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## **Total Synthesis of (–)-Nakadomarin A**

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## **Abstract**



A concise diastereoselective total synthesis of (–)-nakadomarin A has been completed in 21 steps from D-pyroglutamic acid. Key steps include an enecarbamate Michael addition/furan-Nacyliminium ion cascade cyclization to provide the tetracyclic core, and ring-closing alkyne and alkene metatheses to construct the 15- and 8-membered azacycles, respectively.

> The manzamines are a class of architecturally fascinating, biologically active marine alkaloids.<sup>1</sup> Perhaps the most structurally intriguing member is nakadomarin A (**1**), isolated by Kobayashi2a from an Okinawan sponge *Amphimedon* sp. Its structure consists of an unprecedented 6/5/5/5/8/15 hexacyclic ring system and is the only manzamine alkaloid that embodies a furan ring. A biosynthetic pathway from ircinal A has been proposed by Kobayashi2b (Scheme 1).

> This limited availability, coupled with an inspiring structure have made nakadomarin A the target of a number of synthetic groups.<sup>3</sup> Nishida and co-workers have reported pioneering, though lengthy, total syntheses of both **1** 4a and *ent***-1**4b (36 and 38 steps, respectively). Young and Kerr completed the total synthesis of *ent*-**1** 5 in 29 steps from D–mannitol. Most recently, the Dixon group reported the shortest synthesis of **1** to date, requiring only 16 total steps.<sup>6</sup> A strategy common to each of these syntheses is the utilization of ring-closing alkene metathesis to construct the 15-membered macrocycle. While attractive in its efficiency, this approach yielded a mixture of configurational isomers in each case, often favoring the undesired *E*-isomer.<sup>7</sup>

> We recently reported our strategy for the rapid construction of the tetracyclic core of nakadomarin A.3h In the key step, enecarbamate **2** underwent a stereoselective intramolecular Lewis acid-catalyzed Michael addition, and the resulting *N*-acyliminium ion<sup>8</sup> **3** was trapped with the proximate furan to provide tetracycle **4** (Scheme 2). Based on the success of the model system study, we directed our efforts toward the preparation of a tetracycle analogous to **4** that would be fully functionalized for the completion of the total synthesis. Our retrosynthetic analysis is outlined in Scheme 3. Based on precedents from the previous total syntheses, ring-closing metathesis (RCM) could provide the eight- and fifteen-membered azacycles. We hoped to circumvent the *E*/*Z* selectivity problem in the

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Supporting Information Available: Spectroscopic data and experimental details for the preparation of all new compounds. This material is available free of charge via the Internet at<http://pubs.acs.org>.

construction of the macrocycle by utilizing a ring-closing alkyne metathesis  $(RCAM)^9$ /semihydrogenation strategy to deliver the *Z*-cycloalkene as a single configurational isomer. The viability of this strategy was documented by Fürstner and co-workers in their synthesis of the macrocyclic perimeter of nakadomarin A.3b The key cyclization of enecarbamate **7** to the tetracycle **5** should follow the conjectured pathway in our model system. In this case, it was hoped that the Lewis acid-activated conjugated double bond would approach the enecarbamate from the face away from the bulky TIPS group<sup>10</sup> through an *anti* conformer to provide the *N-*acyliminium ion **6** that would subsequently undergo closure with the now more electron rich furan substituent to deliver lactam **5**. The key cyclization substrate **7** could be obtained as the thermodynamic product of Knoevenagel condensation of furaldehyde **8** with β– amido ester **9** (vida infra).

The preparation of amide **9** proceeded in a straightforward fashion (Scheme 4). Enecarbamate **11** was prepared from optically pure imide **10**11 using the one-pot method recently reported by Yu.12 Vilsmeier- Haack formylation of enecarbamate **11** followed by reductive amination of the resultant vinylogous imide **12** with 5-heptynylamine provided the corresponding secondary amine. *N*-acylation with methyl malonyl chloride gave amide **9**.

The preparation of furaldehyde **8** took advantage of Maldonado's methodology for the synthesis of 2,4- disubstituted furans from γ, γ'-diacetoxyenones.<sup>13</sup> Thus, metalation of dimethyl methyl phosphonate followed by treatment with methyl 4-hexynoate gave β– ketophosphonate **13** (Scheme 5). Horner-Wadsworth- Emmons reaction between phosphonate **13** and 1,3- diacetoxyacetone provided γ, γ ′-diacetoxyenone **14**. This enone underwent smooth, acid-catalyzed cyclization in methanol at 50 °C to furanmethanol **15** in 82% yield. Finally, Swern oxidation of alcohol **15** furnished the desired furaldehyde **8**.

Based on observations made during our model system study, we anticipated that Knoevenagel condensation between amide **9** and furaldehyde **8** would provide unsaturated amide **7**, possessing the requisite *E*-geometry for the cyclization to tetracycle **5**. This thermodynamic control has been previously observed in related systems<sup>14</sup> and can be rationalized by better overlap of the ester and adjacent double bond in *E*-amide **7**, in contrast to the carbonyls of the corresponding Z-amide, both of which would be twisted out of conjugation. Much to our delight, heating amide **9** and furaldehyde **8** in benzene in the presence of benzoic acid and piperidine delivered the *E*-configurational isomer **7** as the sole product in 87% yield (Scheme 6). Subjection of enecarbamate **7** to our standard cyclization conditions (10 mol % Sc(OTf)<sub>3</sub>, CH<sub>2</sub>Cl<sub>2</sub>, rt) resulted in cyclization to the desired tetracycle. However, these reaction conditions also caused partial cleavage of the silyl ether, resulting in the isolation of a mixture of tetracyclic silyl ether **5** and the corresponding alcohol. This minor setback could be avoided altogether if  $InCl<sub>3</sub>$  was employed as the catalyst and the reaction was run at reflux, providing tetracycle **5** in 79% yield. Saponification of ester **5** followed by thermally-promoted decarboxylation gave the lactam **16**. The diyne functionality of lactam **16** was subjected to several different alkyne metathesis systems with varying levels of success. Grela's optimized conditions<sup>15</sup> of the Mortreux alkyne metathesis system<sup>16</sup> (Mo(CO)<sub>6</sub>, 2- fluorophenol, 3-hexyne, 1,2-diphenoxyethane in chlorobenzene at 140 °C, 3 h) gave cycloalkyne 17 in 41% yield (54% brsm). The Schrock carbyne catalyst<sup>17</sup> in chlorobenzene (25 mol %, 80 °C, 3 h) proved more effective, allowing cyclization to **17** in 77% yield on a gram scale. It should be noted the air-stable molybdenum nitride complex  $[(pyridine)(Ph<sub>3</sub>SiO)<sub>3</sub>Mo=NI$  recently developed by Fürstner<sup>18a</sup> gave comparable results with a slightly lower catalyst loading (20 mol %, toluene at 80 °C, 16 h, 80% yield). The *cis* double bond of 15- membered ring was then introduced by straightforward Lindlar reduction of cycloalkyne **17** and unaccompanied by the *E*-olefin stereoisomer (Scheme 7).

Synthetic efforts were then directed to the construction of the azocine ring and completion of the total synthesis (Scheme 7). Deprotection of the TIPS ether **17** furnished alcohol **18**. Oxidation of alcohol **18** with IBX in DMSO followed by Tebbe olefination (Wittig, Peterson and Nysted protocols were ineffective) proceeded uneventfully to yield vinyl pyrrolidine **19**. Deprotection of the Boc carbamate with TFA and *N-*acylation with 5-hexenoyl chloride gave alkene metathesis substrate **20**. Ring-closing metathesis of diene **20** to azocine **21** proved problematic. The best yield was obtained when diene **20** was treated with an equimolar amount of Grubbs 1<sup>st</sup>-generation catalyst in refluxing methylene chloride. Reduction of the resultant bis-lactam **21** with alane provided (–)-nakadomarin A in 58% overall yield from diene **20**. Spectral data (NMR, IR, MS) was identical to that of natural **1**. The optical rotation confirmed its absolute configuration ( $[\alpha]_D = -72.7$  ( $c = 0.12$ , MeOH), lit.<sup>4a</sup>  $\alpha|_D = -73.0$  ( $c = 0.08$ , MeOH)).

In conclusion, we have completed a concise total synthesis of (–)-nakadomarin A in 21 steps from D– pyroglutamic acid. Our previously reported strategy for the rapid assembly of the tetracyclic core, which features a tandem enecarbamate Michael addition/furan-*N*acyliminium ion cyclization, has now been modified to incorporate functionality for the completion of a completely diastereoselective total synthesis. Moreover, a sequential ringclosing alkyne metathesis/semi-hydrogenation strategy was utilized to obtain the 15 membered azacycle as a single configurational isomer. The flexibility of this route allows for the preparation of nakadomarin A structural analogs. Indeed, the cyclization of a pyrrole analogous to furan **2** was successful. These studies are currently underway and will be reported in due course.

## **Supplementary Material**

Refer to Web version on PubMed Central for supplementary material.

## **Acknowledgments**

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## **References**

- 1. Nishida A, Nagata T, Nakadawa M. Top Heterocyc Chem. 2006; 5:255. and references therein.
- 2. (a) Kobayashi J, Watanabe D, Kawasaki N, Tsuda M. J Org Chem. 1997; 62:9236.(b) Kobayashi J, Tsuda M, Ishibashi M. Pure Appl Chem. 1999; 71:1123.
- 3. (a) Fürstner A, Guth O, Rumbo A, Seidel G. J Am Chem Soc. 1999; 121:11108.(b) Fürstner A, Guth O, Duffels A, Seidel G, Liebl M, Gabor B, Mynott R. Chem-Eur J. 2001; 7:4811.(c) Nagata T, Nishida A, Nakagawa M. Tetrahedron Lett. 2001; 42:8345.(d) Magnus P, Fielding MR, Wells C, Lynch V. Tetrahedron Lett. 2002; 43:947.(e) Leclerc E, Tius MA. Org Lett. 2003; 5:1171. [PubMed: 12688711] (f) Ahrendt KA, Williams RM. Org Lett. 2004; 6:4539. [PubMed: 15548070] (g) Young IS, Williams JL, Kerr MA. Org Lett. 2005; 7:953. [PubMed: 15727483] (h) Nilson MG, Funk RL. Org Lett. 2006; 8:3833. [PubMed: 16898829] (i) Deng H, Yang X, Tong Z, Li Z, Zhai H. Org Lett. 2008; 10:1791. [PubMed: 18393518] (j) Inagaki F, Kinebuchi M, Miyakoshi N, Mukai C. Org Lett. 2010; 12:1800. [PubMed: 20232853]
- 4. (a) Nagata T, Nakagawa M, Nishida A. J Am Chem Soc. 2003; 125:7484. [PubMed: 12812466] (b) Ono K, Nakagawa M, Nishida A. Angew Chem, Int Ed. 2004; 42:2020.
- 5. Young IS, Kerr MA. J Am Chem Soc. 2007; 129:1465. [PubMed: 17263433]
- 6. Jakubec P, Cockfield DM, Dixon DJ. J Am Chem Soc. 2009; 131:16632. [PubMed: 19883080]
- 7. Nishida and Kerr obtained a 1:1.5 and 1:1.7 ratio of *Z:E* isomers, respectively. Dixon however obtained a favorable 1.7:1 *Z:E* ratio by including an excess of either enantiomer of camphorsulfonic acid in the metathesis reaction mixture.
- 8. For a review of N-acyliminium ion cyclizations, see: Maryanoff BE, Zhang HC, Cohen JH, Turchi TJ, Maryanoff CA. Chem Rev. 2004; 104:1431. [PubMed: 15008627]
- 9. For reviews, see: (a) Zhang W, Moore J. Adv Synth Catal. 2007; 349:93.(b) Fürstner A, Davies PW. Chem Commun. 2005; 18:2307.
- 10. A vinyl group was initially investigated, but its steric influence was found to be insufficient, as a 1:1 mixture of diastereomers was obtained from the cyclization to the corresponding tetracycle. These stereoisomers are presumably epimeric at C14:



- 11. Prepared in four steps from D-pyroglutamic acid in 67% overall yield. For details, see Supporting Information.
- 12. Yu J, Truc V, Riebel P, Hierl E, Mudryk B. Tetrahedron Lett. 2005; 46:4011.
- 13. Díaz-Cortés R, Silva AL, Maldonado LA. Tetrahedron Lett. 1997; 38:2207.
- 14. (a) Brown JM, Guiry PJ, Laing JCP, Hursthouse MB, Malik KMA. Tetrahedron. 1995; 51:7423.(b) Hamper BC, Kolodziej SA, Scates AM. Tetrahedron Lett. 1998; 39:2047.(c) Tietze LF, Schünke C. Eur J Org Chem. 1998:2089.
- 15. Sashuk V, Ignatowska J, Grela K. J Org Chem. 2004; 69:7748. [PubMed: 15498008]
- 16. (a) Mortreux A, Blanchard M. J Chem Soc, Chem Commun. 1974:786.(b) Mortreux A, Dy N, Blanchard M. J Mol Catal. 1975/1976; 1:101.
- 17. (a) Schrock RR, Clark DN, Sancho J, Wengrovius JH, Rocklage SM, Pedersen SF. Organometallics. 1982; 1:1645.(b) Schrock RR. Polyhedron. 1995; 14:3177.
- 18. (a) Bindl M, Stade R, Heilmann EK, Picot A, Goddard R, Fürstner A. J Am Chem Soc. 2009; 131:9468. [PubMed: 19534524] Another catalyst has been recently reported, but was not investigated. See: (b) Heppekausen J, Stade R, Goddard R, Fürstner A. J Am Chem Soc. 2010; 132:11045. [PubMed: 20698671]



**Scheme 1.** Proposed biosynthesis of nakadomarin A.



**Scheme 2.** Model study of the construction of the core of **1** .



**Scheme 3.** Retrosynthetic analysis of (–)-nakadomarin A ( **1**).



**Scheme 4.** Preparation of amide **9** .



**Scheme 5.** Synthesis of furaldehyde **8** .





Construction of the pentacyclic core of **1** .



