

Structure, Mechanistic Action, and Essential Residues of a GH-64 Enzyme, Laminaripentaose-producing β -1,3-Glucanase*

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Laminaripentaose-producing β -1,3-glucanase (LPHase), a member of glycoside hydrolase family 64, cleaves a long-chain polysaccharide β -1,3-glucan into specific pentasaccharide oligomers. The crystal structure of LPHase from *Streptomyces matensis* DIC-108 was solved to 1.62 Å resolution using multiple-wavelength anomalous dispersion methods. The LPHase structure reveals a novel crescent-like fold; it consists of a barrel domain and a mixed (α/β) domain, forming a wide-open groove between the two domains. The liganded crystal structure was also solved to 1.80 Å, showing limited conformational changes. Within the wide groove, a laminaritetraose molecule is found to sit in an electronegatively charged central region and is proximal to several conserved residues including two carboxylates (Glu¹⁵⁴ and Asp¹⁷⁰) and four other sugar-binding residues (Thr¹⁵⁶, Asn¹⁵⁸, Trp¹⁶³, and Thr¹⁶⁷). Molecular modeling using a laminarihexaose as a substrate suggests roles for Glu¹⁵⁴ and Asp¹⁷⁰ as acid and base catalysts, respectively, whereas the side chains of Thr¹⁵⁶, Asn¹⁵⁸, and Trp¹⁶³ demarcate subsite +5. Site-directed mutagenesis of Glu¹⁵⁴ and Asp¹⁷⁰ confirms that both carboxylates are essential for catalysis. Together, our results suggest that LPHase uses a direct displacement mechanism involving Glu¹⁵⁴ and Asp¹⁷⁰ to cleave a β -1,3-glucan into specific α -pentasaccharide oligomers.

Glycoside hydrolases (GHs,³ EC 3.2.1.x) hydrolyze the glycosidic bond between two or more carbohydrates or between a carbohydrate and non-carbohydrate moiety (1). These enzymes play diverse roles in nature; they breakdown cellulose into smaller carbohydrates (*i.e.* during biomass degradation by cellulases), they function during pathogenesis such as the activity of influenza virus neuraminidase (2), and they

are engaged in normal cellular metabolic processes that involve the formation and breakage of glycosidic bonds along with glycosyl transferase (3). GHs can be classified as exo- or endo-type of glycoside hydrolases that catalyze the hydrolysis of the glycosidic bond from the end or at the middle, respectively, of a polysaccharide chain. GHs can also be classified as the inverting or the retaining enzymes with respect to their distinct stereochemical mechanisms during catalysis (3). Sequence-based classification of GHs into various families has been proposed by Henrissat *et al.* (4, 5). Additionally, numerous structures of GHs have revealed details of their catalytic mechanisms as well as the basis for their diverse substrate specificity (6). Based on sequence comparisons and structural analyses, the carbohydrate-active enzymes data base (CAZy) provides continuously updated information on the GH families (7).

Two key residues among GHs, generally found as carboxylates, are involved in the hydrolysis of the glycosidic bond: a proton donor and a nucleophile/base (3). In either the retaining or the inverting enzymes, the position of the proton donor is found within hydrogen-bonding distance of the glycosidic oxygen. Active sites that consist of the key residues have been classified into three topologies by Davies and Henrissat (1): (i) a pocket or a crater that preferentially recognizes a saccharide molecule with a non-reducing end, presenting the exo-type hydrolysis, (ii) a cleft or groove that accommodates a large substrate for endo-type cleavage, and (iii) a tunnel that enables a polysaccharide chain to be threaded through for efficient endo-hydrolysis processivity.

Among the current 114 families of GHs, β -1,3-glucanases, namely exo- β -1,3-glucanases (E.C. 3.2.1.58) and endo- β -1,3-glucanases (E.C. 3.2.1.39) that degrade β -1,3-glucans into smaller biological response modifiers (8) are found in seven families. GH-3 and GH-5 are found to be exo-type; GH-16 and GH-17 are in the endo-type category. Both endo- and exo-type enzymes are found in GH-55. However, GH-64 and GH-81 remain unclear (7). Three-dimensional structures of members from GH-3, GH-5, GH-16, GH-17, and GH-55 have been solved, providing detailed structure-activity information (9–13). Members of the GH-5 and GH-17 families contain a (β/α)₈ architecture, whereas GH-16 family members fold as a β -jelly roll. Barley β -D-glucan exohydrolase, a member of the GH-3 family has an N-terminal TIM-barrel domain and a C-terminal 6-stranded β -sandwich. Notably, glucanases from

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³ The abbreviations used are: GH, glycoside hydrolase; LPHase, laminaripentaose-producing β -1,3-glucanase; PEG, polyethyleneglycol; MES, 4-morpholineethanesulfonic acid.

these families are retaining enzymes. The newly resolved structure of Lam55A, an inverting enzyme in GH-55 family (13), has two β -helical domains, separated by a long linker region, that form a ribcage-like structure. To date, no structure has been reported for GH-64 and GH-81, and thus detailed information on the mode of catalysis is lacking.

Laminaripentaose-producing β -1,3-glucanase (LPHase) cleaves a long-chain polysaccharide, β -1,3-glucan, including laminarin, into a specific pentasaccharide oligomer "laminaripentaose" (14, 15). Of interest, β -1,3-glucans such as laminarin, which constitute the cell walls of plants and fungi have interesting biological roles in immune modulation (8, 16, 17). Biochemical characterization of LPHase from *Streptomyces matensis* DIC-108 showed that the enzyme belongs to the GH-64 family and is an inverting enzyme (14, 15). This enzyme is unique that it releases mainly laminaripentaose as the end product, as compared with that exo-type enzymes produce much smaller sugars (monosaccharides or disaccharides) (18–20) while endo-type enzymes yield heterogeneous forms of oligosaccharides. This atypical product specificity, to our knowledge, has not been reported for other glucanases. Here we report the three-dimensional structures for the apo and complex LPHase of *S. matensis*. Structural analysis, modeling, and mutagenesis results revealed a novel crescent-fold structure containing Glu¹⁵⁴ and Asp¹⁷⁰ involved in the cleavage of a long-chain oligosaccharide from the reducing end.

EXPERIMENTAL PROCEDURES

Expression and Purification of LPHase—The artificial LPHase gene without the sequence of signal peptide (the first 35 residues) was reconstructed by PCR using a set of primers (sequences not shown). The amplified DNA fragment was inserted into pRSET A (designated pRSET_ *lph*) and expressed in *Escherichia coli* BL21 (DE3). The resulting protein had an extra methionine residue at the N terminus. After overnight induction by isopropyl-1-thio- β -D-galactopyranoside, cells (from 1 liter of culture) were harvested and resuspended in 20 ml of acetate buffer (20 mM, pH 5.8). Cells were then disrupted by sonication for 10 min with 30 s for each interval. After centrifugation to remove the cell debris, the supernatant (20 ml) was loaded onto a cation exchange HiTrap SP column (2.6 \times 40 cm, Amersham Biosciences) that was pre-equilibrated with the acetate buffer, pH 5.8. Column was eluted by a linear gradient of 0–1 M NaCl with 10 mM/min at a flow rate of 3 ml/min. The active fractions eluted in the range of 100–250 mM NaCl were collected and concentrated 20-fold before loading onto an anion exchange HiTrap Q column (1.6 \times 20 cm, Amersham Biosciences), which was equilibrated with phosphate buffer (20 mM, pH 6.8). LPHase was eluted by a linear gradient of 0–1 M NaCl with 5 mM/min at a flow rate of 1.5 ml/min. The active fractions eluted in the range of 200–350 mM NaCl were collected and concentrated 10-fold. The finally step of purification was to load 5 ml of crude protein (derived from the previous step) onto high performance CM ion exchange column (1.6 \times 20 cm, Amersham Biosciences) that was equilibrated with sodium acetate buffer (20 mM, pH 5.0). LPHase (in 20 mM acetate buffer, pH 5.0) was eluted by a linear pH gradient with phosphate buffer (0.1 M, pH 7.2) at a flow rate of 1.5 ml/min.

The active fractions were eluted at the range of pH 5.5–6.5. The eluted protein was concentrated for 25-fold and kept in 4 °C for further study. The purity of LPHase was verified by SDS-PAGE analysis. Expression of selenomethionine-labeled LPHase (Se-Met-LPHase) was done essentially described by Yang *et al.* (29) and Van Duyne *et al.* (30) and purified with the same protocol described above. The protein sample was concentrated to 20 mg/ml in a solution containing 40 mM sodium acetate (pH 6.5) by centrifugal filter device, Amicon Ultra (10000 MWCO) (Millipore, Carrigtwohill, Co. Cork, Ireland).

Crystallization—Crystallization was performed by the hanging-drop vapor-diffusion method at 20 °C by mixing equal volumes of protein sample and reservoir solution. Initial crystallization conditions were screened using Crystal Screen I and II kits (Hampton Research) and Clear Strategy Screen I and II kits (Molecular Dimension). Crystals were initially observed from two conditions: (1) using 0.2 M ammonium sulfate, 30% polyethylene glycol (PEG) 8000, 0.1 M sodium cacodylate (pH 6.5), and (2) 0.2 M ammonium sulfate, 30% PEG monomethyl ether (MME) 5000, 0.1 M MES (pH 6.5). Optimized crystals grew in a modified solution containing 0.15 M ammonium sulfate, 30% PEG MME 5000, and 0.1 M MES (pH 6.5). The native crystal diffracted to 1.62 Å and belonged to space group $P2_12_12_1$ with the unit cell dimensions $a = 46.16$, $b = 60.68$, $c = 149.40$ Å. There was one molecule per asymmetric unit.

Crystals of the Se-Met-derivatized protein were obtained in a solution containing 0.1 M lithium sulfate, 25% PEG MME 2000, and 0.1 M Tris (pH 7.0) using Se-Met-LPHase (20 mg/ml) at 4 °C. The SeMet-LPHase crystal also belonged to space group $P2_12_12_1$ with unit cell dimensions $a = 46.27$, $b = 60.36$, $c = 150.13$, and one molecule per asymmetric unit.

The liganded LPHase crystals were obtained by the co-crystallization method in a solution containing 0.05 M ammonium sulfate, 20% PEG 8000, 0.1 M sodium cacodylate (pH 6.5), 20 mg/ml LPHase, and 20 mM laminaripentaose. The complex crystals were characterized as space group $P2_12_12_1$, with unit cell dimensions $a = 45.98$, $b = 60.35$, $c = 149.12$, and one molecule per asymmetric unit.

Construction of Mutants—Mutagenesis was performed by the QuikChange method (Stratagene Co.), with the pRSET_ *lph* vector as template and the following oligonucleotides as primers: E154Q, 5'-CTTCAACTGGTCCCAGTACACGCTCAACG-3' and 5'-CGTTGAGCGTGTACTGGACCAGTTGAAG-3'; E154D, 5'-CTTCAACTGGTCCGACTACACGCTCAACG-3' and 5'-CGTTGAGCGTGTAGTCGGACCAGTTGAAG-3'; D170N, 5'-CAGTACGCAGGTCAACATGTTCTCAGCTC-3' and 5'-GAGCTGAGAACATGTTGACCTGCGTACTG-3'; D170E, 5'-CAGTACGCAGGTTCGAGATGTTCTCAGCTC-3' and 5'-GAGCTGAGAACATCTCGACCTGCGTACTG-3' (underlining shows the location of the mutation). All mutations were confirmed by DNA sequencing. For those mutations creating significant losses in enzymatic activity, the entire mutated genes were sequenced to confirm that only the intended mutations had occurred.

Activity Assays—For preparation of gel-form curdlan, 5 g of curdlan was dissolved in 150 ml phosphate buffer (100 mM, pH 10.5) and stirred at room temperature for 30 min. The solution

Structure and Catalysis of LPHase

was neutralized by 0.1 N HCl. The precipitant, collected by centrifugation, was then resuspended in 100 ml of water and heated at 70 °C for 2 h. The resulting solution was subjected to centrifugation to obtain the precipitant, which was then resuspended in 200 ml of phosphate buffer (50 mM, pH 7.5) to form 2% (w/v) gel-form curdlan solution for subsequent assay. The activity of LPHase was analyzed by determining the amount of reducing ends of sugars using 3,5-dinitrosalicylic acid (DNS) method (31). In brief, an appropriate amount of enzyme (0.2 ml) was incubated with curdlan (0.3 ml, 2%, pH 7.5) at 37 °C for 2 h, followed by adding 0.5 ml of 3,5-dinitrosalicylic acid reagent to stop the reaction. The mixture was boiled for 15 min, chilled, and centrifuged to separate the precipitates. The resulting adducts were measured spectrophotometrically at 540 nm. The absorption coefficient of the resulting adducts was determined to be $780 \text{ M}^{-1}\cdot\text{cm}^{-1}$ when D-glucose was used as reducing sugar.

X-ray Data Collection and Processing—Prior to data collection, crystals were dipped into Fomblin[®] cryoprotectant oil for several seconds and then flash-frozen in a liquid nitrogen stream. All protein crystals were initially screened and characterized using an RU-300 rotating-anode x-ray generator (Rigaku/MSB Inc.) at the Macromolecular x-ray Crystallographic Laboratory of National Tsing Hua University, Taiwan. The 1.62 Å native dataset and the 1.80 Å liganded dataset were collected on the SPXF beamline BL13C1 and BL13B1, respectively, at the National Synchrotron Radiation Research Center in Taiwan using an ADSC Quantum 4R CCD detector. The multi-wavelength anomalous dispersion (MAD) data set was collected from a single Se-Met-LPHase crystal on the BL12B2 Taiwan beamline at SPring-8 in Japan using an ADSC Quantum 4R CCD detector. Data were collected at three wavelengths: peak (0.9794 Å), inflection of selenium K-edge (0.9796 Å), and remote (0.9642 Å). All data sets were indexed, integrated, and scaled using HKL-2000 (32). Data collection statistics are shown in Table 1.

Structure Determination and Refinement—The apo-LPHase structure was solved by the MAD method. Four selenium atom sites were located with the program PHENIX (33). PHENIX was further used for phasing, density modification, and automated model building. The final model included 98.6% of the residues. Five residues (Met³⁵, Ala³⁶, Asn³¹⁴, Asp³¹⁵, and Gln³¹⁶) were not built into the model because of the negative or broken density map. $2F_o - F_c$ and $F_o - F_c$ maps were produced and inspected manually using the program "O", version 9.0.7 (34). Crystallographic refinement was carried out using the maximum likelihood target function embedded in program REFMAC5 (35, 36). The apo-LPHase model (1.62 Å) was obtained using MOLREP (35, 37) and refined using REFMAC5 (35) coupled to ARP/wARP (38), which was used to add water molecules automatically.

The LPHase complex structure was determined by molecular replacement methods using the apo-LPHase structure as a search model by MOLREP and refined by REFMAC5 coupled to ARP/wARP. The apo and complex LPHase models were validated using the program PROCHECK (39). The final models of the apo and liganded LPHase had *R* values of 17.9% ($R_{\text{free}} =$

21.8%) and 16.6% ($R_{\text{free}} = 21.7\%$), respectively. Refinement statistics of the final models are given in Table 1.

Structural Comparisons and Modeling—The structure of LPHase was compared with all protein structures available in the DALI server. Structural comparisons with the allergenic thaumatin-like protein (PDB code: 1Z3Q; Ref. 23), and the sweet protein, thaumatin (PDB code: 1THW, Ref. 22), were carried out using the program LSQMAN in O (34) to superimpose C_α atoms. Combined sequence and secondary structure alignments and figure preparation were done with the program ESPript (40). Structural figures were prepared with the program PyMol. To build the LPHase-substrate complex, the substrate laminaripentaose was docked using the program GOLD, version 2.1 (CCDC Software Limited, Cambridge, UK) (41). The x-ray structure of the LPHase complex was chosen as the template for docking study. The binding pocket for the docking study was defined as a 10 Å radius sphere centered on the O ϵ 2 atom of Glu¹⁵⁴. The three glucose moieties from the reducing end of laminaripentaose were manually placed at the site surrounding by Glu¹⁵⁴, Thr¹⁵⁶, and Thr¹⁶⁷ according to the tetraose seen in the LPHase-tetraose structure, and the glycosidic oxygen was positioned between Glu¹⁵⁴ and Asp¹⁷⁰. The default parameters of GOLD were used. The three best solutions were obtained until a root mean square tolerance of 1.5 Å. This complex model was subjected to energy minimization using the Tripos force field (42) with the SYBYL 8.0 program (The Tripos Associates, St. Louis, MO).

RESULTS AND DISCUSSION

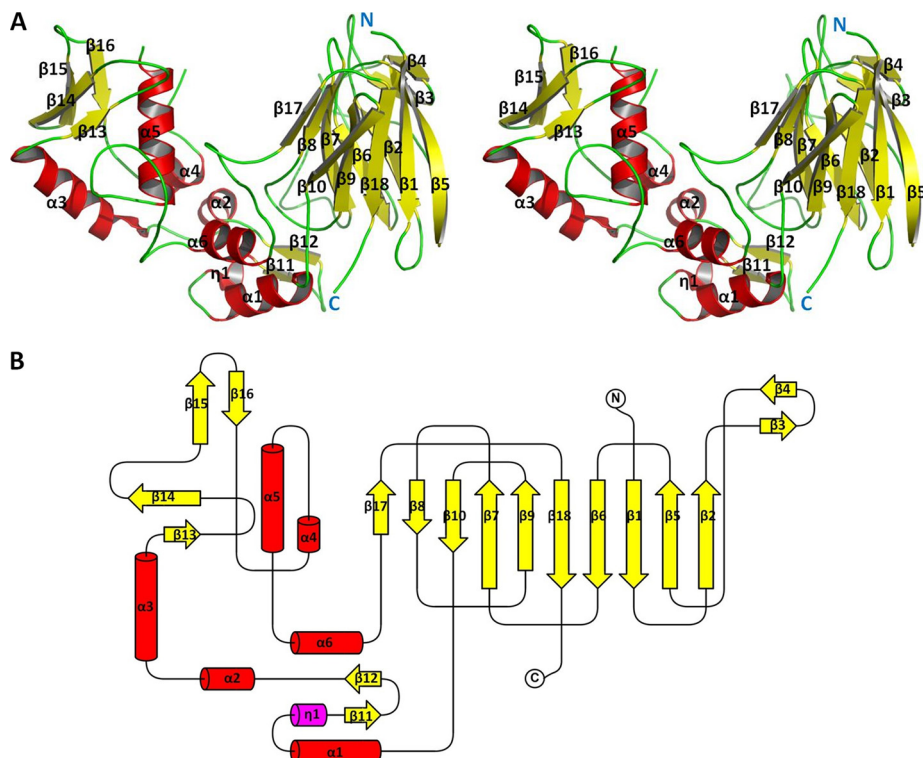
Structure Description—The 1.62 Å resolution electron-density map of apo LPHase reveals one molecule per asymmetric unit. The main-chain and side-chain atoms are well defined except for two segments having negative or weak density: the Met³⁵-Ala³⁶ segment and a loop region (residues 314–316). Only one residue, Arg³⁰⁸, lies in the disallowed region of the Ramachandran plot, despite its low B factor. The unusual geometry of this region is stabilized by three hydrogen bonds: Arg³⁰⁸ O \cdots N Leu³¹⁰, Arg³⁰⁸ O \cdots N Asp³¹¹, and Arg³⁰⁸ O \cdots O δ 1 Asp³¹¹.

The final model was refined to an *R* value of 17.9% ($R_{\text{free}} = 21.8\%$). The mean B-factor of protein atoms is 17.1 Å² (Table 1). The apo-LPHase crystal structure is composed of two domains, a barrel domain and a mixed (α/β) domain (Fig. 1, A and B). The barrel domain starts from the N terminus and consists of 10 consecutive β -strands (β 1- β 10). Two additional strands from the C-terminal region (β 17- β 18) join up, together assembling into a barrel structure. The mixed (α/β) domain contains six helices packed inside. A β -sheet (β 13- β 16) covers one exterior face of the α/β domain, and a β 11 β 12 sheet occupies the other side of the α/β domain. Along with three helices (α 1, α 4, and α 6), this region interacts with residues from one end of the barrel domain, forming a crescent-like architecture.

Crystal Structure of LPHase Complex—To obtain liganded structures, extensive crystallization trials were attempted in the presence of various concentrations of laminaripentaose. The complex structure was only obtained by the co-crystallization method in a solution containing 20 mM laminaripentaose. The structure was solved by the molecular replacement method.

TABLE 1
X-ray data and refinement statistics

Data set	Native LPHase	SeMet-LPHase (MAD)			LPHase Complex
		Peak	Edge	High remote	
Beamline	NSRRC_BL13C1	JASRI_SPring-8_BL12B2			NSRRC_BL13B1
Space group	P2 ₁ 2 ₁ 2 ₁	P2 ₁ 2 ₁ 2 ₁	P2 ₁ 2 ₁ 2 ₁	P2 ₁ 2 ₁ 2 ₁	P2 ₁ 2 ₁ 2 ₁
Cell dimensions					
<i>a</i> (Å)	46.16	46.27	46.27	46.26	45.98
<i>b</i> (Å)	60.68	60.36	60.36	60.36	60.35
<i>c</i> (Å)	149.40	150.13	150.13	150.11	149.12
Wavelength (Å)	0.9762	0.9794	0.9796	0.9642	0.9795
Resolution (Å)	30.00–1.62	30.00–2.29	30.00–2.30	30.00–2.26	30.00–1.80
Highest resolution shell (Å)	1.68–1.62	2.37–2.29	2.38–2.30	2.34–2.26	1.86–1.80
Completeness (%) ^a	100.0 (100.0)	98.5 (95.4)	98.7 (98.1)	98.4 (96.2)	97.0 (89.2)
Average <i>I</i> / σ (<i>I</i>) ^a	43.9 (8.4)	36.4 (34.0)	38.1 (35.1)	41.0 (34.1)	41.7 (19.5)
No. of unique reflections	54,410	19,889	19,770	20,788	39,457
Redundancy ^a	14.0 (12.3)	10.3 (9.2)	10.3 (9.3)	10.1 (9.0)	8.3 (8.2)
<i>R</i> _{merge} (%) ^{a,b}	6.0 (33.6)	5.1 (6.7)	4.9 (6.2)	4.8 (6.1)	4.0 (11.7)
Overall figure of merit ^c			0.87		
Solvent content (%)	42.2				42.1
Refinement					
Resolution range (Å)	30.00–1.62				30.00–1.80
Number of atoms					
Protein	2757				2757
Solvent	730				652
Average <i>B</i> -factor (Å ²)	17.1				18.7
<i>R</i> factor ^d	0.179				0.166
<i>R</i> _{free} ^e	0.218				0.217
R.m.s.d. bond lengths (Å) ^f	0.010				0.014
R.m.s.d. bond angles (°) ^f	1.323				1.185
Estimated coordinate error (Å)	0.053				0.071
Ramachandran analysis (%) ^g					
Favored/Allowed/Generous/Disallowed	89.2/10.4/0.0/0.3				89.6/9.7/0.3/0.3

^a Values in parentheses refer to statistics in the highest-resolution shell.^b $R_{\text{merge}} = \sum |I_{\text{obs}} - \langle I \rangle| / \sum I_{\text{obs}}$.^c Figure of merit = $|F_{\text{best}}| / |F|$.^d $r = \sum |F_{\text{obs}} - F_{\text{calc}}| / \sum F_{\text{obs}}$, where F_{obs} and F_{calc} are the observed and calculated structure-factor amplitudes, respectively.^e R_{free} was computed using 5% of the data assigned randomly.^f R.m.s.d., root mean square deviation.^g Estimated standard uncertainties based on maximum likelihood.**FIGURE 1. Structure of LPHase.** *A*, stereo view of the LPHase crystal structure. N and C termini are labeled in blue as N or C, and α -helices and β -sheets are labeled with α or β and numbered. *B*, topology of LPHase. β -sheets (yellow) and α -helices (red) are numbered.

Residues that could not be defined include Met³⁵, Ala³⁶, and a loop region (314–316). As seen in Fig. 2, the electron density map clearly shows a large piece of non-peptide electron density that consists of several rings inside the groove of the crescent shape LPHase. The final complex model was refined to an *R* value of 16.6% ($R_{\text{free}} = 21.7\%$) (Table 1). Only four glucoses (laminaritetraose) that are linked by three β -1,3 glycosidic bonds could be built into this density. The absence of the density for the fifth glucose moiety in the pentose may be due to its flexibility in this region. The other possibility is that laminaripentaose was slowly cleaved by LPHase during the crystallization process. In support of this view, we found that LPHase had slight residual activity toward laminaripentaose, which produced tetraose in an overnight reaction (data not shown).

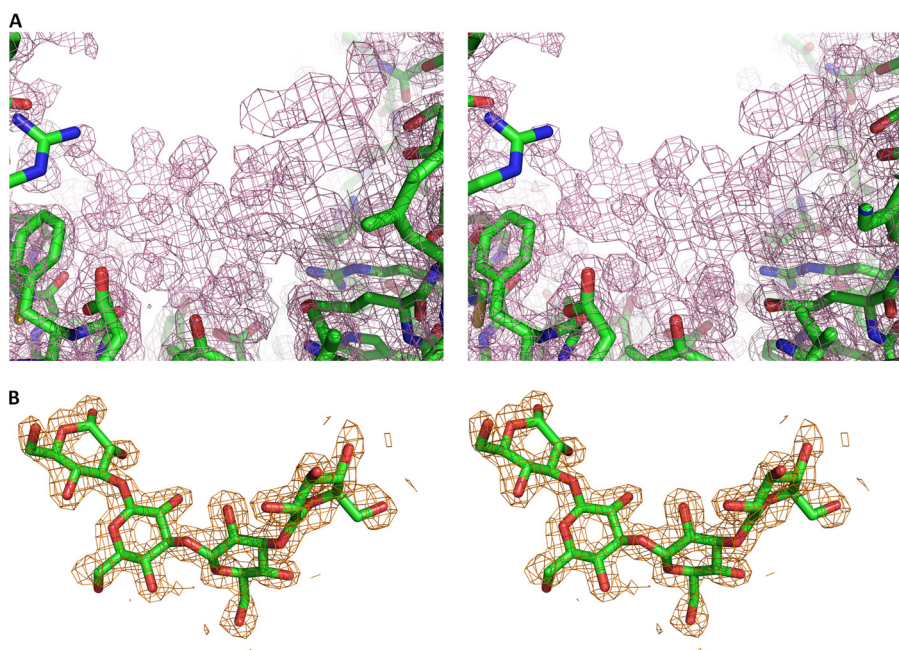


FIGURE 2. **The electron density map of LPHase-complex structural models.** *A*, stereo view of the initial $F_o - F_c$ electron density map of the groove region, contoured at the $1.0\text{-}\sigma$ level. *B*, stereo view of the $2 F_o - F_c$ electron density map, contoured at the $1.0\text{-}\sigma$ level, shows the bound laminaritetraose. Carbon, oxygen, and nitrogen atoms are colored green, red, and blue, respectively.

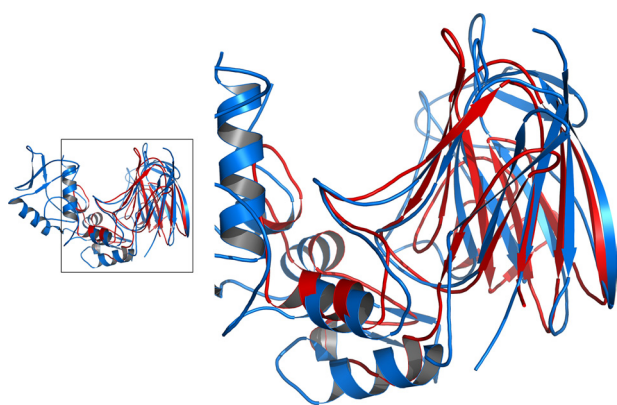


FIGURE 3. **Structural comparison of LPHase with thaumatin.** The structures of LPHase and thaumatin-like protein (PDB code: 1Z3Q) are superimposed and plotted in blue, and red, respectively. The framed region is magnified in the right panel.

The apo and liganded LPHases share the two-domain crescent architecture. Superposition of the structures reveals a root mean square deviation of 0.24 \AA for the C_α atoms, demonstrating an overall identical conformation. Based on analysis with DALI (21), there is no strong structural homology with any GH family member. Only structures of thaumatin (22) or thaumatin-like protein (23) show similarity with a partial region of LPHase, particularly the barrel domain (Fig. 3). These proteins are noted to present the surface with an electronegative cleft, implicating a possible evolutionary relationship.

The Binding Site—Based on the complex structure, laminaritetraose is situated at a negatively charged cleft of the wide groove between the β -barrel and mixed α/β domains (Fig. 4A). A number of water molecules are observed with or without laminaritetraose within the groove. The first three glucoses (Glc^1 - Glc^2 - Glc^3) at the reducing end of laminaritetraose inter-

act with a few residues (Glu^{154} , Thr^{156} , Asn^{158} , Trp^{163} , Asn^{165} , Thr^{167} , and Val^{169}) of the barrel domain (Fig. 4B). Glu^{154} and Thr^{156} from $\beta 7$ as well as Thr^{167} from the $\beta 8$ - $\beta 9$ loop form hydrogen bonds with O4 or O6 of Glc^1 , Glc^2 , and Glc^3 ($\leq 3.6\text{ \AA}$) ($\text{Glc}^1\text{ O4}\cdots\text{O}\gamma 1\text{ Thr}^{156}$, $\text{Glc}^2\text{ O4}\cdots\text{O}\epsilon 1\text{ Glu}^{154}$, $\text{Glc}^2\text{ O6}\cdots\text{O Glu}^{154}$, $\text{Glc}^3\text{ O6}\cdots\text{O}\epsilon 1\text{ Glu}^{154}$, and $\text{Glc}^3\text{ O6}\cdots\text{O}\gamma 1\text{ Thr}^{167}$). The outward side chain of Trp^{163} makes several contacts with Glc^1 including three polar interactions ($\text{Glc}^1\text{ O1}\cdots\text{N}\epsilon 1\text{ Trp}^{163}$: 4.30 \AA ; $\text{Glc}^1\text{ O}_2\cdots\text{N}\epsilon 1\text{ Trp}^{163}$: 4.38 \AA ; $\text{Glc}^1\text{ O5}\cdots\text{N}\epsilon 1\text{ Trp}^{163}$: 4.01 \AA). The side chain of Tyr^{371} from the mixed α/β domain also makes contacts with Glc^3 including a polar interaction ($\text{Glc}^3\text{ O5}\cdots\text{O Tyr}^{371}$: 4.23 \AA). Glc^4 , the last moiety, dangles from the barrel domain and faces toward the $\alpha 4$ region of the α/β domain.

The farthest O6 atom of Glc^4 forms a hydrogen bond with the main-chain O atoms of Gly^{305} and Cys^{306} , respectively ($\text{Glc}^4\text{ O6}\cdots\text{O Gly}^{305}$ and $\text{Glc}^4\text{ O6}\cdots\text{O Cys}^{306}$).

Two strictly conserved carboxylates, Glu^{154} and Asp^{170} lie at the center of the cleft and a little to the right of the barrel axis and are in the vicinity of laminaritetraose (Fig. 4B); as described above, Glu^{154} interacts extensively with laminaritetraose including three hydrogen bonds with Glc^2 (O4 and O6) and Glc^3 (O6), and Asp^{170} , despite being more distant from laminaritetraose ($\text{Glc}^3\text{ O6}\cdots\text{O}\delta 2\text{ Asp}^{170}$: 7.03 \AA ; $\text{Glc}^3\text{ O4}\cdots\text{O}\delta 1\text{ Asp}^{170}$: 6.69 \AA), is surrounded by several water molecules. Notably, two waters are situated between laminaritetraose and Asp^{170} ($\text{Asp}^{170}\text{ O}\delta 2\cdots\text{O Wat}^{236}$: 2.62 \AA ; $\text{Asp}^{170}\text{ O}\delta 1\cdots\text{O Wat}^{456}$: 2.40 \AA). Tyr^{232} that is nearby Asp^{170} ($\text{Tyr}^{232}\text{ O}\eta\cdots\text{O}\delta 1\text{ Asp}^{170}$: 3.77 \AA) also interacts with Wat^{456} ($\text{Tyr}^{232}\text{ O}\eta\cdots\text{O Wat}^{456}$: 2.93 \AA). Glu^{154} and Asp^{170} adopt very similar conformations in the apo and liganded forms, in which the distance of the side chains is in the range $5.8\text{--}8.2\text{ \AA}$, prospectively acting as the catalysts for this LPHase because inverting enzymes require a proton donor and a base, respectively (1).

In parallel with the structural analysis, site-directed mutagenesis studies of Glu^{154} and Asp^{170} were carried out. Enzymatic analysis on curdlan showed no detectable activity for E154Q and D170N, suggesting that the presence of the carboxyl moiety is crucial at these sites. We next evaluated the substitution with the other carboxylate for residues 154 and 170, respectively. E154D showed significantly lower relative activity (19%) as compared with the wild-type enzyme, while there was 81% relative activity for D170E. These results suggest the importance of these two carboxylates and that a shorter side chain greatly reduced the catalytic power even with the same carboxyl moiety at residue 154.

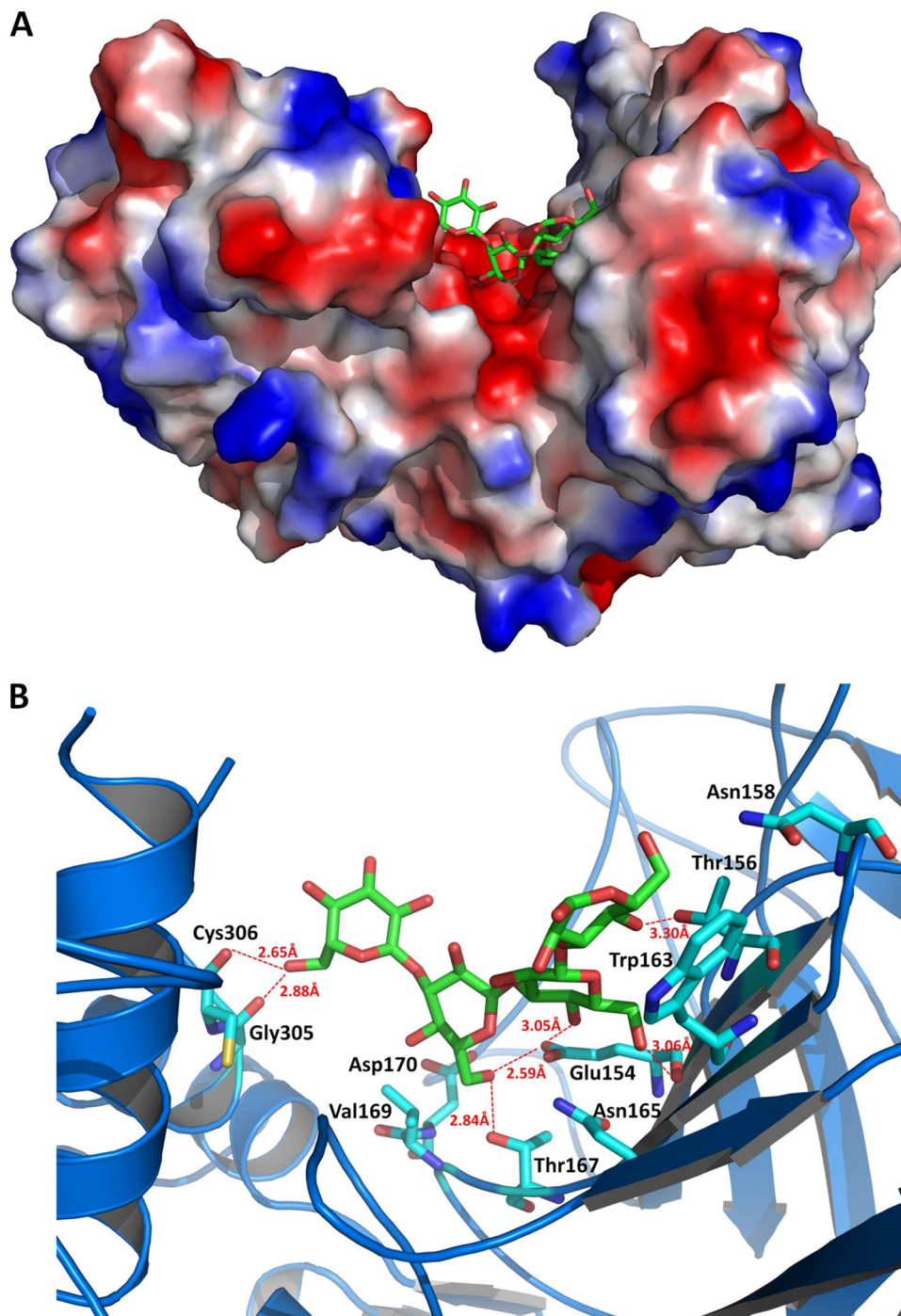


FIGURE 4. *A*, molecular surface of the LPHase-laminaritetraose complex. This figure was prepared with the program PyMol. *B*, binding pocket of the LPHase-laminaritetraose complex. The bound laminaritetraose is drawn as green sticks, whereas ligand-binding residues of LPHase are shown as cyan sticks. Oxygen, nitrogen, and sulfur atoms are colored red, blue, and yellow, respectively.

The LPHase-Laminarihexaose Model—The topologies of the active sites in the GH superfamily can be classified into three general categories: pocket, tunnel, and cleft (1, 24). The structure of LPHase shows a wide cleft topology, which is an “open” structural feature, allowing binding of polysaccharide substrates and presumably performing an endo-cleavage reaction.

To understand the catalytic mechanism, we performed a computer simulation to model an enzyme-laminarihexaose structure. The three glucose moieties from the reducing end of

laminarihexaose are placed at the site surrounded by Glu¹⁵⁴, Thr¹⁵⁶, and Thr¹⁶⁷ according to the sugar seen in the LPHase-tetraose structure. Assuming that Glu¹⁵⁴ and Asp¹⁷⁰ mediate catalysis, the glycosidic oxygen will be positioned between these two residues, in which protonation of the glycosidic oxygen by the catalytic acid residue is followed by the nucleophilic attack by water molecule, which is deprotonated and activated by the general base. This mechanism would ensure that hydrolysis yields an inverted laminaripentaose structure that can be released from the active site (1, 15). The best models we obtained had the following features (Fig. 5A): (i) only Glu¹⁵⁴ is within hydrogen-bonding distance of the glycosidic oxygen (1.94 Å), suggesting its role as the proton donor; (ii) Asp¹⁷⁰ is relatively more distant to the anomeric carbon (~6 Å), allowing the accommodation of a water molecule between Asp¹⁷⁰ and the sugar, thus ideally serving as the basic catalyst; (iii) the protruding side chains of Thr¹⁵⁶, Asn¹⁵⁸, and Trp¹⁶³ demarcate subsite +5; and (iv) Glu¹⁵⁴ and Thr¹⁶⁷ form hydrogen bonds with the substrate at subsites +4 and +3, respectively.

The Catalysis and Substrate Binding Topology—Our structural analyses and modeling suggest a catalytic mechanism for LPHase whereby Glu¹⁵⁴ acts as a proton donor to the glycosidic oxygen, and then Asp¹⁷⁰ facilitates a base-assisted nucleophilic attack by a water molecule from the opposite side of the sugar ring (Fig. 5B). When a long-chain polysaccharide β -1,3-glucan such as laminarin diffuses into the groove, the glucan is positioned with the reducing end at the +5 subsite, consisting of Thr¹⁵⁶, Asn¹⁵⁸, and Trp¹⁶³, and then is bound by the negatively charged region via interactions with several residues including Glu¹⁵⁴ and Thr¹⁶⁷. Hydrolysis of a β -glycosidic bond from the reducing end of sugar chain by Glu¹⁵⁴ and Asp¹⁷⁰ thus creates a laminaripentaose product with the α -anomeric configuration. By use of methyl laminarihexaoside and methyl laminariheptaoside as substrate, respectively, methyl laminaripentaoside was indeed detected after enzymatic digestion (data not shown), confirming that the hydrolysis of sugar chain is from the reduc-

Structure and Catalysis of LPHase

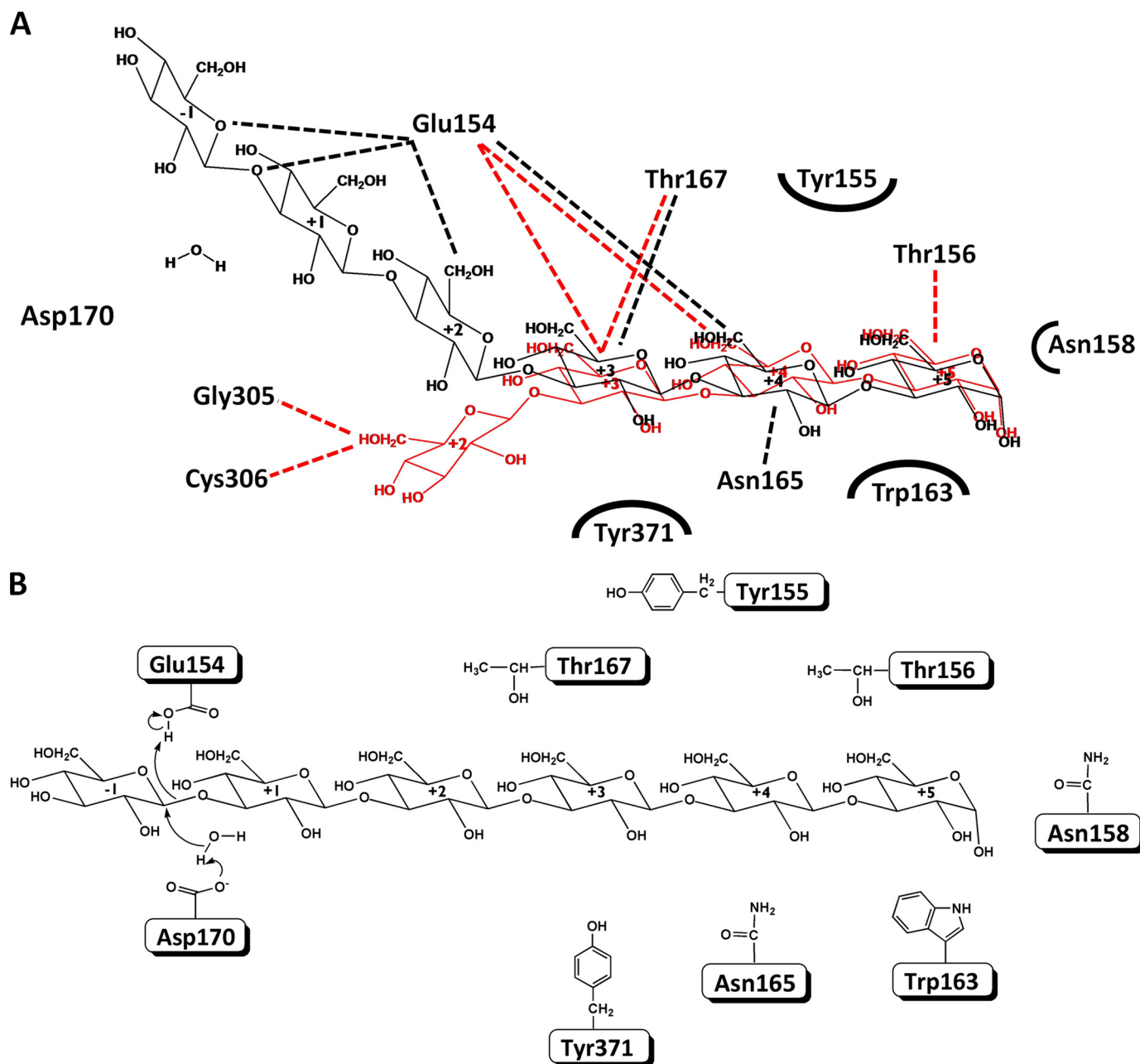


FIGURE 5. *A*, schematic representation of LPHase-laminarihexaose and LPHase-laminaritetraose interactions. The docked laminarihexaose (*black*) and laminaritetraose (*red*) are superimposed and drawn as *sticks*. The hydrogen-bonding interactions are shown as *broken lines*. Asp¹⁷⁰ that is relatively more distant to the anomeric carbon (~ 6 Å) of the docked laminarihexaose interacts with laminarihexaose/laminaritetraose via water molecules. *B*, scheme of the proposed catalysis of LPHase.

ing end. A striking feature of LPHase is its unique product specificity that predominantly releases pentaose as product throughout the enzymatic digestion of curdlan, exhibiting an exo-type cleavage manner. Yet, LPHase structure shows a wide groove, which is commonly observed among endo-type GHs. This puzzle remains unresolved even with the LPHase-tetraose complex structure. Early studies on curdlan revealed that the long-chain curdlan forms more complex tertiary structures due to intramolecular and intermolecular hydrogen bonding interactions; the simplest β -1,3-glucan (or curdlan) usually adopts a triple-stranded helical structure in nature (25–27). Treatment of the helical curdlan at high temperature or with alkaline conditions, it rendered single, double, or triple helical structures at

its gel form (26, 28). It is thus speculated that the higher-order structure of curdlan may play an important role in binding to LPHase, hence contributing to its product specificity during the catalysis. As shown in the LPHase-tetraose complex structure (Fig. 4) and the simulated laminarihexaose-containing structure (Fig. 5*A*), the sugar chains are observed to form a distinct kink to fit into the binding site, suggesting that the tertiary structure of curdlan may be a key factor for substrate recognition. The unusually wide groove in LPHase structure also discloses the possibility for a helical curdlan to bind with and consequently results in an exo-type cleavage.

In conclusion, we have determined the structure of LPHase in its apo and liganded forms. The LPHase structure consists of

a barrel domain and a mixed (α/β) domain. The most prominent structural feature of the crescent-like LPHase is a wide groove with a predominantly electronegative charge running between the two domains. Analysis of the apo and complex LPHase structures demonstrates a relatively rigid catalytic framework consisting of the $\beta 7$ - $\beta 9$ region in the barrel domain including two strictly conserved carboxylates (Glu¹⁵⁴ and Asp¹⁷⁰) and other saccharide-linked residues (Thr¹⁵⁶, Thr¹⁶⁷, Trp¹⁶³, Asn¹⁶⁵, and Val¹⁶⁹). A plausible direct displacement mechanism that involves Glu¹⁵⁴ and Asp¹⁷⁰ as catalysts is thus proposed to yield a pentaose product with the inverted anomeric configuration.

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