

Acta Crystallographica Section E

Structure Reports

Online

ISSN 1600-5368

***N*-tert-Butyl-3-mesitylpropanamide**Abel M. Maharramov,^a Ali N. Khalilov,^a Atash V. Gurbanov^a and Iván Brito^{b*}^aDepartment of Organic Chemistry, Baku State University, Baku, Azerbaijan, and^bDepartamento de Química, Facultad de Ciencias Básicas, Universidad de Antofagasta, Casilla 170, Antofagasta - Chile

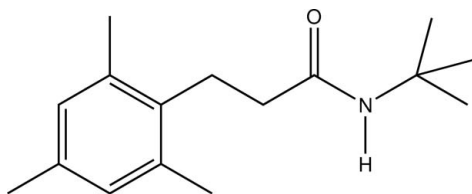
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Received 18 April 2011; accepted 26 April 2011

Key indicators: single-crystal X-ray study; $T = 296$ K; mean $\sigma(\text{C}-\text{C}) = 0.003$ Å; R factor = 0.065; wR factor = 0.173; data-to-parameter ratio = 22.9.

In the title compound, $\text{C}_{16}\text{H}_{25}\text{NO}$, the *N*-tert-butylpropanamide fragment is essentially planar, with the exception of two C atoms of the *tert*-butyl group (r.m.s. deviation = 0.005 Å), forming a dihedral angle of 84.09 (10)° with the plane of the mesityl fragment (r.m.s. deviation = 0.002 Å). The crystal packing is stabilized by an intermolecular $\text{N}-\text{H}\cdots\text{O}$ hydrogen bond, which links the molecules into chains with graph-set notation $C(4)$ running parallel to the c axis.

Related literature

For graph-set notation, see: Bernstein *et al.* (1995).

Experimental

Crystal data

$\text{C}_{16}\text{H}_{25}\text{NO}$
 $M_r = 247.37$
 Monoclinic, $P2_1/c$

$a = 12.8851$ (11) Å
 $b = 13.3441$ (11) Å
 $c = 9.4741$ (8) Å

$\beta = 106.540$ (2)°
 $V = 1561.6$ (2) Å³
 $Z = 4$
 Mo $K\alpha$ radiation

$\mu = 0.06$ mm⁻¹
 $T = 296$ K
 $0.20 \times 0.20 \times 0.20$ mm

Data collection

Bruker APEXII CCD diffractometer
 Absorption correction: multi-scan (SADABS; Sheldrick, 2003)
 $T_{\min} = 0.987$, $T_{\max} = 0.987$

11870 measured reflections
 3870 independent reflections
 1738 reflections with $I > 2\sigma(I)$
 $R_{\text{int}} = 0.052$

Refinement

$R[F^2 > 2\sigma(F^2)] = 0.065$
 $wR(F^2) = 0.173$
 $S = 1.00$
 3870 reflections

169 parameters
 H-atom parameters constrained
 $\Delta\rho_{\text{max}} = 0.15$ e Å⁻³
 $\Delta\rho_{\text{min}} = -0.14$ e Å⁻³

Table 1

Hydrogen-bond geometry (Å, °).

$D-H\cdots A$	$D-H$	$H\cdots A$	$D\cdots A$	$D-H\cdots A$
$\text{N1}-\text{H1N}\cdots\text{O1}^i$	0.83	2.17	2.979 (2)	165

Symmetry code: (i) $x, -y + \frac{1}{2}, z - \frac{1}{2}$.

Data collection: APEX2 (Bruker, 2005); cell refinement: SAINT (Bruker, 2001); data reduction: SAINT; program(s) used to solve structure: SHELXTL (Sheldrick, 2008); program(s) used to refine structure: SHELXTL; molecular graphics: SHELXTL and PLATON (Spek, 2009); software used to prepare material for publication: SHELXTL.

The authors are grateful to Baku State University for supporting this study. IB thanks the Spanish Research Council (CSIC) for the provision of a free-of-charge license to the Cambridge Structural Database.

Supplementary data and figures for this paper are available from the IUCr electronic archives (Reference: OM2425).

References

- Bernstein, J., Davis, R. E., Shimoni, L. & Chang, N.-L. (1995). *Angew. Chem. Int. Ed. Engl.* **34**, 1555–1573.
 Bruker (2001). SAINT. Bruker AXS Inc., Madison, Wisconsin, USA.
 Bruker (2005). APEX2. Bruker AXS Inc., Madison, Wisconsin, USA.
 Sheldrick, G. M. (2003). SADABS. University of Göttingen, Germany.
 Sheldrick, G. M. (2008). *Acta Cryst.* **A64**, 112–122.
 Spek, A. L. (2009). *Acta Cryst.* **D65**, 148–155.

supplementary materials

Acta Cryst. (2011). E67, o1307 [doi:10.1107/S1600536811015856]

N-*tert*-Butyl-3-mesitylpropanamide

A. M. Maharramov, A. N. Khalilov, A. V. Gurbanov and I. Brito

Comment

Fig. 1 shows the structure of title compound. Bond lengths and angles are unexceptional. The dihedral angle between the mesityl fragment and the C7/C6/C5/O1/N1/C4/C3 plane is 84.09 (10)°. Methyl groups of the benzene ring are into the same plane (r.m.s. deviation = 0.002 Å). In the crystal, molecules are linked by N—H···O interactions into chains with graph-set notation C(4) along [001], Figure 2, Table 1 (Bernstein *et al.*, 1995).

Experimental

A mixture of 0.001 mol of 1-chloro-3-(2,4,6-trimethylphenyl)propan-2-one and 0.001 mol of *tert*-butylamine was stirred in water in presence of sodium hydroxide (0.003 mol) for 35–40 minutes. The crystals were recrystallized from ethanol solution (Yield 86%, melting point 143°C). ¹H NMR spectrum, DMSO-d₆, δ, p.p.m.: 1.25 (s, 9H, 3CH₃), 2.15 (s, 2H, CH₂CO), 2.25 (s, 9H, 3CH₃), 2.75 (t, 2H, CH₂Ar), 6.75 (s, 2H, 2CH_{Ar}), 7.45 (s, 1H, NHCO). ¹³C NMR spectrum, DMSO-d₆, δ, p.p.m.: 19 (2CH₃), 21 (CH₃), 23(CH₂CO), 25 [(CH₃)₃], 37 (CH₂Ar), 50 (C_i), 129 (CH_{Ar}), 136 (C_i), 137 (C_i), 162 (CONH). IR spectrum, ν (cm⁻¹): 3360, 3170, 3005, 2928, 2878, 1645, 1615, 1470, 1430, 715, 605.

Refinement

All H-atoms were placed in calculated positions [C—H = 0.93 to 0.97 Å, $U_{\text{iso}}(\text{H}) = 1.2$ to $1.5 U_{\text{eq}}(\text{C})$ and N—H = 0.83 Å, $U_{\text{iso}}(\text{H}) = 1.5 U_{\text{eq}}(\text{N})$] and were included in the refinement in the riding model approximation. Due to weak diffracting ability of the crystal the ratio observed/unique reflections is low (45%).

Figures

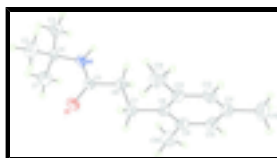


Fig. 1. The molecular structure of the title compound, showing the atom labelling scheme. Displacement ellipsoids are drawn at the 30% probability level. H atoms are shown as circles of arbitrary radius.

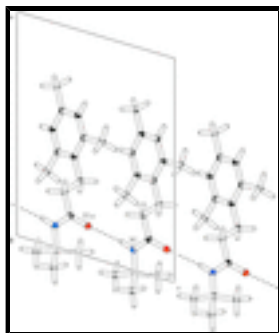


Fig. 2. Part of the crystal structure showing the formation of a C(4) chain along [001]. Hydrogen bond shown as dashed lines. Symmetry code: (a) $x, 1/2 - y, -1/2 + z$.

N-tert-Butyl-3-mesitylpropanamide

Crystal data

$C_{16}H_{25}NO$

$M_r = 247.37$

Monoclinic, $P2_1/c$

Hall symbol: -P 2ybc

$a = 12.8851$ (11) Å

$b = 13.3441$ (11) Å

$c = 9.4741$ (8) Å

$\beta = 106.540$ (2)°

$V = 1561.6$ (2) Å³

$Z = 4$

$F(000) = 544$

$D_x = 1.052$ Mg m⁻³

Mo $K\alpha$ radiation, $\lambda = 0.71073$ Å

Cell parameters from 1996 reflections

$\theta = 2.3$ – 28.0 °

$\mu = 0.06$ mm⁻¹

$T = 296$ K

Prism, colourless

$0.20 \times 0.20 \times 0.20$ mm

Data collection

Bruker APEXII CCD
diffractometer

Radiation source: fine-focus sealed tube
graphite

φ and ω scans

Absorption correction: multi-scan
(*SADABS*; Sheldrick, 2003)

$T_{\min} = 0.987$, $T_{\max} = 0.987$

11870 measured reflections

3870 independent reflections

1738 reflections with $I > 2\sigma(I)$

$R_{\text{int}} = 0.052$

$\theta_{\max} = 28.3$ °, $\theta_{\min} = 2.3$ °

$h = -17 \rightarrow 17$

$k = -15 \rightarrow 17$

$l = -12 \rightarrow 12$

Refinement

Refinement on F^2

Least-squares matrix: full

$R[F^2 > 2\sigma(F^2)] = 0.065$

$wR(F^2) = 0.173$

$S = 1.00$

3870 reflections

Primary atom site location: structure-invariant direct methods

Secondary atom site location: difference Fourier map

Hydrogen site location: difference Fourier map

H-atom parameters constrained

$w = 1/[\sigma^2(F_o^2) + (0.0732P)^2 + 0.0483P]$

where $P = (F_o^2 + 2F_c^2)/3$

$(\Delta/\sigma)_{\max} = 0.001$

169 parameters

$$\Delta\rho_{\max} = 0.15 \text{ e } \text{\AA}^{-3}$$

0 restraints

$$\Delta\rho_{\min} = -0.14 \text{ e } \text{\AA}^{-3}$$

Special details

Geometry. All e.s.d.'s (except the e.s.d. in the dihedral angle between two l.s. planes) are estimated using the full covariance matrix. The cell e.s.d.'s are taken into account individually in the estimation of e.s.d.'s in distances, angles and torsion angles; correlations between e.s.d.'s in cell parameters are only used when they are defined by crystal symmetry. An approximate (isotropic) treatment of cell e.s.d.'s is used for estimating e.s.d.'s involving l.s. planes.

Refinement. Refinement of F^2 against ALL reflections. The weighted R -factor wR and goodness of fit S are based on F^2 , conventional R -factors R are based on F , with F set to zero for negative F^2 . The threshold expression of $F^2 > \sigma(F^2)$ is used only for calculating R -factors(gt) *etc.* and is not relevant to the choice of reflections for refinement. R -factors based on F^2 are statistically about twice as large as those based on F , and R -factors based on ALL data will be even larger.

Fractional atomic coordinates and isotropic or equivalent isotropic displacement parameters (\AA^2)

	x	y	z	$U_{\text{iso}}^*/U_{\text{eq}}$
O1	0.13712 (13)	0.29736 (13)	0.94818 (15)	0.0737 (5)
N1	0.09333 (13)	0.20507 (13)	0.74037 (17)	0.0506 (5)
H1N	0.1102	0.1935	0.6635	0.076*
C1	0.0057 (2)	0.1074 (2)	0.8929 (3)	0.0898 (9)
H1A	0.0617	0.0580	0.9039	0.135*
H1B	-0.0606	0.0750	0.8940	0.135*
H1C	0.0263	0.1545	0.9727	0.135*
C2	-0.0933 (2)	0.2444 (2)	0.7318 (3)	0.0853 (8)
H2A	-0.0668	0.2942	0.8065	0.128*
H2B	-0.1597	0.2168	0.7417	0.128*
H2C	-0.1062	0.2746	0.6364	0.128*
C3	-0.0456 (2)	0.0878 (2)	0.6233 (3)	0.0851 (8)
H3A	0.0075	0.0357	0.6355	0.128*
H3B	-0.0529	0.1218	0.5316	0.128*
H3C	-0.1140	0.0592	0.6231	0.128*
C4	-0.01024 (17)	0.16181 (16)	0.7488 (2)	0.0521 (6)
C5	0.15762 (17)	0.26650 (16)	0.8371 (2)	0.0510 (5)
C6	0.25971 (18)	0.29603 (19)	0.8001 (2)	0.0676 (7)
H6A	0.2405	0.3213	0.7000	0.081*
H6B	0.3040	0.2368	0.8039	0.081*
C7	0.32616 (18)	0.37485 (17)	0.9021 (2)	0.0607 (6)
H7A	0.2826	0.4347	0.8977	0.073*
H7B	0.3454	0.3500	1.0025	0.073*
C8	0.42772 (17)	0.40167 (16)	0.8626 (2)	0.0505 (5)
C9	0.5242 (2)	0.35047 (15)	0.9257 (2)	0.0570 (6)
C10	0.61689 (19)	0.37656 (17)	0.8878 (2)	0.0611 (6)
H10	0.6810	0.3426	0.9316	0.073*
C11	0.61718 (18)	0.45079 (17)	0.7879 (3)	0.0587 (6)
C12	0.52139 (19)	0.49934 (17)	0.7257 (2)	0.0630 (6)
H12	0.5199	0.5495	0.6570	0.076*

supplementary materials

C13	0.42684 (17)	0.47697 (17)	0.7607 (2)	0.0572 (6)
C14	0.3253 (2)	0.5357 (2)	0.6890 (3)	0.0936 (9)
H14A	0.3410	0.5858	0.6252	0.140*
H14B	0.2998	0.5676	0.7636	0.140*
H14C	0.2706	0.4912	0.6326	0.140*
C15	0.7201 (2)	0.4793 (2)	0.7516 (3)	0.0929 (9)
H15A	0.7026	0.5186	0.6631	0.139*
H15B	0.7576	0.4197	0.7374	0.139*
H15C	0.7655	0.5177	0.8311	0.139*
C16	0.5308 (2)	0.26788 (19)	1.0379 (3)	0.0885 (9)
H16A	0.6026	0.2403	1.0664	0.133*
H16B	0.4797	0.2162	0.9954	0.133*
H16C	0.5144	0.2949	1.1229	0.133*

Atomic displacement parameters (\AA^2)

	U^{11}	U^{22}	U^{33}	U^{12}	U^{13}	U^{23}
O1	0.0738 (12)	0.1073 (13)	0.0501 (8)	-0.0259 (10)	0.0340 (8)	-0.0230 (9)
N1	0.0494 (11)	0.0637 (11)	0.0448 (9)	-0.0096 (9)	0.0234 (8)	-0.0049 (9)
C1	0.088 (2)	0.106 (2)	0.0798 (17)	-0.0271 (17)	0.0309 (15)	0.0263 (16)
C2	0.0554 (16)	0.092 (2)	0.112 (2)	0.0023 (14)	0.0289 (15)	-0.0023 (16)
C3	0.0772 (18)	0.098 (2)	0.0874 (18)	-0.0387 (16)	0.0342 (15)	-0.0283 (16)
C4	0.0465 (13)	0.0615 (13)	0.0528 (12)	-0.0089 (11)	0.0211 (10)	0.0006 (11)
C5	0.0506 (13)	0.0637 (14)	0.0426 (11)	-0.0100 (11)	0.0193 (10)	-0.0045 (11)
C6	0.0587 (15)	0.0892 (17)	0.0620 (13)	-0.0250 (13)	0.0287 (12)	-0.0247 (13)
C7	0.0588 (15)	0.0677 (14)	0.0581 (13)	-0.0112 (12)	0.0204 (11)	-0.0143 (12)
C8	0.0481 (13)	0.0538 (12)	0.0503 (11)	-0.0108 (11)	0.0151 (10)	-0.0124 (11)
C9	0.0636 (16)	0.0493 (13)	0.0551 (12)	-0.0051 (12)	0.0120 (11)	-0.0082 (11)
C10	0.0482 (14)	0.0597 (14)	0.0708 (15)	0.0019 (11)	0.0097 (12)	-0.0091 (13)
C11	0.0501 (14)	0.0606 (14)	0.0681 (14)	-0.0077 (12)	0.0211 (11)	-0.0114 (12)
C12	0.0674 (17)	0.0595 (14)	0.0639 (14)	-0.0038 (13)	0.0216 (12)	0.0052 (12)
C13	0.0487 (13)	0.0610 (14)	0.0592 (13)	-0.0004 (11)	0.0113 (11)	-0.0035 (12)
C14	0.0691 (19)	0.103 (2)	0.101 (2)	0.0138 (16)	0.0129 (16)	0.0256 (18)
C15	0.0692 (18)	0.105 (2)	0.116 (2)	-0.0129 (16)	0.0455 (16)	-0.0076 (18)
C16	0.095 (2)	0.0770 (18)	0.0887 (18)	-0.0019 (16)	0.0175 (16)	0.0204 (15)

Geometric parameters (\AA , $^\circ$)

O1—C5	1.226 (2)	C7—H7B	0.9700
N1—C5	1.329 (2)	C8—C13	1.391 (3)
N1—C4	1.477 (2)	C8—C9	1.395 (3)
N1—H1N	0.8310	C9—C10	1.386 (3)
C1—C4	1.508 (3)	C9—C16	1.517 (3)
C1—H1A	0.9600	C10—C11	1.371 (3)
C1—H1B	0.9600	C10—H10	0.9300
C1—H1C	0.9600	C11—C12	1.370 (3)
C2—C4	1.513 (3)	C11—C15	1.510 (3)
C2—H2A	0.9600	C12—C13	1.383 (3)
C2—H2B	0.9600	C12—H12	0.9300

C2—H2C	0.9600	C13—C14	1.511 (3)
C3—C4	1.512 (3)	C14—H14A	0.9600
C3—H3A	0.9600	C14—H14B	0.9600
C3—H3B	0.9600	C14—H14C	0.9600
C3—H3C	0.9600	C15—H15A	0.9600
C5—C6	1.507 (3)	C15—H15B	0.9600
C6—C7	1.517 (3)	C15—H15C	0.9600
C6—H6A	0.9700	C16—H16A	0.9600
C6—H6B	0.9700	C16—H16B	0.9600
C7—C8	1.503 (3)	C16—H16C	0.9600
C7—H7A	0.9700		
C5—N1—C4	126.81 (16)	C8—C7—H7B	109.1
C5—N1—H1N	116.8	C6—C7—H7B	109.1
C4—N1—H1N	116.1	H7A—C7—H7B	107.9
C4—C1—H1A	109.5	C13—C8—C9	118.8 (2)
C4—C1—H1B	109.5	C13—C8—C7	120.6 (2)
H1A—C1—H1B	109.5	C9—C8—C7	120.6 (2)
C4—C1—H1C	109.5	C10—C9—C8	119.6 (2)
H1A—C1—H1C	109.5	C10—C9—C16	118.9 (2)
H1B—C1—H1C	109.5	C8—C9—C16	121.4 (2)
C4—C2—H2A	109.5	C11—C10—C9	122.2 (2)
C4—C2—H2B	109.5	C11—C10—H10	118.9
H2A—C2—H2B	109.5	C9—C10—H10	118.9
C4—C2—H2C	109.5	C12—C11—C10	117.4 (2)
H2A—C2—H2C	109.5	C12—C11—C15	121.7 (2)
H2B—C2—H2C	109.5	C10—C11—C15	121.0 (2)
C4—C3—H3A	109.5	C11—C12—C13	122.8 (2)
C4—C3—H3B	109.5	C11—C12—H12	118.6
H3A—C3—H3B	109.5	C13—C12—H12	118.6
C4—C3—H3C	109.5	C12—C13—C8	119.2 (2)
H3A—C3—H3C	109.5	C12—C13—C14	119.3 (2)
H3B—C3—H3C	109.5	C8—C13—C14	121.5 (2)
N1—C4—C1	110.16 (18)	C13—C14—H14A	109.5
N1—C4—C3	106.71 (16)	C13—C14—H14B	109.5
C1—C4—C3	109.3 (2)	H14A—C14—H14B	109.5
N1—C4—C2	109.41 (18)	C13—C14—H14C	109.5
C1—C4—C2	110.9 (2)	H14A—C14—H14C	109.5
C3—C4—C2	110.2 (2)	H14B—C14—H14C	109.5
O1—C5—N1	123.68 (19)	C11—C15—H15A	109.5
O1—C5—C6	121.82 (19)	C11—C15—H15B	109.5
N1—C5—C6	114.49 (17)	H15A—C15—H15B	109.5
C5—C6—C7	113.87 (17)	C11—C15—H15C	109.5
C5—C6—H6A	108.8	H15A—C15—H15C	109.5
C7—C6—H6A	108.8	H15B—C15—H15C	109.5
C5—C6—H6B	108.8	C9—C16—H16A	109.5
C7—C6—H6B	108.8	C9—C16—H16B	109.5
H6A—C6—H6B	107.7	H16A—C16—H16B	109.5
C8—C7—C6	112.32 (17)	C9—C16—H16C	109.5
C8—C7—H7A	109.1	H16A—C16—H16C	109.5

supplementary materials

C6—C7—H7A	109.1	H16B—C16—H16C	109.5
C5—N1—C4—C1	-54.9 (3)	C7—C8—C9—C16	-1.2 (3)
C5—N1—C4—C3	-173.5 (2)	C8—C9—C10—C11	-0.9 (3)
C5—N1—C4—C2	67.3 (3)	C16—C9—C10—C11	-179.6 (2)
C4—N1—C5—O1	-1.4 (3)	C9—C10—C11—C12	0.1 (3)
C4—N1—C5—C6	178.5 (2)	C9—C10—C11—C15	178.5 (2)
O1—C5—C6—C7	-6.5 (3)	C10—C11—C12—C13	0.6 (3)
N1—C5—C6—C7	173.65 (19)	C15—C11—C12—C13	-177.8 (2)
C5—C6—C7—C8	179.5 (2)	C11—C12—C13—C8	-0.4 (3)
C6—C7—C8—C13	88.1 (2)	C11—C12—C13—C14	178.8 (2)
C6—C7—C8—C9	-90.9 (2)	C9—C8—C13—C12	-0.5 (3)
C13—C8—C9—C10	1.1 (3)	C7—C8—C13—C12	-179.47 (19)
C7—C8—C9—C10	-179.89 (18)	C9—C8—C13—C14	-179.6 (2)
C13—C8—C9—C16	179.8 (2)	C7—C8—C13—C14	1.4 (3)

Hydrogen-bond geometry (\AA , $^\circ$)

$D-H\cdots A$	$D-H$	$H\cdots A$	$D\cdots A$	$D-H\cdots A$
N1—H1N \cdots O1 ⁱ	0.83	2.17	2.979 (2)	165

Symmetry codes: (i) $x, -y+1/2, z-1/2$.

Fig. 1

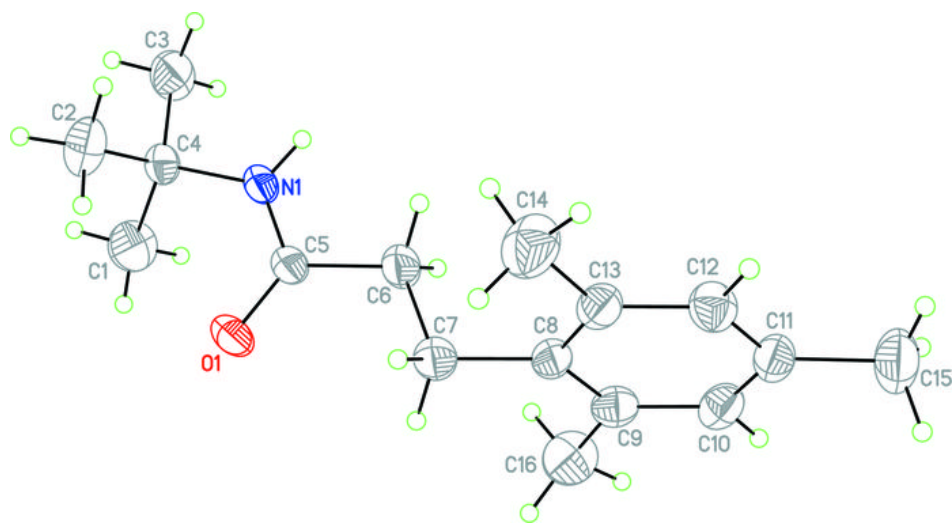


Fig. 2

