Phenazine-1-Carboxylic Acid Promotes Bacterial Biofilm Development via Ferrous Iron Acquisition^v†

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The opportunistic pathogen *Pseudomonas aeruginosa* **forms biofilms, which render it more resistant to antimicrobial agents. Levels of iron in excess of what is required for planktonic growth have been shown to promote biofilm formation, and therapies that interfere with ferric iron [Fe(III)] uptake combined with antibiotics may help treat** *P. aeruginosa* **infections. However, use of these therapies presumes that iron is in the Fe(III) state in the context of infection. Here we report the ability of phenazine-1-carboxylic acid (PCA), a common phenazine made by all phenazine-producing pseudomonads, to help** *P. aeruginosa* **alleviate Fe(III) limitation by reducing Fe(III) to ferrous iron [Fe(II)]. In the presence of PCA, a** *P. aeruginosa* **mutant lacking the ability to produce the siderophores pyoverdine and pyochelin can still develop into a biofilm. As has been previously reported (P. K. Singh, M. R. Parsek, E. P. Greenberg, and M. J. Welsh, Nature 417:552–555, 2002), biofilm formation by the wild type is blocked by subinhibitory concentrations of the Fe(III)-binding innateimmunity protein conalbumin, but here we show that this blockage can be rescued by PCA. FeoB, an Fe(II) uptake protein, is required for PCA to enable this rescue. Unlike PCA, the phenazine pyocyanin (PYO) can facilitate biofilm formation via an iron-independent pathway. While siderophore-mediated Fe(III) uptake is undoubtedly important at early stages of infection, these results suggest that at later stages of infection, PCA present in infected tissues may shift the redox equilibrium between Fe(III) and Fe(II), thereby making iron more bioavailable.**

Iron is an essential component of many metabolically important proteins in living cells. Because we live in an aerobic world, it is often assumed that biologically available iron is present in its ferric [Fe(III)] form and thus extremely limiting; this is thought to be the case in diverse environments, ranging from seawater to soils to the human body (47, 58). This limitation arises from the fact that under aerobic, circumneutral pH conditions, Fe(III) is present as highly insoluble Fe(III) minerals or is tightly bound to mammalian host Fe(III)-sequestering proteins (1, 58). In the context of bacterial infections, much attention has been placed on controlling bacterial growth by limiting Fe(III) acquisition (4, 30, 68). This is especially relevant for *Pseudomonas aeruginosa*, an opportunistic pathogen that infects the lungs of people with cystic fibrosis (CF) (25). Not only is iron essential for its growth, but iron at levels in excess of what is required for growth promotes biofilm formation by signaling a transition from a motile to a sessile state (5, 7, 51, 62, 68). As a biofilm, *P. aeruginosa* becomes significantly more resistant to antimicrobial agents and much more difficult to eradicate (13), which leads to an onslaught of problems that compromise proper lung functioning, commonly resulting in the morbidity and mortality of CF patients (13, 45).

Traditionally, studies of iron acquisition by *P. aeruginosa* have focused on the Fe(III) uptake pathway mediated by heterogeneous siderophores, i.e., small molecules with high affinity for binding Fe(III) (53, 64). Although in laboratory experiments lactoferrin, an innate-immunity protein, was shown to inhibit *P. aeruginosa* biofilm development by sequestering Fe(III) from siderophores (5, 68), it is unclear whether this occurs *in vivo*, because such inhibition may be compromised due to cleavage of lactoferrin by cathepsins (62). Intriguingly, *P. aeruginosa* strains present in advanced CF infections can lose the ability to make the siderophore pyoverdine (16, 69), and the iron content of CF sputum is often highly elevated compared to that in the sputum of healthy controls (60). Nonetheless, the actual chemistry of iron in CF sputum is poorly defined and likely varies temporally for any given patient, from patient-to-patient, and within any given microenvironment. This is due to a complex set of interactions between the host and microbial cells, some of which are just beginning to be elucidated (23), that can affect the chemistry of iron and its release (59). That said, it is reasonable to assume that if lactoferrin is present in sufficient amounts to bind Fe(III) on mucosal surfaces, it would block biofilm formation at early stages of infection (5, 68). Building on this finding, a novel coculture system to permit the study of biofilm formation on CF lung epithelial cells showed that cells harboring the Δ F508 mutation (the most common mutation of the cystic fibrosis transmembrane conductance regulator, CFTR) release significantly more iron than wild-type CFTR cells, stimulating bio-

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film formation by *P. aeruginosa.* In this system, chelation of Fe(III) by conalbumin (a lactoferrin surrogate protein) dramatically reduced biofilm formation (45). A recent follow-up study using this cell culture system showed that the combined use of tobramycin and FDA-approved iron chelators reduced biofilm biomass by approximately 90% (46).

While there is good reason to be optimistic regarding the efficacy of therapeutics that interfere with Fe(III) acquisition at early stages of infection (4, 30, 46), an understudied and potentially important aspect of the problem is whether iron always exists in the ferric state as infections progress, especially as inflammation is stimulated, cell densities increase, and oxygen tensions in CF sputum decline (78). Although the total amount of extracellular iron can reach high values in CF sputum (59), the chemical nature of this iron (e.g., free, chelated to siderophores or some other microbial ligand, or heme) and what fraction is oxidized or reduced is unknown. What is known, however, is that many *P. aeruginosa* strains isolated from CF sputum make redox-active phenazine compounds (21, 69), that phenazines are present in CF sputum at concentrations in the 1 to 100 μ M range (77), and that their concentration increases as lung function declines across a broad patient cohort (R. C. Hunter and D. K. Newman, unpublished data). Phenazines are capable of reducing a variety of high-reductionpotential oxidants, including Fe(III), through extracellular electron transfer (14, 27, 28, 54, 74). *In vitro* experiments have shown that phenazines can liberate Fe(III) from transferrin by reducing it to ferrous iron $[Fe(II)]$ (14), as well as reducing highly insoluble Fe(III) (hydr)oxides to Fe(II) (27, 74). Thus, there is reason to speculate that at least some fraction of the total iron in CF sputum may be in the ferrous form and that efforts to iron limit *P. aeruginosa* through Fe(III) sequestration may be compromised by phenazine-promoted reduction of $Fe(III)$ to $Fe(II)$.

Although it has long been appreciated that *P. aeruginosa* may acquire iron via siderophore-independent pathways (48), whether phenazines might play a role in iron acquisition in the context of biofilm formation has not been studied. Here, we sought to test the hypothesis that phenazine-mediated Fe(II) acquisition through the ferrous iron uptake system (Feo [11, 40]) could alleviate Fe(III) limitation and stimulate biofilm formation.

MATERIALS AND METHODS

Chemicals. Phenazine-1-carboxylate (PCA) was purified from aerobic stationary-phase cultures of *Pseudomonas fluorescens* strain 2-79 (NRRL B-15132) (70) grown in King's A medium (32) at 30°C, as previously described (74). Pyocyanin (PYO) was purified from aerobic stationary-phase cultures of *P. aeruginosa* strain UCBPP-PA14 (56) grown in Luria-Bertani (LB) broth at 37°C, as previously described (74). Pyoverdine, purified according to the method described in Albrecht-Gary et al. (2), and pyochelin, characterized and synthesized according to published procedures (80, 81), were provided by Schalk's group (Institut de Recherche de l'Ecole de Biotechnologie de Strasbourg, IREBS FRE3211 CNRS/ Université de Strasbourg, France). Fe(OH)₃(s), referred to as the Fe(III) mineral ferrihydrite, was synthesized according to the method by Schwertmann and Cornell (66), as previously described (74). Substantially iron-free conalbumin, 1,10-phenanthroline, hydroxylamine hydrochloride, ammonium acetate, ferrous ammonium sulfate, and carrier DNA for yeast transformation were purchased from Sigma-Aldrich. All enzymes used for DNA manipulation were purchased from New England BioLabs.

Strains, plasmids, primers, and growth conditions. Tables 1 and 2 list the strains, plasmids, and primers used in this study. For planktonic and biofilm experiments with *P. aeruginosa* PA14 strains, 0.3 g/liter of Bacto tryptic soy broth (TSB; 1% TSB, where 1% is relative to the usual concentration of TSB medium [30 g/liter]; Becton Dickinson) was used as the medium. Where indicated, for the PA14 siderophore-null strain (Δ*pvdA ΔpchE*), 1.0 μM Fe(OH)₃(s), 10 μM PCA or PYO, or 1.0 μ M Fe(OH)₃(s) together with 10 μ M PCA or PYO was added to 1% TSB. Also as indicated, for the PA14 wild type and the *feoB*::*MAR2xT7* mutant (from the nonredundant PA14 mutant library [38]), $10 \mu M$ PCA or PYO alone or 40 μ g/ml of iron-free conalbumin alone or together with 10 μ M PCA or PYO was added to 1% TSB. Flow cell biofilm experiments with the PA14 wild type and phenazine deletion (Δphz) strains were also performed in 1% TSB supplemented with 10 μ M PCA. To confirm that in contrast to the PA14 wild type, the Δ*pvdA* Δ*pchE* strain was unable to produce pyoverdine and pyochelin during planktonic growth, an iron-deficient morpholinepropanesulfonic acid (MOPS)-based medium (100 mM MOPS at pH 7.2, 20 mM succinate, 93 mM NH₄Cl, 43 mM NaCl, 2.2 mM KH₂PO₄, and 1 mM MgSO₄ [modified from reference 50]) was used. These planktonic growth experiments were performed in acid-washed iron-free sterile glass culture tubes or polypropylene flasks at 37°C with vigorous shaking at 250 rpm to generate aerobic conditions. To confirm that the PA14 *feoB*::*MAR2xT7* mutant was disrupted in ferrous iron transport, cells were incubated, with anaerobic shaking, in Amberlite-treated 1% TSB containing 100 mM KNO₃, 50 mM glutamate, 1% glycerol, and 100 μ M iron source [either $(NH_4)_2Fe(II)(SO_4)_2$ or $Fe(III)Cl_3$] at 37°C for 22 h. Culture densities were monitored at an optical density at 500 nm $(OD₅₀₀)$ in a Thermo Scientific Spectronic $20D^+$ or Shimadzu UV-2450 spectrophotometer.

P. aeruginosa PA14 strains and *Escherichia coli* strains used for mutant construction were cultured in LB broth (Fisher Scientific) at 37°C. *E. coli* BW29427 and β -2155 were supplemented with 0.3 mM diaminopimelic acid. The yeast *Saccharomyces cerevisiae* uracil-auxotrophic strain InvSc1 (Invitrogen), used for gap repair cloning (10, 19), was grown with yeast extract-peptone-dextrose medium (YPD medium; 1% Bacto yeast extract, 2% Bacto peptone, and 2% dextrose) at 30°C, and selections were performed with synthetic defined agar (SDA) medium lacking uracil (URA; Qbiogene 4813-065). For selection and maintenance of plasmids pMQ30 and its derivatives, as well as pAKN69, gentamicin was used at 15 μ g/ml for *E. coli* and 75 to 100 μ g/ml for *P. aeruginosa*, respectively. Selection and maintenance of *E. coli* containing pUX-BF13 was carried out on $100 \mu g/ml$ of ampicillin.

Strain construction. *P. aeruginosa* synthesizes two known siderophores, the stronger Fe(III)-binding pyoverdine and the weaker Fe(III)-binding pyochelin (53). We constructed the siderophore-null strain by generating unmarked deletions of the pyoverdine and pyochelin biosynthetic genes *pvdA* (71) and *pchE* (61), respectively, in PA14 wild type. Analogously, we constructed the phenazinesiderophore-null strain in the PA14 phenazine-null strain (Δphz) . We first deleted *pvdA* and then *pchE*. Here we describe the protocol for in-frame deletion of *pvdA* using yeast gap repair cloning based on previously developed methods (10, 19, 49, 67). The 5' and 3' regions (both \sim 1 kb in length) of the sequence flanking *pvdA* were amplified using primer pairs pvdAKO1/pvdAKO2 and pvdAKO3/pvdAKO4, respectively (Table 2). These 5'- and 3'-flanking DNA fragments and the gapped plasmid vector pMQ30 were simultaneously introduced into the yeast *S. cerevisiae* uracil auxotrophic strain InvSc1 (Invitrogen) for *in vivo* recombination (10, 19). The plasmid pMQ30 is an allelic exchange vector for Gram-negative bacteria unable to support replication of the ColE1 origin and contains CEN6/ARSH4 DNA sequences to support replication in *S. cerevisiae*, a URA3 yeast-selectable marker, a multicloning site in a $lacZ\alpha$ allele for bluewhite screening, an *oriT* for conjugation-mediated plasmid transfer, a gentamicin-resistance gene (*aacC1*), and the counterselectable marker *sacB* (67). Recombinant yeast cells were selected for on medium deficient in uracil. Plasmids were liberated from recombinant yeast and electroporated into *E. coli* UQ950, which was used as a host strain for plasmid replication (3). Transformants containing recombination products of pMQ30 with the PCR products were isolated by blue-white screening and gentamicin resistance (3), yielding the construct for deleting the PA14 *pvdA* gene. This plasmid, which we call pYW01 (Table 1), was purified from *E. coli* UQ950, transformed by heat shock into *E. coli* BW29497 (a diaminopimelic acid auxotroph), and then mobilized into PA14 (the wild type or the Δphz mutant) via biparental conjugation (76). PA14 (the wild type and the Δphz mutant) single recombinants (merodiploid containing the intact and the deleted *pvdA* gene) were isolated by selecting for gentamicin resistance. Resolution of the merodiploid was performed by selection on 10% sucrose, followed by PCR-based screening for loss of the wild-type gene to isolate the *pvdA* deletion mutants (referred to as $\Delta pvdA$ and $\Delta phz \Delta pvdA$). The deletion of *pchE* from the $\Delta pvdA$ and $\Delta phz \Delta pvdA$ strains was performed the same way, using primer pairs pchEKO1/pchEKO2 and pchEKO3/pchEKO4 (Table 2). Besides using PCR-based diagnosis, we confirmed the $\Delta pvdA \Delta pchE$ and Δphz *<u>ApvdA</u> ΔpchE* mutants by their inability to produce pyoverdine or pyochelin when grown in iron-deficient MOPS-based medium, in contrast to the wild-type

TABLE 1. Strains and plasmids used in this study

^a ORF, open reading frame.

b UIUC, University of Illinois at Chicago.

and Δphz strains, respectively (see "Analyses of phenazines and siderophores" for pyoverdine and pyochelin detection methods).

To follow *P. aeruginosa* biofilm development using confocal microscopy, we generated YFP-labeled PA14 strains by introducing a chromosomally encoded constitutive enhanced yellow fluorescent protein (EYFP) into the wild type, *phz*, *ΔpvdA ΔpchE*, and *Δphz ΔpvdA ΔpchE* strains. Plasmid pAKN69, containing the mini-Tn7(Gm)P_{A1/04/03}-*eyfp* fusion, was used for this purpose (35). This plasmid was cloned into *E. coli* BW29427 and then mobilized into each PA14 strain via triparental mating with the helper plasmid pUX-BF13 (carrying the transposase genes) in $E.$ *coli* β -2155, as described previously $(6, 33)$. PA14 transformants with YFP constructs were selected with gentamicin and confirmed by YFP fluorescence.

Biofilm experiments. A flow cell system was constructed for biofilm experiments. The size of each flow channel was 1.5 by 4 by 34 mm; continuous flow of 1% TSB (with or without the additives detailed in the Results section) at the rate of 3 ml/h was supplied with a Watson-Marlow peristaltic pump. The temperature for biofilm growth was 22°C. An early-stationary-phase culture grown in 10% TSB was diluted to an OD_{500} of 0.1 in biofilm control medium (1% TSB). Each

flow cell was then inoculated with 300μ of the diluted culture by injection with a 1-ml syringe. In order to allow cells to attach to the glass surface, the flow was arrested for 1.5 h and then resumed throughout the length of each experiment (up to 6 days).

To image biofilms, confocal laser scanning microscopy (CLSM) with a Leica TCS SPE inverted microscope was used. For PA14 strains constitutively expressing EYFP (the wild type, Δphz , $\Delta pvdA$ *DpchE*, and $\Delta phz \Delta pvdA$ $\Delta pchE$ strains), three-dimensional fluorescence images were acquired using an excitation wavelength of 488 nm with constant intensity and collecting emissions in the range of 510 to 618 nm. For the PA14 *feoB*::*MAR2xT7* mutant, images were obtained using differential interference contrast (DIC) mode. To ensure images used for comparisons of biofilm formation were representative and reproducible, multiple fields of view were acquired over time with a $10\times$ dry lens objective in each flow cell within a single experimental set, and at least four independent experimental sets were performed. Fluorescence and DIC images were processed using Bitplane Imaris and NIH ImageJ software. In most cases, fluorescence-based multiple biofilm image stacks (spaced 1 to 2 μ m apart) were analyzed using AutoCOMSTAT software, a modified version of the COMSTAT biofilm

TABLE 2. PCR primers used in this study

evaluation package by Heydorn et al. (29, 44). For each image, a global threshold was calculated using the robust automated threshold selection algorithm with a critical-size setting of 20 μ m, and connected volume filtering was performed with a connectivity setting of 18 to remove free-floating biomasses. Substratum coverage calculations were based on the first $3 \mu m$ above the substratum. The area of each analyzed image was $3.03 \times 10^5 \,\mathrm{\upmu m^2}$. The results from measurements of one to six images for each strain and treatment were averaged, and the standard error of the mean was calculated. The biofilm parameters reported here are biovolume per image area (referred to as biomass), substratum coverage, maximum height, and average height of the biomass, which excludes any area not covered by cells.

Iron analysis. The total concentrations of iron in TSB-based media and suspensions of the mineral ferrihydrite were analyzed by the phenanthroline assay according to published protocols (34). In summary, complete reduction of Fe(III) (soluble and/or mineral forms) to soluble Fe(II) was achieved by adding the reductant hydroxylamine hydrochloride to acidified samples. 1,10-Phenanthroline and the pH buffer ammonium acetate were then added and allowed enough time to fully develop an orange-red Fe(II)-phenanthroline complex at pH 3.5. The total iron concentrations reflected by the colored Fe(II) complex were calculated based on the absorbance readings at 510 nm in a Shimadzu UV-2450 spectrophotometer. Iron in sterile, aerobic, TSB-based media should be present as the oxidation state $+3$, even though the specific Fe(III) forms are unknown.

Analyses of phenazines and siderophores. Filtrates (passed through a 0.2- μ mpore-size filter) were prepared from biofilm effluents or planktonic cultures. For characterizing and quantifying phenazines and the siderophore pyochelin, filtrates were directly loaded onto a Beckman System Gold reverse-phase high-pressure liquid chromatography (HPLC) instrument with a diode array UV-visible light (Vis) detector and a Waters Symmetry C_{18} analytical column $(5-\mu m$ -particle size; 4.6 by 250 mm). Analysis was performed in a gradient of water–0.1% trifluoroacetic acid (TFA; solvent A) to acetonitrile–0.1% TFA (solvent B) at a flow rate of 1.0 ml/min using the following method. For 0 to 1 min, chromatography was performed in a linear gradient from 100% solvent A to 15% solvent B, for 1 to 12 min in a linear gradient to 58% solvent B, for 12 to 13 min in a linear gradient to 70% solvent B, for 13 to 25 min in a linear gradient to 85% solvent B, for 25 to 26 min in a linear gradient to 100% solvent A, and for 26 to 29 min in 100% solvent A. Phenazines (e.g., PCA, PYO, phenazine-1 carboxamide [PCN], and 1-hydroxyphenazine [1-OHPHZ]) and pyochelin, which are known to be potentially produced by *P. aeruginosa*, can all be detected based on their characteristic absorption wavelengths and retention times as long as their concentrations are higher than 0.05 to $0.1 \mu M$ (17, 43, 65, 74).

The same HPLC instrument was used for analyzing the siderophore pyoverdine, with gradient profiling and sample preparation as described in detail by Bultreys et al. (9). In summary, filtrates with Fe(III)-chelated pyoverdine(s) at pH 5.0 were prepared for HPLC by adding FeCl₃ into the samples, followed by filtration (0.2-µm-pore-size filter) and pH adjustment. Analysis was carried out in a gradient profiling with solvent A as water–17 mM NaOH–acetic acid at pH 5.0 and solvent B as acetonitrile (solvent B) (9). For samples with pyoverdine being released at concentrations higher than $0.1 \mu M$, a single Fe(III)-pyoverdine peak was detected at 403 nm. In addition, iron-free pyoverdine in filtered (0.2- -m-pore-size filter) biofilm effluents was analyzed using a fast fluorescencebased method by a BioTek Synergy 4 fluorescence plate reader with a xenon flash light source at the specific excitation/emission wavelength set of 405 nm/455 nm (20). Control experiments and independent HPLC analyses confirmed that the measured fluorescence signal was contributed predominantly by pyoverdine and hence can be used for its detection and quantification.

RESULTS

PCA can work together with the siderophore pyoverdine in promoting biofilm development. We first followed biofilm development in the *P. aeruginosa* PA14 wild type, a phenazinenull strain (Δ*phz*), a siderophore-null strain (Δ*pvdA* Δ*pchE*), and a phenazine-siderophore-null strain ($\Delta phz \Delta pvdA \Delta pchE$) under a flow of 1% TSB medium over 4 days. In addition, we used AutoCOMSTAT software to analyze biofilm images for quantitative comparisons. As illustrated in Fig. 1, wild-type biofilm formation proceeded in the typical stages such that bacteria initially attached to the abiotic glass surface had by day 1 clustered into microcolonies, which enlarged over time and eventually matured by day 4. The Δphz mutant, despite initially attaching equally well, formed fewer and smaller microcolonies over time than the wild type, as was seen before in LB-based medium (57). The Δ*pvdA* Δ*pchE* and Δ*phz* Δ*pvdA* $ΔpchE$ mutants showed equally severe biofilm defects and failed to develop microcolonies even 4 days after initial attachment (Fig. 1A). More quantitatively, the wild-type biomass was only marginally higher than those of three other strains at day 1, but by day 4 was about 3 to 4 times greater than that of Δphz and 40 to 60 times greater than those of the $\Delta pvdA \Delta pchE$ and *Δphz* $ΔpvdA ΔpchE$ *strains (Fig. 1B; Table 3). By day 4, the* difference in total biomass between the wild-type and Δphz mutant biofilms reflected variation mainly in thickness. On the other hand, the biofilm thickness observed for the $\Delta pvdA$ $\Delta pchE$ and $\Delta phz \Delta pvdA \Delta pchE$ strains was about the same as that of the Δphz strain; their extremely low total biomass results primarily from a lack of surface coverage (Table 3).

To further correlate phenazine and/or siderophore production with biofilm development, we simultaneously analyzed the compounds being released into the biofilm effluents for the wild type and the Δphz and $\Delta pvdA \Delta pchE$ strains (Fig. 1C). The wild type started to release measurable levels of PCA (the precursor phenazine produced by all phenazinemaking pseudomonads [43, 54]) and the siderophore pyoverdine beginning on day 2. Concomitant with biofilm development, both PCA and pyoverdine concentrations increased with time, with a dramatic rise in the PCA concentration by day 4. Surprisingly, no other phenazines, including the blue-colored PYO, ever reached concentrations above their respective HPLC detection limits (0.05 to 0.1 μ M) under our biofilm growth conditions. Given that PYO is a highly diffusible zwitterion at circumneutral pH, we interpret the absence of PYO in the biofilm effluents to reflect little or no production rather than its retention in the biofilm matrix. The Δphz mutant started to release a measurable level of pyoverdine on day 2, and the pyoverdine concentration increased along with biofilm development. Despite releasing more pyoverdine than the wild type, the Δphz mutant generated 3 to 4 times less biofilm biomass by day 4. Adding $10 \mu M$ PCA (comparable to the levels detected in wild-type day 4 biofilm effluents) to the medium of the Δphz mutant throughout the experimental timescale increased the biofilm biomass 4 times by day 6, resulting in a biomass similar to that of the wild type (Fig. S1). The $\Delta pvdA \Delta pchE$ mutant that failed to form biofilms did not release phenazines until day 4, and even then only at trace levels. Because phenazines are typically released only at a high cell density due to quorum-sensing regulation (17, 54), the lack of phenazine production by this strain in light of its sparse growth is not surprising. The lack of phenazine production by the $\Delta pvdA$ $\Delta pchE$ strain matches our observation that the *<u>λ</u>pvdA ΔpchE* and *Δphz ΔpvdA ΔpchE* strains show equally severe biofilm defects. Collectively, these data indicate that both siderophores and phenazines (particularly PCA under these conditions) promote biofilm formation.

PCA can promote biofilm formation by generating Fe(II) in the absence of siderophores. It is well established that pyoverdine promotes *P. aeruginosa* biofilm formation by facilitating iron acquisition (5). Effective iron acquisition is essential for biofilm development because more iron is required than for planktonic growth (68). This is due to the fact that iron acts as

FIG. 1. The phenazine PCA can work together with the siderophore pyoverdine in promoting *P. aeruginosa* biofilm formation. (A) Confocal microscopic images of biofilms at day 4 (A), total biomass over time as inferred by Auto-COMSTAT analysis (B), and the corresponding release of phenazine(s) and/or siderophore(s) into the biofilm effluents for the YFP-labeled PA14 wild type (WT), the phenazine-null strain $(\Delta p h z)$, the siderophore-null strain (*pvdA pchE*), and the phenazine-siderophore-null strain (*phz pvdA pchE*) under a flow of 1% TSB medium at 22°C (C) are shown. Confocal images consist of top-down views (*x*-*y* plane, top images) and side views (*x*-*z* plane, bottom images, enlarged and truncated to emphasize differences in the *z* dimension). Results are representative of six experiments. Data reported in panels B and C represent means \pm standard errors of the means (SEMs). Related quantitative data can be found in Table 3. Scale bars, $100 \mu m$ for images with a top-down view and $50 \mu m$ for side-view images.

a signal to trigger the transition from a motile to a sessile state (68). From the phenanthroline assay, we calculated the total iron concentration to be 0.2 μ M in 1% TSB. Because the medium was prepared aerobically at pH 7, the oxidation state of iron originally present in the medium should be $+3$, even though the specific Fe(III) forms are unknown. It is commonly believed that bioavailable iron in the micromolar concentration range is required for optimal bacterial growth (53, 58), and hence the iron content in 1% TSB verges on the lower threshold. If the observed PCA-promoted biofilm formation (Fig. 1) were due to a stimulation in iron acquisition as a result of phenazine-facilitated Fe(III) reduction to Fe(II), we would

a Biofilms were formed under a flow of 1% TSB medium at 22°C for 4 days. Representative images from this data set are shown in Fig. 1. All values are means of the results of *n* images \pm the standard error of the mean. Each image covers an area of 3.03 \times 10⁵ μ m². WT, wild type.
b Calculated for the area covered by the biomass.

FIG. 2. PCA and PYO can circumvent the siderophore pathway for promoting *P. aeruginosa* biofilm development via Fe(II) uptake-dependent and -independent mechanisms, respectively. The effectively insoluble Fe(III) mineral ferrihydrite $[Fe(OH)_3(s)]$ was the Fe(III) source. Confocal microscopic images of the YFP-labeled *P. aeruginosa* PA14 siderophore-null strain ($\Delta p v dA \Delta p c h E$) incubated in biofilm flow cells at 22 $^{\circ}$ C for 6 days with no additions or with the addition of 1.0 μ M Fe(OH)₃(s), 10 μ M phenazine (PCA or PYO), or 1.0 μ M Fe(OH)₃(s) together with 10 μ M phenazine (PCA or PYO) to 1% TSB medium. Images are top-down views (*x*-*y* plane). Results are representative of four experiments. Related quantitative data can be found in Table 4. Scale bar, $100 \mu m$.

expect that PCA would be able to circumvent the pyoverdine pathway if enough Fe(III) were provided. To test whether PCA could rescue the biofilm defect in the PA14 $\Delta pvdA \Delta pchE$ mutant in the presence of sparingly soluble Fe(III), we investigated the effect of adding $1.0 \mu M$ ferrihydrite mineral suspension [Fe(OH)₃(s); K_{sp} [the solubility product constant] = $10^{-38.8}$ M (47)] alone, 10 μ M PCA alone, or 1.0 μ M $Fe(OH)_{3}(s)$ together with 10 μ M PCA to the base medium (1% TSB). Adding exogenous PCA was necessary because, unlike the wild type, the $\Delta pvdA \Delta pchE$ strain produced very little PCA on its own (Fig. 1C). We found that adding $Fe(OH)₃(s)$ or PCA alone could not rescue biofilm formation by the $\Delta pvdA \Delta pchE$ mutant strain with statistical significance (Fig. 2; Table 4). By contrast, when $Fe(OH)_3(s)$ and PCA were added together, dramatic rescue was observed (Fig. 2; Table 4). This is consistent with our prior abiotic experiments which showed that some phenazines, including reduced PCA, can promote $Fe(OH)_{3}(s)$ reduction over a broad pH range and liberate bioavailable Fe(II) $(27, 74)$. The fact that adding PCA to 1% TSB can promote biofilm formation by the Δphz mutant (which makes pyoverdine but not phenazines) but not the *ΔpvdA ΔpchE* mutant (which cannot make pyoverdine and is severely delayed in producing phenazines; Fig. 1C) further supports the conclusion that PCA and pyoverdine both contribute to biofilm formation. However, even in the absence of pyoverdine, PCA-promoted rescue can be achieved by adding just 1 μ M Fe(III) to the base medium with 0.2 μ M total Fe (Fig. 2; Table 4). This rescue occurred even with the total iron concentration (1.2 μ M) being more than 10 times lower than that reported in CF sputum (generally greater than 10 μ M, ranging from 17 to 200 μ M) (60). Therefore, PCA can stimulate *P. aeruginosa* biofilm formation in the presence of otherwise biologically unavailable Fe(III), even in the absence of pyoverdine if sufficient iron levels can be attained.

PCA can promote biofilm formation in the presence of conalbumin by facilitating Fe(II) uptake. The previous experiments used the effectively insoluble mineral ferrihydrite to limit cells for Fe(III). While this may be relevant to understanding Fe(III) acquisition by *P. aeruginosa* in soil environments, in a clinical context, Fe(III) limitation results from binding by host-produced proteins of the transferrin family (including lactoferrin, serotransferrin, and conalbumin; K_d [dissociation constant] $\sim 10^{20}$ to ²³ M⁻¹) (1, 58). We note,

TABLE 4. Quantitative analyses of flow cell biofilms formed by the *P. aeruginosa* PA14 $\Delta pvdA \Delta pchE$ -YFP strain^a

Day	$Additive(s)$ to biofilm control medium (1% TSB)	No. of images	Total biomass $(\mu m^3/\mu m^2)$	Substratum coverage $(\%)$	Avg thickness of biomass $(\mu m)^b$	Maximum biofilm thickness (μm)
2	None		ND			
	1.0 μ M Fe(OH) ₃ (s)	4	0.03 ± 0.00	0.4 ± 0.1	10.7 ± 2.0	30
	$10 \mu M PCA$	4	0.05 ± 0.01	0.6 ± 0.2	12.6 ± 1.4	32
	1.0 μM Fe(OH) ₃ (s), 10 μM PCA	4	0.45 ± 0.21	6.3 ± 3.3	11.1 ± 1.2	30
	$10 \mu M$ PYO	3	0.09 ± 0.03	0.8 ± 0.3	13.5 ± 0.4	34
	1.0 μM Fe(OH) ₃ (s), 10 μM PYO	4	0.57 ± 0.29	7.3 ± 3.7	10.3 ± 0.5	34
$\overline{4}$	None		ND			
	1.0 μ M Fe(OH) ₃ (s)	3	0.05 ± 0.01	0.5 ± 0.2	12.1 ± 2.3	30
	$10 \mu M PCA$		0.19 ± 0.09	2.9 ± 1.4	10.8 ± 0.9	34
	1.0 μM Fe(OH) ₃ (s), 10 μM PCA		1.90 ± 1.08	12.5 ± 6.4	15.8 ± 1.8	44
	$10 \mu M$ PYO	4	0.83 ± 0.22	9.9 ± 2.4	10.1 ± 0.4	42
	1.0 μM Fe(OH) ₃ (s), 10 μM PYO	4	3.1 ± 1.76	19.2 ± 9.2	10.8 ± 2.4	48
6	None	4	0.63 ± 0.16	6.0 ± 1.0	10.4 ± 0.8	40
	1.0 μ M Fe(OH) ₃ (s)	4	0.75 ± 0.33	3.3 ± 1.3	24.3 ± 1.6	82
	$10 \mu M PCA$	3	0.21 ± 0.09	3.3 ± 2.1	16.1 ± 4.4	66
	1.0 μM Fe(OH) ₃ (s), 10 μM PCA		4.87 ± 1.32	29.3 ± 8.5	18.9 ± 1.8	80
	$10 \mu M$ PYO	3	9.54 ± 0.18	44.3 ± 5.1	24.1 ± 2.1	94
	1.0 μM Fe(OH) ₃ (s), 10 μM PYO	3	4.20 ± 0.17	31.8 ± 3.5	19.9 ± 2.6	56

a Biofilms were formed at 22°C for 6 days with no addition or with the addition of 1.0 μ M Fe(OH)₃(s), 10 μ M phenazine (PCA or PYO), or 1.0 μ M Fe(OH)₃(s) together with 10 μ M phenazine (PCA or PYO) to 1% TSB medium. Representative day 6 images from this data set are shown in Fig. 2. All values are means of the results for *n* images \pm the standard error of the mean. Each image covers an area of 3.03×10^5 μ m². ND, not determined. *b* Calculated for the area covered by the biomass.

however, that recent studies have documented formation of nanoparticles of Fe(III) (hydr)oxides on human serum transferrin fibrils (22), so the two may not be as distinct as previously thought. Regardless, it has been suggested that the Fe(III)-sequestering function of lactoferrin presents an innate defense against *P. aeruginosa*; by competing against high-affinity siderophore(s) for binding Fe(III), a subinhibitory concentration of lactoferrin that does not impose a planktonic growth defect prevents *P. aeruginosa* from accessing the level of iron required to signal the transition to the biofilm mode of growth (5, 68). Furthermore, earlier experiments performed with whole-bacterial-cell suspensions showed that PYO could mediate Fe(II) uptake by iron-limited *P. aeruginosa* by reducing transferrin-bound Fe(III) with concomitant liberation of Fe(II), due to its much weaker affinity for Fe(II) $({\sim}10^3 \text{ M}^{-1})$ (58). This led us to hypothesize that PCA could stimulate biofilm development by facilitating Fe(II) uptake through reductive liberation of Fe(II) from host Fe(III)-binding proteins, a pathway that is independent of siderophore-mediated Fe(III) uptake.

To test this hypothesis, we compared biofilm development under a flow of 1% TSB between the wild type and the *feoB*::*MAR2xT7* mutant, a strain disrupted in gene PA14_56680, which encodes the cytoplasmic membrane protein FeoB (11, 40). We confirmed the identity of this mutant by sequencing and performing a diagnostic phenotypic test, as follows. Because FeoB was recently shown to be the transporter required for energy-dependent Fe(II) uptake across the cytoplasmic membrane of *P. aeruginosa* (11, 40), distinct from the more extensively studied TonB1-dependent ABC transport system for siderophore-mediated Fe(III) uptake (53, 64, 82), we predicted that the *feoB*::*MAR2xT7* mutant would not grow on Fe(II). To test this prediction, we performed experiments in iron-free medium to which we added back either Fe(II) or Fe(III). As expected, the *feoB*::*MAR2xT7* mutant could not grow anaerobically in planktonic batch cultures when given Fe(II) as its sole iron source, yet it could grow in the presence of Fe(III); under the same conditions, the wild type grew regardless of whether iron was in the ferric or ferrous form (Fig. 3). In 1% TSB medium, both strains developed into mature biofilms over time, as expected (Fig. 4B). However, when we reduced the amount of available iron by adding the iron chelator conalbumin (in its iron-free form), neither strain formed biofilms. Previous researchers have also used conalbumin as a lactoferrin surrogate in these types of experiments (45, 68) because they have similar structures, Fe(III)-binding capacities, and effects on biofilm formation. To confirm that the amounts of conalbumin and PCA used in our biofilm experiments specifically affected the amount of iron required to signal biofilm formation and not the amount of iron required for planktonic growth, we measured planktonic growth by the wild type and the *feoB*::*MAR2xT7* mutant; the additive(s) had no effect on the growth of either strain (Fig. 4A). When we added both conalbumin and PCA, making the acquisition of sufficient iron to signal biofilm formation dependent upon PCA-mediated reduction of conalbumin-bound Fe(III) to Fe(II), the wild type could form biofilms but the *feoB*::*MAR2xT7* mutant could not.

For both the wild type and the *feoB*::*MAR2xT7* mutant, in contrast to the mature biofilms formed in medium without

FIG. 3. The *P. aeruginosa* PA14 *feoB*::*MAR2xT7* mutant (*feoB*::tn) cannot grow when given Fe(II) as its sole iron source, yet it can grow in the presence of Fe(III); the wild type (WT) can grow regardless of whether iron is in the ferric or ferrous form. Cells were incubated, with anaerobic shaking, in Amberlite-treated 1% TSB medium containing100 mM KNO₃, 50 mM glutamate, 1% glycerol, and 100 μ M iron source [either $(NH_4)_2Fe(II)(SO_4)_2$ or $Fe(III)Cl_3$] at 37°C for 22 h. Data reported are the means of triplicate experiments \pm standard deviations (SDs).

additives, in medium with added conalbumin, attached bacteria mostly remained as separated individual cells and failed to form clusters even after 6 days (Fig. 4B). This is similar to what was observed by other investigators studying *P. aeruginosa* strain PAO1 in the presence of lactoferrin (5, 30, 68). In accord with the conalbumin-induced severe biofilm defect, little PCA was released into the biofilm effluents by either strain (~ 0.05) to $0.1 \mu M$), as expected for density-dependent phenazine production (17, 54). For the wild type, the addition of PCA to conalbumin-treated medium rescued the defect by promoting biofilm growth into a uniformly distributed lawn type structure, distinct from the mushroom-like structures observed in control medium (Fig. 4B), within 6 days. With PCA-induced rescue, by day 6, the biofilm biomass increased by a factor of 13 compared to biofilms treated with conalbumin alone; the total biofilm biomass was comparable to that in medium without additives (Table 5). The wild-type biofilm developed similarly with respect to structure and biomass regardless of whether PCA was added to the control medium (data not shown), indicating that PCA-induced rescue was specific to the biofilm defect caused by the conalbumin treatment. On the other hand, for the *feoB*::*MAR2xT7* mutant with a disrupted Fe(II) transporter, PCA addition failed to rescue the conalbumin-induced biofilm defect (Fig. 4B). Together, these results indicate that PCA's ability to rescue biofilm formation in the presence of the Fe(III)-binding protein conalbumin is due to the fact that it makes Fe(II) bioavailable by reducing protein-sequestered Fe(III) through extracellular electron transfer.

The phenazine PYO can affect biofilm formation independently of facilitating Fe(II) uptake. PYO, a well-studied phenazine produced by *P. aeruginosa* and also detected in CF sputum (77), has been previously shown to affect biofilm formation in LB-based medium (57). Even though PYO was not released under our 1% TSB-based biofilm medium conditions, we performed analogous experiments to examine whether PYO might promote biofilm development in the same

FIG. 4. (A) The presence of subinhibitory levels of conalbumin alone or together with phenazine (PCA or PYO) does not inhibit planktonic growth of *P. aeruginosa* PA14 strains. Experiments were performed in batch cultures in 1% TSB-based biofilm medium at 37°C. Data reported are the means of triplicate experiments \pm SDs. (B) PCA but not PYO can rescue the conalbumin-induced *P. aeruginosa* biofilm defect by reducing protein-sequestered Fe(III) with concomitant release of Fe(II). Shown are confocal microscopic images of YFP-labeled *P. aeruginosa* PA14 wild type (WT) and DIC microscopic images of the *P. aeruginosa* PA14 *feoB*::*MAR2xT7* mutant (*feoB*::tn) disrupted in Fe²⁺ transport into the cytoplasm and incubated in biofilm flow cells at 22°C for 6 days, with no addition or with the addition of 40 μ g/ml of iron-free conalbumin alone or together with 10 μ M phenazine (PCA or PYO) to 1% TSB medium. Confocal images consist of top-down views (*x-y* plane, top images) and side views (*x-z* plane, bottom images, enlarged and truncated to emphasize differences in the *z* dimension). DIC images are top-down views (*x*-*y* plane). Results are representative of four experiments. Related quantitative data can be found in Table 5. Scale bars, 100 μ m for images with top-down views and $50 \mu m$ for side-view images.

manner as PCA. We first tested whether PYO could rescue the biofilm defect in the PA14 Δ*pvdA* Δ*pchE* mutant by comparing the effect of adding 10 μ M PYO alone to the effect of adding 1.0 μ M Fe(OH)₃(s) and 10 μ M PYO together to the base medium (1% TSB). In contrast to PCA's rescue occurring only together with the addition of 1.0 μ M Fe(OH)₃(s), PYO rescued the biofilm defect regardless of whether $Fe(OH)_3(s)$ was present; the rescue without $Fe(OH)_{3}(s)$ was 2 times greater than that with the addition of $Fe(OH)_3(s)$ (Fig. 2; Table 4).

We next tested whether the addition of PYO to conalbumintreated 1% TSB medium could rescue the PA14 wild-type biofilm defect. Unlike PCA, PYO did not rescue the conalbumin-induced biofilm defect when added to the medium (Fig. 4B; Table 5). This implies that PYO cannot efficiently reduce conalbumin-bound Fe(III) under these conditions, which is consistent with PYO being a thermodynamically less favorable reductant than PCA (74). However, this result was somewhat unexpected, given an earlier study by Cox that reported that *P. aeruginosa* could reduce transferrin-bound Fe(III) (14). We speculate that the reason PYO did not stimulate conalbuminbound Fe(III) reduction for us, but did stimulate transferrinbound Fe(II) reduction for Cox, is that one or more of the following caveats apply: (i) even though proteins within the transferrin family (such as conalbumin and transferrin) have

TABLE 5. Quantitative analysis of flow cell biofilms formed by WT *P. aeruginosa* PA14 with different additives by day 6*^a*

Additive(s) to biofilm control medium $(1\%$ TSB)	No. of images	Total biomass $(\mu m^3/\mu m^2)$	Substratum coverage $(\%)$	Avg thickness of biomass $(\mu m)^b$	Maximum biofilm thickness (μm)
None		7.20	46.7	16.4	-62
40 μ g/ml conalbumin		0.80 ± 0.08	11.8 ± 1.3	9.0 ± 0.7	32
40 μg/ml conalbumin, 10 μM PCA		10.70 ± 0.49	80.0 ± 4.3	13.2 ± 0.8	36
40 μg/ml conalbumin, 10 μM PYO		0.34	4.2		32

a Biofilms were formed at 22°C for 6 days with no addition or with the addition of 40 μ g/ml conalbumin alone or together with 10 μ M phenazine (PCA or PYO) to 1% TSB medium. Representative day 6 images from this data set are shown in Fig. 4B. All values are means of the results for *n* images \pm the standard error of the mean. Each image covers an area of 3.03×10^5 μ m²

. *^b* Calculated for the area covered by the biomass.

similarly strong Fe(III)-binding capacities, the K_d values can be different by 2 to 3 orders of magnitude (1) ; (ii) a given phenazine's Fe(III) reduction activity depends on factors such as concentration ratios between reactants [e.g., phenazine versus chelated Fe(III)] and/or the presence of other oxidants that could react more readily with reduced phenazines. For example, reduced PYO has been shown to be much more reactive toward oxygen than reduced PCA, making PYO more subject to oxygen competing with Fe(III) as an electron acceptor (74). Consistent with this, in the whole bacterial cell suspension system studied by Cox, PYO was demonstrated to be efficient at reducing transferring-bound Fe(III) only under a strict anaerobic condition (14). Considering that our flow cell biofilm system is very different from the one studied by Cox both with respect to oxygen content and other factors, it makes sense that phenazine reactivity would be different. Interestingly, PYO did not stimulate iron-independent rescue under these conditions either, in contrast to what we observed for the $\Delta pvdA \Delta pchE$ mutant under different iron-limited conditions (Fig. 2).

DISCUSSION

Phenazines can affect cells in multiple ways, ranging from deleterious (generating toxic superoxide species; see references 26, 36, 37, and 39) to beneficial (serving as signaling molecules that foster biofilm development [18] and promote redox homeostasis [55] and survival when oxidants are limited [73]). To this list we now add another function: the potential for phenazines to enhance biofilm formation by making iron more available. While it is well known that all organisms need iron to grow, it has become clear that an amount of iron in excess of this need is required for biofilm formation (5, 7, 51, 62, 68). In this context, more attention has been placed on ferric iron acquisition by *P. aeruginosa* than on ferrous iron acquisition (59); however, our results suggest that the latter might be important in environments where phenazines are present, ranging from the rhizosphere (24) to the CF lung (13). Here we have shown that in the presence of Fe(III) and the phenazine PCA, biofilm formation can proceed in the absence of siderophores and in the presence of host iron-binding proteins such as conalbumin. While we have focused on the electron-shuttling properties of the phenazine PCA in this study, we note that similar iron acquisition mechanisms have been proposed for phenazine-like riboflavin produced by *Shewanella* species (41) and *Helicobacter pylori* (79) and for humic-like pyomelanin produced by *Legionella pneumophila* (12), although these have not been tested in biofilm systems.

By highlighting the importance of phenazines in changing the redox state of iron, we do not mean to imply that other means of iron acquisition are somehow less essential for *P. aeruginosa*. On the contrary, it is well established that siderophore production is crucial for Fe(III) acquisition by *P. aeruginosa* (as well as many other bacteria) and that this is necessary for their growth in many contexts. Moreover, we hasten to point out that phenazine production is relevant for *P. aeruginosa* only once it has achieved a sufficiently high density for phenazine biosynthesis to be initiated. What we suggest is that the relative importance of different types of iron acquisition strategies likely change over time in response to changes in the chemistry of the microenvironment that *P. aeruginosa* and other bacteria inhabit. In environments where iron is primarily in the ferric form, growth of *P. aeruginosa* necessitates siderophores. However, as cultures grow in cell density, inflammation progresses, and phenazine production begins, siderophores would be expected to decline in importance for iron acquisition as phenazines mediate the reduction of $Fe(III)$ to $Fe(II)$ and/or heme becomes more available. Consistent with this, we have observed that one of the major effects of phenazine production at the transcriptional level is the repression of genes involved in Fe(III) acquisition (e.g., downregulation of siderophore biosynthesis and transport) (18). While for the sake of clarity we genetically separated siderophore from phenazine production in this study, in the wild type, both siderophores and phenazines contribute to iron acquisition: siderophores play a role early in biofilm development, but likely become less important once phenazine production begins.

Regardless of how Fe(III) is bound—be it in the form of insoluble ferric (hydr)oxides or chelated to host proteins—it is susceptible to reductive liberation to Fe(II) by reacting with phenazines, rendering it more bioavailable to *P. aeruginosa* (Fig. 5). Our results suggest that reduced PCA can efficiently reduce host protein-bound Fe(III) even though, paradoxically, this is predicted to be thermodynamically unfavorable at pH 7: the redox potentials for Fe(III)/Fe(II) and PCA (oxidized versus reduced) redox pairs are reported as -500 mV (52) and -116 mV (74), respectively. Note that these numbers correspond to equilibrium conditions, which likely are quite different from those within a biofilm. For example, biofilms are known to experience oxygen limitation (28), which would shift the redox equilibrium of PCA toward its reduced form. In addition, Fe(II) librated by reducing Fe(III) can be taken up by cells. Both of these effects would drive redox reactions between host protein-bound Fe(III) with reduced PCA in a more thermodynamically favorable direction.

FIG. 5. A highly simplified schematic view of *P. aeruginosa* iron transport via phenazine-facilitated Fe(II) uptake (right), siderophore-mediated Fe(III) uptake (middle), and heme uptake (left). In Fe(III) uptake, a siderophore (e.g., pyoverdine, indicated by the star) binds extracellular Fe(III) and crosses the outer membrane (OM) via a TonB-dependent transporter (53, 64). In the periplasm, Fe(III) is released from the siderophore, which can then be recycled; Fe(III) is reduced by an unknown mechanism to Fe(II) in the periplasm (P) and transported across the cytoplasmic membrane (CM), presumably by an ABC transport system (53, 64). By contrast, phenazines can reduce extracellular Fe(III) to Fe(II). After entering the periplasm, presumably via an OM porin, Fe(II) is transported across the CM via FeoB (11, 40). Phenazines themselves are recycled (27, 73) and enter and leave the cell through various transporters (not drawn, for simplicity). Intracellularly, phenazine reduction is coupled to NADH oxidation to NAD⁺ (55), although whether this reduction is enzyme mediated is unknown. Reduced phenazine is indicated by the open oval and oxidized phenazine by the filled oval. Additionally, *P. aeruginosa* can acquire iron from heme and heme-containing proteins (e.g., hemoglobin), which are transported through a specific outer membrane receptor channel in a TonB-dependent manner (31, 48, 72). In *P. aeruginosa*, two distinct heme uptake systems have been found, although the mechanistic details of heme capture and delivery and the fate of heme after entering the cytoplasm are not fully understood (31, 48).

Given their high redox potentials, ferric (hydr)oxides can be easily reduced by PCA even under equilibrium conditions. This has broad relevance, given that PCA is the precursor phenazine produced by all phenazine-making pseudomonads (43, 54). Other natural phenazines, such as PCN and 1-OHPHZ, can react with Fe(III) at rates comparable to that of PCA (74). Interestingly, however, PYO is the one phenazine whose ability to reduce Fe(III) is poor (74). We therefore think it is unlikely that the observed PYO-enhanced biofilm rescue of our siderophore mutant in the presence of $Fe(OH)₃(s)$ was due to more efficient iron reduction under these conditions than that achieved by PCA. In fact, PYO is known to be a signaling molecule that helps coordinate *P. aeruginosa* biofilm development (18, 57). Thus, PYO's enhancement of biofilm formation under these conditions could be due to an iron-independent mechanism. Curiously, the enhancement of biofilm formation observed in the presence of PCA plus $Fe(OH)$ ₃ or PYO plus $Fe(OH)$ ₃ was less than that achieved in the presence of PYO alone (Table 4). This suggests that PCA- or PYO-mediated iron reduction is a double-edged sword: while it can promote biofilm formation relative to that achieved in the presence of $Fe(OH)$ ₃ alone, it can also exact a toxic price. The detailed mechanisms by which different phenazines affect biofilm formation remain to be elucidated, but what is clear is that different phenazines can affect biofilm formation in different ways under different environmental conditions.

Although it has long been recognized that the microbial community that forms in CF sputum is complex (25), what controls the structure and dynamics of this population is unclear. While many factors can be anticipated to impact the ecological success of different species in the lung, one important parameter is competition for iron (59). In coculture experiments pitting *Staphylococcus aureus* against *P. aeruginosa*, it was found that in the presence of *S. aureus*, *P. aeruginosa* decreased transcription of genes known to be regulated by iron

availability (42). It was concluded that this effect was most likely due to lysis of *S. aureus* by *P. aeruginosa*, leading to iron release. While the iron source was not identified in this study, one reasonable possibility is heme, which is thought to be an important nutrient for *Pseudomonas* and other bacteria in CF sputum, particularly as inflammation progresses (31, 48, 72). In contrast, *P. aeruginosa* can lose out in the Fe(III) battle to other organisms such as *Burkholderia* spp., which produce competing siderophores, such as ornibactin (75), as well as phenazines. Such positive and negative interactions involving iron may help explain the dynamic takeovers between different pathogens in the lungs of CF patients over time. Yet given that phenazines can alleviate the dependence of *Pseudomonas* on Fe(III) by generating Fe(II), competition for Fe(III) may be relevant only when phenazines are not present. Indeed, it seems likely that phenazine-enabled Fe(II) generation could benefit not only phenazine producers like *P. aeruginosa* but also other organisms, much in the same way that siderophores can be used by different species (53, 64).

The full complexity of the microbial community in the CF lung and the interactions among its members are only beginning to be appreciated (8, 25, 63). Systematic studies are needed to understand how iron availability—impacted by both host cells and the microbiome—affects this ecosystem. As recently pointed out by Reid et al. (59), we do not have a clear understanding of iron chemistry in CF sputum, and this must be determined before we can properly understand how iron might be acquired by *P. aeruginosa* cells or any other cell at different stages in lung function decline. In the future, it will be interesting to determine whether phenazines—possibly by way of promoting Fe(II) acquisition, among other effects (17, 18, 27, 28, 54, 55, 57, 73, 74)—influence the composition of the microbial community present in infections and its stability.

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