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Development of (S)-N⁶-(2-(4-(Isoquinolin-1-yl)piperazin-1-yl)ethyl)-N⁶-propyl-4,5,6,7-tetrahydrobenzo[d]thiazole-2,6-diamine and its analogue as a D3 receptor preferring agonist: Potent in vivo activity in Parkinson's disease animal models

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Abstract

Here we report structure-activity relationship study of a novel hybrid series of compounds where structural alteration of aromatic hydrophobic moieties connected to the piperazine ring and bioisosteric replacement of the aromatic tetralin moieties were carried out. Binding assays were carried out with HEK-293 cells expressing either D2 or D3 receptors with tritiated spiperone to evaluate inhibition constants (K_i). Functional activity of selected compounds in stimulating GTPγS binding was assessed with CHO cells expressing human D2 receptors and AtT-20 cells expressing human D3 receptors. SAR results identified compound (–)-**24c** (D-301) as one of the lead molecules with preferential agonist activity for D3 receptor (EC₅₀ (GTPγS); D3 = 0.52 nM; D2/D3 (EC₅₀): 223). Compounds (–)-**24b** and (–)-**24c** exhibited potent radical scavenging activity. The two lead compounds (–)-**24b** and (–)-**24c** exhibited high in vivo activity in two Parkinson's disease (PD) animal models, reserpinized rat model and 6-OH-DA induced unilaterally lesioned rat model. Future studies will explore potential use of these compounds in the neuroprotective therapy for PD.

Introduction

Parkinson's Disease (PD) is a progressive neurodegenerative disorder characterized by degeneration of the nigrostriatal dopaminergic pathway. It is estimated that PD affects approximately 1 % of people older than 65 years of age. It is primarily a sporadic disorder although a rare subset of population (<10%) acquires this disease due to several genetic defects 1-2. Some of the symptoms associated with PD involve rigidity, bradykinesia, resting tremor and postural instability along with cognitive and psychiatric complications 3-5. The neuropathological hallmark of PD is the presence of Lewy bodies (LB) in the surviving neurons of the substantia nigra 6, 7. The etiology of PD is not fully understood. Both oxidative stress and mitochondrial dysfunction have been strongly implicated in cell death 8-10. Recently, α -synuclein, a presynaptic protein involved in fibrilization, has been implicated in the pathogenesis of PD 7, 11. In a rare familial form of PD, a mutation in the α -synuclein gene has been linked to autosomal dominant PD 12, 13.

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Supporting Information Available: Elemental analysis data for all final targets is available. This material is available free of charge via the internet at <http://pubs.acs.org>.

Levo-DOPA (L-DOPA) has proven to be one of the mainstay therapy for PD. However, prolong use of L-DOPA gives rise to motor fluctuations with dyskinesias and the decrease in duration of response to a given L-dopa dose¹⁴. Prolong use of L-dopa also gives rise to “on” and “off” episodes resulting in additional complications. Dopaminergic agents have been used more extensively in the therapy for PD than any other class of drug molecules besides L-DOPA^{15–19}. The dopamine receptor belongs to the class of seven-member G-protein coupled receptors and has been divided into two main classes as D1-like and D2-like. D1-like receptors include D1 and D5 subtypes whereas D2, D3 and D4 subtype receptors fall in the D2-like category^{20–26}. This classification is based on distinct physiological and pharmacological properties of these two receptors as stimulation of the D1-type receptor leads to activation of adenylate cyclase, which promotes synthesis of cAMP, whereas D2-type receptor activation leads to inhibition of adenylate cyclase activity. In the CNS, D1-type receptors are located post-synaptically, whereas D2-type receptors are located both pre- and post-synaptically and have a high affinity for dopamine²⁷.

Due to the high degree of homology between D2 and D3 receptors, especially in the transmembrane domain binding regions for agonists, it has been challenging to develop agonist with highly selectivity for D3 receptors.^{28, 29} D3 preferring agonists have been shown to be neuroprotective in both cell culture and in vivo experiments.^{30–33} Some of the most selective agonist are pramipexole and ropinirole which are used as therapeutic agents in PD.^{34, 35} However, the neuroprotective property of pramipexole has been shown to be complex and mediated by both dopamine- and non-dopamine-dependent pathways.^{33, 36} Recently another D3 preferring agonist, [(+)-*trans*-3,4,4a,5,6,10*b*-hexahydro-9-carbamoyl-4-propyl-2*H*-naphth [1,2-*b*]-1,4-oxazine] (R,R-S32504) has been developed and shown to possess antiparkinsonian and neuroprotective effect.³⁷ Interestingly, pramipexole was also shown to be a potent antidepressant agent in PD patient, speculated to be due to its interaction with the D3 receptor in the mesolimbic dopamine pathway which controls behavior and mood^{38, 39}. Such antidepressant effect was found to be beneficial for PD patients. Besides developing agonists, there is now a growing interest in developing selective antagonist for the D3 receptor as therapeutic agents for psychiatric diseases and drugs of abuse. Consequently, a large number of antagonists have been developed so far which exhibited varying degree of selectivities.^{28–40}

It is increasingly evident that for a complex disease such as PD, a drug targeting only one target site will only partially address the therapeutic need of the disease. Thus it is hypothesized that multi-functional drugs having multiple pharmacological activities will be an effective disease modifying agent in case of PD.⁴¹ In our effort to produce such agent, we have undertaken one of our goals of introducing antioxidant property in our D2/D3 hybrid agonist template which we established earlier^{42, 43}. Recently, we have reported the development of potent D3 preferring agonists, which exhibited potent in vivo activity in PD animal model experiments^{44, 45}. It is believed that the neuroprotective property of pramipexole might originate from its preferential D3 agonist property coupled with its antioxidant activity, although some other mechanisms have also been implicated^{46, 47}. We hypothesize that a combination of D3 preferential agonist property along with antioxidant activity in a compound enhances its neuroprotective effect as it will address an underlying oxidative stress in the Parkinsonian brain. The current SAR study expands this line of inquiry by including compounds having bulky hydrophobic N-aryl substitutions in the piperazine moiety to explore and map out the binding site interaction further. Our earlier report described the development of a 2-aminothiazole based, high-affinity D3 selective agonist **2** (D-264)⁴⁴, which is currently undergoing in vivo evaluation to determine its neuroprotective effect. We introduced quinolines and isoquinolines in the hybrid structure as analogues of **2**. Future studies will indicate whether these derivatives will have better pharmacokinetics properties compared to **2**.

Chemistry

Scheme 1 outlines the syntheses of different aryl piperazines **8a-f** used in the following schemes to synthesize the target compounds. Compound **8a** and **8b** were synthesized by palladium (II) catalyzed amination reaction with two corresponding aryl halides 1 or 2-bromonaphthalene with excess piperazine⁴⁸. **8c** was prepared from commercially available 4-hydroxyquinoline by chlorination reaction with POCl₃ followed by reflux in *i*-propanol with excess piperazine for 4 h⁴⁹. Compound **8d** was synthesized from commercially available 5-nitroquinoline by reduction in presence of SnCl₂ in ethanol under refluxing condition for 1.5 h⁵⁰ followed by reaction of the amino compound **6** with bis-(2-chloroethyl)amine HCl in presence of K₂CO₃ under reflux for 48 h⁵¹. **8e** was synthesized from 1-chloroisoquinoline by *i*-propanol reflux in presence of excess piperazine. Compound **8f** was synthesized by reaction of boronic acid intermediate **7b** with 4-bromopyridine in presence of palladium catalyst.

Scheme 2 depicts the synthesis of two target compounds **14a-b**. Aryl piperazine **8a-b** were *N*-alkylated with chloroacetonitrile under reflux in presence of K₂CO₃ as base to synthesize intermediate **9a-b**, followed by raney nickel catalyzed hydrogenation to yield **10a-b** which was subjected to reductive amination condition in presence of 7-methoxy-2-tetralone to yield intermediates **11a-b**. *N*-alkylation of **11a-b** with propionyl chloride in presence of triethyl amine as base yielded **12a-b** which was reduced to the corresponding amines **13a-b** using LiAlH₄ as reducing agent. The target compounds **14a-b** were obtained by demethylation of **13a-b** using 1M solution of borontribromide in dichloromethane at -40 °C to room temperature for overnight.

The syntheses of compounds **20a-d** are shown in Scheme 3. 5- or 7- Methoxy 2-tetralone was reacted with *n*-propyl amine under reductive amination condition to yield **16a-b**. Enantiomerically pure (-)-**16b** was obtained from racemic **16b** using a synthetic chiral resolving agent following previously reported procedure^{52,45}. *N*-alkylation of **16a-b** and (-)-**16b** using chloroacetyl chloride in presence of triethyl amine produced intermediate **17a-b** and (-)-**17b** which were coupled with different aryl piperazines **8c-e** to yield **18a-d** and (-)-**18d** in moderate to good yield. Reduction of the amide group of **18a-d** using LiAlH₄ followed by demethylation using boron tribromide at -40 °C yielded the target compounds **20a-d** and (-)-**20d**.

Scheme 4 depicts the synthesis of (S)-(-)-pramipexole hybrid analogs **24a-24d**. Alkylation of **8c-f** with 2-bromoethanol produced **21a-d**, which were oxidized under Swern oxidation conditions to the arylpiperazine acetaldehydes **22a-d**. S-(-)-Pramipexole (**23**) was further condensed with **22a-d** under reductive amination conditions (NaCNBH₃ as the reducing agent) to give hybrid analogs **24a-d**.

Discussion

Our earlier SAR studies with hybrid compounds produced ligands which exhibited potent and selective agonist activity at D3 receptor. Our two starting piperazine hybrid compounds based on 7-OH-DPAT and 5-OH-DPAT were **1** (D-315)⁵³ and **4** (D-237)⁴⁵. These first generation hybrid compounds maintained similar affinities and selectivities for D2/D3 receptors as their parent non-hybrid counter part. Further expansion of SAR studies led to the development of compounds **2** and **3** (D-366)⁵³ which exhibited high and selective affinity for the D3 receptor. **4** and **3** were among the most potent and selective agonist for the D3 receptor developed so far. Our SAR studies indicated that the *N*-piperazinyl substituent could be modified by introducing biphenyl and indole substituents without compromising affinity for the D3 receptor. It is important to point out here that for development of D3 antagonist one of the frequently used molecular template consists of amide-piperazine moiety separated via a linker

where bulky aromatic substituents have been introduced mainly on the amide moiety.²⁸ The compound BP 897 is one of the prototypical example of a compound developed from this template.

In our current SAR study we wanted to see the influence of different piperazynyl *N*-aryl or heteroaryl moieties on interaction with dopamine D2 and D3 receptors. Our previous report indicated that bulky hydrophobic substituents attached to the piperazine ring are well tolerated as **2** and **3** retained affinity and were selective for D3. Here we wanted to explore this further with naphthalene and quinoline substituents. Our initial design to incorporate naphthalen-1-yl and naphthalene-2-yl substituents led to development of compounds **14a** and **14b**. Compound **14a** showed a 3- to 4-fold increase in binding affinity for D2/D3 (K_i for D2 = 13.5 nM, K_i for D3 = 0.435 nM) accompanied by increased selectivity for D3 (D2/D3 = 31.0) compared to **1**. Compound **14b** exhibited a binding affinity (K_i for D2 = 8.85 nM, K_i for D3 = 0.685 nM) similar to that of **14a**, however, its selectivity (D2/D3 = 12.9) was reduced compared to that of **1** and **14a**. This result indicated that a bulkier *N*-4-substituted group in the piperazine ring is well tolerated by dopamine D2/D3 receptors. Next, we wanted to introduce a heteroatom in the naphthalene moiety producing different quinoline derivatives. Compound **20a** with an *N*-quinoline-4-yl substitution in the piperazine ring displayed no further increase in binding potency (K_i for D2 = 7.08 ± 0.34 , K_i for D3 = 2.65 ± 0.65 nM) compared to **14a** or **14b**, but its binding potency was still greater than that of **1** although its selectivity for D3 over D2 was reduced (D2/D3 = 2.67). The other regioisomer **20b** which is a quinoline-5-yl derivative, exhibited comparable low nanomolar binding affinity for D2/D3 receptors (K_i for D2 = 5.22 nM, K_i for D3 = 1.27 nM). Both **20a** and **20b** were not highly selective for the D3 receptor. Another compound of this series was **20c**, an isomeric isoquinoline-1-yl derivative displaying subnanomolar binding potency at D3 receptors (K_i = 0.44 nM) while it was less potent at D2 receptors (K_i = 21.7 nM) compared to **20a** and **20b**. Consequently, **20c** exhibited the highest selectivity for D3 receptors (D2/D3 = 49.2) in this series of compounds. Subsequently, the positional effect of an aromatic 5-hydroxyl functionality in the aminotetraline moiety was evaluated on binding affinity and selectivity. Thus, compounds **20d** and its *s*(-) enantiomers (**20e**) were synthesized. Compound **20d** showed increased binding potency at both D2 and D3 receptors (K_i for D2 = 4.89 nM, K_i for D3 = 0.40 nM) although its selectivity for D3 receptors (D2/D3 = 12.22) did not increase appreciably. The enantiomerically pure compound **20e** exhibited increased potency at both D2 (K_i = 3.74 nM) and D3 (K_i = 0.19 nM) receptors compared with its racemic counter part, with comparable selectivity for D3 over D2 receptors (D2/D3 = 19.68).

In one of our earlier publications, we reported compound **2** as one of the most potent and selective compounds for the D3 receptor; this compound contains a 2-aminothiazolidinium moiety as bio-isosteric replacement of the hydroxyl phenyl moiety in the aminotetraline fragment of the hybrid structure. In designing the next set of compounds **24a-d**, we incorporated a 2-aminothiazolidinium moiety instead. In addition to probing the D3 receptor selectivity for these derivatives, we aimed for improved pharmacokinetic properties by the introduction of the more polar quinoline moiety. Specifically, the (-)-isomers of these compounds were made as we have shown previously that activity is increased in the (-)-isomers in the 2-aminothiazolidinium series of compounds⁴⁴. Compound **24a** which is a quinoline-4-yl analog, exhibited moderate binding potency for D2 receptors (K_i = 109 nM) while the binding affinity toward D3 (K_i = 2.61 nM) receptor was in the low nanomolar range with selectivity for D3 over D2 receptors (D2/D3 = 41.8) higher than that for the corresponding **20a**. Compound **24c**, an isoquinoline-1-yl derivative, exhibited approximately 2-fold less potency at D2 (K_i = 269 nM) and identical affinity at D3 (K_i = 2.23 nM) receptors compared to **24a**. Thus, compound **24c** displayed the highest selectivity (D2/D3 = 121) for D3 receptor in the current series of compounds. The other thiazolidinium compound in this series was quinoline-5-yl derivative **24b**, which displayed high binding affinity for both D2 and D3 receptors (K_i for D2

= 57.7 nM, K_i for D3 = 1.21 nM) with appreciable selectivity for D3 over D2 receptors (D2/D3 = 47.7). Next we designed and synthesized compound **24d** consisting of a linearly fused 4-(pyridine-4-yl)-phenyl moiety. This compound could retain its binding affinity at D2 and D3 receptors (K_i = 270 nM and 4.78 nM for D2 and D3 receptors, respectively) and was moderately selective for D3 over D2 receptors (D2/D3 = 56.5). Collectively, the current results are consonant with the existence of hydrophobic interactions resulting from bulky substitutions on the piperazine N-atom.

Following binding analysis, selected compounds **24c**, **24b** and **24a** were subjected to the [³⁵S]GTP γ S functional assay for D2 and D3 receptors and compared with the full agonist dopamine. The assays were carried out with the cloned human D2 and D3 receptors expressed in CHO and AtT cells as described by us earlier⁴⁵. All three compounds exhibited high potency for the D3 receptor in the subnanomolar range (EC₅₀; 0.52, 0.52 and 0.49 nM, respectively) (Table 2). Similar to what was found in the D2 binding assays, isoquinoline derivative **24c** exhibited the lowest potency for D2 receptors in the functional assays with [³⁵S]GTP γ S (EC₅₀; 116 nM); it was also the most selective for D3 (D2/D3 (EC₅₀ ratio) = 223) and this selectivity was comparable to our previously developed **2** (D2/D3 (EC₅₀ ratio) = 248). Like compounds **24b** and **24c**, compound **24a** was potent at D3 (EC₅₀: 0.49 nM) and was less potent at D2 (EC₅₀: 77 nM). Compound **24a** was second best selective compound for D3 receptor. Compound **24b** displayed the highest potency in activating D2 receptors (EC₅₀; 13.5 nM) and was the least selective in activating D3 receptors. In comparison, the reference compound ropinirole displayed much less selectivity for and potency at D3 receptors. As demonstrated by plateaus of activation of 80–119% of maximal stimulation by DA, all compounds appeared to exert close to full agonism in the [³⁵S]GTP γ S assay (Table 2).

Evaluation of free radical scavenging activity

Scavenging of DPPH (1,1-Diphenyl-2-picryl-hydrazyl) radical by **24c**, **24b** and ascorbic acid was carried out based on published method and is shown in Figure 2.⁵⁴ The scavenging effect is expressed as percent of control. As shown in Figure 2, all three compounds inhibited DPPH radical activity dose dependently. The standard compound ascorbic acid had an IC₅₀ of 34.6 ± 3.6 μ M in this assay procedure whereas the IC₅₀ value for **24b** was 39.67 ± 6.23 μ M and for **24c** 45.67 ± 6.89 μ M (n=3 for all compounds). It is evident from the results that the two test compounds are as potent as potent known antioxidant ascorbic acid.

Reversal of reserpine-Induced hypolocomotion in Rats by **24a**, **24b**, **24c** and ropinirole

Reserpine induces depletion of catecholamines in nerve terminals resulting in a cataleptic condition in rats, which is a well established animal model for PD⁵⁵. Significant reduction of locomotion of the rats was observed 18 h after the administration of reserpine (5 mg/kg, s.c.) which indicated the development of akinesia in rats. Compounds **24c** and **24b** were highly efficacious in producing complete reversal of reserpine induced hypolocomotion. At a dose of 10 μ mol/kg (6.5 mg/kg), s.c., **24b** not only reversed the reserpine-induced hypokinetic condition to a level that was similar or higher than that in control (vehicle treated non-reserpinized rats) rats, but also maintained an increased level of locomotor activity of the reserpinized rats significantly throughout the 6-h period (Fig. 3A). Subcutaneous administration of 10 μ Mol/kg (6.3 mg/kg) **24c** and 10 μ Mol/kg (2.96 mg/kg) ropinirole could restore the locomotion to the normal state. The reference drug ropinirole at a dose of 10 μ Mol/kg s.c. exhibited a much shorter duration of action compared to **24b**; the peak of action of ropinirole was reached within 30 min and the pharmacological action ceased after 150 min. It is evident that out of the three compounds, **24b** and **24c** exhibited the highest CNS locomotor activity (Figure 3A). On the other hand, compound **24a** (10 μ Mol/kg) was much less active

compared to **24b** and **24c**. Cumulative horizontal activity data for 6 h period of observation for saline-control meaning animals receive saline instead of reserpine, reserpinized control, and 10 $\mu\text{Mol/kg}$ of ropinirole, **24a**, **24b** and **24c** is presented as a bar graph in Figure 3B.

Effect of **24b**, **24c** and ropinirole in 6-OHDA lesioned rats

Based on the above locomotor activity results, compounds **24b** and **24c** were selected for in vivo evaluation in rats carrying a unilateral lesion in the medial forebrain bundle induced by application of the neurotoxin 6-hydroxydopamine (6-OHDA). This results in destruction of dopamine neurons and development of supersensitivity of dopamine receptors on the lesioned side. Such surgically modified rats when challenged with direct acting dopamine agonists, produce contralateral rotations away from the lesioned side. This rat model is considered to be one of the standard models for preclinical screening of drugs for possible antiparkinsonian property.⁵⁶ Both compounds **24b** and **24c** produced potent rotational activity in a dose dependent manner when administered subcutaneously. Thus, **24b** produced dose dependently contralateral rotations with the highest dose (10 $\mu\text{Mol/kg}$, 6.5 mg/kg) exhibiting the longest duration of action (> 10 h). At a lower dose (5 $\mu\text{Mol/kg}$, 3.28 mg/kg), **24b** reached its peak effect at around 5 h and produced a number of total rotations of 2295 whereas at a higher dose (10 $\mu\text{Mol/kg}$) the peak effect was reached at around 7.5 h with 2648 total rotations (Figure 4). It was clear that **24b** was more efficacious in producing rotations at a higher dose. However, in the first 360 min, the lower dose of **24b** produced significant higher rotations compared to higher dose, indicating possible involvement of stereotypy at a higher dose. In comparison, compound **24c** when tested at a dose of 10 $\mu\text{Mol/kg}$ (6.3 mg/kg), produced an even higher number of total rotations of 3688 with a long duration of action (>8 h), Figure 5. The peak activity for **24c** was reached at about 7 h. At the lower dose of 5 $\mu\text{Mol/kg}$ (3.16 mg/kg), compound **24c** produced a lower number of rotations (2805) which lasted for 7 h (Figure 5). It is evident that contrary to stimulating locomotor activity in reserpinized rats, **24c** was more efficacious compared to **24b** in regards to total number of rotations produced by respective doses. However, **24b** exhibited relatively longer duration of action. The reference drug ropinirole at 5 and 10 $\mu\text{Mol/kg}$ exhibited shorter duration of action and was less active in producing rotations compared to **24b** and **24c** at the same respective doses.

Conclusion

The present study describes the development of compounds with high affinity and selectivity for dopamine D3 receptors. SAR results have demonstrated that Isoquinoline derivatives have higher selectivity for D3 receptor. In both binding and functional assays compound isoquinoline derivative **24c** exhibited the highest selectivity for the D3 over D2 receptor. Lead compounds **24b** and **24c** also exhibited potent free radical quenching property, possibly indicating antioxidant activity. The two lead compounds were tested in two PD animal models. Compound **24b** was the most potent in the reserpinized animal model where it exhibited appreciable locomotor activity, while compound **24c** was as active as ropinirole in this animal model. However, in the 6-OH-DA animal model studies, **24c** produced the highest number of rotations and was the most active among the three compounds. Compound **24b** produced a slightly longer duration of action. Both compounds were more efficacious than ropinirole. Relative contribution of D2 receptor vs. D3 receptor in reversing reserpine effect and to induce rotations by compounds **24b** and **24c** will be assessed in future. Compounds **24b** and **24c** will next be studied in the in vitro and in vivo neuroprotection experiment model to evaluate their potential as a neuroprotective treatment agent for PD.

Experimental Section

Reagents and solvents were purchased from commercial suppliers and used as received unless otherwise indicated. Dry solvent was obtained according to the standard procedure. All

reactions were performed under inert atmosphere (N_2) unless otherwise noted. Analytical silica gel 60 F₂₅₄-coated TLC plates were obtained from EMD Chemicals, Inc. and were visualized with UV light or by treatment with phosphomolybdic acid (PMA), Dragendorff's reagent or ninhydrin. Flash column chromatographic purifications were performed using Whatman Purasil® 60A silica gel 230–400 mesh. The proton nuclear magnetic resonance (1H NMR) spectra were measured on Varian 400 MHz FT NMR spectrometer using tetramethylsilane (TMS) as an internal standard. The NMR solvent used was $CDCl_3$ as indicated. Optical rotations were recorded on Perkin-Elmer 241 polarimeter. Melting points were recorded using MEL-TEMP II (Laboratory Devices Inc., USA) capillary melting point apparatus and were uncorrected. Elemental analyses were performed by Atlantic Microlab, Inc. and were within $\pm 0.4\%$ of the theoretical value.

Procedure A. Preparation of 1-(naphthalen-1-yl)piperazine (8a)

Into a solution of 1-bromo naphthalene (3 g, 14.5 mmol) and piperazine (9.98 g, 115.6 mmol) in 50 ml of diglyme was added Na-*t*-butoxide (4.89 g, 43.46 mmol). The reaction mixture was stirred for few minutes before the addition of palladium catalyst, dichlorobis(tri-*o*-tolylphosphine) palladium(II) (0.724 g, 0.724 mmol) and refluxed at 170 °C for 48 h. The reaction mixture was cooled and the diglyme was evaporated under reduced pressure. The solid residue was partitioned between ethyl acetate and water. The aqueous layer was extracted with ethyl acetate (3 \times 100 ml). The combined organic layer was dried (Na_2SO_4), evaporated under reduced pressure to get the crude product which was purified by column chromatography (ethyl acetate : MeOH 9:1) to yield 1.94 g of pure yellowish oily compound **8a** (63 %). 1H NMR (400 MHz, $CDCl_3$): δ ppm 7.68–7.74 (m, 2H), 7.54–7.56 (d, 1H, $J = 8$ Hz), 7.45–7.48 (t, 1H, $J = 6$ Hz), 7.38–7.42 (m, 1H), 7.26–7.30 (t, 1H, $J = 8$ Hz), 7.08–7.14 (dd, 1H, $J = 8$ Hz); 3.08–3.33 (m, 8H).

1-(naphthalen-2-yl)piperazine (8b)

Compound **8b** was synthesized from 2-bromo naphthalene (1.8 g, 8.9 mmol) and piperazine (6.13 g, 71.17 mmol) according to the procedure A to afford 1.88 g of oily pure compound **8b** (63.55 %). 1H NMR (400 MHz, $CDCl_3$): δ ppm 7.72–7.73 (d, 1H, $J = 4$ Hz); 7.70–7.71 (d, 1H, $J = 4$ Hz); 7.68 (s, 1H); 7.37–7.41 (t, 1H, $J = 8$ Hz); 7.25–7.30 (m, 2H); 7.11–7.12 (d, 1H, $J = 4$ Hz); 3.09–3.33 (m, 8H).

Procedure B. Preparation of 2-(4-(naphthalen-1-yl)piperazin-1-yl)acetonitrile (9a)

A suspension of 1-(naphthalen-1-yl)-piperazine (**8a**) (1.94 g, 9.13 mmol), potassium carbonate (3.77 g, 27.39 mmol), and 2-chloroacetonitrile (1.16 mL, 18.26 mmol) in toluene was refluxed for 3 h. Toluene was removed under reduced pressure, and the residue was diluted with ethyl acetate, washed with water and brine, dried over sodium sulfate (Na_2SO_4), concentrated, and purified by column chromatography (ethyl acetate/hexane = 1:1) to afford the product **9a** as a thick yellow solid (1.95 g, 85%): 1H NMR (400 MHz, $CDCl_3$) δ ppm 7.68–7.64 (m, 2H); 7.54–7.56 (d, 1H, $J = 8$ Hz); 7.45–7.48 (t, 1H, $J = 6$ Hz); 7.38–7.42 (m, 1H); 7.26–7.30 (t, 1H, $J = 8$ Hz); 7.08–7.14 (dd, 1H, $J = 8$ Hz); 3.60 (s, 2H); 3.33–3.36 (t, 4H, $J = 6$ Hz); 2.81–2.83 (t, 4H, $J = 4$ Hz).

2-(4-(naphthalen-2-yl)piperazin-1-yl)acetonitrile (9b)

Compound **9b** was synthesized from 1-(naphthalen-2-yl)-piperazine (**8b**) (1.2 g, 5.6 mmol) and 2-chloropropionitrile (1.08 mL, 16.96 mmol) according to the procedure B to afford product **9b** as a thick yellow solid (1.0 g, 83%): 1H NMR (400 MHz, $CDCl_3$) δ ppm 7.72–7.73 (d, 1H, $J = 4$ Hz); 7.70–7.71 (d, 1H, $J = 4$ Hz); 7.68 (s, 1H); 7.37–7.41 (t, 1H, $J = 8$ Hz); 7.25–7.30 (m, 2H); 7.11–7.12 (d, 1H, $J = 4$ Hz); 3.33–3.36 (t, 4H, $J = 6$ Hz); 2.81–2.83 (t, 4H, $J = 4$ Hz).

Procedure C. Preparation of 2-(4-(naphthalen-1-yl)piperazin-1-yl)ethanamine (10a)

A solution of compound **9a** in methanol (0.86 g, 3.42 mmol) was hydrogenated in a parr hydrogenator apparatus in the presence of raney nickel catalyst at a pressure of 60 psi for 8 h. The reaction mixture was passed through celite, dried over Na₂SO₄, evaporated, and purified over a silica gel column using the solvent system ethyl acetate/methanol/triethylamine (80:15:5) to afford compound **10a** as transparent thick liquid (0.823 g, 94%): ¹H NMR (400 MHz, CDCl₃) δ ppm 7.68–7.64 (m, 2H); 7.54–7.56 (d, 1H, *J* = 8 Hz); 7.45–7.48 (t, 1H, *J* = 6 Hz); 7.38–7.42 (m, 1H); 7.26–7.30 (t, 1H, *J* = 8 Hz); 7.08–7.14 (dd, 1H, *J* = 8 Hz); 3.31 (t, 4H, *J* = 4 Hz); 3.11–3.15 (m, 2H); 2.91–2.96 (m, 2H); 2.71–2.74 (t, 4H, *J* = 6 Hz).

2-(4-(naphthalen-2-yl)piperazin-1-yl)ethanamine (10b)

Compound **10b** was synthesized from compound **9b** (1 g, 3.98 mmol) according to the procedure C to afford **10b** as transparent thick liquid (0.90 g, 88%): ¹H NMR (400 MHz, CDCl₃) δ ppm 7.72–7.73 (d, 1H, *J* = 4 Hz); 7.70–7.71 (d, 1H, *J* = 4 Hz); 7.68 (s, 1H); 7.37–7.41 (t, 1H, *J* = 8 Hz); 7.25–7.30 (m, 2H); 7.11–7.12 (d, 1H, *J* = 4 Hz); 3.30–3.34 (t, 4H, *J* = 6 Hz); 3.13–3.14 (bs, 2H); 2.90–2.92 (bs, 2H); 2.62–2.73 (m, 4H).

Procedure D. Preparation of 7-methoxy-N-(2-(4-(naphthalen-1-yl)piperazin-1-yl)ethyl)-1,2,3,4-tetrahydronaphthalen-2-amine (11a)

A mixture of compound **10a** (0.823 g, 3.22 mmol), 7-methoxy-2-tetralone (0.624 g, 3.5 mmol), and glacial acetic acid (HOAc) (0.19 mL) in 1,2-dichloroethane (50 mL) was stirred at room temperature under N₂ atmosphere for 20 min. Sodium cyanoborohydride (NaCNBH₃) (0.81 g, 12.88 mmol) dissolved in a minimum volume of methanol was added to the reaction mixture. The reaction mixture was stirred at room temperature under nitrogen atmosphere for 12 h. The solvent was evaporated, and saturated NaHCO₃/H₂O (50 mL) was added to the mixture, which was then extracted with ethyl acetate (3 × 100 mL). The combined organic phase was dried over Na₂SO₄ and evaporated to afford the crude product, which was purified by flash chromatography (EtOAc/MeOH/Et₃N = 95:4:1) to give the product **11a** as brown wax (0.84 g, 56%): ¹H NMR (400 MHz, CDCl₃) δ ppm 7.68–7.64 (m, 2H); 7.54–7.56 (d, 1H, *J* = 8 Hz); 7.45–7.48 (t, 1H, *J* = 6 Hz); 7.38–7.42 (m, 1H); 7.26–7.30 (t, 1H, *J* = 8 Hz); 7.08–7.14 (dd, 1H, *J* = 8 Hz); 6.69–7.01 (d, 1H, *J* = 8 Hz); 6.67–6.70 (d, 1H, *J* = 12 Hz); 6.62 (s, 1H); 3.76 (s, 3H); 3.27–3.33 (t, 4H, *J* = 6 Hz); 2.48–2.93 (m, 13H); 1.65–1.68 (m, 2H).

7-methoxy-N-(2-(4-(naphthalen-2-yl)piperazin-1-yl)ethyl)-1,2,3,4-tetrahydronaphthalen-2-amine (11b)

Compound **10b** (0.90 g, 3.5 mmol) was reacted with 7-methoxy-2-tetralone (0.745 g, 4.23 mmol), NaCNBH₃ (0.866 g, 14.99 mmol), and HOAc (0.2 mL) in 1,2-dichloroethane (50 mL) to yield **11b** brown wax (0.855 g, 59%) (procedure D): ¹H NMR (400 MHz, CDCl₃) δ ppm 7.72–7.73 (d, 1H, *J* = 4 Hz); 7.70–7.71 (d, 1H, *J* = 4 Hz); 7.68 (s, 1H); 7.37–7.41 (t, 1H, *J* = 8 Hz); 7.25–7.30 (m, 2H); 7.11–7.12 (d, 1H, *J* = 4 Hz); 6.69–7.01 (d, 1H, *J* = 8 Hz); 6.67–6.70 (d, 1H, *J* = 12 Hz); 6.62 (s, 1H); 3.76 (s, 3H); 3.27–3.33 (t, 4H, *J* = 6 Hz); 2.48–2.93 (m, 13H); 1.65–1.68 (m, 2H).

Procedure E. Preparation of N-(7-methoxy-1,2,3,4-tetrahydronaphthalen-2-yl)-N-(2-(4-(naphthalen-1-yl)piperazin-1-yl)ethyl)propionamide (12a)

Propionyl chloride (0.22 mL, 2.5 mmol) was added into a solution of compound **11a** (0.343 g, 0.83 mmol) and Et₃N (1.0 mL) in anhydrous methylene chloride at 0 °C under N₂ atmosphere and then stirred at room temperature for 4 h. The reaction was diluted with CH₂Cl₂ and washed with water and brine, and the organic layer was dried over Na₂SO₄, evaporated, and purified by flash chromatography (EtOAc/MeOH/Et₃N = 95:4:1) to yield **12a** as yellow oil (0.318 g, 82%): ¹H NMR (400 MHz, CDCl₃) δ ppm 7.68–7.64 (m, 2H); 7.54–7.56 (d, 1H, *J* = 8 Hz);

7.45–7.48 (t, 1H, $J = 6$ Hz); 7.38–7.42 (m, 1H); 7.26–7.30 (t, 1H, $J = 8$ Hz); 7.08–7.14 (dd, 1H, $J = 8$ Hz); 6.69–7.01 (d, 1H, $J = 8$ Hz); 6.67–6.70 (d, 1H, $J = 12$ Hz); 6.62 (s, 1H); 3.76 (s, 3H); 3.27–3.33 (t, 4H, $J = 6$ Hz); 2.48–2.93 (m, 13H); 1.95–1.99 (m, 2H); 1.65–1.68 (m, 2H); 1.15–1.19 (t, 3H, $J = 8$ Hz).

N-(7-methoxy-1,2,3,4-tetrahydronaphthalen-2-yl)-N-(2-(4-(naphthalen-2-yl)piperazin-1-yl)ethyl)propionamide (12b)

Compound **11b** (0.855 g, 2.07 mmol) was reacted with propionyl chloride (0.54 mL, 6.22 mmol) and Et₃N (2.0 mL) in CH₂Cl₂ (20 mL) (procedure E). The crude product was purified by flash chromatography using solvent system EtOAc/MeOH = 90:10 to yield pure compound **12b** as yellow oil (0.5 g, 51.5%): ¹H NMR (400 MHz, CDCl₃) δ ppm 7.72–7.73 (d, 1H, $J = 4$ Hz); 7.70–7.71 (d, 1H, $J = 4$ Hz); 7.68 (s, 1H); 7.37–7.41 (t, 1H, $J = 8$ Hz); 7.25–7.30 (m, 2H); 7.11–7.12 (d, 1H, $J = 4$ Hz); 6.69–7.01 (d, 1H, $J = 8$ Hz); 6.67–6.70 (d, 1H, $J = 12$ Hz); 6.62 (s, 1H); 3.76 (s, 3H); 3.27–3.33 (t, 4H, $J = 6$ Hz); 2.48–2.93 (m, 13H); 1.95–1.99 (m, 2H); 1.65–1.68 (m, 2H); 1.15–1.19 (t, 3H, $J = 8$ Hz).

Procedure F. Preparation of 7-methoxy-N-(2-(4-(naphthalen-1-yl)piperazin-1-yl)ethyl)-N-propyl-1,2,3,4-tetrahydronaphthalen-2-amine (13a)

Compound **12a** (0.318 g, 0.68 mmol) in anhydrous THF (20 mL) was added dropwise into a suspension of lithium aluminum hydride (LiAlH₄) (0.0154 g, 4.07 mmol) in anhydrous THF (15 mL) at 0 °C under N₂ atmosphere. The reaction mixture was refluxed for 4 h, cooled to room temperature, and then cooled further to 0 °C. Saturated NaOH/H₂O (3 mL) was added dropwise to quench excess LiAlH₄. The mixture was filtered, and the reaction mixture was dried over Na₂SO₄. The solvent was removed under vacuum to afford compound **13a** as brown wax (0.246 g, 79%): ¹H NMR (400 MHz, CDCl₃) δ ppm 7.68–7.64 (m, 2H); 7.54–7.56 (d, 1H, $J = 8$ Hz); 7.45–7.48 (t, 1H, $J = 6$ Hz); 7.38–7.42 (m, 1H); 7.26–7.30 (t, 1H, $J = 8$ Hz); 7.08–7.14 (dd, 1H, $J = 8$ Hz); 6.69–7.01 (d, 1H, $J = 8$ Hz); 6.67–6.70 (d, 1H, $J = 12$ Hz); 6.62 (s, 1H); 3.76 (s, 3H); 3.27–3.33 (t, 4H, $J = 6$ Hz); 2.48–2.93 (m, 15H); 1.95–1.99 (m, 2H); 1.65–1.68 (m, 2H); 0.89–0.92 (t, 3H, $J = 6$ Hz).

7-methoxy-N-(2-(4-(naphthalen-2-yl)piperazin-1-yl)ethyl)-N-propyl-1,2,3,4-tetrahydronaphthalen-2-amine (13b)

Compound **12b** (0.50 g, 1.07 mmol) was reacted with LiAlH₄ (0.243 g, 6.4 mmol) in THF (20 mL) by following the procedure F. The crude product was purified by flash chromatography using solvent system EtOAc/MeOH/Et₃N = 95:4:1 to yield compound **13b** as brown wax (0.393 g, 80.7%): ¹H NMR (400 MHz, CDCl₃) δ ppm 7.72–7.73 (d, 1H, $J = 4$ Hz); 7.70–7.71 (d, 1H, $J = 4$ Hz); 7.68 (s, 1H); 7.37–7.41 (t, 1H, $J = 8$ Hz); 7.25–7.30 (m, 2H); 7.11–7.12 (d, 1H, $J = 4$ Hz); 6.69–7.01 (d, 1H, $J = 8$ Hz); 6.67–6.70 (d, 1H, $J = 12$ Hz); 6.62 (s, 1H); 3.76 (s, 3H); 3.27–3.33 (t, 4H, $J = 6$ Hz); 2.48–2.93 (m, 15H); 1.95–1.99 (m, 2H); 1.65–1.68 (m, 2H); 0.89–0.92 (t, 3H, $J = 6$ Hz).

Procedure G. Preparation of 7-((2-(4-(naphthalen-1-yl)piperazin-1-yl)ethyl)(propyl)amino)-5,6,7,8-tetrahydronaphthalen-2-ol (14a, hydrochloride salt)

Boron tribromide (1 M solution in dichloromethane) (1.62 mL, 1.62 mmol) was added into a solution of **13a** (0.246 g, 0.54 mmol) in anhydrous methylene chloride (CH₂Cl₂) (20 mL) at –40 °C under N₂ atmosphere. The reaction mixture was stirred at –40 °C for 2 h and was continued overnight at room temperature. The reaction was quenched by the addition of saturated NaHCO₃ solution, and the mixture was extracted with CH₂Cl₂. The combined organic layer was dried over Na₂SO₄ and evaporated under vacuum, and the crude product was purified by flash chromatography (EtOAc/MeOH = 95:5) to afford compound **14a** as yellowish thick oily (0.075 g, 32%): ¹H NMR (400 MHz, CDCl₃) δ ppm 7.68–7.64 (m, 2H);

7.54–7.56 (d, 1H, $J = 8$ Hz); 7.45–7.48 (t, 1H, $J = 6$ Hz); 7.38–7.42 (m, 1H); 7.26–7.30 (t, 1H, $J = 8$ Hz); 7.08–7.14 (dd, 1H, $J = 8$ Hz); 6.87–6.89 (d, 1H, $J = 8$ Hz); 6.55–6.58 (d, 1H, $J = 12$ Hz); 6.46 (s, 1H); 3.27–3.33 (t, 4H, $J = 6$ Hz); 2.48–2.93 (m, 15H); 1.95–1.99 (m, 2H); 1.65–1.68 (m, 2H); 0.89–0.92 (t, 3H, $J = 6$ Hz). The product was converted into the corresponding trihydrochloride salt as white solid, mp 202–205°C. Anal. Calcd for (C₂₉H₄₀N₃Cl₃O, 2H₂O) C, H, N.

7-((2-(4-(naphthalen-2-yl)piperazin-1-yl)ethyl)(propyl)amino)-5,6,7,8-tetrahydronaphthalen-2-ol (14b, hydrochloride salt)

Compound **13b** (0.393 g, 0.86 mmol) was reacted with 1 M BBr₃/CH₂Cl₂ (2.6 mL, 2.6 mmol) in CH₂Cl₂ (20 mL) by following the procedure G to furnish **14b** as yellow wax (0.227 g, 59.4%): ¹H NMR (400 MHz, CDCl₃) δ ppm 7.72–7.73 (d, 1H, $J = 4$ Hz); 7.70–7.71 (d, 1H, $J = 4$ Hz); 7.68 (s, 1H); 7.37–7.41 (t, 1H, $J = 8$ Hz); 7.25–7.30 (m, 2H); 7.11–7.12 (d, 1H, $J = 4$ Hz); 6.87–6.89 (d, 1H, $J = 8$ Hz); 6.55–6.58 (d, 1H, $J = 12$ Hz); 6.46 (s, 1H); 3.27–3.33 (t, 4H, $J = 6$ Hz); 2.48–2.93 (m, 15H); 1.95–1.99 (m, 2H); 1.65–1.68 (m, 2H); 0.89–0.92 (t, 3H, $J = 6$ Hz). The product was converted into the corresponding trihydrochloride salt as white solid, mp 208–211 °C. Anal. Calcd for (C₂₉H₄₀Cl₃N₃O) C, H, N.

4-chloroquinoline (5)⁵⁷

This intermediate was made according to the reported procedure⁴⁹. Commercially available 4-hydroxyquinoline (1.0 g, 6.9 mmol) was dissolved in POCl₃ (3.2 mL, 34.4 mmol) under nitrogen atmosphere and refluxed for 2 h. It was then cooled and concentrated under vacuo. The solid concentrate was taken in a beaker with 100 mL water, neutralized with NaHCO₃ powder and extracted with EtOAc. The solution was dried over Na₂SO₄, filtered, and concentrated. This crude product was then purified by flash chromatography (EtOAc/Hexan = 50:50) to afford compound **5** as off white solid (0.92 g, 81.6%). ¹H NMR (400 MHz, CDCl₃) δ ppm 8.74–8.75 (d, 1H, $J = 4.8$ Hz); 8.16–8.18 (d, 1H, $J = 8$ Hz); 8.10–8.12 (d, 1H, $J = 8$ Hz); 7.71–7.75 (t, 1H, $J = 8$ Hz); 7.57–7.61 (t, 1H, $J = 8$ Hz); 7.43–7.44 (d, 1H, $J = 4$ Hz).

Procedure H. Preparation of 4-(Piperazin-1-yl)quinoline (8c)

A mixture of 4-chloroquinoline **5** (0.92 g, 5.6 mmol) and piperazine (4.8 g, 56.2 mmol) in isopropanol (30 mL) was refluxed for 4 h. The solvent was evaporated in vacuo and the solid obtained was dissolved in dichloromethane (20 mL) and washed with sat. NaHCO₃ solution followed by water (2×20 mL). The organic layer was dried (Na₂SO₄) and the solvent was evaporated in vacuo to get yellow oily liquid in quantitative yield. It was used without further purification in the next step. ¹H NMR (400 MHz, CDCl₃) δ ppm 8.73–8.74 (d, 1H, $J = 4$ Hz); 8.02–8.06 (t, 2H, $J = 8$ Hz); 7.63–7.67 (t, 1H, $J = 8$ Hz); 7.46–7.50 (t, 1H, $J = 8$ Hz); 6.83–6.84 (d, 1H, $J = 4$ Hz); 3.16–3.18 (m, 8H).

Quinolin-5-amine (6)

Into the solution of 5-nitroquinoline (1.5 g, 8.6 mmol) in absolute ethanol was added stannous chloride dihydrate (10.92 g, 43.06 mmol). The mixture was refluxed at 60 °C for 1 h. NaBH₄ (0.022 g, 4.3 mmol) was added to it and refluxed for another 30 min. The reaction mixture was cooled, ethanol was evaporated and the concentrate was dissolved in water. The reaction mixture was made alkaline with 40% aq. NaOH and extracted with ethyl acetate (3×100 mL). The organic layer was evaporated under reduced pressure, dried (Na₂SO₄) to get a yellow liquid of almost quantitative yield. It was used in the next step without further purification. ¹H NMR (400 MHz, CDCl₃) δ ppm 8.89–8.90 (d, 1H, $J = 4$ Hz); 8.17–8.19 (d, 1H, $J = 8$ Hz); 7.57–7.59 (d, 1H, $J = 8$ Hz); 7.49–7.53 (t, 1H, $J = 8$ Hz); 7.34–7.37 (dd, 1H, $J = 4$ Hz); 6.82–6.84 (d, 1H, $J = 8$ Hz).

5-(Piperazin-1-yl)quinoline (8d)

5-Aminoquinoline **6** (2.2 g, 15.3 mmol) and bis(2-chloroethyl)amine HCl (3.27 g, 18.3 mmol, 1.2 equiv) were refluxed in diglyme for 2 days. The reaction mixture was cooled and diglyme was evaporated in vacuo. To the solid residue water and ethyl acetate were added and extracted with (3×100 ml) ethyl acetate. The combined organic layers were washed with water, dried (Na₂SO₄) and the solvent was evaporated to get the crude product which was purified by column chromatography (dichloromethane/MeOH = 4:1) to afford a resinous pure mass. Yield: 56 %. ¹H NMR (400 MHz, CDCl₃) δ ppm 8.85–8.86 (d, 1H, *J* = 4 Hz); 8.70–8.71 (d, 1H, *J* = 4 Hz); 7.80–7.82 (d, 1H, *J* = 8 Hz); 7.72–7.76 (t, 1H, *J* = 8 Hz); 7.57–7.60 (dd, 1H, *J* = 4 Hz), 7.34–7.36 (d, 1H, *J* = 8 Hz), 3.51–3.54 (t, 4H, *J* = 6 Hz), 3.30–3.34 (m, 4H).

1-(Piperazin-1-yl)isoquinoline (8e)

It was synthesized following the procedure H to furnish **8e** using 1-chloroisoquinoline **7** (0.95 g, 5.8 mmol) and piperazine (2.5 g, 29 mmol) in isopropanol (50 ml) with quantitative yield of **8e** as a yellow semisolid. It was used without further purification in the next step. ¹H NMR (400 MHz, CDCl₃) δ ppm 8.14–8.16 (d, 1H, *J* = 8 Hz); 8.08–8.10 (d, 1H, *J* = 8 Hz); 7.72–7.74 (d, 1H, *J* = 8 Hz); 7.57–7.61 (t, 1H, *J* = 8 Hz); 7.48–7.52 (t, 1H, *J* = 8 Hz); 7.23–7.24 (d, 1H, *J* = 4 Hz); 3.39–3.41 (t, 4H, *J* = 4 Hz); 3.15–3.18 (t, 4H, *J* = 6 Hz); 2.72 (brs, 1H).

Procedure I. Preparation of (7-Methoxy-1,2,3,4-tetrahydro-naphthalen-2-yl)-propyl-amine (16a).⁴⁵

7-Methoxy-2-tetralone (10g, 56.75 mmol) and acetic acid (13.5 ml, 226.9 mmol) were dissolved in dichloroethane (150 ml) and cooled to 0° C. Propyl amine (11.7 ml, 141.87 mmol) was added and the mixture was stirred under a N₂ atmosphere for 30 min. NaCNBH₃ (8.915 g, 141.87 mmol) in anhydrous MeOH (15 ml) was then added to the mixture and allowed to stir overnight at ambient temperature. The volatiles were then evaporated and saturated NaHCO₃ solution was added into the mixture. The compound was then extracted into dichloromethane, dried over Na₂SO₄, filtered, and concentrated. The crude residue was dissolved in EtOAc and ethereal HCl was added. The crude salt was filtered and dried in a vacuum oven. The crude salt was then recrystallized from ethanol. The white salt was collected via filtration and dried to yield 9.5 g (65%) and used in the subsequent transformations. ¹H NMR (free base) (400 MHz, CDCl₃) 6.95–6.98 (d, 1H, *J* = 8.8 Hz), 6.65–6.78 (m, 1H), 6.60–6.61 (dd, 1H, *J* = 1.6 Hz), 3.81 (s, 3H), 2.97–3.04 (m, 1H), 2.88–2.92 (m, 2H), 2.67–2.71 (t, 3H, *J* = 7.6 Hz), 2.54–2.62 (m, 2H), 2.04–2.09 (m, 1H), 1.38 (bs, 1H), 1.48–1.60 (m, 3H), 0.91–0.95 (t, 3H, *J* = 7.6 Hz),

(5-Methoxy-1,2,3,4-tetrahydro-naphthalen-2-yl)-propyl-amine (16b)

Compound **16b** was prepared by following the procedure I from 5-methoxy 2-tetralone (64%). ¹H NMR (free base) (400 MHz, CDCl₃) 7.07–7.11 (t, 1H, *J* = 7.2 Hz), 6.96–6.71 (d, 1H, *J* = 8 Hz), 6.65–6.67 (d, 1H, *J* = 8 Hz); 3.81 (s, 3H), 2.98–3.03 (m, 1H), 2.87–2.94 (m, 2H), 2.66–2.70 (t, 3H, *J* = 7.6 Hz), 2.53–2.62 (m, 2H), 2.05–2.10 (m, 1H), 1.49–1.61 (m, 3H), 1.39 (bs, 1H), 0.92–0.964 (t, 3H, *J* = 7.6 Hz).

Resolution of 5-Methoxy-N-propyl-1,2,3,4-tetrahydronaphthalen-2-amine (–)-16b

Racemic compound **16b** was resolved into its (–) isomer by using the (+) isomer of a synthetic resolving agent 4-(2-chlorophenyl)-5,5-dimethyl-2-hydroxy-1,3,2-dioxaphosphorinane 2-oxide. The (+) and the (–) resolving agents were prepared according to the method of Ten Hoeve and Wynberg. **16b** (free base 14.77 g, 67.36 mmol) and (+)-4-(2-chlorophenyl)-5,5-dimethyl-2-hydroxy-1,3,2-dioxaphosphorinane 2-oxide (20.5 g, 74.1 mmol) were dissolved by warming in 100 ml of ethanol. The solution was cooled to room temperature and then at 0° C. The precipitated crystals were filtered off, washed with cold ether to yield 17.4 g of the salt

($[\alpha]_D = (-)1.2^\circ$, $c = 1$ in methanol). Further recrystallization (2 times) from hot ethanol yielded the salt (12.9 g, $[\alpha]_D = (-)14.1^\circ$, $c = 1$ in methanol). Further crystallization of the salt from hot ethanol did not change the optical rotation to a significant extent. The salt was then neutralized in presence of 20% NaOH solution in water under stirred condition for 2 h at room temperature. The aqueous layer was extracted with dichloromethane (3×100 ml), dried over Na_2SO_4 and evaporated to dryness to yield (-)-**16b** as transparent liquid (5.8 g, $[\alpha]_D$ of the white solid, HCl salt of (-)-**16b** = (-)71.5°, ($c = 1$ in methanol) Yield. 78.5 %.

Procedure J. Preparation of 2-Chloro-N-(7-methoxy-1,2,3,4-tetrahydro-naphthalen-2-yl)-N-propyl-acetamide (17a)

Compound **16a** (HCl salt, 3.117 g, 12.18 mmol) and Et_3N (8.4 ml, 60.9 mmol) was stirred at 0° C in CH_2Cl_2 (75 ml) for 30 min. Chloroacetyl chloride (1.94 ml, 24.37 mmol) was added dropwise and the resulting solution was stirred at room temperature for 20 min, at which time the reaction mixture was poured onto a 1M solution of NaOH (50 ml) and the product was extracted with dichloromethane, dried (Na_2SO_4), filtered, and concentrated. The crude material was purified by column chromatography (Hex:EtOAc, 3:1) to give 3.42 g (95%) of **17a** as a transparent liquid. $^1\text{H NMR}$ (400 MHz, CDCl_3) 0.90–0.98 (m, 3H), 1.64–1.72 (m, 2H), 3.19–3.27 (m, 2H), 4.00 (s, 3H), 6.61–6.62 (dd, 1H, $J = 1.6$ Hz), 6.64–6.77 (m, 1H), 6.96–6.99 (d, 1H, $J = 8.8$ Hz).

2-Chloro-N-(5-methoxy-1,2,3,4-tetrahydro-naphthalen-2-yl)-N-propyl-acetamide (17b)

This compound was prepared from **16b** by following the procedure J (93 %). $^1\text{H NMR}$ (400 MHz, CDCl_3) 6.96–7.04 (m, 1H), 6.61–6.68 (m, 2H), 4.00 (s, 3H), 3.19–3.27 (m, 2H), 1.64–1.72 (m, 2H), 0.90–0.98 (m, 3H).

(-)-2-Chloro-N-(5-methoxy-1,2,3,4-tetrahydronaphthalen-2-yl)-N-propylacetamide (-)-17b

Compound (-)-**16b** (1.20 g, 5.47 mmol) was reacted under similar conditions as reported above (procedure J) to afford the optically pure (-)-**17b** as transparent liquid (1.58 g, 97%): $^1\text{H NMR}$ (400 MHz, CDCl_3) δ 7.07–7.15 (m, 1H), 6.65–6.71 (m, 2H), 4.08–4.12 (m, 2H), 3.95–4.03 (m, 1H), 3.80–3.82 (d, 3H, -OCH₃), 3.15–3.26 (m, 2H), 3.00–3.10 (m, 2H), 2.84–2.89 (dd, 1H, $J_1 = 16.0$ Hz, $J_2 = 4.8$ Hz), 1.83–2.12 (m, 2H), 2.58–2.70 (m, 1H), 1.61–1.73 (m, 2H), 0.90–0.97 (m, 3H, CH_2CH_3),

Procedure K. Preparation of N-(7-methoxy-1,2,3,4-tetrahydronaphthalen-2-yl)-N-propyl-2-(4-(quinolin-4-yl)piperazin-1-yl)acetamide (18a)

Compound **17a** (0.70 g, 2.4 mmol) and **8c** (0.614 g, 2.84 mmol) were refluxed in CH_3CN (100 ml) in presence of K_2CO_3 (1.635 g, 11.83 mmol) for 2 h. The solution was cooled, filtered, and concentrated. The crude material was then partitioned between EtOAc and H_2O , and the organic layer was separated, dried (Na_2SO_4), and concentrated. The crude mixture was purified by column chromatography (EtOAc:MeOH; 4:1) to give 0.45 g (41 %) of **18a** as yellow sticky mass. $^1\text{H NMR}$ (400 MHz, CDCl_3) δ ppm 8.73–8.74 (d, 1H, $J = 4$ Hz); 8.04–8.06 (d, 1H, $J = 8$ Hz); 7.99–8.02 (d, 1H, $J = 12$ Hz); 7.64–7.68 (t, 1H, $J = 8$ Hz); 7.46–7.50 (t, 1H, $J = 8$ Hz); 7.01–7.03 (d, 1H, $J = 8$ Hz); 6.83–6.84 (d, 1H, $J = 4$ Hz); 6.62–6.64 (d, 1H, $J = 8$ Hz); 6.66 (s, 1H); 3.76 (s, 3H); 3.49 (s, 2H); 2.82–3.30 (m, 13H); 2.55 (brs, 2H); 1.85–2.07 (m, 2H); 1.63 (brs, 2H); 0.88–0.91 (t, 3H, $J = 6$ Hz).

N-(7-methoxy-1,2,3,4-tetrahydronaphthalen-2-yl)-N-propyl-2-(4-(quinolin-5-yl)piperazin-1-yl)acetamide (18b)

Compound **17a** (0.472 g, 1.6 mmol) was reacted with **8d** (0.23 g, 1.1 mmol) in CH_3CN (50 ml) by following the procedure **K** to furnish **18b** as a yellow sticky mass (0.240 g, 23%). $^1\text{H NMR}$ (400 MHz, CDCl_3) δ ppm 8.89–8.90 (d, 1H, $J = 4$ Hz); 8.48–8.54 (t, 1H, $J = 12$ Hz);

7.80–7.82 (d, 1H, $J = 8$ Hz); 7.61–7.65 (t, 1H, $J = 8$ Hz); 7.37–7.40 (dd, 1H, $J = 4$ Hz), 7.11–7.13 (d, 1H, $J = 8$ Hz), 6.99–7.00 (dd, 1H, $J = 4$ Hz); 6.65–6.75 (m, 1H), 6.60 (s, 1H); 3.76 (s, 3H); 2.80–3.45 (m, 15H); 2.55 (brs, 2H); 1.85–2.07 (m, 2H); 1.63 (brs, 2H); 0.88–0.91 (t, 3H, $J = 6$ Hz).

2-(4-(isoquinolin-1-yl)piperazin-1-yl)-N-(7-methoxy-1,2,3,4-tetrahydronaphthalen-2-yl)-N-propylacetamide (18c)

Compound **17a** (1.56 g, 5.27 mmol) was reacted with **8e** (0.760 g, 3.5 mmol) in CH₃CN (100 ml) by following the procedure **K** to furnish **18c** as a yellow sticky mass (1.45 g, 87%). ¹H NMR (400 MHz, CDCl₃) δ ppm 8.14–8.16 (d, 1H, $J = 8$ Hz); 8.08–8.10 (d, 1H, $J = 8$ Hz); 7.72–7.74 (d, 1H, $J = 8$ Hz); 7.57–7.61 (t, 1H, $J = 8$ Hz); 7.48–7.52 (t, 1H, $J = 8$ Hz); 7.23–7.24 (d, 1H, $J = 4$ Hz); 6.98–7.02 (t, 1H, 8 Hz); 6.64–6.77 (m, 1H); 6.63 (s, 1H); 3.78 (s, 3H); 2.78–3.84 (m, 16H); 1.63–2.04(m, 5H); 0.90–0.98 (m, 3H).

N-(5-methoxy-1,2,3,4-tetrahydronaphthalen-2-yl)-N-propyl-2-(4-(quinolin-4-yl)piperazin-1-yl)acetamide (18d)

Compound **17b** (1.1 g, 3.7 mmol) was reacted with **8c** (0.855 g, 4.1 mmol) in CH₃CN (100 ml) by following the procedure **K** to furnish **18d** as a yellow sticky mass (1.63 g, 92.7%). ¹H NMR (400 MHz, CDCl₃) δ ppm 8.73–8.74 (d, 1H, $J = 4$ Hz); 7.99–8.06 (m, 2H Hz); 7.63–7.67 (t, 1H, $J = 8$ Hz); 7.46–7.50 (t, 1H, $J = 8$ Hz); 7.07–7.16 (m, 1H); 6.83–6.87 (m, 1H); 6.65–6.73 (m, 2H), 3.83 (s, 3H); 2.80–3.45 (m, 15H), 2.63 (brs, 1H), 2.05 (m, 2H); 1.65–1.71 (m, 2H); 1.24–1.28 (t, 1H, $J = 4$ Hz); 0.90–0.98 (m, 3H).

(S)-N-(5-methoxy-1,2,3,4-tetrahydronaphthalen-2-yl)-N-propyl-2-(4-(quinolin-4-yl)piperazin-1-yl)acetamide (–)-18d

Compound (–)-**17b** (0.67 g, 2.25 mmol) was reacted with **8c** (0.539 g, 2.5 mmol) in CH₃CN (100 ml) by following procedure **K** to furnish (–)-**18d** as a yellow sticky mass (0.854 g, 79.8%). ¹H NMR (400 MHz, CDCl₃) δ ppm 8.73–8.74 (d, 1H, $J = 4$ Hz); 7.99–8.06 (m, 2H Hz); 7.63–7.67 (t, 1H, $J = 8$ Hz); 7.46–7.50 (t, 1H, $J = 8$ Hz); 7.07–7.16 (m, 1H); 6.83–6.87 (m, 1H); 6.65–6.73 (m, 2H), 3.83 (s, 3H); 2.80–3.45 (m, 15H), 2.63 (brs, 1H), 2.05 (m, 2H); 1.65–1.71 (m, 2H); 1.24–1.28 (t, 1H, $J = 4$ Hz); 0.90–0.98 (m, 3H).

7-methoxy-N-propyl-N-(2-(4-(quinolin-4-yl)piperazin-1-yl)ethyl)-1,2,3,4-tetrahydronaphthalen-2-amine (19a)

Compound **18a** (0.45 g, 0.95 mmol) was reacted with LiAlH₄ (0.18 g, 4.76 mmol) in THF (20 mL) by following procedure **F** to furnish **19a** as a yellow thick liquid (0.14 g, 35%). ¹H NMR (400 MHz, CDCl₃) δ ppm 8.72–8.73 (d, 1H, $J = 4$ Hz); 8.04–8.06 (d, 1H, $J = 8$ Hz); 8.01–8.03 (d, 1H, $J = 8$ Hz); 7.63–7.67 (t, 1H, $J = 8$ Hz); 7.45–7.49 (t, 1H, $J = 8$ Hz); 6.98–7.00 (d, 1H, $J = 8$ Hz); 6.83–6.84 (d, 1H, $J = 4$ Hz); 6.67–6.70 (d, 1H, $J = 12$ Hz); 6.64 (s, 1H); 3.77 (s, 3H); 3.26 (brs, 4H); 2.99–3.02 (m, 1H); 2.73–2.88 (m, 10H); 2.54–2.64 (m, 4H); 2.05 (brs, 1H); 1.49–1.66 (m, 3H); 0.90–0.94 (t, 3H, $J = 8$ Hz).

7-Methoxy-N-propyl-N-(2-(4-(quinolin-5-yl)piperazin-1-yl)ethyl)-1,2,3,4-tetrahydronaphthalen-2-amine (19b)

Compound **18b** (0.24 g, 0.51 mmol) was reacted with LiAlH₄ (0.097, 2.5 mmol) in THF (20 mL) by following procedure **F** to furnish **19b** as a yellow thick liquid (0.200 g, 85.9%). ¹H NMR (400 MHz, CDCl₃) δ ppm 8.87–8.89 (d, 1H, $J = 8$ Hz); 8.50–8.52 (d, 1H, $J = 8$ Hz); 7.79–7.81 (d, 1H, $J = 8$ Hz); 7.60–7.64 (t, 1H, $J = 8$ Hz); 7.36–7.39 (dd, 1H, $J = 4$ Hz), 7.12–7.14 (d, 1H, $J = 8$ Hz), 6.97–7.00 (dd, 1H, $J = 4$ Hz); 6.65–6.75 (m, 1H), 6.60 (s, 1H); 3.76 (s, 3H); 2.80–3.45 (m, 15H); 2.04 (brs, 2H); 1.47–1.1.68 (m, 4H); 1.63 (brs, 2H); 0.88–0.91 (t, 3H, $J = 6$ Hz).

N-(2-(4-(Isoquinolin-1-yl)piperazin-1-yl)ethyl)-7-methoxy-N-propyl-1,2,3,4-tetrahydronaphthalen-2-amine (19c)

Compound **18c** (1.42 g, 3.0 mmol) was reacted with LiAlH₄ (0.570 g, 15.02 mmol) in THF (30 mL) by following procedure F to furnish **19c** as a yellow thick liquid (1.2 g, 87%). ¹H NMR (400 MHz, CDCl₃) δ ppm 8.14–8.16 (d, 1H, *J* = 8 Hz); 8.08–8.10 (d, 1H, *J* = 8 Hz); 7.72–7.74 (d, 1H, *J* = 8 Hz); 7.57–7.61 (t, 1H, *J* = 8 Hz); 7.48–7.52 (t, 1H, *J* = 8 Hz); 7.23–7.24 (d, 1H, *J* = 4 Hz); 6.98–7.02 (t, 1H, 8 Hz); 6.64–6.77 (m, 1H); 6.63 (s, 1H); 3.77 (s, 3H); 3.45 (brs, 4H); 2.78–3.00 (m, 11H); 2.57 (m, 5H); 2.05 (s, 1H); 1.51 (m, 2H); 0.90–0.93 (t, 3H, 6Hz).

5-Methoxy-N-propyl-N-(2-(4-(quinolin-4-yl)piperazin-1-yl)ethyl)-1,2,3,4-tetrahydronaphthalen-2-amine (19d)

Compound **18d** (1.63 g, 3.45 mmol) was reacted with LiAlH₄ (0.654 g, 17.24 mmol) in THF (30 mL) by following procedure F to furnish **19d** as a yellow thick liquid (0.325 g, 21%). ¹H NMR (400 MHz, CDCl₃) δ ppm 8.73–8.74 (d, 1H, *J* = 4 Hz); 7.99–8.06 (m, 2H); 7.63–7.67 (t, 1H, *J* = 8 Hz); 7.46–7.50 (t, 1H, *J* = 8 Hz); 7.08–7.12 (m, 2H); 6.83–6.84 (d, 1H, *J* = 4 Hz); 6.71–6.73 (d, 1H, *J* = 8 Hz), 3.83 (s, 3H); 2.80–3.45 (m, 17H), 2.63 (brs, 1H), 2.05 (m, 2H); 1.65–1.71 (m, 2H); 1.24–1.28 (t, 1H, *J* = 4 Hz); 0.90–0.98 (m, 3H).

(S)-5-Methoxy-N-propyl-N-(2-(4-(quinolin-4-yl)piperazin-1-yl)ethyl)-1,2,3,4-tetrahydronaphthalen-2-amine (-)-19d

Compound (-)-**18d** (0.83 g, 1.76 mmol) was reacted with LiAlH₄ (0.333 g, 8.8 mmol) in THF (20 mL) by following procedure F to furnish (-)-**19d** as a yellow thick liquid (0.20 g, 24%). ¹H NMR (400 MHz, CDCl₃) δ ppm 8.73–8.74 (d, 1H, *J* = 4 Hz); 7.99–8.06 (m, 2H); 7.63–7.67 (t, 1H, *J* = 8 Hz); 7.46–7.50 (t, 1H, *J* = 8 Hz); 7.08–7.12 (m, 2H); 6.83–6.84 (d, 1H, *J* = 4 Hz); 6.71–6.73 (d, 1H, *J* = 8 Hz), 3.83 (s, 3H); 2.80–3.45 (m, 17H), 2.63 (brs, 1H), 2.05 (m, 2H); 1.65–1.71 (m, 2H); 1.24–1.28 (t, 1H, *J* = 4 Hz); 0.90–0.98 (m, 3H).

7-(Propyl(2-(4-(quinolin-4-yl)piperazin-1-yl)ethyl)amino)-5,6,7,8-tetrahydronaphthalen-2-ol (20a, hydrochloride salt)

Compound **19a** (0.14 g, 0.31 mmol) was reacted with 1 M BBr₃/CH₂Cl₂ (1.22 mL, 1.22 mmol) in CH₂Cl₂ (20 mL) by following procedure G to furnish **20a** as a white wax (0.030 g, 31%). ¹H NMR (400 MHz, CDCl₃) δ ppm 8.72–8.73 (d, 1H, *J* = 4 Hz); 8.04–8.06 (d, 1H, *J* = 8 Hz); 8.01–8.03 (d, 1H, *J* = 8 Hz); 7.63–7.67 (t, 1H, *J* = 8 Hz); 7.45–7.49 (t, 1H, *J* = 8 Hz); 6.89–6.91 (d, 1H, *J* = 4 Hz); 6.84–6.85 (d, 1H, *J* = 4 Hz); 6.63–6.65 (d, 1H, *J* = 8 Hz); 6.58 (s, 1H); 3.26 (brs, 4H); 2.99–3.02 (m, 1H); 2.73–2.88 (m, 10H); 2.54–2.64 (m, 4H); 2.05 (brs, 1H); 1.49–1.66 (m, 3H); 0.90–0.94 (t, 3H, *J* = 8 Hz).

The product was converted into the corresponding hydrochloride salt as white solid, mp 225–228 °C. Anal. Calcd for (C₂₈H₄₀Cl₄N₄O, 2.6H₂O) C, H, N

7-(Propyl(2-(4-(quinolin-5-yl)piperazin-1-yl)ethyl)amino)-5,6,7,8-tetrahydronaphthalen-2-ol (20b, oxalate salt)

Compound **19b** (0.20 g, 0.44 mmol) was reacted with 1 M BBr₃/CH₂Cl₂ (1.74 mL, 1.74 mmol) in CH₂Cl₂ (20 mL) by following procedure G to furnish **20b** as a white wax (0.123 g, 63.4%). ¹H NMR (400 MHz, CDCl₃) δ ppm 8.87–8.89 (d, 1H, *J* = 8 Hz); 8.50–8.52 (d, 1H, *J* = 8 Hz); 7.79–7.81 (d, 1H, *J* = 8 Hz); 7.60–7.64 (t, 1H, *J* = 8 Hz); 7.36–7.39 (dd, 1H, *J* = 4 Hz), 7.12–7.14 (d, 1H, *J* = 8 Hz), 6.88–6.90 (d, 1H, *J* = 8 Hz); 6.62–6.64 (m, 1H), 6.55 (s, 1H); 2.80–3.45 (m, 15H); 2.04 (brs, 2H); 1.47–1.1.68 (m, 4H); 1.63 (brs, 2H); 0.88–0.91 (t, 3H, *J* = 6 Hz).

The product was converted into the corresponding oxalate salt as a light yellowish solid, mp 155–158 °C. Anal. Calcd for (C₃₄H₄₂N₄O₁₃, 1.5H₂O) C, H, N

7-((2-(4-(Isoquinolin-1-yl)piperazin-1-yl)ethyl)(propyl)amino)-5,6,7,8-tetrahydronaphthalen-2-ol (20c, hydrochloride salt)

Compound **19c** (1.0 g, 2.18 mmol) was reacted with 1 M BBr₃/CH₂Cl₂ (8.7 mL, 8.7 mmol) in CH₂Cl₂ (50 mL) by following procedure G to furnish **20c** as white wax (0.600 g, 62%). ¹H NMR (400 MHz, CDCl₃) δ ppm 8.10–8.12 (d, 1H, *J* = 8 Hz); 8.05–8.07 (d, 1H, *J* = 8 Hz); 7.72–7.74 (d, 1H, *J* = 8 Hz); 7.59–7.63 (t, 1H, *J* = 8 Hz); 7.49–7.53 (t, 1H, *J* = 8 Hz); 7.24–7.26 (d, 1H, *J* = 8 Hz); 6.84–6.86 (d, 1H, 8 Hz); 6.64–6.66 (d, 1H, *J* = 8 Hz); 6.57 (s, 1H); 3.47 (brs, 4H); 3.28 (brs, 1H); 3.04–2.66 (m, 14H); 2.21 (m, 1H); 2.04 (s, 1H); 1.69 (brs, 2H); 0.90–0.93 (t, 3H, 6Hz).

The product was converted into the corresponding trihydrochloride salt as white solid, mp 140–142 °C. Anal. Calcd for (C₂₈H₃₉Cl₃N₄O, 1.4H₂O) C, H, N

6-(Propyl(2-(4-(quinolin-4-yl)piperazin-1-yl)ethyl)amino)-5,6,7,8-tetrahydronaphthalen-1-ol (20d, hydrochloride salt)

Compound **19d** (0.30 g, 0.65 mmol) was reacted with 1 M BBr₃/CH₂Cl₂ (3.3 mL, 3.3 mmol) in CH₂Cl₂ (20 mL) by following procedure G to furnish **20d** as white wax (0.175 g, 45%). ¹H NMR (400 MHz, CDCl₃) δ ppm 8.70–8.71 (d, 1H, *J* = 4 Hz); 8.05–8.07 (d, 1H, *J* = 8 Hz); 7.99–8.01 (d, 1H, *J* = 8 Hz); 7.63–7.67 (t, 1H, *J* = 8 Hz); 7.46–7.50 (t, 2H, *J* = 8 Hz); 6.93–6.97 (t, 1H, *J* = 8 Hz); 6.83–6.84 (d, 1H, *J* = 4 Hz); 6.70–6.72 (m, 1H); 6.59–6.60 (d, 1H, *J* = 4 Hz); 2.80–3.45 (m, 17H); 2.63 (brs, 1H); 2.05 (m, 2H); 1.65–1.71 (m, 2H); 1.24–1.28 (t, 1H, *J* = 4 Hz); 0.94–0.97 (t, 3H, *J* = 6 Hz).

The product was converted into the corresponding hydrochloride salt as white solid, mp 223–225 °C. Anal. Calcd for (C₂₈H₄₀Cl₄N₄O, C₂H₅OH) C, H, N.

(S)-6-(Propyl(2-(4-(quinolin-4-yl)piperazin-1-yl)ethyl)amino)-5,6,7,8-tetrahydronaphthalen-1-ol. [(–)-20d (hydrochloride salt)]

Compound (–)-**19d** (0.193 g, 0.42 mmol) was reacted with 1 M BBr₃/CH₂Cl₂ (2.1 mL, 2.1 mmol) in CH₂Cl₂ (20 mL) by following procedure G to furnish (–)-**20d** as white wax (0.83 g, 45%). ¹H NMR (400 MHz, CDCl₃) δ ppm 8.70–8.71 (d, 1H, *J* = 4 Hz); 8.05–8.07 (d, 1H, *J* = 8 Hz); 7.99–8.01 (d, 1H, *J* = 8 Hz); 7.63–7.67 (t, 1H, *J* = 8 Hz); 7.46–7.50 (t, 2H, *J* = 8 Hz); 6.93–6.97 (t, 1H, *J* = 8 Hz); 6.83–6.84 (d, 1H, *J* = 4 Hz); 6.70–6.72 (m, 1H); 6.59–6.60 (d, 1H, *J* = 4 Hz); 2.80–3.45 (m, 17H); 2.63 (brs, 1H); 2.05 (m, 2H); 1.65–1.71 (m, 2H); 1.24–1.28 (t, 1H, *J* = 4 Hz); 0.94–0.97 (t, 3H, *J* = 6 Hz). [α]_D²⁵ = –36° (c=1, CH₃OH). The product was converted into the corresponding hydrochloride salt as white solid, mp 219–222 °C. Anal. Calcd for C₂₈H₄₀Cl₄N₄O, C₂H₅OH) C, H, N.

4-(4-(Tert-Butoxycarbonyl)piperazin-1-yl)phenylboronic acid (7b)

1-(4-Iodophenyl)-piperazine HCl (0.5 g, 1.54 mmol) was suspended in dichloromethane (15 ml) when Et₃N (1 equiv) was added drop-wise. The clear solution was washed with water, dried (Na₂SO₄) and the solvent evaporated to obtain pale brown semisolid which was dissolved in absolute EtOH. Di-tert-butyl dicarbonate (0.34 g, 1.54 mmol) was added to the above ethanolic solution and the reaction mixture stirred for 30 min. The solvent was evaporated and the residue taken in dichloromethane, washed with water, dried (Na₂SO₄) with subsequent removal of the solvent to yield tert-butyl 4-(4-iodophenyl)piperazine-1-carboxylate (0.56 g, 93 %) which was used without further purification in the next step.

To a solution of tert-butyl 4-(4-iodophenyl)piperazine-1-carboxylate (0.5 g, 2.601 mmol) in dry THF was added dropwise 2.5 M nBuLi in hexane solution (1.36 ml, 3.381 mmol, 1.3 equiv) at -78°C and the mixture stirred at the same temperature for 30 min. Trimethyl borate (0.38 ml, 3.381 mmol, 1.3 equiv) was added dropwise over 15 min period. The mixture was stirred at -78°C for 1 hr and then gradually warmed to RT when it was quenched with 1.2 M HCl solution (pH 5–7) and then extracted with diethyl ether (3×20 ml). The combined organic layers were washed with brine, dried (Na_2SO_4) and the solvent evaporated to get the boronic acid **7b** (0.23 g, 58 %). ^1H NMR(400 MHz, CDCl_3) δ ppm 8.11–8.13 (d, 2H, $J = 8$ Hz), 6.98–7.0 (d, 2H, $J = 8$ Hz), 3.61–3.63 (t, 4H, $J = 4$ Hz), 3.31–3.33 (t, 4H, $J = 4$ Hz), 1.50 (s, 9H).

1-(4-(Pyridin-4-yl)phenyl)piperazine (8f)

A mixture of $\text{Pd}(\text{PPh}_3)_4$ (31.5 mg, 5 mol %) and 4-bromopyridine HCl (0.106 g, 0.544 mmol) in DME was stirred at RT for 30 min. The suspension of boronic acid **7b** (0.2 g, 0.653 mmol, 1.2 equiv) in DME was added to the above solution followed by addition of 2 M Na_2CO_3 solution (0.55 ml, 2 equiv). The reaction mixture was refluxed overnight, cooled to RT and the solids filtered. To the filtrate was added sat. aq. NH_4Cl solution with subsequent extraction by dichloromethane (3×20 ml). The combined organic layers were dried (Na_2SO_4) and the solvent evaporated in vacuo to get tert-butyl 4-(4-(pyridin-4-yl)phenyl)piperazine-1-carboxylate (0.15 g, 68 %).

The deprotection of the N-Boc group was done by stirring tert-butyl 4-(4-(pyridin-4-yl)phenyl)piperazine-1-carboxylate (0.15 g) in trifluoroacetic acid (2 ml) and chloroform (2 ml). The trifluoro-acetate salt (obtained after the removal of the solvent in vacuo) was converted to the free base using 1 M NaOH aq. solution to yield **8f** (0.1 g) as an oil. ^1H NMR(400 MHz, CDCl_3) δ ppm 8.51–8.52 (d, 2H, $J = 4$ Hz), 7.50–7.53 (d, 2H, $J = 12$ Hz), 7.39–7.40 (d, 2H, $J = 4$ Hz), 6.92–6.94 (d, 2H, $J = 8$ Hz), 3.16–3.18 (t, 4H, $J = 4$ Hz), 2.98–3.0 (t, 4H, $J = 4$ Hz), 1.89 (brs, 1H).

General Procedure for the Preparation of 2-(4-Arylpiperazin-1-yl)ethanols 21a–21d

Into a stirring suspension of 1-arylpiperazine **8c–f** and anhydrous K_2CO_3 (2 equiv) in acetonitrile, bromoethanol (1.2 equiv) and KI (catalytic amt) were added. The mixture was refluxed overnight and cooled to RT. The solvent was evaporated in vacuo and the residue was partitioned between ethyl acetate and water. The aqueous layer was further extracted with ethyl acetate (2×20 ml). The combined organic layers were dried (MgSO_4) and the solvent evaporated in vacuo to yield oily liquid which was purified by column chromatography (dichloromethane:MeOH 95:5) to obtain pure **21a–d**.

2-(4-(Quinolin-4-yl)piperazin-1-yl)ethanol (21a)—Yield: 71 %, yellow oil, starting with 0.7 g (3.28 mmol) of 4-(piperazin-1-yl)quinoline **8c**. ^1H NMR(400 MHz, CDCl_3) δ ppm 8.73–8.74 (d, 1H, $J = 4$ Hz), 8.06–8.08 (d, 1H, $J = 8$ Hz), 8.01–8.03 (d, 1H, $J = 8$ Hz), 7.65–7.69 (m, 1H), 7.47–7.51 (m, 1H), 6.85–6.87 (d, 1H, $J = 8$ Hz), 3.70–3.73 (t, 2H, $J = 6$ Hz), 3.30 (brs, 4H), 3.01 (brs, 1H), 2.84 (brs, 2H), 2.70–2.73 (t, 2H, $J = 6$ Hz).

2-(4-(Quinolin-5-yl)piperazin-1-yl)ethanol (21b)—Yield: 67 %, starting with 1.0 g (4.69 mmol) of 5-(piperazin-1-yl)quinoline **8d**. ^1H NMR(400 MHz, CDCl_3) δ ppm 8.92–8.96 (m, 1H), 8.53–8.59 (m, 1H), 7.87–7.92 (t, 1H, $J = 10$ Hz), 7.64–7.71 (m, 1H), 7.40–7.47 (m, 1H), 7.12–7.19 (m, 1H), 4.09 (s, 1H), 3.80–3.84 (t, 2H, $J = 8$ Hz), 3.13 (brs, 4H), 2.83 (brs, 4H), 2.72–2.78 (t, 2H, $J = 12$ Hz).

2-(4-(Isoquinolin-1-yl)piperazin-1-yl)ethanol (21c)—Yield: 64 %, oil, starting with 1.3 g (6.1 mmol) of 1-(piperazin-1-yl)isoquinoline **8e**. ^1H NMR(400 MHz, CDCl_3) ppm 8.13–8.14 (d, 1H, $J = 4$ Hz), 8.07–8.09 (d, 1H, $J = 8$ Hz), 7.73–7.75 (d, 1H, $J = 8$ Hz), 7.58–7.62 (t,

1H, $J = 8$ Hz), 7.48–7.52 (t, 1H, $J = 8$ Hz), 7.24–7.25 (d, 1H, $J = 4$ Hz), 3.68–3.7 (t, 2H, $J = 4$ Hz), 3.46 (brs, 4H), 2.80–2.82 (t, 4H, $J = 4$ Hz), 2.67–2.70 (t, 2H, $J = 6$ Hz), 2.56 (brs, 1H).

2-(4-(4-(Pyridin-4-yl)phenyl)piperazin-1-yl)ethanol (21d)—Yield: 64 %, sticky oil, starting with 0.1 g (0.43 mmol) of 1-(4-(pyridin-4-yl)phenyl)piperazine **8f**. $^1\text{H NMR}$ (400 MHz, CDCl_3) δ ppm 8.51–8.52 (d, 2H, $J = 4$ Hz), 7.50–7.52 (d, 2H, $J = 8$ Hz), 7.39–7.40 (d, 2H, $J = 4$ Hz), 6.91–6.93 (d, 2H, $J = 8$ Hz), 3.61–3.64 (t, 2H, $J = 6$ Hz), 3.23–3.25 (t, 4H, $J = 4$ Hz), 3.0 (brs, 1H), 2.63–2.66 (t, 4H, $J = 6$ Hz), 2.56–2.58 (t, 2H, $J = 4$ Hz).

General Procedure for the Preparation of 2-(4-Arylpiperazin-1-yl)acetaldehydes 22a–22d

To a solution of oxalyl chloride (2 equiv) in dichloromethane (20 ml) at -78 °C was added dimethyl sulfoxide (4 equiv). The mixture was stirred and a solution of 2-(4-arylpiperazin-1-yl)ethanol **21** in dichloromethane (10 ml) was added dropwise within 5 min. The stirring was continued for 15–20 min at -78 °C. After the addition of Et_3N (6 equiv), the mixture was allowed to reach to RT and then poured into water (25 ml). The aqueous phase was further extracted with dichloromethane (2×20 ml). The combined organic phases were concentrated in vacuo and the residue obtained was stirred with ether (20 ml). The ether solution was dried (Na_2SO_4) and concentrated in vacuo to yield 2-(4-arylpiperazin-1-yl)acetaldehyde which was used without further purification in the next step.

2-(4-(Quinolin-4-yl)piperazin-1-yl)acetaldehyde (22a)—Yield: 73 %, starting with 0.6 g (2.33 mmol) of 2-(4-(quinolin-4-yl)piperazin-1-yl)ethanol **21a**. This compound was found highly unstable on exposure to air and was used in the next step without further purification.

2-(4-(Quinolin-5-yl)piperazin-1-yl)acetaldehyde (22b)—Yield: 65 %, starting with 0.8 g (3.1 mmol) of 2-(4-(quinolin-5-yl)piperazin-1-yl)ethanol **21b**. $^1\text{H NMR}$ (400 MHz, CDCl_3) δ ppm 9.69 (s, 1H), 8.80–8.81 (d, 1H, $J = 4$ Hz), 8.40–8.42 (d, 1H, $J = 8$ Hz), 7.74–7.76 (d, 1H, $J = 8$ Hz), 7.53–7.57 (t, 1H, $J = 6$ Hz), 7.28–7.31 (dd, 1H, $J_1 = 4$ Hz, $J_2 = 8$ Hz), 7.05–7.07 (d, 1H, $J = 8$ Hz), 3.24 (s, 2H), 3.0–3.1 (t, 4H, $J = 20$ Hz), 2.74–2.77 (t, 4H, $J = 6$ Hz).

2-(4-(Isoquinolin-1-yl)piperazin-1-yl)acetaldehyde (22c)—Yield: 52 %, starting with 0.8 g (3.1 mmol) of 2-(4-(isoquinolin-1-yl)piperazin-1-yl)ethanol **21c**. $^1\text{H NMR}$ (400 MHz, CDCl_3) δ ppm 9.72 (s, 1H), 8.07–8.09 (d, 1H, $J = 8$ Hz), 7.99–8.01 (d, 1H, $J = 4$ Hz), 7.67–7.69 (d, 1H, $J = 8$ Hz), 7.52–7.54 (t, 1H, $J = 4$ Hz), 7.43–7.46 (t, 1H, $J = 6$ Hz), 7.18–7.20 (d, 1H, $J = 8$ Hz), 3.38–3.44 (t, 4H, $J = 12$ Hz), 3.23 (s, 2H), 2.75–2.77 (t, 4H, $J = 4$ Hz).

2-(4-(4-(Pyridin-4-yl)phenyl)piperazin-1-yl)acetaldehyde (22d)—Yield: 70 %, starting with 0.6 g (2.12 mmol) of 2-(4-(4-(pyridin-4-yl)phenyl)piperazin-1-yl)ethanol **21d**. $^1\text{H NMR}$ (400 MHz, CDCl_3) δ ppm 9.67 (s, 1H), 8.50–8.52 (d, 2H, $J = 8$ Hz), 7.50–7.52 (d, 2H, $J = 8$ Hz), 7.38–7.40 (d, 2H, $J = 8$ Hz), 6.91–6.93 (d, 2H, $J = 8$ Hz), 3.25–3.28 (t, 4H, $J = 6$ Hz), 3.19 (s, 2H), 2.62–2.65 (t, 4H, $J = 6$ Hz).

General Procedure for the Preparation of (S)-N⁶-Propyl-N⁶-(2-(4-arylpiperazin-1-yl)ethyl)-4,5,6,7-tetrahydrobenzo[d]thiazole-2,6-diamine 24a–24d

To a solution of 2-(4-arylpiperazin-1-yl)acetaldehyde **22** (1.5 equiv) in dichloroethane, (*S*)-(-)-pramipexole **23** [$[\alpha]_{\text{D}} = -94^\circ$ ($c = 0.5$, CH_3OH)] was added followed by acetic acid (1 equiv). (*S*)-(-) pramipexole was made according to the reported procedure⁴⁴. The mixture was stirred for 1 hr at RT. NaCNBH_3 (3 equiv) was added to the reaction mixture under vigorous stirring. The contents of the flask were stirred overnight. Saturated NaHCO_3 was added and the layers separated. Aqueous layer was extracted with dichloromethane (2×20 ml). The combined organic layers were dried (Na_2SO_4) and the solvent evaporated in vacuo to yield the crude product which was purified by column chromatography.

(S)-N⁶-Propyl-N⁶-(2-(4-(quinolin-4-yl)piperazin-1-yl)ethyl)-4,5,6,7-tetrahydrobenzo[d]thiazole-2,6-diamine (24a)—Yield: 35 mg, starting with 100 mg of (S)-(-)-pramipexole **23**. Mobile phase used for column chromatographic purification was dichloromethane:MeOH (95:5). ¹H NMR(400 MHz, CDCl₃) δ ppm 8.65–8.66 (d, 1H, *J* = 4 Hz), 7.94–7.99 (m, 2H), 7.57–7.60 (t, 1H, *J* = 6 Hz), 7.39–7.43 (t, 1H, *J* = 8 Hz), 6.77–6.78 (d, 1H, *J* = 4 Hz), 4.89 (brs, 2H), 3.20 (brs, 4H), 3.01 (brs, 1H), 2.62–2.73 (m, 9H), 2.40–2.55 (m, 5H), 1.94–1.96 (brm, 1H), 1.65–1.71 (m, 1H), 1.39–1.46 (m, 2H), 0.82–0.86 (t, 3H, *J* = 8 Hz). [α]_D²⁵ = -26.0° (c=1.0, CHCl₃). The free base was converted to HCl salt as white solid, mp=212–214 °C. Analysis calculated for (C₂₅H₃₉Cl₅N₆S.0.5C₂H₅OC₂H₅) C, H, N.

(S)-N⁶-Propyl-N⁶-(2-(4-(quinolin-5-yl)piperazin-1-yl)ethyl)-4,5,6,7-tetrahydrobenzo[d]thiazole-2,6-diamine (24b)—Yield: 24 mg, starting with 50 mg of (S)-(-)-pramipexole **23**. Mobile phase used for column chromatographic purification was dichloromethane:MeOH (95:5). ¹H NMR(400 MHz, CDCl₃) δ ppm 8.87–8.89 (dd, 1H, *J*₁ = 1.6 Hz, *J*₂ = 4 Hz), 8.48–8.51 (d, 1H, *J* = 12 Hz), 7.79–7.81 (d, 1H, *J* = 8 Hz), 7.60–7.64 (t, 1H, *J* = 8 Hz), 7.36–7.39 (dd, 1H, *J*₁ = 4 Hz, *J*₂ = 8.4 Hz), 7.12–7.14 (d, 1H, *J* = 8 Hz), 4.84 (brs, 2H), 3.14 (brs, 5H), 2.59–2.81 (m, 14H), 2.04–2.07 (brm, 1H), 1.73–1.82 (m, 1H), 1.54–1.56 (m, 2H), 0.9–0.94 (t, 3H, *J* = 8 Hz). [α]_D²⁵ = -23.85° (c=0.9, CHCl₃). The free base was converted to HCl salt as yellow solid, mp = 112–114 °C. Analysis calculated for (C₂₅H₃₈Cl₄N₆S) C, H, N.

(S)-N⁶-(2-(4-(isoquinolin-1-yl)piperazin-1-yl)ethyl)-N⁶-propyl-4,5,6,7-tetrahydrobenzo[d]thiazole-2,6-diamine (24c)—Yield: 48 mg, starting with 50 mg of (S)-(-)-pramipexole **23**. Mobile phase used for column chromatographic purification was dichloromethane:MeOH (95:5). ¹H NMR(400 MHz, CDCl₃) δ ppm 8.06–8.07 (d, 1H, *J* = 4 Hz), 8.0–8.02 (d, 1H, *J* = 8 Hz), 7.66–7.68 (d, 1H, *J* = 8 Hz), 7.51–7.55 (t, 1H, *J* = 8 Hz), 7.41–7.45 (t, 1H, *J* = 8 Hz), 7.16–7.18 (d, 1H, *J* = 8 Hz), 4.79 (brs, 2H), 3.40 (brs, 4H), 3.04 (brm, 1H), 2.62–2.74 (m, 9H), 2.43–2.58 (m, 5H), 1.95–1.97 (brm, 1H), 1.64–1.73 (m, 1H), 1.40–1.50 (m, 2H), 0.82–0.86 (t, 3H, *J* = 8 Hz). [α]_D²⁵ = -30.56° (c=0.82, CHCl₃). The free base was converted to oxalate salt as white solid, mp=100–102 °C. Analysis calculated for (C₃₅H₄₄N₆O₂₀S.H₂O) C, H, N.

(S)-N⁶-Propyl-N⁶-(2-(4-(4-(pyridin-4-yl)phenyl)piperazin-1-yl)ethyl)-4,5,6,7-tetrahydrobenzo[d]thiazole-2,6-diamine (24d)—Yield: 20 mg, starting with 35 mg of (S)-(-)-pramipexole **23**. Mobile phase used for column chromatographic purification was dichloromethane:MeOH (95:5). ¹H NMR(400 MHz, CDCl₃) δ ppm 8.51–8.52 (d, 2H, *J* = 8 Hz), 7.50–7.52 (d, 2H, *J* = 8 Hz), 7.39–7.40 (d, 2H, *J* = 4 Hz), 6.91–6.93 (d, 2H, *J* = 8 Hz), 4.65 (brs, 2H), 3.21–3.23 (t, 4H, *J* = 4 Hz), 3.0 (brs, 1H), 2.44–2.66 (m, 14H), 1.92–1.99 (brm, 1H), 1.61–1.72 (m, 1H), 1.39–1.45 (m, 2H), 0.81–0.85 (t, 3H, *J* = 8 Hz). [α]_D²⁵ = -22.08° (c=0.56, CHCl₃). The free base was converted to HCl salt, off white solid, mp = 206–210 °C. Analysis calculated for (C₂₇H₄₁Cl₅N₆S.0.8H₂O.0.5C₂H₅OC₂H₅) C, H, N.

Evaluation of potency in binding to and activating dopamine D2 and D3 receptors

Binding potency was monitored by inhibition of [³H]spiperone (15.0 Ci/mmole, Perkin-Elmer) binding to dopamine rD2 and rD3 receptors expressed in HEK-293 cells, in a buffer containing 0.9% NaCl as described by us previously⁴⁴. Functional activity of test compounds in activating dopamine hD2 and hD3 receptors expressed in CHO and AtT cells, respectively, was measured by stimulation of [35S]GTPγS (1250 Ci/mmole, Perkin-Elmer) binding in comparison to stimulation by the full agonist dopamine as described by us previously⁴⁴.

Evaluation of antioxidant activity

DPPH Radical Scavenging Assay—To a 96 well plate 40 μL of different concentrations of methanolic drug solutions ranging from 20 μM to 250 μM were added. 200 μL of 100 μM methanolic solution of DPPH (1,1-Diphenyl-2-picryl-hydrazyl) was added, shaken vigorously at 30 $^{\circ}\text{C}$ for 20 min. Control wells received 40 μL methanol and 200 μL methanolic DPPH solution. Wells containing only 240 μL methanol served as back ground for base line correction. The changes in the absorbance of all the samples and standard (ascorbic acid) were measured at 517 nm. the radical scavenging activity was expressed as the inhibition percentage and was calculated using the following formula:- % scavenging activity = [(absorbance of control- absorbance of sample)/absorbance of control] \times 100.⁵⁴

Animal Experiment

Drugs and chemicals—The following commercially available drugs were used in the experiment: reserpine hydrochloride (Alfa Aesar), Ropinirole (Sigma Aldrich), The hydrochloride salts of compounds **24a**, **24b**, **24c** and ropinirole were dissolved in water for both locomotor and 6-OH-DA rotational experiments. Reserpine was dissolved in 10–25 μL of glacial acetic acid and further diluted with 5.5% glucose solution. All compounds for this study were administered s.c. in a volume of 0.1 to 0.2 mL into each rat.

Animals—In rodent studies, animals were male Sprague-Dawley rats from Harlan (Indianapolis, IN) weighing 220–225 g unless otherwise specified. The lesioned rats (290–320 g) were purchased from Taconic Biotechnology (Rensselaer, NY) and their unilateral lesion was checked twice by apomorphine challenge following the surgery. Animals were maintained in sawdust-lined cages in a temperature and humidity controlled environment at $22 \pm 1^{\circ}\text{C}$ and $60 \pm 5\%$ respectively, with a 12-h light/dark cycle, with lights on from 6:00 AM to 6:00 PM. They were group housed with unrestricted access to food and water. All experiment was performed during the light component. All animal use procedures were in compliance with the Wayne State University Animal Investigation Committee consistent with AALAC guidelines.

Reversal of Reserpine-Induced Hypolocomotion in Rats—Administration of reserpine induces catalepsy in rodents primarily by blocking the vesicular monoamine transporter (VMAT) which helps in the internalization of monoamines into vesicles, resulting in metabolism of unprotected monoamines in the cytosol that ultimately causes depletion of monoamines in the synapse of the peripheral sympathetic nerve terminals^{55, 58}. The ability of the compounds (**24a**, **24b** and **24c**) to reverse the reserpine induced hypolocomotion was investigated⁵⁹. Ropinirole was used as standard reference compound in this study. Reserpine (5.0 mg/kg, s.c.) or saline (s.c.) were administered 18h before the injection of drug or vehicle (s.c.). The rats were placed individually in chambers for 1 h for acclimatization purpose before the administration of the test drug, standard drug or vehicle. Immediately after administration of drug or vehicle, animals were individually placed in versamax animal activity monitor chamber (45 \times 30 \times 20 cm) (AccuScan Instruments, Inc. Columbus, OH) to start measuring locomotor activity. Locomotion was monitored for 6 h. Consecutive interruption of two infrared beams situated 24 cm apart and 4 cm above the cage floor in the monitor chamber recorded movement. The data were presented as horizontal counts (HACTV). The effect of the individual doses of drugs on locomotor activity was compared with respect to saline treated controls (mean \pm S.E.M.). The data were analyzed by one way analysis of variance (ANOVA) followed by Dunnett's post-hoc test. The effect was considered significant if the difference from control group was observed at $p < 0.05$.

In vivo Rotational experiment with 6-OH-DA lesioned rats

The first 14 days post-lesion challenge with apomorphine was done with lesioned animals to observe a complete rotation session post administration. In the second challenge with apomorphine (0.05 mg/kg) 21 days post lesion, contralateral rotations were recorded for 30 min; apomorphine produced rotations in all four rats (average rotation > 250) indicating successful unilateral lesion. In these rats, lesion was performed on the left side of the medial forebrain bundle in the brain and the coordinates used from Bregma are: AP-4.3, ML +1.2, DV -8.3. The rotations produced upon agonist challenge occurring clockwise. In this study, apomorphine was also used as a reference compound. The test drugs including ropinirole were dissolved in saline and were administered SC. The rats were brought to the test room 1 h before the administration of the test drug, standard drug or vehicle for acclimatization purpose. The rotations were measured over 7–10 hours. For control, vehicle was administered alone. Rotations were measured in the Rotomax Rotometry System (AccuScan Instruments, Inc. Columbus, Ohio) equipped with Rotomax Analyser, high resolution sensor and animal chambers with harnesses. Data were analyzed with Rotomax Window software program. Test drugs **24c** (5 and 10 $\mu\text{Mol/kg}$), **24b** (5 and 10 $\mu\text{Mol/kg}$) and ropinirole (5 and 10 $\mu\text{Mol/kg}$) dissolved in saline were administered sc. The rotations were measured in a rotational chamber immediately after administration of drugs. The data were collected at every 30 min. Data were analyzed by Graph Pad (Version 4, San Diego) program. All drugs produced contralateral rotations in all lesioned rats which lasted over 3–10 hours.

Supplementary Material

Refer to Web version on PubMed Central for supplementary material.

Abbreviations

GTP γ S	Guanosine 5'-[γ -thio]triphosphate
5-OH-DPAT	5-Hydroxy-2-(dipropylamino)tetralin
6-OHDA	6-Hydroxy dopamine
CHO	Chinese Hamster Ovary
HEK	Human embryonic kidney
L-DOPA	(S)-(3,4-Dihydroxyphenyl) Alanine
DPPH	1,1-Diphenyl-2-picryl-hydrazyl
PD	Parkinson's Disease

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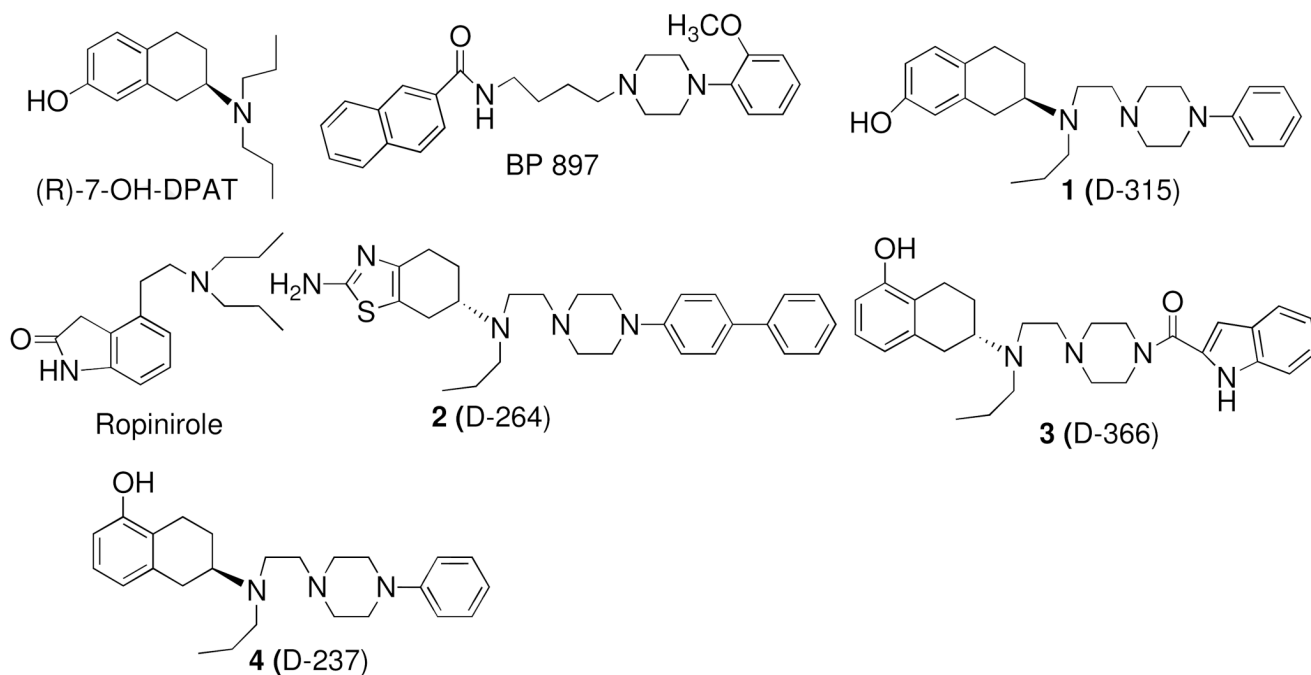


Figure 1.
Molecular structure of dopamine D3 receptor preferring agonists and antagonists.

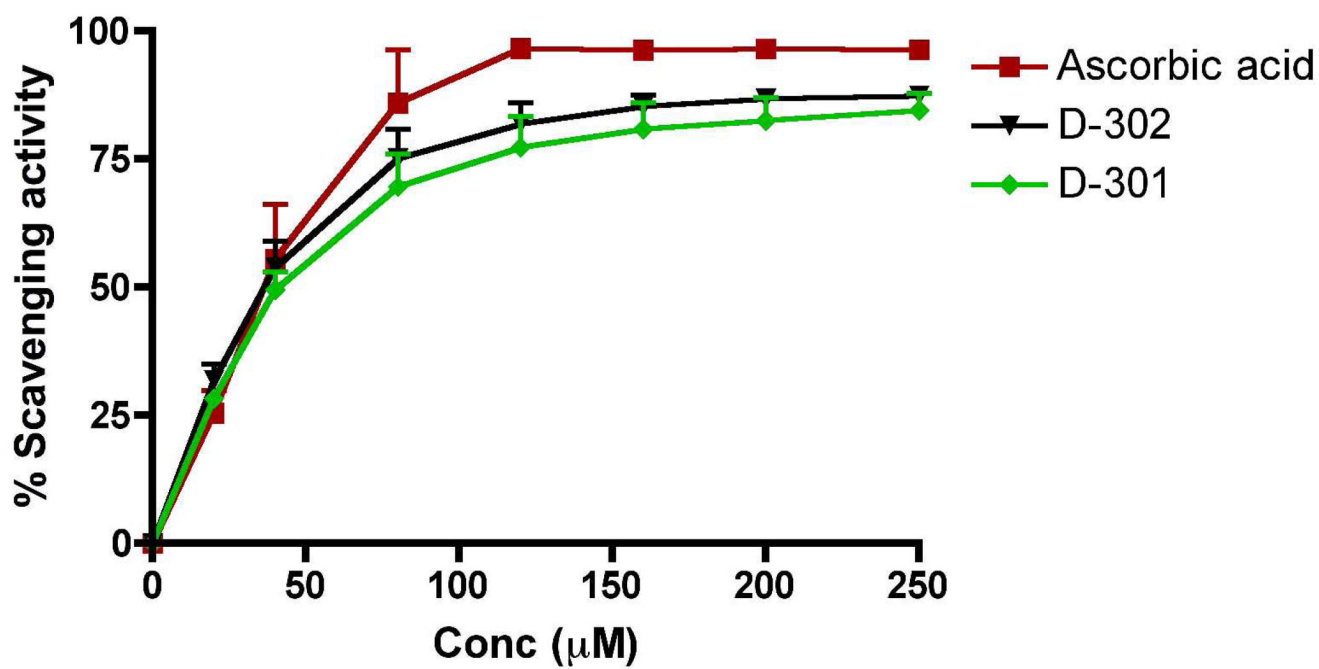
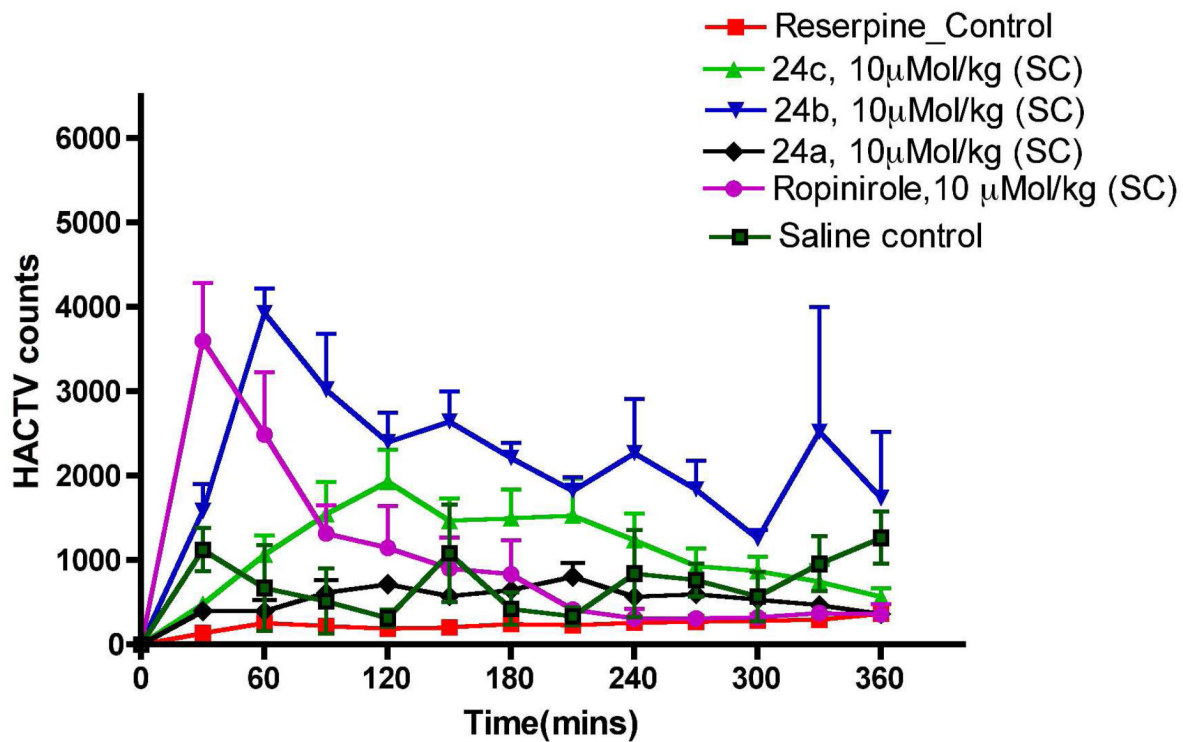


Figure 2.
DPPH radical scavenging activity by **24b**, **24c** and ascorbic acid.

A**B**

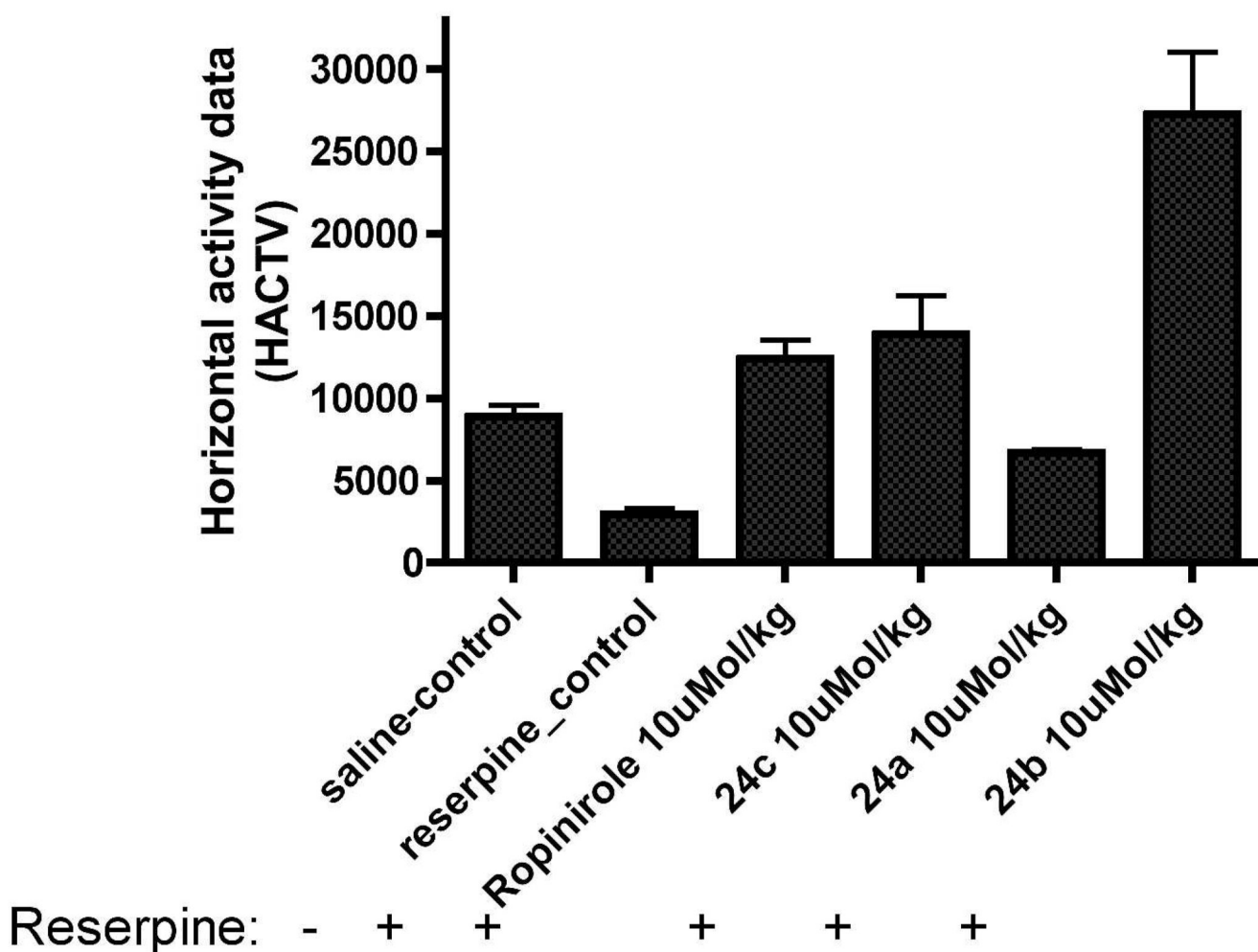


Figure 3.

Effects of different drugs (administered s.c.) upon reserpine (5.0 mg/Kg, s.c., 18 h pretreatment)-induced hypolocomotion in rats. Each point is the mean \pm S.E.M for six rats. Horizontal activity was measured as described under materials and methods section. **Panel A.** Representation of horizontal locomotor activity at discrete 30-min intervals after the administration of **24a**, **24b**, **24c** and ropinirole at the dose of 10 μ Mol/kg compared to control rats in 18 h reserpine post treatment. **Panel B.** Representation of cumulative horizontal activity data for 6 h period of observation for saline-control meaning animals receive saline instead of reserpine, reserpinized control, and 10 μ Mol/kg of ropinirole, **24a**, **24b** and **24c**. One way ANOVA analysis demonstrates significant effect among treatments: **Panel A**, $F(6,95) = 13.69$ ($P < 0.0001$). Dunnett's analysis following ANOVA showed that the effects of **24b** ($P < 0.01$), **24c** ($P < 0.01$) and ropinirole ($P < 0.05$) were significantly different statistically compared to reserpine control but for **24a** was not statistically significant ($P > 0.05$).

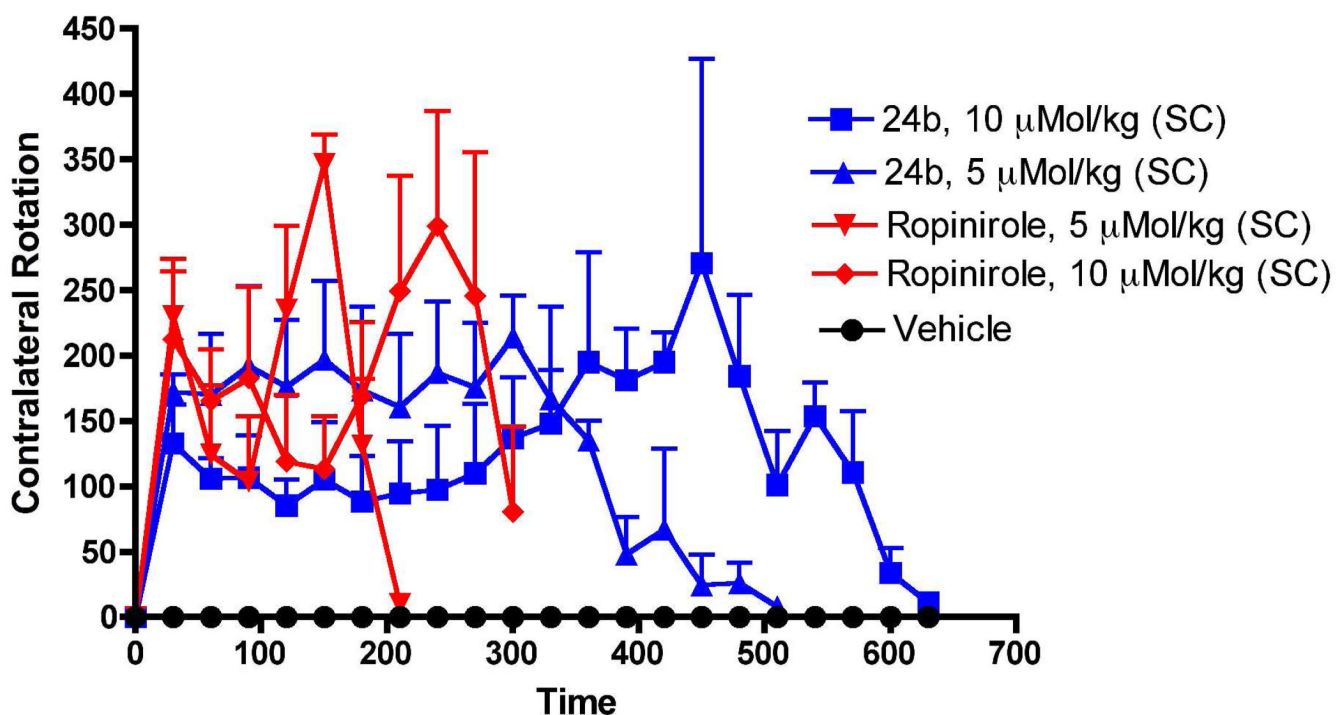


Figure 4. Effect on turning behavior of (–)-**24b** and ropinirole in 6-OH-DA unilaterally lesioned rats studied over ten hours. Each point is the mean \pm SEM for four rats. All drugs were administered s.c. immediately before counting rotational activity. One way ANOVA analysis demonstrates a significant effect among treatments: $F(5,95) = 16.93$ ($P < 0.0001$). Dunnett's analysis shows that the effect of (–)-**24b** at two doses (5 and 10 μ Mol/kg) and ropinirole at two doses (5 and 10 μ Mol/kg) are significantly different compared to vehicle ($P < 0.01$).

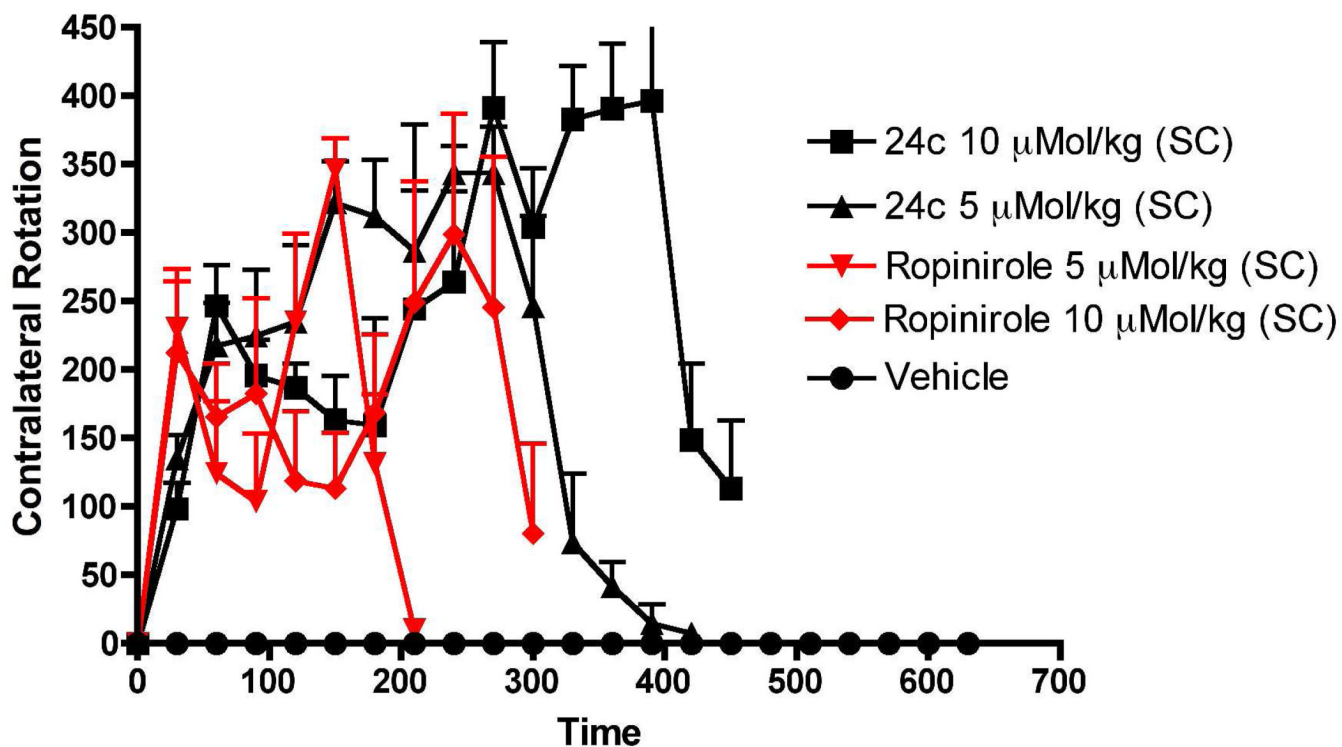
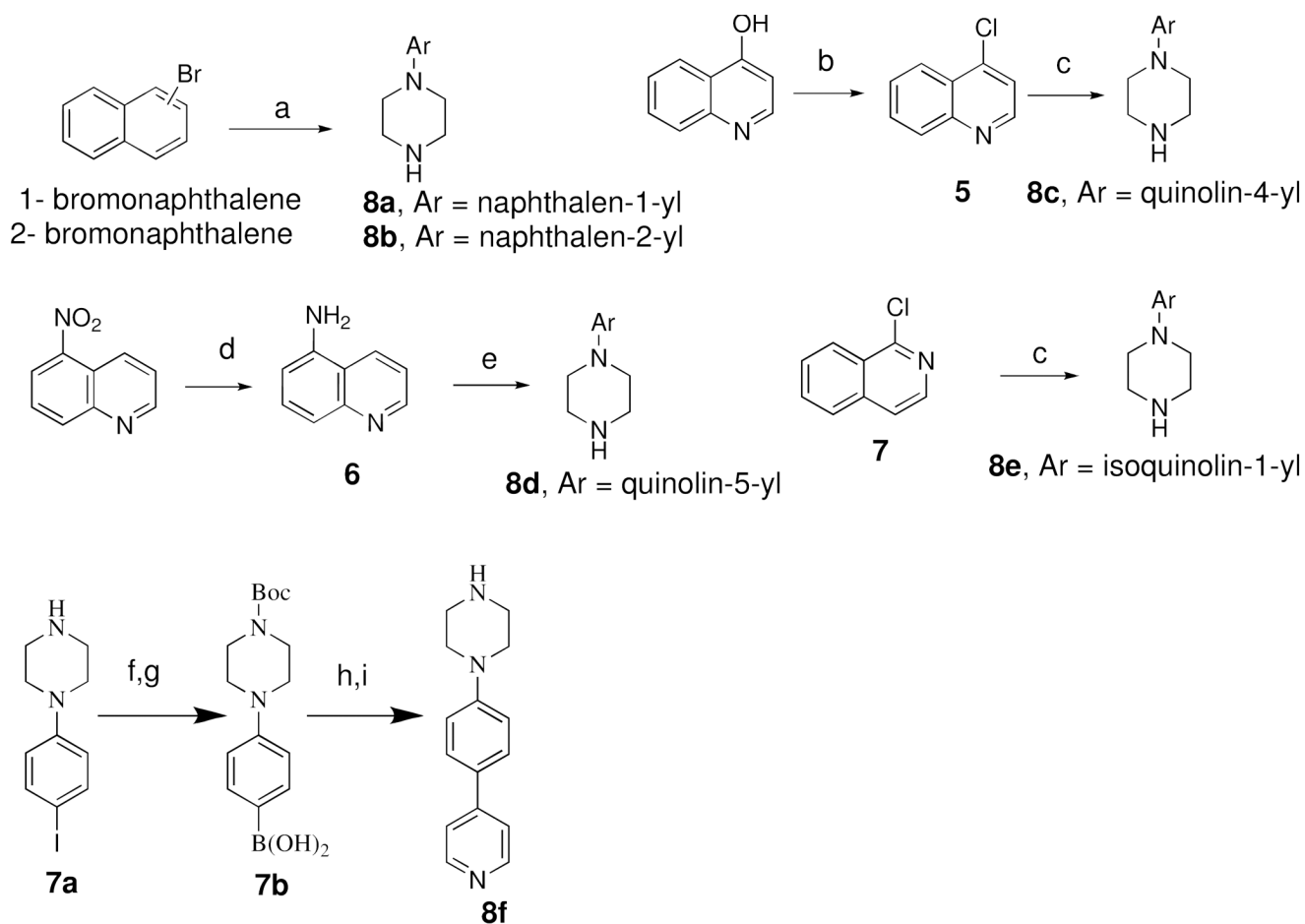
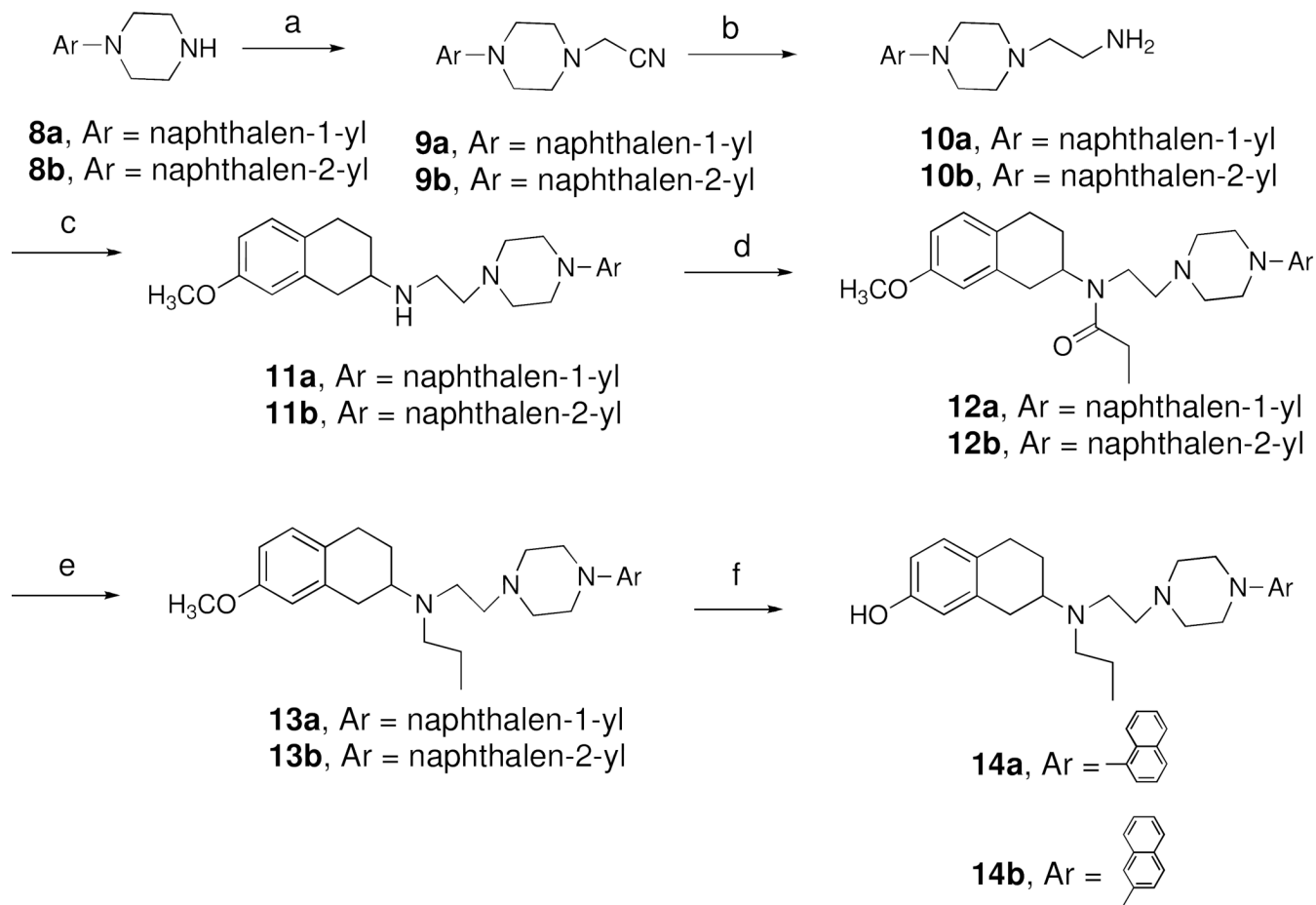


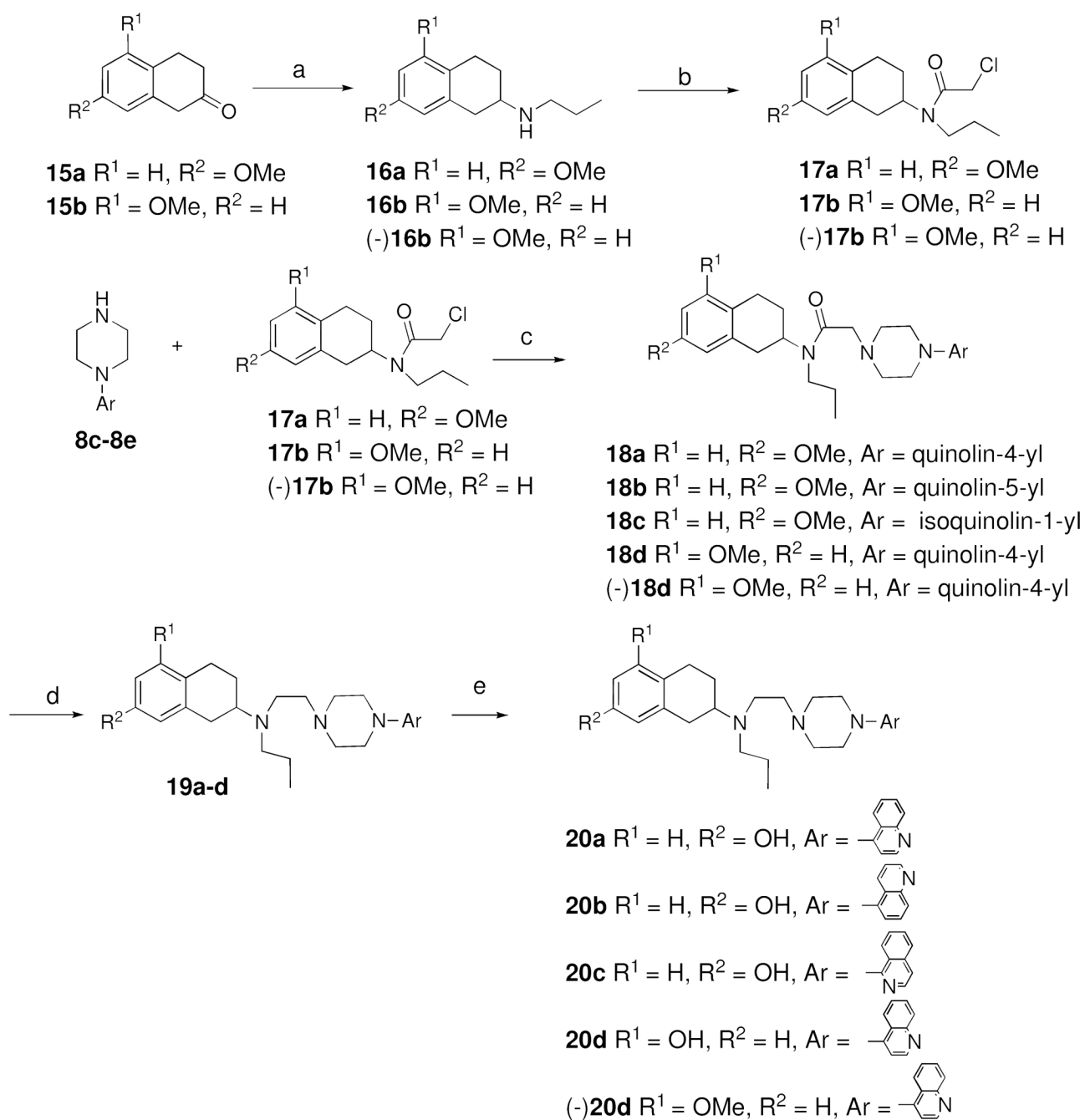
Figure 5. Effect on turning behavior of (–)-**24c** and ropinirole in 6-OH-DA unilaterally lesioned rats studied over seven and half hours. Each point is the mean \pm SEM for four rats. All drugs were administered s.c. immediately before counting rotational activity. One way ANOVA analysis demonstrates a significant effect among treatments: $F(5,95) = 16.23$ ($P < 0.0001$). Dunnett's analysis shows that the effect of (–)-**24c** at two doses (5 and 10 $\mu\text{Mol/kg}$) and ropinirole at two doses (5 and 10 $\mu\text{Mol/kg}$) are significantly different compared to vehicle ($P < 0.01$).

**Scheme 1a.**

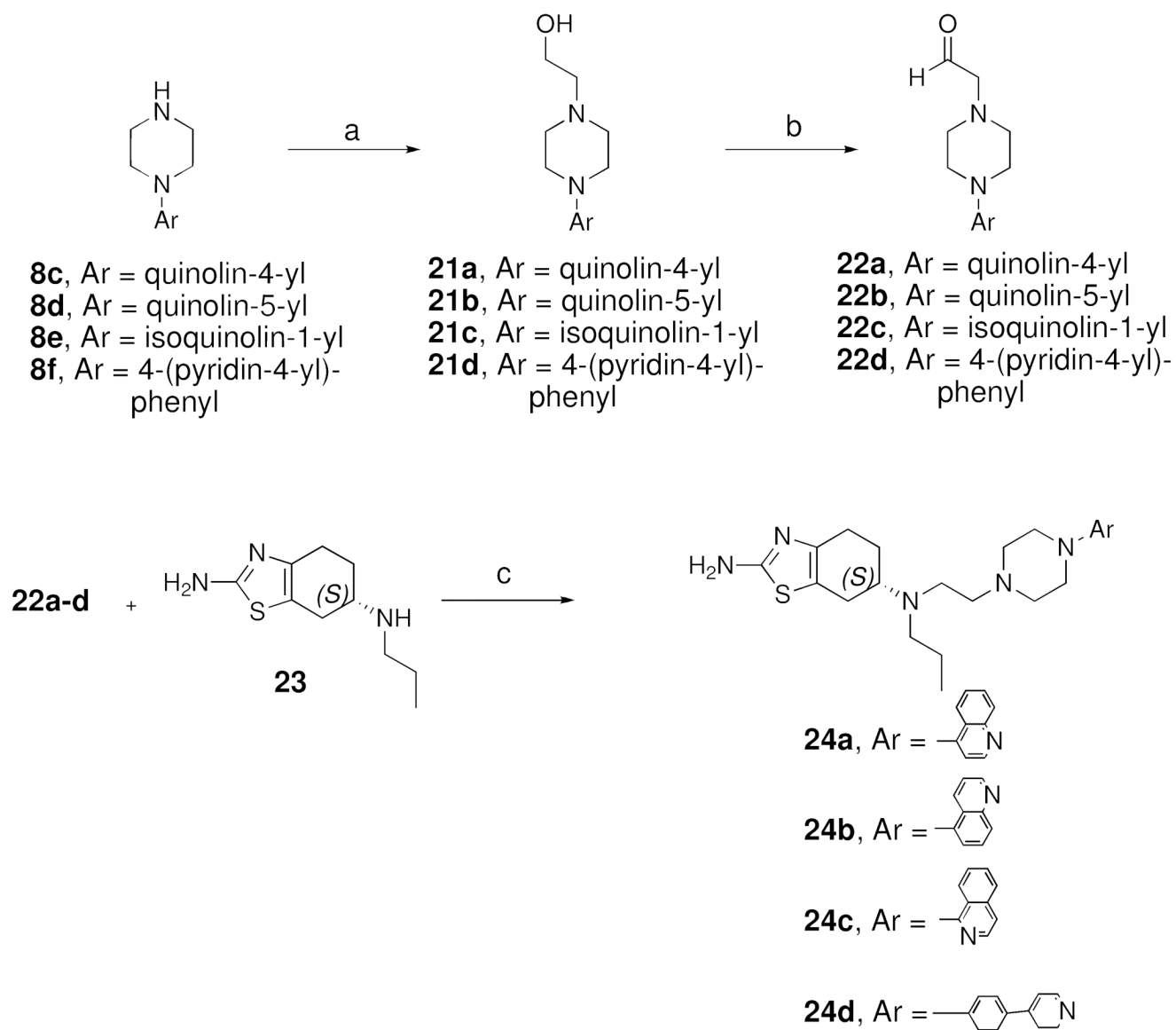
^a Reagents and Conditions : a. 3–5 mol% PdCl₂[P(*o*-tol)₃]₂, NaOt-Bu, Diglyme, reflux, 48h; b. POCl₃, reflux, 2h; c. Piperazine, *i*-propanol, reflux, 4h; d. SnCl₂·2H₂O, ethanol, NaBH₄, reflux 55 °C, 1.5h; e. bis-(2-chloroethyl)amine HCl, diglyme, K₂CO₃, reflux, 48h; f. Boc₂O, EtOH; g. n-BuLi, trimethyl borate, –78 °C; h. 4-bromopyridine, Pd(PPh₃)₄, Na₂CO₃; i. TFA/CHCl₃;

**Scheme 2a.**

^a Reagents and Conditions : a. Chloroacetonitrile, K_2CO_3 , toluene, reflux, 3h; b. Raney nickel, H_2 , 60 psi, 8 h; c. 7-methoxy-2-tetralone, $NaCNBH_3$, AcOH, dichloroethane, RT, overnight; d. propionyl chloride, Et_3N , CH_2Cl_2 , 0 °C to RT, 4h; e. $LiAlH_4$, THF, reflux, 4 h; f. BBr_3 , CH_2Cl_2 , -40 °C to RT, overnight.

**Scheme 3a.**

^a Reagents and Conditions: a. n-Propylamine, NaCNBH₃, CH₃COOH, dichloroethane, RT, overnight; b. chloroacetyl chloride, TEA, dichloromethane, 0°C, 30 min; c. K₂CO₃, CH₃CN, 80 °C, 2 h; d. LiAlH₄, THF, reflux, 2 h; e. BBr₃, -78° C, CH₂Cl₂, overnight

**Scheme 4a.**

^a Reagents and Conditions : a. 2-bromoethanol, K₂CO₃, AcCN, reflux; b. oxalyl chloride, DMSO, Et₃N, -78 °C; c. NaCNBH₃, AcOH, 1,2-dichloroethane

Table 1

Inhibition constants for competing for [³H]spiperone binding to cloned rat D_{2L} and D₃ receptors expressed in HEK cells. Results are means ± SEM for 3–5 independent experiments each performed in triplicate.

Compound	K _i , (nM), D _{2L} [³ H]Spiperone	K _i , (nM), D ₃ [³ H]Spiperone	D _{2L} /D ₃
7-OH-DPAT	202 ± 34	2.35 ± 0.29	86
(-)-5-OH-DPAT	58.8 ± 11.0	1.36 ± 0.28	43
4^a	26.0 ± 7.5	0.825 ± 0.136	31.5
2^b	264 ± 40	0.92 ± 0.23	253
1^c	40.6 ± 3.6	1.77 ± 0.42	22.9
3^c	47.5 ± 6.2	0.570 ± 0.094	83
14a	13.5 ± 3.0	0.435 ± 0.13	31.0
14b	8.85 ± 1.75	0.685 ± 0.217	12.9
20a	7.08 ± 0.34	2.65 ± 0.65	2.67
20b	5.22 ± 0.71	1.27 ± 0.52	4.11
20c	21.7 ± 4.1	0.441 ± 0.101	49.2
20d	4.89 ± 0.20	0.40 ± 0.09	12.23
(-)- 20d	3.74 ± 0.70	0.19 ± 0.03	19.68
(-)- 24a	109 ± 14	2.61 ± 0.18	41.8
(-)- 24b	57.7 ± 3.3	1.21 ± 0.16	47.7
(-)- 24c	269 ± 16	2.23 ± 0.60	121
(-)- 24d	270 ± 28	4.78 ± 0.89	56.5

^aSee Ref # 45,

^bSee Ref # 44,

^cSee Ref # 53

Table 2

Stimulation of [³⁵S]GTPγS binding to cloned human D2 receptor expressed in CHO cells and cloned human D3 receptor expressed in AtT-20 cells. EC50 is the concentration producing half-maximal stimulation; for each compound, maximal stimulation (Emax) is expressed as percent of the Emax observed with 1 mM (D2) or 100 μM (D3) of the full agonist DA (%Emax). Results are means ± SEM for 3–5 independent experiments each

Compound	CHO-D2		AtT-D3		D2/D3
	EC50 (nM) [³⁵ S]GTPγS	%Emax	EC50 (nM) [³⁵ S]GTPγS	%Emax	
Dopamine	209 (4) ± 29	100	8.53 ± 0.62	100	
Ropinirole	304 ± 11	83.9 ± 0.3	19.2 ± 5.1	91.6 ± 0.5	15.83
24c	116 ± 16	88.4 ± 3.9	0.520 ± 0.16	95.1 ± 2.1	223
24b	13.5 ± 4.3	104 ± 2	0.528 ± 0.077	101 ± 2	25
24a	77.2 ± 8.3	118 ± 4	0.493 ± 0.12	96.8 ± 4.4	156
5-OH-DPAT	41.2 ± 6.0	80.0 ± 4.4	1.23 ± 0.53	91.2 ± 1.0	33.49
2	19.9 ± 0.9	119 ± 6	0.085 ± 0.016	102 ± 19	248