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Rapid Assembly of Oligosaccharides: a Highly Convergent Strategy for the Assembly of a Glycosylated Amino Acid Derived from PSGL-1

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Abstract

P-selectin and P-selectin glycoprotein ligand 1 (PSGL-1) are vascular adhesion molecules that play an important role in the recruitment of leukocytes to inflamed tissue by establishing leukocyte– endothelial and leukocyte–platelet interaction. P-selectin binds to the amino-terminus of PSGL-1 through recognition of a sialyl Lewis^x (SLe^x) moiety linked to a properly positioned core-2 *O*-glycan and three tyrosine sulfate residues. We have developed a highly convergent synthesis of the PSGL-1 oligosaccharide linked to threonine based on the use of trichoroacetimidate donors and thioglycosyl acceptors that give products that can immediately be employed in a subsequent glycosylation step without the need for protecting group manipulations. Furthermore, by employing one-pot multi-step glycosylation sequences the number of purification steps could be minimized. The process of oligosaccharide assembly was further streamlined by combining protecting group manipulations and glycosylations as one-pot multi-step synthetic procedure. The resulting PSGL-1 oligosaccharide is properly protected for glycopeptide assembly. It is to be expected that the strategic principles employed for the synthesis of the target compound can be applied for the preparation of other complex oligosaccharides of biological and medical importance.

Introduction

The selectins are a family of three Ca^{2+} dependent membrane-bound glycoproteins that mediate the adhesion of leukocytes and platelets to vascular surfaces.^{1,2} Several studies have demonstrated that they play important roles in inflammation, immune responses, hemeostasis and wound repair.³ Selectins also contribute to a broad spectrum of diseases such as arteriosclerosis, thrombosis, organ-transplant rejection, arthritis, sickle cell anemia and tumor metastasis.4-⁶

Although there are many candidates for selectin ligands, only P-selectin glycoprotein ligand-1 (PSGL-1) has clearly been demonstrated to mediate the adhesion of leukocytes to selectins under flow. P-selectin binds to the amino-terminus of PSGL-1 through recognition of a sialyl Lewis^x (SLe^x) moiety linked to a properly positioned core-2 O-glycan and three tyrosine sulfate residues.7,⁸

Inhibitors of selectins may possess therapeutic properties for the treatment of a number of diseases.⁹ In this respect, a recombinant truncated form of a PSGL-1 immunoglobulin fusion

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Supporting Information **Available:** 1H-NMR spectra and HSQC of all synthesized compounds are being furnished. This material is available free of charge via the internet at<http://pubs.acs.org>.

protein has already demonstrated effectiveness as such an inhibitor.¹⁰⁻¹³ This glycoprotein can, however, only be produced in mammalian cells that are co-transfected with fucosyl- and core-2 GlcNAc transferases, making production of even small amounts of glycoprotein difficult. The Davis laboratory is beginning to address these problems by employing a PSGL-1 mimetic by incorporation of azidohomoalanine and cysteine in PSGL-1 using *E. coli B834* as a Met-auxotrophic expression system and selectively attachment of sialic acid containing oligosaccharides and a sulfated tyrosine mimics by employing the thiol and azide of the protein as chemical tags.¹⁴ Also, it has been shown that conjugation of sialyl Lewis^x and sulfates tyrosine to a polyacrylamide gave a polymer with high affinity of L-selectin.¹⁵

The *N*-terminal glycosulfopeptide of PGSL-1 has been obtained by chemo-enzymatic approaches.14,16,17 In these procedures, a glycosulfopeptide that contains an *N*-acetyl galactosamine linked to a threonine moiety was chemically assembled. Subsequently, glycosyl transferases were employed to assemble the complete oligosaccharide. The problems of this approach include difficulties of preparing sufficient quantities of glycosyltransferases, which often require a eukaryotic cell expression system and the need of expensive sugar nucleotides or the use of a complicated *in-situ* recycling system. Furthermore, the high selectivity of glycosyltransferases also complicates the preparation of analogs that may exhibit more desirable pharmacological properties.

Recent progress in chemical oligosaccharide synthesis is beginning to provide opportunities for the efficient and large-scale synthesis of complex oligosaccharides¹⁸⁻²³ and several laboratories are pursuing the preparation of the oligosaccharide of PSGL-1. $^{24-30}$ In this respect, Kunz and coworkers have reported the chemical synthesis of a properly protected oligosaccharide of PSGL-1, which was attached to threonine for the preparation of a glycopeptide.26 Although synthetic problems such as anomeric selectivity and the acid and base sensitivity of the PSGL-1 glycopeptide were addressed by employing properly protected saccharide building blocks, the synthetic approach suffered from poor regioselectivity in key glycosylations and a need for replacement of protecting groups at an advanced stage of synthesis. It is to be expected that a highly convergent approach for the synthesis of PSGL-1 will make it possible to prepare a wide range of glycopeptides structural analogs for structure activity relationships. Furthermore, it may offer an opportunity to make mimetics that have improved pharmacokinetic properties.

As part of a program to prepare the PGSL-1 analogs with improved properties, we report here a highly efficient and convergent synthesis of a properly protected oligosaccharide of PSGL-1 linked to threonine (**1**) that is appropriately protected for solid-phase glycopeptide synthesis. Key features of the approach include an orchestrated use of thioglycosides and trichoroacetimidates $3\overline{1}$ for oligosaccharide assembly, which minimized protecting group manipulations and made it possible to employ one-pot multi-step glycosylations. The process of oligosaccharide assembly was further streamlined by combining protecting group manipulations and glycosylations as one-pot multi-step synthetic procedure.^{22,23,32-34} It is to be expected that the strategic principles employed for the synthesis of the target compound can be applied for the preparation of other complex oligosaccharides of biological and medical importance.

Results and Discussion

The synthesis of target compound **1** is complicated by the fact that *O*-glycosylated peptides are sensitive to acidic and basic conditions. In addition, sufficient quantities of such a complex glycosylated amino acid for glycosulfopeptide assembly can only be obtained by employing a highly convergent synthetic strategy, which uses properly protected monosaccharide building blocks that can be assembled into the target using a minimal number of synthetic steps. In this

respect, strategies such as chemoselective, orthogonal, two-directional and one-pot multi-step glycosylations^{22,23,32-34} have engendered an increased efficiency of oligosaccharide synthesis by minimizing the number of protecting group manipulations on advanced intermediates. Furthermore, combining protecting group manipulations with glycosylations as a one-pot procedure can further expand the scope of these procedures.³⁵⁻³⁹

It was envisaged that **1** could be prepared by a combination of chemo- and regioselective glycosylations and one-pot multi-step protocols that combine glycosylations and protecting group manipulations. Thus, the core-2 disaccharide linked to a properly protected threonine (Thr) was prepared by a chemoselective glycosylation of trichloroacetimidate **2** with thioglycoside **3** to give a disaccharide, which immediately can be activated with a thiophilic reagent for coupling with threonine acceptor **4** ⁴⁰ (Figure 1). Removal of the benzylidene acetal of the resulting compound will give an acceptor for a regioselective coupling with a properly protected SLe^{x} derivative. It was envisaged the latter compound could be obtained by chemoselective glycosylations and one-pot reactions using compounds 5^{41} 6^{42} and 7 .

Unfortunately, coupling of galactosyl trichloroacetimidate **2a**43 with thioglycosyl acceptor **3a**44 in the presence of TMSOTf led to a complex mixture of products that included the required disaccharide, the corresponding ortho-ester 45 and a thiophenyl galactoside derived from aglycon-transfer⁴⁶ (Scheme 1). It is well known that the use of C -2 benzoyl esters will suppress ortho-ester formation, however, this protecting group is not compatible with glycopeptide synthesis because the rather strong basic conditions required for its removal⁴⁷ will result in β elimination of the *O*-glycopeptide linkage. It has been shown that a 2,5-di-fluoro-benzoyl esters $(dFBz)$ is an efficient neighboring group participant that suppresses ortho-ester formation.^{48,} 49 This protecting group has, however, as an advantage that it can be removed under mild basic conditions without affecting threonine and serine glycosides. Thus, dFBz-protected glycosyl donor **2b**,48,⁴⁹ having a 2,5-di-fluoro-benzoyl (dFBz) ester at C-2 and acetyl esters at C-3, C-4 and C-6, was coupled with thioglycosyl acceptor **3a**50 using TMSOTf as the catalyst.51 Although orthoester formation was suppressed, the aglycon-transfer byproduct was still formed. Recently, it was reported that aglycon transfer of thioglycosyl acceptors can be avoided by employing a 2,6-dimethylthiophenyl glycoside.52 The rationale of this observation is that the bulky 2,6-dimethylthiophenyl hinders reaction with an activated glycosyl donor, thereby reducing aglycon transfer. Indeed, trimethylsilyl triflate (TMSOTf) promoted glycosylation of **2b** with **3b**52 gave the corresponding disaccharide **8** in an excellent yield of 90% as only the β-anomer. Next, the core-2 *O*-glycan **9** was obtained in high yield with exclusively α-selectivity by a diphenylsulfoxide/triflic anhydride mediated glycosylation53 of thioglycoside **8** with threonine derivative **4.**

Having established efficient reaction conditions for the synthesis of **9**, attention was focused on its preparation by a one-pot procedure. Thus, coupling of galactosyl trichloroacetimidate **2b** with the galactosyl acceptor **3b** in presence of TMSOTf followed by activation of the resulting thio-disaccharide **8** by addition of diphenyl sulfoxide and triflic anhydride in presence of DTBMP53 and coupling with threonine **4** gave oligosaccharide **9** in an overall yield of 61%. Finally, glycosyl acceptor **10** was obtained by the removal of the benzylidene acetal of **9** using aqueous acetic acid at 70°C.

The next stage of the synthesis entailed the preparation of properly protected SLe^x glycosyl donor **15** for coupling with glycosyl acceptor **10** to give protected PSGL-1 **16**. Compound **15** was prepared from the readily available saccharide building blocks **5**, **6**, and **7** (Schemes 2 and 3). Thus, fucosyl trichloroacetimidate **5** was coupled with phenyl thioglycosyl acceptor **6** using TMSOTf as the promoter to give the corresponding disaccharide, exclusively as the α-anomer, which was then treated with triethylsilane and trifluoromethanesulfonic acid (TfOH) for regioselective opening of the benzylidene acetal³⁹,54 to provide glycosyl acceptor 11 in an

overall yield of 84% with excellent stereo- and regio-selectivity. The regioselectivity of the latter reaction was confirmed by acetylation of compound 11 and the ${}^{1}H\text{-NMR}$ of the resulting derivative showed a significant down field shift for H-4 (4.94 ppm).

Glycosyl donor **7** could be obtained in facile manner from the known disaccharide **12**55 by a four-step reaction sequence. Thus, hydrogenation of **12** over Pd/C to remove the benzylidene acetal was followed by acetylation of the hydroxyls of the resulting compound **13** to give **14** in a quantitative overall yield. Next, the anomeric trimethylsilylethyl moiety of **14** was cleaved by treatment with trifluoroacetic acid in dichloromethane and the resulting lactol was converted into trichloroacetimidate **7** by reaction with trichloroacetonitrile and DBU in dichloromethane.

Next, a TMSOTf mediated coupling of trichloroactimidate **7** with **11** gave the properly protected SLe^x tetrasaccharide 15 in good yield. Up to this stage of the synthesis, the thiophenyl moiety of **15** has functioned as an effective anomeric-protecting group. However, in the next step it was activated with the thiophilic promoter NIS/TfOH for coupling with **10** to give the hexasaccharide **16** in a yield of 55%. As expected, no glycosylation of the less reactive C-4 hydroxyl of **11** was observed, which was confirmed by a range of two-dimensional NMR experiments. Thus, Heteronuclear Multiple Bond Correlation NMR Spectroscopy (HMBC) of **16** showed a cross peak between H-1 of β-GluNTroc (4.62 ppm) and C-6 of α -GalN₃ (69.3) ppm), confirming that the glycosylation had occurred at the C-6 hydroxyl of **10**. The latter was also supported by Nuclear Overhauser Enhancement Spectroscopy (NOESY), which revealed cross peaks between the H-1 of β-GluNTroc and H-6a and H-6b of the α-GalN₃ moiety. Furthermore, due to effective neighboring group participation of the *N*-Troc group of **15** only the β-glycoside was formed, which was confirmed by a large coupling constant between H-1 and H-2 (10.0 Hz).

Finally, the Troc and the azido moiety of **16** were converted into acetamido functions by reduction with $Zn/CuSO_4$ in a mixture of THF, acetic acid and acetic anhydride⁵⁶ to give the target compound **1**. It is envisaged that the benzyl ester of compound **1** can be removed by performing hydrogenolysis over palladium in a mixture of isopropanol and pyridine.57 The benzyl ethers in the glycan will be removed after glycopeptide assembly using the previously described "low TfOH" method.58,⁵⁹

In conclusion, a properly protected PSGL-1 oligosaccharide linked to threonine has been described that is appropriately protected for glycosulfopeptide assembly. A highly convergent strategy utilizing six strategically protected building blocks, combined with one-pot, chemoselective and regioselective glycosylations was employed to minimize the number of protecting group manipulations and purifications during oligosaccharide assembly. The longest linear sequence entailed only seven chemical steps and gave the target compounds in an excellent overall yield of 17%. Previous attempts to chemically synthesize the PSGL-1 oligosaccharide suffered from extensive replacement of protecting groups at advanced stages of the synthesis and poor regioselectivities in crucial glycosylation steps compromising the poor overall yield of target compound.26 It is to be expected that the strategic principles employed for the synthesis of **1** will be relevant for the synthesis of many other complex oligosaccharides of biological and medical importance.

Experimental Section

2,6-Dimethylphenyl [2-Azido-4,6-*O***-benzylidene-2-deoxy-3-***O***-(3,4,6-tri-***O***-acetyl-2-***O***-(2,5 difluorobenzoyl)-β-D-galactopyranosyl)]-1-thio-α-D-galactopyranoside (8)**

A mixture of galactosyl acceptor **3b** (93 mg, 0.22 mmol), galactosyl trichloroacetaimidate donor 2b (200 mg, 0.34 mmol), and $4\AA$ MS in CH₂Cl₂ (3 ml) was placed under an atmosphere of argon and stirred at room temperature for 1 h. The reaction mixture was then cooled to -78

°C. TMSOTf (0.023 mmol, 0.23_M solution in CH₂Cl₂) was added and the temperature was raised to -15 °C with stirring over a period of 2 h. The progress of the reaction was monitored by TLC and MALDI-ToF MS. The reaction mixture was diluted with CH_2Cl_2 (20 ml), filtered, and washed with sat. aq. NaHCO₃ solution (10 ml), water (10 ml), and brine (10 ml). The organic layer was dried (MgSO4), filtered, and the filtrate was concentrated *in vacuo*. The residue was purified by silica gel column chromatography (Hexanes:EtOAc, 2:1, v:v) to afford compound **8** (170 mg, 90%) as a white foam. Analytical data for **8**: R*^f* = 0.35 (Hexanes:EtOAc, 2:1, v:v); ¹H-NMR (500 MHz, CDCl₃) : δ = 7.52-6.97 (m, 6H, aromatic), 5.49 (dd, 1H, $J_{1'2'}$ = 8.2 Hz, *J*2′,3′ = 10.2 Hz, H-2′), 5.43 (s, 1H, C*H*Ph), 5.38 (bd, 1H, *J* = 3.2 Hz, H-4′), 5.12 (dd, 1H, *J*3′,4′ = 3.1 Hz, *J*2′,3′ = 10.5 Hz, H-3′), 4.96 (d, 1H, *J*1′,2′ = 7.9 Hz, H-1′), 4.18 (m, 2H, H-1, H-4), 4.15-4.05 (m, 3H, H-6a, H-6a′, H-6b′), 3.91-3.85 (m, 2H, H-5′, H-6b), 3.72 (t, 1H, *J*1,2 = *J*2,3 = 9.9 Hz, H-2), 3.46 (dd, 1H, *J*3,4 = 3.1 Hz, *J*2,3 = 9.9 Hz, H-3), 3.17 (bs, 1H, H-5), 2.51 (s, 6H, 2 × C*H*3, SDMP), 2.11 (s, 3H, COC*H*3), 1.98 (s, 3H, COC*H*3), 1.87 (s, 3H, COC*H*3) ppm. 13C from HSQC (125.7 MHz, CDCl3) : δ = 102.3 (C-1′), 101.1 (*CH*Ph), 89.4 (C-1), 80.9 (C-3), 75.3 (C-4), 71.4 (C-5′), 71.2 (C-3′), 70.3 (C-2′), 70.0 (C-5), 69.6 (C-6), 67.3 (C-4′), 62.4 (C-2), 61.8 (C-6′), 22.9 (*CH*3-SDMP), 21.0, 20.9, 20.8 (3 × OAc); HR-MALDI-ToF/MS: *m/ z*: calc. for C₄₀H₄₁F₂N₃O₁₃S [M+Na]⁺: 864.2226; found 864.2231.

*N***-(9-Fluorenylmethyloxycarbonyl)-***O***-[2-Azido-4,6-***O***-benzylidene-2-deoxy-3-***O***-(3,4,6-tri-***O***acetyl-2-***O***-(2,5-difluorobenzoyl)-β-D-galactopyranosyl)-α-D-galactopyranosyl]-L-threonine benzylester (9)**

Method A—A mixture of disaccharide donor **8** (170 mg, 0.20 mmol), Ph₂SO (114 mg, 0.56) mmol), and $4\AA$ MS in CH₂Cl₂ (5 ml) was placed under an atmosphere of argon and stirred at room temperature for 1 h. The reaction mixture was then cooled to -60 °C after the addition of 2,6-di-*tert*-butyl-4-methylpyridine (124 mg, 0.60 mmol). Stirring was continued for 10 min. at the same temperature followed by the addition of $Tf₂O$ (47 µL, 0.28 mmol). Stirring was continued for another 15 min at the same temperature followed by the addition of a solution of threonine acceptor $4(173 \text{ mg}, 0.40 \text{ mmol})$ in CH₂Cl₂ (2 ml). The temperature of the reaction mixture was raised to 0 °C over a period of 1 h. The progress of reaction was monitored by TLC and MALDI-ToF MS. The reaction mixture was diluted with CH_2Cl_2 (30 ml), filtered, and washed with sat. aq. NaHCO₃ solution (15 ml), water (15 ml), and brine (15 ml). The organic layer was dried (MgSO4), filtered, and the filtrate was concentrated *in vacuo*. The residue was purified by silica gel column chromatography (Hexanes:EtOAc, 2:1, v:v) to afford compound **9** (156 mg, 70%) as a white amorphous solid.

Method B—A mixture of galactosyl acceptor **3b** (65 mg, 0.16 mmol), galactosyl trichloroacetaimidate donor $2b(120 \text{ mg}, 0.20 \text{ mmol})$, and 4Å MS in CH₂Cl₂ (2 ml) was placed under an atmosphere of argon and stirred at room temperature for 1 h. The reaction mixture was then cooled to -60 °C. TMSOTf (0.016 mmol, 0.16M solution in CH₂Cl₂) was added and stirring was continued for 1 h at the same temperature. The reaction mixture was then cooled to -78 °C followed by addition of Ph₂SO (88 mg, 0.44 mmol) and 2,6-di-*tert*-butyl-4methylpyridine (112 mg, 0.55 mmol). After stirring for 10 min. at the same temperature, Tf₂O (37 µL, 0.22 mmol) was added followed by increasing the temperature to -60 °C over a period of 15 min. The reaction mixture was again cooled to -78 °C followed by addition of a solution of threonine acceptor $4(100 \text{ mg}, 0.23 \text{ mmol})$ in CH₂Cl₂ (1 ml). The temperature of the reaction mixture was raised to 0° C over a period of 1 h. The progress of reaction was monitored by TLC and MALDI-ToF MS. The reaction mixture was diluted with CH_2Cl_2 (20 ml), filtered, and washed with sat. aq. NaHCO₃ solution (10 ml), water (10 ml), and brine (10 ml). The organic layer was dried (MgSO4), filtered, and the filtrate was concentrated *in vacuo*. The residue was purified by silica gel column chromatography (Hexanes:EtOAc, 2:1, v:v) to afford compound **9** (106 mg, 61%) as a white amorphous solid. Analytical data for **9**: R_f = 0.25 (Hexanes:EtOAc, 2:1, v:v); ¹H-NMR (500 MHz, CDCl₃) : δ = 7.72-6.87 (m, 21H,

aromatic), 5.66 (d, 1H, *J* = 9.4, N*H*Fmoc), 5.51-5.45 (m, 2H, C*H*Ph, H-2′), 5.40 (bd, 1H, H-4′), 5.15 (dd, 1H, *J*3′,4′ = 3.2 Hz, *J*2′,3′ = 10.2 Hz, H-3′), 5.10 (bt, 2H, C*H*2, Bn), 4.88 (d, 1H, *J*1′,2′ $= 7.9$ Hz, H-1'), 4.83 (d, 1H, $J_{1,2} = 3.5$, H-1), 4.46-4.25 (m, 5H, OCHCH₃ threonine, CH₂CH-Fmoc, C*H*COOBn threonine, H-4,), 4.21-4.08 (m, 4H, H-6a′, CH2C*H*-Fmoc, H-6b′, H-6a), 3.96-3.92 (m, 3H, H-6b, H-3, H-5′), 3.69 (dd, 1H, *J*1,2 = 3.5 Hz, *J*2,3 = 10.8 Hz, H-2), 3.56 (bs, 1H, H-5), 2.11 (s, 3H, OC*H*3), 1.97 (s, 3H, COC*H*3), 1.87 (s, 3H, COC*H*3), 1.23 (d, 3H, OCHCH₃ threonine) ppm. ¹³C from HSQC (125.7 MHz, CDCl₃) : δ = 102.3 (C-1'), 100.8 (*CH*Ph), 99.4 (C-1), 76.2 (O*CH*CH3 threonine), 75.9 (C-4), 75.8 (C-3), 71.2 (C-5′), 71.1 (C-3′), 70.3 (C-2′), 69.3 (C-6), 68.0 (*CH2*Ph), 67.6 (*CH2*CH-Fmoc), 67.3 (C-4′), 63.7 (C-5), 61.5 (C-6′), 59.5 (C-2), 58.9 (*CHCOOBn* threonine), 47.4 (*CH₂CH*-Fmoc), 21.0, 20.9, 20.7 (3 \times OAc), 19.0 (OCHCH₃ threonine); HR-MALDI-ToF/MS: m/z : calc. for C₅₈H₅₆F₂N₄O₁₈ [M $+Na$ ⁺: 1157.3455; found 1157.3460 [M+Na]⁺.

*N***-(9-Fluorenylmethyloxycarbonyl)-***O***-[2-azido-2-deoxy-3-***O***-(3,4,6-tri-***O***-acetyl-2-***O***-(2,5 difluorobenzoyl)-β-D-galactopyranosyl)-α-D-galactopyranosyl]-L-threonine benzyl ester (10)**

A solution of compound **9** (80 mg, 0.072 mmol) in 5 ml of 70% aq. acetic acid was heated at 70 °C for 3 h. The progress of the reaction was monitored by TLC and MALDI-ToF MS. The reaction was cooled to rt and concentrated by co-evaporation with toluene *in vacuo*. The residue was purified by silica gel column chromatography (Hexanes:EtOAc, 1:2, v:v) to afford compound **10** (59 mg, 92%) as a white amorphous solid. Analytical data for **10**: R*^f* = 0.25 (Hexanes:EtOAc, 1:2, v:v); ¹H-NMR (500 MHz, CDCl₃) : δ = 7.72-6.94 (m, 16H, aromatic), 5.60 (d, 1H, *J* = 9.5 Hz, N*H*Fmoc), 5.50-5.46 (q, 1H, *J*1′,2′ = 8.6 Hz, *J*3′,2′ = 10.3 Hz, H-2′), 5.40 (bd, 1H, H-4′), 5.16 (q, 1H, *J*3′,4′ = 2.9 Hz, *J*2′,3′ = 10.3 Hz, H-3′), 5.12-5.01 (dd, 2H, *CH2*Ph), 4.81 (d, 1H, *J*1′,2′ = 7.8 Hz, H-1′), 4.76 (d, 1H, *J*1,2 = 3.4 Hz, H-1), 4.41-4.35 (m, 3H, OC*H*CH3 threonine, C*H*COOBn threonine, C*H*H-Fmoc), 4.25-4.21 (m, 1H, C*H*H-Fmoc), 4.16-4.05 (m, 4H, H-6a′, H-6b′, H-4, CH2C*H*-Fmoc), 3.96 (m, 1H, H-5′), 3.91-3.88 (dd, 1H, *J*3,4 = 2.7 Hz, *J*2,3 = 10.7 Hz, H-3), 3.83-3.70 (m, 3H, H-5, H-6a, H-6b), 3.46-3.44 (dd, 1H, *J*1,2 = 3.7 Hz, *J*2,3 = 10.5 Hz, H-2), 2.75 (bs, 1H, C4-O*H*), 2.32 (bs, 1H, C6-O*H*), 2.13 (s, 3H, OC*H*3), 2.00 (s, 3H, COC*H*3), 1.90 (s, 3H, COC*H*3), 1.24 (d, 3H, OCHC*H*3) ppm. 13C from HSQC (125.7 MHz, CDCl3) : δ = 101.8 (C-1′), 99.2 (C-1), 78.1 (C-3), 76.2 (O*CH*CH³ threonine), 71.7 (C-5′), 70.7 (C-3′), 69.8 (C-2′), 69.7 (C-5), 69.3 (C-4), 67.8 (*CH2*Ph), 67.4 (*CH2*CH-Fmoc), 67.2 (C-4′), 62.7 (C-6), 61.6 (C-6′), 59.0 (C-2), 58.7 (*CH*COOBn threonine), 47.4 (CH2*CH*-Fmoc), 20.8, 20.7, 20.6 (3 × OAc), 18.6 (OCH*CH3* threonine); HR-MALDI-ToF/MS: m/z : calc. for C₅₁H₅₂F₂N₄O₁₈ [M+Na]⁺: 1069.3142; found 1069.3140.

Phenyl 3,4-di-*O***-acetyl-2-***O***-benzyl-6-deoxy-5-methyl-α-L-fucopyranosyl-(1→3)-2-(2,2,2 trichloroethoxy)carbonyl amino-6-***O***-benzyl-2-deoxy-1-thio-β-D-glucopyranoside (11)**

A mixture of glycosyl acceptor **6** (50 mg, 0.09 mmol), trichloroacetimidate donor **5** (63 mg, 0.13 mmol), and 4\AA MS in CH₂Cl₂ (1 ml) was placed under an atmosphere of argon and stirred at room temperature for 1 h. The reaction mixture was then cooled to 0° C. TMSOTf (0.018) mmol, 0.18_M solution in CH₂Cl₂) was added and stirring was continued for 30 min at the same temperature. The reaction mixture was then cooled to -78 °C followed by addition of TfOH (23 μ L, 0.26 mmol) and triethylsilane (48 μ L, 0.30 mmol). The reaction mixture was then stirred at -78 °C for 30 min. The progress of the reaction was monitored by TLC and MALDI-ToF MS. The reaction was quenched by the addition of pyridine (25 μL) and MeOH (0.2 ml), diluted with CH_2Cl_2 (20 ml), and washed with sat. aq. NaHCO₃ solution (10 ml), water (10 ml), and brine (10 ml). The organic layer was dried $(MgSO₄)$, filtered, and the filtrate was concentrated *in vacuo*. The residue was purified by silica gel column chromatography (Hexanes:EtOAc, 2:1, v:v) to afford compound **11** (67 mg, 84%) as a white amorphous solid. Analytical data for **11**: R_f = 0.40 (Hexanes:EtOAc, 1:1, v:v); ¹H-NMR (500 MHz, CDCl₃) : δ = 7.50-7.17 (m, 15H, aromatic), 5.46 (d, 1H, *J* = 6.6 Hz, N*H*Troc), 5.30-5.27 (m, 2H, H-3′, H-4′), 5.09 (m, 2H, H-1, H-1′), 4.70-4.57 (m, 6H, 2 × CH₂, Bn, OCH₂CCl₃), 4.42 (m, 1H,

H-5′), 3.89-3.83 (m, 3H, H-2′, H-3, H-6a), 3.78-3.75 (dd, 1H, H-6b), 3.68 (bs, 1H, C4-OH), 3.58-3.55 (m, 2H, H-4, H-5), 3.31 (m, 1H, H-2), 2.13 (s, 3H, COC*H*3), 1.98 (s, 3H, COC*H*3), 1.12 (d, 3H, $J = 6.6$ Hz, CH₃ fucose) ppm. ¹³C from HSQC (125.7 MHz, CDCl₃) : $\delta = 98.6$ (C-1′), 85.9 (C-1), 83.8 (C-3), 78.7 (C-5), 74.1-73.6 (2 × *CH2*Ph, *CH2*Troc), 73.8 (C-2′), 71.2 $(C-3')$, 71.0 $(C-4)$, 70.3 $(C-4')$, 70.2 $(C-6)$, 66.0 $(C-5')$, 55.7 $(C-3)$, 21.0, 20.8 $(2 \times OAC)$, 16.4 (C-6'); HR-MALDI-ToF/MS: m/z : calc. for C₃₉H₄₄Cl₃NO₁₂S [M+Na]⁺: 878.1547; found 878.1543.

2-(Trimethylsilyl)ethyl [methyl 5-(*N*-acetylacetamido)-4,7,8,9-tetra-*O*-acetyl-3,5-dideoxy-_p**glycero-α-D-galacto-2-non-ulopyranosylonate]-(2→3)-***O***-β-D-galactopyranoside (13)**

To a solution of **12** (250 mg, 0.283 mmol) in CH2Cl2:MeOH (30:1, v:v; 15 mL) under an argon atmosphere was added Pd, 10 wt. % on activated carbon, (150 mg) and the mixture stirred for 20 min. at room temperature. The argon was replaced with $H_{2(g)}$, and the reaction stirred for 8 h. The solution was diluted with CH₂Cl₂ (50 mL) and filtered through celite. The solvent was removed by evaporation under reduced pressure and the residue purified by silica gel column chromatography (Toluene:Acetone, 5:2, v:v) to afford compound **13** (220 mg, 98%) as a white amorphous solid. Analytical data for 13 : R_f = 0.48 (Toluene:Acetone, 1:1, v:v); ¹H-NMR (500 MHz, CDCl3) : δ = 5.52 (ddd, 1H, H-4′), 5.34 (dd, 1H, H-8′), 5.15 (dd, 1H, *J* = 8.2 Hz, H-7′), 4.94 (d, 1H, *J* = 10.3 Hz, H-6′), 4.42 (d, 1H, *J* = 7.6 Hz, H-1), 4.33 (dd, 1H, H-9a′), 4.21 (t, 1H, *J* = 10.2 Hz, H-5′), 4.12 (dd, 1H, *J2,3* = 9.4 Hz, *J3,4* = 3.1 Hz, H-3), 4.08 (dd, 1H, H-9b′), 4.02 (m, 1H, OC*H*H), 3.91 (dd, 1H, H-6), 3.92-3.82 (m, 5H, H-6a, H-6b,COOC*H*3), 3.70-3.60 (m, 3H, H-4, H-2, OC*H*H), 3.55 (t, 1H, H-5), 2.86 (dd, 1H, H-3′eq), 2.35, 2.28 (2 × s, 6H, N(COCH₃)₂), 2.10, 2.09, 2.00, 1.97 (4 × s, 12H, 4 × COCH₃), 1.93 (dd, 1H, H-3'ax), 1.13-0.85 (m, 2H, C*H*2SiMe3), 0.01 (s, 9H, Si(C*H*3)3); 13C from HSQC (125.7 MHz, $CDCl₃$) : $\delta = 102.7$ (C-1), 77.2 (C-3), 73.8 (C-5), 70.4 (C-6'), 69.5 (C-2), 68.9 (C-4), 68.7 (C-8′), 67.3 (*CH2*CH2SiMe3), 66.9 (C-7′), 66.8 (C-4′), 62.6 (C-6), 62.2 (C-9′), 56.9 (C-5′), 53.7 (COO*CH3*), 38.6 (C-3′), 28.3, 26.3 (NAc2), 21.4, 21.1, 21.0 (3 × OAc), 18.3 (*CH2*SiMe3); HR-MALDI-ToF/MS: m/z : calc. for C₃₃H₅₃NO₁₉Si [M+Na]⁺: 818.2879; found 818.2880.

2-(Trimethylsilyl)ethyl [methyl 5-(*N***-acetylacetamido)-4,7,8,9-tetra-***O***-acetyl-3,5-dideoxyglycero-α-D-galacto-2-non-ulopyranosylonate]-(2→3)-***O***-(2,4,6-tri-***O***-acetyl-βgalactopyranoside) (14)**

Compound **13** (215 mg, 0.270 mmol) was dissolved in pyridine (10 mL) and acetic anhydride (5 mL) and the reaction stirred for 14 h at room temperature. The solvent was removed by coevaporation with toluene (3×50 mL). Silica gel column chromatography (Hexanes: EtOAc, 1:1, v:v) of the residue afforded compound **14** (246 mg, 99%) as a white solid. Analytical data for **14**: R_f = 0.55 (Hexanes:EtOAc, 1:3, v:v); ¹H-NMR (500 MHz, CDCl₃) : δ = 5.53-5.47 (m, 2H, H-4′, H-8′), 5.14 (dd, 1H, H-7′, *J* = 9.3 Hz, 2.4 Hz), 4.98-4.94 (m, 2H, H-3, H-4), 4.58-4.53 (m, 3H, H-2, H-6′, H-1), 4.28-4.25 (m, 2H, H-5′, H-9a′), 4.08-3.98 (m, 3H, H-6a, H-6b, H-9b ′), 3.97-3.92 (dt, 1H, OC*H*HCH2Si(CH3)3), 3.85 (s, 3H, COOCH3), 3.83 (t, 1H, H-5), 3.59-3.54 (dt, 1H, OCHHCH₂Si(CH₃)₃), 2.63 (dd, 1H, $J = 5.4$ Hz, 12.7 Hz, H-3'eq), 2.32, 2.25 (2 × s, 6H, N(COCH₃)₂), 2.17, 2.15, 2.04, 2.01, 2.00, 1.99, 1.91 (7 \times s, 21H, 7 \times COCH₃), 1.60 (t, 1H, $J = 12.2$ Hz, H-3'ax), 1.01-0.86 (m, 2H, CH₂Si(CH₃)₃), 0.00 (s, 9H, Si(CH₃)₃); ¹³C from HSQC (125.7 MHz, CDCl₃) : δ = 100.8 (C-1), 71.8 (C-2), 70.6 (C-5), 70.4 (C-3), 69.6 (C-6'), 67.9 (C-4), 67.8 (C-4′), 67.6 (*CH2*CH2SiMe3), 67.4 (C-7′), 67.3 (C-8′), 62.7 (C-6), 62.4 (C-9′), 56.5 (C-5′), 53.3 (COO*CH3*), 38.7 (C-3′), 28.4, 27.0 (NAc2), 22.0-20.6 (7 × OAc), 18.4 (CH_2SiMe_3) , 1.3 ($SiMe_3$); HR HR-MALDI-ToF/MS: m/z : calc. for C₃₉H₅₉NO₂₂Si [M +Na]+: 944.3196; found 944.3194.

Methyl 5-(N-acetylacetamido)-4,7,8,9-tetra-O-acetyl-3,5-dideoxy-_p-glycero-α-_p-galacto-2-nonulopyranosylonate-(2→3)-*O***-(2,4,6-tri-***O***-acetyl-β-D-galactopyranosyl) trichloroacetimidate (7)**

TFA (2 mL) was added to a solution of compound 14 $(240 \text{ mg}, 0.260 \text{ mmol})$ in CH₂Cl₂ (10) mL) at 0 ^oC and the reaction stirred for 4 h at the same temperature. The solvent was removed by co-evaporation with toluene (5×20 mL). The residue was purified by silica gel column chromatography (Hexanes:EtOAc, 2:5, v:v). Trichloroacetonitrile (130 μL, 1.26 mmol) and 1,8-diazabicyclo[5.4.0]-undec-7-ene (14 μL, 94.8 μmol) were added to a solution of methyl 5- (*N*-acetylacetamido)-4,7,8,9-tetra-*O*-acetyl-3,5-dideoxy-_D-glycero-α-_D-galacto-2- nonulopyranosylonate- $(2\rightarrow 3)$ -*O*-(2,4,6-tri-*O*-acetyl-β-_D-galactopyranoside) (207 mg, 0.252 mmol), in CH2Cl2 (5 mL). The reaction mixture was stirred for 1 h and then concentrated in *vacuo*. Silica gel column chromatography (Hexanes:EtOAc, 1:2, v:v) of the syrup afforded compound **7** (220 mg, 87% over two steps, 3:2 α:β) as a white foam. Analytical data for **7**: R*f* = 0.30 (Hexanes:EtOAc, 1:2, v:v); 1H-NMR (500 MHz, CDCl3) : δ = 8.65 (s, 1H, NH), 8.61 (s, 1H, NH), 6.48 (d, 1H, *J1,2* = 3.8 Hz, H-1α), 5.93 (d, 1H, *J1,2* = 8.2 Hz, H-1β), 5.56-5.50 (m, 2H, H-4′α, H-8′α), 5.37-5.35 (m, 2H, H-4′β, H-8′β), 5.27-5.24 (m, 2H, H-2α, H-2β), 5.15-5.12 (m, 2H, H-7′α, H-7′β), 5.04 (bd, 1H, H-4β), 4.99 (dd, 1H, *J* = 3.4 Hz, 10.5 Hz, H-3α), 4.77 (dd, 1H, *J* = 3.4 Hz, 10.0 Hz, H-3β), 4.62 (m, 2H, H-6′α, H-6′β), 4.32-4.28 (m, 2H, H-5′α, H-6α), 4.23-4.06 (m, 4H, H-5′β, H-6b, H-9′aα, H-9′bα), 4.02-3.95 (m, 2H, H-9′ aβ, H-9′bβ), 3.88 (s, 3H, COOC*H*3α), 3.85 (s, 3H, COOC*H*3β) 2.70 (dt, 1H, H-3′eq), 2.35, 2.33 $(2 \times s, 6H, NCOCH₃\alpha)$, 2.28, 2.27 (2 × s, 6H, NCOC*H*₃β), 2.16-1.93 (7 × s, 21H, 7 × COCH₃), 1.68 (dd, 1H, H-3'ax); ¹³C from HSQC (125.7 MHz, CDCl₃) : δ = 96.3 (C-1 β), 94.2 C-1α), 72.0 (C-5), 71.2 (C-3β), 69.8 (C-6′), 68.4 (C-2), 68.3 (C-3α), 67.9 (C-4′α), 67.8 (C-8′ α), 67.5 (C-4β), 67.3 (C-4′β), 67.2 (C-8′β), 67.1 (C-7′), 62.5 (C-6), 62.3 (C-9′β), 61.8 (C-9′α), 56.6 (C-5′α), 56.1 (C-5′β), 53.3 (COO*CH3*), 38.9 (C-3′), 28.4, 28.3 (NAc2α), 26.9, 26.8 (NAc₂ β), 23.0-21.8 (7 × OAc α/β);

Phenyl [O-methyl 5-(A-acetylacetamido)-4,7,8,9-tetra-O-acetyl-3,5-dideoxy-_p-glycero-α-_pgalacto-2-non-ulopyranosylonate]-(2→3)-*O***-(2,4,6-tri-***O***-acetyl-β-D-galactopyranosyl)-(1→4)-** *O***-[(3,4-di-***O***-acetyl-2-***O***-benzyl-α-L-fucopyranosyl)-(1→3)]-***O***-[6-***O***-benzyl-2-deoxy-1-thio-2- (2,2,2-trichloroethoxycarbonylamino)-β-D-glucopyranoside] (15)**

A mixture of disaccharide acceptor **11** (30 mg, 0.035 mmol) and trichloroacetimidate donor **7** (50 mg, 0.052 mmol) in CH₂Cl₂ (4 ml) was placed under an atmosphere of argon and stirred at room temperature with 4\AA MS for 1 h. The reaction mixture was then cooled to 0 °C. TMSOTf (3.0 µmol, 0.035M solution in CH₂Cl₂) was added and stirring was continued for 1 h at the same temperature. The progress of reaction was monitored by TLC and MALDI-ToF MS. The reaction was quenched by the addition of pyridine (25 μ L), diluted with CH₂Cl₂ (10 ml), filtered, and washed with sat. aq. NaHCO₃ solution (10 ml) , water (10 ml) , and brine (10 ml) ml). The organic layer was dried (MgSO4), filtered, and the filtrate was concentrated *in vacuo*. The residue was purified by silica gel column chromatography (CHCl₃:EtOAc, 1:1, v:v) to afford compound **15** (35 mg, 61%) as a white amorphous solid. Analytical data for **15**: $R_f = 0.25$ (Acetone:Toluene, 1:3, v:v); ¹H-NMR (500 MHz, CDCl₃) : $\delta = 7.43$ -7.13 (m, 15H, aromatic), 5.56-5.51 (m, 2H, H-8‴, H-4‴), 5.32 (d, 1H, *J* = 7.3 Hz, N*H*Troc), 5.27 (bs, 1H, H-4″), 5.21-5.14 (m, 3H, H-1, H-1″, H-3″), 4.99 (d, 1H, *J* = 1.0 Hz, H-4′), 4.91-4.86 (m, 2H, H-5″, H-2′), 4.83 (d, 1H, *J1′,2′* = 8.2 Hz, H-1′), 4.79 (d, 1H, C*H*HPh), 4.68-4.90 (m, 7H, C*H*2Ph, C*H*HPh, C*H2*CCl3, H-3′, H-6″), 4.29 (t, *J*4′,5′ = *J5′,6′* = 10.1 Hz, H-5′), 4.23-4.14 (m, 4H, H-6a′, H-6b′, H-9a‴, H-3), 4.03 (m, 1H, H-9b‴), 3.96 (t, 1H, *J3,4* = *J5,4* = 9.0 Hz, H-4), 3.87-3.80 (m, 5H, COOC*H3*, H-6a, H-6b), 3.78 (m, 1H, H-5′), 3.52(bd, 1H, H-5), 3.10 (bm, 1H, H-2), 2.62 (m, 1H, H-3a′), 2.33, 2.26 (2s, 6H, 2× NCOC*H3*), 2.15-1.92 (9s, 27H, 9× COC*H*₃), 1.62 (m, 1H, H-3b^{'''}), 1.17 (d, 3H, $J = 6.6$ Hz, C*H*₃ fucose) ppm; ¹³C from HSQC $(125.7 \text{ MHz}, \text{CDCl}_3)$: $\delta = 99.6 \text{ (C-1'}, 97.7 \text{ (C-1'')}, 84.5 \text{ (C-1)}, 79.7 \text{ (C-5)}, 75.7 \text{ (C-3)}, 74.6$ (*CH2*Ph), 74.4 (C-2″), 73.9 (C-4), 73.2 (*CH2*Ph), 72.3 (C-4″), 71.8 (C-3′), 71.2 (C-5′), 70.8

(C-5″), 70.6 (C-3″), 69.7 (C-6′), 68.7 (C-6), 67.8 (C-4′), 67.5 (C-8′), 67.4 (C-7′), 67.3 (C-4‴), 64.8 (C-2′), 62.4 (C-9′), 62.0 (C-6′), 57.9 (C-2), 56.2 (C-5′), 53.4 (COO*CH3*), 38.7 (C-3‴), 28.6, 27.2 (NAc2), 21.6-20.5 (9 × OAc), 16.3 (C-6″); HR-MALDI-ToF/MS: *m/z*: calc. for $C_{73}H_{89}Cl_3N_2O_{33}S$ [M+Na]⁺: 1681.4032; found 1681.4029.

*N***-(9-Fluorenylmethyloxycarbonyl)-***O***-[2-azido-2-deoxy-3-***O***-(3,4,6-tri-***O***-acetyl-2-***O***-(2,5 difluorobenzoyl)-β-D-galactopyranosyl)-6-***O***-(***O***-methyl 5-(***N***-acetylacetamido)-4,7,8,9-tetra-***O***-acetyl-3,5-dideoxy-D-glycero-α-D-galacto-2-non-ulopyranosylonate)-(2→3)-***O***-(2,4,6-tri-***O***acetyl-β-D-galactopyranosyl)-(1→4)-***O***-[(3,4-di-***O***-acetyl-2-***O***-benzyl-α-L-fucopyranosyl)- (1→3)]-***O***-(6-***O***-benzyl-2-deoxy-2-(2,2,2-trichloroethoxycarbonylamino)-β-D-glucopyranosyl) α-D-galactopyranosyl]-L-threonine benzyl ester (16)**

A mixture of disaccharide acceptor **10** (22 mg, 0.025 mmol) and tetrasaccharide donor **15** (32 mg, 0.019 mmol) in CH_2Cl_2 (2 ml) was placed under an atmosphere of argon and stirred at room temperature with 4Å MS for 1 h. The reaction mixture was then cooled to 0 °C. *N*iodosuccinimide (22 mg, 0.096 mmol) and TfOH (0.019 mmol, 0.20 μ solution in CH₂Cl₂) were added sequentially and stirring was continued for 1 h at the same temperature. The progress of reaction was monitored by TLC and MALDI-ToF MS. The reaction was diluted with CH_2Cl_2 (10 ml), filtered, and washed with sat. aq. NaHCO₃ solution (10 ml), water (10 ml), and brine (10 ml). The organic layer was dried $(MgSO₄)$, filtered, and the filtrate was concentrated *in vacuo*. The residue was purified by silica gel column chromatography (Hexanes:EtOAc, 1:2, v:v) to afford compound **16** (29 mg, 55%) as a white amorphous solid. Analytical data for **16**: R_f = 0.30 (Hexanes:EtOAc, 1:2, v:v); ¹H-NMR (500 MHz, CDCl₃) : δ $= 7.77-6.42$ (m, 26H, $3 \times$ Bn, Fmoc, dFBz), 5.92-5.88 (m, 2H, H-4"'", H-8"'"), 5.84 (t, 1H, H-2′), 5.74 (m, 2H, H-3‴, H-4‴), 5.61 (d, 1H, *J* = 3 Hz, H-1‴), 5.47-5.41 (m, 4H, H-2‴′, H-7‴″, H-4′, H-4‴), 5.34 (m, 1H, H-5‴), 5.27 (m, 1H, H-3′), 5.19 (d, 1H, *J* = 8.1 Hz, H-1‴ &prime), 5.04-4.99 (m, 3H, H-6‴″, H-3‴′, COOC*H*HPh), 4.95 (d, 1H, OC*H*HPh), 4.86-4.77 (m, 3H, COOCH*H*Ph, COOC*H*HCCl3, OC*H*HPh), 4.73-4.49 (m, 9H, COOC*H*HCCl3, H-6a‴ ′, OC*H*HPh, H-1″, H-1, H-5‴″, OC*H*CH3 threonine, OC*H*HPh, H-6b‴′), 4.44-4.18 (m, 9H, C*H*2Fmoc, H-9a‴″, H-9b‴″, H-5‴′, C*H*COOBn threonine, H-1′, H-3‴, H-4′), 4.15-3.79 (m, 11H, H-6a, H-6b, H-2‴, H-6a′, H-6b′, CH2C*H*Fmoc, H-5, H-4, H-6a′, H-6b′, H-3), 3.77 (s, 3H, COOC*H*3), 3.46 (m, 4H, H-2″, H-2, H-5′, H-5″), 2.85 (dd, 1H, H-3‴″), 2.26, 2.22 (2s, 6H, 2× NCOC*H3*), 1.87 (m, 1H, H-3‴″), 1.88-1.65 (11s, 33H, 11× COC*H3*), 1.60 (d, 1.623H, *J* = 6.5 Hz, C*H*3 fucose), 1.57 (s, 3H, COC*H3*), 1.34 (m, 3H, C*H*3 threonine) ppm; 13C from HSQC $(125.7 \text{ MHz}, \text{CDCl}_3)$: $\delta = 101.6 \, (\text{C-1}'), 100.7 \, (\text{C-1}''), 100.5 \, (\text{C-1}'''), 99.9 \, (\text{C-1}), 97.4$ (C-1‴), 78.4 (C-3), 76.5 (O*CH*CH3 threonine), 75.6 (C-5″), 74.9 (C-2‴), 74.8 (C-3″), 74.7 (C-4″), 74.5 (*CH2*Ph), 73.6(*CH2*Ph), 73.1 (*CH2*Troc), 72.6 (C-4‴), 72.4 (C-3‴′), 71.7 (C-5′), 71.5 (C-5‴′), 71.1 (C-3′), 71.0 (C-2‴′), 70.8 (C-7‴″), 70.6 (C-3‴), 70.5 (C-6‴″), 70.3 (C-2′), 69.9 (C-5), 69.3 (C-6), 69.1 (C-6″), 68.1 (C-4), 67.9 C-4‴′), 67.7 (*CH2*Fmoc), 67.6 (COO*CH2*Ph threonine), 67.5 (C-4′), 67.4 (C-4‴″), 67.3 (C-8‴″), 65.2 (C-5‴), 62.5 (C-9‴ ″), 62.1 (C-6‴′), 61.6 (C-6′), 59.5 (O*CH*CH3 threonine), 59.3 (C-2), 58.9 (C-2″), 56.3 (C-5‴ γ , 52.9 (COO*CH*₃), 47.6 (CH₂*CH*Fmoc), 39.0 (C-3^{"'}"), 21.0, 20.9 (NAc₂), 20.8-20.1 (12 \times OAc), 18.9 (*CH3* threonine), 16.5 (C-6‴); HR-MALDI-ToF/MS: *m/z*: calc. for $C_{118}H_{135}Cl_3F_2N_6O_{51}$ [M+Na]⁺: 2617.7086; found 2617.7091.

*N***-(9-Fluorenylmethyloxycarbonyl)-***O***-[2-(***N***-acetamido)-2-deoxy-3-***O***-(3,4,6-tri-***O***-acetyl-2-***O***- (2,5-difluorobenzoyl)-β-D-galactopyranosyl)-6-***O***-(***O***-methyl 5-(***N***-acetylacetamido)-4,7,8,9 tetra-***O***-acetyl-3,5-dideoxy-D-glycero-α-D-galacto-2-non-ulopyranosylonate)-(2→3)-***O***-(2,4,6 tri-***O***-acetyl-β-D-galactopyranosyl)-(1→4)-***O***-[(3,4-di-***O***-acetyl-2-***O***-benzyl-α-L-fucopyranosyl)- (1→3)]-***O***-(6-***O***-benzyl-2-deoxy-2-(N-acetamido)-β-D-glucopyranosyl)-α-D-galactopyranosyl]- ^L-threonine benzyl ester (1)**

> Zn dust (400 mg, 6.12 mmol) and saturated aq. $CuSO₄$ (25 µL) were added to a solution of **16** (20 mg, 7.70 μmol) in THF (3 mL), Ac₂O (2 mL), and AcOH (1 mL) and the reaction stirred

at rt for 3 h. The reaction mixture was filtered and co-evaporated with toluene $(3 \times 5 \text{ mL})$. The residue was purified by silica gel column chromatography (CHCl₃:Acetone, $2/1$, v/v) to afford compound **1** (11.5 mg, 60%) as a white amorphous solid. Analytical data for **1**: R*^f* = 0.30 (CHCl₃:Acetone, 2/1, v/v); ¹H-NMR (600 MHz, acetone-D₆) : δ = 7.75-7.08 (m, 26H, 3 × Bn, Fmoc, dFBz), 6.99 (d, 1H, N*H*, GlcNAc), 6.51 (d, 1H, N*H*, GalNAc), 6.42 (d, 1H, N*H*Fmoc, threonine), 5.5 (m, 1H, H-8‴″), 5.45 (m, 1H, H-4‴″), 5.35 (d, 1H, *J* = 3.7 Hz, H-1‴), 5.29 (d, 1H, *J* = 3.5 Hz, H-4′), 5.25 (dd, 1H, *J* = 8.1 Hz, *J* = 10.5 Hz, H-2′), 5.19 (bd, 1H, *J* = 2.6 Hz, H-4‴), 5.12-5.08 (m, 3H, H-3‴, H-7‴″, H-3′), 4.99 (bd, 1H, *J* = 3.7 Hz, H-4‴′), 4.97-4.93 (dd, 2H, COOC*H2*Ph, threonine), 4.88 (q, 1H, H-5‴), 4.84-4.81 (m, 2H, H-2‴′, H-1‴′), 4.79 (d, 1H, *J* = 8.1 Hz, H-1″), 4.67 (d, 2H, 2 × C*H*HPh), 4.60 (m, 1H, H-3‴′), 4.57 (d, 1H, *J* = 2.8 Hz, H-1), 4.55 (dd, 1H, H-6‴″), 4.49 (d, 1H, *J* = 7.5 Hz, H-1″), 4.45 (d, 1H, CH*H*Ph), 4.38 (d, 1H, CH*H*Ph), 4.35 (m, C*H*HFmoc), 4.27-4.22 (m, 2H, CH*H*Fmoc, H-2), 4.20 (m, 1H, H-5‴ ″), 4.15-4.05 (m, 8H, OC*H*CH3 threonine, H-6a‴′, C*H*COOBn threonine, C*H*Fmoc, H-6b‴′, H-9a‴″, H-5′ H-6a′), 4.03-3.91(m, 5H, H-4″, H-6b′, H-9b‴″, H-4, H-3″), 3.87-3.81 (m, 3H, H-6a″, H-2″, H-5), 3.76-3.73 (m, 6H, H-5‴′, COOC*H*3, H-6b″, H-3), 3.70 (dd, 1H, H-2‴), 3.54 (m, 2H, H-6a, H-6b), 3.47 (m, 2H, C4-O*H*, H-5″), 2.48 (dd, 1H, H-3‴″), 2.24, 2.20 (2 × s, 6H, NAc₂), 2.14-1.74 (14 × s, 42H, 12 × OAc, 2 × NHAc), 1.43 (t, 1H, H-3^{"'}"), 1.16 (d, 3H, CH₃ threonine), 1.03 (d, 3H, $J = 6.6$ Hz, CH₃ fucose) ppm; ¹³C from HSQC (150.9 MHz, CDCl₃) : δ = 104.3 (C-1'), 104.1 (C-1''), 102.2 (C-1'''), 101.9 (C-1), 98.6 (C-1'''), 81.0 C-3), 77.6 (C-5″), 77.4 (O*CH*CH3 threonine), 77.3 (C-3″), 76.3 (C-2‴), 76.1 (C-4‴), 75.1 (*CH2*Ph), 74.6 (C-4‴), 74.2 (C-3‴′), 73.8 (*CH2*Ph), 73.6 (C-5‴′), 73.5 (C-5′), 73.4 (C-3′), 73.1 (C-2‴ ′), 72.8 (C-2′), 72.5 (C-3‴), 72.3 (C-6‴″), 72.2 (C-6), 72.1 (C-5), 71.4 (C-4), 71.2 (C-6″), 70.4 (C-8‴″), 70.2 (C-4‴′), 70.0 (C-4′), 69.8 (C-7‴″), 69.5 (COO*CH2*Ph), 69.2 (C-4‴″), 69.1 (*CH2*CHFmoc), 66.4 (C-5‴), 64.7 (C-9‴″), 64.1 (C-6‴′), 63.9 (C-6′), 61.8 (O*CH*CH³ threonine), 58.6 (C-2"), 58.5 (C-5""), 55.2 (COO*CH*₃), 50.5 (C-2), 49.8 (CH₂*CH*Fmoc), 41.1 $(C-3''''$, 29.8, 28.4 (NAc₂), 25.5-22.4 (12 5 OAc, 2 × NHAc), 21.4 (CH_3 threonine), 18.3 (C-6^{$''$}); HR-MALDI-ToF/MS: m/z : calc. for C₁₁₉H₁₄₀F₂N₄O₅₁ [M+Na]⁺: 2501.8350; found 2501.8353.

Supplementary Material

Refer to Web version on PubMed Central for supplementary material.

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Figure 1. Target molecule and building blocks.

Scheme 1.

One-pot three component reaction; $SDMP = 2.6$ -dimethylthiophenyl, $dFBz = 2.5$ difluorobenzyl.

a) Reaction sequence for one-pot reaction:

Scheme 2.

One-pot glycosylation followed by reductive opening of the benzylidene acetal.

