

# Mechanism for the inhibition of the carboxyltransferase domain of acetyl-coenzyme A carboxylase by pinoxaden

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**Acetyl-CoA carboxylases (ACCs) are crucial metabolic enzymes and have been targeted for drug development against obesity, diabetes, and other diseases. The carboxyltransferase (CT) domain of this enzyme is the site of action for three different classes of herbicides, as represented by haloxyfop, tepraloxymid, and pinoxaden. Our earlier studies have demonstrated that haloxyfop and tepraloxymid bind in the CT active site at the interface of its dimer. However, the two compounds probe distinct regions of the dimer interface, sharing primarily only two common anchoring points of interaction with the enzyme. We report here the crystal structure of the CT domain of yeast ACC in complex with pinoxaden at 2.8-Å resolution. Despite their chemical diversity, pinoxaden has a similar binding mode as tepraloxymid and requires a small conformational change in the dimer interface for binding. Crystal structures of the CT domain in complex with all three classes of herbicides confirm the importance of the two anchoring points for herbicide binding. The structures also provide a foundation for understanding the molecular basis of the herbicide resistance mutations and cross resistance among the herbicides, as well as for the design and development of new inhibitors against plant and human ACCs.**

fatty acid metabolism | metabolic syndrome | structure-based drug design

**A**cetyl-CoA carboxylases (ACCs or ACCases) are biotin-dependent enzymes required for fatty acid metabolism. ACCs catalyze the carboxylation of acetyl-CoA to malonyl-CoA in two steps, through the action of three distinct protein components (1, 2). In the first step, a biotin moiety covalently linked to the biotin carboxyl carrier protein (BCCP) component enters the biotin carboxylase (BC) active site for carboxylation of the biotin. In the second step, the carboxylated biotin translocates to the carboxyltransferase (CT) active site to transfer the carboxyl group to the acetyl-CoA substrate, producing malonyl-CoA. In bacterial species, BC, BCCP, and CT reside in separate subunits, and ACC is a multisubunit enzyme. In contrast, ACCs from most eukaryotes are large, multidomain enzymes, with all three domains residing on a single polypeptide.

Two isoforms of ACC exist in mammals. ACC1 is a cytosolic enzyme that catalyzes the first and committed step in long-chain fatty acid biosynthesis. ACC2 is associated with the outer mitochondrial membrane, where its malonyl-CoA product is an inhibitor of carnitine palmitoyltransferase I (CPT-I), which controls the transport of long-chain acyl-CoAs into the mitochondria for oxidation (3, 4). Mice deficient in ACC2 have higher rates of fatty acid oxidation and reduced body fat (5), implicating ACC2 as a drug discovery target against obesity, type 2 diabetes, and other diseases associated with the metabolic syndrome (6, 7).

Plant ACCs are found in both the plastids, for fatty acid biosynthesis, and the cytosol, for synthesis of very long-chain fatty acids and flavonoids. The cytosolic ACCs exist in the multidomain form, while the plastid ACCs exist in either the multidomain (grasses) or the multisubunit (dicots) form. Three classes of commercially available herbicides target the CT activity of the multidomain form of plastid ACCs (Fig. 1A). The aryloxyphenoxy

propionates (APPs or FOPs, represented by haloxyfop in Fig. 1A) and the cyclohexandiones (CHDs or DIMs, represented by tepraloxymid in Fig. 1A) have been widely used in the field since the late 1970s. The phenylpyrazolines (DENs) are relatively new, with pinoxaden being introduced in 2006 (8, 9).

We have reported the crystal structures of the CT domain of yeast ACC and its complex with CoA (10), haloxyfop (11), and tepraloxymid (12). We have also reported the structure of the CT domain in complex with CP-640186 (13), a structurally distinct class of compounds that were identified as potent inhibitors of mammalian ACCs (14). The inhibitors occupy different regions of the CT active site, located at the interface of a dimer of this domain, and block catalysis by competing with the acetyl-CoA (haloxyfop and tepraloxymid) or the carboxybiotin (CP-640186) substrate. A large conformational change in the dimer interface and the active site is required for haloxyfop binding, while a much smaller change is observed for the tepraloxymid complex. Despite their chemical diversity (Fig. 1A), haloxyfop and tepraloxymid share two common points of contact (or anchoring points) with the CT domain (12), which are crucial for the inhibitory activity of these two classes of herbicides.

We report here the crystal structure at 2.8-Å resolution of the CT domain of yeast ACC in complex with pinoxaden, the third class of commercial herbicides. Even though pinoxaden is chemically rather distinct from the other two herbicides (Fig. 1A), it has a similar binding mode as tepraloxymid and requires a small conformational change in the dimer interface for binding. Crystal structures of the CT domain in complex with all three classes of herbicides are now available. They confirm the importance of the two anchoring points for herbicide binding. They also provide a foundation for understanding the molecular basis of the herbicide resistance mutations and cross resistance among the herbicides, as well as for the design and development of new ACC inhibitors.

## Results and Discussion

**The Overall Structure.** The crystal structure of the CT domain of yeast ACC in complex with pinoxaden (Fig. 1A) has been determined at 2.8-Å resolution (Table 1 and Fig. 1B). The current atomic model contains residues 1480–2051 and 2080–2195 in chain A, 1480–2047 and 2081–2192 in chain B, 1491–2051 and 2080–2193 in chain C, for the three CT domain molecules as well as three pinoxaden molecules in the crystallographic asymmetric unit. The atomic model has good agreement with the X-ray diffraction data and the expected values for geometric parameters (Table 1). The majority of the residues (88%) lie in the most

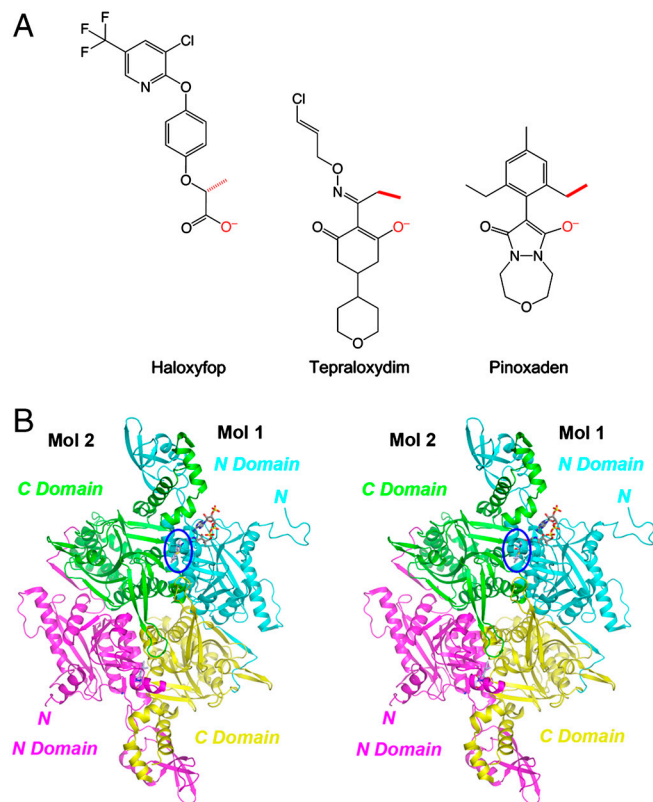
Author contributions: L.P.C.Y., Y.S.K., and L.T. designed research; L.P.C.Y., Y.S.K., and L.T. performed research; L.P.C.Y., Y.S.K., and L.T. analyzed data; and L.P.C.Y. and L.T. wrote the paper.

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Data deposition: The atomic coordinates and structure factors have been deposited in the Protein Data Bank, [www.pdb.org](http://www.pdb.org) (PDB ID code 3PGQ).

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**Fig. 1.** Crystal structure of the CT domain in complex with pinoxaden. (A) Chemical structures of the herbicides haloxyfop, tepraloxymid, and pinoxaden. The two anchoring points for interactions with the CT domain (see text for details) are colored in red. (B) Schematic drawing of the structure of yeast CT domain dimer in complex with pinoxaden. The N domains of the two monomers are colored in cyan and magenta, while the C domains are colored in yellow and green. The inhibitor is shown in stick models, in slate blue for carbon atoms, and one inhibitor molecule is highlighted by the blue oval. The CoA molecule is shown for reference in gray (13). The structure figures were produced with PyMOL ([www.pymol.org](http://www.pymol.org)).

favored region of the Ramachandran plot. Only residue Gln1744 of all three monomers is in the disallowed region, as is the case for all other structures of this domain.

The three CT domain monomers in the asymmetric unit have essentially the same conformation, with rms distance of about 0.5 Å for their equivalent C $\alpha$  atoms. The conformations of the inhibitors in the three CT domain molecules are highly similar to each other as well.

**Binding Mode of Pinoxaden.** The crystal structure of the pinoxaden complex was obtained by soaking free enzyme crystals of yeast CT domain with the inhibitor at 1-mM concentration for 80 min. Longer soaking times and/or higher inhibitor concentrations

**Table 1. Summary of crystallographic information**

Resolution range for refinement (Å)	30–2.8 (2.9–2.8) *
No. of observations	495,940
$R_{\text{merge}}$ (%)	9.9 (43.8)
Redundancy	4.7 (4.4)
$I/\sigma$	9.8 (2.1)
No. of reflections	106,428
Completeness (%)	99 (99)
$R$ factor (%)	19.7 (31.3)
Free $R$ factor (%)	23.9 (37.3)
rms deviation in bond lengths (Å)	0.012
rms deviation in bond angles (°)	1.3

\*The numbers in parentheses are for the highest resolution shell.

(up to 2.5 mM, due to solubility limits) invariably led to significant reduction in the diffraction quality of the crystals. Clear electron density for the inhibitor molecules was observed based on the crystallographic data (Fig. 2A). The average temperature factor values of the three inhibitor molecules are 70, 74, and 94 Å<sup>2</sup>, while that for all three protein molecules is 33 Å<sup>2</sup>. This suggests that the occupancy of the inhibitor may not be 100% in the crystal, especially for chain C. The lower occupancy is also consistent with our kinetic studies, which showed only minor inhibition of the CT domain of yeast ACC by pinoxaden at 1-mM concentration. In comparison, the  $K_i$  of haloxyfop for this CT domain is about 0.5 mM (10).

Pinoxaden is bound in the active site of the CT domain at its dimer interface (Fig. 2B). The central pyrazoline ring is situated between residues Gly1734 and Val2024' (primed residue numbers indicate the other monomer). One oxygen atom on this ring is hydrogen-bonded to the main-chain amides of Ala1627 (2.9 Å) and Ile1735 (3.0 Å) (Fig. 2C). The other oxygen is hydrogen-bonded to the main-chain amide of Gly1998' (2.8 Å). The electron density shows that this ring is planar (Fig. 2A), indicating that one of the carbonyl groups is in the enol form (Fig. 1A). It is likely that this enol, in its ionized state (enolate), is the one hydrogen-bonded to the main-chain amides of Ala1627 and Ile1735, because an oxyanion in haloxyfop (11) and tepraloxymid (12) is also positioned here (see below).

The 4-methyl-(2,6)-diethyl phenyl ring lies between the amide bonds of Gly1734-Ile1735 and Gly1997'-Gly1998', showing  $\pi$ - $\pi$  interactions (Fig. 2B). One of the ethyl substituents points in the direction of a hydrophobic pocket that is composed of the side chains from Ala1627, Leu1705, Ser1708, Val2001', and Val2024', with its closest neighbor being the side-chain CD2 atom of Leu1705. The other ethyl group is positioned near the side chains of Tyr1738, Leu1756, and Phe1956'. The methyl group is located deepest in the dimer interface, surrounded by residues Ile1735, Tyr1738, Val2001', and Val2002'.

The oxadiazepane ring is mostly solvent exposed and has few strong interactions with the enzyme. This ring, especially its oxygen atom, is likely to be flexible, because it has weaker electron density compared to the rest of the inhibitor (Fig. 2A). This region of the binding site is unlikely to be covered by the BC domain in the full-length ACC enzyme, based on observations on the structure of the propionyl-CoA carboxylase holoenzyme (15).

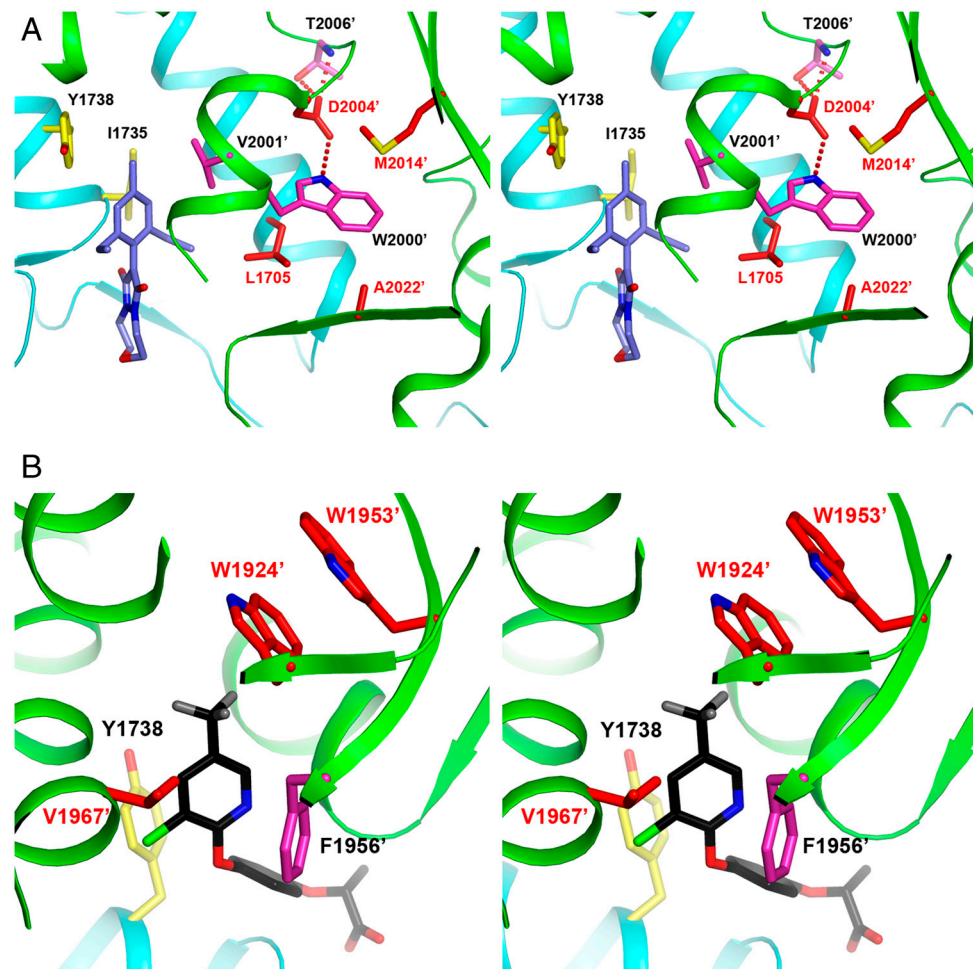
The observed binding mode of pinoxaden is consistent with the structure-activity relationships of this series of compounds (9). For example, replacing the 2,6-diethyl substituents with dimethyl groups or removing the 4-methyl group on the phenyl ring both led to large losses in activity. The presence of an enol in this compound is also supported by the fact that pinoxaden is actually marketed as a pivaloyl ester prodrug with one of the oxygen atoms on the pyrazoline ring (8, 9).

**Pinoxaden Has a Similar Binding Mode as Tepraloxymid.** The binding mode of pinoxaden is similar to that of tepraloxymid (12) (Fig. 3A), even though the two compounds are rather distinct from each other chemically (Fig. 1A). The pyrazoline ring of pinoxaden is comparable to the central cyclohexanedione ring of tepraloxymid. The two oxygen atoms on the cyclic structures in both compounds are tethered to the enzyme in exactly the same way, with the most important interactions being those between the main-chain amides of Ala1627 and Ile1735 with one of the oxygen atoms, located within 0.5 Å of each other in pinoxaden and tepraloxymid (Fig. 3A). The position of this oxygen atom is also occupied by one of the carboxylate oxygen atoms of haloxyfop (Fig. 3B) (12), suggesting that it is an oxyanion in all three inhibitors (enolate in tepraloxymid and pinoxaden, Fig. 1A). In fact, a hydroxyl group at this position is unlikely to be able to maintain hydrogen bonds to both main-chain amides.









**Fig. 4.** Molecular basis for herbicide resistance mutations. (A) Stereographic drawing showing the structure of yeast CT domain near the mutation sites I1781L (L1705 in yeast ACC), D2078G (D2004'), C2088R (M2014'), and G2096A (A2022'). The bound position of pinoxaden is also shown. (B) Stereographic drawing showing the structure of yeast CT domain near the mutation sites W1999C (W1924' in yeast ACC), W2027C (W1953'), and I2041N (V1967'). The bound position of haloxyfop is also shown.

fore, it is likely that the C2088R mutation will also confer resistance to pinoxaden. In addition, another resistance mutation, G2096A (A2022' in yeast CT), is also located near the Trp2000' side chain (Fig. 4A).

Therefore, four of the resistance mutations, I1781L (Leu1705 in yeast ACC), D2078G (Asp2004'), C2088R (Met2014'), and G2096A (A2022'), are clustered close to each other in the structure, distributed around the side chain of the strictly conserved Trp2000' residue (Fig. 4A). Three of the mutations in this cluster, I1781L, D2078G, and C2088R, confer resistance to all three classes of herbicides, and Ile1781 (Leu1705) has direct contacts with these compounds. The other three resistance mutations, W1999C (W1924'), W2027C (W1953'), and I2041N (V1967'), are clustered near the first aryl ring of haloxyfop (11) (Fig. 4B). This is a unique feature of that class of inhibitors (Fig. 1A), which may explain why these mutations primarily confer resistance to the FOPs while having only small effects on the DIMs. A recent computational study on the W2027C, I2041N, D2078G, and G2096A mutations suggests that they may introduce large conformational changes in the binding pocket, which may also contribute to the reduced sensitivity of these mutants to the herbicides (24).

Among the seven mutation sites that are known to confer resistance to herbicides, yeast ACC has a different residue at four of them compared to plant ACC—Leu1705, Val1967, Met2014, and Ala2022—and therefore behaves mostly like a resistant ACC. We have used the CT domain of yeast ACC as a model system to

define the binding modes of the three classes of herbicides by crystallographic analysis because we have so far not been able to obtain crystals of the CT domain of a sensitive ACC. It may be expected, however, that the overall binding modes are likely similar between the two types of ACCs.

Our attempts at introducing mutations into yeast CT to make it more sensitive to herbicides have so far not been successful. Our earlier studies showed that the L1705I/V1967I double mutant, which converts two of the four residues to their equivalents in plant ACCs, was not more sensitive to haloxyfop (11). We produced the L1705I/V1967I/M2014C/A2022G quadruple mutant of the yeast CT domain but found that it was not more sensitive either. Like the L1705I/V1967I double mutant (11), the quadruple mutant also had much lower catalytic activity compared to the wild-type enzyme. Compensatory mutations in other regions of the CT domain may be important for restoring the catalytic activity of the mutants. It is likely that additional differences between the CT domains of plant and yeast ACCs are also determinants of herbicide sensitivity, and some of these positions may be identified as sites of resistance mutations in the future.

#### Materials and Methods

**Protein production and crystallization.** The CT domain (residues 1476–2233) of *Saccharomyces cerevisiae* ACC was expressed and purified according to protocols described previously (11). Free enzyme at 10 mg/ml concentration was crystallized using the hanging-drop method at 4 °C. The reservoir solution contained 0.1 M sodium citrate (pH 5.5), 9% (w/v) PEG8000, 0.2 M NaCl,

and 10% (v/v) glycerol. The pinoxaden complex was obtained by soaking the free enzyme crystal with 1 mM of the compound for 80 min. Cryoprotection was achieved by the addition of 25% (v/v) glycerol and the crystal was then flash frozen in liquid nitrogen for data collection at 100 K.

**Data Collection and Structure Determination.** X-ray diffraction data were collected at the X29A beamline of the National Synchrotron Light Source (NSLS). The diffraction images were processed with the HKL package (25). The crystal belongs to space group *C2*, with unit cell parameters of  $a = 247.2 \text{ \AA}$ ,  $b = 123.4 \text{ \AA}$ ,  $c = 145.7 \text{ \AA}$ , and  $\beta = 94.3^\circ$ . There are three CT molecules in the asymmetric unit, forming a noncrystallographic dimer and a crystallo-

graphic dimer. The structure refinement was carried out with the programs CNS (26) and Refmac (27). The atomic model was built with the programs O (28) and Coot (29). The crystallographic information is summarized in Table 1.

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