

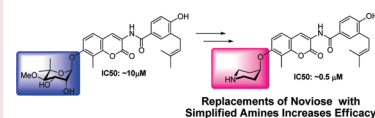
Synthesis and Evaluation of Noviose Replacements on Novobiocin That Manifest Antiproliferative Activity

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ABSTRACT Structural modifications to the coumarin core and benzamide side chain of novobiocin have successfully transformed the natural product from a selective DNA gyrase inhibitor into a potent inhibitor of the Hsp90 C-terminus. However, no structure–activity relationship studies have been conducted on the noviose appendage, which represents the rate-limiting synthon in the preparation of analogues. Therefore, a series of sugar mimics and nonsugar derivatives were synthesized and evaluated to identify simplified compounds that exhibit Hsp90 inhibition. Evaluation against two breast cancer cell lines demonstrated that replacement of the stereochemical complex noviose with simplified alkyl amines increased antiproliferative activity, resulting in novobiocin analogues that manifest IC₅₀ values in the midnanomolar range.

KEYWORDS Heat shock protein 90, Hsp90 inhibitors, novobiocin, structure–activity relationships, breast cancer



The 90 kDa heat shock proteins (Hsp90) are responsible for the conformational maturation of more than 200 Hsp90-dependent client proteins,^{1,2} of which Her2, Src family kinases, Raf, PLK, RIP, AKT, telomerase, and Met are directly associated with the six hallmarks of cancer.^{3,4} Consequently, inhibition of the Hsp90 protein folding machinery simultaneously disrupts multiple oncogenic pathways, leading to cell death.^{5,6} Since the first proof-of-concept drug, 17-AAG (a synthetic analogue of geldanamycin), entered clinical trials and demonstrated therapeutic benefit at tolerable doses,⁷ extensive research has led to more than 20 subsequent clinical trials,⁸ highlighting Hsp90 as a promising therapeutic target for the development of anticancer agents.^{9–11}

Novobiocin, a natural product comprised of a noviose sugar, a coumarin core, and a prenylated benzamide side chain, is isolated from *Streptomyces* strains¹² and is known to exhibit antimicrobial activity by binding to the DNA gyrase ATP-binding pocket,¹³ a unique nucleotide-binding motif shared only by members of the GHKL superfamily.¹⁴ Because of the similar bent conformation exhibited by ADP bound to both DNA gyrase and the Hsp90 N-terminal domain, Neckers and co-workers hypothesized that novobiocin may manifest anti-cancer activity through Hsp90 inhibition.¹⁵ Their pioneering studies revealed novobiocin to bind Hsp90, but instead of binding to the well-recognized N-terminal domain, it bound to a previously unrecognized C-terminal region, albeit with low affinity (~700 μM in SKBr3 cells).¹⁵ Since this study, structural modifications of novobiocin have been pursued to identify molecules that exhibit increased inhibitory activity.^{16–22}

The first library of such compounds was designed and synthesized by Yu and co-workers to identify functionalities

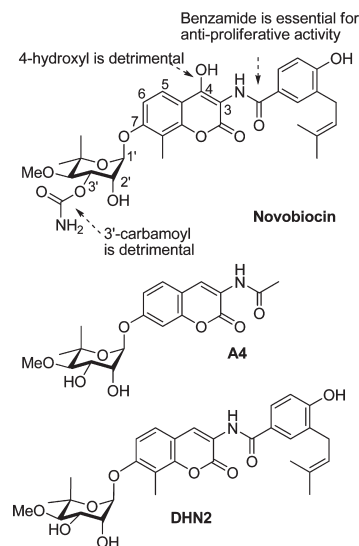


Figure 1. SAR generated from previous investigations.

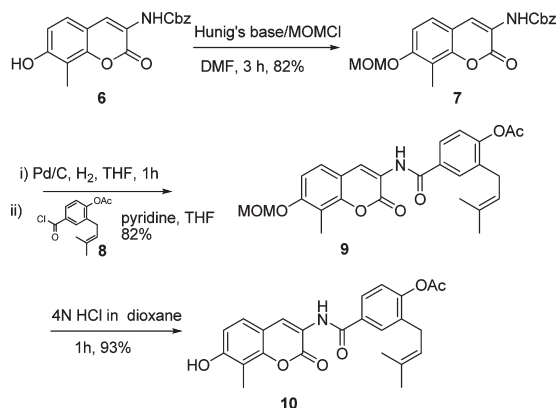
necessary for Hsp90 inhibition on the coumarin ring and benzamide side chain of novobiocin.¹⁷ Their studies revealed that attachment of the noviose appendage to the 7-position of the coumarin ring and an amide linker at the 3-position is critical, while the 4-hydroxy substituent and the 3'-carbamoyl are detrimental. The most efficacious compound identified

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Scheme 1. Preparation of Coumarin Phenol 10



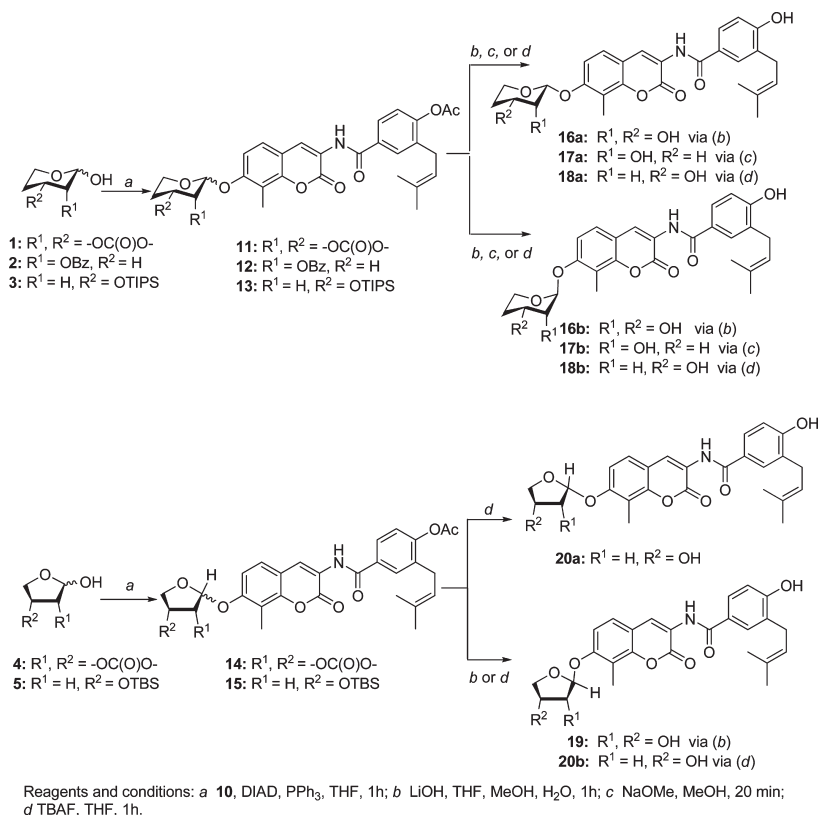
from this library was compound **A4**, which induced degradation of Hsp90-dependent client proteins at ~70-fold lower concentration than novobiocin. Intriguingly, compound **A4** induced the heat shock response at concentrations ~1000-fold lower than that required for client protein degradation.^{23,24}

To confirm whether structure–activity relationship (SAR) trends observed by Yu and co-workers conformed to the natural product, DHN2 was synthesized to delineate functionalities responsible for DNA gyrase versus Hsp90 inhibition.¹⁸ This novobiocin analogue confirmed that the 4-hydroxyl and the 3'-carbamate are detrimental to Hsp90 inhibitory activity but critical for DNA gyrase inhibition, thus confirming the SAR trends observed for **A4**.

Subsequent structural modifications and SAR studies explored the coumarin ring and benzamide side chain, and several lead-like compounds were identified and remain under investigation.^{19,21,22} As shown in Figure 1, the analogues prepared thus far retain the noviose appendage. However, the synthesis of noviose is laborious and hinders analogue development, as it requires more than 10 steps to prepare and activate for subsequent coupling with the coumarin phenol.^{25,26} Acknowledging the limited SAR for this moiety and its cumbersome preparation, simplified analogues were pursued in an effort to increase activity while simultaneously increasing solubility. In this article, we provide the first biologically active substitutions for the noviose appendage on novobiocin and the first nonsugar mimics that exhibit increased antiproliferative activity.

It is well understood that sugar moieties in natural products play a key role in solubility, activity, and bioavailability for these compounds. Furthermore, the ring size can impart significant affinity toward their cognate protein. With these considerations in mind, a series of mono- and dihydroxylated furanose and pyranose sugars (**1–5**, Scheme 2) were synthesized according to previously disclosed procedures²⁶ for incorporation onto the novobiocin scaffold, **10**. The preparation of **10** is described in Scheme 1. The coumarin phenol **6** was converted to the methoxymethyl (MOM) ether using MOM chloride and Hunig's base in dimethylformamide. The free aniline, liberated through hydrogenolysis with 10% Pd/C and hydrogen in tetrahydrofuran from **7**, was coupled with acid chloride **8** to give benzamide **9**.

Scheme 2. Synthesis of Novobiocin Analogues Containing Mono- and Dihydroxylated Furanoses and Pyranoses



Subsequent cleavage of the MOM ether with 4 N hydrochloride in dioxane provided phenol **10** in high yield.

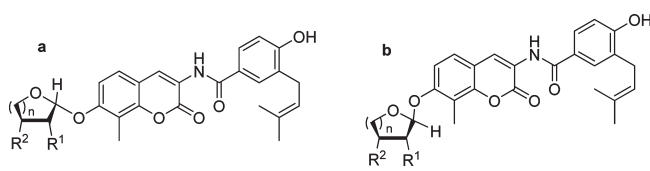
Once prepared, the phenol of **10** was coupled with sugars **1–5** under Mitsunobu conditions to give an inseparable diastereomeric mixture of **11–13** and **15** (Scheme 2). In the case of compound **14**, a single diastereomer was formed. Subsequent hydrolysis of the cyclic carbonates and acetyl esters of **11a,b** and **14** with lithium hydroxide in THF/MeOH/H₂O (3:2:2, v/v) afforded a diastereomeric mixture of **16a,b** and **19**, respectively. At this stage, diastereomers **16a** and **16b** were separated by silica chromatography. The assignment of stereochemistry at the anomeric center was established through two-dimensional NMR studies utilizing nuclear Overhauser effect spectroscopy. Hydrolysis of the benzoyl and acetyl ester of **12** with basic methanol yielded **17a** and **17b**, which could be separated by silica chromatography. The tri-isopropylsilyl of **13** and *tert*-butyldimethylsilyl groups of **15** were removed by the addition of tetrabutylammonium fluoride to give separable **18a** and **18b**, and **20a** and **20b**, respectively.

Upon construction of the noviose surrogates, compounds were subjected to evaluation of anti-proliferative activity against SKBr3 (estrogen receptor negative, Her2 overexpressing breast cancer cells) and MCF-7 (estrogen receptor positive breast cancer cells) cell lines. As shown in Table 1, the six-membered sugar mimics (**16a–18a** and **16b–18b**) were found to be more potent than their five-membered counterparts (**19** and **20a,b**). Compound **18b** displayed an IC₅₀ value of 3.11 ± 0.03 and 1.56 ± 0.20 μM against SKBr3 and MCF-7 cell lines, respectively, which is ~3–8-fold more active than DHN2 (Table 1) and ~200 times greater than the activity manifested by novobiocin. Surprisingly, the stereochemistry of these sugar mimics was not critical for the observed increase in anti-proliferative activity, as both the α- (**16b**) and the β-anomers (**17a**) produced similar activities. Placement of the hydroxyl group (3'-OH or 4'-OH) on the etheral ring also did not impart preferential activity.

Although simplified sugar mimics were found to increase anti-proliferative activity as compared to DHN2 and novobiocin, more simplified analogues exhibiting enhanced solubility and activity were desired. N-Heterocycles are found in a variety of biologically active compounds and, in contrast to carbohydrates, are generally ionized at physiological pH.²⁷ Upon review of the first set of studies, we proposed that the noviose appendage played a significant role in solubilizing the relatively hydrophobic coumarin core and benzamide side chain. Thus, commercially available amines, **21–27** (Scheme 3, secondary amines were protected with Boc), were selected as potential replacements for the noviose moiety. These alkylamines and heterocyclic analogues contain an ionizable amine located at various positions within the structure to afford potential hydrogen-bonding interactions while simultaneously enhancing solubility through their ionized counterparts.

Originally, coupling of these amines with phenol **10** was expected to easily afford the desired products. However, the acetyl ester on the benzamide side chain was hydrolyzed under these conditions and resulted in an inseparable mixture of mono- or dialkylated products. To circumvent this

Table 1. Anti-proliferative Activities of Sugar-Derived Novobiocin Analogues



compound	n	R ²	R ¹	SKBr3 (μM)	MCF7 (μM)
16a	2	OH	OH	10.68 ± 0.05 ^d	10.04 ± 0.03
16b	2	OH	OH	6.96 ± 0.06	13.30 ± 0.21
17a	2	H	OH	7.87 ± 0.04	6.45 ± 0.13
17b	2	H	OH	29.98 ± 2.07	10.24 ± 0.21
18a	2	OH	H	5.07 ± 0.26	1.34 ± 0.18
18b	2	OH	H	3.11 ± 0.03	1.56 ± 0.20
19	1	OH	OH	14.37 ± 0.52	14.31 ± 0.40
20a	1	OH	H	22.16 ± 0.94	> 100
20b	1	OH	H	21.46 ± 2.28	22.50 ± 0.40
DHN2				10.86 ± 0.47	11.29 ± 0.41

^a Values represent mean ± standard deviation for at least two separate experiments performed in triplicate.

issue, the amine was first coupled with the coumarin ring and subsequently with the benzamide side chain to afford the desired analogues. The detailed synthesis is described as follows: Tertiary amines or Boc-protected secondary amines were reacted with Cbz-protected coumarin **6** in the presence of 2 equiv of triphenylphosphine and diisopropylazodicarboxylate in tetrahydrofuran to give amine-derived coumarins, **28a–g**. The Cbz-protecting group was removed by hydrogenolysis to give the free amines, which were then coupled with acid chloride **8** to give compounds **29a–g** in good yield. Removal of the Boc protecting group with trifluoroacetic acid in methylene chloride afforded the secondary amine analogues **30b**, **30d**, and **30g**. Hydrolysis of the phenolic ester with methanolic triethyl amine gave analogues, **31a–g**, in good to excellent yields.

The anti-proliferative activity manifested by these analogues was assessed against SKBr3 and MCF-7 cell lines. As shown in Table 2, the IC₅₀ values for the secondary and tertiary amines varied between 0.4 and 1.5 μM, making them 1500-fold more potent than novobiocin. Generally, the ester series exhibited comparable anti-proliferative activity to their phenol counterparts, suggesting that the ester analogues may rapidly hydrolyze in cells due to esterases. With regard to the piperidine analogues, 4-substituted analogues exhibited greater potency than the 3-substituted analogues against both cell lines. For example, against the SKBr3 cell line, compound **29a** is ~3-fold more active than compound **29c** and compounds **30b** and **31b** are ~2 times more active than compounds **30d** and **31d**, respectively. The same trend was observed for the noncyclic amino analogues as well (**29f** vs **29e** and **31f** vs **31e**). These results indicate that the location of the amine is important for binding/manifesting inhibitory activity. A surprising finding from this work was that analogues containing noncyclic amines exhibit equivalent potencies to their piperidine counterparts,

Scheme 3. Synthesis of Amine Analogues

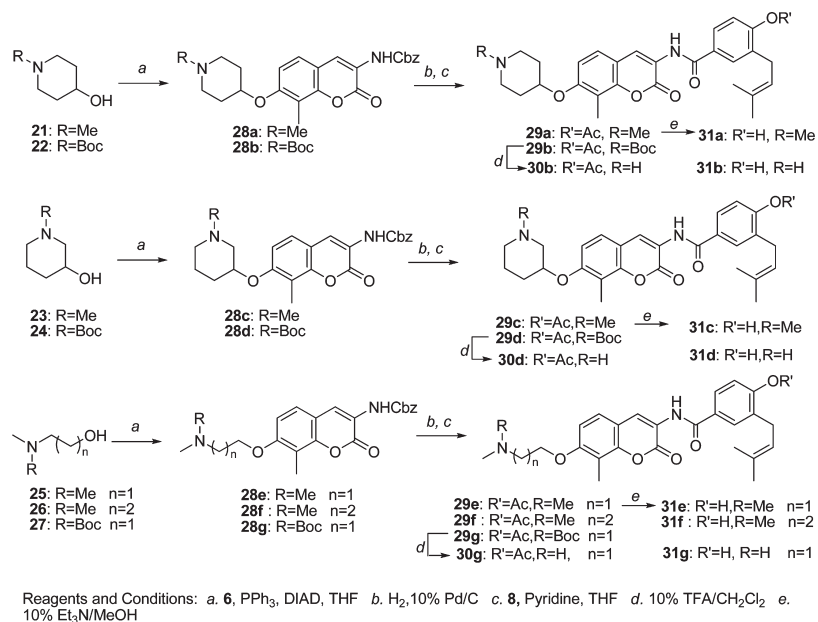


Table 2. Anti-proliferative Activities of Amine Analogues

compound	R	R'	R''	SKBr3 (μM)	MCF-7 (μM)
29a	A	Ac	Me	0.58 ± 0.05 ^a	1.18 ± 0.20
29c	B	Ac	Me	1.42 ± 0.02	1.57 ± 0.05
29e	C	Ac	Me	1.32 ± 0.23	4.76 ± 0.52
29f	D	Ac		0.46 ± 0.19	1.18 ± 0.03
30b	A	Ac	H	0.56 ± 0.05	1.53 ± 0.14
30d	B	Ac	H	1.23 ± 0.00	1.54 ± 0.41
30g	C	Ac	H	0.91 ± 0.21	2.08 ± 0.13
31a	A	H	Me	0.76 ± 0.17	1.09 ± 0.10
31b	A	H	H	0.47 ± 0.10	0.85 ± 0.09
31c	B	H	Me	4.69 ± 0.16	10.12 ± 0.17
31d	B	H	H	0.79 ± 0.11	1.68 ± 0.25
31e	C	H	Me	9.45 ± 0.22	13.48 ± 0.38
31f	D	H		0.44 ± 0.02	1.35 ± 0.38
31g	C	H	H	0.75 ± 0.12	1.33 ± 0.01

^a Values represent mean ± standard deviation for at least two separate experiments performed in triplicate.

specifically, **29a** and **29f** and **31a** and **31f**, indicating that inclusion of a constrained ring is not required.

To confirm that replacement of noviose with a sugar or amino surrogate did not alter inhibitory activity against Hsp90, Western blot analyses of cell lysates following the administration of **18b** or **31b** were performed. As shown in Figure 2, the Hsp90-dependent client proteins, Her2, Raf, and Akt, were degraded in MCF-7 cells in a concentration-

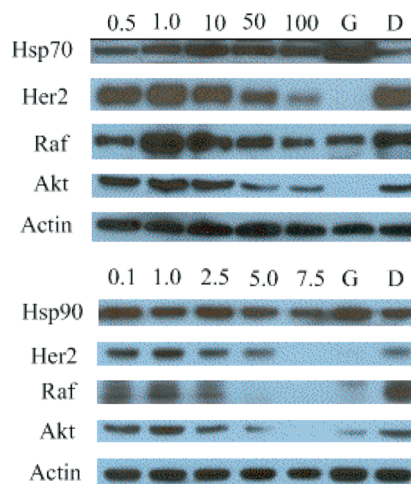


Figure 2. Western blot analyses of Hsp90-dependent client proteins from MCF-7 breast cancer cell lysates upon treatment with **18b** (top) or **31b** (bottom). Concentrations (in μM) were indicated above each line, and geldanamycin (G, 0.5 μM) and dimethyl sulfoxide (D) were employed as positive and negative controls.

dependent manner upon treatment with **18b** or **31b**. The non-Hsp90-dependent protein, actin, was not altered upon administration of **18b** or **31b**, indicating that selective degradation of Hsp90-dependent proteins takes place in the presence of **18b** or **31b**. In addition, neither of these two compounds induced the heat shock response, which is a characteristic shared by benzamide-containing novobiocin analogues that bind the Hsp90 C-terminus.^{28,29}

In a conclusion, sugar mimics and amino analogues of the noviose appendage on the coumarin ring of novobiocin that exhibit improved solubility and anti-proliferative activity have been produced. The cyclic and acyclic amino surrogates can be synthesized expeditiously and will enable rapid identification of

novobiocin analogues that may provide clinical opportunities for the treatment of cancer. The development of improved compounds that exhibit such activity is underway, and the results from those studies will be reported in due course.

SUPPORTING INFORMATION AVAILABLE Experimental procedures for the synthesis and characterization of new compounds (^1H and ^{13}C NMR and HR-MS). This material is available free of charge via the Internet at <http://pubs.acs.org>.

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