Stimulation of Active and Passive Sodium Absorption by Sugars in the Human Jejunum

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ABSTRACT The effects of glucose and fructose on water and sodium absorption in the human jejunum were compared to assess the relative contribution of active and passive sugar stimulation of sodium transport. The effect of fructose is assumed to be entirely passive, and the difference between the effects of fructose and glucose is assumed to be a measure of sugarstimulated, active sodium absorption. Water and sodium movement with mannitol was the base line. Three sets of test solutions with differing sugar concentrations were studied. Fructose stimulated 66-100% as much net sodium and water absorption as glucose. Fructose stimulated potassium absorption, whereas glucose stimulated potassium secretion. Urea absorption was stimulated by both sugars. Glucose and fructose stimulated sodium absorption when chloride was the major anion, but they had relatively little effect on net sodium movement when chloride was replaced by bicarbonate or sulfate.

It is concluded that glucose stimulates passive and active sodium transport in the human jejunum. Stimulated active sodium absorption generates an electrical potential across the mucosa that causes sodium (and potassium) secretion and partly or completely nullifies the effect of active sodium transport on net sodium movement. Net sodium absorption stimulated by glucose is mainly (66-100%) the passive consequence of solvent flow. The accompanying anion determines the degree to which sugars stimulate sodium absorption $(Cl > SO_4 > HCO_8)$. The effects of bicarbonate and sugars on jejunal sodium absorption are not additive.

INTRODUCTION

Two mechanisms have been elucidated to explain glucose-stimulated sodium absorption in the intestine. The

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first, derived mainly from short-circuited in vitro experiments (1-4), is that glucose and sodium interact with a common carrier in the brush border membrane; glucose stimulates the attachment of sodium to the carrier and thereby increases sodium entry into the cell. The sodium is removed from the cell by an active pump located on the lateral membrane. Enhanced sodium pumping increases net sodium absorption and generates a potential difference (serosal positive) across the mucosa (or increases the short-circuit current if the preparation is short-circuited). The second mechanism for glucose-stimulated sodium absorption, derived primarily from experiments carried out in vivo (5), is that glucose absorption stimulates water absorption and that water movement across the mucosa stimulates the passive absorption of sodium by solvent drag.

These two postulated mechanisms are not mutually exclusive, and in all likelihood both are correct. It is important, however, to establish the relative contribution that each plays in the stimulation of net sodium absorption by glucose in the human intestine. The problem is basic to an understanding of how the small intestine absorbs the 120–140 meq of sodium delivered to it after each meal (6) and how oral solutions containing glucose reduce water and sodium losses in severe diarrheal diseases such as cholera (7).

In the present paper the carrier interaction and solvent drag components of sugar-stimulated sodium absorption have been evaluated by comparing the effects of mannitol, fructose, and glucose on jejunal absorption of water, sodium, potassium, and urea. Mannitol is a nonabsorbable solute, and water and electrolyte movement with mannitol forms a base line for quantitating the effects of fructose and glucose. Urea and potassium are markers for passive movement of noncharged and charged solutes, respectively (5, 8). Fructose is rapidly absorbed by carrier-mediated diffusion. However, fructose does not stimulate active sodium absorption,

and the fructose carrier does not react with sodium (9, 10). Fructose-stimulated absorption of sodium is therefore assumed to be passive, secondary to solvent drag. The difference between the effects of glucose and fructose is a measure of the active carrier interaction component of glucose-stimulated absorption. This assumption implies that the products of glucose and fructose metabolism within the intestinal mucosa have the same osmotic effectiveness, so that the absorption of 1 mol of glucose and fructose would result in the same solvent drag. Data of Holdsworth and Dawson (11) and Cook (12) suggest that the error resulting from metabolic degradation would be negligible.

To determine the role of the anion that accompanies sodium, the effects of glucose and fructose were measured with sodium chloride, sodium bicarbonate, and sodium sulfate. These experiments were prompted by the previous observation that substitution of bicarbonate for chloride enhances sodium absorption (5, 13) and by the desire to learn if the stimulatory effects of sugars and bicarbonate on sodium absorption are additive.

All of these experiments were carried out in the jejunum, since in man, glucose and bicarbonate increase net sodium absorption only in the proximal small intestine (5). In the human ileum glucose increases the potential difference across the mucosa but has no effect on net sodium absorption (5, 14–16).

METHODS

Normal subjects were studied with the triple-lumen perfusion system as previously described (5). Polyethylene glycol was the nonabsorbable volume marker. The infusion aperture was at the ligament of Treitz, and the proximal and distal aspiration sites were located 10 and 30 cm distally. The different test solutions were infused at appropriate rates (between 10 and 15 ml/min) so that the mean flow rate and solute concentrations would be equal in the test segment, regardless of the rate of net water absorption or secretion with the various sugars and electrolytes (5). A 30-min equilibration period preceded the collection of samples; each study period lasted 1 h. Collections from the proximal aspirating site began and ended 10 min earlier than collections from the distal site, in accord with the mean transit time of fluid through a 20-cm segment of proximal jejunum. Two to four test solutions were studied in each subject, depending on the number of variables in a particular experiment. The order of infusion of different test solutions was randomized.

Samples were analyzed for polyethylene glycol, electrolytes, glucose, fructose, mannitol, and urea by methods that have been previously described (5, 13, 14, 16). Absorption rates were calculated from the perfusion rate, the change in concentration of the nonabsorbable marker, and the change in concentration of the test substance. Results are expressed as mean \pm SE and were analyzed statistically by means of paired t test. Solute and water movements have dimensions of milliliters, milliequivalents, microequivalents, millimoles, or micromoles per hour per 20 cm of jejunum. For convenience, the intestinal segment length will be omitted when results are given in the text.

RESULTS

Effect of mannitol, glucose, and fructose on water and sodium (chloride) movement

130 mM glucose and fructose. Three test solutions containing 80 mM sodium chloride and either 130 mM glucose, fructose, or mannitol were perfused in nine subjects, and the results are shown in Table I, part A. Absorption rates of the sugars were 0.4 mmol/h for mannitol, 35 mmol/h for glucose, and 19 mmol/h for fructose. With the solution containing mannitol, sodium was secreted at a rate of 7.4 meg/h. With the glucose solution, sodium was absorbed at a rate of 4.2 meq/h, whereas with the fructose solution, sodium was secreted at a rate of 0.9 meg/h. The difference in the rate of sodium movement with glucose and mannitol and between fructose and mannitol is equal to the quantity of sodium absorption stimulated by these two sugars, designated ^{\Delta}Na in Table I. Stimulated sodium absorption was 11.6 meq/h with glucose and 6.5 meq/h with fructose. Dividing these values by the quantity of sugar absorbed gives the stimulated sodium absorption rate per millimole of sugar absorbed: 0.32 meg/mmol for glucose and 0.35 meq/mmol for fructose. By a similar series of calculations, stimulated water absorption rates were 5.5 and 5.7 ml/mmol of glucose and fructose, respectively. These results are depicted diagrammatically in Fig. 1. It is concluded that at these concentrations glucose and fructose stimulate water and sodium absorption equally per millimole of sugar absorbed.

65 mM glucose and 130 mM fructose. These test solutions were designed so that glucose and fructose absorption rates would be approximately equal. Each of the three test solutions (mannitol, glucose, and fructose) contained 80 mM sodium chloride, and the glucose solution also contained 65 mM mannitol to achieve isotonicity.

The results are shown in Table I, part B. Glucose and fructose were absorbed at approximately equal rates. The stimulated rates of water and sodium absorption per millimole of sugar absorbed were about 30% lower with fructose than with glucose (Fig. 1 and Table I, part B), and the differences were statistically significant.

30 mM glucose and 40 mM fructose. To study the effect of lower concentrations of sugars on water and sodium absorption, test solutions contained 120 mM NaCl and either 40 mM mannitol, 40 mM fructose, or 30 mM glucose plus 10 mM mannitol. Table I, part C shows that the absorption rates of glucose and fructose were similar at about 7 mmol/h. The stimulated water absorption rate was 13.2 ml/mmol with glucose and 13.3 ml/mmol with fructose (P > 0.4). The stimulated sodium absorption rates were 1.4 meg/mmol with

TABLE I

Effect of Glucose and Fructose on Sodium and Water Movement

Test solution	[S] Mean	Net S movement	[Na] Mean	Net Na movement	ΔNa‡		Net H ₂ O movement	$\Delta \mathrm{H}_2\mathrm{O}$ ‡	
	mM	mmol/h/20 cm	meq/liter	meq/h/20 cm	meq/h/20 cm	meq/mmol S	ml/h/20 cm	ml/h/20 cm	ml/mmol S
A) Mannitol									
130 mM	98 ± 8	-0.4 ± 2	92 ± 2	$+7.4 \pm 1$			$+48 \pm 13$		_
Glucose									
130 mM	87 ± 4	-35 ± 3	88 ± 3	-4.2 ± 1	-11.6 ± 2	-0.32 ± 0.08	-143 ± 20	-191 ± 20	-5.5 ± 0.6
Fructose									
130 mM	84 ± 5	-19±1*	$93 \pm 2*$	$+0.9 \pm 1*$	$-6.5 \pm 4*$	-0.35 ± 0.08	$-68 \pm 6*$	$-116 \pm 13*$	-5.7 ± 0.6
B) Mannitol									
130 mM			97 ± 1	$+8.3 \pm 1$			$+50 \pm 9$		
Glucose									
65 mM	24 ± 2	-24 ± 2	88 ± 1	-2.0 ± 1	-10.3 ± 1	-0.44 ± 0.06	-97 ± 9	-147 ± 10	-7.3 ± 0.5
Fructose									
130 mM	83±2*	-21 ± 1	95 ± 1*	$+1.9 \pm 1*$	$-6.4 \pm 1*$	$-0.29\pm0.06*$	$-61 \pm 8*$	$-111 \pm 11*$	-5.4 ± 0.6
C) Mannitol									
40 mM			122 ± 1	$+0.8 \pm 1$			$+4 \pm 11$		
Glucose									
30 mM	6.9 ± 1	-6.7 ± 1	128 ± 1	-9.0 ± 2	-9.8 ± 2	-1.4 ± 0.2	-88 ± 18	-92 ± 18	-13.2 ± 2
Fructose									
40 mM	23.3 ± 2	-7.1 ± 1	128 ± 1	-8.1 ± 2	-8.9 ± 2	-1.3 ± 0.2	-88 ± 19	-92 ± 18	-13.3 ± 2

Number of studies: (A) 9, (B) 16, and (C) 9. If sugar concentrations in test solutions not equal, the difference was made up by mannitol.

In studies B and C mannitol movement assumed to be zero. Sodium concentrations are the mean of concentrations in fluid from proximal and distal ends of the 20-cm test segment.

glucose and 1.3 meq/mmol with fructose (P > 0.4). As shown in Fig. 1, these values were much higher than was noted in the previous two experiments (Table I, parts A and B), and the difference is due to the higher sodium concentration in the perfusion solutions.

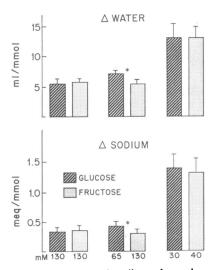


Figure 1 Rate of water and sodium absorption stimulated by the absorption of 1 mmol of glucose and fructose. Water and sodium movement with mannitol was the base line. The concentrations of sugars for each of three sets of test solutions are shown at the bottom of the figure. The asterisk indicates P < 0.05 by paired t test.

Effect of glucose and fructose on chloride movement

This comparison was made with the test solutions containing 30 mM glucose or 40 mM fructose, and 120 mM sodium chloride, and the results are shown in Table II. Glucose and fructose stimulated chloride absorption to approximately the same extent and in proportion to the stimulated water and sodium absorption rates (compare with Table I, part C). None of the differences between fructose and glucose were statistically significant by paired t test.

Chloride was also analyzed in 6 of the 16 studies reported in Table I, part B and the effects of glucose and fructose on chloride movement were similar to their influence on sodium movement, i.e., glucose stimulated more chloride absorption per millimole of sugar absorbed than did fructose.

Effect of glucose and fructose on urea and potassium movement

This comparison was made with the 65 mM glucose and 130 mM fructose solutions, which resulted in approximately equal rates of sugar absorption and showed that glucose was more effective than fructose in stimulating sodium and water absorption. Each of the three test solutions (mannitol, glucose, and fructose) contained urea and potassium (as KCl) in amounts cal-

S, sugar; -, absorption; +, secretion.

^{* =} Difference between glucose and fructose solution significant at P < 0.05 by paired t test.

[‡] Values in these columns were calculated by averaging the results in individual subjects. Hence, ΔNa and ΔH₂O may not agree exactly with the ratio of averages of the same data.

TABLE II

Effect of Glucose and Fructose on Chloride Movement (Nine Studies)

Test solution	Net sugar movement	[Cl] Mean	Net Cl movement	Δ	CI
•	mmol/h/20 cm	meq/liter	meq/h/20 cm	meq/h/20 cm	meq/mmol
Mannitol 40 mM	-	118±1.2	$+1.0\pm1.4$		
Glucose 30 mM*	-6.7 ± 0.9	123±0.8	-9.2 ± 1.9	-10.2 ± 2.2	-1.5 ± 0.3
Fructose 40 mM	-7.1 ± 0.9	122±1.9	-7.9 ± 1.4	-8.9 ± 1.4	-1.3 ± 0.2

^{*} Plus 10 mM mannitol to achieve isotonicity. Mannitol absorption assumed to be zero.

culated to achieve mean concentrations in the 20-cm test segment approximately equal to their concentrations in plasma. As shown in Fig. 2, glucose and fructose both stimulated the absorption of water, sodium and urea, with glucose having a 20–30% greater effect than fructose on the absorption of each of these substances. In contrast, glucose and fructose had opposite effects on potassium movement; glucose-stimulated potassium secretion, and fructose-stimulated potassium with mannitol, glucose, and fructose are shown in detail in Table III. Potassium concentrations in the test solutions were varied to compensate for different rates and directions of water movement with mannitol, glucose, and fructose (5), hoping to achieve approximately

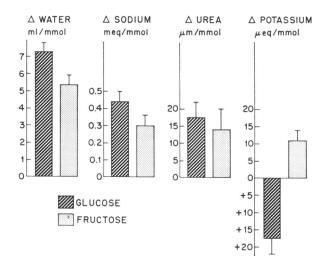


FIGURE 2 Effect of absorption of 1 mmol of glucose and fructose on the net movement of water, sodium, urea, and potassium. Water and solute movement with mannitol was the base line. The plus signs indicate net secretion (accumulation in the lumen).

equal mean potassium concentration in the 20-cm test segment. With mannitol and fructose, potassium concentration fell progressively, whereas with glucose, it increased progressively between the infusion site and the end of the study segment. The potential difference (PD) across the mucosa was estimated from a comparison of luminal potassium concentration in the distal sample and serum potassium concentration by means of the Nernst equation, assuming that potassium movement is passive (8) and had reached equilibrium in accord with electrochemical gradients by the time the perfusion solution reached the distal sampling site (17). Calculated in this manner, PD was 2.2 mV (mucosal side positive) with mannitol and fructose and 5.3 mV (mucosal side negative) with glucose. According to these calculations, glucose produces a 7.5-mV change in PD across the jejunal mucosa. This value is in good agreement with previous measurements of the effect of glucose on PD measured with an electrode (5, 13). Fructose, by contrast, had no effect on calculated PD, even though fructose stimulates almost as much water

TABLE III

Potassium Concentrations and Calculated PD from
the Nernst Equation*

T	Pota			
Test solutions	Infusion	Proximal	Distal	PD
Mannitol		meq/liter		m V
130 mM	5.17 ± 0.04	4.65 ± 0.15	4.11 ± 0.08	$+2.2 \pm 1.2$
Glucose 65 mM	3.80±0.08	4.84 ± 0.17	5.41 ±0.18	-5.3 ± 0.8
Fructose 130 mM	4.45 ± 0.04	4.32±0.11	4.11±0.15	+2.2±1.0

^{*}PD was calculated in individual subjects by the following equation: PD = $61.5 \times log serum [K]/lumen [K]$. A minus sign means the lumen is negative to the serosa, and vice versa.

^{-,} net absorption; +, net secretion.

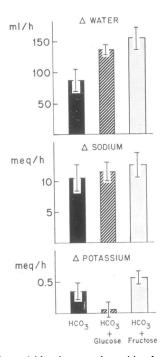


FIGURE 3 Effect of bicarbonate alone, bicarbonate plus glucose, and bicarbonate plus fructose on water, sodium, and potassium movement. The base-line solution contained mannitol and sulfate, and the bicarbonate solution contained mannitol.

and sodium absorption as glucose (Fig. 2). Qualitatively similar results were noted with the other pairs of sugar concentrations that were studied and reported in Table I.

Effect of bicarbonate, bicarbonate plus glucose, and bicarbonate plus fructose

In all results presented so far, chloride was the only anion in the perfused solutions. The results showed that jejunal perfusion of 65 mM glucose and 130 mM fructose resulted in approximately equal rates of sugar absorption, and only with this set of test solutions was glucose more effective than fructose in stimulating water and sodium absorption. These sugar concentrations were selected for further study in experiments designed to assess the interaction of bicarbonate and sugars on water, sodium, and potassium movement.

Four test solutions were perfused in each of 10 subjects. One contained mannitol and sodium sulfate (this was the reference solution). The second contained 70 meq/liter of bicarbonate to replace the sulfate. The third and fourth contained 70 meq/liter of bicarbonate plus either 65 mM glucose or 130 mM fructose. All test solutions were isotonic to plasma and contained 80 meq/liter of sodium and 5 mM potassium chloride. The stimulatory effects of bicarbonate and bicarbonate

plus the sugars were calculated exactly as demonstrated in Table I, and the results are shown in Fig. 3. The addition of bicarbonate stimulated the absorption of 87 ml/h water, 10.4 meq/h of sodium, and 0.3 meq/h of potassium. (See Table IV for bicarbonate absorption rates with mannitol, glucose and fructose.) When, in addition to the absorption of bicarbonate, 17 mmol/h glucose or 21 mmol/h fructose were absorbed, water absorption was stimulated further, to 135 ml/h (P < 0.005) and 155 ml/h (P < 0.05), respectively. In contrast, absorption of glucose and fructose did not significantly enhance sodium absorption (above that noted with bicarbonate alone). Glucose depressed (P < 0.05) and fructose enhanced (P < 0.05) the stimulated potassium absorption that was noted with bicarbonate alone.

Effect of chloride, bicarbonate, and sulfate anions on glucose-stimulated water and sodium absorption

The data in the previous section demonstrated that glucose and fructose had no significant effect on sodium movement when bicarbonate was the major anion present in the perfusing solution. To evaluate further the importance of the anionic composition of test solutions on glucose-stimulated absorption, four sets of experiments were carried out. Each set consisted of a comparison of the effect of 65 mM glucose and mannitol on sodium and water absorption. In the first set, chloride was the only anion present. In the second, chloride and bicarbonate were equal at 40 meq/liter. In the third, bicarbonate was 70 and chloride was 10 meq/liter, and in the fourth set chloride and bicarbonate were completely replaced by sulfate. In each of the sets

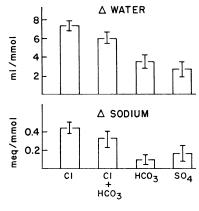


FIGURE 4 Effect of anion composition on net water and sodium absorption stimulated by the absorption of 1 mmol of glucose. Water and sodium movement with mannitol was the base line. Total anion concentration was 80 meq/liter with all solutions. The Cl + HCO₃ solution contained 40 meq/liter of each anion.

TABLE IV

Effect of Glucose and Fructose on Bicarbonate Movement

Test solutions*	Net S‡	[HCO3] Mean	Net HCO₃‡
	mmol/h/20 cm	meq/liter	meq/h/20 cm
(A) 70 meq/lite	r Bicarbonate, 1	i = 10	
Mannitol		29 ± 3	-9.0 ± 0.9
Glucose	-17 ± 1.3	44 ± 3	-10.6 ± 0.9
Fructose	-21 ± 2.5	44 ± 2	-11.1 ± 1.6
(B) 40 meq/lite	r Bicarbonate, 1	i = 13	
Mannitol	_	16 ± 1	-4.8 ± 0.8
Glucose	-18 ± 1	24 ± 1	-5.9 ± 0.9
Fructose	-21 ± 2	20 ± 1	-5.8 ± 0.9

^{*} Sodium concentration in A and B test solutions = 80 meg/liter.

Glucose concentration = 65 mM; fructose concentration = 130 mM.

of experiments the sodium concentration was 80 meq/liter, and all solutions had an osmolality of 290 mosmol/kg.

The results of these experiments are shown in Fig. 4. They demonstrate that glucose has progressively less effect on sodium and water absorption as the chloride concentration is reduced by substitution of either bicarbonate or sulfate.

Effect of glucose and fructose on bicarbonate absorption

As shown in Table IV, neither glucose nor fructose significantly enhanced bicarbonate absorption in spite of the fact that both sugars stimulated water absorption (Fig. 3). This suggests that solvent drag has no effect on bicarbonate movement in the human jejunum. Since bicarbonate absorption was the same in the presence of glucose and fructose and since these sugars have different effects on transmembrane PD (Table III), bicarbonate movement is apparently not affected by electrical gradients. However, bicarbonate absorption is influenced by concentration gradients, as illustrated by the fact that absorption rate was almost twice as rapid when luminal bicarbonate concentration was 29–44 meq/liter as when luminal concentration was 16–24 meq/liter (Table IV).

Absolute rates of sodium and water movement with various glucose-containing solutions

In previous sections of this paper the effect of sugars has been expressed in terms of a stimulated absorption rate relative to a control solution containing mannitol. This is necessary to show the effect of the sugars,

but the results may be misleading in certain situations, particularly if the mannitol base-line values are different with different solutions. For example, assume that glucose reduced water secretion rate from 100 to 50 ml/h with solution A and stimulated absorption from 100 to 150 ml/h with solution B. The glucose effect is 50 ml/h with both solutions, but this result fails to convey the fact that in the presence of glucose the jejunum was secreting with solution A and absorbing with solution B. Therefore, in Table V the results of various solutions containing glucose are shown in terms of absolute absorption or secretion rates. When the test solution contained 80 meg/liter sodium and 65 mM glucose, sodium was absorbed only if chloride was the predominant anion, although water was absorbed with the sulfate and bicarbonate-containing solutions. When sodium and chloride concentrations were kept at 80 meq/liter and glucose concentration increased to 130 mM, sodium absorption rate increased from 2 to 4 meg/h and water absorption increased from 97 to 143 ml/h. Highest sodium absorption rates occurred when the sodium and chloride concentrations were increased to 120 meg/liter and the glucose concentration was reduced to 30 mM.

DISCUSSION

Most intestinal preparations absorb water and this water flow theoretically could drag sodium chloride and other solutes with it. However, relatively few experiments have been conducted to evaluate the extent to which solvent drag actually does influence salt transport. Green, Seshadri, and Matty (18) found that net absorption and unidirectional flux of sodium was not reduced when water absorption in the proximal small intestine 1 of the rat in vitro was stopped by placing

TABLE V

Comparison of the Makeup Infusion Solution on Absolute
Rates of Sodium and Water Movement in
the Presence of Glucose

	Test solut	ion			
[Na]	[Glucose]	Major anion	Net Na*	Net H ₂ O*	
meq/liter	mM		meq/h/20 cm	ml/h/20 cm	
80	65	SO ₄	$+10 \pm 0.6$	-16 ± 7	
80	65	HCO ₃	+2+0.9	-64 + 7	
80	65	HCO3/CI	0±1.3	-72 + 12	
80	65	Cl	-2+1.0	-97 + 9	
80	130	CI	-4+1.6	-143 + 20	
120	30	Cl	-9 ± 1.8	-88±18	

^{* -,} absorption; +, secretion.

^{‡ -,} net absorption.

¹ Reference 18 reports studies exclusively in the proximal small intestine in spite of its title, which suggests that the experiments were carried out in the lower small bowel.

hypertonic mannitol in the mucosal solution. This experiment and others reported by Schultz and Zalusky (19) and Schultz, Zalusky, and Gass (20) in in vitro rabbit ileum were interpreted originally and in subsequent reviews (1, 21) as indicating that solvent drag is probably not a major mechanism of sodium or chloride absorption. Starting in 1965, a series of experiments by Fordtran, Rector, and associates (22, 23, 5, 13) were reported which suggested that solvent drag was a major mechanism for sodium chloride absorption in the proximal but not in the distal small bowel of normal human subjects. Our data suggested that the entire increment in sodium absorption in response to glucose could, in the human jejunum, be explained as a passive consequence of solvent flow (5, 24). Although it was specifically pointed out that active sodium transport might also play a role (reference 5, p. 898), criteria for active sodium absorption in the human jejunum could not be met.

Our conclusion that solvent drag was a major mechanism for sodium absorption in the human jejunum was questioned by Schultz and Curran (25, 26). While recognizing that our data were compatible with our solvent flow model, Schultz and Curran (26) pointed out that the results of Holdsworth and Dawson (27) and Sladen and Dawson (28) were in major disagreement with our results and were exceedingly difficult to reconcile with our solvent drag model. They favored the concept of a direct link between sodium, sugar, and a transport site in the brush border membrane as the mechanism by which glucose stimulates sodium absorption in the human jejunum. I agree that the experimental results of Holdsworth and Dawson and Sladen and Dawson (27, 28) are difficult to reconcile with data obtained previously in this laboratory.2 The only explanation I can offer is that the perfusion methods they used in these particular experiments may not reliably measure net water and sodium movement

More recently, Modigliani and Bernier (15) concluded that the effect of glucose on net sodium movement in the jejunum could be accounted for by both solvent drag and stimulation of active sodium absorption, and this conclusion was accepted by Schultz, Frizzell, and Nellans (4). However, the extent to which these two mechanisms actually contribute to the net enhancement of sodium and water absorption by

actively transported sugars is not known, and there is no way to guess from previous work whether active sodium transport (or solvent drag) accounts for 1, 50, or 99% of the observed response.

The major purpose of the present series of experiments was to estimate the relative importance of these two mechanisms by comparing water and sodium movements with mannitol, fructose, and glucose. The results conclusively demonstrate that fructose stimulates sodium absorption in the human jejunum, compared with the rate of sodium movement when fructose is replaced by a nonabsorbable solute (mannitol). Since previous studies have not suggested an interaction of fructose and sodium with a common carrier (9, 10) it seems likely that the effect of fructose is mediated passively by solvent flow. The fact that fructose stimulates urea and potassium absorption is further evidence in favor of a passive connection between fructose and sodium absorption. These results thus strengthen our previous suggestion that solvent flow stimulates sodium absorption in the human jejunum.

Since glucose and fructose are both absorbed by carrier-mediated processes located in the brush border membrane, it is reasonable to assume that glucose absorption stimulates the same amount of passive water and sodium absorption as fructose, per millimole of sugar absorbed (see Introduction). However, two pieces of evidence indicate that glucose also stimulates active sodium absorption in the human jejunum, First, when glucose and fructose absorption rates are 21-24 mmol/h (Table I, part B), glucose stimulates significantly more sodium chloride and water absorption than fructose.3 Assuming that the effect of fructose is passive and that glucose has the same passive consequence as fructose, it seems reasonable to conclude that under these experimental conditions each millimole of absorbed glucose stimulates the absorption by solvent drag of 0.29 meq of sodium and 5.4 ml of water (equal to the fructose effect). In addition, each millimole of absorbed glucose stimulated the absorption of 0.15 meg of sodium and 1.9 ml of water (the difference between the fructose and the glucose effects), and this component may be assumed to be secondary to stimulation of active sodium transport. The apparent sodium concentration in the active component (0.15 meq sodium and 1.9 ml water) of the stimulated absorbate is 79 meq/liter. This is fairly close to the sodium concentration in the luminal contents (88 meg/liter). For the passive component (0.29 meg sodium and 5.4 ml water), the concentration of sodium in the stimulated absorbate is 54 meg/liter. Since the luminal sodium concentration in

² The results of Holdsworth and Dawson (27) also disagree with data obtained in the present paper. For example, they found that with 2.5% solutions (140 mM) of glucose and fructose, water absorption was 2.5 times higher with glucose than with fructose per gram of sugar absorbed (their Table II); in contrast, with 130 mM sugar solutions we found water absorption was equal with glucose and fructose per millimole of sugar absorbed (Table I, this paper).

³ In this instance our results are in qualitative agreement with those of Holdsworth and Dawson (27) and Sladen and Dawson (28).

these studies was about 95 meq/liter, there was a moderate sieving out of sodium through the solvent flow pathway, consistent with previous estimates of the reflection coefficient for sodium chloride in the human jejunum (5, 22).

The second piece of evidence indicating that glucose stimulates active sodium absorption comes from the study of the effect of these sugars on potassium movement. Although glucose and fructose have qualitatively similar effects on water, sodium, and urea movement, these sugars have opposite effects on potassium (Fig. 2). Since potassium movement is generally assumed to be passive (8), and since the opposite effects of glucose and fructose cannot be explained by differences in luminal potassium concentration (Table III), glucose stimulation of potassium secretion suggests that glucose absorption generates a lumen-negative PD across the jejunal mucosa. The fact that fructose stimulates almost as much water and sodium absorption as glucose, but stimulates potassium absorption rather than secretion, indicates that solvent drag does not generate a similar PD change. In fact, fructosestimulated water and sodium absorption does not cause the PD to change at all according to estimates based on the Nernst equation and the potassium concentration in fluid aspirated from the distal end of a 20-cm distal segment (Table III).

Therefore, glucose stimulates both active and passive sodium transport in the jejunum, and the relative importance of these two mechanisms in the overall net enhancement of sodium absorption apparently depends on the sugar and sodium concentrations in the jejunal lumen. For example, when a 130 mM sugar and 80 mM NaCl solution was perfused, glucose and fructose stimulated sodium and water absorption equally per millimole of sugar absorbed, suggesting that the entire enhancement of net sodium absorption was passive, secondary to solvent drag. A similar conclusion was reached with 30 mM sugar and 120 mM NaCl solution (Fig. 1). These results indicate that glucose stimulation of active sodium absorption was manifested only by the generation of a PD across the mucosa, In other words, glucose stimulated active sodium absorption, and this generated a PD that caused sodium (and potassium) secretion; net sodium movement was not affected. A similar mechanism probably explains why glucose in the human ileum increases transmucosal PD (5, 14) and increases unidirectional sodium flux rates from lumen to plasma and from plasma to lumen (15) but has no effect on net sodium absorption (5, 15, 16).

In only one of our test solutions (Table I, part B) did glucose induce a greater stimulation of net sodium absorption than fructose per millimole of sugar absorbed and even here fructose was 66% as effective

as glucose (Fig. 1).⁴ I therefore conclude that solvent drag is quantitatively the most important mechanism by which glucose stimulates net sodium absorption in the human jejunum. Glucose stimulation of active sodium absorption is, for the most part, nullified by passive sodium excretion in response to the electrical gradients generated by active sodium pumping. Of course, if the intestine were studied under experimental conditions that short-circuited the PD, glucose stimulation of active sodium transport would contribute in a major way to the observed increase in net sodium absorption (32).

Another purpose of these experiments was to determine whether or not glucose and bicarbonate are additive in their stimulatory effects on net sodium absorption. With test solutions in which bicarbonate was virtually the only anion present in the luminal fluid, bicarbonate and glucose were found to be additive insofar as stimulation of water absorption is concerned, but net sodium absorption stimulated by bicarbonate was not further enhanced by the addition of glucose or fructose. These results suggest that the solvent flow pathway markedly restrict the movement of sodium bicarbonate. Further studies showed that the accompanying anion determines the degree to which sugars stimulate sodium absorption ($C1 > SO_4 > HCO_3$).

A final point of physiologic interest concerns the absorption of bicarbonate. It is known from previous work that bicarbonate absorption in the human jejunum is an active process (13), that bicarbonate movement is minimally affected by solvent drag (5), and that absorption rate increases in response to increasing luminal bicarbonate concentrations between 0 and 60 meq/liter (13). Although bicarbonate absorption is, therefore, sensitive to concentration gradients across the mucosa, the present studies show that bicarbonate absorption is not influenced by electrical gradients. This is most dramatically illustrated by the fact that glucose stimulates a PD (mucosal side negative) that induces potassium and sodium secretion but does not accelerate absorption of the bicarbonate anion.

From the practical standpoint of designing oral sugar solutions for therapy of diarrheal diseases, several conclusions seem justified from these results. First, glucose is preferable to fructose since under certain conditions glucose stimulates more absorption of sodium and water than fructose per millimole of sugar absorbed. Furthermore, glucose is more rapidly absorbed than fructose

⁴ The reason why glucose was more effective than fructose with one but not with two other test solutions is not entirely clear, but it indicates that the quantitative jejunal response to glucose in terms of enhancing sodium and water absorption is complicated and is determined by absolute rates of sugar absorption as well as by the concentration of sodium in the luminal contents.

at sugar concentrations likely to be employed in oral test solutions. Consequently, a given rate of sugar absorption can be obtained with a lower glucose concentration, leaving room to elevate the sodium concentration and still keep the solution isotonic with plasma. The disadvantage of glucose (compared to fructose) is that the electrical gradients it generates tends to stimulate potassium secretion, whereas fructose stimulates potassium absorption. Second, the sodium concentration in the jejunal lumen is of critical importance in determining the rate of sugar enhancement of sodium absorption. For example, when the luminal sodium concentration was 120 meq/liter, sodium absorption was more than twice as fast as when the luminal sodium concentration was 80 meq/liter (Table V). It would seem worthwhile to evaluate whether maltose or starch could partially replace glucose in oral test solutions; if so, higher sodium concentrations could be used without exceeding the lower limits of sugar requirement or the upper limits of tolerable hypertonicity. Third, if the major purpose of oral glucose solutions is to stimulate sodium absorption, chloride should be the only anion. Of course, some patients are severely acidotic, and bicarbonate may be needed to correct this acid-base abnormality. And finally, caution must be exercised in extrapolating these results with a 20-cm test segment to an intact patient who has approximately 120 cm of proximal small intestine capable of responding to glucose by enhancing its rate of net sodium absorption. For example, the fact that sodium absorption is more than twice as rapid with a 30 mM glucose and 120 meq/liter NaCl solution than with solutions containing 130 mM glucose and 80 meq/liter NaCl does not necessarily mean that qualitatively similar results would be obtained if absorption over a longer segment of intestine were studied.

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