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## Annulation of Thioimidates and Vinyl Carbodiimides to Prepare 2-Aminopyrimidines, Competent Nucleophiles for Intramolecular Alkyne Hydroamination. Synthesis of (–)-Crambidine

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Certain members of the crambescidin natural products,<sup>1–5</sup> derived from the marine sponge *Crambe crambe*, have exhibited remarkable biological properties, including anticancer, anti-HIV, antifungal, and Ca<sup>2+</sup> ion channel blocking activities. Many compounds within this class are characterized by a polycyclic guanidine core linked to a hydroxyspermidine moiety by a linear ω-hydroxy fatty acid. Crambidine (**1**; Figure 1) is atypical within this family of alkaloids in that it possesses a fused *pyrimidine* heterocyclic core, while most of its other congeners exist in more highly reduced forms. Several elegant strategies have been reported for the synthesis of crambescidin alkaloids.<sup>6–9</sup> However, only a single reported synthesis of crambidine (**1**) has appeared,<sup>10</sup> involving dihydropyrimidine construction via Biginelli condensation, followed by oxidation. We report herein a synthesis of (–)-crambidine that capitalizes on two key processes, including a [4+2] annulation of thioimidates with vinyl carbodiimides and a hydroamination of alkynes with 2-aminopyrimidine nucleophiles.

Vinylcarbodiimides derived from pyrimidine-diones have been shown to converge with *O*-methyl imidates to furnish 2-aminopyrimidines.<sup>11</sup> However, when this annulation process was applied to vinylcarbodiimides and *O*-imidates more appropriate to the synthesis of **1**, the reaction was inefficient. For example, heating a mixture of *O*-methyl imidate **2** (Scheme 1A) with benzylvinylcarbodiimide **4** at elevated temperature in C<sub>6</sub>D<sub>6</sub> for 24 h provided only 24% conversion to the 2-aminopyrimidine **5**. By contrast, when the corresponding thiomethyl imidate **3** was exposed to carbodiimide **4** under otherwise identical conditions, conversion to the pyrimidine **5** was accomplished with marked improvement (59%). Addition of the thiol scavenger AgOTf further enhanced the efficiency of the thioimide annulation, allowing for the preparation of a variety of bicyclic 2-aminopyrimidines (Scheme 1B, **8–10**).

Extension of this [4+2] annulation reaction to access a complex substrate more suited to the synthesis of **1** commenced with the synthesis of a fully elaborated vinyl carbodiimide **14** (Scheme 2). Silyl ether **11**, obtained in enantiopure form according to the procedure of Campagne and coworkers,<sup>12</sup> was acylated with allyl chloroformate to afford the propargylic ester **12**. This intermediate was subjected to conjugate addition with azide to afford a separable *E/Z* mixture (2:1) of the corresponding β-azidoacrylate **13**. Reduction of the azide in (*E*)-**13** with PPh<sub>3</sub>, followed by condensation with TBSOCH<sub>2</sub>N=C=O, provided the vinyl carbodiimide **14**. Importantly, advancement of (*Z*)-**13** under analogous conditions also led to the formation of the (*E*)-isomer of **14** in similar yield, wherein stereochemical convergence occurs at the iminophosphorane stage.

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 Supporting Information Available: Experimental details (PDF). This information is available at <http://pubs.acs.org>.

Synthesis of an initial crambidine-relevant thioimide employed methyl thioimide **15**<sup>13</sup> (Scheme 3). This served as a suitable acylation agent for the alkyl lithium species derived from the homoallylic iodide **16**, prepared in a manner similar to that described by Overman.<sup>14</sup> The resulting keto-thioimide **17** was then heated with vinyl carbodiimide **14** to effect [4+2] annulation. Unfortunately, the bicyclic pyrimidine **18** was not observed; instead, the sole isolable product was pyrimidine **19**, presumably a result of C9–C10  $\beta$ -elimination of the heterocycle in **18**.<sup>7</sup>

Given that the fragmentation (**18**→**19**) and concomitant destruction of the C10 stereoconfiguration could not be avoided, attention turned to thioimide **26** (Scheme 4) as an alternate annulation substrate in which the internal alkyne would serve as a less acidic ketone surrogate. Thus, *Z*-selective Wittig olefination of (*S*)-2-(*t*-butylsilyloxy)butyraldehyde (**20**) with the phosphonium ylide derived from 4-(triphenylphosphonium)but-1-yne bromide (**21**) afforded enyne **22**. Following iodination of the terminal alkyne, the alkynyl iodide **23** was subjected to a sequence involving: (1) Cu-mediated coupling with pyrrolidinone **24** to afford internal alkyne **25**; (2) a two step conversion of the lactam **25** to the thioimide **26** via carbonyl thionation and *S*-alkylation; (3) [4+2] annulation with vinylcarbodiimide **14** to furnish bicyclic pyrimidine **27**; and (4) chemoselective *N*-deprotection to provide the free 2-aminopyrimidine **28**, a substrate poised for intramolecular alkyne hydroamination.

Transition metal catalyzed hydroamination of alkynes<sup>15, 16</sup> is a powerful reaction in synthesis; <sup>17–19</sup> however, the paucity of guanidine or 2-aminopyrimidine nucleophiles engaging in this reaction is notable. After extensive experimentation, this transformation was validated by treatment of alkyne **28** (Scheme 4) with 10 mol% AuCl<sub>3</sub><sup>20</sup> at 40 °C, leading to efficient production of the tricyclic pyrimidine **29** as a single isomer (78%).

Subsequent spiroaminal formation at C8 in enamine **29** was conducted under carefully controlled acidic conditions, being mindful of the possibility of undesired C10–*N* bond rupture via potential C8-iminium reactivity. This liability was precluded by treatment with TsOH, effecting TBS removal and spirocyclization to provide the tetracyclic pyrimidinium **30** (77%).<sup>21</sup> The final stages of the synthesis involved conversion of the allyl ester **30** to its Cs-carboxylate. This nucleophile, obtained from carboxylic acid **31**, was amenable to selective alkylation with iodide **32**.<sup>13</sup> The resulting ester **33** was then subjected to *t*-butylcarbamate removal to afford (–)-crambidine (**1**).

A convergent synthesis of crambidine has been described, showcasing a [4+2] thioimide-vinyl carbodiimide annulation and an intramolecular alkyne-guanidine hydroamination. This strategy should not only prove useful for preparing other members of the crambescidins, but also provide an attractive means with which to access complex *N*-heterocycles in general.

## Supplementary Material

Refer to Web version on PubMed Central for supplementary material.

## Acknowledgments

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21. This and subsequent aminopyrimidine intermediates were purified by RP-HPLC (water/MeCN/TFA).

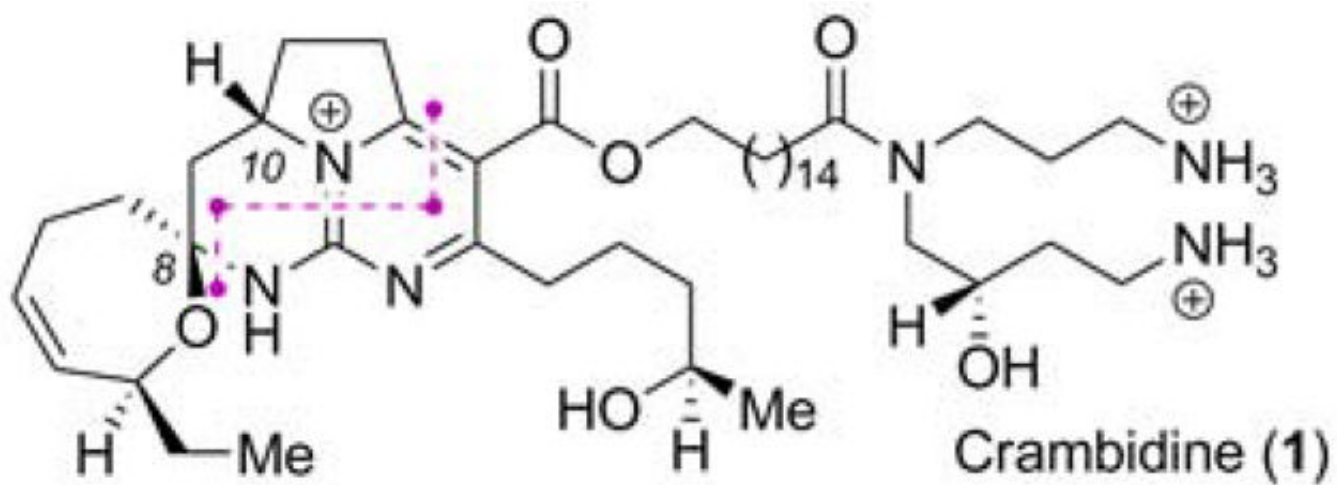
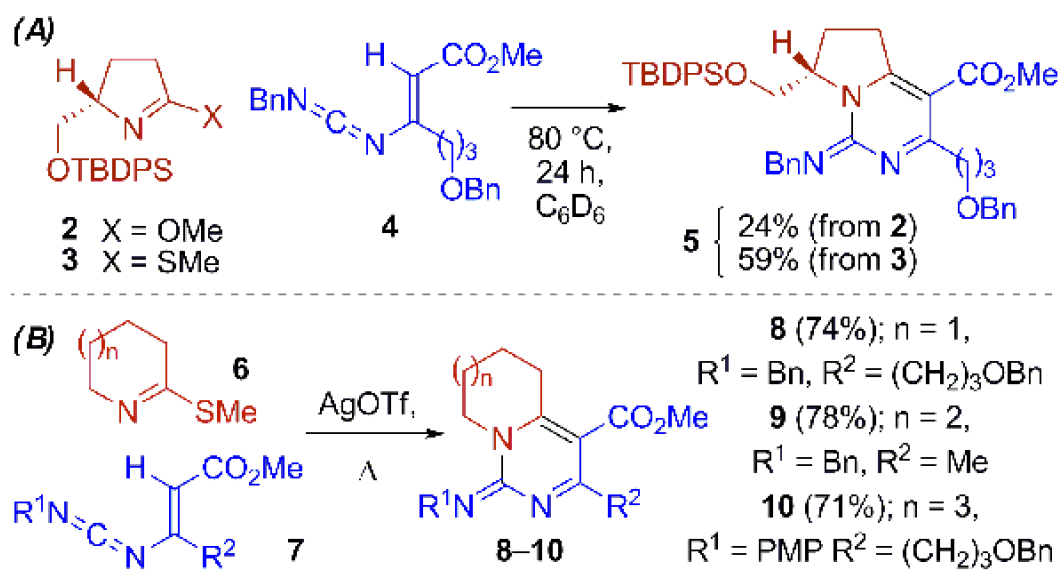
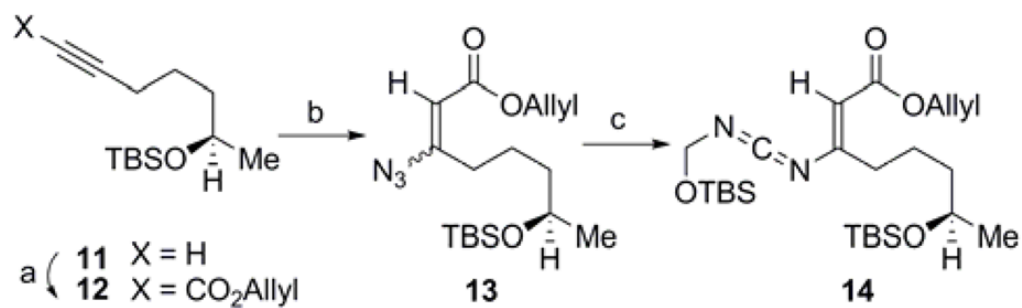


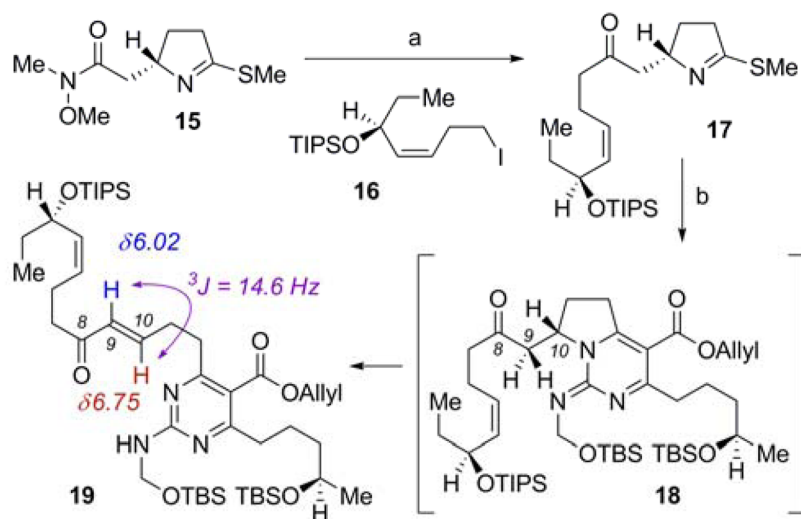
Figure 1.



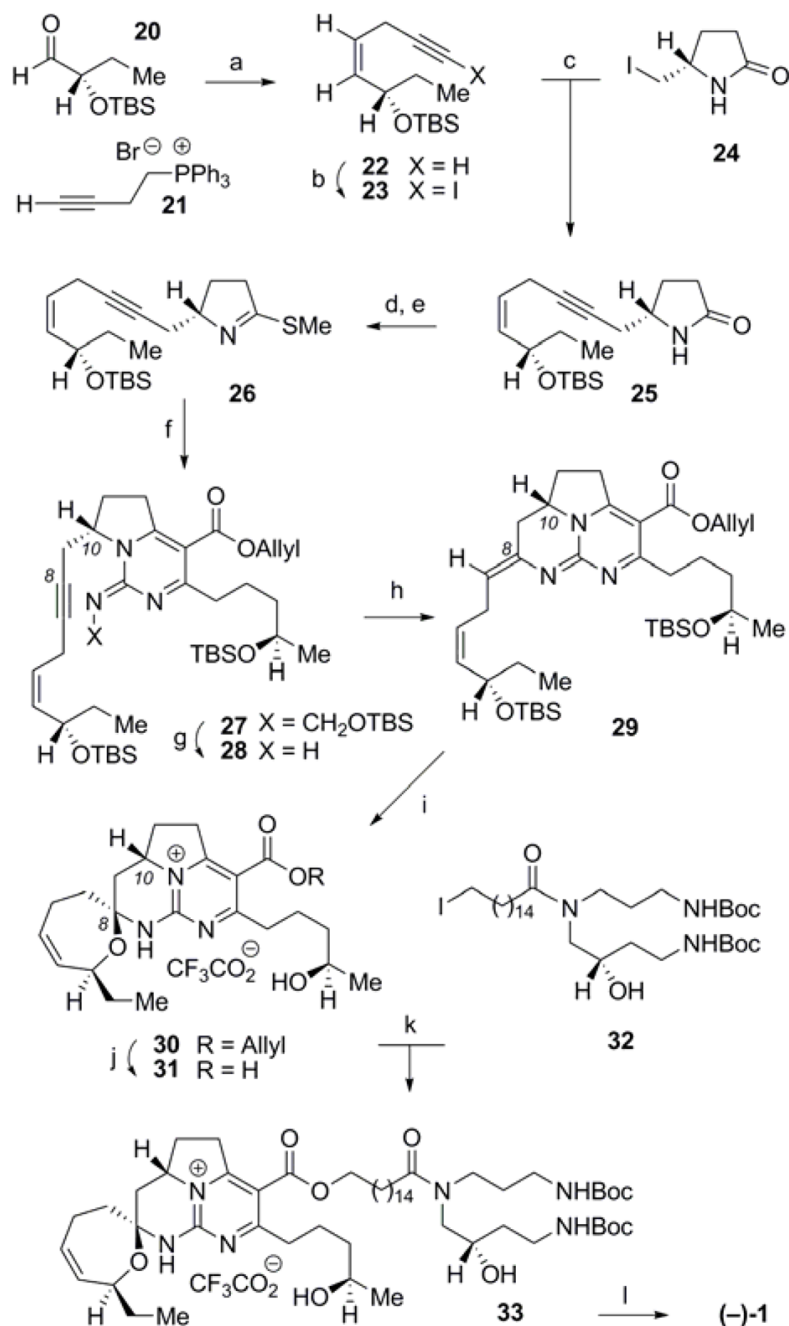
Scheme 1.

**Scheme 2a.**

<sup>a</sup>Reagents and conditions: (a) *n*-BuLi, THF,  $-78$  °C; AllylOCOC1, 90%; (b)  $(\text{Me}_2\text{N})_2\text{C}=\text{NH}_2\text{N}_3$ ,  $\text{CHCl}_3$ ,  $23$  °C, 69% (2:1, *E*:*Z*); (c)  $\text{PPh}_3$ ,  $\text{CH}_2\text{Cl}_2$ ,  $23$  °C; add  $\text{TBSOCH}_2\text{N}=\text{C}=\text{O}$ , PhH,  $80$  °C, 49% from (*E*)-**13**, 43% from (*Z*)-**13**.

**Scheme 3a.**

<sup>a</sup>Reagents and conditions: (a) **16**, *t*-BuLi, Et<sub>2</sub>O, Hexanes, -78 °C; add **15**, -78 → 23 °C, 67%;  
 (b) **14** (2 equiv), (CH<sub>2</sub>Cl)<sub>2</sub>, 60 °C, 70%.

**Scheme 4a.**

<sup>a</sup>Reagents and conditions: (a) **21**, *n*-BuLi, THF, -78  $\rightarrow$  0  $^{\circ}\text{C}$ ; add **20**, -78  $\rightarrow$  23  $^{\circ}\text{C}$ , 85%; (b) NIS, AgNO<sub>3</sub>, Me<sub>2</sub>CO, 23  $^{\circ}\text{C}$ , 91%; (c) **24**, Zn, DMF, 0  $^{\circ}\text{C}$ ; CuCN, LiCl, THF, DMF, -40  $\rightarrow$  23  $^{\circ}\text{C}$ ; add **23**, -40  $\rightarrow$  23  $^{\circ}\text{C}$ , 54% (d) Lawesson's Rgt, THF, 0  $^{\circ}\text{C}$ , 94%; (e) MeI, K<sub>2</sub>CO<sub>3</sub>, THF, 23  $^{\circ}\text{C}$ , 95%; (f) **14** (2 equiv), (CH<sub>2</sub>Cl)<sub>2</sub>, 23  $^{\circ}\text{C}$ , 65%; (g) NH<sub>4</sub>F, MeOH, 23  $^{\circ}\text{C}$ , 79%; (h) AuCl<sub>3</sub>, MeCN, 40  $^{\circ}\text{C}$ , 78%; (i) *p*-TsOH·H<sub>2</sub>O, MeCN, 23  $^{\circ}\text{C}$ , 77%; (j) Pd(PPh<sub>3</sub>)<sub>4</sub>, pyrrolidine, MeCN, 23  $^{\circ}\text{C}$ , 81%; (k) Cs<sub>2</sub>CO<sub>3</sub>, DMF, 23  $^{\circ}\text{C}$ , 88%; (l) HCl, Et<sub>2</sub>O, 0  $^{\circ}\text{C}$ , 77%.