

Direct observation of stepped proteolipid ring rotation in *E. coli* F_0F_1 -ATP synthase

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Although single-molecule experiments have provided mechanistic insight for several molecular motors, these approaches have proved difficult for membrane bound molecular motors like the FoF1-ATP synthase, in which proton transport across a membrane is used to synthesize ATP. Resolution of smaller steps in F_0 has been particularly hampered by signal-to-noise and time resolution. Here, we show the presence of a transient dwell between Fo subunits a and c by improving the time resolution to 10 µs at unprecedented S/N, and by using Escherichia coli F₀F₁ embedded in lipid bilayer nanodiscs. The transient dwell interaction requires 163 µs to form and 175 µs to dissociate, is independent of proton transport residues aR210 and cD61, and behaves as a leash that allows rotary motion of the c-ring to a limit of $\sim 36^{\circ}$ while engaged. This leash behaviour satisfies a requirement of a Brownian ratchet mechanism for the Fo motor where c-ring rotational diffusion is limited to 36°.

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Introduction

The F_0F_1 -ATP synthase is composed of two opposed rotary molecular motors connected by a common axle of $\gamma\epsilon$ -subunits (Stock *et al*, 1999). The integral membrane F_0 motor, which has a subunit stoichiometry of ab_2c_{10} in *Escherichia coli* (Jiang *et al*, 2001), uses the electrochemical potentialdriven flux of protons across a membrane (proton-motive force (PMF)) to drive clockwise rotation of the ring of 10 c-subunits as viewed from the periplasm (Börsch *et al*, 2002). The c-ring is docked to the $\gamma\epsilon$ -subunits that extend into the hexameric ring of α - and β -subunits in the F_1 peripheral membrane motor. Rotation of this axle drives conformational changes in each of the three catalytic $\alpha\beta$ heterodimers resulting in ATP synthesis (Boyer, 1997). The F_1 motor can also hydrolyze ATP resulting in counterclockwise $\gamma\epsilon$ -subunit rotation and proton translocation via F_0 (Börsch *et al*, 2002). When solubilized away from F_o and the membrane, *E. coli* F_1 -ATPase-driven rotation at saturating ATP concentrations occurs in three 120° power strokes (Sabbert *et al*, 1996; Noji *et al*, 1999; Spetzler *et al*, 2006), separated by 8.3 ms dwells comparable to the turnover time of the rate-limiting step of ATP hydrolysis (Spetzler *et al*, 2006; Hornung *et al*, 2008). In the absence of drag on the F_1 motor, the velocity of the power stroke is ~0.5° µs⁻¹ (Spetzler *et al*, 2006).

In vivo, F_oF_1 uses the PMF across the membrane to maintain the [ATP]/[ADP][Pi] ratio (Q) far from equilibrium so that the high-ATP concentration provides an energy source to drive other cellular processes. Energetically, this means that at steady state, cellular PMF \cong 2.3RTlogQ. In other words, the driving force of the F_o motor (PMF) is in equilibrium with the driving force of the F_1 motor (logQ). In *E. coli*, the cytoplasm typically contains 3 mM ATP, 0.4 mM ADP, and 6 mM Pi such that logQ \cong 0.1 (Weber and Senior, 1997).

The maximum reported rate of *E. coli* F_oF_1 ATP synthesis (Senior *et al*, 2002) is about 100 s⁻¹ (10 ms ATP⁻¹), although rates of 27 s⁻¹ (37 ms ATP⁻¹) are more common with *E. coli* F_oF_1 in proteolipisomes (Fischer *et al*, 1994). Proton translocation can occur at faster rates when powered by ATP hydrolysis (Feniouk and Junge, 2008), or by membrane potentials imposed on *E. coli* F_o -embedded membranes after removal of F_1 (Franklin *et al*, 2004; Wiedenmann *et al*, 2008). In the absence of F_0 saturates at high-driving force (Feniouk and Junge, 2008), which suggests that the proton translocation step is not rate limiting to the mechanism.

Proton translocation across the membrane occurs in F_o when subunit-a residue aR210 deprotonates the cD61 carboxyl on each c-subunit as the c-ring rotates (Fillingame *et al*, 1984; Lightowlers *et al*, 1987; Angevine *et al*, 2003; Ishmukhametov *et al*, 2008). A Brownian ratchet mechanism has been postulated to power F_o rotation that must meet two requirements to function (Junge *et al*, 1997; Oster *et al*, 2000): first, that there are two noncolinear proton access half-channels from each side of the membrane leading to the cD61 carboxyl; and second, that rotational diffusion of the c-ring relative to subunit-a is periodically restricted in some manner. However, experimental evidence that provides a molecular basis for the latter requirement is scarce.

Initial single-molecule c-ring rotation measurements of F_oF_1 driven by ATP hydrolysis, or by an electrochemical potential, resolved only 120° steps (Sambongi *et al*, 1999; Pänke *et al*, 2000; Börsch *et al*, 2002; Kaim *et al*, 2002; Nishio *et al*, 2002; Ueno *et al*, 2005). C-ring rotation with step sizes of 36°, 72°, 108°, and 144° occurring 48, 37, 12, and 3% of the time, respectively, have now been observed using *E. coli* F_oF_1 proteoliposomes that synthesized an ATP every 37 ms in response to a membrane potential of >200 mV (Düser *et al*, 2009). As the proton translocation rate of F_o does not saturate at high-driving force (Feniouk and Junge, 2008), the movement of c-subunits past subunit-a during the 72°, 108°, and 144° steps may not have caused a dwell in rotation.

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We now report the observation of a previously unknown interaction between F_o subunits a and c of F_oF₁ when ATPasedriven rotation is slowed by a viscosity-induced load. A striking feature of this interaction is that it forms a leash that limits rotation to $\sim 36^{\circ}$ in a manner that can satisfy the restricted motion requirement in the Fo Brownian ratchet mechanism. As the transient dwells do not form when c-ring rotation and proton transport occur at high rates, the practical advantage of using the leash is anticipated to be under steady-state conditions in which the cellular ATP concentration is high relative to ADP and Pi. Under these conditions, when the free energy of the proton gradient approaches equilibrium with the chemical potential of ATP, the Fo motor could use this leash as part of a Brownian ratchet to bias rotation for ATP synthesis (clockwise) against an F₁ motor-imposed load.

Results

F_oF_1 nanodiscs are fully assembled and retain complete activity during single-molecule measurements

To stabilize the hydrophobic F_0 complex, we inserted solubilized F_0F_1 into phospholipid bilayer nanodiscs. The particle size of nanodiscs is constrained by the membrane scaffold protein (MSP) construct MSP-1E3D1 that forms a 13-nm diameter ring of α -helices around a bilayer of phospholipid molecules, and has been shown to provide a good model for lipid bilayers (Bayburt et al, 2007). The nanodiscs are large enough to allow the incorporation of the Fo complex and a few hundred lipid molecules, yet are on the same scale as the F_0F_1 complex. Assembly of stable nanodisc- F_0F_1 complexes $(n-F_0F_1)$ from MSP, lipids, and detergent solubilized F_0F_1 was verified by 2D electrophoresis (Figure 1). The first nondenaturing gel dimension contained one prominent band. This band contained both MSP and the FoF1 subunits when separated in the second denaturing gel dimension. The absence of other bands in the nondenaturing gel corresponding to incomplete n-F_oF₁ constructs suggests that the majority of proteins contain the full complement of subunits.

We made the c2 ∇ C mutation to a cys-free *E. coli* F₀F₁ enzyme that inserted a sulfhydryl group on each c-subunit for covalent modification with biotin maleimide, which is designated here as F₀F₁. Figure 1C shows the ATPase activity of n- F_0F_1 versus detergent solubilized F_0F_1 as a function of time. Detergent solubilized FoF1 lost all activity and aggregated within a few hours at room temperature. In comparison, the activity of n-FoF1 was initially higher and did not decline significantly after the preparation had been at 25° for 8 h. Modification of cD61 in the c-ring of $n-F_0F_1$ by N,N'-dicyclohexylcarbodiimide (DCCD) inhibited ATPase activity by as much as 85% (Table I), indicating that there was strong coupling between hydrolysis and proton transport. This extent of inhibition is comparable to that reported by Ueno et al (2005) for detergent solubilized F_oF₁ used in single-molecule rotation studies. The rapid loss of activity of detergent solubilized enzyme may explain why Ueno et al (2005) observed rotation of only 60 detergent solubilized F₀F₁ molecules in 840 fields of view.

Biotinylated $n-F_0F_1$ was attached to a cover slip via 6xHistags on the β -subunit N-terminus. Subsequent addition of avidin-coated gold nanorods then became bound to the biotins positioned on the c-ring distal from F_1 (Figure 2A).

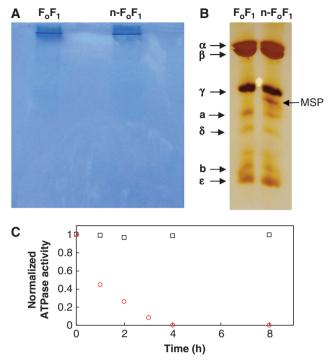


Figure 1 Incorporation of F_oF_1 into nanodiscs. Two-dimensional electrophoresis gel of purified F_oF_1 before (left lane) and after (right lane) incorporation into nanodiscs. (A) Dimension 1: Coomassiestained 5–15% nondenaturing gel. (B) Dimension 2: Silver-stained 15% denaturing gel separating proteins in the single-band excised from the nondenaturing gel. (C) ATPase activity of detergent solubilized F_oF_1 (\bigcirc) and $n-F_oF_1$ (\bigcirc) versus time at 25°C normalized to the initial activity of 110 and 145 s⁻¹, respectively.

Nanorods observed in Figure 2B were specifically bound to the c-ring of F_0F_1 on the microscope slide as $n-F_0F_1$ that lacked the $c2\nabla C$ mutation failed to bind nanorods (Figure 2C). The stability of ATPase activity of the $n-F_0F_1$ complex (Figure 1C) is important due to the time required to complete single-molecule experiments. The abundance of $n-F_0F_1$ observed to rotate was at least 25% of the molecules in an average field of view that typically contained about 250 molecules (Figure 2B), which was comparable to the abundance observed using purified F_1 -ATPase (York *et al*, 2007).

High-speed rotational power stroke measurement using gold nanorods

As shown in Figure 3, gold nanorod rotation results in a change in the intensity of red light scattered from the nanorod when viewed through a polarizing filter (Sönnichsen and Alivisatos, 2005; Spetzler et al, 2006). The intensity of red scattered light from a nanorod changes in a sinusoidal manner as a function of the rotary position of the nanorod relative to the plane of polarization with minimal and maximal intensities separated by 90° (Spetzler *et al*, 2006). Figure 3B shows the distribution of scattered red light intensities from a single-nanorod immobilized to the surface of a microscope slide as a function of the rotational position of the polarizing filter. At each position of the polarizer, the scattered light intensity was sampled 3520 times under conditions comparable to that used to measure rotation of n-F₀F₁ molecules. The sample number of 3520 was used because it corresponds to the average number of rotational power stroke

Strain	$k_{\text{cat}} \stackrel{\text{ATPase}}{(s^{-1})}$	k_{cat} ATPase + DCCD (s ⁻¹)	ATPase in SBP μ mol ATP min ⁻¹ mg protein ⁻¹	ATPase-dependent proton pumping (% of WT)
$n-F_0F_1$ (WT)	140	21	1.8	100
n-F _o F ₁ -cD62G ^a	130	130	0.8	0
n-FoF1-aR210G	20	20	1.1	0
$n-F_{o}F_{1}-a\nabla 14$	90	90	0.4	0

Table I Biochemical characterization of FoF1 mutants that lack transient dwells

^a cD62 is so named due to the $c2\nabla C$ insert mutant used for biotinylation.

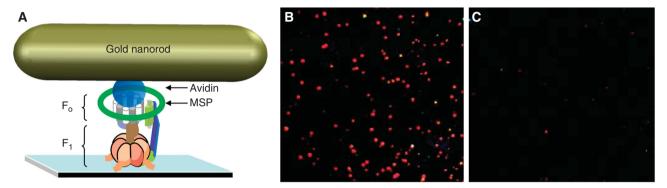


Figure 2 F_0F_1 nanodiscs $(n-F_0F_1)$ in single-molecule rotation studies. (**A**) Microscope slide bound $n-F_0F_1$ attached via β -subunit N-terminus $6 \times$ His tags attached to an avidin-coated 77×39 nm² nanorod via a biotinylated subunit-c cys. (**B**, **C**) Microscope fields-of-view of gold nanorods (red and green dots) bound to a slide coated with $n-F_0F_1$ in which subunit-c contained (**B**) or lacked (**C**) the cys insertion mutation.

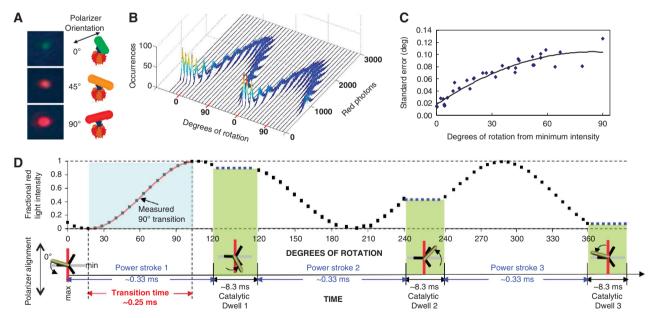


Figure 3 Use of nanorods to measure c-ring rotation of $n-F_0F_1$. (A) Micrographs of white light scattered from a single-gold nanorod viewed through a polarizing filter, and as a schematic showing the orientation of the nanorod to the polarizer. Measurement of rotation position is made only with the intensity of polarized red light scattered from the nanorod. (B) Histograms of the intensity of red light scattered from a single nonrotating nanorod fixed to a slide as a function of the rotational position of the polarizer. Each histogram contains 3520 measurements at each position of the polarizer obtained with the data acquisition speeds used to collect data points for c-ring rotation. The polarizer was then rotated counterclockwise by 10°, and data collection was repeated. (C) Standard error of nanorod rotational position versus degrees of rotation of the polarizer from the minimum intensity of light scattered from the nanorod as determined by Equation (1). (D) Relationship between a 120° power stroke and a 90° measured rotational transition. Theoretical plot of the intensity of scattered red light from a nanorod during one complete revolution that involves three consecutive power strokes and three consecutive catalytic dwells separated by exactly 120°. The nanorod is initially positioned almost, but not exactly perpendicular to the orientation of the polarizer such that the scattered light intensity goes through a minimum then a maximum prior to catalytic dwell 1. A transition includes the data between the minimum and each of the successive power strokes is exactly 120°, the algorithm selects transitions for power strokes 1 (min to max) and 3 (max to min).

events measured for each $n\mathchar`F_oF_1$ molecule during the 50-s data acquisition period used for all measurements reported here.

The scattered light intensity from the nanorod in Figure 3B varied between maximum and minimum values of 2500 and 500. The difference between these values resulted in a dynamic range of about 2000 photons per sample, which determined the sensitivity of the measurement. This was the minimum dynamic range used to measure rotation (the average range was ~ 3000 photons per sample), and thus serves as the upper limit for determining the error in the measurement of rotational position. The error in the determination of rotational position primarily results from variations in the intensity of scattered photons from the nanorod. The distribution of light intensity scattered from the nanorod was narrower at polarizer angles in which the intensity was at a minimum than that observed at the maximum. The degrees of rotation during a transition were derived from the arcsine of the fractional intensity of light scattered from the nanorod by Equation 1:

$$\theta = (a\sin I)180\pi^{-1} \tag{1}$$

where θ is degrees of rotation, and I is the fractional intensity of scattered light. Consequently, the standard error in the measurements of Figure 3B varied between about 0.02 and 0.12 degrees as the scattered light intensity varied between minimum and maximum values (Figure 3C).

A saturating concentration of 1 mM Mg²⁺-ATP was used for all rotational measurements reported here. Under this condition, F1-ATPase-dependent power strokes occur in uninterrupted 120° rotational events separated by 8.3 ms catalytic dwells (Spetzler et al, 2006). A schematic of scattered light intensity during three consecutive power strokes (one complete revolution) is shown in Figure 3D when the nanorod was initially aligned nearly, but not exactly, perpendicular to the polarizer. As the stochastic nature of the enzyme results in a variation in the rotational position of each catalytic dwell (Yasuda et al, 2001), the alignment of the nanorod with the polarizer will show small variations during the data collection period. Consequently, the most sensitive and precise measure of rotational position during a power stroke is obtained when the nanorod rotates through the parallel and perpendicular alignment of the polarizer during a single 120° power stroke. This was measured as a change between maximum and minimum intensities of the scattered light, which corresponds to 90° of rotation, and an algorithm was used to collect these data as described previously (Spetzler et al, 2006).

The time required for 90° of continuous rotation to occur is defined as the transition time (Figure 3D). If the nanorod is initially aligned perpendicular to the polarizer and the three consecutive power strokes are exactly 120° during a single revolution such that the nanorod is also perpendicular during catalytic dwell 3, the algorithm will analyse transitions from power strokes one and three. In practice, the number of consecutive power strokes analysed is randomized by the stochastic nature of the molecular motor. Due to the randomization, there is an equal probability that the 90° increments of rotation measured as transitions represents the beginning, the middle, and the end of each 120° power stroke such that the entire power stroke is sampled in the course of the $\sim\!3520$ power stroke events monitored for each molecule during the 50-s of data acquisition.

Appearance of transient dwells independent of proton translocation

In single-molecule studies of 320 n-F_oF₁ molecules (~3520 transitions molecule⁻¹), we observed two populations of n-F_oF₁ molecules based on differences in the transitions (Figure 4A, open and solid black squares) acquired during the 50-s of rotation measured for each molecule. The transitions in one population appeared nearly identical to those observed with isolated F₁-ATPase-driven rotation (circles). These transitions were similar in that the power strokes of both rotated continuously for the full 90° of the transition and achieved equivalent velocities. The other population of n-F_oF₁ molecules took much longer to complete a 90° transition due to the appearance of transient dwells. The transient dwells were present in >90% of the power strokes of any n-F_oF₁ molecule in this latter population, whereas <1% of the F₁ power strokes had anomalies that appeared similar to

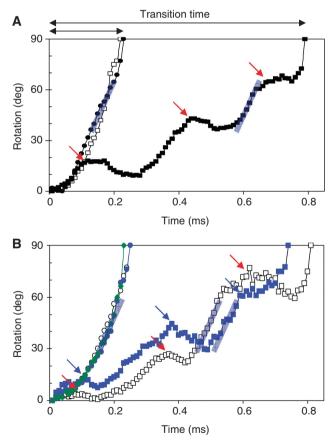


Figure 4 Power stroke events with and without transient dwells due to ATPase-driven rotation of single molecules. Arrows indicate transient dwells. (A) Example transitions for F_1 (\bullet), as well as for n-F_oF₁ with (\blacksquare) and without (\square) transient dwells. (B) Transitions obtained from n-F_oF₁-a ∇ 14 (\bullet), n-F_oF₁-aR210G with (\square) and without (\bigcirc) transient dwells, and n-F_oF₁-cD62G with (\square) and without (\bullet) transient dwells. Nanorod attachment occurred via the γ subunit for F₁ or via the c-ring for n-F_oF₁. Data were acquired at 100kHz in the presence of 15% PEG400 (v/v) and 1 mM MgATP, and were converted from scattered light intensity to degrees of rotation by equation (1). Grey lines indicating the power stroke velocity have the same slope.

Table II Periodicity of transient dwells observed during rotational transitions in the presence of 15 and 30% PEG400

Strain	Rotation between transient dwells	Transient dwells per 90° rotation		Transitions examined
WT	$37^{\circ}\pm0.5$	2.48 ± 0.043	45	36 421
cD62G	$38^{\circ} \pm 0.8$	2.45 ± 0.065	17	11 458
aR210G	$37^{\circ} \pm 0.5$	2.50 ± 0.039	17	5856

these dwells. The power stroke velocity was not significantly altered by the presence of transient dwells as indicated by the blue lines in Figure 4 that have the same slopes.

During the transient dwells, the c-ring often rotates a few degrees in the reverse direction (Figure 4). An average interval of $37.3^{\circ} \pm 0.75$ between transient dwells in a transition was observed (Table II) as derived from Equation (1). The 90° transitions were also found to contain an average of ~2.5 transient dwells. Both of these measurements translate into an average of ~10 transient dwells for each complete revolution of the c-ring, indicating that the transient dwells result from an interaction between subunit-a and each c-subunit in the 10 c-subunit *E. coli* ring (i.e. every 36°). These data also show that the occurrence of longer rotational stepping that skip one or more c-subunits is rare.

To determine whether the appearance of transient dwells resulted from the aR210-cD61 interaction essential to F_o proton translocation, we studied the mutants $n-F_oF_1$ -aR210G and $n-F_oF_1$ -cD62G (instead of cD61 due to the c2 ∇ C insertion) that lack these charged groups. Transitions from molecules of both mutants contained transient dwells (Figure 4B) that had the same periodicity and duration of those lacking these mutations (Table II). This indicates that the transient dwells do not result from the periodic cD61–aR210 interaction.

The transient dwells were eliminated by a mutation to subunit-a (a ∇ 14) as shown in Figure 4A. All 219 n-F₀F₁-a ∇ 14 molecules examined had continuous power strokes that lacked transient dwells, and were comparable to those of F_1 . The subunit-a mutation designated $n-F_0F_1-a\nabla 14$ was formed by site-directed mutagenesis during PCR at suboptimal conditions when making the aR210G mutant (see Materials and methods), and was identified by sequencing. This mutant has a 14 amino acid insert that duplicates the sequence in transmembrane helix-4 (TMH4) between residues 204 and 217, except that aR210 has been converted to valine and glycine in the repeated sequences (Figure 5A). The mutation did not alter the subunit composition of F_0F_1 as determined by PAGE (Figure 5B). However, subunit-a in F_0F_1 -a ∇ 14 appears to run as a slightly higher molecular weight band. The F_0F_1 -a $\nabla 14$ had a k_{cat} of 90 s⁻¹ for ATPase activity that was not susceptible to inhibition by DCCD (Table I), and membranes containing this mutant were unable to catalyse ATPase-dependent proton translocation. These results are similar to those obtained for other aR210 mutations (Lightowlers et al, 1987; Ishmukhametov et al, 2008).

The data in Figures 4 and 5 indicate that the transient dwells result from a periodic interaction between subunits-a and c in F_o that has not been previously described. Figure 5C uses the well-established subunit-a folding pattern (Zhang

and Vik, 2003b; Moore and Fillingame, 2008) to show the approximate locations of the residues affected by the a ∇ 14 mutation. Due to the restraints imposed by TMH3 and TMH5, the hydrophobic insert likely extends TMH4 to displace helix 4C and to a smaller extent helix 3C. In the crystal structure of the chloroplast F_0F_1 c-ring (Figure 5D), residues analogous to *E. coli* cD44 and cR50 form a ring of charged residues (Vollmar *et al*, 2009) that is also evident in the c-ring structures from *Ilyobacter tartaricus* (Meier *et al*, 2009; Pogoryelov *et al*, 2009) and *Saccharomyces cerevisiae* (Dautant *et al*, 2010). Other than aR210 and cD61, the interface between helices 3C and 4C and the cytoplasmic side of the c-ring is the only location where conserved charged residues capable of forming salt bridges are juxtaposed.

Effects of nanodiscs and F_o mutations on Torque

Figure 6 shows transition times as a function of PEG400 concentration, which was used to increase the viscosity of the solution as a means to impose a greater load on the motor. PEG400 molecules were chosen because they were determined to behave as a Newtonian fluid (Hornung et al, 2008). As such, they are too small to be pulled along by the rotating nanorod, and thus do not make secondary nonlinear contributions to the drag. The increased drag on the gold nanorod due to the viscosity of the PEG400 solution exerts a load on the motor that slows the power stroke velocity, which can be used to determine the torque (Hornung et al, 2008). For a given size of nanorod, the angular velocity is determined by the rotational distance (arc distance) divided by time. In the absence of the transient dwell, the average angular velocity is calculated using the arc distance of the rod moving 90° divided by the transition time. In the presence of the transient dwell, the average angular velocity is the arc distance of the rod moving 36° divided by the average time between transient dwells.

Torque was calculated from the drag and the velocity by Equation (2):

$$T = \Gamma w, \tag{2}$$

where Γ is the drag force and ω is the angular velocity of the power stroke. The dependence of the purified F₁-ATPase power stroke velocity on the PEG400 concentration was about the same as that of n-F₀F₁, and was not significantly altered by any of the mutations examined (Figure 6), resulting in ~62 pN nm of torque. This value is closely similar to that reported previously for purified *E. coli* F₁-ATPase (Hornung *et al*, 2008). Thus, incorporation of F₀F₁ into nanodiscs did not significantly alter the abundance of molecules observed to rotate, or the power stroke velocity (i.e. the torque). Nanodiscs also increased the stability of the ATPase activity of the enzyme. Based on these results, we conclude that incorporation of the F₀ into nanodiscs does not influence the speed or efficiency of the rotor.

Transient dwells form when viscous drag slows rotation below a threshold speed

The proportion of F_0F_1 molecules that exhibited transient dwells increased with the viscosity of the medium (Figure 7A). All molecules measured at PEG400 concentrations below 15% had the same rotational profile of power stroke transitions as that of F_1 . The abundance of molecules with transient dwells increased from 27 to >80% for PEG400

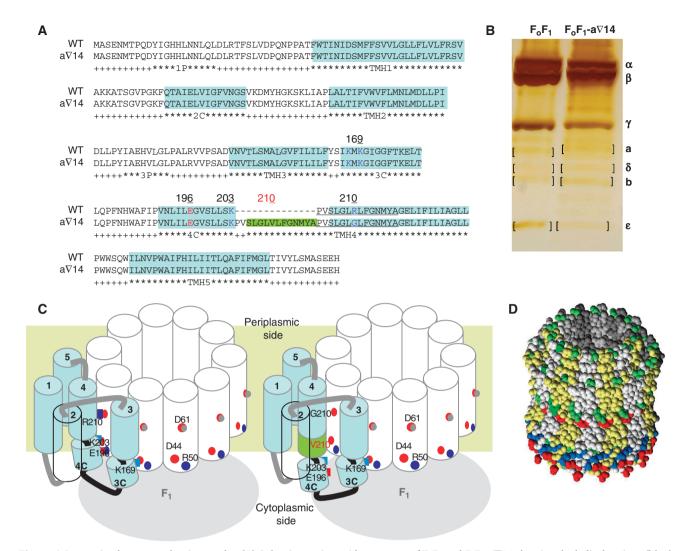


Figure 5 Interaction between subunits a and c. (A) Subunit-a amino acid sequences of F_0F_1 and F_0F_1 -a ∇ 14 showing the helical regions (blue) as established by Zhang and Vik (2003b); and Moore and Fillingame (2008). In the n- F_0F_1 -a ∇ 14 mutant, a 14 amino acid sequence was duplicated (green bar and underlined) that included arginine at position 210, which was replaced by value and by glycine in the duplicated region. (**B**) Silver-stained 8–16% gradient gel (Bio-Rad) of F_0F_1 and F_0F_1 -a ∇ 14 subunits separated by SDS-PAGE. (**C**) Folding of the suburit-a transmembrane helices (TMH) in the membrane relative to the position of residues cD44, cR50, and cD61 in the c-ring as established by Zhang and Vik (2003b); and Moore and Fillingame (2008). (**D**) C-ring crystal structure from chloroplast F_0F_1 (Vollmar *et al*, 2009). Hydrophobic residues (yellow, grey) distinguish c-subunits, whereas polar, positive, and negative residues are green, blue, and red.

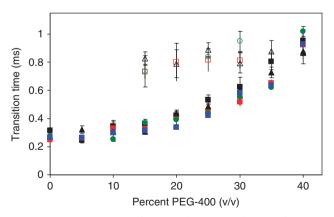


Figure 6 Transition times of rotational power strokes as a function of PEG400 concentration. Average transition times with (open symbols) and without (closed symbols) transient dwells for n-F₀F₁ (Δ , \blacktriangle), n-F₀F₁-aR210G (\square , \blacksquare), n-F₀F₁-cD62G (\bigcirc , \bigcirc), n-F₀F₁-a ∇ 14 (\square), and F₁ (\blacksquare) at each concentration of PEG400. Approximately 3520 transitions were acquired for analysis from each molecule examined.

concentrations between 15 and 35%. A similar trend was observed with $n-F_oF_1$ -aR210G and $n-F_oF_1$ -cD62G molecules as a function of PEG400 concentration although the total abundance was somewhat lower than that observed with $n-F_oF_1$.

The average transient dwell duration was $\sim 200 \,\mu s$ at 15% PEG400, but decreased with increasing PEG400 (Figure 7B). This was determined by the difference in the average transition times with and without transient dwells divided by the average number of transient dwells per transition (Figure 6), and was confirmed by direct measurement of the duration of about 200 transient dwells at each PEG400 concentration. It is noteworthy that the decreases in transient dwell duration were compensated by slower power stroke velocities at higher PEG400 concentrations (Figure 7C). Consequently, for those molecules that exhibited transient dwells, the total transition time did not change as a function of PEG400 (Figure 6). At 40% PEG400, the power stroke velocity was too slow to distinguish changes in the slope that identify the existence of transient dwells. The same relationship between

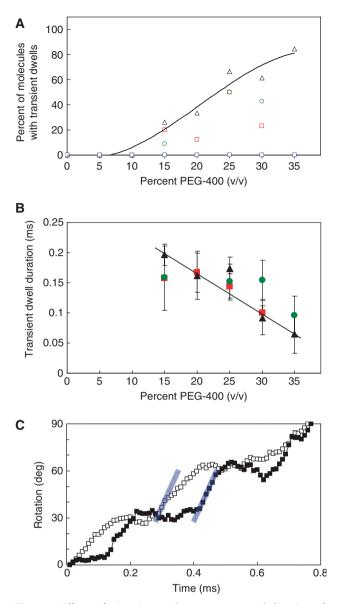


Figure 7 Effects of viscosity on the appearance and duration of transient dwells. (A) Abundance of $n-F_0F_1$ (Δ), $n-F_0F_1-aR210G$ (\square), $F_0F_1-cD62G$ (\bigcirc), and $n-F_0F_1-a\nabla14$ (\square) molecules with transient dwells as a function of PEG400 concentration. (B) Transient dwell duration of $n-F_0F_1$ (\blacktriangle), $n-F_0F_1-aR210G$ (\square), and $n-F_0F_1-cD62G$ (\bigcirc) as a function of PEG400. (C) Transitions with transient dwells at 15% PEG400 (\blacksquare) and 30% PEG400 (\square) have the same transition time, but different transient dwell durations. Grey lines indicating the power stroke velocity in the presence of 15% PEG400 have the same slope.

transient dwell duration and transition time was also observed with the $n-F_0F_1$ -aR210G and $n-F_0F_1$ -cD62G mutants.

The appearance of transient dwells is not an effect of the binding of PEG400 to the enzyme as none of the n-F₀F₁-a ∇ 14 molecules contained these dwells regardless of the PEG400 concentration. Alternatively, the transient dwells may appear when PEG400 increases the viscous drag on the motor beyond a threshold value. To test this hypothesis, the maximum and minimum drag generated by a 77 × 39 nm² gold nanorod was calculated as a function of PEG400 concentration when the axis of rotation was at the end or in the middle of the nanorod, respectively (Figure 8A). Comparison of

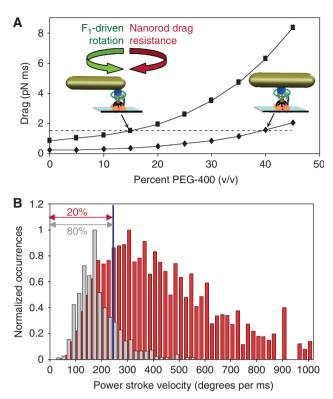


Figure 8 Requirements for transient dwell formation. (A) Minimum and maximum drag imposed by a $77 \times 39 \text{ nm}^2$ gold nanorod on the motor as a function of PEG400 based upon c-ring attachment in the middle (\blacklozenge) or the end (\blacksquare) of the nanorod as determined by Equation (2). Dotted line: threshold of drag for the appearance of transient dwells in Figure 4A. (B) Distributions of power stroke velocities from single molecules with $77 \times 39 \text{ nm}^2$ nanorods at 15% (red) and 35% (grey) PEG400. Velocity at blue line is 220° ms⁻¹. The slowest 20 and 80% transitions at 15 and 35% PEG400 is indicated.

direct measurements of the drag on the nanorod (Hornung *et al*, 2008) showed that the propeller model provided a close approximation of the drag force as a function of PEG400 concentration. Based on this model, the drag force is approximated by equation (3),

$$\Gamma = \frac{4\pi\mu(L_1^3 + L_2^3)}{3\cosh^{-1}(h/r)},$$
(3)

where L_1 and L_2 are the length of the propeller extending from the rotational axis, *r* is the radius of the rod, μ is the viscosity of the medium, and h is the height of the cylinder axis relative to the surface. Using Equation (3), the load exceeds ~ 1.4 pN ms for the subset of molecules in which the c-ring is attached to the end of the nanorod when transient dwells first become apparent (15% PEG400). This same load on the motor occurs at 40% PEG400 when the rotation axis is in the middle of the nanorod. This suggests that the appearance of transient dwells occurs when the load on any $n-F_0F_1$ molecule exceeds the threshold of 1.4 pN ms. As the binding position of the c-ring along the length of the nanorod is random, the percentage of molecules that exceed this threshold increases as a function of PEG400, consistent with the trend observed in Figure 7A. Moreover, when the transient dwell is present for a particular molecule, it is evident in every observed transition, suggesting that the cause of the transient dwell is constant for any given

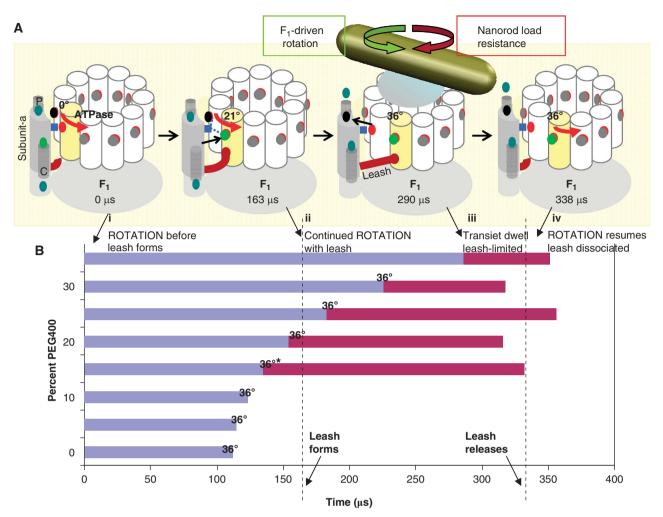


Figure 9 Timing of formation and release of the leash interaction relative to c-ring rotation. (**A**) Model of leash interaction between subunits a and c at 35% PEG400. Green dots: H^+ in subunit-a cytoplasmic (C) and periplasmic (P) half-channels. Grey/black dots: H^+ on cD61 (red dots). Periplasmic half-channel in subunit-a (P). (i) Rotation without leash; (ii) Leash forms, but rotation continues; (iii) Rotation at end of leash; (iv) Leash dissociates, rotation resumes. (**B**) Time course of c-ring rotation and transient dwell duration as a function of PEG400 concentration. Blue bars indicate average time for the c-ring to rotate 36°. At 15% PEG400, the subpopulation of molecules that gave rise to the transient dwells rotated more slowly (36°*) than the average rotational velocity (Figure 8B). Red bars indicate average transient dwell duration.

molecule, as is the case for the binding orientation of the rod to the motor.

The distributions of power stroke velocities observed in the presence of 15 (red bars) and 35% (grey bars) PEG400 is shown in Figure 8B, where the fraction of molecules that exhibit transient dwells was 20 and 80%, respectively. Thus, the molecules subject to the drag on the motor exceeding the 1.4 pN ms threshold will have power stroke velocities in the slowest 20 and 80% of the two distributions at these PEG400 concentrations. This corresponds to molecules with velocities < 220° ms⁻¹ (blue line) for both distributions. At this velocity, the c-ring rotates 36° in about 163 µs. As the interaction responsible for the transient dwells occurs about every 36° (i.e. between subunit-a and each c-subunit, Table II), the time constant for formation of the transient interaction is $\sim 160 \,\mu s$. Thus, the drag threshold needed to observe transient dwells is actually the extent to which the power stroke velocity must be decreased in order to allow the interaction to form within the time that the c-ring rotates 36°. Any molecule that rotates 36° in <163 µs does not exhibit transient dwells.

The interaction responsible for the transient dwell behaves as a leash

Figure 9 shows the average time courses of $n-F_0F_1$ molecules as a function of PEG400 concentration. Blue bars represent the average time required for the c-ring to rotate 36° based on distributions like those in Figure 8B, and red bars show the average transient dwell duration (Figure 7B). The rotational velocity of all the molecules in 0-10% PEG400 was too fast for the transient interaction to form. At 15% PEG400, the 20% of molecules subject to a load sufficient to allow the transient interaction to form (indicated as $36^{\circ*}$) had a velocity less than the average of the distribution (Figure 8B). For all the molecules exhibiting transient dwells regardless of PEG400 concentration, the average total time required to rotate 36° and complete the transient dwell was 338 µs independent of the power stroke velocity. As the time constant to form the subunit a-c transient interaction is 163 µs, these results indicate that the time constant for the termination of the interaction is $\sim 175 \,\mu s$, independent of PEG400 concentrations. The turnover time of the

interaction responsible for the transient dwells is then $\sim\!338\,\mu s.$

The extent to which the c-ring has rotated after $163 \,\mu s$ (transient interaction formation time) decreases as the rotational velocity is slowed by increased load on the motor. For example, at 25 and 35% PEG400, the interaction forms after the c-ring has rotated ~31 and 21°, respectively (Figure 9A). As transient dwells appear about every 36°, the decrease in observed transient dwell duration (161 and 59 μs at 25 and 35% PEG400, respectively) indicates that formation of the transient interaction acts as a leash. At the slowest velocities in the distribution of any given molecule (e.g. Figure 8B), the transient interaction forms after the c-ring has rotated only a few degrees such that the leash must allow rotation to continue to a limit of ~36°.

Discussion

Model for the leash mechanism

A schematic of F_0F_1 illustrating the steps in the F_0 leash mechanism is summarized in Figure 9A for F₁-ATPase-driven rotation in 35% PEG400 based on the data presented here. In this model, all cD61 carboxyl groups (red) dots are protonated (grey/black dots) except in the yellow c-subunit where cD61 interacts with aR210 (blue square). (i) At the start of an F₁-ATPase 120° power stroke, constant torque is applied to the c-ring, which begins to rotate in the absence of the leash. Residue cD61 of the yellow c-subunit becomes protonated (green dot) from the cytoplasmic half-channel (C) as it rotates away from aR210. (ii) Formation of the transient leash occurs at ~163 μ s, at which point the c-ring has rotated 21°, which does not interfere with rotation. (iii) Rotation is interrupted at $290\,\mu s$ when the leash becomes fully extended upon rotation of the c-ring by 36°, at which point the transient dwell becomes apparent. A proton (black dot) moves to the periplasmic half-channel (P) as aR210 deprotonates cD61 on the adjacent c-subunit. (iv) Rotation resumes at 338 µs upon dissociation of the leash.

As the E. coli F1-ATPase generates 62 pN nm of torque (Hornung et al, 2008), the leash is likely comprised of multiple salt bridges between subunit-a and the c-ring to be of sufficient strength to cause the transient dwells. With the exception of aR210 and cD61, conserved charged residues on both subunit-a and the c-ring that could participate in such intersubunit salt bridges occur only on the cytoplasmic side of the membrane (Figure 5). At this location, available crystal structures of c-rings (Meier et al, 2009; Pogoryelov et al, 2009; Vollmar et al, 2009; Dautant et al, 2010) all show a band of charged residues analogous to cD44 and cR50 in E. coli. Cross-linking studies between subunits-a and c support the juxtaposition of aE196 and cR50 (Jiang and Fillingame, 1998; Zhang and Vik, 2003a, b), and aqueous accessibility studies position aK169, aK203, aE196, and cR50 close to the cytoplasmic surface of the membrane facing subunit-a (Fillingame et al, 2000; Zhang and Vik, 2003b; Steed and Fillingame, 2009). This suggests that the axis of subunit-a, helix 4C may be more parallel to the membrane surface than is represented in Figure 5C. At this location, the deprotonated aE196 and cD44 carboxyl groups that could participate in salt bridges between subunits-a and c are not subjected to the large free energy penalty that would result if they were buried in the hydrophobic core of the membrane-like cD61 (Elston *et al*, 1998). Thus, the formation/dissociation kinetics of salt bridges formed from these groups is anticipated to be consistent with the transient dwells observed here.

Site-directed mutations that altered the charge of aE196 were found to significantly alter E. coli Fo-dependent proton translocation (Vik et al, 1988), and aE196 has been identified as one of three sites leading to oligomycin resistance in mitochondria (Breen et al, 1986; John and Nagley, 1986; Ray et al, 1988). Mutational studies have implicated that aE196 (Zhang and Vik, 2003a; Moore et al, 2008) and aK203 (Moore et al, 2008) participate in the proton half-channel between cD61 and the cytoplasm. Mutations aK203C (Angevine et al, 2007), aK167Q, and/or aK169Q (Lightowlers *et al*, 1987) do not effect the activity of F_0F_1 significantly. However, in each case, polar groups were created by these mutations, which would not eliminate the ability to form hydrogen bonds between subunits-a and c. It is noteworthy that participation in a proton half-channel and in the formation of the transient dwells observed here need not be mutually exclusive functions of a particular residue. These functions could be linked under conditions where the free energy from the PMF and the logQ are close to equilibrium. The results of the F_0F_1 -a ∇ 14 mutant reported here (Figure 4B) that eliminate the transient dwells are consistent with the juxtaposition of these residues on subunits-a and c. However, given that this mutant includes a 14 amino acid insertion, more experiments are required to pinpoint the specific interaction responsible for the transient dwells.

The results presented here support a mechanism for c-ring rotation when the F_1 motor is subjected to a load (Figure 9A). Under these conditions, the cycle of Figure 9A repeats with each c-subunit in 36° increments until the F1-ATPase power stroke completes 120° to begin the next catalytic dwell (Figure 10A). This conclusion is supported by the observations of 2.5 transient dwells per transition with an average of 37° between transient dwells within a transition (Table II). These data suggest that double, triple, or multiple steps between transient dwells are rare. This means that on average, each 120° power stroke includes rotation of the c-ring by three complete c-subunit steps (36° each) and 12° of a fourth step. This partial fourth step can be accommodated by the observation presented here that the leash can form before the c-subunit has rotated the full 36°. The well-established compliance of the γ -subunit drive shaft (Sielaff *et al*, 2008) and the stochastic nature of F₁-ATPase-driven rotation (Yasuda et al, 1998) will also mediate the stoichiometric differences between F_o and F₁.

The lack of an effect of aR210G and cD62G mutations on the transient dwells reported here demonstrates that proton translocation is independent of, and not rate limiting to, the interactions responsible for the transient dwells during ATPase-driven rotation. Thus, even though the power stroke velocity at $\leq 10\%$ PEG400 occurs too quickly for the transient dwell to form (Figure 7A), ATPase-driven proton translocation across the lipid bilayer of the nanodisc can still occur (Figure 10B). This conclusion is consistent with the much higher rates of F₀-dependent proton translocation that can occur in response to a PMF after F₁ has been removed (Franklin *et al*, 2004; Wiedenmann *et al*, 2008), and the fact that the proton translocation rate of F₀ does not saturate at high-driving force (Feniouk and Junge, 2008). Based on the leash formation rate of 163 µs/H⁺ reported here, the leash

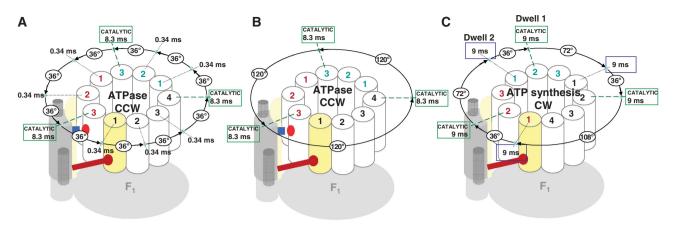


Figure 10 Summary of F_oF_1 rotational stepping events powered by F_1 during ATP hydrolysis and by F_o during ATP synthesis. (**A**, **B**) F_oF_1 c-ring transient dwells during F_1 -ATPase-driven rotation with 1 mM MgATP in the presence (**A**) and absence (**B**) of viscosity-induced load. The ATP-waiting dwells were not observed in the data presented here or by Düser *et al* (2009) due to the use of the saturating substrate concentrations. (**C**) Model for the grouping of F_oF_1 c-ring rotation events during ATP synthesis powered by F_o with a $\Delta pH = 4.1$ at saturating [ADP] and [Pi] consistent with (Düser *et al*, 2009).

will form until F_o proton translocation rates exceed 6135 H⁺ s⁻¹. Chloroplast F_o was observed to exceed this rate, and a PMF could not be found that was large enough to maximize the proton translocation rate (Feniouk and Junge, 2008). However, direct comparison of chloroplast and *E. coli* F_o indicates that the latter translocates protons more slowly (Wiedenmann *et al*, 2008), such that leash engagement will occur in *E. coli* F_oF_1 even at the highest reported rates of 3100 H⁺ s⁻¹ (Franklin *et al*, 2004).

Although the highest reported ATP synthesis rates of 100 s^{-1} correspond to 3.3 ms/H^+ , or ~10 ms per 120° (Senior et al, 2002), a significant amount of that time is consumed by the catalytic dwell. Düser et al (2009) reported 9 ms dwells during ATP synthesis with *E. coli* F_0F_1 , which is consistent with the 8.3-ms catalytic dwell observed for E. coli F₁-ATPase (Spetzler et al, 2006). Due to the F₁ requirement of 120° steps for each catalytic dwell, the rotational stepping of the c-ring during ATP synthesis must also occur in approximate increments that sum to ~120°. As Düser et al (2009) observed that about 50% of the c-ring substeps were 36°, each 120° step is likely the sum of one 36° substep and one substep of a larger step size (Figure 10C). This corresponds to a $36^{\circ} + 108^{\circ}$ substep combination between catalytic dwells involving c-ring rotation by four c-subunits, and a $36^{\circ} + 72^{\circ}$ substep combination between the two catalytic dwells involving rotation by three c-subunits. This hypothesis predicts a 1:2 ratio for the 72° and 108° substeps, which is similar to the observed ratio of 1:3. This difference can be compensated by the stochastic nature of the enzyme, as evidenced by occasional (3%) 144° steps (Düser *et al*, 2009), and by compliance of the γ -subunit (Sielaff *et al*, 2008).

As 9 ms dwells were observed after each 36° substep and each multiple (double or triple) subunit-c step (Düser *et al*, 2009), the model of Figure 8C has two 9 ms dwells per each 120° of rotation that results in ATP synthesis. If transient dwells occur under these conditions as they are suggested to do by the overall proton translocation rate, then these dwells that result from the leash with a 335-µs turnover time likely occurred during the multiple subunit-c steps during ATP synthesis and were not resolved by the 2-ms time resolution of the FRET measurements. Although one of the two 9 ms dwells during ATP synthesis is likely the catalytic dwell, more work is required to understand the relationship between the 9-ms, 36° substep dwell observed during ATP synthesis and the much shorter 36° transient dwells that occur during ATP hydrolysis reported here.

Transient dwells occur under conditions common for ATP synthesis in vivo

The experiments presented here indicate that the leash forms when a load on the F₁-ATPase motor is sufficient to slow the rotational velocity below a threshold value to provide enough time for the interaction between subunits a and c to occur. *In vivo*, proton translocation driven by a transmembrane electrochemical gradient powers c-ring rotation in the opposite (CW) direction from that driven by ATP hydrolysis (Börsch *et al*, 2002). The rate of ATP synthesis only becomes significant above an electric potential threshold, which for *E. coli* F₀F₁, is ~40 mV (Kaim and Dimroth, 1998). However, this threshold is dependent on Q, the ATP/ADP \cdot P₁ chemical potential (Turina *et al*, 2003). Thus, a small transmembrane proton gradient will cause the F₀ motor to apply a load on F₁-ATPase-driven rotation similar to that induced by the viscous drag of PEG400 reported here.

Conversely, F_1 imposes a load on the F_o motor during ATP synthesis that is dependent on the ATP/ADP·P_i chemical potential. Although ATP synthesis rates of 100 s^{-1} (3.3 ms per H⁺) can be achieved upon addition of saturating concentrations of ADP and phosphate in the absence of ATP, at steady state the enzyme sustains a concentration of 3 mM ATP at a $\log Q \cong 0.1$ in *E. coli* cells (Senior *et al*, 2002). Under these conditions, when the energetics of the PMF and Q approach equilibrium, the rate of c-ring rotation will be sufficiently slow for the leash to form. In so doing, the leash would limit the backward (ATPase) rotation of the c-ring to ~36° at steady-state conditions when ATP concentration is high, and thereby minimize ATP hydrolysis events relative to synthesis.

Potential relationship of the leash to the Brownian ratchet mechanism

A model for the role of the leash in PMF-powered CW c-ring rotation during ATP synthesis is shown in Figure 11. The PMF is represented by the disproportionate H^+ abundance aligned

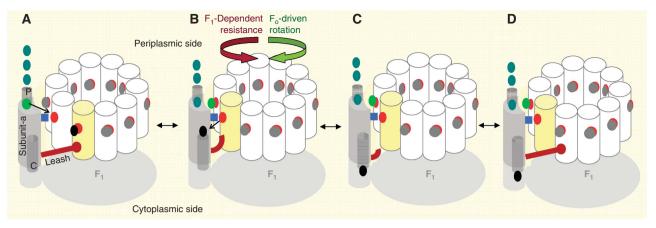


Figure 11 A model for a leash-based F_o ratchet mechanism during F_oF_1 -ATP synthesis. (A) The end of the leash prevents further counterclockwise rotation. Excess H⁺ (green dots) on the periplasmic side indicates a PMF. (B) The leash limits Brownian-dependent back and forth c-ring motion to 36°. During clockwise rotation, subunit-a aR210 (blue square) displaces the H⁺ (black dot) from yellow subunit-c cD61 (red dot) to the cytoplasmic half-channel (C), whereas the adjacent white subunit-c cD61 becomes protonated (light green dot) from the periplasmic half-channel (P). (C) The interaction responsible for the transient dwell dissociates. (D) Formation of the interaction at the end of the leash with the adjacent subunit-c ratchets the c-ring clockwise with H⁺ translocation to power ATP synthesis.

with the half-channels (green dots). During the 36° of CW movement permitted by the leash $(A \rightarrow B)$, the c-ring accepts a proton from half-channel-P and donates a proton to halfchannel-C, which is reversed by CCW rotation $(B \rightarrow A)$. Upon dissociation of the leash (C), the probability that the leash will reform with the adjacent c-subunit that ratchets the c-ring clockwise to catalyse ATP synthesis $(C \rightarrow D)$ versus reforming with the same c-subunit $(C \rightarrow B)$ depends on the relative probabilities of c-ring protonation by each half-channel due to the availability of H⁺ at each half-channel at that moment. The PMF promotes a higher probability of CW rotation. At high steady-state ATP concentrations, we anticipate that several $C \rightarrow B \rightarrow C$ cycles will occur before a successful $C \rightarrow D$ step takes place. Based on this model, the $C \rightarrow D$ step would have occurred once for every 27 $C \rightarrow B \rightarrow C$ cycles (given a 0.335-ms leash turnover time) for the 9-ms ATP synthesis-dependent 36° substep observed previously (Düser *et al*, 2009). In the absence of F_1 , the F_0 proton translocation rate in response to a 73-mV electric potential is two-fold greater when the potential is in the hydrolysis direction than in the synthesis direction (Wiedenmann et al, 2008). This difference might be explained by anisotropic behaviour of the leash.

The F_o motor has been postulated to use a Brownian ratchet mechanism that requires, first, that there are two noncolinear proton access half-channels from each side of the membrane leading to the cD61 carboxyl, and second, that rotational diffusion of the c-ring relative to subunit-a is restricted in some manner (Junge *et al*, 1997; Oster *et al*, 2000). The leash behaviour reported here can fulfill the role of the second requirement of the ratchet to restrict the rotational diffusion of the c-ring to 36°. We anticipate that the use of nanodiscs and the time resolution that is possible from the single-molecule approach presented here will have a broad application for the study of other integral membrane motors and proteins that undergo similar conformational changes.

Materials and methods

Preparation of F₁

 $F_1\text{-}ATPase$ containing a His_6 tag on the N-terminus of the $\beta\text{-subunit}$ and γS193C as described previously (Greene and Frasch, 2003; York

et al, 2007) was purified from the *E. coli* XL-10 strain. Additional details are provided in Supplementary data.

Preparation of F_oF₁

The pFV2 plasmid encoding cysteine-free F_0F_1 (Ishmukhametov *et al*, 2005) was used to introduce $c2\nabla C$, cD62G, aR210G, and $a\nabla 14$ mutations with the QuikChange®II-XL Site-Directed Mutagenesis Kit (Stratagene) using oligonucleotides synthesized by Invitrogen. All mutations were confirmed by sequencing. The $F_0F_1-a\nabla 14$ mutation was accidentally formed during PCR using a nonoptimal annealing temperature in the process of making the aR210G mutation.

The F_oF_1 isolation was similar to that described in Ishmukhametov *et al* (2005) with modifications. The DK8 *unc* operon deletion strain of *E. coli* (Klionsky *et al*, 1984) containing the plasmid pFV2 were grown with shaking at 37°C in 1 l of LB medium with 50 µg/ml of ampicillin. About 4–5 g wet weight of bacteria was harvested by centrifugation at 7700 g for 15 min at 4°C and stored as cell pellets at -80° C. Cell pellets thawed at 25°C were immediately resuspended in 100 mM KCl, 5 mM MgCl₂, 0.1 mM EDTA, 2.5% glycerol, and 200 mM Tris/HCl, pH 8.0, and the cells were broken using a French press at 16 000 psi. Unbroken cells were collected as a pellet at 7700 g for 15 min at 4°C and discarded.

All subsequent steps were performed at 4°C. The supernatant fraction containing membrane vesicles was centrifuged at 184 000 *g* for 1 h. The pellet of membranes containing F_oF_1 was resuspended in extraction buffer (EB; 100 mM NaCl, 40 mM ϵ -aminocaproic acid, 15 mM *p*-aminobenzamidine, 5 mM MgCl₂, 0.03% phosphatidyl-choline, 1.0% octyl glucopyranoside, 0.5% sodium deoxycholate, 0.5% sodium cholate, 6% glycerol, 30 mM imidazole, and 50 mM Tris/HCl, pH 8.0). About 5 ml of EB was added to 1 g of membranes, which was shaken for 90 min at 4°C, then centrifuged at 184 000 *g* for 1 h. The supernatant was applied to a Ni-NTA column containing 1.5 ml of resin equilibrated with EB. The resin containing bound F_oF_1 was washed with about 20 ml of EB, and F_oF_1 was eluted with 3 ml of EB containing 180 mM imidazole. After determination of protein concentration, the solubilized F_oF_1 was immediately incorporated into nanodiscs.

Preparation of n-F_oF₁

The MSP construct MSP1E3D1 was used that is composed of scaffold protein contains three 22-mer amphipathic helices and a cleavable his-tag as the result of an introduced TEV protease site to facilitate purification (Denisov *et al*, 2004; Bayburt *et al*, 2007). Nanodiscs formed from this construct contain a lipid bilayer that is about 13 nm in diameter surrounded by a double belt of the MSP helices. The His-tag of the purified MSP-1E3D1 was cleaved by overnight incubation with TEV protease (at 25:1 ratio, w/w) at 25°C and passed through a Ni-NTA column. We assembled n-F₀F₁ by mixing MSP in Buffer D (50 mM Tris, pH 8.0, 100 mM NaCl, 4 mM 4-aminobenzamidine, 5 mM MgCl₂ and 5% (v/v) glycerol) with

10% sodium cholate in Buffer D, and F_0F_1 in Elution Buffer to achieve a 1:7 molar ratio of F_0F_1 :MSP in 1% sodium cholate with a final volume not exceeding 1 ml, and adjusted with Buffer D. To make biotinylated n- F_0F_1 , a 10-fold molar excess of biotin maleimide was added to this mixture. The mixture was incubated at 4°C for 15 min with gentle shaking, then passed through a 2-ml Sephadex G-50 column equilibrated with Buffer D from which 3 ml of effluent were collected. The effluent was diluted with Buffer D to 6 ml to decrease the imidazole concentration to <30 mM, and passed through a 1.5-ml Ni-NTA column. The column was washed with 15 ml of Buffer D and eluted with Buffer D, containing 150 mM imidazole. The yield of n- F_0F_1 was ~60–70% of the amount of F_0F_1 starting material as measured with the BCA protein assay (Sigma).

2D electrophoresis

Proteins ($_{\mu}$ g total per sample) were first separated on a 5–15% native polyacrylamide gradient gel for 4.5 h at room temperature. The gel slice with the single-dominant band was excised from the gel and transferred to a glass plate, which was covered with 12% denaturing gel (SDS–PAGE). The sample was run for 4.5 h at 25°C and silver stained as described (Nesterenko *et al*, 1994).

ATP hydrolysis assay and DCCD modification conditions

To modify the enzyme by DCCD, 10–30 μ g of protein was incubated with 50 μ M DCCD at pH 6.5. at 28°C in a 2-ml cuvette for 30 min. This solution was diluted into the reaction mixture used to measure ATPase activity. The rate of ATP hydrolysis was measured with an ATP-regenerating coupled assay that resulted in a final concentration of 50 mM Tris–HCl (pH 8.0), 10 mM KCl, 2.5 mM phosphoenolpyruvate, 0.3 mM NADH, 50 μ g/ml pyruvate kinase, 50 μ g/ml lactate dehydrogenase, and 2 mM MgCl₂ and 1 mM ATP. The rate was determined as the change in absorbance at 340 nm using a Cary 100 spectrophotometer with Peltier temperature control. ATPasedriven proton translocation was measured by ACMA quenching using F₀F₁ containing membranes as described (Ishmukhametov *et al*, 2005).

Single-molecule studies

To assemble $n-F_oF_1$ with a nanorod on the slide, the slide was spotted with $5 \mu l$ of about $85 \mu g/ml$ of $n-F_oF_1$ and incubated for 5 min. The slide was washed with assay buffer (10 mM KCl and 50 mM Tris, pH 8.0) for 30 s, and excess liquid was removed. Immobilized $n-F_oF_1$ was exposed to $5 \mu l$ avidin-coated gold nanorods prepared as described (Spetzler *et al*, 2006) for 5 min, then washed with assay buffer. After excess liquid was removed, 5 ml of assay buffer containing the desired amount of PEG400,

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 $2\,mM$ ATP and $1\,mM$ MgCl_ was added, a cover slip was applied, and the slide was placed on the microscope.

Rotation measurements were performed as described in Spetzler *et al* (2009) and Spetzler *et al* (2006) using a Leica DMIRE II inverted dark field microscope illuminated with a Sutter LB-17 Xenon light source with a custom Chroma cold mirror coupled with a series 2000 Lumatec light guide to deliver 400–925 nm collimated light to the dark field condenser. Light not scattered by gold nanorods was blocked by an iris in the \times 63 variable aperture objective. Colour photos of fields of view under the microscope were obtained with a Zeiss Axiocam HSC series-2 camera with a refresh rate of 53 fps.

To measure transitions, a nanorod observed to blink red and green was positioned to a 100-µm pinhole to allow light scattered from that nanorod passed through a polarizing filter and a high pass 600 nm cutoff filter (to permit only red light), and focused onto a single-photon counting avalanche photodiode (Perkin Elmer SPCM-AQR-15) that has a dark count of ~50 photons s⁻¹ with a temporal resolution of 50 ns. Detector output was fed to a National Instruments DAQ PCI-6602 counter/timer board. Photons were recorded and binned into different time intervals that provided rotational data with various time resolutions. The rotation of each molecule was monitored for 50 s at a data acquisition speed of 100-200 kHz to provide a minimal time resolution of 20–10 µs. Custom software was written in LabView 7.1 to control data acquisition and storage. Additional custom software was written in Matlab 6.5 to compute transition times.

Supplementary data

Supplementary data are available at *The EMBO Journal* Online (http://www.embojournal.org).

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Conflict of interest

The authors declare that they have no conflict of interest.

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