loop comes close to the P domain. In the subsequent change, $E1Ca_2 \cdot BeF_x \rightarrow E2 \cdot BeF_3^-$, the A domain rotates further (by the A/M3-linker strain as predicted to be motive force (18, 19, 43, 44)) and pro-



FIGURE 2. **Structure of SERCA1a and its change during processing of phosphorylated intermediate.** $E1Ca_2 \cdot AIF_4^- \cdot ADP$ (the transition state analog for phosphorylation $E1PCa_2 \cdot AIF_4^- \cdot ADP$ (the transition state analog for phosphorylation $E1PCa_2 \cdot ADP \cdot Mg^+$) and $E2 \cdot BeF_3^-$ (the ground state E2P analog (25)) were obtained from the Protein Data Bank (PDB accession code 1T5T (17) and 2ZBE (21), respectively). Cytoplasmic domains N (nucleotide binding), P (phosphorylation), and A (actuator), and 10 transmembrane helices (M1–M10) are indicated. The *arrows* on the domains, M1' and M2 (Tyr¹²²) in $E1Ca_2 \cdot AIF_4^- \cdot ADP$, indicate their approximate motions predicted for $E1PCa_2 \cdot ADP \cdot Mg^+ \rightarrow E2P \cdot Mg$. The phosphorylation site Asp³⁵¹, TGES¹⁸⁴ of the A domain, Arg¹⁹⁸ (tryptic T2 site) on the Val²⁰⁰ loop (DPR¹⁹⁸AV²⁰⁰NQD) of the A domain, and Thr²⁴² (proteinase K site) on the A/M3-linker are shown. Seven hydrophobic residues gather in the *E2P* state to form the Tyr¹²²-hydrophobic cluster (Y122-HC); Tyr¹²²/Leu¹¹⁹ on the top part of M2, IIa¹⁷⁹/Leu¹⁸⁰/IIe²³² of the A domain, and Val⁷⁰⁵/Val⁷²⁶ of the P domain. The overall structure of $E1Ca_2 \cdot AIF_4^- \cdot ADP$ is virtually the same as those of $E1Ca_2 \cdot Ca \cdot AMP - PCP$ and $E1PCa_2 \cdot Ca \cdot AMP - PN$ (17, 18, 22).



FIGURE 4. Ca^{2+} binding and occlusion at transport sites. The $E1Ca_2$ state ATPase of SR vesicles in various concentrations of ${}^{45}Ca^{2+}$ and 15 mM MgCl₂ was incubated with BeF_{xx} AIF_{xx} and AIF_x·ADP or without these compounds ($E1Ca_2$) for 30 min at 25 °C. The amounts of bound (*upper panel*) and occluded (*lower panel*) ${}^{45}Ca^{2+}$ were determined without and with the perfusion of the membrane filter with a 5 mM EGTA-containing washing solution (without CaCl₂ and fluoride compounds otherwise as the above incubation solution). The nonspecific Ca²⁺ binding was determined by including 10 μ M thapsigargin before the addition of fluoride compounds, and subtracted. When ADP was used for $E1Ca_2$ ·AIF₄⁻·ADP, 5 μ M A23187 was included to avoid Ca²⁺ accumulation in the vesicles by ATP produced from ADP due to adenylate kinase in the vesicles. In the *inset*, the stoichiometries of bound Ca²⁺ (*open bars*) and occluded Ca²⁺ (*closed bars*) to the phosphorylation site (P-site) were determined at saturating 50 μ M ${}^{45}Ca^{2+}$. Solid lines in the upper panel show the least squares fit to the Hill equation. $K_{0.5}$ of Ca²⁺ and Hill coefficients obtained were 1.3 μ M and 1.7 ($E1Ca_2$ ·AIF_x), and 0.4 μ M and 1.4 ($E1Ca_2$ ·AIF₄⁻·ADP). In the *lower panel*, the values for E1Ca₂·BeF_x, and E1Ca₂·AIF₄⁻·ADP are essentially not altered by EGTA washing (4.7 μ M and 1.9, and 0.4 μ M and 1.9, respectively).

duces its tight association with the P domain at the Val²⁰⁰ loop, mimicking $E1PCa_2 \cdot Mg \rightarrow E2P \cdot Mg + 2Ca^{2+}$.

 Ca^{2+} Ligation at the Catalytic Mg^{2+} Site—The proteolysis further revealed that $E1Ca_2 \cdot BeF_x$ was not produced from $E1Ca_2$ in 5 mM Ca²⁺ without Mg^{2+} ($E1Ca_2 + 5 \text{ mM } Ca^{2+} + BeF_x$ in Table 1), and that $E1Ca_2 \cdot BeF_x$ produced in 15 mM Mg^{2+} and 50–100 μ M Ca²⁺ was decomposed to the $E1Ca_2$ by 5 mM Ca²⁺ ($E1Ca_2 \cdot BeF_x + 5 \text{ mM } Ca^{2+}$), as shown by the rapid cleavage rates at the T2 and proteinase K sites. In E1PCa₂·Ca and E1Ca₂· CaAMP-PCP formed in 5 mM Ca²⁺ without Mg²⁺ (Table 1),³ the T2 site was also rapidly cleaved, in contrast to its substantially slowed cleavage in $E1Ca_2 \cdot BeF_r$ formed with Mg²⁺. Thus, for organization of the cytoplasmic domains at the T2 site (Arg¹⁹⁸), E1Ca₂·CaAMP-PCP and E1PCa₂·Ca are very similar to $E1Ca_2 \cdot AlF_4 \cdot ADP$, but differ from $E1Ca_2 \cdot BeF_x$. The close similarity between E1Ca2·CaAMP-PCP and $E1Ca_2 \cdot AlF_4 \cdot ADP$ is in agreement with their nearly same atomic structures and previous observations (17, 18, 45). Also notably, structure E1PCa₂·Ca·AMP-PN formed by CaAMP-PNP in 10 mM Ca²⁺ without Mg²⁺ (22) is almost identical with those of E1Ca₂·CaAMP-PCP and $E1Ca_2 \cdot AIF_4 \cdot ADP$ (see also Table 1).

In *E*1Ca₂·CaAMP-PCP and E1PCa₂·Ca·AMP-PN, the N-P domain cross-linked state is stabilized by Ca²⁺ bound at catalytic Mg²⁺ site I (Asp³⁵¹/Thr³⁵³/Asp⁷⁰³ and the phosphate) and by the nucleotide to be nearly identical to the state stabilized with AlF₄⁻ plus ADP in *E*1Ca₂·AlF₄⁻·ADP (17, 18, 22). The results on E1PCa2. Ca further indicated that such an N-P domain closed state is stabilized solely by site I Ca²⁺ ligation without the nucleotide. The stabilization of this state in E1PCa2. Ca is consistent with its markedly retarded isomerization to E2P (27), because isomerization requires the A domain rotation into the space between the N and P domains. In E1Ca₂·BeF_x formed with Mg^{2+} at the catalytic site (site I), such a Ca²⁺ ligation effect is obviously not present. Therefore, the N and P domains are probably more easily separated from each other, and the A domain

can rotate into the space between the N and P domains to some extent thus resulting in partial T2 resistance (but not yet as

aseme

³ The proteolytic analysis of *E*1PCa₂·Ca was made possible by its markedly retarded decay due to Ca²⁺ ligation at the catalytic site (27) and feedback inhibition by the high lumenal Ca²⁺. Analysis of *E*1PCa₂·Mg formed from MgATP was not feasible because of its very rapid turnover, thus of a very rapid ATP exhaustion.

TABLE 1

Summary of fluorescence changes and proteolysis rates

Maximal TNP-AMP fluorescence intensity at saturating 4 μ M TNP-AMP is given as % value of that of $E2\cdot\text{BeF}_3^-$ without A23187. Tryptophan fluorescence change upon complex formation with the ligand from $E1\text{Ca}_2$ (or from other state when indicated in parentheses) is shown as % value of the intensity of $E1\text{Ca}_2$ (see also supplemental Fig. S3*E*). The cleavage rate at the T2 site (Arg¹⁹⁸) with trypsin and the digestion rate of the 110-kDa ATPase chain with proteinase K were obtained by the detailed time course analysis in the initial 1 (*T2*) and 30 min (*proteinase K*), and shown as % values of those determined with $E1\text{Ca}_2$ in 0.1 mM Ca²⁺. Some of these experiments were done at 0.05 mM Ca²⁺ instead of 0.1 mM, but the results were virtually the same and therefore are represented with 0.1 mM Ca²⁺ for simplicity. It should also be mentioned that the digestion rates in $E1\text{Ca}_2$ were not altered by 5 or 0.7 mM Ca²⁺ or by A23187 (being 97–101% of the rates of $E1\text{Ca}_2$ in 0.1 mM Ca²⁺ without A23187), and that the ligand-free E2 state was also rapidly digested by trypsin and proteinase K, and TG binding to E2 and A23187 did not alter essentially the rapid cleavage rates (Refs. 23–25). The other Ca²⁺-ATPase complexes were produced under the same buffer conditions as those for $E1\text{Ca}_2\cdot\text{BEF}_x$ formation, otherwise as follows and noted below: $E1\text{Ca}_2\cdot\text{MgAMP-PCP}$ by 5 mM MgAMP-PCP; $E11\text{Ca}_2\cdot\text{Ca}\cdot\text{AMP-PCP}$ by 5 mM CaAMP-PCP by 5 mM CaAMP-PCP by 5 mM CaAMP-PCP by 5 mM CaAMP-PCP by 5 mM Ca²⁺ and 100 μ M Ca²⁺ and then incubated with the subsequently added 5 mM Ca²⁺ for 3 h; $E1\text{Ca}_2 + 5 \text{ mM} \text{Ca}^{2+}$, the $E1\text{Ca}_2$ state ATPase was incubated with BEF_x for 10 min in the presence of 5 mM Ca²⁺ without Mg²⁺.

ATPase state $(\rightarrow \text{consequent state, if altered})$	Ca ²⁺ (mM)/Mg ²⁺ (mM)	Relative TNP-AMP fluorescence intensity, see Fig. 5 <i>A</i> and supplemental Fig. S5	Change in tryptophan fluorescence from <i>E</i> 1Ca ₂ (or from another state), supplemental Fig. S3 ↑ increase, ↓ decrease	Relative digestion rate	
				Trypsin (T2), supplemental Fig. S2	Proteinase K, supplemental Fig. S2
	тм	%	% of E1Ca2 level	%	
E1Ca ₂	0.1/15	7	$(3.27 \uparrow \text{from } E2)$	100	100
E1Ca, MgAMP-PCP	0.1/15		0	64	25
E1Ca ₂ ·CaAMP-PCP	5/0		0.80 ↑	61	9
$E1Ca_{2} \cdot AlF_{4} \cdot ADP$	0.1/15		0.84 🕆	65	3
E1PCa ₂ ·Ca·AMP-PN	5/0		0.77 🛉	65	4
$E1Ca_2 \cdot AlF_x$	0.1/15	7	0	70	10
E1PCa ₂ ·Ca	5/0		0.89 ↓	80	6
E1PCa ₂ ·Mg	0.1/15		1.18 ↓		
$E1Ca_2 \cdot BeF_x$	0.1/15	75	1.27 ↓	35	13
$E1Ca_2 \cdot BeF_x + 5 \text{ mM } Ca^{2+} (\rightarrow \text{ partially } E1Ca_2)$	$0.1/15 + 5 \text{ mm Ca}^{2+}$	30		70	32
$E1Ca_2 + 5 \text{ mM} Ca^{2+} + BeF_x$	5/0	16	0	116	90
$E2 \cdot BeF_3^-$	0/15	100	$(0.66 \uparrow \text{from } E2)$	0	0
$E1Ca_2 \cdot BeF_x + TG (\rightarrow E2 \cdot BeF_3^-(TG))$	0.1/15	64	$(5.35 \downarrow \text{from } E1\text{Ca}_2 \cdot \text{BeF}_x)$	0	0
$E2 \cdot BeF_3^- + TG (\rightarrow E2 \cdot BeF_3^-(TG))$	0/15	65	$(4.62 \downarrow \text{from } E2 \cdot \text{BeF}_3^-)$	0	0
$E1Ca_2 \cdot BeF_x + A23187 (\rightarrow E2 \cdot BeF_3^-)$	0.1/15	100		10	0
$E2 \cdot BeF_{3}^{-} + A23187$	0/15	100		0	0
$E1PCa_2 \cdot Mg (0.7 \text{ mM } Ca^{2+})$	0.7/15		1.19 ↓		
$E1Ca_2 \cdot BeF_x (0.7 \text{ mm } Ca^{2+})$	0.7/15	75	1.27 ↓	39	15
$E1Ca_2 \cdot BeF_x$ (0.7 mm Ca^{2+} + A23187)	0.7/15	75		32	8
$E2 \cdot BeF_3^- + 0.7 \text{ mm Ca}^{2+}$	$0/15 \pm 0.7 \text{ mm Ca}^{2+}$	100	$(0 \text{ from } E2 \cdot BeF_3^{-})$	0	2
$E2 \cdot BeF_3^- + 0.7 \text{ mm Ca}^{2+} + A23187$	$0/15 \pm 0.7 \text{ mm } \text{Ca}^{2+}$	75		38	16
$(\rightarrow E1Ca_2 \cdot BeF_3^-)$					

completely as in E2·BeF₃⁻). As the cause of the Ca²⁺-induced E1Ca₂·BeF_x to E1Ca₂ decomposition, Ca²⁺ replacement of Mg²⁺ at site I altered the domain organization state and made the BeF_x ligation unfavorable (see "Discussion").

TNP-AMP Superfluorescence—TNP-AMP binds to the nucleotide binding site with an extremely high affinity (46–48), and in the *E*2P ground state and its analog *E*2·BeF₃⁻, the bound TNP-AMP develops its extremely high fluorescence "superfluorescence" (25), which reflects a strongly hydrophobic atmosphere around Asp³⁵¹ (49, 50). On the other hand, it has been controversial whether *E*1PCa₂ develops TNP-AMP superfluorescence, mostly because its tight binding to the nucleotide binding site prevents phosphorylation to form *E*1PCa₂·Mg, so the TNP-AMP·*E*1PCa₂·Mg complex is not formed in significant amounts. Nakamoto and Inesi (47), nevertheless, predicted the development of superfluorescence in *E*1PCa₂.

Here, with $E1Ca_2 \cdot BeF_x$ as the $E1PCa_2 \cdot Mg$ analog, we examined the superfluorescence without ATP. In Fig. 5, *A* and *B*, we first formed $E1Ca_2 \cdot BeF_x$ in 15 mM Mg²⁺ and 50 μ M Ca²⁺, then TNP-AMP was added to give a saturating level of 4 μ M. $E1Ca_2 \cdot BeF_x$ rapidly developed superfluorescence, and then the fluorescence decreased slowly (much more slowly and extensively at 4 °C).⁴ The proteolysis after the loss of superfluores-

cence revealed that $E1Ca_2 \cdot BeF_x$ was decomposed to $E1Ca_2$ (data not shown). Thus, $E1Ca_2 \cdot BeF_x$ develops superfluorescence, and the TNP-AMP binding *per se* causes its decomposition to $E1Ca_2$, in sharp contrast to the completely stable $E2 \cdot BeF_3^-$ even after TNP-AMP binding. The maximum superfluorescence level of $E1Ca_2 \cdot BeF_x$ was slightly lower than that of $E2 \cdot BeF_3^-$ (Fig. 5, *A*-*C*), which is the same as that of $E2P \cdot Mg$ formed from P_i (25). The results clearly revealed that the atmosphere around Asp³⁵¹ in $E1Ca_2 \cdot BeF_x$ is strongly hydrophobic, similar to $E2 \cdot BeF_3^-$, although the cytoplasmic domain organization in $E1Ca_2 \cdot BeF_x$ distinctly differs from and did not yet reach the most compactly organized state in $E2 \cdot BeF_3^-$.

 $E1Ca_2 \cdot AlF_x$ as well as E2 and $E1Ca_2$ did not develop superfluorescence despite high affinity TNP-AMP binding. Therefore the catalytic site in $E1Ca_2 \cdot AlF_x$ is hydrophilic and differs critically from the strongly hydrophobic site in $E1Ca_2 \cdot BeF_x$. Note also that $E2 \cdot AlF_4^-$ ($E2 \cdot P^{\dagger}$) and $E2 \cdot MgF_4^{2-}$ ($E2 \cdot P_i$) do not develop TNP-AMP superfluorescence (25). The superfluorescence therefore develops solely with $Ca^{2+} \cdot ATP$ ase complexed with BeF_x ; $E1Ca_2 \cdot BeF_x$ and $E2 \cdot BeF_3^-$.

The superfluorescence development of $E1Ca_2 \cdot BeF_x$ in the presence of 15 mM Mg²⁺ and 50 μ M Ca²⁺ was rapidly diminished with increasing Ca²⁺ over millimolar concentrations of



⁴ The fluorescence level after the decrease of the transient superfluorescence of E1Ca₂:BeF_x was somewhat higher than the non-superfluorescent low level of TNP-AMP bound to E1Ca₂ especially at 25 °C. We obtained the results indicating that a small fraction of E2·BeF₃⁻ was produced even in the

presence of 50 μ M Ca²⁺ (more at 25 than at 4 °C) after the TNP-AMP-induced *E*1Ca₂·BeF_x decomposition to *E*1Ca₂, and therefore remained somewhat superfluorescence (data not shown).



FIGURE 5. **TNP-AMP superfluorescence of E1Ca_2 \cdot BeF_x and E2 \cdot BeF_3.** $E1Ca_2 \cdot BeF_x$ and $E1Ca_2 \cdot AIF_x$ were produced in 50 μ m Ca²⁺ and 15 mm MgCl₂, then at 25 (A) and 4 °C (B), a small volume of TNP-AMP was added to give saturating 4 μ m, and the TNP-AMP fluorescence was followed. The fluorescence of $E2 \cdot BeF_3$ produced with BeF_x in the absence of Ca^{2+} , the $E1Ca_2$ and E2 states without the fluoride compounds, and without SR vesicles (*no SRV*) were also followed. *C*, the TNP-AMP fluorescence intensities were measured at various concentrations of TNP-AMP at 25 °C, otherwise as described for *A*. For $E1Ca_2 \cdot BeF_x$, the maximum level of transient superfluorescence was determined by extrapolating its decrease to the time of TNP-AMP addition. In the *lower panel* in *C*, the low fluorescence was replotted on the expanded scale. The maximum fluorescence intensities at saturating 4 μ m TNP-AMP were obtained by subtracting the background level without TNP-AMP and the level of 4 μ m TNP-AMP without the SR vesicles, and given as the relative values in Table 1.

 Ca^{2+} (see Fig. 8 and Table 1). Also, inclusion of 5 mM Ca^{2+} without Mg^{2+} in the $E1Ca_2 \cdot BeF_x$ formation mixture abolished the superfluorescence (Table 1). The results agree with the above findings that $E1Ca_2 \cdot BeF_x$ is not produced from and decomposed to $E1Ca_2$ by Ca^{2+} ligation at the catalytic Mg^{2+} site (site I).

Transmembrane Domain Structure—The 12 tryptophan residues among 13 in the Ca²⁺-ATPase are located at the transmembrane region. The tryptophan fluorescence changes in fact reflect the transmembrane domain structural changes, *i.e.* rearrangements of the transmembrane helices upon Ca²⁺ binding to the high affinity transport sites and during the transport

cycle (28, 51, 52) as found originally by Dupont and Leigh (53). As summarized in Table 1 with typical fluorescence traces in supplemental Fig. S3, the fluorescence changes were determined at 4 °C upon formation of the E1PCa₂ analogs by the addition of fluoride compounds to $E1Ca_2$ in 15 mM Mg^{2+} and 100 μ M Ca^{2+} . $E1Ca_2 \cdot BeF_x$ formation decreased fluorescence by 1.3% very similar to the decrease in $E1PCa_2$ ·Mg formation from $E1Ca_2$ by MgATP, *i.e.* in $E1Ca_2$ ·MgATP \rightarrow $E1PCa_2 \cdot Mg$ (52). In contrast, $E1Ca_2 \cdot Mg$ AlF_x formation did not cause any change. The E1Ca₂·AlF₄·ADP formation increased the fluorescence by 0.8%. (F^- alone and ADP alone did not cause any change, except the dilution (F⁻) and absorption of excitation light (ADP).) Thus the transmembrane structure of $E1Ca_2 \cdot BeF_x$ mimics that of $E1PCa_2 \cdot$ Mg, but those of $E1Ca_2 \cdot AIF_4 \cdot ADP$ and $E1Ca_2 \cdot AIF_r$ differ substantially although the Ca²⁺ ions are occluded at the transport sites (or less strongly in $E1Ca_2 \cdot AlF_x$, Fig. 4). This observation is consistent with the finding in proteolysis and TNP-AMP superfluorescence that organization of the cytoplasmic domains and structure at the catalytic site in $E1Ca_2 \cdot BeF_x$ substantially differ from those in $E1Ca_2 \cdot AlF_4 \cdot ADP$ and $E1Ca_2 \cdot AlF_{x}$. It is concluded that the transmembrane structure with the occluded Ca^{2+} adopts not simply one state, but changes with the change in the cytoplasmic region during phosphoryl transfer and ADP release (see the diagram of tryptophan fluorescence change in supplemental Fig. S3E (with Ref. 54)).

Upon formation of E1Ca₂·

CaAMP-PCP and $E1PCa_2 \cdot Ca \cdot AMP-PN$ by CaAMP-PCP and CaAMP-PNP, respectively, the fluorescence increased by 0.8% equal to that upon $E1Ca_2 \cdot AIF_4^- \cdot ADP$ formation (Table 1), in agreement with their essentially identical structures with occluded Ca²⁺ (17, 18, 22, 45). By contrast, the fluorescence did not change upon formation of the $E1Ca_2 \cdot MgAMP$ -PCP, which is the Ca²⁺-unoccluded state (28, 45), and in rapid equilibrium with $E1Ca_2$.

Upon the exclusive accumulation of $E1PCa_2 \cdot Ca$ by CaATP without Mg²⁺, tryptophan fluorescence decreased by 0.9%, being slightly less than that by formation of $E1PCa_2 \cdot Mg$ and $E1Ca_2 \cdot BeF_x$ (Table 1). Thus in the overall structure,





FIGURE 6. **ADP causes the loss of Ca²⁺ occlusion in E1Ca₂·BeF_x.** E1Ca₂·BeF_x or *E*1Ca₂ were produced in 50 μ M⁴⁵Ca²⁺ and 15 mM Mg²⁺ as described in the legend to Fig. 4, then subjected to membrane filtration without and with perfusion for 20 s by the same buffer containing 50 μ M non-radioactive Ca²⁺ (in place of ⁴⁵Ca²⁺) with and without 1 mM ADP. The amounts of ⁴⁵Ca²⁺ specifically bound and occluded in the Ca²⁺-ATPase were determined by subtracting the nonspecific background level (2.03 \pm 0.04 nmol/mg) obtained with the above perfusion without ADP of the *E*1Ca₂ state. The error bars show the S. D. of three independent experiments.

*E*1PCa₂·Ca may be between $E1Ca_2 \cdot CaAMP$ -PCP and $E1PCa_2 \cdot Mg$ ($E1Ca_2 \cdot BeF_x$), and closer to the latter state. Although Ca^{2+} ligation at catalytic Mg^{2+} site I in $E1PCa_2 \cdot Ca$ favors the N-P domain closed state, similar to $E1Ca_2 \cdot CaAMP$ -PCP, the absence of the N-P domain cross-linking nucleotide in $E1PCa_2 \cdot Ca$ likely altered the overall structure slightly.

Upon formation of $E2 \cdot BeF_3^-$ from E2 by BeF_x and Mg^{2+} without Ca^{2+} , the fluorescence increased by 0.7%, mimicking the change upon $E2P \cdot Mg$ formation from E2 with P_i and Mg^{2+} , and reflecting the opening of the lumenal gate from the closed state (25). As a consequence, the fluorescence of $E1Ca_2 \cdot BeF_x$ was definitely higher by $\sim 1.3\%$ than that of $E2 \cdot BeF_3^-$, showing their distinct difference in the transmembrane structure. In agreement, the previous kinetic analysis have shown (28) that tryptophan fluorescence decreases by $\sim 1\%$ in the isomerization/ Ca^{2+} release, $E1PCa_2 \cdot Mg \rightarrow E2P \cdot Mg + 2Ca^{2+}$, reflecting the transmembrane structural change from the Ca^{2+} -occluded state to the Ca^{2+} -released and lumenally opened state.

Upon the addition of thapsigargin (TG) to $E1Ca_2 \cdot BeF_x$ and $E2 \cdot BeF_3^-$, tryptophan fluorescence decreased rapidly by 5.4 and 4.6%, respectively, and reached the level of $E2 \cdot BeF_3^-$ with bound TG ($E2 \cdot BeF_3^-$ (TG), see Table 1). TNP-AMP superfluorescence (supplemental Fig. S5, *A* and *B*) and proteolysis (Table 1) also demonstrated that $E1Ca_2 \cdot BeF_x$ was converted by TG to $E2 \cdot BeF_3^-$ (TG). Importantly, as described under supplemental Fig. S4, two Ca²⁺ occluded in $E1Ca_2 \cdot BeF_x$ are most likely released into the lumen by the TG-induced structural perturbation and trapped in the lumen by the bound TG, as TG fixes the lumenal gate in the closed state and suppresses Ca²⁺ leakage (16, 55).

 $E1Ca_3$ ·BeF_x Is ADP-sensitive—In Fig. 6, two ⁴⁵Ca²⁺ occluded in $E1Ca_2$ ·BeF_x were rapidly removed by washing with 1 mM ADP, whereas the occluded ⁴⁵Ca²⁺ remained completely without ADP. Thus ADP caused the loss of Ca²⁺ occlusion. In agreement, ADP binding to $E1Ca_2$ ·BeF_x increased tryptophan



FIGURE 7. **Conversion of E1Ca₂·BeF_x to E2·BeF₃⁻ with Ca²⁺ release.** The complex E1Ca₂·BeF_x was produced at 25 °C in 50 μ M ⁴⁵Ca²⁺ and 15 mM MgCl₂ as described in the legend to Fig. 4. The complex was then further incubated in the presence of these ligands at 25 and 4 °C for various periods, and the bound ⁴⁵Ca²⁺ was determined with and without EGTA washing as described in the legend to Fig. 4. In the control, the bound ⁴⁵Ca²⁺ in E1Ca₂ without BeF_x was determined.

fluorescence to the $E1Ca_2$ level, and resulted in the tryptic T2 site cleavage as $E1Ca_2$ with bound ADP (data not shown). By contrast, ADP binding to $E2 \cdot BeF_3^-$ did not alter its structure (data not shown). The ADP-induced decomposition of $E1Ca_2 \cdot BeF_x$ to $E1Ca_2$ was also demonstrated with the ADP-induced loss of TNP-AMP superfluorescence, in contrast to normal superfluorescence development in $E2 \cdot BeF_3^-$ after ADP incubation (data not shown). Thus $E1Ca_2 \cdot BeF_x$ is ADP-sensitive as $E1PCa_2 \cdot Mg$, and $E2 \cdot BeF_3^-$ is ADP-insensitive as $E2P \cdot Mg$.

Conversion of $E1Ca_2$ ·BeF_x to E2·BeF⁻₃ at 50 μ M Ca²⁺—In Fig. 7, $E1Ca_2 \cdot BeF_x$ was first formed in SR vesicles with BeF_x at 25 °C in 50 μ M Ca²⁺ and 15 mM Mg²⁺, then further incubated at 25 and 4 °C in the presence of these ligands. The amount of bound and occluded Ca²⁺ was lost slowly ($t_{1/2} = -2$ h at 25 °C and -7h at 4 °C). TNP-AMP superfluorescence (Fig. 8) and tryptic and proteinase K proteolyses (data not shown) revealed that $E1Ca_2 \cdot BeF_x$ turned to $E2 \cdot BeF_3^-$ with Ca^{2+} loss. Thus $E1Ca_2 \cdot BeF_3^ BeF_x$ proceeded its spontaneous slow conversion to $E2 \cdot BeF_3^-$, as the autoisomerization of $E1PCa_2$ ·Mg to E2P·Mg. The Ca²⁺ ions released into the lumen may leak out during such long periods. In $E1Ca_2 \cdot AIF_x$ and $E1Ca_2 \cdot AIF_4 \cdot ADP$, the amount of bound (occluded) Ca²⁺ was not decreased during the above 10-h incubation at 25 °C (data not shown). The proteolysis showed that these complexes were not converted to the Ca²⁺-released forms, $E2 \cdot AlF_4^-$ ($E2 \cdot AlF_x$) with and without ADP (data not shown). The results indicate that the product state $E1PCa_2$ ·Mg in the phosphoryl transfer acquires the structure ready for autoisomerization to E2P·Mg releasing Ca²⁺, whereas the transition state structure is not yet fully prepared for autoisomerization to the Ca^{2+} -released E2P form. Interestingly, as described in supplemental Figs. S4 and S5 (with Refs. 56 and 57), the conversion $E1Ca_2 \cdot BeF_x \rightarrow E2 \cdot BeF_3^-$ was markedly accelerated by the transmembrane structural perturbation with hydrophobic reagents such as A23187, lasalocid, and $C_{12}E_8$, as





FIGURE 8. **Stability of E1Ca₂:BeF_x in various Ca²⁺ concentrations.** *A*, *E*1Ca₂:BeF_x was first produced in SR vesicles in 0.1 mM Ca²⁺ and 15 mM MgCl₂ with BeF_x. Subsequently, Ca²⁺ was changed to 0.1 (unchanged), 0.4, 0.7, 1.1, 3.1, and 10.1 mM, and the incubations continued at 25 °C for 72 h. At the indicated periods, the superfluorescence with 4 μ M TNP-AMP was examined at 25 °C, and the maximum levels obtained as described in the legend to Fig. 6 are shown. *E2*:BeF₃ formed without Ca²⁺, *E*1Ca₂ in 0.1–10 mM Ca²⁺, and *E2* without Ca²⁺ were also incubated. *B*, in the presence of 0.7 mM Ca²⁺, *E*1Ca₂ in SR vesicles was first incubated with and without 1.2 μ M A23187 (*A23*), then BeF_x was added (*E1Ca₂:BeF_x*). *E*1Ca₂:BeF_x (*DOC-E1Ca₂:BeF_x*) was also produced from *E*1Ca₂ in 0.7 mM Ca²⁺ of the Ca²⁺-ATPase purified and delipidated from SR vesicles by deoxycholate (*DOC*) treatment (30). The incubation was continued for 12 days, otherwise as indicated in *A*. *E2*:BeF₃ (*E2:BeF₃*, *DOC-E2:BeF₃*) without Ca²⁺ and *E*1Ca₂ in 0.7 mM Ca²⁺ (*E1Ca₂*, *DOC-E1Ca₂*) and *E2* without Ca²⁺ (*E2, DOC-E2:BeF₃*) without Ca²⁺ both with and without A23187, the development of the representative samples with incubation periods and Ca²⁺ concentration (mM). Note that at 0.7 mM Ca²⁺, both with and without A23187, the development of the *E*1Ca₂:BeF_x characteristic transient superfluorescence remained periods. By contrast, the transient superfluorescence was converted to the stable and higher superfluorescence characteristic of *E2*:BeF₃⁻ at 0.1 and 0.4 mM Ca²⁺, and it was markedly reduced by 3 and 10 mM Ca²⁺ due to decomposition to *E*1Ca₂.

well as TG. In contrast, $E1Ca_2 \cdot AIF_x$ and $E1Ca_2 \cdot AIF_4^- \cdot ADP$ were resistant against these reagents.

 $E1Ca_2$ ·BeF_x Is Perfectly Stabilized at 0.7 mM Ca²⁺—As found here, Ca^{2+} binding at high affinity transport sites in *E*1Ca₂ is obligatorily required for $E1Ca_2 \cdot BeF_x$ formation, whereas millimolar high Ca^{2+} (Ca^{2+} ligation at the catalytic Mg^{2+} site I) decomposes this complex to $E1Ca_2$. Furthermore, $E1Ca_2$ ·BeF_x at 50 μ M Ca²⁺ is spontaneously and slowly converted to E2·BeF_x releasing Ca²⁺, and the conversion is markedly accelerated by transmembrane perturbation with hydrophobic reagents such as C12E8 and A23187 (see supplemental materials). The results showed that $E1Ca_2 \cdot BeF_x$ as the $E1PCa_2 \cdot Mg$ analog possesses the structure prepared for its isomerization to $E2 \cdot BeF_3^-$ with Ca^{2+} release as $E1PCa_2 \cdot Mg \rightarrow E2P \cdot Mg + 2Ca^{2+}$. On the other hand, it is essential for crystallographic studies to find conditions to perfectly stabilize the $E1Ca_2 \cdot BeF_x$ complex. In Fig. 8A, $E1Ca_2$ ·BeF_x was first formed in 0.1 mM Ca²⁺ and 15 mM Mg²⁺, then further incubated with various concentrations of Ca²⁺ with and without A23187. The structural state was monitored by TNP-AMP superfluorescence (Fig. 8), proteolysis, and tryptophan fluorescence (see Table 1 for representative data). Then we successfully found that Ca^{2+} at a very narrow concentration range, 0.7 mM, perfectly stabilizes $E1Ca_2 \cdot BeF_x$ and maintains this complex for at least 12 days at 25 °C (and 4 °C) even in the presence of A23187. The ${}^{45}Ca^{2+}$ binding measurements on $E1Ca_2 \cdot BeF_x$ in 0.7 mM ${}^{45}Ca^{2+}$ demonstrated that two Ca^{2+} ions are bound and occluded in this complex (Fig. 9A).

The perfectly stable $E1Ca_2 \cdot BeF_x$ was produced from $E1Ca_2$ even in the presence of A23187 if 0.7 mM Ca²⁺ was included before BeF_x addition (Fig. 8, *B* and *F*, and Table 1). Also, $E1Ca_2 \cdot BeF_x$ was successfully produced with the Ca²⁺-ATPase purified from SR vesicles by delipidation with deoxycholate (30); in this case again, by including 0.7 mM Ca²⁺ before BeF_x addition. $E1Ca_2 \cdot BeF_x$ thus produced with the purified and delipidated Ca²⁺-ATPase was perfectly stable at least for 12 days at 4 and 25 °C in 0.7 mM Ca²⁺ (Fig. 8, *B* and *G*, at 25 °C).

 $E1Ca_2 \cdot BeF_3^-$ Is Produced from $E2 \cdot BeF_3^-$ by Lumenal Ca^{2+} Binding—We successfully found also that $E1Ca_2 \cdot BeF_x$ $(E1Ca_2 \cdot BeF_3^-)$ can be produced from $E2 \cdot BeF_3^-$ by lumenal Ca^{2+} binding, as mimicking the lumenal Ca^{2+} -induced reverse transition, $E2P \cdot Mg + 2Ca^{2+} \rightarrow E1PCa_2 \cdot Mg$. In Fig. 10, we added





FIGURE 9. Two Ca²⁺ are bound in E1Ca₂·BeF_x formed from E1Ca₂ and from **E2·BeF**₃ at 0.7 mM Ca²⁺. A, SR vesicles in 0.7 mM ⁴⁵Ca²⁺ were incubated with BeF_x in the absence or presence of 5 μ M A23187, otherwise as described in the legend to Fig. 4. The amount of bound Ca²⁺ was obtained with subtraction of the background level (10.1 \pm 0.5 nmol/mg (n = 6)) determined by EGTA washing the vesicles incubated without BeF_x and A23187. The occluded Ca²⁻ was determined by EGTA washing in the absence of A23187 and by subtraction of the background level. (Here the determination of occluded Ca²⁻ in A23187 by EGTA washing was not feasible, because in the absence of (or even in 0.1 mm) $(Ca^{2+}, A23187 converts E1Ca_2BeF_x very rapidly to E2·BeF_3^- releasing Ca^{2+} (supplemental Figs. 4 and 5). B, E2·BeF_3^- was first produced in SR vesicles without A23187 and Ca^{2+}. Subsequently, the samples were diluted 10 times$ with the buffer containing 45 CaCl₂ and BeF_x with and without 5 μ M A23187, to give 0.7 mm $^{45}Ca^{2+}$ and the same buffer conditions as in A. At 15 s after dilution, the amount of bound $^{45}Ca^{2+}$ was determined without EGTA washing and by subtracting the nonspecific Ca^{2+} binding (1.0 ± 0.1 nmol/mg (n =6)) determined by EGTA washing the sample incubated without BeF_x and A23187.

various concentrations of Ca^{2+} to $E2 \cdot BeF_3^-$ formed in SR vesicles in 15 mM Mg^{2+} without Ca^{2+} in the presence and absence of A23187, then at 10 s after Ca^{2+} addition the structural state was examined by TNP-AMP superfluorescence. With increasing Ca²⁺ to 0.7 mM in the presence of A23187, the stable superfluorescence of $E2 \cdot BeF_3^-$ was converted to the transient and slightly lower superfluorescence characteristic of $E1Ca_2$ ·BeF_x with $K_{0.5}$ of 0.4 mM Ca²⁺ and a Hill coefficient of 4 (Fig. 10, A and D). A further Ca^{2+} increase in the millimolar range caused the marked loss of superfluorescence with $K_{0.5}$ of 1.7 mM and a Hill coefficient of 1 (Fig. 10, B and D). The proteolysis also clearly showed that $E2 \cdot BeF_3^-$ was converted to $E1Ca_2 \cdot BeF_x$ by $0.7 \text{ mM} \text{ Ca}^{2+}$ in A23187 (Table 1), and this complex was further decomposed to $E1Ca_2$ by 10 mM Ca^{2+} (data not shown). In Fig. 9*B*, two ${}^{45}\text{Ca}^{2+}$ were shown to be bound producing *E*1Ca₂·BeF_x, when 0.7 mM ${}^{45}Ca^{2+}$ was added to $E2 \cdot BeF_3^-$ in the presence of A23187. In contrast, in the absence of A23187, $E2 \cdot BeF_3^-$ was neither converted to $E1Ca_2 \cdot BeF_x$ nor decomposed to $E1Ca_2$ even at 10 mM Ca²⁺ (Figs. 9B and 10, C and E, and Table 1 (proteolysis at 0.7 mM Ca^{2+})).

The results demonstrated that $E1Ca_2 \cdot BeF_x$, most probably $E1Ca_2 \cdot BeF_3^-$, was produced from $E2 \cdot BeF_3^-$ by the lumenal Ca^{2+} binding at the lumenally oriented low affinity transport sites, and further that Ca^{2+} substitution of Mg^{2+} at the catalytic site in $E1Ca_2 \cdot BeF_3^-$ produced from $E2 \cdot BeF_3^-$ caused its decomposition to $E1Ca_2$, therefore as the change $E2 \cdot BeF_3^- + 2Ca^{2+} \rightarrow E1Ca_2 \cdot BeF_3^- \rightarrow E1Ca_2$. Note that Mg^{2+} bound at the catalytic site in $E2P \cdot Mg$ is occluded, whereas it is not and therefore is

Structural Analog of E1PCa₂·Mg Intermediate of Ca²⁺-ATPase

exchangeable in $E1PCa_2 \cdot Mg$ (42). Thus, these two distinctly different states of the ligated Mg^{2+} at the catalytic site (site I) in $E2P \cdot Mg$ and $E1PCa_2 \cdot Mg$ are obviously mimicked here by the respective analogs $E2 \cdot BeF_3^-$ and $E1Ca_2 \cdot BeF_3^-$. The perfect stabilization of $E1Ca_2 \cdot BeF_3^-$ achieved by 0.7 mM Ca^{2+} (Fig. 10) obviously involves lumenal Ca^{2+} binding and prevention of the Ca^{2+} release into the lumen. The stabilization by 0.7 mM Ca^{2+} in the absence of A23187 is probably due to Ca^{2+} moved passively into the vesicle lumen during the long incubation periods. All these findings show that the forward and reverse transition, $E1PCa_2 \cdot Mg \leftrightarrow E2P \cdot Mg + 2Ca^{2+}$, is mimicked by the forward and reverse conversion, $E1Ca_2 \cdot BeF_3^- \leftrightarrow E2 \cdot BeF_3^- + 2Ca^{2+}$.

It is of interest to note the Hill coefficient of 4 in the lumenal Ca^{2+} -induced reverse conversion, $E2 \cdot BeF_3^- + 2Ca^{2+} \rightarrow E1Ca_2 \cdot BeF_3^-$ at 0.1–0.7 mM Ca^{2+} in Fig. 10*A*. This might be indicative of the existence of lumenal Ca^{2+} access sites in addition to transport sites and their possible cooperative involvement in lumenal Ca^{2+} access to the transport sites. In fact, two such sites besides the two transport sites have been suggested by the kinetics and protein-chemical study on the lumenal loops (58, 59).

DISCUSSION

Formation of $E1Ca_2 \cdot BeF_3^-$ —As a structural analog of the physiological intermediate $E1PCa_2 \cdot Mg$, the $E1Ca_2 \cdot BeF_x$ complex was successfully produced by BeF_x binding to the $E1Ca_2$ state Ca^{2+} -ATPase and from $E2 \cdot BeF_3^-$ by lumenal Ca^{2+} binding to the lumenally oriented low affinity transport sites. All the revealed properties of $E1Ca_2 \cdot BeF_x$ met the requirements for the $E1PCa_2 \cdot Mg$ analog; *i.e.* two Ca^{2+} occluded at the transport sites, Mg^{2+} bound (but not occluded) at the catalytic site, the ADP-released but still ADP-sensitive state, and its isomerization to the ADP-insensitive Ca^{2+} -released state $E2P \cdot Mg$ ($E2 \cdot BeF_3^-$) and reversal by lumenal Ca^{2+} binding, $E1PCa_2 \cdot Mg \leftrightarrow E2P \cdot Mg + 2Ca^{2+}$.

Furthermore, the coordination chemistry of beryllium fluoride, actually BeF_3^- , fulfills the requirement of $E1\mathrm{Ca}_2\cdot\mathrm{BeF}_x$ as the E1PCa₂·Mg analog. In chemistry, beryllium fluoride compounds are known to adopt tetrahedral geometry with the Be-F 1.55-Å bond length, thereby making them strictly isomorphous to the tetrahedral phosphate group (60). Moreover, because of the high charge density due to the small size, beryllium is able to coordinate the aspartate-oxygen in addition to F⁻. The -O- BeF_3^- thus produced with Asp³⁵¹-oxygen in fact possesses the tetrahedral geometry superimposable with the covalently bound phosphate at the aspartate, as actually seen in $E2 \cdot BeF_3^-$, the *E*2P·Mg ground-state analog (21, 22, 25). MgF_4^{2-} also possesses the tetrahedral geometry, but magnesium is not able to be coordinated directly with the Asp³⁵¹-oxygen, as seen in $E2 \cdot MgF_4^{2-}$, the $E2 \cdot P_i$ analog. AlF_4^{-} in $E1Ca_2 \cdot AlF_4^{-} \cdot ADP$ and E2·AlF₄ (17, 18, 20) (or AlF₃ in some cases in the haloacid dehalogenase superfamily (61)) possesses planar geometry, in which Asp^{351} -oxygen and ADP β -phosphate or the hydrolytic water $(E2 \cdot AlF_4^-)$ coordinate the aluminum at apical positions producing the bipyramidal structure superimposable to the penta-coordinated phosphorus in the transition state of in-line phosphoryl transfer E1PCa₂·ADP·Mg⁺ and acylphosphate hydrolysis E2-P·Mg^{*}. Thus all chemical properties of the P_i ana-





FIGURE 10. Formation of $E1Ca_2:BeF_3^-$ from $E2:BeF_3^-$ by lumenal Ca^{2+} binding. $E2:BeF_3^-$ was first produced in SR vesicles with BeF_x in the presence and absence of 1.2 μ m A23187 at 25 °C in 0.5 mm EGTA, 15 mm MgCl₂, 0.1 m KCl, and 30 mm MOPS/Tris (pH 7.0). Then a small volume of CaCl₂ was added to give various Ca²⁺ concentrations as indicated. At 10 s after this Ca²⁺ addition, TNP-AMP fluorescence was monitored with 4 μ m TNP-AMP. In *A* and *B*, the maximum intensity in the presence of A23187 was plotted in semi-log scale at 0–0.9 mm Ca²⁺ (A) and 0.7–15 mm Ca²⁺ (B). Note also the different scales in the ordinates. The inset in *A* is the linear plot at 0–0.9 mm Ca²⁺ to clearly show the saturation of the first phase of the Ca²⁺-dependent change, *i.e.* the formation of *E*1Ca₂:BeF₃⁻ from *E*2:BeF₃⁻. In *C*, the maximum superfluorescence intensity in the absence of A23187 was plotted at 0–15 mm Ca²⁺. In *D* and *E*, the traces of superfluorescence intensity in the absence of A23187 was plotted at 0–15 mm Ca²⁺. In *D* and *E*, the traces of superfluorescence intensity in the absence (*E*) were shown at the representative Ca²⁺ concentrations (0, 0.7 or 1.0, and 10 mm). Solid lines in *A* and *B* show the least squares fit to the Hill equation. Apparent Ca²⁺ affinity (K_{0.5}) and Hill coefficients at 0–0.9 mm Ca²⁺ are 0.43 mm and 3.8 with the intensity decrease from 170 to 142, and those at Ca²⁺ over 0.7 mm are 1.7 mm and 1.0 with a further intensity decrease to 36.

logs agree with the conclusion that $E1Ca_2 \cdot BeF_x$ is the analog for $E1PCa_2 \cdot Mg$, and BeF_x is most probably BeF_3^- , *i.e.* $E1Ca_2 \cdot BeF_3^-$.

Here note that the replacement of phosphate with BeF_3^- produces stabilization of the $E1PCa_2$ ·Mg structure with the same geometry of BeF_3^- as phosphate, and therefore probably with the same binding residues for them within the catalytic site. The $E1Ca_2$ ·Be F_3^- stability is likely brought about by the specific chemical nature of fluoride. Namely, it possesses a significantly higher electronegativity than oxygen (actually the highest among all atoms) and a small size, therefore producing stronger BeF_3^- binding in the catalytic site and fixing the intermediate structure.

Structure of $E1Ca_2 \cdot BeF_3^-$ —Then with the newly developed $E1Ca_2 \cdot BeF_3^-$, we explored its structural properties and uncovered the hitherto unknown nature of the physiological intermediate $E1PCa_2 \cdot Mg$ and structural changes during the phosphoryl transfer/ADP release and subsequent *E*P isomerization/Ca²⁺ release. The observed proteinase K resistance of Thr²⁴² on the A/M3-linker revealed that, in $E1PCa_2 \cdot Mg$ ($E1Ca_2 \cdot BeF_3^-$) the A domain is already rotated perpendicular to the membrane plane from the position in $E1Ca_2$, thereby bringing up its junction with the A/M3-linker and imposing a strain on this linker, similarly to $E1Ca_2 \cdot AIF_4^- \cdot ADP$ and $E1Ca_2 \cdot CaAMP$ -PCP (17, 18). As described for the $E1Ca_2 \cdot CaAMP$ -PCP structure (18), the

A/M3-linker strain, *i.e.* the A domain perpendicular rotation is brought about by bending the P domain due to binding of the phosphate moiety and Ca²⁺ at the catalytic site (Mg²⁺-site I, Asp³⁵¹/Thr³⁵³/Asp⁷⁰³) on the P domain (see Figs. 4 and 5 in Ref. 18).⁵ Our results revealed that such a strained state is achieved even without the N-P domain cross-linking nucleotide but solely with BeF₃⁻ and Mg²⁺ binding at the catalytic site, and therefore remains in *E*1PCa₂·Mg after ADP release.

The strain of the A/M3-linker thus imposed has been predicted with the atomic structure (18, 19) to function as a motive force for the A domain rotation parallel to the membrane in the *E*1P to *E*2P isomerization. The partial resistance at T2 site Arg^{198} in *E*1Ca₂·BeF₃⁻ (as compared with the rapid cleavage in *E*1Ca₂·CaAMP-PCP/*E*1Ca₂·AlF₄⁻·ADP and *E*1Ca₂·AlF_x) further indicated that in *E*1PCa₂·Mg, the A domain is already likely

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⁵ As depicted in Figs. 4 and 5 by Toyoshima *et al.* (18) for the change *E*1Ca₂ → *E*1Ca₂·CaAMP-PCP, the top part of the first half of the P domain (Pβ1–Pβ4) moves together as a result of γ-phosphate and Ca²⁺ (Mg²⁺) binding, because Thr³⁵³ just above Pβ1 coordinates to both ligands. Furthermore, Pβ5 twists upon binding of Ca²⁺ (Mg²⁺) because of the coordination by Asp⁷⁰³, which causes Pα5–Pα7 tilting. Thus the P domain is bent. This bending causes the perpendicular A domain rotation because the P7 helix moves upwards and tilts so that Gly¹⁵⁶–Lys¹⁵⁸ on the A domain is brought up as they are in contact with Ala⁷²⁵–Val⁷²⁶ on top of the P7 helix.

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rotated parallel to the membrane to some extent from the position in $E1Ca_2 \cdot MgATP^*/E1PCa_2 \cdot ADP \cdot Mg^{\ddagger}$. Thus, the A/M3linker strain is likely functioning for this partial A domain rotation during the phosphoryl transfer/ADP release to produce $E1PCa_2 \cdot Mg$, and further for the large and complete rotation to achieve the tight A-P domain association at Arg^{198} on the Val^{200} loop in the Ca^{2+} -released state $E2P \cdot Mg$ ($E2 \cdot BeF_3^-$). The A-P domain interaction at the Val^{200} loop is actually critical for formation of the proper Ca^{2+} -released structure, $E2P \cdot Mg$ and its analog $E2 \cdot BeF_3^-$ (25, 62, 63).

Here, it is of interest to note that residues Asp³⁵¹, Thr³⁵³, and Asp⁷⁰³ ligating Mg²⁺ and phosphate will come more proximate to each other during $E1PCa_2 \cdot ADP \cdot Mg^* \rightarrow E1PCa_2 \cdot Mg + ADP$ $(E1Ca_2 \cdot AlF_4 \cdot ADP/E1Ca_2 \cdot AlF_x \rightarrow E1Ca_2 \cdot BeF_x)$. As a consequence, a further P domain bending and more strain for the A/M3-linker will likely be induced by this coordination-chemical change, thereby contributing to inducing the A domain rotations during E1PCa₂·Mg formation and subsequent isomerization to E2PCa₂·Mg (besides the release of the N-P domain cross-linking nucleotide ADP). In any case, our results show that $E1PCa_2 \cdot Mg(E1Ca_2 \cdot BeF_3^-)$ as the product of the phosphorylation reaction acquires the structure ready for isomerization and Ca²⁺ deocclusion/release (*i.e.* ready for the large A domain rotation to produce $E2P \cdot Mg (E2 \cdot BeF_3^-)$), whereas the transition state structure in the phosphorylation $(E1Ca_2 \cdot AIF_4 \cdot ADP \text{ and } E1Ca_2 \cdot AIF_x)$ is not yet fully prepared. Note again that the $E1PCa_2$ ·Mg structure before such motions for its isomerization to *E*2P·Mg is stabilized with replacement of phosphate with BeF_3^- in $E1Ca_2 \cdot BeF_3^-$.

Important also, we found that the Ca²⁺-occluded transmembrane structure adopts not simply one state, but will proceed through changes during the phosphoryl transfer and ADP release to form *E*1PCa₂·Mg (see supplemental Fig. S3*E*). The structural change is probably coupled with the above described motions of the P and A domains (more bending and rotation) during this process. In the subsequent Ca²⁺ deocclusion/release in *E*1PCa₂·Mg \rightarrow *E*2P·Mg + 2Ca²⁺, the transmembrane structure changes further (52), which was also clearly mimicked here in the change *E*1Ca₂·BeF₃⁻ \rightarrow *E*2·BeF₃⁻. Thus, the structures of the transmembrane domain as well as the cytoplasmic domains in *E*1Ca₂·ABF₃⁻ (*E*1PCa₂·Mg) are intermediate between those of *E*1Ca₂·AIF₄⁻·ADP (*E*1PCa₂·ADP·Mg⁺) and *E*2·BeF₃⁻ (*E*2P·Mg).

 Mg^{2^+} as Physiological Catalytic Cation—Important questions regarding the $E1Ca_2 \cdot BeF_3^-$ structure are why Ca^{2^+} coordination at the catalytic Mg^{2^+} site (site I, $Asp^{351}/Thr^{353}/$ Asp^{703}) is absolutely unfavorable for $E1Ca_2 \cdot BeF_3^-$ formation, and why the Mg^{2^+} -coordinated structure $E1Ca_2 \cdot BeF_3^-$ differs from Ca^{2^+} -coordinated $E1PCa_2 \cdot Ca$, $E1Ca_2 \cdot CaAMP$ -PCP, and $E1PCa_2 \cdot Ca \cdot AMP$ -PN structures as well as from $E1Ca_2 \cdot$ $AIF_4^- \cdot ADP/E1Ca_2 \cdot AIF_x$, especially in A domain positioning. These questions may be relevant to the questions of why forward isomerization of $E1PCa_2 \cdot Ca$ to E2P is markedly retarded in contrast to $E1PCa_2 \cdot Mg$ and thus why Mg^{2^+} is preferred as the catalytic cation. In stringent coordination chemistry, the coordination distance of Mg^{2^+} is shorter than Ca^{2^+} , typically by 0.2 Å (e.g. 2.1 versus 2.3 Å (64, 65)). As a consequence, in the case of $E1Ca_2 \cdot CaAMP$ -PCP, the distance between the γ -phosphate and Asp³⁵¹-oxygen becomes 3.24 Å, being greater by 0.3 Å than that predicted in E1Ca₂·MgAMP-PCP. Therefore MgAMP-PCP (MgATP) binding would result in steric clash, and E1Ca₂·CaAMP-PCP is more stable than E1Ca₂·MgAMP-PCP, and therefore has less tendency to decompose to *E*1Ca₂ (also in the forward direction to the EP formation and its decay in the case of $E1Ca_2 \cdot CaATP$ (45). In $E1Ca_2 \cdot BeF_3^-$ formed here with Mg^{2+} , the direct coordination between Asp^{351} and the beryllium and their proximate positioning would probably favor the closely positioned ligand residues (Thr³⁵³/Asp⁷⁰³/ Asp³⁵¹) for BeF₃⁻ and Mg²⁺ but not for Ca²⁺. Therefore Ca²⁺ substitution of Mg²⁺ probably disrupted the precise geometry and decomposed the $E1Ca_2 \cdot BeF_3^-$ complex. Also, a possible difference in the coordination number might be involved; Mg²⁺ prefers definitely six, whereas Ca²⁺ can accommodate seven or eight ligands (65-69).

Furthermore, the difference in A domain positioning between the $Mg^{2+}\mbox{-}coordinated$ state ${\it E1Ca}_2\mbox{-}BeF_3^-$ and the Ca²⁺-coordinated states may be reasonably understood by the consequence of the stringent coordination chemistry. Namely, because the shorter coordination distance of Mg²⁺, P domain bending, and the resulting A domain rotation perpendicular to the membrane will be greater in the Mg²⁺-coordinated state. Therefore the strain of the A/M3-linker and A domain rotation parallel to the membrane will be more in the Mg^{2+} state $E1Ca_2 \cdot BeF_x$. In this context, it is also reasonable that $E1PCa_2$ ·Mg is more rapidly isomerized to E2P with less energy barrier for the large A domain rotation, in contrast to the retarded isomerization in E1PCa₂·Ca that is stabilized by the likely conformational inadequacy. Here note that the cause of the *E*1PCa₂·Ca stabilization is obviously different from that of E1PCa₂·Mg stabilization produced by replacement of phosphate with BeF_3^- (see the above discussion for $E1Ca_2 \cdot BeF_3^$ stabilization).

Previously it was documented (45, 64, 70) that destabilization of the non-covalent complex $E1Ca_2 \cdot MgATP$ by Mg^{2+} (as found with MgAMP-PCP versus CaAMP-PCP) together with stabilization of the transition state by Mg^{2+} (as found with $E1Ca_2 \cdot AIF_4^- \cdot ADP$ bound Mg^{2+} at both sites I and II) leads to a decrease of the activation energy and a rapid phosphoryl transfer. As another critical reason for Mg^{2+} preference for catalysis, we predict here by exploring the property of the $E1Ca_2 \cdot BeF_x$ that the Mg^{2+} bound at the catalytic site produces the proper $E1PCa_2$ structure, which is ready for rapid transition to E2P in this rate-limiting process of the transport cycle.

Hydrophobic Catalytic Site in $E1Ca_2 \cdot BeF_3^-$ —The microenvironment around Asp³⁵¹ in $E1PCa_2 \cdot Mg$ was further predicted by TNP-AMP superfluorescence in $E1Ca_2 \cdot BeF_3^-$ to be strongly hydrophobic and thus a closed state, and this will become even more in the change $E1PCa_2 \cdot Mg \rightarrow E2P \cdot Mg + 2Ca^{2+}$ ($E1Ca_2 \cdot BeF_3^- \rightarrow E2 \cdot BeF_3^-$). The observed distinct difference between $E1Ca_2 \cdot BeF_3^-$ and $E2 \cdot BeF_3^-$ (transient *versus* stable and slightly higher superfluorescence) is probably ascribed to the distinct difference in the organization state of cytoplasmic domains. The superfluorescence, nevertheless, developed solely in the Ca^{2+} -ATPase complexed with beryllium fluoride, $E1Ca_2 \cdot BeF_3^-$ and $E2 \cdot BeF_3^-$ ($E2P \cdot Mg$), but no development in $E1Ca_2 \cdot AlF_x$ and in $E2 \cdot AlF_4^-$ and $E2 \cdot MgF_4^{2-}$ (25). Therefore, the



hydrophobic closed catalytic site is accomplished by the direct coordination and close proximity of the beryllium with Asp³⁵¹-oxygen and by the specific coordination of the tetrahedral -O-BeF₃⁻, *i.e.* Asp³⁵¹-acylphosphate within the catalytic site. This is obviously not the case in AlF₄⁻ (the penta-coordinated phosphorus of the transition states) and MgF₄²⁻ (non-covalently bound P_i), thus in these states, the catalytic site is more accessible to nonspecific water molecules.

Whether E1PCa₂·Mg develops the superfluorescence had been controversial. In addition to the obvious problem of TNP-AMP competition against ATP for phosphorylation, the observed TNP-AMP-induced decomposition of E1Ca₂·BeF₃ further revealed that the E1PCa₂·Mg structure may be similarly disrupted rapidly by TNP-AMP binding, therefore making it virtually impossible to examine the superfluorescence development in E1PCa₂·Mg. The TNP-AMP-induced $E1Ca_2$ ·BeF₃⁻ decomposition might have occurred by means of a similar structural change as the ADP-induced one, i.e. disruption of the cytoplasmic domain organization and possible BeF₃⁻ release. The most important conclusion here is that the hydrophobic and closed property of the phosphorylated catalytic site both in $E1PCa_2$ ·Mg and E2P·Mg may be requisite to avoid a possible attack of nonspecific water molecules on the Asp³⁵¹-acylphosphate thus accomplishing Ca²⁺ release into the lumen and energy coupling.

Formation of $E1Ca_2 \cdot BeF_3^-$ from $E2 \cdot BeF_3^-$ and Perfect Stabili*zation of* $E1Ca_2 \cdot BeF_3^-$ – $E1Ca_2 \cdot BeF_3^-$ was produced also from $E2 \cdot BeF_3^-$ by binding two lumenal Ca²⁺ to the lumenally oriented low affinity transport sites at 0.7 mM Ca²⁺ and 15 mM Mg^{2+} , as mimicking the reverse transition $E2P \cdot Mg + 2Ca^{2+} \rightarrow$ *E*1PCa₂·Mg. At the critical concentration of 0.7 mM Ca^{2+} in 15 mM Mg²⁺, $E1Ca_2 \cdot BeF_3^-$ is perfectly stabilized without decomposition to $E1Ca_2$ or conversion to $E2 \cdot BeF_3^-$. The perfect $E1Ca_2 \cdot BeF_3^-$ stabilization is obviously achieved by preventing Ca²⁺ release into the lumen and by avoiding the absolutely unfavorable Ca²⁺ substitution of Mg²⁺ in site I at the most appropriately balanced concentrations of Ca²⁺ and Mg²⁺. As noted in the last paragraph under "Results," stabilization of $E1Ca_2 \cdot BeF_3^-$ might possibly involve lumenal Ca^{2+} access at the putative lumenal gating sites besides the transport sites. If this is the case, the gate-opening and Ca^{2+} release into the lumen takes place when the lumenal Ca^{2+} is low enough to avoid the possible lumenal Ca^{2+} access to the gate.

Integrated Picture of EP Processing—Recently, we successfully identified and trapped for the first time the intermediate state $E2PCa_2$ ·Mg, ADP-insensitive EP with two Ca^{2+} occluded at transport sites, by elongating the A/M1'-linker (71), and revealed that the proper length of this linker is critical for inducing structural changes for Ca^{2+} deocclusion and release from $E2PCa_2$ ·Mg. This dependence on the length of the linker is probably because the length controls the extent of strain between the A domain and M1', which causes motions of the cytoplasmic A and P domains thereby transmitting the structural signal to the transmembrane transport sites. In trapped $E2PCa_2$ ·Mg, the A domain is already largely rotated, and A-P domain associations at Val²⁰⁰ and TGES¹⁸⁴ loops are already produced, although the interaction network is not produced properly at the Tyr¹²²-hydrophobic cluster (71), which is critical for Ca²⁺ deocclusion/release and *E*2P hydrolysis (72–74). In the Ca²⁺-released *E*2P·Mg, this cluster is formed from seven residues of the A ($Ile^{179}/Leu^{180}/Ile^{232}$) and P (Val^{705}/Val^{726}) domains and the top part of M2 (Leu^{119}/Tyr^{122}) (see Fig. 2).

The results indicated that the successive structural changes take place as follows: in $E1PCa_2 \cdot Mg \rightarrow E2PCa_2 \cdot Mg$, the A domain rotates largely (further from the position in $E1PCa_2 \cdot Mg$) into the space between the N and P domains and docks onto the Asp³⁵¹-acylphosphate of the P domain, thereby causing loss of ADP sensitivity and also the strain of the A/M1'linker (because the A domain is brought above the P domain). The strain thus imposed will cause inclinations of the A and P domains and the connected M2 and M4/M5 thereby rearranging the helices to destroy Ca²⁺ sites and open the lumenal gate thus to release Ca^{2+} . Upon these motions, the Tyr¹²²-hydrophobic cluster is produced from the inclined A and P domains and M2. Hence, interactions at this cluster and at the Val²⁰⁰ loop stabilize the Ca^{2+} -released structure *E*2P·Mg, and also produce the catalytic site for the acylphosphate hydrolysis to occur after Ca^{2+} release, ensuring energy coupling (63, 72–74). Atomic level structural studies of $E1Ca_2 \cdot BeF_3^-$ as $E1PCa_2 \cdot Mg$ and the trapped intermediate state E2PCa₂·Mg will contribute to further understanding of *EP* processing, Ca²⁺ handling, and E2P hydrolysis.

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REFERENCES

- 1. Hasselbach, W., and Makinose, M. (1961) Biochem. Z. 333, 518-528
- 2. Ebashi, S., and Lipmann, F. (1962) J. Cell Biol. 14, 389-400
- 3. Inesi, G., Sumbilla, C., and Kirtley, M. E. (1990) Physiol. Rev. 70, 749-760
- 4. Møller, J. V., Juul, B., and le Maire, M. (1996) *Biochim. Biophys. Acta* **1286**, 1–51
- MacLennan, D. H., Rice, W. J., and Green, N. M. (1997) J. Biol. Chem. 272, 28815–28818
- 6. McIntosh, D. B. (1998) Adv. Mol. Cell. Biol. 23A, 33-99
- 7. Toyoshima, C., and Inesi, G. (2004) Annu. Rev. Biochem. 73, 269-292
- 8. Toyoshima, C. (2008) Arch. Biochem. Biophys. 476, 3-11
- 9. Toyoshima, C. (2009) Biochim. Biophys. Acta 1793, 941-946
- 10. Yamamoto, T., and Tonomura, Y. (1968) J. Biochem. 64, 137-145
- 11. de Meis, L., and Masuda, H. (1974) *Biochemistry* 13, 2057–2062
- 12. Masuda, H., and de Meis, L. (1973) Biochemistry 12, 4581-4585
- 13. Kanazawa, T., and Boyer, P. D. (1973) J. Biol. Chem. 248, 3163-3172
- 14. Inesi, G., Kurzmack, M., Coan, C., and Lewis, D. E. (1980) J. Biol. Chem.
- **255**, 3025–3031
- Toyoshima, C., Nakasako, M., Nomura, H., and Ogawa, H. (2000) Nature 405, 647–655
- 16. Toyoshima, C., and Nomura, H. (2002) Nature 418, 605-611
- 17. Sørensen, T. L., Møller, J. V., and Nissen, P. (2004) Science 304, 1672-1675
- 18. Toyoshima, C., and Mizutani, T. (2004) Nature 430, 529-535
- 19. Toyoshima, C., Nomura, H., and Tsuda, T. (2004) Nature 432, 361-368
- Olesen, C., Sørensen, T. L., Nielsen, R. C., Møller, J. V., and Nissen, P. (2004) Science 306, 2251–2255
- 21. Toyoshima, C., Norimatsu, Y., Iwasawa, S., Tsuda, T., and Ogawa, H. (2007) *Proc. Natl. Acad. Sci. U.S.A.* **104**, 19831–19836
- Olesen, C., Picard, M., Winther, A. M., Gyrup, C., Morth, J. P., Oxvig, C., Møller, J. V., and Nissen, P. (2007) *Nature* 450, 1036–1042
- Danko, S., Daiho, T., Yamasaki, K., Kamidochi, M., Suzuki, H., and Toyoshima, C. (2001) FEBS Lett. 489, 277–282
- 24. Danko, S., Yamasaki, K., Daiho, T., Suzuki, H., and Toyoshima, C. (2001) *FEBS Lett.* **505**, 129–135

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- Danko, S., Yamasaki, K., Daiho, T., and Suzuki, H. (2004) J. Biol. Chem. 279, 14991–14998
- Shigekawa, M., Wakabayashi, S., and Nakamura, H. (1983) J. Biol. Chem. 258, 8698 – 8707
- 27. Wakabayashi, S., and Shigekawa, M. (1987) J. Biol. Chem. 262, 11524–11531
- Nakamura, S., Suzuki, H., and Kanazawa, T. (1994) J. Biol. Chem. 269, 16015–16019
- Barrabin, H., Scofano, H. M., and Inesi, G. (1984) Biochemistry 23, 1542–1548
- 30. Meissner, G., and Fleischer, S. (1974) Methods Enzymol. 32, 475-481
- Kubota, T., Daiho, T., and Kanazawa, T. (1993) *Biochim. Biophys. Acta* 1163, 131–143
- Troullier, A., Girardet, J. L., and Dupont, Y. (1992) J. Biol. Chem. 267, 22821–22829
- 33. Laemmli, U. K. (1970) Nature 227, 680-685
- 34. Hiratsuka, T. (1982) Biochim. Biophys. Acta 719, 509-517
- Lowry, O. H., Rosebrough, N. J., Farr, A. L., and Randall, R. J. (1951) *J. Biol.* Chem. 193, 265–275
- Humphrey, W., Dalke, A., and Schulten, K. (1996) J. Mol. Graph. 14, 33–38
- 37. Webb, M. R., and Trentham, D. R. (1981) J. Biol. Chem. 256, 4884-4887
- Hasselbach, W., Fassold, E., Migala, A., and Rauch, B. (1981) Fed. Proc. 40, 2657–2661
- González, D. A., Ostuni, M. A., Lacapère, J. J., and Alonso, G. L. (2006) Biophys. Chem. 124, 27–34
- 40. Yamada, S., and Ikemoto, N. (1980) J. Biol. Chem. 255, 3108-3119
- 41. Kanazawa, T. (1975) J. Biol. Chem. 250, 113-119
- Ogurusu, T., Wakabayashi, S., and Shigekawa, M. (1991) J. Biochem. 109, 472–476
- Möller, J. V., Lenoir, G., Marchand, C., Montigny, C., le Maire, M., Toyoshima, C., Juul, B. S., and Champeil, P. (2002) *J. Biol. Chem.* 277, 38647–38659
- Holdensen, A. N., and Andersen, J. P. (2009) J. Biol. Chem. 284, 12258-12265
- Picard, M., Toyoshima, C., and Champeil, P. (2005) J. Biol. Chem. 280, 18745–18754
- 46. Watanabe, T., and Inesi, G. (1982) J. Biol. Chem. 257, 11510-11516
- 47. Nakamoto, R. K., and Inesi, G. (1984) J. Biol. Chem. 259, 2961-2970
- Dupont, Y., Chapron, Y., and Pougeois, R. (1982) *Biochem. Biophys. Res.* Commun. 106, 1272–1279
- de Meis, L., Martins, O. B., and Alves, E. W. (1980) *Biochemistry* 19, 4252–4261
- 50. Dupont, Y., and Pougeois, R. (1983) FEBS Lett. 156, 93-98

- Champeil, P., Le Maire, M., Moller, J. V., Riollet, S., Guillain, F., and Green, N. M. (1986) *FEBS Lett.* **206**, 93–98
- 52. Suzuki, H., and Kanazawa, T. (1995) J. Biol. Chem. 270, 3089-3093
- 53. Dupont, Y., and Leigh, J. B. (1978) Nature 273, 396-398
- Obara, M., Suzuki, H., and Kanazawa, T. (1988) J. Biol. Chem. 263, 3690-3697
- 55. Inesi, G., Lewis, D., Toyoshima, C., Hirata, A., and de Meis, L. (2008) *J. Biol. Chem.* **283**, 1189–1196
- 56. Murphy, A. J., and Coll, R. J. (1993) J. Biol. Chem. 268, 23307-23310
- 57. Lakowitz, J. R. (1983) *Principles of Fluorescence Spectroscopy*, Plenum Press, New York
- 58. Myung, J., and Jencks, W. P. (1994) Biochemistry 33, 8775-8785
- Webb, R. J., Khan, Y. M., East, J. M., and Lee, A. G. (2000) J. Biol. Chem. 275, 977–982
- 60. Petsko, G. A. (2000) Proc. Natl. Acad. Sci. U.S.A. 97, 538-540
- Wang, W., Cho, H. S., Kim, R., Jancarik, J., Yokota, H., Nguyen, H. H., Grigoriev, I. V., Wemmer, D. E., and Kim, S. H. (2002) *J. Mol. Biol.* 319, 421–431
- Daiho, T., Suzuki, H., Yamasaki, K., Saino, T., and Kanazawa, T. (1999) FEBS Lett. 444, 54–58
- Kato, S., Kamidochi, M., Daiho, T., Yamasaki, K., Gouli, W., and Suzuki, H. (2003) J. Biol. Chem. 278, 9624–9629
- Picard, M., Jensen, A. M., Sørensen, T. L., Champeil, P., Møller, J. V., and Nissen, P. (2007) J. Mol. Biol. 368, 1–7
- Peeraer, Y., Rabijns, A., Collet, J. F., Van Schaftingen, E., and De Ranter, C. (2004) *Eur. J. Biochem.* 271, 3421–3427
- Stokes, D. L., and Green, N. M. (2003) Annu. Rev. Biophys. Biomol. Struct. 32, 445–468
- Yang, W., Lee, H. W., Hellinga, H., and Yang, J. J. (2002) Proteins Struct. Funct. Genet. 47, 344–356
- 68. Shannon, R. D. (1976) Acta Crystallogr. Sect. A 32, 751-767
- Falke, J. J., Drake, S. K., Hazard, A. L., and Peersen, O. B. (1994) *Q. Rev. Biophys.* 27, 219–290
- Suzuki, H., Nakamura, S., and Kanazawa, T. (1994) *Biochemistry* 33, 8240-8246
- Daiho, T., Yamasaki, K., Danko, S., and Suzuki, H. (2007) J. Biol. Chem. 282, 34429–34447
- Yamasaki, K., Daiho, T., Danko, S., and Suzuki, H. (2004) J. Biol. Chem. 279, 2202–2210
- Wang, G., Yamasaki, K., Daiho, T., and Suzuki, H. (2005) J. Biol. Chem. 280, 26508–26516
- Yamasaki, K., Wang, G., Daiho, T., Danko, S., and Suzuki, H. (2008) J. Biol. Chem. 283, 29144–29155

