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# **Potassium trifluoroborate salts as convenient, stable reagents for difficult alkyl transfers**

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# **1. Introduction**

The ability to incorporate alkyl groups selectively and conveniently into molecular substructures at a late stage is of critical strategic value in complex molecule synthesis. In the quest for a universal alkylating platform, various nucleophilic organometallic reagents have surfaced that are capable of facilitating such desired transformations [i]. However, the highly reactive nature of many of these reagents limits their use for polyfunctional molecules. The Suzuki-Miyaura cross-coupling reaction has emerged as one of the most powerful means to effect this type of transformation [ii]. The organoboron reagents employed in this reaction can be prepared readily using a variety of routes and have a high compatibility with a broad range of functional groups. The reaction is also particularly advantageous because, unlike the byproducts generated in other cross-coupling bond strategies, those generated via the Suzuki-Miyaura cross-coupling reaction are relatively benign inorganic salts that can be removed via an aqueous wash. As a testament to its effectiveness, strategies using the Suzuki-Miyaura cross-coupling reaction for the transfer of alkyl groups have been employed in the syntheses of various natural products and biologically significant analogues [iii].

The Suzuki-Miyaura reaction most often employs trialkylboranes as the nucleophilic partner because they are easily accessed via hydroboration reactions of the corresponding alkenes with dialkylboranes (i.e., 9-BBN, Chx<sub>2</sub>BH, Sia<sub>2</sub>BH). Many effective protocols have been established and optimized for the cross-coupling of trialkylboranes with various types of electrophilic partners. However, the reaction scope is limited by the incompatibility of these dialkylborane hydroborating agents with a variety of functional groups. The air- and

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The past decade has witnessed the emergence of potassium organotrifluoroborates as a new class of nucleophilic boron reagents for the Suzuki-Miyaura cross-coupling and Rh-catalyzed addition reactions. These substrates are easily prepared and most of them are indefinitely stable to air and moisture. Recent advances have focused on the utility of alkyltrifluoroborates in appending alkyl groups selectively and conveniently onto appropriate molecular substructures. Strategies employing these reagents are described herein.

moisture sensitivity of the resulting trialkylborane species requires them to be prepared and used *in situ*, making small-scale optimization an arduous task, in turn limiting their use in synthetic sequences.

Alkylboronic acids and alkylboronate esters can also be prepared and employed as the nucleophilic coupling partners. Reactions involving the cross-coupling of alkylboronic acids suffer from competitive protodeboronation, and as a result, significant excesses of the boronic acids are often employed to ensure complete consumption of the electrophilic coupling partner [iv]. One report details the preformation and isolation of an "activated borate" by treating an alkylboronic acid with sodium hydroxide. Once prepared, the alkyl trihydroxyborate salt is able to undergo a base-free cross-coupling with an aryl bromide, but only a moderate yield is obtained even with the use of an excess of the activated species [v].

Alkylboronate esters have a decreased propensity for protodeboronation and thus can also serve as suitable boron reagents. Their enhanced stability, however, comes at the cost of sluggish reactivity in many cases, as the use of these compounds generally leads to low yields unless highly toxic thallium bases (TIOH or  $Tl_2CO_3$ ) are employed to facilitate the transmetalation step [vi]. A single report demonstrates a high yielding alkyl transfer reaction using a boronate ester without the use of a thallium base, but treatment *in situ* with *sec*butyllithium was required to generate a lithium *n*-alkylborate reagent, the activated form required for transmetalation [vii].

Most notably, the aforementioned classes of organoboron compounds are all tricoordinate species, each possessing an empty p-orbital on boron, leaving them susceptible to reactions with important classes of reagents such as organic acids, oxidants, bases, and nucleophiles. As a result, these tricoordinate alkylborons are most often prepared and used directly, rather than carried though synthetic steps.

Since the discovery of the Suzuki-Miyaura cross-coupling reaction, much of the research has focused on optimization of the metal/ligand catalyst systems incorporating the classes of organoboron reagents outlined above [4,viii]. These studies have also incorporated the use of additives to facilitate product formation [ix]. Although important contributions have been made through these efforts, until recently little consideration had been given to modifying the organoboron reagent, which could lead to a more stable nucleophilic partner and potentially enhanced levels of reactivity.

Throughout the past decade, organotrifluoroborates have emerged as alternative nucleophilic partners in Suzuki-Miyaura cross-coupling [x,xi]. Their tetracoordinate nature masks the inherent reactivity of the C-B bond, making them protected forms of boronic acids that can endure a variety of synthetic transformations. Their reactivity can subsequently be unveiled under the same conditions required for cross-coupling.

Potassium alkyltrifluoroborates are easily prepared from commercially available, inexpensive starting materials {e.g., via hydroboration of the corresponding alkenes [xii,xiii,xiv,xv], transmetalation from other organometallics [xvi], or metalation reactions [xvii], followed by treatment with inexpensive potassium hydrogen fluoride  $(KHF_2)$  [xviii]]. Once prepared, most of them are indefinitely stable to air and moisture (Scheme 1).

These traits, partnered with their demonstrated ability to undergo transmetalation with limited interference from competitive protodeboronation, reinforce their value as appealing reagents for alkyl transfer reactions.

# **2. Cross-Coupling of Primary Alkyltrifluoroborates with Aryl- and Alkenyl Electrophiles**

Various studies have focused on the incorporation of alkyl groups into aryl- or alkenyl compounds via the Suzuki-Miyaura cross-coupling reaction. The earliest reports used alkylboronic acids as the nucleophilic partner and commonly resulted in very low yields [xix]. Later studies by Falck and coworkers demonstrated vast improvements in the crosscoupling of alkylboronic acids, revealing high yields of the cross-coupled product. However, expensive silver oxide (2.5 equiv) was required as an additive to facilitate product formation [9]. More recently, the cross-coupling of alkylboronic acids with aryl halides and -triflates has been demonstrated using  $PdCl<sub>2</sub>(dppf) \cdot CH<sub>2</sub>Cl<sub>2</sub>$  and  $Cs<sub>2</sub>CO<sub>3</sub> [xx]$ , but the known instability of boronic acids prompted investigations into the use of the highly crystalline and indefinitely stable organotrifluoroborate salts as the nucleophilic partner.

#### **2.1 Cross-Coupling of Primary Alkyltrifluoroborates with Aryl- and Alkenyl Electrophiles**

Primary alkyltrifluoroborates can be accessed via the addition of alkyl Grignard reagents to trimethyl borate and by the hydroboration of alkenes employing several hydroborating protocols, followed in both cases by treatment with KHF<sub>2</sub>.

Initial investigations focused on the cross-coupling of these alkyltrifluoroborates with aryland 1-alkenyl trifluoromethanesulfonates (Tf). This study illustrated that  $Cs_2CO_3$  (3 equiv) was the most effective base to promote the transformation when used in combination with 9 mol % of PdCl<sub>2</sub>(dppf)·CH<sub>2</sub>Cl<sub>2</sub> and a THF/H<sub>2</sub>O solvent system. These optimized conditions were also found to be suitable for the cross-coupling of alkylboronic acids and -esters with aryl electrophiles [xxi]. A subsequent study utilizing these same conditions expanded the scope of the cross-coupling reaction of alkyltrifluoroborates to include aryl bromides as the electrophilic partner [xxii] (Scheme 2).

These conditions were also employed in the cross-coupling of oxiranylethyltrifluoroborate with 4-bromobenzonitrile. However, in this case a base-catalyzed epoxide opening occurred, and the cross-coupled product was isolated as a diol [xxiii]. Simply reducing the amount of water in the solvent mixture to a 40:1 THF/H<sub>2</sub>O mixture allowed the oxiranyl product to be obtained (Scheme 3).

A more recent account was published detailing the cross-coupling of potassium alkyltrifluoroborate substrates with aryl chlorides [xxiv]. Using a parallel microscale experimentation approach, optimal conditions of 2 mol % of  $Pd(OAc)_2$  and 4 mol % of RuPhos (2-dicyclohexylphosphino-2',6'-diisopropoxybiphenyl) in combination with  $K_2CO_3$  $(3$  equiv) and a toluene/H<sub>2</sub>O solvent system were identified that were capable of crosscoupling a wide variety of aryl- and heteroaryl chloride substrates with alkyltrifluoroborates. This finding with chloride electrophiles represents an important extension of the alkyltrifluoroborate cross-coupling method because it allows the use of more stable and less expensive electrophilic partners to be employed. These newly optimized conditions were also found to be amenable to bromides, iodides, and triflates. Of particular note is the tolerance of a silyl ether derivative through not only the preparation of the trifluoroborate (with exposure to a fluoride source  $-KHF_2$ ), but also the cross-coupling to 4-chloroanisole (Scheme 4).

The cross-coupling of potassium primary alkyltrifluoroborates has also been demonstrated with alkenyl bromides [xxv]. Using 10 mol % of  $PdCl<sub>2</sub>(dppf) \cdot CH<sub>2</sub>Cl<sub>2</sub>$  and  $Cs<sub>2</sub>CO<sub>3</sub>$  (3 equiv) and a toluene/H2O solvent system, various alkenyl bromides underwent conversion to generate the cross-coupled products in good yields. Using these conditions, a dihydroxylated

potassium alkyltrifluoroborate was also found to be a suitable nucleophilic partner with alkenyl bromides [xxvi] (Scheme 5).

A one-pot synthesis of trisubstituted conjugated dienes was also demonstrated via sequential Suzuki-Miyaura cross-coupling reactions [xxvii]. A *gem*-dibromide could be partnered with an alkenyl trifluoroborate in the presence of 7 mol % of Pd(PPh<sub>3</sub>)<sub>4</sub> and Cs<sub>2</sub>CO<sub>3</sub> followed by reaction with a variety of alkyltrifluoroborates to furnish trisubstituted alkenes in excellent yields (Scheme 6).

#### **2.2 Potassium Methyltrifluoroborate**

In addition to the development of general methods to append primary alkyl groups onto organic molecules, strategies to incorporate specific organic moieties, particularly methyl groups, are of importance. Historically, methylation strategies using organometallics have proved to be less than ideal. One report details the cross-coupling of methylboronic acid, but necessitates the use of 40 mol % of triphenylarsine [xxviii]. In a more recent publication, a methylation procedure was described using methylboronic acid and aryl bromides with sodium dodecyl sulfate (SDS) [xxix]. Another means of introducing methyl groups into systems is through the use of trimethylboroxine (the cyclic trimer of methyl boronic acid). However, significant excesses of these reagents are often required (3 equiv of the boroxine) [xxx].

Molander and coworkers have reported the cross-coupling of potassium methyltrifluoroborate with bromide and triflate electrophiles using 9 mol % of  $PdCl<sub>2</sub>(dppf)$  $\cdot$ CH<sub>2</sub>Cl<sub>2</sub> or with chlorides using a combination of 2 mol % of Pd(OAc)<sub>2</sub> and 4 mol % of RuPhos to yield the methylated compounds in good yields [22,24]. Another recent publication outlined the cross-coupling of halopurines with potassium methyltrifluoroborate to generate substituted purines, which represent an important class of biologically active compounds [xxxi]. Notably, the use of  $Pd(OAc)_2$  and trisodium triphenylphosphine-3,3',3"trisulfonate (TPPTS) under microwave conditions facilitated the cross-coupling of potassium methyltrifluoroborate with 6-chloropurine (Scheme 7).

# **3. Cross-Coupling of Secondary Alkyltrifluoroborates with Aryl Electrophiles**

Beyond the exceptional reactivity of cyclopropylboron derivatives (which can be rationalized by the partial  $sp^2$  character these compounds possess), there has been little precedent for the cross-coupling of secondary alkylboron derivatives [8,xxxii,xxxiii]. In the few published examples, there was no significant development toward identifying a general solution for this significant, unsolved problem. The difficulty in this cross-coupling reaction derives from two key steps of the mechanistic cycle – the transmetalation that is more difficult for secondary organoboron derivatives than other organic moieties, and the reductive elimination step, which competes with β-hydride elimination.

#### **3.1 Cyclopropylation**

Incorporating cyclopropyl groups into complex molecules is becoming increasingly important owing to the prevalence of cyclopropyl-substituted natural products and biologically active molecules. The ease of preparation of cyclopropylboron compounds makes the Suzuki-Miyaura cross-coupling reaction an attractive cyclopropylating platform. Much of the research in this area has focused on the cross-coupling of cyclopropylboronic acid. However, the propensity of this reagent to protodeboronate upon prolonged storage renders it unstable and unsafe. The trifluoroborate analogs serve as more suitable alternatives because they are more resistant to protodeboronation, and thus efforts in this

area focused on the preparation and cross-coupling of various cyclopropyltrifluoroborates with aryl halides.

Stereodefined substrates were first prepared from alkenylboronic acids through a sequence of reactions in which the pinacol boronates were generated in situ and treated with  $CH_2N_2$  in the presence of  $Pd(OAc)_{2}$ . Subsequent treatment with  $KHF_{2}$  afforded the cyclopropyltrifluoroborates[xxxiv]. These substrates were subsequently cross-coupled with aryl bromides using 2 mol % of Pd(PPh<sub>3</sub>)<sub>4</sub> and K<sub>3</sub>PO<sub>4</sub>·3H<sub>2</sub>O (3 equiv) in a toluene/H<sub>2</sub>O solvent system, in each case retaining the configurations of the cyclopropane stereocenters during the cross-coupling event. A subsequent report detailed the synthesis of 1,2,3-*syn*-*cis*substituted potassium cyclopropyltrifluoroborates from allylic alcohols, a readily prepared *gem*-dizinc carbenoid, and trimethyl borate with subsequent addition of KHF<sub>2</sub> [xxxv]. Using 3 mol % of  $Pd(OAc)_2$  and 6 mol % of CyJohnPhos (2-biphenyldicyclohexylphosphine),  $K_3PO_4$  (3.3 equiv) and a toluene/H<sub>2</sub>O solvent system, these substrates could be crosscoupled with a variety of aryl bromides (Scheme 8).

A later report described the preparation potassium cyclopropyltrifluoroborate by treating the corresponding boronic acid with KHF<sub>2</sub> [xxxvi]. Using 3 mol % of Pd(OAc)<sub>2</sub> and 6 mol % of XPhos (2-dicyclohexylphosphino-2',4',6'-triisopropylbiphenyl),  $K_2CO_3$  (3 equiv) and a cyclopentyl methyl ether (CMPE)/H2O solvent system, the organotrifluoroborate could be cross-coupled with both aryl- and heteroaryl chlorides.

Additionally, the use of Pd(OAc)<sub>2</sub> and di(1-adamantyl)-*n*-butyl-phosphine (*n*-BuPAd<sub>2</sub>) facilitated the cross-coupling of potassium cyclopropyltrifluoroborate and 6-chloropurine to generate the cyclopropyl-substituted product in good yield [31] (Scheme 9).

#### **3.2 Cross-Coupling of Other Secondary Alkyltrifluoroborate Substrates**

The earliest development in this area using organotrifluoroborates is reported via a RuPhosmediated cross-coupling of aryl- and heteroaryl bromides with potassium cyclopentyltrifluoroborate [xxxvii]. Although the yields were moderate, this clearly demonstrated the first example of a cross-coupling reaction with a secondary alkyltrifluoroborate. A few examples with *sec*-butyltrifluoroborate were also described, in each case generating the secondary alkylated derivatives in moderate yields (Scheme 10).

At the same time, another account was published describing the cross-coupling of potassium secondary alkyltrifluoroborates with a variety of aryl- and heteroaryl chlorides and bromides. The optimized reaction conditions of  $Pd(OAc)$ <sub>2</sub> and *n*-BuPAd<sub>2</sub> were identified through a microscale parallel experimentation screen and found to be suitable for cyclic, symmetrical trifluoroborate derivatives [xxxviii]. Additionally, a strategy was identified using these conditions to incorporate cyclobutyl rings into molecules via the cross-coupling of cyclobutyltrifluoroborate [36]. These conditions proved to be somewhat substrate dependent when applied to this system. However, various aryl- and heteroaryl chlorides were shown to be suitable electrophiles, providing the aryl cyclobutane products (Scheme 11).

When these conditions were applied to secondary acyclic derivatives, such as potassium isopropyltrifluoroborate, a β-hydride elimination and subsequent isomerization occurred to generate a mixture of the desired branched product to an undesired straight chain product. Subsequent investigations showed that more sterically hindered ligands, *t*-Bu<sub>2</sub>PPh and *t*-Bu3P, were capable of inhibiting isomerization to generate higher ratios of secondary alkylated products to the undesired straight-chain products [38].

# **4. Cross-Coupling of Specialized Alkyltrifluoroborate Reagents with Aryl Electrophiles**

In addition to the cross-couplings of primary and secondary alkyltrifluoroborates, the crosscouplings of several specialized alkyltrifluoroborates with aryl halides have also been described.

#### **4.1 Aminoethyltrifluoroborates**

Preliminary investigations in this area focused on the preparation of βaminoethyltrifluoroborates via the Sniekus hydroboration with di(isopropylprenyl)borane [(*i*-PP)<sub>2</sub>BH] [xxxix]. Once prepared, the cross-coupling of these substrates with a variety of aryl- and heteroaryl bromides was accomplished using  $PdCl<sub>2</sub>(dppf) \cdot CH<sub>2</sub>Cl<sub>2</sub>$ . This method provided a convenient way to prepare a variety of phenethylamines.

A subsequent publication described advances in the scope of this aminoethylation reaction, now extending the preparation of the aminoethyltrifluoroborate partner to enecarbamate substrates, including Boc and Cbz protected derivatives [xl]. Investigations of substrate scope with the Cbz carbamate derivative and electron-poor aryl electrophiles demonstrated that the originally optimized conditions for aminoethylations were suitable for the crosscoupling. However, preliminary experiments with electron-rich electrophiles failed to provide the desired cross coupled products in satisfactory yields, demonstrating the need for further optimization. The utilization of a catalytic system of  $Pd(OAc)_2$  and RuPhos offered significantly improved results, and also proved to be an appropriate catalyst component for the cross coupling with a variety of heteroaryl bromides as well. The Boc-protected aminoethyltrifluoroborate derivative also cross-coupled with equal facility. Once crosscoupled, these trifluoroborate substrates offer the potential for further functionalization and incorporation of the phenethylamine subunit into complex molecules (Scheme 12).

#### **4.2 Aminomethyltrifluoroborates**

In addition to the importance of installing aminoethyl subunits into molecules, there are a variety of biologically active molecules that possess aminomethyl aryl linkages as well. There is ample precedent for the installation of this subunit via other bond-forming techniques (nucleophilic substitution, reductive amination, alkylation of iminium ions, etc.). However, a complementary cross-coupling strategy has been employed only to a limited extent to introduce this subunit. A singular example existed that demonstrated the crosscoupling of a highly specialized aminomethylorganostannane, but the perceived toxicity of tin reagents makes the use of the Stille reaction a less than ideal aminomethylating platform  $[x<sub>li</sub>]$ .

A variety of *N*,*N*-dialkylaminomethyltrifluoroborates were prepared via reaction of secondary amines with bromomethyltrifluoroborate [xlii], and conditions involving 3 mol % of Pd(OAc)<sub>2</sub> and 6 mol % of XPhos were identified to be capable of facilitating the pairing of these substrates first with aryl- and heteroaryl bromides [xliii] and then further extended to include aryl chlorides [xliv]. In many cases, the results with aryl chlorides were actually superior to those obtained with bromides (Scheme 13).

The method was also extended to alkenyl electrophiles, albeit with limited success because the conditions optimized for this cross-coupling seemed to be substrate dependent. Nonetheless, these examples represent the possible utility of this reaction to incorporate an aminomethyl alkenyl subunit in complex molecule synthesis [44].

#### **4.3 Alkoxymethyltrifluoroborates**

A similar strategy was developed for the incorporation of an alkoxymethyl subunit into organic molecules [xlv]. Ethers are one of the most utilized protecting groups for alcohols and as a result, this strategy could provide a simplified synthesis of protected aryl- and heteroaryl alcohol derivatives, avoiding common multistep processes generally employed to access the compounds (i.e., carbonylation/carboxylation, reduction, and protection steps).

Alkoxymethylstannanes have been employed to a limited extent as hydroxymethylating agents, but the minimal use of this class of reagents is undoubtedly a result of the perceived toxicity of these reagents in combination with the known difficulty in purification after cross-coupling [xlvi]. The use of potassium alkoxymethyltrifluoroborates represents a more practical and convenient method of installing this moiety into organic molecules. Like their aminomethyl counterparts, these substrates were prepared via substitution reactions (in this case using nucleophilic alkoxides) with bromomethyltrifluoroborate as an electrophile. After an extensive optimization,  $Pd(OAc)$  and RuPhos were identified as a capable catalytic system for the cross-coupling of these substrates with aryl- and heteroaryl chlorides, generating the desired cross-coupled products in good yields (Scheme 14).

#### **4.4 3-Oxoalkyltrifluoroborates**

Zinc ketohomoenolates have been used in palladium catalyzed reactions with acid chlorides. However, prior to the work Molander and coworkers performed in this area, there was no precedent of zinc, tin, or boro homoenolates having been employed in cross-coupling reactions with aryl electrophiles. Initial investigations in this area dealt with the preparation of potassium 3-oxoalkyltrifluoroborates via conjugate addition of bis(pinacolato)diboron to unsaturated carbonyl compounds [xlvii] or by alkylation of commercially available methyl ketones (via lithium enolates) with iodomethylpinacol boronate [xlviii] followed in both cases by treatment with KHF2. Conditions were then identified allowing cross-coupling of the 3-oxoalkyltrifluoroborates with a variety of aryl bromides. Aryl chlorides and -triflates and an alkenyl bromide also served as suitable electrophilic partners in this reaction [xlix].

The scope and generality of this reaction was further investigated, and through the same procedures a family of trifluoroboratoketohomoenolates was prepared [l]. Although aliphatic ketohomoenolates were shown to cross-couple in good yields with a variety of aryl- and heteroaryl chlorides (Scheme 15), aromatic ketohomoenolates were significantly more challenging substrates to cross-couple.

A subsequent investigation extended the method to include the cross-coupling of potassium β-trifluoroborato amides with aryl- and heteroaryl chlorides (Scheme 16) [li].

This method could be applied to the synthesis of more highly functionalized molecules; in particular, functionalized dipeptides.

#### **4.5 4-Oxoalkyltrifluoroborates**

Another related class of compounds has recently emerged, prepared via the nickel-catalyzed borylation of aryl cyclopropyl ketones with bis(pinacolato)diboron [lii]. The resulting 4 oxoalkylboronates can be readily transformed into the corresponding trifluoroborate derivatives by treatment with KHF2. This 4-oxoalkyltrifluoroborate can be cross-coupled with 4-bromoanisole using 5 mol % of Pd(OAc)<sub>2</sub>, 10 mol % of RuPhos, and  $K_2CO_3$  in a toluene/H<sub>2</sub>O solvent system to provide the product in good yield (Scheme 17).

## **5. Rhodium-Catalyzed 1,2-Additions to Aldehydes**

There is ample precedent for the Rh-catalyzed 1,2- and 1,4-additions of organoboron compounds to aldehydes and Michael acceptors; however, the difficulty in transmetalation and problems with β-hydride elimination associated with alkylboron compounds has restricted the method to the addition of  $sp^2$ -hybridized species. Recently, Aggarwal and coworkers have circumvented these limitations and described the first Rh-catalyzed 1,2 addition of chiral secondary and tertiary potassium alkyltrifluoroborates to aldehydes [liii]. Using 2.5 mol %  $[RhCl(cod)]_2$  in a dioxane/H<sub>2</sub>O solvent system, a variety of chiral trifluoroborates were found to add to aldehydes with complete stereoretention (Scheme 18).

## **6. Conclusions**

Throughout the past decade, potassium alkyltrifluoroborates have emerged as exceptional reagents for difficult alkyl transfers. This class of organoboron reagents offers a superior shelf life and stability and in some cases enhanced reactivity compared to their tricoordinate organoboron analogs. Conditions have been identified for both primary (including trifluoroboratohomoenolates) and secondary trifluoroborate substrates for their crosscoupling with a variety of electrophilic species. Additionally, a number of specialized alkyltrifluoroborate substrates have been shown to cross-couple under various conditions to append aminomethyl-, aminoethyl-, and alkoxymethyl linkages to aryl electrophiles. The recent advances in Rh-catalyzed addition chemistry further strengthen the value of alkyltrifluoroborate reagents. The developments in this area represent an important progression in organometallic chemistry, because a number of alkyl groups can be readily transferred to increase structural complexity in highly functionalized molecules via C-C bond-forming reactions.

## **Abbreviationsp**



## **References**

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**Scheme 1.** Synthetic Routes to Alkyltrifluoroborates





Cross-Coupling of Primary Alkyltrifluoroborates with Triflate and Bromide Electrophiles



**Scheme 3.** Cross-Coupling of Epoxytrifluoroborates





Cross-Coupling of Primary Alkyltrifluoroborates with Aryl Chlorides











#### **Scheme 7.**

Methylation via Cross-Coupling of Potassium Methyltrifluoroborate



#### **Scheme 8.**

Cyclopropylation via Cross-Coupling with Retention of Configuration



**Scheme 9.** Cross-Coupling of Cyclopropyltrifluoroborate











#### **Scheme 12.**

Cross-Coupling of Aminoethyltrifluoroborates with Aryl- and Heteroaryl Electrophiles

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**Scheme 13.** Aminomethylations via Cross-Coupling Reactions



#### **Scheme 14.**

Cross-Coupling of Alkoxymethyltrifluoroborates with Aryl and Heteroaryl Electrophiles



**Scheme 15.** Cross-Coupling of Trifluoroboratoketohomoenolates with Various Electrophiles







**Scheme 17.** Cross-Coupling of 4-Oxoalkyltrilfuoroborate with 4-Bromoanisole



**Scheme 18.** Rh-Catalyzed 1,2-Addition of Secondary- and Tertiary Alkyltrifluoroborates to Aldehydes