

NIH Public Access

Author Manuscript

Chem Commun (Camb). Author manuscript; available in PMC 2011 March 28

Published in final edited form as:

Chem Commun (Camb). 2010 March 28; 46(12): 2016–2018. doi:10.1039/b921664b.

Azlactone-Functionalized Polymers as Reactive Templates for Parallel Polymer Synthesis: Synthesis and Screening of a Small Library of Cationic Polymers in the Context of DNA Delivery

Bin Sun^a, Xianghui Liu^b, Maren E. Buck^a, and David M. Lynn^{a,b}

David M. Lynn: dlynn@engr.wisc.edu

^a Department of Chemistry, University of Wisconsin – Madison, 1101 University Avenue, Madison, WI 53706

^b Department of Chemical and Biological Engineering, University of Wisconsin – Madison, 1415 Engineering Drive, Madison, WI 53706

Abstract

Azlactone-functionalized polymers are used as reactive templates for the synthesis of a library of amine-functionalized polymers of interest in the context of DNA delivery and other applications.

Methods for the parallel synthesis and high-throughput screening of polymeric materials can accelerate dramatically the rates at which new and useful materials are identified in a variety of fundamental and applied contexts.^{1–3} The synthesis of libraries of synthetic cationic polymers, for example, has contributed significantly to the discovery of new agents for the delivery of nucleic acids *in vitro* and *in vivo*.¹ The parallel synthesis/fabrication of spot-based arrays of polymers has also lead to the rapid identification of polymers that promote the adhesion and/or differentiation of stem cells and other cell types.² Other examples of this approach in these contexts and a range of other potential applications have been reported recently.³

In addition to accelerating the rate at which new materials can be identified, approaches based on parallel synthesis and screening can also be used to identify key structural features that govern the behaviours of new classes of materials. Identifying structure/function relationships for libraries of synthetic polymers can be complicated, however, if systematic changes in some structural features (such as side chain or end group structure) cannot be made without holding other structural features (e.g., molecular weights or molecular weight distributions) constant. For example, step growth polymerization processes can be used to introduce systematic changes in side chain structure, but also often result in libraries of polymers with a broad range of molecular weights.^{1a-c,f} Synthetic approaches that provide access to libraries of polymers that are structurally diverse – but that also have uniform molecular weights and molecular weight distributions – would provide opportunities to exploit further the potential of this approach and identify the key features of new materials that are important for function.

Several of the issues described above can be addressed, at least in part, by adopting a synthetic approach based on the functionalization of reactive 'template' polymers.⁴ The most apparent practical advantage of this approach is that the synthesis of multiple polymers using a single,

 $Correspondence \ to: \ David \ M. \ Lynn, \ \texttt{dlynn@engr.wisc.edu}.$

[†]Electronic Supplementary Information (ESI) available: Details of polymer synthesis and characterization; results of gel electrophoresis, cell transfection, and cytotoxicity assays.

'universal' backbone can, in principle, provide access to libraries of structurally diverse polymers having one common molecular weight and one common polydispersity. Several different reactive polymers have been used to explore the feasibility of this approach.^{1h,3c,f,4} The work reported here sought to investigate the use of azlactone-functionalized polymers as reactive templates for parallel polymer synthesis.





Polymers containing azlactone functionality appear well-suited to this approach because azlactones react readily with a variety of nucleophiles through simple ring-opening reactions. ^{5a} In the case of primary amine nucleophiles, these reactions proceed rapidly, in the absence of a catalyst, and without the generation of byproducts. We and others have reported the rational design of side-chain functionalized polymers by the reaction of poly(2-vinyl-4,4-dimethylazlactone) (PVDMA) with different amine-based nucleophiles (Eq. 1).⁵ This investigation sought to investigate the suitability of this approach for the parallel synthesis of a small library of amine-functionalized polymers of potential interest in the context of DNA delivery and other applications for which cationic polymers are useful.

To explore the feasibility of this approach, we targeted the synthesis of a small test library of amine-functionalized polymers using PVDMA (M_n =74,000; PDI=2.65) and 12 different amine-functionalized small molecules. We selected compounds **1–12** possessing both primary and tertiary amine functionality for two reasons: (i) as stated above, primary amines react rapidly with azlactone functionality, and (ii) tertiary amines do not react with azlactone functionality.^{5a} The library of polymers resulting from the reaction of PVDMA with these compounds would thus consist of tertiary-amine containing polymers (hereafter referred to as polymers **P1–P12**). Compounds **1–12** were selected on the basis of differences in steric environments and hydrophobicity as well as the resulting distance of the tertiary amines from the amide/amide linker in the backbone of the resulting polymer (e.g., two versus three carbons in the structures of polymers **P1** and **P2**, etc).



Sun et al.

























P11

NIH-PA Author Manuscript



Exhaustive functionalization of PVDMA in 12 individual reactions resulted in polymers **P1– P12** in near quantitative yield (additional details related to synthesis and characterization are included as Supporting Information). All polymers were soluble in aqueous media and organic solvents such as THF. Characterization of polymers **P1–P10** by gel permeation chromatography (GPC) revealed average molecular weights (M_n) ranging from ~30,000 to ~43,000 with molecular weight distributions that were substantially overlapping (PDI's ranging from 1.4–2.0; see Supporting Information). These differences are small compared to the large differences reported in past studies on the synthesis of polymer libraries using step growth methods, ^{1a–c,f} and could arise from precipitation steps used during isolation of the polymers or from small variations in the interactions of these polyamines with the GPC columns used to characterize molecular weight. The molecular weights of polymers **P11** and **P12** were significantly higher (Mn ~58,000) with higher PDI's (PDI ~3.5). Although the reasons for these larger differences are not clear, these two polymers were also included in all subsequent experiments described below.

The results above demonstrate that azlactone-functionalized polymers can be used as reactive templates for the synthesis of a library of tertiary amine-functionalized polymers. As stated above, these polymers are of potential interest in a range of different applications for which polyamines or cationic polymers are commonly used. We sought to evaluate the potential of this small library to identify new polymers of interest in the context of DNA delivery and cell transfection.⁶ In addition to the potential to accelerate the discovery of new polymers for this application, this work is also of fundamental interest in this context because the 'amide/amide' side chain structural motif present in polymers **P1–P12** has not, to our knowledge, been explored previously for the design of polymer-based DNA delivery agents.

Polymers **P1–P12** were first characterized with respect to their ability to form electrostatic complexes with plasmid DNA using agarose gel electrophoresis retardation assays. Each polymer was used to formulate DNA/polymer complexes (or 'polyplexes') at 10 different DNA/polymer ratios (w/w) ranging from 1:1 to 1:10, as described in past studies.^{1a–c} All 12 polymers formed complexes with DNA at levels sufficient to retard the migration of DNA at ratios as low as 1:1. The full results of these experiments are included as Supp. Info.

We next conducted a series of cell-based screening experiments to characterize the ability of polymers **P1–P12** to promote cell transfection *in vitro*. These initial experiments were performed using COS-7 cells and polyplexes formed using a plasmid DNA construct (pCMV-Luc) encoding firefly luciferase to permit quantitative characterization of transgene expression using a bioluminescence-based assay.^{1a–c}Fig. 1 shows the results (expressed as relative light units per mg of protein) of transfection experiments using polyplexes formed using DNA and polymers **P1–P12** or linear or branched poly(ethyleneimine) (LPEI or BPEI; two well-established gene delivery polymers⁶ used here as positive controls). We note here that initial rounds of screening were performed in parallel to evaluate each polymer over a range of different DNA/polymer ratios (e.g., from 1:1 to 1:10). However, Fig. 1 shows only those results for DNA/polymer ratios identified to yield the highest levels of transfection for each polymer (i.e., the 'optimal' formulations observed for each polymer) after these initial rounds of screening. The results of initial screens and characterization of cytotoxicity are included as Supporting Information.

The data in Fig. 1 reveal that polymers **P1–P12** can promote cell transfection, but at levels that vary significantly as a function of polymer structure. Three polymers in this library (**P3**, **P7**, and **P8**) mediated levels of transfection that were on the order of those mediated by BPEI and LPEI. Figs. 1B–D show additional representative fluorescence microscopy images of cells treated with polyplexes formed using these three polymers and plasmid DNA encoding enhanced green fluorescent protein (EGFP).

Fig. 1 also demonstrates that all other polymers mediated levels of transfection that were lower than BPEI, LPEI, or polymers **P3**, **P7**, and **P8**. However, further inspection of these data permits the identification of structural motifs that appear to influence the gene delivery properties of these materials. For example, the best performing polymers in this library all have side chains with tertiary amine groups that are, in general, (i) more sterically hindered (e.g., the isopropyl groups or cyclic structures of **P3**, **P7**, and **P8**), and (ii) positioned two carbons away from the amide/amide linker to the backbones of the polymers. The apparent importance of this latter feature is further suggested by a comparison of the results for polymers **P1** and **P2**, **P4** and **P5**, **P8** and **P9**, which each differ in structure only by the number of carbons between the tertiary amine group and the amide/amide linker. It is difficult to draw quantitative conclusions using the results for these less effective polymers. However, the data in Fig. 1, when combined, suggest that polymers with amine groups two carbons removed from the amide/amide linker promote levels of transfection that are, in general, higher than those with amine groups that are three carbons removed.

Our results demonstrate that the addition of amine-containing functionality to azlactonefunctionalized polymers can be used to design cationic polymers useful for the delivery of DNA to cells. Our initial screens using this relatively small polymer library have also uncovered structural features that appear important with respect to the gene delivery behaviours of this new class of materials. It is important to note, however, that the synthesis and additional screening of larger and more diverse polymer libraries will be required to establish these and other structure/property relationships more clearly. In this context, the azlactone-based approach described here presents a robust platform for the elaboration of larger, more diverse libraries of polymers by selecting potential side chain functionality from (i) a larger pool of

primary amine-functionalized molecules, and/or (ii) a range of other alcohol-or thiol-based nucleophiles (which can also react readily with azlactones).^{5a} This latter approach would also permit comparisons of the properties of polymers bearing the amide/amide structure of polymers **P1–P12** with those having analogous amide/ester or amide/thioester structures, as well as the combinatorial synthesis of copolymers having mixtures of different side chain functionality.^{1h}

Finally, we note that the synthesis of azlactone-functionalized polymers can be performed under living/controlled conditions⁷ that give rise to reactive polymers with specified molecular weights and lower polydispersities (thereby increasing the potential 'structural space' that could be designed into a given polymer library). Initial characterization of analogs of polymers **P3**, **P7**, and **P8** synthesized using lower molecular weight PVDMA (M_n =5,800; PDI=1.11) demonstrated that these polymers do not mediate high levels of transfection (see Supp. Info.), suggesting that molecular weight also plays an important role in governing the ability of these polymers to transfect cells. In addition to the structural influences noted above, differences in the pKa values of the amines in polymers P1–P12 could lead to differences in endosomal escape or the release of DNA that could also lead to differences in transfection. Additional characterization of polyplexes arising from these libraries with respect to other physicochemical properties (e.g., size, zeta potential) known to be important in the context of cell transfection⁶ will provide additional insight into factors that govern the gene delivery behaviours of these new materials. More generally, access to libraries of new polymers with well defined structures should prove useful in a wide range of other fundamental and applied contexts.

Supplementary Material

Refer to Web version on PubMed Central for supplementary material.

Acknowledgments

Support was provided by the Arnold and Mabel Beckman Foundation, the NIH (EB006820), and the Alfred P. Sloan Foundation. M.E.B was funded in part by an NIH Chemistry Biology Interface Training Grant (NIGMS T32 GM008505).

Notes and references

- (a) Lynn DM, Anderson DG, Putnam D, Langer R. J Am Chem Soc 2001;123:8155–8156. [PubMed: 11506588]
 (b) Akinc A, Lynn DM, Anderson DG, Langer R. J Am Chem Soc 2003;125:5316–5323. [PubMed: 12720443]
 (c) Anderson DG, Lynn DM, Langer R. Angew Chem, Int Ed 2003;42:3153–3158.
 (d) Anderson DG, Peng WD, Akinc A, Hossain N, Kohn A, Padera R, Langer R, Sawicki JA. Proc Natl Acad Sci U S A 2004;101:16028–16033. [PubMed: 15520369]
 (e) Chen DJ, Majors BS, Zelikin A, Putnam D. J Controlled Release 2005;103:273–283.
 (f) Gan L, Olson JL, Ragsdale CW, Yu LP. Chem Commun 2008:573–575.
 (g) Barua S, Joshi A, Banerjee A, Matthews D, Sharfstein ST, Cramer SM, Kane RS, Rege K. Mol Pharm 2009;6:86–97. [PubMed: 19102694]
 (h) Wong SY, Sood N, Putnam D. Mol Ther 2009;17:480–490. [PubMed: 19142180]
- 2. (a) Anderson DG, Levenberg S, Langer R. Nat Biotechnol 2004;22:863–866. [PubMed: 15195101]
 (b) Bailey SN, Sabatini DM, Stockwell BR. Proc Natl Acad Sci U S A 2004;101:16144–16149.
 [PubMed: 15534212] (c) Anderson DG, Putnam D, Lavik EB, Mahmood TA, Langer R. Biomaterials 2005;26:4892–4897. [PubMed: 15763269] (d) Tweedie CA, Anderson DG, Langer R, Van Vliet KJ. Adv Mater 2005;17:2599–2604.
- (a) Brocchini S, James K, Tangpasuthadol V, Kohn J. J Am Chem Soc 1997;119:4553–4554. (b) Brocchini S, James K, Tangpasuthadol V, Kohn J. J Biomed Mater Res 1998;42:66–75. [PubMed: 9740008] (c) Pedone E, Li XW, Koseva N, Alpar O, Brocchini S. J Mater Chem 2003;13:2825–2837. (d) Rickerby J, Prabhakar R, Patel A, Knowles J, Brocchini S. J Controlled Release 2005;101:21–34.

(e) Yang Y, Bolikal D, Becker ML, Kohn J, Zeiger DN, Simon CG. Adv Mater 2008;20:2037–2043.(f) Gibson MI, Frohlich E, Klok HA. J Polym Sci, Part A: Polym Chem 2009;47:4332–4345.

- 4. Theato P. J Polym Sci, Part A: Polym Chem 2008;46:6677–6687.
- 5. (a) Heilmann SM, Rasmussen JK, Krepski LR. J Polym Sci, Part A: Polym Chem 2001;39:3655–3677.
 (b) Guichard B, Noel C, Reyx D, Thomas M, Chevalier S, Senet JP. Macromol Chem Phys 1998;199:1657–1674. (c) Zhang JT, Lynn DM. Adv Mater 2007;19:4218–4223. (d) Kinsinger MI, Buck ME, Campos F, Lynn DM, Abbott NL. Langmuir 2008;24:13231–13236. [PubMed: 18991416]
 (e) Messman JM, Lokitz BS, Pickel JM, Kilbey SM. Macromolecules 2009;42:3933–3941.
- 6. (a) Pack DW, Hoffman AS, Pun S, Stayton PS. Nat Rev Drug Discovery 2005;4:581–593. (b) Putnam D. Nat Mater 2006;5:439–451. [PubMed: 16738681]
- 7. Fournier D, Pascual S, Fontaine L. Macromolecules 2004;37:330-335.



Figure 1.

(A) Levels of transgene expression as a function of polymer structure for COS-7 cells treated with polyplexes formed using DNA encoding firefly luciferase. Light units are arbitrary and normalized to total cell protein; experiments were performed in triplicate. Results correspond to experiments using polyplexes formed at DNA/polymer ratios of 1:2 for **P2**, **P5**, **P10**, and **P11**; 1:4 for **P8**; 1:5 for **P1** and **P3**; 1:8 for **P4**, **P6**, **P7**, and **P12**; 1:9 for **P9** chosen on the basis of initial broader screens (see text). Experiments performed in the absence of polymer (DNA only) and using polyplexes formed using LPEI and BPEI as controls are shown for comparison. (B–D) Fluorescence micrographs showing EGFP expression in cells treated with polyplexes formed using DNA encoding EGFP and polymer **P3** (B), **P7** (C), or **P8** (D) at the DNA/polymer ratios used in (A).