

NIH Public Access

Author Manuscript

J Magn Reson. Author manuscript; available in PMC 2011 February 1

Published in final edited form as:

J Magn Reson. 2010 February ; 202(2): 267. doi:10.1016/j.jmr.2009.11.017.

VARIABLE FIELD PROTON-ELECTRON DOUBLE-RESONANCE IMAGING: APPLICATION TO pH-MAPPING OF AQUEOUS SAMPLES

Valery V. Khramtsov, George L. Caia, Keerthi Shet, Eric Kesselring, Sergey Petryakov, Jay L. Zweier, and Alexandre Samouilov^{*}

Davis Heart and Lung Research Institute and the Division of Cardiovascular Medicine, The Ohio State University, College of Medicine, Columbus, Ohio 43210, USA

Abstract

A new concept of Variable Field Proton-Electron Double-Resonance Imaging (VF PEDRI) is proposed. This allows for functional mapping using specifically designed paramagnetic probes (e.g. oxygen or pH mapping) with MRI high quality spatial resolution and short acquisition time. Studies performed at 200 G field MRI with phantoms show that a pH map of the sample can be extracted using only two PEDRI images acquired in 140 s at pre-selected EPR excitation fields providing pH resolution of 0.1 pH units and a spatial resolution of 1.25 mm. Note that while concept of functional VF PEDRI was demonstrated using the pH probe, it can be applied for studies of other biologically relevant parameters of the medium such as redox state, concentrations of oxygen or glutathione using specifically designed EPR probes.

Keywords

proton-electron double-resonance imaging, PEDRI; pH; functional imaging; dynamic nuclear polarization, DNP; nitroxide; spin pH probe

INTRODUCTION

Broad biomedical applications of nuclear magnetic resonance, NMR and magnetic resonance imaging, MRI, are possible due to the existence of endogenous NMR sensitive nuclei, with concentrations up to 110 M in the case of water protons. MRI has found numerous clinical applications but still suffers from limited functional resolution. On the other hand, electron paramagnetic resonance, EPR, has the unique advantage over NMR in functional specificity due to the absence of overlap with endogenous EPR signals, and has greater sensitivity for the same probe concentration due to the 658 times larger magnetic moment of the electron compared with that of the proton. However, EPR and EPR imaging techniques are far from attaining their maximum potential because of technical limitations which include low depth of penetration of the microwaves in the aqueous sample and short relaxation times of the EPR probes. Since EPR linewidths are 3 orders of magnitude larger than those of NMR, EPRI requires much more powerful gradients[1;2]. Because of the very short electron relaxation times, typically microsecond, pulsed approaches have been limited to paramagnetic probes with long relaxation times[3;4]. Nevertheless, recent advances in pulsed EPR techniques operating at 300 MHz frequency allowed for the first time in vivo imaging of the nitroxide

^{*}Address for correspondence: Alexandre Samouilov, Davis Heart and Lung Research Institute, The Ohio State University, 420 West 12th Ave, Room 611B, Columbus, OH 43210. alex.samouilov@osumc.edu.

with narrow EPR line[5]. To date, to obtain the highest sensitivity and best quality images, minutes to hours of time are required with common continuous wave (CW) EPR which utilizes stepped gradients. The images obtained give information only on the location of the probe and often lack the anatomic structure required for interpretation[6]. Special efforts are needed to co-register EPRI with anatomic structure[7;8]. Spectral-spatial EPRI is also possible with the CW approach and can provide valuable physiological and functional information[9–14], however even more lengthy acquisition times are required, limiting its applicability, especially for *in vivo* applications.

An alternative imaging modality which also employs unpaired electrons is PEDRI (Proton-Electron Double-Resonance Imaging)[15] or OMRI (Overhauser-enhanced Magnetic Resonance Imaging)[16]. PEDRI as MRI-based technique for *in vivo* imaging of free radicals was first developed in Aberdeen, Scotland, by Dr. D. Lurie and his colleagues[15;17]. Using the PEDRI approach, the EPR signal amplitude spatial distribution is reconstructed from the NMR signal of water protons after irradiation of the paramagnetic solute with an EPR frequency microwave. Under the proper conditions, a transfer of polarization from the electrons to protons occurs by Overhauser effect[18], resulting in enhancement of the NMR signal up to 150 times as measured in biological systems[16;19]. The Overhauser enhancement depends on the RF power, the line width and concentration of the paramagnetic agent, with theoretical maximum enhancement factor of 328. However, practically in in vivo experiments, considerations of limitation of RF power deposition usually preferred over the maximizing the signal intensity; enhancement of several tens usually preferred for optimal results [20–22].

Recent developments in PEDRI demonstrated that this method allowed simultaneous coregistration of free radical distribution and anatomic information[22;23]. Since PEDRI is based on the proton MRI, it circumvents the resolution limitations of EPRI that occur due to very broad linewidths of most paramagnetic labels, and inherently offers high spatial resolution and rapid image data collection. PEDRI has been used by a few research groups to image free radicals *in vivo*[20;22;24;25] and it is now recognized as a powerful alternative to conventional EPRI. One of the successful applications of functional PEDRI is oxygen mapping. It is based on the paramagnetic character of dissolved oxygen which affects the EPR linewidth of free radical probes and, as a consequence, alters radio frequency power saturability of the probes. Therefore, PEDRI with variable saturation power offers a reliable method of imaging oxygen concentrations *in vivo* using triarylmethyl (TAM) probes[19;20;26;27]. In general, the capacity of PEDRI to reflect EPR spectral properties was demonstrated by distinguishing localizations of ¹⁴N and ¹⁵N isotope-labeled nitroxides by PEDRI using different EPR irradiation magnetic fields[28].

Previously we applied Field Cycling DNP (FC DNP) and Field Cycling PEDRI (FC PEDRI) techniques in vivo for obtaining spectral characteristics and spatial distribution of the pH sensitive probe, correspondently [29]. FCDNP approach provided spectral information of the whole sample lacking in spatial resolution. In a complimentary way, FC PEDRI allowed for imaging probe distribution while lacking in spectral information.

In this work we proposed a new modality of functional imaging in living tissues with enhanced functional and temporal resolution using a PEDRI approach in combination with the original concept of Variable Field (VF) PEDRI. We hypothesized that valuable spectral parameters at each pixel can be extracted from a limited number of PEDRI acquisitions acquired at preselected EPR excitation fields. This allows for functional mapping using specifically designed paramagnetic probes (e.g. oxygen or pH mapping) with MRI quality spatial resolution and short acquisition time. The hypothesis has been verified using VF PEDRI and a pH sensitive nitroxide probe. The images were acquired at two pre-selected EPR excitation fields which coincide with EPR spectral peak positions of protonated and nonpotonated forms of the probe.

A pH map of the sample was extracted from these two PEDRI images providing pH resolution of 0.1 pH units and a spatial resolution of 1.25 mm. The obtained data shows several fold decrease in acquisition time is possible for VF PEDRI compared with EPRI. This is particularly important for *in vivo* applications where stability of the paramagnetic probes is limited.

MATERIALS AND METHODS

Chemicals

The pH-sensitive nitroxyl radical, 4-amino-2,2,5,5-tetramethyl-3-imidazoline-1-yloxy (R1, Fig.1) was synthesized as previously described [30].

FC-DNP and FC-PEDRI measurements

Field-Cycled (FC)-PEDRI images and FC-Dynamic Nuclear Polarization (DNP) spectra were obtained using a home-built imager/spectrometer at the Ohio State University[31]. To obtain EPR spectral characteristics, partial cancellation of the detection field B_0^{NMR} is required to alter the evolution field B_0^{EPR} at which the electron paramagnetic resonance (EPR) is excited. This partial cancelation of B_0^{NMR} is achieved by using a secondary electromagnet added to a 0.38 T clinical MRI magnet. The secondary electromagnet built into the gap of the primary magnet provides a vertical magnetic field offset of up to 0.1 T to perform EPR irradiation at the low field followed by high field NMR detection. The field cancellation coils are actively shielded to minimize the effect of eddy currents that occur in the primary magnet. In order to saturate electron spins, a long EPR pulse of relatively high power is needed. For the double resonance used in PEDRI, a modified Alderman-Grant design resonator with capacitive coupling have been constructed for the EPR excitation channel along with a typical solenoidal coil for the NMR channel[31]. The system is capable of performing fixed-field PEDRI along with two field-cycling modes, FC-DNP and FC-PEDRI. DNP spectra and images of the phantom samples were collected using field-cycling techniques with an EPR irradiation frequency of 562 MHz (200 G) or 282 MHz (100 G) applied for 500 ms before each collection of a proton signal, and an NMR frequency of 856 kHz (200 G). The repetition time (TR) of the pulse sequence was 1100 ms. The average incident power during an acquisition was 3.3. W. Spectra were obtained by means of a field-cycled DNP pulse sequence in which the evolution field strength was stepped. The EPR irradiation frequency was maintained constant (as was the NMR frequency), so each step of the evolution field was equivalent to the sampling of an EPR signal at a different magnetic field value. The number of steps and their separation defined the overall width of the observed spectrum and its resolution. In this study, spectra of 850 points over a field range of 34 G centered on 200 G (resolution 0.04 G) provided the full three-line spectrum.

pH titration using FC DNP

The solutions of the R1, 0.5 mM, in water or in 50 mM phosphate buffer were titrated with solutions of HCl or KOH to the required pH, placed in glass tubes of 12 mm diameter for FC DNP spectra acquisition. The observed hyperfine splitting (hfs) measured as the distance between center- and high-field DNP spectral line positions of the triplet has been used as a pH-sensitive parameter. The EPR irradiation field positions that correspond to low- or high-field spectral line positions of fully protonated, RH⁺ (pH«pKa), and fully nonprotonated, R (pH»pKa), radical forms were measured and used as two pre-selected magnetic fields for pH mapping using VF PEDRI.

pH mapping using VF PEDRI

The phantom samples were prepared from the glass tubes of 9.5 or 12 mm diameter filled with 0.5 mM solution of the R1 radical in water or in 50 mM phosphate buffer titrated with HCl or

KOH to desired pH. This concentration corresponds to administration of 14μ mole of the nitroxide in 28 g mouse, which could be done by bolus i.p. or i.v. injection of 0.4 cc of 35 mM solution. Previously we have shown that a bolus injection of 0.5 cc of 100 mM nitroxide solution was tolerated by the mice used in the imaging experiments [22].

Field-cycling PEDRI images were collected as 80 mm×80 mm projective images, with matrices' size of 64×64 giving voxel size of 1.25 mm. The field-cycling capability of the system

was used to perform EPR irradiation at two pre-selected EPR magnetic fields, B_{RH+}^{EPR} and B_{R}^{EPR} corresponding to the peak positions of RH⁺ and R forms of the R1 radical, respectively (see Fig.2). The ratio of NMR signals at each pixel of these two images is pH dependent and was converted to a pH map using a corresponding calibration curve.

RESULTS

Figure 3 shows a typical DNP spectra of the nitroxide R1 ($pK_a=6.1[32]$) acquired in acid, pH 4.98, and slightly alkaline, pH 7.62, aqueous solutions corresponding to dominant contributions of the protonated, RH⁺, and nonprotonated, R, forms of the radical, correspondingly. Significantly larger distance between outer lines of the triplet spectra observed for the alkaline solution of the nitroxide is in agreement with previously reported larger nitrogen hyperfine splitting for the R form[32]. Figure 1 illustrates the effect of protonation of the atom N-3 of the radical heterocycle resulting in decreasing unpaired electron density at the nitrogen of the N-O fragment, and, as consequence, lowered hfs for the RH⁺ form. Figure 4 shows corresponding pH dependent reversible changes of the hfs of the R1 around its $pK_a=6.1$. Note that at low EPR frequency, 563.2 MHz in Figure 3, the three lines of the spectrum are unequally spaced due to the Breit–Rabi effect[29;33]. Therefore the hfs values presented in Figure 4 were measured as the distance between the positions of the center- and high-field spectral lines.

The pH dependent DNP spectral changes allow for preferred excitation of electron paramagnetic resonances of R or RH⁺ forms of the nitroxide R1 as illustrated in Figure 5 for the phantom sample of two tubes filled with the aqueous solution of R1 titrated to alkaline and acidic pHs. The stepped variation of the EPR irradiation field, B^{EPR} , resulted in the subsequent changes in the image intensity with the maximal image intensities of alkaline (predominantly R form) and acidic (predominantly RH⁺ form) solutions when B^{EPR} is equal to 83.2 G and 84 G, respectively. Note that the observed 0.8 G difference in B^{EPR} values between the brightest images of R and RH⁺ forms is in excellent agreement with maximal pH dependent change of hyperfine splitting shown in figure 4.

Based on the data shown in Figure 5, we hypothesized that in general pH values at each pixel can be extracted from only two PEDRI acquisitions with EPR irradiation at pre-selected EPR fields. Taking into account that the ratio of concentrations of protonated and nonprotonated forms of the probe is directly related to pH ($[RH^+]/[R]=[H^+]/K_a$), we selected the values of EPR excitation fields to coincide with DNP spectral peak positions of RH⁺ and R forms of the probe.

Figure 6 shows pH dependence of the ratio of the corresponding high-field DNP signal amplitudes of RH⁺ and R forms of the R1 probe, which allows for ratiometric pH quantification in pH range from 5 to 7. To demonstrate the capacity of VF PEDRI for pH mapping the measurements were performed on the phantom consisting of four tubes, 9.5 mm inner diameter, filled with R1 solution titrated to different pH values. Figure 7a shows two PEDRI images of the phantom acquired at EPR frequency 562 MHz (200 G for the EPR center field) and excitation fields, $B_{RH^+}^{EPR} = 214.16$ G and $B_{R}^{EPR} = 214.88$ G, which correspond to the positions of high-field DNP spectral lines of RH⁺ and R forms, respectively.

The ratio of NMR signals at each pixel of these two images (pixel size of 1.25 mm) is pH dependent and was converted to the pH map shown in Fig. 7b using calibration curve. Average pH values extracted from the image data of the individual tube are in very good agreement with actual pH values. Difference between measured and actual pH does not exceed 0.07. Functional resolution was determined from standard error calculated from the variations of the pH values inside of the individual tube and did not exceed 0.1 unit of pH. Spatial resolution of 1.25 mm was calculated as ratio between image field of view and matrices size (64×64) .

DISCUSSION

The critical role of pH status in physiology and pathophysiology of living organisms is well recognized. At the microscopic level local pH drastically affects the vital activities of the cell, cellular organelles and enzymes. Recently extracellular pH_e has been identified as a significant prognostic factor not only in experimental transplantable tumor models but also in spontaneous tumors[34]. The acidic pH_e in tumors has a number of important consequences, playing a role in tumor initiation, progression, and therapy[35]. Upon therapeutic intervention, the delivery, absorption and pharmacological effectiveness of drugs can be altered by changing the pH of their local environment. Therefore, spatially and temporarily addressed pH measurements *in vivo* are of considerable clinical relevance.

For *in vivo* pH measurements, ³¹P-NMR has proven to be the most suitable noninvasive approach. However pH assessment using ³¹P-NMR and inorganic phosphate, P_i has its own limitations, which are rarely discussed, including the lack of resolution (about 0.2–0.3 pH units and even less at lower pH), or the fact that P_i concentrations vary with metabolism and ischemia, and its chemical shift depends on ionic strength[36;37]. Moreover, ³¹P-NMR using endogenous phosphate reports intracellular pH_i but is practically insensitive to extracellular pH_e[35]. Therefore several exogenous phosphorus- and fluorine-containing NMR probes were developed for pH measurement using ³¹P NMR or ¹⁹F NMR[35].

Application of exogenous probes using EPR spectroscopy has an advantage over exogenous NMR probes in sensitivity and reasonable depth of penetration in living tissues (about 1 cm for commercially available L-band spectrometers). A number of pH sensitive nontoxic nitroxide probes were developed over the last two decades[13;32;38], and have been recently applied to various biological systems, including *in vivo* pH measurements in rodents[29;39; 40]. A capacity for pH mapping of aqueous samples using pH sensitive nitroxides and spectral-spatial CW EPRI was also demonstrated for phantom samples[13;41;42]. However the requirement of long acquisition times (typically > 1 hr for 4D acquisition), high gradients (\geq 30 G/cm), and low spatial resolution make application of CW EPRI for pH mapping of living tissues difficult.

In this work we developed an approach to pH mapping of aqueous samples using pH sensitive nitroxides and VF PEDRI. In general, PEDRI with variable field EPR pre-excitation allows EPR spectroscopic information to be obtained along with the spatial information on the structure of the object and the distribution of the radical within the object, from the value of the enhancements observed at each pixel. The information obtained is equivalent to that of a 3D or 4D spectral-spatial EPR image along with a superimposed proton MRI. However, complete (EPR) spectral-spatial reconstruction from VF PEDRI requires multiple MRI acquisitions with a corresponding increase of acquisition time by tens or hundreds fold, making it comparable to 4D spectral-spatial EPRI. Fortunately, valuable spectral parameters at each pixel can be extracted from a limited number of selected PEDRI acquisitions (as little as two) with acquisition time of a few minutes or less. VF PEDRI acquisitions only at two pre-selected EPR excitation fields (70 s each, Fig. 7A) were sufficient to extract pH map with good functional (0.1 pH units) and spatial (1.25 mm) resolutions for the phantom sample with

aqueous solutions of pH sensitive nitroxide. An improvement in acquisition time is particularly important for *in vivo* applications where the experimental window and stability of the nitroxides are limited. Note also that VF PEDRI allows for slice selectivity of the functional image which is unavailable in EPRI and possesses the capacity for functional and anatomical resolution in one experimental set-up (otherwise available only in EPR/NMR co-imaging[8]).

In this work the ratiometric approach to mapping pH non-invasively has been demonstrated using a VF PEDRI and pH sensitive nitroxide that has a pK = 6.1 (Fig. 3). This is useful for pH monitoring in the physiological range or in slightly acidic conditions which are characteristic for ischemic hearts[43] or extracellular tumor microenvironments[35]. A series of pH sensitive nitroxides with enhanced stability against bioreduction and various range of pH sensitivity have been reported and might be used in specific applications[29;44;45]. Moreover, recently the first pH sensitive triarylmethyl (TAM) radical derivatives which possess an extraordinary stability *in vivo* were developed [46;47]. The amino derivatives of TAM report aqueous acidity in physiological pH range. However, further synthetic efforts are required to develop TAM derivatives based on more hydrophilic structures, such as Oxo63, to improve their aqueous solubility and avoid possible toxicity.

While the concept of functional VF PEDRI was proved using the pH probe, it can be applied for studies of other biologically relevant parameters of the medium such as redox state, concentrations of oxygen or glutathione using specifically designed probes. Recently developed TAM probe with doublet EPR spectrum resulted in enhanced sensitivity to low oxygen concentration[48]. The characteristic oxygen-sensitive EPR field positions corresponding to maximal and minimal peak intensities of the TAM doublet spectrum provide another opportunity for functional (oxygen) mapping using VF PEDRI similar to that for pH with two characteristic pH-sensitive EPR field positions. Recent spectroscopic *in vivo* application of a disulfide biradical probe to report tissue glutathione (GSH) content[49] might also be extended for GSH mapping using VF PEDRI due to the presence of characteristic "biradical" and "monoradical" components in the EPR spectrum.

In this work we employed the FC PEDRI approach with the same NMR detection field (B_0^{NMR}) and EPR evolution field which were only slightly shifted from B_0^{NMR} up or down to fit resonance EPR excitation fields of RH⁺ and R forms of the nitroxide. This avoids large field jumps from EPR evolution field to NMR detection field adding stability and decreasing ramping and stabilization times (see Fig. 2). In general, an alternative approach with stationary magnetic field but slightly different EPR radio frequencies can be proposed for functional mapping using specific paramagnetic probes. This approach, which we termed variable radio frequency (VRF) PEDRI, requires minimal instrumental modification of the fixed-field PEDRI system by inclusion of either (i) a wide-band resonator allowing EPR irradiation at close frequencies; or (ii) a dual-frequency switchable resonator. On the other hand, it improves magnetic field homogeneity and stability and decreases acquisition time by eliminating periods of ramping and stabilization of the magnetic field. Moreover, future development of a simplified VRF PEDRI system may allow for elimination of the field cycling coil and its power supplies, an increased gap in the magnet system, and the possibility to use conventional NMR gradients and gradient power supplies.

CONCLUSION

A new concept of functional mapping using VF PEDRI is proposed and experimentally verified for pH mapping using a pH sensitive nitroxide. The proposed general VF PEDRI approach can be modified for specific studies of other biologically relevant parameters of the medium such as redox state, concentrations of oxygen or glutathione using specifically designed probes.

Acknowledgments

This work was partly supported by NIH grants EB009433, CA132068, EB03519, EB00890 and EB004900.

REFERENCES

- 1. Eaton, GR.; Eaton, SE.; Ohno, K., editors. EPR imaging and in vivo EPR. Boca Raton: CRC Press; 1991.
- Kuppusamy P, Zweier JL. Cardiac applications of EPR imaging. NMR Biomed 2004;17:226–239. [PubMed: 15366025]
- Subramanian S, Krishna MC. Dancing with the Electrons: Time-Domain and CW In Vivo EPR Imaging. Magnetic Resonance Insights 2008;2:43–74.
- Matsumoto K, Subramanian S, Murugesan R, Mitchell JB, Krishna MC. Spatially resolved biologic information from in vivo EPRI, OMRI, and MRI. Antioxidants & Redox Signaling 2007;9:1125–1141. [PubMed: 17571957]
- Hyodo F, Matsumoto S, Devasahayam N, Dharmaraj C, Subramanian S, Mitchell JB, Krishna MC. Pulsed EPR imaging of nitroxides in mice. Journal of Magnetic Resonance 2009;197:181–185. [PubMed: 19157932]
- Berliner JL, Fujii H. Magnetic resonance imaging of biological specimens by electron paramagnetic resonance of nitroxide spin labels. Science 1985;227:517–519. [PubMed: 2981437]
- He G, Deng Y, Li H, Kuppusamy P, Zweier JL. EPR/NMR co-imaging for anatomic registration of free-radical images. Magnetic Resonance in Medicine 2002;47:571–578. [PubMed: 11870845]
- Samouilov A, Caia GL, Kesselring E, Petryakov S, Wasowicz T, Zweier JL. Development of a hybrid EPR/NMR coimaging system. Magn Reson Med 2007;58:156–166. [PubMed: 17659621]
- He G, Samouilov A, Kuppusamy P, Zweier JL. In vivo imaging of free radicals: applications from mouse to man. Molecular & Cellular Biochemistry 2002:359–367. 234–235. [PubMed: 12162454]
- Eaton SS, Maltempo MM, Stemp EDA, Eaton GR. 3-dimensional Electron-Paramagnetic-Res imaging with one spectral and 2 spatial dimensions. Chemical Physics Letters 1987;142:567–569.
- Halpern HJ, Yu C, Peric M, Barth E, Grdina DJ, Teicher BA. Oxymetry deep in tissues with lowfrequency electron paramagnetic resonance. Proc Natl Acad Sci U S A 1994;91:13047–13051. [PubMed: 7809170]
- Krishna MC, Kuppusamy P, Afeworki M, Zweier JL, Cook JA, Subramanian S, Mitchell JB. Development of functional electron paramagnetic resonance imaging. Breast Dis 1998;10:209–220. [PubMed: 15687576]
- 13. Khramtsov VV. Biological imaging and spectroscopy of pH. Curr. Org. Chem 2005;9:909–923.
- He G, Samouilov A, Kuppusamy P, Zweier JL. In vivo EPR imaging of the distribution and metabolism of nitroxide radicals in human skin. Journal of Magnetic Resonance 2001;148:155–164. [PubMed: 11133289]
- 15. Lurie DJ, Bussell DM, Bell LH, Mallard JR. Proton electron double magnetic resonance imaging of free radical solutions. J Magn Reson 1988;76:366–370.
- Golman K, Leunbach I, Ardenkjaer-Larsen JH, Ehnholm GJ, Wistrand LG, Petersson JS, Jarvi A, Vahasalo S. Overhauser-enhanced MR imaging (OMRI). Acta Radiol 1998;39:10–17. [PubMed: 9498861]
- Lurie DJ, Hutchison JMS, Bell LH, Nicholson I, Bussell DM, Mallard JR. Field-cycled protonelectron double resonance imaging of free radicals in large aqueous samples. J Magn Reson 1989;84:431–437.
- 18. Overhauser AW. Polarization of nuclei in metals. Phys Rev 1953;92:411-415.
- Ardenkjaer-Larsen JH, Laursen I, Leunbach I, Ehnholm G, Wistrand LG, Petersson JS, Golman K. EPR and DNP properties of certain novel single electron contrast agents intended for oximetric imaging. J Magn Reson 1998;133:1–12. [PubMed: 9654463]
- 20. Krishna MC, English S, Yamada K, Yoo J, Murugesan R, Devasahayam N, Cook JA, Golman K, Ardenkjaer-Larsen JH, Subramanian S, Mitchell JB. Overhauser enhanced magnetic resonance imaging for tumor oximetry: coregistration of tumor anatomy and tissue oxygen concentration. Proc Natl Acad Sci U S A 2002;99:2216–2221. [PubMed: 11854518]

- Li H, Deng Y, He G, Kuppusamy P, Lurie DJ, Zweier JL. Proton electron double resonance imaging of the in vivo distribution and clearance of a triaryl methyl radical in mice. Magnetic Resonance in Medicine 2002;48:530–534. [PubMed: 12210919]
- 22. Li H, He G, Deng Y, Kuppusamy P, Zweier JL. In vivo proton electron double resonance imaging of the distribution and clearance of nitroxide radicals in mice. Magnetic Resonance in Medicine 2006;55:669–675. [PubMed: 16463344]
- Lurie DJ, Davies GR, Foster MA, Hutchison JM. Field-cycled PEDRI imaging of free radicals with detection at 450 mT. Magn Reson Imaging 2005;23:175–181. [PubMed: 15833609]
- 24. Foster MA, Seimenis I, Lurie DJ. The application of PEDRI to the study of free radicals in vivo. Phys Med Biol 1998;43:1893–1897. [PubMed: 9703052]
- 25. Utsumi H. Molecular imaging of in-vivo ROS generation in oxidative diseases using ESRI and OMRI. Nippon Ronen Igakkai Zasshi 2007;44:11–16. [PubMed: 17342822]
- Grucker D, Chambron J. Oxygen imaging in perfused hearts by dynamic nuclear polarization. Magn Reson Imaging 1993;11:691–696. [PubMed: 8393951]
- Golman K, Petersson JS, Ardenkjaer-Larsen JH, Leunbach I, Wistrand LG, Ehnholm G, Liu KJ. Dynamic in vivo oxymetry using overhauser enhanced MR imaging. J Magn Reson Imaging 2000;12:929–938. [PubMed: 11105032]
- 28. Utsumi H, Yamada K, Ichikawa K, Sakai K, Kinoshita Y, Matsumoto S, Nagai M. Simultaneous molecular imaging of redox reactions monitored by Overhauser-enhanced MRI with 14N- and 15Nlabeled nitroxyl radicals. Proc Natl Acad Sci U S A 2006;103:1463–1468. [PubMed: 16432234]
- Potapenko DI, Foster MA, Lurie DJ, Kirilyuk IA, Hutchison JM, Grigor'ev IA, Bagryanskaya EG, Khramtsov VV. Real-time monitoring of drug-induced changes in the stomach acidity of living rats using improved pH-sensitive nitroxides and low-field EPR techniques. J Magn Reson 2006;182:1– 11. [PubMed: 16798033]
- Volodarsky, LB.; Grigor'ev, IA. Synthesis of heterocyclic nitroxides. In: Volodarsky, LB., editor. Imidazoline Nitroxides. Boca Raton: CRC Press; 1988. p. 5-28.
- Petryakov S, Samouilov A, Roytenberg M, Li H, Zweier JL. Modified Alderman-Grant resonator with high-power stability for proton electron double resonance imaging. Magn Reson Med 2006;56:654–659. [PubMed: 16902975]
- Khramtsov VV, Grigor'ev IA, Foster MA, Lurie DJ, Nicholson I. Biological applications of spin pH probes. Cell Mol Biol 2000;46:1361–1374. [PubMed: 11156481]
- 33. Breit G, Rabi II. Measurement of nuclear spin. Phys Rev 1931;38:2082-2083.
- 34. Lora-Michiels M, Yu D, Sanders L, Poulson JM, Azuma C, Case B, Vujaskovic Z, Thrall DE, Charles HC, Dewhirst MW. Extracellular pH and P-31 magnetic resonance spectroscopic variables are related to outcome in canine soft tissue sarcomas treated with thermoradiotherapy. Clin Cancer Res 2006;12:5733–5740. [PubMed: 17020978]
- 35. Gillies RJ, Raghunand N, Garcia-Martin ML, Gatenby RA. pH imaging. A review of pH measurement methods and applications in cancers. IEEE Eng Med Biol Mag 2004;23:57–64. [PubMed: 15565800]
- 36. Pietri S, Martel S, Culcasi M, Delmas-Beauvieux MC, Canioni P, Gallis JL. Use of diethyl(2methylpyrrolidin-2-yl)phosphonate as a highly sensitive extra- and intracellular 31P NMR pH indicator in isolated organs. Direct NMR evidence of acidic compartments in the ischemic and reperfused rat liver. J Biol Chem 2001;276:1750–1758. [PubMed: 11013264]
- 37. Gillies, RJ.; Alger, JR.; den Hollander, JA.; Shulman, RG. Intracellular pH measured by NMR: methods and results. In: Nuccitelli, R.; Deamer, DW., editors. Intracellular pH: Its Measurement, Regulation and Utilization in Cellular Functions. New York: Alan R. Liss; 1982. p. 79-104.
- 38. Khramtsov, VV.; Volodarsky, LB. Use of imidazoline nitroxides in studies of chemical reactions. ESR measurements of the concentration and reactivity of protons, thiols and nitric oxide. In: Berliner, LJ., editor. Spin labeling. The next Millennium. New York: Plenum Press; 1998. p. 109-180.
- Mader K, Gallez B, Liu KJ, Swartz HM. Non-invasive in vivo characterization of release processes in biodegradable polymers by low-frequency electron paramagnetic resonance spectroscopy. Biomaterials 1996;17:457–461. [PubMed: 8938242]
- 40. Foster MA, Grigor'ev IA, Lurie DJ, Khramtsov VV, McCallum S, Panagiotelis I, Hutchison JM, Koptioug A, Nicholson I. In vivo detection of a pH-sensitive nitroxide in the rat stomach by low-field ESR-based techniques. Magn Reson Med 2003;49:558–567. [PubMed: 12594760]

- Sotgiu A, Mader K, Placidi G, Colacicchi S, Ursini CL, Alecci M. pH-sensitive imaging by lowfrequency EPR: a model study for biological applications. Phys Med Biol 1998;43:1921–1930. [PubMed: 9703055]
- 42. Khramtsov VV, Grigor'ev IA, Foster MA, Lurie DJ, Zweier JL, Kuppusamy P. Spin pH and SH probes: enhancing functionality of EPR-based techniques. Spectroscopy 2004;18:213–225.
- Zweier JL, Wang P, Samouilov A, Kuppusamy P. Enzyme-independent formation of nitric oxide in biological tissues. Nat Med 1995;1:804–809. [PubMed: 7585184]
- 44. Kirilyuk IA, Bobko AA, Grigor'ev IA, Khramtsov VV. Synthesis of the tetraethyl substituted pHsensitive nitroxides of imidazole series with enhanced stability towards reduction. Org Biomol Chem 2004;2:1025–1030. [PubMed: 15034626]
- 45. Woldman YY, Semenov SV, Bobko AA, Kirilyuk IA, Polienko JF, Voinov MA, Bagryanskaya EG, Khramtsov VV. Design of liposome-based pH sensitive nanoSPIN probes: nano-sized particles with incorporated nitroxides. Analyst. 2009 DOI: 10.1039/b818184e.
- 46. Bobko AA, Dhimitruka I, Zweier JL, Khramtsov VV. Trityl radicals as persistent dual function pH and oxygen probes for in vivo electron paramagnetic resonance spectroscopy and imaging: concept and experiment. J Am Chem Soc 2007;129:7240–7241. [PubMed: 17511458]
- Dhimitruka I, Bobko AA, Hadad CM, Zweier JL, Khramtsov VV. Synthesis and characterization of amino derivatives of persistent trityl radicals as dual function pH and oxygen paramagnetic probes. J Am Chem Soc 2008;130:10780–10787. [PubMed: 18636723]
- Bobko AA, Dhimitruka I, Eubank TD, Marsh CB, Zweier JL, Khramtsov VV. Synthesis and characterization of trityl-based EPR probe with enhanced sensitivity to oxygen. Free Rad. Biol. Med. 2009 submitted.
- Roshchupkina GI, Bobko AA, Bratasz A, Reznikov VA, Kuppusamy P, Khramtsov VV. In vivo EPR measurement of glutathione in tumor-bearing mice using improved disulfide biradical probe, Free Radic. Biol. Medicine. Free Rad. Biol. Med 2008;45:312–320. [PubMed: 18468522]

Khramtsov et al.



Figure 1.

Protonation of imidazoline pH-sensitive radical, R1. Two main resonance structures are shown illustrating the favored structure with higher unpaired electron density on nitrogen atom N-1 in the nonprotonated form.









Figure 3.

FC-DNP spectra of the nitroxide R1 obtained in phosphate–citrate buffer (10 mM each) at pH 7.62 (a) and pH 4.98 (b). Sample volume was 5 ml. Frequency of EPR irradiation 563.2 MHz, 500 ms, TR 1140ms; NEX 1; Step size = 0.04 G; P =0.8 W. All other settings were as described in Materials and Methods. A dotted line is extended from low- and high-field peaks of the spectra (b) to aid the eye.



Figure 4.

pH dependence of the hypefine splitting of the nitroxide R1 measured as the distance between the center- and high-field spectral lines of the DNP spectra. The solid line is a nonlinear least-square fit of the data to a conventional titration curve yielding hfs(RH⁺)=14.21 G, hfs(R)=15.03 G and $pK_a=6.1$.



pH=3.0 pH=9.2



Figure 5.

Sequence of PEDRI images of the phantom sample of a pair of tubes, 12 mm diameter, containing 1 mM aqueous solutions of the R1 probe at pH 9 (left tube) and pH 2 (right tube). Images were acquired at EPR frequency 282 MHz (\approx 100 G for the EPR center field) with evolution field stepped in the range from 83.0 G to 84.4 G around the position of the low-field EPR component of the R1 triplet spectrum. The observed variation of the image intensity with the shift in EPR irradiation field, B^{EPR} (see Fig.1), illustrates the subsequent changes with the maximal image intensity of R form (left tube) and RH⁺ form (right tube) when B^{EPR} is equal to 83.2 G and 84 G, respectively. The PEDRI scan parameters were: TR, 1.1 s; TE, 20 ms; flip angle, 90⁰; matrix, 64×64; NEX, 1; FOV, 80 × 80 mm; acquisition time, 70 s.

Khramtsov et al.

Page 15



Figure 6.

pH dependence of the ratio of the high-field DNP signal amplitudes measured at EPR excitation field corresponding to maximal intensities of the RH⁺ and R forms of the R1 probe. DNP spectra were acquired at EPR frequency 562 MHz (≈ 200 G for the EPR center field), EPR excitation fields were equal to $B_{_{RH^+}}^{EPR}$ =214.16 G and $B_{_{R}}^{EPR}$ =214.88 G.



Figure 7.

A. pH phantom and its PEDRI images acquired at B_{RH+}^{EPR} =214.16 G (Left) and B_{R}^{EPR} =214.88 G (Right). The PEDRI scan parameters were: TR, 1.1 s; TE, 14 ms; flip angle, 90⁰; matrix, 64×64; NEX, 1; FOV, 80×80 mm, acquisition time, 71 s; NMR frequency, 856 KHz. **B.** VF PEDRI: proof of concept of functional imaging. pH map of phantom was calculated from two PEDRI images acquired at pre-selected EPR excitation fields as shown on panel A. Averaged values of pH are given near corresponding tube; functional resolution was determined from standard error calculated from the variations of the pH values inside of the individual tube and did not

exceed 0.1 unit of pH. Spatial resolution of 1.25 mm was calculated as ratio between image field of view and matrices size (64×64) .