

NIH Public Access

Author Manuscript

Tetrahedron Lett. Author manuscript; available in PMC 2011 January 6.

Published in final edited form as:

Tetrahedron Lett. 2010 January 6; 51(1): 164–166. doi:10.1016/j.tetlet.2009.10.117.

Total Syntheses of Naamidine G and 14-Methoxynaamidine G

Panduka B. Koswatta and Carl J. Lovely*

Department of Chemistry and Biochemistry, The University of Texas at Arlington, Arlington TX 76019-0065

Abstract

Simple total syntheses of two *Leucetta*-derived marine alkaloids have been developed using position specific halogen-metal exchange of polyhaloimidazoles to introduce the benzyl substituted sidechains. Introduction of the C2 amine group by lithiation and trapping with tosyl azide provides amines on catalytic hydrogenation, which can be converted to naamidine G and 14-methoxynaamidine G using a procedure described in the literature.

2-Aminoimidazole alkaloids have recently attracted the attention of the synthetic community, in particular several members of the oroidin family of alkaloids.^{1, 2} A less well explored group of 2-aminoimidazole-containing natural products are found in sponges of the *Leucettidae* family of sponges. The *Leucetta* family of alkaloids are characterized by the presence of a 2-aminoimidazole moiety which is substituted at various positions around the heterocycle and possess typically one or two benzylic fragments (1–6, Figure 1).^{1b} They range in complexity from the fairly simple clathridine A (1)³ *via* naamine/naamidine type systems (e.g., naamine D (2), naamidine G (3), 14-methoxynaamidine G (5))⁴ to the more elaborate and highly oxygenated members such as calcaridine A (4) and spiroleucettadine (6).⁵ From a medicinal chemistry perspective, several of these alkaloids have been reported to exhibit activity as antibiotics,⁶ nitric oxide synthase inhibitors,⁷ and cytotoxicity,⁸ but in general their broad scale pharmacological evaluation has not been addressed. Although these molecules are not overly complex or challenging targets, they are potentially attractive scaffolds for evaluation in high throughput screening programs.^{1b}

Our lab has developed a number of synthetic methods for the total synthesis of 2aminoimidazole alkaloids using polyhaloimidazoles as the starting point, utilizing this strategy en route to the total synthesis of members of both the oroidin and *Leucetta* families of alkaloids. 2·9 Following earlier reports,¹⁰ we and others have demonstrated that 4,5-dihaloimidazoles can be functionalized in a sequential and controlled manner in the order of C5 \rightarrow C4 \rightarrow C2 using Grignard reagents (for functionalization at the 4- and 5-positions) and n-BuLi (for C2).⁹,¹¹, 12 This manuscript describes the application of this strategy in the total syntheses of naamidine G (**3**) and 14-methoxynaamidine G (**5**).

Naamidine G (**3**) and 14-methoxynaamidine G (**5**) were isolated and described by Pietra in 1995, but little in terms of biological activity of these natural products was reported.^{4b} It is mentioned that the crude extract possess mild cytotoxicity (against KB cells) and activity against *Candida albicans*, but apparently the purified compounds were not further evaluated. ^{4b} Our approach to these natural products was patterned after a similar sequence of reactions

^{*}Corresponding author. Tel.: 817-272-5446; fax: 817-272-3808; lovely@uta.edu.

Publisher's Disclaimer: This is a PDF file of an unedited manuscript that has been accepted for publication. As a service to our customers we are providing this early version of the manuscript. The manuscript will undergo copyediting, typesetting, and review of the resulting proof before it is published in its final citable form. Please note that during the production process errors may be discovered which could affect the content, and all legal disclaimers that apply to the journal pertain.

used en route to calcaridine A^{9a} starting from 4,5-diiodo-1-methylimidazole (7). Specifically, 7 was converted to the corresponding 4-iodoimidazole **8** by treatment with EtMgBr and then *p*-anisaldehyde (Scheme 1).^{9a} After ionic reduction of the benzylic hydroxyl group in **8** with Et₃SiH and TFA, the resulting 4-iodoimidazole was reacted with EtMgBr to effect formation of the imidazol-4-yl Grignard reagent which was treated with *N*-methylformanilide (**10**) to provide the aldehyde **11**. Reaction of this aldehyde with *p*-MeOC₆H₄MgBr provided the desired alcohol **12**, which is the key intermediate for both naamidine G and 14methoxynaamidine G (Scheme 2). Removal of the doubly benzylic alcohol in **12** was achieved by reduction with Et₃SiH and TFA at room temperature affording **13**. Lithiation at C2 was accomplished with *n*-BuLi and the resulting organolithium species was treated with TsN₃ to install an azide moiety, which was subjected to the catalytic hydrogenation to provide 1methylnaamine D (**15**). This was then converted to naamidine G (**3**) using the procedure reported earlier with TMS-activated methyl parabanic acid^{13,9b} to complete the total synthesis of **3** in 41% overall yields in 8 steps.

The total synthesis of 14-methoxynaamidine G (**5**) commenced with the conversion of alcohol **12** to methyl ether **16** with methanol and TFA (Scheme 2). Introduction of C2 azide was accomplished by metalation and reaction with TsN_3 provided **17**, which was converted to amine **18** by catalytic hydrogenation. Following the same reaction as above, TMS-activated methyl parabanic acid provided 14-methoxynaamidine G (**5**) in low yield. Initially it was assumed that used TMS source might cause the decomposition of the starting material. Therefore, the introduction of hydantoin moiety was attempted with other methods reported in the literature. ¹⁴ However, these methods failed to give the product and only the decomposition of the starting material was observed. The reason for the low yield from these reactions is still under investigation, but presumably is a result of competitive silylation and ionization of the methyl ether.

The spectroscopic data of the synthetic compounds were in agreement with the data reported for the natural products. However the melting point of the synthetic naamidine G was 195–196 °C in contrast to the natural product (94 °C).^{4b}

In summary, we have developed high yielding eight-step syntheses of the *Leucetta* alkaloids naamidine G (**3**) and 14-methoxynaamidine G (**5**) using a sequential chemoselective halogenmagnesium exchange, permitting the selective installation of the C5 and C4-benzyl groups without protection of the more acidic C2-position. Subsequent lithiation at C2 and electrophilic trapping leads to introduction 2-amino moiety. We are actively exploring the synthesis of other members of the *Leucetta* family of alkaloids, and their analogs for medicinal chemistry programs.

Acknowledgments

This work was supported by the Robert A. Welch Foundation (Y-1362) and in part the NIH (GM066503). The NSF (CHE-9601771, CHE-0234811) is thanked for partial funding of the purchases of the NMR spectrometers used in this work.

References

- (a) Arndt HD, Riedrich M. Angew Chem, Int Ed 2008;47:4785. (b) Sullivan JD, Giles RL, Looper RE. Curr Bioact Compd 2009;5:39.See also (c) Giles RL, Sullivan JD, Steiner AM, Looper RE. Angew Chem, Int Ed 2009;48:3116.Int Ed 2009;48:3116.
- 2. He Y, Du H, Sivappa R, Lovely CJ. Synlett 2006:965.
- 3. (a) Ciminiello P, Fattorusso E, Magno S, Mangoni A. Tetrahedron 1989;45:3873. (b) Ciminiello P, Fattorusso E, Mangoni A, Di Blasio B, Pavone V. Tetrahedron 1990;46:4387.

Tetrahedron Lett. Author manuscript; available in PMC 2011 January 6.

- For isolation see: (a) Dunbar DC, Rimoldi JM, Clark AM, Kelly M, Hamann MT. Tetrahedron 2000;56:8795–8798. (b) Mancini I, Guella G, Debitus C, Pietra F. Helv Chim Acta 1995;78:1178.
- (a) Edrada RA, Stessman CC, Crews P. J Nat Prod 2003;66:939. [PubMed: 12880310] (b) Ralifo P, Crews P. J Org Chem 2004;69:9025. [PubMed: 15609934] (c) White KN, Amagata T, Oliver AG, Tenney K, Wenzel PJ, Crews P. J Org Chem 2008;73:8719. [PubMed: 18925788]
- 6. Akee R, Carroll TR, Yoshida WY, Scheuer PJ, Stout TJ, Clardy J. J Org Chem 1990;55:1944.
- 7. Dunbar DC, Rimoldi JM, Clark AM, Kelly M, Hamann MT. Tetrahedron 2000;56:8795.
- (a) Copp BR, Fairchild CR, Cornell L, Casazza AM, Robinson S, Ireland CM. J Med Chem 1998;41:3909. [PubMed: 9748366] (b) Aberle NS, Catimel J, Nice EC, Watson KG. Bioorg Med Chem Lett 2007;17:3741. [PubMed: 17462892]
- 9. (a) Koswatta PB, Sivappa R, Dias HV, Lovely CJ. Org Lett 2008;10:5055–5058. [PubMed: 18816134]
 (b) Koswatta PB, Lovely CJ. Tetrahedron Lett 2009;50:4998. (c) Koswatta PB, Sivappa R, Dias HVR, Lovely CJ. Synthesis 2009:2970.
- 10. Carver DS, Lindell SD, Saville-Stones EA. Tetrahedron 1997;53:14481.
- (a) Lovely CJ, Du H, Sivappa R, Bhandari MK, He Y, Dias HVR. J Org Chem 2007;72:3741.
 [PubMed: 17425369] (b) Bhandari MR, Sivappa R, Lovely CJ. Org Lett 2009;11:1535. [PubMed: 19278243]
- (a) Dehmel F, Abarbri M, Knochel P. Synlett 2000:345. (b) Abarbri M, Thibonnet J, Bérillon L, Dehmel F, Rottländer M, Knochel P. J Org Chem 2000;65:4618. [PubMed: 10959867] (c) Knochel P, Dohle W, Gommermann N, Kneisel F, Kopp F, Korn T, Sapountzis I, Vu V. Angew Chem Int Ed 2003;42:4302. (d) Yang X, Knochel P. Chem Commun 2006:2170.
- 13. Aberle NS, Lessene G, Watson KG. Org Lett 2006;8:419. [PubMed: 16435849]
- (a) Kawasaki I, Taguchi N, Yoneda Y, Yamashita M, Ohta S. Heterocycles 1996;43:1375. (b) Ohta S, Tsuno N, Maeda K, Nakamura S, Taguchi N, Yamashita M, Kawasaki I. Tetrahedron Lett 2000;41:4623. (c) Nakamura S, Kawasaki I, Yamashita M, Ohta S. Heterocycles 2003;60:583.

Koswatta and Lovely



5: 14-Methoxynaamidine G

6: (-)-Spiroleucettadine

Figure 1. Selected *Leucetta* Alkaloids Koswatta and Lovely



Scheme 1. Total synthesis of naamidine G (3)

Tetrahedron Lett. Author manuscript; available in PMC 2011 January 6.

Koswatta and Lovely



Scheme 2. Total synthesis of 14-methoxynaamidine G (**5**)