

# Variability of serum oxidative stress biomarkers relative to biochemical data and clinical parameters of glaucoma patients

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Purpose: The importance of oxidative stress in both the formation and the course of glaucoma has been known. Among the antioxidants, vitamin E possesses the specific effects and regulatory mechanisms of a neurohormone. The serum oxidant/antioxidant profile is reportedly altered in ocular pathologies. In this study, we analyzed the effect of the clinical parameters of glaucoma and biochemical data on antioxidants and serum oxidative stress markers as oxidation degradation products.

Methods: In this multicenter case control study, control and patient groups consisted of 31 healthy individuals and 160 glaucoma patients with no known additional abnormalities, respectively. We analyzed the oxidation degradation products malonyl dialdehyde (MDA), advanced oxidation protein products (AOPP), antioxidants, vitamins E and A, Serine (Ser), superoxide dismutase (SOD), glutathione peroxidase (Gpx), transferrine (TF), and total antioxidant capacity (TADA). All of these parameters and their relationships with serum cholesterol, glucose, protein, albumin, triglyceride levels, age, gender, visual acuities, intraocular pressure (IOP), c/d ratio, gonioscopic findings, medications, presence of pseudoexfoliation (px), central visual field and Optical Coherence Tomography (OCT) data, pachymetry, and Laplace values, were evaluated individually. Statistical comparisons were performed among them, and with the control group as well.

Results: TADA, AOPP, SOD, and Gpx were found to be decreased, and MDA, Ser, TF, vitamins A and E increased in the patient group. All data, excluding AOPP, varied significantly. Vitamin E was the most consistent parameter. Conclusions: In this study, the association between glaucoma and lipid oxidation was shown on a systematic basis, and the significance of vitamin E as a neuroprotective agent has been revealed once more.

Although glaucoma (recognized as the most frequent cause of irreversible blindness), is characterized by progressive retinal ganglion cell loss, it was regarded for years as "a disease associated with an increase in the intraocular pressure (IOP)" [1]. Even if increased IOP has been excluded from the definition of glaucoma, considered a major risk factor, and glaucoma has been defined as an optic neuropathy, current clinical applications strongly aim to decrease IOP. Though we have effective medical and surgical therapies at hand, progressive visual loss is still a prevalent symptom in glaucoma cases [2]. In light of current knowledge, a valid hypothesis is that ganglion cell death (apoptosis) observed in glaucoma is caused by a special type ischemia. Indeed, the clinical manifestations of glaucoma are different from other ischemic pathologies. Beyond animal experiments, ischemia and glaucoma can be induced by an increase in IOP, and guite different abnormalities have been observed using various methods, such as carotid occlusion [3]. In another study, the

role of oxidative damage in the etiopathogenesis of glaucoma was explained by the production of reactive oxygen species secondary to a complex trabecular mitochondrial defect [4].

Tissue damage due to oxidation is a chain reaction, which is mainly initiated by the production of free oxygen radicals. Though these molecules interact with intracellular signal conduction and regulation mechanisms, they demonstrate their main effects as destructive changes induced by a series of DNA reactions and macromolecules, such as proteins and lipids [5]. Oxidation degradation products, which are tissue and serum markers of this destructive process, consist of malonyl dialdehyde (MDA), advanced lipid oxidation endproducts (ALEs) for lipids, and advanced oxidation protein products (AOPPs) for proteins [6]. Vitamins E, C, and A, molecules like homocysteine, and transferrine (TF) bind oxygen ions and transform into steady-state compounds with their resultant antioxidant effects. Serine (Ser) is an amino acid used in the effect pathway of vitamin E. These buffer compounds that are formed offer their ions to the downregulating (velocity-limiting) systems, which consist of superoxide dismutase (SOD) and glutathione peroxidase (Gpx), to curtail chain reactions [7]. The nervous system, which also includes retina ganglion cells, is rich in lipids. Further, metabolic rate, oxygen degradation, and synthesis of

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Тав	LE 1. COMPARISON OF CONTROL AND PATIENT GRO	UPS.
Parameters	Control group	Patient group
Mean age	44.87±10.78	50.96±14.19
Gender (F/M)	15/16	106/54
Cholesterol	173±46.67	201.23±43.25
Glucose	86±5.66	119.76±58.54
Triglyceride	118±38.18	170.46±86.84
Protein	0.78±7.95	8.05±0.6
Albumin	4.67±0.21	4.71±0.25

ATP are increased in these cells, while the cellular regeneration rate is restricted. Dopamine oxidation and chemical factors (e.g., glutamate) are also important. Secondary to all of these factors, nerve cells are quite sensitive to oxidative damage [5].

The use of antioxidants for the prevention of glaucomatous decay is also addressed. Higher lipid contents of nerve cells has enhanced the importance of lipid-soluble vitamin E, especially  $\alpha$ -tocopherol, which has hormone-like regulatory mechanisms with its unique transporter proteins and receptors, exerting neuromodulatory effects on the eye and other tissues. Neuroprotective effects of vitamin E compounds in retinal diseases and glaucoma have been clincally demonstrated [8]. Ginkgo biloba extracts are also neuroprotective antioxidants [9]. Both vitamin E compounds [10], and ginkgo biloba extracts [11] also manifest vasoregulatory activities in the retina, which are significant for the prevention of ischemia.

As a form of optic neuropathy, glaucoma has also demonstrated central nervous system pathologies in experimental [12], and clinical [13,14] studies. In addition to the pathogenesis of all types of glaucoma, oxidative damage plays a key role [5,15]. In our study, we investigated the effects of the clinical parameters of glaucoma, and relevant biochemical parameters on various oxidative stress indicators, such as increments in various antioxidants and oxidative degradation products in serum samples.

# **METHODS**

Control and patient groups consisted of 31 and 160 individuals, respectively (Table 1). With routine examination, there were no findings implying ophthalmic pathologies, including glaucoma or ocular hypertension, in the control group. Patients with no known ocular or systemic concomitant disorders, previous glaucoma surgeries, and antioxidant usage, who received follow-up in our glaucoma polyclinics, were selected for the patient group. For both groups, oxidation degradation products (MDA and AOPP), antioxidants (e.g., vitamins E and A), Gpx, and total antioxidant capacity (TADA) were studied, in addition to routine blood biochemical tests for cholesterol, glucose, protein, albumin, and triglyceride. All of these parameteres with their relationships to blood cholesterol, glucose, protein, albumin and triglyceride levels, age, gender, visual acuity, intraocular pressure, c/d ratio, gonioscopic findings, drugs used, the presence of pseudoexfoliation (px), central field of vision, Mean deviation (MD)- Pattern Standart Deviation (PSD), pachymetry, and eye wall stress (Laplace's value), were evaluated individually. The patients were examined on the day of blood sample collection. The patients with visual acuities less than 5/10 in one eye were considered to have lower visual acuity. IOPs and c/d ratios were taken with the Pascal Dynamic Contour tonometry (Nidek Inc., Fremont, CA) and RTVue-100 fourier domain Optical Coherens Tomography (OCT) (Nidek Inc.), respectively. Patients with IOPs higher than 21 mmHg in one or both eyes were evaluated as monoor bilateral higher IOP groups, respectively, while those with c/d ratios more than 0.5 in one or both eyes were assessed as mono-or bilaterally increased IOP groups, respectively. Visual fields were taken with the Humphrey Field Analyzer Model 740i (Carl Zeiss Inc. Dublin, CA). Patients with glaucomateus visual field defects in one or both eyes were evaluated as mono-or bilateral visual field defect groups, respectively.

Corneal thickness <555 or >558 constituted groups with thinner or thicker corneas, respectively [16]. In gonioscopic examination, patients with grade  $\leq 2$  consisted of narrow-angle glaucoma patients, and the patients were divided into those using single (prostaglandin analogs, beta blockers), 2 and 3 drops, or patients without medications. Prostaglandin analoges and beta blockers were included in all of the combinations, and fixed combinations were considered as single drops. Corneal thickness of the patients were measured with the Pocket II pachymeter device (Quantel medical inc. Bozeman, MO). Axial diameters of corneas were measured with the AB5500+ (Sonomed Inc., L Success, NY) A scan mode, and together with the results of the pachymetre, the Laplace formula was applied [17]. Patients with IOPs higher or lower than normal eye wall stress values constituted higher and lower Laplace groups.

These results were compared statistically among one another, as well as with those of the control group. When sample sizes in all groups were appropriate for parametric tests, the Student *t*-test and Mann–Whitney U tests were used. Test results were evaluated as moderately (p<0.01), highly (p<0.005), or extremely (p<0.001) significant.

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*Biochemical analyses:* Routine biochemical analyses for serum glucose, cholesterol, triglyceride, total protein, and albumin were performed using the Roche autoanalyser DP (Roche Diagnostics Ltd. W Sussex, UK) modular calorimetric analytical method, while for other measurements were done manually.

Measurement of advanced oxidation protein products (AOPP)—AOPP levels were studied in the AU 2700 autoanalyser (Olympus Diagnostics Inc. Melville, NY). Blood samples were drawn from cubital veins in tubes with EDTA (EDTA), and centrifuged at 1660× g and 4 °C for 10 min. The blood samples were divided in aliquots and kept in Eppendorf tubes at -20 °C. All samples were analyzed within approximately 30 days. After preparation of chloramine-T stock solution (10 mmol/l), they were diluted 100 times with PBS (20 mmol/l, pH:7.4) to obtain a main standard solution at 100 µmol/l concentration (standard 1). Chloramine-T main standard solution (standard 1:100 µmol/l) was diluted with PBS (20 mmol/l, pH 7.4) to get a 5-point calibration curve, and lso to prepare Chloramine-T standards at 75, 50, 25, and 12.5 µmol/l concentrations. PBS (160 µl) was added to 40 µl standard or plasma, mixed, and incubated for 25 s. The absorbance of the mixture was read at 340 nm, then 20 µl of acetic acid was added and incubated for 25 s. Finally, 10 µl KI solution was added and reincubated for 25 s, and its absorbance was read again. All steps were completed at 37 °C in a single cuvette. Time intervals were arranged at every step as 25 s or longer based on the program characteristics of the analyzer. A calibration curve was formulated using absorbance (A) values corresponding to the concentrations of 5 standard solutions. AOPP concentration was reported as µmol Chloramine-T/l, corresponding to the absorbance measured.

Measurement of malonyl dialdehyde (MDA)-Blood samples drawn from the cubital vein were placed in blood tubes containing EDTA as an anticoagulant. Plasma samples were separated rapidly, and cryopreserved at -70 °C Samples were not thawed and refrozen, and they were also not exposed to light to avoid photooxidation. A 140 µl standard, sample, and reagent blank, were placed individually into microcentrifuge tubes. "Reagent" (455 µl) was then added into each tube and vortexed. HCl (105 µl; 12 N [37%]) was added into each tube as well. The tubes were stirred throughly, tapped close, and incubated at 95 °C for 60 min in a milleu of acidity provided by HCl. Then, 1 molecule of MDA and 2 molecules of reagent (N-methyl-2-phenylenilindol) reacted with each other to yield a stable chromophor product (colored product), which might provide a maximal absorbance spectrum at a 586 nm wavelength. Centrifugation at 13,000× g for 15 min yielded a clear supernatant sample. This sample (150 µl) was placed in each well. Their fluorescent activities were measured on a microplate reader (Synergy<sup>™</sup> 2 Multi-Mode; BioTek Instruments, Inc., Vinooski, VT) at 500 nm  $(\pm 30)$  excitation, and 586 nm  $(\pm 30)$  emission. Using a y=ax+b formula derived from the absorbance-concentration correlation of standards used for the construction of the MDA standard curve, and absorbance data obtained, the analyzer automatically calculated MDA concentrations, and the results were expressed as "µmol/l."

**Measurement of TADA**—Blood samples drawn from the cubital vein were taken into gelatinized tubes with no anticoagulant, centrifuged at  $3,000 \times \text{g}$ , and 4 °C for 12 min, and their plasma components were separated into two aliquots. Dilution buffer, copper, and stop solutions were preserved at 2–8 °C, and working samples were kept as "aliquots" at –70 °C. Before measurements, dilution buffer, copper, and stop solutions were kept under room temperature for 30 min. To obtain a standard solution, 1.5 ml deionized water, and uric acid standard were added, and the prepared solution was vortexed until it dissolved thoroughly to obtain a "uric acid standard" with a final concentration of 2 nM.

From this stock standard solution, 5 working standards were obtained using serial dilutions. A dilution buffer was used in 1:4 dilutions of standard and sample solutions. Standard and sample solutions were diluted with the dilution buffer. Diluted standard, samples, reagent blank, and 200 µl buffer were pipetted into every well. The reagent blank contained buffer solution for dilution and the standard/sample solution. At 490 nm wavelength using a Roche COBAS MIRA Plus Chemistry Analyzer (Roche Diagnostics Ltd. W Sussex, UK), the absorbances of reagent blank, standard, and samples were read. Copper solution (50 µl) was added in each well, excluding wells containing reagent blank, incubated under room temperature for 3 min, and the reaction was finalized after adding 50 µl stop solution in each well. A second reading was done for absorbance at 490 nm wavelength. Under the combined effects of the sample of antioxidants in the standard solution, Cu<sup>+</sup> (ferric form) in the copper solution is reduced into Cu<sup>2+</sup> (ferro form), and a color reaction yielding maximal absorbance at 490 nm wavelength is formed. Using spectrophotometry, total antioxidant capacity (TAC) was automatically estimated based on the equation y=ax+b, derived from uric acid standard curve, and the results were recorded.

Determination of superoxide dismutase (SOD) activity—Blood samples were drawn from the cubital vein, placed into tubes containing EDTA as an anticoagulant, and centrifuged at  $1,000 \times$  g and 4 °C for 10 min. Supernatant plasma was drawn using a pipette, and then discarded. The samples were mixed 4 times with ice water to disintegrate RBCs. A 250 µl sample and 400 µl of a ethanol/chloroform (62.5/37.5) solution were mixed to measure Cu-Zn SOD activity. Thus, inactivation of mitochondrial Mn-SOD and Fe-SOD by ethanol/chloroform mixture was achieved. Then, this mixture was centrifuged at 3,000× g and 4 °C for a minimum of 30 s. The samples were frozen at -70 °C, and analyzed within 1 month. The Roche COBAS MIRA Plus Chemistry Analyzer (Roche Diagnostics Ltd. W Sussex, UK) was used



for manual measurement procedure. The results were evaluated in comparison with hemoglobin concentrations.

The determination of erythrocyte Gpx activity was achieved using a Cellular Glutathione Peroxidase kit (Cayman Chemical Ann Arbor, MI), modified for the Roche COBAS MIRA Plus analyzer (Roche Diagnostica). Gpx concentrations were calculated based on the Equation 1 mU/ ml=1 nmol NADPH/ml/min=(A3, 40min)/0.00622. The results were evaluated in comparison with hemoglobin concentrations.

Transferrine (TF) analyses were performed using a Beckman kit.

Levels of vitamins E and A were measured using High Perfusion Lipid Chromatography (HPLC). The patient's serum (200 µl) was placed into a tube and reacted with 200 µl (1 g/l) ethanol ascorbate to deproteinize. Asetonitrile  $(24 \mu l)$ was added to the mixture, and then vortexed for 1 min. Next, 500 µl of HPLC grade hexane was added to this solution, vortexed for 2 min, centrifuged at 770× g for 1 min,and a supernatant hexane phase was collected. This procedure was repeated 3 times. Vitamins A and E were derived from approximately 1,500 µl of hexane solution in these procedures. A completely clear 1,250 µl solution was drawn from the tube, removing a small amount of precipitated sediment that as present in the bottom of the tube. The hexane was completely evaporated under liquid nitrogen in a water bath at 36 °C. The tubes were closed watertight with paraffin, and prepared for the test. To perform the HPLC procedure,  $10 \mu l$  (1 g/l) ethanol ascorbate was added to the sediment at the bottom of the tube, and then vortexed. Then, 150 µl of mobile phase (methanol:water; 95.5) was added, vortexed for 1 min, and degassed by sonication for 1 min. The mixture was sieved through a 45 nm filter. A 50  $\mu$ l sample was removed from this filtered solution, injected into the HPLC Chromesystem column at a flow rate of 1.5 ml/min, and using the Knauer UV detector (Knauer Inc. Berlin, Germany), vitamins E and A were read at 295 nm and 320 nm, respectively. Peaks of vitamins A and E, which were derived individually from chromatograms obtained from the Spectra-Physics integrator (Triad Scientific Inc. Lakewood, NJ), were compared to calculate concentrations expressed as  $\mu$ g/ml.

### RESULTS

In comparison to the control group, TADA, AOPP, SOD, and Gpx were found to be decreased, while MDA, Ser, TF, and vitamins A and E increased in the patient group (Figure 1). Excluding AOPP, all data varied significantly.

For vitamin E and MDA, this increase was found to be extremely significant. In the hypertriglyceridemia group, MDA, vitamin E, TADA, Gpx, SOD, and TF levels varied significantly. In the hypo-and hyperproteinemia groups, MDA, and TADA levels varied significantly, and in the hypercholesterolemia group, all data (excluding AOPP) varied significantly. Vitamin E demonstrated extremely significant increases among all glycemia groups, in hypercholesterolemia, and hypertriglyceridemia, as well as in hyper-and normoproteinemia groups. In the hyperproteinemia group, an increase in Ser was found to be extremely significant. MDA increments were of utmost significance in the hypercholesterolemia while in group, the

$ \begin{array}{cccc} Cholesterol & H \ (n=13) & 1.63\pm0.09 & 23\\ 1 \ (n=44) & 1.62\pm0.12 & 34.8\\ N \ (n=103) & 1.63\pm0.11 & 31.2\\ Glucose & H \ (n=8) & 1.58\pm0.19 & 28.5\\ 1 \ (n=32) & 1.64\pm0.1 & 36.1\\ N \ (n=120) & 1.62\pm0.11 & 31.1\\ Trygliceride & H \ (n=17) & 1.62\pm0.17 & 53.1\\ 1 \ (n=36) & 1.63\pm0.08 & 29. \end{array} $	23±7.38 34 87+30 15		GPX	MDA	SEK	H.I.	VITA	VITE
$ \begin{array}{ccccccc} I \ (n=44) & 1.62\pm0.12 & 34.8 \\ N \ (n=103) & 1.63\pm0.11 & 31.2 \\ R \ (n=8) & 1.58\pm0.19 & 28.5 \\ I \ (n=32) & 1.64\pm0.1 & 36.1 \\ N \ (n=120) & 1.62\pm0.11 & 31.1 \\ N \ (n=17) & 1.62\pm0.17 & 53.1 \\ I \ (n=36) & 1.63\pm0.08 & 29. \\ I \ (n=36) & 1.63\pm0.08 & 29. \\ \end{array} $	34 87+30 15	$3.03 \pm 0.67$	$1.97 \pm 0.67$	$1.34 \pm 0.27$	$28.9\pm 12.2$	$276.92 \pm 72.15$	$0.67 \pm 0.35$	$17.4 \pm 4.87$
$ \begin{array}{cccc} N \ (n=103) & 1.63\pm0.11 & 31.2 \\ Glucose & H \ (n=8) & 1.58\pm0.19 & 28.5 \\ I \ (n=32) & 1.64\pm0.1 & 36.1 \\ N \ (n=120) & 1.62\pm0.11 & 31.1 \\ Trygliceride & H \ (n=17) & 1.62\pm0.17 & 53.1 \\ I \ (n=36) & 1.63\pm0.08 & 29. \\ \end{array} $	01.00-10.10	$3.06 \pm 0.83$	$1.91 \pm 0.63$	$1.46 \pm 0.97$	37.7±13.25	$259.93\pm59.68$	$0.9 \pm 0.58$	$20.58\pm 8.65$
$ \begin{array}{cccc} Glucose & H \ (n=8) & 1.58\pm0.19 & 28.9 \\ I \ (n=32) & 1.64\pm0.1 & 36.1 \\ N \ (n=120) & 1.62\pm0.11 & 31.1 \\ Trygliceride & H \ (n=17) & 1.62\pm0.17 & 53.1 \\ I \ (n=36) & 1.63\pm0.08 & 29. \\ \end{array} $	$31.24\pm16.31$	$3.03 \pm 0.83$	$2.04\pm0.75$	$1.33 \pm 0.3$	$37.12\pm13.82$	$280.62\pm 84.05$	$0.88 \pm 0.62$	$20.89 \pm 15.16$
$ \begin{array}{cccc} I \ (n=32) & 1.64\pm0.1 & 36.1 \\ N \ (n=120) & 1.62\pm0.11 & 31.1 \\ Trygliceride & H \ (n=17) & 1.62\pm0.17 & 53.1 \\ I \ (n=36) & 1.63\pm0.08 & 29. \\ \end{array} $	$28.98 \pm 13.58$	$3.25 \pm 1.13$	$2.23\pm0.34$	$1.31\pm0.21$	42.34±6.55	$264.8 \pm 55.04$	$0.66 \pm 0.23$	$20.83\pm8.13$
N (n=120) 1.62 $\pm$ 0.11 31.1 Trygliceride H (n=17) 1.62 $\pm$ 0.17 53.1 1 (n=36) 1.63 $\pm$ 0.08 29.	$36.15\pm 25.82$	$3.19\pm0.89$	$1.97 \pm 0.64$	$1.26 \pm 0.23$	$34.72\pm10.11$	$270.52\pm 59.16$	$0.87 \pm 0.54$	20.77±12.88
Trygliceride H (n=17) 1.62±0.17 53.1 1 (n=36) 1.63±0.08 29.	$31.17\pm 25.38$	$2.98 \pm 0.76$	$1.91 \pm 0.7$	2.37±9.8	37.44±14.75	267.72±75.83	$0.87 \pm 0.61$	$20.2\pm10.55$
I $(n=36)$ 1.63±0.08 29.	53.12±32.17	$3.24 \pm 1.11$	$0.7 \pm 0.58$	$1.25 \pm 0.3$	$39.29\pm10.4$	$277.94\pm 52.53$	$0.86 \pm 0.44$	21.85±7.75
	$29.1 \pm 14.79$	$2.95 \pm 0.75$	$1.88 \pm 0.7$	$1.42 \pm 0.27$	$32.13\pm12.53$	$250.58\pm61.66$	$0.83 \pm 0.47$	$17.99 \pm 5.43$
N(n=107) 1.63±0.1 30.0	$30.61 \pm 25.56$	$3.12 \pm 0.78$	$1.97 \pm 0.67$	$1.31 \pm 0.26$	$36.67\pm 14.5$	$275.11 \pm 78.05$	$0.79 \pm 0.5$	$19.49 \pm 11.05$
Protein I (n=19) 1.61±0.12 37.5	37.56±27.37	$3.18 \pm 0.97$	$2.23\pm0.63$	$1.37 \pm 0.22$	$36.02 \pm 9.01$	$315.78 \pm 103.32$	$0.69 \pm 0.34$	17.24±4.55
N (n=139) 1.63±0.11 31.1	$31.18\pm 24.65$	$3.07 \pm 0.79$	$1.9 \pm 0.67$	$1.4 \pm 0.86$	$36.17\pm14.23$	$262.48\pm65.6$	$0.82 \pm 0.51$	$19.46 \pm 10.48$
D $(n=2)$ 1.68±0.1 51.	$51.16\pm1.02$	$2.63 \pm 0.78$	$1.35 \pm 0.03$	$1.48 \pm 0.22$	$20.7\pm6.79$	$268 \pm 98.99$	$0.62 \pm 0.11$	$16.8 \pm 2.54$
Albumin I (n=3) 1.65±0.05 40.1	$40.14\pm21.55$	$2.89 \pm 1.04$	$2.43\pm0.22$	$1.49 \pm 0.1$	37.57±13.08	448±225.8	$0.55 \pm 0.22$	$16.37\pm6.3$
N (n=141) 1.63±0.11 32.1	32.16±25.17	$3.08 \pm 0.83$	$1.16\pm 1.92$	$2.18 \pm 9.11$	$35.2\pm 13.66$	$264.35\pm63.67$	$0.77 \pm 0.45$	$18.89 \pm 9.77$
D (n=16) 1.62±0.12 38.0	38.07±26.81	3.26±0.6	$2.42\pm0.8$	$1.23 \pm 0.18$	42.12±13.43	$279.44 \pm 65.67$	$1.18 \pm 0.72$	22.99±10.98
TADA: total antioxidant canacity AOPP: 3	PP. advanced o	xidation nrotei	n nroducts S	OD <sup>.</sup> suneroxid	e dismutase Gnx	·· olutathione nerox	idase MDA <sup>,</sup> m	alonvl

nyl	
: malc	
MDA	
peroxidase,	: number.
glutathione	N: Normal, n
Gpx:	ased, N
dismutase,	l, D: Decrei
superoxide	I: Increased
SOD:	ry high
products,	E, H: Ver
protein 1	3: vitamin
oxidation	ı A, Vit: I
dvanced o	A: vitamin
)PP: a	e, Vit /
ty, AC	sferrine
capaci	F: trans
ntioxidant	: Serine, Tl
otal ar	e, Ser:
DA: t(	dehyd
TA	dial

TABLE 2. VARIABILITY OF SERUM OXIDATIVE STRESS MARKERS BASED ON BIOCHEMICAL PARAMETERS.

	TABLE 5. STATIST	ICAL SIGNIFIC	ANCE OF SERUI		STRESS MAP	KKEKS BASED	ON BIOCHEM	IICAL PARA	METERS.	
Parameters	Groups	TADA	AOPP	SOD	Gpx	MDA	Ser	TF	Vit A	Vit E
Cholesterol	Н					*				**
	Ι	***		*	*	***	*		*	***
	Ν	**		*		**		*		**
Glucose										
	Н						**			*
	Ι	*				*		*		***
	Ν	***		*	*	***	*			***
Triglyceride	Н					*	*	*		*
	Ι					***				***
	Ν	*				*		*		***
Protein	Ι	*				***	***	*		***
	Ν	***			*	***				***
	D					*				
Albumin	Ι								*	***
	Ν					*				
	D	***				***				***

TABLE 3. STATISTICAL SIGNIFICANCE OF SERUM OXIDATIVE STRESS MARKERS BASED ON BIOCHEMICAL PARAMETERS

TADA: total antioxidant capacity, AOPP: advanced oxidation protein products, SOD: superoxide dismutase, Gpx: glutathione peroxidase, MDA: malonyl dialdehyde, Ser: Serine, TF: transferrine, Vit A: vitamin A, Vit: E: vitamin E, H: Very high I: Increased, D: Decreased, N: Normal, n: number., \*:p<0.01 moderately significant,\*\*: p<0.005 very significant, \*\*\*: p<0.001 extremely significant.

hypertriglyceridemia group, they were observed to be extremely significant (Table 2 and Table 3). The presence of px induces decreases in TADA and AOPP, and increases in MDA and vitamin E levels, while in the wide-angle glaucoma patients, TADA was lower. In both wide and narrow-angle glaucoma cases, MDA and vitamin E levels were found to be higher. In the lower visual acuity group, MDA and vitamin E increased and SOD decreased, while in the higher IOP group, TADA was found to be significantly lower. Higher Laplace and c/d values significantly influenced TADA and vitamin E. Vitamin E demonstrated extremely higher levels in both genders, all ages, IOPs, and age groups, as well as in patients manifesting bilateral c/d abnormalities. In addition to having extremely higher decreases in TADA levels in 40- to 60-yearold patients and normal IOPs, and wide-angle glaucomas, and extremely higher increases in MDA concentrations in agematched cases, normal visual acuities and higher IOPs were observed (Table 4 and Table 5). Furthermore, vitamin E showed extremely higher levels in all MD and PSD groups, and thicker cornea groups as well. In addition, vitamins E and A were found to be extremely higher in groups of patients with increased Laplace values. Significant differences were observed in MDA, vitamin E, SOD, and TADA levels in groups with higher MD and PSD values (Table 6 and Table 7).

When compared with the normal control groups, MDA levels in the group of patients with hyperglycemia, px, and Gpx levels, in cases with bilaterally higher c/d, ratios were found to be extremely higher and lower, respectively. In addition, in the group with lower visual acuity, MDA and vitamin E levels were moderately higher and SOD levels were

moderately lower when compared to the patients with normal visual acuities. Vitamins E and A were found to be significantly increased in the group with higher IOP Laplace scores as well.

### DISCUSSION

Glaucoma, being a nonsystemic disease, is the most frequent etiology of irreversible blindness worldwide is not only an ocular pathology [13]. However, studies related to its effects on serum oxidative stress markers are quite limited in the existing literature. In ocular pathologies, such as Behçet's disease [18] and cataract [19], the serum oxidant/antioxidant profile was reportedly altered. However, in glaucoma, it was determined that the serum glutathion concentration decreased [20], and composition of blood fatty acids were altered [21]. In this study, the effects of 5 biochemical and 12 clinical parameters on serum levels of oxidation degradation products and 8 antioxidants were examined.

In the main comparison between the patient and control groups, the observation of significant differences in all data (excluding AOPP) should be emphasized. In this study, TADA, protein oxidation end product AOPP, and the enzymes SOD and Gpx, were found to be decreased. On the other hand, lipid oxidation end product MDA, Ser and TF, and liposoluble vitamins A and E increased in the patient group. Glaucoma is an optic neuropathy, and oxidative stress plays an important role in its etiopathogenesis. Thus, adipose tissue damage is its predominant characteristic [5]. Oxidative reactions occurring within tissues lead to the formation of different end-products, according to the molecules they affect. Among these end-products that can be traced in serum, AAOP

			Table 4. Variabil	ITY OF SERUM OXIDA	TIVE STRESS MARKERS	BASED ON CLINICAL	PARAMETERS.			
Davamotone	Cunne	TADA	4OP	COD	CDV	MDA	CED	ŢĘ	VIT A	VIT F
Gender	F(n=106)	1.63±0.11	$29.93\pm 25.03$	3.08±0.76	$1.98 \pm 0.72$	1.43±0.96	36.26±14.14	$269.55\pm62.68$	0.78±0.45	$19.04\pm10.79$
	M (n=54)	$1.63 \pm 0.11$	$37.73\pm 24.51$	$3.12 \pm 0.91$	$1.87 \pm 0.56$	$1.31 \pm 0.26$	35.44±12.95	$269.11\pm 89.68$	$0.87 \pm 0.57$	19.69±7.85
Age (years)	<40 (n=24)	$1.66\pm 0.11$	$30.44 \pm 14.72$	2.75±0.67	$2.04 \pm 0.74$	$1.34 \pm 0.35$	39.69±14.61	$303.21 \pm 104.34$	$0.74 \pm 0.38$	$18.16\pm8.02$
•	40-60 (n=92)	$1.63 \pm 0.11$	$31.9\pm 28.26$	$3.18 \pm 0.82$	$1.96 \pm 0.7$	$1.34 \pm 0.26$	$36.31\pm14.26$	267.54±62.11	$0.84 \pm 0.56$	$20.22 \pm 11.25$
	>60 (n=34)	$1.63 \pm 0.09$	$34.03\pm 26.02$	$3.08 \pm 0.86$	$1.8 \pm 0.59$	$1.29 \pm 0.26$	$33.38\pm11.63$	$259.33\pm69.8$	$0.78 \pm 0.35$	$18.24 \pm 6.87$
Visual acuity	N (n=138)	$1.63 \pm 0.11$	$31.99\pm 26.76$	$3.12 \pm 0.84$	$1.9 \pm 0.69$	$1.33 \pm 0.28$	$37.21\pm13.44$	275.69±74.65	$0.81 \pm 0.5$	$20.01\pm10.5$
	D (n=22)	$1.63 \pm 0.09$	33.39±20.45	$2.95 \pm 0.67$	$2.08 \pm 0.63$	$1.32 \pm 0.25$	$30.15\pm 14.91$	244.95±54.64	$0.78 \pm 0.41$	$16.16 \pm 4.61$
IOP	N (n=105)	$1.64 \pm 0.1$	31.17±21.2	$3.06 \pm 0.82$	$1.96 \pm 0.73$	$1.29 \pm 0.29$	35.82±11.17	270.69±77.97	$0.82 \pm 0.48$	$19.65 \pm 11.02$
	U (n=25)	$1.64 \pm 0.11$	$28.31 \pm 13.95$	$3.25 \pm 0.89$	$1.8 \pm 0.66$	$1.37 \pm 0.22$	35.78±15.12	$273.18 \pm 74.05$	$0.8 \pm 0.56$	$20.38 \pm 9$
	B (n=30)	$1.61 \pm 0.1$	38.7±42.96	$3.02 \pm .69$	$1.96 \pm .54$	$1.4 \pm 0.24$	37.76±12.98	$272.1\pm60.96$	$0.78 \pm 0.47$	$17.97 \pm 6.88$
C/d	N (n=89)	$1.63 \pm 0.1$	$37.34 \pm 32.53$	$3.07\pm0.73$	$1.74 \pm 0.54$	$1.35 \pm 0.29$	$37.43\pm 14.4$	272.46±70.2	$0.83 \pm 0.47$	$19.83 \pm 8.5$
	U (n=26)	$1.65 \pm 0.1$	$30.02 \pm 14.12$	$3.01 \pm 0.95$	$1.87 \pm 0.71$	$1.32 \pm 0.22$	32.73±13.92	295.67±105.5	$0.92 \pm 0.63$	21.84±17.45
	B (n=45)	$1.63 \pm 0.12$	$25.93 \pm 13.99$	$3.22 \pm 0.89$	2.25±0.77	$1.29 \pm 0.27$	35.65±12.98	255.07±51.4	$0.72 \pm 0.45$	$17.68 \pm 6.47$
Angle	Wide (n=140)	$1.67 \pm 0.1$	$31.67\pm0.1$	$3.01 \pm 0.88$	$1.93 \pm 0.68$	$1.31\pm0.19$	$35.23\pm12.89$	$285.5\pm 81.96$	$1.25 \pm 0.84$	$29.08 \pm 25.01$
I	Narrow (n=20)	$1.63 \pm 0.11$	$31.64 \pm 25.36$	$3.12 \pm 0.81$	$2.17\pm0.82$	$1.33 \pm 0.28$	52.31±17.72	271.54±72.54	$0.78 \pm 0.44$	$18.88 \pm 7.71$
Drugs	(-) (n=49)	$1.63 \pm 0.1$	$30.97 \pm 25.85$	$3.06 \pm 0.8$	$1.9 \pm 0.61$	$1.34 \pm 0.28$	$37.01\pm14.14$	$286.22\pm 83.19$	$0.74 \pm 0.47$	$18.94 \pm 8.14$
	PG (n=43)	$1.67 \pm 0.09$	$41.23\pm36.53$	$3.15\pm0.71$	$1.83 \pm 0.74$	$1.29 \pm 0.27$	38.13±14.58	256.36±53.32	$0.97 \pm 0.56$	$21.99 \pm 14.85$
	b-b (n=13)	$1.62 \pm 0.13$	$24.88 \pm 12.79$	$3.01 \pm 0.94$	$1.82 \pm 0.74$	$1.37\pm0.25$	33.86±16.76	$248.65\pm 68.93$	$0.91 \pm 0.6$	$19.99 \pm 9.97$
	2 drugs (n=44)	$1.61 \pm 0.07$	32.99±19.38	$3.1 \pm 0.77$	2.26±0.77	$1.25 \pm 0.3$	31.59±9.18	245.67±37.29	$0.71 \pm 0.35$	17.58±7.6
	3 drugs (n=11)	$1.57 \pm 0.18$	35.32±21.43	$3.73\pm1.08$	$2.35 \pm 0.87$	$1.32 \pm 0.21$	36.58±7.93	$295.17 \pm 85.6$	$0.87 \pm 0.13$	$17.63 \pm 3.1$
Px	(+) (n=21)	$1.62 \pm 0.11$	$31.23\pm 8.93$	$3.06 \pm 0.7$	$1.94 \pm 0.89$	$1.4 \pm 0.3$	29.82±12.2	249±52.52	$0.688 \pm 0.19$	15.54±3.29
	(-) (n=139)	$1.67 \pm 0.08$	$32.98 \pm 18.38$	$2.56 \pm 0.81$	$2.23\pm0.13$	$1.35 \pm 0.23$	28.63±5.71	217.33±44.66	$0.61 \pm 0.34$	$16.9\pm6.41$
TADA: to	tal antioxidant caps	acity. AOPP: a	idvanced oxidati	on protein pro	ducts. SOD: s	uperoxide dis	mutase. Gpx: gl	utathione peroxids	ase. MDA: ma	lonvl
		· · · · · · · · · · · · · · · · · · ·			TT. TT.: 1-4-1				1 1. 1.	

# dialdehyde, Ser: Serine, TF: transferrine, Vit A: vitamin A, Vit: E: vitamin E, U: Unilateral, B: Bilateral, D: Decreased, N: Normal, PG: Prostaglandin analogous, β-b: β-blocker, n: number.

	I ABLE 5. STATISI	ICAL SIGNIFICA	NCE OF SERUM	OXIDATIVE S	TRESS MARK	KERS BASED OF	N CLINICAL	PARAMETI	ERS.	
Parameters	Groups	TADA	AOPP	SOD	Gpx	MDA	Ser	TF	Vit A	Vit E
Gender	Female	**				**				***
	Male	**				**			*	***
Age (years)	<40	*		***		*		*		***
	40-60	***				***		*		***
	>60				*	*				***
Visual acuity	Ν	*				***	*	*		***
	D			*		*				*
IOP	Ν	***		*		*	*		*	***
	U	*				***				***
	В			*		***	*			***
c/d	Ν	***		*	***	***	*	*		
	U					**		*		**
	В	*			***	*				***
Angle	Wide	***			*	***		*		***
	Narrow					*	***		*	***
Drugs	(-)			*	*	***		*		***
•	PG	*				*			*	**
	β-b			*		**				**
	2 drugs									
	3 drugs									
Px	(+)		*			***				**
	(-)				*	**				*

TADA: total antioxidant capacity, AOPP: advanced oxidation protein products, SOD: superoxide dismutase, Gpx: glutathione peroxidase, MDA: malonyl dialdehyde, Ser: Serine, TF: transferrine, Vit A: vitamin A, Vit: E: vitamin E, U: Unilateral, B: Bilateral, D: Decreased, N: Normal, PG: Prostaglandin analogous, β-b: β-blocker, n: number, \*:p<0.01 moderately significant,\*\*: p<0.005 very significant, \*\*\*: p<0.001 extremely significant.

is formed as a result of protein degradation [22]. Further, the increments in levels of vitamin E and MDA in the patient group were found to be extremely significant. Vitamins A and E are lipid soluble [23], and MDA is a lipid oxidation degradation product [24]. In glaucoma patients, MDA increases twofold in the anterior chamber [25], and in cases with cataract, its serum levels are also enhanced [19]. Since retina and nervous tissue are rich in lipid, these results are significant. In addition, in the hypertriglyceridemia group, TADA, Gpx, and SOD were statistically significantly lower, and TF, MDA, and vitamin E were statistically significantly higher. On the other hand, in the hypercholesterolemia group, significant differences in all parameters, except AOPP, were observed, which demonstrates the importance of lipid metabolism in the pathogenesis of glaucoma. This is also consistent with previous findings [21]. Additionally, the most significant and consistent results were observed for all parameters in the levels of vitamin E. However, in intergroup comparisons between MDA and TADA, as well as with the control group, the most frequently encountered significant elevations were noted for oxidative stress indicators.

When compared with the control group, vitamin E was significantly elevated in all patients except for those with normal corneal thickness, lower Laplace scores, normal c/d, px(-), and hypoproteinemia. However, lipid-soluble antioxidant vitamin A only exhibited extremely higher levels in patients with higher Laplace scores. These results emphasize vitamin E as an antioxidant and a neurohormone.

It is recognized that oral or parenteral vitamin E accumulates more frequently inside the retina [26]. A study conducted on human eyes reported that retinal vitamin E levels were higher than those found in choroidal and vitreal tissues, and were directly proportional to serum vitamin E levels [27]. In animal trials, oral or parenteral administration of 100 mg/kg dl-a-tocopheril acetate were reported to similarly induce 3-6 fold increases in plasma levels, while after oral intake, retinal and vitreal levels were achieved later [28]. Apart from other antioxidants, vitamin E contains sensitive mechanisms for the regulation of tissue surface. Discovery of the tocopherol transfer protein, specific membrane receptors, and cytosolic transfer proteins, has suggested that this molecule might have other functions apart from its antioxidant effects. Indeed, many in vivo and in vitro studies performed in normal and neoplastic cells have demonstrated that  $\alpha$ -tocopherol had specific effects, including gene regulation [29]. We have specific evidence that this phenomenon also applies to ocular tissues [30]. Again, it has been established that some vitamin E derivatives act as neurohormones, and initiate various intracellular conduction pathways with a lock -and-key model. Cell culture and animal studies have confirmed that, among these specific effects, those on PKC were unique for PKC, and closer isomers (e.g.,

Parameters	Groups	TADA	AOPP	SOD	GPX	MDA	SER	TF	VITA	VITE
MD	N $(n=107)$	$1.64 \pm 0.1$	$31\pm 25.18$	$3.03 \pm 0.85$	$1.9 \pm 0.73$	$1.31 \pm 0.28$	$36.8 \pm 15.02$	266.7±63.2	$0.88 \pm 0.54$	20±8i77
	U $(n=27)$	$1.59 \pm 0.16$	$48.05\pm 24.16$	$3.22 \pm 0.9$	$2.6 \pm 0.68$	$1.35 \pm 0.21$	$40\pm 13.68$	278.08±64.49	$1.61 \pm 0.76$	26.7±22.97
	B (n=26)	$1.62 \pm 0.12$	$26.25 \pm 9.26$	$3.22 \pm 0.79$	$1.88 \pm 0.48$	$1.22 \pm 0.3$	42.4±19.62	$283.73\pm58.29$	$0.84 \pm 0.41$	$20.51 \pm 8.58$
PSD	>10 (n=112)	$1.61 \pm 0.12$	$24.77 \pm 10.36$	$3.29 \pm 0.91$	$2.07\pm0.49$	$1.37 \pm 0.24$	$39.39 \pm 13.5$	289.71±70.67	$0.95 \pm 0.68$	22.64±17.92
	<10 (n=48)	$1.64 \pm 0.1$	34.6±27.02	$3.03 \pm 0.82$	$1.97 \pm 0.78$	$1.28 \pm 0.29$	38.07±15.48	$263.43\pm58.28$	$0.9 \pm 0.54$	$20.49 \pm 9.04$
Corneal thickness	H (n=32)	$1.63 \pm 0.06$	32.2±12.69	$2.92 \pm 0.66$	$1.72 \pm 0.75$	$1.36 \pm 0.3$	$26.72 \pm 3.14$	238.2±69.67	$0.64 \pm 0.21$	$15.48\pm 1.26$
	N (n=68)	$1.64 \pm 0.1$	29.65±17.47	$3.3\pm0.72$	$2.07 \pm 0.8$	$1.24 \pm 0.28$	$39.8 \pm 14.91$	$262.04 \pm 52.67$	$0.86 \pm 0.58$	$21.61 \pm 18.8$
	L (n=60)	$1.61 \pm 0.12$	$33.35 \pm 31.53$	$3.02 \pm 0.81$	$1.83 \pm 0.76$	$1.22 \pm 0.29$	$31.13\pm 14.23$	262.71±52.24	$0.98 \pm 0.59$	$20.44 \pm 9.3$
Laplace value	I(n=76)	$1.64 \pm 0.07$	$31.01 \pm 19.18$	$3.03 \pm 0.55$	$1.82 \pm 0.43$	$1.32 \pm 0.31$	$36.54\pm21.05$	266.22±77.98	$0.79 \pm 0.32$	$18.18 \pm 7.79$
	D (n=84)	$1.68 \pm 0.08$	$36.25\pm 24.13$	<b>3.65±0.86</b>	2.66±0.58	$1.19 \pm 0.1$	40.47±11.42	$289.11 \pm 44.84$	$1.55 \pm 0.74$	31.05±10.95

TABLE 6. VARIABILITY OF SERUM OXIDATIVE STRESS MARKERS BASED ON VISUAL FIELD, PACHYMETRIC PARAMETERS AND LAPLACE SCORES.

dialdehyde, Ser: Serine, TF: transferrine, Vit A: vitamin A, Vit: E: vitamin E, H: Very high I: Increased, D: Decreased, N: Normal, n: number.

Parameters	Groups	TADA	AOPP	SOD	Gpx	MDA	Ser	TF	Vit A	Vit E
MD	Z	*		*		* *	*			**
	N					* *	*		*	* *
	В									* * *
PSD	>10	*		* *		* * *		*	*	* *
	<10	*				*	*		*	* *
Corneal thickness (µm)	Н					*				*
	Z									
	Γ	* *							*	* *
Laplace value	Ι	*						*	* *	***

": p<0.001 extremely significant. \*: p<0.005 very significant, significant,\*

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 $\beta$ -tocopherol), and distinct antioxidants (e.g., probucol), lacked these effects [31]. Various trials have reported that retinal vascular dysfunction secondary to hyperglycemia is corrected by  $\alpha$ -tocopherol through the DAG-PKC pathway [32].

Regarding glaucoma, the PKC pathway also has significant effects on non-vascular ocular smooth muscles, including trabecular meshork of the eye [33], PGF<sub>2a</sub>, and matrix metalloproteinases [34]. In addition, some publications have stated that glutamine transporter activity, which plays a key role in neurodegeneration, is regulated by PKC [35]. Retinal blood flow regulating and neuroprotective effects of  $\alpha$ -tocopherol in glaucoma patients have been clinically demonstrated [10]. Moreover,  $\alpha$ -tocopherol is recognized for prolonging life span in retinal cell cultures [36]. Taking all of these issues into consideration, significant and consistent elevation manifested by vitamin E, which has a special role among antioxidants [8] in glaucoma patients, is very important.

Elevations in Ser in the hyperproteinemia group were found to be extremely significant. Ser is an aminoacid that plays an important role in the antioxidant and neuromodulator effects of vitamin E [37]. Another important point is that PKC, which interacts with  $\alpha$ -tocopherol, is a type of Ser/Treonin kinase [38].

Another elevated liposoluble vitamin was vitamin A. Similar to tocopherols, vitamin A derivatives are also known to be neuromodulators in the retina [39], to improve endothelial function by reducing the concentration of reactive oxygen species in the vessel wall [40], and possess specific receptors in the nervous system [41].

In addition, Vitamin E and MDA elevations observed in the px(+) group (which was more significant than that of px(-)group, but meaningful relative to the control group) are consistent with the literature. The study performed by Yılmaz et al., which demonstrated that MDA levels in cataract patients with px were higher when compared with other forms of cataract, has been among trials indicating that an ocular pathology could affect serum oxidative stress markers [42].

In this study, clinical findings on the effects of glaucoma on oxidative stress indicators have been demonstrated to shed light on the pathogenesis of glaucoma. The association between glaucoma and lipid oxidation has been revealed as well. Another point to be emphasized is that, in addition to being an antioxidant, vitamin E has unique neurohormonelike activities and regulatory mechanisms, and its serum levels increase in conjunction with the severity of clinical findings of glaucoma. Further studies are needed to conclude that this molecule is the sole predictor of glaucoma. However, based on the results of this study, the significance of vitamin E as a neuroprotective agent with neurohormone-like activities, independent to being an antioxidant, has been revealed once more.

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## REFERENCES

- Goldberg I. How common is glaucoma worldwide? Weinreb RN, Kitazawa Y, Krieglstein GK, editors. 1st ed. Glaucoma in the 21st century. Harcourt Health Communications (London): Mosby Int Ltd; 2000. p. 1-9.
- Gupta N, Yücel YH. Glaucoma as a neurodegenerative disease. Curr Opin Ophthalmol 2007; 18:110-4. [PMID: 10416754]
- Osborne NN, Ugarte M, Chao M, Chidlow G, Bae JH, Wood JP, Nash MS. Neuroprotection in relation to retinal ischemia and relevance to glaucoma. Surv Ophthalmol 1999; 43:S102-28. [PMID: 10416754]
- He Y, Leung KW, Zhang YH, Duan S, Zhong XF, Jiang RZ, Peng Z, Tombran-Tink J, Ge J. Mitochondrial complex I defect induces ROS release and degeneration in trabecular meshwork cells of POAG patients: protection by antioxidants. Invest Ophthalmol Vis Sci 2008; 49:1447-58. [PMID: 18385062]
- Tezel G. Oxidative Stress in Glaucomatous Neurodegeneration: Mechanisms and Consequences. Prog Retin Eye Res 2006; 25:490-513. [PMID: 16962364]
- Kalousova M, Zima T, Tesar V, Dusilová-Sulková S, Skrha J. Advanced glycoxidation end products in chronic diseasesclinical chemistry and genetic background. Mutat Res 2005; 579:37-46. [PMID: 16084533]
- Elliott WH, Elliott DC. Enzymic protective mechanisms in the body. In: Elliott WH, Elliott DC editors. Biochemistry and molecular biology. New York: Oxford University press; 1997. p. 213–220.
- Engin KN. Alpha Tocopherol: Looking beyond an antioxidant. Mol Vis 2009; 15:855-60. [PMID: 19390643]
- Quaranta L, Bettelli S, Uva MG, Semeraro F, Turano R, Gandolfo E. Effect of Ginkgo biloba extract on preexisting visual field damage in normal tension glaucoma. Ophthalmology 2003; 110:359-62. [PMID: 12578781]
- Engin KN, Engin G, Kücükşahin H, Oncu M, Engin G, Guvener B. Clinical evaluation of the neuroprotective effect of αtocopherol on retina against glaucomatous damage. Eur J Ophthalmol 2007; 17:528-33. [PMID: 17671926]
- Chung HS, Harris A, Kristinsson JK, Ciulla TA, Kagemann C, Ritch R. Ginkgo biloba extract increases ocular blood flow velocity. J Ocul Pharmacol Ther 1999; 15:233-40. [PMID: 10385132]
- Yücel YH, Zhang Q, Weinreb RN, Kaufman PL, Gupta N. Effects of retinal ganglion cell loss on magno-, parvo-, koniocellular pathways in the lateral geniculate nucleus and visual cortex in glaucoma. Prog Retin Eye Res 2003; 22:465-81. [PMID: 12742392]
- Gupta N, Ang LC, Noël de Tilly L, Bidaisee L, Yücel YH. Human glaucoma and neural degeneration in intracranial optic nerve, lateral geniculate nucleus, and visual cortex. Br J Ophthalmol 2006; 90:674-8. [PMID: 16714257]
- Duncan RO, Sample PA, Weinreb RN, Bowd C, Zangwill LM. Retinotopic organization of primary visual cortex in

glaucoma: Comparing fMRI measurements of cortical function with visual field loss. Prog Retin Eye Res 2007; 26:38-56. [PMID: 17126063]

- He Y, Ge J, Tombran-Tink J. Mitochondrial defects and dysfunction in calcium regulation in glaucomatous trabecular meshwork cells. Invest Ophthalmol Vis Sci 2008; 49:4912-22. [PMID: 18614807]
- Alías EG, Ferreras A, Polo V, Larrosa JM, Pueyo V, Honrubia FM. Importance of central corneal thickness when studying ocular hypertensive eyes, glaucoma suspects and preperimetric glaucomatous eyes. Arch Soc Esp Oftalmol 2007; 82:615-22. [PMID: 17929204]
- Szczudlowski K. Glaucoma hypothesis: application of the law of Laplace. Med Hypotheses 1979; 5:481-6. [PMID: 459994]
- Taysi S, Kocer I, Memisogullari R, Kiziltunc A. Serum Oxidant/Antioxidant Status in Patients with Behçet's Disease. Ann Clin Lab Sci 2002; 32:377-82. [PMID: 12458889]
- Kłos-Rola J, Zagórski Z. Peroxidation of lipids in patients with senile cataract. Klin Oczna 2004; 106:416-8. [PMID: 15636220]
- Bunin AI, Filina AA, Erichev VP. A glutathione deficiency in open-angle glaucoma and the approaches to its correction. Vestn Oftalmol 1992; 108:13-5. [PMID: 1295181]
- Ren H, Magulike N, Ghebremeskel K, Crawford M. Primary open-angle glaucoma patients have reduced levels of blood docosahexaenoic and eicosapentaenoic acids. Prostaglandins Leukot Essent Fatty Acids 2006; 74:157-63. [PMID: 16410047]
- Stitt AW, Frizzell N, Thorpe SR. Advanced glycation and advanced lipoxidation: possible role in initiation and progression of diabetic retinopathy. Curr Pharm Des 2004; 10:3349-60. [PMID: 15544520]
- 23. Packer L. Protective role of Vitamin E in biological systems. Am J Clin Nutr 1991; 53:10508-58. [PMID: 2012017]
- Popov B, Gadjeva V, Valkanov P, Popova S, Tolekova A. Lipid peroxidation, superoxide dismutase and catalase activities in brain tumor tissues. Arch Physiol Biochem 2003; 111:455-9. [PMID: 16026034]
- Kurysheva NI, Vinetskaia MI, Erichev VP, Demchuk ML, Kuryshev SI. Contribution of free-radical reactions of chamber humor to the development of primary open-angle glaucoma. Vestn Oftalmol 1996; 112:3-5. [PMID: 9019910]
- Nagata M, Kawazu K, Midori Y, Kojima M, Shirasawa E, Sasaki K. Intracameral and lenticular penetration of locally applied stable isotope-labeled vitamin E. Jpn J Ophthalmol 2001; 45:125-7. [PMID: 11313042]
- Bhat R. Serum, retinal, choroidal vitreal Vitamin E concentrations in human infants. Pediatrics 1986; 78:866-70. [PMID: 3763301]
- Bhat R, Raju T, Barrada A, Evans M. Disposition of Vitamin E in the eye. Pediatr Res 1987; 22:16-20. [PMID: 3627865]

- Traber MG, Packer L. Vitamin E beyond antioxidant function. Am J Clin Nutr 1995; 62:1501s-9s. [PMID: 7495251]
- Alvarez RA, Liou GI, Fong SL, Bridges CD. Levels of α- and γ-tocopherol in human eyes: evaluation of the possible role of IRBP in intraocular α-tocopherol transport. Am J Clin Nutr 1987; 46:481-7. [PMID: 3630966]
- 31. Özer NK, Şirikçi Ö, Taha S, Engin KN, Boscobionik D, Clement S, Stocker A, Azzi A. Prevention of atherosclerosis by α-tocopherol in smooth muscle cells by a mechanism involving signal transduction modulation. In: Özben T, editors. Free radicals, oxidative stress and antioxidants. New York: Plenum press; 1998. p. 333–42.
- 32. KunisakiMBursellSClermontACIshiiHBallasLMJirousekMR UmedaFNawataHKingGLVitamin E prevents diabetesinduced abnormal retinal blood flow via the diacylglycerolprotein kinase C pathway.Am J Physiol1995269E23946 7653541
- Wiederholt M, Thieme H, Stumpff F. The regulation of trabecular meshwork and ciliary muscle contractility. Prog Retin Eye Res 2000; 19:271-95. [PMID: 10749378]
- Alexander JP, Acott TS. Involvement of protein kinase C in TNFalpha regulation of trabecular matrix metalloproteinases and TIMPs. Invest Ophthalmol Vis Sci 2001; 42:2831-8. [PMID: 11687525]
- Bull ND, Barnett NL. Antagonists of protein kinase C inhibit rat retinal glutamate transport activity in situ. J Neurochem 2002; 81:472-80. [PMID: 12065656]
- Rego AC, Santos MS, Proenca MT, Oliveira CR. Influence of Vitamin E succinate on retinal cell survival. Toxicology 1998; 128:113-24. [PMID: 9710152]
- Wang X, Fan Z, Wang B, Luo J, Ke ZJ. Activation of doublestranded RNA-activated protein kinase by mild impairment of oxidative metabolism in neurons. J Neurochem 2007; 103:2380-90. [PMID: 17953670]
- Carter CA, Kane CJ. Therapeutic potential of natural compounds that regulate the activity of protein kinase C. Curr Med Chem 2004; 11:2883-902. [PMID: 15544481]
- Weiler R, Pottek M, Schultz K, Janssen-Bienhold U. Retinoic acid, a neuromodulator in the retina. Prog Brain Res 2001; 131:309-18. [PMID: 11420951]
- Duvall WL. Endothelial dysfunction and antioxidants. Mt Sinai J Med 2005; 72:71-80. [PMID: 15770336]
- Bremner JD, McCaffery P. The neurobiology of retinoic acid in affective disorders. Prog Neuropsychopharmacol Biol Psychiatry 2008; 32:315-31. [PMID: 17707566]
- Yimaz A, Adigüzel U, Tamer L, Yildirim O, Oz O, Vatansever H, Ercan B, Değirmenci US, Atik U. Serum oxidant/ antioxidant balance in exfoliation syndrome. Clin Experiment Ophthalmol 2005; 33:63-6. [PMID: 15670081]

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