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The emerging role of iron dyshomeostasis in the mitochondrial decay of aging

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1. Introduction

Iron is a trace metal essential for several life-sustaining functions, while excess iron, by virtue of its ability to catalyze the formation of reactive oxygen species (ROS), has the potential to be a causative factor in the age-related mitochondrial deterioration (Kohgo et al. 2008;Duvigneau et al. 2008;Liang et al. 2008). Iron accumulates in senescent cells and most nonhematopoietic tissues with age (Killilea et al. 2003;Killilea et al. 2004;Dunaief 2006;Hofer et al. 2008;Jung et al. 2008). Rapidly emerging evidence suggests that iron accumulation and loss of mitochondrial iron homeostasis may contribute to mitochondrial decay, which subsequently leads to aging (Table 1) (Bitar and Weiner 1983;Atamna et al. 2001;Atamna et al. 2009;Cantu et al. 2006;Doulias et al. 2009;Chen et al. 2009). Although studies in both yeast and mammalian systems support the conclusion that iron homeostasis may be disrupted with age (Zacharski et al. 2000;Gau et al. 2001), the mechanisms underlying this phenomenon are still unclear. Here, we discuss important features of iron dyshomeostasis with a particular emphasis on its effects on mitochondrial decay and aging.

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2. Iron crisis and mitochondrial decay

2.1. Labile iron in mitochondria

Iron taken up by eukaryotic cells must reach mitochondria, the unique site for heme and ironsulfur cluster (ISC) biosynthesis (Dunn et al. 2007;Levi and Rovida 2009). Since mitochondria are also the major source of intracellular ROS and excess iron has a strong catalytic potential to enhance ROS generation, it is important that mitochondrial iron concentration is maintained within a tightly controlled range. In cells and tissues, iron exists in two pools. Ferritin and ironcontaining prosthetic groups in various proteins sequester "non-chelatable" iron that conventional iron chelators like deferoxamine are unable to chelate. The other iron pool is socalled "chelatable or labile" iron that represents both free and loosely bound iron. In hepatocytes, this labile iron is estimated to be about 5 μ M (Ma et al. 2006). Most iron transferred from cytoplasm to mitochondria or delivered from late endosomes and lysosomes to mitochondria is sequestered efficiently by the iron storage proteins, frataxin and mitochondrial ferritin (MtF) (Scheme 1) (Kaur and Andersen 2004; Zhang et al. 2005). With aging, a minor amount of mitochondrial iron, either loosely bound to proteins or escaped from storage sites becomes redox-active, and may be harmful, particularly in the presence of high concentration of hydrogen peroxide within the same compartment (Sohal et al. 1999;Kakhlon and Cabantchik 2002; Doulias et al. 2008). Several studies reported iron accumulation with age in mitochondria in rat substantia nigra (Schipper et al. 1998) and skeletal muscle (Seo et al. 2008) as well as human subcortical brain tissue (Schipper and Cisse 1995). Given the fact that labile iron has a strong catalytic potential to generate ROS, iron overload may result in catastrophic cellular damage via increasing oxidative stress accrual.

Since labile iron is transient and exists in dynamic equilibrium with various cellular components, early attempts to identify the labile iron pool were based on cell-disruptive methods, which in turn alter the equilibrium between free and bound iron, as well as the $Fe^{II}/$ Fe^{III} redox state (Rothman et al. 1992;Kozlov et al. 1992;Sohal et al. 1999). Nondisruptive techniques that rely on the application of fluorescent metalosensors have been developed to estimate the intracellular chelatable iron (Epsztejn et al. 1997;Kakhlon and Cabantchik 2002). Changes in labile iron in cells and tissues can be visualized by fluorescent probes, including phen green SK and calcein (Petrat et al. 2001;Petrat et al. 2002a) since ferrous iron quenches these fluorescence. However, these fluorophores also bind to other divalent cations like Cu, Ni and Co, which raises an issue of their selectivity (Breuer et al. 1995). The development of iron-sensitive fluorescent probes specifically targeting mitochondria has allowed significant advances in the study of labile iron (Petrat et al. 2002b;Rauen et al. 2007). Probes are comprised of a fluorescent group coupled with a high-affinity iron chelator and must fulfill several requirements. Firstly, probes must be lipophilic and highly membranepermeable (Petrat et al. 2001). Secondly, fluorescent groups must possess a net positive charge and be electophoretically driven into mitochondria due to the inside negative membrane potential (Dykens and Stout 2001). Thirdly, chelators must possess relatively high affinity for iron and be able to compete with endogenous ligands (i.e., pyruvate, phosphate, and polypeptides) (Rauen et al. 2007). In light of previous studies using the iron indicator rhodamine B 4-[(1,10-phenanthrolin-5-yl)aminocarbonyl]benzyl ester (RPA) to determine chelatable iron concentration in mitochondria of rat hepatocytes and endothelial cells (Petrat et al. 2001;Petrat et al. 2002b), Rauen et al. (2007) have developed an additional selective mitochondrial iron indicator, rhodamine B 4-[(2,2'-bipyridin-4-yl)aminocarbonyl]benzyl ester (RDA), which detected the same iron concentration $(16.0 \pm 1.9 \,\mu\text{M})$ in rat hepatocyte mitochondria as RPA did (17.0 \pm 1.0 μ M). A recent study by Cantu *et al.* (2009) employing an adenoviral construct technique and the iron indicator RPA, has shown that mitochondrialaconitase (m-aconitase) releases labile iron under oxidative stress in primary ventral mesencephalic cells. This event was followed by mitochondrial dysfunction and cell death. In

addition, the observation in the same study that removing labile iron using an iron chelator mitigated mitochondrial damage and cell death strongly suggests its role in oxidative damage and mitochondrial dysfunction during the aging process. A recent study by Ma et al. (2009) also suggests that iron-sulfur containing proteins are targets of ROS in aging, and that m-aconitase is a major target of ROS under conditions of cellular stress. Upon exposure to oxidants, m-aconitase is disassembled and releases labile iron. Moreover, an age-associated decrease of aconitase expression has been observed in brain, heart, and muscle mitochondria in rodents, which contributes to the sensitivity of the enzyme to oxidative stress, as well as the overload of labile iron observed in aging (Dencher et al. 2007;O'Connell et al. 2007;Prokai et al. 2007). In further support of the role of labile iron, H₂O₂-induced collapse of mitochondrial membrane potential was completely prevented by pre-treatment with the lipophilic iron chelator salicylaldehyde isonicotinoyl hydrazone (SIH) in cultured H9c2 cardiac myoblasts (Simunek et al. 2005;Kurz et al. 2006). This finding indicates that hydrogen peroxide *per se* is not harmful, but it may become highly toxic if labile iron coexists.

2.2. Loss of iron homeostasis with aging

Heme and ISCs are important for the assembly of electron transfer complexes. Alterations in mitochondrial iron homeostasis cause iron accumulation in this compartment, which may be responsible for the age-related mitochondrial deterioration. Interestingly, multiple defects in mitochondrial heme and ISC biosynthesis have been demonstrated with aging, which may result in bioenergetic crisis and genomic instability (Rouault and Tong 2005;Veatch et al. 2009).

In the early 1980's, Bitar and Walter (1983) observed that heme biosynthesis declined with aging. This phenomenon was investigated extensively by the Ames group, leading to important findings in this area (Atamna et al. 2001;Atamna et al. 2002a;Atamna et al. 2002b). For instance, heme deficiency in senescent human fibroblasts was found to selectively decrease the expression and activity of cytochrome *c* oxidase (complex IV) (Atamna et al. 2001). Complex IV, the terminal oxidase in mitochondrial electron transfer chain, is responsible for generating a transmembrane proton gradient across the inner mitochondrial membrane (Saraste 1999;Atamna et al. 2002b). Therefore, age-associated heme deficiency leads to impaired mitochondrial energy production through depressing complex IV. In a follow-up study, a decreased activity of complex IV resulting from defective heme biosynthesis was also observed in human brain cell lines and rat primary hippocampal neurons (Atamna et al. 2002a), further suggesting an association among age-related heme deficiency, mitochondrial decay and the aging process.

A recent study in yeast established a link between defects in ISC biogenesis and age-associated genomic instability (Veatch et al. 2009). Aging yeast cells lose their mitochondrial DNA (mtDNA) over time, leading to decreased inner mitochondrial membrane potential and mitochondrial dysfunction. The reduction of mitochondrial membrane potential, in turn, contributes to the defect in the transport of iron-sulfur proteins into and out of the mitochondria, which is required for mitochondrial ISC assembly and maturation of cytosolic and nuclear iron-sulfur proteins (reviewed in Lill et al., 2008). Defects in the mitochondrial ISC machineries result in an impaired iron homeostasis with increased cellular iron acquisition, iron regulon activation and iron accumulation in mitochondria. The function of iron-sulfur proteins in both the mitochondrial compartment and throughout the cell is either reduced or lost, and eventually cells fail to maintain nuclear genome intergrity.

2.3. Iron manipulation and longevity

The first indication that inhibition of iron absorption could extend life span was provided by Massie et al. (1993), who studied male *Drosophila* fed an iron-rich diet and tea extracts. The

study demonstrated that increased dietary iron reduced longevity and dietary tea prevented the age-related iron accumulation in Drosophila. Brune et al. (1989) and Rossander-Hulthen et al. (1996) reported that the galloyl group in tea polyphenols was responsible for the inhibitory effects, suggesting a unique role of tea polyphenols in iron binding. However, it is possible that polyphenolic compounds in tea extracts exerted a direct antioxidant effect in addition to the inhibition of iron absorption as proposed and that both mechanisms contributed to the reported increase in lifespan. In fact, other studies suggest different mechanisms that may underlie the life extension effects of tea polyphenolics in Drosophila (Li et al. 2007;Berletch et al. 2008; Peng et al. 2009). Additional genetic studies in C. elegans and Drosophila have established a link between frataxin and longevity by demonstrating that frataxin deficiency in C. elegans led to shorter lifespan (Vazquez-Manrique et al. 2006;Zarse et al. 2007) and that frataxin overexpression in Drosophila extended life span (Runko et al. 2008). By stressing Drosophila with paraquat and hydrogen peroxide, Runko et al. (2008) found that transgenic flies were resistant to oxidative stress and that frataxin overexpression extended the median and maximum life span as much as 35% and 28%, respectively. However, the observation that reducing the expression of frataxin by utilizing a bacterial feeding RNAi against the nematode ortholog frh-1 prolongs life span in nematodes (Ventura et al. 2005) provides evidence that there is a threshold effect on life extension in nematodes. Activation of compensatory pathways may allow nematodes with frataxin reduction to exhibit a life-extension phenotype (Rea et al. 2007). These observations also highlight a fundamental difference in frataxin requirements between mammals and nematodes and suggest that the other findings linking reduced mitochondrial function and life extension in invertebrates may not apply to mammals. Aside from the studies in Drosophila (Massie et al. 1993;Runko et al. 2008), little work has been conducted to investigate the effects of frataxin overexpression and iron restriction on longevity in rodents. Such investigations are required to determine whether frataxin overexpression and iron restriction may eventually be employed as a strategy to modulate aging in humans.

2.4. Diseases related to mitochondrial iron overload

Although the precise molecular components regulating the mitochondrial iron pool are still unknown, frataxin, a 17-KDa mitochondrial protein, is considered central to mitochondrial iron homeostasis (see Scheme 1). Studies in yeast models showed that frataxin was involved in heme biosynthesis (Lange et al. 1999;Seguin et al. 2009), ISC assembly (Muhlenhoff et al. 2002), aconitase repair (Bulteau et al. 2004), and iron storage (Gakh et al. 2006;Gakh et al. 2008;Correia et al. 2010). Ablation of frataxin results in mitochondrial iron overload and Friedreich's ataxia, a major inheritable neurodegenerative disorder (Pandolfo and Pastore 2009). MtF is another mitochondrial iron storage protein that has been identified so far in testis, neuronal cells and islets of Langherans, and is thought to play an important role in iron storage and regulation in mitochondria (Levi and Arosio 2004; Santambrogio et al. 2007). Levels of MtF are significantly elevated in erythroblasts from patients with sideroblastic anemia, implying that MtF may be induced by iron overloading to sequester excess iron in mitochondria (Drysdale et al. 2002). In agreement with its protective role, studies in frataxin-deficient yeast (Campanella et al. 2004; Campanella et al. 2009) and frataxin-silenced HeLa cells (Zanella et al. 2008) showed that human MtF expression prevented the accumulation of mitochondrial iron, maintained mtDNA integrity, and increased the resistance to oxidative stress by rescuing mitochondrial respiration. Recently, mitochondrial glutaredoxin 5, a thiol-disulfide oxidoreductase, has been shown to be essential for ISC biogenesis and the maintenance of normal mitochondrial and cytosolic iron homeostasis in human RD4 and COS cells (Tong and Rouault 2000). In glutaredoxin 5 deficient HeLa cells, ISC biosynthesis and mitochondrial iron trafficking were impaired, causing mitochondrial iron overload and concomitant cytosolic iron depletion (Ye et al. 2010). In agreement with these observations, glutaredoxin 5 deficiency was associated with sideroblastic anemia in human patients (Camaschella et al. 2007). Besides importer-mediated iron transport to mitochondria, physical interaction between endosomes and

mitochondria has also been proposed as a possible mechanism delivering cellular labile iron into the mitochondrion (Zhang et al. 2005;Sheftel et al. 2007). *Sec1511*, a mammalian ortholog of the yeast SEC15, has been suggested as a component of the docking machinery between the endosome and the mitochondrion (White et al. 2005). The absence of *Sec1511* causes anemia in mice due to improper iron trafficking in the erythroid transferrin cycle (Lim et al. 2005). Unlike iron import pathways, most iron is exported from mitochondria in the form of heme and ISCs through various transporters (see Scheme 1) (Levi and Rovida 2009). The breast cancer resistance protein (ABCG2), the feline leukemic virus subgroup C receptor (FLVCR), and the ABC-mitochondrial erythroid (ABC-me) are necessary for heme export, whereas ABCB7 is needed for ISCs transport from the mitochondrion to the cytoplasm (Shirihai et al. 2000;Cavadini et al. 2007;Zutz et al. 2009). Defects in these transporters impair mitochondrial iron homeostasis and lead to cellular degeneration and death (Dunn et al. 2007;Rouault and Tong 2008).

Taken together, recent studies on mitochondrial iron overload diseases suggest that redoxactive iron accumulation in mitochondria with age may be responsible for increases in cellular and mitochondrial oxidative stress and mitochondrial function decline. These alterations, in turn, may be involved in the pathogenesis of age-related neuro-muscular degeneration as well as in the aging process as a whole.

2.5. Iron accumulation and sarcopenia

The age-related loss of muscle mass and strength, referred to as sarcopenia of aging, is a highly prevalent condition among older adults and severely impacts functional status, quality of life and mortality (Marzetti and Leeuwenburgh 2006). Older individuals are also especially vulnerable to disuse-induced muscle atrophy (Edgerton et al. 2002), the recovery from which is impaired at old age (Zarzhevsky et al. 2001). The socio-economic burden associated with sarcopenia and disuse muscle atrophy has instigated an intensive research on the etiopathogenesis of these conditions, leading to the discovery of several potential contributing factors (Marzetti et al. 2009b). In particular, recent experimental evidence indicates that abnormal iron homeostasis may be involved in the pathogenesis of both sarcopenia (Altun et al. 2007;Hofer et al. 2008;Seo et al. 2008;Jung et al. 2008;Xu et al. 2008) and acute muscle atrophy in old animals (Kondo et al. 1992;Hofer et al. 2008). Hofer et al. (2008) recently demonstrated that non-heme iron levels were over 2-fold higher in the gastrocnemius muscle of old rats compared to younger controls. Acute atrophy induced by hind limb suspension resulted in a further elevation in muscle non-heme iron content in old, but not young rats. Notably, histochemical analysis revealed that iron accumulated in atrophied rather than normal fibers, suggesting a mechanistic link between iron overload and the loss of muscle mass. In addition, Kondo et al. (1992) demonstrated that the accrual of oxidative damage and the severity of muscle atrophy in hind limb immobilized rats were significantly attenuated by the administration of the iron chelator deferoxamine. More recently, Xu et al. (2008) found that total non-heme iron levels increased progressively over the course of aging in the rat gastrocnemius muscle and correlated with decreased muscle mass and grip strength. Importantly, lifelong 40% calorie restriction, an intervention known to mitigate the severity of sarcopenia in rodents, preserved muscle iron homeostasis into very old age (Marzetti et al. 2009b). This adaptation was paralleled by a reduced decline in muscle mass and strength, further supporting a role for muscle iron overload in the pathogenesis of sarcopenia.

Since the mitochondrion is one of the most important sites for cellular iron utilization (Levi and Rovida 2009), Seo *et al.* (2008) investigated whether mitochondrial iron homeostasis was altered in the aged muscle. Advanced age was associated with increased mitochondrial nonheme iron levels in the rat quadriceps muscle, which correlated with levels of mitochondrial RNA (mtRNA) oxidation and susceptibility to mPTP opening. This finding is of particular

relevance, since mPTP opening is considered a central event in the induction of apoptosis, which in turn is thought to be involved in the pathogenesis of sarcopenia and acute muscle atrophy (Marzetti et al. 2009a). Further research is warranted to unveil the mechanisms underlying muscle iron dyshomeostasis during aging as well as in other atrophying conditions. This knowledge will likely allow the development of new therapeutic tools for the prevention and treatment of sarcopenia and disuse muscle atrophy.

2.6. Iron accumulation and brain aging

During aging a progressive structural derangement of biomolecules and cellular compartments takes places in the brain, causing a decline in motor plasticity and cognitive performance (Moos and Morgan 2004;Droge and Schipper 2007;Stankiewicz and Brass 2009). The mechanisms responsible for brain aging are complex and not yet completely understood; however, growing evidence indicates that the aging process partly stems from the accumulation of damage to macromolecules and cell components caused by distorted cellular redox balance and aberrant metal homeostasis (Levenson 2005;Lee and Andersen 2010). In fact, oxidant-mediated damages of macromolecules disturb redox-sensitive signaling (Altamura and Muckenthaler 2009) and cause mitochondrial dysfunction (Lin and Beal 2006) as well as brain cell death (Mattson 2006). Additionally, accumulation of metal ions, especially iron, has been observed in the aged central nervous system in humans (Aquino et al. 2009) and animal models (Hahn et al. 2009) as well as in disorders such as Alzheimer's disease (Honda et al. 2007;Bartzokis et al. 2007), Parkinson's disease (Lee and Andersen 2010) and Friedreich's ataxia (Pandolfo and Pastore 2009).

In the 1950s, Hallgren and Sourander (1958) reported non-heme iron accumulation in several regions of the aged human brain such as putamen, motor cortex, prefrontal cortex, sensory cortex and thalamus (Hallgren and Sourander 1958). In addition to these observations, studies in post-mortem brains showed an age-dependent accumulation of ferritin, indicative of increased storage iron, in different brain regions (Thomas et al. 1993;Connor et al. 1995;Zecca et al. 2001). Recent studies using high-field magnetic resonance imaging revealed an increase in iron levels in various regions of the aged brain (Pfefferbaum et al. 2009;Aquino et al. 2009;Peran et al. 2009;Cherubini et al. 2009). Excessive iron content in the aged brain may generate cellular toxic stress, which partly explains the age-related decline in cognitive performance and other neurodegenerative disorders (Polla et al. 2003;Zecca et al. 2004;Sen et al. 2007;Hahn et al. 2009).

Neurodegenerative diseases associated with iron accumulation in the brain, especially in the basal ganglia, are caused by specific gene mutations (Gregory et al. 2009). For instance, a defect in the *PANK2* gene encoding pantothenate kinase 2 causes pantothenate kinase-associated neurodegeneration, in which pathologic accumulation of iron in the brain is observed (Zhou et al. 2001;Gregory and Hayflick 2005). Patients lacking circulating serum ceruloplasmin and ferritin light polypeptide bear mutations in the *CP* and *FTL* genes, respectively (Curtis et al. 2001;Texel et al. 2008). About half of cases of infantile neuroaxonal dystrophy and atypical neuroaxonal dystrophy exhibit iron deposition in the brain, which is associated with mutations in the gene *PLA2G6*. Much research effort is directed at exploration of noninvasive therapeutics to combat these neurodegenerative diseases associated with brain iron accumulation (Miyajima et al. 1997;Gregory et al. 2009).

3. Future research

If indeed iron accumulation and alterations in labile iron are significant factors in the aging process, iron scavenging and removal might therefore prevent cellular and mitochondrial oxidative damage and attenuate age-related mitochondrial decay. This goal could be achieved

iron chelation may be beneficial in the treatment of iron overload diseases, such as Alzheimer's disease (Weinberg and Miklossy 2008;Liu et al. 2010), Parkinson's disease (Kaur and Andersen 2004;Ghosh et al. 2010), Friedreich's ataxia (Whitnall et al. 2008;Goncalves et al. 2008) and retinal disease (Dunaief 2006;Lukinova et al. 2009). An important issue associated with iron chelation therapy is that compounds available to date do not possess enough selectivity for organs or macromolecular structures, and once penetrated in tissues chelate iron indiscriminately. This may dramatically limit the use of iron chelators in elderly people, in whom iron-deficient anemia is highly prevalent (Darnton-Hill et al. 2005). Therefore, we need safer therapeutic interventions with a high selectivity to reduce labile iron levels while maintaining the bioavailable iron pool.

Calorie restriction, the only non-genetic intervention extending life and health span in all organisms studied to date, has been shown to be effective in alleviating the age-associated iron accumulation in rat muscle, liver, brain and kidney (Cook and Yu 1998;Xu et al. 2008). However, one major issue associated with calorie restriction involves feasibility and tolerability, especially for the old frail elderly. Tea polyphenolics, on the other hand, show promise in terms of feasibility and may provide a valid alternative to calorie restriction to reduce iron absorption, but this intervention has not been fully tested. Thus, a major research challenge will be the development of novel, safe and feasible interventions to preserve iron homeostasis into old age.

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Scheme 1. Schematic illustration of mitochondrial iron dyshomeostasis with aging

Iron is transported into the mitochondrial matrix by iron importers (e.g. mitoferrin) where it can be directed to different pathways, including storage in frataxin, iron-sulfur cluster (ISC) biosynthesis, heme metabolism, mitochondrial ferritin (MtF) or other currently unknown pathways. The ISCs can be exported to the cytoplasma by ABCB7. Heme is thought to be exported from the mitochondrion by several pathways, including ABCG2, the feline leukemia virus subgroup-C receptor (FLVCR) and ABC-me. Defects in these transporters or defective biosynthesis of heme and ISCs with age impair mitochondrial iron homeostasis and lead to cellular degeneration. Increased labile iron with age, especially in mitochondria, has a strong potential to catalyze the generation of reactive oxygen species (ROS), resulting in cellular damage.

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Table 1

Summary of studies reporting an association of iron accumulation, mitochondrial dysfunction and aging.

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Stress	Pathway/mechanism	Effects	Prevention	Models	Selected references
Frataxin deficiency	Mitochondrial ISCs defects, decreased Mn-SOD activity	Increased iron levels, increased protein carbonylation	Copper and manganese treatments	Yeast	Napoli et al. (2006), Irazusta et al. (2009)
Paraquat	Iron release from m-aconitase	Increased mitochondrial iron, cell death	Iron chelation	Cell line	Cantu <i>et al.</i> (2009)
Aging	Mitochondrial ISCs defects trigger the activation of the iron regulon	Increased iron uptake, genomic instability	N/A	Yeast	Veatch <i>et al.</i> (2009)
Aging	N/A	Increased mitochondrial iron, permeability transition	CR	Rat	Seo et al. (2008), Xu et al. (2008)
Aging	Inactivation or degradation of IRPs	Free iron overload, increased ferritin, decreased transferritin, decreased lysosomal activity	N/A	Rat and mouse	Chen <i>et al.</i> (2009)
Aging	Deficiency in mitochondrial heme synthesis	Increased iron levels, increased ferrochelatase, loss of complex IV	N/A	Cell line and rat	Bitar and Weiner (1983), Atamna <i>et al.</i> (2001, 2002a)
Aging	N/A	Increased labile iron	N/A	Human	Doulias et al. (2008)
1-14					

Abbreviations: m-aconitase, mitochondrial aconitase; IRPs, iron-regulatory proteins; SOD, superoxide dismutase; CR, calorie restriction; N/A, not available; ISCs, iron-sulfur clusters.