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A1, ³-strain enabled retention of chirality during bis-cyclization of β-ketoamides: total synthesis of (−)-salinosporamide A and (−) homosalinosporamide A†

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Abstract

A concise, enantioselective synthesis of the Phase I anticancer agent, (−)-salinosporamide A, is described. The brevity of the described strategy stems from a key bis-cyclization of a β-keto tertiary amide, accomplished on gram scale, which retains optical purity enabled by $A¹$, 3-strain rendering epimerization slow relative to the rate of bis-cyclization. The versatility of the strategy for derivative synthesis is demonstrated by the synthesis of (−)-homosalinosporamide A.

> Inhibitors of the human 20S proteasome are of continued intense interest due to their potential as anticancer therapeutics and the recent FDA approval of bortezomib (Velcade) validates the proteasome as a target for cancer chemotherapy.¹ Salinosporamides (salino A/B, **1a–b**) are unique bicyclo [3.2.0] β-lactone-containing natural products of marine bacterial origin isolated by Fenical and coworkers² from the marine actinomycete, *Salinispora tropica* (Fig. 1). Salino A is a potent nanomolar inhibitor of the proteasome and is currently in Phase I clinical studies for multiple myeloma having shown potential in mouse models toward several cancers when administered intravenously despite the potentially labile β -lactone.³ Salinosporamide A and congeners, which bear close structural similarities to the terrestrial metabolites, the cinnabaramides (*e.g.* cinnabaramide A, **2**) and lactacystin-β-lactone (omuralide),⁴ have been targets for total5 and formal6 syntheses, structure–activity studies,7 biosynthetic engineering, ⁸ and crystallographic studies with the 20S proteasome.⁹ The latter studies revealed an intriguing mode of action for salino A involving *O*-acylation of the *N*-terminus threonine of the proteasome by the β-lactone with concomitant cyclization of the incipient alkoxide with the C13 chloro substituent leading to a tetrahydrofuran. Herein we disclose a versatile and concise, enantioselective strategy to both (−)-salino A and (−)-homosalino A made possible by the $A¹,³$ -strain of β-keto tertiary amides¹⁰ which enables retention of optical purity during a key bis-cyclization process that simultaneously forms the γ-lactam and fused β-lactone core.

> Building on our previously reported racemic synthesis of salino A,^{5e} a key step in our strategy is a diastereoselective, nucleophile-promoted, bis-cyclization (Fig. 2). This process delivers the bicyclic-β-lactone pharmacophore **4** in a single operation from acyclic precursors and the optical purity of the β-lactone would reflect the diastereomeric purity of β-ketoamide **5**, derived from acylation of serine derivatives with racemic ketene dimers **7**. The C4 stereocenter in β-

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ketoamide **5** is lost during the bis-cyclization process, thus the integrity of the C2 stereocenter dictates the optical purity of β-lactone 4 which we predicted would be preserved by $A^{1,3}$ strain.

The synthesis of salino A began with reductive amination of commercially available (*R*)-*O*benzyl serine (**8**) (99% ee) with *p*-anisaldehyde, followed by esterification to provide allyl ester (+)-**9** (Scheme 1). After extensive optimization by careful control of reaction temperature and time, the desired ester (+)-**9** could be obtained with minimal racemization (98% ee) through these two steps in 79% yield.[§] The required unsymmetrical ketene dimer (\pm) -7a was obtained by heterodimerization of acetyl chloride and 4-chlorobutanoyl chloride following the procedure of Sauer.¶11 Acylation of serine derivative (+)-**9** with this ketene dimer under microwave conditions gave diastereomeric β-ketoamides **10a**/**10a**′ (dr 1 : 1) in 80% yield. Separation of the diastereomers by MPLC provided the required (*R,R*)-β-ketoamide **10a** (45%, dr 30 : 1, 98% ee). Conditions were identified to recycle the undesired diastereomer, (*R,S*)-**10a** ′, *via* epimerization (98% yield, dr 1 : 1) of the C2 but not the C4 stereocenter thus allowing an effective resolution of ketene dimer (\pm) -**7a** and greater material throughput.^{||}

Towards the required ketoacid substrate for bis-cyclization, mild and brief exposure of βketoamide (*R,R*)-**10a** to Pd(0)-mediated deprotection provided keto acid (*R,R*)-**5** with only slight erosion of the diastereomeric purity (Table 2). At the outset, it was unclear whether the A¹,³-strain induced conformational bias in β-keto tertiary amides (~4 kcal mol⁻¹) would be sufficient to avoid epimerization of this center in the time frame and under the basic conditions of the bis-cyclization. As expected, applying bis-cyclization conditions that were previously developed for the racemic series to keto acid (R, R) -**5** (dr 29 : 1, 98% ee) led to similar yields and diastereoselectivity (40% yield, dr 2.5 : 1),^{5e} however erosion of optical purity of adduct **3a** was observed (85% ee). Towards exploring the impact of the chlorine atom on the acidity of the C2-proton and the versatility of this strategy for derivative synthesis, we studied variations of the C2-substituent by employing other heteroketene dimers for the synthesis of ketoamide substrates **5b–c** (see ESI† for details). Bis-cyclization of ketoacid **5b** bearing a γsiloxy group (Table 1, entry 2) led to similar results as ketoacid **5a** bearing a γ-chloro substituent, however the δ-chloro ketoacid **5c** led to a significant increase in yield (Table 1, entry 3). These results are mirrored by the ¹H NMR chemical shifts of the C2 protons of βketoamides **10a–c**, which show a Δ*δ* of 0.4 ppm between γ- and δ-substituted substrates, pointing to the inductive effects and resultant increased acidity imposed by these βheteroatoms.

Towards optimization of the synthesis of β-lactone **3a** required for salino A, reaction parameters were optimized extensively including alternative carboxylic acid activating agents. Ultimately we found that mesyl chloride at lower temperature and less polar solvents led to dramatic increases in both diastereoselectivity (7 : 1) and retained enantiopurity of bicyclic βlactone **3a** (92% ee, 35% yield, 28% recovered keto acid, Table 2, entry 1). In addition, a yield of 60% was obtained with longer reaction times under these conditions, however this led to reduced diastereoselectivity and enantiopurity (dr 4 : 1, 88% ee, Table 2, entry 2). *Importantly, this reaction could be performed on gram scale with comparable diastereoselectivity and retention of enantiopurity* (52%, dr 5 : 1, 90% ee, Table 2, entry 3).

Completion of the salino A synthesis entailed hydrogenolysis, which facilitated separation of the minor diastereomer **3a**′ from the bis-cyclization, to provide alcohol (−)-**12a** in 75% yield

[§]Optical purities were determined by chiral HPLC analysis. See ESI.†

[¶]While ketene dimer (±)-**7a** was obtained in low statistical yield (13%, 52% of theoretical), its synthesis is readily performed on multigram scale. See ESI.†

^{||}Chiral HPLC analysis verified that epimerization occurred only at C2. See ESI.[†]

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(Scheme 2). Modified Moffatt oxidation¹² and addition of zinc reagent 13^{5a} gave diastereomeric alcohols (dr 11 : 3 : 1 : 1) in 62% yield (2 steps) and final PMB deprotection of the mixture of diastereomers led to pure (−)-salinosporamide A following recrystallization (syn. $\lceil \alpha \rceil_D - 71.3$; lit. $\lceil \alpha \rceil_D - 72.9$).

The synthesis of homosalino A (**14**), a derivative with the potential to form a tetrahydropyran upon acylation of the proteasome, followed an identical route as for salinosporamide A but using heteroketene dimer (\pm) -**7c** with serine derivative $(+)$ -**9** to provide the homologous ketoester (−)-**10c** (94% ee, dr 25 : 1) after MPLC separation (Scheme 3). Ester deprotection and subsequent bis-cyclization provided the bicyclic-β-lactone (−)-**3c**/**3c**′ in 60% yield (dr 3.5 : 1, 2 steps). Subsequent hydrogenolysis again enabled separation of the minor diastereomer to give alcohol (−)-**12c** in 62% yield (82% ee) and recrystallization led to enrichment of enantiopurity to 89% ee (52% yield). Using an identical sequence as described for salino A, the β-lactone core (−)-**12c** was converted to (−)-homosalino A (**14**). The structure and relative stereochemistry of **14** was confirmed by X-ray analysis (Fig. 3).

In summary, we have developed a concise, 9-step enantioselective route to (−)-salinosporamide A from *O*-benzyl serine. The key bis-cyclization of a β-ketoamide, amenable to gram scale, constructs both the γ-lactam and the fused-β-lactone in one operation contributing to the brevity of the synthesis. The ability of the described β-keto *tertiary* amide substrates to maintain stereochemical integrity by virtue of $A¹$, 3 strain raises the intriguing question of how such stereochemical integrity is maintained with β-keto *secondary* amides, known salino A precursors, prior to and during related biosynthetic cyclizations or if a dynamic kinetic resolution is operative.** The flexibility of the described strategy derives from the versatility of bicyclic cores *e.g.* **3a–c** obtained from alternative ketene dimers **7** which enable variation of the C2 side chain as demonstrated in the synthesis of (−)-homosalinosporamide A. Furthermore, the strategy allows for changes in the C4 side chain by addition of various organometallic reagents, notably in the presence of the β-lactone. The synthesis of additional hypothesis-driven salino A derivatives by the described strategy and their bioactivity will be reported in due course.

Supplementary Material

Refer to Web version on PubMed Central for supplementary material.

Acknowledgments

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^{**}C2-Epimers of salinosporamides have been isolated during large-scale fermentation of *S. tropica* (see ref. 7*a*).

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Fig. 3. X-Ray structure (ORTEP) of (−)-homosalino A (**14**).

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Scheme 1.

Reagents and conditions: (a) p-anisaldehyde, MeOH, 0 °C, 8 h; NaBH4, MeOH, 12 h, −10 ° C (97%); (b) p-TsOH, allyl alcohol, PhH, 100 °C, 8 h (82%); (c) (\pm)-7a, ClCH₂CH₂Cl, pyridone, μW, 50 °C, 2 h (80%); (d) p-TsOH, MeOH : EtOAc (1 : 4), 45 °C, 48 h (98%).

Scheme 2.

Reagents and conditions: (a) H_2 , Pd/C THF, 23 °C, 12 h (75%); (b) (i) EDCl, Cl₂CHCO₂H, DMSO/PhMe, 23 °C, 5 h; (ii) 2-cyclohexenylzinc chloride (**13**), THF, −78 °C, 4 h (62%, dr = 11 : 3 : 1 : 1); (c) CAN, MeOH/H2O, 0 °C, 6 h (43%, dr 15 : 1).

Scheme 3.

Reagents and conditions: (a) (−)-7c, pyridone, ClCH₂CH₂Cl, μW, 53 °C, 2 h (80%, dr 1 : 1); (b) Pd(PPh3)4, morpholine, THF, 1 h, −5 °C; (c) TsCl, 4-PPy, PhMe, −5 °C, 3.5 h (60%, dr 3.5 : 1); (d) H2, Pd/C, THF, 23 °C, 12 h (62%, 82% ee; following recrystallization: 52%, 89% ee).

Table 1

Variation of the C2-side chain: inductive effects on efficiency of bis-cyclization with β-ketoacids **5a–c**

^{*a*} Yield for 2 steps (deprotection and bis-cyclization).

b Yields refer to isolated, purified β-lactones.

c Chemical shift of C2 protons (Ha/b) of diastereomeric ketoamide esters **10a–c**.

Table 2

Varying conditions for the bis-cyclization leading to β -lactone (-)-3a including a gram scale synthesis −)-**3a** including a gram scale synthesis Varying conditions for the bis-cyclization leading to β-lactone (

 $\alpha_{\rm Yield}$ for 2 steps (deprotection and bis-cyclization). *a*Yield for 2 steps (deprotection and bis-cyclization).

3*c*

 $c = 3.5$ 52 90 5:1 11

 52

 3.5

 Ξ

 $5:1$

 ∞

 $b_{\mbox{\small{Yields}}}$ refer to isolated, purified β-lactones. *b*Yields refer to isolated, purified β-lactones.

 ${}^{\prime} \! \mathrm{This}$ run was performed on gram scale. *c*This run was performed on gram scale.

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