

Visualization of nitric oxide production in the mouse main olfactory bulb by a cell-trappable copper(II) fluorescent probe

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We report the visualization of NO production using fluorescence in tissue slices of the mouse main olfactory bulb. This discovery was possible through the use of a novel, cell-trappable probe for intracellular nitric oxide detection based on a symmetric scaffold with two NO-reactive sites. Ester moieties installed onto the fluorescent probe are cleaved by intracellular esterases to yield the corresponding negatively charged, cell-impermeable acids. The trappable probe Cu₂(FLZE) and the membrane-impermeable acid derivative Cu₂(FLZA) respond rapidly and selectively to NO in buffers that simulate biological conditions, and application of Cu₂(FLZE) leads to detection of endogenously produced NO in cell cultures and olfactory bulb brain slices.

fluorescent sensing | NO | olfaction | trappable probe | fluorescence microscopy

Nitric oxide (NO) is important for biological signaling. It activates soluble guanylyl cyclase, initiating a signaling cascade that promotes vascular smooth muscle dilation (1–3). Nitric oxide produced in the nervous system has been implicated in neurotransmission (4), and the immune system generates NO as a defense against pathogens (5). Unregulated nitric oxide production has been associated with pathological conditions such as cancer, ischemia, septic shock, inflammation, and neurodegeneration (6).

Because of its various biological consequences, investigating the generation, translocation, and utilization of NO continues to be an active area of research. A major limitation to advances in the field, however, has been the dearth of selective tools for visualization of biological NO, for example by fluorescence microscopy. Detection of nitric oxide offers many challenges. NO reacts rapidly in vivo with dioxygen, oxygen-generated radicals such as superoxide, amines, thiols, and metal centers (7, 8). It also diffuses readily from its point of origin (9), making rapid detection desirable for uncovering both its production site and function. Moreover, NO is produced at concentrations as low as ~100 picomolar, so it is essential to have an NO probe with a low detection limit (10). Fluorescent sensors can be designed to accommodate the properties of NO under physiological conditions, making this technique particularly valuable for in vivo nitric oxide imaging.

Transition metal complexes have been investigated as platforms for NO detection (11). The strategy is to incorporate a fluorophore into a ligand that is quenched either by intracellular photoinduced electron transfer and/or by coordination to a paramagnetic or heavy metal ion. Fluorescence is restored by reduction of the metal and/or displacement of the ligand upon reaction of the probe with NO. CuFL1 (Fig. 1) is an excellent example of such a metal-based cellular NO imaging agent (12, 13). CuFL1 satisfies many of the requirements of a good sensor. It is nontoxic, cell membrane permeable, has low energy excitation and emission wavelengths, responds directly and selectively to NO, and exhibits dramatic fluorescence enhancement upon reaction with NO. When the nonemissive CuFL1 reacts with NO, Cu(II) is reduced to Cu(I) and the secondary amine of the ligand

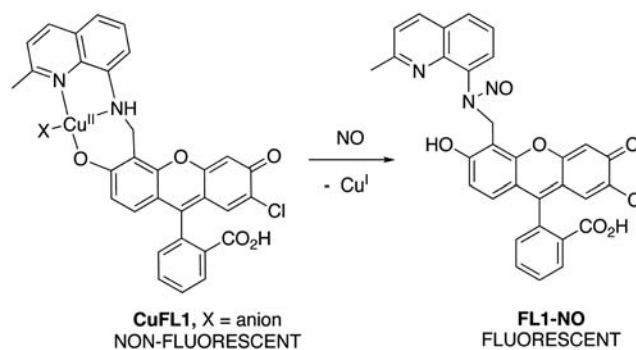


Fig. 1. Structure and NO detection scheme of CuFL1.

is *N*-nitrosated to produce the fluorescent species, FL1-NO (Fig. 1). CuFL1 has been used to monitor NO production in bacterial cell cultures with great success (14, 15), but to improve its applicability a cell-trappable version of the ligand is required. Cell-trappability is a key feature for biological experiments in intact tissues for which continual perfusion of tissue slices with artificial physiological saline solution is required. Under perfusion conditions, CuFL1 diffuses out of cells, making analysis of the results difficult. Incorporating an ester moiety onto the ligand is one way to confer trappability to a probe (16). The probe remains cell membrane permeable until cytosolic esterases cleave the ester to yield a carboxylate, which is negatively charged at physiological pH. This transformation traps the probe within the cell.

The mammalian olfactory bulb (OB), the first site for synaptic evaluation of olfactory receptor input in the CNS, is an ideal site to elucidate the functions of NO. Immunocytochemistry experiments confirm that the OB is rich in NO-producing cells (17). Direct measurements of NO in OBs, both in vivo and in vitro, using an NO-selective microprobe, reveal significant resting levels of NO and also odor-stimulated increases in NO (18). These results are of interest because of literature linking NO levels in the OB to odor learning and memory storage (19–21). To determine the effects of NO on synaptic interactions in the OB circuit, it would be of great value to localize NO sources and sinks on the seconds time scale with cellular resolution in living brain slices of the OB suitable for electrophysiological analysis (22, 23).

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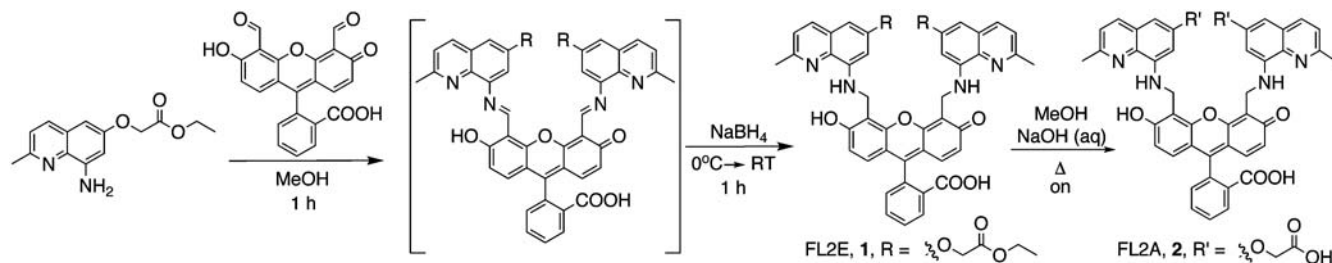


Fig. 2. Syntheses of FL2E and FL2A.

The development of the cell-trappable NO probe described here is an important contribution not only to these analyses, but potentially to studies of the cellular and synaptic actions of NO in many other brain regions, e.g., the hippocampus (24–26) and cerebellum (27, 28), where NO has also been implicated in processes of synaptic plasticity, learning, and memory. Ultimately, one might envision application of the trappable NO probe to anaesthetized mice exposed to odorants at the nose while monitoring NO production in the OB by fluorescence optical imaging.

Herein we describe the synthesis and characterization of a set of fluorescein-based symmetrical ligands and their Cu(II) complexes as NO-specific fluorescent probes. These second-generation probes are based on CuFL1 and employ the ester/acid strategy for cell-trappability. The Cu(II) complexes, generated in situ, respond rapidly and selectively to nitric oxide over other biologically relevant reactive oxygen and nitrogen species (RONS). We present data showing fluorescence detection of endogenously produced NO in macrophage and neuroblastoma cells, together with mouse olfactory bulb slice experiments that visually confirm and extend previous electrochemical results.

Results

Ligand Construction and Properties. The synthetic procedures that we used to generate the modified fluoresceins are outlined in Fig. 2. Condensation of ethyl[8-amino-2-methylquinolin-6-yl]oxyacetate] with 4',5'-fluorescein dialdehyde in a 2:1 ratio in methanol followed by reduction using sodium borohydride afforded FL2E, **1**. The acid form of the ligand was prepared by saponification of FL2E in methanol and water to yield FL2A, **2**.

The free ligands emit weakly with maxima at 522 and 516 nm for FL2E and FL2A, respectively, with quantum yields of $0.37 \pm$

0.05% and $1.8 \pm 0.2\%$. When $1 \mu\text{M}$ buffered sensor solutions, prepared in situ by combining CuCl_2 and FL2E or FL2A in a 2:1 ratio, were exposed to excess NO ($\sim 1,300$ equiv) under anaerobic conditions, the emission maxima shifted to 526 nm for both probes, in accord with formation of the *N*-nitrosated ligand (12). Accompanying this wavelength shift was a 17 ± 2 -fold and 27 ± 3 -fold increase in the integrated fluorescence emission relative to the initial copper complex for $\text{Cu}_2(\text{FL2E})$ and $\text{Cu}_2(\text{FL2A})$, respectively. The final quantum yields were $40 \pm 8\%$ and $36 \pm 5\%$, respectively. A fluorescence response was also observed when the probes were exposed to *S*-nitrosothiols (RSNO), such as *S*-nitroso-*N*-acetyl-DL-penicillamine (SNAP). The observed fluorescence was reduced compared to that generated using gaseous NO, because SNAP must decompose to produce NO before NO can react with the probe, and the decomposition kinetics are relatively slow under the conditions of the experiment. The probes are selective for NO or RSNO over other biologically relevant RONS, exhibiting only modest changes in fluorescence when exposed for 1 h to excess NO_2^- , NO_3^- , H_2O_2 , ClO^- , ONOO^- or HNO (Fig. S1).

Fluorescence Microscopic Imaging of Biologically Derived NO Production. We first evaluated the ability of the trappable probe to detect endogenously produced NO. Macrophage cells of the immune system produce micromolar quantities of NO when inducible nitric oxide synthase (iNOS) is activated in response to a pathogenic attack (5). Endotoxins from pathogens, such as lipopolysaccharide (LPS), and proinflammatory cytokines, such as interferon- γ (IFN- γ), induce expression of the iNOS gene, which results in NO production several hours later. To determine whether $\text{Cu}_2(\text{FL2E})$ can detect NO under these conditions, RAW 264.7 macrophages were stimulated with LPS (500 ng/mL)

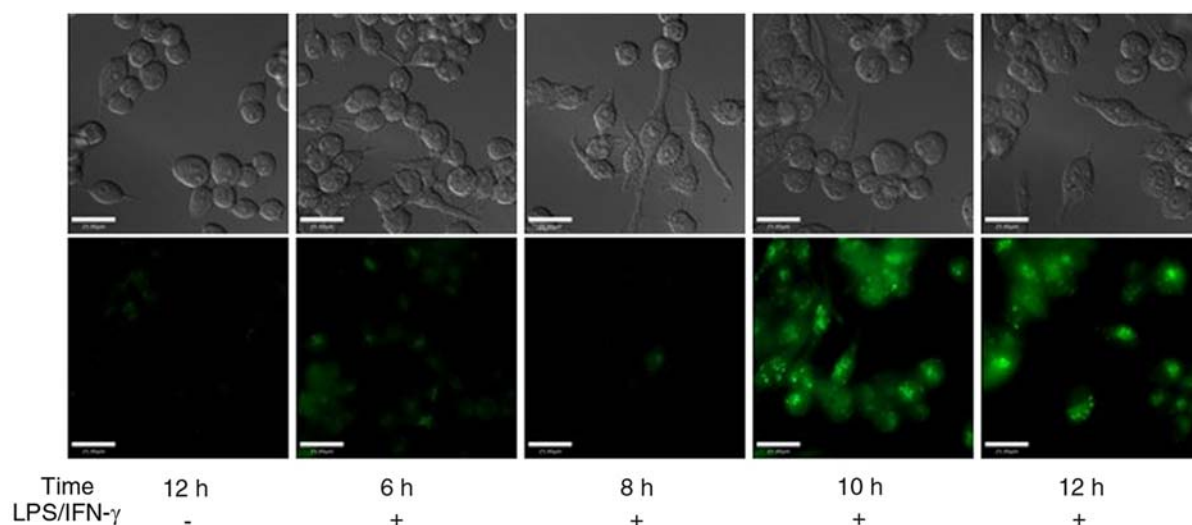


Fig. 3. RAW 264.7 macrophages coincubated with $1 \mu\text{M}$ $\text{Cu}_2(\text{FL2E})$, 500 ng/mL LPS, and 250–1,000 U/mL IFN- γ for 6–12 h. (Top) DIC images; (Bottom) emission from the probe in the green channel. Scale bars, 25 μm .

of the fluorescence signal increased by a factor of 12 ± 7 (range of increase 3.9–26.2, $n = 8$ trials from 4 cells, $p = 0.0003$), from which we estimate that NO levels in the slice were boosted from a basal value of ~ 85 nM to ~ 1 μ M for 1–2 min after KCl addition (18).

Discussion

Efficient synthetic procedures for generating unique ligands for preparing the NO-specific probes FL2E and FL2A have been devised. Cu(II) complexes assembled by combining CuCl₂ with either ligand in a 2:1 ratio display photophysical characteristics that are consistent with the asymmetric probe CuFL1 (12). The resulting complexes are excellent nitric oxide probes and are highly selective for NO and RSNO over other biologically relevant RONS.

The trappable probe detects endogenously produced NO from cNOS and iNOS, consistent with the capabilities of CuFL1 (13). Like CuFL1, the trappable probe is minimally toxic. Unlike CuFL1, however, the trappable probe is maintained within cells despite numerous changes of the extracellular medium, confirming that installing acetoxyethyl esters confers cell trappability. Cu₂(FL2E) localizes to the mitochondria, a feature also observed for the Zn(II) sensor ZS5 (31). The structure of ZS5 resembles that of FL2 up to the secondary amine of FL2E. In ZS5 the amine is tertiary and is appended to methyl pyridyl and methyl thiophene units. The fluorescein platform alone is not responsible for the mitochondrial localization, because fluorescein-based probes sometimes localize to other regions of the cell (31). The upper-ring substituents are not similar in ZS5 and FL2E, suggesting that some other, as yet unknown, factor is responsible for directing these sensors to mitochondria.

Our ultimate purpose in devising NO-specific probes is to apply them in experiments pertaining to the time-dependent production of NO in the mammalian brain. Because CuFL1 diffuses out of cells under conditions of tissue perfusion, no conclusive results were obtained from experiments using CuFL1 in mouse OB tissue slices. This shortcoming led to the present design of the cell-trappable probe, and with it OB tissue slices were successfully loaded and fluorescence caused by NO production visualized.

The detection of NO probe fluorescence in mouse OB slices is consistent with results obtained using an NO-selective microprobe to monitor robust basal NO production in the mitral and granule cell layers of slices (18). The NO microprobe technique provides fast (seconds time scale), sensitive, dynamic detection of extracellular NO signals, but is limited in spatial resolution to a single site ~ 10 μ m in diameter sampling the extracellular environment of only one or a few neurons. By contrast, the fluorescent probe described herein offers slower (minutes time scale), parallel, time-integrated detection of intracellular NO signals from dozens to hundreds of 10–15 μ m sized cells in slices that can trap the fluorophore. Extensive fluorescent labeling of slices suggests that tonic NO production is not confined to the granule cell layer, but is widespread throughout the OB. The most intense fluorescence occurs in the glomerular and granule cell layers, with weaker labeling of the external plexiform layer. This result mirrors the large-scale distribution of nNOS immunoreactive cells and fibers that has been documented in the mouse OB (17, 33–36). The higher density of nNOS expression in the glomerular and granule cell layers might result in elevated

binding of NO by locally trapped probe molecules. However, at the cellular level the patterns of fluorescence and NOS immunoreactivity should not be directly correlated, because the probe can be loaded into both NOS-positive and NOS-negative cells. For example, mitral cells in mouse OB are NOS-negative, but their somata displayed strong fluorescence in the present experiments. Probe trapped in mitral cells may be reporting NO that was synthesized in the granule cell layer and subsequently diffused to the mitral cell layer. The basal NO level was previously measured to be ~ 85 nM in the vicinity of mitral cell somata (18). One potential caveat of this interpretation is that NO may be continuously generated by cells during the loading period, and therefore some fraction of the sensor could bind NO before entry into cells. FL2E-NO could then reach and load cells beyond the normal spatial signaling range of NO in living tissue. To block this effect, the competitive NOS inhibitor *N*^ω-nitro-L-arginine was added and exogenous L-arginine was omitted during loading. NO production was not completely suppressed, as indicated by persistent probe fluorescence. However, the kinetics of post-loading fluorescence should depend on reception of NO signals by target cells because extracellular probe will be removed by perfusion (Fig. 6C), and thus postloading fluorescence signal increases will accurately report cellular patterns of NO synthesis and neurotransmission in the neural circuit.

We expected that our fluorescent probe could provide spatial information about the location of NO sources if extracellular NO were intercepted by an NO scavenger. We tested commonly used scavengers, 2-phenyl-4,4,5,5-tetramethylimidazole-1-oxyl 3-oxide (PTIO), its derivative carboxy-PTIO, hemoglobin, and myoglobin, but these were not compatible with the fluorescent probe: PTIO is membrane permeant and competes with the probe for NO binding, carboxy-PTIO reacts with NO to yield NO₂ which binds the probe, and both hemoglobin and myoglobin have strong optical absorption at the emission wavelengths of the probe. However, the door is open for synthesis of new scavengers that would be compatible with the probe and enable localization of NO generating cells.

In conclusion, optical recordings of steady state or stimulus-driven increases in fluorescence in slices promise to yield insights into spatiotemporal properties and physiological regulators of NO signaling in the central nervous system. Future improvements in probe design to limit the fluorescence response to intracellular, rather than tonic tissue sources, of NO should move the field toward functional imaging of neuronal signaling at the chemical, in addition to the electrophysiological, level.

Materials and Methods

Details describing the chemical reagents; synthetic materials and methods; synthetic procedures; spectroscopic materials and methods; cell culture; diffusion studies; NO detection experiments; imaging methods; cytotoxicity assays; olfactory bulb imaging; olfactory bulb preparation; nitric oxide probe loading; and animal care can be found in *SI Text*. Ligand NMR spectra are reported in *Figs. S5* and *S6*.

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