Multiple translational products from a five-nucleotide ribozyme

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Edited by Harry F. Noller, University of California Santa Cruz, Santa Cruz, CA, and approved January 27, 2010 (received for review November 14, 2009)

An indispensable step in protein biosynthesis is the 2'(3') aminoacylation of tRNA by aminoacyl-tRNA synthetases. Here we show that a similar activity exists in a tiny, 5-nt-long RNA enzyme with a 3-nt active center. The small ribozyme initially *trans*-phenylalanylates a partially complementary 4-nt RNA selectively at its terminal 2'-ribose hydroxyl using PheAMP, the natural form for activated amino acid. The initial 2' Phe-RNA product can be elaborated into multiple peptidyl-RNAs. Reactions do not require divalent cations, and have limited dependence on monovalent cations. Small size and minimal requirements for regiospecific translational activity strongly support the hypothesis that minuscule RNA enzymes participated in early forms of translation.

aminoacyl-RNA | enzyme | evolution | peptidyl-RNA | RNA

A mino acids enter modern translation via attachment to a 2'(3') tRNA terminus, a reaction catalyzed by a protein aminoacyl-tRNA synthetase. Because it is implausible that primitive peptides were synthesized using already-formed protein catalysts, the RNA world hypothesis (1, 2) requires peptide synthetic reactions performed by RNA enzymes (3, 4). Indeed, a number of RNAs have been isolated which accelerate related translational reactions (5).

Several ribozymes capable of catalyzing the same chemical group transfer that is today carried out by aminoacyl-tRNA synthetases (6–9) have been isolated using Systematic Evolution of Ligands by Exponential Enrichment (10, 11). However, none possess all desirable characteristics. First, an RNA world enzyme should be small, accessible after rudimentary RNA synthesis. In addition, it should act *in trans*, and should use universal biological, water-soluble substrates. Previously isolated ribozymes employ appropriate substrates [amino acids activated as acyladenylates (6, 12)], but are not true enzymes, as they are self-aminoacylators modified by their own reaction. Other ribozymes aminoacylate RNA with turnover; however, the amino acids must be activated as cyanomethyl, 3,5-dinitrobenzyl, or *p*-chlorobenzyl esters (8, 13), and hence they do not facilitate the biological reaction.

The family of self-aminoacylating ribozymes exemplified by truncate C3 RNA (Fig. 1*A*) presented the intriguing possibility of very simple aminoacyl transfer (6). Mutational analyses as well as molecular dynamics and energy minimization of the reactants suggested a tiny active center consisting of only three essential nucleotides—a 3'-terminal U, and a 5'-GU-3' sequence in a loop apposed to the unpaired 3'-terminal U. Although the C3 RNA family possessed helical elements, adjacent helices appeared nonspecific in sequence, perhaps required only for assembly of the active center (6).

Here we present a radically modified version of C3 ribozyme, unique in three ways: It functions *in trans*; it has been minimized to a tiny, five-nucleotide ribozyme; and it also supports peptidyl-RNA synthesis. The result is the smallest *trans*-aminoacylator, and arguably the smallest true ribozyme, ever observed (Fig. S1).

Results

Initially, RNAC3 (6) was separated into "enzyme" and "substrate" modules (Fig. 1B). Subsequently, a succession of progressively

smaller complexes were tested for activity, shortening the helix that linked the ribozymic fragment to the substrate, and also discarding the right-hand hairpin completely (Fig. 1*B*). Covalent continuity, half of the substrate helix, and the entire rightward structure, proved dispensable. Each of these modified RNAs made Phe-RNA from PheAMP (Fig. S2). These results are consistent with lack of conservation of the discarded structures in the initially selected RNAs (6). The terminus of this molecular minimization is the 5-nt ribozyme/4-nt substrate complex, GUGGC/ GCCU (Fig. 1*C*).

The 5-nt ribozyme/4-nt substrate complex was the smallest that effectively produced aminoacyl-RNA under our standard conditions. A smaller complex with only two base pairs between the enzymic and substrate elements (GUGG/CCU) yielded no observable product(s). The additional GC base pair in GUGGC/GCCU apparently provides marginal stability required for substrate acquisition and/or aminoacyl transfer.

Ribozyme 5'-<u>GU</u>GGC-3' (GUGGC) catalyzes the addition of phenylalanine to the 3'-terminal uridine of RNA substrate 5'-GCC<u>U</u>-3' (GCCU), where the underlined ribonucleotides comprise the specific nucleotides of the active site. When GUGGC and ³²P-radiolabeled GCCU are incubated with PheAMP, product bands are evident on an acid urea polyacrylamide gel (Fig. 2). These multiple products are shown below to be RNA-Phe, RNA-Phe₂, and RNA-Phe₃. Thus GUGGC/GCCU complex hosts two of three essential chemical group transfers that occur in biological peptide synthesis: aminoacylation of RNA and subsequent peptide extension.

Shortened substrate lacking unpaired 3'-terminal U (GCC) did not produce detectable product bands when incubated with GUGGC and PheAMP, suggesting the importance of the unpaired 3'-terminal substrate nucleotide in the reaction (Fig. 2, lane 5). Further, no product was formed when GCCU was pretreated with periodate (NaIO₄), which selectively oxidizes the 2'(3')-terminal *cis*-glycol of ribose, converting the sugar to an open dialdehyde (Fig. 2, lane 4). Gel resolution of the unoxidized GCCU band from the periodate product shows that oxidation was quantitative. These data suggest that phenylalanine transfer is to the ribose hydroxyl(s), as in modern enzymatic aminoacyl-tRNA synthesis.

This transfer was confirmed and defined by use of a synthetic substrate oligomer lacking the terminal 2'-hydroxyl group (GCC2'dU; Fig. 2, lane 2). This molecule yields no visible products under the conditions used here, suggesting that phenylalanine is initially transferred to the 2'-terminal –OH. Initial 2' aminoacylation was further confirmed by facile Phe-RNA formation when an

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Author contributions: R.M.T. and M.Y. designed research; R.M.T. performed research; N.V.C. contributed new reagents/analytic tools; M.Y. analyzed data; and R.M.T. and M.Y. wrote the paper.

The authors declare no conflict of interest.

This article is a PNAS Direct Submission.

Freely available online through the PNAS open access option.

This article contains supporting information online at www.pnas.org/cgi/content/full/ 0912895107/DCSupplemental.



Fig. 1. Small trans-aminoacylating RNA complexes. (A) C3 RNA. (B) Intermediate trans complexes. (C) Final GUGGC/GCCU complex.

alternative substrate with a single 2'-terminal hydroxyl, GCC3'dU, was aminoacylated (Fig. 2, lane 3). This reactivity verifies that the 2'-hydroxyl group, but not the 3', is required. Further, lanes 2, 3, 4, 5, and 6 confirm that internal 2'-OH groups are not the site of aminoacyl transfer: for example, a DNA-RNA hybrid substrate dGdCdCrU retained activity (Fig. 2, lane 6). Accordingly, the terminal ribose 2'-OH is the unique site of initial aminoacyl trans-



Fig. 2. Initial reaction occurs only at 2'-OH. Lanes: 1, substrate = GCCU; 2, GCC2'dU; 3, GCC3'dU; 4, GCCU + NaIO₄; 5, GCC; 6, dGdCdCrU; 7, GCCU + PheAMP, no GUGGC; 8, GCCU + GUGGC, no PheAMP. Reaction conditions: 10 μ M ribozyme GUGGC, 20 μ M RNA substrate, 100 mM KCl, 5 mM MgCl₂, 100 mM Hepes pH 7.0, 2.0 mM PheAMP; incubation at 4 °C for 30 min.

fer. These results parallel similar findings for the parental C3 ribozyme (6), and will be relevant below to discussion of multiple products, all now shown to depend on the earliest transfer to terminal 2'-OH.

The rate of Phe-RNA formation from 3'-dU substrate $(k_{2nd \text{ order}} \cong k_{cat}/K_m = 1.2 \text{ M}^{-1} \text{ min}^{-1})$ is 18% that of the original RNA GCCU $(k_{2nd \text{ order}} = 6.5 \text{ M}^{-1} \text{ min}^{-1})$ under comparable conditions and where PheAMP instability has been accounted for (Fig. S3). This implies a minor catalytic role for the 3'-hydroxyl group, though it is not required for aminoacylation or peptide formation. This GUGGC/GCCU rate is about an order slower than other small ribozyme self-acylators (12). This minor difference is remarkable in view of the lack of higher-order structure at the active center. In many respects, the small active center appears unchanged from that of parental C3 RNA. The 5'-terminal guanosine of GUGGC is indispensable for the reaction. Both mutation of 5'-G to C, and elimination of 5'-G altogether, completely abolished product formation (Fig. S4). This parallels prior mutational and computational analyses of the active center in C3 RNA, where the analogous G supports critical intermolecular interactions during aminoacyl transfer (6).

GUGGC/GCCU accepts varied aminoacylation substrates, activated as mixed phosphoric anhydrides. Both PheUMP and MetAMP are substrates as monitored by gel electrophoresis (Fig. S5). Both these substrates were also reactive with the C3 ribozyme (6). GUGGC/GCCU therefore tolerates alternative substitution on the phosphate oxygen (A vs. U) and aminoacyl residue (Phe vs. Met), as did C3 RNA. Therefore, both the 2' regioselectivity and substrate tolerance of the original C3 RNA active center are still demonstrable in GUGGC, even though most of the primary and secondary structure of the initially selected RNA catalyst have been discarded. Indeed, the originally selected RNA molecule (the selected parent of C3 RNA) was a specifically



Fig. 3. Aminoacylation reaction does not require divalent cations. All reactions contained 10 μ M GUGGC, 20 μ M GCCU, 1.9 mM PheAMP, 100 mM Hepes pH 6.5, 10 mM NaOH. Reactions contained 100 mM KCl, 5 mM MgCl₂, or 100 mM NaCl where indicated. Incubation at 4 °C for 60 min.

3'-truncated 69-mer; thus in the present work, 87% of the initial RNA has been deleted.

Unlike most ribozymes, GUGGC does not require divalent cations for function. Removal of divalent cations from the reaction reduced activity by only 6.5% (Fig. 3). The implied ionic tolerance is perhaps not surprising because the GUGGC/GCCU complex probably has a relatively simple 3D structure, and thus does not require divalent cations for folding. Magnesium is somewhat stimulatory, however, as are potassium and sodium. We observed slower but significant product formation at 10 mM Na⁺, but did not test the enzyme's function in the complete absence of monovalents as some monovalents were introduced during adjustment of the pH of the reaction buffer. Nondiscrimination toward metallocations further supports such enzyme function in primitive environments, where particular divalents and monovalents may have been variably available.

MALDI mass spectrometric analysis of Product 1 confirmed the presence of a compound with the expected mass of quadruply protonated GCCU-Phe (Fig. 4*A* and Fig. S6). To summarize, the carboxyl-activated substrate PheAMP transfers the atoms of phenylalanine to ribose 2'-hydroxyl, paralleling the parental self-aminoacylating RNA. Therefore, the first product is the ester Phe-RNA. To further show that the first product formed is an ester, Product 1 was purified and subjected to mild base hydrolysis (pH 8.5, 37 °C). Product 1 hydrolyzed to GCCU (as monitored by gel electrophoresis) at a rate of (0.036 min⁻¹). This rate is close to the previously reported rate of tRNA-Phe hydrolysis under similar conditions (0.042 min⁻¹), confirming that the initial product is RNA-Phe ester (14).

Reaction of GUGGC with GCCU and PheAMP under these conditions yields multiple slower-migrating products, which are RNA peptides. MALDI analysis of Product 2 confirmed a compound with the same mass as quadruply protonated GCCU-Phe₂ peptidyl-RNA (Fig. 4*B* and Fig. S6). Products 2 and 3 are also susceptible to digestion by exopeptidase, which confirms that they are peptidyl-RNAs (Fig. 5). Treatment of standard reaction products with leucine aminopeptidase, which hydrolyzes N-terminal peptidyl amino acids but not ester bonds, resulted in the degradation of all products to Product 1 (GCCU-Phe ester) (Fig. 5). Thus, the observed products (Figs. 2 and 5) are GCCU-Phe through GCCU-Phe₃ peptidyl-RNA. The absence of any products in reactions employing GCC and GCC2'dU instead of GCCU

GUGGC 3000 GCCU-Phe GCCU (1583.25) (1425.23) (1278.16) 1583.10 1425.19 Intensity 1278.21 2000 1000 0 B 2000 GCCU-Phe (1572.30)1572.54 Intensity 1000 n 1200 1500 1600 1300 1400 1700 Mass (m/z)

4000

Fig. 4. MALDI-TOF mass spectrometric analysis of RNA and Products 1 and 2. (*A*) Product 1 and RNA. (*B*) Product 2. Numbers in parentheses indicate theoretical mass of protonated products; outside parentheses, observed mass.

(Fig. 2) is consistent with higher products formed exclusively from initially produced 2'-aminoacyl-RNA.

Catalysis on GUGGC/GCCU is seen against a lower background of spontaneous reaction with PheAMP. Under certain



Fig. 5. Higher products are RNA peptides. Aminoacylation reaction conditions: 10 μ M GUGGC, 20 μ M GCCU, 100 mM KCl, 5 mM MgCl₂, 100 mM Hepes pH 7.0, 1 mM PheAMP; incubation at 4 °C for 30 min. LAP, leucine aminopeptidase, 23 ng/ μ L.

conditions (increased pH and/or PheAMP concentration), a dim band of RNA-Phe can be visualized even in reactions that do not contain ribozyme GUGGC. Quantitation of such reactions suggests that GUGGC accelerates formation of RNA-Phe by 25-fold at pH 7.5.

Discussion

These data strongly and collectively support the active site model first suggested for the C3 ribozyme (6): RNA 2'(3') aminoacylation requires only an unusually simple center composed of three nucleotides.

Simple metal-catalyzed RNA activity is already known. Every ribophosphodiester is subject to hydrolysis catalyzed by divalent cations, and this can be enhanced by small RNA structures that bind metals. The small RNA GAAACp/UUU is cleaved between G and A in the presence of Mn^{2+} or Cd^{2+} (15). Further, newly selected RNAs routinely contain nucleotides that can be deleted without abolishing function (16).

Nonetheless, GUGGC/GCCU is unique: It binds a complex heteroatomic substrate and facilitates group transfer from a multipart biochemical. This implies a more complex catalytic interface than in metal-catalyzed hydrolysis employing already poised 2'-hydroxyl and ribonucleotide phosphate. In addition, reactions presented here do not depend on ions infrequent in biological systems, and, in fact, show little ion dependence. Thus, it is more surprising that aminoacyl transfer is accelerated by two normal Us and a G nucleotide, poised at a helix terminus.

Further, these particular reactions are central to metabolism, resembling the substrate and product of biological aminoacyltRNA synthesis. *Trans*aminoacylation in this work is performed more simply than elsewhere (9, 12, 17). In addition, observation of RNA-peptide products provides the simplest polypeptide synthesis from aminoacyl adenylate (12, 18, 19), or from any other substrate, in the absence of protein catalysts. Essential intermediates in protein biosynthesis therefore arise surprisingly easily in the presence of very short RNAs.

The ultimate importance of these observations may lie partly in the unknown number of other reactions that can be accelerated by comparably small RNAs. This is because for each such minuscule RNA reaction, there is a prima facie case that it would become accessible even after the most primitive ribonucleotide polymerization.

To see this, consider that, to pick every possible RNA pentamer sequence from arbitrary pentamers (with probability 0.9975), one needs only accumulate 4.1×10^{-18} gm of RNA. To possess every tetramer (with probability 0.9975) from a pool of arbitrary tetramers, one would need 3.4×10^{-18} gm RNA. In a real polymerization, one would have a distribution of lengths; nonetheless, with only attograms of total RNA of distributed short lengths from some geochemical source, one would have not only our ribozyme, but *every* activity of comparable size.

As an illustration, the ribozymic complexes characterized here demonstrate that aminoacyl-RNA and peptidyl-RNAs could have appeared in the presence of ≥ 9 nucleotides of polymeric RNA, with six of these free to vary to other base pairs. We have previously estimated that a population containing about 1 ng of arbitrary-sequence RNA would be required before useful ribozymes and other active RNA structures would probably occur among this population (20). This follows the so-called axiom of origin (21), which estimates that the RNA world would begin when the amount of RNA exceeds the threshold for occurrence of ribozymes. The finding of nine-nucleotide active centers reduces the threshold for ribozyme activity about 7 orders of magnitude, to a level much more easily breached by undirected geochemical syntheses, or by RNA-catalyzed RNA synthesis itself (22–24).

The most intriguing possibility raised by these results is that an RNA reaction center for phosphoester transfer may exist somewhere near this size. This would make the polymerase/replicase needed to initiate Darwinian evolution of RNAs, the founding event of the RNA world, much more likely. On one hand, with this few ribonucleotides to dispose in space, there may not be other similar nucleotide structures that are both stable and capable of catalysis. On the other hand, for obvious reasons, it will be extraordinarily important to look for other tiny RNA active centers, now knowing they can exist.

Methods

RNA and Aminoacyl Substrate Synthesis. RNA was synthesized by Dharmacon. GUGGC = 5'-GUGGC-3'; GCCU = 5'P-GCCU-3'; 5'OH-GCCU = 5'-GCCU-3';GCCU2'dU = 5'-GCC-2'-dU; GCC = 5'-GCC-3'; dGdCdCrU = 5'-dGdCdCU-3'.

RNA GCC3'dU was prepared by first synthesizing 5'-O-(4,4'-Dimethoxytrityl)3'-deoxyuridine as follows: 3'-deoxyuridine (MP Biomedicals; 991 mg, 0.434 mmol) was dissolved in 5 mL anhydrous pyridine and pyridine was then removed under vacuum while stirring. Solid was then redissolved in 2 mL pyridine. Dimethoxytrityl chloride (170 mg, 0.499 mmol) was dissolved in 12 mL pyridine and slowly added to 3'-deoxyuridine solution. Solution was stirred at room temperature for 4 h. All solutions were sequestered from exposure to air throughout.

Reaction was then quenched by addition of 5 mL methanol, and solvent was removed by rotary evaporation. Remaining solvent evaporated overnight in a vacuum chamber. Product was then dissolved in 1 mL acetonitrile and purified through a silica column (acetonitrile elution). Final product fractions (confirmed through TLC, 1:1 hexane:acetonitrile) were pooled and rotary evaporated. Yield was 71%.

Dimethoxytrityl-protected 3'dU was then sent to Dharmacon for immobilization of 3'-dU on glass and synthesis of 5'-GCC-3'-dU.

PheAMP, PheUMP, and MetAMP were synthesized by the method of Berg (25) with modifications and purification as described in ref. 6. Yield was as follows: PheAMP 85%, PheUMP 67%, and MetAMP 36%.

Aminoacylation Assays. For aminoacylation assays, substrate RNA was labeled with ³²P using T4 polynucleotide kinase (New England BioLabs). Kinase conditions were 5 μ M substrate RNA, 1X T4 polynucleotide kinase forward reaction buffer (Invitrogen), 0.625–1.25 μ M (7.5 mCi/mL) γ -³²P-ATP, 0.5 units/ μ L T4 polynucleotide kinase; 37 °C for 20–30 min. Kinase reaction was then passed through a Micro Bio-Spin 6 chromatography column (Bio-Rad) to remove excess γ -³²P-ATP or purified using polyacrylamide gel electrophoresis followed by ethanol precipitation.

Standard aminoacylation conditions were as follows: GUGGC 10 μ M, GCCU 20 μ M, γ^{-32} P-GCCU \leq 0.3 μ M, KCl 100 mM, MgCl_2 5 mM, Hepes pH 7.0 100 mM, PheAMP 1–2 mM; incubation at 4 °C for 30 min. RNA, salts, and buffer were first incubated at 4 °C for 10 min. PheAMP (1–10 μ mol) was dissolved in cold water, filtered, and added to RNA solution. After incubation, the reaction was stopped by addition of xylene cyanol/bromophenol blue dye containing sodium acetate, and either frozen on dry ice or loaded directly into gel for electrophoresis. Conditions for product isolation experiments for MS used 25 μ M GUGGC, 50 μ M GCCU, and 2.5 mM PheAMP.

Periodate reaction contained 100 μ M 5'OH-GCCU, \leq 0.3 μ M $\gamma^{-32}P$ -GCCU, 60 mM KCl, 5 mM MgCl₂, 30 mM NaOAc pH 5.2, 8.2 mM NaIO₄ (total volume 100 μ L). Reaction was incubated on ice in the dark for 2 h. Then glucose was added to 14.9 mM to consume excess periodate and the reaction sat on ice for ~10 min. Reaction was then passed through a Micro Bio-Spin 6 column (Bio-Rad) and lyophilized. Product was redissolved in 15 μ L water and 1 μ L was mixed with GUGGC, KCl, MgCl₂, Hepes, and PheAMP as in standard assay conditions.

Gel electrophoresis was performed using polyacrylamide gel from 12% acrylamide, 6 M urea, 0.1 M sodium acetate pH 5.2. Gels were run from 6–16 h at 4 °C until bromophenol blue had migrated at least 15 cm. Gels were dried, exposed to a phosphorimager screen, and analyzed with a Bio-Rad Molecular Imager FX and QuantityOne software (Bio-Rad).

Purification and Identification of Products 1 and 2. Removal of salts and buffer from products were necessary for MALDI mass spectrometric analysis. Product 1, GUGGC, and GCCU were purified from a standard aminoacylation reaction using a Waters Sep-Pak C18 cartridge. Sep-Pak (Waters) was equilibrated with 1.8 mL methanol followed by 7 mL water. Completed aminoacylation reaction (100 μ L, aqueous) was added to Sep-Pak and washed with 3.5 mL water followed by 200 μ L methanol. Then 600 μ L methanol was added, collected, and concentrated with a SpeedVac (Savant). Gel electrophoresis and analysis confirmed the presence of radiolabeled GCCU and Product 1 in collected fractions. A parallel Sep-Pak purification with a scaled up (10x volume), cold reaction was performed for mass spectrometric analysis.

A Waters Atlantis 5 μ m dC18 column was used to separate Product 2 from other reaction components through HPLC. The gradient was 95% A/5% C for 10 min, ramp to 70% A/30% C for 33 min. The flow was 1 mL/min. Solvent A = 0.1 M NH₄OAc, pH 4.5; solvent C = acetonitrile. Fractions were collected (one per minute) and counted using a scintillation counter to detect elution of products. Product 2 (confirmed by gel electrophoresis) was present in fractions eluted between 32 and 34 min. A parallel HPLC purification was done with a scaled-up cold reaction and fractions 32–34 for both hot and cold reactions were collected, lyophilized, and resuspended in 100 μ L water.

A second HPLC purification was necessary to remove ammonium acetate buffer from Product 2. The gradient was 95% B/5% C for 20 min, ramp to 100% C for 10 min. Solvent B = water. Counts were highest in fractions 7–13 for hot reaction; corresponding fractions were collected for scaledup cold reaction, pooled, and lyophilized.

To measure the rate of ester hydrolysis, Product 1 was first purified through HPLC. Lyophilized product was resuspended in 0.1 M Tris \cdot HCl, pH 8.5, and incubated at 37 °C. Hydrolysis rate was determined from the slope of the natural log of fraction P1 remaining vs. time.

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Mass spectrometry. MS was carried out using a Voyager-DE STR MALDI-TOF mass spectrometer (Applied Biosystems). Dried samples were dissolved in acetonitrile and cospotted with α -cyano-4-hydroxycinnamic acid matrix solution. Analysis was performed in the negative, reflector mode and masses were calibrated using a three-peptide standard. Spectra were averaged over 50 laser shots.

Peptidase reactions. Leucine aminopeptidase (microsomal from porcine kidney) was purchased from Sigma-Aldrich. Peptidase reactions were carried out in the presence of 0.1 M Hepes pH 7.0.

ACKNOWLEDGMENTS. The authors thank Dr. Bruce Eaton and his lab for contribution of equipment and reagents, Dr. Shuji Kato for aiding with mass spectrometry, Dr. Mali Illangasekare for assistance with kinetic experiments, and Dr. Irene Majerfeld and Dr. Teresa Janas for comments on this manuscript. This work was supported by National Institutes of Health research Grant R01 GM 48080 (to M.Y.).

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