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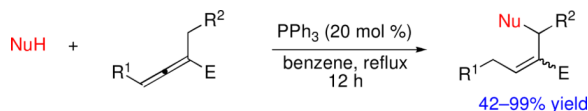
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## Phosphine-Catalyzed $\beta'$ -Umpolung Addition of Nucleophiles to Activated $\alpha$ -Alkyl Allenes

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### Abstract



NuH = ROH, RR'NH, CH<sub>2</sub>(CN)<sub>2</sub>, RCH(CN)CO<sub>2</sub>Et, PhSH  
 R<sup>1</sup> = H, Me; R<sup>2</sup> = H, Ph, 4-pentenyl, CO<sub>2</sub>Et; E = CO<sub>2</sub>Et, CN

Highly functionalized alkenes can be prepared through phosphine-catalyzed  $\beta'$ -umpolung additions of nucleophiles (carbon-, oxygen-, nitrogen-, and sulfur-centered) to activated  $\alpha$ -disubstituted allenenes, providing many potentially useful synthetic intermediates in good to excellent yields, often with high levels of stereoselectivity for the product olefin geometry. Various substitution patterns around the allene are compatible with the process, showcasing the synthetic utility of allenenes under the conditions of nucleophilic phosphine catalysis.

Nucleophilic phosphine catalysis is firmly established as a reliable platform for a variety of transformations involving activated allenenes as starting materials.<sup>1</sup> Within this field,  $\gamma$ -umpolung addition of nucleophiles to activated allenenes and acetylenes allows the formation of a myriad of substrates. Since the first report by Trost,<sup>2</sup> there has been increasing interest in phosphine-catalyzed  $\gamma$ -umpolung additions of pronucleophiles to electron-deficient alkynes and allenenes (Scheme 1; eq 1).<sup>3</sup> Combining the mechanistic insights gained from studies of the  $\gamma$ -umpolung additions and nucleophilic phosphine-catalyzed reactions of  $\alpha$ -substituted allenenoates developed in our laboratory<sup>4</sup> and by others,<sup>5</sup> we envisaged the possibility of a  $\beta'$ -umpolung addition to  $\alpha$ -alkyl allenenoates (Scheme 1; eq 2). This process facilitates functionalization of the seemingly unactivated  $\beta'$ -C–H bond, allowing the generation of unique, highly versatile olefinic products. The observed activation of the  $\beta'$ -carbon atom of  $\alpha$ -alkyl allenenoates stems from the equilibrium between the  $\beta$ -phosphonium dienolate and the vinylogous ylide (see the mechanistic discussion below).<sup>4d,f,g</sup>

We began our studies by treating ethyl  $\alpha$ -methyl allenenoate with a variety of substituted phenols and catalytic PPh<sub>3</sub> in benzene under reflux (Table 1). The reaction with phenol proceeded smoothly to produce the  $\beta'$ -umpolung adduct **1a** in quantitative yield as a single geometrical isomer (entry 1). To our delight, we detected no  $\gamma$ -umpolung addition product. Notably, unlike the requirements of the  $\gamma$ -umpolung addition, no basic or acidic/basic additives were needed for the  $\beta'$ -umpolung reaction,<sup>2,3</sup> which provided an expedient route to the allylphenyl ether **1a**, a known intermediate in the functionalized coumarin synthesis.<sup>6</sup> We also obtained high yields from reactions of phenols bearing electron-donating methyl or

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**Supporting Information Available** Experimental procedures and characterization data for new compounds (PDF). This information is available free of charge via the Internet at <http://pubs.acs.org>.

methoxy substituents (entries 2 and 3). The allylaryl ethers **1d–1h** derived from phenols with electron-withdrawing substituents were, however, isolated in diminished yields (entries 4–8), the result of a competitive fragmentation, producing ethyl 2-methylenebut-3-enoate, which underwent known dimerization through a Diels–Alder reaction.<sup>7</sup> The presence of carboxylic ester and protected amino groups on the phenols was tolerated well (entries 9 and 10). Intriguingly, a phenol featuring a Boc-protected amino substituent produced a mixture of *E* and *Z* isomers (entry 10). Ortho and meta substitutions on the phenol ring were also well tolerated (entries 11–13), although we observed a slight decrease in stereoselectivity in the case of *o*-cresol (entry 12). Extreme steric hindrance around the phenolic nucleophile diminished the reaction efficiency. For instance, 2,6-dimethylphenol provided a relatively low yield (entry 14) and 2,6-di-*tert*-butylphenol gave no reaction (entry 15).

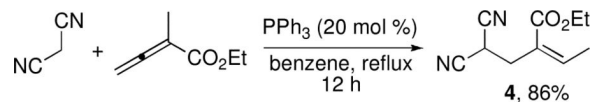
Further substitution at the  $\beta'$ -position of the allene produced *Z*-olefins, presumably because of steric clash between the methyl and alkyl groups (entries 16–18). Furthermore, the  $\beta'$ -substitution protected the arylallyl ether **1q** from undergoing fragmentation to produce the diene,<sup>7</sup> providing **1q** in 97% isolated yield (entry 17; cf. entry 8, 73%). Allenitrile also worked well as the electrophile: the arylallyl ether **1r** was obtained in excellent yield (entry 18).<sup>8</sup>

Because our initial investigations with phenol pronucleophiles were highly successful, we expanded our exploration of the  $\beta'$ -umpolung additions to amine-based pronucleophiles (Table 2). *N*-Phenyl-*p*-toluenesulfonamide gave its expected product in excellent yield, albeit as a 2.5:1 mixture of *E* and *Z* olefins (entry 1). *N*-Benzyl-*p*-toluenesulfonamide and *tert*-butyl *N*-tosylcarbamate pronucleophiles were also viable substrates (entries 2 and 3). *p*-Toluenesulfonamide itself resulted in lower product yield and selectivity, favoring the *Z*-olefin geometry (entry 4). The use of *tert*-butyl *N*-phenylcarbamate resulted in no reaction product, due to the diminished acidity of the pronucleophile (entry 5).<sup>3g</sup> The allylic amine product derived from phthalimide was obtained in 71% yield as a single *E* isomer (entry 6). *p*-Toluenesulfonylhydrazine gave its  $\beta$ -amino acid product in moderate yield (entry 7). *N*-Tosyl-protected ethyl phenylalanine was also a suitable pronucleophile, producing the amide **2h** in excellent yield (entry 8). Employing protected amines provides access to  $\alpha$ -substituted  $\beta$ -amino esters, which have great utility in  $\beta$ -peptide chemistry;<sup>9</sup> in addition,  $\alpha$ -Methylene- $\beta$ -alanine is a naturally occurring herbicidal amino acid.<sup>10</sup>

Carboxylic acids also turned out to be excellent pronucleophiles for this  $\beta'$ -umpolung reaction and worked well with a variety of diversely substituted allenoates (Table 3). With ethyl  $\alpha$ -methyl allenoate, both acetic acid and benzoic acid provided their corresponding acylated allylic alcohols in excellent yields, as *E* isomers exclusively (entries 1 and 2).<sup>11</sup> Complementary to the N-terminus addition of phenylalanine (Table 2, entry 8), the C-terminus addition of *N*-*tert*-butoxycarbonyl phenylalanine was just as efficient (Table 3, entry 3). The mild and neutral conditions ensured that no epimerization occurred during the reaction.<sup>12</sup> Benzoic acid reacted as anticipated with the  $\gamma$ -methyl-substituted allenoate (entry 4), but the excessively large  $\gamma$ -*tert*-butyl substituent inhibited the reaction (entry 5). Ethyl  $\alpha$ -(5-hexenyl)allenoate was compatible with the transformation (entry 6). In accordance with our observations regarding phenol pronucleophiles, the  $\alpha$ -benzyl allenoate produced its *Z*-olefin in excellent yield (entry 7). Diethyl 2-vinylidenesuccinate also proved to be a viable substrate for the  $\beta'$ -addition of benzoic acid, producing a mixture of its two olefinic isomers in excellent yield (entry 8).

We further expanded the reaction scope to consider carbon-centered pronucleophiles. The reaction of malonitrile with ethyl  $\alpha$ -methyl allenoate gave a *Z*-configured olefin product, which structurally resembles a mixed Rauhut–Currier adduct (**4**, eq 3), in 86% yield.<sup>13</sup> Although dimethyl malonate and ethyl cyanoacetate did not undergo corresponding  $\beta'$ -

umpolung additions,  $\alpha$ -alkylcyanoacetates did.<sup>14</sup> The actual pronucleophiles employed were alkylidencyanoacetates, which underwent  $\gamma$ -deprotonation and addition at the  $\alpha$ -carbon atom (Table 4). We obtained the resulting dienes as mixtures of *E* and *Z*  $\alpha,\beta$ -unsaturated enoates, with the  $\beta,\gamma$ -olefin unit from the cyanoacetate existing exclusively in *E* form. Although we obtained these products in relatively modest yields, the alkylcyanoacetate pronucleophiles provided access to highly functionalized all-carbon diene substrates.

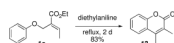


eq 3

Furfuryl alcohol, oxime, and thiol functionalities were compatible with the  $\beta'$ -umpolung addition (Table 5). While most aliphatic alcohols did not participate in the umpolung reaction, due to their high values of  $pK_a$ ,<sup>15</sup> the relatively low  $pK_a$  of furfuryl alcohol (9.55) made it a prime candidate for the  $\beta'$ -addition reaction (entry 1). Oximes and thiols also possess suitable values of  $pK_a$  for this transformation. Indeed, (*E*)-benzaldehyde oxime ( $pK_a$  11.3) and benzenethiol ( $pK_a$  7) underwent the reaction smoothly (entries 2 and 3). *O*-Allyl oximes are useful synthetic precursors for allylation of activated methylene compounds<sup>16</sup> and [2,3]-sigmatropic N,O rearrangements.<sup>17</sup> When treated with palladium(II), oxime *O*-allyl ether undergoes formal [2,3]-sigmatropic rearrangement to form *N*-allyl nitron, which can then be engaged in a dipolar cycloaddition reaction in situ.<sup>18</sup> *O*-Allyl oximes have also been converted in one step into pyrroles under iridium catalysis.<sup>19</sup>

Scheme 2 presents a plausible mechanism for this reaction. Phosphine addition to the allenolate gives the phosphonium dienolate **9**. A well-established equilibrium between the phosphonium enolate **9** and the vinylogous phosphorous ylide **10** is facilitated through proton transfer steps.<sup>4d,f,g</sup> Deprotonation of the pronucleophile by the ylide **11** forms the phosphonium enoate electrophile **12**. Conjugate addition of the anionic nucleophile to the enoate and subsequent  $\beta$ -elimination of the phosphine yields the observed  $\beta'$ -umpolung addition product.

As mentioned previously, many of these substrates are synthetically useful. For example, we performed a further transformation of the allylphenyl ether **1a**. Upon heating in diethylaniline under reflux, the allylaryl ether **1a** underwent a facile Claisen rearrangement, lactonization, and isomerization cascade to produce 3,4-dimethylcoumarin (**13**, eq 4).<sup>20</sup>



eq 4

This paper reports the first examples of  $\beta'$ -umpolung additions of nucleophiles (carbon-, oxygen-, nitrogen-, and sulfur-centered) to activated allenes, producing many useful synthetic intermediates in good to excellent yields—often with high levels of selectivity for the olefin geometry. Whereas activated  $\alpha$ -monosubstituted allenes undergo the well-established  $\gamma$ -umpolung addition, activated  $\alpha$ -disubstituted allenes undergo  $\beta'$ -umpolung addition exclusively, underlining the synthetic utility of this versatile class of molecules. As a showcase of the power of this reported reaction, we performed a one-step cascade process from an allylphenyl ether intermediate to a coumarin product. We are currently exploring applications of this methodology to other syntheses of small molecules.

## Supplementary Material

Refer to Web version on PubMed Central for supplementary material.

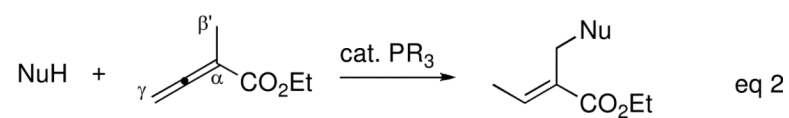
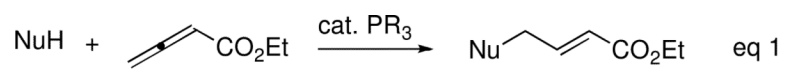
## Acknowledgments

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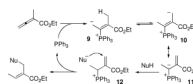
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**Scheme 1.**  
 $\gamma$ -Addition and  $\beta'$ -Umpolung Addition of Allenes



**Scheme 2.**  
Mechanism of the  $\beta'$ -Umpolung Addition

Table 1

Addition of Phenols to Activated Allenes<sup>a</sup>


entry	R <sup>1</sup>	R <sup>2</sup>	E	1	yield (%) <sup>b</sup>	E:Z <sup>c</sup>
1	H	H	CO <sub>2</sub> Et	<b>1a</b>	>99	<i>E</i> only
2	4-Me	H	CO <sub>2</sub> Et	<b>1b</b>	97	<i>E</i> only
3	4-OMe	H	CO <sub>2</sub> Et	<b>1c</b>	92	<i>E</i> only
4	4-I	H	CO <sub>2</sub> Et	<b>1d</b>	69	<i>E</i> only
5	4-Br	H	CO <sub>2</sub> Et	<b>1e</b>	72	<i>E</i> only
6	4-Cl	H	CO <sub>2</sub> Et	<b>1f</b>	62	<i>E</i> only
7	4-F	H	CO <sub>2</sub> Et	<b>1g</b>	78	<i>E</i> only
8	4-CF <sub>3</sub>	H	CO <sub>2</sub> Et	<b>1h</b>	73	<i>E</i> only
9	4-CO <sub>2</sub> Et	H	CO <sub>2</sub> Et	<b>1i</b>	90	<i>E</i> only
10	4-NHBoc	H	CO <sub>2</sub> Et	<b>1j</b>	89	4:1
11	3-Me	H	CO <sub>2</sub> Et	<b>1k</b>	95	<i>E</i> only
12	2-Me	H	CO <sub>2</sub> Et	<b>1l</b>	96	20:1
13	2-I	H	CO <sub>2</sub> Et	<b>1m</b>	91	<i>E</i> only
14	2,6-di-Me	H	CO <sub>2</sub> Et	<b>1n</b>	66	<i>E</i> only
15	2,6-di- <i>t</i> -Bu	H	CO <sub>2</sub> Et	<b>1o</b>	0	N/A
16	4-OMe	Ph	CO <sub>2</sub> Et	<b>1p</b>	89	<i>Z</i> only
17	4-CF <sub>3</sub>	Ph	CO <sub>2</sub> Et	<b>1q</b>	97	<i>Z</i> only
18	4-CF <sub>3</sub>	Ph	CN	<b>1r</b>	88	<i>Z</i> only

<sup>a</sup>Reactions were performed using 1.1 equiv of allene.<sup>b</sup>Isolated yield.<sup>c</sup>Determined through <sup>1</sup>H NMR spectroscopic analysis.



Table 2

Addition of Amines to  $\alpha$ -Methyl Allenolate<sup>a</sup>

entry	R <sup>1</sup>	R <sup>2</sup>	yield (%) <sup>b</sup>	2	E:Z <sup>c</sup>
1	Ph	<i>p</i> -Ts	94	<b>2a</b>	2.5:1
2	Bn	<i>p</i> -Ts	94	<b>2b</b>	2.5:1
3	Boc	<i>p</i> -Ts	83	<b>2c</b>	2:1
4	H	<i>p</i> -Ts	53	<b>2d</b>	1:1.75
5	Ph	Boc	0	<b>2e</b>	N/A
6	phthalimide		71	<b>2f</b>	<i>E</i> only
7	TsHN-NH <sub>2</sub>		57	<b>2g</b>	1.5:1
8	TsHN	CO <sub>2</sub> Et Ph	92	<b>2h</b>	1:1.2

<sup>a</sup>Reactions were performed using 1.1 equiv of the allenolate.<sup>b</sup>Isolated yield.<sup>c</sup>Determined through <sup>1</sup>H NMR spectroscopic analysis.

Table 3

Addition of Carboxylic Acids to Allenates<sup>a</sup>

entry	R <sup>1</sup>	R <sup>2</sup>	R <sup>3</sup>	3	yield (%) <sup>b</sup>	E/Z <sup>c</sup>
1	Me	H	H	<b>3a</b>	97	<i>E</i> only
2	Ph	H	H	<b>3b</b>	95	<i>E</i> only
3	BoCHN	H	H	<b>3c</b>	96	<i>E</i> only
4	Ph	Me	H	<b>3d</b>	90	<i>E</i> only
5	Ph	<i>t</i> -Bu	H	<b>3e</b>	0	N/A
6	Ph	H	4-pentenyl	<b>3f</b>	85	<i>E</i> only
7	Ph	H	Ph	<b>3g</b>	95	<i>Z</i> only
8	Ph	H	CO <sub>2</sub> Et	<b>3h</b>	96	1:1.5

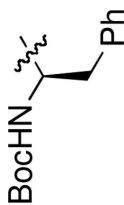

<sup>a</sup>Reactions were performed using 1.1 equiv of allenolate.<sup>b</sup>Isolated yield.<sup>c</sup>Determined through <sup>1</sup>H NMR spectroscopic analysis.

Table 4

Addition of Cyanoacetates to Activated Allenes<sup>a</sup>


entry	R <sup>1</sup>	R <sup>2</sup>	5	yield (%) <sup>b</sup>	E:Z <sup>c</sup>
1	<i>n</i> -Bu	H	<b>5a</b>	57	1:5
2	Me	H	<b>5b</b>	42	1:5
3	Me	Me	<b>5c</b>	79	1:2 <sup>d</sup>

<sup>a</sup>Reactions were performed using 1.1 equiv of the allenolate.<sup>b</sup>Isolated yield.<sup>c</sup>Determined through <sup>1</sup>H NMR spectroscopic analysis.<sup>d</sup>Isomers were separated and characterized; details provided in the Supporting Information.

**Table 5**Addition of Pronucleophiles to Ethyl  $\alpha$ -Methylallenoate<sup>a</sup>

entry	NuH	yield (%) <sup>b</sup>	product	E/Z <sup>c</sup>
1	furfuryl alcohol	88	<b>6</b>	1:1
2	( <i>E</i> )-benzaldehyde oxime	77	<b>7</b>	4:1
3	PhSH	65	<b>8</b>	5:1

<sup>a</sup>Reactions were performed using 1.1 equiv of allenoate.<sup>b</sup>Isolated yield.<sup>c</sup>Determined through <sup>1</sup>H NMR spectroscopic analysis.