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2-Formyl-3-hydroxy-9,10-anthroquinone

Nor Hadiani Ismail,[a] Che Puteh Osman,[a] Khalijah Awang,[b] Sri Nurestri Abdul Malek[c] and Seik Weng Ng[b]*

[a]Faculty of Applied Sciences, Universiti Teknologi MARA, 40450 Shah Alam, Malaysia, [b]Department of Chemistry, University of Malaya, 50603 Kuala Lumpur, Malaysia, and [c]Institute of Biological Sciences, University of Malaya, 50603 Kuala Lumpur, Malaysia
Correspondence e-mail: seikweng@um.edu.my

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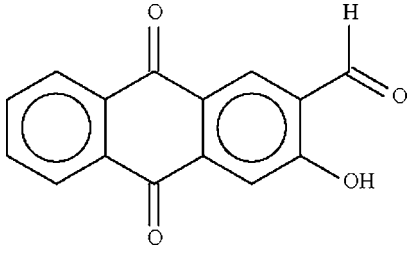
Key indicators: single-crystal X-ray study; T = 100 K; mean σ(C–C) = 0.006 Å; R factor = 0.087; wR factor = 0.343; data-to-parameter ratio = 14.0.

The molecule of the title compound, $C_{15}H_8O_4$, is approximately planar. An intramolecular O—H···O hydrogen bond is observed between the hydroxy and formyl groups. The crystal used was a nonmerohedral twin, with a minor twin component of 15.9%.

Related literature

For antileshmanial and antiplasmodial activities, see: Sittie *et al.* (1999). For the treatment of twinned diffraction data, see: Spek (2003).



Experimental

Crystal data

$C_{15}H_8O_4$
M_r = 252.21
Triclinic, $P\overline{1}$
a = 6.9194 (2) Å
b = 8.0650 (2) Å
c = 10.7601 (3) Å
α = 86.250 (2)°
β = 83.214 (2)°
γ = 64.692 (2)°
V = 538.96 (3) Å^3
Z = 2
Mo $K\alpha$ radiation
μ = 0.11 mm^{-1}
T = 100 (2) K
0.22 × 0.04 × 0.04 mm

Data collection

Bruker SMART APEXII area-detector diffractometer
Absorption correction: none
4946 measured reflections
2419 independent reflections
1880 reflections with $I > 2\sigma(I)$
R_{int} = 0.024

Refinement

$R[F^2 > 2\sigma(F^2)]$ = 0.087
$wR(F^2)$ = 0.343
S = 1.11
2419 reflections
173 parameters
H-atom parameters constrained
$\Delta\rho_{max}$ = 0.49 e Å^{-3}
$\Delta\rho_{min}$ = −0.44 e Å^{-3}

Table 1
Hydrogen-bond geometry (Å, °).

D—H···A	D—H	H···A	D···A	D—H···A
O2—H2···O1	0.84	2.00	2.635 (5)	132

Data collection: *APEX2* (Bruker, 2007); cell refinement: *SAINT* (Bruker, 2007); data reduction: *SAINT*; program(s) used to solve structure: *SHELXS97* (Sheldrick, 2008); program(s) used to refine structure: *SHELXL97* (Sheldrick, 2008); molecular graphics: *X-SEED* (Barbour, 2001); software used to prepare material for publication: *publCIF* (Westrip, 2008).

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Supplementary data and figures for this paper are available from the IUCr electronic archives (Reference: CI2681).

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supplementary materials

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2-Formyl-3-hydroxy-9,10-anthroquinone

N. H. Ismail, C. P. Osman, K. Awang, S. N. Abdul Malek and S. W. Ng

Experimental

Rennellia elliptica Korth from the Rubiaceae family was collected from Kuala Keniam, Pahang, Malaysia. The root was chopped into small pieces and dried. The dried sample (1 kg) was ground and then extracted successively with hexane, dichloromethane and methanol. The dichloromethane extract was concentrated *in vacuo* to give 27 g crude extract. The crude extract was fractionated by column chromatography. The column (60 cm *X* 5 cm) was packed with acid-washed silica gel and eluted with hexane, dichloromethane and methanol. Nine fractions were obtained, and 3-hydroxy-2-formyl-9,10-anthraquinone (41.5 mg) was isolated from the third fraction (hexane:dichloromethane, 30:70) by slow evaporation of the solvent mixture. The yellow crystals obtained were washed with acetone.

Refinement

Carbon- and oxygen-bound H-atoms were placed in calculated positions (C—H = 0.95 Å and O—H = 0.84 Å) and were included in the refinement in the riding model approximation, with U(H) set to 1.2–1.5U_{eq}(C,*O*). The crystal studied was a non-merohedral twin. The *TwinRotMat* in *PLATON* (Spek, 2003) gave the twin law as (-1 0 0, 0 - 1 0, -0.343 - 0.049 1), whose inclusion in the refinement lowered the *R* index from 11.3 to 8.7%. The twin component refined to 18.9%. The refinement is deemed satisfactory although the wR_2 value for all reflections is somewhat high. The structure has a long C5–C14 bond; as the anisotropic displacement parameters are normal, the likely reason is localization of the double bonds in the ring. On the other hand, the C13–C14 bond is somewhat short.

Figures

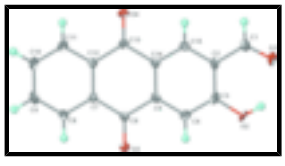

Fig. 1. The molecular structure of the title compound, showing 70% probability displacement ellipsoids. H atoms are drawn as spheres of arbitrary radii.

2-Formyl-3-hydroxy-9,10-anthroquinone

Crystal data

$C_{15}H_8O_4$	$Z = 2$
$M_r = 252.21$	$F_{000} = 260$
Triclinic, $P\bar{1}$	$D_x = 1.554$ Mg m^{-3}
Hall symbol: -P 1	Mo $K\alpha$ radiation $\lambda = 0.71073$ Å
$a = 6.9194$ (2) Å	Cell parameters from 1888 reflections
$b = 8.0650$ (2) Å	$\theta = 3.3$–$28.3°$

c = 10.7601 (3) Å	μ = 0.11 mm^{-1}
α = 86.250 (2)°	T = 100 (2) K
β = 83.214 (2)°	Block, yellow
γ = 64.692 (2)°	0.22 × 0.04 × 0.04 mm
V = 538.96 (3) Å^3	

Data collection

Bruker SMART APEXII area-detector diffractometer	1880 reflections with $I > 2\sigma(I)$
Radiation source: fine-focus sealed tube	R_{int} = 0.024
Monochromator: graphite	θ_{max} = 27.5°
T = 100(2) K	θ_{min} = 2.8°
ω scans	h = −8→8
Absorption correction: none	k = −10→10
4946 measured reflections	l = −13→13
2419 independent reflections	

Refinement

Refinement on F^2	Secondary atom site location: difference Fourier map
Least-squares matrix: full	Hydrogen site location: inferred from neighbouring sites
$R[F^2 > 2\sigma(F^2)]$ = 0.087	H-atom parameters constrained
$wR(F^2)$ = 0.343	$w = 1/[\sigma^2(F_o^2) + (0.1778P)^2 + 2.2381P]$ where $P = (F_o^2 + 2F_c^2)/3$
S = 1.11	$(\Delta/\sigma)_{max}$ = 0.001
2419 reflections	$\Delta\rho_{max}$ = 0.49 e Å^{-3}
173 parameters	$\Delta\rho_{min}$ = −0.44 e Å^{-3}
Primary atom site location: structure-invariant direct methods	Extinction correction: none

Fractional atomic coordinates and isotropic or equivalent isotropic displacement parameters (Å^2)

	x	y	z	U_{iso}*/U_{eq}
O1	0.3155 (6)	0.6486 (5)	0.0888 (3)	0.0251 (8)
O2	0.2501 (6)	0.3493 (5)	0.0941 (3)	0.0242 (8)
H2	0.2611	0.4345	0.0490	0.029*
O3	0.2366 (5)	−0.0898 (4)	0.4383 (3)	0.0153 (7)
O4	0.2779 (5)	0.4648 (4)	0.6688 (3)	0.0172 (7)
C1	0.3060 (7)	0.6395 (6)	0.2041 (4)	0.0178 (9)
H1	0.3171	0.7336	0.2474	0.021*
C2	0.2786 (7)	0.4901 (6)	0.2772 (4)	0.0143 (9)
C3	0.2544 (7)	0.3500 (6)	0.2187 (4)	0.0161 (9)
C4	0.2356 (7)	0.2059 (6)	0.2894 (4)	0.0152 (9)
H4	0.2187	0.1116	0.2500	0.018*
C5	0.2419 (6)	0.2016 (5)	0.4183 (4)	0.0123 (8)

C6	0.2335 (6)	0.0406 (5)	0.4916 (4)	0.0112 (8)
C7	0.2285 (6)	0.0417 (5)	0.6299 (4)	0.0112 (8)
C8	0.2117 (7)	−0.1043 (6)	0.7003 (4)	0.0139 (8)
H8	0.1986	−0.1999	0.6601	0.017*
C9	0.2144 (7)	−0.1091 (6)	0.8286 (4)	0.0179 (9)
H9	0.2031	−0.2083	0.8766	0.021*
C10	0.2336 (7)	0.0310 (6)	0.8883 (4)	0.0188 (9)
H10	0.2353	0.0269	0.9767	0.023*
C11	0.2503 (7)	0.1762 (6)	0.8187 (4)	0.0173 (9)
H11	0.2645	0.2709	0.8594	0.021*
C12	0.2462 (6)	0.1835 (5)	0.6895 (4)	0.0113 (8)
C13	0.2641 (6)	0.3403 (6)	0.6165 (4)	0.0126 (8)
C14	0.2621 (6)	0.3426 (5)	0.4787 (4)	0.0120 (8)
C15	0.2815 (7)	0.4851 (6)	0.4072 (4)	0.0133 (8)
H15	0.2968	0.5801	0.4468	0.016*

Atomic displacement parameters (Å^2)

	U^{11}	U^{22}	U^{33}	U^{12}	U^{13}	U^{23}
O1	0.038 (2)	0.0258 (18)	0.0125 (16)	−0.0155 (16)	−0.0033 (13)	0.0054 (13)
O2	0.046 (2)	0.0283 (18)	0.0063 (15)	−0.0226 (17)	−0.0062 (13)	0.0020 (12)
O3	0.0194 (15)	0.0131 (14)	0.0147 (15)	−0.0079 (12)	−0.0028 (11)	−0.0009 (11)
O4	0.0228 (16)	0.0148 (15)	0.0165 (15)	−0.0103 (13)	−0.0003 (12)	−0.0039 (11)
C1	0.021 (2)	0.017 (2)	0.016 (2)	−0.0078 (17)	−0.0040 (16)	0.0035 (16)
C2	0.0146 (19)	0.0149 (19)	0.0119 (19)	−0.0053 (16)	−0.0011 (14)	0.0030 (15)
C3	0.017 (2)	0.021 (2)	0.0103 (19)	−0.0080 (17)	−0.0030 (15)	0.0010 (15)
C4	0.016 (2)	0.016 (2)	0.014 (2)	−0.0074 (16)	−0.0007 (15)	−0.0025 (15)
C5	0.0106 (18)	0.0107 (18)	0.0135 (19)	−0.0032 (14)	0.0010 (14)	0.0003 (14)
C6	0.0099 (17)	0.0105 (18)	0.0121 (19)	−0.0036 (14)	−0.0001 (14)	−0.0004 (14)
C7	0.0109 (18)	0.0114 (18)	0.0103 (18)	−0.0038 (14)	−0.0025 (13)	0.0011 (14)
C8	0.0145 (19)	0.0118 (18)	0.015 (2)	−0.0053 (15)	−0.0023 (15)	0.0013 (14)
C9	0.019 (2)	0.017 (2)	0.017 (2)	−0.0081 (17)	−0.0012 (16)	0.0055 (16)
C10	0.024 (2)	0.022 (2)	0.0110 (19)	−0.0100 (18)	−0.0009 (16)	0.0026 (16)
C11	0.021 (2)	0.018 (2)	0.013 (2)	−0.0084 (17)	−0.0008 (16)	−0.0009 (15)
C12	0.0108 (18)	0.0112 (18)	0.0109 (18)	−0.0039 (14)	−0.0002 (14)	−0.0005 (14)
C13	0.0112 (18)	0.0125 (19)	0.0139 (19)	−0.0046 (15)	−0.0015 (14)	−0.0010 (14)
C14	0.0116 (18)	0.0112 (18)	0.0127 (19)	−0.0044 (14)	0.0001 (14)	−0.0002 (14)
C15	0.0138 (19)	0.0112 (18)	0.014 (2)	−0.0049 (15)	0.0006 (14)	−0.0017 (14)

Geometric parameters (Å, °)

O1—C1	1.234 (5)	C7—C8	1.398 (5)
O2—C3	1.345 (5)	C7—C12	1.403 (5)
O2—H2	0.84	C8—C9	1.380 (6)
O3—C6	1.222 (5)	C8—H8	0.95
O4—C13	1.226 (5)	C9—C10	1.397 (6)
C1—C2	1.464 (6)	C9—H9	0.95
C1—H1	0.95	C10—C11	1.388 (6)
C2—C15	1.400 (6)	C10—H10	0.95

C2—C3	1.407 (6)	C11—C12	1.391 (6)
C3—C4	1.391 (6)	C11—H11	0.95
C4—C5	1.390 (6)	C12—C13	1.487 (5)
C4—H4	0.95	C13—C14	1.483 (6)
C5—C14	1.410 (6)	C14—C15	1.387 (6)
C5—C6	1.494 (5)	C15—H15	0.95
C6—C7	1.485 (5)		
C3—O2—H2	120.0	C9—C8—H8	120.1
O1—C1—C2	122.8 (4)	C7—C8—H8	120.1
O1—C1—H1	118.6	C8—C9—C10	120.4 (4)
C2—C1—H1	118.6	C8—C9—H9	119.8
C15—C2—C3	119.7 (4)	C10—C9—H9	119.8
C15—C2—C1	119.1 (4)	C11—C10—C9	120.0 (4)
C3—C2—C1	121.2 (4)	C11—C10—H10	120.0
O2—C3—C4	117.9 (4)	C9—C10—H10	120.0
O2—C3—C2	121.8 (4)	C10—C11—C12	120.2 (4)
C4—C3—C2	120.3 (4)	C10—C11—H11	119.9
C5—C4—C3	119.2 (4)	C12—C11—H11	119.9
C5—C4—H4	120.4	C11—C12—C7	119.6 (4)
C3—C4—H4	120.4	C11—C12—C13	119.4 (4)
C4—C5—C14	121.2 (4)	C7—C12—C13	121.1 (3)
C4—C5—C6	118.4 (4)	O4—C13—C14	121.0 (4)
C14—C5—C6	120.3 (4)	O4—C13—C12	121.0 (4)
O3—C6—C7	121.3 (4)	C14—C13—C12	118.0 (3)
O3—C6—C5	120.5 (4)	C15—C14—C5	119.0 (4)
C7—C6—C5	118.1 (3)	C15—C14—C13	119.7 (4)
C8—C7—C12	120.1 (4)	C5—C14—C13	121.3 (4)
C8—C7—C6	118.9 (4)	C14—C15—C2	120.5 (4)
C12—C7—C6	121.0 (3)	C14—C15—H15	119.8
C9—C8—C7	119.7 (4)	C2—C15—H15	119.8
O1—C1—C2—C15	176.5 (4)	C10—C11—C12—C7	−1.0 (6)
O1—C1—C2—C3	−2.1 (7)	C10—C11—C12—C13	180.0 (4)
C15—C2—C3—O2	179.8 (4)	C8—C7—C12—C11	1.0 (6)
C1—C2—C3—O2	−1.7 (7)	C6—C7—C12—C11	−177.2 (4)
C15—C2—C3—C4	−0.7 (7)	C8—C7—C12—C13	−180.0 (3)
C1—C2—C3—C4	177.9 (4)	C6—C7—C12—C13	1.8 (6)
O2—C3—C4—C5	179.3 (4)	C11—C12—C13—O4	−1.5 (6)
C2—C3—C4—C5	−0.3 (7)	C7—C12—C13—O4	179.4 (4)
C3—C4—C5—C14	1.4 (6)	C11—C12—C13—C14	179.2 (4)
C3—C4—C5—C6	−176.7 (4)	C7—C12—C13—C14	0.2 (6)
C4—C5—C6—O3	5.1 (6)	C4—C5—C14—C15	−1.6 (6)
C14—C5—C6—O3	−173.0 (4)	C6—C5—C14—C15	176.4 (3)
C4—C5—C6—C7	−176.8 (4)	C4—C5—C14—C13	178.7 (4)
C14—C5—C6—C7	5.1 (6)	C6—C5—C14—C13	−3.2 (6)
O3—C6—C7—C8	−4.6 (6)	O4—C13—C14—C15	1.6 (6)
C5—C6—C7—C8	177.4 (4)	C12—C13—C14—C15	−179.1 (4)
O3—C6—C7—C12	173.7 (4)	O4—C13—C14—C5	−178.7 (4)
C5—C6—C7—C12	−4.4 (6)	C12—C13—C14—C5	0.6 (6)

C12—C7—C8—C9	−0.5 (6)	C5—C14—C15—C2	0.7 (6)
C6—C7—C8—C9	177.7 (4)	C13—C14—C15—C2	−179.6 (4)
C7—C8—C9—C10	0.0 (7)	C3—C2—C15—C14	0.4 (6)
C8—C9—C10—C11	0.0 (7)	C1—C2—C15—C14	−178.1 (4)
C9—C10—C11—C12	0.5 (7)		

Hydrogen-bond geometry (Å, °)

D—H···*A*	*D*—H	H···*A*	*D*···*A*	*D*—H···*A*
O2—H2···O1	0.84	2.00	2.635 (5)	132

Fig. 1