

NIH Public Access

Author Manuscript

Tetrahedron. Author manuscript; available in PMC 2011 August 1.

Published in final edited form as:

Tetrahedron. 2010 August 1; 66(33): 6391–6398. doi:10.1016/j.tet.2010.04.094.

A Diels-Alder Route to Angularly Functionalized Bicyclic Structures

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Abstract

A Diels-Alder based route to *trans*-fused angularly functionalized bicyclic structures has been developed. This transformation features the use of a tetrasubstituted dienophile in the cycloaddition step.

1. Introduction

The development of improved methods for the stereoselective construction of substituted decalin and hydrindane systems remains an important focus of organic synthesis.¹ Over the years, our laboratory has been exploring extensions of the Diels-Alder reaction, with a view to gaining access to substructural patterns which are not traditionally filed under Diels-Alder logic. Toward this end, we recently developed a two-step "*trans*-Diels-Alder" paradigm, which provides access to *trans*-fused decalin and hydrindane systems from 1 nitrocycloalkene dienophiles (**2**) and simple dienes (**1**). As is shown in Figure 1, cycloaddition provides a *cis*-fused bicyclic adduct, bearing a nitro group in the ring junction (**3**). Subsequent radical denitration furnishes the target *trans*-fused system (**4**) with good selectivity.² In the case of the hydrindane series, the Diels-Alder step, per se, is even more straightforward; however, denitration results in nearly 1:1 mixtures of diastereomeric products.

To further expand upon this concept, we considered the possibility of achieving cycloaddition between a tetrasubstituted dienophile of the type 5 and a diene (1) .³ It was hoped that the *cis-*fused nitro-substituted Diels-Alder adduct, **6**, following radical-induced denitration, might progress to a *trans*-fused system bearing substitution (A) at the ring junction (cf. **7**). Alternatively, **6** could be converted to a *cis*-fused angular amine product (cf.

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Supplementary data

 1_H and 13_C NMR spectra of all new compounds are available. Crystallographic data for the structures reported in this article have been deposited with the Cambridge Crystallographic Data Centre as supplementary publication numbers CCDC-763857 (**13**), CCDC-763854 (**14**), CCDC-763855 (**19**), and CCDC-763856 (**25**). Copies of the data can be obtained free of charge at www.ccdc.cam.ac.uk/conts/retrieving.html or from the Cambridge Crystallographic Data Centre, 12, Union Road, Cambridge CB2 1EZ, UK [Tel: (+44)1223 336 408, Fax: (+44)1223 336 033 or deposit@ccdc.cam.ac.uk].

8) through reduction of the nitro functionality. We were not unmindful of the challenges inherent in this proposed sequence. Indeed, tetrasubstituted olefins are known to act as poor dienophiles in the Diels-Alder reaction,⁴ and relatively few examples of tetrasubstituted cyclic dienophiles bearing nitro functionality have been reported.⁵ To our knowledge, compound **5** had not been demonstrated to function as a competent Diels-Alder dienophile. We were hopeful that, by installing a second electron-withdrawing group (i.e. CN) as substituent A, we could realize the desired cycloaddition. We describe herein the development of two-step, Diels-Alder-based routes to *trans-*fused adducts containing angular substitution, such as **7**, and to disubstituted, *cis-*fused adducts of the type **8**. 6

2. Results and discussion

We first examined the cycloaddition-denitration sequence in the context of dienophile **11**⁷ (itself prepared in one step from **9**, as shown) and synergistic diene **10** (Scheme 2).⁸ In the event, **10** and **11** did undergo cycloaddition to afford a 1:1 (*exo*:*endo*) mixture of the readily separable adducts **12** and **13**. The structure of **13** (*endo*) was confirmed by X-ray crystallographic analysis. Each adduct was then separately subjected to conditions which favor radical-mediated denitration.⁹ Following hydrolysis, *trans*-fused adduct **14** was isolated as the predominant product; its structure was also unambiguously determined by Xray crystallography. The slightly higher levels of stereocontrol achieved in the denitration of *exo* cycloadduct **12** (12:1 vs. 10:1) can perhaps be attributed to the axial disposition of the methoxy group, which serves to more effectively shield the "front" face of the transient free radical, leaving the back face exposed to attack by the hydrogen-donor agent.

We next examined the two-step sequence in the context of a cyclopentenyl dienophile. Here, the tetrasubstituted dienophile, **17**, was prepared from compound **16**, as shown (Scheme 3). Under thermal conditions, **17** underwent cycloaddition with diene **10** to afford the *cis*-fused adduct, **18**, as a mixture of *endo* and *exo* isomers, in 95% yield. Upon denitration and subsequent hydrolysis, a 2:1 mixture of *trans-*fused **19** and *cis*-fused **20** was isolated. Obviously, this lack of stereocontrol would compromise the value of the method. We also examined the denitration and hydrolysis of hydrindane **22**, arising from the Diels-Alder reaction of **21** and **17**. ¹⁰ As shown in Scheme 3, it was observed that, when denitration preceded hydrolysis, the *trans*-fused adduct (**23**) was obtained as the predominant product, with only moderate selectivity. By contrast, when hydrolysis was carried out at the stage of compound **22**, denitration of the resultant ketone selectively delivered the *cis*-fused adduct **24** (*cis:trans* = 7:1).¹¹

Having explored the cycloaddition-denitration sequence with activated diene coupling partners, we next investigated the use of less reactive hydrocarbon-based dienes in the Diels-Alder step. Our initial efforts to effect Diels-Alder reaction between 2,3-dimethyl-1,3 butadiene and dienophile **11** were unsuccessful. Under moderate thermal conditions (<130 °C), no reaction was observed. At higher temperatures (150 °C), compound **25** was formed in 25-40% yield.12 The structure of adduct **25** was confirmed by X-ray crystallography. It was eventually determined that cycloadduct **26** could be generated in moderate yield at 100 $°C$, through the addition of 5.0 M LiClO₄ in THF,¹³ which serves to accelerate the Diels-Alder reaction (Table 1, entry 1). In contrast to the moderate Diels-Alder yields obtained with cyclohexenyl dienophile, **11** (Table 1, entries 1-2), the cyclopentenyl dienophile, **17**, readily undergoes cycloaddition with hydrocarbon dienes, furnishing *cis*-fused hydrindane adducts in excellent yields (Table 1, entries 3-4).

We now turned to an assessment of the feasibility of the denitration step, which we hoped would provide access to substituted, *trans*-fused hydrindane and decalin systems. As outlined in Table 2, entries 1 and 2, the *cis*-fused decalin systems (**26** and **27**) underwent

radical-based denitration to provide the corresponding *trans*-fused adducts in good yield and with excellent levels of diastereocontrol (>20:1 *trans*: *cis*).¹⁴ However, denitration of hydrindanes **28** and **29** proceeded with quite poor levels of stereoselectivity (Table 2, entries 3 and 4). 15

Finally, we sought to accomplish reduction of the *cis*-fused Diels-Alder adducts (**26** and **28**), to provide the corresponding *cis*-fused amine derivatives. As outlined in Scheme 5, treatment16 of compounds **26** and **28** with zinc dust in the presence of AcOH, followed by Alloc protection of the resultant angular amines, provided compounds **34** and **35** in good yield, and with the cyanide functionality intact. The two-step Diels-Alder/reduction sequence represents a straightforward synthesis of angular amines.

3. Experimental section

3.1. General

Experiments involving moisture- and/or air-sensitive compounds were performed in ovenor flame-dried glassware with rubber septa under a positive pressure of nitrogen or argon using standard Schlenk techniques. Heating was accomplished by heating mantle or silicon oil bath using a temperature controller. Organic solutions were concentrated under reduced pressure using a Büchi rotatory evaporator, unless otherwise noted. Reactions were magnetically stirred and monitored by thin-layer chromatography (TLC) carried out on 0.25 mm E. Merck silica gel plates (60- F_{254}) using UV light as a visualizing agent and a KMnO₄ solution, a vanillin solution, an anisaldehyde solution, or a ceric ammonium molybdate (CAM) solution, and heat as developing agent. Flash chromatography was carried out with EM silica gel 60 (230-240 mesh) or Sorbent Technology silica gel 60 (particle size 32-63 μ m) according to the method of Still.¹⁷ Yields refer to chromatographically and spectroscopically (^1H-NMR) homogeneous materials, unless otherwise stated. Unless otherwise noted, all reagents were purchased at the highest commercial quality from commercial suppliers and used without further purification. Lithium perchlorate (battery grade, dry, 99.99%) was purchased from Aldrich and dried under high vacuum at 130 $^{\circ}$ C for at least 12 h just prior to use. Tetrahydrofuran (THF) was freshly distilled from sodium/ benzophenone ketyl under an atmosphere of argon. Benzene, dichloromethane (CH_2Cl_2) and toluene were freshly distilled over CaH2 or filtered through a column of activated alumina under an atmosphere of argon. Diethyl ether was filtered through a column of activated alumina under an atmosphere of argon. Microwave reactions were performed on a Biotage microwave reactor. ¹H and ¹³C NMR spectra were recorded on a Bruker DRX-400, a Bruker DRX-500 or a Bruker DRX-600 spectrometer at ambient temperature (300 K), unless otherwise stated. Chemical shifts of the ¹H NMR (CDCl₃: 7.26 ppm) and ¹³C NMR (CDCl₃: 77.0 ppm) spectra were referenced to residual solvent peaks. ¹H NMR spectra are reported as follows: chemical shift, multiplicity ($br = broad$, $s = singlet$, $d = doublet$, $m = multiplet$), coupling constant and integration. ¹³C NMR spectra were recorded with ¹H decoupling. The multiplicities of the carbons of *trans*-3,4-dimethylbicyclo[4.4.0]dec-3-en-1-carbonitrile (**30a**) and *trans-*1,2,3,4,4a,5,6,7,8,9,9a,10-dodecahydroanthracene-4a-carbonitrile (**31a**) were determined by DEPT (Distortionless Enhancement by Polarization Transfer) experiments. IR spectra were recorded on a Jasco FT/IR-6100 or a Nicolet AVATAR 370 DTGS spectrometer and are reported in terms of frequency of absorption (cm−¹). Melting points were measured using MELTEMP® and are uncorrected. Low resolution mass spectra were acquired on a ZQ Micromass spectrometer using the technique of ESI (electrospray ionization). High-resolution mass spectra (HRMS) were obtained from the Columbia University Mass Spectral Core Facility on a JEOL HX 110 mass spectrometer using the technique of EI (electron impact ionization) or FAB (fast atom bombardment).

3.2. General Procedure

3.2.1. General procedure for the nitration of cyloalkenecarbonitriles (9 and 16). ⁷—To a stirred solution of cycloalkenecarbonitrile (20.0 mmol, 1.0 equiv) in anhydrous CH₃CN (100 mL) at 0 °C was added NaNO₂ (4.14 g, 60.0 mmol, 3.0 equiv) followed by ammonium cerium nitrate (32.8 g, 60.0 mmol, 3.0 equiv). The reaction mixture was stirred at room temperature for 24 h, diluted with H₂O, and extracted with EtOAc (20 mL \times 3). The organic layer was treated with a saturated solution of NaHCO_3 (100 mL) and the resulting mixture was stirred at room temperature for 12 h. The mixture was diluted with H_2O and extracted with EtOAc (20 mL \times 3). The combined extract was dried over anhydrous MgSO₄, filtered, and concentrated on a rotatory evaporator. Purification of the residue by flash column chromatography on silica gel gave the title compound.

3.2.2. 2-Nitrocyclohexenecarbonitrile (11)—Prepared using cyclohexenecarbonitrile $(9, 2.14 \text{ g}, 20.0 \text{ mmol})$, NaNO₂ $(4.14 \text{ g}, 60.0 \text{ mmol})$ and ammonium cerium nitrate $(32.8 \text{ g},$ 60.0 mmol) in CH₃CN (100 mL), 11 was isolated as a yellow oil (1.70 g, 56%) after column chromatography (hexanes/EtOAc=4:1). ¹H NMR (500 MHz, CDCl₃) δ 2.75-2.72 (m, 2H), 2.61-2.59 (m, 2H), 1.86-1.82 (m, 2H), 1.77-1.72 (m, 2H); ¹³C NMR (125 MHz, CDCl₃) δ 157.9, 115.1, 114.2, 29.4, 25.7, 20.7, 20.3; IR (neat) 2952, 2869, 2221, 1652, 1528, 1427, 1342, 1331 cm−¹ ; HRMS (EI) exact mass calculated for [M]+ (C7H8N2O2) required *m*/*z* 152.0586, found *m*/*z* 152.0583.

3.2.3. 2-Nitrocyclopentenecarbonitrile (17)—Prepared using cyclopentenecarbonitrile $(16, 3.00 \text{ g}, 32.2 \text{ mmol})$, NaNO₂ $(6.67 \text{ g}, 96.6 \text{ mmol})$ and ammonium cerium nitrate $(53.0 \text{ g},$ 96.6 mmol) in CH3CN (200 mL), **17** was isolated as a yellow oil (2.70 g, 61%) after column chromatography (hexanes/EtOAc=4:1). ¹H NMR (600 MHz, CDCl₃) δ 3.08-3.04 (m, 2H), 2.96-2.92 (m, 2H), 2.24-2.19 (m, 2H); 13C NMR (150 MHz, CDCl3) δ 160.4, 117.2, 112.4, 34.6, 31.2, 20.7; IR (neat) 2964, 2919, 2850, 2229, 1644, 1521, 1435, 1348, 1321, 1177 cm⁻¹. HRMS (EI) exact mass calculated for [M]⁺ (C₆H₆N₂O₂) required *m*/*z* 138.0429, found *m*/*z* 138.0432.

3.2.4. General procedure for the Diels-Alder reaction of 2-

nitrocycloalkenecarbonitriles (11 and 17) with electron-rich dienes (10 and 21) —A sealable vial reactor equipped with a magnetic stirring bar was charged with 2 nitrocycloalkenecarbonitriles (1.0 equiv), an electron-rich diene (2.0 equiv) and anhydrous toluene. The reactor was sealed and the mixture was stirred at 100° C in a pre-heated oil bath for 10-80 h. At this point, the mixture was cooled to room temperature and the solvent was evaporated. The resulting residue was purified by flash column chromatography on silica gel to afford the desired Diels-Alder adduct as a mixture of *endo* and *exo* diastereomers.

3.2.5. A 1:1 mixture of silyl enol ethers 12 and 13—Prepared using **11** (200 mg, 1.31 mmol) and *trans*-2-(*tert*-butyldimethylsilyloxy)-4-methoxy-1,3-butadiene (**10**, 564 mg, 2.63 mmol) in toluene (3.0 mL) at 100 °C for 80 h, an approximate 1:1 mixture of **12** and **13** was isolated as a pale yellow oil (350 mg, 78%) after column chromatography (hexanes/ EtOAc=12:1). Diastereomers **12** and **13** were separated by careful column chromatography. *Exo* isomer 12: ¹H NMR (600 MHz, CDCl₃) δ 5.14 (d, *J* = 4.7 Hz, 1H), 4.04 (d, *J* = 4.9 Hz, 1H), 3.32 (s, 3H), 2.61 (br s, 2H), 2.45-2.44 (m, 1H), 2.37-2.34 (m, 1H), 1.89-1.85 (m, 1H), 1.74-1.71 (m, 2H), 1.65-1.56 (m, 2H), 1.25 (br s, 1H), 0.94 (s, 9H), 0.23 (s, 3H), 0.21 (s, 3H); IR (neat) 2951, 2932, 2884, 2859, 2239, 1752, 1673, 1554, 1454, 1370, 1254, 1230, 1210, 1089 cm−¹ ; MS (ESI) 389.3 [M+Na]+. *Endo* isomer **13**: 1H NMR (600 MHz, CDCl3) δ 4.93 (d, *J* = 1.7 Hz, 1H), 4.58 (s, 1H), 3.36 (s, 3H), 2.81 (d, *J* = 17.9 Hz, 1H), 2.66 (d, *J* = 15.4 Hz, 1H), 2.42 (d, *J* = 17.9 Hz, 1H), 1.93-1.90 (m, 2H), 1.77-1.75 (m, 1H), 1.67-1.58

(m, 3H), 1.52-1.46 (m, 1H), 0.93 (s, 9H), 0.22 (s, 3H), 0.20 (s, 3H); 13C NMR (125 MHz, CDCl3) δ 141.8, 121.8, 102.3, 92.1, 81.0, 58.5, 39.3, 38.1, 30.6, 29.5, 19.4, 18.5, 17.9, −4.4; IR (neat) 2953, 2932, 2884, 2858, 2237, 1674, 1548, 1455, 1366, 1254, 1226, 1198, 1096, 914 cm−¹ ; HRMS (FAB) exact mass calculated for [M+1]+ (C18H31N2O4Si) required *m*/*z* 367.2053, found *m*/*z* 367.2063. The structure of **13** was also confirmed by x-ray crystallographic analysis (see supplementary data).

3.2.6. A 1.8:1 mixture of silyl enol ethers 18 and 18'—Prepared using **17** (500 mg, 3.62 mmol) and *trans*-2-(*tert*-butyldimethylsilyloxy)-4-methoxy-1,3-butadiene (**10**, 1.55 g, 7.24 mmol) in toluene (4.0 mL) at 100 °C for 10 h, an approximate 1.8:1 mixture of **18** and **18'** was isolated as a pale yellow oil (1.21 g, 95%) after column chromatography (hexanes/ EtOAc=10:1). Diastereomers **18** and **18'** were separated by careful column chromatography. Compound **18**: 1H NMR (600 MHz, CDCl3) δ 4.96-4.93 (m, 1 H), 4.91-4.88 (m, 1 H), 3.32 (s, 3H), 2.73-2.69 (m, 1H), 2.62-2.59 (m, 1H), 2.32-2.28 (m, 1H), 2.14-2.01 (m, 4H), 1.95-1.89 (m, 1H), 0.90 (s, 9H), 0.19 (s, 3H), 0.17 (s, 3H); 13C NMR (150 MHz, CDCl3) δ 147.5, 119.7, 101.0, 100.7, 78.6, 57.7, 45.0, 38.3, 35.8, 28.6, 25.4, 20.0, 17.8, −4.6, −4.7; IR (neat) 2955, 2931, 2887, 2858, 2830, 2241, 1675, 1547, 1462, 1363, 1255, 1218, 1094, 835 cm⁻¹; HRMS (FAB) exact mass calculated for [M+1]⁺ (C₁₇H₂₉N₂O₄Si) required *m*/*z* 353.1897, found m/z 353.1918. Compound 18': ¹H NMR (600 MHz, CDCl₃) δ 5.11 (d, *J* = 4.9 Hz, 1H), 4.26 (d, *J* = 5.0 Hz, 1H), 3.32 (s, 3H), 2.79 (d of AB pattern, *J* = 18.3 Hz, 1H), 2.62-2.51 (m, 2H), 2.26 (d of AB pattern, *J* = 18.3 Hz, 1H), 2.17-2.12 (m, 1H), 2.06-2.01 (m, 1H), 1.93-1.88 (m, 2H), 0.91 (s, 9H), 0.20 (s, 3H), 0.18 (s, 3H); 13C NMR (150 MHz, CDCl3) δ 149.7, 120.8, 99.2, 98.7, 75.6, 57.3, 40.9, 39.8, 35.7, 33.5, 25.4, 18.6, 17.8, −4.6, −4.7; IR (neat) 2955, 2931, 2893, 2858, 2825, 2243, 1671, 1553, 1461, 1369, 1256, 1221, 1087, 835 cm⁻¹; HRMS (FAB) exact mass calculated for [M+1]⁺ (C₁₇H₂₉N₂O₄Si) required *m*/*z* 353.1897, found *m*/*z* 353.1880.

3.2.7. A 4:1 mixture of silyl enol ethers 22 and 22'—Prepared using **17** (75.0 mg, 0.543 mmol) and 2-(*tert*-butyldimethylsilyloxy)-1,3-butadiene (**21**, 200 mg, 1.08 mmol) in toluene (1.0 mL) at 100 °C for 36 h, an approximate 4:1 regioisomeric mixture of **22** and **22'** was isolated as a pale yellow oil (110 mg, 63%) after column chromatography (hexanes/ EtOAc=14:1). Regioisomers **22** and **22'** were inseparable and characterization was carried out in the next step. 1H and 13C NMR spectra of the mixture of **22** and **22'** are available in the supplementary data.

3.2.8. General procedure for the preparation of bicyclic ketones (14, 15, 19, 20, 23, and 24)—To a stirred solution of a *cis***-fused silyl enol ether (1.0 equiv) in anhydrous** benzene was added tri-*n*-butyltin hydride (2.0 equiv) followed by azobisisobutyronitrile (0.5 equiv) and the mixture was heated under reflux for 2 h. At this point, the mixture was cooled to room temperature and the solvent was evaporated. In a separate Falcon® tube, the resulting residue was dissolved in CH₃CN and treated with HF (48% in H₂O) and the mixture was stirred at room temperature for 1 h. The mixture was cooled in an ice-water bath and the reaction was quenched by dropwise addition of saturated $NAHCO₃$ solution. The precipitate thus formed was filtered off and the filtrate was diluted with H_2O and extracted with CH_2Cl_2 (3 × 30 mL). The combined extract was dried over anhydrous Na2SO4, filtered, and concentrated on a rotatory evaporator. The residue was purified by flash column chromatography on silica gel to give the title compound.

3.2.9. A 12:1 mixture of *trans***-3-oxobicyclo[4.4.0]dec-4-enecarbonitrile (14) and** *cis***-3-oxobicyclo[4.4.0]dec-4-enecarbonitrile (15)—**Prepared using an *exo* isomer **12** (35.0 mg, 95.5 μmol), tri-*n*-butyltin hydride (55.3 mg, 190 μmol), azobisisobutyronitrile (8.0 mg, 48 μmol) in benzene (4.0 mL), a 12:1 mixture of *trans*-fused **14** and *cis*-fused **15** was

isolated as a pale yellow oil (13 mg, 78% over two steps) after column chromatography (hexanes/EtOAc=4:1). The *trans*-fused **14** was isolated in pure form by careful column chromatography. Compound 14: ¹H NMR (600 MHz, CDCl₃) δ 6.68 (dd, *J* = 10.1, 1.5 Hz, 1 H), 6.14-6.12 (m, 1H), 2.79 (d of AB pattern, *J* = 16.7 Hz, 1H), 2.41 (d of AB pattern, *J* = 16.7 Hz, 1H), 2.38-2.37 (m, 1H), 2.04-2.02 (m, 1H), 1.98-1.94 (m, 2H), 1.83-1.80 (m, 1H), 1.78-1.63 (m, 2H), 1.55-1.50 (m, 1H), 1.48-1.46 (m, 1H); ¹³C NMR (150 MHz, CDCl₃) δ 194.4, 151.4, 130.0, 120.8, 48.3, 43.6, 43.6, 35.9, 28.2, 25.5, 22.0; IR (neat) 2936, 2861, 2232, 1683, 1448, 1384, 1244, 1159, 982 cm− HRMS (EI) exact mass calculated for [M]⁺ $(C_{11}H_{13}NO)$ required m/z 175.0997, found m/z 175.0998. The structure of 14 was also confirmed by x-ray crystallographic analysis (see supplementary data).

3.2.10. A 10:1 mixture of *trans***-3-oxobicyclo[4.4.0]dec-4-enecarbonitrile (14) and** *cis***-3-oxobicyclo[4.4.0]dec-4-enecarbonitrile (15)—**Prepared using an *endo* isomer **13** (50.0 mg, 0.136 mmol), tri-*n*-butyltin hydride (78.6 mg, 0.270 mmol), azobisisobutyronitrile (11.0 mg, 68.0 μmol) in benzene (5.0 mL), a 10:1 mixture of *trans*fused **14** and *cis*-fused **15** was isolated as a pale yellow oil (19 mg, 80% over two steps) after column chromatography (hexanes/EtOAc=4:1).

3.2.11. A 2:1 mixture of *trans***-3-oxobicyclo[4.3.0]non-4-enecarbonitrile (19) and** *cis***-3-oxobicyclo[4.3.0]non-4-enecarbonitrile (20)—**Prepared using a distereomeric mixture of **18** and **18′** (200 mg, 0.567 mmol), tri-*n*-butyltin hydride (329 mg, 1.13 mmol), azobisisobutyronitrile (47.0 mg, 0.284 mmol) in benzene (6.0 mL), a 2:1 mixture of *trans*-fused **19** and *cis*-fused **20** was isolated as a pale yellow oil (47 mg, 51% over two steps) after column chromatography (hexanes/EtOAc=8:1). Diastereomers **19** and **20** were separated by careful column chromatography. Compound **19**: 1H NMR (600 MHz, CDCl3) δ 7.04 (dd, *J* = 10.0, 1.3 Hz, 1H), 6.14 (dd, *J* = 10.0, 2.8 Hz, 1H), 3.05 (d of AB pattern, *J* = 16.7 Hz, 1H), 2.63-2.59 (m, 1H), 2.39 (d of AB pattern, *J* = 16.7 Hz, 1H), 2.30-2.26 (m, 1H), 2.18-2.12 (m, 1H), 2.11-2.05 (m, 1H), 2.01-1.93 (m, 1H), 1.86-1.80 (m, 1H), 1.77-1.71 (m, 1H); 13C NMR (150 MHz, CDCl3) δ 195.3, 147.9, 130.8, 121.6, 48.1, 47.4, 47.4, 35.6, 26.2, 21.0; IR (neat) 2960, 2879, 2231, 1683, 1455, 1381, 1240, 1159 cm−¹ ; HRMS (EI) exact mass calculated for [M]+ (C10H11NO) required *m*/*z* 161.0841, found *m*/*z* 161.0850. The structure of **19** was also confirmed by x-ray crystallographic analysis (see supplementary data). Compound 20: ¹H NMR (600 MHz, CDCl₃) δ 6.73 (dd, *J* = 10.3, 3.7 Hz, 1H), 6.05 (dd, *J* = 10.3, 1.9 Hz, 1H), 3.17-3.14 (m, 1H), 2.83 (d of AB pattern, *J* = 16.6 Hz, 1H), 2.73 (d of AB pattern, *J* = 16.6 Hz, 1H), 2.35-2.23 (m, 2H), 2.00-1.93 (m, 1H), 1.90-1.82 (m, 3H); 13C NMR (150 MHz, CDCl3) δ 193.8, 149.5, 128.5, 123.3, 44.8, 42.0, 41.9, 35.9, 31.0, 23.0; IR (neat) 2923, 2853, 1679, 1446, 1386, 1244, 1151, 1055, 842 cm−¹ ; HRMS (EI) exact mass calculated for [M]+ (C10H11NO) required *m*/ *z* 161.0841, found *m*/*z* 161.0836.

3.2.12. A 2:1 mixture of *trans***-3-oxobicyclo[4.3.0]nonanecarbonitrile (23) and** *cis***-3-oxobicyclo[4.3.0]nonanecarbonitrile (24)—**Prepared using a silyl enol ether **22** (60.0 mg, 0.186 mmol), tri-*n*-butyltin hydride (81.2 mg, 0.279 mmol), azobisisobutyronitrile (10.0 mg, 56.0 μmol) in benzene (3.0 mL), a 2:1 mixture of *trans*-fused **23** and *cis*-fused **24** was isolated as a pale yellow oil (10.3 mg, 34% over two steps) after column chromatography (hexanes/EtOAc=4:1). Diastereomers **23** and **24** were separated by careful column chromatography. Compound 23: ¹H NMR (600 MHz, CDCl₃) δ 2.93-2.90 (m, 1H), 2.59-2.56 (m, 1H), 2.35-2.26 (m, 3H), 2.19-2.16 (m, 1H), 2.08-1.87 (m, 5H), 1.70-1.61 (m, 2H); 13C NMR (150 MHz, CDCl3) δ 206.3, 120.9, 49.7, 48.3, 47.7, 39.8, 37.0, 27.6, 25.7, 21.9; IR (neat) 2960, 2873, 2229, 1715, 1457, 1423, 1187, 913 cm⁻¹; HRMS (EI) exact mass calculated for $[M]^+$ (C₁₀H₁₃NO) required m/z 163.0997, found m/z 163.0988. Compound 24: ¹H NMR (500 MHz, CDCl₃) δ 2.71 (d of AB pattern, $J = 18.0$ Hz, 1H),

2.65-2.60 (m, 1H), 2.53 (d of AB pattern, *J* = 18.0 Hz, 1H), 2.42-2.33 (m, 2H), 2.28-2.22 (m, 1H), 2.18-2.10 (m, 2H), 1.92-1.83 (m, 2H), 1.77-1.70 (m, 2H), 1.66-1.56 (m, 1H); ¹³C NMR (125 MHz, CDCl3) δ 207.3, 124.3, 45.2, 43.5, 42.2, 38.2, 37.0, 30.4, 26.0, 23.0; IR (neat) 2957, 2872, 2231, 1719, 1526, 1373, 1350, 1241, 1045 cm−¹ ; HRMS (EI) exact mass calculated for $[M]^+$ (C₁₀H₁₃NO) required m/z 163.0997, found m/z 163.0988.

3.2.13. Stereochemical confirmation of *cis***-3-**

oxobicyclo[4.3.0]nonanecarbonitrile (24)—To a solution of **20** (20.0 mg, 0.124 mmol) in degassed CH₃OH (4.0 mL) was added Pd(OH)₂ (20.0 mg, 20% w/w). A hydrogenfilled balloon (1 atm) was placed over the solution and the mixture was stirred at room temperature for 2 h. At this point, TLC analysis indicated that the reaction was completed. The mixture was filtered through a pad of Celite®, washed thoroughly with EtOAc, and the solvents were evaporated. The crude mixture was purified by flash column chromatography (hexanse/EtOAc=2:1) on silica gel to afford *cis*-fused **24** (19 mg, 94%), whose spectroscopic properties were in complete accord with the authentic sample.

3.2.14. Preparation of a 1:7 mixture of *trans***-3 oxobicyclo[4.3.0]nonanecarbonitrile (23) and** *cis***-3-**

oxobicyclo[4.3.0]nonanecarbonitrile (24)—A Falcon® tube was charged with a silyl enol ether **22** (10.0 mg, 31.0 μmol) and CH3CN (2.0 mL) and HF (48% in H2O, 0.010 mL) was added dropwise. The mixture was stirred at room temperature for 8 h. The mixture was cooled in an ice-water bath and the reaction was quenched by dropwise addition of saturated NaHCO₃ solution. The mixture was diluted with H₂O and extracted with CH₂Cl₂ (3×4) mL). The combined extract was dried over an hydrous Na₂SO₄, filtered, and concentrated on a rotatory evaporator. The residue was purified by flash column chromatography (hexanes/ EtOAc=4:1) on silica gel to afford *cis*-6-nitro-3-oxobicyclo[4.3.0]nonanecarbonitrile (4.1 mg, 64%) as a pale yellow oil. *cis*-6-Nitro-3-oxobicyclo[4.3.0]nonanecarbonitrile: 1H NMR (500 MHz, CDCl3) δ 3.32 (d of AB pattern, *J* = 18.8 Hz, 1H), 2.84-2.78 (m, 1H), 2.72 (d of AB pattern, *J* = 18.8 Hz, 1H), 2.70-2.59 (m, 2H), 2.50-2.45 (m, 1H), 2.44-2.38 (m, 1H), 2.35-2.14 (m, 3H), 1.97-1.88 (m, 2H); 13C NMR (125 MHz, CDCl3) δ 203.0, 119.6, 93.5, 47.5, 43.9, 36.0, 35.7, 34.4, 32.3, 18.1; IR (neat) 2962, 2919, 2889, 2850, 2239, 1724, 1675, 1539, 1419 cm−¹ ; HRMS (EI) exact mass calculated for [M]+ (C10H12N2O3) required *m*/*z* 208.0848, found *m*/*z* 208.0855. To a stirred solution of *cis*-6-nitro-3 oxobicyclo[4.3.0]nonanecarbonitrile (27.0 mg, 0.130 mmol, 1.0 equiv) in anhydrous benzene (2.0 mL) was added tri-*n*-butyltin hydride (75.7 mg, 0.260 mmol, 2.0 equiv) followed by azobisisobutyronitrile (11.0 mg, 70.0 μmol, 0.5 equiv) and the mixture was heated under reflux for 1.5 h. At this point, the mixture was cooled to room temperature and the solvent was evaporated. The residue was purified by flash column chromatography (hexanes/EtOAc=6:1) on silica gel to give a 1:7 mixture of *trans*-fused **23** and *cis*-fused **24** (16 mg, 75%).

3.2.15. Preparation of tricyclic compound 25—A sealable vial reactor equipped with a magnetic stirring bar was charged with 2-nitrocyclohexenecarbonitrile (**11**, 70.0 mg, 0.460 mmol, 1.0 equiv), 2,3-dimethyl-1,3-butadiene (379 mg, 4.60 mmol, 10 equiv), 2,6-di-*tert*butyl-4-methylphenol (10.1 mg, 46.0 μmol, 0.1 equiv) and anhydrous toluene (2.0 mL). The reactor was sealed and the mixture was stirred at 150 °C in a pre-heated oil bath for 24 h. At this p oint, the mixture was cooled to room temperature and the solvent was evaporated. The resulting residue was purified by flash column chromatography (hexanes/EtOAc=14:1) on silica gel to afford the title compound (32.0 mg, 33%) as a tan/orange sticky solid. ¹H NMR (500 MHz, CDCl3) δ 2.81-2.74 (m, 1H), 2.67-2.64 (m, 1H), 2.41-2.29 (m, 2H), 2.12-1.88 (m, 5H), 1.70 (s, 3H), 1.65 (s, 3H), 1.56-1.54 (m, 2H); ¹³C NMR (125 MHz, CDCl₃) δ 180.0, 172.2, 124.2, 117.1, 47.4, 38.3, 32.4, 30.5, 27.7, 26.3, 24.2, 19.4, 18.5; IR (neat)

2931, 2863, 1785, 1715, 1452, 1386, 1086 cm−¹ ; mp 80-86 °C; HRMS (EI) exact mass calculated for $[M]^+$ (C₁₃H₁₇NO₂) required m/z 219.1259, found m/z 219.1250. The structure of **25** was also confirmed by x-ray crystallographic analysis (see supplementary data).

3.2.16. General procedure for the Diels-Alder reaction of 2-

nitrocyclohexenecarbonitriles (11) with simple dienes—A sealable tube equipped with a magnetic stirring bar was charged with 2-nitrocyclohexenecarbonitrile (**11**, 76.1 mg, 0.500 mmol, 1.0 equiv), a simple diene (5-10 equiv), 2,6-di-*tert*-butyl-4-methylphenol (5.5 mg, 25 mmol, 5.0 mol%) and anhydrous lithium perchlorate (532 mg, 5.00 mmol, 10 equiv). Under an atmosphere of argon, the mixture was cooled at 0° C and anhydrous THF (1.0 mL) was added slowly. After stirring of the mixture for 10 min, the cooling bath was removed and the tube was sealed. The mixture was then stirred at 100 °C in a pre-heated oil bath for 60 h. At this p oint, the mixture was cooled to room temperature, diluted with H_2O , and extracted with CH₂Cl₂ (3×15 mL). The combined extract was washed with brine, dried over anhydrous Na2SO4, filtered, and concentrated on a rotatory evaporator. The resulting residue was purified by gradient column chromatography (hexanes/EtOAc=9:1→8:1) on silica gel to afford the title compound.

3.2.17. *cis***-3,4-Dimethyl-6-nitrobicyclo[4.4.0]dec-3-en-1-carbonitrile (26)—**

Prepared using **11** (76.1 mg, 0.500 mmol) and 2,3-dimethyl-1,3-butadiene (410 mg, 5.00 mmol), **26** was isolated as an off-white solid (30.5 mg, 26%) after careful column chromatography. 1H NMR (CDCl3, 400 MHz, 333.15 K) δ 2.85 (d, *J* = 18.0 Hz, 1H), 2.76 (d of AB pattern, *J* = 18.8 Hz, 1H), 2.60 (d of AB pattern, *J* = 18.8 Hz, 1H), 2.32-2.24 (m, 2H), 2.10-2.02 (m, 2H), 1.86-1.55 (m, 5H), 1.64 (s, 6H); 13C NMR (CDCl3, 100 MHz, 333.15 K) δ 121.8, 121.4, 120.8, 90.4, 40.5, 39.7, 37.1, 33.9, 32.7, 22.0, 20.9, 18.4, 18.2; IR (KBr) 2933, 2865, 2232, 1789, 1544, 1455, 851 cm−¹ ; HRMS (EI) exact mass calculated for $[M]^+$ (C₁₃H₁₈NO₂) required m/z 234.1368, found m/z 234.1383.

3.2.18. *cis***-9a-Nitro-1,2,3,4,4a,5,6,7,8,9,9a,10-dodecahydroanthracene-4a-**

carbonitrile (27)—Prepared using **11** (76.1 mg, 0.500 mmol) and 1,2 dimethylenecyclohexane¹⁸ (270 mg, 2.50 mmol), 27 was isolated as an off-white solid (45.6) mg, 35%) after careful column chromatography. ¹H NMR (CDCl₃, 400 MHz, 333.15 K) δ 2.79 (d, *J* = 17.6 Hz, 1H), 2.68 (d of AB pattern, *J* = 18.8 Hz, 1H), 2.54 (d of AB pattern, *J* = 18.8 Hz, 1H), 2.31-2.24 (m, 1H), 2.20 (d, *J* = 18.0 Hz, 1H), 2.10-2.02 (m, 2H), 1.92-1.56 (m, 13H); 13C NMR (CDCl3, 100 MHz, 333.15 K) δ 124.2, 123.1, 121.5, 90.4, 39.7, 39.4, 36.1, 34.0, 32.8, 29.7, 29.5, 29.3, 22.6, 22.1, 21.0; IR (KBr) 2930, 2236, 1789, 1546, 1439, 851 cm−¹ ; HRMS (EI) exact mass calculated for [M+1]+ (C15H21N2O2) required *m*/*z* 261.1603, found *m*/*z* 261.1607.

3.2.19. General procedure for the Diels-Alder reaction of 2-

nitrocyclopentenecarbonitriles (17) with simple dienes—A sealable vial reactor equipped with a magnetic stirring bar was charged with 2-nitrocyclopentenecarbonitriles (**17**, 34.5 mg, 0.250 mmol, 1.0 equiv), a simple diene (0.750 mmol, 3.0 equiv), 2,6-di-*tert*butyl-4-methylphenol (2.8 mg, 13 μmol, 5.0 mol%) and anhydrous toluene (0.5 mL). The reactor was sealed and the mixture was stirred at 100 °C in a pre-heated oil bath for 24 h. At this point, the mixture was cooled to room temperature and the solvent was evaporated. The resulting residue was purified by column chromatography (hexanes/EtOAc=9:1) on silica gel to afford the title compound.

3.2.20. *cis***-3,4-Dimethyl-6-nitrobicyclo[4.3.0]non-3-en-1-carbonitrile (28)—**

Prepared using **17** (34.5 mg, 0.250 mmol) and 2,3-dimethyl-1,3-butadiene (61.6 mg, 0.750 mmol), **28** was isolated as a pale yellow oil (52.3 mg, 95%) after column

chromatography. ¹H NMR (600 MHz, CDCl₃) δ 2.80 (d of AB pattern, $J = 18.4$ Hz, 1H), 2.72 (d of AB pattern, *J* = 18.0 Hz, 1H), 2.64-2.60 (m, 1H), 2.49 (d of AB pattern, *J* = 18.4 Hz, 1H), 2.35-2.31 (m, 1H), 2.28 (d of AB pattern, *J* = 18.0 Hz, 1H), 2.15-2.10 (m, 1H), 2.09-1.02 (m, 1H), 1.99-1.94 (m, 1H), 1.91-1.85 (m, 1H), 1.66 (s, 6H); 13C NMR (150 MHz, CDCl3) δ 121.4, 121.3, 121.2, 95.0, 43.4, 38.8, 37.6, 35.0, 34.5, 18.5, 18.3; IR (neat) 2986, 2964, 2915, 2891, 2861, 2237, 1537, 1457, 1431, 1354, 1133 cm⁻¹; HRMS (EI) exact mass calculated for $[M]^+$ (C₁₂H₁₆N₂O₂) required m/z 220.1212, found m/z 220.1214.

3.2.21. *cis***-9a-Nitro-2,3,3a,4,5,6,7,8,9,9a-decahydro-1H-**

cyclopenta[b]naphthalene-3a-carbonitrile (29)—Prepared using **17** (34.5 mg, 0.250 mmol) and 1,2-dimethylenecyclohexane¹⁸ (81.1 mg, 0.750 mmol), **29** was isolated as a pale yellow oil (56.0 mg, 91%) after column chromatography. ¹H NMR (CDCl₃, 400 MHz) δ 2.75 (d of AB pattern, *J* = 18.0 Hz, 1H), 2.70-2.60 (m, 2H), 2.44 (d of AB pattern, *J* = 18.0 Hz, 1H), 2.38-2.31 (m, 1H), 2.23 (d, *J* = 17.6 Hz, 1H), 2.19-1.82 (m, 8H), 1.67-1.57 (m, 4H); 13C NMR (CDCl3, 100 MHz) δ 123.6, 123.5, 121.3, 94.8, 43.2, 37.8, 36.6, 35.0, 34.5, 29.3, 29.2, 22.5, 22.5, 18.4; IR (neat) 2924, 2239, 1726, 1549, 1439, 1361, 845 cm⁻¹; HRMS (EI) exact mass calculated for $[M]^+(C_{14}H_{18}N_2O_2)$ required m/z 246.1368, found m/z 246.1383.

3.2.22. General procedure for the radical denitration of *cis***-fused decalin (26 and 27) and hydrindane (28 and 29) systems—**A 15-mL Schlenk-type flask

equipped with a magnetic stirring bar and a reflux condenser was charged with a *cis*-fused decalin or hydrindane derivative (1.0 equiv), tri-*n*-butyltin hydride (2.0 equiv), azobisisobutyronitrile (0.3-0.5 equiv) and anhydrous benzene. The mixture was heated under reflux for 2 h. At this point, the mixture was cooled to room temperature and the solvent was evaporated. The resulting residue was purified by gradient column chromatography $(hexanes/EtOAc=1:0\rightarrow 97:3\rightarrow 19:1)$ on silica gel to afford the title compound.

3.2.23. *trans***-3,4-Dimethylbicyclo[4.4.0]dec-3-en-1-carbonitrile (30a)—**Prepared using **26** (41.0 mg, 0.175 mmol), tri-*n*-butyltin hydride (102 mg, 0.350 mmol) and azobisisobutyronitrile (8.6 mg, 52 μmol) in benzene (1.8 mL), *trans*-fused **30a** was isolated as a pale yellow oil (29.5 mg, 89%) after column chromatography. The minor isomer, *cis*fused **30b** was not detected within the limits of ¹H and ¹³C NMR. ¹H NMR (CDCl₃, 400) MHz) δ 2.21 (d, *J* = 16.8 Hz, 1H), 2.05-1.89 (m, 3H), 1.80-1.75 (m, 1H), 1.72-1.58 (m, 3H), 1.62 (s, 6H), 1.49-1.41 (m, 1H), 1.37-1.23 (m, 4H); 13C NMR (CDCl3, 100 MHz) δ 125.8, 122.6 , 121.9 , 43.2 (CH₂), 39.9 (CH), 39.6 , 36.9 (CH₂), 36.6 (CH₂), 29.9 (CH₂), 25.5 (CH₂), 23.1 (CH₂), 18.7 (CH₃), 18.6 (CH₃); IR (neat) 2926, 2230, 1738, 1448, 1382 cm⁻¹; HRMS (EI) exact mass calculated for $[M]^+(C_{13}H_{19}N)$ required m/z 189.1517, found m/z 189.1516.

3.2.24. *trans***-1,2,3,4,4a,5,6,7,8,9,9a,10-Dodecahydroanthracene-4a-carbonitrile (31a)—**Prepared using **27** (21.0 mg, 80.7 μmol), tri-*n*-butyltin hydride (47 mg, 161 μmol) and azobisisobutyronitrile (6.6 mg, 40 μmol) in benzene (0.8 mL), *trans*-fused **31a** was isolated as an off-white solid (14.8 mg, 85%) after column chromatography. The minor isomer, *cis*-fused 31b was not detected within the limits of ¹H and ¹³C NMR. ¹H NMR (CDCl3, 400 MHz) δ 2.15 (d, *J* = 16.8 Hz, 1H), 2.05-1.95 (m, 2H), 1.88-1.45 (m, 14H), 1.37-1.24 (m, 4H); 13C NMR (CDCl3, 100 MHz) δ 128.0, 124.3, 122.7, 42.1 (CH2), 39.8 $(CH_1, 39.4, 36.8$ (CH₂), 35.7 (CH₂), 30.0 (CH₂), 29.7 (CH₂), 29.6 (CH₂), 25.5 (CH₂), 23.2 (CH_2) , 22.8 (CH₂), 22.7 (CH₂); IR (KBr) 2925, 2830, 2228, 1738, 1449, 1320 cm⁻¹; HRMS (EI) exact mass calculated for $[M]^+$ (C₁₅H₂₁N) required m/z 215.1674, found m/z 215.1658.

3.2.25. A 1.5:1 mixture of *trans***-3,4-dimethylbicyclo[4.3.0]non-3-enecarbonitrile (32a) and** *cis-***3,4-dimethylbicyclo[4.3.0]non-3-enecarbonitrile (32b)—**Prepared

using **28** (30.8 mg, 0.140 mmol), tri-*n*-butyltin hydride (81.5 mg, 0.280 mmol) and azobisisobutyronitrile (11.5 mg, 70.0 μmol) in benzene (1.4 mL), an inseparable 1.5:1 mixture of *trans*-fused **32a** and *cis*-fused **32b** was isolated as an pale yellow oil (18.9 mg, 77%) after column chromatography. The relative stereochemistry of the major isomers was assigned according to Casadevall's observation and our previous study.² Diastereomeric ratios were estimated by the integration of the methine carbon signals. ¹H and ¹³C NMR spectra of the mixture of **32a** and **32b** are available in the supplementary data. IR (neat) 2959, 2914, 2836, 2231, 1727, 1452, 1275, 1124, 1073 cm⁻¹; HRMS (EI) exact mass calculated for $[M]^+$ (C₁₂H₁₇N) required m/z 175.1361, found m/z 175.1358.

3.2.26. A 1.7:1 mixture of *trans***-2,3,3a,4,5,6,7,8,9,9a-decahydro-1Hcyclopenta[b]naphthalene-3a-carbonitrile (33a) and** *cis***-2,3,3a,4,5,6,7,8,9,9adecahydro-1H-cyclopenta[b]naphthalene-3a-carbonitrile (33b)—**Prepared using **29** (34.5 mg, 0.140 mmol), tri-*n*-butyltin hydride (81.5 mg, 0.280 mmol) and azobisisobutyronitrile (11.5 mg, 70.0 μmol) in benzene (1.4 mL), an inseparable 1.7:1 mixture of *trans*-fused **33a** and *cis*-fused **33b** was isolated as an pale yellow oil (25.2 mg, 89%) after column chromatography. The relative stereochemistry of the major isomers was assigned according to Casadevall's observation and our previous study.² Diastereomeric ratios were estimated by the integration of the methine carbon signals. ¹H and ¹³C NMR spectra of the mixture of **32a** and **32b** are available in the supplementary data. IR (neat) 2927, 2833, 2230, 1729, 1440, 1312 cm⁻¹; HRMS (EI) exact mass calculated for [M]⁺ (C14H19N) required *m*/*z* 201.1517, found *m*/*z* 201.1523.

3.2.27. General procedure for the preparation of *cis***-fused angular amine**

derivatives—To a stirred solution of a *cis*-fused decalin or hydrindane (1.0 equiv) in a mixture of glacial acetic acid and THF (1:2, *v*/*v*) at −20 °C was added Zn dust (10 equiv). After stirring for 3 h, the mixture was filtered through a pad of Celite® and the filtrate was concentrated on a rotatory evaporator. The resulting residue was dissolved in THF and the mixture was treated successively with a saturated solution of $NAHCO₃$ and allyl chloroformate (2.5 equiv). The mixture was stirred for 3 h, diluted with H_2O , and extracted with EtOAc $(3 \times 10 \text{ mL})$. The combined extract was washed with brine, dried over anhydrous MgSO4, filtered, and concentrated on a rotatory evaporator. The resulting residue was purified by gradient column chromatography on silica gel to afford the title compound.

3.2.28. *cis***-Allyl 8a-cyano-6,7-dimethyl-1,2,3,4,4a,5,8,8a-**

octahydronaphthalen-4a-ylcarbamate (34)—Prepared using **26** (20.0 mg, 85.4 μmol), Zn dust (55.8 mg, 854 μmol) and allyl chloroformate (25.7 mg, 214 μmol), **34** was isolated as a pale yellow oil (10 mg, 40%) after column chromatography. ${}^{1}H$ NMR (600 MHz, CDCl3) δ 5.95-5.88 (m, 1H), 5.32-5.28 (m, 1H), 5.23-5.21 (m, 1H), 4.92 (s, 1H), 4.54-4.48 (m, 2H), 2.67 (d of AB pattern, *J* = 17.9 Hz, 1H), 2.50-2.42 (m, 2H), 2.25 (d of AB pattern, *J* = 17.9 Hz, 1H), 2.12-1.98 (m, 2H), 1.81-1.77 (m, 1H), 1.72-1.62 (m, 4H), 1.63 (s, 3H), 1.62 (s, 3H), 1.52-1.47 (m, 1H); 13C NMR (150 MHz, CDCl3) δ 154.4, 132.8, 123.2, 126.7, 119.3, 117.8, 65.3, 55.0, 41.5, 38.5, 36.2, 31.3, 30.2, 29.7, 21.2, 18.7, 18.3; IR (neat) 3346, 2933, 2865, 2232, 1724, 1509, 1450, 1294, 1231, 1111, 1046, 985 cm⁻¹; HRMS (FAB) exact mass calculated for $[M+1]^+$ (C₁₇H₂₅N₂O₂) required m/z 289.1916, found m/z 289.1919.

3.2.29. *cis***-Allyl 7a-cyano-5,6-dimethyl-2,3,3a,4,7,7a-hexahydro-1H-inden-3aylcarbamate (35)—**Prepared using **28** (20.0 mg, 90.8 μmol), Zn dust (59.4 mg, 908 μmol) and allyl chloroformate (27.4 mg, 227 μmol), **35** was isolated as a pale yellow oil (13 mg, 52%) after column chromatography. 1H NMR (600 MHz, CDCl3) δ 5.95-5.89 (m, 1H), 5.33-5.30 (m, 1H), 5.23 (dd, *J* = 10.4, 1.3 Hz, 1H), 5.01 (s, 1H), 4.54 (d, *J* = 5.5 Hz, 2H),

2.50-2.39 (m, 4H), 2.25-2.16 (m, 2H), 2.05-1.91 (m, 2H), 1.88-1.75 (m, 2H), 1.63 (s, 3H), 1.62 (s, 3H); 13C NMR (150 MHz, CDCl3) δ 155.0, 132.6, 123.9, 122.7, 119.8, 118.0, 65.5, 62.5, 45.2, 37.7, 37.6, 34.7, 29.7, 19.3, 18.8, 18.4; IR (neat) 3332, 2916, 2886, 2849, 2233, 1724, 1510, 1445, 1274, 1238, 1086, 916 cm−¹ ; HRMS (FAB) exact mass calculated for [M $+1$ ⁺ (C₁₆H₂₃N₂O₂) required *m*/*z* 275.1760, found *m*/*z* 275.1772.

Supplementary Material

Refer to Web version on PubMed Central for supplementary material.

Acknowledgments

Support was provided by the NIH (HL25848 to SJD). W.H.K. is grateful for a Korea Research Foundation Grant funded by the Korean government (KRF-2007-357-c00060). We thank Rebecca Wilson for assistance with the preparation of the manuscript and Dr. Peter K. Park (Columbia University), Dr. Bernhard Fasching (MSKCC), Dr. Jennifer Stockdill (MSKCC) and Dr. Pavel Nagorny (MSKCC) for helpful discussions. We also thank Dr. George Sukenick, Ms. Hui Fang, Sylvi Rusli (NMR Core Facility, Sloan-Kettering Institute) and Dr. Yasuhiro Itagaki (Mass Spectral Core Facility, Columbia University) for mass spectral and NMR.

We dedicate this paper to Professor Steven Ley on the occasion of his receipt of the Tetrahedron Prize for Creativity in Organic Chemistry, honoring his pioneering accomplishments in organic synthesis.

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Figure 1.

Scheme 2.

^aKey: (a) NaNO₂, CAN, CH₃CN, RT, 24 h, 56%; (b) toluene, 100 °C, 80 h, 78%; (c) *n*-Bu3SnH, AIBN, benzene, reflux, 2 h; (d) HF, CH3CN, 1 h.

Scheme 3.

^aKey: (a) NaNO₂, CAN, CH₃CN, RT, 24 h, 61%; (b) toluene, 100 °C, 10 h, 95%; (c) *n*-Bu3SnH, AIBN, benzene, reflux, 2 h; (d) HF, CH3CN, 1 h; (e) toluene, 100 C, 36 h, 63%, d.r. = 4:1; (f) HF, CH3CN, RT, 8 h; (g) *n*-Bu3SnH, AIBN, benzene, reflux, 1.5 h, 48% for two steps.

Scheme 4.

^aKey: (a) 2,6-di-*tert*-butyl-4-methylphenol, toluene, 130 °C, 24 h. (b) 2,6-di-*tert*-butyl-4 methylphenol, toluene, 150 °C, 24 h.

Scheme 5.

^aKey: (a) Zn dust, THF/AcOH (2:1), -20 °C, 3 h; (b) AllocCl, THF, sat. NaHCO₃, RT, 3 h.

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