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## Nitrones as Therapeutics

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### Abstract

Nitrones have the general chemical formula X-CH=NO-Y. They were first used to trap free radicals in chemical systems and then subsequently in biochemical systems. More recently several nitrones including PBN (*α*-phenyl-*tert*-butylnitron) have been shown to have potent biological activity in many experimental animal models. Many diseases of aging including stroke, cancer development, Parkinson's disease and Alzheimer's disease are known to have enhanced levels of free radicals and oxidative stress. Some derivatives of PBN are significantly more potent than PBN and have undergone extensive commercial development in stroke. Recent research has shown that PBN-related nitrones also have anti-cancer activity in several experimental cancer models and have potential as therapeutics in some cancers. Also in recent observations nitrones have been shown to act synergistically in combination with antioxidants in the prevention of acute acoustic noise induced hearing loss. The mechanistic basis of the potent biological activity of PBN-related nitrones is not known. Even though PBN-related nitrones do decrease oxidative stress and oxidative damage, their potent biological anti-inflammatory activity and their ability to alter cellular signaling processes can not readily be explained by conventional notions of free radical trapping biochemistry. This review is focused on our observations and others where the use of selected nitrones as novel therapeutics have been evaluated in experimental models in the context of free radical biochemical and cellular processes considered important in pathologic conditions and age-related diseases.

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### Financial Disclosures

RAF has been active in the commercial development of NXY-059 in stroke. He's a cofounder of Centaur Pharmaceuticals, Inc. and then a consultant of Renovis, Inc. He and RAT are cofounders of Onconos, Inc. a start up company involved in cancer therapeutics.

## Keywords

Nitrones; Stroke; Glioma; Hepatocellular Carcinoma; Hearing Loss; Acoustical Trauma

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## Introduction

Scientific advancements regarding our understanding of free radicals in biological systems, their importance in disease processes and now the possibility of using novel therapeutics that act by influencing free radical processes has changed dramatically over the last several years. The possibility that free radicals actually may exist in biological systems was only taken seriously after radiation chemists had shown: A) that oxygen free radicals were formed by ionizing radiation and B) that ionizing radiation also causes cancer. These facts were established over 50 years ago and many discoveries since then have occurred which have made this area an important field of research. Important findings include the following demonstrations: A) that biological systems actively produce reactive oxygen species (ROS) and reactive nitric oxide species (RNS), B) that biological systems have enzymatic systems specifically to degrade these reactive species, C) that specific oxidation products are formed when biological molecules react with ROS and RNS, D) that cellular stress-related processes are communicated using ROS and RNS as signaling agents and E) that ROS and RNS are agents that play an important role in many normal processes as well as pathologic disease progression.

Following this over arching massive growth in our understanding of ROS and RNS in biological systems, an important research area in novel therapeutics has arisen. Namely that potential novel therapeutics have been discovered which act by altering the course of disease progression by acting upon ROS and RNS mediated processes. We have spent considerable effort in exploring this area mostly focused on the use of nitrones in age-related diseases. In this effort much has been learned but much more is yet to be discovered. This review is presented as an aid to enhance future effort in the pursuit of goals along this path of research.

## Nitrones – From Spin Traps in Free Radical Chemistry to Experimental Animals

The nitrone chemical structure in its simplest form can be represented as  $X-CH=NO-Y$ . Nitrones began to be used in analytical chemistry applications in the late 1960s. Nitrones will react with and “trap” and stabilize free radical intermediates (Figure 1). Many chemical reactions have free radical intermediates. Since many of these intermediates exist for only a brief time they are therefore very difficult to characterize and study. Chemists showed in 1967 that some free radicals will react with nitrones to produce nitroxide free radicals [1]. The nitrone trapping of a free radical intermediate is represented in the following simple reaction where the free radical intermediate ( $R\bullet$ ) is trapped by a nitrone:  $X-CH=NO-Y+R\bullet \rightarrow X-CHR-NO\bullet-Y$ . Janzen and Blackburn advanced the field significantly and utilized nitrones to trap free radicals and also coined the term “spin trap” for trapping free radical intermediates [2]. The reaction of the free radical species with a nitrone yields a product termed the spin-adduct (Figure 1). The nitroxyl free radical spin-adduct is usually much more stable than the free radical therefore making it possible in principle to characterize the original free radical trapped using electron paramagnetic resonance methods [3]. Shortly after beginning the use of nitrones to trap free radicals in analytical chemistry, scientists then began to use nitrones to trap free radicals in biochemical systems with reports first appearing in 1975 [4] and 1976 [5].

After the original early reports in the mid 1970s demonstrated the potential for the use of nitrone spin traps to detect and characterize free radical intermediates in subcellular biological systems,

increasing research was conducted using nitrones in biochemical research for the next 10–20 years. In this regard excellent research came from several major laboratories including, Mason's laboratory [6–10], Piette's laboratory [11–14], Kalyanaraman's laboratory [15–18], our laboratory [19–24], as well as many others. McCay's laboratory from the very early days began to focus more on the use of nitrones to trap free radicals in experimental animals. For example, trapped radicals were demonstrated in rats administered  $\text{CCl}_4$  [25,26], experimental animals given ethanol [27], as well as during exposure to gamma radiation [28]. Bolli, Janzen and McCay studied the free radicals produced in "stunned" hearts of rats [29–33] and McCay, Janzen and Bray made seminal observations on the pharmacokinetics and metabolism of the commonly used spin-trap PBN i.e.  $\alpha$ -phenyl-tert-butyl nitron [33–35].

### Early Experiments Testing Nitrones as Pharmaceutical Agents

It was in the context of the increasing use of nitrones to trap free radicals in experimental animals undergoing oxidative stress that the extrapolation to the experimental testing of nitrones as pharmaceutical agents began. We and others have reviewed the early phases of the use of nitrones as pharmaceutical agents previously [36–39]. The first early reports of the use of nitrones as potential pharmaceutical agents were those reported by Noveli in 1985 and 1986 [40–42] demonstrating that administered PBN protected rats from the trauma and death brought on by subjecting them to confinement in a rotating drum as well as the shock trauma brought on by LPS injection. The protective activity of PBN was replicated and extended by more rigorous experiments of septic shock using LPS by McKechnie et al [43] Hamburger and McCay [44] and Pogrebniak et al [45]. PBN was found to be protective in these models only if administered prior to the LPS challenge. It was soon after these observations (Dec, 1988) that Floyd and Carney made the first observations demonstrating that PBN had neuroprotective activity in the Mongolian gerbil stroke model [46–48]. This discovery arose from follow-up experiments where we had demonstrated using salicylate to trap hydroxyl free radicals that ischemia/reperfused gerbil brains produced reactive oxygen species [49]. PBN was then utilized in an attempt to trap putative secondary free radicals induced by hydroxyl free radical reactions with biological molecules. This approach failed because the secondary free radicals that may have been trapped to form the nitroxyl PBN spin adducts were rendered paramagnetically silent due to their reduction to the hydroxylamine products. Despite the failure to realize our experimental goals, serendipitous success was soon realized since the follow-up experiments showed that PBN was protective against stroke-induced brain injury even if it was administered up to 1 hr after the stroked brain was reperfused [48]. These results were soon reproduced [50] and extended [51,52].

### Observations Extending the Biological Activity of PBN

Following these early observations other laboratories as well as our own extended the neuroprotective activity of PBN to models involving excitotoxicity induced neuronal damage [53], toxin induced brain seizures [54,55], and bacterial meningitis caused by group B streptococci [56]. Hillered's group also extensively studied the effects of traumatic brain injury in animal models and demonstrated that PBN and the 2-sulfonyl-PBN derivative i.e. S-PBN have protective roles [57–62]. In addition to brain, PBN was shown to be protective in hearing loss in rats caused by the combined exposure to carbon monoxide and loud noise [63]. PBN was also shown to be effective in preventing streptozotocin-induced diabetes [64], thalidomide-induced birth defects [65], ischemia/reperfusion-induced acute renal failure [66] and in light-induced damage to eyes affecting vision [67–69]. Several studies have also suggested that PBN increased life-span in normal [70] and senescence-accelerated mice [71] and in rats [72] and also prevented age-related decline in behavioral parameters in rats [72] as well as gerbils [47,73]. Pertinent to the anti-aging effect of PBN, in rigorously controlled and replicated studies at three different experimental sites using a genetic model that simulates outbred wild mice, it

was demonstrated that the PBN metabolite 4-hydroxy-PBN administered in the diet from a young age did not cause an increase in life span [74,75].

### Wide-Spread Biological Activity of PBN – What General Mechanism is Involved?

The studies noted above as well as those not cited in this review clearly demonstrate that PBN-related nitrones have potent biological activity. In fact, it can be generalized that these compounds have shown activity in the majority of experimental models of either pathologic conditions or age-related diseases where they have been tested. New observations continue to be made. For example, recent observations include the demonstrations that: A) PBN ameliorates hippocampal injury and improved learning and memory in juvenile rats following neonatal exposure to LPS [76], B) PBN decreased the incidence of external malformations in a mouse model of LPS-induced teratogenesis [77] and C) PBN decreased the damage to striatal neurons and loss in motor skills in neonatal rats exposed to hypoxia-ischemia [78].

The wide-spread activity suggests that there is a universal mechanism that can explain the mechanistic-basis of the general activity of PBN. If this is true it has not been elucidated thus far. Since PBN was brought to the attention of scientists because of its ability to trap free radicals it is rational to assume that its biological action is due to its spin-trapping activity. This can not be completely ruled out, however the conventional notions of mass action solution chemistry spin-trapping is highly unlikely to explain the biological action of PBN and related compounds. This is because the general reaction rate of PBN with most free radicals is quite slow, i.e. usually about  $10^5$ – $10^7$  M<sup>-1</sup>sec<sup>-1</sup> [3]. This dictates, as has been demonstrated in solution chemistry trapping studies, that PBN then must be present at very high concentrations (i.e. 10–50 mM) in order for PBN to trap a significant fraction of the free radicals being produced. As we have argued previously [36,<sup>79,80</sup>], PBN is rarely higher than 10–50 μM and almost never more than 0.5 mM in the target tissue of systems where it has been shown to be active. Based on a solution chemistry based perspective, the low target tissue PBN levels strongly suggest that PBN would be capable of spin-trapping only a very small fraction of free radical species produced which strongly argues against this explanation to account for its biological action. If it however is taken into consideration that the interior of cells has physical properties of a gel rather than bulk water and that PBN may have specific strong affinities for specific proteins or other biological molecules then it is possible that PBN may spin-trap critical free radicals on the surface of these macromolecules which may be of importance in the biological processes investigated. PBN has been shown to trap protein thiyl free radicals [81] but it has not been proven that these or other protein radicals are of critical importance in biological processes where PBN has been shown to have a significant effect. In the case of the spin trap DMPO (5, 5-dimethyl-1-pyrroline N-oxide) new methodology has shown that many proteins retain a DMPO molecule attached covalently as revealed by mass spectroscopy [82–86] and by immunological methods [87–89]. Modifications of this new methodology may become available to help address the questions of the importance of PBN reacting with free radicals on biological macromolecules.

### PBN-Possible Mechanisms Other than Spin-Trapping

Mechanisms other than spin-trapping have been investigated to account for the biological activity of PBN and related compounds. In general PBN has been shown to have some activity and therefore influence many biological processes. As discussed earlier various notions other than spin-trapping to explain the mechanistic action of PBN have been considered. These include A) its antioxidant properties, B) its action on important membrane enzymes including ion transport proteins and C) its action as an anti-inflammatory agent. With regard to PBN as an antioxidant, at least two studies have demonstrated that it is a very poor antioxidant in simple systems [90,91]. In biological lipid peroxidation systems it was shown that PBN at ~5 mM was an effective antioxidant but the antioxidant BHT or γ-tocopherol was as active at about 5 μM

or less, i.e. they were about thousand-fold more effective than PBN [90]. Since PBN solubilizes into lipid bilayers it is not surprising that it may influence enzymes especially if the concentration is in the high range. In this regard PBN has been shown to block calcium channels that are important in pulmonary artery relaxation [92]. The ED<sub>50</sub> for PBN was shown to be 1.93 mM in this regard. PBN also has been shown to inhibit acetylcholinesterase activity with a K<sub>i</sub> of 0.58 mM [93]. It was shown that PBN has no influence on muscarinic or glutamate receptors though [94,<sup>95</sup>]. Many studies in our laboratory and Kotake's laboratory have demonstrated that PBN has general anti-inflammatory activity. The anti-inflammatory activity of PBN is manifested as the curtailment of inflammatory cytokine expression and the accompanying expression of genes associated with the action of proinflammatory cytokines. Examples of gene induction products which are in general suppressed by PBN include inducible nitric oxide synthase (iNOS) and COX-2. The induction of these genes have been shown to be curtailed by PBN administration in a wide range of biological models and systems [37,<sup>64,96-103</sup>]. The resultant curtailment of the production of enhanced nitric oxide (NO) and inflammatory prostaglandins is expected to have significant influence on biological parameters important in pathologic conditions and age-related diseases as has been noted in several neurodegenerative models [55,<sup>96,99</sup>] and in cancer development [80,<sup>104,105</sup>].

## Nitrones as Therapeutics for Stroke

### Basic Science Discoveries Leading to Commercial Development

The activity of PBN-related nitrones has been a subject of intense research and commercial development for over 20 years. As noted in the Introduction section, Floyd and Carney in 1988–1992 made the original discoveries in experimental stroke that led to most of this activity [46–48]. The fact that PBN administered up to 1 hr after an ischemia/reperfusion insult to the brain of Mongolian gerbils showed protective activity was surprising and clearly suggested possible therapeutic potential for the indication of acute ischemic stroke in humans. Centaur Pharmaceuticals beginning in 1993 aided by NIH SBIR grants helped implement much of the commercial research and development activity in this arena until 2000 when Renovis, Inc became the primary commercial company guiding this effort. Centaur partnered with Astra that later became AstraZeneca which funded all of the major development research and clinical studies from Phase I through Phase III stages from mid 1995 through 2007. The PBN derivative 2,4-disulfophenyl-*N-tert*-butylnitron, referred to as NXY-059, was the candidate drug that was taken through the many stages of clinical development for the acute ischemic stroke indication.

### Nitron Stroke Drug NXY-059 Preclinical Studies

Green et al provided a most extensive review of the nitron data (NXY-059, PBN, as well as 2-sulfophenyl-*N-tert*-butylnitron) in relation to the preclinical research on NXY-059 in stroke [106]. NXY-059 is a less competitive compound for trapping free radicals in vitro systems than PBN or S-PBN [107] It should be noted however that PBN is a poor antioxidant when compared to the well known antioxidants  $\alpha$ -tocopherol and butylated hydroxytoluene [90,<sup>91</sup>]. However as a protective agent in experimental stroke models, NXY-059 is much more effective than PBN in the amount of compound needed as well as the time window after the stroke when administration was started [106]. NXY-059 showed effectiveness when administered at a dose of 0.3 mg/kg/hr starting 1 hr after a 2 hr occlusion of the middle cerebral artery allowing regional brain reperfusion in a rat model. However PBN administered at a comparable dose rate showed no neuroprotective effect. In addition to the transient focal ischemia models in rats, NXY-059 was also shown to have significant neuroprotective activity in the permanent focal ischemia model in rat [106] as well as in the marmoset primate model [106,<sup>108,109</sup>]. In the rat middle cerebral artery permanent focal ischemia model, the degree of neuroprotection as judged by decreased infarct volume was directly proportional to the plasma free

concentration of NXY-059 in the 0–120  $\mu\text{mol/L}$  range [106]. In this model neuroprotection was significant even if NXY-059 was administered starting 4 hrs after middle cerebral artery ligation. Probably the most critical preclinical experiments were conducted in the marmoset model. This permanent focal middle cerebral ischemia model was studied when NXY-059 was administered either immediately after artery ligation [108] or starting at 4 hrs after artery ligation [110]. NXY-059 had significant neuroprotective activity in both studies as judged by behavior parameters as well as by prevention of brain death. Behavioral parameters obtained from both the hill and valley tests were used. In these tests the use of the lateral and contralateral arms was evaluated in trained animals at 3 and 10 weeks after artery ligation. Significant neuroprotection was afforded by NXY-059 in all the neuroprotection parameters measured. These critical experiments in addition to all the rodent studies conducted in several independent laboratories as well as the fact that NXY-059 had shown no significant toxicity in humans in several Phase I and Phase II clinical trials [106,<sup>111</sup>] were important in the evaluation conducted by AstraZeneca that led to proceeding into Phase III clinical trials.

### Phase I and II Clinical Trials

Much of the background development research leading up to Phase II clinical trials has been thoroughly summarized previously [39,<sup>106,108,109,111</sup>]. NXY-059 was shown to be a very safe drug in all the clinical trials [106,<sup>111–113</sup>]. If a stroke drug alters hemostasis parameters, this may have an adverse effect. NXY-059 was studied in 30 healthy adults where it was demonstrated that it had no effect on bleeding time or platelet aggregation or adhesion [112]. It was also shown that NXY-059 has no effect on the action of recombinant tissue-type plasminogen activation on the clearance of human thrombi in vitro [113]. Studies of NXY-059 pharmacokinetics administered intravenously in healthy young and elderly humans showed that the plasma elimination half-life was short, being 2–3 hrs, renal elimination was predominant, it was not reabsorbed by the kidney and the recovery of the unchanged drug in the urine was 80–90% of the total dose irrespective of age of the subjects [111]. A follow up study confirmed that renal elimination of NXY-059 was dominant and that there was no reabsorption by the drug. It was also demonstrated that NXY-059 excretion in the kidney was due to active elimination by the organic ion transporter [114]. The drug clearance from plasma did fit a 2 compartment model indicating that it did have some penetration into tissue. Studies in rats showed that there was some penetration in brain. Studies in sham operated rats as well as those receiving a permanent middle cerebral artery occlusion demonstrated that NXY-059 did penetrate into the brain cortex region of sham operated rats ( $6.26 \pm 1.26$  nmol/g) but it penetrated even less in the stroked rats in the region of the artery ligation ( $3.84 \pm 0.80$  nmol/g) than the region of the contralateral side ( $6.14 \pm 2.18$  nmol/g). This is expected if the main supply of blood is blocked from coming into the stroked region. Another study important to the penetration of NXY-059 into stroked brain was conducted by Dehouck et al [110]. In this study the penetration of PBN, S-PBN as well as NXY-059 was examined in an in vitro model of the blood-brain barrier. As expected, the uptake of S-PBN and NXY-059 was low but PBN readily penetrated. However, importantly the penetration rate of NXY-059 increased several-fold if the blood-brain barrier was subjected to hypoxia or ischemia [110].

### Phase III Clinical Trials

As noted earlier, the marmoset experiments plus the safety profile of NXY-059 contributed to the decision to begin Phase III clinical trials. Phase III clinical trials were conducted as a series of two major studies. The two trials were referred to as the Stroke-Acute Ischemic NXY Treatment (SAINT) trials or SAINT I and SAINT II trials. The SAINT I trial ran from May 2003 through November 2004 at 158 hospitals in 24 countries [115]. The patients were older than 18 yrs and had a clinical diagnosis of acute stroke that commenced within 6 hrs before entry into the study. It was required that the patients had limb weakness and had a score of at least 6 on the National Institutes of Health Stroke Scale (NIHSS). Patients were randomly

assigned to receive an i.v. infusion of NXY-059 or placebo within 6 hrs after the onset of stroke. The initial infusion rate was 2270 mg per hour and then reduced to 480 to 960 mg for a further 71 hours with the aim of maintaining a free NXY-059 plasma concentration of 260  $\mu$ mol per liter. Treatment outcome was based on a clinical assessment which included functional measures that were dependent primarily on the modified Rankin scale (range 0–5) and the Barthel index [115]. Therefore the SAINT I trial was a randomized, double-blind, placebo-controlled trial involving 1722 patients with acute ischemic stroke and randomly assigned to receive 72-hr infusion of placebo or intravenous NXY-059 beginning within 6 hrs after the onset of stroke. In the final outcome, 1699 subjects were included in the efficacy analysis. It was shown that NXY-059 significantly improved the overall distribution of scores on the modified Rankin scale, as compared with placebo ( $P = 0.038$ ) by the Cochran-Martel-Haenszel test [115]. The common odds ratio for improvement across all categories of the modified Rankin scale was 1.20 (95 percent confidence interval, 1.01 to 1.42). In contrast to the modified Rankin scale noted above, NXY-059 did not show significant improvement in neurological functioning as measured by the NIHSS test. There was no difference in mortality or adverse events between the treated and untreated group.

### Results of SAINT I Clinical Trial

The first report of the success of NXY-059 in the SAINT I trial was hailed as the first experimental drug to receive this distinction amid a large number of failures for the treatment of acute ischemic stroke. After the first report there was further analysis of the data by others leading to conclusions that differed. For instance one critical review concluded that in general terms that for every 100 stroke patients with NXY-059 about 10 would benefit and none would be harmed by the treatment [116]. Another review concluded that the treatment provided little evidence for the demonstration of efficacy [117]. Actually the primary investigator's conducted a much more thorough analysis of the data where the outcome parameters were evaluated at 7 days, 30 days and 90 days [118]. The authors concluded from this more detailed analysis that NXY-059 provided beneficial effects at 7 and 30 days after the stroke but not at 90 days. They also concluded that the SAINT I trial was underpowered to measure effects on the neurological examination. These conclusions were apparently important in causing the initiation of another Phase III clinical trial containing many more stroke patients which was referred to as the SAINT II trial.

### Results of SAINT II Clinical Trial and Pooled Summary

The SAINT II trial was conducted during the period of May 2003 through June 2006 and enrolled 3306 patients having the same criteria as in SAINT I trial except the study involved 362 centers from 31 countries [119]. The primary endpoint was the distribution of disability scores on the modified Rankin scale at 90 days. In addition scores on neurologic and activities of daily living scales were secondary endpoints. The efficacy analysis was based on 3195 patients. The results showed that the distribution on the modified Rankin scale did not differ between the NXY-059 treatment group and the placebo group [119]. In addition there was no evidence of efficacy for any of the secondary endpoints either. Following the publication of the SAINT II results a pooled analysis of the SAINT I and SAINT II trials was then published [120] which concluded that NXY-059 was not effective in the combined trials. The details provided of the pooled trials shows that these trials included 498 centers from 38 countries and that the design and content of the 2 SAINT studies were developed by a steering committee comprising stroke experts from Europe, North America and Australia. The steering committee had complete access to the data and was responsible for writing the manuscript [120]. These results were clearly disappointing to those in the stroke treatment community who had expected a positive outcome based on the SAINT I trials. It should be noted that the trials, both preclinical and clinical, had been developed in close accordance with the guidelines proposed by an academic-industry roundtable group (STAIR) [121–123] and therefore the failure of NXY-059

was considered a serious blow to the neuroprotection concept that guided stroke scientists for some time [124].

### **Critiques of Clinical Trials**

There have been many reviews and critiques of the steps in the commercial development of NXY-059 [125] as well as questioning the neuroprotection concept that has guided stroke scientists for some time. Since NXY-059 is a compound that has been grouped into the loosely defined category of antioxidants, it is important for the future of potential therapeutics that may be placed in this category to learn from the important points revealed by the lessons to be gleaned from the NXY-059 development history. It is important to list deficiencies in the preclinical trials that scientists have pointed out. Savitz has pointed out that the preclinical data have several short comings including a lack of strenuous testing that reproducibly showed robust protection in extended time windows in clinically relevant stroke models at several different academic research laboratories [125]. He also states that the clinical trials of NXY-059 were inadequately designed, in part, because of inappropriate treatment windows and inclusion of diverse stroke patients. Others seem to agree with this assessment [122, 126]. Another important failure pointed out by Feuerstein et al was that more attention was needed to characterize the nature of the drug candidate-target interaction and its relationship to pharmacodynamic treatment end points [123].

### **Lessons Learned Pertinent to Antioxidant Therapeutics**

An insightful review focused specifically on free radical scavengers and stroke was provided by Wang and Shuaib [127]. They noted that in addition to NXY-059 only 3 other therapeutic antioxidant medications had progressed into clinical trials for stroke; these include Ebselen, Tirilazad and Edavarone. Both Ebselen and Tirilazad were tested in stroked patients but their development has been terminated and therefore only Edavarone has succeeded in being used for stroke but only in Japan. They conclude that antioxidants as potential therapeutics is still valid as a goal, but that certain key elements need to be addressed in future development. First, there is a need to more thoroughly understand mechanistically how neurons die especially in human cerebral ischemia. Second, ischemic cell death is the result of complex noxious processes including oxidative stress, excitotoxicity, functional failure of ionic pumps, inflammatory reactions and activation of apoptotic death pathways and that termination of one of these cascades may not be sufficient to reduce the brain damage that occurs. Thirdly, therefore a combination therapy with different compounds targeting different pathways may offer better chance than single medications. Two other elements were suggested, these included the concept that DNA regulatory elements in the promoter regions of genes that were upregulated by ischemic tolerance of cells may be a valid target and finally the need to test compounds in patients that had been more carefully selected using newer technologies such as MRI screening. These are excellent suggestions and are clearly issues that need to be considered in future drug development for stroke and probably other pathologic conditions where exacerbated oxidative stress is considered an etiological agent.

## **Anti-Cancer Activity of PBN-Related Nitrones**

### **Early Studies on Hepatocellular Carcinoma**

We discovered that PBN has anti-cancer activity in the well known dietary choline deficiency rat liver cancer model [128–130]. Our research on the influence of PBN on hepatocarcinogenesis induced by a choline-deficiency diet started in 1997. Development of hepatocellular carcinoma in many animal models begins as preneoplastic lesions that appear as “small islands” of “altered cells” in the liver. In the case of the choline deficiency model as well as in several other models there are hundreds to thousands of preneoplastic nodules (islands) that develop in the liver. With time and sustained oncogenic pressure a few adenomas



arise from some of these islands and also with time carcinomas may arise from the adenomas or from the preneoplastic lesions per se. The choline-deficiency model was tested because reactive oxygen species and lipid peroxidation had been shown to be very important primary agents in this model by several laboratories [131–135]. Since PBN had been shown to suppress oxidative damage in several biological systems with very little toxicity, it seemed a reasonable choice. The first experiments were conducted on the Wistar rat model and involved three levels of PBN administered in the drinking water for 12 weeks while the rats were on the choline deficiency amino acid (CDAA) defined diet. At the end of this time period the rats were sacrificed and the livers examined to assess several parameters about preneoplastic nodule growth, size, etc. In this study we first noted that PBN in increasing amounts administered in the drinking water caused dramatic decreases in the size of preneoplastic nodules in the liver. In the same study we also noted a slight decrease, more so with increasing dosage, of the number of preneoplastic nodules in the liver. Additionally we also demonstrated that PBN at increased dosage, caused a decrease in the amount of oxidized DNA adducts as measured by 8-OHdG content in the livers [136]. The research results are summarized in Table 1.

The dramatic decrease in the size of the preneoplastic nodules was evident even at very low levels (6 mg/kg-day) of PBN in the drinking water and this spurred us to explore the anti-cancer activity in more detail. Therefore we then followed up with a long term (70 week) study where rats were placed on a choline deficient amino acid defined diet (CDAA) or a control diet (choline sufficient amino acid defined diet (CSAA) and PBN was administered in the drinking water. The results of this study clearly demonstrated that PBN had anti-cancer activity. The results have been summarized in 3 publications [128–130] and also we were granted a patent on these observations [137]. These results clearly show that PBN when given in the drinking water for the first 26 weeks or for the last 44 weeks to the rats given the CDAA diet caused the complete prevention of hepatocellular carcinoma (HCC) formation assessed after 70 weeks. We also found that when PBN was given in the control diet (CSAA), no tumor formation was observed. Therefore PBN is not carcinogenic per se. This study with PBN was followed up by a new study testing the anti-cancer activity of 4-hydroxy-PBN i.e. 4-OHPBN, which is the natural p450 hydroxylated derivative of PBN. In this case 4-OHPBN was given in the diet as opposed to PBN which was given in the drinking water. A dietary administration route for 4-OHPBN was used simply because of convenience. The results of this 70 week study of 4-OHPBN demonstrated that 4-OHPBN had anti-cancer activity in this model also. Even though the activity of 4-OHPBN and PBN were not directly compared in this experiment they were judged to be approximately equal in efficacy.

The data presented in Table 1 suggested that PBN in some unknown manner caused decreased growth of the cells in the preneoplastic lesions. A clear demonstration that this was due to increased selective apoptosis of these cells was demonstrated in an experiment where PBN as well as some of its chemical derivatives (2-OHPBN, 3-OHPBN, 4-OHPBN and 2-Sulfate PBN) were tested in a short term study in the same model although in this case PBN and derivatives were administered in the CDAA diet [129]. Important data concerned with the effects of PBN and the two active derivatives on preneoplastic lesions are presented in Figure 2. This data is a summary of our published results [138]. Note that the highest levels of PBN as well as 4-OHPBN and 3-OHPBN caused significant decreases in the size of the preneoplastic lesions as well as the content of 8-OHdG in the DNA and the amount of lipid peroxidation as measured by thio barbituric acid reactive species (TBARS) in the tissue. As was noted in the earlier study [136] the numbers of preneoplastic lesions were decreased by the highest levels of PBN (as well as 4-OHPBN and 3-OHPBN in this study) although not as marked as that noted in the change in size of the preneoplastic lesions brought on by the nitrones. The data collected on apoptosis of cells within the lesions versus the surrounding “normal” tissue cells are presented in Table 2. These data show that treatment with the nitrones caused a large significant increase in the apoptosis of the cells within the preneoplastic lesions but had the opposite effect of the

cells in the surrounding tissue. These results clearly show that the nitrones decreased the size of the preneoplastic lesions by enhancing the selective apoptosis of these cells. This is a remarkable finding and may be the key to understanding the anti-cancer activity of the nitrones in this model.

Presently we do not understand the mechanisms involved as to why the active nitrones caused the enhanced selective apoptosis of the cells within the preneoplastic lesions. Many potential mechanisms can be envisioned. However a set of observations we made suggests that enhanced expression of iNOS and thus the enhanced level of NO produced could play an important role. Our observations and rationale implicating the role of iNOS in cancer development has been summarized previously [80, 105] and therefore only the major points will be presented here. We observed that isolated hepatocytes from rats fed a choline deficiency diet had enhanced NO production [104] and that this was due to enhanced levels of iNOS expression and that PBN administration decreased NO production and iNOS expression by the isolated hepatocytes. These observations clearly implicate that iNOS and enhanced NO production is correlated with the growth of liver tumor in the choline deficiency model and that PBN administration caused the suppression of iNOS/NO production which may help to explain the anti-cancer activity of these compounds in this model. The scientific literature reinforces the notion that iNOS/NO is important in cancer development in several experimental models and in human studies as we summarized previously [105]. The mechanisms involved in the iNOS/NO mediated cancer development can be rationalized and are briefly summarized below.

Rigorous studies in the literature with several tumor cell lines [139–142], as well as isolated hepatocytes [143–149] demonstrated that NO prevents apoptosis by forming critical S-nitrosylation bonds with specific caspases, probably Caspase 8 [147] or Caspase 9 or Procaspase 9 [139] and thus preventing the initiation of the apoptosis processes. Recent studies have also demonstrated that NO mediates S-nitrosylation of Bcl-2 which prevents proteosomal degradation thereby enhancing the buildup of this protein and therefore preventing apoptosis of the cancer cells [150]. In addition to forming S-nitrosylation adducts of caspases and Bcl-2 thereby inactivating these enzymes and preventing apoptosis, NO also mediates the S-nitrosylation of OGG1 thus causing the inactivation of this DNA repair enzyme [151]. OGG1 inactivation is then expected to cause the buildup of 8-OHdG, as we observed [136]. The enhanced 8-OHdG is expected to then enhance cellular mutation events. Additionally NO mediates S-nitrosylation of PTEN inactivating this tumor suppressor protein [152] which then is expected to cause enhanced oncogenic cell cycle processes to proceed [153]. In effect then because NO is produced at sustained increased levels it causes the S-nitrosylation of OGG1, PTEN and specific caspases and Bcl-2 which then respectively causes the enhancement of cellular mutational rates and increased oncogenic cell growth as well as decreased apoptosis of preneoplastic and cancer cells. Therefore based on this rationale we postulate that PBN administration then shuts down NO production and this in effect causes cessation of these three very important carcinogenic processes by the mechanisms outlined above and therefore PBN slows down the development of cancer by enhancing the death of the preneoplastic cells and possible cancer cells. Although this rationale helps explain our results in this model, these notions have not been rigorously proven by direct experimental methods in this model. Additionally there may be other mechanisms involved in the anti-cancer action of the nitrones which supersede or encompass the iNOS/NO notions discussed above.

### Studies on Gliomas

The efficacy of PBN as a potential anti-glioma therapeutic agent was assessed in a rat C6 intracerebral glioma model. Magnetic resonance (MR) imaging (MRI) and MR angiography (MRA) was used to follow tumor morphology, such as tumor heterogeneity, size and volume, and to determine alterations in tumor vascularity, such as associated with angiogenesis,

respectively [154]. PBN was administered both as a prophylaxis and a post-tumor treatment agent. For prophylaxis, PBN treatment was administered in the drinking water (75 mg/kg rat/day or 0.065% w/w, for an estimated water uptake rate of 110–120 ml/kg rat/day) continuously starting 5 days before intracerebral implantation of C6 rat glioma cells ( $10^6$  cells/ml in Dulbecco's modified Eagle's medium (DMEM)) [154]. For post-tumor treatment, PBN was administered in the drinking water (as above) starting 14 days after cell implantation, which corresponded to a tumor volume of  $\sim 50$  mm<sup>3</sup>, as assessed by MRI. Prophylaxis PBN treatment was able to significantly decrease tumor growth (from  $0.14 \pm 0.028$  day<sup>-1</sup> and a doubling time ( $T_d$ ) of 2.6 days to  $0.05 \pm 0.007$  day<sup>-1</sup> and a  $T_d$  of 7.4 days) and result in partial or even complete tumor regression over the course of 40 days [154]. Post-tumor PBN treatment was found to alter tumor growth and result in tumor regression in 40% of treated rats [154]. As shown in Fig. 3 treatment with PBN 5 days before implantation seems to clearly inhibit glioma formation, although post-tumor PBN treatment also seems to recede glioma formation in a significant proportion of the tumors. Histological examination of rat brain slices in animals with complete tumor regression indicated no evidence of tumor cells following PBN treatment.

MRA analysis of rat brain vasculature, quantified using a Mathematica-based program developed in-house, indicated that PBN also decreased angiogenesis associated with the gliomas (see Figure 4) [154]. Non-treated C6 gliomas had a relative blood volume ratio (tumor blood volume at  $23 \pm 3$  days following cell implantation compared to the first day of assessment at 7 days following cell implantation when no tumor was visible by MRI) of  $1.26 \pm 0.157$  compared to  $0.98 \pm 0.072$  for rats implanted with non-neoplastic primary astrocytes [154]. Prophylaxis PBN treatment decreased the blood volume ratio to  $0.79 \pm 0.144$  (with an overall  $50 \pm 8\%$  decrease in blood vessel diameters), whereas post-tumor PBN treatment in responsive animals was found to decrease the blood volume ratio and vessel length by  $20 \pm 7\%$  [154]. Immunostaining for the proangiogenic factor VEGF (vascular endothelial growth factor) and a marker of endothelial cell proliferation, the von Willebrand factor (vWF), following PBN treatment indicated decreased expression of VEGF and vWF which was similar to normal brain tissue [154]. The neuroprotective [99], anti-inflammatory [100], anti-ischemic [155] and anti-carcinogenic [136] properties of PBN, coupled with its ability to freely cross the blood-brain-barrier, and also being both lipophilic and hydrophilic, makes PBN an interesting therapeutic candidate that may potentially be used for gliomas in humans, particularly following surgery and/or radiation to prevent tumor recurrence. PBN has been previously shown to decrease iNOS activity [102]. Increased iNOS levels have been found to occur in human brain tumors [156] and is strongly correlated with the degree of malignancy of a carcinoma [157]. Nitric oxide and VEGF tightly cooperate to promote angiogenesis. PBN has been found to down-regulate cytokines and NF- $\kappa$ B expression [45], which promotes iNOS expression [158], and therefore PBN may prevent or reduce angiogenesis in gliomas through NF- $\kappa$ B and/or suppression of iNOS, which would result in decreased NO production and subsequent reduced VEGF expression.

Diffusion-weighted imaging (DWI) and diffusion tensor imaging (DTI) was also recently used to assess the white matter neuronal fiber tracts following glioma formation and treatment with PBN. In this study, we found that the displacement, loss of connectivity and disappearance of neuronal fiber tracts resulting from C6 glioma formation, was prevented if PBN treatment was implemented [159]. In some of these animals the C6 gliomas completely disappeared in 6 weeks after cell implantation, resulting in a total disappearance of the compressed neuronal fibers observed in non-treated rats [159]. In non-treated animals, the C6 glioma growth compressed surrounding tissues, inducing a hematogenous disorder such as ischemia, which led to destruction of the neuronal cells and fibers [159]. It is thought that perhaps PBN may be protecting the neuronal fibers from compression-induced ischemic injury [159]. PBN has been previously shown to increase cortical cerebral blood flow in rats by inhibiting the breakdown of NO [160].

## Nitrones and Hearing Loss

### Acute acoustic trauma (AAT) and oxidative stress

An abundance of scientific data is accumulating supporting the thesis that oxidative stress plays a substantial role in the induction of cochlear injury and permanent hearing loss after excessive acute noise exposure [161–167]. Acute acoustic trauma (AAT) has been associated with a variety of changes in mammalian cochleae, including excessive release of glutamate, ischemia-reperfusion, mitochondrial injury, glutathione depletion, and ionic fluxes leading to an increase in oxidative stress [165,167].

AAT induced by steady state noise and, presumably, a variety of other types of noises, is associated with the production of reactive oxygen species (ROS), reactive nitrogen species (RNS), and lipid peroxidation species both with the onset of the noise trauma as well as some days after cessation of noise exposure [168–172]. Additionally, recent evidence suggests that there is a significant upregulation of iNOS activity in the mammalian cochlea after intense steady state noise exposure. In one study mice were exposed to high level steady state noise and tissues of the lateral cochlear wall and stria vascularis were examined [173]. This region of the cochlea is highly vascularized and very metabolically active as it generates endocochlear potential responsible for sound transduction. The investigators found that in the noise exposure condition there was a greater expression of iNOS immunoreactivity as well as increased detection of NO production and ROS activity leading to apoptosis of cells in these tissues. These changes were attenuated with the administration of an iNOS inhibitor [173]. An additional study reported increased iNOS immunoreactivity in afferent nerve fibers, outer hair cells and a variety of supporting cells in the cochlear organ of Corti as well as cells of blood vessels and marginal cells of the stria vascularis in the lateral cochlear wall [174]. Importantly, one of the consequences of the excessive oxidative stress induced by excessive noise is the onset of programmed cell death associated with mitochondrial release of cytochrome c, activation of caspases and the c-jun NH<sub>2</sub>-terminal MAP kinase pathway [175,176].

### PBN and toxin-potentiated, noise-induced hearing loss (NIHL)

Further evidence for the importance of oxidative stress in producing the cochlear injury associated with AAT includes the extensive experimental therapeutic benefit demonstrated with compounds having antioxidant properties. For example, free radical scavengers, such as mannitol, salicylate and ebselen, have been reported to decrease permanent hearing loss due to acoustic overexposure [167]. Also, approaches to enhance glutathione replenishment have been successful strategies to reduce NIHL in animal models [171,172,177–179]. Glutathione replenishment with NAC (N-acetylcysteine) has been reported to be effective in the prevention and treatment of NIHL in several animal models from several different independent laboratories and has been studied in an initial pilot study in humans [178].

A variety of early experiments were conducted with PBN. Fechter and colleagues reported that several compounds, such as carbon monoxide (CO), hydrogen cyanide (HCN) and acrylonitrile (ACN), can potentiate the hearing loss associated with acoustic overexposure [63,180–183]. Fechter and colleagues hypothesized that the mechanism of the potentiation of NIHL by these toxins involved increasing cochlear oxidative stress. They noted that PBN given systemically before and after high level steady state noise was able to decrease the toxin-induced potentiation of the noise-induced threshold shifts [181,182]. However, PBN was unable to reduce the auditory threshold shifts induced by noise alone [181,182].

### 4-OHPBN and AAT

Surprisingly, a compound closely related to PBN, 4-hydroxy phenyl-*n*-tert-butyl-nitron (4-OHPBN), has recently been reported to decrease permanent hearing loss induced by AAT in

a chinchilla model [184]. The compound 4-OHPBN is a major metabolite of PBN. This was surprising based on the previously noted work by Fechter and colleagues that PBN was ineffective in decreasing noise-induced auditory threshold shifts. In the study reported by Choi et al., chinchilla were exposed to high level steady state noise for 6 hours and treated with intraperitoneal injections of 4-OHPBN alone or in combination with NAC or NAC plus acetyl-L-carnitine (ALCAR) four hours after cessation of noise exposure and continuing twice daily for another 48 hours. Treatment with 4-OHPBN alone reduced permanent hearing threshold shifts and outer hair cell loss in a dose-dependent manner. A summary of these recently published results is presented in Figure 5. The results show that when 4-OHPBN is combined with either NAC or NAC plus ALCAR permanent hearing loss was reduced from 35 dB to 5 dB or less, and outer hair cell loss was reduced from 60% to 10% or less. Current studies suggest that substantial reduction in permanent hearing loss can still be obtained using the combination of 4-OHPBN plus NAC and ALCAR when given even 24 hours after the noise exposure (Choi et al, unpublished results). Utilizing these compounds in combination with 4-OHPBN allowed for a substantial reduction in doses of compounds required while still affecting a maximal reduction in hearing and hair cell loss.

The exact mechanisms by which 4-OHPBN reduces cochlear injury associated with AAT are still unknown. Possibilities include free radical scavenging, inhibition of iNOS activation, suppression of ROS and RNS formation, decreased mitochondrial ROS production and reduced neuroinflammation and activation of MAP kinase cascades [185]. Preclinical testing to further define toxicity, mechanism of action, pharmacokinetics, and oral dosing parameters are now underway.

## Conclusions and Perspective on Antioxidant Therapeutics

### What is the Candidate Drug-Therapeutic Target?

Our experience with the PBN-related nitrones has taught us many lessons in terms of the development of therapeutics that are generally considered antioxidants. We have focused on nitrones simply because the PBN-related nitrones have been shown to have potent biologic activity in preclinical models of several pathologic conditions and in age-related diseases where exacerbated oxidative stress is generally considered an etiological factor. Even though PBN and related compounds are generally considered antioxidants, the truth is they are poor antioxidants in several in vitro systems but in fact generally they do act in biological systems to decrease oxidative stress at least in those systems where the condition has been evaluated. Nitrones became widely known because of their ability to spin-trap free radicals yet the biologic action of PBN, especially, is most likely not due to its ability to spin-trap radicals in biological systems. It should be noted that in the case of NXY-059 it might act as a spin-trap simply because its level in blood is so high (260  $\mu\text{M}$ ) under treatment conditions that it could act as a spin-trap. But there certainly was no agreement as to its mechanism of action in preclinical models [39,106]. Therefore there is considerable uncertainty and confusion about the mechanism of action of the nitrones.

Our research has shown that the nitrones have potent anti-inflammatory properties and this is probably important. Therefore, at first glance, in the case of the nitrones there is considerable confusion regarding the drug-therapeutic target when compared to other well known drugs such as the statin's for instance. In truth, however, many well known drugs such as aspirin, the statins and the non steroidal anti-inflammation drugs have action on many other processes and biological parameters other than those that they are widely considered to act upon. Conversely some well known drugs are also very potent antioxidants. This is the case for the well known  $\beta$  blocker D-propranolol which has very potent antioxidant activity and this property is considered to be an important reason for its therapeutic mechanism of action [186–188].

## Confusion over Antioxidants and Their Action

There is another widely unrecognized problem with defining the mechanism of action or drug-therapeutic target for antioxidants. Simply stated, oxidative stress as well as the activity of antioxidants in quelling oxidative stress is an area of active research and as such, a well defined drug target is difficult to present at the present time when the therapeutic under question is categorized as an antioxidant. It is therefore clear from the foregoing discussion that much development research effort is necessary to understand and then clearly define the candidate drug-therapeutic target for a specific "antioxidant" therapeutic. The case of alpha-tocopherol (vitamin E) is instructive. Vitamin E was considered an essential vitamin since the 1920's and only later found to possess antioxidant properties. But its true mechanism of action, especially how its position in the membrane and interfaces with ascorbate in the cytosolic phase to quell oxidative stress and oxidative damage was only elucidated in the last ten years. It is probable that in the case of many so-called antioxidants that the therapeutic target may be very different than what is generally considered an antioxidant property. As noted above this task is probably harder for "antioxidant drugs" than many other drugs. Therefore in many cases the best that can be done is to precisely define which biological processes are of vital importance in the development of the pathologic indication where the therapeutic agent acts and then use this as a first line screening tool. In most cases the basic science underling the processes involved in causing the pathologic condition may be a major limitation as for instance in the case of stroke. Also as in the case of stroke, several different processes combine to play a synergistic role leading to the neuronal damage that occurs. Blocking only one of these processes is probably not enough. In this case there is strong rationale for using a combination of compounds as was pointed out as a suggestion for stroke [127]. Our observations on the synergistic effectiveness of NAC combined with 4-OHPBN is a good illustration of this point in the case of acute acoustical trauma-induced hearing loss [184]. It is possible that the active plant constituents could be useful in this regard.

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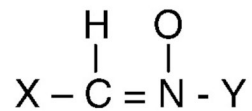
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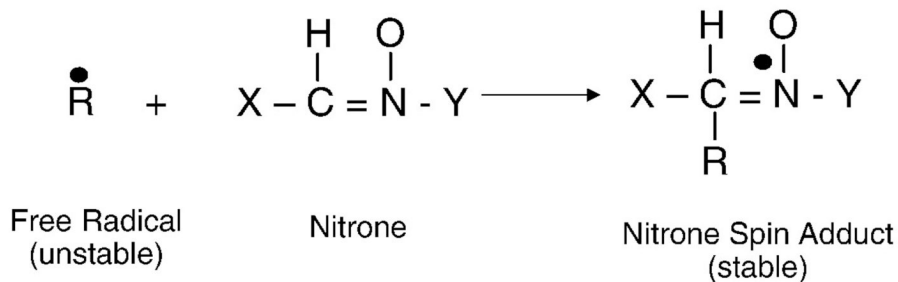


# Nitrones and Their Properties

Nitronone Functionality

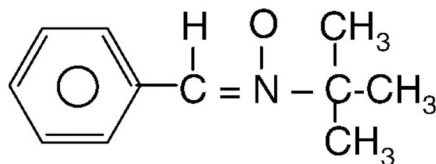


Spin Trapping Activity



Specific Nitronone

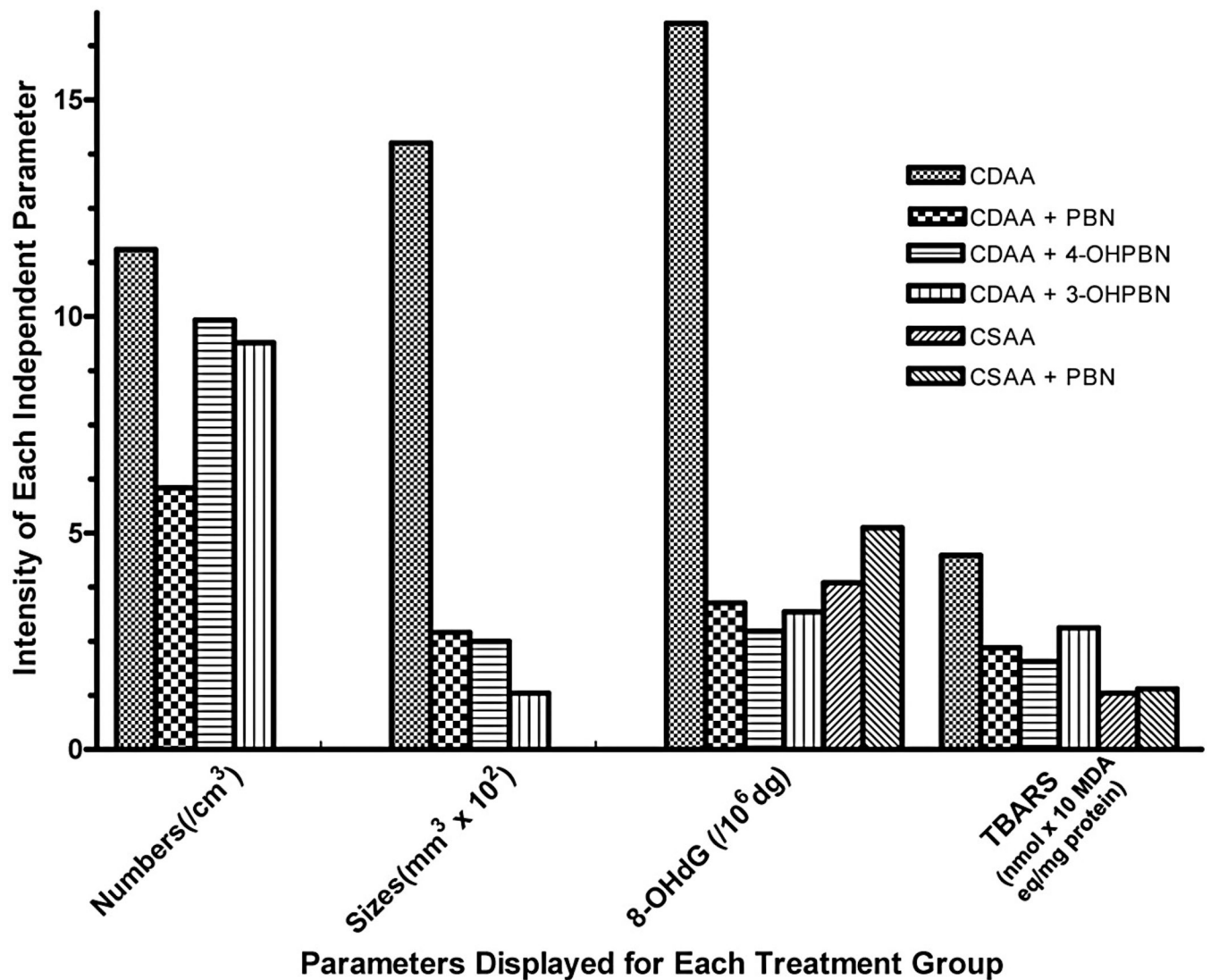
$\alpha$ -Phenyl-*tert*-butyl Nitronone  
(PBN)



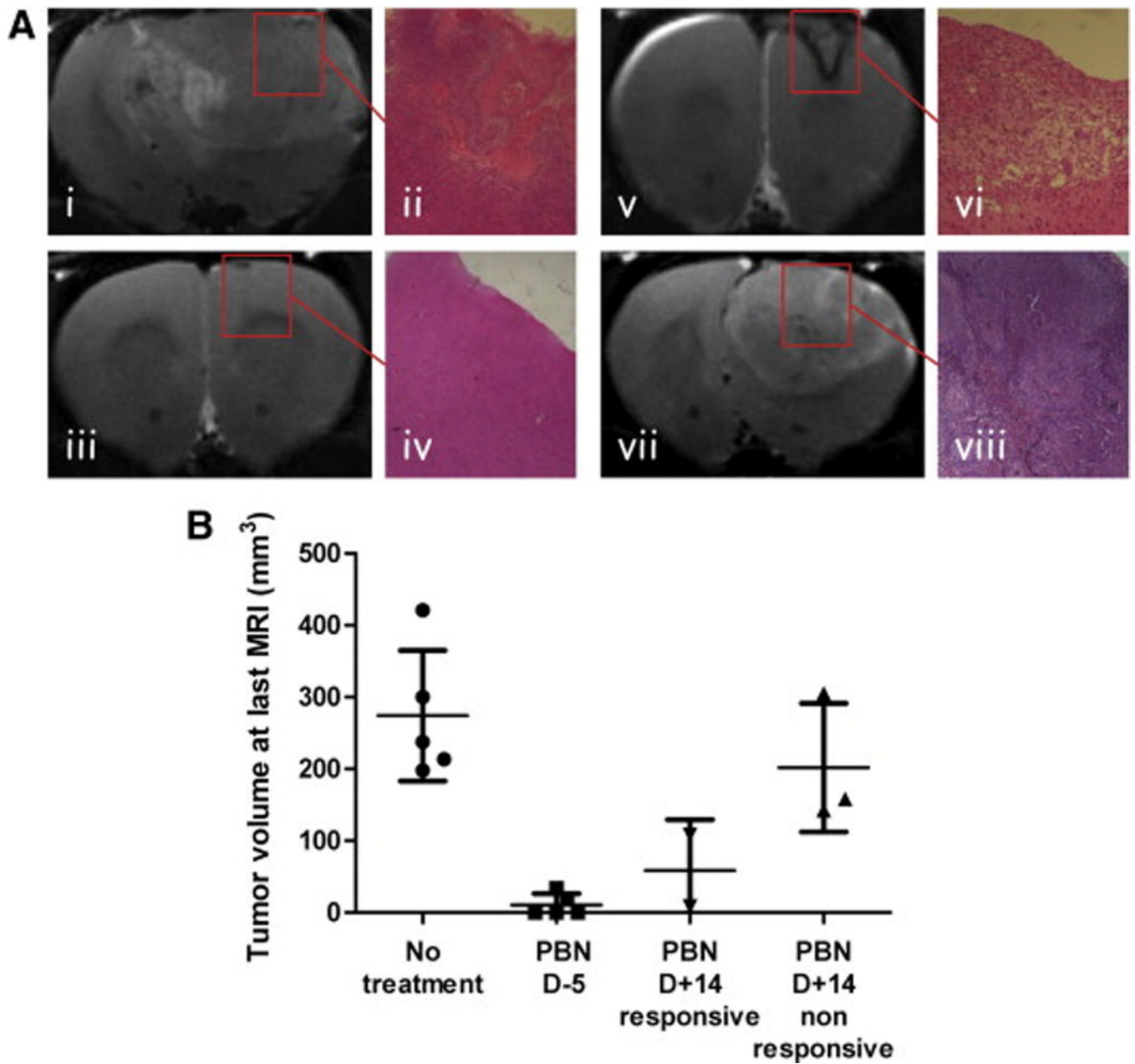
**Figure 1.**

Chemical structures of nitronones, their reaction to trap free radicals and form spin adducts and the chemical structure of PBN.

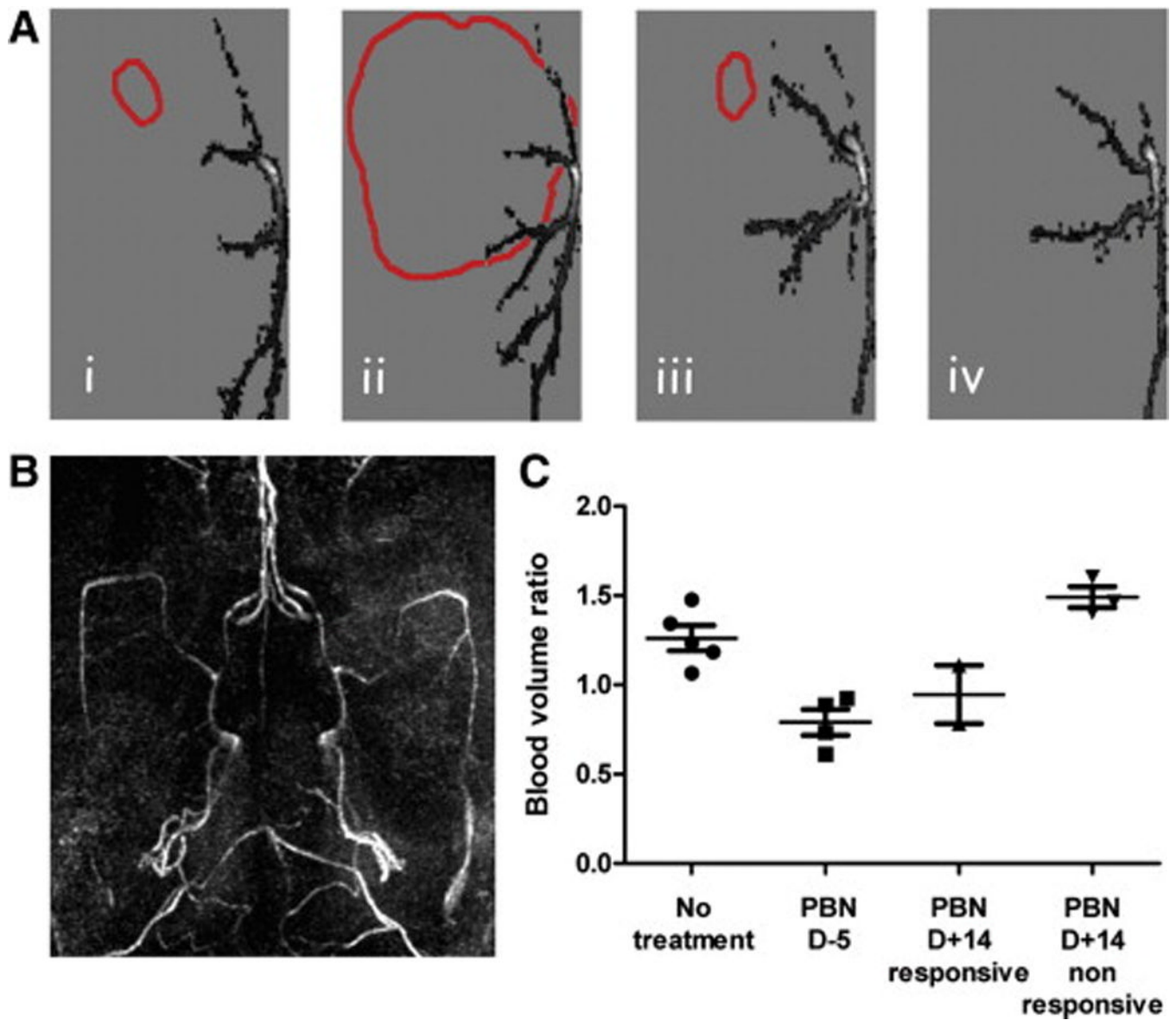
## Influence of PBN and Derivatives on Parameters of Preneoplastic Lesions



**Figure 2.** The influence of PBN and its 3-hydroxy and 4-hydroxy derivatives on the parameters of preneoplastic lesions in the liver of rats administered these chemicals in a choline deficient and choline sufficient diet. The values were calculated from the data presented in Floyd et al [138].

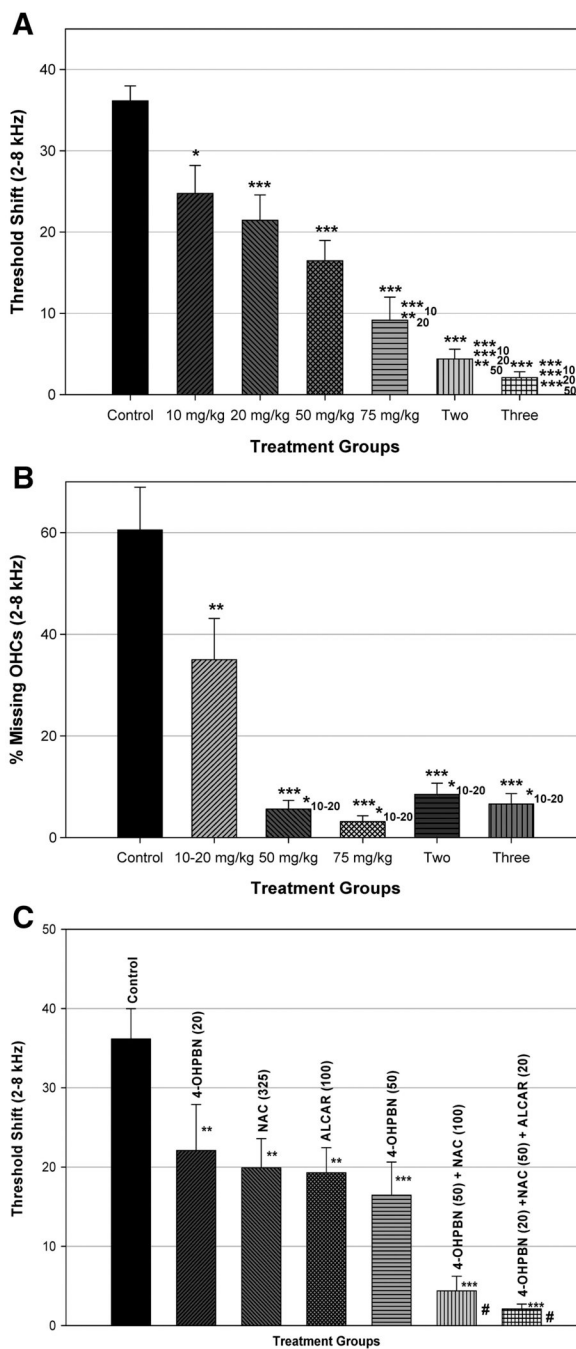


**Figure 3.** (A) Representative T2-weighted images and histology (H&E, x4) staining of C6 gliomas from the different treatment groups. (i–ii) Untreated C6 glioma at day 18 following cell implantation. (iii–iv) PBN pre-treated C6 glioma at day 27. (v–vi) Post-tumor PBN treated C6 glioma undergoing regression, shown at day 32. (vii–viii) Nonresponsive post-tumor PBN treated C6 glioma shown at day 16. (B) Tumor volumes (mm<sup>3</sup>) measured at the last MRI timepoint, represented as means  $\pm$  standard deviation.



**Figure 4.**

(A) 2D vasculature projections in the horizontal plane of a representative non-treated C6 glioma at day 8 (i) and 20 (ii) after cell implantation, and of a PBN pretreated C6 glioma at day 9 (iii) and 39 (iv). The tumor boundaries are also represented (*red*). (B) 3D maximum intensity projection (MIP) rendering of the brain vasculature. (C) Blood volume ratios for each treatment group at the last time point (days 17–21 for untreated rats and non-responsive rats (PBN D +14), and days 39–40 for treated rats (PBN D–5, PBN D+14 responsive), represented as means  $\pm$  standard deviation.



**Figure 5.** Figure 5A. Auditory brainstem response (ABR) threshold shifts averaged at frequencies of 2–8 kHz for control group, different 4-OHPBN dosage group, and different drug combination group. The threshold shifts of control group were reduced with increases of 4-OHPBN dosage and the number of drug combination. Asterisks of \* and \*\*\* represent statistically significant differences in ABR threshold shifts between control group and each experimental group at  $p < 0.05$  and  $p < 0.001$ , respectively. Asterisks of \*\*\*10, \*\*20, \*\*\*20, \*\*50, and \*\*\*50 indicate statistically significant differences in ABR threshold shifts between each experimental group (75 mg/kg, two, and three) and each different dosage of 4-OHPBN (10, 20, and 50 mg/kg) at

$p < 0.001$ ,  $p < 0.01$ ,  $p < 0.001$ ,  $p < 0.01$ , and  $p < 0.001$ , respectively. This figure has been published previously by Choi et al [184] and is reproduced with the permission of the publisher.

Figure 5B. Percentage of missing OHC averaged at cochlear frequency regions corresponding to 2–8 kHz for four 4-OHPBN dosage groups, different drug combination groups, and control group. Outer hair cell (OHC) losses of control group were reduced with increases of 4-OHPBN dosage and the number of drug combination. Asterisks of \*\* and \*\*\* represent statistically significant differences in OHC loss between each experimental group and control group at  $p < 0.01$ , and  $p < 0.001$ , respectively. Asterisks of \*10–20 indicate statistically significant differences in OHC loss between each experimental group and 4-OHPBN of 10–20 mg/kg at  $p < 0.05$ . This figure has been published previously by Choi et al [184] and is reproduced with the permission of the publisher.

Figure 5C. Auditory brainstem response (ABR) threshold shifts averaged at frequencies of 2–8 kHz for different antioxidant drugs and their combinations. Asterisks of \*\* and \*\*\* indicate statistically significant differences between control group and each experimental group at  $p < 0.01$  and  $p < 0.001$ , respectively. The symbol # shown in the two-drug combination group indicates significant differences in averaged ABR threshold shifts between the two-drug combination and other experimental groups [4-OHPBN(20), NAC(325), ALCAR(100)] at  $p < 0.001$ , respectively while the symbol # shown in the three-drug combination groups shows significant differences between the three-drug combination and other experimental groups [4-OHPBN(20), NAC(325), ALCAR(100)] at  $p < 0.001$  and the three-drug combination and an experimental group [4-OHPBN(50)] at  $p < 0.05$ , respectively. Data for NAC (325 mg/kg) and ALCAR (100mg/kg) were excerpted from Coleman et al [189]. This figure has been published previously by Choi [184] and is reproduced with the permission of the publisher.

Table 1

PBN Effect on Choline Deficiency Induced Preneoplastic Lesions\*

Group	Treatment(s)	no./cm <sup>3</sup>	GST-P-positive lesions			Av. PBN exposure (mg/kg body wt/day)
			Size(mm <sup>3</sup> )	8-OHG (8-OHdG/10 <sup>6</sup> dG)		
1	CDAA	190 ± 10	1.92 ± 0.31	18.92 ± 1.96	0	
2	CDAA + PBN (low)	170 ± 28	0.33 ± 0.20**	13.10 ± 2.56**	6.06 ± 1.07	
3	CDAA + PBN (middle)	149 ± 15**	0.17 ± 0.06**	5.11 ± 1.24**	32.86 ± 4.81	
4	CDAA + PBN (high)	142 ± 10**	0.10 ± 0.05**	2.55 ± 0.83**	61.76 ± 8.90	
5	CSAA	0		1.00 ± 0.10	0	
6	CSAA + PBN (high)	0		1.00 ± 0.17	64.45 ± 9.53	

\* PBN was administered in the drinking water, data taken at 12 weeks; Summary of data taken from Table 1 and Table 2 of Nakae, D. et al.(131)

\*\* Significantly different from group p&lt;0.01.

**Table 2**

Effect of PBN Derivations on Apoptotic Index of Preneoplastic Lesion Cells and Surrounding Tissue Cells\*

Treatment	% of Cells in Apoptosis	
	Preneoplastic Lesions	Surrounding Cells
CDA A Diet Only	3.24	6.48
PBN Added	12.65	2.40
4-OHPBN Added	8.97	2.65
3-OHPBN Added	10.90	2.40
Control Diet Only	-**	0.29

\* Experimental results obtained on liver tissue in the dietary choline deficiency rat model of hepatocellular carcinoma conducted by Floyd et al(133).

\*\* There were no preneoplastic lesions observed in the livers of rats fed the control (choline sufficient) diet.