

Poly[[μ_2 -aqua-tetraaqua-hexakis(μ_4 -naphthalene-2,6-dicarboxylato)tetraholmium(III)] 1.75-hydrate]

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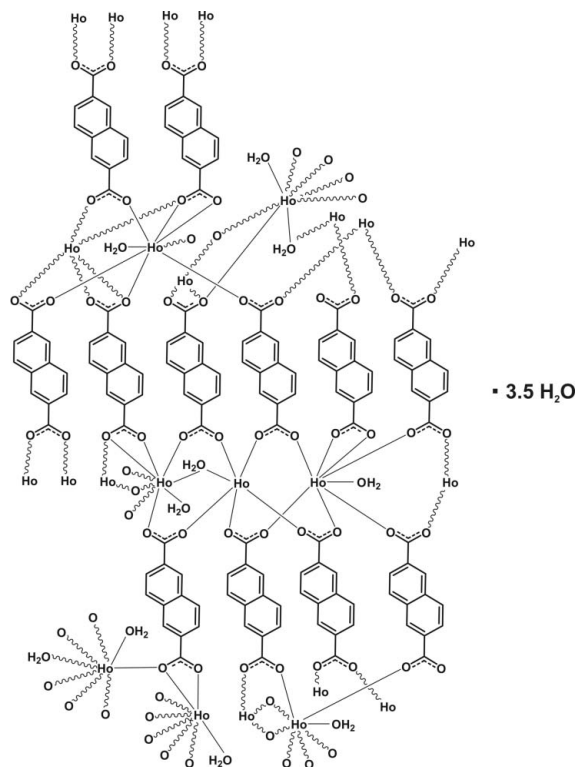
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Key indicators: single-crystal X-ray study; $T = 180$ K; mean $\sigma(\text{C}-\text{C}) = 0.016$ Å; H-atom completeness 73%; disorder in solvent or counterion; R factor = 0.062; wR factor = 0.172; data-to-parameter ratio = 11.2.

In the title compound, $\{[\text{Ho}_4(\text{C}_{12}\text{H}_6\text{O}_4)_6(\text{H}_2\text{O})_5] \cdot 1.75\text{H}_2\text{O}\}_n$, which is isostructural with its Tb^{3+} - and Eu^{3+} -containing analogues, there are four crystallographically independent Ho^{3+} centres, each exhibiting a highly distorted HoO_8 bicapped trigonal-prismatic coordination environment. Adjacent polyhedra are interconnected *via* the carboxylate groups and one μ_2 -bridging water molecule, forming one-dimensional chains propagating along [100]. The naphthalene-2,6-dicarboxylate ligands further interconnect these chains into a three-dimensional framework, which has zigzag channels housing the water molecules. Two naphthalene-2,6-dicarboxylate bridging ligands have their centroids located on crystallographic centres of inversion. One water O atom has a fixed site occupancy factor of 0.75.

Related literature

For isostructural materials, see: Min & Lee (2002); Zheng, Sun *et al.* (2004). For related structures, see: Zheng, Wang *et al.* (2004); Paz & Klinowski (2003); Almeida Paz & Klinowski (2008); Wang *et al.* (2002). For general background, see: Shi *et al.* (2008); Cunha-Silva *et al.* (2007). For bond-length data, see: Allen (2002).



Experimental

Crystal data

$[\text{Ho}_4(\text{C}_{12}\text{H}_6\text{O}_4)_6(\text{H}_2\text{O})_5] \cdot 1.75\text{H}_2\text{O}$
 $M_r = 2066.34$
 Monoclinic, $P2_1/n$
 $a = 17.0505$ (4) Å
 $b = 15.1728$ (4) Å
 $c = 24.9142$ (6) Å
 $\beta = 106.126$ (1)°

$V = 6191.8$ (3) Å³
 $Z = 4$
 $\text{Mo } K\alpha$ radiation
 $\mu = 5.16$ mm⁻¹
 $T = 180$ (2) K
 $0.12 \times 0.12 \times 0.01$ mm

Data collection

Nonius KappaCCD diffractometer
 Absorption correction: multi-scan
 (SORTAV; Blessing, 1995)
 $T_{\text{min}} = 0.576$, $T_{\text{max}} = 0.950$

27842 measured reflections
 10664 independent reflections
 7421 reflections with $I > 2\sigma(I)$
 $R_{\text{int}} = 0.060$

Refinement

$R[F^2 > 2\sigma(F^2)] = 0.062$
 $wR(F^2) = 0.172$
 $S = 1.04$
 10664 reflections
 951 parameters

4 restraints
 H-atom parameters constrained
 $\Delta\rho_{\text{max}} = 9.72$ e Å⁻³
 $\Delta\rho_{\text{min}} = -2.00$ e Å⁻³

Table 1

Selected bond lengths (Å).

Ho1—O1	2.313 (8)	Ho3—O8	2.410 (8)
Ho1—O3 ⁱ	2.335 (7)	Ho3—O12	2.337 (8)
Ho1—O5	2.277 (8)	Ho3—O14	2.356 (8)
Ho1—O10 ⁱⁱ	2.314 (8)	Ho3—O17	2.299 (7)
Ho1—O15 ⁱⁱⁱ	2.309 (8)	Ho3—O20 ⁱⁱ	2.299 (8)
Ho1—O23 ^{iv}	2.847 (9)	Ho3—O21	2.300 (8)
Ho1—O24 ^{iv}	2.311 (8)	Ho3—O4W	2.469 (8)
Ho1—O2W	2.370 (8)	Ho3—O5W	2.640 (6)
Ho2—O2	2.299 (8)	Ho4—O4 ^v	2.360 (8)
Ho2—O6	2.316 (8)	Ho4—O9 ^{vi}	2.382 (7)
Ho2—O7	2.338 (8)	Ho4—O16 ^{vii}	2.342 (8)
Ho2—O11	2.264 (8)	Ho4—O18	2.337 (7)
Ho2—O13	2.347 (8)	Ho4—O22	2.276 (7)
Ho2—O19 ⁱⁱ	2.338 (8)	Ho4—O23 ^{viii}	2.296 (8)
Ho2—O20 ⁱⁱ	2.876 (8)	Ho4—O5W	2.671 (6)
Ho2—O3W	2.423 (9)	Ho4—O6W	2.453 (10)

Symmetry codes: (i) $-x+1, -y+1, -z+2$; (ii) $x-\frac{1}{2}, -y+\frac{1}{2}, z-\frac{1}{2}$; (iii) $-x+1, -y+1, -z+1$; (iv) $x-\frac{1}{2}, -y+\frac{1}{2}, z+\frac{1}{2}$; (v) $-x+2, -y+1, -z+2$; (vi) $x+\frac{1}{2}, -y+\frac{1}{2}, z-\frac{1}{2}$; (vii) $-x+2, -y+1, -z+1$; (viii) $x+\frac{1}{2}, -y+\frac{1}{2}, z+\frac{1}{2}$.

Data collection: *COLLECT* (Nonius 1998); cell refinement: *HKL SCALEPACK* (Otwinowski & Minor 1997); data reduction: *HKL DENZO* (Otwinowski & Minor 1997) and *SCALEPACK*; program(s) used to solve structure: *SIR92* (Altomare *et al.* 1994); program(s) used to refine structure: *SHELXTL* (Bruker, 2001); molecular graphics: *DIAMOND* (Brandenburg, 2006); software used to prepare material for publication: *SHELXTL*.

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Supplementary data and figures for this paper are available from the IUCr electronic archives (Reference: HB2668).

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Poly[[μ_2 -aqua-tetraaqua-hexakis(μ_4 -naphthalene-2,6-dicarboxylato)tetraholmium(III)] 1.75-hydrate]

F. A. Almeida Paz and J. Klinowski

Comment

Multi-dimensional (*i.e.*, one-dimensional, two-dimensional or three-dimensional) networks, known as coordination polymers or metal-organic frameworks (MOFs), in which metallic centres are bridged *via* organic ligands, are of considerable interest. Even though structural diversity can be achieved by selecting different metallic centres (which implies a variation in the number and type of the coordination geometry), fascinating structural architectures are often produced by using uncommon bridging ligands. To reconcile the robustness and crystallinity of the synthesized networks, crystal engineers usually employ *exo*-carboxylate derivatives as the bridging ligands, usually associated with aromatic rings. It is therefore surprising that only a handful of papers reporting lanthanide centres coordinated to residues of naphthalene-2,6-dicarboxylic acid (H₂NDC) have been published (Paz & Klinowski, 2008; Zheng, Sun *et al.*, 2004; Zheng, Wang *et al.*, 2004; Paz & Klinowski, 2003; Wang *et al.*, 2002; Min & Lee, 2002), as confirmed by a search in the Cambridge Structural Database (CSD, Version 5.28 with three updates - August 2007; Allen, 2002).

Following our interest in the hydrothermal synthesis of MOFs, (*e.g.* Shi *et al.*, 2008; Cunha-Silva *et al.*, 2007), we report here the low temperature crystal structure of the title compound, (I), a three-dimensional MOF containing the naphthalene-2,6-dicarboxylate dianion (NDC²⁻) bound to Ho³⁺: [Ho₄(NDC)₆(H₂O)₅] 1.75H₂O. Despite being analogous the frameworks reported by Min & Lee (2002) (with Tb³⁺) and Zheng, Sun *et al.* (2004) (with Eu³⁺), this new crystal structure provides further insights into the self-assembly process. Thus crystals of a two-dimensional network, [Ho₂(NDC)₃(H₂O)₆], could also be isolated from the same synthetic batch (Paz & Klinowski, 2008). We infer that the ionic radius of the lanthanide employed determines whether a three-dimensional (for the lighter series of lanthanides - up to Dy³⁺) or a two-dimensional network (for lanthanides after and including Er³⁺) is obtained. Ho³⁺ always produces a mixture of the two materials, even though it is possible to vary the amount of each framework in the product by adjusting the composition of the synthesis mixture.

Compound (I) contains four crystallographically independent Ho³⁺ centres (Ho1 to Ho4) which are coordinated to a total of six NDC²⁻ ligands (two of these have their centroids located at crystallographic inversion centres) and five water molecules. The coordination sphere of each metallic centre is composed by one unidentate water molecule, with the fifth water (O5W) bridging two neighbouring metallic centres (Ho3 and Ho4 - see Figure 1). Despite the large number of crystallographically independent moieties, the NDC²⁻ moieties coordinate to the Ho³⁺ centres through only two distinct coordination fashions: a *syn,syn*-chelate coupled to a *syn,syn*- μ_2 -bridge (for the C56 and C68 carboxylate groups), and simple *syn,syn*- μ_2 -bridges (for all remaining carboxylate moieties). The {HoO₈} coordination geometries for the Ho³⁺ centres remain strikingly similar, resembling highly distorted bicapped trigonal prisms (Figures 2a to 2d), with the capping positions being either water molecules or the O-atoms involved in the *syn,syn*- μ_2 -bridges coupled to *syn,syn*-chelate mentioned above (O20 and O23 - Figure 2). Disregarding the Ho—O distances related to the O20, O23 and O5W atoms which occupy the capping positions of the coordination polyhedra, the remaining Ho—O distances are typical and well within the ranges registered

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for related materials (as revealed by a search in the CSD - 77 entries, range of 2.20–2.82 Å with a median of 2.34 Å): for Ho1 to Ho4, respectively, 2.277 (8)–2.370 (8) Å, 2.264 (8)–2.423 (9) Å, 2.299 (7)–2.469 (8) Å and 2.276 (7)–2.453 (10) Å (Table 1). We emphasize that even though the Ho—O distances associated with these capping positions are unusually long, they are still within the feasible range found in related materials. Moreover, we also note that the longest values of Ho—O for Ho1 to Ho4 found in the ranges given above are those with the coordinated water molecules. In fact, by restricting the search in the CSD to the geometrical parameters for coordinated water molecules to Ho³⁺ centres, the expected range is from 2.28 to 2.55 Å, which is in good agreement with the experimental data for the title compound.

The connection between neighbouring {HoO₈} polyhedra *via* the carboxylate groups and water molecules leads to the formation of a one-dimensional chain of metallic centres running along the [100] crystallographic direction (Figure 2 e). The Ho···Ho distances range from 4.0258 (1) to 5.2585 (1) Å. These chains are interconnected along the [001] direction *via* the NDC²⁻ bridges forming a three-dimensional MOF (Figure 3). There is structural evidence that such connectivity creates small one-dimensional zigzag channels parallel to the *a*-axis, distributed in a typical brick-wall fashion in the *bc* plane containing the water molecules of crystallization O1W and O7W. Although the water H atoms could not be located in the present study, presumably O—H···O hydrogen bonds from the water molecules (both coordinated and uncoordinated) interconnect adjacent chains (not shown).

Experimental

To a solution of HoCl₃·6H₂O (1.062 g, 2.799 mmol) in distilled water (6.04 g), naphthalene-2,6-dicarboxylic acid (0.102 g, 0.472 mmol) and triethylamine (0.089 g, 0.880 mmol) were added and the mixture was stirred thoroughly for 5 minutes at ambient temperature. The suspension, with a molar composition of 5.93 Ho³⁺: 1.00 H₂NDC: 1.86 TEA: 120 H₂O, was transferred to a Parr teflon-lined stainless steel vessel (*ca* 21 cm³) and placed for 8 h at 418 K in a preheated oven. Before opening, the reaction vessel was allowed to cool slowly to ambient temperature at a rate of 10 K per hour over a period of 14 h. Colourless plates of (I) were manually selected from the product which also contains [Ho₂(NDC)₃(H₂O)₆] (Paz & Klinowski, 2008).

Refinement

The water molecules O1W, O5W and O7W were refined isotropically. Following structural evidence from unrestrained refinement cycles, the O7W water molecule was given a fixed occupancy of 75% in the final structural model.

It is important to stress that a considerable smeared-out electron density was found surrounding the water molecules O1W and O5W. Attempts to model this disorder (during the last stages of the overall structural refinement) over two (or more) partially occupied sites (for each water molecule) did not produce satisfactory models, with large shifts associated with these chemical moieties being observed. In order to achieve full convergence the positions of O1W and O5W were restrained to be equally distant from, respectively, Ho1 and Ho2, and Ho3 and Ho4 (one free variable for each pair of distances). The difficulties while modelling these two water molecules are attributed to the quality of the crystal used for data collection, which was a very small and thin colourless plate diffracting rather weakly at high angles [*e.g.*, almost no reflections were observed for resolutions higher than 0.80 Å]. The highest difference peak is 0.78 Å from O5W.

H atoms associated with all water molecules could not be located from difference Fourier maps, and attempts to place these atoms in calculated positions in order to maximize hydrogen bonding interactions did not lead to chemically reasonable

structural models and they were omitted from the refinement. The H atoms bound to carbon were placed at idealized positions ($C-H = 0.95 \text{ \AA}$) and refined as riding with $U_{iso} = 1.2U_{eq}(C)$.

Figures

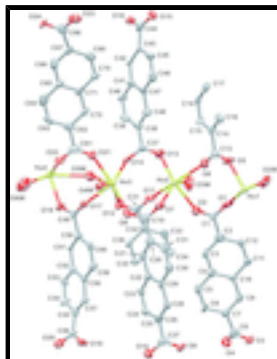


Fig. 1. Simplified representation of the asymmetric unit of (I) with displacement ellipsoids drawn at the 80% probability level. Water molecules O1W and O7W, and hydrogen atoms have been omitted for clarity.

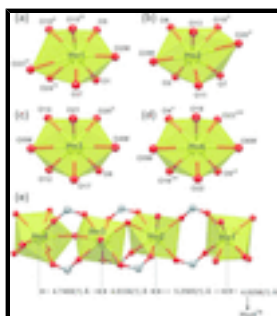


Fig. 2. Polyhedral representation of the $\{HoO_8\}$ coordination environments, which resemble highly distorted bicapped trigonal prisms, for: (a) Ho1, (b) Ho2, (c) Ho3 and (d) Ho4. (e) Interconnection of the individual $\{HoO_8\}$ polyhedra along the $[100]$ crystallographic direction leading to the formation of one-dimensional chains. For selected bond lengths (in \AA) see the dedicated Table in the main paper. Symmetry codes used to generate equivalent atoms: (i) $-x + 1, -y + 1, -z + 2$; (ii) $x - 1/2, -y + 1/2, z - 1/2$; (iii) $-x + 1, -y + 1, -z + 1$; (iv) $x - 1/2, -y + 1/2, z + 1/2$; (v) $-x + 2, -y + 1, -z + 2$; (vi) $x + 1/2, -y + 1/2, z - 1/2$; (vii) $-x + 2, -y + 1, -z + 1$; (viii) $x + 1/2, -y + 1/2, z + 1/2$; (ix) $x - 1, y, z$.

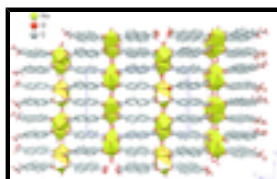


Fig. 3. Crystal packing of the title compound viewed in perspective along the $[100]$ direction of the unit cell. Hydrogen atoms have been omitted for clarity.

poly[[μ_2 -aqua-tetraaqua-hexakis(μ_4 -naphthalene-2,6-dicarboxylato) tetraholmium(III)] 1.75-hydrate]

Crystal data

$[Ho_4(C_{12}H_6O_4)_6(H_2O)_5] \cdot 1.75H_2O$

$M_r = 2066.34$

Monoclinic, $P2_1/n$

Hall symbol: $-P 2yn$

$a = 17.0505 (4) \text{ \AA}$

$b = 15.1728 (4) \text{ \AA}$

$c = 24.9142 (6) \text{ \AA}$

$\beta = 106.126 (1)^\circ$

$V = 6191.8 (3) \text{ \AA}^3$

$Z = 4$

$F(000) = 3982$

$D_x = 2.217 \text{ Mg m}^{-3}$

Mo $K\alpha$ radiation, $\lambda = 0.71073 \text{ \AA}$

Cell parameters from 70959 reflections

$\theta = 1.0\text{--}25.0^\circ$

$\mu = 5.16 \text{ mm}^{-1}$

$T = 180 \text{ K}$

Plate, colourless

$0.12 \times 0.12 \times 0.01 \text{ mm}$

Data collection

Nonius Kappa CCD diffractometer	7421 reflections with $I > 2\sigma(I)$
Radiation source: fine-focus sealed tube	$R_{\text{int}} = 0.060$
Thin slice ω and φ scans	$\theta_{\text{max}} = 25.0^\circ$, $\theta_{\text{min}} = 3.6^\circ$
Absorption correction: multi-scan (SORTAV; Blessing, 1995)	$h = -20 \rightarrow 20$
$T_{\text{min}} = 0.576$, $T_{\text{max}} = 0.950$	$k = -18 \rightarrow 18$
27842 measured reflections	$l = -29 \rightarrow 29$
10664 independent reflections	

Refinement

Refinement on F^2	Primary atom site location: structure-invariant direct methods
Least-squares matrix: full	Secondary atom site location: difference Fourier map
$R[F^2 > 2\sigma(F^2)] = 0.062$	Hydrogen site location: inferred from neighbouring sites
$wR(F^2) = 0.172$	H-atom parameters constrained
$S = 1.04$	$w = 1/[\sigma^2(F_o^2) + (0.0951P)^2 + 49.5291P]$
10664 reflections	where $P = (F_o^2 + 2F_c^2)/3$
951 parameters	$(\Delta/\sigma)_{\text{max}} = 0.001$
4 restraints	$\Delta\rho_{\text{max}} = 9.73 \text{ e } \text{\AA}^{-3}$
	$\Delta\rho_{\text{min}} = -2.01 \text{ e } \text{\AA}^{-3}$

Special details

Experimental. See dedicated section in the main paper

Geometry. All e.s.d.'s (except the e.s.d. in the dihedral angle between two l.s. planes) are estimated using the full covariance matrix. The cell e.s.d.'s are taken into account individually in the estimation of e.s.d.'s in distances, angles and torsion angles; correlations between e.s.d.'s in cell parameters are only used when they are defined by crystal symmetry. An approximate (isotropic) treatment of cell e.s.d.'s is used for estimating e.s.d.'s involving l.s. planes.

Refinement. Refinement of F^2 against ALL reflections. The weighted R -factor wR and goodness of fit S are based on F^2 , conventional R -factors R are based on F , with F set to zero for negative F^2 . The threshold expression of $F^2 > \sigma(F^2)$ is used only for calculating R -factors(gt) *etc.* and is not relevant to the choice of reflections for refinement. R -factors based on F^2 are statistically about twice as large as those based on F , and R -factors based on ALL data will be even larger.

Fractional atomic coordinates and isotropic or equivalent isotropic displacement parameters (\AA^2)

	x	y	z	$U_{\text{iso}}^*/U_{\text{eq}}$	Occ. (<1)
Ho1	0.43036 (3)	0.43328 (3)	0.74931 (2)	0.01641 (16)	
Ho2	0.73944 (3)	0.44550 (4)	0.75148 (2)	0.01864 (16)	
Ho3	0.94596 (3)	0.31502 (3)	0.75250 (2)	0.01782 (16)	
Ho4	1.22082 (3)	0.30713 (4)	0.74454 (2)	0.02020 (16)	
O1W	0.5958 (6)	0.2081 (11)	0.7441 (7)	0.108 (5)*	

O2W	0.4125 (5)	0.5883 (5)	0.7439 (4)	0.028 (2)	
O3W	0.7616 (5)	0.6017 (6)	0.7697 (4)	0.039 (2)	
O4W	0.9089 (5)	0.1577 (5)	0.7388 (3)	0.0251 (18)	
O5W	1.0839 (4)	0.3895 (5)	0.7479 (3)	0.0263 (19)*	
O6W	1.2552 (6)	0.1500 (7)	0.7565 (4)	0.046 (3)	
O7W	0.0661 (11)	0.6098 (13)	0.7506 (7)	0.080 (5)*	0.75
O1	0.5535 (5)	0.4870 (6)	0.8042 (3)	0.0273 (19)	
O2	0.6872 (5)	0.4687 (6)	0.8260 (4)	0.030 (2)	
O3	0.6279 (5)	0.5218 (5)	1.1807 (3)	0.0226 (18)	
O4	0.7584 (5)	0.5611 (5)	1.2030 (3)	0.0265 (19)	
O5	0.4886 (5)	0.4701 (6)	0.6805 (3)	0.0274 (19)	
O6	0.6189 (5)	0.5100 (5)	0.7000 (3)	0.0251 (18)	
O7	0.7555 (5)	0.3124 (5)	0.8006 (3)	0.0257 (19)	
O8	0.8854 (5)	0.2691 (5)	0.8246 (3)	0.0224 (17)	
O9	0.7805 (5)	0.2343 (6)	1.1724 (3)	0.0272 (19)	
O10	0.9123 (5)	0.1926 (5)	1.1948 (3)	0.0271 (19)	
O11	0.8714 (5)	0.4522 (5)	0.8033 (3)	0.0221 (18)	
O12	0.9998 (5)	0.4138 (6)	0.8254 (3)	0.0263 (19)	
O13	0.7983 (5)	0.4924 (6)	0.6817 (3)	0.0286 (19)	
O14	0.9277 (5)	0.4469 (5)	0.7005 (3)	0.0255 (19)	
O15	0.7017 (5)	0.5512 (6)	0.3068 (3)	0.029 (2)	
O16	0.8295 (5)	0.5917 (6)	0.3278 (3)	0.0275 (19)	
O17	1.0540 (4)	0.2363 (5)	0.8072 (3)	0.0193 (17)	
O18	1.1862 (5)	0.2690 (6)	0.8260 (3)	0.0255 (19)	
O19	1.1803 (5)	0.1574 (6)	1.1822 (3)	0.0244 (18)	
O20	1.3118 (5)	0.1900 (5)	1.2001 (3)	0.0279 (19)	
O21	0.9784 (4)	0.2705 (6)	0.6730 (3)	0.0237 (18)	
O22	1.1109 (4)	0.2322 (5)	0.6914 (3)	0.0205 (17)	
O23	0.8546 (5)	0.2013 (5)	0.2970 (3)	0.029 (2)	
O24	0.9861 (5)	0.1763 (6)	0.3129 (3)	0.027 (2)	
C1	0.6227 (6)	0.4803 (8)	0.8388 (5)	0.018 (2)	
C2	0.6267 (7)	0.4936 (7)	0.8990 (5)	0.017 (2)	
C3	0.6942 (6)	0.4698 (7)	0.9413 (5)	0.017 (2)	
H3	0.7390	0.4426	0.9321	0.020*	
C4	0.6985 (6)	0.4851 (8)	0.9988 (5)	0.018 (2)	
C5	0.7657 (7)	0.4613 (7)	1.0441 (5)	0.018 (2)	
H5	0.8115	0.4337	1.0367	0.021*	
C6	0.7666 (7)	0.4770 (8)	1.0988 (5)	0.021 (3)	
H6	0.8137	0.4634	1.1284	0.025*	
C7	0.6966 (7)	0.5137 (7)	1.1106 (5)	0.019 (2)	
C8	0.6950 (7)	0.5322 (8)	1.1689 (5)	0.023 (3)	
C9	0.6303 (7)	0.5372 (7)	1.0675 (5)	0.017 (2)	
H9	0.5837	0.5621	1.0756	0.021*	
C10	0.6301 (6)	0.5250 (7)	1.0112 (4)	0.013 (2)	
C11	0.5608 (7)	0.5494 (7)	0.9665 (5)	0.019 (2)	
H11	0.5152	0.5765	0.9746	0.023*	
C12	0.5597 (6)	0.5342 (7)	0.9130 (5)	0.016 (2)	
H12	0.5131	0.5510	0.8839	0.019*	
C13	0.5503 (6)	0.4905 (7)	0.6661 (5)	0.017 (2)	

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C14	0.5434 (6)	0.5002 (7)	0.6056 (5)	0.016 (2)
C15	0.6068 (6)	0.5430 (7)	0.5885 (4)	0.017 (2)
H15	0.6541	0.5631	0.6158	0.020*
C16	0.5998 (6)	0.5554 (7)	0.5329 (5)	0.017 (2)
H16	0.6416	0.5861	0.5222	0.020*
C17	0.5312 (6)	0.5233 (7)	0.4910 (5)	0.017 (2)
C18	0.4772 (6)	0.4677 (7)	0.5661 (5)	0.015 (2)
H18	0.4357	0.4382	0.5778	0.018*
C19	0.8176 (7)	0.2826 (8)	0.8349 (5)	0.021 (3)
C20	0.8111 (7)	0.2621 (7)	0.8932 (5)	0.019 (2)
C21	0.8757 (6)	0.2863 (7)	0.9394 (5)	0.016 (2)
H21	0.9242	0.3103	0.9337	0.019*
C22	0.8690 (7)	0.2752 (8)	0.9943 (5)	0.021 (3)
C23	0.9341 (7)	0.2967 (8)	1.0427 (5)	0.020 (2)
H23	0.9828	0.3228	1.0387	0.024*
C24	0.9259 (7)	0.2794 (7)	1.0947 (4)	0.018 (2)
H24	0.9704	0.2916	1.1264	0.021*
C25	0.8527 (7)	0.2439 (8)	1.1029 (5)	0.024 (3)
C26	0.8468 (7)	0.2230 (7)	1.1597 (5)	0.021 (3)
C27	0.7881 (6)	0.2240 (7)	1.0570 (5)	0.019 (2)
H27	0.7388	0.2011	1.0621	0.023*
C28	0.7956 (7)	0.2380 (7)	1.0023 (5)	0.018 (2)
C29	0.7310 (7)	0.2155 (8)	0.9532 (5)	0.021 (3)
H29	0.6820	0.1906	0.9577	0.026*
C30	0.7375 (6)	0.2283 (8)	0.9012 (5)	0.023 (3)
H30	0.6929	0.2148	0.8698	0.027*
C31	0.9407 (7)	0.4500 (7)	0.8369 (5)	0.019 (2)
C32	0.9522 (7)	0.4816 (7)	0.8958 (5)	0.018 (2)
C33	0.8895 (7)	0.5255 (7)	0.9109 (5)	0.023 (3)
H33	0.8419	0.5425	0.8827	0.028*
C34	0.8955 (7)	0.5444 (7)	0.9651 (5)	0.018 (2)
H34	0.8516	0.5726	0.9747	0.021*
C35	1.0323 (7)	0.4782 (7)	0.9923 (4)	0.017 (2)
C36	1.0245 (7)	0.4606 (8)	0.9367 (5)	0.021 (3)
H36	1.0684	0.4341	0.9261	0.025*
C37	0.8618 (7)	0.4769 (7)	0.6682 (4)	0.017 (2)
C38	0.8630 (7)	0.4925 (7)	0.6092 (4)	0.019 (2)
C39	0.9333 (7)	0.5273 (7)	0.5949 (5)	0.021 (3)
H39	0.9816	0.5403	0.6236	0.026*
C40	0.9309 (6)	0.5418 (7)	0.5409 (4)	0.016 (2)
H40	0.9775	0.5658	0.5325	0.019*
C41	0.8605 (6)	0.5219 (7)	0.4964 (4)	0.015 (2)
C42	0.8534 (6)	0.5405 (7)	0.4398 (4)	0.013 (2)
H42	0.8984	0.5668	0.4303	0.015*
C43	0.7835 (6)	0.5220 (7)	0.3975 (4)	0.015 (2)
C44	0.7714 (7)	0.5548 (8)	0.3403 (5)	0.021 (3)
C45	0.7181 (7)	0.4779 (7)	0.4120 (4)	0.017 (2)
H45	0.6710	0.4612	0.3832	0.021*
C46	0.7218 (6)	0.4594 (7)	0.4660 (5)	0.018 (2)

H46	0.6772	0.4307	0.4745	0.021*
C47	0.7923 (6)	0.4830 (7)	0.5103 (4)	0.015 (2)
C48	0.7949 (7)	0.4712 (7)	0.5676 (5)	0.018 (2)
H48	0.7485	0.4481	0.5768	0.021*
C49	1.1229 (6)	0.2538 (7)	0.8407 (5)	0.017 (2)
C50	1.1273 (6)	0.2519 (7)	0.9011 (5)	0.017 (2)
C51	1.1979 (6)	0.2820 (7)	0.9410 (4)	0.013 (2)
H51	1.2405	0.3098	0.9294	0.016*
C52	1.2048 (6)	0.2704 (7)	0.9985 (4)	0.014 (2)
C53	1.2777 (6)	0.2962 (7)	1.0406 (5)	0.017 (2)
H53	1.3196	0.3276	1.0303	0.020*
C54	1.2864 (7)	0.2755 (7)	1.0951 (4)	0.018 (2)
H54	1.3353	0.2908	1.1226	0.022*
C55	1.2227 (6)	0.2308 (8)	1.1113 (5)	0.020 (2)
C56	1.2393 (7)	0.1930 (7)	1.1688 (4)	0.017 (2)
C57	1.1509 (6)	0.2115 (7)	1.0724 (4)	0.016 (2)
H57	1.1080	0.1843	1.0838	0.019*
C58	1.1396 (6)	0.2314 (7)	1.0155 (4)	0.015 (2)
C59	1.0675 (6)	0.2059 (7)	0.9728 (5)	0.018 (2)
H59	1.0224	0.1819	0.9832	0.021*
C60	1.0627 (6)	0.2154 (7)	0.9185 (5)	0.017 (2)
H60	1.0147	0.1970	0.8912	0.020*
C61	1.0416 (7)	0.2537 (7)	0.6579 (4)	0.020 (3)
C62	1.0362 (7)	0.2519 (7)	0.5969 (5)	0.021 (3)
C63	1.1013 (6)	0.2124 (7)	0.5797 (5)	0.016 (2)
H63	1.1487	0.1917	0.6067	0.019*
C64	1.0947 (7)	0.2049 (7)	0.5240 (5)	0.023 (3)
H64	1.1384	0.1796	0.5125	0.027*
C65	1.0234 (6)	0.2345 (7)	0.4826 (4)	0.017 (2)
C66	1.0120 (7)	0.2199 (7)	0.4252 (5)	0.021 (3)
H66	1.0551	0.1945	0.4131	0.025*
C67	0.9396 (7)	0.2414 (7)	0.3858 (5)	0.018 (2)
C68	0.9257 (7)	0.2056 (8)	0.3278 (5)	0.023 (3)
C69	0.8791 (7)	0.2877 (7)	0.4044 (5)	0.021 (2)
H69	0.8306	0.3070	0.3778	0.025*
C70	0.8896 (7)	0.3041 (7)	0.4580 (5)	0.019 (2)
H70	0.8482	0.3351	0.4690	0.023*
C71	0.9601 (6)	0.2772 (7)	0.4996 (4)	0.015 (2)
C72	0.9682 (6)	0.2855 (7)	0.5576 (5)	0.016 (2)
H72	0.9266	0.3143	0.5698	0.019*

Atomic displacement parameters (\AA^2)

	U^{11}	U^{22}	U^{33}	U^{12}	U^{13}	U^{23}
Ho1	0.0159 (3)	0.0228 (3)	0.0110 (3)	-0.0031 (2)	0.00474 (19)	-0.0011 (2)
Ho2	0.0156 (3)	0.0290 (3)	0.0126 (3)	0.0039 (2)	0.0060 (2)	-0.0003 (2)
Ho3	0.0158 (3)	0.0273 (3)	0.0115 (3)	0.0032 (2)	0.0057 (2)	-0.0002 (2)
Ho4	0.0173 (3)	0.0331 (3)	0.0116 (3)	-0.0074 (2)	0.0063 (2)	-0.0006 (2)

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O2W	0.029 (5)	0.023 (4)	0.043 (5)	-0.006 (4)	0.026 (4)	-0.002 (4)
O3W	0.026 (5)	0.026 (5)	0.065 (7)	0.000 (4)	0.012 (4)	-0.011 (5)
O4W	0.026 (5)	0.026 (4)	0.026 (5)	-0.004 (4)	0.012 (4)	-0.003 (4)
O6W	0.050 (6)	0.044 (6)	0.052 (7)	-0.012 (5)	0.030 (5)	-0.006 (5)
O1	0.027 (5)	0.044 (5)	0.015 (4)	-0.011 (4)	0.013 (4)	-0.003 (4)
O2	0.024 (5)	0.045 (5)	0.024 (5)	0.008 (4)	0.009 (4)	-0.001 (4)
O3	0.022 (4)	0.031 (5)	0.018 (4)	0.000 (3)	0.011 (3)	-0.004 (3)
O4	0.021 (4)	0.031 (5)	0.026 (5)	-0.003 (4)	0.003 (4)	-0.006 (4)
O5	0.023 (4)	0.046 (5)	0.016 (4)	-0.008 (4)	0.011 (3)	0.003 (4)
O6	0.021 (4)	0.030 (5)	0.027 (5)	0.007 (4)	0.009 (4)	0.004 (4)
O7	0.027 (5)	0.035 (5)	0.018 (4)	0.002 (4)	0.010 (4)	0.005 (4)
O8	0.021 (4)	0.027 (4)	0.019 (4)	0.001 (3)	0.005 (3)	-0.002 (3)
O9	0.028 (5)	0.041 (5)	0.020 (4)	0.009 (4)	0.018 (4)	0.004 (4)
O10	0.020 (4)	0.037 (5)	0.024 (5)	0.006 (4)	0.005 (4)	0.006 (4)
O11	0.022 (4)	0.031 (5)	0.012 (4)	-0.009 (4)	0.002 (3)	-0.003 (3)
O12	0.019 (4)	0.040 (5)	0.022 (4)	-0.002 (4)	0.009 (3)	-0.010 (4)
O13	0.027 (5)	0.039 (5)	0.024 (5)	0.006 (4)	0.014 (4)	0.007 (4)
O14	0.027 (5)	0.039 (5)	0.016 (4)	0.006 (4)	0.016 (4)	0.002 (4)
O15	0.013 (4)	0.053 (6)	0.018 (4)	0.002 (4)	0.000 (3)	-0.002 (4)
O16	0.022 (4)	0.047 (5)	0.015 (4)	0.001 (4)	0.007 (3)	0.004 (4)
O17	0.024 (4)	0.020 (4)	0.017 (4)	0.002 (3)	0.012 (3)	0.004 (3)
O18	0.018 (4)	0.044 (5)	0.018 (4)	0.002 (4)	0.011 (3)	0.006 (4)
O19	0.021 (4)	0.036 (5)	0.018 (4)	-0.002 (4)	0.007 (3)	0.011 (4)
O20	0.030 (5)	0.034 (5)	0.022 (5)	0.003 (4)	0.011 (4)	-0.002 (4)
O21	0.015 (4)	0.038 (5)	0.019 (4)	-0.002 (4)	0.005 (3)	-0.005 (4)
O22	0.013 (4)	0.033 (5)	0.014 (4)	-0.007 (3)	0.001 (3)	-0.005 (3)
O23	0.031 (5)	0.036 (5)	0.019 (4)	-0.010 (4)	0.006 (4)	0.001 (4)
O24	0.021 (4)	0.046 (5)	0.018 (4)	0.009 (4)	0.010 (3)	-0.009 (4)
C1	0.010 (5)	0.033 (7)	0.014 (6)	-0.003 (5)	0.005 (4)	-0.007 (5)
C2	0.019 (6)	0.015 (5)	0.019 (6)	-0.007 (5)	0.008 (5)	0.003 (5)
C3	0.016 (6)	0.016 (6)	0.024 (6)	0.000 (4)	0.013 (5)	-0.006 (5)
C4	0.010 (5)	0.027 (6)	0.019 (6)	-0.004 (5)	0.007 (4)	0.002 (5)
C5	0.015 (6)	0.017 (6)	0.024 (7)	0.001 (4)	0.009 (5)	0.000 (5)
C6	0.014 (6)	0.033 (7)	0.021 (6)	0.004 (5)	0.014 (5)	0.004 (5)
C7	0.021 (6)	0.022 (6)	0.019 (6)	-0.001 (5)	0.012 (5)	-0.002 (5)
C8	0.024 (7)	0.023 (6)	0.027 (7)	-0.003 (5)	0.014 (5)	0.002 (5)
C9	0.018 (6)	0.016 (6)	0.021 (6)	0.001 (4)	0.011 (5)	0.000 (5)
C10	0.018 (6)	0.013 (5)	0.010 (5)	-0.002 (4)	0.008 (4)	0.000 (4)
C11	0.021 (6)	0.023 (6)	0.017 (6)	0.005 (5)	0.011 (5)	0.003 (5)
C12	0.015 (6)	0.019 (6)	0.015 (6)	0.005 (4)	0.006 (4)	0.002 (4)
C13	0.010 (5)	0.021 (6)	0.017 (6)	-0.003 (5)	-0.001 (4)	-0.005 (5)
C14	0.012 (5)	0.020 (6)	0.018 (6)	0.000 (4)	0.006 (4)	-0.003 (5)
C15	0.013 (5)	0.023 (6)	0.013 (6)	-0.002 (5)	0.003 (4)	-0.012 (5)
C16	0.009 (5)	0.022 (6)	0.022 (6)	0.002 (4)	0.008 (4)	0.001 (5)
C17	0.014 (5)	0.017 (6)	0.022 (6)	-0.001 (4)	0.008 (5)	-0.006 (5)
C18	0.013 (5)	0.009 (5)	0.021 (6)	-0.002 (4)	0.002 (4)	-0.002 (4)
C19	0.016 (6)	0.026 (6)	0.019 (6)	-0.002 (5)	0.002 (5)	-0.001 (5)
C20	0.023 (6)	0.011 (5)	0.022 (6)	0.000 (5)	0.006 (5)	-0.004 (5)
C21	0.014 (5)	0.013 (5)	0.021 (6)	0.001 (4)	0.006 (4)	0.001 (4)

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C22	0.019 (6)	0.032 (7)	0.015 (6)	0.008 (5)	0.009 (5)	0.003 (5)
C23	0.017 (6)	0.029 (6)	0.013 (6)	0.010 (5)	0.004 (4)	0.006 (5)
C24	0.025 (6)	0.016 (6)	0.012 (6)	0.003 (5)	0.004 (4)	0.002 (4)
C25	0.028 (7)	0.023 (6)	0.028 (7)	0.007 (5)	0.019 (5)	0.011 (5)
C26	0.036 (7)	0.014 (6)	0.015 (6)	0.003 (5)	0.010 (5)	-0.004 (5)
C27	0.014 (6)	0.025 (6)	0.019 (6)	0.008 (5)	0.005 (4)	0.002 (5)
C28	0.021 (6)	0.011 (5)	0.027 (7)	0.008 (4)	0.015 (5)	0.001 (5)
C29	0.017 (6)	0.024 (6)	0.025 (7)	0.003 (5)	0.007 (5)	0.002 (5)
C30	0.010 (5)	0.034 (7)	0.022 (6)	-0.001 (5)	0.001 (4)	-0.005 (5)
C31	0.022 (6)	0.020 (6)	0.023 (6)	0.000 (5)	0.019 (5)	0.001 (5)
C32	0.017 (6)	0.019 (6)	0.019 (6)	-0.001 (5)	0.006 (5)	0.006 (5)
C33	0.018 (6)	0.022 (6)	0.031 (7)	0.005 (5)	0.011 (5)	0.009 (5)
C34	0.018 (6)	0.020 (6)	0.016 (6)	-0.002 (5)	0.006 (5)	-0.004 (5)
C35	0.021 (6)	0.018 (6)	0.013 (6)	-0.006 (5)	0.006 (5)	0.002 (4)
C36	0.019 (6)	0.028 (6)	0.021 (6)	-0.003 (5)	0.014 (5)	-0.008 (5)
C37	0.021 (6)	0.017 (6)	0.014 (6)	0.002 (5)	0.007 (5)	0.001 (4)
C38	0.024 (6)	0.020 (6)	0.011 (6)	0.001 (5)	0.001 (5)	0.004 (4)
C39	0.019 (6)	0.025 (6)	0.016 (6)	-0.003 (5)	-0.001 (5)	0.000 (5)
C40	0.015 (6)	0.023 (6)	0.012 (6)	-0.003 (5)	0.006 (4)	0.004 (4)
C41	0.017 (6)	0.014 (5)	0.015 (6)	0.000 (4)	0.005 (4)	0.005 (4)
C42	0.011 (5)	0.019 (5)	0.011 (6)	-0.002 (4)	0.008 (4)	-0.002 (4)
C43	0.012 (5)	0.022 (6)	0.010 (5)	0.011 (4)	0.004 (4)	-0.001 (4)
C44	0.019 (6)	0.026 (6)	0.024 (7)	-0.001 (5)	0.014 (5)	-0.002 (5)
C45	0.020 (6)	0.020 (6)	0.012 (6)	0.001 (5)	0.005 (4)	0.002 (4)
C46	0.011 (5)	0.016 (6)	0.027 (7)	-0.005 (4)	0.007 (5)	0.001 (5)
C47	0.014 (5)	0.018 (6)	0.015 (6)	0.003 (4)	0.006 (4)	0.008 (4)
C48	0.017 (6)	0.018 (6)	0.017 (6)	-0.005 (5)	0.004 (5)	-0.005 (5)
C49	0.014 (6)	0.021 (6)	0.019 (6)	0.002 (5)	0.009 (5)	-0.002 (5)
C50	0.014 (5)	0.024 (6)	0.016 (6)	0.000 (5)	0.008 (4)	0.002 (5)
C51	0.013 (5)	0.020 (6)	0.010 (5)	0.006 (4)	0.007 (4)	0.005 (4)
C52	0.012 (5)	0.013 (5)	0.017 (6)	-0.001 (4)	0.003 (4)	-0.002 (4)
C53	0.013 (5)	0.016 (6)	0.022 (6)	0.000 (4)	0.005 (4)	0.003 (5)
C54	0.021 (6)	0.024 (6)	0.012 (6)	-0.007 (5)	0.010 (4)	-0.006 (5)
C55	0.015 (6)	0.026 (6)	0.020 (6)	0.002 (5)	0.006 (5)	-0.001 (5)
C56	0.017 (6)	0.026 (6)	0.011 (5)	-0.003 (5)	0.010 (4)	-0.002 (4)
C57	0.013 (5)	0.023 (6)	0.013 (6)	0.000 (5)	0.005 (4)	0.000 (5)
C58	0.015 (5)	0.016 (5)	0.017 (6)	0.005 (4)	0.006 (4)	0.002 (4)
C59	0.015 (6)	0.021 (6)	0.018 (6)	-0.002 (5)	0.006 (4)	0.001 (5)
C60	0.008 (5)	0.023 (6)	0.019 (6)	0.000 (4)	0.003 (4)	0.005 (5)
C61	0.019 (6)	0.025 (6)	0.014 (6)	-0.015 (5)	0.001 (5)	-0.006 (5)
C62	0.023 (6)	0.021 (6)	0.020 (6)	-0.005 (5)	0.008 (5)	-0.001 (5)
C63	0.012 (5)	0.014 (5)	0.019 (6)	-0.002 (4)	0.001 (4)	0.009 (4)
C64	0.020 (6)	0.018 (6)	0.035 (7)	-0.004 (5)	0.015 (5)	0.003 (5)
C65	0.017 (6)	0.024 (6)	0.014 (6)	-0.007 (5)	0.010 (4)	-0.007 (5)
C66	0.022 (6)	0.013 (6)	0.030 (7)	0.000 (5)	0.009 (5)	-0.002 (5)
C67	0.019 (6)	0.018 (6)	0.020 (6)	0.001 (5)	0.012 (5)	-0.001 (5)
C68	0.016 (6)	0.028 (6)	0.023 (6)	-0.003 (5)	0.003 (5)	0.001 (5)
C69	0.018 (6)	0.022 (6)	0.021 (6)	-0.002 (5)	0.004 (5)	0.004 (5)
C70	0.019 (6)	0.023 (6)	0.017 (6)	0.003 (5)	0.008 (5)	-0.003 (5)

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C71	0.013 (5)	0.018 (6)	0.019 (6)	-0.009 (4)	0.013 (4)	-0.005 (5)
C72	0.014 (5)	0.019 (6)	0.019 (6)	-0.005 (4)	0.011 (4)	-0.006 (5)

Geometric parameters (Å, °)

Ho1—O1	2.313 (8)	C17—C18 ⁱⁱⁱ	1.395 (16)
Ho1—O3 ⁱ	2.335 (7)	C17—C17 ⁱⁱⁱ	1.45 (2)
Ho1—O5	2.277 (8)	C18—C17 ⁱⁱⁱ	1.395 (16)
Ho1—O10 ⁱⁱ	2.314 (8)	C18—H18	0.9500
Ho1—O15 ⁱⁱⁱ	2.309 (8)	C19—C20	1.519 (16)
Ho1—O23 ^{iv}	2.847 (9)	C20—C21	1.402 (15)
Ho1—O24 ^{iv}	2.311 (8)	C20—C30	1.420 (16)
Ho1—O2W	2.370 (8)	C21—C22	1.415 (15)
Ho2—O2	2.299 (8)	C21—H21	0.9500
Ho2—O6	2.316 (8)	C22—C23	1.431 (16)
Ho2—O7	2.338 (8)	C22—C28	1.437 (16)
Ho2—O11	2.264 (8)	C23—C24	1.365 (15)
Ho2—O13	2.347 (8)	C23—H23	0.9500
Ho2—O19 ⁱⁱ	2.338 (8)	C24—C25	1.425 (16)
Ho2—O20 ⁱⁱ	2.876 (8)	C24—H24	0.9500
Ho2—O3W	2.423 (9)	C25—C27	1.383 (16)
Ho3—O8	2.410 (8)	C25—C26	1.482 (16)
Ho3—O12	2.337 (8)	C27—C28	1.420 (16)
Ho3—O14	2.356 (8)	C27—H27	0.9500
Ho3—O17	2.299 (7)	C28—C29	1.441 (16)
Ho3—O20 ⁱⁱ	2.299 (8)	C29—C30	1.346 (16)
Ho3—O21	2.300 (8)	C29—H29	0.9500
Ho3—O4W	2.469 (8)	C30—H30	0.9500
Ho3—O5W	2.640 (6)	C31—C32	1.503 (16)
Ho4—O4 ^v	2.360 (8)	C32—C33	1.397 (16)
Ho4—O9 ^{vi}	2.382 (7)	C32—C36	1.400 (16)
Ho4—O16 ^{vii}	2.342 (8)	C33—C34	1.357 (16)
Ho4—O18	2.337 (7)	C33—H33	0.9500
Ho4—O22	2.276 (7)	C34—C35 ^v	1.426 (15)
Ho4—O23 ^{viii}	2.296 (8)	C34—H34	0.9500
Ho4—O5W	2.671 (6)	C35—C36	1.380 (16)
Ho4—O6W	2.453 (10)	C35—C35 ^v	1.43 (2)
O1—C1	1.258 (13)	C35—C34 ^v	1.426 (15)
O2—C1	1.239 (13)	C36—H36	0.9500
O3—C8	1.269 (14)	C37—C38	1.494 (15)
O3—Ho1 ⁱ	2.335 (7)	C38—C48	1.364 (15)
O4—C8	1.254 (14)	C38—C39	1.443 (16)
O4—Ho4 ^v	2.360 (8)	C39—C40	1.352 (16)
O5—C13	1.240 (13)	C39—H39	0.9500
O6—C13	1.272 (13)	C40—C41	1.421 (15)

O7—C19	1.246 (14)	C40—H40	0.9500
O8—C19	1.269 (14)	C41—C42	1.410 (15)
O9—C26	1.268 (14)	C41—C47	1.429 (15)
O9—Ho4 ^{iv}	2.382 (7)	C42—C43	1.383 (15)
O10—C26	1.297 (14)	C42—H42	0.9500
O10—Ho1 ^{viii}	2.314 (8)	C43—C45	1.429 (15)
O11—C31	1.245 (14)	C43—C44	1.470 (16)
O12—C31	1.249 (13)	C44—Ho4 ^{vii}	3.005 (12)
O13—C37	1.241 (13)	C45—C46	1.359 (16)
O14—C37	1.271 (13)	C45—H45	0.9500
O15—C44	1.249 (14)	C46—C47	1.433 (15)
O15—Ho1 ⁱⁱⁱ	2.309 (7)	C46—H46	0.9500
O16—C44	1.250 (13)	C47—C48	1.428 (15)
O16—Ho4 ^{vii}	2.342 (8)	C48—H48	0.9500
O17—C49	1.266 (13)	C49—C50	1.486 (15)
O18—C49	1.255 (13)	C50—C60	1.404 (14)
O19—C56	1.266 (13)	C50—C51	1.408 (15)
O19—Ho2 ^{viii}	2.338 (8)	C51—C52	1.417 (14)
O20—C56	1.266 (14)	C51—H51	0.9500
O20—Ho3 ^{viii}	2.299 (8)	C52—C58	1.424 (15)
O20—Ho2 ^{viii}	2.876 (8)	C52—C53	1.439 (15)
O21—C61	1.263 (13)	C53—C54	1.361 (15)
O22—C61	1.285 (13)	C53—H53	0.9500
O23—C68	1.243 (14)	C54—C55	1.431 (15)
O23—Ho4 ⁱⁱ	2.296 (8)	C54—H54	0.9500
O23—Ho1 ^{vi}	2.847 (9)	C55—C57	1.366 (15)
O24—C68	1.267 (14)	C55—C56	1.495 (15)
O24—Ho1 ^{vi}	2.311 (8)	C56—Ho2 ^{viii}	2.942 (11)
C1—C2	1.496 (15)	C57—C58	1.409 (15)
C2—C3	1.375 (16)	C57—H57	0.9500
C2—C12	1.422 (15)	C58—C59	1.438 (15)
C3—C4	1.433 (16)	C59—C60	1.340 (15)
C3—H3	0.9500	C59—H59	0.9500
C4—C5	1.414 (16)	C60—H60	0.9500
C4—C10	1.422 (15)	C61—C62	1.496 (16)
C5—C6	1.379 (16)	C62—C72	1.390 (16)
C5—H5	0.9500	C62—C63	1.428 (15)
C6—C7	1.419 (15)	C63—C64	1.366 (16)
C6—H6	0.9500	C63—H63	0.9500
C7—C9	1.373 (16)	C64—C65	1.431 (16)
C7—C8	1.485 (16)	C64—H64	0.9500
C9—C10	1.415 (15)	C65—C66	1.406 (16)
C9—H9	0.9500	C65—C71	1.421 (14)
C10—C11	1.428 (15)	C66—C67	1.386 (16)
C11—C12	1.349 (15)	C66—H66	0.9500
C11—H11	0.9500	C67—C69	1.428 (15)
C12—H12	0.9500	C67—C68	1.500 (16)

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C13—C14	1.487 (15)	C68—Ho1 ^{vi}	2.890 (12)
C14—C18	1.367 (15)	C69—C70	1.322 (16)
C14—C15	1.423 (15)	C69—H69	0.9500
C15—C16	1.371 (16)	C70—C71	1.412 (15)
C15—H15	0.9500	C70—H70	0.9500
C16—C17	1.421 (15)	C71—C72	1.419 (15)
C16—H16	0.9500	C72—H72	0.9500
O5—Ho1—O15 ⁱⁱⁱ	94.3 (3)	C9—C10—C11	120.8 (10)
O5—Ho1—O24 ^{iv}	120.7 (3)	C4—C10—C11	119.5 (10)
O15 ⁱⁱⁱ —Ho1—O24 ^{iv}	129.4 (3)	C12—C11—C10	120.5 (10)
O5—Ho1—O1	81.0 (3)	C12—C11—H11	119.8
O15 ⁱⁱⁱ —Ho1—O1	153.5 (3)	C10—C11—H11	119.8
O24 ^{iv} —Ho1—O1	73.5 (3)	C11—C12—C2	121.6 (10)
O5—Ho1—O10 ⁱⁱ	77.0 (3)	C11—C12—H12	119.2
O15 ⁱⁱⁱ —Ho1—O10 ⁱⁱ	77.3 (3)	C2—C12—H12	119.2
O24 ^{iv} —Ho1—O10 ⁱⁱ	76.7 (3)	O5—C13—O6	124.3 (10)
O1—Ho1—O10 ⁱⁱ	126.1 (3)	O5—C13—C14	118.9 (9)
O5—Ho1—O3 ⁱ	148.8 (3)	O6—C13—C14	116.6 (9)
O15 ⁱⁱⁱ —Ho1—O3 ⁱ	82.9 (3)	C18—C14—C15	119.5 (10)
O24 ^{iv} —Ho1—O3 ⁱ	83.0 (3)	C18—C14—C13	120.7 (10)
O1—Ho1—O3 ⁱ	87.7 (3)	C15—C14—C13	119.7 (9)
O10 ⁱⁱ —Ho1—O3 ⁱ	131.6 (3)	C16—C15—C14	120.3 (10)
O5—Ho1—O2W	77.9 (3)	C16—C15—H15	119.8
O15 ⁱⁱⁱ —Ho1—O2W	77.1 (3)	C14—C15—H15	119.8
O24 ^{iv} —Ho1—O2W	140.8 (3)	C15—C16—C17	121.2 (10)
O1—Ho1—O2W	76.4 (3)	C15—C16—H16	119.4
O10 ⁱⁱ —Ho1—O2W	142.3 (3)	C17—C16—H16	119.4
O3 ⁱ —Ho1—O2W	71.2 (3)	C18 ⁱⁱⁱ —C17—C16	123.1 (10)
O5—Ho1—O23 ^{iv}	146.5 (3)	C18 ⁱⁱⁱ —C17—C17 ⁱⁱⁱ	119.2 (12)
O15 ⁱⁱⁱ —Ho1—O23 ^{iv}	81.2 (3)	C16—C17—C17 ⁱⁱⁱ	117.7 (13)
O24 ^{iv} —Ho1—O23 ^{iv}	49.1 (2)	C14—C18—C17 ⁱⁱⁱ	121.9 (10)
O1—Ho1—O23 ^{iv}	116.7 (3)	C14—C18—H18	119.0
O10 ⁱⁱ —Ho1—O23 ^{iv}	69.6 (3)	C17 ⁱⁱⁱ —C18—H18	119.0
O3 ⁱ —Ho1—O23 ^{iv}	64.0 (3)	O7—C19—O8	124.6 (11)
O2W—Ho1—O23 ^{iv}	132.0 (2)	O7—C19—C20	117.2 (10)
O5—Ho1—C68 ^{iv}	142.0 (3)	O8—C19—C20	118.2 (10)
O15 ⁱⁱⁱ —Ho1—C68 ^{iv}	106.2 (3)	C21—C20—C30	120.2 (10)
O24 ^{iv} —Ho1—C68 ^{iv}	25.2 (3)	C21—C20—C19	118.8 (10)
O1—Ho1—C68 ^{iv}	92.7 (3)	C30—C20—C19	120.6 (10)
O10 ⁱⁱ —Ho1—C68 ^{iv}	76.8 (3)	C20—C21—C22	120.5 (10)
O3 ⁱ —Ho1—C68 ^{iv}	67.1 (3)	C20—C21—H21	119.8
O2W—Ho1—C68 ^{iv}	137.3 (3)	C22—C21—H21	119.8

O23 ^{iv} —Ho1—C68 ^{iv}	25.0 (3)	C21—C22—C23	122.5 (10)
O11—Ho2—O2	94.6 (3)	C21—C22—C28	119.2 (10)
O11—Ho2—O6	152.3 (3)	C23—C22—C28	118.3 (10)
O2—Ho2—O6	84.2 (3)	C24—C23—C22	119.7 (11)
O11—Ho2—O19 ⁱⁱ	127.4 (3)	C24—C23—H23	120.1
O2—Ho2—O19 ⁱⁱ	120.9 (3)	C22—C23—H23	120.1
O6—Ho2—O19 ⁱⁱ	74.8 (3)	C23—C24—C25	122.2 (11)
O11—Ho2—O7	77.7 (3)	C23—C24—H24	118.9
O2—Ho2—O7	74.0 (3)	C25—C24—H24	118.9
O6—Ho2—O7	127.7 (3)	C27—C25—C24	119.5 (10)
O19 ⁱⁱ —Ho2—O7	76.9 (3)	C27—C25—C26	119.5 (10)
O11—Ho2—O13	81.2 (3)	C24—C25—C26	120.9 (11)
O2—Ho2—O13	153.4 (3)	O9—C26—O10	122.8 (10)
O6—Ho2—O13	87.6 (3)	O9—C26—C25	120.8 (10)
O19 ⁱⁱ —Ho2—O13	80.9 (3)	O10—C26—C25	116.4 (10)
O7—Ho2—O13	129.6 (3)	C25—C27—C28	119.8 (10)
O11—Ho2—O3W	76.8 (3)	C25—C27—H27	120.1
O2—Ho2—O3W	76.9 (3)	C28—C27—H27	120.1
O6—Ho2—O3W	76.0 (3)	C27—C28—C22	120.4 (10)
O19 ⁱⁱ —Ho2—O3W	143.6 (3)	C27—C28—C29	121.8 (10)
O7—Ho2—O3W	139.2 (3)	C22—C28—C29	117.8 (10)
O13—Ho2—O3W	76.5 (3)	C30—C29—C28	122.3 (11)
O11—Ho2—O20 ⁱⁱ	78.7 (3)	C30—C29—H29	118.9
O2—Ho2—O20 ⁱⁱ	141.2 (3)	C28—C29—H29	118.9
O6—Ho2—O20 ⁱⁱ	118.7 (3)	C29—C30—C20	120.0 (10)
O19 ⁱⁱ —Ho2—O20 ⁱⁱ	49.0 (2)	C29—C30—H30	120.0
O7—Ho2—O20 ⁱⁱ	67.2 (3)	C20—C30—H30	120.0
O13—Ho2—O20 ⁱⁱ	64.0 (3)	O11—C31—O12	122.1 (11)
O3W—Ho2—O20 ⁱⁱ	136.0 (3)	O11—C31—C32	119.4 (9)
O11—Ho2—C56 ⁱⁱ	103.8 (3)	O12—C31—C32	118.0 (10)
O2—Ho2—C56 ⁱⁱ	138.5 (3)	O11—C31—Ho3	75.5 (6)
O6—Ho2—C56 ⁱⁱ	94.9 (3)	O12—C31—Ho3	49.1 (6)
O19 ⁱⁱ —Ho2—C56 ⁱⁱ	24.5 (3)	C32—C31—Ho3	153.4 (7)
O7—Ho2—C56 ⁱⁱ	74.1 (3)	C33—C32—C36	119.8 (11)
O13—Ho2—C56 ⁱⁱ	67.4 (3)	C33—C32—C31	120.8 (10)
O3W—Ho2—C56 ⁱⁱ	143.1 (3)	C36—C32—C31	119.2 (10)
O20 ⁱⁱ —Ho2—C56 ⁱⁱ	25.1 (3)	C34—C33—C32	121.1 (11)
O20 ⁱⁱ —Ho3—O17	145.8 (3)	C34—C33—H33	119.5
O20 ⁱⁱ —Ho3—O21	86.6 (3)	C32—C33—H33	119.5
O17—Ho3—O21	90.7 (3)	C33—C34—C35 ^v	120.1 (10)
O20 ⁱⁱ —Ho3—O12	125.3 (3)	C33—C34—H34	119.9
O17—Ho3—O12	78.2 (3)	C35 ^v —C34—H34	119.9
O21—Ho3—O12	136.1 (3)	C36—C35—C35 ^v	119.6 (13)

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O20 ⁱⁱ —Ho3—O14	75.9 (3)	C36—C35—C34 ^v	121.6 (10)
O17—Ho3—O14	136.8 (3)	C35 ^v —C35—C34 ^v	118.8 (12)
O21—Ho3—O14	78.6 (3)	C35—C36—C32	120.4 (10)
O12—Ho3—O14	81.1 (3)	C35—C36—H36	119.8
O20 ⁱⁱ —Ho3—O8	81.3 (3)	C32—C36—H36	119.8
O17—Ho3—O8	81.2 (3)	O13—C37—O14	125.4 (10)
O21—Ho3—O8	144.2 (3)	O13—C37—C38	119.2 (10)
O12—Ho3—O8	76.3 (3)	O14—C37—C38	115.4 (9)
O14—Ho3—O8	129.5 (3)	C48—C38—C39	119.3 (10)
O20 ⁱⁱ —Ho3—O4W	73.3 (3)	C48—C38—C37	118.0 (10)
O17—Ho3—O4W	73.2 (3)	C39—C38—C37	122.7 (10)
O21—Ho3—O4W	73.2 (3)	C40—C39—C38	120.5 (10)
O12—Ho3—O4W	139.3 (3)	C40—C39—H39	119.7
O14—Ho3—O4W	139.1 (3)	C38—C39—H39	119.7
O8—Ho3—O4W	71.1 (3)	C39—C40—C41	121.8 (10)
O20 ⁱⁱ —Ho3—O5W	138.4 (3)	C39—C40—H40	119.1
O17—Ho3—O5W	70.8 (2)	C41—C40—H40	119.1
O21—Ho3—O5W	70.1 (3)	C42—C41—C40	123.8 (10)
O12—Ho3—O5W	66.2 (3)	C42—C41—C47	118.4 (10)
O14—Ho3—O5W	66.2 (3)	C40—C41—C47	117.8 (10)
O8—Ho3—O5W	136.6 (2)	C43—C42—C41	122.5 (9)
O4W—Ho3—O5W	127.3 (2)	C43—C42—H42	118.8
O20 ⁱⁱ —Ho3—C31	101.9 (3)	C41—C42—H42	118.8
O17—Ho3—C31	96.7 (3)	C42—C43—C45	118.1 (10)
O21—Ho3—C31	151.5 (3)	C42—C43—C44	121.9 (10)
O12—Ho3—C31	23.8 (3)	C45—C43—C44	119.6 (9)
O14—Ho3—C31	77.1 (3)	O15—C44—O16	121.7 (11)
O8—Ho3—C31	64.3 (3)	O15—C44—C43	119.0 (10)
O4W—Ho3—C31	135.3 (3)	O16—C44—C43	119.0 (10)
O5W—Ho3—C31	86.4 (3)	O15—C44—Ho4 ^{vii}	77.3 (7)
O22—Ho4—O23 ^{viii}	146.1 (3)	O16—C44—Ho4 ^{vii}	47.1 (6)
O22—Ho4—O18	90.9 (3)	C43—C44—Ho4 ^{vii}	153.7 (8)
O23 ^{viii} —Ho4—O18	86.7 (3)	C46—C45—C43	121.4 (10)
O22—Ho4—O16 ^{vii}	79.3 (3)	C46—C45—H45	119.3
O23 ^{viii} —Ho4—O16 ^{vii}	124.6 (3)	C43—C45—H45	119.3
O18—Ho4—O16 ^{vii}	134.8 (3)	C45—C46—C47	120.6 (10)
O22—Ho4—O4 ^v	135.9 (3)	C45—C46—H46	119.7
O23 ^{viii} —Ho4—O4 ^v	76.1 (3)	C47—C46—H46	119.7
O18—Ho4—O4 ^v	76.4 (3)	C48—C47—C41	119.5 (10)
O16 ^{vii} —Ho4—O4 ^v	80.7 (3)	C48—C47—C46	121.6 (10)
O22—Ho4—O9 ^{vi}	83.4 (3)	C41—C47—C46	118.8 (10)
O23 ^{viii} —Ho4—O9 ^{vi}	81.4 (3)	C38—C48—C47	120.9 (10)
O18—Ho4—O9 ^{vi}	148.5 (3)	C38—C48—H48	119.5
O16 ^{vii} —Ho4—O9 ^{vi}	74.6 (3)	C47—C48—H48	119.5

O4 ^v —Ho4—O9 ^{vi}	127.6 (3)	O18—C49—O17	124.2 (10)
O22—Ho4—O6W	73.5 (3)	O18—C49—C50	119.6 (10)
O23 ^{viii} —Ho4—O6W	73.2 (3)	O17—C49—C50	116.1 (9)
O18—Ho4—O6W	76.3 (3)	C60—C50—C51	120.0 (10)
O16 ^{vii} —Ho4—O6W	139.0 (3)	C60—C50—C49	119.8 (10)
O4 ^v —Ho4—O6W	139.6 (3)	C51—C50—C49	120.1 (9)
O9 ^{vi} —Ho4—O6W	72.4 (3)	C50—C51—C52	119.3 (9)
O22—Ho4—O5W	70.5 (3)	C50—C51—H51	120.4
O23 ^{viii} —Ho4—O5W	138.4 (3)	C52—C51—H51	120.4
O18—Ho4—O5W	69.8 (3)	C51—C52—C58	120.0 (9)
O16 ^{vii} —Ho4—O5W	65.3 (3)	C51—C52—C53	121.0 (9)
O4 ^v —Ho4—O5W	65.4 (3)	C58—C52—C53	119.0 (10)
O9 ^{vi} —Ho4—O5W	135.1 (3)	C54—C53—C52	119.7 (10)
O6W—Ho4—O5W	129.2 (3)	C54—C53—H53	120.1
O22—Ho4—C44 ^{vii}	97.5 (3)	C52—C53—H53	120.1
O23 ^{viii} —Ho4—C44 ^{vii}	102.0 (3)	C53—C54—C55	120.7 (10)
O18—Ho4—C44 ^{vii}	148.7 (3)	C53—C54—H54	119.6
O16 ^{vii} —Ho4—C44 ^{vii}	23.0 (3)	C55—C54—H54	119.6
O4 ^v —Ho4—C44 ^{vii}	76.7 (3)	C57—C55—C54	120.1 (10)
O9 ^{vi} —Ho4—C44 ^{vii}	62.7 (3)	C57—C55—C56	119.4 (10)
O6W—Ho4—C44 ^{vii}	135.0 (3)	C54—C55—C56	119.8 (10)
O5W—Ho4—C44 ^{vii}	84.6 (3)	O19—C56—O20	122.3 (10)
Ho3—O5W—Ho4	126.7 (3)	O19—C56—C55	117.6 (10)
C1—O1—Ho1	154.4 (8)	O20—C56—C55	119.7 (9)
C1—O2—Ho2	143.2 (8)	O19—C56—Ho2 ^{viii}	50.0 (5)
C8—O3—Ho1 ⁱ	137.4 (8)	O20—C56—Ho2 ^{viii}	74.5 (6)
C8—O4—Ho4 ^v	129.9 (8)	C55—C56—Ho2 ^{viii}	155.2 (8)
C13—O5—Ho1	149.4 (7)	C55—C57—C58	120.9 (10)
C13—O6—Ho2	141.3 (7)	C55—C57—H57	119.5
C19—O7—Ho2	128.7 (8)	C58—C57—H57	119.5
C19—O8—Ho3	135.0 (7)	C57—C58—C52	119.2 (10)
C26—O9—Ho4 ^{iv}	138.6 (7)	C57—C58—C59	122.5 (10)
C26—O10—Ho1 ^{viii}	129.6 (7)	C52—C58—C59	118.0 (10)
C31—O11—Ho2	171.8 (7)	C60—C59—C58	121.2 (10)
C31—O12—Ho3	107.0 (7)	C60—C59—H59	119.4
C37—O13—Ho2	136.7 (7)	C58—C59—H59	119.4
C37—O14—Ho3	127.7 (7)	C59—C60—C50	121.4 (10)
C44—O15—Ho1 ⁱⁱⁱ	170.7 (8)	C59—C60—H60	119.3
C44—O16—Ho4 ^{vii}	109.8 (7)	C50—C60—H60	119.3
C49—O17—Ho3	136.5 (7)	O21—C61—O22	124.3 (10)
C49—O18—Ho4	138.1 (7)	O21—C61—C62	119.3 (10)
C56—O19—Ho2 ^{viii}	105.6 (7)	O22—C61—C62	116.2 (10)
C56—O20—Ho3 ^{viii}	176.9 (7)	C72—C62—C63	120.5 (10)

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C56—O20—Ho2 ^{viii}	80.4 (6)	C72—C62—C61	120.8 (10)
Ho3 ^{viii} —O20—Ho2 ^{viii}	101.8 (3)	C63—C62—C61	118.6 (10)
C61—O21—Ho3	138.2 (7)	C64—C63—C62	119.3 (10)
C61—O22—Ho4	135.3 (7)	C64—C63—H63	120.4
C68—O23—Ho4 ⁱⁱ	176.8 (8)	C62—C63—H63	120.4
C68—O23—Ho1 ^{vi}	79.4 (7)	C63—C64—C65	121.3 (10)
Ho4 ⁱⁱ —O23—Ho1 ^{vi}	102.5 (3)	C63—C64—H64	119.3
C68—O24—Ho1 ^{vi}	103.9 (7)	C65—C64—H64	119.3
O2—C1—O1	124.4 (10)	C66—C65—C71	118.3 (10)
O2—C1—C2	118.8 (10)	C66—C65—C64	122.2 (10)
O1—C1—C2	116.7 (9)	C71—C65—C64	119.4 (10)
C3—C2—C12	119.0 (10)	C67—C66—C65	121.9 (10)
C3—C2—C1	121.8 (10)	C67—C66—H66	119.1
C12—C2—C1	119.2 (10)	C65—C66—H66	119.1
C2—C3—C4	121.5 (10)	C66—C67—C69	117.9 (10)
C2—C3—H3	119.2	C66—C67—C68	117.9 (10)
C4—C3—H3	119.2	C69—C67—C68	123.8 (10)
C5—C4—C10	117.8 (10)	O23—C68—O24	122.3 (11)
C5—C4—C3	124.3 (10)	O23—C68—C67	118.6 (10)
C10—C4—C3	118.0 (10)	O24—C68—C67	119.1 (10)
C6—C5—C4	122.0 (10)	O23—C68—Ho1 ^{vi}	75.6 (7)
C6—C5—H5	119.0	O24—C68—Ho1 ^{vi}	50.9 (6)
C4—C5—H5	119.0	C67—C68—Ho1 ^{vi}	152.8 (8)
C5—C6—C7	119.6 (11)	C70—C69—C67	120.8 (10)
C5—C6—H6	120.2	C70—C69—H69	119.6
C7—C6—H6	120.2	C67—C69—H69	119.6
C9—C7—C6	119.8 (10)	C69—C70—C71	122.5 (10)
C9—C7—C8	118.7 (10)	C69—C70—H70	118.8
C6—C7—C8	121.4 (10)	C71—C70—H70	118.8
O4—C8—O3	123.2 (11)	C70—C71—C72	122.9 (10)
O4—C8—C7	118.9 (10)	C70—C71—C65	118.3 (10)
O3—C8—C7	117.8 (10)	C72—C71—C65	118.6 (10)
C7—C9—C10	121.1 (10)	C62—C72—C71	120.7 (10)
C7—C9—H9	119.5	C62—C72—H72	119.7
C10—C9—H9	119.5	C71—C72—H72	119.7
C9—C10—C4	119.7 (10)		
O20 ⁱⁱ —Ho3—O5W—Ho4	-119.2 (4)	C20—C21—C22—C28	0.6 (16)
O17—Ho3—O5W—Ho4	38.6 (3)	C21—C22—C23—C24	-176.0 (10)
O21—Ho3—O5W—Ho4	-59.5 (4)	C28—C22—C23—C24	1.7 (16)
O12—Ho3—O5W—Ho4	123.8 (4)	C22—C23—C24—C25	-2.8 (17)
O14—Ho3—O5W—Ho4	-145.6 (4)	C23—C24—C25—C27	1.5 (17)
O8—Ho3—O5W—Ho4	91.3 (5)	C23—C24—C25—C26	177.8 (10)
O4W—Ho3—O5W—Ho4	-11.0 (5)	Ho4 ^{iv} —O9—C26—O10	-14.1 (19)
C31—Ho3—O5W—Ho4	136.9 (4)	Ho4 ^{iv} —O9—C26—C25	165.6 (8)
O22—Ho4—O5W—Ho3	38.8 (4)	Ho1 ^{viii} —O10—C26—O9	52.1 (15)
O23 ^{viii} —Ho4—O5W—Ho3	-119.3 (4)	Ho1 ^{viii} —O10—C26—C25	-127.6 (9)

O18—Ho4—O5W—Ho3	-59.7 (4)	C27—C25—C26—O9	-39.3 (16)
O16 ^{vii} —Ho4—O5W—Ho3	125.6 (5)	C24—C25—C26—O9	144.3 (11)
O4 ^v —Ho4—O5W—Ho3	-143.5 (5)	C27—C25—C26—O10	140.3 (11)
O9 ^{vi} —Ho4—O5W—Ho3	96.9 (5)	C24—C25—C26—O10	-36.0 (15)
O6W—Ho4—O5W—Ho3	-8.5 (6)	C24—C25—C27—C28	0.9 (17)
C44 ^{vii} —Ho4—O5W—Ho3	138.7 (4)	C26—C25—C27—C28	-175.5 (10)
O5—Ho1—O1—C1	-111.2 (17)	C25—C27—C28—C22	-1.9 (16)
O15 ⁱⁱⁱ —Ho1—O1—C1	167.1 (15)	C25—C27—C28—C29	177.8 (11)
O24 ^{iv} —Ho1—O1—C1	14.6 (16)	C21—C22—C28—C27	178.4 (10)
O10 ⁱⁱ —Ho1—O1—C1	-44.6 (18)	C23—C22—C28—C27	0.6 (16)
O3 ⁱ —Ho1—O1—C1	97.9 (17)	C21—C22—C28—C29	-1.3 (16)
O2W—Ho1—O1—C1	169.2 (17)	C23—C22—C28—C29	-179.2 (10)
O23 ^{iv} —Ho1—O1—C1	38.7 (17)	C27—C28—C29—C30	-179.9 (11)
C68 ^{iv} —Ho1—O1—C1	31.0 (17)	C22—C28—C29—C30	-0.2 (17)
O11—Ho2—O2—C1	174.7 (14)	C28—C29—C30—C20	2.4 (18)
O6—Ho2—O2—C1	22.5 (14)	C21—C20—C30—C29	-3.2 (17)
O19 ⁱⁱ —Ho2—O2—C1	-46.0 (15)	C19—C20—C30—C29	-176.0 (10)
O7—Ho2—O2—C1	-109.5 (14)	Ho3—O12—C31—O11	-20.8 (13)
O13—Ho2—O2—C1	95.3 (15)	Ho3—O12—C31—C32	151.5 (8)
O3W—Ho2—O2—C1	99.4 (14)	O20 ⁱⁱ —Ho3—C31—O11	-7.7 (7)
O20 ⁱⁱ —Ho2—O2—C1	-107.6 (13)	O17—Ho3—C31—O11	-158.8 (6)
C56 ⁱⁱ —Ho2—O2—C1	-68.4 (15)	O21—Ho3—C31—O11	97.2 (8)
O15 ⁱⁱⁱ —Ho1—O5—C13	168.4 (16)	O12—Ho3—C31—O11	161.9 (11)
O24 ^{iv} —Ho1—O5—C13	-50.0 (16)	O14—Ho3—C31—O11	64.7 (6)
O1—Ho1—O5—C13	14.7 (15)	O8—Ho3—C31—O11	-81.9 (6)
O10 ⁱⁱ —Ho1—O5—C13	-115.7 (16)	O4W—Ho3—C31—O11	-85.9 (7)
O3 ⁱ —Ho1—O5—C13	84.9 (17)	O5W—Ho3—C31—O11	131.0 (6)
O2W—Ho1—O5—C13	92.6 (16)	O20 ⁱⁱ —Ho3—C31—O12	-169.6 (7)
O23 ^{iv} —Ho1—O5—C13	-111.1 (15)	O17—Ho3—C31—O12	39.3 (8)
C68 ^{iv} —Ho1—O5—C13	-68.3 (17)	O21—Ho3—C31—O12	-64.7 (10)
O11—Ho2—O6—C13	166.5 (10)	O14—Ho3—C31—O12	-97.2 (7)
O2—Ho2—O6—C13	-104.7 (12)	O8—Ho3—C31—O12	116.2 (8)
O19 ⁱⁱ —Ho2—O6—C13	19.5 (11)	O4W—Ho3—C31—O12	112.2 (7)
O7—Ho2—O6—C13	-40.0 (13)	O5W—Ho3—C31—O12	-30.9 (7)
O13—Ho2—O6—C13	100.7 (12)	O20 ⁱⁱ —Ho3—C31—C32	120.3 (16)
O3W—Ho2—O6—C13	177.4 (12)	O17—Ho3—C31—C32	-30.8 (17)
O20 ⁱⁱ —Ho2—O6—C13	42.3 (12)	O21—Ho3—C31—C32	-134.8 (15)
C56 ⁱⁱ —Ho2—O6—C13	33.7 (12)	O12—Ho3—C31—C32	-70.1 (17)
O11—Ho2—O7—C19	-0.8 (9)	O14—Ho3—C31—C32	-167.3 (17)
O2—Ho2—O7—C19	-99.2 (10)	O8—Ho3—C31—C32	46.1 (16)
O6—Ho2—O7—C19	-168.5 (9)	O4W—Ho3—C31—C32	42.1 (18)
O19 ⁱⁱ —Ho2—O7—C19	132.8 (10)	O5W—Ho3—C31—C32	-100.9 (16)
O13—Ho2—O7—C19	66.7 (10)	O11—C31—C32—C33	-9.9 (16)
O3W—Ho2—O7—C19	-53.0 (11)	O12—C31—C32—C33	177.6 (10)

supplementary materials

O20 ⁱⁱ —Ho2—O7—C19	82.0 (9)	Ho3—C31—C32—C33	-128.7 (14)
C56 ⁱⁱ —Ho2—O7—C19	107.6 (10)	O11—C31—C32—C36	165.2 (10)
O20 ⁱⁱ —Ho3—O8—C19	-32.2 (10)	O12—C31—C32—C36	-7.3 (16)
O17—Ho3—O8—C19	177.4 (11)	Ho3—C31—C32—C36	46 (2)
O21—Ho3—O8—C19	-103.7 (11)	C36—C32—C33—C34	-3.4 (17)
O12—Ho3—O8—C19	97.5 (10)	C31—C32—C33—C34	171.6 (10)
O14—Ho3—O8—C19	31.5 (11)	C32—C33—C34—C35 ^v	2.0 (17)
O4W—Ho3—O8—C19	-107.4 (11)	C35 ^v —C35—C36—C32	-3.0 (19)
O5W—Ho3—O8—C19	127.9 (10)	C34 ^v —C35—C36—C32	178.6 (10)
C31—Ho3—O8—C19	75.6 (10)	C33—C32—C36—C35	3.9 (17)
O20 ⁱⁱ —Ho3—O12—C31	12.5 (9)	C31—C32—C36—C35	-171.2 (10)
O17—Ho3—O12—C31	-140.0 (8)	Ho2—O13—C37—O14	-22.7 (19)
O21—Ho3—O12—C31	141.4 (7)	Ho2—O13—C37—C38	156.1 (8)
O14—Ho3—O12—C31	78.2 (7)	Ho3—O14—C37—O13	57.0 (15)
O8—Ho3—O12—C31	-56.3 (7)	Ho3—O14—C37—C38	-121.8 (9)
O4W—Ho3—O12—C31	-93.9 (8)	O13—C37—C38—C48	-38.8 (15)
O5W—Ho3—O12—C31	146.0 (8)	O14—C37—C38—C48	140.0 (11)
O11—Ho2—O13—C37	38.7 (11)	O13—C37—C38—C39	142.3 (11)
O2—Ho2—O13—C37	121.3 (11)	O14—C37—C38—C39	-38.9 (15)
O6—Ho2—O13—C37	-166.7 (11)	C48—C38—C39—C40	2.3 (17)
O19 ⁱⁱ —Ho2—O13—C37	-91.7 (11)	C37—C38—C39—C40	-178.8 (10)
O7—Ho2—O13—C37	-27.2 (13)	C38—C39—C40—C41	-1.2 (17)
O3W—Ho2—O13—C37	117.1 (12)	C39—C40—C41—C42	176.2 (11)
O20 ⁱⁱ —Ho2—O13—C37	-43.0 (11)	C39—C40—C41—C47	-2.1 (16)
C56 ⁱⁱ —Ho2—O13—C37	-70.4 (11)	C40—C41—C42—C43	-178.7 (10)
O20 ⁱⁱ —Ho3—O14—C37	5.3 (8)	C47—C41—C42—C43	-0.4 (16)
O17—Ho3—O14—C37	173.3 (8)	C41—C42—C43—C45	-3.5 (15)
O21—Ho3—O14—C37	94.6 (9)	C41—C42—C43—C44	169.5 (10)
O12—Ho3—O14—C37	-124.6 (9)	Ho4 ^{vii} —O16—C44—O15	22.1 (14)
O8—Ho3—O14—C37	-60.6 (10)	Ho4 ^{vii} —O16—C44—C43	-152.1 (8)
O4W—Ho3—O14—C37	47.6 (10)	C42—C43—C44—O15	-166.0 (10)
O5W—Ho3—O14—C37	167.6 (9)	C45—C43—C44—O15	7.0 (16)
C31—Ho3—O14—C37	-100.6 (9)	C42—C43—C44—O16	8.4 (16)
O20 ⁱⁱ —Ho3—O17—C49	-166.4 (9)	C45—C43—C44—O16	-178.7 (10)
O21—Ho3—O17—C49	108.6 (10)	C42—C43—C44—Ho4 ^{vii}	-42 (2)
O12—Ho3—O17—C49	-28.6 (10)	C45—C43—C44—Ho4 ^{vii}	130.7 (14)
O14—Ho3—O17—C49	34.6 (11)	C42—C43—C45—C46	4.1 (16)
O8—Ho3—O17—C49	-106.3 (10)	C44—C43—C45—C46	-169.1 (10)
O4W—Ho3—O17—C49	-179.2 (10)	C43—C45—C46—C47	-0.7 (16)
O5W—Ho3—O17—C49	40.1 (9)	C42—C41—C47—C48	-174.1 (9)
C31—Ho3—O17—C49	-43.7 (10)	C40—C41—C47—C48	4.3 (15)
O22—Ho4—O18—C49	-31.4 (12)	C42—C41—C47—C46	3.8 (15)
O23 ^{viii} —Ho4—O18—C49	-177.7 (12)	C40—C41—C47—C46	-177.8 (10)
O16 ^{vii} —Ho4—O18—C49	44.2 (13)	C45—C46—C47—C48	174.6 (10)
O4 ^v —Ho4—O18—C49	105.9 (12)	C45—C46—C47—C41	-3.3 (16)

O9 ^{vi} —Ho4—O18—C49	-110.2 (11)	C39—C38—C48—C47	-0.1 (17)
O6W—Ho4—O18—C49	-104.2 (12)	C37—C38—C48—C47	-179.0 (10)
O5W—Ho4—O18—C49	37.4 (11)	C41—C47—C48—C38	-3.3 (16)
C44 ^{vii} —Ho4—O18—C49	74.5 (13)	C46—C47—C48—C38	178.9 (11)
O20 ⁱⁱ —Ho3—O21—C61	179.4 (12)	Ho4—O18—C49—O17	17.4 (18)
O17—Ho3—O21—C61	-34.7 (11)	Ho4—O18—C49—C50	-165.5 (8)
O12—Ho3—O21—C61	38.9 (13)	Ho3—O17—C49—O18	-83.7 (14)
O14—Ho3—O21—C61	103.1 (12)	Ho3—O17—C49—C50	99.1 (11)
O8—Ho3—O21—C61	-110.6 (11)	O18—C49—C50—C60	-165.1 (10)
O4W—Ho3—O21—C61	-107.0 (12)	O17—C49—C50—C60	12.2 (15)
O5W—Ho3—O21—C61	34.5 (11)	O18—C49—C50—C51	11.2 (16)
C31—Ho3—O21—C61	70.7 (14)	O17—C49—C50—C51	-171.4 (10)
O23 ^{viii} —Ho4—O22—C61	-167.4 (9)	C60—C50—C51—C52	3.5 (16)
O18—Ho4—O22—C61	107.3 (10)	C49—C50—C51—C52	-172.8 (9)
O16 ^{vii} —Ho4—O22—C61	-28.3 (10)	C50—C51—C52—C58	-2.0 (15)
O4 ^v —Ho4—O22—C61	36.1 (11)	C50—C51—C52—C53	176.9 (10)
O9 ^{vi} —Ho4—O22—C61	-103.8 (10)	C51—C52—C53—C54	-172.7 (10)
O6W—Ho4—O22—C61	-177.3 (10)	C58—C52—C53—C54	6.2 (16)
O5W—Ho4—O22—C61	39.1 (10)	C52—C53—C54—C55	-2.1 (16)
C44 ^{vii} —Ho4—O22—C61	-42.5 (10)	C53—C54—C55—C57	-2.5 (17)
Ho2—O2—C1—O1	-4(2)	C53—C54—C55—C56	168.0 (10)
Ho2—O2—C1—C2	-179.4 (8)	Ho2 ^{viii} —O19—C56—O20	-19.2 (12)
Ho1—O1—C1—O2	92 (2)	Ho2 ^{viii} —O19—C56—C55	153.9 (8)
Ho1—O1—C1—C2	-92.3 (18)	Ho2 ^{viii} —O20—C56—O19	15.1 (10)
O2—C1—C2—C3	-18.2 (17)	Ho2 ^{viii} —O20—C56—C55	-157.9 (10)
O1—C1—C2—C3	165.8 (11)	C57—C55—C56—O19	-12.4 (16)
O2—C1—C2—C12	160.1 (11)	C54—C55—C56—O19	177.0 (10)
O1—C1—C2—C12	-15.9 (15)	C57—C55—C56—O20	161.0 (10)
C12—C2—C3—C4	-0.2 (16)	C54—C55—C56—O20	-9.6 (16)
C1—C2—C3—C4	178.1 (10)	C57—C55—C56—Ho2 ^{viii}	41 (2)
C2—C3—C4—C5	179.2 (11)	C54—C55—C56—Ho2 ^{viii}	-129.7 (15)
C2—C3—C4—C10	0.8 (16)	C54—C55—C57—C58	2.9 (17)
C10—C4—C5—C6	-1.1 (16)	C56—C55—C57—C58	-167.6 (10)
C3—C4—C5—C6	-179.5 (11)	C55—C57—C58—C52	1.3 (16)
C4—C5—C6—C7	3.6 (17)	C55—C57—C58—C59	175.0 (11)
C5—C6—C7—C9	-3.2 (17)	C51—C52—C58—C57	173.1 (9)
C5—C6—C7—C8	-179.6 (10)	C53—C52—C58—C57	-5.8 (15)
Ho4 ^v —O4—C8—O3	-55.0 (16)	C51—C52—C58—C59	-0.9 (15)
Ho4 ^v —O4—C8—C7	121.4 (10)	C53—C52—C58—C59	-179.9 (9)
Ho1 ⁱ —O3—C8—O4	14.9 (19)	C57—C58—C59—C60	-171.3 (10)
Ho1 ⁱ —O3—C8—C7	-161.5 (8)	C52—C58—C59—C60	2.6 (16)
C9—C7—C8—O4	-138.4 (12)	C58—C59—C60—C50	-1.2 (17)
C6—C7—C8—O4	38.0 (16)	C51—C50—C60—C59	-1.9 (17)
C9—C7—C8—O3	38.2 (16)	C49—C50—C60—C59	174.4 (10)
C6—C7—C8—O3	-145.4 (11)	Ho3—O21—C61—O22	22.6 (18)

supplementary materials

C6—C7—C9—C10	0.4 (17)	Ho3—O21—C61—C62	-161.9 (8)
C8—C7—C9—C10	176.9 (10)	Ho4—O22—C61—O21	-85.9 (14)
C7—C9—C10—C4	2.1 (16)	Ho4—O22—C61—C62	98.5 (11)
C7—C9—C10—C11	179.8 (10)	O21—C61—C62—C72	13.2 (16)
C5—C4—C10—C9	-1.7 (16)	O22—C61—C62—C72	-171.0 (10)
C3—C4—C10—C9	176.8 (9)	O21—C61—C62—C63	-165.0 (10)
C5—C4—C10—C11	-179.5 (10)	O22—C61—C62—C63	10.9 (15)
C3—C4—C10—C11	-0.9 (16)	C72—C62—C63—C64	-2.6 (16)
C9—C10—C11—C12	-177.2 (10)	C61—C62—C63—C64	175.6 (10)
C4—C10—C11—C12	0.5 (16)	C62—C63—C64—C65	-1.0 (16)
C10—C11—C12—C2	0.1 (16)	C63—C64—C65—C66	-173.8 (10)
C3—C2—C12—C11	-0.3 (16)	C63—C64—C65—C71	3.8 (16)
C1—C2—C12—C11	-178.6 (10)	C71—C65—C66—C67	-3.6 (16)
Ho1—O5—C13—O6	-10 (2)	C64—C65—C66—C67	173.9 (10)
Ho1—O5—C13—C14	175.0 (10)	C65—C66—C67—C69	6.3 (16)
Ho2—O6—C13—O5	88.6 (15)	C65—C66—C67—C68	-166.6 (10)
Ho2—O6—C13—C14	-96.4 (13)	Ho1 ^{vi} —O23—C68—O24	21.2 (11)
O5—C13—C14—C18	-15.3 (16)	Ho1 ^{vi} —O23—C68—C67	-154.9 (10)
O6—C13—C14—C18	169.3 (10)	Ho1 ^{vi} —O24—C68—O23	-26.8 (13)
O5—C13—C14—C15	165.3 (10)	C66—C67—C68—O23	159.0 (11)
O6—C13—C14—C15	-10.0 (15)	C69—C67—C68—O23	-13.5 (17)
C18—C14—C15—C16	3.1 (16)	C66—C67—C68—O24	-17.2 (16)
C13—C14—C15—C16	-177.5 (10)	C69—C67—C68—O24	170.3 (11)
C14—C15—C16—C17	-2.4 (16)	C66—C67—C68—Ho1 ^{vi}	43 (2)
C15—C16—C17—C18 ⁱⁱⁱ	-178.1 (10)	C69—C67—C68—Ho1 ^{vi}	-129.5 (15)
C15—C16—C17—C17 ⁱⁱⁱ	-0.1 (18)	C66—C67—C69—C70	-4.5 (16)
C15—C14—C18—C17 ⁱⁱⁱ	-1.3 (16)	C68—C67—C69—C70	168.0 (11)
C13—C14—C18—C17 ⁱⁱⁱ	179.3 (10)	C67—C69—C70—C71	-0.1 (17)
Ho2—O7—C19—O8	-57.9 (16)	C69—C70—C71—C72	-172.6 (11)
Ho2—O7—C19—C20	121.5 (9)	C69—C70—C71—C65	2.9 (16)
Ho3—O8—C19—O7	15.9 (18)	C66—C65—C71—C70	-1.0 (15)
Ho3—O8—C19—C20	-163.5 (7)	C64—C65—C71—C70	-178.7 (10)
O7—C19—C20—C21	-137.2 (11)	C66—C65—C71—C72	174.7 (10)
O8—C19—C20—C21	42.2 (15)	C64—C65—C71—C72	-3.0 (15)
O7—C19—C20—C30	35.8 (16)	C63—C62—C72—C71	3.3 (16)
O8—C19—C20—C30	-144.8 (11)	C61—C62—C72—C71	-174.8 (9)
C30—C20—C21—C22	1.7 (16)	C70—C71—C72—C62	175.0 (10)
C19—C20—C21—C22	174.6 (10)	C65—C71—C72—C62	-0.5 (15)
C20—C21—C22—C23	178.3 (11)		

Symmetry codes: (i) $-x+1, -y+1, -z+2$; (ii) $x-1/2, -y+1/2, z-1/2$; (iii) $-x+1, -y+1, -z+1$; (iv) $x-1/2, -y+1/2, z+1/2$; (v) $-x+2, -y+1, -z+2$; (vi) $x+1/2, -y+1/2, z-1/2$; (vii) $-x+2, -y+1, -z+1$; (viii) $x+1/2, -y+1/2, z+1/2$.

Fig. 1

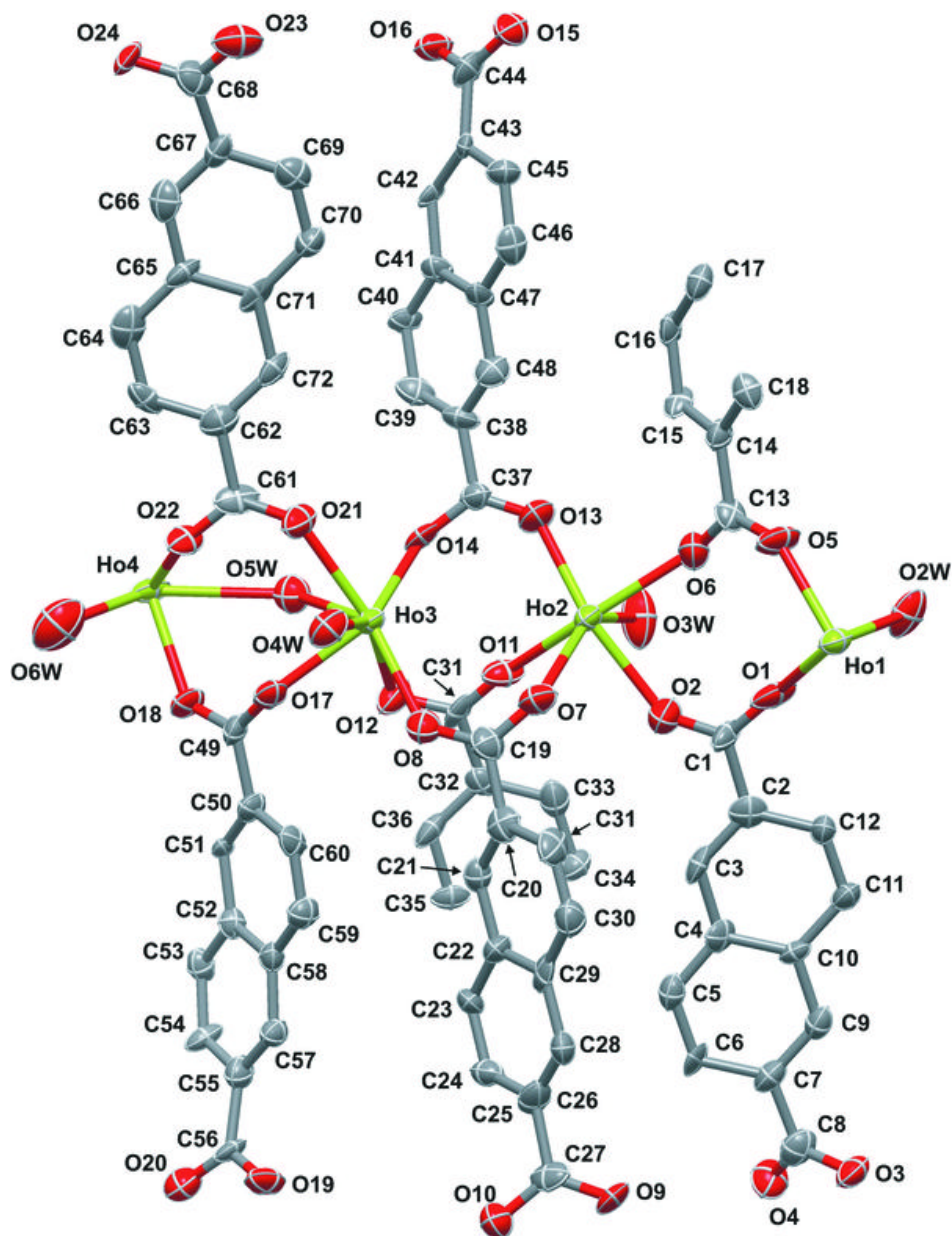


Fig. 2

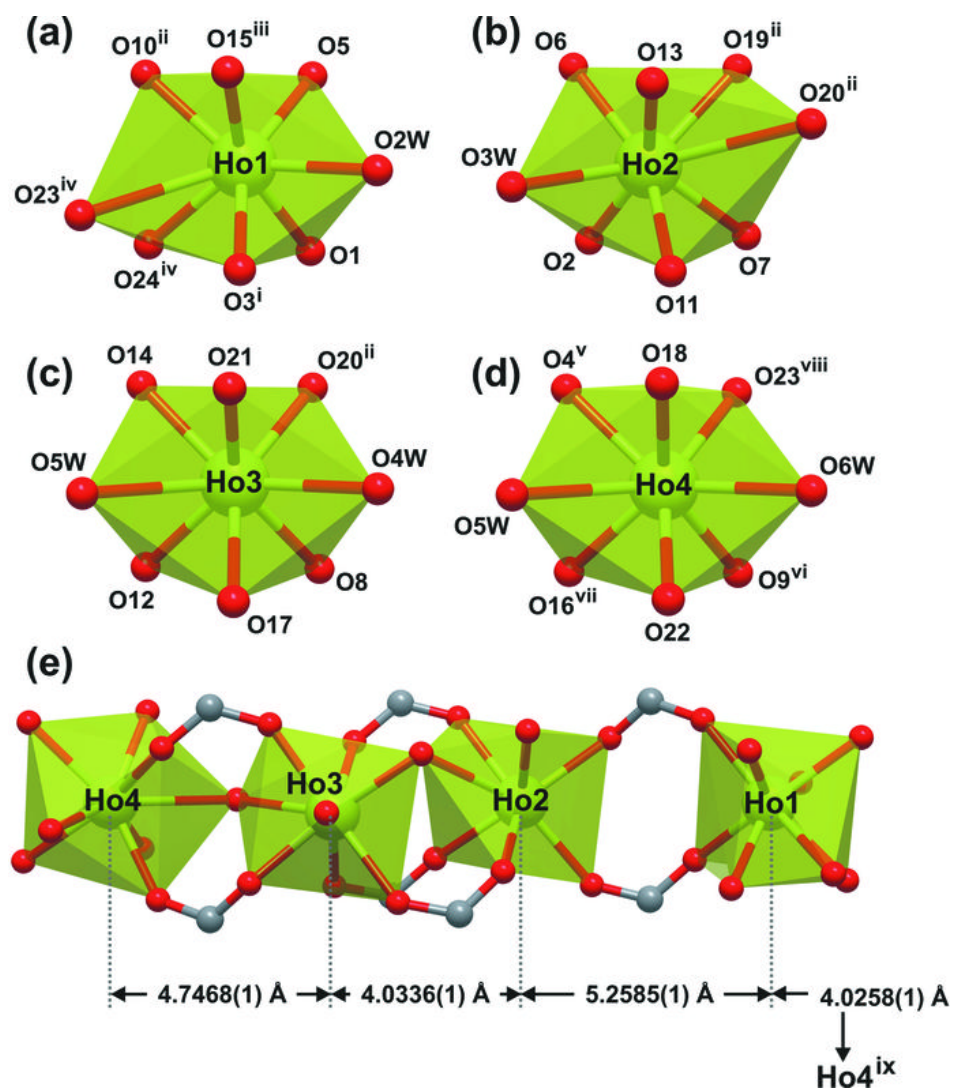


Fig. 3

