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Structure-based design of kinetic stabilizers that ameliorate the transthyretin amyloidoses

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Abstract

Small molecules that bind to normally unoccupied thyroxine (T₄) binding sites within transthyretin (TTR) in the blood stabilize the tetrameric ground state of TTR relative to the dissociative transition state and dramatically slow tetramer dissociation, the rate-limiting step for the process of amyloid fibril formation linked to neurodegeneration and cell death. These so-called TTR kinetic stabilizers have been designed using structure-based principles and one of these has recently been shown to halt the progression of a human TTR amyloid disease in a clinical trial, providing the first pharmacologic evidence that the process of amyloid fibril formation is causative. Structure-based design has now progressed to the point where highly selective, high affinity TTR kinetic stabilizers that lack undesirable off-target activities can be produced with high frequency.

Transthyretin aggregation appears to cause the transthyretin amyloidoses

The TTR amyloid diseases (amyloidoses) appear to be caused by extracellular TTR tetramer dissociation, monomer misfolding and misassembly into a variety of aggregate morphologies [1,2,3•] (Figure 1). The field has hypothesized that oligomers formed during the process of amyloidogenesis lead to cellular toxicity [4,5]. The displacement of tissue by extracellular amyloid is also thought to exacerbate pathology [6]. The TTR amyloidoses include senile systemic amyloidosis (SSA), familial amyloid cardiomyopathy (FAC), familial amyloid polyneuropathy (FAP), and central nervous system selective amyloidoses (CNSA) [3•]. Those amyloidoses impacting the most patients include SSA (10–25% of the elderly population) [7] and FAC (3–4% of African Americans), leading to congestive heart failure [8] (Figure 2). Compelling genetic evidence is supportive of the amyloid hypothesis in the TTR amyloidoses, where the notion is that the process of amyloidogenesis is causatively linked to disease pathology [9•,10].

TTR is a homotetrameric protein, comprising 127-amino-acid, β -sheet-rich subunits [11,12], (Figure 3A). Amyloidogenesis occurs when the thermodynamically linked equilibrium between tetramer, natively folded monomer and partially denatured monomer is shifted toward

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Appendix A. Supplementary data: Supplementary data associated with this article can be found, in the online version, at doi:10.1016/j.sbi.2009.12.009.

the partially denatured TTR monomer that is amyloidogenic (Figure 1) [13]. The energetically weaker dimer–dimer interface of TTR (i, ii/iii, iv) creates the two T₄ binding sites (Figure 3A and B) that are bisected by the crystallographic two-fold axis (C₂) [14•]. In humans, the vast majority (>99.5%) of the TTR T₄ binding sites in blood are unoccupied and available for small molecule binding [3•]. Occupancy of these sites by small molecule kinetic stabilizers has recently been shown to halt neurodegeneration in FAP patients (foldrx.com). TTR kinetic stabilizers differentially stabilize the tetramer over the dissociative transition state by selectively binding to the tetrameric state. This binding increases the free energy of activation for dissociation, dramatically slowing tetramer dissociation, which is rate limiting for TTR fibril formation (Figure 3C). This review focuses on recent developments, and outlines the current state of structure-based design of TTR ligands that kinetically stabilize TTR and prevent the process of amyloid fibril formation.

Transthyretin's thyroxine binding sites

Each T₄ binding site is characterized by a series of subsites [15•] (Figure 3D), an outer binding subsite, an inner binding subsite, and an intervening interface that are all composed of pairs of symmetric hydrophobic depressions referred to as halogen binding pockets (HBPs), wherein the iodine atoms of T₄ reside (Figure 3B) [16]. HBPs 1 and 1' in the outer binding subsite comprise the side chains of Ala108/108', Thr106/106', Met13/13' and Lys15/15' with the pocket lined by methyl and methylene groups of Lys15/15', Ala108/108' and Thr106/106', whereas HBPs 2 and 2', are made up of the side chains from Leu110/110', Ala109/109', Lys15/15' and Leu17/17'. Finally, HBPs 3 and 3' are composed of the methyl and methylene groups of Ser117/117', Thr119/119', Ala109/109' and Leu110/110' (Figure 3B) [16]. T₄ and most ligands bind to TTR with negative cooperativity, apparently resulting from conformational changes within the tetramer upon binding to the first T₄ site [3•], although this has yet to be definitively confirmed crystallographically. Comparative studies of TTR from human and rat in complex with T₄ suggest that conformational changes and hydrogen bonding to Lys15/15' in the outer subsite and Ser117/117' in the inner subsite affect both binding orientation and ligand affinity [17].

Small molecule TTR ligands and the basis for kinetic stabilization

Over one thousand aromatic small molecules exhibiting structural complementarity to the T₄ binding sites within TTR have been synthesized [18–25,26•,27•,28•,29,30] and are typically composed of two aromatic rings occupying the inner and outer T₄ binding subsites (Figure 3D). The aromatic rings composing TTR kinetic stabilizers can either be linked directly (e.g. biphenyls) [19,20] or tethered through short hydrophobic linkers [26•,27•] (Figure 3D). Most of the kinetic stabilizers were conceived of by structure-based drug design principles, taking advantage of > 100 high resolution TTR•(small molecule)₂ crystal structures [11,16,19,21,²², 26•,27•,28•,31–54] (PDB accession codes and compounds are listed in Appendix A, Table S1).

Ligands that bind to the T₄ sites do so utilizing two dominant forces, the hydrophobic effect and electrostatic interactions. Proper choice of the aryl substitution pattern and hydrophobic substituents enable maximal occupancy of the HBPs, thus facilitating ligand interactions with neighboring monomeric subunits through hydrophobic interactions. These bridging hydrophobic interactions help impose kinetic stabilization on the weaker dimer–dimer interface of TTR by stabilizing the tetramer over the dissociative transition state, thereby raising the barrier for tetramer dissociation [9•], Figure 3C. Bridging hydrogen bonds between the ligand substructure in the inner binding subsite and the Thr116/116' or Ser117/117' residues of TTR further differentially stabilize the native non-amyloidogenic tetramer. Lysine and glutamic acid residues (Lys15/15', Glu54/54') on the periphery of the outer T₄ binding subsite electrostatically complement polar substituents on the aryl rings further stabilizing the native

state as also observed in the binding of T₄ whose ammonium groups salt bridge with Glu54/54' [16] (Figure 3B and D). These tetramer stabilizing forces combine to make the dissociation barrier insurmountable under physiological conditions (Figure 3C) [3•,9•,47].

Structure-based design of TTR kinetic stabilizers

Kinetic stabilizers for the treatment of the TTR amyloidoses must be both highly potent and selective. The ligands must not interact with the thyroid hormone receptor (THR), a major concern given the similarity of some kinetic stabilizers with triiodothyronine (T₃, the primary thyroid hormone) and T₄ (the prohormone). Nonsteroidal anti-inflammatory drug (NSAID) activity is also contraindicated for treating TTR amyloidosis patients; thus, TTR kinetic stabilizers should exhibit minimal NSAID activity [55]. While sometimes unpredictable, these activities can usually be minimized by maximizing TTR amyloid inhibition potency, TTR binding affinity and binding selectivity to TTR in human plasma, which we have accomplished using a combination of structure-based drug design and assays to evaluate the above mentioned criteria [3•].

A wide variety of TTR kinetic stabilizers have been identified including naturally derived flavonoid and xanthone derivatives [56–58], as well as synthetic compounds belonging to five families including: bisaryloxime ethers, biphenyls, 1-aryl-4,6-biscarboxydibenzofurans, 2-phenylbenzoxazoles and biphenylamines [18–25,26••,27••,28••,29,30]. Kinetic stabilizers have also been identified through halogenation of NSAIDs, such as salicylic acid [52], diflunisal [21,48,59,60], and analogues of flufenamic [53,60–62]. More recently, additional ligands, such as isatin [63] and β-aminoxypionic acid linked aryl or fluorenyl derivatives have been identified [54,56]. As of 2005, the primary candidates for the most potent and selective kinetic stabilizers were biphenyls, 2-phenylbenzoxazoles and dibenzofurans, since some bisaryloxime ethers have lower than desirable chemical stability [18]. This relative lack of selective, structurally diverse TTR kinetic stabilizers is largely due to the fact that, until recently, no systematic optimization had been attempted on the three structural elements composing a typical TTR amyloidogenesis inhibitor [26••,27••,28••].

Optimization of the ideal aryl substituents and their substitution pattern on aromatic rings **X** and **Z**, as well as the linker substructure joining the rings (Figure 3D), revealed numerous high affinity solutions for occupancy of TTR's thyroxine binding sites [26••,27••,28••]. On the basis of these new SAR data, it is now possible to make accurate predictions of which structures will exhibit TTR kinetic stabilizer potency, TTR plasma binding selectivity and the desired binding orientation [64••]. Detailed herein is a summary of the emerging trends from this structure-based, TTR kinetic stabilizer design, with an exhaustive tabulated list of recently synthesized TTR kinetic stabilizers, complete with so-called efficacy scores (Appendix A, equation S2), integrating TTR amyloidogenesis inhibitor potency and TTR binding selectivity in plasma (Appendix A, Tables S3, S5 and S7).

Optimal aryl-X ring substructures

To optimize the aryl-**X** ring, a library of 2-arylbenzoxazoles was synthesized in which the benzoxazole ring was purposefully unmodified and the 2-aryl ring was systematically substituted (Figure 4A). The library members were evaluated for TTR amyloid inhibition potency, plasma TTR binding selectivity, as well as undesired COX-1 and THR binding activity (Appendix A, Table S4). Substituents at the 3 and 4 positions and, especially, the 3, 4 and 5 positions afforded the most potent and selective TTR amyloidogenesis inhibitors (Appendix A, Table S3). The most potent inhibitors had a 4-hydroxyl substituent and hydrophobic substituents at the 3 and 5 positions [27••,64••,65]. The results showed that hydrogen and fluorine substituents at the 3 and 5 positions tended to afford poorer TTR kinetic

stabilizers than their chloro, bromo, iodo, and methyl counterparts, probably due to less efficient interactions with the HBPs (Figure 4A). Halogens are known to contribute more to ligand binding affinity than would be expected on the basis of their hydrophobicity alone. This enhanced binding is probably due to formation of a charge-transfer complex between the halogen and a neighboring oxygen atom [65,66]. 3,5-substituents are generally superior to 3-substituents only. Substitution at both the 3 and the 5 position enables simultaneous interactions with HBPs on adjacent TTR subunits, whereas mono-*meta*-substitution enables occupancy of only one HBP on one subunit, less efficiently stabilizing the weaker dimer-dimer interface [27••]. Recent studies concerning the iodination of salicylic acid show that 3,5-diiodo salicylic acid binds in the inner T₄ binding subsite, interacting with HBP 3 and 3' as would be predicted by these results. However, these data demonstrate that a second ring is needed to occupy the outer T₄ binding subsite for high binding affinity, also consistent with our observations [52].

The electrostatic interactions made by the 4-hydroxyl group also appear to be important in maximally stabilizing the tetrameric, non-amyloidogenic state of TTR. In the case of the 2-(3,5-dimethyl-hydroxyphenyl)-benzoxazole, the phenol forms hydrogen bonds that bridge Ser117/117' hydroxyls deep in the inner binding subsite (Figure 4A, middle panel), mimicking the structural water oxygen placement seen in the apo TTR structure 2QGB or in 2QGE where the ligand is devoid of a 4-hydroxyl group [27••] (Figure 4A, left panel). Electron-donating 3,5-dimethyl substituents result in a higher 4-hydroxyl pK_a (calculated to be 9.21), enabling it to be protonated, thus favoring hydrogen bonding with the Ser117/117' hydroxyls in the so-called 'forward' binding mode. By contrast, in the isostructural 2-(3,5-dibromo-4-hydroxyphenyl)-benzoxazole, the substituted ring is placed in the outer binding subsite in 'reverse' binding mode with the putative phenolate substituent engaged in an electrostatic interaction with the ε-NH₃⁺ of Lys15/15' (Figure 4A, right panel), apparently owing to the lower pK_a of the phenol (calculated to be 5.35) [27••]. Thus, the 3,5-X₂-4-hydroxyphenyl ring binding orientation in the context of 2-arylbenzoxazoles is predicted by the pK_a of the phenol. These amyloidogenesis inhibitors display little, if any, inhibition of COX-1 activity, and with the exception of the thyroxine-like 3,5-I₂-4-OH substituted 2-aryl ring, the majority of the most potent TTR amyloidogenesis inhibitors do not significantly interact with the THR either (Appendix A, Table S4).

Optimal linker substructures

We next evaluated possible substructures to link the aryls occupying the inner and outer TTR T₄ binding subsites, including CH=CH, -CH₂CH₂-, polar 1-3 atom linkers and fused five-membered ring heteroaromatic linkers. One aromatic ring is always unsubstituted in analogues made for this purpose, whilst the other ring comprises a 3,5-X₂ or a 3,5-X₂-4-hydroxyphenyl substructure, where X is Br or Me (Figure 4B and Appendix A, Table S5). Analysis of the numerous TTR•(ligand)₂ structures in the Protein Data Bank (PDB) reveals that the two aryl rings are typically bound out-of-plane with respect to each other, thus optimal linkers will probably have to permit this [18-22,26••,27••,28••,47]. Co-consideration of TTR amyloid inhibition potency and plasma TTR binding selectivity revealed that the direct linkage of two aryl rings (i.e. biphenyls or 2-arylbenzoxazole), or linkage through a nonpolar substructure such as an *E*-olefin or a -CH₂CH₂- bridge, afforded the most potent and selective TTR amyloidogenesis inhibitors [26••] (Appendix A, Table S5). Irrespective of whether a compound binds in the 'forward' or 'reverse' orientation, the linker is positioned within the hydrophobic vicinity of HBPs 2 and 2' (Figure 4B, left and middle panels). Stilbenes composed of a *trans*-CH=CH linker exhibit excellent structural complementarity to the hydrophobic micro-environments in the vicinity of HBPs 2 and 2' (Figure 4B, left and middle panels), unlike polar amide linkers (Figure 4B, right panel). The CH=CH linker positioned proximal to HBPs 2 and 2' interacts with Leu17/17', Ala108/108', Leu110/110' and Val121/121' side chains as seen in the TTR•(Resveratrol)₂ structure (Figure S9) [22]. Of the 40 compounds tested, only four

displayed appreciable inhibition of COX-1 activity (Appendix A, Table S6). All four were composed of a common 3,5-dimethyl-4-hydroxyphenyl substructure but contained distinct linkers, suggesting the linker contribution to COX-1 activity is negligible.

Optimal aryl-Z ring substructures

Evaluation of aryl-Z ring candidates was carried out using a library composed of a fixed aryl-X substructure (*N*-(3,5-dibromo-4-hydroxyphenyl) as well as a fixed and suboptimal amide linker (Figure 4C) [28••]. Regardless of the diverse range of aryl-Z substituents evaluated, most were capable of increasing inhibitor potency and plasma TTR binding selectivity (Appendix A, Table S7). Independent of the substituent, 2,6-substituents are generally more potent and selective than 2,5-substituents that are more potent and selective than 2-substituents [3,18–25,26••,27••,28••,29,30,47]. This effect is probably due to more efficient hydrophobic interactions with HBPs from adjacent TTR subunits in the case of bis-alkyl and halide substituted aromatics (Figure 4C). Of 56 compounds synthesized, only 5 were observed to displace >10% of T₃ from the THR (Appendix A, Table S8). The random distribution of substitution patterns in thyroid-displacing TTR kinetic stabilizers suggests that this undesirable characteristic could be prevented, perhaps by employing an optimal linker.

Substructure combination strategy is effective

Using SAR data from the aryl-X, aryl-Z and linker-Y substructure optimization studies [26••, 27••,28••], we envisioned that a library rich in highly potent and selective TTR kinetic stabilizers could be generated by simply combining the most highly ranked substructures.

As a demonstration, we utilized the highly ranked 3,5-dibromo-4-hydroxyphenyl aryl-X ring (and isosteric variants replacing the OH functional group with a NH₂), a variety of highly ranked aryl-Z rings and two hydrophobic linkers (*trans*-CH=CH- and -CH₂CH₂-) that previously yielded potent TTR kinetic stabilizers [22,26••] (Figure 5).

Of the 88 3,5-dibromo-4-hydroxyphenyl-based stilbene and dihydrostilbene compounds evaluated, 80 were excellent amyloidogenesis inhibitors allowing < 10% WT-TTR fibril formation at a concentration of 7.2 μM (minimal concentration required to occupy both T₄ sites) and between 15 and 30% fibril formation at a concentration of 3.6 μM. These potent compounds were further evaluated for their binding selectivity for TTR in human plasma. Notably, 37 of these exhibited exceptional selectivity to plasma TTR with >1.5 equivalents bound per tetramer (the maximum being 2). Nonetheless, it was disappointing that more compounds than would have been predicted displaced T₃ at the THR, with 70% of the promising stilbene compounds displacing >20% of T₃. In the case of the stilbenes that displace T₃ from the THR, replacing the CH=CH linker with a dihydrostilbene (CH₂-CH₂) linker yielded a library that was largely free of T₃ THR displacement (24% displacing >20% of T₃) [64••].

The 4-amino group was found to be equivalent to the 4-hydroxyl group in terms of potency and exhibited excellent TTR binding selectivity in plasma. Notably, none of the library members composed of a 4-NH₂ displaced T₃ from the THR, suggesting that further aryl-X optimization may be an effective strategy to reduce THR binding [64••]. While the 3,5-Br₂-4-OH ring prefers to occupy the outer binding subsite in the 'reverse' binding mode [26••, 27••,28••], the 3,5-dibromo-4-aminophenyl ring strongly prefers to occupy the inner binding subsite in the 'forward' binding mode, and identifies a key piece of new data in that it is desirable to have more substructures that bind with subsite selectivity.

The kinetic stabilizer efficacy scores (integrating TTR amyloid inhibition efficacy and TTR binding selectivity data in human plasma) reveal that the substructure combination strategy is

very effective for efficiently producing TTR kinetic stabilizers, in that nearly 70% of the stilbenes examined and 93% of the dihydrostilbenes examined exhibit exceptional TTR kinetic-stabilizer activity [64••].

Conclusions

The systematic optimization of the three substructures comprising a typical TTR kinetic stabilizer has enabled rank ordering of these substructural components and these data now enable us to confidently predict the structures of diverse, potent and highly selective TTR kinetic stabilizers that inhibit TTR amyloidogenesis *in vivo*. From the high-resolution X-ray crystallographic data, we now know why some aryls prefer the inner binding subsite and others prefer the outer binding subsite. We also now appreciate that those that maximize HBP occupancy and electrostatic interactions with one or more TTR subunits are the most potent kinetic stabilizers [26••,27••,28••]. The substructure subsite preferences revealed by X-ray crystallography now inform us regarding which aryl-*X*, aryl-*Y* and linker substructures should be combined to yield potent and selective TTR kinetic stabilizers.

Supplementary Material

Refer to Web version on PubMed Central for supplementary material.

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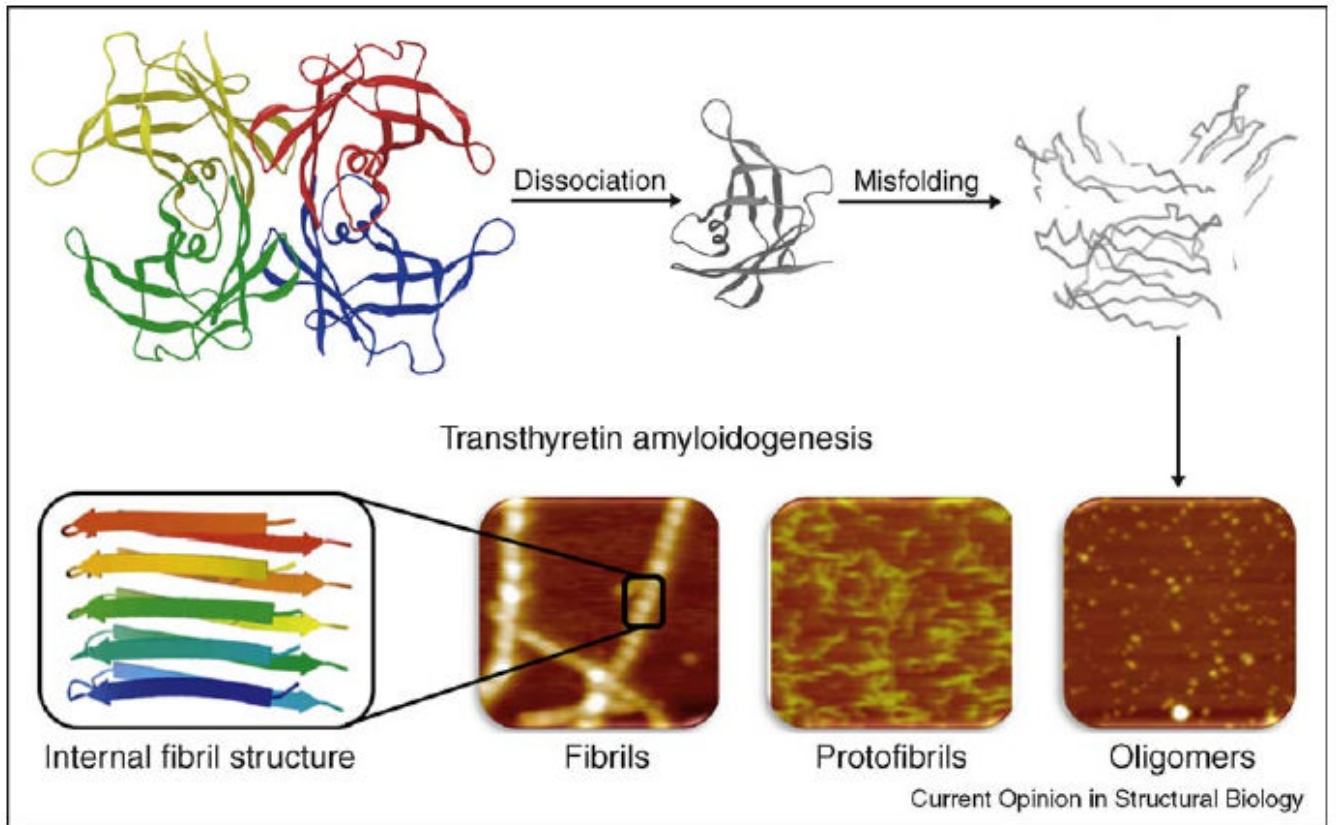


Figure 1.

The transthyretin (TTR) amyloidogenesis cascade. For amyloidogenesis to occur, the TTR tetramer must dissociate into four folded monomers and undergo partial denaturation in order to subsequently misassemble into a spectrum of aggregate structures including cross- β -sheet amyloid fibrils.

Disease	Clinical Classification	Amyloid Constituent	Age of Onset	Affected Population
Senile Systemic Amyloidosis (SSA)	Cardiomyopathy	WT TTR	> 60 y of age	10-25% over 80 y of age
Familial Amyloid Cardiomyopathy (FAC)	Cardiomyopathy	V122I mutant of TTR	> 65 y of age	3-4% of the African American population; clinical penetrance likely high
		T60A mutant of TTR L111M mutant of TTR	> 60 y of age	Geographic clusters
Familial Amyloid Polyneuropathy (FAP)	Peripheral neuropathy	V30M Mutant of TTR	30-80 y of age	European (Portugal, France etc), and Japan with high penetrance (early and late onset); Sweden, low penetrance
Non-V30M Familial Amyloid Polyneuropathy (FAP)	Peripheral neuropathy +/- cardiomyopathy	Mutant TTR	15-70 y of age	Worldwide
Central Nervous System Selective Amyloidoses (CNSA)	CNS amyloidosis	Highly destabilized A25T and D18G mutants of TTR	< 50 y of age	Rare

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Figure 2.
A table summarizing the transthyretin amyloidoses.

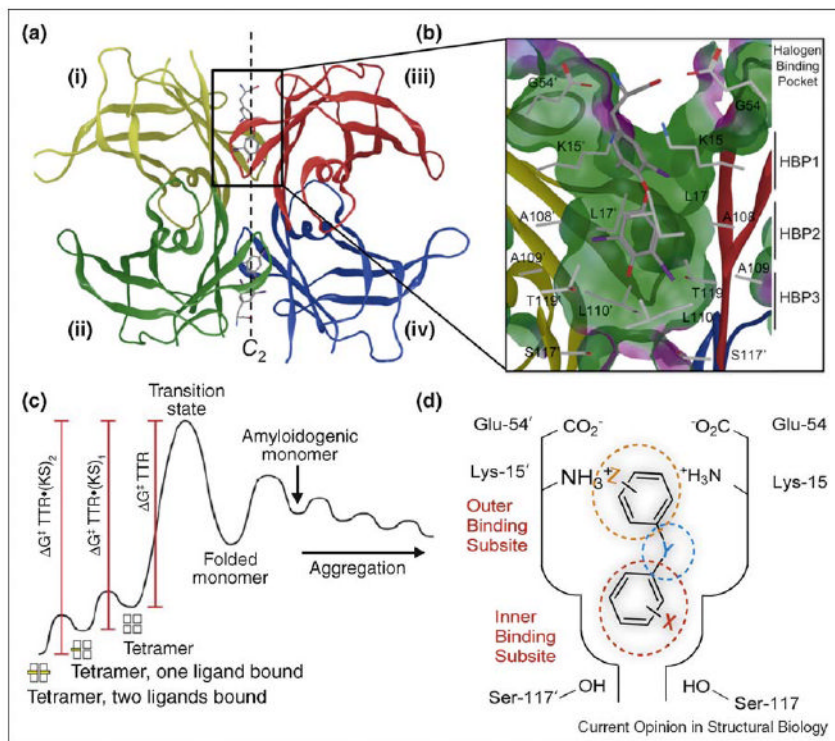


Figure 3. Structure of transthyretin•(T₄)₂ complex and the basis for TTR kinetic stabilization. **(a)** Crystal structure of TTR with T₄ bound to each T₄ binding site (PDB accession code 2ROX [16]). **(b)** Close-up view of one T₄ binding site with T₄ bound. Primed amino acids refer to those comprising symmetry-related halogen binding pockets (HBPs) of TTR monomers. **(c)** Kinetic stabilization of TTR by small molecule binding to the natively folded tetramer stabilizes the ground state and increases the tetramer dissociation barrier, preventing amyloidogenesis and pathology. **(d)** Schematic depiction of one of the two T₄ binding pockets within TTR occupied by a small molecule TTR kinetic stabilizer, where *Y* represents a linker of variable chemical structure (e.g. NH, O, CH=CH, C(O)NH, etc.) joining the two aryl rings, which typically bear a combination of alkyl, carboxyl, halide, trifluoromethyl, or hydroxyl substituents (*X* and *Z*). The inner and outer binding subsites are indicated.

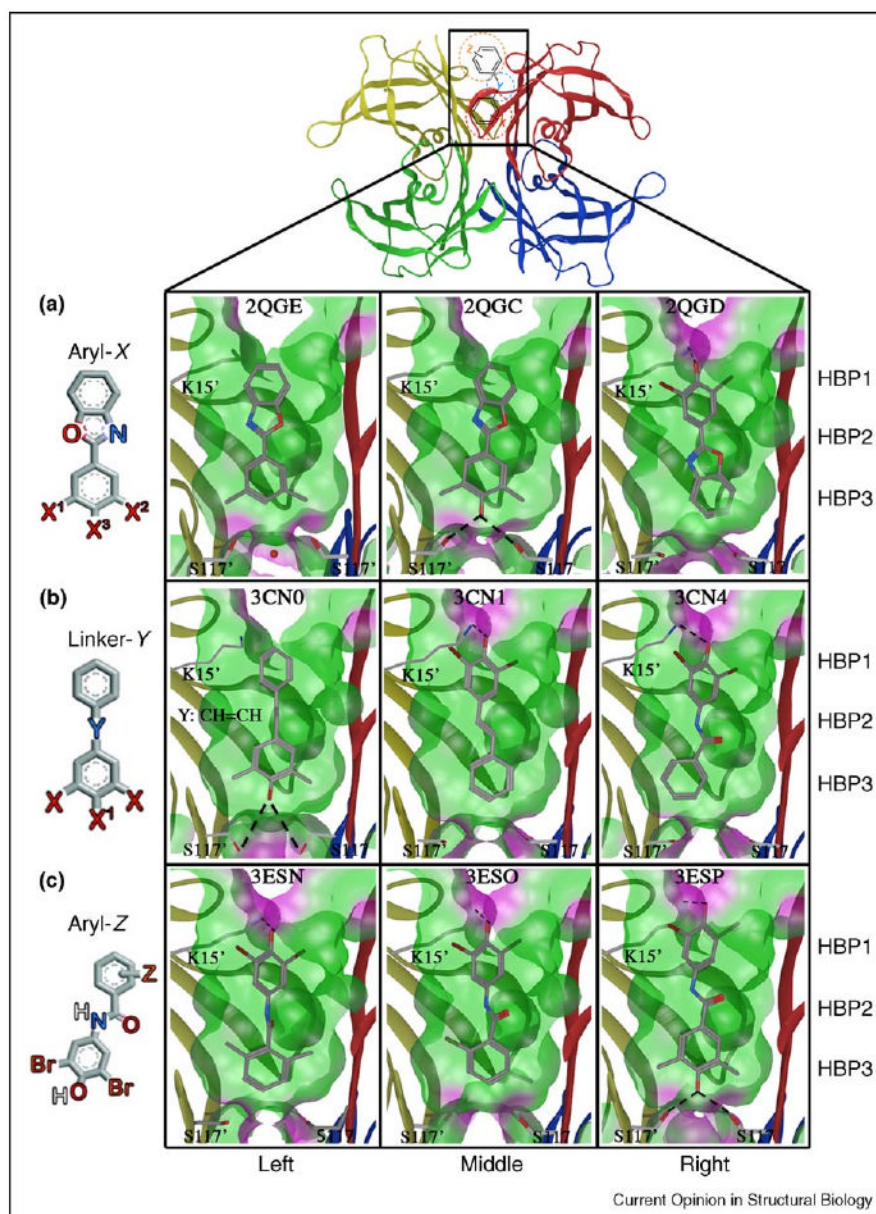


Figure 4. Crystal structures of WT-TTR in complex with kinetic stabilizers from substructure **X**, **Y** and **Z** optimization studies. Ribbon diagram depiction of WT-TTR highlighting one of the two identical T₄ binding sites containing a schematic representation of a typical TTR kinetic stabilizer. (a) Aryl-**X** substructure optimization, PDB accession codes 2QGE, 2QGC and 2QGD, as indicated. (b) Linker-**Y** substructure optimization, PDB accession codes 3CN0, 3CN1 and 3CN4, as indicated. (c) Aryl-**Z** substructure optimization, PDB accession codes 3ESN, 3ESO and 3ESP, as indicated. All structures have ‘Connolly analytical’ surface representation of the pocket (green = hydrophobic, purple = polar), defined as all residues within 8 Å of ligand. Lysine 15' and serines 117' and 117 are shown with corresponding bonds with ligand. Figure is generated using the program MOE (2008.10), Chemical Computing Group, Montreal, Canada.

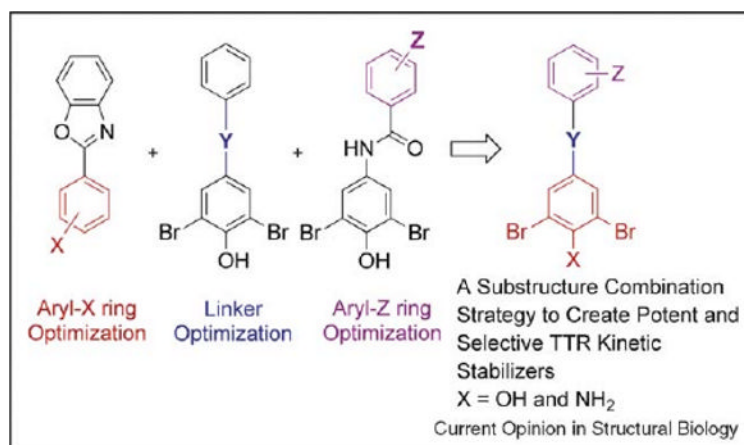


Figure 5. Overview of the substructure combination strategy. Potent and selective TTR kinetic stabilizers were created by combining highly ranked aryl-*X* and aryl-*Z* rings with appropriate linker-*Y*.