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Mechanism-Based Therapeutic Approaches to Rhabdomyolysis-Induced Renal Failure

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Abstract

Rhabdomyolysis-induced renal failure represents up to 15% of all cases of acute renal failure. Many studies over the past four decades have demonstrated that accumulation of myoglobin in the kidney is central in the mechanism leading to kidney injury. However, some discussion exists regarding the mechanism mediating this oxidant injury. Although free iron-catalyzed fenton reaction has been proposed to explain the tissue injury, more recent evidence strongly suggests that the main cause of oxidant injury is myoglobin redox cycling and generation of oxidized lipids. These molecules can propagate tissue injury and cause renal vasoconstriction, two of the three main conditions associated with acute renal failure. This review presents the evidence supporting the two mechanisms of oxidative injury, describes the central role of myoglobin redox cycling in the pathology of renal failure associated with rhabdomyolysis, and discuss the value of therapeutic interventions aiming at inhibiting myoglobin redox cycling for the treatment of rhabdomyolysis-induced renal failure.

Introduction

Rhabdomyolysis affects about 1 in 10,000 persons in the USA and accounts for an estimated 8 to 15 % of all cases of acute renal failure (www.rhabdomyolysis.org). About 5% of rhabdomyolysis cases result in death (about 1,500 for the USA). Some data suggests that the number of patients with rhabdomyolysis following trauma may be underestimated and that the current treatments may not improve the outcome in a large proportion of patients with renal failure following rhabdomyolysis [1]. In this review we will describe the current treatments and discuss possible new therapeutic venues derived from recent new advances in the understanding of the pathophysiology of rhabdomyolysis-induced renal failure.

Rhabdomyolysis

Rhabdomyolysis, the breakdown of striated muscle, results in the release of potentially toxic compounds in the circulation that may affect kidney function. The syndrome has been recognized for a few thousands years (Old Testament, Book of Numbers, Ch. 11, verses 31–35). However, the link between renal failure and rhabdomyolysis was clearly established for

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the first time by the classic description of the crush syndrome following London bombardment during the Second World War [2]. The authors later determined the presence of abnormal levels of myoglobin (Mb) in the urine of these patients [3].

Etiology

The causes of rhabdomyolysis are diverse ranging from crush injury to genetic disorders or drug induced (Table 1). These disorders can be due to mechanical or electrical trauma [4–8], muscle injury following intense exercise, heatstroke, malignant hyperthermia, seizure or long term immobilization [8–16]. Lysis of myocytes can also occur following ischemia [17–20] or be caused by metabolic disorders leading to hypokalemia, hypernatremia or hypophosphatemia [21–26]. Many drugs administered in overdose but also in chronic normal dose administration have been shown to cause rhabdomyolysis [27–45]. Finally, genetic disorders leading to deregulations of enzymes involved in metabolic pathways have also been linked to rhabdomyolysis [46–50].

Pathophysiology

Although the causes are numerous, the etiology of the disease results from the lysis of the myocytes and the release of their content in the circulation (Table 1, for review see [51,52]). Following lysis of myocytes, large amounts of salts, enzymes (aldolase, Creatine kinase (CK), lactate dehydrogenase) and Mb, an 18,800 Da oxygen carrier [53–55] are released in the circulation. Circulating Mb then becomes deposited in the kidney causing renal tubular obstruction and necrosis, which is accompanied by intense renal vasoconstriction [51,56–59].

Renal failure

Although it is well accepted that renal failure is caused by Mb deposition in the kidney, the mechanism by which it occurs is still debated. Over time, number of mechanisms have been proposed to explain the pathophysiology of rhabdomyolysis-induced renal failure, including decreased delivery of blood to the glomerulus [56,60], reduced glomerular filtration rate [61], leakage of filtrate across a damaged tubular epithelium [62] or tubular obstruction by Mb casts [62]. The current consensus is that renal failure is due to the combined effects of hypovolemia, aciduria, and direct cytotoxicity due to accumulation of renal tubular Mb (for review, see [63,64]).

Oxidant injury and rhabdomyolysis-induced renal failure

There is accumulating evidence for a causative role of Mb-mediated oxidative injury to the kidney in the development of rhabdomyolysis-induced renal failure. Two hypotheses have been proposed to explain the mechanism by which Mb can cause oxidative injury to the kidney: 1) release of free iron from the Mb, catalyzing Fenton reactions, and 2) Mb redox cycling induced lipid peroxidation. In this section we will analyze the merit of both theories.

Free iron-mediated Fenton reaction

Most data suggested that free iron released from Mb degradation in the kidney is involved in the generation of oxidative species through catalysis of Fenton reaction. The strongest evidence for this mechanism came from data showing that desferrioxamine, an iron chelator, decreased rhabdomyolysis-induced renal injury in the rat [59] and prevented cell toxicity induced by direct exposure to Mb [65,66]. Although the role of free iron in renal injury can not be totally excluded, there are reasons to think that this is unlikely to be a major contributor. First, several mechanisms are in place in the body to bind free iron and prevent its toxic effect *in vivo*. Experiments designed to measure formation of hydroxyl radical ($\cdot\text{OH}$), the reactive oxygen species produced by the Fenton reaction, in proximal tubular

segments of the rat model of glycerol-induced rhabdomyolysis demonstrated that levels of OH^\bullet are decreased after glycerol injection compared with control animals [67]. Using cultured kidney cells subjected to Mb, Zager *et al.* demonstrated that compounds that scavenge $\bullet\text{OH}$ were unable to protect the cells from the Mb-induced injury [66].

Cytochrome P-450s present in the kidney were proposed as an other source for free iron [68,69]. This hypothesis was based on observations that kidney content of P-450 was decreased following rhabdomyolysis in rats and that P-450 specific inhibitors decreased tissue injury [68]. Although several lines of evidence suggest that P-450s are a source of free iron in chronic kidney disease such as minimal change nephrotic syndrome [70–73], their roles in rhabdomyolysis-induced renal failure is more controversial [68,69,74].

The effects of deferoxamine observed both *in vivo* and in cells in culture [59,65,66] can be explained in view of the evidence that deferoxamine is known to be a reductant of ferryl Mb [75,76], the species responsible for the pseudo-peroxidase activity of Mb which mediates lipid peroxidation (see section below). Finally, degradation of hemoproteins by induction of heme oxygenase is known to prevent renal failure following induction of rhabdomyolysis in the rat [77,78] and following injection of hemoglobin in mouse [79], processes that release the heme from the proteins. Not only does this evidence support a role for Mb in rhabdomyolysis-induced renal failure, it also argues against the role of free iron in the oxidative damage, as degradation of the heme by heme oxygenase generates free iron [80].

Myoglobin redox cycling and lipid peroxidation

The normal function of Mb is to transport oxygen in muscles. It does so because of its ability, in its ferrous form (Fe^{II}), to bind reversibly molecular oxygen. In pathologic conditions in which this hemoprotein is released from the reducing environment of cells, the ferrous heme undergoes oxidation to the ferric state (Fe^{III}). In the presence of peroxides, such as hydrogen peroxide or lipid hydroperoxides, hemoproteins are oxidized to the ferryl form ($\text{Fe}^{\text{IV}}=\text{O}$), which often accompanied by the formation of a globin-based free radical [81–84]. The ferryl form of these hemoproteins can generate lipid-based radical species through abstraction of a lipid hydrogen atom and the globin radical can propagate to exogenous substrates [85]. In the absence of antioxidants, these reactions can initiate a cascade in which free and membrane-bound lipids become oxidized (Figure 1).

Several lines of evidence support the role of Mb redox cycling in the development of rhabdomyolysis-induced renal failure. Mb by itself can catalyze peroxidation of arachidonic acid [81]. The high levels of Mb accumulating in the kidney have been shown to cause lipid peroxidation in that target organ as reflected by markedly increased F_2 -isoprostanes (F_2 -IsoPs) excretion in the urine and increased levels of F_2 -IsoPs in the kidney following rhabdomyolysis in the rat [74]. Moreover, the urine of humans with rhabdomyolysis contains increased levels of F_2 -IsoPs and hemeprotein crosslinks of Mb [86]. These data unambiguously demonstrate that Mb redox cycling occurs in the kidney of patients with rhabdomyolysis, as the heme-protein crosslink only forms by the reaction of the ferryl heme and the globin radical [87].

Supporting evidence for the role of Mb redox cycling in renal failure is also shown by the attenuation of renal failure by alkalization [74] a process that is known to inhibit Mb redox cycling [74,88]. Classically the beneficial effects of alkalization were thought to be due to enhanced solubilization of Mb, which increased the excretion of Mb from the kidney. Moreover, arachidonic acid oxidation catalyzed by Mb has been shown to be increased at acidic pH [89]. Inhibitors of lipid peroxidation also has been shown to protect rhabdomyolysis-induced injury to the kidney [90] and deferoxamine can inhibit lipid peroxidation by reducing ferryl Mb to its ferric state [75,91,92] without chelating free iron

[91]. All these data support the mechanism by which ferric Mb deposited in the kidney undergoes redox cycling by reacting with peroxides present in the milieu, which in turn induces an oxidant injury to the kidney as a result of enhanced lipid peroxidation.

Further support for this hypothesis that Mb redox cycling is responsible for renal injury associated with rhabdomyolysis comes from the recent evidence that acetaminophen (ApAP) inhibits lipid peroxidation and significantly improves renal function in a rat model of rhabdomyolysis, with a reduction in the amount of structural renal damage [93]. The basis for considering that ApAP might inhibit the pseudoperoxidase-initiated formation of heme-protein radicals derived from our work that addressed the cellular selectivity of inhibition of the prostaglandin H synthases (PGHSs) by ApAP. The PGHSs are bifunctional enzymes that contain both cyclooxygenase and peroxidase catalytic sites [94]. The catalytic activity is initiated by reduction of a hydroperoxide in the peroxidase site leading to formation of a protoporphyrin radical cation similar to what happens with Mb. Abundant evidence indicates that ApAP is a reducing cosubstrate of the PGHS peroxidase [95–98]. ApAP also has been found to inhibit myeloperoxidase-induced lipid peroxidation [99]. The analogy of the PGHS catalytic cycle and that of ferryl Mb-mediated lipid peroxidation led us to hypothesize that ApAP may be able to reduce ferryl Mb and prevent its redox cycling, similarly to what happens in PGHS (Figure 1). In this manuscript [93], we demonstrated that ApAP inhibits the oxidation of free arachidonic acid catalyzed by Mb *in vitro*, via reduction of ferryl Mb to its ferric state. All the *in vitro* experiments were done in presence of EDTA, a free-iron chelator, ruling out the participation of free-iron in this process. We also have shown that ApAP prevents formation of the protein radicals, which result from incubation of Mb with hydrogen peroxide as well as the formation of heme-protein crosslinks. We then showed that treatment of rats with ApAP before or after induction of rhabdomyolysis by glycerol injection in skeletal muscle, led to a decrease urinary F₂-isoprostanes levels as well as inhibition of heme-protein crosslinks in the urine of the animals when compared to controls. Notably, heme-protein crosslinking can only occur if Mb has passed through the ferryl state [87]. Taken together, our results clearly demonstrate that compounds that can reduce the ferryl form of Mb can be used *in vivo* to protect the kidney from injury following rhabdomyolysis, establishing a basis for new therapeutic hypotheses.

Current treatments

The standard of care of rhabdomyolysis is aggressive IV rehydration with or without addition of mannitol and bicarbonate. Mannitol is thought to protect the kidney by decreasing the intratubular accumulation of Mb while bicarbonate increases the urine pH above 6.5, which is thought to help solubilize Mb thus enhancing its excretion from the kidney. This treatment was first described in 1984 by Ron *et al.* in patients suffering from crush injury [100]. However this study did not include a control group, thus undermining its interpretation. Since then, several studies, both prospective and retrospective, suggested that large volume infusion of fluid with balanced salt solution is sufficient to protect the kidney from injury and that bicarbonate seems to be unnecessary [65,101,102]. In 2004 in a retrospective study evaluating over 2,000 cases, Brown *et al.* found that patients with CK levels over 5,000 U/l do not benefit from administration of bicarbonate and mannitol. The authors concluded that “the standard of administering bicarbonate and mannitol to patients with post-traumatic rhabdomyolysis should be reevaluated” [1]. The conclusion of that study stresses the necessity to explore new therapeutic venues for the prevention of rhabdomyolysis-induced renal failure.

New hypothetical pharmacologic interventions

The three major mechanisms by which Mb induces acute renal failure include renal vasoconstriction, tubular obstruction by Mb casts and lipid peroxidation-mediated tubular injury. Tubular obstruction seems to be a result of the kidney injury [65,101,102] and, as such, may not necessitate specific interventions. On the other hand, targeting either of the other two mechanisms or both together may have profound effects against acute renal failure.

Inhibition of glomerular vasoconstriction

One of the other two mechanisms participating in the development of acute renal failure is the decrease in blood flow in the glomerulus (Figure 2). The current treatment with large volume infusion of fluid with balanced salts supports the idea that restoring glomerular filtration and flushing the intra-tubular Mb helps protecting the kidney.

Several vasoactive factors seem to play an important role in modifying renal blood flow. Nitric oxide (NO) production is decreased during rhabdomyolysis-induced acute renal failure as measured by levels of urinary nitrite and kidney injury is increased by treatment of the animals with L-NAME (an inhibitor of NO synthesis) [103]. This can potentially be explained by scavenging of NO by Mb in the kidney. However, the kidneys are protected from injury by treatment with L-arginine (a precursor of NO) or by molsidomine (a NO donor) [103]. These results suggest that increasing renal production of NO may help protect the kidney against oxidant injury following rhabdomyolysis.

Overproduction of F₂-isoprostanes in the kidney in rhabdomyolysis induced renal failure may also be an important contributor to the renal vasoconstriction. F₂-isoprostanes [104] have been shown to be remarkably potent renal vasoconstrictors [105–107], which is mediated by activation of vascular thromboxane receptors [108]. The notion that F₂-isoprostanes are an important of the renal vasoconstriction is supported by the findings that they are generated in the kidney following rhabdomyolysis both in an animal model [74] and in patients [109], and that thromboxane receptor antagonists have been shown to reduce renal vasoconstriction following rhabdomyolysis in the rat [110]. This evidence that Mb-induced isoprostanes can cause renal vasoconstriction, which can be prevented by antagonists of the thromboxane receptor, provides a mechanistic basis for developing additional therapeutic strategies to improve renal function in rhabdomyolysis induced renal failure.

Inhibition of myoglobin redox-cycling

Because the products of lipid peroxidation can cause both oxidant injury and vasoconstriction (figure 2), inhibiting Mb redox cycling might possibly represent the most efficient therapeutic intervention.

Lipid soluble antioxidants, if distributed to the target lipids, would be expected to break the chain reaction triggered by interaction of the hemoprotein with a lipid interface and could thereby attenuate the propagation of radicals that escape inhibition by a hemoprotein reductant. Although antioxidants would inhibit propagation of the free radical oxidation reaction, it would not eliminate the formation of fatty acid hydroperoxides that can serve as substrates for the pseudoperoxidase activity of Mb, sustaining the initiation phase of lipid-based radicals that would decrease the efficiency of the antioxidant. Indeed, the importance of hydroperoxides in the mechanism of renal failure caused by hemoproteins can be inferred from experiments demonstrating that scavenging hydrogen peroxide protects the kidney from injury following rhabdomyolysis induction in rats [111]. Similarly, a therapeutic intervention that increases the level in the aqueous phase of the glutathione peroxidase co-

substrate, glutathione, could accelerate disposition of the lipid hydroperoxides and attenuate this additional positive feedback on radical generation. Thus, a therapy combining chain breaking antioxidants and glutathione precursor, such as N-acetyl cysteine, or hydrogen peroxide scavengers may provide a synergistic effect on acute kidney failure.

Optimal therapy to inhibit lipid-peroxidation catalyzed by hemoproteins clinically may well be based on a drug with the properties to block radical initiation, with the likelihood that interruption of the more distal steps in the chain reaction by lipid soluble anti-oxidants could be additive or synergistic in inhibiting the feed-forward process of hemoprotein-catalyzed lipid peroxidation. The peroxidation of lipids by Mb is a multistep process, initiated by oxidation of hemoproteins in an aqueous environment that is accessible poorly if at all by highly lipid soluble antioxidants such as vitamin E. Thus, aqueous soluble inhibitors of formation of hemoprotein radicals may be optimal for blocking the initiation of the radical cascade.

Our findings that ApAP can inhibit oxidative injury of the kidney and protect rats from renal failure following rhabdomyolysis provides proof of the concept that the inhibition of Mb-induced lipid peroxidation can alter a pathophysiologic process in vivo. Importantly, this effect was seen when ApAP was given not only before induction of rhabdomyolysis but also when administered after rhabdomyolysis was induced [93]. Although ApAP is the only compound presently available for human use with a potency within its range of safety, many compounds that are peroxidase substrates [98,112] may potentially exert similar effects as ApAP.

Conclusions

In conclusion, this review describes the central role of Mb redox cycling in the pathology of renal failure associated with rhabdomyolysis. The standard of care for rhabdomyolysis-induced renal failure consists in aggressive rehydration and may be sufficiently effective for patients with a mild form of the disease. However, the evidence suggests that pharmacologic agents that inhibit Mb redox cycling may represent the best additive therapeutic intervention for patients with a more severe form of this disease. The report that ApAP prevents renal failure in an animal model of rhabdomyolysis-induced renal failure at plasma concentrations that are within the therapeutic concentrations in humans, provides the basis for future studies aiming at evaluating the therapeutic value of ApAP in treating patients with rhabdomyolysis. The data also supports the development of novel drugs that would be better reducing agents and/or have less hepatotoxicity than ApAP. Finally, although this review focuses specifically on rhabdomyolysis-induced renal failure, the approach described above could be extended to other conditions caused by hemoprotein-mediated oxidant injury such as myocardial infarction, subarachnoid hemorrhage, *Plasmodium falciparum* malaria, and sickle cell disease.

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List of Abbreviations

ApAP	acetaminophen
CK	Creatine kinase
EDTA	Ethylenediaminetetraacetic acid

F₂-IsoPs	F ₂ -isoprostanes
L-NAME	N (G)-nitro-L- arginine methyl ester
NO	Nitric oxide
OH[•]	hydroxyl radical
PGHSs	prostaglandin H synthases

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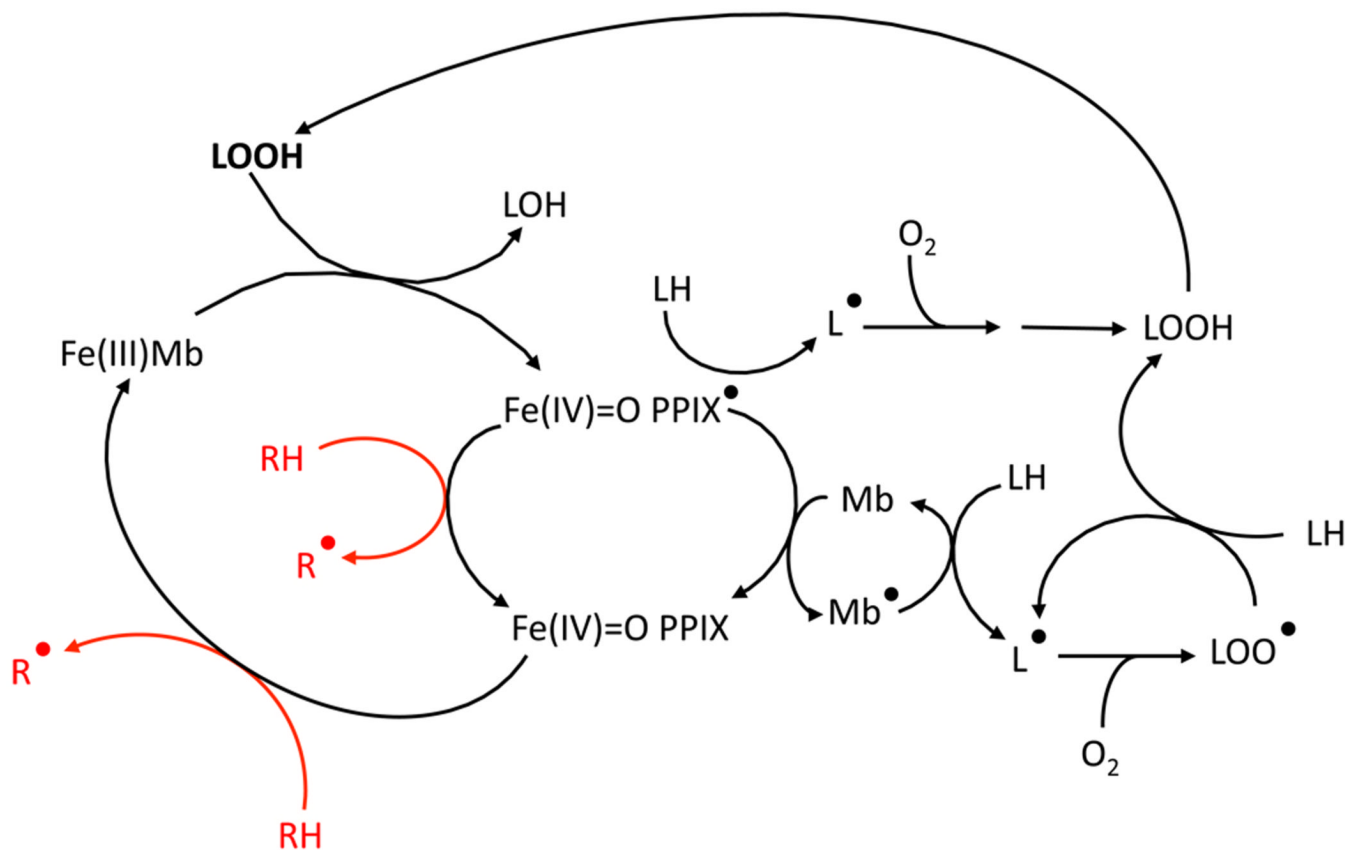


Figure 1. Mechanism for Mb-induced lipid peroxidation and its inhibition by acetaminophen
 Oxidation of ferric Mb (Fe(III)Mb) by a lipid hydroperoxide (LOOH) yields the ferryl Mb protoporphyrin radical (Fe(IV)=O PPIX[•]). This radical can delocalize to the protein to form a globin radical (Mb[•]), which can in turn catalyze lipid peroxidation by abstracting an electron from a lipid (LH). The lipid radical (L[•]) can then react to molecular oxygen to form the peroxy radical (LOO[•]), which can propagate lipid peroxidation. Reducing co-substrates such as ApAP (RH) reduce the ferryl Mb back to its ferric state, thus inhibiting Mb-catalyzed lipid peroxidation.

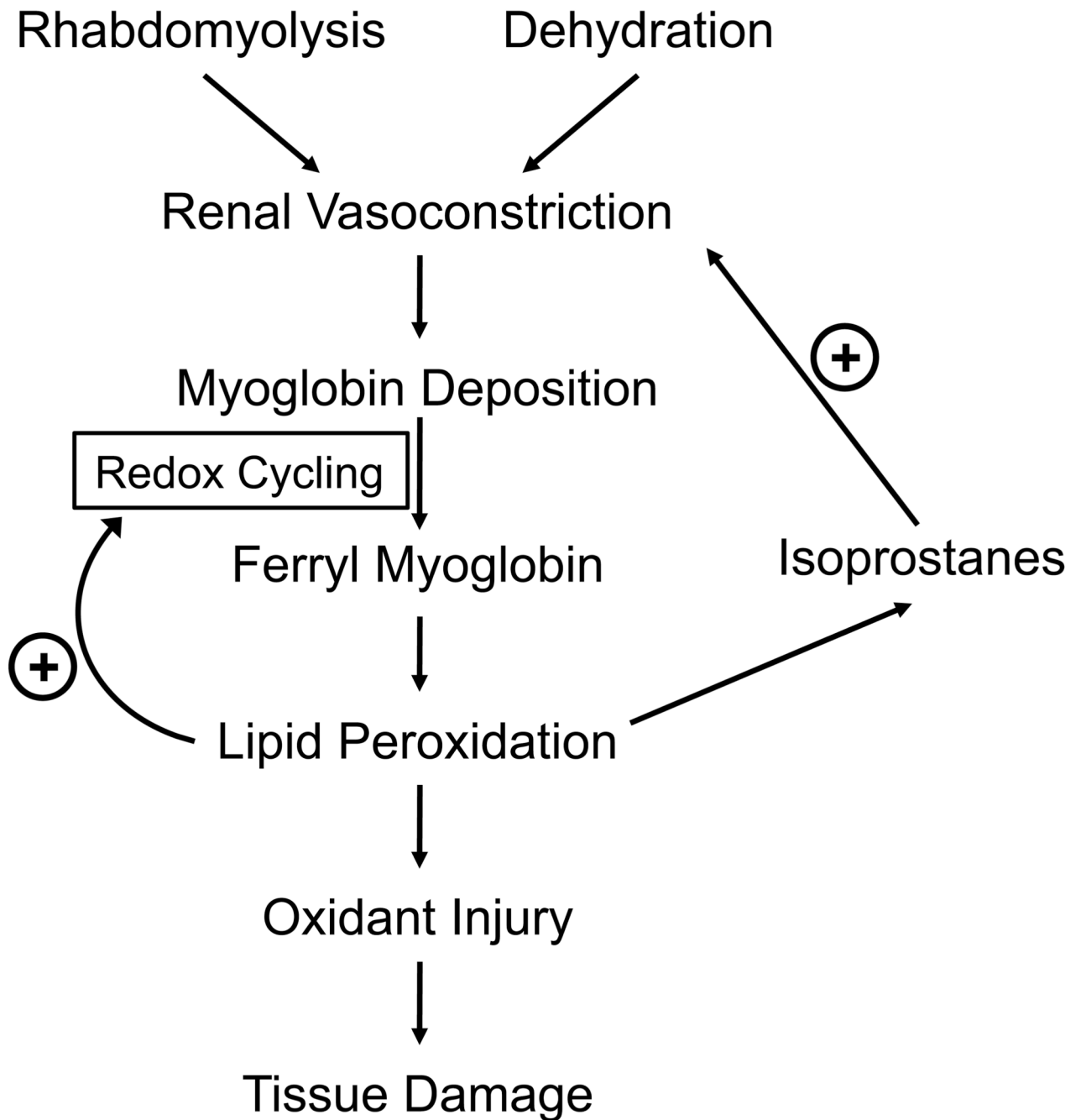


Figure 2. Pathophysiologic mechanism leading to kidney injury following rhabdomyolysis
 Release of Mb following rhabdomyolysis associated with hypovolemia induces renal vasoconstriction and Mb deposition in the kidney. Mb gets oxidized to its ferryl state inducing lipid peroxidation. Products of lipid peroxidation play a central role in the pathology via three different mechanisms: 1) they exacerbate lipid peroxidation by increasing the rate of formation of ferryl Mb; 2) they contribute to the oxidative injury of the kidney; and 3) they cause vasoconstriction preventing normal function of the kidney.

Table 1
Causes of rhabdomyolysis

(www.nlm.nih.gov/medlineplus, for review Khan, 2009, NJM, 272).

Crush injury	
Muscle injuries	Severe exertion, muscle ischemia, heat stroke, prolonged immobilization.
Drugs	Cocaine, heroin, alcohol, statins.
Electrolyte imbalance	Hyponatremia, hypokalemia and hypophosphatemia.
Genetic disorders	Deficiencies of glycolytic enzymes, abnormal lipid metabolism and other.