



Published in final edited form as:

J Am Chem Soc. 2010 June 2; 132(21): 7312–7320. doi:10.1021/ja906700x.

Development of a rotamer library for use in β -peptide foldamer computational design

Scott J. Shandler¹, Maxim V. Shapovalov², Roland L. Dunbrack Jr.², and William F. DeGrado¹

¹ Department of Biochemistry and Molecular Biology, University of Pennsylvania School of Medicine

² Institute for Cancer Research, Fox Chase Cancer Center, 333 Cottman Avenue, Philadelphia PA 19111

Abstract

Foldamers present a particularly difficult challenge for accurate computational design compared to the case for conventional peptide and protein design due to the lack of a large body of structural data to allow parameterization of rotamer libraries and energies. We therefore explored the use of molecular mechanics for constructing rotamer libraries for non-natural foldamer backbones. We first evaluated the accuracy of molecular mechanics (MM) for the prediction of rotamer probability distributions in the crystal structures of proteins is explored. The van der Waals radius, dielectric constant and effective Boltzmann temperature were systematically varied to maximize agreement with experimental data. Boltzmann-weighted probabilities from these molecular mechanics energies compare well with database-derived probabilities for both an idealized α -helix ($R = 0.95$) as well as β -strand conformations ($R = 0.92$). Based on these parameters, *de novo* rotamer probabilities for secondary structures of peptides built from β -amino acids were determined. To limit computational complexity, it is useful to establish a residue-specific criterion for excluding rare, high-energy rotamers from the library. This is accomplished by including only those rotamers with probability greater than a given threshold (e.g. 10%) of the random value, defined as $1/n$ where n is the number of potential rotamers for each residue type.

Introduction

In the past decade the field of computational protein design has rapidly expanded, providing an approach to test the structural basis for function as well as a tool for designing useful molecules.¹⁻³ In parallel, chemists have demonstrated the ability to design and synthesize foldamers, short sequence-specific oligomers of non-natural building blocks that adopt secondary and tertiary structures.⁴ Recent studies from the labs of Dill,⁵⁻⁷ Zuckermann,⁸⁻¹⁰ Seebach,¹¹⁻¹⁴ Gellman,¹⁵⁻²¹ and Schepartz²²⁻²⁶ have demonstrated the feasibility of designing protein-like tertiary structures, which has set the stage for application of computational methods akin to those used in protein design to the problem of foldamer design.

Protein design depends strongly on rotamer libraries that define available conformations for side chains and their relative frequencies.²⁷ Rotamers reflect the low-energy conformations

Corresponding Author: William F. DeGrado (wdegrado@mail.med.upenn.edu).

Supporting Information: Computer readable versions of the β 314 and β 312 rotamer libraries for C₃ and C₂, along with file format information can be found at <http://degrado.med.upenn.edu/rotamers>. Complete references can be found in the supporting information.

of amino acid side chains and a rotamer library consists of those rotamers that are energetically reasonable for structure prediction or design. While molecular mechanics²⁸⁻³¹ (MM) and quantum mechanics (QM) approaches have been successful at defining the geometry of amino acid rotamers, they have been less successful at capturing their relative energetics. Handel and coworkers have shown both the challenges and limitations of using physics-based force fields.³² Therefore, protein design typically uses rotamer libraries obtained from analysis of the very large database of high-resolution crystallographic structures of proteins.^{27,33,34}

Many computational approaches to protein design³⁵⁻³⁹ and comparative modeling⁴⁰⁻⁴² combine these statistically defined rotamer libraries with energy functions that are carefully parameterized to best reproduce the native rotamers of proteins of known three-dimensional structure. As the number of available structures has increased, rotamer definitions have been refined,^{27,34,43,44} energy equations optimized,⁴⁵⁻⁴⁸ and force-field parameters improved.^{47,49-51} Recent efforts have resulted in libraries and energy functions that have been able to achieve high recovery rates of χ_1 (~90%) and χ_{1+2} (~80%).⁵² These systematic improvements have been a result of both the growing database of protein structures as well as the increased stringency for inclusion into a library. For example, recent work has shown that electron-density maps generated by X-ray crystallography experiments leave some room for interpretation.⁵³ The Penultimate Rotamer library³⁴ and more recently the backbone-dependent rotamer library (R. Dunbrack and M. Shapovalov, unpublished) have demonstrated substantial improvement in rotamer library construction by careful selection and refinement of the underlying database of rotamers used to generate the respective libraries. The use of such rotamer libraries in protein design has substantially decreased the possible search space to a large, yet tractable problem.

The challenge becomes substantially greater when a non-natural backbone is employed and there is no database to guide selection of low-energy rotamers as well as the avoidance of improbable, high-energy rotamers. In this paper, we therefore explore the scope and limitations of using molecular mechanics for developing rotamer libraries, initially calibrating the approach using conventional proteins (α -amino acids, α AA) and then extending this approach to develop libraries for β^3 substituted β -amino acid peptides (β AA).

β AA peptides are similar to their α -amino acid counterparts, the difference being an additional methylene group in the backbone between $C\alpha$ and the carboxylic acid group, as shown in Figure 1. This backbone modification has several interesting effects on β AA peptides specifically: protease resistance,⁵⁴⁻⁵⁹ an additional branch-point (C_2) for R-group substitution of each residue,⁴ and additional degrees of backbone freedom. Each of these features makes this field of growing interest for applications in medicinal chemistry and material science.

The additional degree of backbone freedom of β AA peptides contributes to their ability to adopt a family of stable helical backbone conformations, each forming a unique hydrogen-bonding pattern. The most studied backbone conformations include the 3_{14} , 3_{12} , $3_{10/12}$ helices, each of which has a different number of heavy atoms enclosed within a hydrogen bonded ring.^{4,60} The traditional α -helix has 13 atoms encompassing four residues within a hydrogen bond, while the β AA $\beta 3_{14}$ helix spans three residues and 14 atoms (Figures 1B and 1C).

The $\beta 3_{14}$ helix has been extensively studied because this is a preferred conformation for water soluble peptides composed of β^3 amino acids, which are easily obtained by homologation of the corresponding α AAs.⁶¹ Sequence-directed approaches have been successfully employed to design secondary and tertiary architectures.^{23,24,62-64} In several

cases the β -peptides were designed by exchanging residues from known α AA peptides⁶³ as well as charge patterning along the β AA helical wheel.^{22,25} These studies have not only demonstrated the feasibility of β -peptide foldamer design, but also the need for more computationally intensive approaches that can combine the advantages of traditional computational protein design with the structural versatility of foldamer scaffolds.

We have therefore developed a program to facilitate foldamer design that allows the input of a variety of backbones with a suite of options including the construction of secondary and tertiary structures, side-chain repacking using either Monte Carlo⁶⁵ or Dead End Elimination,⁶⁶ flexible and rigid-body docking, coiled-coil generation⁶⁷ together with a customizable energy equation that draws upon parameters from either AMBER⁶⁸ or CHARMM.⁶⁹ An essential component of the repacking algorithm is an accurate rotamer library with associated energies. Here we explore the accuracy of molecular mechanics (MM) computed rotamer library probabilities in the context of α AAs and extend this approach to β AAs, and evaluate the library for repacking of a tertiary structure.

Methods

Rotamer energetics and repacking experiments were calculated using a custom C++ implementation of the CHARMM software package (TIGER). All atom, bonded and non-bonded parameters used to generate structures were taken from the CHARMM27 force field.^{45,69} Hydrogen atoms were placed according to atom type and bonding patterns, using bond lengths and angles specified in the CHARMM27 parameter file.

Calibration of α -amino acid rotamer calculations against experimental data

An idealized poly-glycine 20 residue α -helix as well as a three residue β -sheet scaffold were generated using bond lengths and angles from the CHARMM27 force field, followed by backbone dihedral angle modifications to arrive at the desired homogeneous secondary structure. The central amino acid of the secondary structure was mutated to each residue type in turn.

Each side chain torsional angle for 17 of the 20 natural amino acids (glycine and alanine as well as proline were excluded) was rotated in 10° steps to generate candidate rotamers. The energetics for each candidate rotamer were then evaluated. Energetic evaluation included: (1) the Lennard-Jones van der Waals potential with atomic radii scaled at 90%, 95%, or 100% with a 12 Å distance cutoff and no switching function; (2) Coulombic electrostatics with a distance-dependent dielectric constant set to either 10 or 20, also with a 12 Å distance cutoff and no switching function; and (3) dihedral angle terms. Polar hydrogen positions for serine, threonine and tyrosine were optimized by selecting the minimum energy as a function of polar hydrogen dihedral angle searched in 10° steps.

Rotamer energetics for a particular amino acid in the α -helix were then converted into their equivalent Boltzmann probabilities, using Equation 1 (see below). Individual probabilities were then summed within angular bins used to define each rotamer (e.g., $0^\circ \leq \chi < 120^\circ$ for g^+ ; $120^\circ \leq \chi < 240^\circ$ for $trans$; $240^\circ \leq \chi < 360^\circ$ for g^-) to arrive at a final probability for a given rotamer. The temperature used for the Boltzmann conversion was optimized to obtain the highest correlation with a statistical database of rotameric probabilities for a subset of the amino acids (ILVFYSTV).

β -amino acid rotamer library generation

Various idealized β -amino acid helices of length 40 were generated using bond lengths and angles from the CHARMM27 force field, followed by backbone dihedral angle modifications to arrive at a homogeneous atomic structure. All side chains were set to β -

homoglycine for each candidate scaffold. Side chains were placed at the C³ position of the central residue on each scaffold to create an L- β -amino acid.

Each side chain torsional angle for 17 of the 20 amino acids (non-rotameric glycine and alanine as well as proline were excluded) was rotated in 10° steps to generate candidate rotamers. The energetics for each candidate rotamer were evaluated using the process optimized during the α -helix calibration.

Rotamer energetics for a particular amino acid were then converted into their equivalent Boltzmann probabilities using the α -helix optimized conversion temperature of 450K, in a manner similar to the α amino acids. It is of interest to note that the minimum energy within a given rotameric state is not necessarily the center of the rotameric bin, however there was never more than a single distinct minimum within each bin (some bins did not have any minimum due to poor energetics).

This process was repeated for each of the candidate β -peptide helix scaffolds, including the use of 3-residue and 20-residue helices. The correlation coefficient between the β_{314} 3.0 helices of these lengths is 0.857, indicating that 26.6% (1-R²) of the variance in the data is independent of the neighboring atoms.

Backbone-dependent rotamer libraries for each secondary structure type

A backbone (ϕ, ψ) dependent rotamer library was created for both α -helices and β -sheets from a set of 2028 protein chains of resolution ≤ 1.7 Å, R-factor ≤ 0.20 , and less than 50% mutual sequence identity using the PISCES server^{70,71}. Secondary structure of chains was determined with the program Stride⁷². For α -helices, only helices of length 12 or longer were used and the first four and last four residues of each helix were excluded from the data set for the rotamer library. For β -sheets, only strands of length 8 or longer were used and the first two and last two residues of each strand were excluded from the data set for the rotamer library. The resulting data sets consisted of 51,290 α -helix residues and 22,853 β -strand residues.

Rotamer frequencies were determined using kernel density estimates of the ϕ, ψ probability distribution function (pdf) for each rotamer, $p(\phi, \psi/r)$, and Bayes' rule to determine $p(r/\phi, \psi)$ on a grid every 10° from these pdfs (R. Dunbrack and M. Shapovalov, unpublished). The kernels used were von Mises functions with the bandwidth parameter κ set to 50.

β -peptide scaffold repacking

Using in-house foldamer repacking software we conducted several different experiments using the crystallographic representation (five octameric bundles, each a dimer of tetramers generated using symmetry information from the structure) of the Zwit1f bundle. For each exercise, native sidechains were removed and candidate sidechains were evaluated using the CHARMM27 molecular mechanics force-field. Candidate sidechains were drawn from the generated rotameric preferences for idealized β -3₁₄ helix presented in this work. Each candidate sidechain was evaluated using the same energetic form used in the calibration of α -amino acid sidechain energetics as described above. Experiments included the preservation of sequence identity allowing rotameric freedom, conservation of position specific charge (positive: KOR, negative: DE, neutral: ILVAFW) and finally a general sequence recovery experiment (core: ILVFA, polar exposed: DENQSTKOR, neutral exposed: ILVFW). Symmetric constraints were applied, linking identity and rotamer selections between related chains. A Monte Carlo search algorithm was applied to the available rotameric search space and concluded after 50,000 steps.

Graphics & Regressions

The PyMol 1.1beta3 graphics package, MatLab (R2008a), Microsoft Excel 2004, Gnuplot 4.0 and ChemBioDraw Ultra 2008 were used to generate figures. Microsoft Excel 2004 was used to perform regression analysis.

Results

Prediction of α AA rotamer probability distributions

To develop reliable and accurate rotamer libraries and associated energies and probabilities for β AA libraries, we first asked how well the CHARMM27^{45,69} force field, as implemented in our general-purpose program for foldamer design, predicts α AA rotamer libraries. Specifically we examined two parameters that are typically varied in computational protein design. To compensate for the rigid backbone employed in protein/foldamer design, one often dampens the Lennard Jones potential⁶⁹ by scaling the van der Waals (vdw) radii for the atoms. Furthermore, the effective dielectric constant in the electrostatic term is often scaled in a distance-dependent manner to simulate screening of electrostatic effects. We apply the Boltzmann equation:

$$P_i = \frac{\exp\left(\frac{-E_i}{k_B T}\right)}{\sum_{j=1}^n \exp\left(\frac{-E_j}{k_B T}\right)} \quad (1)$$

to compute rotamer probability distributions from computed MM energies. The Boltzmann equation relates the energy (E) of a given state (i) to its probability at a given temperature (T) in the context of all n possible states. k_B is the Boltzmann constant. An important parameter in this equation is the temperature, which is not necessarily a physically meaningful quantity. This parameter relates the energy difference between distinct rotamers to their probabilities in the overall population; for a given energy difference, a very low temperature would give rise to a single rotamer, while at limiting high temperature all rotamers would be equally populated. Therefore, we systematically varied this parameter to determine the temperature range that yields the best agreement with observed populations.

We computed the rotameric energies considering different vdw radii scaling factors (90%-100%) and distance-dependent electrostatic treatments ($10 < \epsilon < 20$) in conjunction with torsional potentials. We then used these calculated energies to compute the Boltzmann-weighted probabilities for comparison with experimental frequency distributions. An initial analysis focused on the short, neutral α AA (ILVFYST) on an α -helical backbone with surrounding residues set to glycine. The experimental frequencies came from the most recent backbone-dependent rotamer library²⁷ with backbone dihedrals ϕ and ψ near the standard values of -60° and -50° respectively. We observe an excellent correlation over a range of temperatures (300-500 K) with an optimum correlation coefficient $R=0.946$ at 450K with vdw scaling of 100% and distance-dependent dielectric constant of $\epsilon = 10$, as shown in Figure 2.

The excellent correlation with experiment for the α -helix bodes well for prediction of rotamers for the helical conformations of β AA. However, we also wished to explore the ability of this approach to predict the distributions within the β -sheet conformation of conventional proteins, which has relevance to the general problem of α AA protein design. Rotameric probabilities were calculated using an idealized three residue β -strand ($\phi=-120^\circ$, $\psi=110^\circ$), which provides enough side-chain/backbone interactions for all intra-strand

rotameric interactions. The results using MM together with a ϕ, ψ based backbone-dependent rotamer library²⁷ as a benchmark resulted in an R of 0.900 (VDW scaling 95%, $\epsilon = 20$, T=450K). However, many residues in a ϕ, ψ defined library are in extended conformations but not necessarily in β -sheets. To overcome this limitation, we developed a backbone-dependent rotamer library based only on residues in by β -sheet strands of at least 8 consecutive residues, excluding the first two and last two residues. The ϕ, ψ -dependent rotamer library frequencies were then determined from the filtered data using kernel density estimates (R. Dunbrack and M. Shapovalov, unpublished). The R for the β -sheet motif improved to 0.916 (VDW scaling 90%, $\epsilon = 20$, T= 450K, $\phi=-120^\circ$, $\psi=110^\circ$).

Criteria for constructing a rotamer library for protein design

In the context of computational protein design, we considered the effect of a minimum rotamer probability inclusion cutoff of 5% on the various amino acid types. For short side chains with one degree of freedom, this results in a reasonable representation of the rotameric possibilities. For valine, for instance, the unfavorable g^+ rotamer occurs at a frequency of about 5% in α -helices. However, for long side chains such as lysine, the 5% criteria excludes a large portion of the available rotamers. For example, lysine has potentially 81 rotameric states (3 states for each of four degrees of freedom); thus if each were equally populated they would occur at 1/81 or 1.2% abundance. In fact, three of these 81 rotamers occur with greater than 5% probability in the crystallographic database in the α -helix. Moreover, less than half (47.8%) of the lysine residues in the crystallographic database adopt one of these three rotameric states. Thus, a majority of observed α -helical lysine states are excluded in design projects that use a simple 5% cutoff criteria. Clearly the use of a constant probability inclusion cutoff eliminates a very large number of possible rotamers given that the random expectation for this identity would be 1.2% (1/81). We therefore adopted the ratio of the predicted probability, p_i , or observed frequency, f_i , of a given rotamer i and the frequency expected if rotamers were equally distributed (Ratio to the null hypothesis, RN) as our inclusion criterion:

$$RN_{PRED}(i) = \frac{p_i}{\left(\frac{1}{n}\right)} \quad (2)$$

$$RN_{OBS}(i) = \frac{f_i}{\left(\frac{1}{n}\right)} \quad (3)$$

where n is the number of rotamers available to a specific residue type, generally 3^d where d is the number of degrees of freedom of the side chain. Using the RN approach, with a cutoff of 0.1 (log RN) would include 9 additional lysine rotamers (12 total, additional 34.7% of observed states, 82.5% of total), more accurately representing the diversity of this larger amino acid. In contrast, including 95% of all observed rotamers from the structural database would require the inclusion of 25 lysine rotamers in a given design project, placing undue demands on conformational search and including relatively rare (high energy) rotamers.

In constructing a rotamer library, it is particularly important to exclude highly unfavorable states. In Figure 3, we show predicted versus observed RN for residues in α helices for the residues ILFVSTD. Here we define a bad rotamer as one with an $RN_{OBS}(i) < 0.1$ (log value -1). Of 48 possible rotamers for ILFVSTD, 22 are “bad” rotamers based on their value of $RN_{OBS}(i)$. All 22 bad rotamers were correctly predicted by MM equations based on their

having a value of $RN_{pred}(i) < 0.1$. A second consideration is to capture as many highly favorable rotamers as possible. Along these lines, again we find that the MM force field correctly predicts all 15 favorable rotamers whose values of $RN_{obs}(i)$ are greater than 0.5 (log value -0.30). Furthermore, there is a good linear correlation between $\log(RN_{obs}(i))$ and $\log(RN_{pred}(i))$ for this set of favorable rotamers ($R = 0.67$), which are most suited for protein design. The borderline cases, consisting of moderately unfavorable rotamers whose values of $RN_{obs}(i)$ lie between 0.1 and 0.5 (log values -1 and -0.30), are more difficult to predict (highlighted in yellow in figure 3). Using the best combination of vdw radii-scaling, dielectric constant and Boltzmann temperature, we were able to capture only two of the eleven rotamers in this regime. Thus, for the purposes of protein design, we would lose some of these moderately unfavorable rotamers based on a molecular mechanics rather than a structural bioinformatics approach. These borderline rotamers in total make up 5.2% (1868) of the total residues in the structural database, while the 'good' rotamers encompass 93.4% (33,443) of the residues, with the 'bad' rotamers covering the remaining 1.4% (505) of residues. These data are summarized for α -helices in Table 1 for different combinations of MM parameters.

Construction of secondary structure dependent rotamer library for β AA

There have been several definitions of the minimum energy conformations of the β_{314} helix, which vary slightly in their backbone torsional angles. We examine three such sets of β_{314} backbone angles, one derived from a recent crystal structure,²² one the result of quantum mechanics optimization,⁷³ and one derived from an energy-optimized backbone that maintains a precise 3.0 periodicity (Figure 4). Using these backbones we computed rotamer energies and the associated probabilities. In 10° steps, all possible side-chain χ angles for the natural residue types (proline was excluded) were evaluated on each scaffold, which resulted in a set of allowed rotameric states for each identity and backbone combination. The energetics of these states were then converted into Boltzmann weighted probabilities using the predetermined optimal conversion temperature (Equation 1) and vdw and electrostatic parameters optimized for α -helices and rank ordered to enable more effective rotamer searches in typical design programs. These files can be downloaded from <http://degrado.med.upenn.edu/rotamers>.

Comparison of the isoleucine side-chain χ_1 rotameric preferences across backbone scaffolds and motifs provides further insight into the differences between the secondary structures of α AA and β AA scaffolds. The α -helix and β_{314} helix demonstrate a strong preference for the *gauche*⁻ rotamer ($g^- = -60^\circ$), while the β -sheet preferences are predominantly spread between the g^- and *trans* ($t = 180^\circ$) rotamers, with a small population for the g^+ ($+60^\circ$) rotamer. However, Ile in the β_{312} helix favors the g^- rotamer with a 58% preference for the g^+ rotamer, as shown in Figure 5.

Using rotamer probabilities calculated using a crystal structure of a protein-like β -peptide²² backbone as a reference, we find the best agreement with the minimized 3.0 helix amongst the computed log probabilities for each of the idealized candidate backbones. In each case, there is reasonable agreement with correlation coefficients (R) in the range of 0.8-0.9 (Table 2). Nevertheless this exercise demonstrates the subtle interdependence of backbone and rotamer preferences. Even greater variation in rotamer probabilities is observed if one compares different secondary structures of α AA and β AA. Again taking the rotamer probabilities calculated on the β_{314} crystallographically observed helix backbone as a reference we see correlations of 0.6-0.9 when comparing rotamer probabilities of the β_{310} , β_{312} , the α -helix and β -sheet motifs (Table 3). These data further demonstrate the importance of computing backbone dependent rotamer libraries in the context of the correct secondary structure.

Finally, to confirm that the rotamer probabilities were accurately predicted for the β_{314} helix, we compared predicted Leu and Phe energy surface with the dihedral angles of residues observed in the crystal structure. These observed data points fall in minimum energy wells that are associated with the highest probability rotameric states, as shown in Figure 6.

To evaluate the β -amino acid rotamer library for repacking of a complex tertiary structure, we repacked the crystallographic structure of the 8-helix bundle formed by the β -peptide Zwit1f²². The CHARMM27 force field was used to evaluate energies, and the computation conducted as described in methods. We first evaluated the recovery of amino acid rotamers while keeping the sequence held fixed; the χ_1 and χ_1/χ_2 recoveries from this treatment were to 95% and 65% respectively. This compares quite favorably with the accuracy achieved using the most advanced repacking algorithms for natural proteins⁵². This is a significant accomplishment because repacking of natural proteins employs sidechain rotamer libraries defined using crystallographic structures rather than computed energetics. We next addressed the extent to which the native sequence would be recapitulated when the amino acids identities were allowed to vary.

Figure 8 compares the sequence and structures of the starting Zwit1f sequence with that of the computed sequence. Because of asymmetry in the crystallographic structure, the sidechain repacking was conducted using a dimer as the asymmetric unit. Of eight Leu residues in the starting sequence, six were recovered as leucine in the crystallographically observed χ_1/χ_2 rotamers. The other two residues were converted to Ala, due to small clashes with other residues when these positions were held Leu. The four Phe residues were converted to three Tyr residues, whose hydroxyl groups were on the surface of the protein. This finding is consistent with the observation that *para*-iodo-phenylalanine can substitute for Phe in Zwit1f. The remaining starting hPhe was converted to hLeu, which shares a large hydrophobic sidechain. The remaining solvent-exposed residues were more variable, because the positions of amino acids in salt-bridging positions can interchange with retention of energetics. Nevertheless, the retention of χ_1 in the repacked structure was 75%. A further repacking in which the charge of the sidechains was retained from the original sequence, the χ_1 recovery increased to 83.3% and sequence recovery was 33.3% for the entire structure, in-line with the other modern repacking algorithms⁷⁴.

We note that the structure of zwit1f was not designed to form the observed octameric structure. Thus the resulting sequence of zwit1f is not necessarily optimal to maximize stability, and some of the changes to polar residues might in fact improve stability of the structure, although it is unlikely that the above-mentioned replacement of hleu for hala would be favorable.

Discussion

In the area of protein design, Handel and coworkers have shown both the challenges and limitations of using physics-based force fields.³² The challenge becomes substantially greater when a non-natural backbone is employed and there is no database to guide selection of low-energy rotamers as well as the avoidance of improbable high-energy rotamers. There is also the issue of setting energetic limits on the inclusion of rotamers in a library for non-natural backbones. For example, including high-energy rotamers will significantly slow convergence in any repacking algorithm, while the exclusion of too many rotamers decreases the complexity of the library. Furthermore, if the energy of a rotamer is overestimated, there is a danger of excluding this state, while if it is underestimated, a physically unreasonable structure would be obtained.

Therefore, before computing rotamer libraries for non-natural backbones, we began by asking what set of parameters in the CHARMM force-field would provide the best prediction of the relative abundance of rotamers as seen in the crystallographic structures of natural proteins. We computed the rotameric energies considering different van der Waals radii scaling factors and electrostatic treatments in conjunction with torsional interactions. The resulting energies were used to compute the Boltzmann-weighted probabilities for comparison with experimental frequency distributions. A good overall correlation was obtained, particularly in the α -helix.

Using the α -helix parameters, we developed rotameric distributions for several β AA scaffolds. In the context of β_{314} helices, our analysis suggests that the backbone details of the modeled β_{314} 3.0 helix correlate most closely with experimental evidence. Computed rotameric probabilities generated from the β_{314} scaffold are most similar to the β_{312} helix followed by the β_{310} helix and are dramatically different from their α AA counterparts.

Using this technique we have addressed how the additional methylene in the backbone influences the rotamer distributions. Rotamer frequencies are predominantly determined by interactions of atoms separate by three and four atomic bonds⁷⁵. In α AAs, the γ heavy atom of a side chain has ϕ -dependent steric interactions with the carbonyl carbon of preceding residue, placing restrictions on g^+ and g^- rotamers. Also, in α AAs, the γ heavy atom may have ψ -dependent unfavorable interactions with the adjacent backbone amide, affecting the g^+ and t rotamers.

The β_{314} helix g^+ rotamer has steric interactions with its own amide NH and carbonyl group which interacts with the C_2 group. The g^- rotamer has no unfavorable interactions with the local backbone in β_{314} helices and is therefore expected to be the most probable rotamer, followed by *trans* at moderate frequency, and g^+ at very low frequency.

This analysis is in good accord with the rotamer frequencies predicted by our molecular mechanics calculations. For instance, Met has χ_1 predicted frequencies of 77%, 22%, and 1% for the g^- , *trans*, and g^+ rotamers respectively. The numbers for Phe are 86%, 13%, and <1% and for Cys they are 70%, 30%, and 0%. As with α AAs, Val and Ile have only one viable rotamer since there are two γ heavy atoms to interact with the local backbone. This analysis predicts that Val (CG1 at χ_1 and CG2 at χ_1+120°) should exist only in the *trans* rotamer in β_{314} helices, which is what is predicted by the MM calculations. For Ile (CG1 at χ_1 and CG2 at χ_1-120°), only the g^- rotamer is expected and predicted to be observable. To confirm our hypothesis, probabilities were computed using a three-residue β_{314} scaffold. As expected, the *trans* and g^- rotamers are both accessible to valine (56.6% vs. 43.3%), while the g^+ rotamer is highly disfavored (0.09%).

There are additional interactions with the helical structure, outside of the local ϕ and ψ dependent distances. Good examples of this are the aromatic ring side chains of β^3 homophenylalanine, β^3 homotyrosine and β^3 homotryptophan. These rings demonstrate a preference for a single rotamer ($\chi_1, \chi_2 = g^-, g^+$), which affords them significant side-chain-backbone interaction against the β_{314} helix. If they were to occupy the same rotameric state in the α -helix, they would extend away from the backbone and forego favorable intra-helical interactions. However, these extended rotamers could prove fruitful in higher-order structures where side chains participate in side-chain-side-chain interactions.

In addition, since the backbone carbonyl points in the opposite direction in β_{314} helices compared to α -helices, they have a different geometry for the hydrogen bonding side chains Ser and Thr. This causes a significant increase in the g^+ rotamer of Ser relative to other side chain types, and a change in the rotameric preferences for threonine relative to Ile. Thr has a predicted frequency of 78% and 22% for the g^+ and g^- rotamers respectively, due to the

favorable hydrogen bond for the g^+ rotamer (Figure 7). By contrast, in the α -helix, Thr and Ser make an inter-residue hydrogen bond by assuming the -60° (g^-) rotamer.

While local atomic interactions are useful for development of rotamer libraries, critical evaluation of more complex sequence-dependent and tertiary structure-dependent effects require a full forcefield, as exemplified in the repacking of a protein-like β -peptide crystal structure²². The results presented here show the first attempt to overcome this challenging problem through the rigorous generation of a rotameric library, for use in the computational design of β -peptides.

Supplementary Material

Refer to Web version on PubMed Central for supplementary material.

Acknowledgments

The authors would like to thank the anonymous reviewers for their useful input. We thank NIH (GM54616 to W.F.D, P20 GM76222 and R01 GM84453 to R.L.D.) and DOE (DE-FG-02-04ER46156-A006 to W.F.D) for financial support.

References

1. Jiang L, Althoff EA, Clemente FR, Doyle L, Röthlisberger D, Zanghellini A, Gallaher JL, Betker JL, Tanaka F, Barbas CF, Hilvert D, Houk KN, Stoddard BL, Baker D. *Science*. 2008; 319:1387–91. [PubMed: 18323453]
2. Yin H, Slusky JS, Berger BW, Walters RS, Vilaire G, Litvinov RI, Lear JD, Caputo GA, Bennett JS, Degrado WF. *Science*. 2007; 315:1817–22. [PubMed: 17395823]
3. Slovic AM, Kono H, Lear JD, Saven JG, Degrado WF. *Proc Natl Acad Sci USA*. 2004; 101:1828–33. [PubMed: 14766985]
4. Goodman CM, Choi S, Shandler S, Degrado WF. *Nat Chem Biol*. 2007; 3:252–262. [PubMed: 17438550]
5. Lee BC, Zuckermann RN, Dill KA. *J Am Chem Soc*. 2005; 127:10999–1009. [PubMed: 16076207]
6. Armand P, Kirshenbaum K, Falicov A, Dunbrack RL Jr, Dill KA, Zuckermann RN, Cohen FE. *Fold Des*. 1997; 2:369–75. [PubMed: 9427011]
7. Kirshenbaum K, Zuckermann RN, Dill KA. *Curr Opin Struct Biol*. 1999; 9:530–5. [PubMed: 10449369]
8. Kirshenbaum K, Barron AE, Goldsmith RA, Armand P, Bradley EK, Truong KT, Dill KA, Cohen FE, Zuckermann RN. *Proc Natl Acad Sci U S A*. 1998; 95:4303–8. [PubMed: 9539732]
9. Horn T, Lee BC, Dill KA, Zuckermann RN. *Bioconjugate Chemistry*. 2004; 15:428–435. [PubMed: 15025542]
10. Lee BC, Chu TK, Dill KA, Zuckermann RN. *Journal of the American Chemical Society*. 2008; 130:8847–8855. [PubMed: 18597438]
11. Seebach D, Beck AK, Bierbaum DJ. *Chem Biodivers*. 2006; 1:1111–239. [PubMed: 17191902]
12. Rueping M, Mahajan YR, Jaun B, Seebach D. *Chemistry (Weinheim an der Bergstrasse, Germany)*. 2004; 10:1607–15.
13. Gademann K, Häne A, Rueping M, Jaun B, Seebach D. *Angew Chem Int Ed Engl*. 2003; 42:1534–7. [PubMed: 12698493]
14. Rueping M, Schreiber JV, Lelais G, Jaun B, Seebach D. *Helv Chim Acta*. 2002; 85:2577–2593.
15. Vaz E, Pomerantz WC, Geyer M, Gellman SH, Brunsveld L. *Chembiochem*. 2008; 9:2254–2259. [PubMed: 18756554]
16. Horne WS, Price JL, Gellman SH. *Proc Natl Acad Sci USA*. 2008; 105:9151–6. [PubMed: 18587049]
17. Sadowsky JD, Murray JK, Tomita Y, Gellman SH. *Chembiochem*. 2007

18. Lee MR, Raguse TL, Schinnerl M, Pomerantz WC, Wang X, Wipf P, Gellman SH. *Org Lett*. 2007; 9:1801–4. [PubMed: 17394351]
19. Choi SH, Guzei IA, Gellman SH. *J Am Chem Soc*. 2007; 129:13780–1. [PubMed: 17949002]
20. Porter EA, Wang X, Schmitt MA, Gellman SH. *Org Lett*. 2002; 4:3317–9. [PubMed: 12227778]
21. Cheng RP, Gellman SH, Degrado WF. *Proc Natl Acad Sci USA*. 2001; 101:3219–32.
22. Daniels DS, Petersson EJ, Qiu JX, Schepartz A. *J Am Chem Soc*. 2007; 129:1532–3. [PubMed: 17283998]
23. Hart SA, Bahadoor AB, Matthews EE, Qiu XJ, Schepartz A. *J Am Chem Soc*. 2003; 125:4022–3. [PubMed: 12670203]
24. Kritzer JA, Tirado-Rives J, Hart SA, Lear JD, Jorgensen WL, Schepartz A. *J Am Chem Soc*. 2005; 127:167–78. [PubMed: 15631466]
25. Petersson EJ, Schepartz A. *J Am Chem Soc*. 2008; 130:821–3. [PubMed: 18166055]
26. Qiu JX, Petersson EJ, Matthews EE, Schepartz A. *J Am Chem Soc*. 2006; 128:11338–9. [PubMed: 16939241]
27. Dunbrack RL. *Curr Opin Struct Biol*. 2002; 12:431–40. [PubMed: 12163064]
28. Brooks BR, Bruccoleri RE, Olafson DJ, States DJ, Swaminathan S, Karplus M. *Journal of Computational Chemistry*. 1983; 4:187–217.
29. Weiner SJ, Kollman PA, Case DA, Singh UC, Ghio C, Alagona G, Profeta S, Weiner P. *J Am Chem Soc*. 1984; 106:765–784.
30. Jorgensen WL, Tiradorives J. *Journal of the American Chemical Society*. 1988; 110:1657–1666.
31. Renfrew PD, Butterfoss GL, Kuhlman B. *Proteins*. 2008; 71:1637–46. [PubMed: 18076032]
32. Pokala N, Handel TM. *J Mol Biol*. 2005; 347:203–27. [PubMed: 15733929]
33. Dunbrack RL, Cohen FE. *Protein Sci*. 1997; 6:1661–81. [PubMed: 9260279]
34. Lovell SC, Word JM, Richardson JS, Richardson DC. *Proteins*. 2000; 40:389–408. [PubMed: 10861930]
35. Baker D, Degrado WF. *Curr Opin Struct Biol*. 1999; 9:485–6. [PubMed: 10449378]
36. Kuhlman B, Baker D. *Proc Natl Acad Sci USA*. 2000; 97:10383–8. [PubMed: 10984534]
37. Dahiyat BI, Mayo SL. *Protein Sci*. 1996; 5:895–903. [PubMed: 8732761]
38. Voigt CA, Gordon DB, Mayo SL. *J Mol Biol*. 2000; 299:789–803. [PubMed: 10835284]
39. Gordon DB, Hom GK, Mayo SL, Pierce NA. *Journal of computational chemistry*. 2003; 24:232–43. [PubMed: 12497602]
40. Kuhlman B, Dantas G, Ireton GC, Varani G, Stoddard BL, Baker D. *Science*. 2003; 302:1364–8. [PubMed: 14631033]
41. Levitt M. *J Mol Biol*. 1992; 226:507–33. [PubMed: 1640463]
42. Marti-Renom MA, Stuart AC, Fiser A, Sanchez R, Melo F, Sali A. *Annu Rev Biophys Biomol Struct*. 2000; 29:291–325. [PubMed: 10940251]
43. Ponder JW, Richards FM. *J Mol Biol*. 1987; 193:775–91. [PubMed: 2441069]
44. Jiang L, Kuhlman B, Kortemme T, Baker D. *Proteins*. 2005; 58:893–904. [PubMed: 15651050]
45. Mackerell AD. *Journal of computational chemistry*. 2004; 25:1584–604. [PubMed: 15264253]
46. Kortemme T, Baker D. *Proc Natl Acad Sci USA*. 2002; 99:14116–21. [PubMed: 12381794]
47. Lazaridis T, Karplus M. *Proteins*. 1999; 35:133–52. [PubMed: 10223287]
48. Lazaridis T, Karplus M. *Curr Opin Struct Biol*. 2000; 10:139–45. [PubMed: 10753811]
49. Senes A, Chadi DC, Law PB, Walters RF, Nanda V, Degrado WF. *J Mol Biol*. 2006; 366:436–48. [PubMed: 17174324]
50. Summa CM, Levitt M, Degrado WF. *J Mol Biol*. 2005; 352:986–1001. [PubMed: 16126228]
51. Zhou H, Zhou Y. *Protein Sci*. 2002; 11:2714–26. [PubMed: 12381853]
52. Krivov GG, Shapovalov MV, Dunbrack RL Jr. *Proteins*. 2009
53. Shapovalov MV, Dunbrack RL. *Proteins*. 2007; 66:279–303. [PubMed: 17080462]
54. Gopi HN, Ravindra G, Pal PP, Pattanaik P, Balaram H, Balaram P. *FEBS Lett*. 2003; 535:175–8. [PubMed: 12560099]
55. Saikumari YK, Ravindra G, Balaram P. *Protein Pept Lett*. 2006; 13:471–6. [PubMed: 16800800]

56. Hook DF, Bindschädler P, Mahajan YR, Sebesta R, Kast P, Seebach D. *Chem Biodivers.* 2006; 2:591–632. [PubMed: 17192006]
57. Hook DF, Gessier F, Noti C, Kast P, Seebach D. *Chembiochem.* 2004; 5:691–706. [PubMed: 15122642]
58. Wiegand H, Wirz B, Schweitzer A, Camenisch GP, Rodriguez Perez MI, Gross G, Woessner R, Voges R, Arvidsson PI, Frackenhohl J, Seebach D. *Biopharmaceutics & drug disposition.* 2002; 23:251–62.
59. Frackenhohl J, Arvidsson PI, Schreiber JV, Seebach D. *Chembiochem.* 2002; 2:445–55. [PubMed: 11828476]
60. Chatterjee S, Roy RS, Balaram P. *J Nat Prod.* 2007
61. Seebach D, Ciceri PE, Overhand M, Jaun B, Rigo D, Oberer L, Hommel U, Amstutz R, Widmer H. *Helvetica Chimica Acta.* 1996; 79:2043–2066.
62. Guarracino DA, Chiang HR, Banks TN, Lear JD, Hodsdon ME, Schepartz A. *Org Lett.* 2006; 8:807–10. [PubMed: 16494446]
63. Kritzer JA, Lear JD, Hodsdon ME, Schepartz A. *J Am Chem Soc.* 2004; 126:9468–9. [PubMed: 15291512]
64. Petersson EJ, Craig CJ, Daniels DS, Qiu JX, Schepartz A. *J Am Chem Soc.* 2007; 129:5344–5. [PubMed: 17425318]
65. Metropolis N, Ulam S. *Journal Of The American Statistical Association.* 1949; 44:335–341. [PubMed: 18139350]
66. Pierce NA, Spriet JA, Desmet J, Mayo SL. *Journal of computational chemistry.* 2000; 21:999–1009.
67. Dieckmann GR, Degrado WF. *Curr Opin Struct Biol.* 1997; 7:486–94. [PubMed: 9266169]
68. Case DA, Cheatham TE 3rd, Darden T, Gohlke H, Luo R, Merz KM Jr, Onufriev A, Simmerling C, Wang B, Woods RJ. *J Comput Chem.* 2005; 26:1668–88. [PubMed: 16200636]
69. MacKerell AD Jr, et al. *J Phys Chem.* 1998; B102:3586–3616.
70. Wang G, Dunbrack RL Jr. *Bioinformatics.* 2003; 19:1589–91. [PubMed: 12912846]
71. Wang G, Dunbrack RL Jr. *Nucleic Acids Res.* 2005; 33:W94–8. [PubMed: 15980589]
72. Frishman D, Argos P. *Proteins.* 1995; 23:566–79. [PubMed: 8749853]
73. Bode KA, Applequist J. *Macromolecules.* 1997; 30:2144–2150.
74. Barth P, Schonbrun J, Baker D. *Proc Natl Acad Sci USA.* 2007
75. Dunbrack RL, Karplus M. *Nature Structural Biology.* 1994
76. Gunther R, Hofmann HJ, Kuczera K. *The Journal of Physical Chemistry B.* 2001; 105:5559–5567.
77. Applequist J, Bode KA, Appella DH, Christianson LA, Gellman SH. *J Am Chem Soc.* 1998; 120:4891–4892.

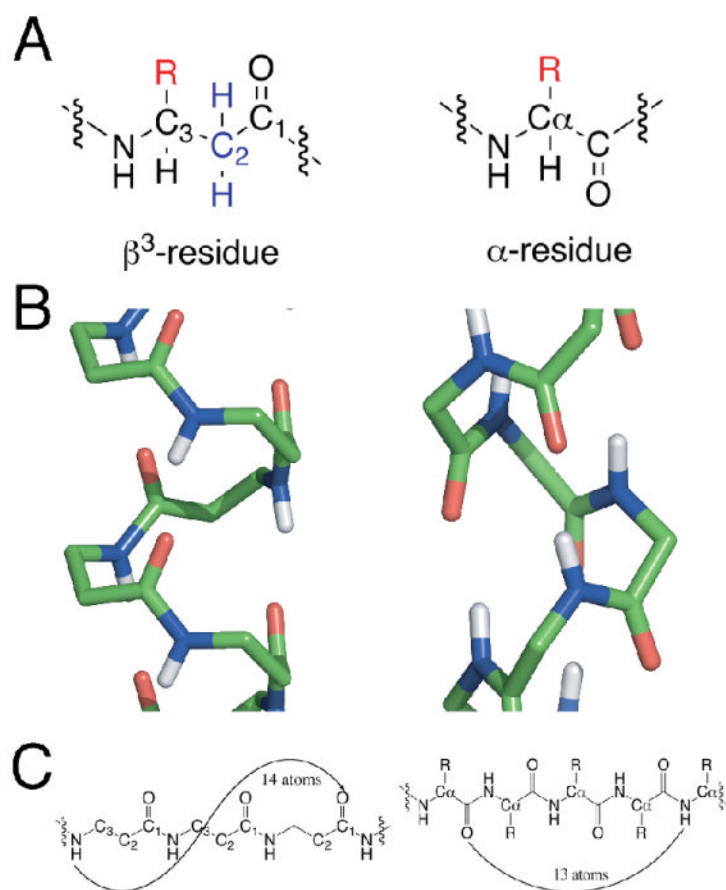


Figure 1. **A.** backbone units for a β^3 AA (left) and α AA (right) peptides. **B.** Wireframe of β^3_{14} helix (left) and α -helix (right, non-polar hydrogens omitted for clarity, atoms colored by elemental type: carbon (green), oxygen (red), nitrogen (blue), hydrogen (white)). **C.** Hydrogen-bonding patterns for each structural motif. Note the direction of the carbonyl group in each subunit. The carbonyl points in the opposite direction in some β^3 AA peptide helices from that in the α -helix.

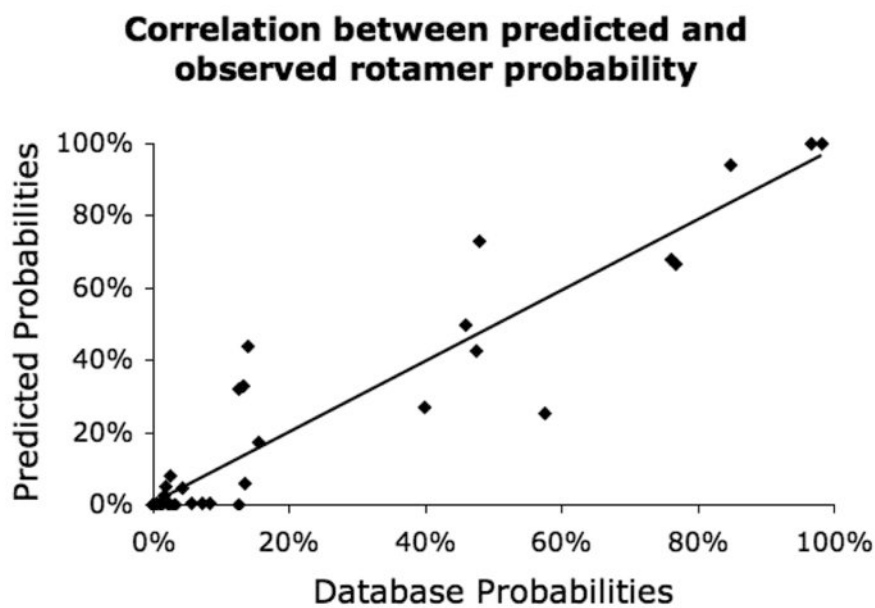


Figure 2. Comparison between predicted and observed rotamer probabilities of α -helical rotamers (ILVFYST), $R=0.946$. Optimized Boltzmann temperature of 450 K shown. Correlation of remaining α -helical rotamers (CMWDENQHKR) which include extended sidechains, $R = 0.568$.

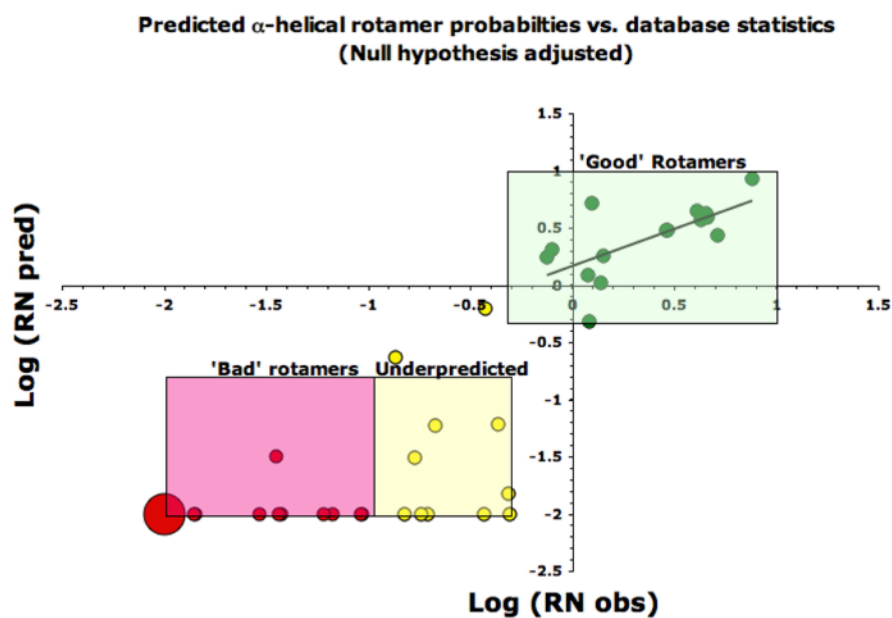


Figure 3.

Log (Predicted probabilities / Null probability) vs. Log (Database probabilities / Null probability) for the α -helix (α AA residues VILSTFY, Boltzmann temperature 450° for predicted probabilities). Green points are 'good' predicted rotamers, $RN_{obs} > 0.5$ (the majority of which are over-represented when compared to the null hypothesis expectations); yellow points are 'underpredicted' rotamers, $RN_{obs} = 0.1 - 0.5$; red points are rotamers that are ranked poorly, $RN_{obs} < 0.1$ (the point at -2, -2 represents 11 unique rotamer evaluations and is sized accordingly). Predicted probabilities with $RN_{obs} < 0.01$ are plotted at $\log(RN_{obs}) = -2$.

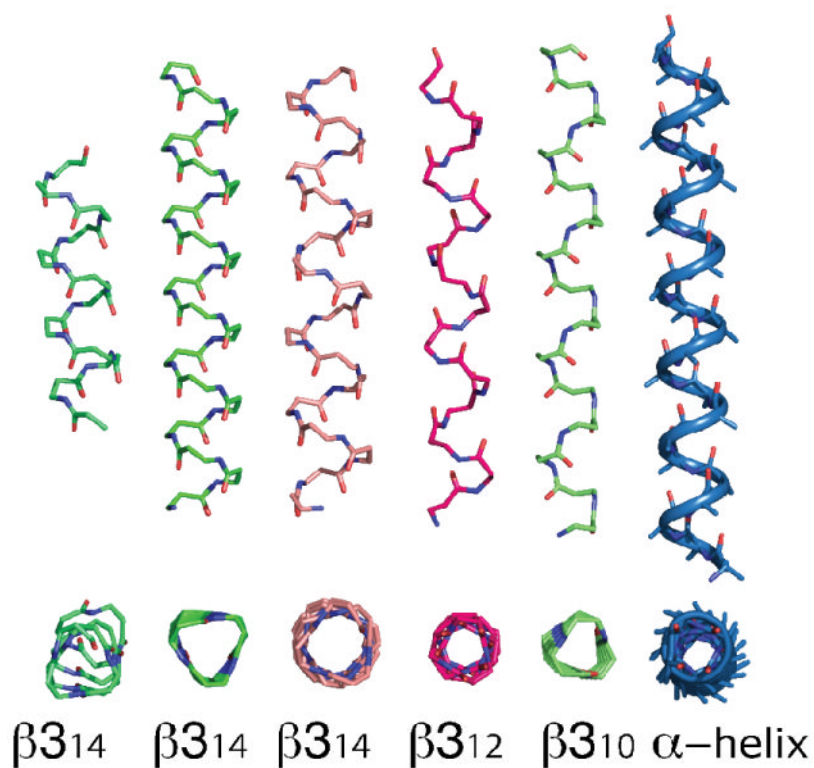


Figure 4.

Various β -peptide helix motifs. From left to right: $\beta 3_{14}$ helix from crystal structure²², energy minimized $\beta 3_{14}$ 3.0 helix, $\beta 3_{14}$ QM minimized $\beta 3_{14}$ helix⁷³, $\beta 3_{12}$ helix⁷⁷, $\beta 3_{10}$ helix⁷⁶. Note the direction of the carbonyl groups in the $\beta 3_{12}$ helix. Side-chain and hydrogen atoms omitted for clarity. α -helix shown for comparison at right.

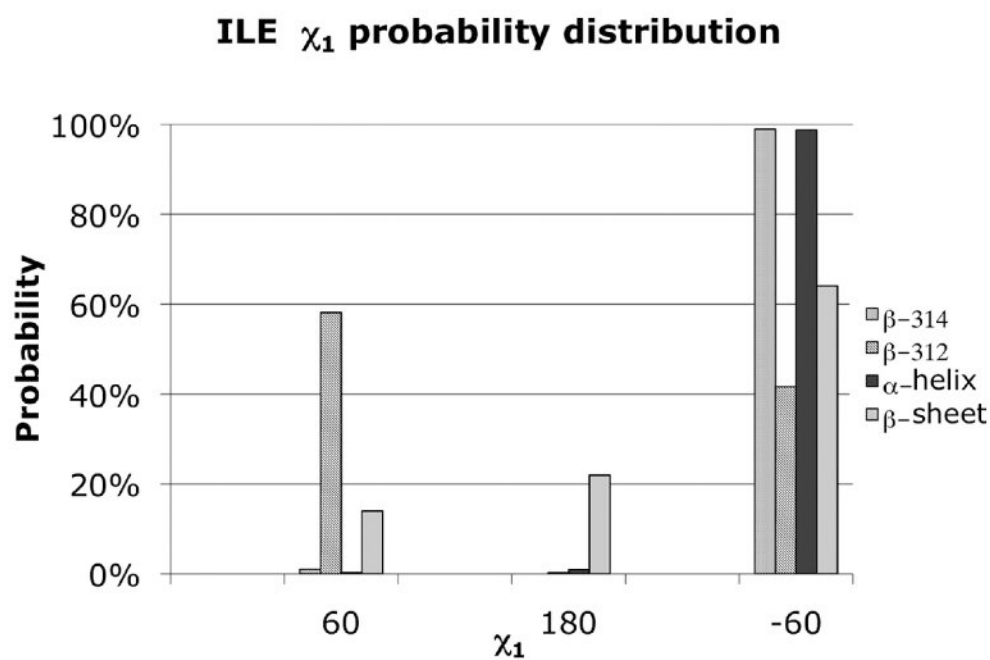


Figure 5. χ_1 rotamer distributions for the isoleucine side chain in the context of the α -helix, β -sheet, β_{314} helix, and the β_{312} helix. α -helix and β -sheet rotameric preference derived from a binned backbone dependent rotamer library for each secondary structure.

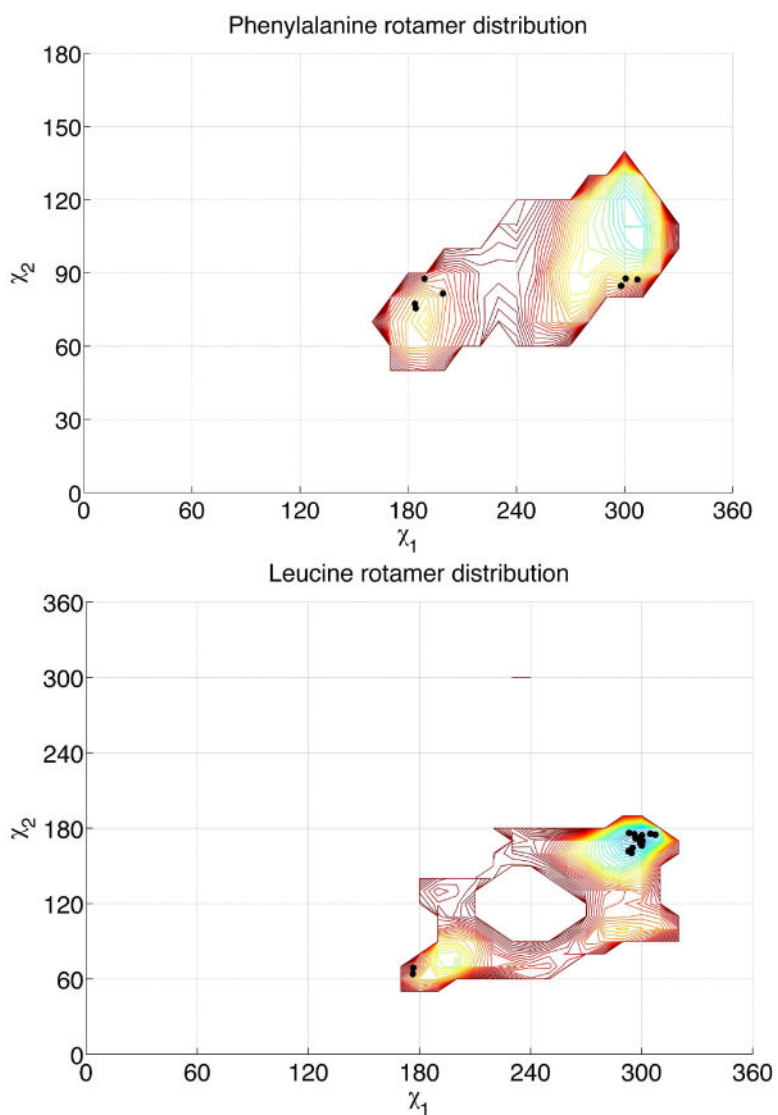
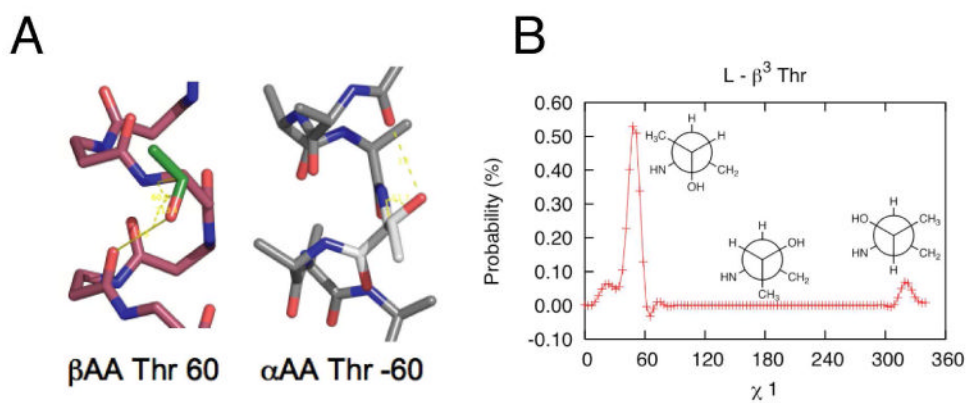


Figure 6. Contour plot of χ_1/χ_2 dependent rotamer probabilities for β^3 homoleucine and β^3 homophenylalanine in the $\beta 3_{14}$ 3.0 rotamer library. The negative logs of rotamer probabilities are indicated via the contour map, in $10^\circ \times 10^\circ$ bins. Experimental rotamers from the crystal structure²² are overlaid (black dots).

**Figure 7.**

A. β_{314} helix and α -helix each with the threonine side chain. The preferred rotamer for each scaffold is shown, g^+ for β_{314} helix and g^- for the α -helix. **B.** Rotameric probabilities for β^3 substituted threonine. (Newman projections shown in inset)

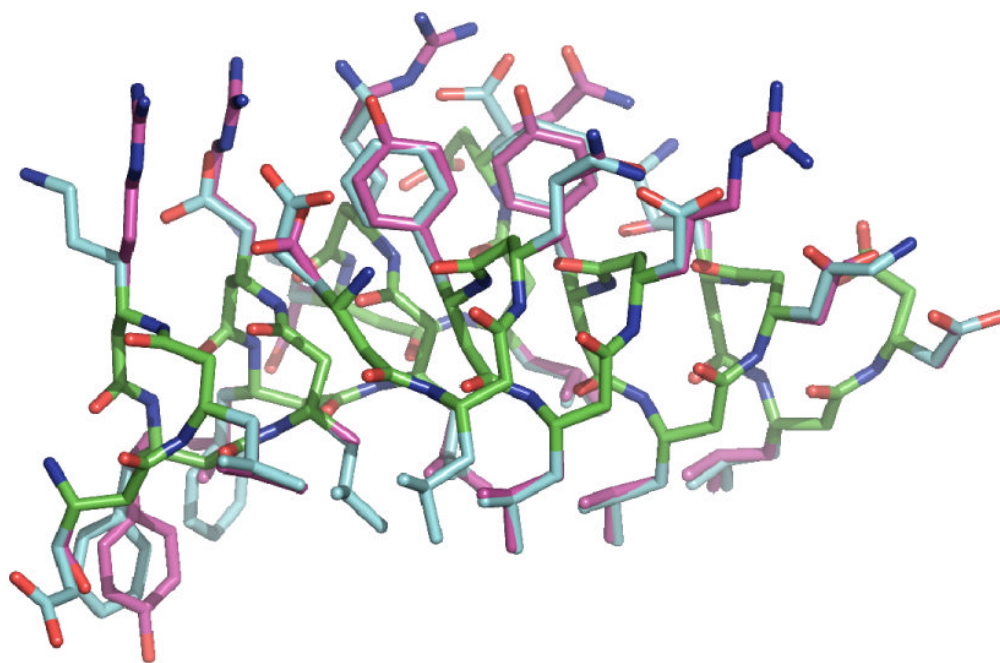


Figure 8.

Comparison between experimental and computationally selected rotamers for the Zwi1f β -peptide scaffold using *denovo* β -peptide rotamer library. β_{314} backbone carbon atoms are shown in green, crystallographic sidechain carbon atoms shown in blue and computationally selected rotamers shown in magenta. Oxygen and nitrogen atoms are shown in red and blue respectively. Note the good identity and rotameric agreement between 6 of the 8 core β -homoLeucines and 3 of 4 aromatic residues.

Table 1
Performance of RN (odds ratio) metric in rotamer library design for α AA residues

		VDW=90% $\epsilon = 20$	VDW=90% $\epsilon = 10$	VDW=95% $\epsilon = 20$	VDW=95% $\epsilon = 10$	VDW=100% $\epsilon = 20$	VDW=100% $\epsilon = 10$
False Negatives							
α -Helix	3	6	5	7	5	8	5
β -Sheet	2	3	4	4	5	5	1
False Positives							
α -Helix	0	0	0	0	0	0	0
β -Sheet	5	5	1	1	1	1	1

VDW is the vdW radii-scaling factor, ϵ is the distance-dependent dielectric constant. False positives and false negatives are listed when compared to knowledge-based α AA rotamer library of 48 rotameric states. 'Good' (positive) and 'bad' (negative) rotamers have odds ratio RN_{obs} of >0.1 and <0.01 respectively.

Table 2
The dependence of rotamer probabilities on the backbone conformation of the $\beta_{3,14}$ helix

Helix	ϕ^{\ddagger}	ψ^{\ddagger}	θ^{\ddagger}	$\beta_{3,14}$ Crystal (R)	$\beta_{3,14}$ Literature (R)	$\beta_{3,14}$ 3.0 Idealized (R)
$\beta_{3,14}$ Crystal ²²	54.2	-135.1	-132.3	1		
$\beta_{3,14}$ Literature ⁷³	60.0	-134.3	-139.9	0.889	1	
$\beta_{3,14}$ 3.0 Idealized	55.0	-123.4	-139.3	0.929	0.824	1

[†] Backbone torsional angles in degrees.

[‡] The correlation coefficients in the three rightmost columns are based on the linear regression of the log of the calculated rotamer probabilities of residues ILVFYSTD between the helical backbones under consideration.

Table 3

Rotamer library similarity for different scaffolds

Helix	ϕ^{\ddagger}	ψ^{\ddagger}	θ^{\ddagger}	Crystal (R)	$\ddagger\ddagger$	Available Rotameric States	Polarity	Ref
3 ₁₀	64	59	75	0.771		14	N	76
3 ₁₂	-90	90	-110	0.932		19	C	24
α Helix (α AA)	-57	-47	-	0.634		19	C	33
β -sheet (α AA)	-119	113	-	0.667		18	C	

Available rotameric states reflect the total size of the rotamer library for amino acid side chains (ILVFYSTD) after excluding rotamers with less than a 5% probability.

\ddagger Backbone torsional angles in degrees.

$\ddagger\ddagger$ Correlation between the log of the null hypothesis adjusted probabilities of the library based on the average β 314 crystal backbone and libraries for the same amino acids derived from the respective scaffolds.²² The α -helix and β -sheet data are derived from a backbone dependent rotamer library database determined using the indicated backbone dihedral angles.