



Published in final edited form as:

J Am Chem Soc. 2011 March 16; 133(10): 3276–3279. doi:10.1021/ja110748s.

Spectral Diffusion at the Water/Lipid Interface Revealed by Two-Dimensional Fourth-Order Optical Spectroscopy: A Classical Simulation Study

Yuki Nagata and Shaul Mukamel

Department of Chemistry, University of California, Irvine, Irvine, California, 92617, USA

Abstract

Using a classical simulation protocol for nonlinear optical signals, we predict the two-dimensional (2D) spectra of water near a monolayer of [1,2-dimyristoyl-*sn*-glycero-3-phosphatidylcholine] (DMPC) generated by three infrared and one visible probe pulses. Sum frequency generation 1D spectra show two peaks of the OH stretch representing two environments of near-bulk water non-adjacent to DMPC and top-layer water adjacent to DMPC. These peaks create a 2D pattern in the fourth-order signal. The asymmetric cross-peak pattern with respect to the diagonal line suggests coherence transfer from the higher to the lower frequency modes. The nodal lines in the imaginary part of the 2D spectrum show that the near-bulk water has fast spectral diffusion, resembling bulk water despite the orientation by the strong electrostatic field of DMPC. The top-layer water has slower spectral diffusion.

Interfacial water plays an essential role in the stabilization and functionality of membranes and of proteins attached to membranes. Since the cell membrane of living organisms consists primarily of phospholipid bilayers, the water/phospholipid monolayer interface is a good model system for studying the water-membrane interactions. A $\sim 10\text{\AA}$ layer of water is oriented by the strong electrostatic interactions with the lipid.¹ Vibrational spectra of interfacial water are distinct from bulk water. A single $\sim 3400\text{cm}^{-1}$ OH stretch in the infrared (IR) spectra of bulk water is split into two peaks at $3100\text{--}3300\text{cm}^{-1}$ (*L*, lower) and $3300\text{--}3500\text{cm}^{-1}$ (*H*, higher) in the second-order sum-frequency generation (SFG) spectra.² Time-resolved SFG experiments revealed that the dynamics was different as well; the *H*-peak exhibited longer relaxation time ($\sim 600\text{fs}$) than bulk ($\sim 200\text{fs}$).³ Our recent molecular dynamics (MD) simulation of the SFG spectra at the water/[1,2-dimyristoyl-*sn*-glycero-3-phosphatidylcholine] (DMPC) interface has shown that the *L*- and *H*-peaks originate from the OH stretches of near-bulk water non-adjacent to DMPC and the top-layer water adjacent to DMPC, respectively.⁴ An important open issue is how fast do the configurations of the interfacial water vary. A good experimentally accessible indicator for ultrafast configuration change is provided by the fluctuations of frequencies (spectral diffusion). Spectral diffusion probes the surrounding molecular structures and local dynamics of water at lipid and biological interfaces. Even though both the top-layer and the near-bulk water are oriented to the DMPC layer by the strong electrostatic interactions with the hydrophilic part of DMPC, it is not clear whether their dynamics is heterogeneous.

nagatay@uci.edu, smukamel@uci.edu.

SUPPORTING INFORMATION AVAILABLE: The derivation of the classical response function for the fourth-order signal, the protocols for the MD simulation and the calculation of the optical response, and the simulated SFG spectra are included in the supporting information. This material is available free of charge via the internet at <http://pubs.acs.org>.

Water heterogeneity near DMPC has been examined by two-color pump-probe IR and two-dimensional (2D) IR experiments in low-hydrated DMPC⁵ and DNA;⁶ water attached to DMPC was found to be more heterogeneous than bulk. Since low-hydrated DMPC does not have bulk water, the top-layer water adjacent to DMPC is dominant and the near-bulk water connecting the top-layer water with bulk is not observed.⁵ At higher levels of hydration, the interfacial contribution is masked by the bulk contribution in odd-order optical signals. Thus, probing spectral diffusion of interfacial water requires even-order interface-specific 2D techniques. Second-order 2D IR-IR surface-specific spectroscopy which had two time delays analogous to 2D Raman spectroscopy⁷ has been proposed.^{8,9} An alternative 2D surface-specific spectroscopy uses the fourth-order IR-IR-IR-visible (IIIV)¹⁰ pulse configuration shown schematically in Figure 1. This technique has three time delays, as in 2D IR spectroscopy.¹¹ Since the IIIV signal is in the visible region, its detection is easier than the second-order IR-IR signal.

Here we examine the spectral diffusion of interfacial water near DMPC by a simulation study of the IIIV signal. This is the first reported simulation of the fourth-order IIIV signal. The marriage of homodyne-detected fourth-order spectroscopy technique¹⁰ with heterodyne-detected SFG¹² should make the proposed heterodyne-detected fourth-order spectroscopy possible in the near future. MD trajectories are used to compute the classical optical response. Our main findings are an asymmetric cross peak pattern in the 2D spectra representing coherence transfer, a steep nodal slope for the *H*-mode indicating slow spectral diffusion of top-layer water, and a shallow nodal slope for the *L*-mode indicating fast spectral diffusion of near-bulk water. The gradient of the nodal slope of the *L*-mode indicates that the spectral diffusion of near-bulk water is as fast as bulk water, even though the near-bulk water is oriented to DMPC unlike bulk water.

A system consists of 8 DMPC and 462 water molecules. Details of the MD simulation protocols are given in the supporting information. The classical approximation holds when the vibrational frequency is low compared to kT or for weakly anharmonic vibrations at any temperature, where T is the system temperature. The OH stretch is weakly anharmonic, which makes the classical protocol adequate for the present application. The merits of this approach are given in the supporting information. The fourth-order IIIV response function was calculated classically using the stability matrix formalism^{13,14}

$$R_{abcde}^{(4)}(\tau_3, \tau_2, \tau_1) = -\beta \left\langle \{A_{ab}(t_3), M_c(t_2)\} \right. \\ \left. \times \left(\beta \dot{M}_d(t_1), \dot{M}_e(t_0) - \{M_d(t_1), \dot{M}_e(t_0)\} \right) \right\rangle, \quad (1)$$

where $\beta=1/kT$, $T=300\text{K}$, $t_3=\tau_3+\tau_2/2$, $t_2=\tau_2/2$, $t_1=-\tau_2/2$, $t_0=-\tau_1-\tau_2/2$, and a, b, c, d, e represent the pulse polarization directions. τ_1, τ_2 , and τ_3 are the delays between pulses (Figure 1). $M(t)$ and $A(t)$ are the total dipole moment and the total polarizability of the entire system at time t , and $\{\dots\}$ is the Poisson bracket. Calculation details for $M(t)$ and $A(t)$ are given in the supporting information.

In all simulations, we have set the waiting time (τ_2) to zero. Variation of the signal with τ_2 should reveal directly vibrational relaxation. This goes beyond the present study. The time-domain fourth-order response function, $R_{xxzz}^{(4)}(\tau_3, \tau_2=0, \tau_1)$, is displayed in Figure 2(a). The signal decays in $\sim 80\text{fs}$ along the τ_3 axis. To suppress the small fluctuations at long τ_3 and τ_1 , we employed the window function^{9,14}

$$f_s(\tau_s) = \begin{cases} 1 & \text{for } \tau_s < t_{c1} \\ \cos^2(\pi(\tau_s - t_{c1})/2(t_{c2} - t_{c1})) & \text{for } t_{c1} \leq \tau_s < t_{c2} \\ 0 & \text{for } t_{c2} \leq \tau_s \end{cases}, \quad (2)$$

where $\tau_s = \sqrt{\tau_1^2 + \tau_3^2}$, $t_{c1} = 112.5$ fs, and $t_{c2} = 150$ fs. The smoothed response function $R_{s,xxxx}^{(4)}(\tau_3, \tau_2=0, \tau_1) = R_{s,xxxx}^{(4)}(\tau_3, \tau_2=0, \tau_1) f_s(\tau_s)$ is compared with the bare response function in Figure 2(b).

The frequency-domain 2D spectra were obtained by the double Fourier transform of the time-domain response function

$$(R_{abcde}^{(4)}(\omega_3, \tau_2=0, \omega_1) = \int_0^\infty d\tau_1 \int_0^\infty d\tau_3 e^{-i\omega_1\tau_1 - i\omega_3\tau_3} R_{s,abcde}^{(4)}(\tau_3, \tau_2=0, \tau_1). \quad (3)$$

$R_{s,xxxx}^{(4)}(\omega_3, \tau_2, \omega_1)$ includes all possible Liouville space pathways that contribute to different signals. Denoting the wavevectors of the IR pulses by \mathbf{k}_1 , \mathbf{k}_2 , and \mathbf{k}_3 and that of the visible pulse by \mathbf{k}_4 , we have the rephasing signal at $\mathbf{k}_I = -\mathbf{k}_1 + \mathbf{k}_2 + \mathbf{k}_3 + \mathbf{k}_4$, the non-rephasing signal at $\mathbf{k}_{II} = \mathbf{k}_1 - \mathbf{k}_2 + \mathbf{k}_3 + \mathbf{k}_4$, and the double quantum coherence signal at $\mathbf{k}_{III} = \mathbf{k}_1 + \mathbf{k}_2 - \mathbf{k}_3 + \mathbf{k}_4$. In this study, we focus on the \mathbf{k}_I signal, which appears in the region of $\omega_1 < 0$ and $\omega_3 > 0$. The \mathbf{k}_{II} and \mathbf{k}_{III} signals are easily separated since they appear in the region of $\omega_1 > 0$ and $\omega_3 > 0$.

The vibrational energy level diagram for the water OH stretch at the DMPC interface is presented in Figure 3. $|g\rangle$ is the ground state and the singly excited L - and H -states are denoted as $|e_I\rangle$ and $|e_2\rangle$, while the doubly excited states of the L - and H -modes and the combination state are denoted as $|f_I\rangle$, $|f_2\rangle$, and $|f_3\rangle$. The three Liouville space pathways contributing to the \mathbf{k}_I signal are shown in Figure 4. Peaks corresponding to diagrams A and B have opposite signs compared to diagram C in $\text{Im}(R_{s,xxxx}^{(4)}(\omega_3, \tau_2, \omega_1))$. When the frequency shifts of doubly excited states or combination state are zero, these opposite sign contributions exactly cancel as in 2D IR spectra. The information on the molecular orientation contained in $\text{Im}(R_{xxz}^{(2)}(\omega))$ shows up also in $\text{Im}(R_{s,xxxx}^{(4)}(\omega_3, \tau_2=0, \omega_1))$: Peaks corresponding to diagram A and B in $\text{Im}(R_{s,xxxx}^{(4)}(\omega_3, \tau_2=0, \omega_1))$ have opposite signs than in $\text{Im}(R_{xxz}^{(2)}(\omega))$.

The frequency-domain 2D spectra, $|R_{s,xxxx}^{(4)}(\omega_3, \tau_2=0, \omega_1)|^2$ and $\text{Im}(R_{s,xxxx}^{(4)}(\omega_3, \tau_2=0, \omega_1))$, are displayed in Figure 5(a) and (b), respectively. In Figure 5(a), the H -peak is on the diagonal line $-\omega_1 = \omega_3$, whereas the L -peak appears in the $-\omega_1 > \omega_3$ region. In addition, both signals show weak cross peaks in the $-\omega_1 > \omega_3$ region (marked by asterisks), but no cross peak in the $-\omega_1 < \omega_3$ region.

The diagonal peaks in Figure 5(b) show that the frequency shifts were $\Delta_1 = 180\text{cm}^{-1}$ for the L -mode and $\Delta_2 = 150\text{cm}^{-1}$ for the H -mode. These are close to the frequency shift of 153cm^{-1} due to the intramolecular anharmonicity of water OH stretch.¹⁵ The peaks overlap, since Δ_1 and Δ_2 are comparable with frequency splitting of the L - and H -peaks of $\Delta\omega = 280\text{cm}^{-1}$. In particular, the positive peak of the H -mode and the negative peak of the L -mode cancel out, because the latter peak is elongated along the ω_1 axis as schematically illustrated in Figure 6. This cancellation weakens the negative peak of the L -mode and shifts the peak in

$|R_{s,xxxx}^{(4)}(\omega_3, \tau_2=0, \omega_1)|^2$ to the $-\omega_1 > \omega_3$ region.

We next turn to the cross peaks. There are two possible origins for the cross peaks. One is the frequency shift for the combination band of the *L*- and *H*-modes. If the *L*- and *H*-modes are coupled and this shifts the combination mode to the lower frequency by Δ_3 , the cross peaks appear in both the $-\omega_1 > \omega_3$ and $-\omega_1 < \omega_3$ regions with the same intensities. The absence of a cross peak in the $-\omega_1 < \omega_3$ region means that the direct intermode coupling is weak compared with the intramolecular anharmonicity and $\Delta_3 \ll \Delta_1 \approx \Delta_2$. Cross peaks can also originate from coherence transfer from $|g\rangle\langle e_2|$ to $|g\rangle\langle e_1|$ and from $|g\rangle\langle e_1|$ to $|g\rangle\langle e_2|$ during τ_1 and τ_3 . The $\Delta\omega=280\text{cm}^{-1}$ energy splitting of the *L*- and *H*-modes indicates that the transition from $|g\rangle\langle e_2|$ to $|g\rangle\langle e_1|$ is $\exp(\hbar\Delta\omega/kT) \approx 3.8$ times faster than the transition from $|g\rangle\langle e_1|$ to $|g\rangle\langle e_2|$. Coherence transfer from $|g\rangle\langle e_2|$ to $|g\rangle\langle e_1|$ creates the cross peak in the $-\omega_1 > \omega_3$ region, while the cross peak in the $-\omega_1 < \omega_3$ region arises from the transition from $|g\rangle\langle e_1|$ to $|g\rangle\langle e_2|$. Thus, both $|R_{xxzz}^{(4)}(\omega_3, \tau_2=0, \omega_1)|^2$ and $\text{Im}(R_{xxzz}^{(4)}(\omega_3, \tau_2=0, \omega_1))$ show a cross peak in the $-\omega_1 > \omega_3$ region, while the cross peak in the $-\omega_1 < \omega_3$ region is weak and is not clearly observed, as illustrated in Figure 6.

Spectral diffusion of water at the DMPC interface can be inferred from the angle between the nodal lines and the ω_1 axis in Figure 5(b); the nodal line parallel to the diagonal indicates slow spectral diffusion compared with the decay time of the signal of $\sim 80\text{fs}$, while the nodal line parallel to the ω_1 axis indicates fast spectral diffusion.¹⁶ In Figure 5, the angles are 29° for the *L*-mode and 44° for the *H*-mode. Since the *H*-mode was assigned to the top-layer OH stretch adjacent to DMPC,⁴ we conclude that this OH stretch has slow spectral diffusion due to the strong electrostatic interactions between water and the hydrophilic part of DMPC. The heterogeneous nature of water at DMPC with low hydration was also reported by Volkov *et al.*⁵ Figure 5(b) also shows that the *L*-mode has a shallow nodal slope compared with the *H*-mode, indicating that the near-bulk OH stretch non-adjacent to DMPC has faster spectral diffusion than the OH stretch adjacent to DMPC. The angle of 29° in

$\text{Im}(R_{xxzz}^{(4)}(\omega_3, \tau_2=0, \omega_1))$ is compatible with $30^\circ \pm 5^\circ$ in 2D IR spectrum of bulk water.¹⁷ This means that the interfacial water corresponding to the *L*-mode changes the frequency of its OH stretch rapidly due to the ultrafast conformational change at the similar speed as bulk water. The near-bulk water non-adjacent to DMPC is oriented to DMPC unlike bulk water. Nevertheless the water dynamics is similar in both cases.

In conclusion, we have reported the first simulation study of the 2D fourth-order IIIV rephrasing $\mathbf{k}_1=\mathbf{k}_1-\mathbf{k}_2+\mathbf{k}_3+\mathbf{k}_4$ signal at the water/DMPC interface. Both simulated signals, $|R_{xxzz}^{(4)}(\omega_3, \tau_2=0, \omega_1)|^2$ and $\text{Im}(R_{xxzz}^{(4)}(\omega_3, \tau_2=0, \omega_1))$, show a cross peak in the $-\omega_1 > \omega_3$ region and no cross peak in the $-\omega_1 < \omega_3$ region, indicating coherence transfer from the *H*- to *L*-modes. $\text{Im}(R_{xxzz}^{(4)}(\omega_3, \tau_2=0, \omega_1))$ shows that the *L*-mode has a shallower nodal slope than the *H*-mode and the gradient of the nodal line for the *L*-mode is similar to that of the bulk water in 2D IR spectra. These indicate that the *H*- and *L*-modes have fast and slow spectral diffusion, respectively. By using the peak assignments of the *H*- and *L*-modes in our previous work,⁴ we can conclude that the near-bulk water non-adjacent to DMPC has faster spectral diffusion than the top-layer water adjacent to DMPC and its spectral diffusion timescale is similar to bulk water.

Supplementary Material

Refer to Web version on PubMed Central for supplementary material.

Acknowledgments

This work was supported by the National Institute of Health (Grant No. GM-59230) and the National Science Foundation (Grant No. CHE-0745892).

REFERENCES

- (1). Egberts E, Marrink SJ, Berendsen HJC. *Eur. Biophys. J.* 1994; 22:423–436. [PubMed: 8149924]
- (2) (a). Mondal JA, Nihonyanagi S, Yamaguchi S, Tahara T. *J. Am. Chem. Soc.* 2010; 132:10656. [PubMed: 20681689] (b) Chen XK, Hua W, Huang ZS, Allen HC. *J. Am. Chem. Soc.* 2010; 132:11336–11342. [PubMed: 20698700]
- (3) (a). Ghosh A, Smits M, Bredenbeck J, Bonn M. *J. Am. Chem. Soc.* 2007; 129:9608. [PubMed: 17637056] (b) Ghosh A, Smits M, Sovago M, Bredenbeck J, Bonn M. *Chem. Phys. Lett.* 2008; 350:23–30.
- (4). Nagata Y, Mukamel S. *J. Am. Chem. Soc.* 2010; 132:6434–6442. [PubMed: 20394423]
- (5) (a). Volkov VV, Palmer DJ, Righini R. *J. Phys. Chem. B.* 2007; 111:1377–1382. [PubMed: 17249718] (b) Volkov VV, Palmer DJ, Righini R. *Phys. Rev. Lett.* 2007; 99:078302. [PubMed: 17930929] (c) Volkov VV, Kataoka Y, Righini R. *J. Phys. Chem. B.* 2009; 113:4119–4124. [PubMed: 19228033]
- (6). Szyc L, Yang M, Nibbering ETJ, Elsaesser T. *Angew. Chem. Int. Ed.* 2010; 49:3598–3610.
- (7). Tanimura Y, Mukamel S. *J. Chem. Phys.* 1993; 99:9496–9511.
- (8). Cho M. *J. Chem. Phys.* 2000; 112:9978–9985.
- (9). Nagata Y, Tanimura Y, Mukamel S. *J. Chem. Phys.* 2007; 126:204703. [PubMed: 17552785]
- (10) (a). Bredenbeck J, Ghosh A, Smits M, Bonn M. *J. Am. Chem. Soc.* 2008; 130:2152. [PubMed: 18225904] (b) Bredenbeck J, Ghosh A, Nienhuys HK, Bonn M. *Acc. Chem. Res.* 2010; 42:1332–1342. [PubMed: 19441810]
- (11). Mukamel S, Abramavicius D. *Chem. Rev.* 2004; 104:2073–2098. [PubMed: 15080721]
- (12) (a). Stiofkin IV, Jayathilake HD, Bordenyuk AN, Benderskii AV. *J. Am. Chem. Soc.* 2008; 130:2271–2275. [PubMed: 18217755] (b) Nihonyanagi S, Yamaguchi S, Tahara T. *J. Chem. Phys.* 2009; 130:204704. [PubMed: 19485472]
- (13) (a). Mukamel S, Khidekel V, Chernyak V. *Phys. Rev. E.* 1996; 53:R1–R4. (b) Saito S, Ohmine I. *J. Chem. Phys.* 1998; 108:240–251.
- (14). Jeon J, Cho M. *New J. Phys.* 2010; 12:065001.
- (15). Hayashi T, Jansen T. I. C. Zhuang W, Mukamel S. *J. Phys. Chem. A.* 2005; 109:64–82. [PubMed: 16839090]
- (16). Roberts ST, Ramasesha K, Tokmakoff A. *Acc. Chem. Res.* 2009; 42:1239–1249. [PubMed: 19585982]
- (17) (a). Asbury JB, Steinel T, Stromberg C, Corcelli SA, Lawrence CP, Skinner JL, Fayer MD. *J. Phys. Chem. A.* 2004; 108:1107–1119. (b) Eaves JD, Loparo JJ, Fecko CJ, Roberts ST, Tokmakoff A, Geissler PL. *Proc. Natl. Acad. Sci.* 2005; 102:13019–13022. [PubMed: 16135564] (c) Jansen, T. I. C.; Auer, BM.; Yang, M.; Skinner, JL. *J. Chem. Phys.* 2010; 132:224503. [PubMed: 20550404]

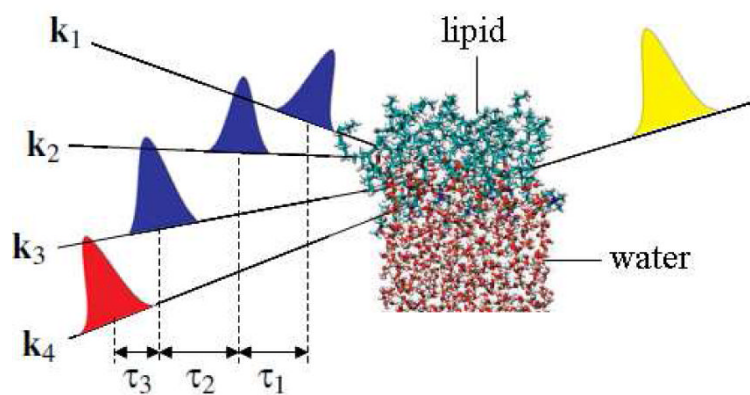


Figure 1. The fourth-order IR-IR-IR-visible experiment. The blue, red, and yellow pulses represent the IR, visible, and detection pulses. k_1 , k_2 , k_3 , and k_4 are their wavevectors, and τ_1 , τ_2 , and τ_3 represent the delays between pulses.

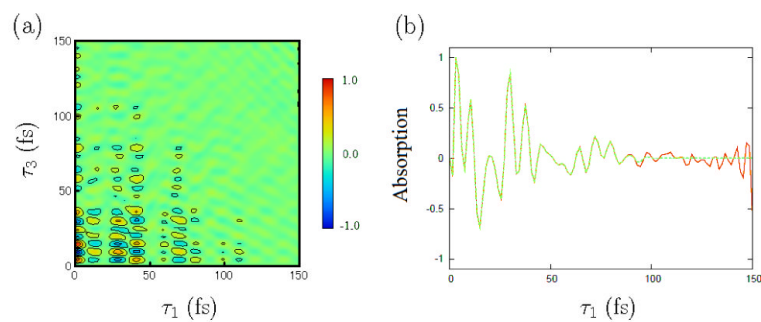


Figure 2.

(a) Time-domain 2D response function $(R_{xxzz}^{(4)}(\tau_3, \tau_2=0, \tau_1))$. (b) Diagonal elements of time-domain 2D response function. Red and green lines represent bare $(R_{xxzz}^{(4)}(\tau_1, \tau_2=0, \tau_1))$ and smoothed $(R_{xxzz}^{(4)}(\tau_1, \tau_2=0, \tau_1))$, respectively.

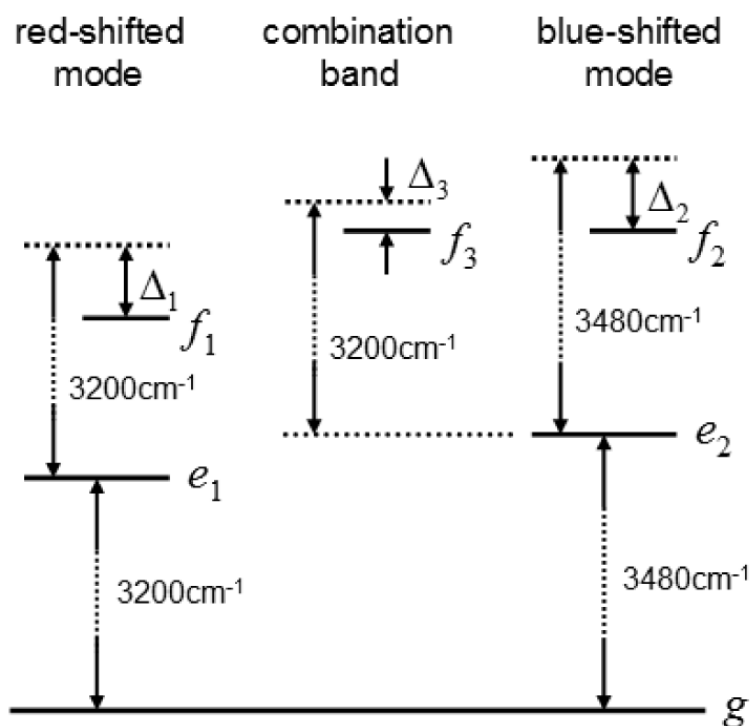


Figure 3. Energy level scheme for water OH stretch at the DMPC interface. State g , e , and f represent the ground, one-exciton, and two-exciton states. Δ_1 , Δ_2 , and Δ_3 denote the frequency shifts of the two exciton states.

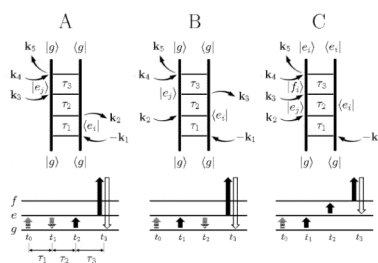


Figure 4. Top panels; Feynman diagrams for the fourth-order IIIV signal at $\mathbf{k}_I = -\mathbf{k}_1 + \mathbf{k}_2 + \mathbf{k}_3 + \mathbf{k}_4$. Bottom panels; energy level scheme. The transition of the bra and ket are drawn in solid and shadowed arrows. The detection pulse is denoted by the open arrows.

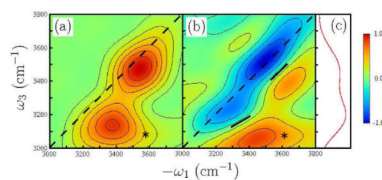


Figure 5.

(a) Simulated fourth-order spectra in the direction $\mathbf{k}_1 = -\mathbf{k}_1 + \mathbf{k}_2 + \mathbf{k}_3 + \mathbf{k}_4$, $|R_{xxzz}^{(4)}(\omega_3, \tau_2=0, \omega_1)|^2$ and (b) $\text{Im}(R_{xxzz}^{(4)}(\omega_3, \tau_2=0, \omega_1))$. The cross peaks are marked by asterisks (*). The black solid lines represent the nodal lines between negative and positive peaks. The angles between the nodal lines and the ω_1 axis are 29° for the *L*-mode and 44° for the *H*-mode. The frequencies of the *L*- and *H*-mode are 3480cm^{-1} and 3200cm^{-1} (c) SFG spectrum $\text{Im}(R_{xxz}^{(2)}(\omega_3))$.

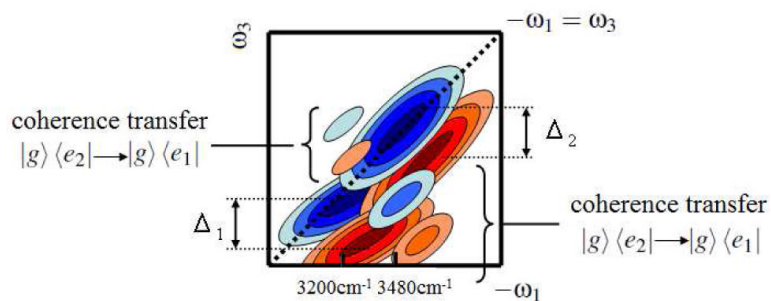


Figure 6.

Schematic 2D spectrum for $\text{Im}\left(R_{xvzz}^{(4)}(\omega_3, \tau_2=0, \omega_1)\right)$ which helps the peak assignment. Coherence transfer from $|g\rangle\langle e_2|$ to $|g\rangle\langle e_1|$ is responsible for the triangular cross peak in the $-\omega_1 > \omega_3$ region, while coherence transfer from $|g\rangle\langle e_1|$ to $|g\rangle\langle e_2|$ causes cross peak in the $-\omega_1 < \omega_3$ region.