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Anti-Tuberculosis Constituents from the Stem Bark of

Micromelum hirsutum

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Abstract

Anti-TB bioassay-directed fractionation led to the isolation of six carbazole alkaloids, as well as the γ -lactone derivative of oleic acid, from the CH₂Cl₂ extract of the stem bark of *Micromelum hirsutum*. The carbazoles include the new micromeline (**2**) and five known alkaloids: lansine (**3**), 3-methylcarbazole (**4**), methyl carbazole-3-carboxylate (**5**), 3-formylcarbazole (**6**), and 3-formyl-6-methoxycarbazole (**7**). Compound **1** was identified as the lactone derivative of oleic acid, (–)-Z-9-octadecene-4-olide, for which the trivial name micromolide (**1**) is suggested. It showed potent *in vitro* anti-TB activity against H37Rv (MIC: $1.5 \,\mu$ g/mL), a selectivity index (SI) of 63, and exhibited activity against the Erdman strain of *M. tuberculosis* in a J774 mouse macrophage model (EC₉₀: 5.6 μ g/mL). Thus, **1** appears worthy of further evaluation as a potential new anti-TB agent. Isolates **2**, **3**, **6** and **7** had anti-TB MIC values between 14.3 and 42.3 μ g/mL, while compounds **4** and **5** were considered inactive (MIC > 128 μ g/mL). Structure elucidation and identification were based on spectroscopic analysis, including MS, 1D/2D NMR, and a full ¹H spin system analysis of **1**.

Keywords

Micromelum hirsutum; Rutaceae; (–)-Z-9-octadecene-4-olide; micromeline; micromolide; carbazole alkaloids; anti-TB activity

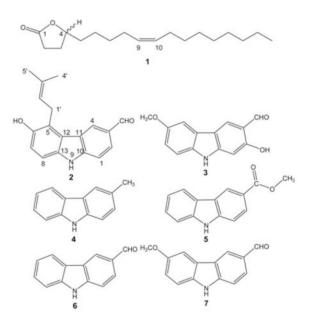
Introduction

With approximately 3 million annual deaths in the 1990s [1], tuberculosis remains a leading cause of mortality worldwide. The current treatment of an infection, a cocktail of drugs including, for example, isoniazid, rifampin, ethambutol and pyrazinamide is prescribed for 2 months, followed by a continuation phase in which isoniazid and rifampin are taken for an

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additional 4 months. However, failure to successfully complete this treatment has frequently led to the emergence of multi-drug-resistant tuberculosis (MDR-TB) [1].

Both the cost of treating an MDR-TB patient and the treatment failure rate are much greater than for drug-sensitive TB. Therefore, there is an urgent need for new anti-mycobacterial drugs, particularly those effective against a persistent infection [1]. Our International Cooperative Biodiversity Group (ICBG) research project [2], involving the collaboration of institutions in Vietnam, Laos and USA, has as one of its objectives the discovery of anti-TB compounds from plants. Micromelum hirsutum Oliv. (Rutaceae) was found to be a promising lead against Mycobacterium tuberculosis in vitro. A review of the literature revealed no prior phytochemical reports on this plant. In the present study, anti-TB bioassay-directed fractionation of the CH₂Cl₂ extract of the stem bark of *M. hirsutum* led to the isolation of the γ -lactone derivative of oleic acid (1) beside six carbazole alkaloids. One of them, micromeline (2), was determined to be a new structure, whereas lansine (3), 3-methylcarbazole (4), methyl carbazole-3-carboxylate (5), 3-formylcarbazole (6), and 3-formyl-6-methoxycarbazole (7) were previously described. The lactone 1 was identified as the known compound, (-)-Z-9octadecene-4-olide, and showed promising in vitro anti-TB activity with an MIC value of 1.5 μ g/mL and a selectivity index (SI) of 63. Alkaloids 2,3,6 and 7 showed *in vitro* anti-TB activity with MIC values of 31.5,14.3,42.3 and 15.6 µg/mL, respectively, whereas alkaloids 4 and 5 were considered inactive (MIC > 128 μ g/mL). The present paper reports the isolation, structure elucidation and/or dereplication, as well as the biological evaluation of these isolates.



Materials and Methods

General

Optical rotations were measured on a Perkin-Elmer model 241 polarimeter. IR spectra were run on a Jasco FT/IR-410 spectrometer, equipped with a Specac Silver Gate ATR system by applying a film on a germanium crystal. UV spectra were obtained with a Beckman DU-7 spectrometer. NMR spectra were recorded on Bruker DRX-500 MHz (11.7 Tesla) and DPX-300 MHz (7.0 Tesla) spectrometers. Chemical shifts (δ) were expressed in ppm with reference to the solvent signals. All NMR data were obtained by using standard pulse sequence supplied by the vendor. Column chromatography was carried out on silica gel (200 – 400 mesh, Natland International Corporation). Reversed-phase HPLC was carried out on a Waters 600E

Delivery System pump, equipped with a Waters 996 photodiode detector, and a Phenomenex C18 column (250×50 mm, 10μ m, 120 Å), at a flow rate of 18 mL/min. Thin-layer chromatography was performed on Whatman glass-backed plates coated with 0.25 mm layers of silica gel 60. HRE-SI-MS were recorded on a Micromass QTOF-2 spectrometer. EI-MS were recorded on a JEOL GC mate II spectrometer.

Plant material

The initial stem bark sample (SV-0155) of *Micromelum hirsutum* (a tree of 10 m height with greenish-yellow flowers and aromatic leaves) was collected at the Cuc Phuong National Park, Vietnam (Bong, $20^{\circ} 21'$ N, $105^{\circ} 35'$ E, edge of valley forest) on March 19, 1999, and was documented by voucher herbarium specimens number *Soejarto 10591*. A voucher specimen was deposited each at the herbaria of Cuc Phuong National Park (*CPNP*), Vietnamese Academy of Science and Technology (formerly, National Center for Science and Technology) (*HN*; Hanoi), and the Field Museum (*F*; Chicago). A larger sample (SVA-0155, stem bark, 4.5 kg, documented by voucher specimens number *Nguyen Manh Cuong 524*) was re-collected from the same site at Cuc Phuong on November 2, 2001, for complete isolation work.

Extraction and isolation

The dried, milled stem bark (4.5 kg) of *M. hirsutum* was extracted with CH₂Cl₂ to yield 45.5 g extract, of which 32.1 g were chromatographed by vacuum liquid chromatography over a silica gel column (5.0 kg), which was developed by gradient elution with *n*-hexane/EtOAc/ MeOH/H₂O (H:E:M:W, 300 mL aliquots beginning with 100% H, increasing by 10 mL of the next polar solvent, until 100% MeOH, with three aqueous MeOH washes) to afford 9 fractions (F1 – F9) based on their TLC patterns (SiO₂, H:E, 1: 1, an-isaldehyde reagent). Bioassay localized the anti-TB activity to fractions F2 (H:E 6: 4, 2.7 L V_E) (5.0 g) and F3 (H:E 4: 6, 3.6 $L V_{\rm F}$) (7.8 g). Fraction F2 was chromatographed over a silica gel column (250 g) and eluted with mixtures of petroleum ether-acetone of increasing polarities (P:A, 500 mL aliquots from 100: 0 to 80: 20, increasing by 2% acetone, with 100% acetone wash) as mobile phase to afford 31 combined fractions (F10 – F40) based on their TLC patterns (SiO₂, P:A, 8:2, 10% H₂SO₄ reagent). Fractions F15 (P:A, 9.2: 0.8, 300 mL $V_E)$ and F16 (P:A, 9.1:0.9, 600 mL $V_E)$ showed strong anti-TB activity with MICs of 1.96 and 3.72 µg/mL, respectively. F15 (347 mg) was further fractionated on a 16 g silica gel column. Elution was performed with a gradient of petroleum ether-CHCl₃ (50 mL aliquots from P:C, 9: 1 increasing 10% CHCl₃ to P:C 6:4) to yield compound 4 (2 mg, P:C, 9:1, 50 mL V_E) and one fraction composed primarily of compound 1 (209 mg, P:C, 8:2, 50 mL, P:C, 7:3,100 mLV_F) based on TLC patterns (C-18, MeOH:H₂O,9.8:0.2,10% H₂SO₄ detection reagent). A portion (210 mg) of F16 (772 mg) was subjected to preparative HPLC separation on a Phenomenex C18 column and developed with MeOH-H₂O (9:1), 18 mL/min; t_R 79 – 83 min, 1422–1494 mL V_E to obtain compound 1 (59 mg). Fraction 3 was chromatographed over a Sephadex LH-20 (250 g) column and isocratically eluted at 0.75 mL/min, in 25-mL fractions with CHCl3-MeOH (4:6) to afford 17 fractions (F41 - F57) based on TLC patterns (SiO₂, P:A, 8:2,10% H₂SO₄ reagent). Fractions F53, F54 and F55 showed the same level of anti-TB activity (MIC: $16 \,\mu g/mL$). Compound 3 (70 mg) was crystallized from F55 in acetone as yellow needles. Fractions F53 and F54 were pooled and chromatographed over a silica gel column (25 g), which was eluted with a petroleum etheracetone gradient(500 mL aliquots from 100% P, increasing by 1 % acetone to P:A, 8: 2) to yield compounds 5 (18 mg, P:A, 9.5: 0.5, 60 mL), 7 (9 mg, P:A, 9.3: 0.7, 40 mL) and 2 (6 mg, P:A, 9: 1, 30 mL), in addition to fractions F69 – F75. Fraction 70 was repeatedly separated on silica gel columns using the same solvent system as above to yield pure compound $\mathbf{6}$ (14 mg, P:A, 9.4:0.6, 60 mL).

(-)-Z-9-Octadecene-4-olide [Micromolide] (1)—Colorless oil. $[\alpha]_{D}^{23}$: -19.8° (*c* 0.49, CHCl₃); IR (film, Ge ATR): v_{max} = 2925, 2851,1773, 1163 cm⁻¹; ¹H-NMR (CDCl₃, 300 MHz) and ¹³C-NMR (CDCl₃, 75 MHz): see Table 1; EI-MS: *m/z* (rel. int.) = 280 ([M]⁺, 40), 220 (15), 195 (8), 181 (10), 167 (18), 153 (29), 137 (32), 136 (42), 123 (32), 109 (48), 95 (92), 85 (74), 81 (100), 67 (67).

Micromeline (2)—Yellow microcrystalline powder. UV (acetone): λ_{max} (log ε) = 316 (3.3), 329 (3.7) nm; IR (film): v_{max} = 3394, 3317, 2964, 2929, 2846, 1670, 1600, 1300, 1164, 1058, 812, 670 cm⁻¹; ¹H-NMR (acetone- d_6 , 300 MHz) and ¹³C-NMR (acetone- d_6 , 75 MHz), see Table 2; Negative HRESI-MS: m/z = 278.1183 [M – 1][–] (calcd. for C₁₈H₁₆NO₂: 278.1181).

Lansine (3)—Yellow needles. ¹³C-NMR (acetone- d_6 , 75 MHz): $\delta = 97.6$ (C-1), 162.2 (C-2), 116.5 (C-3), 129.2 (C-4), 104.5 (C-5), 156.3 (C-6), 115.7 (C-7), 113.0 (C-8), 148.0 (C-10), 119.1 (C-11), 125.3 (C-12), 136.8 (C-13), 196.9 (CHO), 56.6 (OCH₃). (See Wu et al. [11] and Prakash et al. [12] for additional molecular data).

Anti-TB assay

Anti-TB activity of crude extracts, fractions and compounds against *M. tuberculosis* (H37Rv) was determined using the fluorometric microplate Alamar blue assay (MABA) as described previously [3]. Percent inhibition was defined as $1 - (\text{test well fluorescence units}/\text{mean FU fluorescence units of triplicate wells containing only bacteria)x 100. The drug concentration effecting an inhibition of > 90% was considered as the MIC.$

Cytotoxicity of compounds

Evaluation of the cytotoxic activity of compounds 1 - 7 in Vero cells (African green monkey kidney cells) was performed as described earlier [4] using the CellTiter 96 aqueous non-radioactive cell proliferation assay (Promega Corp., Madison, WI). The IC₅₀ was defined as the reciprocal dilution resulting in 50% inhibition of the Vero cells.

Intracellular activity

The mouse macrophage cell line J774A.1 (ATCC TIB-67) was used to study the activity of samples against intracellular *M. tuberculosis*. Cells were cultured in Dulbecco's modified Eagle's medium (D-MEM) supplemented with 10% FBS, 1% glutamine, at 37°C and 5% CO₂. The J774A.1 macrophage cell line was prepared using standard procedures (ATCC). Twenty mL of media (D-MEM) were dispensed into a 75 cm³ flask. Cells were detached with a scraper from the flask, centrifuged at 1000 rpm for 5 min, and re-suspended in 10 mL media. Cell number was counted using trypan blue and the cell suspension was adjusted to1–3×10⁵ cells mL⁻¹. The coverslips were placed in the 24-well plates (Fisher Scientific), and 1 mL cells were added to each well. Plates were incubated at 37 °C in a 5% CO₂ incubator for 24 hours until confluent.

M. tuberculosis Erdman (ATCC 35 801) was diluted with the cell culture media to a final concentration of $1-3\times10^5$ CFU mL⁻¹ and 500 μ L were added to each well of another 24-well plate. The coverslips were transferred with the cells from the old plates to the new 24-well plates and infected for 2 – 3 hours at 37 °C in a CO₂ incubator. Extracellular bacteria were removed by washing the coverslips with HBSS (Fisher Scientific). The coverslips were transferred to new 24-well plates and 1 mL of fresh media was added to each well. Cultures were incubated at 37 °C, 5 % CO₂ for 7 days.

Cells were lysed using 0.25 % SDS (Fisher Scientific), and sonicated for 15 seconds. Pretreatment samples (T_0) were diluted (1: 10, 1: 100 and 1: 1000), and 0.1 mL of the undiluted

suspension and the 3 dilutions were plated on 7H11 agar, and incubated for 3 weeks. Stock solutions of test samples were prepared in DMSO, diluted in D-MEM, and added to each well of new 24-well plates. Coverslip cultures were transferred to the plate and incubated for 7 days in the CO_2 incubator.

After a week, cells from control and treated wells were lysed, and 3 dilutions were prepared. One hundred μ L of dilutions were placed on 7H11 agar plates (T₇). Colonies were counted after 16 – 20 days and compared with the control (T₀). By comparing these data with CFU obtained from untreated control macrophages, the concentrations of the test compounds required to achieve a one log reduction (EC₉₀) and a two log reduction (EC₉₉) in viable *M. tuberculosis* were determined using CurveExpert 1.3 (Microsoft).

Results and Discussion

Anti-TB bioassay-directed fractionation of the CH_2Cl_2 extract of the stem bark of *Micromelum hirsutum*, screening against the virulent (H37Rv) strain of *Mycobacterium tuberculosis*, led to the isolation of six carbazole alkaloids (2 – 7) and the C_{18} y-lactone 1.

Compound **1**, was isolated as a colorless gum, $[\alpha]_{D}^{23}$: -19.8° (*c*, 0.49, CHCl₃), and was identified as the previously described γ -lactone derivative of oleic acid ([5], [6], reporting ¹H-NMR, IR and MS only). However, dereplication of 1 based on published spectroscopic/NMR data proved to be an impossible task due to inconsistencies and gaps in the published assignments, but also due to severe spectral overlap in both the ¹H and ¹³C aliphatic regions. Therefore, and in order to substantiate the structure of this potent antimycobacterial agent, an *ab initio* structure elucidation was performed taking into account COSY, HMQC and HMBC correlations beside the 1D¹H- and ¹³C-NMR spectra. The results are compiled in Table 1. The molecular formula C₁₈H₃₂O₂ of **1** was derived on the basis of MS results and ¹³C-NMR data. The ¹³C-NMR data additionally indicated that **1** contained one carbonyl ($\delta = 177.5$) and two olefinic carbons ($\delta =$ 130.7, 129.4). The DEPT spectrum further indicated that it contained a methyl group ($\delta = 14.3$) and an oxymethine carbon ($\delta = 81.2$). Resonances of aliphatic methylene carbons ($\delta = 22.9 -$ 35.7) suggested the presence of a long aliphatic chain with one double bond. The carbonyl carbon chemical shift was consistent with the presence of a saturated γ -lactone ring, supported by the IR absorption band at 1773 cm⁻¹ [8]. Further evidence for the γ -lactone partial structure came from the observation of an ion fragment (m/z = 85) in the EI-MS [5]. Critical to the establishment of the structure were (A) the unequivocal confirmation of the γ -lactone partial structure, (B) the location of the double bond in the aliphatic chain, and (C) the stereochemical assignment.

Regarding (A), several HMBC correlations (see Table 1) of the clearly resolved downfield ¹H and ¹³C resonances led to the definitive assignment of a butanolide ring system. The lactone ring closure gave rise to a ³*J* correlation between H-4 and C-1. Severe spectral overlap made proof for the location of the double bond (B) a complicated task. Ultimately, it was possible through a combination of COSY, HMQC and HMBC data, processed with post-acquisition software. Fig. 1 shows the conclusive ³*J* couplings from the HMBC spectrum that were essential in unambiguously determining the location of the double bond at C-9/10. In contrast to the literature, the olefinic protons are not isochronous. This is due to the asymmetrical substitution of the double bond. Since the asymmetry is minimal, the difference in chemical shifts between carbons C-9 and C-10 is small (0.044 ppm), and, therefore, higher order effects can be seen. The dtt-like shaped multiplet shows a significant roofing effect as a result of these higher order interactions. Similar higher order effects were also observed in the two methylene protons (H₂-8 and H₂-11) immediately adjacent to the double bond. Spectral iteration and simulation, using the PERCH software package, replicated the peak characteristics and assigned the exact shift values and *J* values of the olefinic protons. The

spectral overlap of the methylene, aliphatic resonances (H-12 to -17 and H-6 +-7) limited the achievable precision of the iteration process involved in a full-spin analysis of H-2-H-11. It can be estimated that at least 800 MHz spectra will be required to yield sufficient dispersion to allow successful iteration of H-2-H-11. These findings are in line with the fragmentation pattern observed in EI-MS (Fig. 2). The double bond was determined to be *Z*-configured on the basis of the ¹H,¹H-coupling constant (J < 12 Hz) and the ¹³C-NMR chemical shifts of the two adjacent carbons [9].

In order to determine the enantiomeric composition (C) of the natural isolate 1, a chiral LSR-NMR analysis according to the protocol recently established in our laboratory [7] was carried out using tris[3-(heptafluoropropylhydroxymethylene)-D-camphorato]-praseodymium(III), and concluded that the compound is enantiomerically pure. In order to assign the absolute stereochemistry of 1, we attempted to prepare its Mosher esters. However, after hydrolyzing the γ -lactone, the free alcohol spontaneously converted back into the butanolide during the following purification step. Thus, we are presently unable to report the absolute stereochemistry of the natural isolate 1. Compound 1 appears to be identical with previously described (-)-Z-9-octadecene-4-olide [5], [6], one of which has been reported to be enantiomerically pure [6]. However, because of inconsistencies in the published analytical data, an exact match with 1 cannot be shown. In order to address the stereochemical ambiguity remaining in both the literature [5], [6] and our work, and in following established drug discovery protocols, attempts are currently under way to synthesize 1, and to characterize and isolate minor side components/analogues that are likely to be contained in active fraction F16. Owing to the fact that 1 shows promising potential as a new anti-TB lead (see below), and since no trivial name has previously been suggested, the trivial name micromolide is proposed for **1**.

The carbazole alkaloid 2 was obtained as yellow crystals, and the molecular formula determined as $C_{18}H_{17}NO_2$ by negative HRESI-MS at $m/z = 278.1183 [M - 1]^-$ (calcd.: 278.1181). Based on the analysis of its IR, ¹H-NMR and ¹³C-NMR spectra, 2 was recognized as a carbazole alkaloid [10], [11]. The IR spectrum showed the presence of NH (3317 cm⁻¹), OH (3394 cm⁻¹) and conjugated carbonyl (1670 cm⁻¹) functions, which were also observed in the ¹H- and ¹³C-NMR spectra [δ = 10.666 (br s, NH), 10.054 (s, CHO), 7.990 (br s, OH) and $\delta = 191.9$ (aldehyde CH)]. It was shown to be an analogue of 3-formylcarbazole (6) by comparison with published NMR data [10]. As in 6, the aldehyde group in 2 is attached to C-3 due to the presence of signals due to an AMN spin system centered at $\delta 8. = 648, 7.913$ and 7.596, respectively (Table 2). The presence of HMBC long-range correlations between the carbon signal at $\delta = 191.9$ (CHO) and the proton signals at $\delta = 8.648$ (H-4) and 7.913 (H-2) supported this finding. However, in contrast to the ABMN spin-spin system of 6, 2 only showed a set of AB-type signals at δ = 7.114 (H-7) and 7.293 (H-8) with a shared coupling of 8.5 Hz, suggesting the presence of two additional substituents in the second benzene ring attached to either C-5 and C-6, or C-7 and C-8. One of the substituents was determined to be a 3methylbut-2-enyl (prenyl) group based on in-depth analyses of the COSY, HMQC and HMBC correlations, and due to the fact that the set of ¹H-NMR signals showed the complex J pattern $(A_2MX_3Y_3, \text{ see Table 2})$, which perfectly resembled the fingerprint of the prenyl groups of prenylated chalcones and flavanones recently isolated in our laboratory [16]. The site of linkage of the prenyl residue was determined to be C-5 due to the presence of NOE correlations of the proton resonating at $\delta = 8.648$ (H-4) with the protons at $\delta = 4.023$ (H₂-1') and 1.991 (H-4') (Fig. 3). The second substituent was determined to be a hydroxy group based on the ¹Hand ¹³C-NMR data [$\delta_{\rm H}$ = 7.990 (br s, OH) and $\delta_{\rm C}$ = 149.5]. It could be placed to C-6 due to the presence of HMBC correlations between the methylene signals of the prenyl residue at $\delta_{\rm H} = 4.023$ (H₂-1') and the hydroxylated C-6 carbon $\delta_{\rm C} = 149.5$. Thus, the structure of compound 2 was determined to be 6-hydroxy-5-(3-methylbut-2-enyl)-9H-carbazole-3carbaldehyde, a new carbazole alkaloid named micromeline.

Compounds **3** – **7** were shown to be the known carbazole alkaloids lansine (**3**) [11], [12], 3methylcarbazole (**4**) [13], [14], methyl carbazole-3-carboxylate (**5**) [10], 3-formylcarbazole (**6**) [10] and 3-formyl-6-methoxycarbazole (**7**) [10] by comparison of their NMR data and physical properties to those reported in the literature. Since no ¹³C-NMR data of **3** have been published in the literature, they are presented in the Materials and Methods section of the current report.

The crude CH_2Cl_2 extract, fractions and isolates of *M. hirsutum* were tested against *M*. tuberculosis H₃₇Rv in the MABA assay system (Table 3). All isolates were also tested for cytotoxicity toward Vero cells (derived from African green monkey kidney) in the MTS assay system to determine the selective index value (SI, $SI = IC_{50}$ Vero cell/MIC TB). Rifampin was used as a positive control in the assay. The CH₂Cl₂ extract of *M. hirsutum* had an MIC of 12.5 μ g/mL. Compound **1** showed promising *in vitro* anti-TB activity with an MIC of 1.5 μ g/mL and a selectivity index (SI) of 63 (Table 3). This level of selectivity compares favorably with some of the less potent anti-TB agents such as ethambutol, and is well above the value of 10 that is required to move into advanced testing in a major on-going TB drug discovery program [17]. Compound **1** was therefore further evaluated in the J774 mouse macrophage cell line infected with *M. tuberculosis* Erdman, a strain more virulent than H₃₇Rv (Fig. 4). The concentration of 5.6 µg/mL required to achieve a level of growth inhibition of 90% relative to control cultures, while higher than that required for the same level of inhibition in axenic medium (MIC = $1.5 \,\mu\text{g/mL}$), is still well below the IC₅₀ value of **1** for J774 cells, suggesting that growth inhibition of *M. tuberculosis* is not secondary to host cell toxicity. A three-fold higher concentration of 1 was required to achieve an additional \log_{10} reduction in cfu (EC₉₉); this was consistent with that observed for rifampin and streptomycin (Table 3). This compound appears worthy of further evaluation in in vivo models of tuberculosis infection in mice as well as to serve as a template for an analoging program.

The carbazoles **2**, **3**, **6** and **7** had MICs ranging from 14.3 to 42.3 μ g/mL. Compounds **4** and **5** are inactive. Inhibitory activity of carbazole alkaloids against *M. tuberculosis* was previously reported against the H₃₇R*a* rather than the H₃₇R*v* strain [15] for methyl carbazole-3-carboxylate (**5**) and 3-formylcarbazole (**6**) with MIC values of 50 and 100 μ g/ml, respectively. Although the number of studied carbazole alkaloids evaluated is limited to six compounds, preliminary conclusions about anti-TB structure-activity relationships can be drawn as follows. Carbazole alkaloids possessing an aldehyde group at the C3 position (**2**, **3**, **6** and **7**) exhibit significantly greater activity than those lacking this functionality (**4** and **5**). The presence of a methoxy group in **3** and **7** renders compounds at least twice as active as their free-phenolic counterparts **2** and **6**. Further carbazole alkaloids will have to be biologically evaluated, before definitive structure-activity correlations can be established.

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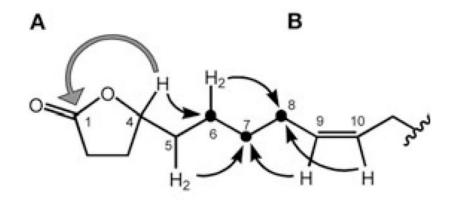


Fig. 1.

Conclusive proof for the structure of compound 1 with regards to (A) the γ -lactone arrangement and (B) the position of the double bond from the essential ³*J* HMBC correlations (see Table 1 for a complete NMR data set).

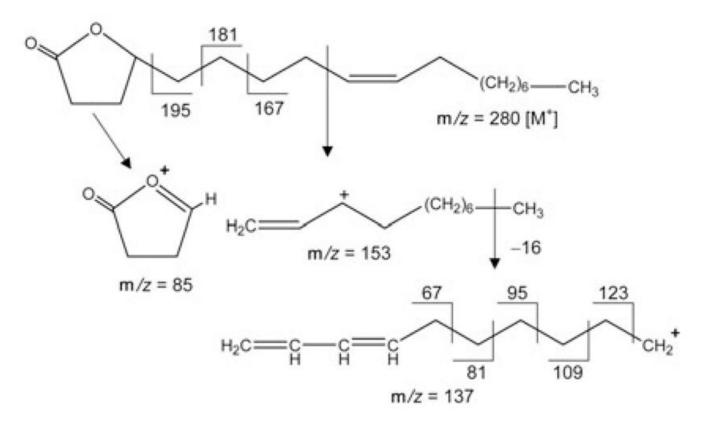
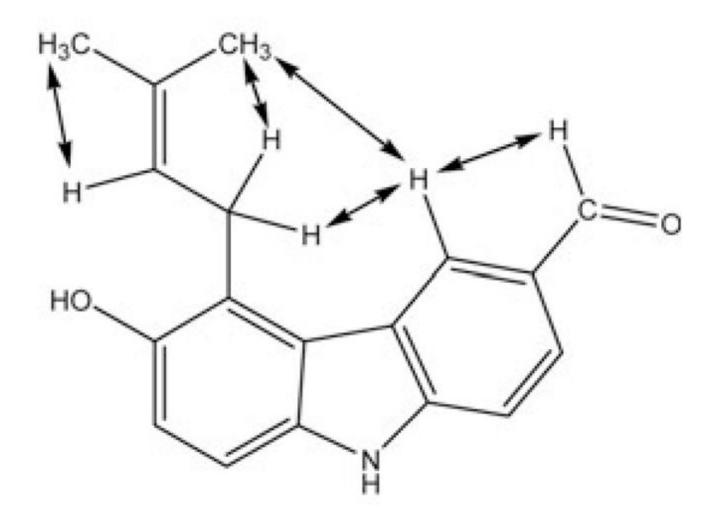


Fig. 2. EI-MS fragmentation pattern of compound **1**.





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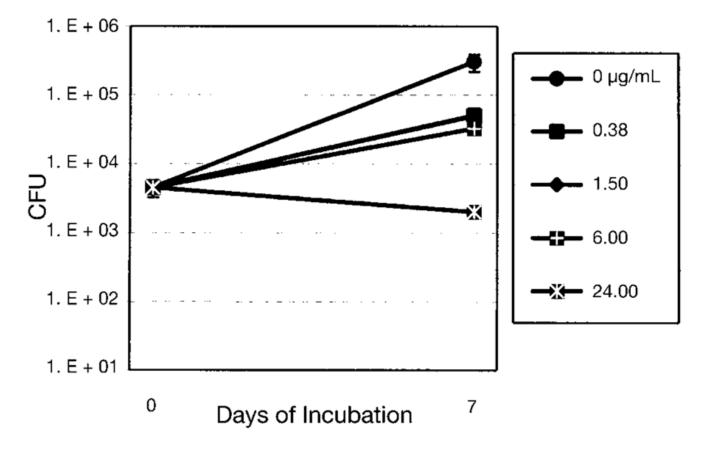


Fig. 4.

Evaluation of **1** in the J774 mouse macrophage cell line infected with *M. tuberculosis* Erdman, a strain somewhat more virulent than H37Rv.

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Position	Carbon			Proton	HMBC (→C)
	$\delta_c[ppm]$	$\delta_{H}[ppm]$	Mult.	J [Hz] (→H)	
1	177.5			-	
2a + b	29.1	2.528	pp	6.84 (3a), 9.51 (3b)	1, 4
3a	28.3	2.319	ddt	6.84 (2), -12.69 (3b), 6.62 (4)	1, 4, 5
q		1.846	ddt	9.51 (2), -12.69 (3a), 5.60 (4)	1, 4, 5
4	81.2	4.484	ddt	6.62 (3a), 5.60 (3b), 6.84 (5a), 7.72 (5b)	1, 5, 6
5a	35.8	1.721	dt	6.84 (4), -11.98 (5b), 5.88 (6)	6, 7
q		1.609	^m a	6.00 (6), 7.72 (4), -11.98 (5a)	
6a + b	25.1	1.408	тa	6.00 (5b), 5.88 (5a), 5.56 (7)	7, 8
7a + b	29.6	1.408	^m a	$5.56(6), 5.50(8), 0.45(9)^b$	6,9
8a + b	27.2	2.039	dt	5.50 (7), 6.81 (9)	6, 10
6	130.7	5.372	dtt	$0.45 (7)^{b}, 6.81 (8), 10.94 (10)$	7, 11
10	129.4	5.328	dtt	$10.94(9), 6.71(11), 0.49(12)^{b}$	8, 12
11a + b	27.5	2.014	dt	6.71 (10)	6
12a + b	29.7	1.269	^m a	0.49~(10)b	
13a + b	29.8	1.269	^m a		
14a + b	29.9	1.269	^m a		
15a + b	30.0	1.269	^m a		

Position Carbon	Carbon				
	$\delta_{c}[ppm]$	ð _c [ppm] ð _H [ppm] Mult.	Mult.	J [Hz] (→H)	
16a + b		32.2 1.269 m ^a	m ^a		18
17a + b	22.9	22.9 1.269 m ^a	_p m		18
18	14.4	14.4 0.881 br t/m	br t/m	6.78 (17)	16, 17

¹Overlapping peaks limited effective iteration, resulting in unresolved exact resonances and J values (see text).

 b An interesting observation is that, instead of the expected allylic⁴J9, 11 and ⁴J8, 10 couplings, W-type ⁴J7, 9 and ⁴J10, 12 long-range couplings are observed.

Table 2

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Position	Cai	Carbon		Proton	HMBC (→C)
	$\delta_c[ppm]$	δ _H [ppm]	Mult. $(spin^b)$	J [Hz] (→H)	
-	111.9	7.596	$dd/m^b(AMN)$	8.5 (2), 0.5 (4)	3, 11
2	126.7	7.913	dd (AMN)	1.6 (4), 8.5 (1)	4, 10, CHO
ю	129.5				
4	127.4	8.648	$dd/m^b(AMN)$	1.6 (2), 0.5 (1)	2, 10, CHO
5	122.8				
9	149.5				
7	116.7	7.114	q	8.5 (8)	5, 6, 13
∞	110.1	7.293	q	8.5 (7)	6, 12
10	145.4				
11	124.1				
12	123.3 ^a	ı			
13	136.1				
1′a + b	26.3	4.023	$\text{dqq}\;(A_2MX_3Y_3)$	6.5 (2'), 1.3 (5'), 0.4 (4')	6, 2'
2,	123.3 <i>a</i>	5.322	tqq $(A_2MX_3Y_3)$	6.5 (1'), 1.2 (5'), 1.5 (4')	
3′	132.9				
4′	18.5	1.991	tdq $(A_2MX_3Y_3)$	0.4 (1'), 1.5 (2'), 0.4 (5')	2', 3'
5'	25.9	1.688	tdq $(A_2MX_3Y_3)$	1.3 (1'), 1.2 (2'), 0.4 (4')	2', 3'
СНО	191.9	10.054	s		33

Position	Carbon	pon		Proton	HMBC $(\rightarrow C)$
	$\delta_{c}[ppm]$	$\delta_{H}[ppm]$	$\delta_c[ppm] \delta_H[ppm] Mult. (spin^b) \qquad J \left[Hz\right] (\rightarrow H)$	J [Hz] (→H)	
HN		10.666 br s	br s		
НО		7.990 br s	br s		

 a Overlapping resonances, assigned to one CH and one C carbon by integration.

b Multiplicities are given under nuclei first order assumptions, and are supplemented by the nomenclatures of the underlying spin system (spin).

Table 3

Anti-tuberculosis and cytotoxic activities, and resulting selectivity indices (SI) of Micromelum constituents 1-7

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		VERO	J774	EC_{90}	EC99	EC ₉₉
CHCl ₃ extract	12.5					
1	1.5 ± 0.4	95	40.71	63	5.6	18.9
2	31.5 ± 0.2	> 102		>3		
3	14.3 ± 0.9	> 102		7<		
4	> 128	> 102				
S	> 128	> 102				
9	42.3 ± 0.5	101				
7	15.6 ± 0.2	> 102		7<		
Rifampin ^e	0.040 ± 0.017	100		2500	0.05	0.18
Streptomycin					0.1	0.3

^bMedium inhibitory concentration (IC50) on Vero cells.

 c Selectivity Index (SI) calculated by the quotient of the VERO IC50 and the MIC90.

d Concentration effecting a reduction of *M. tuberculosis* cfu of 90% (EC90) or 99% relative to untreated control.

 e Positive control; negative control was solvent.