Formation of the Stable Structural Analog of ADP-sensitive Phosphoenzyme of Ca2-ATPase with Occluded Ca2 by Beryllium Fluoride

*STRUCTURAL CHANGES DURING PHOSPHORYLATION AND ISOMERIZATION******□**^S**

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As a stable analog for ADP-sensitive phosphorylated intermediate of sarcoplasmic reticulum Ca2-ATPase *E***1PCa2**-**Mg, a complex of** *E***1Ca2**-**BeF***x***, was successfully developed by addition** of beryllium fluoride and Mg^{2+} to the Ca^{2+} -bound state, $E1Ca_2$. In $E1Ca_2$ BeF_{*x*}</sub>, most probably $E1Ca_2$ BeF₃, two Ca^{2+} are oc**cluded at high affinity transport sites, its formation required Mg2 binding at the catalytic site, and ADP decomposed it to** *E***1Ca2, as in** *E***1PCa2**-**Mg. Organization of cytoplasmic domains** $\inf E1$ Ca₂·BeF_{*x*} was revealed to be intermediate between those in $E1Ca₂$ ^{*AlF₄</sub>* ADP (transition state of $E1PCa₂$ formation) and} *E***2**-**BeF3** --**(ADP-insensitive phosphorylated intermediate** *E***2P**-**Mg). Trinitrophenyl-AMP (TNP-AMP) formed a very fluorescent (superfluorescent) complex with** *E***1Ca2**-**BeF***^x* **in contrast** to no superfluorescence of TNP-AMP bound to $E1Ca_2$ ·AlF_{*x*}. $E1Ca_2$ · BeF_x with bound TNP-AMP slowly decayed to $E1Ca_2$, **being distinct from the superfluorescent complex of TNP-AMP** with $E2 \cdot BeF_3^-$, which was stable. Tryptophan fluorescence re- \mathbf{v} realed that the transmembrane structure of $E1\mathrm{Ca}_2\text{-} \mathrm{BeF}_x$ mimics $E1PCa_2$ ⁻Mg, and between those of $E1Ca_2$ ⁻AlF₄⁻-ADP and $E2$ ^{-BeF}₃⁻. $E1Ca_2$ **·BeF**_{*x*} at low 50–100 μ M Ca^{2+} was converted slowly to $E2 \cdot BeF_3^-$ releasing Ca^{2+} , mimicking $E1PCa_2 \cdot Mg \rightarrow E2P \cdot Mg +$ 2Ca²⁺. Ca²⁺ replacement of Mg²⁺ at the catalytic site at approx i **i mately millimolar** high Ca^{2+} decomposed $E1Ca_2$ ·BeF_{*x*} to $E1\text{Ca}_2$. Notably, $E1\text{Ca}_2$ ·Be F_x was perfectly stabilized for at least **12 days by 0.7 mM lumenal Ca2 with 15 mM Mg2. Also, stable** $E1Ca_2$ ^{*}BeF_{*x*} was produced from $E2$ ^{*}BeF₃ at 0.7 m_M lumenal Ca^{2+} by binding two Ca^{2+} to lumenally oriented low affinity **transport sites, as mimicking the reverse conversion** *E***2P**- $Mg + 2Ca^{2+} \rightarrow E1PCa_2 \cdot Mg.$

Sarcoplasmic reticulum Ca^{2+} -ATPase (SERCA1a),² a representative member of the P-type ion transporting ATPases, cat-

Various intermediate structural states in the transport cycle were fixed as their structural analogs produced by appropriate ligands such as AMP-PCP (non-hydrolyzable ATP analog) or metal fluoride compounds (phosphate analogs), and their crystal structures were solved at the atomic level (15–22). The three cytoplasmic domains, N, P, and A, largely move and change their organization state during the transport cycle, and the changes are coupled with changes in the transport sites. Most remarkably, in the change from $E1Ca₂$ ·AlF₄·ADP (the transition state for $E1PCa_2$ formation, $E1PCa_2$ ADP Mg⁺) to $E2$ Be F_3^- (the ground state *E*2P-Mg) (23–25), the A domain largely rotates by more than 90° approximately parallel to the membrane plane and associates with the P domain, thereby destroying the Ca^{2+} binding sites, and opening the lumenal gate, thus releasing Ca²⁺ into the lumen (see Fig. 2). $E1PCa_2$ ·Ca·AMP-PN formed by CaAMP-PNP without Mg^{2+} is nearly the same as

rylation; A, actuator; AMP-PNP, adenosine 5'-(β , γ -imido)triphosphate; AMP-PCP, adenosine 5'-(β , γ -methylene)triphosphate; AMP-PN, adenosine 5'-diphosphoramidate.

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[supplemental Figs. S1–S5.](http://www.jbc.org/cgi/content/full/M109.029702/DC1) ¹ To whom correspondence should be addressed: Midorigaoka-Higashi, Asahikawa 078-8510, Japan. Tel.: 81-166-68-2350; Fax: 81-166-68-2359; E-mail:

hisuzuki@asahikawa-med.ac.jp. 2 The abbreviations used are: SERCA1a, adult fast-twitch skeletal muscle sarcoplasmic reticulum Ca²⁺-ATPase; SR, sarcoplasmic reticulum; *EP*, phosphoenzyme; *E*1P, ADP-sensitive phosphoenzyme; *E*2P, ADP-insensitive phosphoenzyme; MOPS, 3-(*N*-morpholino)propanesulfonic acid; TG, thapsigargin; TNP-AMP, 2'(3')-O-(trinitrophenyl)-AMP; C₁₂E₈, octaethylene glycol monododecyl ether; N, nucleotide binding; P, phospho-

 $E1Ca₂$ **-AlF₄**</sub>**-ADP** and $E1Ca₂$ **-CaAMP-PCP** in their crystal structures (17, 18, 22).

Despite these atomic structures, yet unsolved is the structure of $E1PCa_2$ Mg, the genuine physiological intermediate $E1PCa_2$ with bound Mg^{2+} at the catalytic site without the nucleotide. Its stable structural analog has yet to be developed. $E1PCa_2 Mg$ is the major intermediate accumulating almost exclusively at steady state under physiological conditions. Its rate-limiting isomerization results in Ca^{2+} deocclusion/release producing *E*2P·Mg as a key event for Ca^{2+} transport. In $E1Ca_2$ ·CaAMP-PCP, $E1Ca_2$ **:**AlF₄⁻**.**ADP, and $E1PCa_2$ **:**Ca**·**AMP-PN, the N and P domains are cross-linked and strongly stabilized by the bound nucleotide and/or Ca^{2+} at the catalytic site, thus they are crystallized (17, 18, 22). Kinetically, $E1PCa_2$ ·Ca formed with CaATP is markedly stabilized due to Ca^{2+} binding at the catalytic Mg²⁺ site, and its isomerization to *E*2P is strongly retarded in contrast to $E1PCa_2$ ·Mg (26, 27). Thus, the bound Ca^{2+} at the catalytic Mg^{2+} site likely produces a significantly different structural state from that with bound Mg^{2+} .

Therefore, it is now essential to develop a genuine $E1PCa_2$ Mg analog without bound nucleotide and thereby gain further insight into the structural mechanism in the Ca^{2+} transport process. It is also crucial to further clarify the structural importance of Mg^{2+} as the physiological catalytic cation. In this study, we successfully developed the complex $E1Ca₂$ -BeF_x, most probably $E1Ca₂$ -BeF₃, as the $E1PCa₂$ -Mg analog by adding beryllium fluoride (BeF_x) to the $E1Ca₂$ state without any nucleotides. For its formation, Mg^{2+} binding at the catalytic site was required and Ca^{2+} substitution for Mg^{2+} was absolutely unfavorable, revealing a likely structural reason for its preference as the physiological cofactor. In $E1Ca₂$ ⁻BeF₃, two Ca^{2+} ions bound at the high affinity transport sites are occluded. It was also produced from $E2$ Be F_3^- by lumenal Ca^{2+} binding at the lumenally oriented low affinity transport sites, mimicking $E2P \cdot Mg + 2Ca^{2+} \rightarrow E1PCa_2 \cdot Mg$. All properties of the newly developed $E1\text{Ca}_2\text{·BeF}_3^-$ fulfilled the requirements as the *E*1PCa₂·Mg analog, and hence we were able to uncover the hitherto unknown nature of $E1PCa_2 \cdot Mg$ as well as structural events occurring in the phosphorylation and isomerization processes. Also, we successfully found the conditions that perfectly stabilize the $E1\text{Ca}_2 \cdot \text{BeF}_3^-$ complex.

Structural Analog of E1PCa2-*Mg Intermediate of Ca2-ATPase*

EXPERIMENTAL PROCEDURES

Preparation of SR Vesicles and Treatment with BeF_x, AlF_x, and AlFx-*ADP*—SR vesicles were prepared from rabbit skeletal muscle as described (28). The content of the phosphorylation site in the vesicles determined according to Barrabin *et al.* (29) was 4.49 ± 0.22 nmol/mg of vesicle protein ($n = 5$). The Ca²⁺dependent ATPase activity determined at 25 °C in a mixture containing 5 μ g/ml microsomal protein, 1 mm ATP, 1.7 μ M A23187, 7 mm MgCl₂, 0.1 m KCl, 50 mm MOPS/Tris (pH 7.0), and 0.6 mm CaCl₂ with 0.5 mm EGTA (or 2 mm EGTA without added CaCl₂) was 1.87 \pm 0.14 μ mol/min/mg of vesicle protein $(n = 3)$. The Ca²⁺-ATPase was purified from the vesicles by deoxycholate as described (30, 31). The *E*1Ca₂ state ATPase was incubated with fluoride compounds, 2 mM potassium fluoride and 100 μ M BeSO₄ or AlCl₃, at 25 °C for 30 min in the presence of 0.1 mm Ca^{2+} , 15 mm MgCl₂, 0.1 m KCl, 30 mm MOPS/Tris buffer (pH 7.0), unless stated otherwise. $E1Ca₂$. $\text{AlF}_{4}^{-}\cdot$ ADP was formed by including 50 μ M ADP in the above AlF_{*x*} incubation mixture as described (32). $E2 \cdot BeF_3^-$, $E2 \cdot AlF_4^-$, and $E2$ ·MgF $_4^{2-}$ were produced as described (23–25).

Determination of EP—*E*P formation was performed with 3 μ м [γ -³²P]ATP in 100 (or 50) μ м Ca²⁺ at 0 °C for 3 s, and terminated by trichloroacetic acid containing carrier P_i. The amount of *E*P formed was determined as described previously (28). The background level determined with excess 5 mM EGTA was less than 0.5% of the phosphorylation sites.

 Ca^{2+} *Binding and Occlusion*—⁴⁵Ca²⁺ binding and occlusion at the transport sites was determined at 25 °C with 2 ml of the SR vesicle mixture (0.2 mg/ml protein) with a 0.45 - μ m nitrocellulose membrane filter (Millipore) as described (31). In some cases, the vesicles on the filter were washed for 10 s by perfusion with 2 ml of a washing solution containing 5 mm EGTA. The amount of Ca^{2+} bound at the transport sites was obtained by subtracting the nonspecific Ca^{2+} binding level determined as described in the figure legends.

Proteolysis—SR vesicles (0.45 mg/ml protein) were treated at 25 °C with trypsin (0.3 mg/ml) or proteinase K (0.1 mg/ml) as described (23, 24) and as noted in the figure legends. The samples were subjected to Laemmli SDS-PAGE (33) and densitometric analyses of the Coomassie Brilliant Blue R-250-stained gels (23, 24).

Fluorescence Measurements—The TNP-AMP fluorescence and intrinsic tryptophan fluorescence of $Ca^{2+}-ATP$ ase (0.06) mg/ml protein) were measured on a RF-5300PC spectrofluorophotometer (Shimadzu, Kyoto, Japan) with excitation and emission wavelengths 408 and 540 nm for TNP-AMP (with band widths 5 and 10 nm), and 290 and 338.4 nm for tryptophan (with bandwidth 1.5 and 5 nm), unless otherwise described (28).

Miscellaneous—Trypsin (L-1-tosylamido-2-phenylethyl chloromethyl ketone-treated) and proteinase K were obtained from Sigma. TNP-AMP was synthesized according to Hiratsuka (34). Protein concentrations were determined by the method of Lowry *et al.* (35) with bovine serum albumin as a standard. Free Ca^{2+} concentrations were calculated by the Calcon program. Data were analyzed by nonlinear regression using the program Origin (Microcal Software, Inc., Northampton,

MA). Three-dimensional models of the enzyme were reproduced by the program VMD (36).

RESULTS

*Formation of E1PCa*₂ *Analogs by Fluoride Compounds*—The Ca2- ATPase of SR vesicles in 100 μ M Ca^{2+} and 15 mm Mg^{2+} was incubated with beryllium fluoride (BeF*x*) or aluminum fluoride (AlF*x*) for 30 min at 25 °C. The ability of *E*P formation from ATP was almost completely lost (actually within 1 min) (Fig. 3), showing the stable complex formation with BeF*^x* and AlF*x*. No inhibition occurred with $\mathrm{Mg^{2+}}$ and fluoride without beryllium and aluminum $(E1Ca₂+(MgF_x))$. Thus, $MgF_x (MgF_4^{2-})$ was not able to produce a complex with $E1Ca₂$, in contrast to $E2 \cdot MgF_4^{2-}$ formation from *E*2, the *E*2-Pi product analog in *E*2P hydrolysis (19, 25). This finding agrees with the in-line phosphorylation of $E1Ca₂$ to $E1PCa₂$ (37), in which there is no state with non-covalently bound P_i.

The binding of Mg^{2+} at the catalytic site as a physiological cation is nevertheless required for *E*P formation. Actually in Fig. 3, Mg^{2+} was required for complex formation with BeF_{*x*} in 100 μ M Ca²⁺. The apparent Mg²⁺ affinity ($K_{0.5}$ of 5 m_M, [supplemental Fig. S1\)](http://www.jbc.org/cgi/content/full/M109.029702/DC1) was consistent with that of the catalytic site in phosphorylation from ATP or Pi (*e.g.* Refs. 11, 12, and 38– 41). The BeF*x*-induced inhibition also occurred with Mn^{2+} with apparent affinity $(K_{0.5}$ of 0.6 m_M) significantly higher than that of Mg^{2+} , as found in *E*1PCa₂ formation and ATP hydrolysis with Mn^{2+} (40, 42).

When Ca^{2+} over millimolar amounts was added in place of Mg^{2+} , the *E*P formation was not inhibited by BeF_x (BeF_x (5 mm Ca)). Therefore, Ca^{2+} substitution probably at the catalytic Mg^{2+} site abolished complex formation with BeF*x*. Although CaATP as a substrate and Ca^{2+} bound at the catalytic Mg²⁺ site are able to function for *E*1PCa₂ formation (26, 27), Ca^{2+} bound at the catalytic site likely produces different structure from the Mg^{2+}

FIGURE 3. Inhibition of *E*P formation by binding of BeF_y and AIF_y to *E*1Ca₂. The $E1Ca₂$ state ATPase of SR vesicles in 0.1 or 5 mm free Ca²⁺ was treated for 30 min at 25 \degree C with fluoride compounds in the presence of the indicated concentrations of the ligands. Subsequently, the samples were cooled to 0 °C and phosphorylated for 3 s by 3 μ M [γ ⁻³²P]ATP. The amount of *E*P formed was shown as the percentage of the control value obtained without F⁻, Be²⁺, and Al³⁺. The *E*P formation was not inhibited when the incubations were performed with Be²⁺ or Al³⁺ but without F⁻. The error bars show the S. D. of five independent experiments.

bound structure (as in fact found, see below). The complex formation of $E1Ca₂$ by AlF_x in 100 μ M Ca²⁺ also required Mg²⁺ or Mn^{2+} with somewhat higher apparent affinities than those for BeF*x*-induced complex formation, and was abolished by $Ca²⁺$ binding at the catalytic site.

Ca2 Binding and Occlusion at Transport Sites—In Fig. 4, Ca^{2+} binding and occlusion in formation of the Ca^{2+} -ATPase complexes with fluoride compounds were determined in 15 mm Mg^{2+} with and without a 10-s EGTA filter washing. In all the cases without washing, Ca^{2+} was bound to the $Ca^{2+}-ATP$ ase with high affinities; $K_{0.5}$ at sub-micromolar to millimolar ranges, Hill coefficient \sim 2, and maximum levels of 9-10 nmol/mg of protein, *i.e.* the stoichiometry of two Ca^{2+} per phosphorylation site (*inset* at 50 μ M Ca²⁺). Therefore, $E1Ca_2$ ·Be F_x and $E1Ca_2$ ·AlF₄·ADP/*E*1Ca₂·AlF_x were produced by cooperative binding of two Ca^{2+} ions at high affinity transport sites. This finding agrees with the property of the sites for $Ca²⁺$ binding and the resulting enzyme activation for phospho-

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rylation by ATP as nicely demonstrated for the first time by Inesi *et al.* (14). Upon the EGTA washing of *E*1Ca₂ that is complexed with BeF_{*x*}, the two bound Ca^{2+} were not removed, and therefore occluded in the complex as " $E1Ca_2$ -BeF_x." The two Ca^{2+} are occluded also in $E1\text{Ca}_2\text{-} \text{AlF}_{4}^{-}\text{-}\text{ADP}$ and less strongly in $E1Ca₂·AlF_x$.

Note that the Ca²⁺ affinity became 2–3-fold lower for BeF_{*x*} and AlF_{*x*}. This may be because the Ca^{2+} -free *E*2 state produces $E2$ **·BeF**₃⁻ and $E2$ **·AlF**₄⁻ (25), and therefore competes with Ca²⁺ binding for formation of $E1Ca_2$ ·BeF_{*x*} and $E1Ca_2$ ·AlF_{*x*}. On the other hand, the observed \sim 3-fold Ca²⁺-affinity increase in formation of $E1Ca₂$ AlF₄. ADP is probably brought about by the fact that ADP together with AlF_x strongly stabilizes the crosslinked N-P domains (17, 18), which is unfavorable for formation of the Ca^{2+} -free *E*2 and *E*2·AlF₄⁻, because for these structures the A domain should rotate into the opened space between the N and P domains and associate with them (19, 23, 24).

Cytoplasmic Structure in E1Ca2-*BeFx Is Intermediate* between Those in E1Ca₂**·**AlF₄</sub> ·ADP and E2·BeF₃ – Proteolytic analysis was made to reveal the organization state of the cytoplasmic domains in the newly developed $E1PCa_2$ ·Mg analog $E1Ca_2$ **E**: BeF_{*x*}, and to compare with $E1Ca_2$ **AlF**₄**·ADP**/ $E1\text{Ca}_2\text{-} \text{AlF}_x$ and $E2\text{-} \text{BeF}_3^-\text{ } (E2\text{P}\text{-}\text{Mg)}$ (see the typical cleavage in [supplemental Fig. S2\)](http://www.jbc.org/cgi/content/full/M109.029702/DC1). The initial rate of the "A1" appearance upon cleavage at the T2 site (Arg¹⁹⁸ on the Val²⁰⁰ loop of the A domain) in $E1\text{Ca}_2\text{·BeF}_x$ was substantially slower than the rapid cleavage of $E1Ca_2$ AlF₄ ADP and $E1Ca_2$ AlF_x as well as $E1Ca_2$ (Table 1). The slowed T2 cleavage was also observed when $E1Ca₂$ -BeF_{*x*} was formed with 3 m_M Mn²⁺ in place of 15 m_M Mg^{2+} (data not shown). Also important was the slow but definitely occurring T2 cleavage in $E1\text{Ca}_2\text{-}8\text{eF}_\text{}_{x^\prime}$ in sharp contrast to its complete resistance in *E*2·BeF₃. Therefore A-P domain organization at the Val²⁰⁰ loop in $E1\text{Ca}_2\text{·BeF}_x$ is intermediate between those in $E1Ca_2$ **:** AlF^{$-$}**.** ADP/ $E1Ca_2$ **:** AlF_{*x*} and $E2$ **·BeF**₃⁻.

All complexes were almost completely resistant to proteinase K at the major site of Thr^{242} on the A/M3-linker that produces the "p83" fragment. Therefore, in $E1\text{Ca}_2\text{·BeF}_\textit{x}$, the A domain is rotated perpendicular to the membrane plane from its position in *E*1Ca₂ thereby causing the A/M3-linker strain, as in $E1Ca₂$ -CaAMP-PCP, $E1Ca_2$ **•AlF**₄**•ADP** (18, 19, 24), and $E1Ca_2$ **•AlF**_x.

These analyses revealed that in the change $E1Ca₂·AlF₄·$ $ADP \rightarrow E1Ca_2 \cdot BeF_x$ (*i.e.* upon the ADP release from the transition state), the A domain moves, *i.e.* probably rotates to some extent parallel to the membrane plane likely due to the A/M3 linker strain, and thereby Arg¹⁹⁸ on the Val²⁰⁰ loop comes close to the P domain. In the subsequent change, $E1Ca_2 \cdot BeF_x \rightarrow$ *E*2·BeF₃, the A domain rotates further (by the A/M3-linker strain as predicted to be motive force (18, 19, 43, 44)) and pro-

FIGURE 2. Structure of SERCA1a and its change during processing of phosphorylated intermediate. E1Ca₂·AIF₄·ADP (the transition state analog for phosphorylation *E*1PCa₂·ADP·Mg[‡]) and *E2·*BeF₃ (the ground state *E*2P analog (25)) were obtained from the Protein Data Bank (PDB accession code 1T5T (17) and 2ZBE (21), respectively). Cytoplasmic domains N (nucleotide binding), P (phosphorylation), and A (actuator), and 10 transmembrane helices (M1–M10) are indicated. The *arrows* on the domains, M1' and M2 (Tyr¹²²) in E1Ca₂·AIFa ·ADP, indicate their approximate motions predicted for E1PCa₂·ADP·Mg⁺ → E2P·Mg.
The phosphorylation site Asp³⁵¹, TGES¹⁸⁴ of the A domain K site) on the A/M3-linker are shown. Seven hydrophobic residues gather in the E2P state to form the Tyr¹²²-hydrophobic cluster (*Y122-HC*); Tyr¹²²/Leu¹¹⁹ on the top part of M2, Ile¹⁷⁹/Leu¹⁸⁰/Ile²³² of the A domain, and Val⁷⁰⁵/Val⁷²⁶ of the P domain. The overall structure of *E*1Ca₂·AIF₄·ADP is virtually the same as those of *E*1Ca2-CaAMP-PCP and *E*1PCa2-Ca-AMP-PN (17, 18, 22).

FIGURE 4. Ca²⁺ binding and occlusion at transport sites. The *E*1Ca₂ state ATPase of SR vesicles in various concentrations of ${}^{45}Ca^{2+}$ and 15 mm MgCl₂ was incubated with BeF_x, AlF_x, and AlF_x-ADP or without these compounds (*E1Ca₂*) for 30 min at 25 °C. The amounts of bound (*upper panel*) and occluded (*lower panel*) ⁴⁵Ca²⁺ were determined without and with the perfusion of the membrane filter with a 5 mm EGTA-containing washing solution (without CaCl₂ and fluoride compounds otherwise as the above incubation solution). The nonspecific
Ca²⁺ binding was determined by including 10 *u*M thapsigargin before the addition of fluoride compounds. binding was determined by including 10 μ m thapsigargin before the addition of fluoride compounds, and subtracted. When ADP was used for E1Ca₂ AlF₄ ADP, 5 μ M A23187 was included to avoid Ca²⁺ accumulation in the vesicles by ATP produced from ADP due to adenylate kinase in the vesicles. In the *inset,* the
stoichiometries of bound Ca²⁺ (*open bars*) and occluded Ca²⁺ (*closed bars*) to the phosphorylation site (Pwere determined at saturating 50 μ ^{M 45}Ca²⁺. Solid lines in the *upper panel* show the least squares fit to the Hill equation. K_{0.5} of Ca²⁺ and Hill coefficients obtained were 1.3 μm and 1.7 (*E1Cα₂), 4.3 μm* and 1.7 (*E1Cα₂·BeF_x), 2.*3 μ м and 1.7 (E1Ca₂·AlF_x), and 0.4 μ м and 1.4 (E1Ca₂·AlF-4·ADP). In the lower panel, the values for E1Ca₂·BeF \hat{x} and $_{E1C a_{2}}$ ·AIF $_{4}^{-}$ ·ADP are essentially not altered by EGTA washing (4.7 μ m and 1.9, and 0.4 μ m and 1.9, respectively).

duces its tight association with the P domain at the Val²⁰⁰ loop, mimicking $E1PCa_2$ ·Mg \rightarrow $E2P$ ·Mg + $2Ca^{2+}$.

 Ca^{2+} *Ligation at the Catalytic Mg²⁺ Site*—The proteolysis further revealed that $E1\text{Ca}_2\text{-} \text{BeF}_x$ was not produced from $E1\text{Ca}_2$ in 5 mm Ca²⁺ without Mg²⁺ (*E1Ca*₂ + 5 mm Ca²⁺ + BeF_x</sub> in Table 1), and that $E1Ca_2$ BeF_{*x*} produced in 15 mm Mg²⁺ and 50–100 μ M Ca²⁺ was decomposed to the *E*1Ca₂ by 5 mM Ca²⁺ $(EICa₂:BeF_x + 5$ *mm* $Ca²⁺$, as shown by the rapid cleavage

rates at the T2 and proteinase K sites. In $E1PCa_2$ **Ca** and $E1Ca_2$ **²** CaAMP-PCP formed in 5 mm Ca^{2+} without Mg^{2+} (Table 1),³ the T2 site was also rapidly cleaved, in contrast to its substantially slowed cleavage in $E1Ca_2$ ·BeF_{*x*} formed with Mg^{2+} . Thus, for organization of the cytoplasmic domains at the T2 site (Arg¹⁹⁸), *E*1Ca₂·CaAMP-PCP and E1PCa₂·Ca are very similar to $E1Ca₂$ -AlF₄-ADP, but differ from $E1Ca₂·BeF_x$. The close similarity between $E1Ca₂$ -CaAMP-PCP and $E1Ca₂$ **AlF**₄**·ADP** is in agreement with their nearly same atomic structures and previous observations (17, 18, 45). Also notably, structure E1PCa₂·Ca·AMP-PN formed by CaAMP-PNP in 10 mm Ca^{2+} without Mg^{2+} (22) is almost identical with those of $E1Ca₂$ -CaAMP-PCP and $E1Ca_2$ **AlF**₄ **ADP** (see also Table 1).

In $E1Ca₂$ -CaAMP-PCP and E1PCa₂·Ca·AMP-PN, the N-P domain cross-linked state is stabilized by Ca^{2+} bound at catalytic Mg^{2+} site I $(Asp^{351}/Thr^{353}/Asp^{703}$ and the phosphate) and by the nucleotide to be nearly identical to the state stabilized with AlF_4^- plus ADP in $E1Ca₂$ -AlF₄ -ADP (17, 18, 22). The results on *E*1PCa₂·Ca further indicated that such an N-P domain closed state is stabilized solely by site I Ca^{2+} ligation without the nucleotide. The stabilization of this state in *E*1PCa₂·Ca is consistent with its markedly retarded isomerization to *E*2P (27), because isomerization requires the A domain rotation into the space between the N and P domains. In $E1Ca_2$ ·BeF_{*x*} formed with Mg^{2+} at the catalytic site (site I), such a Ca^{2+} ligation effect is obviously not present. Therefore, the N and P domains are probably more easily separated from each other, and the A domain

can rotate into the space between the N and P domains to some extent thus resulting in partial T2 resistance (but not yet as

³ The proteolytic analysis of *E*1PCa₂·Ca was made possible by its markedly retarded decay due to Ca^{2+} ligation at the catalytic site (27) and feedback inhibition by the high lumenal Ca²⁺. Analysis of *E*1PCa₂·Mg formed from MgATP was not feasible because of its very rapid turnover, thus of a very rapid ATP exhaustion.

TABLE 1

Summary of fluorescence changes and proteolysis rates

Maximal TNP-AMP fluorescence intensity at saturating 4 μ m TNP-AMP is given as % value of that of $E2\text{-BeF}_3^-$ without A23187. Tryptophan fluorescence change upon complex formation with the ligand from *E*1Ca₂ (or from other state when indicated in parentheses) is shown as % value of the intensity of *E*1Ca₂ (see also [supplemental Fig.](http://www.jbc.org/cgi/content/full/M109.029702/DC1) [S3](http://www.jbc.org/cgi/content/full/M109.029702/DC1)E). The cleavage rate at the T2 site (Arg¹⁹⁸) with trypsin and the digestion rate of the 110-kDa ATPase chain with proteinase K were obtained by the detailed time course
analysis in the initial 1 (72) and 30 min (*prot* mM Ca^{2+} instead of 0.1 mM, but the results were virtually the same and therefore are represented with 0.1 mM Ca^{2+} for simplicity. It should also be mentioned that the digestion rates in *E*1Ca₂ were not altered by 5 or 0.7 mm Ca²⁺ or by A23187 (being 97–101% of the rates of *E*1Ca₂ in 0.1 mm Ca²⁺ without A23187), and that the ligand-free *E*2 state was also rapidly digested by trypsin and proteinase K, and TG binding to *E*2 and A23187 did not alter essentially the rapid cleavage rates (Refs. 23–25). The other Ca²⁺-ATPase complexes were produced under the same buffer conditions as those for $E1Ca_2$ ·BeF_x formation, otherwise as follows and noted below: $E1Ca_2$ ·MgAMP-PCP by 5 mm MgAMP-PCP; *E*1Ca₂·CaAMP-PCP by 5 mm CaAMP-PCP; *E*1PCa₂·Ca AMP-PN by 5 mm CaAMP-PNP; *E*1PCa₂·Ca by 5 mm CaATP. *E*1Ca₂·BeF_x + 5 mm Ca²⁺, the $E1\text{Ca}_2\text{Ber}_x$ complex was formed in 15 mm Mg²⁺ and 100 μ m Ca²⁺ and then incubated with the subsequently added 5 mm Ca²⁺ for 3 h; $E1\text{Ca}_2$ + 5 mm Ca²⁺ + BeF_x, the *E*1Ca₂ state ATPase was incubated with BeF_x for 10 min in the presence of 5 mM Ca²⁺ without Mg²⁺.

completely as in $E2$ ·BeF₃). As the cause of the Ca²⁺-induced $E1\text{Ca}_2$ **BeF**_x to $E1\text{Ca}_2$ decomposition, Ca^{2+} replacement of Mg^{2+} at site I altered the domain organization state and made the BeF*^x* ligation unfavorable (see "Discussion").

TNP-AMP Superfluorescence—TNP-AMP binds to the nucleotide binding site with an extremely high affinity $(46 - 48)$, and in the *E*2P ground state and its analog $E2$ ·BeF₃, the bound TNP-AMP develops its extremely high fluorescence "superfluorescence" (25), which reflects a strongly hydrophobic atmosphere around $Asp³⁵¹$ (49, 50). On the other hand, it has been controversial whether *E*1PCa₂ develops TNP-AMP superfluorescence, mostly because its tight binding to the nucleotide binding site prevents phosphorylation to form $E1PCa_2 Mg$, so the TNP-AMP·E1PCa₂·Mg complex is not formed in significant amounts. Nakamoto and Inesi (47), nevertheless, predicted the development of superfluorescence in *E*1PCa₂.

Here, with $E1Ca_2$ ·Be F_x as the $E1PCa_2$ ·Mg analog, we examined the superfluorescence without ATP. In Fig. 5, *A* and *B*, we first formed $E1\text{Ca}_2$ ·BeF_{*x*} in 15 mm Mg²⁺ and 50 μ m Ca²⁺, then TNP-AMP was added to give a saturating level of 4 μ M. $E1\text{Ca}_2\text{\cdot}\text{BeF}_x$ rapidly developed superfluorescence, and then the fluorescence decreased slowly (much more slowly and extensively at 4° C).⁴ The proteolysis after the loss of superfluores-

cence revealed that $E1Ca_2$ ·BeF_x was decomposed to $E1Ca_2$ (data not shown). Thus, $E1\text{Ca}_2\text{·BeF}_x$ develops superfluorescence, and the TNP-AMP binding *per se* causes its decomposition to $E1Ca₂$, in sharp contrast to the completely stable $E2$ **·BeF**₃ even after TNP-AMP binding. The maximum superfluorescence level of $E1\text{Ca}_2\text{·BeF}_x$ was slightly lower than that of $E2 \cdot BeF_3^-$ (Fig. 5, *A-C*), which is the same as that of $E2P \cdot Mg$ formed from P_i (25). The results clearly revealed that the atmosphere around Asp^{351} in $E1\text{Ca}_2\text{·BeF}_x$ is strongly hydrophobic, similar to $E2$ · BeF_{3}^{-} , although the cytoplasmic domain organization in $E1\text{Ca}_2\text{·BeF}_x$ distinctly differs from and did not yet reach the most compactly organized state in $E2$ ^{-BeF}₃.

 $E1\text{Ca}_2\text{·AlF}_x$ as well as $E2$ and $E1\text{Ca}_2$ did not develop superfluorescence despite high affinity TNP-AMP binding. Therefore the catalytic site in $E1Ca_2$ AlF_{*x*} is hydrophilic and differs critically from the strongly hydrophobic site in $E1Ca_2$ ·BeF_{*x*}. Note also that $E2$ AlF₄ (*E*2-*P*⁺) and $E2$ MgF₄² (*E*2·*P*_i) do not develop TNP-AMP superfluorescence (25). The superfluorescence therefore develops solely with $Ca^{2+}-ATP$ ase complexed with BeF_{x} ; $E1\text{Ca}_{2}$ · BeF_{x} and $E2$ · BeF_{3}^{-} .

The superfluorescence development of $E1Ca_2$ ·Be F_x in the presence of 15 mm Mg^{2+} and 50 μ m Ca²⁺ was rapidly diminished with increasing Ca^{2+} over millimolar concentrations of

⁴ The fluorescence level after the decrease of the transient superfluorescence of *E*1Ca₂·BeF_x was somewhat higher than the non-superfluorescent low level of TNP-AMP bound to *E*1Ca₂ especially at 25 °C. We obtained the results indicating that a small fraction of $E2$ ·BeF $_3^-$ was produced even in the

presence of 50 μ M Ca²⁺ (more at 25 than at 4 °C) after the TNP-AMP-induced *E*1Ca₂·BeF_x decomposition to *E*1Ca₂, and therefore remained somewhat superfluorescence (data not shown).

FIGURE 5. TNP-AMP superfluorescence of *E***1Ca₂·BeF_x and** *E***2·BeF₃ .** *E***1Ca₂·BeF_x and** *E***1Ca₂·AlF_x were pro**duced in 50 μ m Ca²⁺ and 15 mm MgCl₂, then at 25 (A) and $4 \textdegree C$ (B), a small volume of TNP-AMP was added to give saturating 4 μ m, and the TNP-AMP fluorescence was followed. The fluorescence of *E*2-BeF $_3^-$ produced with BeF $_\star$ in the absence of Ca²⁺, the *E*1Ca₂ and *E*2 states without the fluoride compounds, and without SR vesicles (no *SRV*) were also followed. *C*, the TNP-AMP fluorescence intensities were measured at various concentrations of TNP-AMP at 25 °C, otherwise as described for *A*. For *E*1Ca₂·BeF_x, the maximum level of transient superfluorescence was determined by extrapolating its decrease to the time of TNP-AMP addition. In the *lower panel* in *C*, the low fluorescence was replotted on the expanded scale. The maximum fluorescence intensities at saturating 4 μ M TNP-AMP were obtained by subtracting the background level without TNP-AMP and the level of 4 μ M TNP-AMP without the SR vesicles, and given as the relative values in Table 1.

 Ca^{2+} (see Fig. 8 and Table 1). Also, inclusion of 5 mm Ca^{2+} without Mg^{2+} in the $E1\text{Ca}_2\text{·BeF}_x$ formation mixture abolished the superfluorescence (Table 1). The results agree with the above findings that $E1Ca_2$ ·BeF_{*x*} is not produced from and decomposed to $E1Ca₂$ by $Ca²⁺$ ligation at the catalytic $Mg²⁺$ site (site I).

Transmembrane Domain Structure—The 12 tryptophan residues among 13 in the $Ca^{2+}-ATP$ ase are located at the transmembrane region. The tryptophan fluorescence changes in fact reflect the transmembrane domain structural changes, *i.e.*rearrangements of the transmembrane helices upon Ca^{2+} binding to the high affinity transport sites and during the transport

cycle (28, 51, 52) as found originally by Dupont and Leigh (53). As summarized in Table 1 with typical fluorescence traces in [supplemental](http://www.jbc.org/cgi/content/full/M109.029702/DC1) [Fig. S3,](http://www.jbc.org/cgi/content/full/M109.029702/DC1) the fluorescence changes were determined at 4 °C upon formation of the $E1PCa_2$ analogs by the addition of fluoride compounds to *E*1Ca₂ in 15 m_M Mg^{2+} and 100 μ _M $\bar{C}a^{2+}$. $E1Ca_{2}$ ·BeF_{*x*} formation decreased fluorescence by 1.3% very similar to the decrease in $E1PCa_2$ •Mg formation from $E1Ca_2$ by MgATP, *i.e.* in $E1Ca_2$ ·MgATP \rightarrow *E*1PCa₂·Mg (52). In contrast, *E*1Ca₂· AlF*^x* formation did not cause any change. The $E1Ca₂$ AlF₄ ADP formation increased the fluorescence by 0.8%. $(F⁻$ alone and ADP alone did not cause any change, except the dilution (F^{-}) and absorption of excitation light (ADP).) Thus the transmembrane structure of $E1Ca₂BeF_x$ mimics that of $E1PCa₂$. Mg, but those of $E1Ca₂$ AlF₄ ADP and $E1Ca_2$ ·AlF_{*x*} differ substantially although the Ca^{2+} ions are occluded at the transport sites (or less strongly in $E1Ca₂·AlF_x$, Fig. 4). This observation is consistent with the finding in proteolysis and TNP-AMP superfluorescence that organization of the cytoplasmic domains and structure at the catalytic site in $E1Ca₂·BeF_x$ substantially differ from those in $E1Ca_2$ **·**AlF^{$\frac{1}{4}$ **·**ADP and $E1Ca_2$ **·**AlF_{*x*}.} It is concluded that the transmembrane structure with the occluded Ca^{2+} adopts not simply one state, but changes with the change in the cytoplasmic region during phosphoryl transfer and ADP release (see the diagram of tryptophan fluorescence change in [supplemental](http://www.jbc.org/cgi/content/full/M109.029702/DC1) [Fig. S3](http://www.jbc.org/cgi/content/full/M109.029702/DC1)*E* (with Ref. 54)).

Upon formation of *E*1Ca₂.

CaAMP-PCP and $E1PCa_2$ ·Ca·AMP-PN by CaAMP-PCP and CaAMP-PNP, respectively, the fluorescence increased by 0.8% equal to that upon $E1Ca₂$ AlF₄ ADP formation (Table 1), in agreement with their essentially identical structures with occluded Ca^{2+} (17, 18, 22, 45). By contrast, the fluorescence did not change upon formation of the *E*1Ca₂·MgAMP-PCP, which is the Ca^{2+} -unoccluded state (28, 45), and in rapid equilibrium with $E1Ca₂$.

Upon the exclusive accumulation of $E1PCa_2$ ·Ca by CaATP without Mg^{2+} , tryptophan fluorescence decreased by 0.9%, being slightly less than that by formation of $E1PCa_2 \cdot Mg$ and $E1Ca_2$ · BeF_x (Table 1). Thus in the overall structure,

FIGURE 6. **ADP causes the loss of Ca²⁺ occlusion in** *E***1Ca₂·BeF_x.** *E***1Ca₂·BeF_x
or** *E***1Ca₂ were produced in 50** μ **m ⁴⁵Ca²⁺ and 15 mm Mg²⁺ as described in the** legend to Fig. 4, then subjected to membrane filtration without and with perfusion for 20 s by the same buffer containing 50 μ M non-radioactive Ca² (in place of $45Ca^{2+}$) with and without 1 mm ADP. The amounts of $45Ca^{2+}$ specifically bound and occluded in the Ca^{2+} -ATPase were determined by subtracting the nonspecific background level (2.03 \pm 0.04 nmol/mg) obtained with the above perfusion without ADP of the *E*1Ca₂ state. The error bars show the S. D. of three independent experiments.

 $E1PCa_2$ ·Ca may be between $E1Ca_2$ ·CaAMP-PCP and $E1PCa_2$ ·Mg $(E1Ca_2$ ·BeF_{*x*}), and closer to the latter state. Although Ca²⁺ ligation at catalytic Mg²⁺ site I in *E*1PCa₂·Ca favors the N-P domain closed state, similar to $E1\text{Ca}_2\text{-}\text{CaAMP-}$ PCP, the absence of the N-P domain cross-linking nucleotide in $E1PCa_2$ ·Ca likely altered the overall structure slightly.

Upon formation of $E2$ BeF₃ from $E2$ by BeF_x and Mg^{2+} without Ca^{2+} , the fluorescence increased by 0.7%, mimicking the change upon $E2\text{P·Mg}$ formation from $E2$ with P_i and Mg^{2+} , and reflecting the opening of the lumenal gate from the closed state (25). As a consequence, the fluorescence of $E1Ca₂·BeF_x$ was definitely higher by \sim 1.3% than that of $E2$ Be F_3^- , showing their distinct difference in the transmembrane structure. In agreement, the previous kinetic analysis have shown (28) that tryptophan fluorescence decreases by \sim 1% in the isomerization/ Ca^{2+} release, $E1PCa_2$ Mg $\rightarrow E2P$ Mg + $2Ca^{2+}$, reflecting the transmembrane structural change from the Ca^{2+} -occluded state to the Ca^{2+} -released and lumenally opened state.

Upon the addition of thapsigargin (TG) to $E1Ca_2 \cdot BeF_x$ and $E2\text{\cdot}BeF_3^-$, tryptophan fluorescence decreased rapidly by 5.4 and 4.6%, respectively, and reached the level of $E2$ Be F_3^- with bound TG (E2·BeF₃ (TG), see Table 1). TNP-AMP superfluorescence [\(supplemental Fig. S5,](http://www.jbc.org/cgi/content/full/M109.029702/DC1) *A* and *B*) and proteolysis (Table 1) also demonstrated that $E1Ca_2$ ·Be F_x was converted by TG to $E2$ ·Be $F_3^-(TG)$. Importantly, as described under [supplemental](http://www.jbc.org/cgi/content/full/M109.029702/DC1) [Fig. S4,](http://www.jbc.org/cgi/content/full/M109.029702/DC1) two Ca^{2+} occluded in $E1Ca_2$ ·BeF_x are most likely released into the lumen by the TG-induced structural perturbation and trapped in the lumen by the bound TG, as TG fixes the lumenal gate in the closed state and suppresses Ca^{2+} leakage (16, 55).

 $E1Ca_{3}$ · BeF_{x} *Is ADP-sensitive—*In Fig. 6, two ⁴⁵Ca²⁺ occluded in $E1Ca_2$ ·BeF_{*x*} were rapidly removed by washing with 1 mm ADP, whereas the occluded $45Ca^{2+}$ remained completely without ADP. Thus ADP caused the loss of Ca^{2+} occlusion. In agreement, ADP binding to $E1\text{Ca}_2\text{-BeF}_{x}$ increased tryptophan

FIGURE 7. Conversion of $E1Ca₂$: BeF_x to $E2$: BeF₃ with $Ca²⁺$ release. The complex *E*1Ca₂·BeF_x was produced at 25 °C in 50 μ M ⁴⁵Ca²⁺ and 15 mm MgCl₂ as described in the legend to Fig. 4. The complex was then further incubated in the presence of these ligands at 25 and $\overline{4}^{\circ}$ C for various periods, and the bound^{'45}Ca²⁺ was determined with and without EGTA washing as described in the legend to Fig. 4A. In the control, the bound 45 Ca²⁺ in \tilde{E} 1Ca₂ without BeF*^x* was determined.

fluorescence to the $E1Ca₂$ level, and resulted in the tryptic T2 site cleavage as *E*1Ca₂ with bound ADP (data not shown). By contrast, ADP binding to $E2$ ·Be F_3^- did not alter its structure (data not shown). The ADP-induced decomposition of $E1\text{Ca}_2\text{·BeF}_x$ to $E1\text{Ca}_2$ was also demonstrated with the ADPinduced loss of TNP-AMP superfluorescence, in contrast to normal superfluorescence development in $E2$ ·BeF⁻₃ after ADP incubation (data not shown). Thus $E1Ca_2$ ·BeF_x is ADP-sensitive as $E1PCa_2$ Mg, and $E2$ Be F_3^- is ADP-insensitive as $E2P$ Mg.

Conversion of E1Ca₂</sub>. BeF_x</sub> to E2·BeF₃ at 50 μ *M Ca²⁺ - In Fig.* 7, $E1\text{Ca}_2\text{-}\text{BeF}_x$ was first formed in SR vesicles with BeF $_x$ at 25 °C in 50 μ M Ca²⁺ and 15 mM Mg²⁺, then further incubated at 25 and 4 °C in the presence of these ligands. The amount of bound and occluded Ca²⁺ was lost slowly ($t_{1/2} = \sim 2$ h at 25 °C and \sim 7 h at 4 °C). TNP-AMP superfluorescence (Fig. 8) and tryptic and proteinase K proteolyses (data not shown) revealed that $E1Ca₂$ BeF_x turned to $E2$ BeF₃ with $Ca²⁺$ loss. Thus $E1Ca₂$ BeF_x proceeded its spontaneous slow conversion to $E2 \cdot \text{BeF}_3^-$, as the autoisomerization of $E1PCa_2$ Mg to $E2P$ Mg. The Ca²⁺ ions released into the lumen may leak out during such long periods. In $E1Ca_2$ AlF_x and $E1Ca_2$ AlF₄ ADP, the amount of bound (occluded) Ca^{2+} was not decreased during the above 10-h incubation at 25 °C (data not shown). The proteolysis showed that these complexes were not converted to the Ca^{2+} -released forms, $E2$ **·AlF**₄^{$-$} ($E2$ **·AlF**_x $)$ with and without ADP (data not shown). The results indicate that the product state $E1PCa_2 \cdot Mg$ in the phosphoryl transfer acquires the structure ready for autoisomerization to *E*2P·Mg releasing Ca²⁺, whereas the transition state structure is not yet fully prepared for autoisomerization to the Ca^{2+} -released *E*2P form. Interestingly, as described in [supplemental Figs. S4 and S5](http://www.jbc.org/cgi/content/full/M109.029702/DC1) (with Refs. 56 and 57), the conversion $E1Ca₂$ BeF_x $\rightarrow E2$ BeF₃ was markedly accelerated by the transmembrane structural perturbation with hydrophobic reagents such as A23187, lasalocid, and $C_{12}E_8$, as

FIGURE 8. **Stability of E1Ca₂·BeF_x in various Ca²⁺ concentrations.** A, E1Ca₂·BeF_x was first produced in SR vesicles in 0.1 mm Ca²⁺ and 15 mm MgCl₂ with BeF_x. Subsequently, Ca²⁺ was changed to 0.1 (unchanged), 0.4, 0.7, 1.1, 3.1, and 10.1 mm, and the incubations continued at 25 °C for 72 h. At the indicated periods, the superfluorescence with 4 μ m TNP-AMP was examined at 25 °C, and the maximum levels obtained as described in the legend to Fig. 6 are shown. E2·BeF3
formed without Ca²⁺, E1Ca₂ in 0.1–10 mm Ca²⁺, and E2 without incubated with and without 1.2 μ m A23187 (A23), then BeF_x was added (*E1Ca₂·BeF_x*). *E*1Ca₂·BeF_x (*DOC-E1Ca₂·BeF_x*) was also produced from *E*1Ca₂ in 0.7 mm Ca^{2 +} of the Ca²⁺-ATPase purified and delipidated from SR vesicles by deoxycholate (DOC) treatment (30). The incubation was continued for 12 days, otherwise as indicated in A. E2 BeF₃ (E2 BeF₃, DOC-E2 BeF₃) without Ca²⁺ and E1Ca₂ in 0.7 mm Ca²⁺ (E1Ca₂, DOC-E1Ca₂) and E2 without Ca²⁺ (E2, DOC-E2) were also incubated. C–G, the fluorescence traces upon TNP-AMP addition were shown for the representative samples with incubation periods and Ca²⁺ concentration (mM). Note that at 0.7 mm Ca²⁺, both with and without A23187, the development of the *E*1Ca₂·BeF_x characteristic transient superfluorescence remained perfectly the same for 12 days. By contrast, the transient superfluorescence was converted to the stable and higher superfluorescence characteristic of *E*2·BeF₃ at 0.1 and 0.4 mm Ca²⁺, and it was markedly reduced by 3 and 10 mm Ca²⁺ due to decomposition to *E*1Ca₂.

well as TG. In contrast, $E1Ca₂$ AlF_x and $E1Ca₂$ AlF₄ \cdot ADP were resistant against these reagents.

E1Ca2-*BeFx Is Perfectly Stabilized at 0.7 mM Ca2*—As found here, Ca^{2+} binding at high affinity transport sites in $E1Ca₂$ is obligatorily required for $E1\text{Ca}_2\text{·BeF}_x$ formation, whereas millimolar high Ca²⁺ (Ca²⁺ ligation at the catalytic Mg²⁺ site I) decomposes this complex to $E1Ca₂$. Furthermore, $E1Ca₂$ -BeF_{*x*} at 50 μ M Ca²⁺ is spontaneously and slowly converted to $E2$ **·BeF**_{*x*} releasing Ca^{2+} , and the conversion is markedly accelerated by transmembrane perturbation with hydrophobic reagents such as $C_{12}E_8$ and A23187 (see [supplemental materi](http://www.jbc.org/cgi/content/full/M109.029702/DC1)[als\)](http://www.jbc.org/cgi/content/full/M109.029702/DC1). The results showed that $E1Ca_2$ ·Be F_x as the $E1PCa_2$ ·Mg analog possesses the structure prepared for its isomerization to $E2 \cdot BeF_3^-$ with Ca^{2+} release as $E1PCa_2 \cdot Mg \rightarrow E2P \cdot Mg + 2Ca^{2+}$. On the other hand, it is essential for crystallographic studies to find conditions to perfectly stabilize the $E1\text{Ca}_2\text{-}\text{BeF}_x$ complex. In Fig. 8*A*, $E1Ca_2$ BeF_{*x*} was first formed in 0.1 mm Ca^{2+} and 15 mM Mg²⁺, then further incubated with various concentrations of Ca^{2+} with and without A23187. The structural state was monitored by TNP-AMP superfluorescence (Fig. 8), proteolysis, and tryptophan fluorescence (see Table 1 for representative data). Then we successfully found that Ca^{2+} at a very narrow concentration range, 0.7 mm, perfectly stabilizes $E1Ca_2$ ·BeF_{*x*} and maintains this complex for at least 12 days at 25 °C (and 4 °C) even in the presence of A23187. The $45Ca^{2+}$ binding measurements on $E1\text{Ca}_2\text{·BeF}_x$ in 0.7 m_M ⁴⁵Ca²⁺ demonstrated that two Ca^{2+} ions are bound and occluded in this complex (Fig. 9*A*).

The perfectly stable $E1\text{Ca}_2\text{·BeF}_x$ was produced from $E1\text{Ca}_2$ even in the presence of A23187 if 0.7 mm Ca^{2+} was included before BeF*^x* addition (Fig. 8, *B* and *F*, and Table 1). Also, $E1\text{Ca}_2\text{·BeF}_x$ was successfully produced with the Ca^{2+} -ATPase purified from SR vesicles by delipidation with deoxycholate (30); in this case again, by including 0.7 mm Ca^{2+} before BeF_{*x*} addition. $E1Ca₂BeF_x$ thus produced with the purified and delipidated $Ca^{2+}-ATP$ ase was perfectly stable at least for 12 days at 4 and 25 °C in 0.7 mm Ca^{2+} (Fig. 8, *B* and *G*, at 25 °C).

 $E1Ca₂BeF₃⁻$ *Is Produced from E2·BeF₃ by Lumenal Ca*²⁺ $\mathit{ Binding}$ —We successfully found also that $E1\mathrm{Ca}_2\text{·BeF}_x$ $(E1\text{Ca}_2\text{·BeF}_3^-)$ can be produced from $E2\text{·BeF}_3^-$ by lumenal Ca^{2+} binding, as mimicking the lumenal Ca^{2+} -induced reverse transition, $E2P \cdot Mg + 2Ca^{2+} \rightarrow E1PCa_2 \cdot Mg$. In Fig. 10, we added

FIGURE 9. **Two Ca2 are bound in***E***1Ca2**-**BeFx formed from***E***1Ca2 and from** $E2 \cdot BEF_3^-$ at 0.7 mm Ca²⁺. A, SR vesicles in 0.7 mm 45 Ca²⁺ were incubated with BeF_x in the absence or presence of 5 µM A23187, otherwise as described in the
legend to Fig. 4. The amount of bound Ca²⁺ was obtained with subtraction of the background level (10.1 \pm 0.5 nmol/mg ($n = 6$)) determined by EGTA washing the vesicles incubated without BeF_x and A23187. The occluded Ca² was determined by EGTA washing in the absence of A23187 and by subtraction of the background level. (Here the determination of occluded Ca^{2+} in A23187 by EGTA washing was not feasible, because in the absence of (or even in 0.1 mm) Ca²⁺, A23187 converts E1Ca₂·BeF_x very rapidly to E2·BeF₃ releasing
Ca²⁺ [\(supplemental Figs. 4 and 5\)](http://www.jbc.org/cgi/content/full/M109.029702/DC1). B, E2·BeF₃ was first produced in SR vesicles
without A23187 and Ca²⁺. Subsequently, the samples with the buffer containing 45 CaCl₂ and BeF_x with and without 5 μ M A23187, to give 0.7 mm ⁴⁵Ca²⁺ and the same buffer conditions as in A. At 15 s after dilution, the amount of bound $45Ca^{2+}$ was determined without EGTA washing and by subtracting the nonspecific Ca²⁺ binding (1.0 \pm 0.1 nmol/mg (*n* = 6)) determined by EGTA washing the sample incubated without BeF*^x* and A23187.

various concentrations of Ca^{2+} to $E2$ BeF₃ formed in SR vesicles in 15 mm Mg^{2+} without Ca^{2+} in the presence and absence of A23187, then at 10 s after Ca^{2+} addition the structural state was examined by TNP-AMP superfluorescence. With increasing Ca^{2+} to 0.7 mm in the presence of A23187, the stable superfluorescence of $E2$ ^{-BeF₃</sub> was converted to the transient and} slightly lower superfluorescence characteristic of $E1Ca_2$ ·BeF_{*x*} with $K_{0.5}$ of 0.4 mm Ca²⁺ and a Hill coefficient of 4 (Fig. 10, *A* and D). A further Ca²⁺ increase in the millimolar range caused the marked loss of superfluorescence with $K_{0.5}$ of 1.7 mm and a Hill coefficient of 1 (Fig. 10, *B* and *D*). The proteolysis also clearly showed that $E2$ BeF₃ was converted to $E1Ca_2$ BeF_x by 0.7 mm Ca^{2+} in A23187 (Table 1), and this complex was further decomposed to $E1Ca₂$ by 10 mm $Ca²⁺$ (data not shown). In Fig. 9*B*, two ⁴⁵Ca²⁺ were shown to be bound producing $E1\text{Ca}_2\text{·}$ BeF_{*x*}, when 0.7 m_M ⁴⁵Ca²⁺ was added to $E2$ BeF₃ in the presence of A23187. In contrast, in the absence of A23187, $E2 \cdot BeF_3^-$ was neither converted to $E1Ca_2$ ·BeF_{*x*} nor decomposed to $E1Ca_2$ even at 10 mm Ca^{2+} (Figs. 9*B* and 10, *C* and *E*, and Table 1 (proteolysis at 0.7 mm Ca^{2+})).

The results demonstrated that $E1Ca_2$ ·BeF_{*x*}, most probably $E1\text{Ca}_2\text{:} \text{BeF}_3^-$, was produced from $E2\text{:} \text{BeF}_3^-$ by the lumenal Ca^{2+} binding at the lumenally oriented low affinity transport sites, and further that Ca^{2+} substitution of Mg^{2+} at the catalytic site in $E1\text{Ca}_2\text{·BeF}_3^-$ produced from $E2\text{·BeF}_3^-$ caused its decomposition to $E1Ca₂$, therefore as the change $E2 \cdot BeF₃⁻ + 2Ca²⁺ \rightarrow$ $E1\text{Ca}_2 \cdot \text{BeF}_3^- \rightarrow E1\text{Ca}_2$. Note that Mg^{2+} bound at the catalytic site in *E*2P-Mg is occluded, whereas it is not and therefore is

Structural Analog of E1PCa2-*Mg Intermediate of Ca2-ATPase*

exchangeable in $E1PCa_2$ ·Mg (42). Thus, these two distinctly different states of the ligated Mg^{2+} at the catalytic site (site I) in *E*2P·Mg and *E*1PCa₂·Mg are obviously mimicked here by the respective analogs $E2$ Be F_3^- and $E1$ Ca₂ Be F_3^- . The perfect stabilization of $E1\text{Ca}_2\text{·BeF}_3^-$ achieved by 0.7 mm Ca^{2+} (Fig. 10) obviously involves lumenal Ca^{2+} binding and prevention of the Ca^{2+} release into the lumen. The stabilization by 0.7 mm Ca^{2+} in the absence of A23187 is probably due to Ca^{2+} moved passively into the vesicle lumen during the long incubation periods. All these findings show that the forward and reverse transition, $E1PCa_2$ ·Mg \leftrightarrow $E2P$ ·Mg + $2Ca^{2+}$, is mimicked by the forward and reverse conversion, $E1Ca_2$ $BeF_3^- \leftrightarrow E2$ $BeF_3^- + 2Ca^{2+}$.

It is of interest to note the Hill coefficient of 4 in the lumenal Ca^{2+} -induced reverse conversion, $E2 \cdot BeF_3^- + 2Ca^{2+} \rightarrow$ $E1\text{Ca}_2\text{·BeF}_3^-$ at 0.1–0.7 mm Ca^{2+} in Fig. 10*A*. This might be indicative of the existence of lumenal Ca^{2+} access sites in addition to transport sites and their possible cooperative involvement in lumenal Ca^{2+} access to the transport sites. In fact, two such sites besides the two transport sites have been suggested by the kinetics and protein-chemical study on the lumenal loops (58, 59).

DISCUSSION

Formation of $E1Ca₂$ **:** $BeF₃$ - As a structural analog of the physiological intermediate $E1PCa_2$ ·Mg, the $E1Ca_2$ ·BeF_{*x*} complex was successfully produced by BeF_x binding to the $E1Ca₂$ state Ca^{2+} -ATPase and from $E2$ ·BeF₃ by lumenal Ca^{2+} binding to the lumenally oriented low affinity transport sites. All the revealed properties of $E1\text{Ca}_2\text{-}\text{BeF}_x$ met the requirements for the $E1PCa_2$ ·Mg analog; *i.e.* two Ca^{2+} occluded at the transport sites, Mg^{2+} bound (but not occluded) at the catalytic site, the ADP-released but still ADP-sensitive state, and its isomerization to the ADP-insensitive Ca²⁺-released state E2P·Mg $(E2 \cdot \text{BeF}_3^-)$ and reversal by lumenal Ca^{2+} binding, $E1PCa_2 \cdot \text{Mg}$ \leftrightarrow *E*2P·Mg + 2Ca²⁺.

Furthermore, the coordination chemistry of beryllium fluoride, actually BeF_{3}^{-} , fulfills the requirement of $E1\text{Ca}_{2}$ Be F_{x} as the E1PCa₂·Mg analog. In chemistry, beryllium fluoride compounds are known to adopt tetrahedral geometry with the Be-F 1.55-Å bond length, thereby making them strictly isomorphous to the tetrahedral phosphate group (60). Moreover, because of the high charge density due to the small size, beryllium is able to coordinate the aspartate-oxygen in addition to F^- . The -O- BeF_3^- thus produced with Asp^{351} -oxygen in fact possesses the tetrahedral geometry superimposable with the covalently bound phosphate at the aspartate, as actually seen in $E2$ Be F_3^- , the $E2P$ ·Mg ground-state analog (21, 22, 25). Mg F_4^{2-} also possesses the tetrahedral geometry, but magnesium is not able to be coordinated directly with the Asp³⁵¹-oxygen, as seen in $E2 \cdot MgF_4^{2-}$, the $E2 \cdot P_i$ analog. AIF_4^- in $E1Ca_2 \cdot AIF_4^- \cdot ADP$ and $E2$ **·**AlF^{$-$} (17, 18, 20) (or AlF₃ in some cases in the haloacid dehalogenase superfamily (61)) possesses planar geometry, in which Asp³⁵¹-oxygen and ADP β -phosphate or the hydrolytic water $(E2 \cdot AIF₄⁻)$ coordinate the aluminum at apical positions producing the bipyramidal structure superimposable to the penta-coordinated phosphorus in the transition state of in-line phosphoryl transfer $E1PCa_2$ ·ADP·Mg[‡] and acylphosphate hydrolysis *E*2-P·Mg[‡]. Thus all chemical properties of the P_i ana-

FIGURE 10. Formation of E1Ca₂ BeF₃ from E2 BeF₃ by lumenal Ca²⁺ binding. E2 BeF₃ was first produced in SR vesicles with BeF_x in the presence and absence of 1.2 µм A23187 at 25 °C in 0.5 mм EGTA, 15 mм MgCl₂, 0.1 м KCl, and 30 mм MOPS/Tris (pH 7.0). Then a small volume of CaCl₂ was added to give
various Ca²⁺ concentrations as indicated. At 10 s after this Ca intensity in the presence of A23187 was plotted in semi-log scale at 0–0.9 mm Ca²⁺ (A) and 0.7–15 mm Ca²⁺ (B). Note also the different scales in the *ordinates*. The *inset* in A is the linear plot at 0–0.9 mm Ca²⁺ to clearly show the saturation of the first phase of the Ca²⁺-dependent change, *i.e*. the formation of *E*1Ca₂·BeF₃ from *E*2·BeF₃ . In C, the maximum superfluorescence intensity in the absence of A23187 was plotted at 0–15 mm Ca²⁺ . In *D* and *E,* the traces of superfluorescence development upon TNP-AMP addition in the presence of A23187 (*D*) or its absence (*E*) were shown at the representative Ca²⁺ concentrations (0, 0.7 or 1.0, and 10 mm). *Solid lines* in A and B show the least squares fit to the Hill equation. Apparent Ca²⁺ affinity ($K_{0.5}$) and Hill coefficients at 0 – 0.9 mm Ca²⁺ are 0.43 mm and 3.8 with the intensity decrease from 170 to 142, and those at Ca²⁺ over 0.7 mm are 1.7 mm and 1.0 with a further intensity decrease to 36.

 \log s agree with the conclusion that $E1\text{Ca}_2\text{·BeF}_x$ is the analog for $E\overline{\text{PCa}_{2}}$ Mg, and BeF_{x} is most probably $\overline{\text{BeF}_{3}}$, *i.e.* $E1\overline{\text{Ca}_{2}}$ BeF_{3}^{-} .

Here note that the replacement of phosphate with BeF_{3}^{-} produces stabilization of the $E1PCa_2$ ⁻Mg structure with the same geometry of BeF_{3}^{-} as phosphate, and therefore probably with the same binding residues for them within the catalytic site. The $E1\text{Ca}_2\text{·BeF}_3^-$ stability is likely brought about by the specific chemical nature of fluoride. Namely, it possesses a significantly higher electronegativity than oxygen (actually the highest among all atoms) and a small size, therefore producing stronger BeF_{3}^{-} binding in the catalytic site and fixing the intermediate structure.

Structure of $EICa_2$: BeF_3^- – Then with the newly developed $E1\text{Ca}_2$ **BeF**₃, we explored its structural properties and uncovered the hitherto unknown nature of the physiological intermediate *E*1PCa₂·Mg and structural changes during the phosphoryl transfer/ADP release and subsequent *E*P isomerization/ Ca^{2+} release. The observed proteinase K resistance of Thr^{242} on the A/M3-linker revealed that, in *E*1PCa₂·Mg (*E*1Ca₂·BeF₃⁻) the A domain is already rotated perpendicular to the membrane plane from the position in *E*1Ca₂, thereby bringing up its junction with the A/M3-linker and imposing a strain on this linker, similarly to $E1Ca_2$ AlF₄ ADP and $E1Ca_2$ CaAMP-PCP (17, 18). As described for the $E1Ca₂$ -CaAMP-PCP structure (18), the

A/M3-linker strain, *i.e.* the A domain perpendicular rotation is brought about by bending the P domain due to binding of the phosphate moiety and Ca^{2+} at the catalytic site (Mg²⁺-site I, Asp³⁵¹/Thr³⁵³/Asp⁷⁰³) on the P domain (see Figs. 4 and 5 in Ref. 18).⁵ Our results revealed that such a strained state is achieved even without the N-P domain cross-linking nucleotide but solely with BeF_{3}^{-} and Mg^{2+} binding at the catalytic site, and therefore remains in $E1PCa_2 \cdot Mg$ after ADP release.

The strain of the A/M3-linker thus imposed has been predicted with the atomic structure (18, 19) to function as a motive force for the A domain rotation parallel to the membrane in the *E*1P to *E*2P isomerization. The partial resistance at T2 site $\rm Arg^{198}$ in $E1\rm Ca_2$ · $\rm BeF_3^-$ (as compared with the rapid cleavage in $E1Ca_2$ ·CaAMP-PCP/*E*1Ca₂·AlF₄·ADP and $E1Ca_2$ ·AlF_x) further indicated that in $E1PCa_2$ ·Mg, the A domain is already likely

⁵ As depicted in Figs. 4 and 5 by Toyoshima *et al*. (18) for the change $E1Ca₂ \rightarrow$ $E1Ca₂$ CaAMP-PCP, the top part of the first half of the P domain (P β 1–P β 4) moves together as a result of γ -phosphate and Ca²⁺ (Mg²⁺) binding, because Thr³⁵³ just above P β 1 coordinates to both ligands. Furthermore, P β 5 twists upon binding of Ca²⁺ (Mg²⁺) because of the coordination by Asp⁷⁰³, which causes P α 5–P α 7 tilting. Thus the P domain is bent. This bending causes the perpendicular A domain rotation because the P7 helix moves upwards and tilts so that Gly¹⁵⁶–Lys¹⁵⁸ on the A domain is brought up as they are in contact with Ala⁷²⁵–Val⁷²⁶ on top of the P7 helix.

rotated parallel to the membrane to some extent from the position in $E1Ca_2$ ·MgATP^{*}/*E*1PCa₂·ADP·Mg[‡]. Thus, the A/M3linker strain is likely functioning for this partial A domain rotation during the phosphoryl transfer/ADP release to produce $E1PCa_2$ ·Mg, and further for the large and complete rotation to achieve the tight A-P domain association at $Arg¹⁹⁸$ on the Val²⁰⁰ loop in the Ca²⁺-released state $E2P$ ·Mg ($E2$ ·BeF₃). The A-P domain interaction at the Val 200 loop is actually critical for formation of the proper Ca²⁺-released structure, E2P·Mg and its analog $E2$ ·BeF $_3^-$ (25, 62, 63).

Here, it is of interest to note that residues Asp³⁵¹, Thr³⁵³, and Asp⁷⁰³ ligating Mg²⁺ and phosphate will come more proximate to each other during $E1PCa_2$ ADP $Mg^+ \rightarrow E1PCa_2$ $Mg + ADP$ $(E1Ca₂·AIF₄·ADP/E1Ca₂·AIF_x → E1Ca₂·BeF_x$). As a consequence, a further P domain bending and more strain for the A/M3-linker will likely be induced by this coordination-chemical change, thereby contributing to inducing the A domain rotations during $E1PCa_2$ ·Mg formation and subsequent isomerization to *E*2PCa₂·Mg (besides the release of the N-P domain cross-linking nucleotide ADP). In any case, our results show that $E1PCa_2 \cdot Mg (E1Ca_2 \cdot BeF_3^-)$ as the product of the phosphorylation reaction acquires the structure ready for isomerization and Ca^{2+} deocclusion/release (*i.e.* ready for the large A domain rotation to produce $E2P \cdot Mg$ ($E2 \cdot BeF_3^-$)), whereas the transition state structure in the phosphorylation $(E1Ca₂·AIF₄·ADP$ and $E1Ca₂·AIF_x$) is not yet fully prepared. Note again that the *E*1PCa₂·Mg structure before such motions for its isomerization to *E*2P-Mg is stabilized with replacement of phosphate with BeF_{3}^{-} in $E1\mathrm{Ca}_{2}\text{·BeF}_{3}^{-}.$

Important also, we found that the Ca^{2+} -occluded transmembrane structure adopts not simply one state, but will proceed through changes during the phosphoryl transfer and ADP release to form *E*1PCa₂·Mg (see [supplemental Fig. S3](http://www.jbc.org/cgi/content/full/M109.029702/DC1)*E*). The structural change is probably coupled with the above described motions of the P and A domains (more bending and rotation) during this process. In the subsequent Ca^{2+} deocclusion/release in $E1PCa_2$ ·Mg $\rightarrow E2P$ ·Mg + $2Ca^{2+}$, the transmembrane structure changes further (52), which was also clearly mimicked here in the change $E1Ca₂$ $BeF₃⁻$ $\rightarrow E2$ $BeF₃⁻$. Thus, the structures of the transmembrane domain as well as the cytoplasmic domains in $E1Ca₂·BeF₃⁻$ ($E1PCa₂·Mg$) are intermediate between those of $\overline{E1Ca_2}$ AlF₄ ADP ($\overline{E1PCa_2}$ ADP Mg⁺) and $E2$ **·**BeF₃⁻ (*E*2P**·**Mg).

Mg2 as Physiological Catalytic Cation—Important questions regarding the $E1Ca₂$ BeF₃ structure are why $Ca²⁺$ coordination at the catalytic Mg^{2+} site (site I, Asp³⁵¹/Thr³⁵³/ Asp⁷⁰³) is absolutely unfavorable for $E1Ca₂$ ·BeF₃ formation, and why the Mg^{2+} -coordinated structure $\tilde{E1}Ca_2$ -BeF₃ differs from Ca²⁺-coordinated $E1PCa_2$ ·Ca, $E1Ca_2$ ·CaAMP-PCP, and $E1PCa_2$ ·Ca·AMP-PN structures as well as from $E1Ca_2$ · AlF₄ ADP/*E*1Ca₂ AlF_x, especially in A domain positioning. These questions may be relevant to the questions of why forward isomerization of $E1PCa_2$ ·Ca to $E2P$ is markedly retarded in contrast to $E1PCa_2$ Mg and thus why Mg²⁺ is preferred as the catalytic cation. In stringent coordination chemistry, the coordination distance of Mg^{2+} is shorter than Ca^{2+} , typically by 0.2 Å (*e.g.* 2.1 *versus* 2.3 Å (64, 65)). As a consequence, in the case of $E1\text{Ca}_2$ ·CaAMP-PCP, the distance between the γ -phosphate and Asp^{351} -oxygen becomes 3.24 Å, being greater by 0.3 Å than that predicted in $E1Ca₂$ MgAMP-PCP. Therefore MgAMP-PCP (MgATP) binding would result in steric clash, and *E*1Ca₂·CaAMP-PCP is more stable than *E*1Ca₂·MgAMP-PCP, and therefore has less tendency to decompose to $E1Ca₂$ (also in the forward direction to the *E*P formation and its decay in the case of $E1Ca_2$ ·CaATP) (45). In $E1Ca_2$ ·BeF₃ formed here with Mg^{2+} , the direct coordination between Asp^{351} and the beryllium and their proximate positioning would probably favor the closely positioned ligand residues $(Thr^{353}/Asp^{703}/$ Asp³⁵¹) for Be F_3^- and Mg²⁺ but not for Ca²⁺. Therefore Ca²⁺ substitution of Mg^{2+} probably disrupted the precise geometry and decomposed the $E1\text{Ca}_2\text{·BeF}_3^-$ complex. Also, a possible difference in the coordination number might be involved; Mg^{2+} prefers definitely six, whereas Ca^{2+} can accommodate seven or eight ligands (65-69).

Furthermore, the difference in A domain positioning between the Mg^{2+} -coordinated state $E1Ca_2$ -Be F_3^- and the $Ca²⁺$ -coordinated states may be reasonably understood by the consequence of the stringent coordination chemistry. Namely, because the shorter coordination distance of Mg^{2+} , P domain bending, and the resulting A domain rotation perpendicular to the membrane will be greater in the Mg^{2+} -coordinated state. Therefore the strain of the A/M3-linker and A domain rotation parallel to the membrane will be more in the Mg^{2+} state $E1Ca₂·BeF_x$. In this context, it is also reasonable that *E*1PCa2-Mg is more rapidly isomerized to *E*2P with less energy barrier for the large A domain rotation, in contrast to the retarded isomerization in *E*1PCa₂·Ca that is stabilized by the likely conformational inadequacy. Here note that the cause of the *E*1PCa₂·Ca stabilization is obviously different from that of $E1PCa_2$ Mg stabilization produced by replacement of phosphate with BeF_3^- (see the above discussion for $E1\text{Ca}_2\text{·BeF}_3^$ stabilization).

Previously it was documented (45, 64, 70) that destabilization of the non-covalent complex $E1\text{Ca}_2$ MgATP by Mg²⁺ (as found with MgAMP-PCP *versus* CaAMP-PCP) together with stabilization of the transition state by Mg^{2+} (as found with $E1\text{Ca}_2$ **·**AlF⁻₄ ·ADP bound Mg²⁺ at both sites I and II) leads to a decrease of the activation energy and a rapid phosphoryl transfer. As another critical reason for Mg^{2+} preference for catalysis, we predict here by exploring the property of the $E1Ca_2$ ·BeF_{*x*} that the Mg^{2+} bound at the catalytic site produces the proper $E1PCa₂$ structure, which is ready for rapid transition to $E2P$ in this rate-limiting process of the transport cycle.

Hydrophobic Catalytic Site in E1Ca₂.BeF3 — The microenvironment around Asp³⁵¹ in *E*1PCa₂·Mg was further predicted by TNP-AMP superfluorescence in $E1Ca₂BeF₃⁻$ to be strongly hydrophobic and thus a closed state, and this will become even more in the change $E1PCa_2$ ·Mg $\rightarrow E2P$ ·Mg + $2Ca^{2+}$ $(E1Ca₂·BeF₃⁻$ \rightarrow *E*2⁻BeF₃⁻). The observed distinct difference between $E1\rm Ca_2$ ·Be F_3^- and $E2$ ·Be F_3^- (transient *versus* stable and slightly higher superfluorescence) is probably ascribed to the distinct difference in the organization state of cytoplasmic domains. The superfluorescence, nevertheless, developed solely in the Ca^{2+} -ATPase complexed with beryllium fluoride, $E1\text{Ca}_2\text{·BeF}_3^-$ and $E2\text{·BeF}_3^-$ (*E*2P·Mg), but no development in $E1\text{Ca}_2$ AlF_{*x*}</sub> and in $E2$ AlF₄⁻ and $E2$ MgF₄²⁻ (25). Therefore, the

hydrophobic closed catalytic site is accomplished by the direct coordination and close proximity of the beryllium with $Asp³⁵¹$ oxygen and by the specific coordination of the tetrahedral -O- BeF_3^- , *i.e.* Asp³⁵¹-acylphosphate within the catalytic site. This is obviously not the case in $\mathop{\mathrm{AlF}}_4^-$ (the penta-coordinated phosphorus of the transition states) and MgF_4^{2-} (non-covalently bound P_i), thus in these states, the catalytic site is more accessible to nonspecific water molecules.

Whether $E1PCa_2$ ·Mg develops the superfluorescence had been controversial. In addition to the obvious problem of TNP-AMP competition against ATP for phosphorylation, the observed TNP-AMP-induced decomposition of $E1Ca₂BeF₃$ further revealed that the *E*1PCa2-Mg structure may be similarly disrupted rapidly by TNP-AMP binding, therefore making it virtually impossible to examine the superfluorescence development in *E*1PCa₂·Mg. The TNP-AMP-induced *E*1Ca₂·BeF₃ decomposition might have occurred by means of a similar structural change as the ADP-induced one, *i.e.* disruption of the cytoplasmic domain organization and possible BeF_{3}^{-} release. The most important conclusion here is that the hydrophobic and closed property of the phosphorylated catalytic site both in *E*1PCa2-Mg and *E*2P-Mg may be requisite to avoid a possible attack of nonspecific water molecules on the Asp^{351} -acylphosphate thus accomplishing Ca^{2+} release into the lumen and energy coupling.

Formation of E1Ca₂·BeF₃ from E2·BeF₃ and Perfect Stabilization of $E1Ca_2$: BeF_3^- - $E1Ca_2$: BeF_3^- was produced also from $E2 \cdot BeF_3^-$ by binding two lumenal Ca^{2+} to the lumenally oriented low affinity transport sites at 0.7 mm Ca^{2+} and 15 mm Mg^{2+} , as mimicking the reverse transition $E2\text{P·Mg} + 2\text{Ca}^{2+} \rightarrow$ $E1PCa_2$ ² Mg. At the critical concentration of 0.7 mm Ca²⁺ in 15 m_M Mg^{2+} , $E1Ca_2$ BeF_3^- is perfectly stabilized without decomposition to $E1\overline{Ca}_2$ or conversion to $E2\overline{e}F_3$. The perfect $E1\text{Ca}_2\text{·BeF}_3^-$ stabilization is obviously achieved by preventing $Ca²⁺$ release into the lumen and by avoiding the absolutely unfavorable Ca²⁺ substitution of Mg²⁺ in site I at the most appropriately balanced concentrations of Ca^{2+} and Mg^{2+} . As noted in the last paragraph under "Results," stabilization of $E1\text{Ca}_2\text{·BeF}_3^-$ might possibly involve lumenal Ca^{2+} access at the putative lumenal gating sites besides the transport sites. If this is the case, the gate-opening and Ca^{2+} release into the lumen takes place when the lumenal Ca^{2+} is low enough to avoid the possible lumenal Ca^{2+} access to the gate.

Integrated Picture of EP Processing—Recently, we successfully identified and trapped for the first time the intermediate state $E2PCa_2$ Mg, ADP-insensitive EP with two Ca^{2+} occluded at transport sites, by elongating the $A/M1'$ -linker (71), and revealed that the proper length of this linker is critical for inducing structural changes for Ca^{2+} deocclusion and release from *E*2PCa₂·Mg. This dependence on the length of the linker is probably because the length controls the extent of strain between the A domain and M1', which causes motions of the cytoplasmic A and P domains thereby transmitting the structural signal to the transmembrane transport sites. In trapped *E*2PCa₂·Mg, the A domain is already largely rotated, and A-P domain associations at Val²⁰⁰ and TGES¹⁸⁴ loops are already produced, although the interaction network is not produced properly at the Tyr¹²²-hydrophobic cluster (71), which is critical for Ca^{2+} deocclusion/release and *E*2P hydrolysis (72–74). In the Ca²⁺-released *E*2P·Mg, this cluster is formed from seven residues of the A (Ile¹⁷⁹/Leu¹⁸⁰/Ile²³²) and P (Val⁷⁰⁵/Val⁷²⁶) domains and the top part of M2 (Leu¹¹⁹/Tyr¹²²) (see Fig. 2).

The results indicated that the successive structural changes take place as follows: in $E1PCa_2 \cdot Mg \rightarrow E2PCa_2 \cdot Mg$, the A domain rotates largely (further from the position in $E1PCa_2$ ·Mg) into the space between the N and P domains and docks onto the Asp³⁵¹-acylphosphate of the P domain, thereby causing loss of ADP sensitivity and also the strain of the A/M1 linker (because the A domain is brought above the P domain). The strain thus imposed will cause inclinations of the A and P domains and the connected M2 and M4/M5 thereby rearranging the helices to destroy Ca^{2+} sites and open the lumenal gate thus to release Ca^{2+} . Upon these motions, the Tyr¹²²-hydrophobic cluster is produced from the inclined A and P domains and M2. Hence, interactions at this cluster and at the Val²⁰⁰ loop stabilize the Ca²⁺-released structure E2P·Mg, and also produce the catalytic site for the acylphosphate hydrolysis to occur after Ca²⁺ release, ensuring energy coupling (63, 72-74). Atomic level structural studies of $E1Ca₂$ ·BeF₃ as $E1PCa₂$ ·Mg and the trapped intermediate state *E*2PCa₂·Mg will contribute to further understanding of E P processing, Ca^{2+} handling, and *E*2P hydrolysis.

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